Silicon Application Modulates Growth, Physio-Chemicals, and Antioxidants in Wheat (*Triticum aestivum* L.) Exposed to Different Cadmium Regimes

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Abstract

Silicon (Si) application enhanced the tolerance of plants against different environmental stresses. Therefore, objective of the study revealed that foliar applied Si alleviates the adverse effect of Cd by enhancing the growth, metabolite accumulation, strengthening the antioxidant defense system, reducing oxidative injury, improving plant nutrient status, and decreasing the Cd uptake in wheat. The surface sterilized seeds of Sahar-2006 (tolerant) and Inqalab-91 (sensitive) having the differential metal tolerance capacity were sown in plastic pots containing normal and Cd spiked sandy loamy soil. The design of experiments was completely randomized with 3 replicates per treatment. Two weeks after germination, plants were sprayed with different concentrations of Si (1.5 and 3 mM) with 0.1% surfactant in the form of Tween-20. The plants were harvested after 2 weeks of Si application to determine various attributes. High concentration of Cd (25 mg kg⁻¹) decreased growth-related-attributes, essential nutrient uptake and increase the levels of oxidative stress indicators. The application of Si increased the growth-related attributes, photosynthetic pigments, essential nutrient uptake and also enhanced the activities of various antioxidant compounds (superoxide dismutase (SOD), peroxidase (POD, ascorbate peroxidase (APX) and catalase (CAT) by decreasing the contents of oxidative stress indicators and Cd uptake in root and shoot of both wheat cultivars. Sahar-2006 cultivar showed more tolerance to Cd regimes than that of Inqalab-91 as clear from greater plant dry masses. Thus, our results showed that the applied Si level (3 mM) is an efficient strategy for field use in the areas, where slightly Cd polluted soils limit the agriculture production.

Keywords

silicon, oxidative stress, antioxidant compounds, cadmium, wheat, nutrient uptake

Introduction

Metals stress is the major ecological issue, now a days for agricultural soil and arises from the geological sources such as vast industrialization and some agricultural practices.¹ Moreover, anthropogenic activities such as mine and smelters,² rapid industrialization,³ untreated wastewater irrigation,⁴ atmospheric deposition,⁵ disposal of wastes, gasoline and paints, fertilizers,⁶ bio solids,⁷ and pesticides⁸ are major causes to increase the concentrations of various toxic pollutants, especially Cd in the soil.^{9,10} The Cd is a hazardous trace element because of its higher solubility, mobility and toxicity in the nature.¹¹ And is widely recognized as carcinogenic, mutagenic and toxic metal.¹² Plant roots simply absorbed Cd from the soil and then uptake it to vegetative tissues, enters into the food chain and then transferred to the end consumers leading to

human health problems.^{13,14} Excessive Cd influences the many physiological processes in plants such as changes the membrane permeability,¹⁵ inhibits physiologically active enzymes,¹⁶

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inactivates photosystems,17 and disturbing mineral metabolism.^{18,19} Cadmium toxicity induces oxidative stress, disrupts the pigments function and changes in protein activity.¹⁹ Reacting to Cd stress, hyper generation of reactive oxygen species (ROS) in plants that cause damage to cell structures in plants such as oxidation of proteins and lipids,^{20,21} nucleic acid damages,²² enzyme inhibition and ultimately cell death.^{23,24} However, to protect cells and tissues from injury and dysfunction, plants have evolved a broad range antioxidative defense systems containing peroxidase (POD), ascorbate peroxidase (APX), superoxide dismutase (SOD), and catalase (CAT) contents¹² and nonenzymatic antioxidants such as anthocyanin, proline,²⁵ cysteine,²⁶ ascorbic acid,²⁷ and glutathione²⁴ to scavenge the ROS²⁸ and protect the cellular structures from oxidative damage.^{29,30} Hence, higher accumulation of Cd may produce significant change in morpho-physiological and biochemical attributes that ultimately contribute to low yield and productivity.¹⁹

Silicon (Si) is known as quasi-essential element because of its multiple beneficial role.³¹ Silicon is second most abundant metalloid in the soil³² and found in a form of mono-silicic acid.³³ Application of Si gives freshness to the plants and its foliar spray enhance stress tolerance by changing the various biological and physical processes.³⁴⁻³⁶ Therefore, when Si promotes nutrient absorption by the stressed plant, as found in other species such as banana.³⁷ Silicon is taken up from the soil by plants through roots via aquaporin channels³⁸ and accumulated in plants, mostly depends on environmental factors such as water contents, Si availability and temperature.³⁹ Silicon is a universal player to mitigate the metal toxicity by enhancing the metabolic activities in polluted environments.^{32,36} Silicon through polymerization which act as barriers to symplastic transport of Cd^{34,40} and reduce transport from root to xylem.⁴¹ Silicon nutrients play a vital structural and enzymatic role in physiological processes and in the antioxidant defense system, based on the production of compounds, such as carotenoids.^{42,43} Moreover, it accumulates in cell walls,⁴⁴ increases photosynthetic activity^{45,46} and mineral uptakes⁴⁷ and also activates the antioxidant systems⁴⁸ to mitigate the negative effects of reactive oxygen species (ROS) in plant.49,50

Wheat is used as staple food in various regions of the globe and is extremely sensitive to Cd stress. We used 2 wheat cultivars to determine whether the impact of foliar applied Si differs with cultivars. It was hypothesized that Si application may confer the growth improvement in Cd stressed wheat plants by maintaining optimum levels of Cd. Therefore, this study was conducted to assess (i) the effects of different Si levels on plant growth and biomass of wheat under Cd stress by strengthens the antioxidant capacity, (ii) the silicon application may improve essential mineral uptakes by reducing oxidative injury under Cd stress, (iii) the role of Si application in regulating the main metabolic processes in wheat exposed to Cd stress.

 Table I. Mean Values of Physico-Chemical Properties of Dry Loamy

 Soil.

Values	Soil characteristics	Values
Clay loam 34.1	Carbonate (meq L ⁻¹) Bicarbonate (meq L ⁻¹)	Nil 4.0
67	Total nitratas	105
0.7	(mg kg ⁻¹)	10.5
7.7	Total nitrogen (mg kg ⁻¹)	14.5
0.4-0.6	Phosphates	8.1-10.7
3.53 5.67	Potassium (mg kg ⁻¹)	108-204
	Values Clay Ioam 34.1 6.7 7.7 0.4-0.6 3.53 5.67	ValuesSoil characteristicsClay loam 34.1Carbonate (meq L ⁻¹) Bicarbonate (meq L ⁻¹)6.7Total nitrates (mg kg ⁻¹)7.7Total nitrogen (mg kg ⁻¹)0.4-0.6Phosphates Potassium (mg kg ⁻¹)5.67Soil characteristics

