

Design and Synthesis of Neutralizable Fondaparinux

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antidotes. This work suggests that our synthetic biotin conjugate and trimer have potential for the development of neutralizable and safe anticoagulant drugs.

KEYWORDS: anticoagulant agent, heparin, oligosaccharide, glycosylation, synthesis design

INTRODUCTION

Heparin, a highly sulfated linear polysaccharide constituted by alternant α -1,4-D-glucosamine (GlcN) and β -1,4-D-glucuronic acid (GlcA) or α -1,4-L-iduronic acid (IdoA) units, has served as an anticoagulant agent to prevent or treat venous and arterial thrombosis for more than 80 years.^{1,2} Currently, the approved heparin-based anticoagulant drugs for clinical use mainly contain three forms: unfractionated heparin (UFH, average molecular weight ~16,000 Da), low-molecular weight heparin (LMWH, average molecular weight 3500-6000 Da), and ultralow-molecular weight heparin (UMWH, molecular weight 1500-3000 Da) such as synthetic pentasaccharide fondaparinux (molecular weight 1728 Da).³ UFH, isolated from animal sources such as the porcine intestine or bovine lung, exhibits a rapid anticoagulant effect to treat acute thrombotic events, and the effect can be reversed by protamine.⁴ However, the usage of UFH may result in the occurrences of heparin-induced thrombocytopenia and osteoporosis.^{5,6} To improve the pharmacokinetic properties and decrease the risks accompanied by the usage of UFH, LMWH, prepared by chemical or enzymatic degradation of UFH, has been increasingly applied in treatment of thrombotic disorders.⁷ Compared with UFH, LMWH such as enoxaparin shows a longer half-life and can be administered subcutaneously, which contributes to its clinical use. The anticoagulant effect of enoxaparin is only incompletely neutralized by protamine, which may result in the bleeding risks.^{4,8} Although enoxaparin has a significantly larger market share, it lacks an effective neutralizable agent. More importantly, the quality control and safety of heterogeneous animal-soured UFH and LMWH are still concerns due to batch-to-batch differences and potential contaminations. The clinical use of contaminated heparins with over-sulfated chondroitin sulfate resulted in hundreds of patient deaths and other adverse events in 2008.⁹ As a result, a structurally homogeneous heparin pentasaccharide fondaparinux (Arixtra) as an alternative is extensively used to treat thrombotic events.^{10,11}

Unlike heterogeneous animal-sourced heparins, fondaparinux is a pure synthetic pentasaccharide exhibiting antithrombin III-mediated exclusive factor Xa (FXa) inhibition activity.¹¹ Since the pentasaccharide is obtained by chemical synthesis, $^{10-17}$ the quality in terms of purity and reproducibility can be readily controlled. Additionally, fondaparinux exhibits a longer half-life and better antithrombotic efficacy and biosafety than the aforementioned two forms.^{8,18} However, clinical use of fondaparinux is still restricted because the anticoagulant effect cannot be immediately neutralized in the bleeding events. Although many elegant methodologies and strategies $^{20-36}$ have been developed to synthesize heparin oligosaccharides for drug discovery and great endeavors have been dedicated to the development of reversible heparin-based anticoagulant agents such as biotinylated idraparinux derivatives that can be neutralized by non-toxic avidin protein,³⁷

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Figure 1. Structures of fondaparinux and synthetic fondaparinux-based molecules 1-4.

Scheme 1. Building Blocks 5-9 for Synthesis of Aminopentyl-Functionalized Pentasaccharide 1



the design and efficient synthesis of neutralizable fondaparinux still remain challenging. Given the drug/antidote pair by binding-assisted clearance from circulation such as biotin/ avidin^{37–39} and large-size homogeneous heparin oligosaccharide/protamine,⁴⁰ we describe here the chemical preparation of fondaparinux-based biotin conjugate **2** and clusters **3–4** relying on a synthetic aminopentyl-functionalized pentasaccharide **1** (Figure 1). Additionally, all these fondaparinux-based molecules were evaluated to explore the anti-FXa activity and neutralizable anticoagulant activity.

