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Inhibition of Shiga toxin-converting bacteriophage development by novel antioxidant compounds

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ABSTRACT

Oxidative stress may be the major cause of induction of Shiga toxin-converting (Stx) prophages from chromosomes of Shiga toxin-producing *Escherichia coli* (STEC) in human intestine. Thus, we aimed to test a series of novel antioxidant compounds for their activities against prophage induction, thus, preventing pathogenicity of STEC. Forty-six compounds (derivatives of carbazole, indazole, triazole, quinolone, ninhydrine, and indenoindole) were tested. Fifteen of them gave promising results and were further characterized. Eleven compounds had acceptable profiles in cytotoxicity tests with human HEK-293 and HDFa cell lines. Three of them (selected for molecular studies) prevent the prophage induction at the level of expression of specific phage genes. In bacterial cells treated with hydrogen peroxide, expression of genes involved in the oxidative stress response was significantly less efficient in the presence of the tested compounds. Therefore, they apparently reduce the oxidative stress, which prevents induction of Stx prophage in *E. coli.*

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Introduction

The significance of Shiga toxin-producing Escherichia coli (STEC) as a public health problem was recognized in 1982 in USA, during an investigation of the outbreak of hemorrhagic colitis associated with consumption of contaminated hamburgers¹. Since then, STEC strains have been implicated in many outbreaks of diarrhea worldwide. Quite recently (2011), the Shiga toxin-producing E. coli serotype O104:H4 was responsible for a serious epidemic outbreak in Germany^{2,3}. STEC pathogens can cause serious food poisoning with bloody diarrhea in humans⁴. Their main virulence factors are Shiga toxins, encoded by stx genes located in genomes of bacteriophages, which occur in bacteria as prophages⁵. These bacteriophages are called Shiga toxin-converting or shortly Stx phages and belong to the lambdoid family of phages⁶. All phages within this group indicate high similarities in the lifecycle and genomic organization to bacteriophage λ_i , the most reviewed member of this family'. In the prophage state, most of phage genes, including stx genes, are not transcribed due to inhibition caused by the phage cl repressor. Consequently, Shiga toxins are not produced under such conditions. Expression of stx as well as other phage genes occurs effectively only after prophage induction. In most cases, this process requires activation of the RecA-dependent bacterial S.O.S. response, which is provoked by factors causing

appearance of single-stranded DNA fragments. Activated RecA protein stimulates cleavage of both the SOS regulon repressor, the LexA protein, and the cl phage repressor. Prophage induction and subsequent phage lytic development lead to production of progeny phage particles and Shiga toxins, and then to their release from the lysed cell⁸.

Induction of lambdoid prophages is easy and can be provoked by different inducers like treatment with targeted DNA antibiotics (e.g. mitomycin C), UV-treatment⁷, EDTA⁹, ⁶⁰Co irradiation¹⁰, high hydrostatic pressure¹¹, and many others. Such conditions do not naturally occur in the human gut, the place where the STEC infection initiates. Therefore, the oxidative stress conditions, which appear in human intestine as a response to STEC infection, has attracted attention of researchers. Indeed, it has been shown that hydrogen peroxide (which is excreted by neutrophils during infection) increases efficiency of Stx prophage induction and stimulates production of Shiga toxins¹²⁻¹⁴. In bacteria, H₂O₂ acts as a DNAdamaging agent that interacts with free cellular iron and form hydroxyl radicals OH[•] in Fenton reaction. DNA subjected to attack by OH[•] generates a wide range of base and sugar modifications and strand breaks, which in turn activate the RecA protein and SOS response¹⁵. According to hypothesis proposed by two different research groups^{16,17}, STEC strains may obtain advantage due to H₂O₂-mediated production of Shiga toxins. As demonstrated,

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• Supplemental data for this article can be accessed here.

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these toxins are produced in a small fraction of STEC cells (usually less than 1%) which are sacrificed for the good of the whole bacterial population. Released toxins kill H_2O_2 -producing protozoan predator, so allowing the rest of bacteria to survive its attack. It is suspected that STEC response to attack of neutrophils in human intestine might be analogous^{16,17}.

The OxyR and SoxR regulators mainly control E. coli defense against oxidative stress. Interestingly, H₂O₂ activate genes of both regulons: OxyR (about 30 genes) and SoxRS (about 17 genes)¹⁸. It was demonstrated that nitric oxide and its derivatives (NOs) as well as S-nitrosothiols (SNOs) may be involved in signaling in bacteria and may change properties of several bacterial proteins including OxyR, SoxR, NorR and Fur in Escherichia coli^{19–22}, as well as ResDE in Bacillus subtilis²³. As reported previously, certain SNOs are able to provoke a nitrosative stress in E. coli bacteria, evidenced by lowering of intracellular thiol and the transcriptional activation of OxyR by S-nitrosylation²⁴. OxyR is a thiol-containing transcriptional activator of antioxidant genes that are involved in H₂O₂ detoxification. The activation of *E. coli* antioxidant enzymes by OxyR in response to H_2O_2 has been widely described. The genes controlled by OxyR include catalase/hydroperoxidase (katG), glutathione reductase (gorA), and alkyl hydroperoxide reductase (ahpCF). Their expression is essential to ensure resistance to $H_2O_2^{24-26}$. As demonstrated previously, S-nitrosylation, S-hydroxylation, and S-glutathionylation of wild-type OxyR activate this protein and appear to serve as the functional switches in the activation of OxyR-regulated genes^{24,27}. On the other hand, nitric oxide (NO) donors are able to activate SoxRS regulon and induce the expression of enzymes involved in the oxidative stress response of E. coli bacteria, such as the superoxide dismutase (sodA)^{28,29}.

It was previously proposed that OxyR may be responsible for the lambdoid prophage maintenance under conditions of the oxidative stress⁸. Other studies indicated that under such conditions, activated OxyR is able to compete effectively with the cl protein (the main repressor of phage lytic development) for the binding to the o_{R3} operator of the p_M - p_R promoter region³⁰. These results allowed to suggest that OxyR might enhance repression of p_R (the major promoter for expression of genes of phage lytic lifecycle) and enhance activation, and at the same time also downregulate repression, of the transcription of the *cl* gene from the p_M promoter. This would lead to higher (relative to normal growth conditions) activity of p_M , increased production of cl repressor and lower efficiency of prophage induction under the oxidative stress.

In the light of the importance of the oxidative stress in induction of Stx prophages and Shiga toxin production (summarized above), we analyzed in this study impact of the different antioxidant compounds, on viability of STEC and development of the Stx phage $\Phi 24_{B}$. Since virulence of STEC depends on the prophage induction, and in the light of the fact that most of antibiotics should not be used for treatment of STEC-infected patients, development of novel drugs that would impair survival of STEC (without causing prophage induction) or prophage induction/ development or both would be highly desirable. Although development of novel compounds which might be potentially used to combat E. coli infections, exemplified by those based on furanylderived sulfonamides, has been reported³¹, and sophisticated techniques become available to either engineer E. coli surface³² or to use nanoparticles as carriers of compounds³³, the anti-bacterial activity of these molecules were moderate³¹. Therefore, we have tested a series of novel compounds that are derivatives of carbazole, indazole, triazole, quinolone, ninhydrin, and indenoindole.

Materials and methods

Chemistry

Chemical structures of all compounds tested in this report are presented in the supporting information (Supplementary Figures 1-4) with the corresponding references. Seven indenoindole derivatives are described in this paper (CM3159A, MF4, THN10, CM3072B, CM3159B, MF5, MF1). Melting points were determined on an Electrothermal 9200 capillary apparatus. The IR spectra were recorded on a PerkinElmer Spectrum Two IR spectrometer. The ¹H and ¹³C NMR spectra were recorded on a Bruker DRX 400 spectrometer. Chemical shifts are expressed in ppm (δ) downfield from internal tetramethylsilane, and coupling constants J are reported in hertz (Hz). The following abbreviations are used: s, singlet; bs, broad singlet; d, doublet; t, triplet; dd, doubled doublet; q, quartet; sept, septuplet; m, multiplet; Cquat, quaternary carbons. The mass spectra were performed by direct ionization (El or Cl) on a ThermoFinnigan MAT 95 XL apparatus. Chromatographic separations were performed on silica gel columns by column chromatography (Kieselgel 300-400 mesh). All reactions were monitored by TLC on GF254 plates that were visualized under a UV lamp (254 nm). Evaporation of solvent was performed in vacuum with rotating evaporator. The purity of the final compounds (greater than 95%) was determined by uHPLC/MS on an Agilent 1290 system using a Agilent 1290 Infinity ZORBAX Eclipse Plus C18 column $(2.1 \times 50 \text{ mm}, 1.8 \,\mu\text{m} \text{ particle size})$ with a gradient mobile phase of H₂O/CH₃CN (90:10, v/v) with 0.1% of formic acid to H₂O/CH₃CN (10:90, v/v) with 0.1% of formic acid at a flow rate of 0.5 ml/min, with UV monitoring at the wavelength of 254 nm. A run time of 10 min was applied. LC/HRMS data of all new indenoindole derivatives are presented in the supporting information.

