



Phenolic constituents from twigs of *Aleurites fordii* and their biological activities

Kyoung Jin Park^{‡1}, Won Se Suh^{‡1}, Da Hye Yoon², Chung Sub Kim^{1,3}, Sun Yeou Kim² and Kang Ro Lee^{*1,§}

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¹Natural Products Laboratory, School of Pharmacy, Sungkyunkwan University, Suwon 16419, Republic of Korea, ²Laboratory of Pharmacognosy, College of Pharmacy, Gachon University, Incheon, 21936, Republic of Korea and ³Department of Biopharmaceutical Convergence, Sungkyunkwan University, Suwon 16419, Republic of Korea

Email:

Kang Ro Lee* - krlee@skku.edu

* Corresponding author ‡ Equal contributors

§ Phone +82-31-290-7710; fax +82-31-290-7730

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Abstract

Three new neolignan glycosides (**1–3**), a new phenolic glycoside (**15**), and a new cyanoglycoside (**16**) were isolated and characterized from the twigs of *Aleurites fordii* together with 14 known analogues (**4–14** and **17–19**). The structural elucidation of the new compounds was performed through the analysis of their NMR, HRMS, and ECD spectra and by chemical methods. All isolated compounds were tested for their antineuroinflammatory and neuroprotective activities.

Introduction

Aleurites fordii Hemsl. (= *Vernicia fordii* Hemsl., Euphorbiaceae), known as tung oil tree, is widely distributed throughout Northeast Asia [1]. The fruits, leaves, and roots of this plant have been used as a Korean traditional medicine for treating sore throat, respiratory illness, constipation, and diuresis [2,3]. Phytochemical investigations of *A. fordii* reported coumarins, diterpenoid esters, triterpenoids, and tannins [4–7]. Some phorbol diterpenoids isolated from *A. fordii* have shown Epstein–Barr virus activation effects and an enhancement of HTLV-I-induced colony formation of lymphocytes [8].

As an ongoing search for bioactive secondary metabolites from Korean medicinal sources, we investigated the methanolic extract of the twigs of *A. fordii* which resulted in the isolation and characterization of 14 lignan derivatives including three new neolignan glycosides (**1–3**), four phenolic glycosides including a new compound (**15**), and a new cyanoglycoside (**16**) from the organic extracts. The structures of the new compounds were established by NMR analysis (¹H and ¹³C NMR, COSY, HSQC, HMBC, and NOESY), HRMS, and chemical methods. The isolated compounds **1–19** were evaluated for their

antineuroinflammatory and neuroprotective activities. In this paper, we report the isolation and structural elucidation of these phytochemicals and their biological activity.

Results and Discussion

The MeOH extract of *A. fordii* twigs was subjected to liquid–liquid solvent partitioning to yield *n*-hexane, CHCl₃, EtOAc, and *n*-BuOH-soluble fractions. Repeated column chro-

matographic purification of the CHCl₃, EtOAc, and *n*-BuOH-soluble fractions afforded three new neolignan glycosides (**1–3**), a new phenolic glycoside (**15**), a new cyanoglucoside (**16**), and 14 known compounds (**4–14** and **17–19**) (Figure 1).

Compound **1** was obtained as a colorless gum. The molecular formula was determined to be C₂₅H₃₂O₁₁ from the [M + Na]⁺ molecular ion peak in the positive mode HRFABMS. The

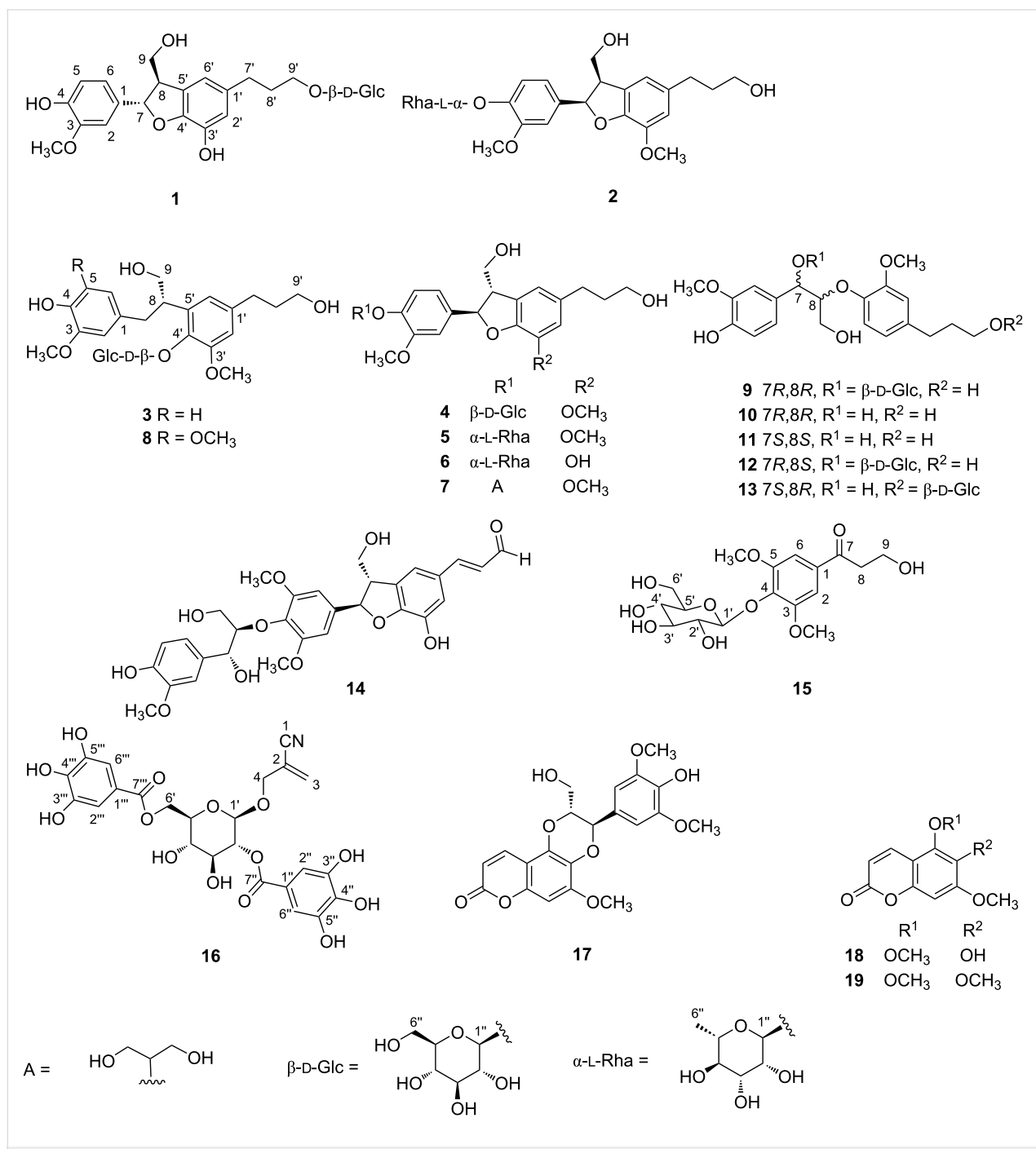


Figure 1: Chemical structures of compounds **1–19**.