RESULTS AND DISCUSSION

Synthesis Plan of Aminopentyl-Functionalized Pentasaccharide 1

The chemical synthesis of **1** is challenging because it requires regioselective introduction of sulfate groups, installment of glycosidic bonds in a stereoselective fashion, and differentiation of amino functionalities for N-sulfation in a saccharide moiety and late-stage conjugation in the artificial spacer. It is envisaged that protected pentasaccharide **12** could be synthesized via a convergent and stereocontrolled [3 + 2]

Scheme 2. Synthesis of Trisaccharide Donor 10^a



^aReagents and conditions: (a) NIS, TMSOTf, CH_2Cl_2 , 0 °C, 85%; (b) (i) hydrazine acetate, DCM/pyridine; (ii) NaH, BnBr, DMF, 88% over two steps; (c) (i) DDQ, CH_2Cl_2/PBS (100 mM, pH 7.4); (ii) Ac₂O, DMAP, pyridine, 80% over two steps; (d) (i) *p*-TsOH'H₂O, $CH_2Cl_2/MeOH$; (ii) TEMPO, BAIB, CH_2Cl_2/H_2O ; (iii) CH_3I , KHCO₃, DMF, 80% over three steps; (e) TMSOTf, CH_2Cl_2 , -20 °C, 66%; (f) (i) HF/pyridine, THF; (ii) $CF_3C(NPh)Cl$, Cs_2CO_3 , CH_2Cl_2 , 85% over two steps. Nap = 2-naphthylmethyl, TDS = thexyldimethylsilyl, NIS = *N*-iodosuccinimide, TMSOTf = trimethylsilyl trifluoromethanesulfonate, DDQ = 2,3-dichloro-5,6-dicyano-1,4-benzoquinone, *p*-TsOH'H₂O = *p*-toluenesulfonic acid monohydrate, TEMPO = 2,2,6,6-tetramethyl-1-piperidinyloxy, BAIB = (diacetoxyiodo)benzene, and $CF_3C(NPh)Cl = N$ -phenyltrifluoroacetimidoyl chloride.

Scheme 3. Synthesis of Disaccharide Acceptor 11^a



^aReagents and conditions: (a) NIS, TMSOTf, DCM, 0 °C, 84%; (b) (i) *p*-TsOH[·]H₂O, CH₂Cl₂/MeOH; (ii) TEMPO, BAIB, CH₂Cl₂/H₂O; (iii) CH₃I, KHCO₃, DMF, 48% over three steps.

method from appropriately protected monosaccharide building blocks 5-9 (Scheme 1). A well-organized protecting group strategy made it possible to achieve O-sulfation, N-sulfation, and highly stereoselective glycosylations. Acetyl (Ac) and benzoyl (Bz) protecting groups were introduced to allow selective O-sulfation at the corresponding hydroxyl groups. Levulinoyl (Lev) and Bz protecting groups at C-2 positions of 6 and 8 were chosen to construct 1,2-trans-glycosidic linkages by neighboring-group participation. Azido groups at C-2 positions of 5, 7, and 9 were employed as amino precursors for selective N-sulfation. Moreover, these azido groups did not perform neighboring-group participation, thus contributing to installing 1,2-cis-glycosidic linkages by the anomeric effect. Also, the 6-O-acetyl or Bz group of D-glucosazide donors facilitated α -directing glycosylation due to the remote participation effect.^{41,42} Additionally, benzyl (Bn) ethers and benzyloxycarbonyl (Cbz) as permanent protecting groups could be easily removed for unmasking the hydroxy groups and amino group to afford fondaparinux-based pentasaccharide 1 with an anomeric aminopentyl linker.

Synthesis of D-GlcN₃- α -(1 \rightarrow 4)-D-GlcA- β -(1 \rightarrow 4)-D-GlcN₃ Trisaccharide Donor 10