Materials and Methods

Plant Materials and Growing Conditions

The seeds of 2 wheat cultivars having the differential metal tolerance capacity⁵¹ namely, Sahar-2006 and Ingalab-91 were collected from the Ayub Agricultural Research Institute (AARI) Faisalabad, Pakistan. In order to contaminate the dry, sandy loamy soil, the calculated amount of Cd (10 and 25 mg kg⁻¹) was applied to soil collected from Agriculture Field Faisalabad for spiking process to make contaminated soil artificially for 3 months. Cadmium chloride (CdCl₂) was used as source for Cd stress. Prior to sowing, seeds were washed with Cd free water and then surface sterilized with 10% H₂O₂ to minimize the microbial contamination and rinsed with distilled water thoroughly. The plastic pots containing 25-cm diameter and 15-cm depth were filled with normal and Cd spiked sandy loamy soil (5 kg). The physico-chemical properties of dry sandy loamy soil are given in Table 1. In each pot, 10 seeds were sown in each pot and placed in an open greenhouse. All pots were arranged in completely randomized design with 3 replicate per treatment. An average of $19 \pm 3/10 \pm 2^{\circ}$ C was the day/ night temperatures, an average of 62.0% to 65.1% was the relative humidity, and an average of 10 to 11 was the day length. Upon full emergence, thinning was done and only 5 seedlings were kept in each pot. Then half strength Hoagland's nutrient solution made in Cd-free water⁵² was supplied to all plants at every 5 days-interval. Soil pH was adjusted back to control, pH of 6.0 to 6.2 with diluted H₂SO₄. A control group without Cd application was maintained for comparison. Two weeks after germination, plants were sprayed with different concentration of Si (1.5 and 3 mM Si) using 30 mL of Na₂SiO₃ (Merck, pH 11.76) solution to each wheat seedling. An equivalent amount of Na⁺ (as NaCl) was added to the controlled plants to compensate the differences resulting from Na2SiO3 application. The controlled plants were sprayed with distilled water in the same amount. The plants were fertilized with a Hoagland nutrient solution to keep the moisture contents. Plants were harvested after 2 weeks of Si application and fresh leaves were kept in a freezer at -20° C for recording of different growth and biochemical attributes.

Growth Attributes

The uprooting of the plants from the pot soil were carried in such a way that there will be no damage to whole intact plant. The uprooted intact plants were washed with the water to remove all the dust and soil particles before being subjected to phenotypic analysis. The root and shoot tissues are completely segregated and the metric scale was used to estimate the root and shoot lengths. Their fresh weight measured immediately after rooted-up and packed some plant samples in a paper bag (which were labelled) and oven dried for 70°C for 72 h to measure plant dry weight.

Photosynthetic Pigments

Uppermost leaves that are fully expanded were harvested at 9 to 10 am and used for the estimation of chlorophyll (*a*, *b*), total chlorophyll and carotenoid contents using the method described by Lichtenthaler (1987).⁵³ The supernatant was used to find out the absorbance at 645, 663, and 480 nm by using UV-VIS (Hitachi U–2910) spectrophotometer. Carotenoids contents were measured by the method of Davies (1976).⁵⁴ The chlorophyll contents and total carotenoid contents were calculated by using the following formulas.

 $\begin{array}{l} \mbox{Chlorophyll a (mg g^{-1}FW) = $12.7 (OD663 - 2.69 (OD645)$ \\ $$\times (V/1000 \times W)$ \\ \mbox{Chlorophyll b (mg g^{-1}FW) = $12.9 (OD645 - 4.68 (OD663)$ \\ $$\times (V/1000 \times W)$ \\ \mbox{Total chlorophyll (mg g^{-1}FW) = $20.2 (OD645 - 8.02 (OD663)$ \\ $$\times (V/1000 \times W)$ \\ \mbox{Carotenoids (mg g^{-1}FW) = $OD480 + (0.114 \times OD663)$ \\ $$- (0.638 \times OD645)$ \\ \end{array}$

Hydrogen Peroxide (H₂O₂) Contents

For determination of H_2O_2 , 0.15 g fresh leaves were homogenized in 0.1% TCA and then centrifuged for 15 min at 12000 × g, the supernatant mixed with 10 mM potassium phosphate buffer (pH 7.0) and 1 mL potassium iodide (1 M). H_2O_2 utilized to establish a standard curve. Absorbance at 390 nm were used to estimate the $H_2O_2^{55}$ by using the UV-VIS spectrophotometer (Hitachi U–2910).

Malondialdehyde (MDA) Contents

The content of the MDA was assessed by using the TBA method as described by Hodge et al (1999).⁵⁶ For analysis, fresh leaves (0.15 g) of wheat were grinded with 5.0 mL of 5% (w/v) TCA in mortar placed on ice bath, centrifuged, and the MDA content was measured at 532 and 600 nm by using UV-VIS (Hitachi U–2910) spectrophotometrically.

Electrolyte Leakage

Electrolyte leakage (EL) was assayed according to the method of Dionisio-Sese and Tobita (1998).⁵⁷ Fresh leaves of wheat seedlings (0.5 g) were cut into uniform sized leaf discs, and put in a test tube with 10 ml deionized water. Leaf discs are incubated in distilled water, and the electrical conductivity (EC₁) was measured before and (EC₂) after boiling for taking the respective electrical conductivities. The electrolyte leakage was estimated by using the following formula

$$EL = EC_1/EC_2 \times 100$$

Total Soluble Protein Contents

Total soluble protein content in fresh leaf materials (0.1 g) of wheat was estimated by using the 2 mL of phosphate buffer saline (pH 7.2) as described by Bradford $(1976)^{58}$ and then homogenized material was centrifuged for 10 minutes at 12000 g. The sample was incubated at room temperature for 30 minutes and measured the absorbance at 595 nm using UV-VIS spectrophotometer (Hitachi U-2910, Tokyo, Japan) and bovine serum albumin (BSA) was used as standard.

Anthocyanin Contents

Total anthocyanin content was measured using the procedure of Hodges and Nozzolillo (1996).⁵⁹ Fresh leaves (0.15 g) of wheat were grinded in in 2 mL of acidified methanol (methanol+1% HCI) by using the mortar and pestle. The grinded material was moved to the test tube and boiled at 100°C in the water bath for 30 min and then homogenate was centrifuged at $12000 \times g$ for 15 min and absorbance was read at 540 and 600 nm on a spectrophotometer.

Antioxidant Enzymes Assay

The fresh leaves of wheat were grinded in phosphate buffer (50 mM; pH 7.0) and 1 mM dithiothreitol. Then centrifuged the material at 15,000 rpm for 15 min and was stored at 4°C until being utilized to assay superoxide dismutase (SOD), peroxidase (POD), catalase (CAT) and ascorbate peroxidase (APX) activities.⁶⁰ Superoxide dismutase (SOD) activity of fresh leaves of wheat was estimated with NBT (Nitro blue tetrazolium) method to assay the activity of SOD.⁶¹ Briefly, 20 mL of enzyme extract was mixed with a reaction mixture containing 50 µM phosphate buffer (pH 7.8), 50 µM NBT, 1.3 µM riboflavin, 75 µM EDTA and methionine. A light source was used to expose the reaction mixture in the test tube and then absorbance was measured at 560 nm using a spectrophotometer (Hitachi U-1800). Accordingly, one Unit SOD activity was equal to the amount of enzyme required to inhibit 50% reduction of NBT as observed at 560 nm. Chance & Maehly method (1995)⁶² was followed for the estimation of CAT and POD activities. Accordingly, the CAT solution mixture (3 mL) was composed of enzyme extract (0.1 mL), phosphate buffer (50 mM; pH 7.0), and H_2O_2 (5.9 mM). The CAT reaction

mixture was irradiate and decrease in absorbance read at 240 nm as an outcome of H₂O₂ consumption after a 20 s interval indicates the activity of the enzyme. Likewise, POD solution mixture (3 mL) contained 50 mM phosphate buffer (pH 5.0), 0.1 mL enzyme extract, 10 mM H₂O₂ and 20 mM guaiacol. Peroxidase solution mixture was irradiated and absorbance after a 20 s interval was measured at 470 nm using a spectrophotometer (Hitachi U-1800). Per-minute change in absorbance of 0.01 Units was equal to one Unit of each CAT and POD activities. The guaiacol oxidation rate in H₂O₂ presence read at 470 nm indicates the enzyme activity. The method of Nakano and Asada (1981)⁶³ was used for the assay of APX activity. The reaction mixture of APX contained 1 mL of 50 mM phosphate buffer (pH 7.5), 0.5 mM ascorbic acid, 0.1 mM H_2O_2 and 200 µL enzyme extract. The absorbance of the reaction mixture of APX was measured spectrophotometerically (Hitachi U-1800) at 290 nm after every 20 s for 2 min and EU mg⁻¹ protein expresses the APX activity. One unit of APX is the quantity of protein used to break down 1.0 µmol of substrate per min at 25°C.