^1H NMR data (Table 1) of compound **1** displayed characteristic resonances for a 1,3,4-trisubstituted benzene ring [δ_{H} 7.00 (d, $J = 1.9$ Hz, H-2), 6.87 (dd, $J = 8.1, 1.9$ Hz, H-6), and 6.78 (d, $J = 8.1$ Hz, H-5)], a 1,3,4,5-tetrasubstituted benzene ring [δ_{H} 6.62 (brs, H-2') and 6.65 (brs, H-6')], an oxygenated methine [δ_{H} 5.51 (d, $J = 6.1$ Hz, H-7)], an anomeric proton of a sugar [δ_{H} 4.27 (d, $J = 7.8$ Hz, H-1'')], and a methoxy group [δ_{H} 3.84 (s, 3-OCH₃)]. The ^{13}C NMR data (Table 1) showed 25 peaks including 12 aromatic carbons [δ_{C} 147.6 (C-3), 145.9 (C-4), 145.1 (C-4'), 140.4 (C-3'), 135.2 (C-1'), 133.7 (C-1), 128.3 (C-5'), 118.2 (C-6), 115.8 (C-2'), 115.3 (C-6'), 114.6 (C-5), and 109.1 (C-2)] and six glucose carbons [δ_{C} 103.0 (C-1''), 76.7 (C-3''), 76.5 (C-5''), 73.7 (C-2''), 70.2 (C-4''), and 61.3 (C-6'')]. The spectroscopic data of compound **1** suggested

that it is a typical dihydrobenzofuran neolignan glycoside [9–11]. The data for compound **1** were similar to those of glochidioboside isolated from *Glochidion obovatum* [12], except for the presence of a hydroxy group instead of the methoxy group at C-3' in **1**. The two-dimensional structure of **1** was elucidated via analysis of COSY, HSQC, and HMBC spectroscopic data (Figure 2). The locations of the glucose unit and the methoxy group were confirmed from the observed HMBC correlations of H-1''/C-9' and 3-OCH₃/C-3, respectively (Figure 2). Acid hydrolysis of **1** was conducted to analyze the aglycone and sugar moiety. The structure of the aglycone (**1a**) was confirmed as demethyldihydrodehydrodiconiferyl alcohol based on the comparison of ^1H NMR and MS data [13]. The relatively large coupling constant of the anomeric proton

Table 1: ^1H and ^{13}C NMR spectroscopic data of compounds **1–3** in CD₃OD.

Pos.	1^a		2^b		3^a	
	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)
1	133.7		134.3		131.9	
2	109.1	7.00, d (1.9)	112.3	7.06, d (2.0)	112.2	6.58, d (2.3)
3	147.6		152		146.9	
4	145.9		146.6		143.9	
5	114.6	6.78, d (8.1)	119.4	7.12, d (8.0)	114.2	6.57 d (8.0)
6	118.2	6.87, dd (8.1, 1.9)	120.1	6.92, dd (8.0, 2.0)	121.2	6.49 dd (8.0, 2.3)
7	87.3	5.51, d (6.1)	88.3	5.85, d (8.8)	37.8	2.99 dd (13.8, 5.5) 2.72 dd (13.8, 9.6)
8	54.3	3.47, dd (12.6, 6.1)	50.2	3.69, m	41.3	3.98, m
9	63.7	3.85, overlap 3.77, dd (11.0, 7.5)	63.6	3.31, m	65.6	3.77 dd (10.7, 6.2) 3.68 dd (10.7, 7.3)
1'	135.2		137.1		138.9	
2'	115.8	6.62, br s	114.3	6.76, br s	110.3	6.73, s
3'	140.4		145.3		151.6	
4'	145.1		147.5		142.2	
5'	128.3		131.9		137.1	
6'	115.3	6.65, br s	119.2	6.85, br s	118.9	6.73, s
7'	31.2	2.63, m	33.1	2.65, t (7.5)	31.7	2.66, m
8'	31.5	1.90, dt (13.5, 6.6)	35.9	1.85, m	34.1	1.84, tt (13.0, 6.5)
9'	68.5	3.93, m 3.56, m	62.4	3.59, t (6.5)	60.8	3.59, td (6.5, 1.9)
1''	103.0	4.27 d (7.8)	101.6	5.37, d (1.8)	104.2	4.63 d (7.6)
2''	73.7	3.22 dd (9.1, 7.8)	72.2	4.08, m	74.5	3.47, m
3''	76.7	3.37 dd (10.8, 9.1)	72.4	3.89, m	76.4	3.42, m
4''	70.2	3.30, m	74	3.47, m	69.8	3.39, m
5''	76.5	3.27 ddd (9.6, 5.6, 2.2)	71	3.83, m	76.6	3.14 ddd (9.2, 5.2, 2.3)
6''	61.3	3.88 dd (12.0, 2.2) 3.69 dd (12.0, 5.6)	18.1	1.24, d (6.2)	61.0	3.80 dd (11.0, 8.7) 3.70, overlap
3-OCH ₃	54.9	3.84, s	56.9	3.89, s	54.8	3.71, s
3'-OCH ₃			56.6	3.83, s	54.9	3.82, s

^aMeasured at 700 (δ_{H}) and 175 (δ_{C}) MHz. ^bMeasured at 500 (δ_{H}) and 125 (δ_{C}) MHz.