General procedure for the synthesis of compound 3

A solution containing equimolar amounts of enaminone **2** and ninhydrin **1** dissolved in MeOH was stirred at room temperature for 19 h. Generally, a precipitate of compound **3** was formed. The solvent was evaporated, and the product triturated, filtrated and washed with EtOAc. A second quantity could be obtained from the filtrate by purification by silica gel column chromatography with $CH_2Cl_2/acetone$ (96:4, v/v) as the eluent.

Mixture of regioisomers 4 b,9 b-Dihydroxy-5-isopropyl-1-methoxy-4 b,5,6,7,8,9 b-hexahydroindeno[1,2-b]indole-9,10-dione (major, MAJ) 4 b,9 b-Dihydroxy-5-isopropyl-4-methoxy-4 b,5,6,7,8,9 b-hexahyand droindeno[1,2-b]indole-9,10-dione (minor, min) CM3072A: beige solid. Yield 77%. IR (v cm⁻¹): 3368, 1714, 1595, 1532, 1491 cm⁻¹. ¹H NMR (300 MHz, DMSO-d₆) δ : 7.73 (t, 1H, J = 8.1 Hz, H-2 MAJ), 7.54 (t, 1H, J = 7.9 Hz, H-3 min), 7.44 (d, 1H, J = 7.5 Hz, H-4 MAJ), 7.38 (d, 1H, J = 7.5 Hz, H-3 min), 7.26 (d, 1H, J = 7.3 Hz, H-2 min), 7.12 (d, 1H, J = 8.3 Hz, H-4 MAJ), 6.66 (s, 1H, OH MAJ), 6.32 (s, 1H, OH min), 5.73 (s, 1H, OH min), 5.54 (s, 1H, OH MAJ), 4.85 (sept, 1H, J = 7.1 Hz, CH(CH₃)₂ min), 4.54 (sept, 1H, J = 7.1 Hz, CH(CH₃)₂ MAJ), 3.95 (s, 3H, OCH₃ min), 3.88 (s, 3H, OCH₃ MAJ), 2.62-2.82 (m, 4H, H-6 MAJ + min), 2.06-2.13 (m, 4H, H-8 MAJ + min), 1.79-1.91 (m, 4H, H-7 MAJ + min), 1.45 (d, 3H, J = 6.8 Hz, CH₃ MAJ), 1.39 (d, 3H, J = 7.2 Hz, CH₃ min), 1.33 (d, 3H, J = 7.0 Hz, CH₃ min), 1.25 (d, 3H, $J = 7.1 \text{ Hz}, \text{ CH}_3 \text{ MAJ}$).

4 b,9 b-Dihydroxy-5-isopropyl-1,4-dimethoxy-4 b,5,6,7,8,9 b-hexahydroindeno[1,2-b]indole-9,10-dione **CM3159A**: beige solid. Yield 66%. Mp 216 °C. IR (v cm⁻¹): 3347, 1717, 1497, 1270, 999. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.30 (d, 1H, J = 8.9 Hz, H-3), 7.05 (d, 1H, J = 8.9 Hz, H-2), 6.27 (s, 1H, OH), 5.65 (s, 1H, OH), 4.82 (sept, 1H, J = 7.1 Hz, CH(CH₃)₂), 3.88 (s, 3H, OCH₃), 3.80 (s, 3H, OCH₃), 2.70–2.79 (m, 1H, CH₂), 2.40–2.50 (m, 1H, CH₂), 2.06 (t, 2H, J = 6.3 Hz, CH₂), 1.75–1.84 (m, 2H, CH₂), 1.36 (d, 3H, J = 7.1 Hz, CH₃), 1.31 (d, 3H, J = 7.1 Hz, CH₃). ¹³C NMR + DEPT (100 MHz, DMSO-d₆) δ : 194.7 (C = O), 188.7 (C = O), 165.5 (Cquat), 150.6 (Cquat), 149.5 (Cquat), 136.1 (Cquat), 124.3 (Cquat), 118.4 (CH), 112.9 (CH), 106.2 (Cquat), 95.8 (Cquat), 82.9 (Cquat), 55.7 (2 OCH₃), 45.1 (CH), 36.7 (CH₂), 24.5 (CH₂), 22.4 (CH₃), 22.2 (CH₃), 21.8 (CH₂). HRMS calcd for C₂₀H₂₄NO₆ [M + H]⁺ 374.1598 found, 374.1599.

4 b,9 b-Dihydroxy-5-isopropyl-2,3-dimethoxy-4 b,5,6,7,8,9 b-hexahydroindeno[1,2-b]indole-9,10-dione **MF4**: white solid. Yield 87%. Mp 206 °C. IR (ν cm⁻¹): 3071, 1727, 1712, 1488, 1284, 1174, 954. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.39 (s, 1H, H-1), 7.09 (s, 1H, H-4), 6.60 (s, 1H, OH), 5.50 (s, 1H, OH), 4.64 (sept, 1H, J=7.1 Hz, CH(CH₃)₂), 3.92 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 2.44–2.70 (m, 2H, CH₂), 2.05 (t, 2H, J=6.3 Hz, CH₂), 1.72–1.85 (m, 2H, CH₂), 1.43 (d, 3H, J=7.1 Hz, CH₃), 1.24 (d, 3H, J=7.1 Hz, CH₃). ¹³C NMR + DEPT (75 MHz, DMSO-d₆) δ : 196.2 (C=O), 188.8 (C=O), 164.2 (Cquat), 155.5 (Cquat), 151.0 (Cquat), 142.9 (Cquat), 127.9 (Cquat), 105.7 (CH), 105.5 (Cquat), 103.6 (CH), 95.4 (Cquat), 82.8 (Cquat), 56.3 (OCH₃), 55.7 (OCH₃), 44.7 (CH), 36.7 (CH₂), 24.3 (CH₂), 22.6 (CH₃), 22.1 (CH₃), 22.0 (CH₂). HRMS calcd for C₂₀H₂₄NO₆ [M + H]⁺ 374.1598, found 374.1603.

4 b,9 b-Dihydroxy-5-isopropyl-1-nitro-4 b,5,6,7,8,9 b-hexahydroindeno[1,2-b]indole-9,10-dione **AM10A**: yellow solid. Yield 92%. Mp 227 °C. IR (ν cm⁻¹): 3553, 1746, 1589, 1536, 1491. ¹H NMR (500 MHz, DMSO-d₆) δ : 8.28 (d, 1H, J=7.6 Hz, H-4), 8.04 (d, 1H, J=7.6 Hz, H-2), 7.99 (t, 1H, J=7.6 Hz, H-3), 7.04 (s, 1H, OH), 5.87 (s, 1H, OH), 4.61 (sept, 1H, J=7.2 Hz, CH₁(CH₃)₂), 2.41–2.77 (m, 2H, CH₂), 2.10 (t, 2H, J=6.0 Hz, CH₂), 1.84 (m, 2H, CH₂), 1.45 (d, 3H, J=6.9 Hz, CH₃), 1.27 (d, 3H, J=6.9 Hz, CH₃). ¹³C NMR + DEPT (125 MHz, DMSO-d₆) δ : 192.8 (C=O), 188.7 (C=O), 164.9 (Cquat), 148.8 (Cquat), 145.1 (Cquat), 136.1 (CH), 128.9 (CH), 126.5 (Cquat), 124.8 (CH), 104.5 (Cquat), 94.6 (Cquat), 83.3 (Cquat), 44.9 (CH(CH₃)₂), 36.7 (CH₂), 24.2 (CH₂), 22.5 (CH₃), 22.4 (CH₃), 21.9 (CH₂). HRMS calcd for C₁₈H₁₉N₂O₆ [M + H]⁺, 359.1238 found: 359.1240.

General procedure for the synthesis of compound 4

A solution containing 5.92 mmol of **3** and 11.5 mmol (2 equiv) of TETA dissolved in 18 ml of DMF and 3.5 ml of AcOH was stirred at room temperature for 22 h. The solution was then poured into 50 ml of ice and water and stirred for 30 min. The resulting precipitate was filtered, washed with water and dried to give a first quantity of **4**. EtOAc was added to the filtrate, and the organic product extracted. The organic phase was washed with water, dried over anhydrous Na₂SO₄ and evaporated in vacuum to give a second quantity of **4** which was purified by silica gel column chromatography with CH₂Cl₂/acetone (95:5, v/v) as the eluent.