Mineral Uptake and Cd Content Accumulation in Root and Shoot

Plant root and shoot dry sample (0.1 g) was added to a test tube having HNO₃ and H₂SO₄ (3:1) and placed it on a hot plate for 2 to 3 hours. The temperature of hot plate was progressively increased to 200°C. The mixture was turned black in test tubes. HClO₄ (0.5 mL) was added to the test tubes, and again heated until the solution become discoloured. Then removed the test tubes from the hot plate for cooling. Then diluted the solution with distilled water up to 50 mL. Nutrients with the procedure of Wolf (1982)⁶⁴ were measured from the solution. Magnesium (Mg²⁺), zinc (Zn²⁺), iron (Fe²⁺) and Cd²⁺ contents in the root and shoot of wheat were determined by using atomic absorption spectrophotometer (Hitachi, Z-2000, Tokyo, Japan).

Silicon Content in Root and Shoot

The plant samples (shoot and root) were dried in an oven at 70°C for 2 to 3 h and then powdered by grinding prior to analysis. The dry sample (0.1 g) was digested microwave digestion system containing a mixture of 2 mL of 30% H₂O₂, 1 mL of 40% hydrofluoric acid and 7 mL HNO₃ (70 percent) for 10 minutes.65 The digested samples were diluted with 4% boric acid up to 50 mL. The Si contents were measured in the digested solution as described below: Digested aliquot (0.5 mL) was shifted to a plastic centrifuge tube, and then added 0.5 mL ammonium molybdate (10%), 3.75 mL of HCl (0.2 N), and amino naphtholsulphonic acid (0.5 mL) and 0.5 mL of tartaric acid (20%) was added and then made the volume with distilled water up to 12.5 ml. Optical density was recorded at 390 nm with a UV visible spectrophotometer after1 h, Standards of Si (0, 0.2, 0.4, 0.8 and 1.2 ppm) were prepared by using Merck Certipur[®] Si standard solution (1000 mgL⁻¹) and then measured according to method of Ma et al (2002).⁶⁶

Statistical Analysis

The experiment was conducted in completely randomized design (CRD) with 3 replicates per treatment. All variables studied in this experiment was analyzed by means of 3-way ANOVA using STATISTIX software (version 8.1). Least significant difference (LSD) test was used for the comparison of the different mean values (control and treatment values), and the significance was estimated at the $P \leq 0.05$.

Results

Growth Attributes

Cadmium stress caused a reduction ($P \le 0.001$) in root length, shoot length, root fresh, shoot fresh, and their dry weights (Table 2). Both Cd treatments (10 and 25 mg kg⁻¹) imposed a remarkable reduction in these growth attributes. Root length (19%, 49% and 13%, 46%), shoot length (23%, 45% and 32%, 61%), shoot fresh weight (12%, 39% and 23%, 49%), shoot dry weight (13%, 40% and 25%, 57%), root fresh weight (15%, 38% and 24%, 59%) and root dry weight (25%, 51% and 29%, 59%) was reduced under Cd stress (10 and 25 mg kg⁻¹) in both Sahar-2006 and Ingalab-91 cultivars, respectively. However, less reduction was observed in Sahar-2006 cultivar as compared to Ingalab-91 cultivar. At the highest level of Cd (25 mg/kg), there was a consistent decline of all the growth attributes in Ingalab-91 cultivar. After Si application (1.5 and 3 mM) without Cd stress, the root (6% and 25%) and shoot lengths (6%), shoot fresh (12% and 7%) and dry weights (11%), and root fresh (8% and 4%) and dry weights (13% and 26%) increased in Sahar-2006 and Ingalab-91, respectively; Si addition to Cd showed a remarkable impact on these variables. For example, higher shoot (25%, 45% and 31%, 57%) and root lengths (18%, 39% and 22%, 51%), shoot fresh (14%, 42% and 22%, 51%) and dry weights (13%, 40% and 24%, 58%) was seen in plants with Si application (1.5 mM) in both Sahar-2006 and Ingalab-91 cultivars, respectively under Cd stress (10 and 25 mg kg⁻¹), whereas values of these parameters greatly enhanced at higher level of Si-application (3 mM) in both cultivars under Cd stress. Sahar-2006 cultivar shows better performance in all growth attributes as compared to Inqalab-91, which has been less performed and recorded as sensitive one (Table 3).

Photosynthetic Pigments

Cadmium toxicity significantly ($P \le 0.001$) reduced the chlorophyll molecules and non-significant ($P \ge 0.05$) carotenoids in both wheat cultivars (Table 2). The decline in photosynthetic pigments was significantly in wheat plants exposed to Cd stress (10 and 25 mg kg⁻¹). Cadmium (10 mg kg⁻¹) reduced the chlorophyll (a, b), total chlorophyll and carotenoid contents in both Sahar-2006 (22, 14, 19 and 19%) and Inqalab-91 (32%, 34%, 33% and 23%) cultivars, respectively, but higher reduction was observed at 25 mg kg⁻¹ Cd in both cultivars. The decline in photosynthetic pigment was greater in Inqalab-91 than that in

Source of Variance	Cultivars (CV), $df = 1$	Cadmium (Cd), df = 2	Silicon (Si), df = 2	$CV \times Cd$, $df = 2$	$CV \times Si$, df = 2	$Cd \times Si$, $df = 4$ (CV $ imes$ Cd $ imes$ Si, df = $ au$	Fron, df = 36
Root length	62.875***	36.622***	3.922***	0.430***	0.281***	0.081**	0.159***	0.0299
Shoot length	49.832***	110.262***	2.399***	0.223*	0.120ns	0.032ns	0.042ns	0.068
Shoot fresh weight	18.963***	31.062***	2.657***	0.054ns	0.146ns	0.113ns	0.038ns	0.0622
Shoot dry weight	I.356***	I.439***	0.128***	0.005*	0.009**	0.004*	0.0007ns	0.00165
Root fresh weight	3.190***	4.461***	0.481***	0.040**	0.004ns	0.018ns	0.003ns	0.00895
Root dry weight	0.199***	0.355***	0.038***	0.000 lns	0.007**	0.0002ns	0.00004ns	0.00140
Chlorophyll a	0.6350***	I.0485***	0.0288***	0.0091**	0.00008ns	0.00068ns	0.00042ns	0.00113
Chlorophyll b	0.9180***	0.6608***	0.0233***	0.01701***	0.00057ns	0.00064ns	0.00019ns	0.00113
Total chlorophyll	3.0816***	3.3665***	0.1036***	0.0472***	0.00103ns	0.00224ns	0.00030ns	0.00127
Total carotenoid	0.0126***	0.0089***	$2.72 \times 10^{-4***}$	$1.73 imes 10^{-5}$ ns	4.31×10^{-6} ns	$3.78 imes 10^{-6}$ ns	3.37×10^{-6} ns	$8.56 imes 10^{-6}$
Hydrogen peroxide	245.08***	95.595***	10.050***	5.845***	I.635***	0.398**	0.111ns	0.128
Malondialdehyde	41.724***	21.929***	0.849***	I.273***	0.1599***	0.042**	0.0325ns	0.0237
Electrolyte leakage	3855.18**	2681.42***	197.67***	I 45.23***	9.62*	I.I6ns	3.12ns	3.28
TSP	6.738***	3.362***	0.357***	0.085***	0.0103ns	0.0032ns	0.00146ns	0.00490
Anthocyanin	I.689***	2.325***	0.268***	0.014**	0.013**	0.006*	0.004ns	0.00277
SOD	484.69***	322.47***	119.47***	2. I 35ns	0.294ns	3.999*	0.389ns	1.513
POD	5162.5***	5271.11***	I 498.09***	25.66ns	15.16ns	8.39ns	10.45ns	33.25
CAT	121793***	37923***	9378***	892***	37ns	8 Ins	I 9ns	79
APX	43896***	25669.7***	6910.3***	404.8***	54.5ns	41.3ns	89.8*	39.1
Root Fe ²⁺	42629***	31543.5***	6487.1***	478.9***	1875.6***	34.6ns	20.8ns	56.9
Shoot Fe ²⁺	20193***	11275.2***	2066.2***	67.2ns	46.8ns	41.6ns	5.8ns	33.7
Root Zn ²⁺	12138**	2681***	333.I***	232.2***	9.6ns	2.5ns	l 4ns	12.6
Shoot Zn ²⁺	1262.4***	497.31***	34.30***	 4.64 ***	0.20ns	0.37ns	0.17ns	0.88
Root Mg ²⁺	116361***	25978***	5828***	770***	I Ons	154**	125**	39
Shoot Mg ²⁺	51746***	25126.2***	3317.7***	109.4**	131.6**	49.7ns	10.0ns	27.6
Root Cd ²⁺	12.033***	I 52.368***	II.655***	2.317***	0.034ns	2.150***	0.034ns	0.048
Shoot Cd ²⁺	2.8921***	17.764***	I.I299***	0.591***	0.036***	0.1734***	0.0218***	0.0050
Root Si	0.6828***	0.9447***	0.7219***	0.0788***	0.0155***	0.0417***	0.0047**	0.00146
Shoot Si	0.2790***	0.2440***	0.0221***	0.0085***	0.0032***	0.0004ns	0.0005ns	0.00051
Abbreviations: TSP, tot:	al soluble protein; SOD, su	uperoxide dismutase; POD,	peroxidase; CAT, catal	ase; APX, ascorbate	peroxidase; Fe ²⁺ , irc	on ion; Zn ²⁺ , zinc ion	; Mg ²⁺ , magnesium ion;	Cd ²⁺ , cadmium; Si,