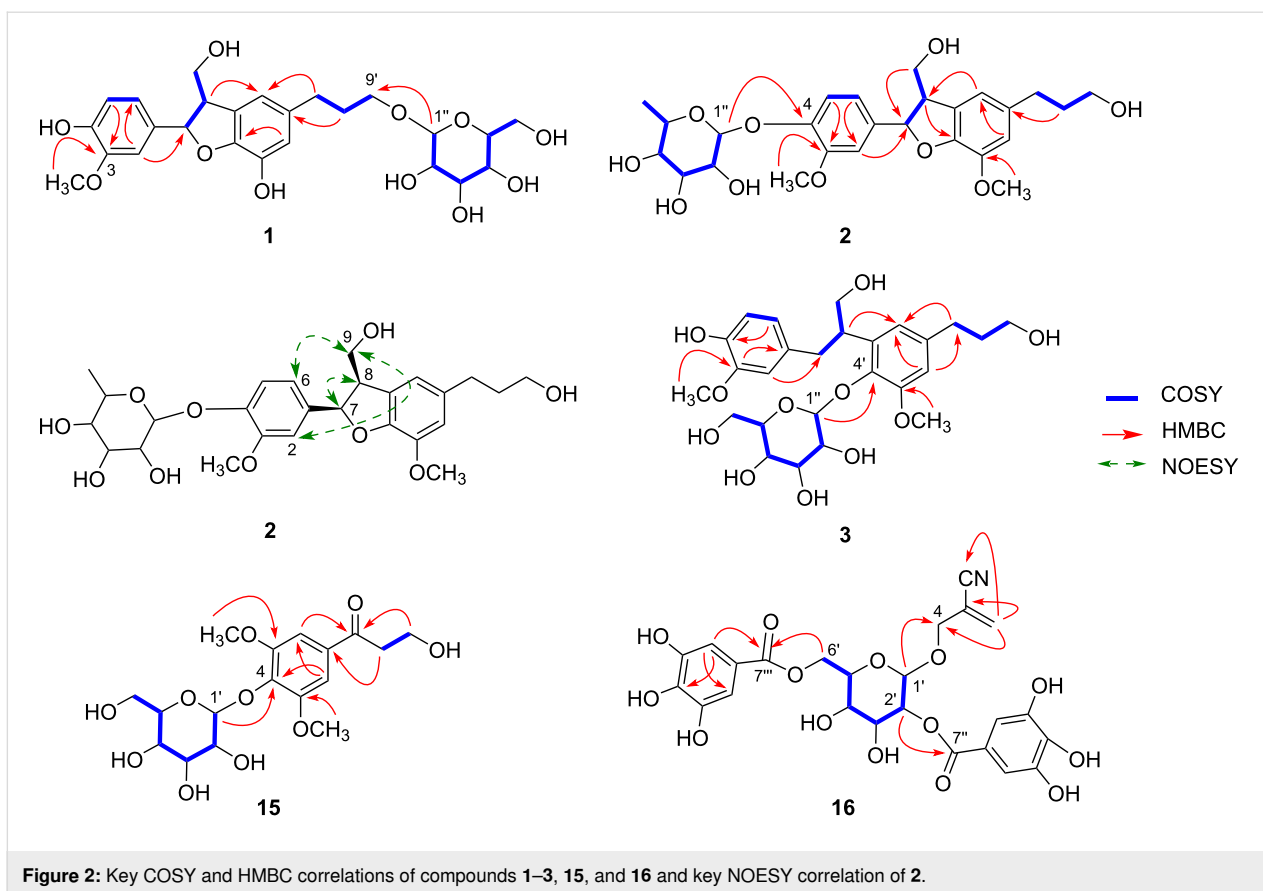


Figure 2: Key COSY and HMBC correlations of compounds 1–3, 15, and 16 and key NOESY correlation of 2.

(7.8 Hz) confirmed that the glucose is combined as β -form [14]. D-Glucose was identified by co-TLC with a standard sample [$\text{CHCl}_3/\text{MeOH}/\text{H}_2\text{O}$ 2:1:0.1, $R_f = 0.3$] and GC–MS analysis [15]. The relative configuration at C-7/C-8 of **1** was established as *trans* through the relatively small coupling constant (6.1 Hz) [10,16]. The analysis of the ECD spectrum of **1** determined the absolute configuration of **1** to be 7*S* and 8*R* (positive Cotton effects (CEs) at 292 and 248 nm, and a negative CE at 221 nm; see Supporting Information File 1, Figure S7) [10]. Thus, the structure of compound **1** was elucidated as (7*S*,8*R*)-3'-demethyl-dihydrodehydrodiconiferyl alcohol 9'-*O*- β -D-glucopyranoside, and was named as aleuritide A.

The molecular formula of compound **2**, isolated as a colorless gum, was confirmed to be $\text{C}_{26}\text{H}_{34}\text{O}_{10}$ from the positive ion mode HRESIMS data. The ^1H and ^{13}C NMR spectra of **2** were very close to that of icaricide E₄ (**5**) [17] with significant differences in the chemical shifts of C-1, C-8, C-9, C-5', and C-6' [**2**: δ_{C} 134.3, 50.2, 63.6, 131.9, and 119.2; **5**: δ_{C} 138.9, 55.7, 65.1, 129.7, and 118.0, respectively], indicating that compound **2** could be a stereoisomer of **5** at C-7 and C-8. The inspection of the COSY, HSQC, and HMBC spectra confirmed the planar structure of **2**. The HMBC correlation of H-1'' to C-4 indicated that the rhamnose unit was linked to the oxygen at C-4 and the

characteristic J value of the anomeric proton (1.5 Hz) confirmed the rhamnose as α -form (Figure 2) [10]. Acid hydrolysis of compound **2** afforded the aglycone, dihydrodehydrodiconiferyl alcohol (**2a**) [18], and L-rhamnose ($[\alpha]_{\text{D}}^{25} +9.0$), which was identified in an identical manner to that of compound **1**. The relatively large coupling constant (8.8 Hz) between H-7 and H-8 in **2**, as opposed to the relatively small coupling constant (6.1 Hz) between H-7 and H-8 in **1**, verified that H-7 and H-8 are *cis*-oriented [10,16], which was supported by the NOESY correlations of H-7/H-8, H-2/H-9, and H-6/H-9 (Figure 2). The ECD spectrum of **2** showed negative CEs at 276 nm and 229 nm and a positive CE at 248 nm, indicating the absolute configuration of C-7 and C-8 as *R* (Figure S15 in Supporting Information File 1) [19]. Therefore, the structure of compound **2** was determined to be (7*R*,8*R*)-dihydrodehydrodiconiferyl alcohol 4-*O*- α -L-rhamnopyranoside and was named as aleuritide B.

Compound **3** was obtained as a colorless gum after purification with a molecular formula of $\text{C}_{26}\text{H}_{36}\text{O}_{11}$ as deduced from the positive molecular ion peak $[\text{M} + \text{Na}]^+$ at m/z 547.2155 (calcd for $\text{C}_{26}\text{H}_{36}\text{O}_{11}\text{Na}$, 547.2155) in the HRESIMS. Analysis of the ^1H and ^{13}C NMR data (Table 1) showed a 1,3,4-trisubstituted benzene ring [δ_{H} 6.58 (d, $J = 2.3$ Hz, H-2), 6.57 (d, $J = 8.0$ Hz,

H-5), and 6.49 (dd, $J = 8.0, 2.3$ Hz, H-6); δ_C 146.9 (C-3), 143.9 (C-4), 131.9 (C-1), 121.2 (C-6), 114.2 (C-5), and 112.2 (C-2)], a 1,3,4,5-tetrasubstituted benzene ring [δ_H 6.73 (s, H-2', 6'); δ_C 151.6 (C-3'), 142.2 (C-4'), 138.9 (C-1'), 137.1 (C-5'), 118.9 (C-6'), and 110.3 (C-2')], a glucopyranose unit [δ_H 4.63 (d, $J = 7.6$ Hz, H-1''); δ_C 104.2 (C-1''), 76.6 (C-5''), 76.4 (C-3''), 74.5 (C-2''), 69.8 (C-4''), and 61.0 (C-6'')], and two methoxy groups [δ_H 3.82 (s, 3'-OCH₃) and 3.71 (s, 3-OCH₃); δ_C 54.9 (s, 3'-OCH₃) and 54.8 (s, 3-OCH₃)]. The spectroscopic data resembled closely to those of icariside E₃, isolated from *Epimedium grandiflorum* var. *thunbergianum* [20], indicating that compound **3** may have the identical planar structure to icariside E₃, which was reported without assignment of the absolute configuration. The planar structure of **3** was further confirmed by analysis of 2D NMR data, including COSY, HSQC, and HMBC (Figure 2). The determination of the stereochemistry for the

sugar unit of **3** was conducted following the same method as for compound **2**. The structure of the aglycone **3a** obtained by acid hydrolysis of **3** was confirmed based on ¹H NMR and MS data [20]. The absolute configuration of **3a** was established as 8*S* (a negative CE at 273 nm) based on the comparison of its ECD spectrum with the reported data [21]. Thus, the structure of compound **3** was determined as 8*S*-tetrahydrodehydrodiconiferyl alcohol 4-*O*- α -L-rhamnopyranoside and was named aleuriticide C.