5-Isopropyl-1-methoxy-5,6,7,8-tetrahydroindeno[*1,2-b*]*indole-9,10-dione* **THN10**: brown solid. Yield 50%. Mp 186 °C. IR (ν cm⁻¹): 1694, 1661, 1590, 1261, 1023, 803. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.38 (dd, 1H, $J_1 = 8.6$ Hz, $J_2 = 7.5$ Hz, H-3), 7.04 (d, 1H, J = 7.3 Hz, H-4), 6.98 (d, 1H, J = 8.8 Hz, H-2), 4.77 (sept, 1H, J = 6.9 Hz, C<u>H</u>(CH₃)₂), 3.86 (s, 3H, OCH₃), 2.93 (m, 2H, CH₂), 2.37 (m, 2H, CH₂), 2.08 (m, 2H, CH₂), 1.57 (d, 6H, J = 7.1 Hz, CH₃). ¹³C NMR + DEPT (100 MHz, DMSO-d₆) δ : 191.2 (C = O), 182.2 (C = O), 156.8 (2 Cquat), 150.1 (2 Cquat), 136.8 (2 Cquat), 135.1 (CH), 121.2 (Cquat), 114.7 (CH), 112.8 (CH), 55.6 (OCH₃), 49.5 (<u>C</u>H(CH₃)₂), 37.7 (CH₂), 22.8 (2 CH₂), 21.3 (2 CH₃). HRMS calcd for C₁₉H₂₀NO₃ [M + H]⁺ 310.1438, found: 310.1438.

5-Isopropyl-4-methoxy-5,6,7,8-tetrahydroindeno[1,2-*b*]*indole-9,10dione* **CM3072B**: red solid. Yield 24%. Mp 224 °C. IR (v cm⁻¹): 3436, 1704, 1660, 1600, 1488, 1270. ¹H NMR (400 MHz, DMSO-*d₆*) δ : 7.19–7.28 (m, 2H, H-1 and H-2), 7.02 (d, 1H, J=8.8 Hz, H-3), 5.67 (m, 1H, CH(CH₃)₂), 2.94 (s, 3H, OCH₃), 3.00 (m, 2H, CH₂), 2.38 (t, 2H, J=6.3 Hz, CH₂), 2.07 (t, 2H, J=6.0 Hz, CH₂), 1.56 (d, 6H, J=7.0 Hz, CH₃). ¹³C NMR + DEPT (100 MHz, DMSO-d₆) δ : 191.4 (C=O), 183.1 (C=O), 150.2 (Cquat), 149.8 (2 Cquat), 139.6 (2 Cquat), 130.6 (CH), 121.2 (2 Cquat), 119.2 (CH), 116.3 (CH), 56.3 (OCH₃), 51.0 (CH(CH₃)₂), 37.7 (CH₂), 24.5 (CH₂), 23.4 (CH₂), 21.6 (2 CH₃). HRMS calcd for C₁₉H₂₀NO₃ [M + H]⁺ 310.1438, found: 310.1435.

1,4-Dimethoxy-5-isopropyl-5,6,7,8-tetrahydroindeno[1,2-b]indole-9,10-dione **CM3159B**: orange-red solid. Yield 75%. Mp 247 °C. IR (ν cm⁻¹): 1693, 1661, 1494, 1262. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.19 (d, 1H, J=9.3 Hz, H-3), 6.99 (d, 1H, J=9.3 Hz, H-2), 5.71 (m, 1H, CH₂(CH₃)₂), 3.90 (s, 3H, OCH₃), 3.81 (s, 3H, OCH₃), 2.99 (m, 2H, CH₂), 2.36 (m, 2H, CH₂), 2.06 (m, 2H, CH₂), 1.54 (d, 6H, J=7.1 Hz, CH₃). ¹³C NMR + DEPT (75 MHz, DMSO-d₆) δ : 191.2 (C=O), 181.7 (C=O), 151.9 (2 Cquat), 149.9 (Cquat), 144.3 (2 Cquat), 122.8 (Cquat), 122.2 (Cquat), 121.5 (CH), 117.1 (CH), 56.0 (2 OCH₃), 50.7 (CH(CH₃)₂), 37.7 (2 CH₂), 23.4 (CH₂), 21.6 (CH₃). HRMS calcd for $\overline{C_{20}H_{21}}$ NNaO₄ [M + Na]⁺ 362.1363, found: 362.1364.

2,3-Dimethoxy-5-isopropyl-5,6,7,8-tetrahydroindeno[1,2-b]indole-9,10-dione **MF5**: red solid. Yield 74%. Mp 245 °C. IR (υ cm⁻¹): 1696, 1659, 1268. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.04 (s, 1H, H-1), 6.97 (s, 1H, H-4), 4.77 (sept, 1H, J = 6.9 Hz, CH(CH₃)₂), 3.93 (s, 3H, OCH₃), 3.84 (s, 3H, OCH₃), 2.94 (m, 2H, CH₂), 2.38 (t, 2H, J = 6.0 Hz, CH₂), 2.10 (t, 2H, J = 5.9 Hz, CH₂), 1.61 (d, 6H, J = 6.5 Hz, CH₃). ¹³ C NMR + DEPT (100 MHz, DMSO-d₆) δ : 190.6 (C = O), 182.6 (C = O), 151.4 (Cquat), 150.6 (Cquat), 148.8 (Cquat), 148.1 (Cquat), 130.8 (Cquat), 128.6 (Cquat), 118.6 (Cquat), 116.9 (Cquat), 109.2 (CH), 106.0 (CH), 56.3 (OCH₃), 55.9 (OCH₃), 48.9 (CH(CH₃)₂), 37.3 (CH₂), 22.6 (CH₂), 22.5 (CH₂), 21.2 (2 CH₃). HRMS calcd for C₂₀H₂₁NNaO₄ [M + Na]⁺ 362.1363, found: 362.1366.

5-Isopropyl-1-nitro-5,6,7,8-tetrahydroindeno[*1,2-b*]*indole-9,10-dione* **AM12B**: red solid. Yield 80%. Mp 244 °C. IR ($v \text{ cm}^{-1}$): 1707, 1666, 1605, 1536. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.63 (d, 1H, *J* = 7.5 Hz, H-2), 7.59 (t, 1H, *J* = 7.5 Hz, H-3), 7.49 (d, 1H, *J* = 7.0 Hz, H-4), 4.80 (sept, 1H, *J* = 7.0 Hz, CH(CH₃)₂), 2.95 (m, 2H, CH₂), 2.37 (t, 2H, *J* = 6.0 Hz, CH₂), 2.06 (m, 2H, CH₂), 1.56 (d, 6H, *J* = 7.0 Hz, CH₃). ¹³ C NMR + DEPT (100 MHz, DMSO-d₆) δ : 190.8 (C = O), 177.8 (C = O), 155.7 (2 Cquat), 149.7 (Cquat), 145.5 (Cquat), 136.0 (2 Cquat), 134.9 (CH), 127.4 (Cquat), 121.6 (CH), 121.0 (CH), 49.5 (CH(CH₃)₂), 37.3 (CH₂), 22.8 (CH₂), 22.5 (CH₂), 21.2 (2 CH₃). HRMS calcd for C₁₈H₁₆N₂NaO₄ [M + Na]⁺, 347.1002 found: 347.0989.

1-Amino-5-isopropyl-5,6,7,8-tetrahydroindeno[1,2-b]indole-9,10dione MF1: To a stirred solution of nitro derivative AM12B (485 mg, 1.50 mmol) in 4.9 ml of glacial acetic and 3.4 ml of water at 90 °C was added iron (686 mg, 12.2 mol) in portions. The resulting suspension was stirred at the same temperature for 45 min. Then, the reaction mixture was cooled to room temperature and basified with saturated NaHCO₃ aqueous solution. The organic phase was extracted with dichloromethane washed with brine, dried with Na2SO4, filtered and evaporated to dryness. The residue was purified by silica gel column chromatography with CH₂Cl₂/ acetone (1:1, v/v) as the eluent to afford 254 mg of **MF1**: yellow solid. Yield 58%. Mp 238 °C. IR (v cm⁻¹): 3448, 3351, 1673, 1625, 1463. ¹H NMR (400 MHz, DMSO-d₆) δ : 7.07 (t, 1H, J = 7.7 Hz, H-3), 6.64 (d, 1H, J=7.1 Hz, H-2), 6.57 (d, 1H, J=8.6 Hz, H-4), 6.34 (bs, 2H, NH₂), 4.72 (sept, 1H, J=7.0 Hz, CH(CH₃)₂), 2.93 (t, 2H, J = 6.0 Hz, CH₂), 2.38 (t, 2H, J = 6.0 Hz, CH₂), 2.07–2.13 (m, 2H, CH₂), 1.57 (d, 6H, J = 6.8 Hz, CH₃). ¹³C NMR + DEPT (100 MHz, DMSO-d₆) δ : 191.0 (C = O), 186.4 (C = O), 149.1 (Cquat), 148.3 (Cquat), 146.8 (Cquat), 134.8 (Cquat), 133.3 (CH + Cquat), 118.8 (CH), 116.3 (Cquat), 114.0 (Cquat), 109.0 (CH), 48.7 (CH(CH₃)₂), 37.5 (CH₂), 22.7 (2 CH₂), 20.9 (2 CH₃). HRMS calcd for $C_{18}H_{18}N_2NaO_2$ [M + Na]⁺ 317.1260, found: 317.1259.