Table 2. Mean Sum of Squares From ANOVA of Data for Growth and Physio-Chemical Attributes of Wheat Cultivars Treated With Si Under Cadmium Stress.

ı silicon. ***, **, * = significant at 0.001, 0.01 and 0.05 levels, respectively.

Cultivars	Cd Stress (mg/kg)	Silicon (Si) (mM)	Root length (cm)	Shoot length (cm)	Root fresh weight (g)	Root dry weight (g)	Shoot fresh weight (g)	Shoot dry weight (g)
Sahar-2006	0	0	7.2 ^c ± 0.057	10.00 ^c ± 0.100	$2.03^{cd} \pm 0.09$	0.54 ^c ± 0.031	5.66 ^{cd} ± 0.24	1.171 ^{cd} ± 0.004
		1.5	$7.63^{b} \pm 0.066$	$10.60^{b} \pm 0.06$	$2.20^{b} \pm 0.057$	0.61 ^b ± 0.032	6.33 ^b \pm 0.20	I.30 ^b ± 0.028
		3.0	$8.23^{a} \pm 0.033$	$11.23^{a} \pm 0.145$	$2.50^{a} \pm 0.115$	$0.68^{a} \pm 0.016$	$7.00^{a} \pm 0.252$	$1.45^{a} \pm 0.029$
	10	0	5.84 ^f ± 0.024	$7.66^{h} \pm 0.088$	$1.73^{ m fg}$ \pm 0.03	$0.40^{ ext{ef}}$ \pm 0.04	$5.00^{ ext{ef}} \pm 0.12$	1.02 ^g ± 0.044
		1.5	$6.27^{e} \pm 0.126$	$7.97^{gh} \pm 0.120$	$1.93^{de} \pm 0.03$	$0.47^{d} \pm 0.006$	$5.47^{cd} \pm 0.12$	1.13 ^{de} ± 0.044
		3.0	$6.75^{d} \pm 0.083$	$8.33^{fg} \pm 0.328$	$2.12^{bc} \pm 0.07$	$0.54^{c} \pm 0.022$	$5.85^{c} \pm 0.08$	1.23 ^{bc} ± 0.033
	25	0	3.69 ^k ± 0.078	5.53 ^k ± 0.152	I.26 ^j ± 0.025	0.26 ^{hi} ± 0.04	3.44 ^k ± 0.25	0.70 ^{kl} ± 0.015
		1.5	4.63 ^h ± 0.328	5.84 ^{jk} ± 0.217	1.31 ^{ij} ± 0.02	$0.33^{g} \pm 0.005$	3.70 ^{jk} ± 0.199	0.77 ^{jk} ± 0.026
		3.0	5.30 ^g ± 0.058	6.28 ⁱ ± 0.217	1.47 ^{hi} ± 0.02	0.38 ^{fg} ± 0.005	4.08 ^{ij} ± 0.06	0.84 ⁱ ± 0.034
Inqalab-91	0	0	4.57 ^h ± 0.053	8.58 ^{ef} ± 0.251	1.69 ^{fg} ± 0.06	$0.46^{de} \pm 0.03$	4.83 ^{fg} ± 0.04	0.94 ^h ± 0.012
		1.5	5.34 ^g ± 0.062	8.93 ^{de} ± 0.123	1.83 ^{ef} ± 0.06	$0.49^{cd} \pm 0.004$	$5.33^{de} \pm 0.03$	1.05 ^{fg} ± 0.005
		3.0	5.67 ^f ± 0.070	9.13 ^d ± 0.017	$2.06^{bcd} \pm 0.05$	$0.51^{cd} \pm 0.031$	$5.50^{cd} \pm 0.06$	1.09 ^{ef} ± 0.005
	10	0	3.98 ^{jk} ± 0.044	5.82 ^{jk} ± 0.197	1.29 ^j ± 0.035	$0.32^{gh} \pm 0.017$	3.71 ^{jk} ± 0.056	0.71 ^{kl} ± 0.012
		1.5	4.179 ^{ij} ± 0.05	6.13 ^{ij} ± 0.007	1.37 ^{ij} ± 0.07	0.35 ^{fg} ± 0.007	4.16 ^{hi} ± 0.05	0.80 ^{ij} ± 0.005
		3.0	4.41 ^{hi} ± 0.036	6.39 ⁱ ± 0.034	1.59 ^{gh} ± 0.03	0.38 ^{fg} ± 0.009	4.50 ^{gh} ± 0.05	0.85 ^{ij} ± 0.004
	25	0	$2.43^{m} \pm 0.048$	$3.37^{m} \pm 0.070$	0.69 ¹ ± 0.043	0.19 ^j ± 0.005	2.46 ¹ ± 0.231	0.40 ⁿ ± 0.022
		1.5	2.59 ^m ± 0.084	$3.80^{lm} \pm 0.026$	0.77 ^{kl} ± 0.020	0.21 ^{ij} ± 0.010	2.59 ^I ± 0.036	0.44 ^{mn} ± 0.015
		3.0	2.95 ¹ ± 0.055	3.99 ¹ ± 0.026	$0.89^{k} \pm 0.016$	0.23 ^{ij} ± 0.009	2.77 ^I ± 0.063	0.48 ^m ± 0.007

Table 3. Effect of Si Spray on Growth Attributes of 2 Wheat Cultivars Under Cd Regimes.

Means are followed by standard errors. Three-way ANOVA was performed and different mean values (control and treatment values) were tested by least significant difference (LSD) at $P \le 0.05$ and different letters following means within the same column indicate the significant differences between the variables in the table.

Table 4. Effect of Si Spray on Photosynthetic Pigments and Biochemical Attributes of 2 Wheat Cultivars Under Cd Regimes.