The trisaccharide *N*-phenyltrifluoroacetimidate (PTFAI) donor **10** was efficiently prepared by coupling glycosyl donor S^{13} with disaccharide acceptor **16** (Scheme 2). The synthesis

of the β -1,4-linked GlcA-GlcN₃ glycosidic linkage in 16 is challenging due to the inherently low reactivity of the glucuronic acid donor⁴³ and the lack of neighboring-group participation in the presence of 2-O-benzyl (Bn) protection. Although our previous work showed that a Ag₂CO₃-mediated coupling of glycosyl bromide with the glucosazide acceptor gave the desired disaccharide as a mixture of anomers (β/α = 7:1),¹³ the separation of both stereoisomers was extremely tedious and time-consuming. Furthermore, we tried different glucuronic acid donors for glycosylation, which did not give satisfactory results.¹³ To circumvent this problem, the integration of flexible orthogonal protecting group manipulations with a late-stage oxidation strategy made it possible to efficiently synthesize disaccharide acceptor 16. First, the coupling of thioglycoside donor 6 with 7 using Niodosuccinimide (NIS) and catalytic trimethylsilyl trifluoromethanesulfonate (TMSOTf) gave exclusively β -linked disaccharide 13 (J_{H1-H2} = 8.0 Hz) in 85% yield owing to the neighboring-group participation of 2-O-Lev. Then, the Lev ester was readily converted into the corresponding benzyl ether for permanent protection of the C-2 hydroxyl to give 14 via a two-step procedure involving the removal of the Lev group by hydrazine acetate to afford a free hydroxyl group,⁴⁴ followed by benzylation of the resulting hydroxyl group using BnBr and NaH. Subsequently, Nap ethers of 14 were oxidatively cleaved Scheme 4. Synthesis of Aminopentyl-Functionalized Pentasaccharide 1 and Fondaparinux-Based Biotin Conjugate 2^{a}



^aReagents and conditions: (a) TMSOTf, CH_2Cl_2 , -20 °C, 89%; (b) (i) 30% H_2O_2 , 1 M LiOH, THF; (ii) 4 M NaOH, 94% over two steps; (c) SO₃·NMe₃, DMF, 50 °C, 94%; (d) (i) PMe₃, THF/H₂O; (ii) SO₃·NMe₃, Et₃N/pyridine, 84%; (e) Pd(OH)₂/C, H₂, *t*-BuOH/H₂O, 83%; (f) biotin-OSu, DMSO, Et₃N, 79%.

using 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) without affecting any other protecting groups to provide a diol intermediate,⁴⁵ which was acetylated using Ac₂O and 4dimethylaminopyridine (DMAP) to give 15 in 80% yield for two steps. The cleavage of the 4,6-O-benzylidene group using p-TsOHH₂O in CH₂Cl₂/MeOH afforded the corresponding 4,6-diol intermediate, which was selectively oxidized using 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO)/ (diacetoxyiodo)benzene (BAIB) to a carboxylic acid at the C-6 position. The resulting carboxylic acid was esterified using MeI/KHCO₃ to give disaccharide acceptor 16. Subsequently, a TMSOTf-mediated coupling of donor 5^{13} with acceptor 16 afforded α -anomer trisaccharide 17 (J_{H1-H2} = 3.8 Hz). The anomeric thexyldimethylsilyl (TDS) ether in 17 was selectively cleaved using HF/pyridine to provide a hemiacetal, followed by treatment with CF₃C(NPh)Cl and Cs₂CO₃ to give the desired trisaccharide donor 10.46,47

Synthesis of L-IdoA- α -(1-4)-D-GlcN₃ Disaccharide Acceptor 11

As illustrated in Scheme 3, the disaccharide acceptor 11 was efficiently prepared by the same late-stage oxidation procedure as that used for the synthesis of 16. An NIS/TMSOTfpromoted coupling of suitably functionalized idosyl donor 8^{13} and acceptor 9 provided only α -anomeric product 18 in a high yield of 84% owing to neighboring-group participation of Bz at the C-2 position. The benzylidene acetal of 18 was cleaved by treating with *p*-TsOH:H₂O in CH₂Cl₂/MeOH to afford the corresponding 4,6-diols, which were reacted with TEMPO/ BAIB for selective oxidation of the primary alcohol to carboxylic acid. The resulting carboxylic acid was esterified using MeI/KHCO₃ to give the desired disaccharide acceptor 11.

Scheme 5. Synthesis of Fondaparinux-Based Dimer 3 and Trimer 4^a



"Reagents and conditions: (a) (i) DMSO, Et_3N ; (ii) DTT, 98% over two steps; (b) PBS buffer (100 mM, pH 7.4)/dioxane/acetonitrile (4:1:1), 65% for 3, 57% for 4.