Prophage induction experiment

E. coli MG1655 strain³⁴ lysogenic for phage $\Phi 24_B \Delta stx2::cat^{35}$ was cultured in LB medium at 37 °C with one of the analyzed compounds (Supplementary Figures S1-S4), in the presence or absence of the prophage inductor: 0.2 µg/ml; 0.5 µg/ml mitomycin C or 1 mM H₂O₂. In control experiments, DMSO (a solvent used for preparation of stock solutions) was added instead of the tested compound. Each analyzed compound was added at the beginning of the culture (time 0), at three different concentrations (50, 100 and 200 μ M), and selected prophage inductor was added after 3 h. Measurements of the bacterial cultures optical density at 600 nm (A600) were performed at indicated times (0, 3, 5, 7, and 9 h). Six hours after an induction (at the ninth hour of the experiment), 50 μ l samples were withdrawn, mixed with $10 \,\mu$ l of chloroform and centrifuged (4,500 rpm, 5 min). Serial dilutions were prepared in TM buffer (10 mM Tris-HCl, 10 mM MgSO₄; pH 7.2). Phage titer (number of phages per ml) was determined by spotting 2.5 μ l of each dilution of the phage lysate on a freshly prepared LB agar (1.5%) with 2.5 μ g/ml chloramphenicol, to obtain visible plaques formed on bacterial lawn (according to a procedure described previously³⁶), with a poured mixture of 1 ml indicator E. coli MG1655 strain culture and 2 ml of 0.7% nutrient agar (prewarmed to 45 °C), supplemented with MgSO₄ and CaCl₂ (to a final concentration of 10 mM each). Plates were incubated at 37 °C overnight. Each experiment was repeated three times. The relative phage titer (PFU/ml) representing ratios of phage titers in induced and noninduced cultures was calculated and presented as percentage of relative phage titer obtained in control experiments, with DMSO added instead of the analyzed compound.

MTT cell viability assay: 3×10^3 HEK-293 (Human Embryonic Kidney) or HDFa (Human Dermal Fibroblast, adult) cells were passaged in each well of 96-well plate, and allowed to attached overnight. Cells were then treated with DMSO (control cells) or 50, 100 or 200 μ M of tested compound at 37 °C. After 48 h incubation, 25 μ l of MTT solution (4 mg/ml) was added to each well. Following 3 h incubation at 37 °C, formazan crystals, formed in living cells, were dissolved in 100 μ l of DMSO. Absorbance was measured at 570 and 620 nm (reference wavelength) in a Victor³ microplate reader.

 Table 1. Primers used in the real-time PCR assay.

	· · · · · · · · · · · · · · · · · · ·
Primer name	Sequence $(5' \rightarrow 3')$
pF_OxyR	GCAGGTAGCGGGATCACTTT
pR_OxyR	GCACGGCAGATAAACAACCC
pF_KatG	GCTCTGCCTGTTCTGGAGAA
pR_KatG	CACACCAGCCAGCACTATGA
pF_AhpC	GCTGGAGCGTCTTCTTCTTCT
pR_AhpC	TAGTGGTCAGCAACGTCACC
pF_CysD	ATTTGCCCGTTGTAGTTGTGC
pR_CysD	TACTCTTTCCGTGACCGCTTC
pF_SodA	CTGCCAGAATTTGCCAACCTG
pR_SodA	GTACGGTTTTCTTGTCTGCTGG
pF_SoxR	AAACAGCTTTCGTCCCAATGG
pR_SoxR	TACATCCGTCCAGTTCGTCAC
pF_SoxS	GCTGGGAGTGCGATCAAACT
pR_SoxS	GCAATGGACCTGGGTTATGTGT
pF_16SrRNA	CCTTACGACCAGGGCTACAC
pR_16SrRNA	TTATGAGGTCCGCTTGCTCT
pF_Ф24 _B _N	AGGCGTTTCGTGAGTACCTT
pR_ Φ24 _B _N	TTACACCGCCCTACTCTAAGC
pF_Ф24 _B _cl	TGCTGTCTCCTTTCACACGA
pR_Ф24 _B _cl	GCGATGGGTGGCTCAAAATT
pF_Ф24 _B _cll	TGATCGCGCAGAAACTGATTTAC
pR_@24 _B _cll	GACAGCCAATCATCTTTGCCA
pF_@24 _B _Q	GGGAGTGAGGCTTGAGATGG
pR_@24 _B _Q	TACAGAGGTTCTCCCTCCCG

Bacterial RNA preparation: For the extraction of total RNA from *E. coli* strain MG1655 lysogenic for $\Phi 24_B$ bacteriophage, the procedure described in previous subsection was employed with slight modification. Three selected compounds: **CM092**, **CM032D**, and **CM3186** were added to the bacterial culture at time 0 to a final concentration 200 μ M. The control experiment was treated with DMSO solution, which allows estimation of impact of tested compounds on prophage induction. After 3 hours of bacterial cultivation at $37 \,^{\circ}$ C, phage lytic development was provoked by addition of 1 mM H₂O₂. At indicated time after prophage induction, 1×10^9 bacterial cells were treated with 10 mM NaN₃ (Sigma-Aldrich), harvested and frozen in a liquid nitrogen. The RNA samples were prepared using the High Pure RNA Isolation Kit (Roche

Table 2. Differences in OD values resulting from comparison of bacterial growth (*E. coli* strain MG1655 lysogenic for $\Phi 24_{B}$) in control culture (DMSO) and cultures carried out with the analyzed compounds. Each experiment was conducted during 9 h, in the presence or absence of mitomycin C (MITC).

			Effects on the lysogenic strain growth ^a					
		50	μΜ	100 μM		200 μM		
No	Compound	-MITC	$+ \operatorname{MITC}$	-MITC	$+ \operatorname{MITC}$	-MITC	+ MITC	
1	BZ23	-12	9	-12	5	-22	12	
2	BZ26	7	52	-1	48	-2	>100	
3	BZ102	-1	12	-22	2	-25	8	
4	BZ105	-23	-16	-20	-29	30	-44	
5	BZ106	-2	-11	-7	-68	-97	-96	
6	BZ86	1	29	-2	24	-18	15	
7	BZ70	0	41	-3	54	-7	40	
8	BZ83	12	12	7	-2	-7	-18	
9	BZ64	0	26	0	12	-18	11	
10	BZ96	1	10	-1	15	-7	-2	
11	BZ89	-2	8	2	-6	2	-36	
12	IA011C	14	83	18	>100	16	>100	
13	CM092	6	>100	6	>100	-11	>100	
14	CM022G	1	48	4	48	-1	68	
15	CM032D	3	17	6	49	0	65	
16	CM2071F	4	>100	4	>100	-7	>100	
17	CM3186B	2	65	9	72	7	>100	
18	CM3141B	9	23	12	57	8	>100	
19	BZA15	8	>100	17	>100	15	>100	
20	AR02	-1	4	-3	20	-8	41	
21	AR09	1	10	2	17	-9	45	
22	BZA23	4	8	3	7	-5	76	
23	CM3116A	0	18	-1	22	-9	20	
24	CM3159A	4	27	5	26	-1	42	
25	CM3146B	-3	4	-4	11	-14	40	
26	CM3129A	0	21	5	28	-30	3	
27	MF1	2	15	0	41	-1	>100	
28	MF27A	-2	9	-2	25	-6	68	
29	THN10	3	21	-1	32	-2	>100	
30	THN6C	11	20	9	51	16	>100	
31	CM3072B	3	10	3	3	2	76	
32	CM3116C	5	1	3	3	-4	77	
33	MF5	-1	44	-1	97	-7	>100	
34	CM3159B	-1	30	1	68	0	>100	
35	AR27	0	-4	-4	-1	-14	69	
36	MQ4	3	-23	-2	-18	-15	45	
37	MQ8	7	-8	-9	-10	-11	18	
38	BZA37	10	-6	9	10	2	75	
39	CM4017A	0	3	-6	19	-11	38	
40	CM3130B	-6	23	-17	22	-26	29	
41	CM4016A	-1	6	-4	18	-11	58	
42	CM3112B	0	40	-15	1	-26	12	
43	SiA5	3	25	-14	25	-17	29	
44	CM032E	-2	19	-4	13	-12	20	
45	MF4	-2	4	-2	41	-10	91	
46	MF6	-2	1	-14	13	-13	92	

^aPresented results are expressed as percent values above (numbers) or under (numbers with minus sign) the control value which is assumed as 100%. SD was below 20% for each point, and it is not shown for clarity of presentation. Compounds selected for further analyses are underlined.