Compound **15** was obtained as a yellow gum. The [M + Na]⁺ ion peak at m/z 411.1260 (calcd for 411.1267) in the HRESIMS corresponded to the molecular formula C₁₇H₂₄O₁₀. The IR spectrum exhibited signals at 3321 cm⁻¹ and 1675 cm⁻¹ suggesting the presence of hydroxy and carbonyl groups, respectively. The ¹H NMR spectrum of compound **15** (Table 2)

Table 2: ¹H and ¹³C NMR spectral data of **15** and **16** in CD₃OD.

pos.	15^a		16^b	
	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)
1	134.6		121.3	
2	107.5	7.23, s	118.2	
3	154.4		133.2	5.97, br s 5.95, br s
4	140.7		69.3	4.34, dd (13.7, 1.5) 4.22, dd (13.7, 1.5)
5	154.4			
6	107.5	7.23, s		
7	200.0			
8	42.1	3.23, t (6.0)		
9	58.8	3.96, t (6.0)		
1'	104.6	5.00, d (7.7)	102.1	4.70, d (8.0)
2'	75.8	3.51, m	75.2	4.99, dd (9.5, 8.0)
3'	78.6	3.23, m	76.0	3.70, m
4'	71.5	3.42, m	71.8	3.57, m
5'	78.1	3.44, m	76.2	3.67, m
6'	62.7	3.76, dd (12.0, 2.0) 3.66, dd (12.0, 5.0)	64.5	4.57, dd (12.0, 2.0) 4.46, dd (12.0, 5.0)
1''			121.4	
2'',6''			110.6	7.10 s
3'',5''			146.6	
4''			140.1	
7''			167.7	
1'''			121.4	
2''',6'''			110.3	7.11s
3''',5'''			146.7	
4'''			140.1	
7'''			168.4	
3,5-OCH ₃	57.3	3.81, s		

^aMeasured at 700 (δ_H) and 175 (δ_C) MHz. ^bMeasured at 500 (δ_H) and 125 (δ_C) MHz.

exhibited signals for a 1,3,4,5-tetrasubstituted aromatic ring [δ_{H} 7.23 (s, 2H, H-2 and H-6)], two methoxy groups [δ_{H} 3.81 (s, 6H, 3,5-OCH₃)], an anomeric proton [δ_{H} 5.00 (d, $J = 7.7$ Hz, 1H, H-1')], and two methylenes [δ_{H} 3.23 (t, $J = 6.0$ Hz, 2H, H-8) and 3.96 (t, $J = 6.0$ Hz, 2H, H-9)]. The ¹³C NMR spectrum of compound **15** (Table 2) revealed 14 peaks for 17 carbons including a ketone carbon (δ_{C} 200.0), a 1,3,4,5-tetrasubstituted aromatic ring [δ_{C} 154.4 (×2), 140.7, 134.6, and 107.5 (×2)], two methoxy groups [57.3 (×2)], and a glucose unit (δ_{C} 104.6, 78.6, 78.1, 75.8, 71.5, and 62.7). The location of the glucose unit was determined to be at C-4 by analysis of the HMBC data showing a correlation from H-1' to C-4. The coupling constant (7.7 Hz) of the anomeric proton of glucose suggested that it was the β -form. Acid hydrolysis of **15** yielded 3-hydroxy-1-(4-hydroxy-3,5-dimethoxyphenyl)-1-propanone (**15a**), whose ¹H NMR spectral data were in good agreement with the values reported previously [22], and β -D-glucopyranose was identified through co-TLC and the specific rotation value $\{[\alpha]_{\text{D}}^{25} + 86.0$ (c 0.03, MeOH)}. Accordingly, the structure of compound **15** was identified as 2,6-dimethoxy-4-(1-oxo-3-hydroxypropyl)phenyl β -D-glucopyranoside and named aleuriteoside A.

Compound **16** was isolated as a colorless gum and the molecular formula was determined to be C₂₄H₂₄NO₁₄ from the [M + Na]⁺ ion in the positive ion HRESIMS. The ¹H NMR spectrum of **16** (Table 2) showed two 1,3,4,5-tetrasubstituted aromatic rings [δ_{H} 7.10 (s, 2H, H-2'' and H-6''), and 7.11 (s, 2H, H-2''' and H-6''')], a terminal olefinic methylene [δ_{H} 5.97 (br s, 1H, H-3a), and 5.95 (br s, 1H, H-3b)], two oxygenated methylenes [δ_{H} 4.34 (dt, $J = 13.7, 1.5$ Hz, 1H, H-4a) and 4.22 (dt, $J = 13.7, 1.5$ Hz, 1H, H-4b); 4.57 (dd, $J = 12.0, 2.0$ Hz, 1H, H-6'a) and 4.46 (dd, $J = 12.0, 5.0$ Hz, 1H, H-6'b)], and an anomeric proton [δ_{H} 4.70 (d, $J = 8.0$ Hz, 1H, H-1')]. The ¹³C NMR spectrum of **16** (Table 2) revealed 24 carbons, including two galloyl moieties [δ_{C} 168.4, 167.7, 146.7 (×2), 146.6 (×2), 140.1 (×2), 121.4 (×2), 110.6 (×2), and 110.3 (×2)], a glucose moiety (δ_{C} 102.1, 76.2, 76.0, 75.2, 71.8, and 64.5), and 2-(hydroxymethyl)acrylonitrile signals (δ_{C} 133.2, 121.3, 118.2, and 69.3). 1D NMR spectra of **16** were similar to those of taxilulide C [23], except for the presence of a 2-(hydroxymethyl)acrylonitrile moiety at C-1' instead of 1,1-dimethylallyl alcohol. Furthermore, the NMR signals of cyanoglucoside moiety of **16** were in good agreement with literature data reported from *Codiaeum variegatum* belonging to the same family [24]. The location of the glucose unit was determined to be at C-4 based on the analysis of the HMBC data showing a correlation from H-1' to C-4 (Figure 2). The HMBC cross-peaks of H-2'/C-7'' and H-6'/C-7''' also indicated the presence of two galloyl groups at C-2' and C-6' of the glucose unit, respectively (Figure 2). Alkaline hydrolysis of **16** yielded codiacyanoglucoside

(**16a**) and gallic acid (**16b**). The identification of **16a** and **16b** was conducted by comparison of their ¹H NMR and MS data [24,25]. Consequently, the structure of **16** was determined to be codiacyano glucosyl-2',6'-O-digallate, named aleucyanoglucoside.

The other known compounds were identified as 7*R*,8*S*-dihydrodehydrodiconiferyl alcohol 4-*O*- β -D-glucopyranoside (**4**) [26], icariside E₄ (**5**) [17], isomassonioside B (**6**) [27], sakuraresinol (**7**) [28], selaginellol 4'-*O*- β -D-glucopyranoside (**8**) [21], 7*R*,8*R*-4,7,9,9'-tetrahydroxy-3,3'-dimethoxy-8-*O*-4'-neolignan-7-*O*- β -D-glucopyranoside (**9**) [29], 7*R*,8*R*-4,9,9'-trihydroxy-3,3'-dimethoxy-8-*O*-4'-neolignan (**10**) [30], 7*S*,8*S*-4,9,9'-trihydroxy-3,3'-dimethoxy-8-*O*-4'-neolignan (**11**) [30], 7*R*,8*S*-4,7,9,9'-tetrahydroxy-3,3'-dimethoxy-8-*O*-4'-neolignan-7-*O*- β -D-glucopyranoside (**12**) [29], 7*S*,8*R*-4,7,9-trihydroxy-3,3'-dimethoxy-8-*O*-4'-neolignan-9'-*O*- β -D-glucopyranoside (**13**) [31], buddlenol A (**14**) [32], aleuritin (**17**) [33], fraxinol (**18**) [34], and 5,6,7-trimethoxycoumarin (**19**) [34] based on the comparison of their spectroscopic data and specific rotation with the reported data.