Synthesis of Fondaparinux-Based Pentasaccharide 1 and Biotin Conjugate 2

With trisaccharide donor 10 and disaccharide acceptor 11 in hand, attention was focused on preparing aminopentylfunctionalized pentasaccharide 1 for late-stage conjugation (Scheme 4). First, a TMSOTf-mediated glycosylation of PTFAI donor 10 and acceptor 11 proceeded in highly stereoselective fashion to afford the fully protected pentasaccharide 12 with a higher yield of 89% than that in our previous result (68%).¹³ Probably, the use of the PTFAI glycosyl donor improved the coupling yield compared with that of the glycosyl trichloroacetimidate donor by eliminating the potential rearrangement reaction of the donor in glycosylation. The highly stereoselective glycosylation is attributed to the dual effects involving the anomeric effect of the azido group at the C-2 position and the remote participation effect of the acetyl (Ac) ester at the C-6 position in the D-glucosazide moiety of donor 10. The anomeric configuration of the newly generated $\alpha(1,4)$ -glycosidic linkage was verified by detailed NMR analysis (J_{H1-H2} = 3.8 Hz). Next, differential manipulations of protecting groups for regioselective installment of sulfate groups were carried out. The saponification of esters to remove acetyl, Bz, and methyl groups provided partially deprotected pentasaccharide 19 with five free hydroxyl groups via a twostep procedure including treatment of 12 with H_2O_2 and LiOH, followed by addition of NaOH. The resulting hydroxy groups in 19 were completely sulfated by the sulfur trioxide trimethylamine complex $(SO_3 \cdot NMe_3)$ to afford pentasulfate compound 20. Subsequently, three azide groups were reduced by trimethyl phosphine (PMe₃) to give an amine intermediate, followed by N-sulfation with SO3·NMe3 in the presence of Et₃N/pyridine. The resulting N-sulfated crude was purified using a reverse-phase C-18 silica gel column, followed by the SCR732 Na⁺-form column to provide 21. Finally, the removal

of Bn and Cbz by catalytic hydrogenolysis using $Pd(OH)_2/C$ in *t*-BuOH and H_2O afforded fondaparinux-based pentasaccharide **1**. The stereochemistry of five glycosidic linkages in compound **1** was confirmed by detailed NMR characterization (Supporting Information Page S20). In order to prepare biotinylated pentasaccharide **2**, compound **1** was conjugated to biotin by an efficient reaction of the free amine of the artificial spacer with the activated biotin succinimide ester (biotin-OSu). The resulting crude was readily purified by size exclusion chromatography using Bio-Gel P4 to afford fondaparinux-based biotin conjugate **2** in a yield of 79%.

Synthesis of Fondaparinux-Based Dimer 3 and Trimer 4

Initially, the coupling of excess aminopentyl-functionalized pentasaccharide 1 (10 equiv) with di-N-hydroxysuccinimidyl suberate in phosphate buffered saline (PBS) buffer (pH 7.4) did not give any dimer product, and in this case, the starting material 1 and the hydrolysis product of the activated monoester were mainly observed by electrospray ionizationmass spectrometry (ESI-MS) analysis (Figure S1). To circumvent this issue, the preparation of fondaparinux-based dimer 3 and trimer 4 was achieved by using thiol-maleimide coupling chemistry with a two-step procedure (Scheme 5).⁴⁸ First, the primary amine from the linker of 1 was treated with an excess of dithiobispropanoic acid bis(N-hydroxysuccinimide ester) and trimethylamine (Et₃N) in dimethyl sulfoxide (DMSO), followed by the treatment with dithiothreitol (DTT) to reduce the disulfide bond to furnish thiolderivatized compound 22. Next, the resulting thiol was reacted with the dimaleimide- and trimaleimide-activated spacer in PBS buffer/dioxane/acetonitrile to afford the desired dimer 3 and trimer 4, which were purified using Bio-Gel P4 and P6, respectively.