Applied Science) and TURBO DNA-free[™] Kit (Life Technologies) according to the manufacturer's guidelines. The amount and quality of RNA were evaluated by using NanoDrop spectrophotometer (Eppendorf). In addition, the band patterns of total RNA was visualized by agarose gel electrophoresis. Moreover, the absence of DNA was verified by PCR amplification, and by quantitative real-time reverse transcription-PCR (qRT-PCR).

cDNA synthesis via reverse transcription: Each cDNA was prepared using Transcriptor Reverse Transcriptase and random hexamer primers (Roche Applied Science) following the protocols supplied from the provider. 1.25 μ g of the total RNA sample was used as a template. The resulting cDNA mixture was diluted 10-fold and analyzed in gRT-PCR.

Real-time PCR assay: qRT-PCR was performed with the LightCycler 480 Real-Time PCR System (Roche Applied Science) according to the procedure described previously^{14,37,38}. Reactions were prepared in Roche 96-well plates containing: $10 \,\mu$ I 2 × SYBR Green I Master Mix (Roche Applied Science), 6.25 μ I/ml cDNA, and 200 nM specific primers (Table 1). Real-time PCR amplification were carried out for 55 cycles and cycling conditions were as follows: an initial incubation at 95 °C for 5 min, followed by denaturation at 95 °C for 10 s, annealing at 60 °C for 15 s, and extension at 72 °C for 5 s. No-template controls and a melting curve analysis were performed for each reaction. The 16S rRNA gene was used as reference gene that showed no changes in expression levels in the presence of H₂O₂ and tested compounds. Each experiment was repeated three times.

Real-time PCR analysis: For precise relative quantification of changes in the gene expression, the E-Method with efficiency correction was chosen. This method has been used and described in detail by us previously^{37–39}. The sample before addition of the induction agent (3 h after inoculation) was a calibrator. The final results were calculated using the following formula: Normalized relative ratio $= Et^{CT(t)}$ calibrator -CT(t) sample/ $Er^{CT(r)}$ calibrator -CT(r) sample, where *E* is efficiency, *t* is target and *r* is reference. The raw run data for *E. coli* strain MG1655 and $\Phi 24_B$ genes were transferred to the LC480Conversion software and the efficiency and PCR efficiency was calculated for each gene by LinRegPCR program.

Results

We have analyzed effects of 46 compounds (Supplementary Figures S1–S4) on growth of *E. coli* MG1655 bacteria lysogenic with phage $\Phi 24_B$ (a Shiga-toxin converting phage). Density of bacterial culture was measured at indicated time after prophage induction with $0.2\,\mu$ g/ml mitomycin C in the presence and absence of analyzed compounds. The results are presented in Table 2.

Based on the analysis of growth inhibition of bacterial cultures, 15 compounds were selected to further studies. Most of them were able to increase OD value of the mitomycin C-treated bacterial culture, above 100% in at least one analyzed concentration.

The 15 compounds selected for further biological exploration are presented in Figure 1. Indazole-4,7-dione IA011C was

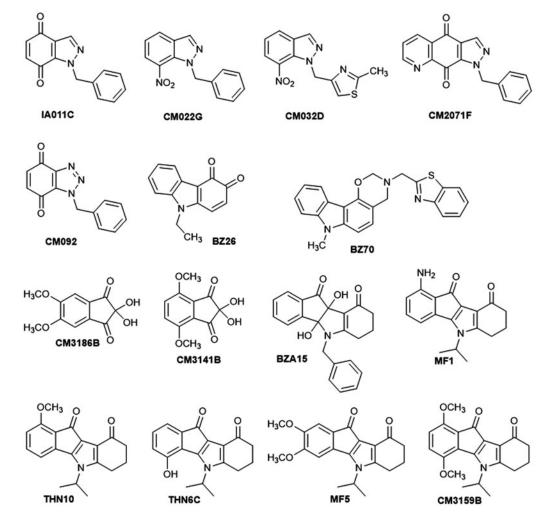


Figure 1. Structures of the selected 15 compounds.

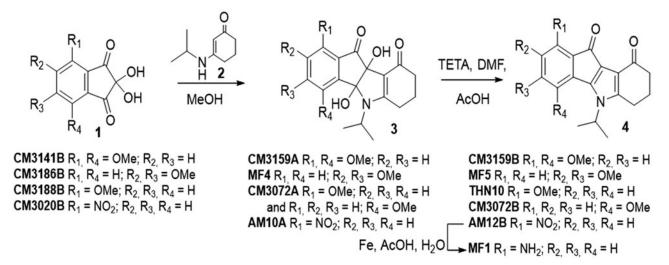


Figure 2. Chemical procedure for the synthesis of targeted indenoindoles.

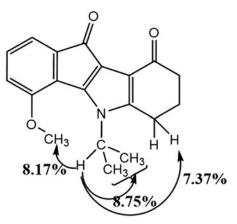


Figure 3. Correlations observed from NOE experiments for CM3072B.

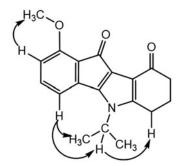
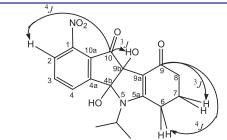


Figure 4. Correlations observed from NOESY experiment for THN10.

synthesized in two steps from CM022G: reduction of the nitro group by hydrogenation in acidic medium followed by oxidation by PIFA⁴⁰. Nitroindazoles CM022G⁴⁰ and CM032D⁴¹ were prepared by N-alkylation of the commercially available 7-nitro-1Hindazole with respectively benzyl bromide or 4-(chloromethyl)-2methylthiazole as alkylating agent. Pyrazolo[4,5-g]quinoline-4,9dione CM2071F was obtained by a Diels-Alder reaction between *N,N*-dimethylhydrazone IA011C⁴². and indazole-4,7-dione Benzotriazole-4,7-dione CM092 was obtained by a 1,3-dipolar cycloaddition of benzyl azide43 on commercial para-benzoquinone⁴⁰. Carbazole-3,4-quinone **BZ26** was obtained in two steps from the commercially available 9-ethylcarbazol-3-amine. The formation of the corresponding diazonium salt at 0°C followed by

Table 3. 2 D ¹H-¹³C HMBC correlations for AM10A (DMSO, 500.13 MHz).



				HMBC [J(C,H)]				
Atom	¹³ C	¹ H	¹ J	² J	³ J	⁴ J		
1	145.1			2-H,	3-H	4-H		
2	124.8	8.04	2-H					
3	136.1	7.99	3-H					
4	128.9	8.28	4-H					
4a	148.8			4-H	3-H, 4b-OH	2-H		
4b	94.6				4-H, 4b-OH, 1'-H,	3-H, 9b-OH		
5a	164.9				7-H,1'-H	8-H		
6	36.7	2.10	6-H	7-H	8-H			
7	21.9	1.84	7-H	6-H, 8-H				
8	24.2	2.41-2.77	8-H	7-H	6-H			
9	188.7				7-H	6-H		
9a	104.5				6-H, 8-H, 9b-OH			
9b	83.3			9b-OH	4b-OH			
10	192.8				9b-OH	2-H		
10a	126.4				2-H, 4-H	3-H		
1′	44.9	4.61	1′-H	CH_3				

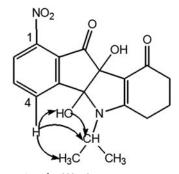


Figure 5. NOESY interactions for AM10A.

heating the latter in highly acidic aqueous solution gave 9-ethylcarbazol-3-ol, which was finally oxidized with Frémy's salt⁴⁴.