Cultivars	Cd stress (mg/kg)	Silicon (Si) (mM)	Chl. a (mg g ¹ FW)	Chl. <i>b</i> (mg g ¹ FW)	Total Chl. (mg g ⁻¹ FW)	Total Car. (mg g ⁻¹ FW)	TSP (mg g ⁻¹ FW)	Anthocyanin (Unit g ⁻¹ FW)
Sahar-2006	0	0	1.128 ^b ± 0.040	0.916 ^b ± 0.027	$2.04^{c} \pm 0.039$	$0.120^{b} \pm 0.003$	1.93 ^b ± 0.042	1.50 ^c ± 0.026
		1.5	$1.185^{\circ} \pm 0.007$	$0.969^{\circ} \pm 0.019$	$2.15^{\circ} \pm 0.017$	$0.123^{\circ} \pm 0.001$	$2.18^{\circ} \pm 0.006$	$1./1^{\circ} \pm 0.029$
		3.0	$1.219^{\circ} \pm 0.005$	$1.026^{\circ} \pm 0.005$	$2.24^{\circ} \pm 0.004$	$0.129^{a} \pm 0.001$	$2.29^{\circ} \pm 0.039$	$1.88^{\circ} \pm 0.069$
	10	0	$0.879^{r} \pm 0.019$	$0.784^{der} \pm 0.005$	$1.66^{n} \pm 0.017$	0.097 ^c ± 0.002	1.62ª ± 0.043	1.15 [°] <u>+</u> 0.019
		1.5	$0.918^{et} \pm 0.012$	$0.813^{cde} \pm 0.003$	1.73 ^{tg} ± 0.011	$0.100^{bc} \pm 0.001$	1.78 ^c ± 0.033	1.28 ^{de} ± 0.023
		3.0	0.972 ^{de} ± 0.005	0.846 ^c ± 0.005	1.82 ^e ± 0.005	0.106 ^b ± 0.002	I.96 ^b ± 0.045	1.46 ^c ± 0.071
	25	0	0.687 ^{hi} ± 0.039	0.608 ^h ± 0.020	1.29 ^{jk} ± 0.019	0.074 ^{fgh} ± 0.001	1.02 ^j ± 0.062	0.85 ^k ± 0.008
		1.5	$0.717^{gh} \pm 0.006$	$0.632^{gh} \pm 0.006$	1.35 ^j ± 0.099	0.077 ^{fg} ± 0.001	1.16 ^{hi} ± 0.048	0.94 ^{ij} ± 0.014
		3.0	0.756 ^g ± 0.010	0.684 ^g ± 0.027	1.44 ⁱ ± 0.031	$0.082^{ m ef}$ \pm 0.001	$1.30^{fg} \pm 0.030$	1.06 ^{gh} ± 0.02
Ingalab-91	0	0	$0.947^{e} \pm 0.013$	0.746 ^f ± 0.042	1.69 ^{gh} ± 0.031	$0.087^{de} \pm 0.004$	1.21 ^{gh} ± 0.027	1.20 ^{ef} ± 0.006
·		1.5	$1.011^{cd} \pm 0.003$	$0.769^{ef} \pm 0.023$	1.78 ^{ef} ± 0.020	$0.091^{cd} \pm 0.002$	1.37 ^{ef} ± 0.023	$1.30^{d} \pm 0.023$
		3.0	1.055° ± 0.010	$0.826^{cd} \pm 0.013$	$1.88^{d} \pm 0.006$	0.097 ^c ± 0.002	I.48 ^e ± 0.076	1.36 ^d ± 0.033
	10	0	0.640 ⁱ ± 0.022	0.490 ^j ± 0.033	1.13 ¹ ± 0.011	$0.067^{h} \pm 0.002$	$0.85^{k} \pm 0.063$	$0.86^{jk} \pm 0.013$
		1.5	$0.650^{i} \pm 0.022$	0.514 ^{ij} ± 0.022	$1.16^{1} \pm 0.018$	0.069 ^{gh} ± 0.001	1.02 ^j ± 0.005	0.99 ^{hi} ± 0.019
		3.0	$0.688^{hi} + 0.007$	$0.550^{i} + 0.014$	$1.24^{k} + 0.010$	0.073 ^{gh} + 0.001	$1.09^{ij} + 0.007$	1.14 ^{fg} + 0.027
	25	0	$0.469^{k} + 0.029$	$0.325^{i} + 0.004$	$0.79^{n} + 0.025$	$0.047^{i} + 0.0005$	$0.51^{m} + 0.041$	$0.53^{m} + 0.014$
		1.5	$0.510^{jk} + 0.016$	$0.342^{k} + 0.010$	$0.85^{mn} + 0.027$	$0.049^{i} + 0.001$	$0.63^{1} + 0.019$	$0.60^{\text{lm}} + 0.008$
		3.0	$0.540^{j} \pm 0.021$	$0.366^{k} \pm 0.026$	$0.91^{m} \pm 0.023$	$0.052^{i} \pm 0.001$	$0.69^{I} \pm 0.029$	$0.67^{1} \pm 0.018$

Abbreviations: Chl., chlorophyll; Car., carotenoids; TSP, total soluble protein.

Means are followed by standard errors. Three-way ANOVA was performed and different mean values (control and treatment values) were tested by least significant difference (LSD) at $P \le 0.05$ and different letters following means within the same column indicate the significant differences between the variables in the table.

the Sahar-2006 cultivar. Exogenous application of Si (1.5 and 3 mM) in control plants had a significant rise in chlorophyll *a* (5%, 8%, and 4%, 10%), chlorophyll *b* (6%, 12%, and 4%, 12%), total chlorophyll contents (5%, 10%, and 4%, 11%) and total carotenoid contents (2%, 7%, and 4%, 11%) in both

Sahar-2006 and Inqalab-91 cultivars, respectively. A Si application under different Cd regimes, significantly increased the chlorophyll (a, b), total chlorophyll and total carotenoid content in the range of 16% to 56% in both wheat cultivars. Generally, the Si (3 mM) induced increases were better in Sahar-2006 than Inqalab-91 cultivars, particularly under stress conditions. However, cultivars difference was also evident; Sahar-2006 cultivar accumulated more photosynthetic pigments as compared Inqalab-91 (Table 4).

Total Soluble Protein (TSP) and Anthocyanin Contents

Compared with the control group, Cd treatments (10 and 25 mg kg⁻¹) significantly ($P \le 0.001$) decreased the TSP and anthocyanin contents in both Sahar-2006 (16%-47% and 23%-43%) and Inqalab-91 (29%-58% and 28%-55%) cultivars. Besides, the enhancement in TSP (14%-47% and 25%-54%) and anthocyanin contents (23%-45% and 17%-54%) were triggered in leaves of Sahar-2006 and Inqalab-91 cultivars, respectively, with Si (1.5 and 3 mM) plus Cd stress (10 and 25 mg kg⁻¹) as compared to the stress treatments alone. Similarly, higher significant increase in TSP and anthocyanin was identified only in wheat plants with Si (3 mM) in both Sahar-2006 (18% and 26%) and Inqalab-91 (27% and 24%) cultivars (Table 4).