Compounds **1–19** were tested for their effects on nitric oxide (NO) production levels in lipopolysaccharide (LPS)-stimulated murine microglial BV-2 cells to evaluate for antineuroinflammatory activities (Table 3). Compound **14** showed relative inhibitory effects on NO production with an IC₅₀ value of 20.9 μ M which was stronger than the positive control (L-NMMA, IC₅₀ 28.8 μ M). Compounds **11**, **17**, and **19** also displayed moderate activity (IC₅₀ 35.5–37.1 μ M), whereas compounds **2**, **4**, **10**, **12**, and **13** exhibited only weak effects (IC₅₀ 42.1–55.0 μ M). Interestingly, compounds **2** and **5** have the same planar structures with only differing in the C-7 stereochemistry, but they showed quite different inhibition effects on NO production (IC₅₀ 55.0 μ M, **2**; IC₅₀ > 500 μ M, **5**). The MTT cell viability test suggested that all the compounds had no cytotoxic effect on BV-2 cell survival at a concentration of 20 μ M.

Compounds **1–19** were also tested for their neuroprotection activity by measuring the secretion of NGF from C6 cells into the medium (Table 4). Compounds **8** and **16** stimulated NGF release, exhibiting stimulation levels of 134.2 \pm 8.1% and 134.6 \pm 5.9%, respectively. Although compounds **3** and **8** have similar structures without or with a methoxy group at C-5, respectively, only compound **8** showed a significant activity (96.2 \pm 1.1% for **3**). The other compounds exhibited moderate or no NGF secretion effect.

The cytotoxicity of compounds **1–19** was also evaluated against four human cancer cell lines (A549, SK-OV-3, SK-MEL-2, and

Table 3: Effects of isolated compounds on NO production in LPS-activated BV-2 cells.

comp.	IC ₅₀ (μM) ^a	cell viability% ^b	comp.	IC ₅₀ (μM) ^a	cell viability% ^b
1	>500	87.5 ± 5.1	11	37.1	89.97 ± 3.2
2	55.0	90.3 ± 3.1	12	47.6	90.1 ± 3.7
3	109.8	93.1 ± 3.5	13	42.1	87.9 ± 4.2
4	48.6	86.9 ± 6.5	14	20.9	105.2 ± 1.5
5	>500	111.9 ± 4.3	15	93.7	93.3 ± 7.4
6	74.6	86.2 ± 6.2	16	>500	95.8 ± 4.2
7	278.8	90.2 ± 4.1	17	35.5	101.8 ± 4.2
8	126.6	87.9 ± 5.1	18	117.0	96.6 ± 4.4
9	321.7	87.9 ± 7.6	19	36.7	99.7 ± 2.9
10	42.9	86.3 ± 4.4	L-NMMA ^c	28.8	99.9 ± 3.6

^aThe IC₅₀ value of each compound was defined as the concentration (μM) that caused 50% inhibition of NO production in LPS-activated BV-2 cells.

^bThe cell viability after treatment with 20 μM of each compound was measured using the MTT assay and is expressed as a percentage (%). Results are the means of three independent experiments, and data are expressed as the means ± SD. ^cPositive control substance.

Table 4: Effects of isolated compounds on NGF secretion in C6 cells.

comp.	NGF secretion ^a (%)	cell viability ^b (%)	comp.	NGF secretion ^a (%)	cell viability ^b (%)
1	117.4 ± 2.8	92.7 ± 1.3	11	102.1 ± 5.4	92.3 ± 0.3
2	101.3 ± 7.6	90.5 ± 0.1	12	110.3 ± 0.8	94.0 ± 1.9
3	96.2 ± 1.1	91.0 ± 0.8	13	104.5 ± 3.2	92.9 ± 0.2
4	99.1 ± 1.0	93.9 ± 1.1	14	12.0 ± 0.4	99.6 ± 3.1
5	102.5 ± 8.0	91.1 ± 2.0	15	111.3 ± 7.6	98.0 ± 4.1
6	113.9 ± 0.9	93.0 ± 0.5	16	134.6 ± 5.9	88.8 ± 3.9
7	101.2 ± 5.8	97.9 ± 0.3	17	102.1 ± 2.5	98.7 ± 2.7
8	134.2 ± 8.1	94.5 ± 3.9	18	99.2 ± 2.7	97.9 ± 4.3
9	121.0 ± 0.6	94.4 ± 1.8	19	104.1 ± 4.6	98.7 ± 3.7
10	109.0 ± 5.3	92.9 ± 4.1	6-shogaol ^c	143.9 ± 12.5	95.6 ± 1.8

^aC6 cells were treated with 20 μM of each test compound. After 24 h, the content of NGF secreted in the C6-conditioned medium was measured by ELISA. The level of secreted NGF is expressed as the percentage of the untreated control (set as 100%). ^bCell viability after treatment with 20 μM of each compound was determined by an MTT assay and is expressed as a percentage (%). Results are the means of three independent experiments, and the data are expressed as means ± SD. ^cPositive control substance.

HCT-15) through an SRB assay. All the tested compounds showed no activity for the cell lines (IC₅₀ > 10 μM).

Conclusion

Isolation of phytochemical constituents from the twigs of *A. fordii* led to the discovery of three new neolignan glycosides **1–3**, a new phenolic glycoside **15**, and a new cyanoglycoside **16** along with 14 known compounds **4–14** and **17–19**. The structural characterization of the new compounds was conducted based on the analysis of their spectroscopic and spectrometric data, and chemical methods. All isolated compounds were tested for their antineuroinflammatory and neuroprotective activities. Compound **14** showed inhibition effects on NO production and the stereoisomers **2** and **5** demonstrated the difference

in activity according to the configuration. Compounds **8** and **16** exhibited neuroprotection effects. Thus, this study indicates that the active phenolic compounds from *A. fordii* would be potential candidates for drug discovery associated with antineurodegenerative diseases.

Experimental

General experimental procedures. Optical rotations were measured on a JASCO P-2000 polarimeter. IR spectra were acquired with a JASCO FT/IR-4600 spectrometer. UV spectra were obtained on a Shimadzu UV-1601 UV–visible spectrophotometer. NMR spectra were recorded on a Bruker AVANCE III 700 NMR spectrometer operating at 700 MHz (¹H) and 175 MHz (¹³C) with chemical shifts given in ppm (δ)

and a Varian UNITY INOVA 500 NMR spectrometer (Varian Palo Alto, CA, USA) operating at 500 MHz (^1H) and 125 MHz (^{13}C). HRESIMS spectra were obtained on a Waters SYNAPT G2 mass spectrometer and semipreparative HPLC was conducted using a Gilson 306 pump with a Shodex refractive index detector and a Phenomenex Luna 10 μm column (250 \times 10 mm). Silica gel 60 (Merck, Darmstadt, 70–230 mesh, and 230–400 mesh) and RP-C₁₈ silica gel (Merck, 230–400 mesh) were used for column chromatography. Low-pressure liquid chromatography was performed over Merck LiChroprep Lobar-A Si gel 60 (240 \times 10 mm) with an FMI QSY-0 pump (ISCO). Merck precoated silica gel F₂₅₄ plates and RP-18 F_{254s} plates were used for TLC. Spots were detected on TLC under UV light or by heating after spraying the samples with anisaldehyde-sulfuric acid.

Plant material. Twigs of *A. fordii* were collected in Chungbuk Goesan, Korea in August 2012 and the plant was identified by Dr. Kang Ro Lee, Professor at Sungkyunkwan University. A voucher specimen (SKKU-NPL 1212) has been deposited in the herbarium of the School of Pharmacy, Sungkyunkwan University, Suwon, Republic of Korea.