Oxazinocarbazole **BZ70** was prepared by a Mannich type condensation of benzo[*d*]thiazol-2-ylmethanamine and formaldehyde with 9-methylcarbazol-4-ol. This latter was obtained by a chemoselective N-alkylation of the commercially available carbazol-4-ol which was achieved by generating the N,O-dianion with NaH in a DMF/THF mixture under argon atmosphere at room temperature and subsequent treatment with methyl iodide⁴⁵. Dimethoxyninhydrins (2,2-dihydroxydimethoxyindane-2,3-diones) **CM3186B** and **CM3141B** were prepared in a single step from the corresponding commercial substituted indan-1-ones by microwave-assisted

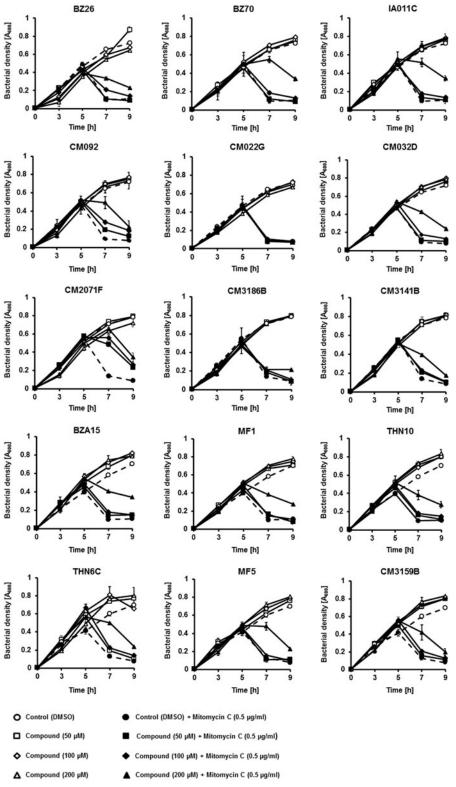


Figure 6. Growth of *E. coli* MG1655 lysogenic with $\Phi 24_B \Delta stx 2::cat$ at 37 °C in LB medium after induction with 0.5 µg/ml mitomycin C (added to the culture at time 3 h) in the absence or presence of tested compounds at indicated concentrations (added to the culture at time 0). Bacterial growth was monitored by measurement of A₆₀₀ at indicated times. Presented results are mean values from three experiments with SD indicated as error bars.

selenium oxidation⁴⁶. Dihydroxyindeno[1,2-*b*]indole-9,10-dione **BZA15**⁴⁷ was prepared by condensation of ninhydrin and 3-(benzylamino)cyclohex-2-en-1-one. 5-Isopropylindeno[1,2-*b*]indole-9,10dione derivatives **MF1**⁴⁸, **THN10**, **THN6C**⁴⁹, **MF5**, and **CM3159B**, were synthesized in two steps according to the method previously described⁴⁷: first, condensation between 3-(isopropylamino)cyclohex-2-en-1-one **2**⁴⁷ and the corresponding substituted ninhydrin, then deoxygenation using tetraethylthionylamide (TETA) (Figure 2).

Use of dimethoxyninhydrins **CM3141B** and **CM3186B** afforded the dihydroxydimethoxyindeno[1,2-*b*]indole-9,10-diones **CM3159A** and **MF4**, respectively, which were easily deoxygenated to give the expected dimethoxyindenoindoles **CM3159B** and **MF5**.

Condensation of enaminone **2** with 4-methoxyninhydrin **CM3188B**⁴⁰ led to a mixture of the two regiosiomers 1- and 4methoxy (ratio 69/31) which could not be separated under classical conditions. After deoxygenation of the mixture, the two regioisomers 1- and 4-methoxyindenoindoles were easily separated by column chromatography and their assignment established by NMR experiments. NOE experiment for the regioisomer **CM3072B** (Figure 3) showed a correlation between the CH of the isopropyl group and the methoxy group.

This correlation was not observed for the other regioisomer, compound **THN10**. NOESY experiment for this latter showed

correlations (Figure 4) between the CH of the isopropyl group and an aromatic proton (H-4), confirming unambiguously the assignment.

Amino derivative **MF1** was prepared from 4-nitroninhydrin **CM3020B**⁴⁰. The first coupling step led to the single 1-nitro regioisomer **AM10A**. Assignment was established by NMR experiments: 2 D NMR HSQC and ¹H-¹³C HMBC correlations, using the characteristic long-range ³J and ⁴J couplings and NOESY. All of the cross couplings observed are reported in the Table 3, and the decisive correlations for the structural assignment are summarized in Figure 5.

The carbonyls C-9 and C-10 were firstly attributed due to the ³J and ⁴J couplings between the protons 6-H and 7-H with C-9 on the one hand, and OH-9 b and 2-H with C-10 on the other hand. Then assignment of the C-4 b and C-9 b was allowed. In fact, only the OH-9 b proton correlated with C-10 (${}^{3}J_{H,C}$). Next, the cross couplings found in the NOESY between the protons 4-H with OH-4 b and the CH of isopropyl group allowed us to identify unambiguously the proposed regiochemistry (Figure 5). This derivative **AM10A** was then dedihydroxylated with TETA to afford **AM12B**. Finally, the nitro group was reduced into amine **MF1** using iron in acidic conditions⁵⁰.

Detailed analysis of effects of the tested 15 compounds on bacterial growth revealed various activities of different substances.

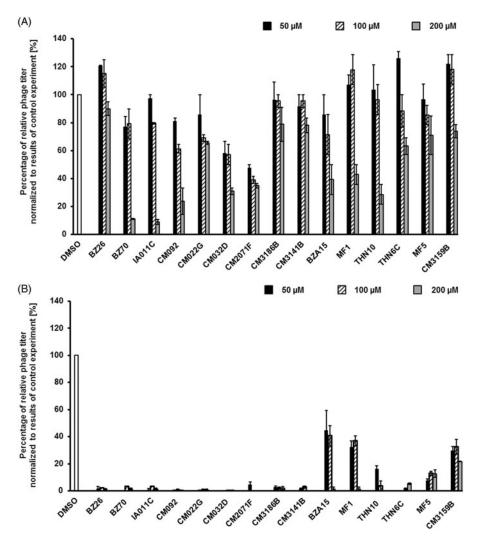


Figure 7. Relative phage titer in cultures of *E. coli* MG1655 lysogenic with $\Phi 24_B \Delta stx2::cat$ treated with $0.5 \mu g/ml$ mitomycin C (A) or 1 mM H_2O_2 (B) (inducers were added to the culture at time 3 h) in the absence (control experiments) or presence of tested compounds at indicted concentrations (added to the culture at time 0). Presented results are mean values from three experiments with SD indicated as error bars.

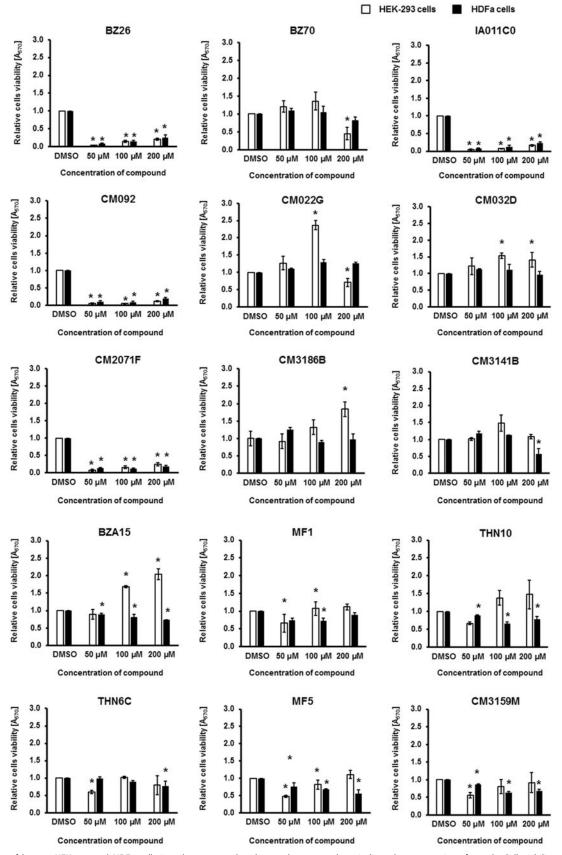


Figure 8. Viability of human HEK-293 and HDFa cells in cultures treated with tested compounds at indicated concentrations for 48 h. Cell viability was tested in the MTT test. Presented results are mean values from three experiments with SD indicated as error bars. The significance of differences between control and cells treated with tested compounds was assessed by the ANOVA test. Differences were marked by asterisks (*) and considered significant when the *p* value was <0.05.

Induction of the lysogenic bacterial culture with mitomycin C caused prophage induction and subsequent lysis of the host cells, seen in Figure 6 as a decrease in the cell culture optical density.

However, majority of the selected compounds (BZ26, BZ70, IA011C, CM092, CM032D, CM2071F, CM3141B, BZA15, MF1, THN10, THN6C, MF5, CM3159B) ameliorated this effect, preventing bacterial cell lysis after treatment with mitomycin C, in at least one used concentration (Figure 6). These results suggested that such compounds might prevent either effective prophage induction or phage lytic development. Hence, we have measured titers of bacteriophages in cultures treated with mitomycin C with and without the presence of tested compounds. In addition, hydrogen peroxide was used as an inducing agent, as this substance was found previously as a prophage inducer occurring naturally in human gut. Contrary to mitomycin C, this inducer did not cause detectable changes in the density of bacterial culture (data not shown). In mitomycin C-treated lysogenic E. coli cell culture, among the 15 selected compounds, the presence of most of them in the culture caused a significant decrease in the titer of bacteriophage (Figure 7(A)). Importantly, in hydrogen peroxide-treated cells, the effects of these compounds on phage titer were dramatic (Figure 7(B)).