Oxidative Stress Attributes

Hydrogen peroxide, MDA and EL are key markers in assessing the degree of oxidative damage in the plants. We have documented, highly significant ($P \le 0.001$) increase in cellular levels of H₂O₂, MDA and EL in plants exposed to Cd toxicity (Table 2). MDA contents were increased by 31% and 74% in Sahar-2006 and 46% and 77% in Ingalab-91 cultivar under Cd stress (10 and 25 mg kg⁻¹), respectively, as compared to their corresponding controls. After Si application (1.5 and 3 mM) without Cd stress, MDA contents (3% and 6%) in Sahar-2006 and Inqalab-91 (1% and 6%), respectively. In different Cd regimes, Si (1.5 and 3 mM) considerably reduced the MDA contents in both cultivars. Sahar-2006 (22%-77%) cultivar was more responsive to foliar applied Si (3 mM) than Inqalab-91 (43%-73%) in terms of MDA in response to Cd stress (Figure 1). Si application (1.5 and 3 mM) in controlled environment, reduced the H₂O₂ contents in Sahar-2006 (6% and 9%) cultivar as well as Ingalab-91 (2% and 3%), respectively. Cadmium stress caused an increased in H₂O₂ and EL in both cultivars. Thus, Inqalab-91 cultivar (41% and 53%) showed significantly more H₂O₂ contents than Sahar-2006 (31% and 48%) under Cd stress (10 and 25 mg kg⁻¹), respectively. Similarly, EL significantly increased (27%-80%) under both levels of Cd-stress in both cultivars, but more permeability was observed in both cultivars under Cd stress (25 mg kg⁻¹). Silicon significantly decreased the H₂O₂ and EL in plants of both Sahar-2006 and Inqalab-91cultivars. The response of both cultivars to interactive effect of Si and Cd stress was almost same for H₂O₂ and EL and the effect of Si (1.5 and 3 mM) in Sahar-2006 (20%-59% and 28%-94\%) cultivar was greater than that of the Inqalab-91 (37%-56% and 50%-93%) cultivar. Of the 2 wheat cultivars, elevated levels of H2O2 and EL were found in Inqalab-91 being a sensitive cultivar under Cd (25 mg kg⁻¹) stress (Figure 1).



Figure 1. Effect of Si spray on MDA contents, H_2O_2 contents and EL in the shoot of 2 wheat cultivars seedlings grown under different levels of Cd stress (n = 3 \pm SE). Three-way ANOVA was performed and mean values (control and treatment values) were tested by least significant difference (LSD) at $P \leq 0.05$ and different lowercase letters indicate significant difference between the treatments.

Antioxidant Enzyme Activity

A marked increase in the activities of SOD, POD, CAT and APX was measured in 2 wheat cultivars subjected to Cd stress (Table 2). Cadmium toxicity (10 and 25 mg kg⁻¹) increased the SOD, POD, CAT and APX contents in both Sahar-2006 (19%-60%, 20%-59%, 31%-56%, and 14%-40%) and Inqalab-91 (28%-81%, 30%-86%, 33%-72%, and 46%-84%) cultivars, respectively as compared to their corresponding controls. However, Si-application at both levels (1.5 and 3 mM) further increased the antioxidant activities to a significant level in both cultivars under control and stressed wheat seedlings. Silicon enhanced the SOD content under Cd stress (17%-48% and 30%-51%) and non-stress conditions (13%-35% and 4%-14%) in both Sahar-2006 and Inqalab-91 cultivars, respectively. Similarly, Si (1.5-3 mM)



Figure 2. Effect of Si spray on SOD, POD, CAT and APX contents in the shoot of 2 wheat cultivars seedlings grown under different levels of Cd stress (n = 3 \pm SE). Three-way ANOVA was performed and mean values (control and treatment values) were tested by least significant difference (LSD) at $P \leq 0.05$ and different lowercase letters indicate the significant difference between the treatments.

significantly increased in POD (11%-44% and 18%-74%), CAT (21%-44% and 16%-51%), and APX (16%-38% and 29%-81%) contents in both cultivars under Cd stress, while values of these attributes greatly enhanced at higher level of Si-application (3 mM) in both cultivars under Cd stress. The cultivar difference was also pertinent, Inqalab-91 (which was sensitive against Cd stress), while in Sahar-2006 cultivar, higher antioxidant activity was recorded and declared as tolerant against cadmium (25 mg kg⁻¹) stress (Figure 2).

Mineral Nutrient Contents

Cadmium stress resulted in a significant (P < 0.001) decrease in iron (Fe²⁺) in root and shoot of Sahar-2006 (25%-35% and 18%-28%) and Ingalab-91 (34%-49% and 23%-44%) cultivars, respectively. However, exogenous Si (1.5 and 3 mM) improved the Fe²⁺ contents in the root (17%-31% and 32%-50%) and shoot (13%-29% and 17%-42%) of both Sahar-2006 and Ingalab-91 cultivars under Cd stress compared with the control plants. However, there was a conspicuous difference between 2cultivars for root and shoot Fe contents, where the Sahar-2006 cultivar had greater values for Fe contents. (Table 5). Similarly, Zn contents considerably reduced ($P \le 0.001$) in the root and shoot of Sahar-2006 (16%-20% and 22%-31%) and Ingalab-91 (18%-48% and 24%-55%) cultivars under Cd stress, while Mg contents are also reduced in different plant parts such as root and shoot of both Sahar-2006 (13%-22% and 17%-31%) and Ingalab-91 (14%-39% and 25%-51%) cultivars, when subjected to Cd toxicity, respectively. Silicon with Cd addition, enhanced the Zn contents in root and shoot of Sahar-2006 (15%-24% and 19%-30%) cultivar, respectively. Whereas Si (1.5-3 mM) mitigated the effect of Cd stress and as a result, Mg content was greater in both root and shoot tissues of Sahar-2006 (9%-22%) and 15%-34%) and Ingalab-91 (21%-43% and 23%-49%) cultivars, respectively. Over all, greater increases in mineral nutrients $(Fe^{2+}Zn^{2+} and Mg^{2+})$ was identified only in both root and shoot tissues with Si (3 mM) in both wheat cultivars (Table 5).

Cadmium and Si Accumulation in Root and Shoot Tissues

The cadmium concentrations were significantly ($P \leq 0.001$) increased in root and shoot tissue of both Sahar-2006 (950%-1456% and 816%-1076%) and Ingalab-91 (980%-1125% and 696%-913%) cultivars under stress condition compared to controls. Exogenous applied Si enhanced the impairment and lessen the Cd uptake and accumulation of Cd in the root and shoot of Sahar-2006 (1076%-2313% and 1062%-1487%) and Ingalab-91 (885%-1390% and 830%-1274%) cultivars, respectively (Figure 3). Under Cd stress. Si concentration was significantly decreased in root and shoot tissues of Sahar-2006 (21%-52% and 17%-52%) and Ingalab-91 (10%-27% and 25%-56%) cultivars, respectively. In normal seedlings, Si (3 mM) resulted in 69% to 76% and 28% to 33% higher root and shoot Si content in both Sahar-2006 and Ingalab-91 cultivars with respect to Si (1.5 mM), respectively. More Si accumulated in the plant roots than leaves. In addition, compared to Si (1.5-3 mM) treated seedlings exhibited more Si content in root (17%-50% and 16%-50%) and shoot (22%-50% and 22%-49%) in both cultivars, respectively under Cd stress. Moreover, Sahar-2006 was more responsive to Si application (3 mM) than Inqalab-91cultivar (Figure 4).