Extraction and isolation. Twigs of *A. fordii* (7.0 kg) were extracted three times with 80% aqueous MeOH (each 10 L \times 1 day) under reflux and filtered. The filtrate was evaporated under vacuum to obtain a crude MeOH extract (325 g), which was suspended in distilled water and successively partitioned with *n*-hexane, CHCl₃, EtOAc, and *n*-BuOH, to yield 15 g, 15 g, 9 g, and 23 g of each residue, respectively. The CHCl₃-soluble layer (15.0 g) was separated by Sephadex LH-20 chromatography (80% aq. MeOH) to yield six fractions (C1–C6). Fraction C1 (2.0 g) was subjected to RP-C₁₈ silica gel chromatography, eluting with gradient solvent system (30 \rightarrow 100% aq. MeOH) to yield four subfractions (C2A–C2D). Fraction C2C (210 mg) was purified by semipreparative HPLC (2 mL/min, 50% aq. MeOH) to yield compounds **17** (5 mg) and **19** (6 mg). Fraction C3 (5.2 g) was subjected to RP-C₁₈ silica gel chromatography, eluting with gradient solvent system (30 \rightarrow 100% aq. MeOH) to yield five subfractions (C3A–C3E). Fraction C3B (554 mg) was subjected to repeated RP-C₁₈ silica gel chromatography and further purified by semipreparative HPLC (50% aq. MeOH) to yield compound **18** (8 mg). Fraction C3D (1.0 g) was subjected to repeated RP-C₁₈ silica gel chromatography and further purified by semipreparative HPLC (23% aq. CH₃CN) to yield compound **14** (5 mg). Fraction C4 (0.5 g) was subjected to RP-C₁₈ silica gel chromatography, eluting with gradient solvent system (30 \rightarrow 100% aq. MeOH) to yield six subfractions (C4A–C4F). Compounds **7** (4 mg), **10** (4 mg), and **11** (4 mg) were obtained by purification of fraction

C4D (43 mg) and C4E (57 mg) using semipreparative HPLC (15% aq. CH₃CN).

The EtOAc-soluble layer (9.0 g) was separated on a silica gel column (CHCl₃/MeOH 15:1 \rightarrow 1:1) to yield eight fractions (E1–E8). Fraction E7 (0.4 g) was subjected to RP-C₁₈ silica gel chromatography, eluting with gradient solvent system (30 \rightarrow 100% aq. MeOH) to yield nine subfractions (E7A–E7I). Fractions E7A (31 mg), E7B (98 mg), E7C (30 mg), and E7D (37 mg) were purified by semipreparative HPLC (15% aq. MeOH and 25–30% aq. CH₃CN) to yield compounds **2** (3 mg), **5** (6 mg), and **15** (3 mg). Fraction E8 (0.9 g) was subjected to RP-C₁₈ silica gel chromatography, eluting with gradient solvent system (30 \rightarrow 100% aq. MeOH) to yield six subfractions (E8A–E8F). Compounds **4** (4 mg), **6** (3 mg), and **16** (4 mg) were obtained by purification of fractions E8B (54 mg) and E8C (120 mg) using semipreparative HPLC (40% aq. MeOH).

The *n*-BuOH-soluble layer (23.0 g) was chromatographed on a Diaion HP-20 column, eluting with an isocratic solvent system of 100% H₂O and 100% MeOH, yielding H₂O and MeOH-soluble fractions. The MeOH fraction was subjected to separation on a silica gel column (CHCl₃/MeOH/H₂O 6:1:0.1 \rightarrow 1:1:0.1) to afford five fractions (BM1–BM5). Fraction BM3 (2.5 g) was fractionated over an RP-C₁₈ silica gel column, eluting with gradient solvent system (25 \rightarrow 100% aq. MeOH) to give nine subfractions (BM3A–BM3I). Subfraction BM3E (61 mg) was purified by semipreparative HPLC (23% aq. CH₃CN) to acquire compound **13** (3 mg). Compounds **3** (8 mg) and **8** (8 mg) were isolated upon purification of subfraction BM3G (52 mg) by semipreparative HPLC (17% aq. CH₃CN). Fraction BM4 (1.0 g) was subjected to passage over an RP-C₁₈ silica gel column, eluting with gradient solvent system (15 \rightarrow 100% aq. MeOH) to acquire 17 subfractions (BM4A–BM4Q). Compounds **1** (5 mg) and **12** (9 mg) were obtained by purification of fraction BM4M (65 mg) using semipreparative HPLC (15% aq. CH₃CN). Fraction BM4N (58 mg) was purified by semipreparative HPLC (30% aq. CH₃CN) to yield compound **9** (10 mg).

Aleuriticide A (1). Colorless gum; $[\alpha]_{\text{D}}^{25}$ -12.1 (*c* 0.05, MeOH); IR (KBr) ν_{max} : 3360, 2943, 2830, 1448, 1033 cm^{-1} ; UV (MeOH) λ_{max} , nm (log ϵ): 282 (1.40), 228 (3.61); ECD (MeOH) λ_{max} , nm ($\Delta\epsilon$): 292 (5.3), 248 (3.3), 221 (-2.1); ^1H and ^{13}C NMR data, see Table 1; positive HRMS–FAB (*m/z*): $[\text{M} + \text{Na}]^+$ calcd for C₂₅H₃₂O₁₁Na, 531.1837; found, 531.1844.

Aleuriticide B (2). Colorless gum; $[\alpha]_{\text{D}}^{25}$ -15.4 (*c* 0.05, MeOH); IR (KBr) ν_{max} : 3355, 2945, 2832, 1453, 1033 cm^{-1} ; UV (MeOH) λ_{max} , nm (log ϵ): 283 (1.31), 230 (3.53); ECD

(MeOH) λ_{\max} , nm ($\Delta\epsilon$): 276 (−3.3), 248 (5.1), 229 (−8.5) nm; ^1H and ^{13}C NMR data, see Table 1; positive HRMS–ESI (m/z): $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{26}\text{H}_{34}\text{O}_{10}\text{Na}$, 529.2050; found, 529.2050.

Aleuriticide C (3). Colorless gum; $[\alpha]_{\text{D}}^{25}$ −23.4 (*c* 0.05, MeOH); IR (KBr) ν_{\max} : 3361, 2946, 2830, 1462, 1029 cm^{-1} ; UV (MeOH) λ_{\max} , nm ($\log \epsilon$): 275 (2.53); ECD (MeOH) λ_{\max} , nm ($\Delta\epsilon$): 273 (−8.1), 236 (−8.3); ^1H and ^{13}C NMR data, see Table 1; positive HRMS–ESI (m/z): $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{26}\text{H}_{36}\text{O}_{11}\text{Na}$, 547.2155; found, 547.2155.

Aleuriteoside A (15). Colorless gum; $[\alpha]_{\text{D}}^{25}$ −13.7 (*c* 0.08, MeOH); IR (KBr) ν_{\max} : 3321, 2975, 1675, 1601, 1453, 1065 cm^{-1} ; UV (MeOH) λ_{\max} , nm ($\log \epsilon$): 280 (2.31); ^1H and ^{13}C NMR data, see Table 2; positive HRMS–ESI (m/z): $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{17}\text{H}_{24}\text{O}_{10}\text{Na}$, 411.1267; found, 411.1260.