Figure 2. Evaluation of anticoagulant activity of compounds 2-4. (A) FXa inhibition curves of fondaparinux-based biotin conjugate 2 and fondaparinux. (B) Neutralization of FXa activity of biotin conjugate 2 by avidin. Fondaparinux was included as a control. (C) FXa inhibition curves of dimer 3, trimer 4, and LMWH enoxaparin. (D) Neutralization of FXa activity of dimer 3 and trimer 4 by protamine. Fondaparinux and enoxaparin were included as controls.

Bioactivity Tests

Since the FDA-approved UMWH fondaparinux and LMWH enoxaparin have exhibited excellent anticoagulant activity and the latter is incompletely neutralized by protamine, we wonder whether our synthetic fondaparinux-based biotin conjugate 2 as UMWH and 3-4 as LMWH can display effective anti-factor Xa (FXa) activity and be neutralized by avidin and protamine, respectively. The anticoagulant activities of compounds 2-4 and corresponding neutralization efficacy were evaluated in vitro with comparison to those of fondaparinux and enoxaparin. Biotinylated pentasaccharide 2 as UMWH exhibited similar anti-FXa activity with fondaparinux (median inhibitory concentration IC₅₀: 6.3 nM for 2 vs 4.4 nM for fondaparinux, Figure 2A), and the anti-FXa activity of 2 was efficiently neutralized by the addition of avidin (Figure 2B). Next, we tested the anti-FXa activities of fondaparinux-based clusters 3-4 and protamine-mediated reversibility. Fondaparinux-based dimer 3 and trimer 4 as LMWH showed similar anti-FXa activity with enoxaparin (Figure 2C). As compared to enoxaparin, which was only partially neutralized by protamine, dimer 3 showed inferior protamine-mediated reversibility than enoxaparin, while trimer 4 was nearly completely reversed by treatment with protamine (Figure 2D). These results suggested that synthetic fondaparinux-based biotin conjugate 2 and trimer 4 are encouraging targets with great potential for the development of neutralizable and safe anticoagulant drugs.

CONCLUSIONS

In summary, we have developed a convergent strategy for efficient synthesis of a fondaparinux-based pentasaccharide **1** with an anomeric aminopentyl linker. The careful selection of protecting groups makes it possible to construct the target

pentasaccharide by stereocontrolled chemical glycosylations, selective O-sulfations, and differentiation of amino functionalities of the saccharide moiety and linker for selective Nsulfations. In particular, the integration of flexible orthogonal protecting group manipulations with the late-stage oxidation strategy enables stereoselective and facile synthesis of challenging β -1,4-linked glucuronic acid modular disaccharide 16. Additionally, efficient preparation of fondaparinux-based biotin conjugate 2 and clusters 3-4 was accomplished by coupling pentasaccharide 1 with functionalized biotin and spacers, respectively. More importantly, the anticoagulant effect of biotinylated pentasaccharide 2 can be efficiently neutralized by avidin, and the anticoagulant activity of trimer 4 can be completely reversed by protamine. Therefore, the synthetically neutralizable fondaparinux-based compounds 2 and 4 could serve as safer anticoagulant agents to solve potential medical problems of current clinically used fondaparinux- and animal-sourced low-molecular weight heparin. Taken together, this work will facilitate efficient synthesis of heparin-like oligosaccharides and development of the next generation of safe and effective heparin-based anticoagulant drugs.

METHODS

Materials

Unless otherwise noted, solvents and reagents were purchased and directly used without further purification. Chemical reagents were purchased from J&K Scientific Ltd. and TCI Shanghai (Shanghai, China). Avidin beads and protamine were purchased from Sangon Biotech (Shanghai, China). Bio-Gel P-2, P-4, or P-6 (45–90 μ m) was purchased from Bio-Rad Laboratories (Hercules, California, USA). C18 silica gel was purchased from Waters Corporation (Milford,

Massachusetts, USA). A Biophen anti-Xa (two-stage heparin assay) kit was purchased from HYPHEN BioMed (Neuville-sur-Oise, France). 96-well plates were purchased from Thermo Fisher Scientific (Waltham, Massachusetts, USA).