Discussion

Metal contamination in agricultural soils has become a main environmental issue nowadays due to its toxic effects on

Cultivars	Cd stress (mg/kg)	Silicon (Si) (mM)	Root Fe ²⁺ (µg g ⁻¹ DW)	Shoot Fe ²⁺ (µg g ⁻¹ DW)	Root Zn ²⁺ (µg g ⁻¹ DW)	Shoot Zn ²⁺ (µg g ⁻¹ DW)	Root Mg ²⁺ (µg g⁻¹ DW)	Shoot Mg ²⁺ (μg g ⁻¹ DW)
Sahar-2006	0	0	201.49 ^d ± 1.70	154.20 ^b ±2.72	87.76 ^b ± I.81	31.06 ^b ± 0.61	278.51 ^{cd} ±4.19	199.62 ^b ±1.91
		1.5	234.82 ^b ± 5.71	164.20 ^a <u>+</u> 3.29	93.37 ^b ±5.04	32.01 ^b ± 0.93	295.90 ^b <u>+</u> 4.37	226.79 ^a ± 3.75
		3.0	$259.11^{a} \pm 1.92$	$172.46^{a} \pm 4.76$	99.26 a \pm 2.54	$33.78^{a} \pm 0.62$	$315.05^{a} \pm 4.08$	$234.54^{a} \pm 2.86$
	10	0	151.50 ^f ± 7.88	27. 3 ^e ±3.7	73.64 ^{de} ±1.37	24.15 ^{de} ± 0.67	242.42 ^e ±4.13	164.84 ^{de} ±2.81
		1.5	183.85 ^e ± 4.87	143.40 ^{cd} ± 3.51	79.17 ^{cd} ±2.37	$25.27^{d} \pm 0.40$	268.73 ^d ± 3.01	182.74 ^c ± 2.05
		3.0	214.97 ^c ± 3.84	150.26 ^{bc} ± 1.91	79.96° ±1.51	27.41 ^c ± 0.64	283.28 ^c ± 3.06	199.30 ^b ±1.67
	25	0	3 .70 ^g ± 7. 3	$111.20^{\text{gh}} \pm 4.48$	$70.30^{ ext{ef}} \pm 1.53$	21.54 ^g ± 0.55	217.13 ^f ±6.15	l 36.79 ^g ± 5.88
		1.5	162.36 ^f ± 6.12	116.40 ^{fg} ±4.89	71.96 ^{ef} ±2.21	22.39 ^{fg} ± 0.48	238.39 ^e ± 3.58	150.19 ^f ±2.25
		3.0	185.75 ^e ± 3.79	124.20 ^{ef} ± 3.25	75.42 ^{cde} ± 0.93	$23.60^{ef} \pm 0.42$	245.00 ^e ± 1.76	64.35 ^e ± .
Inqalab-91	0	0	175.63 ^e ± 4.42	114.00 ^g ±2.14	60.19 ^h ± 2.75	21.31 ^g ± 0.77	185.99 ^g <u>+</u> 2.62	148.79 ^f ±2.10
-		1.5	186.01° ± 3.60	l 29.93 ^e ± 4.75	62.79 ^{gh} ±1.04	22.33 ^{fg} ± 0.38	219.70 ^f ±3.12	165.76 ^{de} ± 2.50
		3.0	199.2 ^d ± 4.28	140.00 ^d ± 3.47	67.11 ^{fg} ±1.71	$23.54^{ m ef} \pm 0.61$	241.48 ^e ±1.79	173.19 ^d ±1.43
	10	0	115.29 ^h ± 3.09	88.24 ⁱ ±2.51	49.60 ⁱ ±1.71	16.24 ⁱ ± 0.45	160.02 ⁱ ± 3.87	2.0 ^h ± .7
		1.5	126.23 ^{gh} ± 4.66	103.72 ^h ± 2.99	53.49 ⁱ ±1.92	17.76 ⁱ ± 0.23	171.72 ^h ±1.62	120.21 ^h ±1.14
		3.0	133.77 ^g ± 2.72	116.79 ^{fg} ±2.50	61.45 ^{gh} ± 0.70	19.55 ^h ± 0.40	189.61 ^g ± 5.90	32.73 ^g ± 4.13
	25	0	89.71 ⁱ ± 1.40	64.37 ^k ±2.02	31.11 ^k ±1.11	9.65 ^k ± 0.19	113.60 ^k ± 4.11	72.52 ^j ± 5.75
		1.5	93.83 ⁱ ± 2.30	74.81 ^j ±3.27	34.36 ^k ± 0.93	11.26 ^j ± 0.36	I 28.55 ^j ± I.50	84.19 ⁱ ±1.97
		3.0	100.48 ⁱ ± 1.67	83.54 ^{ij} ± I.43	40.88 ^j \pm 1.34	12.56 ^j ± 0.49	138.16 ^j ± 0.79	92.56 ⁱ ± 2.99

Table 5. Effect of Si Spray on Mineral Nutrient in 2 Wheat Cultivars Under Cd Stress.

Abbreviations: Fe^{2+} = iron ion; Zn^{2+} = Zinc; Mg^{2+} = magnesium ion.

Means are followed by standard errors. Three-way ANOVA was performed and different mean values (control and treatment values) were tested by least significant difference (LSD) at $P \le 0.05$ and different letters following means within the same column indicate the significant differences between the variables in the table.

soil-plant system, food chain and human beings.⁶⁷ Soil contamination is generally caused by more release of metals from anthropogenic sources such as mining, smelting, irregular agricultural inputs (pesticides and fertilizers) and inadequate battery recycling.⁶⁸ The most abundant metals like the Cd normally co-exist in the environment, and disturb the metabolism of living organisms at the cellular level and affected the plant growth and yield.⁶⁹ Furthermore, Cd interferes with several metabolic processes, resulting inhibited plant growth, stunted development and plant death in mustard.²⁰ This inhibition by Cd stress might be related to the damaged photosynthesis system.^{70,71} The present study examines the role of foliar applied Si in mitigation of Cd stress in 2 wheat cultivars (sensitive and tolerant), and investigates the primary physiochemical mechanism. In this study, Cd stress harshly inhibited the fresh and dry masses of both cultivars, and this decline was more conspicuous in Ingalab-91 (sensitive) cultivar might also be the result of oxidative stress (Table 3). Although, a number of reports are available in the literature, where Cd-induced reduction in wheat growth attributes, and photosynthesis, might be due to reduce cell division and cell expansion owing to oxidative damage and decline in activities of important antioxidant enzyme activities.^{70,72} Cadmium induced decline in chlorophyll and carotenoid contents has been reported in many plants, e.g. strawberry,⁷³ faba bean,⁷⁴ cotton,⁴⁴ and Artemisia annua.⁷⁵ Furthermore, Cd toxicity decreased plant growth and affected photosynthetic efficiency in various crops and also affects the nutrient uptake and disrupt the membrane structure and properties.^{71,76} The similar

trend was found in wheat plant grown under Cd stress in our study.

Silicon application improves the plant growth and biomass might be due to photosynthesis improvement,³⁴ enhanced nutrient uptake⁷⁷ and immobilized the Cd.⁴⁵ Furthermore, Si is known to enhance the internodal elongation of shoot by increasing the cell division and cell expansion,78,79 which could be a reason for better growth performance under stress condition. However, exogenous application of Si was effective in improving the harmful effects of Cd stress on wheat growth and photosynthetic pigments, as shown by Si-induced increase in shoot and root growth (Table 3), chlorophyll pigments (Table 4). Moreover, Si application enhances the chlorophyll pigments might have also contributed in enhancing the growth performance of wheat under stress and non-stress conditions. Previous studies considered the Si, as a growth regulator-like compound that enhanced the cell division and extension in different crops, including maize,⁷² mustard,⁸⁰ alfalfa,⁷⁸ barley⁷⁹ and rice⁸¹ has a defensive role against many biotic and abiotic stresses.⁸² Silicon improved the photosynthetic pigments and protect them from oxidative injury by increasing the synthesis and accumulation of chlorophyll and carotenoid contents.^{83,84} In the present study, Si application lessens the oxidative injury in wheat caused by Cd, which might have provoked the chlorophyll and carotenoid contents accumulation in wheat seedling (Table 4).