These results indicate that in the presence of compounds **BZ26**, **BZ70**, **IA011C**, **CM092**, **CM022G**, **CM032D**, **CM2071F**, **CM3186B**, **CM3141B**, **THN6C**, $\Phi 24_B$ prophage induction and/or phage lytic development is/are impaired. This suggest that at least some of them might be considered as potential anti-STEC drugs.

Since any potential therapeutic must be safe for humans, we have tested the selected 15 compounds for toxicity to human cells. Model HEK-293 and HDFa cells were treated with these compounds for 48 h in cultures, and their viability was assessed in the MTT test (Figure 8).

Compounds **BZ26**, **IA011C**, **CM092**, and **CM2071F** were evidently toxic to human cells which precludes their use as therapeutics (Figure 8). Among the rest, there were substances neutral for human cells' viability or those even stimulating their growth (compounds: **BZ70**, **CM022G**, **CM032D**, **CM3186B**, **CM3141B**, **BZA15**, **MF1**, **THN10**, **THN6C**, **MF5**, and **CM3159B**) (Figure 8).

To learn about molecular mechanisms of impairment of prophage induction and/or phage lytic development, expression of selected phage and host genes was monitored quantitatively by using qRT-PCR. Three compounds belonging to different groups of human cell growth inhibitors (**CM092**), and neutral or weakly stimulating agents (**CM032D**, **CM3186B**), were chosen for further investigations (Figure 9).

In the absence of the prophage inducing agent, the influence of compounds **CM092**, **CM032D**, and **CM3186B** on expression of bacterial genes involved in the oxidative stress response, as well as of bacteriophage genes crucial for either lytic (N, Q) or lysogenic (cl, cll) development was negligible (Figure 9(A)). However, addition of hydrogen peroxide caused activation of the host genes coding for the oxidative stress proteins, and phage genes coding for antitermination proteins (as expected), while having little effects on phage lysogenic proteins (also as expected)

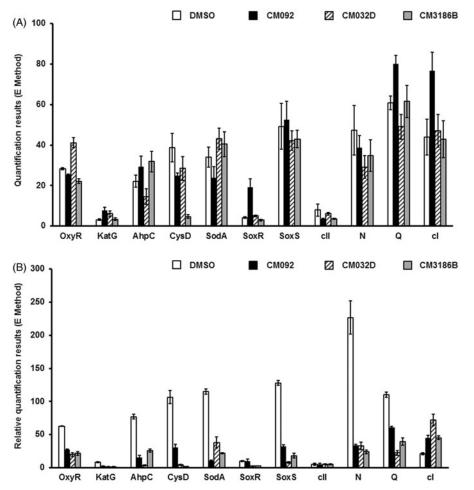


Figure 9. Expression of bacterial genes coding for proteins involved in the oxidative stress response and of selected bacteriophage genes in *E. coli* MG1655 lysogenic with $\Phi 24_B \Delta stx2::cat$ either non-treated (A) or after induction with 1 mM H₂O₂ (B) in the absence (control experiments) or presence of tested compounds added to final concentration of 0.2 mM. Levels of mRNAs were determined by RT-qPCR. Presented results are mean values from three experiments with SD indicated as error bars.

(Figure 9(B)). Importantly, simultaneous presence of hydrogen peroxide and any of the tested compounds caused a significant decrease of the efficiency of expression of the oxidative stress genes as well as phage genes (N, Q) supporting the lytic development (Figure 9(B)). In contrast to these genes, expression of the gene coding for the cl repressor was enhanced under these conditions. These results suggest that compounds **CM092**, **CM032D**, and **CM3186B** impair expression of bacterial genes involved in the oxidative stress response, or ameliorate the effects of oxidative stress. This leads to prevention of the prophage induction by hydrogen peroxide.

Discussion and conclusions

Infections by STEC strains are particularly dangerous due to both production of strong toxins (Shiga toxins) by these pathogens and unavailability of the use of antibiotic therapy because these antibacterial compounds may induce Stx prophage leading to enhanced toxin synthesis and release^{2,3}. Stx prophage induction is required for effective expression of *stx* genes and production of Shiga toxins^{2,17}. Therefore, compounds which interfere with prophage induction might be potential anti-STEC agents. Since it was previously demonstrated that oxidative stress conditions are likely responsible for Stx prophage induction in human intestine infected with STEC (summarized in⁸), we asked whether compounds revealing antioxidant features might be effective in inhibition of Stx prophage induction, thus, diminishing pathogenicity of STEC strains.

We have tested 46 compounds, derivatives of carbazole, indazole, triazole, quinolone, ninhydrine, and indenoindole, and after preliminary experiments, 15 of them were chosen for more detailed analysis. These chemicals prevented or alleviated induction of Stx prophages by hydrogen peroxide, thus, revealing an anti-STEC-pathogenicity potential. Importantly, out of 15 tested compounds, eleven had acceptable profiles in the cytotoxicity MTT test, i.e. they did not cause a decrease in viability of two human cell lines (HEK-293 and HDFa). These might be considered for further studies as potential anti-STEC drugs.

Studies on the mechanism of action of the selected compounds, CM092, CM032D, and CM3186B, indicated that in E. coli lysogenic with an Stx phage they had generally no significant effects on expression of tested host and phage genes under standard growth conditions. However, after addition of hydrogen peroxide to the bacterial culture, all three tested compounds caused drastic decrease in efficiency of expression of phage genes involved in the lytic development and an increase in expression of the cl gene, coding for the phage repressor, responsible for prophage maintenance. These results demonstrate that CM092, CM032D, and CM3186B prevent the prophage induction at the level of expression of specific phage genes. This may arise from interference with the oxidative stress, as in cells treated with hydrogen peroxide, expression of genes involved in the oxidative stress response was significantly less efficient in the presence of the tested compounds. Therefore, the tested compounds apparently reduce the oxidative stress in the presence of hydrogen peroxide, which prevents induction of Stx prophage in E. coli.

In conclusion, several compounds tested in this study, particularly **BZ70**, **CM022G**, **CM032D**, **CM3186B**, **CM3141B**, **BZA15**, **MF1**, **THN10**, **THN6C**, **MF5**, **CM3159B**, might be considered as potential anti-STEC drugs, and qualified for further studies. At least some of them, **CM092**, **CM032D**, and **CM3186B**, act by reduction of the oxidative stress, changes in gene expression and prevention of Stx prophage induction.

Disclosure statement

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References

- Riley LW, Remis RS, Helgerson SD, et al. Hemorrhagic colitis associated with a rare *Escherichia coli* serotype. N Engl J Med 1983;308:681–5.
- Muniesa M, Hammerl JA, Hertwig S, et al. Shiga toxin-producing *Escherichia coli* O104:H4: a new challenge for microbiology. Appl Environ Microbiol 2012;78:4065–73.
- Bloch SK, Felczykowska A, Nejman-Faleńczyk B. Escherichia coli O104:H4 outbreak-have we learnt a lesson from it? Acta Biochim Pol 2012;59:483–8.
- 4. Nataro JP, Kaper JB. Diarrheagenic *Escherichia coli*. Clin Microbiol Rev 1998;11:142–201.
- Mizutani S, Nakazono N, Sugino Y. The so-called chromosomal verotoxin genes are actually carried by defective prophages. DNA Res 1999;6:141–3.
- 6. Schmidt H. Shiga-toxin-converting bacteriophages. Res Microbiol 2001;152:687–95.
- 7. Węgrzyn G, Licznerska K, Węgrzyn A. Phage λ -new insights into regulatory circuits. Adv Virus Res 2012;82:155–78.
- Licznerska K, Nejman-Faleńczyk B, Bloch S, et al. Oxidative stress in Shiga toxin production by enterohemorrhagic *Escherichia coli*. Oxid Med Cell Longev 2016;2016:3578368.
- Imamovic L, Muniesa M. Characterizing RecA-independent induction of Shiga toxin2-encoding phages by EDTA treatment. PLoS One 2012;7:e32393.
- Yamamoto T, Kojio S, Taneike I, et al. 60Co irradiation of Shiga toxin (Stx)-producing *Escherichia coli* induces Stx phage. FEMS Microbiol Lett 2003;222:115–21.
- 11. Aertsen A, Faster D, Michiels CW. Induction of Shiga toxinconverting prophage in *Escherichia coli* by high hydrostatic pressure. Appl Environ Microbiol 2005;71:1155–62.
- Wagner PL, Acheson DWK, Waldor MK. Human neutrophils and their products induce Shiga toxin production by enterohemorrhagic *Escherichia coli*. Infect Immun 2001;69:1934–7.
- Łoś JM, Łoś M, Węgrzyn A, Węgrzyn G. Hydrogen peroxidemediated induction of the Shiga toxin-converting lambdoid prophages ST2-8624 in *Escherichia coli* O157:H7. FEMS Immunol Med Microbiol 2010;58:322–9.
- Licznerska K, Dydecka A, Bloch S, et al. The role of the *exoxis* region in oxidative stress-mediated induction of Shiga toxin-converting prophages. Oxid Med Cell Longev 2016;2016:8453135.