General Methods

¹H NMR spectra were recorded on a Bruker AVANCE 500, 600, or 800 (at 500/600/800 MHz). ¹³C NMR spectra were recorded on a Bruker AVANCE 500, 600, or 800 (at 125/150/200 MHz). Shimadzu LC-MS2020 was used to record ESI-MS data. Agilent 1290 G6460A Q-TOF was used to perform high-resolution mass spectrometry (HRMS). Size-exclusion chromatography was carried out by using a Bio-Gel P-2, P-4, or P6 column. The absorbance was recorded and analyzed at 450 nm on BioTek Synergy 2.

Synthesis of Fondaparinux-Based Biotin Conjugate 2

To a solution of compound 1 (20 mg, 12.7 μ mol) in dry DMSO (300 μ L) were added biotin-OSu (13 mg, 38.1 μ mol) and Et₃N (11 μ L, 76.2 μ mol). The reaction mixture was stirred at room temperature until compound 1 was completely converted to product 2 as seen by ESI-MS analysis. The reaction mixture was lyophilized to give a residue. The resulting residue was purified by Bio-Gel P-4 sizeexclusion chromatography (eluent: 0.1 M NH₄HCO₃) to provide fondaparinux-based biotin conjugate 2 (18 mg, 79%). ¹H NMR (600 MHz, D₂O): δ 5.62 (d, J = 3.9 Hz, 1H, H-1-GlcN3), 5.55 (d, J = 3.4 Hz, 1H, H-1-GlcN2), 5.21 (d, *J* = 4.2 Hz, 1H, H-1-IdoA), 5.16–5.12 (m, 1H, H-1-GlcN1), 4.90-4.92 (m, 1H, H-5-IdoA), 4.68-4.59 (m, 2H, including H-1-GlcA, SCH₂CHNH), 4.5-4.11 (m, 12H, including H-3-GlcN2, 6 x H-6-GlcN, SCHCHNH, H-2-IdoA, H-4-IdoA), 4.06-3.99 (m, 2H), 3.94-3.85 (m, 4H), 3.81-3.52 (m, 6H, including OCH2CH2CH2CH2CH2NH), 3.48-3.42 (m, 2H, H-2-GlcN2, H-2-GlcA), 3.39-3.35 (m, 1H, SCHNH), 3.30-3.26 (m, 2H, H-2-GlcN3, H-2-GlcN1), 3.25-3.19 (m, 2H, OCH₂CH₂CH₂CH₂CH₂CH₂NH), 3.07-3.02 (m, 1H, SCH₂), 2.83-2.81 (m, 1H, SCH₂), 2.30-2.28 (m, 2H, COCH₂), 1.78-1.60 (m, 6H, OCH₂CH₂CH₂CH₂CH₂NH, COCH₂CH₂CH₂CH₂CHS, COCH₂CH₂CH₂CH₂CHS), 1.59-1.54 (m, 2H, OCH₂CH₂CH₂CH₂CH₂NH), 1.49-1.40 (m, 4H, OCH₂CH₂CH₂CH₂CH₂NH, COCH₂CH₂CH₂CH₂CH₂CHS). ¹³C NMR (150 MHz, D₂O): δ 176.21 (NHCOCH₂), 173.73 (COOH-GlcA), 172.91 (COOH-IdoA), 164.86 (NHCONH), 100.70 (C-1-GlcA), 98.99 (C-1-IdoA), 97.33 (C-1-GlcN3), 96.55 (C-1-GlcN1), 95.73 (C-1-GlcN2), 76.64, 75.75, 75.66, 75.51, 75.46, 75.36, 72.63, 72.21 (C-2-GlcA), 70.58, 69.71, 69.51, 69.41, 69.30, 69.18, 68.47, 68.18, 67.92, 66.16 (C-6-GlcN), 65.79 (C-6-GlcN), 65.40 (C-6-GlcN), 61.58 (SCHCHNH), 59.77 (SCH₂CHNH), 57.42 (C-2-GlcN), 57.29 (C-2-GlcN), 56.20 (C-2-GlcN2), 54.89 (SCHNH), 39.26 (SCH2), 38.76 (OCH₂CH₂CH₂CH₂CH₂CH₂NH), 35.05 (COCH₂), 27.73, 27.56, 27.35, 27.20, 24.77, 22.42. HRMS (ESI): $[M-2H]^{2-}$ m/z calcd for $C_{45}H_{74}N_6O_{51}S_{9}\!\!\!\!\!$ 901.0434; found, 901.0433.