The equilibrium between the generation of ROS and its removal from the plant cell or tissue disturb and causes oxidative damage to membrane-bound organelles when a plant



Figure 3. Effect of Si spray on Si contents in the root and shoot of 2 wheat cultivars seedlings grown under different levels of Cd stress (n = 3 \pm SE). Three-way ANOVA was performed and mean values (control and treatment values) were tested by least significant difference (LSD) at $P \leq 0.05$ and different lowercase letters indicate the significant difference between the treatments.

subjected to abiotic stress.^{24,85} Cadmium toxicity inhibits the activities of antioxidant enzymes in plants due to more accumulation of MDA and H_2O_2 in plants, which reflected the damaging of cell membrane and thus, increases the EL in plants.⁸⁶ In present study, greater production of ROS (e.g., H_2O_2) considerably increased the MDA content and EL in wheat cultivars under Cd (25 mg kg⁻¹) stress (Figure 1). Earlier, Cd toxicity induced generation of H_2O_2 and MDA contents has been described in different plant species. For instance, rice,⁸⁷ rapeseed,²⁴ tomato,⁸⁸ and maize.⁸⁹ While the plant has a strong defense mechanism which scavenged ROS production by increasing the activities of various antioxidants and plants with a strong defense system are known as more resistant plant species against oxidative stress.^{77,79,90}

The results of this study have shown that Si application reduced ROS generation and accumulation that afterward limited the lipid peroxidation with the help of increased activities of antioxidant enzymes (Figure 1). Furthermore, Si has a defensive role to inhibit the generation of ROS, maintain the membrane stability, and alleviate the lipid peroxidation indirectly under stress conditions.⁹¹ We also



Figure 4. Effect of Si spray on Cd contents in the root and shoot of 2 wheat cultivars seedlings grown under different levels of Cd stress (n = 3 \pm SE). Three-way ANOVA was performed and mean values (control and treatment values) were tested by least significant difference (LSD) at $P \leq 0.05$ and different lowercase letters indicate the significant difference between the treatments.

recorded a Cd-induced reduction in the accumulation of non-enzymatic antioxidants (low molecular weight) such as phenolics, flavonoids, and anthocyanins. The Sahar-2006 cultivar was superior to Ingalab-91cultivar for nonenzymatic antioxidant compounds. Non-enzymatic antioxidants also protect the plants against pests and diseases, play a role in plant development and regarded as a powerful antioxidant compounds⁹² and protect the plants from ROSinduced oxidative stress.⁹³ Our results have shown that rise in the endogenous levels of anthocyanin in plants with Si application because anthocyanin possess potent antioxidant activity that enables them to play pivotal role in plant abiotic stress tolerance (Table 4). Plants tend to accumulate more anthocyanin contents under abiotic stress.⁹⁴ Metal stress induced the production of anthocyanin contents, which has been reported in many plants, e.g. barley,⁷⁹ and Echium amoenum.95 Anthocyanin contents have strong potential to scavenge the free radicals and prevent the plants from oxidative damage.⁹⁶

In the present study, Cd toxicity significantly reduced the soluble protein contents in wheat (Table 4); similar results have been reported for Tall fescue seedlings.⁸⁶ Osmotic adjustment is also an important way to resist adverse effect of stress on plants. The osmotic adjustment substances such as proline and soluble protein plays vital roles in maintaining the osmotic equilibrium and integrity of membranes.⁹⁷ The reduction in soluble protein contents might be due to inhibition of the uptake of nitrate ions in root by metals, and may partly reflect the changes in various antioxidant enzymes to cope with metal induced oxidative stress.98 Application of Si (3 mM) more increased the soluble protein content in Cd treated wheat plants. This increase in soluble protein may possibly be due to increase of nitrate reductase biosynthesis with the inhibition of their degradation,⁸⁰ thus enhanced the accumulation of soluble protein in leaves of wheat plants.

Cadmium can promote the production of reactive oxygen species (ROS) thus causing oxidative stress in plants.²⁹ Increased activities of antioxidant enzymes also contribute significantly to the antioxidant potential of plants.⁹⁹ Cadmium is generally known to inhibit the activity of various enzymatic antioxidants such as SOD, POD, APX and CAT.^{100,101} For example, Cd stress decreased the SOD and APX activities in cotton¹⁰² and reduced CAT activity in wheat.⁴⁵ Silicon (3 mM) application further increased the activities of SOD, POD, APX and CAT under Cd stress (Figure 2), the increase antioxidant in wheat activated the defense mechanism in plants, thus enhanced the Cd resistance in wheat seedling. These antioxidants coordinately played a key role in alleviating the oxidative damage in Si treatments seedlings. Similar results were reported in rapeseed,²⁴ cucumber,¹⁰³ Brassica chinensis,¹⁰⁴ and soybean¹⁰⁵ under environmental stresses.

The shortage of essential nutrients in plants is directly linked to the growth and yield of plants. Abiotic stresses hinder the uptake of nutrients by down regulating the expression of membrane transporters.¹⁰⁶ Cadmium, being a bivalent cation, usually competes with Mg, Ca, Zn, and Fe during membrane transport.^{107,108} Cation transporters can uptake toxic elements (Cd) potentially.¹⁰⁹ Our results suggested that Cd stress considerably decreased the Mg^{2+} , Zn^{2+} , and Fe²⁺ accumulation in root and shoot of both cultivars (Figure 3). However, Si treatment reduced the Cd uptake and considerably improved the macro- and micronutrient concentration in leaf, stem and root tissues. Increased uptake of mineral elements like Mg, Zn, Fe and Si in Si treated plants may have directly affected the metabolism by regulating the mechanisms such as chlorophyll synthesis, protein synthesis, and enzyme activity.⁴² In the present study, Si alleviated the Cd effect on the mechanisms related to uptake of nutrients, thereby contributing to the maintenance of plant growth and development under Cd stress (Figure 3).

The toxicity of Cd relies on Cd species, which influences its uptake, accumulation, and translocation of Cd. In the current study, significant increase Cd contents in the root and leaf by the addition of Cd, which was then taken up by the wheat plants. In addition, more Cd contents were accumulated in root of both cultivars than those in the shoot (Figure 4). Thind et al (2020)⁸⁴ described that Si inhibited the uptake of Cd and decreased its transport in different plant tissues by chelating mechanism. Si being a non-essential nutrient for crop plants can affect plant growth and development in stressed¹¹⁰ or normal conditions.³⁸ In this study, application of Si (3 mM) decreased the Cd content in root and shoot of both cultivars under Cd stress (Figure 4). Moreover, Si prevent the Cd transport in plants via symplastic pathway. Silicon inhibited the Cd accumulation and increased the yield because of more photosynthetic pigment synthesis.⁴⁹ These findings showed that our easy and practical approach is appropriate for agricultural practices to combat metal stress environment.

Conclusions

Cadmium stress a severely hampered the growth attributes, photosynthetic pigments, and nutrient uptake in wheat cultivars, while enhancing the levels of oxidative stress indicators (MDA and H₂O₂) contents. However, Si application significantly increased the plant growth under Cd stress. Cadmium regime (25 mg kg⁻¹) revealed a substantial increase in the activities of antioxidant enzymes. Higher activities of these antioxidants associated with the lesser oxidative damage in terms of minimal cellular H₂O₂ and MDA contents. The response of both wheat cultivars was significantly variable with respect to Cd tolerance. Inqalab-91 cultivar was found to be sensitive having higher MDA and H₂O₂ contents and poor antioxidative defense system, and more degradation of chlorophyll under Cd stress. The application of Si increased growthrelated-attributes, improved photosynthetic machinery, enhanced essential nutrient uptake and increased the activities of various antioxidant compounds by decreasing the MDA and H₂O₂ contents and Cd uptake in various tissues of plants. In addition, the use of Si might provide interesting information to future researchers about chelating mechanism, which involved in Si induced metal tolerance of plants under stress conditions. Overall, foliar applied Si relieves Cd stress by regulating growth, physiological, and biochemical, and antioxidant attributes in wheat.

Declaration of Conflicting Interests

The author(s) declared no potential conflicts of interest with respect to the research, authorship, and/or publication of this article.

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