- 15. Aruoma OI, Halliwell B, Gajewski E, Dizdaroglu M. Damage to the bases in DNA induced by hydrogen peroxide and ferric ion chelates. J Biol Chem 1989; 264:20509–12.
- 16. Mauro SA, Koudelka GB. Shiga toxin: expression, distribution, and its role in the environment. Toxins (Basel) 2011;3: 608–25.
- Łoś JM, Łoś M, Węgrzyn G. Bacteriophages carrying Shiga toxin genes: genomic variations, detection and potential treatment of pathogenic bacteria. Future Microbiol 2011;6:909–24.
- Semchyshyn H, Bagnyukova T, Storey K, Lushchak V. Hydrogen peroxide increases the activities of soxRS regulon enzymes and the levels of oxidized proteins and lipids in *Escherichia coli*. Cell Biol Int 2005;9:898–902.
- 19. González-Flecha B, Demple B. Role for the oxyS gene in regulation of intracellular hydrogen peroxide in *Escherichia coli*. J Bacteriol 1999;181:3833–6.
- 20. Vasil'eva SV, Stupakova MV, Lobysheva II, et al. Activation of the *Escherichia coli* SoxRS-regulon by nitric oxide and its physiological donors. Biochemistry Mosc 2001;66:984–8.
- Marshall HE, Merchant K, Stamler JS. Nitrosation and oxidation in the regulation of gene expression. Faseb J 2000;14:1889–900.
- 22. Mukhopadhyay P, Zheng M, Bedzyk LA, et al. Prominent roles of the NorR and Fur regulators in the *Escherichia coli* transcriptional response to reactive nitrogen species. Proc Natl Acad Sci USA 2004;101:745–50.
- 23. Nakano MM. Induction of ResDE-dependent gene expression in *Bacillus subtilis* in response to nitric oxide and nitrosative stress. J Bacteriol 2002;184:1783–7.
- 24. Hausladen A, Privalle CT, Keng T, et al. Nitrosative stress: activation of the transcription factor OxyR. Cell 1996;86:719–29.
- 25. Storz G, Tartaglia LA, Ames BN. Transcriptional regulator of oxidative stress-inducible genes: direct activation by oxidation. Science 1990;248:189–94.
- Kullik I, Toledano MB, Tartaglia LA, Storz G. Mutational analysis of the redox-sensitive transcriptional regulator OxyR: regions important for oxidation and transcriptional activation. J Bacteriol 1995;177:1275–84.
- 27. Kim SO, Merchant K, Nudelman R, et al. OxyR: a molecular code for redox-related signaling. Cell 2002;109:383–96.
- Vasil'eva SV, Stupakova MV, Lobysheva II, et al. Activation of the *Escherichia coli* SoxRS-regulon by nitric oxide and its physiological donors. Biochemistry (Mosc) 2001;66:984–8.
- 29. D'Autreaux B, Touati D, Bersch B, et al. Direct inhibition by nitric oxide of the transcriptional ferric uptake regulation protein via nitrosylation of the iron. Proc Natl Acad Sci USA 2002;99:16619–24.
- 30. Glinkowska M, Łoś JM, Szambowska A, et al. Influence of the *Escherichia coli* oxyR gene function on lambda prophage maintenance. Arch Microbiol 2010;192:673–83.
- 31. Chohan ZH, Shaikh AU, Naseer MM, Supuran CT. In-vitro antibacterial, antifungal and cytotoxic properties of metalbased furanyl derived sulfonamides. J Enzyme Inhib Med Chem 2006;21:771–81.
- 32. Del Prete S, Perfetto R, Rossi M, et al. A one-step procedure for immobilising the thermostable carbonic anhydrase (SspCA) on the surface membrane of *Escherichia coli*. J Enzyme Inhib Med Chem 2017;32:1120–8.
- Perfetto R, Del Prete S, Vullo D, et al. Production and covalent immobilisation of the recombinant bacterial carbonic anhydrase (SspCA) onto magnetic nanoparticles. J Enzyme Inhib Med Chem 2017;32:759–66.

- 34. Jensen KF. The *Escherichia coli* K-12 "wild types" W3110 and MG1655 have an *rph* frameshift mutation that leads to pyrimidine starvation due to low *pyrE* expression levels. J Bacteriol 1993;175:3401–7.
- 35. Allison HE. Immunity profiles of wild-type and recombinant Shiga-like toxin-encoding bacteriophages and characterization of novel double lysogens. Infect Immun 2003;71: 3409–18.
- 36. Łoś JM, Golec P, Węgrzyn G, et al. Simple method for plating *Escherichia coli* bacteriophages forming very small plaques or no plaques under standard conditions. Appl Environ Microbiol 2008;74:5113–20.
- 37. Bloch S, Nejman-Faleńczyk B, Dydecka A, et al. Different expression patterns of genes from the exo-xis region of bacteriophage λ and Shiga toxin-converting bacteriophage $\Phi 24_{\rm B}$ following infection or prophage induction in *Escherichia coli*. PLoS One 2014;9:e108233.
- 38. Nejman-Faleńczyk B, Bloch S, Licznerska K, et al. A small, microRNA-size, ribonucleic acid regulating gene expression and development of Shiga toxin-converting bacteriophage $\Phi 24_{B}$. Sci Rep 2015;5:10080.
- Nowicki D, Bloch S, Nejman-Faleńczyk B, et al. Defects in RNA polyadenylation impair both lysogenization by and lytic development of Shiga toxin-converting bacteriophages. J Gen Virol 2015;96:1957–68.
- 40. Marminon C, Gentili J, Barret R, Nebois P. Synthesis of *N*benzylated indole-, indazole- and benzotriazole-4,7-diones. Tetrahedron 2007;63:735–9.
- 41. Zeinyeh W, Xia H, Lawton P, et al. Synthesis and modulation properties of imidazo[4,5-*b*]pyridin-7-one and indazole-4,7-dione derivatives towards the *Cryptosporidium parvum* CpABC3 transporter. Eur J Med Chem 2010;45:2480–8.
- 42. Marminon C, Fenet B, Mignosi V, et al. Diels-Alder reactions between acrolein N,N-dimethylhydrazone and N-benzylated benzotriazole-, indazole- or indole-4,7-diones. Heterocycles 2009;78:2799–809.
- 43. Saxon E, Luchansky SJ, Hang HC, et al. Investigating cellular metabolism of synthetic azidosugars with the Staudinger ligation. J Am Chem Soc 2002;124:14893–902.
- 44. Compain-Batissou M, Latreche D, Gentili J, et al. Synthesis and Diels-Alder reactivity of *ortho*-carbazolequinones. Chem Pharm Bull (Tokyo) 2004;52:1114–16.
- 45. Issa S, Walchshofer N, Kassab I, et al. Synthesis and antiproliferative activity of oxazinocarbazole and N,N-bis (carbazolylmethyl) amine derivatives. Eur J Med Chem 2010;45:2567–77.
- 46. Marminon C, Nacereddine A, Bouaziz Z, et al. Microwaveassisted oxidation of indan-1-ones into ninhydrines. Tetrahedron Lett 2015;56:1840–2.
- Hemmerling HJ, Reiss G. Partially saturated indeno[1,2b]indole derivatives via deoxygenation of heterocyclic α-hydroxy-N,O-hemiaminals. Synthesis 2009;2009:985–99.
- 48. Perret F, Marminon C, Zeinyeh W, et al. Preparation and characterization of CK2 inhibitor-loaded cyclodextrin nano-particles for drug delivery. Int J Pharm 2013;441:491–8.
- 49. Jabor Gozzi G, Bouaziz Z, Winter E, et al. Converting potent indeno[1,2-*b*]indole inhibitors of protein kinase CK2 into selective inhibitors of the breast cancer resistance protein ABCG2. J Med Chem 2015;58:265–77.
- 50. Alcalde E, Mesquida N, Frigola J, et al. Indene-based scaffolds. Design and synthesis of novel serotonin 5-HT6 receptor ligands. Org Biomol Chem 2008;6:3795–810.