Synthesis of Fondaparinux-Based Dimer 3

Compound 1 (50 mg, 31.6 μ mol) was dissolved in dry DMSO (400 μ L), followed by addition of 3,3'-dithiobispropanoic acid bis(Nhydroxysuccinimide ester) (64 mg, 158 μ mol) and Et₃N (44 μ L, 316 μ mol). The reaction mixture was stirred at room temperature for 8 h. Subsequently, DTT (48 mg, 311 μ mol) was added, and the reaction mixture was stirred for another 3 h. The reaction mixture was lyophilized to give a residue. The resulting residue was purified by Bio-Gel P-2 size-exclusion chromatography (eluent: 0.1 M NH₄HCO₃) to give 22 (51.5 mg, 98%), which was immediately used for next thiol-maleimide coupling reactions. The resulting compound 22 (40 mg, 24 μ mol) was dissolved in a mixture solution of degassed PBS buffer (100 mM, pH 7.4)/dioxane/acetonitrile (v/v/ v = 4:1:1, a total volume of 300 μ L), followed by the addition of the dimaleimide-activated spacer (0.5 mg, 1.2 μ mol). Then, NaOH (aq) was added to adjust the mixture to pH 7 \sim 9, and the reaction mixture was stirred at room temperature for 24 h under an Ar atmosphere. Subsequently, DTT (14.8 mg, 96 μ mol) was added, and the reaction mixture was stirred for another 3 h. The reaction mixture was loaded on a Bio-Gel P-4 size-exclusion chromatography column (eluent: 0.1 M NH₄HCO₃) for purification to afford fondaparinux-based dimer 3 (2.9 mg, 65%). ¹H NMR (600 MHz, D₂O): δ 5.56 (d, *J* = 3.7 Hz, 2H, 2× H-1-GlcN3), 5.52 (d, *J* = 2.6 Hz, 2H, 2× H-1-GlcN2), 5.16–5.11 (m, 4H, 2× H-1-GlcN1, 2× H-1-IdoA), 4.94–4.89 (m, 2H, 2× H-5-IdoA), 4.61 (d, *J* = 7.8 Hz, 2H, 2× H-1-GlcA), 4.45 (d, *J* = 10.8 Hz, 2H, 2× H-6-GlcN), 4.42–4.20 (m, 15H), 4.20–4.10 (m, 7H), 4.09–3.92 (m, 9H), 3.92–3.78 (m, 10H), 3.77–3.48 (m, 13H), 3.48–3.11 (m, 10H, 2× H-2-GlcN3, 2× H-2-GlcN2, 2× H-2-GlcN1, 2× H-2-IdoA, 2× H-2-GlcA), 3.09–2.74 (m, 7H), 2.73–2.56 (m, 7H), 2.56–2.38 (m, 8H), 1.68–1.58 (m, 4H), 1.56–1.48 (m, 4H), 1.47–1.33 (m, 4H). HRMS (ESI): [M-3H]^{3–} *m*/*z* calcd for C₉₄H₁₅₁N₁₂O₁₀₇S₁₈, 1245.0650; found, 1245.0558.