Aleucyanoglucoside (16). Colorless gum; $[\alpha]_{\text{D}}^{25}$ −33.5 (*c* 0.30, MeOH); IR (KBr) ν_{\max} : 3535, 3330, 2832, 2218, 1453, 1033 cm^{-1} ; UV (MeOH) λ_{\max} , nm ($\log \epsilon$): 283 (1.31), 230 (3.53); ^1H and ^{13}C NMR data, see Table 2; positive HRMS–ESI (m/z): $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{24}\text{H}_{24}\text{NO}_{14}\text{Na}$, 572.1016; found, 572.1014.

Acid hydrolysis and sugar analysis. In a manner similar as described in [35], compounds **1–3** and **15** (each 1.0–2.0 mg) were refluxed with 1 mL of 1 N HCl for 1 h at 90 °C. The hydrolysate was extracted with EtOAc and the aqueous layer was neutralized by passing it through an Amberlite IRA-67 column to give the sugar. The sugar obtained from the hydrolysis was dissolved in anhydrous pyridine (0.5 mL) followed by adding of L-cysteine methyl ester hydrochloride (2.0 mg, Sigma-Aldrich, St. Louis, MO, USA). The mixture was stirred at 60 °C for 1.5 h and trimethylsilylated with 1-trimethylsilylimidazole (0.1 mL, Sigma-Aldrich, St. Louis, MO, USA) for 2 h. The mixture was partitioned between *n*-hexane and H₂O (1.0 mL each), and the organic layer (1.0 μL) was analyzed by GC–MS. The identification of D-glucose and L-rhamnose was carried out by coinjection of the hydrolysate with standard silylated samples, giving a single peak at 9.730 and 9.712 min, respectively. Authentic samples (Sigma-Aldrich, St. Louis, MO, USA) that were treated in the same way showed a single peak at 9.730 and 9.708 min, respectively.

Alkaline hydrolysis of 16. In a manner similar as described in [36], compound **16** (1.0 mg) was hydrolyzed with 0.1 N KOH (1 mL) at room temperature for 4 h. The reaction mixture was subsequently passed through an ion exchange column (Dowex® 50WX8 hydrogen form, Sigma-Aldrich, St. Louis, MO, USA)

using deionized water to remove KOH. A portion of the reaction product was partitioned between EtOAc/H₂O (each 1.0 mL) and the aglycone **16a** was acquired from the EtOAc-soluble phase.

Measurement of nitric oxide production and cell viability. In a manner similar as described in [37], BV2 cells were used to test the inhibitory effect of the isolated compounds on LPS-stimulated NO production [38,39]. The BV2 cells seeded on a 96-well plate (4×10^4 cells/well) were treated with and without various concentrations of the test compounds. The treated cells were stimulated with LPS (100 ng/mL) and incubated for 24 h. The level of nitrite (NO₂, a soluble oxidation product of NO) in the culture medium was measured using the Griess reagent (0.1% *N*-1-naphthylethylenediamine dihydrochloride and 1% sulfanilamide in 5% phosphoric acid). The supernatant (50 μL) in each well was harvested and mixed with an equal volume of Griess reagent. After 10 min, the absorbance was measured at 570 nm with a microplate reader (Emax, Molecular Devices, Sunnyvale, CA, USA). Graded sodium nitrite solution was used as a standard to gauge NO₂ concentration. Cell viability was assessed by the MTT assay.

Measurement of NGF secretion and cell viability assays. C6 glioma cells (Korean Cell Line Bank, Seoul, Republic of Korea) were used to measure the release of nerve growth factor (NGF) into the culture medium. The C6 cells were seeded onto 24-well plates at a density of 1×10^5 cells/well. After 24 h, the cells were treated with serum-free DMEM and incubated with different concentrations of the test compounds for an additional 24 h. The NGF levels were evaluated in the medium supernatant using an ELISA development kit. Cell viability was measured using the MTT assay and the results were expressed as a percentage of the control group (untreated cells).

Cytotoxicity assessment. The SRB assay was performed to evaluate cytotoxicity of all the isolated compounds against four cultured human cancer cell lines. The cell lines (National Cancer Institute, Bethesda, MD, USA) were used A549 (non-small cell lung adenocarcinoma), SK-OV-3 (ovarian malignant ascites), SK-MEL-2 (skin melanoma), and HCT-15 (colon adenocarcinoma). Each cell line was inoculated over standard 96-well flat-bottom microplates and incubated for 24 h at 37 °C in condition of a humidified atmosphere of 5% CO₂. The attached cells were incubated with various concentrations of the test compounds and the cells were cultured for 48 h. Then, the culture medium was removed from each well and the cells were fixed with 10% cold trichloroacetic acid at 4 °C for 1 h. After the supernatant was discarded and the plates were washed with tap water, the cells were stained with 0.4% SRB solution and incubated for 30 min at room temperature. These cells were

washed to remove the unbound dye and subsequently solubilized with 10 mM unbuffered Tris base solution (pH 10.5). The absorbance was measured spectrophotometrically at 520 nm with a microtiter plate reader. Etoposide (Sigma Chemical Co., ≥98%) was used as a positive control.

Supporting Information

Supporting Information File 1

Copies of NMR spectra including 1D and 2D NMR and HRMS data of compounds **1–3**, **15**, and **16** and ECD spectra of compounds **1–3**.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-17-151-S1.pdf>]

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ORCID® iDs

Kyoung Jin Park - <https://orcid.org/0000-0001-8786-1644>

Kang Ro Lee - <https://orcid.org/0000-0002-3725-5192>

References

- Schönemann, A.; Edwards, H. G. M. *Anal. Bioanal. Chem.* **2011**, *400*, 1173–1180. doi:10.1007/s00216-011-4855-0
- Pencreac'h, G.; Graille, J.; Pina, M.; Verger, R. *Anal. Biochem.* **2002**, *303*, 17–24. doi:10.1006/abio.2001.5427
- Ito, Y.; Yanase, S.; Tokuda, H.; Kishishita, M.; Ohigashi, H.; Hirota, M.; Koshimizu, K. *Cancer Lett.* **1983**, *18*, 87–95. doi:10.1016/0304-3835(83)90121-0
- Okuda, T.; Yoshida, T.; Toh, N. *Phytochemistry* **1975**, *14*, 2513–2514. doi:10.1016/0031-9422(75)80383-9
- Fozdar, B. I.; Khan, S. A.; Shamsuddin, T.; Shamsuddin, K. M.; Kintzinger, J. *Phytochemistry* **1989**, *28*, 2459–2461. doi:10.1016/s0031-9422(00)98005-1
- Taniguchi, S.; Uechi, K.; Kato, R.; Ito, H.; Hatano, T.; Yazaki, K.; Yoshida, T. *Planta Med.* **2002**, *68*, 1145–1146. doi:10.1055/s-2002-36348
- Lee, Y.-C.; Nobles, W. L. *J. Am. Pharm. Assoc., Sci. Ed.* **1959**, *48*, 162–165. doi:10.1002/jps.3030480308
- Matsuda, S.; Nakao, Y.; Ohigashi, H.; Koshimizu, K.; Ito, Y. *Int. J. Cancer* **1986**, *38*, 859–865. doi:10.1002/ijc.2910380613
- Matsumoto, T.; Nakamura, S.; Nakashima, S.; Ohta, T.; Ogawa, K.; Fukaya, M.; Tsukioka, J.; Hasei, T.; Watanabe, T.; Matsuda, H. *Phytochemistry* **2017**, *137*, 101–108. doi:10.1016/j.phytochem.2017.02.007
- Kim, C. S.; Subedi, L.; Kim, S. Y.; Choi, S. U.; Kim, K. H.; Lee, K. R. *J. Nat. Prod.* **2015**, *78*, 1174–1178. doi:10.1021/acs.jnatprod.5b00090
- Suh, W. S.; Kim, K. H.; Kim, H. K.; Choi, S. U.; Lee, K. R. *Helv. Chim. Acta* **2015**, *98*, 1087–1094. doi:10.1002/hlca.201500002
- Takeda, Y.; Mima, C.; Masuda, T.; Hirata, E.; Takushi, A.; Otsuka, H. *Phytochemistry* **1998**, *49*, 2137–2139. doi:10.1016/s0031-9422(98)00362-8
- Kim, T. H.; Ito, H.; Hayashi, K.; Hasegawa, T.; Machiguchi, T.; Yoshida, T. *Chem. Pharm. Bull.* **2005**, *53*, 641–644. doi:10.1248/cpb.53.641
- Kim, C. S.; Subedi, L.; Park, K. J.; Kim, S. Y.; Choi, S. U.; Kim, K. H.; Lee, K. R. *Fitoterapia* **2015**, *106*, 147–152. doi:10.1016/j.fitote.2015.08.013
- Hara, S.; Okabe, H.; Mihashi, K. *Chem. Pharm. Bull.* **1987**, *35*, 501–506. doi:10.1248/cpb.35.501
- García-Muñoz, S.; Álvarez-Corral, M.; Jiménez-González, L.; López-Sánchez, C.; Rosales, A.; Muñoz-Dorado, M.; Rodríguez-García, I. *Tetrahedron* **2006**, *62*, 12182–12190. doi:10.1016/j.tet.2006.10.018
- Nakanishi, T.; Iida, N.; Inatomi, Y.; Murata, H.; Inada, A.; Murata, J.; Lang, F. A.; Iinuma, M.; Tanaka, T. *Phytochemistry* **2004**, *65*, 207–213. doi:10.1016/j.phytochem.2003.10.025
- Tan, R. X.; Jakupovic, J.; Jia, Z. *J. Planta Med.* **1990**, *56*, 475–477. doi:10.1055/s-2006-961015
- Dong, L.-P.; Ni, W.; Dong, J.-Y.; Li, J.-Z.; Chen, C.-X.; Liu, H.-Y. *Molecules* **2006**, *11*, 1009–1014. doi:10.3390/11121009
- Miyase, T.; Ueno, A.; Takizawa, N.; Kobayashi, H.; Oguchi, H. *Chem. Pharm. Bull.* **1988**, *36*, 2475–2484. doi:10.1248/cpb.36.2475
- Zhuo, J.-X.; Wang, Y.-H.; Su, X.-L.; Mei, R.-Q.; Yang, J.; Kong, Y.; Long, C.-L. *Nat. Prod. Bioprospect.* **2016**, *6*, 161–166. doi:10.1007/s13659-016-0095-5
- Lee, T.-H.; Kuo, Y.-C.; Wang, G.-J.; Kuo, Y.-H.; Chang, C.-I.; Lu, C.-K.; Lee, C.-K. *J. Nat. Prod.* **2002**, *65*, 1497–1500. doi:10.1021/np020154n
- Ding, B.; Dai, Y.; Hou, Y.-L.; Wu, X.-M.; Chen, X.; Yao, X.-S. *Fitoterapia* **2013**, *86*, 1–5. doi:10.1016/j.fitote.2013.01.017
- Forero, J. E.; Avila, L.; Taborda, N.; Tabares, P.; López, A.; Torres, F.; Quiñones, W.; Bucio, M. A.; Mora-Pérez, Y.; Rugeles, M. T.; Joseph-Nathan, P.; Echeverri, F. *Phytochemistry* **2008**, *69*, 2815–2819. doi:10.1016/j.phytochem.2008.09.003
- Gottlieb, H. E.; Kumar, S.; Sahai, M.; Ray, A. B. *Phytochemistry* **1991**, *30*, 2435–2438. doi:10.1016/0031-9422(91)83676-c
- Matsuda, N.; Sato, H.; Yaoita, Y.; Kikuchi, M. *Chem. Pharm. Bull.* **1996**, *44*, 1122–1123. doi:10.1248/cpb.44.1122
- He, W. J.; Fu, Z. H.; Han, H. J.; Yan, H.; Zeng, G. Z.; Ji, C. J.; Chu, H. B.; Zhang, Y. M.; Tan, N. H. *Z. Naturforsch., B: J. Chem. Sci.* **2011**, *66*, 733–739. doi:10.1515/znB-2011-0715
- Morikawa, T.; Tao, J.; Ueda, K.; Matsuda, H.; Yoshikawa, M. *Chem. Pharm. Bull.* **2003**, *51*, 62–67. doi:10.1248/cpb.51.62
- Cai, W.-H.; Matsunami, K.; Otsuka, H.; Shinzato, T.; Takeda, Y. *J. Nat. Med.* **2009**, *63*, 408–414. doi:10.1007/s11418-009-0344-6
- Chin, Y.-W.; Chai, H.-B.; Keller, W. J.; Kinghorn, A. D. *J. Agric. Food Chem.* **2008**, *56*, 7759–7764. doi:10.1021/jf801792n
- Matsuda, N.; Kikuchi, M. *Chem. Pharm. Bull.* **1996**, *44*, 1676–1679. doi:10.1248/cpb.44.1676

32. Yang, X.-W.; Zhao, P.-J.; Ma, Y.-L.; Xiao, H.-T.; Zuo, Y.-Q.; He, H.-P.; Li, L.; Hao, X.-J. *J. Nat. Prod.* **2007**, *70*, 521–525.
doi:10.1021/np0603931
33. Xie, Y.-F.; Tao, Z.-M.; Wang, H.-B.; Qin, G.-W. *Zhongguo Tianran Yaowu* **2010**, *8*, 264–266.
doi:10.3724/sp.j.1009.2010.00264
34. Tsukamoto, H.; Hisada, S.; Nishibe, S. *Chem. Pharm. Bull.* **1985**, *33*, 4069–4073. doi:10.1248/cpb.33.4069
35. Kim, C. S.; Subedi, L.; Kim, S. Y.; Choi, S. U.; Choi, S. Z.; Son, M. W.; Kim, K. H.; Lee, K. R. *Phytochem. Lett.* **2015**, *14*, 215–220.
doi:10.1016/j.phytol.2015.10.014
36. Park, K. J.; Subedi, L.; Kim, S. Y.; Choi, S. U.; Lee, K. R. *Bioorg. Chem.* **2018**, *77*, 527–533. doi:10.1016/j.bioorg.2018.02.006
37. Kim, C. S.; Subedi, L.; Oh, J.; Kim, S. Y.; Choi, S. U.; Lee, K. R. *J. Nat. Prod.* **2017**, *80*, 1134–1140. doi:10.1021/acs.jnatprod.7b00111
38. Blasi, E.; Barluzzi, R.; Bocchini, V.; Mazzolla, R.; Bistoni, F. *J. Neuroimmunol.* **1990**, *27*, 229–237.
doi:10.1016/0165-5728(90)90073-v
39. Choi, Y.; Lee, M. K.; Lim, S. Y.; Sung, S. H.; Kim, Y. C. *Br. J. Pharmacol.* **2009**, *156*, 933–940.
doi:10.1111/j.1476-5381.2009.00022.x

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