Synthesis of Fondaparinux-Based Trimer 4

Compound 22 (35.5 mg, 21.3 μ mol) was dissolved in a mixture solution of degassed PBS buffer (100 mM, pH 7.4)/dioxane/ acetonitrile (v/v/v = 4:1:1, a total volume of $300 \ \mu$ L), followed by addition of the trimaleimide-activated spacer (0.5 mg, 0.71 μ mol). Then, NaOH (aq) was added to adjust the mixture to pH 7 \sim 9, and the reaction mixture was stirred at room temperature for 24 h under an Ar atmosphere. Subsequently, DTT (13.1 mg, 85.2 µmol) was added, and the reaction mixture was stirred for another 3 h. The reaction mixture was loaded on a Bio-Gel P-6 size-exclusion chromatography column (eluent: 0.1 M NH₄HCO₃) for purification to provide fondaparinux-based trimer 4 (2.3 mg, 57%). ¹H NMR (800 MHz, D_2O): δ 5.60–5.54 (m, 6H, 3× H-1-GlcN3, 3× H-1-GlcN2), 5.25-5.10 (m, 6H, 3× H-1-GlcN1, 3× H-1-IdoA), 4.98-4.88 (m, 3H, 3× H-5-IdoA), 4.68–4.59 (m, 3H, 3× H-1-GlcA), 4.47–4.07 (m, 31H), 4.06-3.79 (m, 20H), 3.78-3.50 (m, 29H), 3.49-3.16 (m, 34H, including 3× H-2-GlcN3, 3× H-2-GlcN2, 3× H-2-GlcN1, 3× H-2-IdoA, 3× H-2-GlcA), 3.11-2.77 (m, 9H), 2.73-2.66 (m, 3H), 2.64-2.58 (m, 3H), 2.58-2.53 (m, 3H), 2.52-2.47 (m, 3H), 2.46-2.42 (m, 3H), 1.75–1.60 (m, 6H), 1.58–1.50 (m, 6H), 1.46–1.36 (s, 6H), 0.85 (s, 3H, -CH₃). HRMS (ESI): [M-5H]⁵⁻ m/z calcd for $\rm C_{146}H_{235}N_{18}O_{162}S_{27}$, 1139.0711; found, 1139.2380.

Evaluation of In Vitro Anti-FXa Activity of Compounds $2-4^{49}$

The Biophen anti-Xa (two-stage heparin assay) kit was used to evaluate anti-FXa activity of compounds 2-4 according to the procedures provided by the manufacturer. First, antithrombin [anti-Xa reagent 1 (R1)], factor Xa (R2), and factor Xa specific chromogenic substrates (R3) were dissolved using distilled water (1 mL) for reconstitution. Also, the resulting reagents R1, R2, and R3 were diluted five times by using Tris-ethylenediaminetetraacetic acid (EDTA)-NaCl-poly(ethylene glycol) (PEG) buffer (pH 8.4) for immediate use. Fondaparinux and biotinylated fondaparinux conjugate 2 were dissolved in the above-mentioned buffer at various concentrations (0-500 ng/mL), and enoxaparin and fondaparinuxbased dimer 3 and trimer 4 were also dissolved in the same buffer varying from 0 to 5000 ng/mL. 40 μ L of each sample solution was introduced into a 96-well plate, followed by the addition of 40 μ L of R1 and incubation at 37 °C for 2 min. Factor Xa (R2, 40 µL) was then added to the mixture which was incubated for another 2 min. Subsequently, R3 (40 μ L) was added to the reaction mixture which was then incubated at 37 °C for 2 min. The reaction was quenched by the addition of 20% AcOH (80 μ L). BioTek Synergy 2 was used to record the absorbance at 405 nm. The absorbance values were calculated to measure the activity of FXa.

Neutralization of Fondaparinux-Based Biotin Conjugate 2 by Avidin In Vitro³⁷

The fondaparinux and biotinylated fondaparinux conjugate 2 were dissolved in buffer (Tris-EDTA-NaCl-PEG, pH 8.4) at a concentration of 1000 ng/mL. The avidin beads from Sangon Biotech were suspended at various concentrations (0–100 μ g/mL) in the above-mentioned buffer. The avidin beads were incubated with fondaparinux or biotinylated fondaparinux conjugate 2 for 30 min at room temperature (a total volume of 300 μ L). Samples were then put on a magnetic rack, and anti-FXa activity was determined in the supernatant following the anti-FXa measurement protocols as described above.

Neutralization of Fondaparinux-Based Dimer 3 and Trimer 4 by Protamine In Vitro⁵⁰

The enoxaparin and fondaparinux-based dimer 3 and trimer 4 were dissolved in buffer (Tris-EDTA-NaCl-PEG, pH 8.4) at a concentration of 5000 ng/mL. Protamine from Sangon Biotech was dissolved in the same buffer and diluted at various concentrations (0–200 μ g/mL). Samples (40 μ L) and protamine (20 μ L) with various concentrations were added into a 96-well plate and incubated for 10 min at 37 °C; then, the mixture (60 μ L) was subjected to anti-FXa measurement protocols as described above.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacsau.2c00537.

Experimental information, compound synthesis and characterization, bioassay procedures, and NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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