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## **REVIEW**



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# Accessing the synthesis of natural products and their analogues enabled by the Barbier reaction: a review

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The Barbier reaction is significantly referred to as one of the efficient carbon–carbon bond forming reactions which involves the treatment of haloalkanes and carbonyl compounds by utilizing the catalytic role of a diverse range of metals and metalloids. The Barbier reaction is tolerant to a variety of functional groups, allowing a broad substrate scope with the employment of lanthanides, transition metals, amphoteric elements or alkaline earth metals. This reaction is also water-resistant, thereby overcoming the challenges posed by moisture sensitive organometallic species involving C–C bond formation reactions. The Barbier reaction has significantly found its applicability towards the synthesis of intricate and naturally occurring organic compounds. Our review provides an outlook on the synthetic applications of the Barbier reaction and its variants to accomplish the preparation of several natural products, reported since 2020.

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### 1 Introduction

Carbon–carbon bond forming reactions are of substantial interest in organic chemistry to afford the functionalization of chains along with their branching and elongation. These reactions have garnered considerable attention in synthetic organic chemistry, thereby leading to the synthesis of intricate molecules. The Hutil the end of the 19th century, zinc metal was predominantly employed in several organic reactions. For P. Barbier and V. Grignard introduced the utilization of magnesium metal in the addition reactions of carbonyl compounds. This employment led towards the improvisation of carbon–carbon bond forming reactions, thus exploring the use of other metals in these reactions. Initially, the Barbier reaction reported the utilization of magnesium metal-based turnings in the

treatment of methylheptenone **1** with methyl iodide **2** to afford the synthesis of 2,6-dimethylheptenol **3**. The attainment of the target molecule in moderate yield with limited replicability hindered the exploration of the Barbier reaction (Scheme 1).<sup>7</sup>

Later, Barbier's pupil Grignard performed this reaction by modifying the protocol, where he isolated the generated nucleophile *i.e.*, organomagnesium intermediate, rather than exploiting a one-pot reaction. The synthesized nucleophile 5 was then further treated with carbonyl compound 4 resulting in relatively higher yield with consistent replicability (Scheme 1).8 Ever since 1900, the Grignard reaction has held vital importance in synthetic organic chemistry, owing to its facile pathway towards the procurement of various organic molecules.9-11 Based on his significant contribution in synthetic chemistry, Grignard was awarded with the Nobel prize in 1912.

Both Grignard and Barbier reactions are regarded as nucleophilic addition reactions of carbonyl compounds by treating them with alkyl halides in the presence of metal or its salts, thus resulting in the synthesis of higher order alcohols. However, they mainly vary on the way, the nucleophilic specie (organometallic specie) is produced followed by its attack on the electrophilic centre of carbonyl functionality.<sup>12,13</sup> Though, Grignard reaction was more frequently employed in past years, most recent modifications that sustain the original one-pot characteristics of Barbier reaction are referred as "Barbier-type reactions".<sup>14</sup> They have been found to be even more favourable than Grignard reactions as they surpass the certain limitations imposed by Grignard reactions which involve the requirement of anhydrous conditions, withdrawal and handling of

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Scheme 1 Synthesis of tertiary alcohol from Barbier and Grignard reaction.

organometallic specie. This issue is resolved by substituting Mg metal with some other metal or amphoteric element that are resistant to moisture. Thus, there has been continuous advancement regarding the usage of variety of metals, bimetallic system, zero valent metals and benign metal slats in the Barbier reaction. Moreover, water has been employed as solvent in several methodologies, eliminating the use of toxic and expensive solvents.

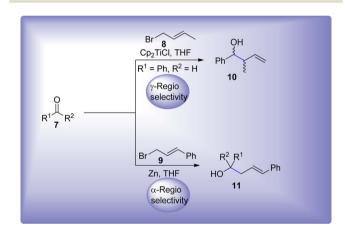
Besides the methodological development, substrate scope of Barbier reaction has also been explored and thoroughly investigated in several protocols to include the treatment of various vinyl, alkyl, propargyl and allyl halides in combination with imines, esters, ketones, nitriles, aldehydes and several azo compounds. The broad range of substrate scope has been analyzed in Barbier reaction to produce a diverse range of complicated products.<sup>18-20</sup>

Stereoselectivity and regioselectivity are some other considerable factors in Barbier reaction, which arise under certain conditions. The  $\alpha$  or  $\gamma$ -adduct emerges as a result of treating  $\gamma$ -substituted allyl compounds. The nature of regioselective product is dependent upon the solvent,  $\gamma$ -substrate bulkiness, utilized additive and metal promoter species.  $\gamma$ -Adduct is usually obtained predominantly by employing In,  $^{21-23}$  Sn,  $^{24-26}$  Zn  $^{27,28}$  or Ti  $^{29,30}$  metals in Barbier reaction. However,  $\alpha$ -adduct is selectively produced by the introduction of praseodymium  $^{31}$  and neodymium  $^{32}$  metals in Barbier reaction utilizing  $\gamma$ -substituted allyl compounds as precursors (Scheme 2).

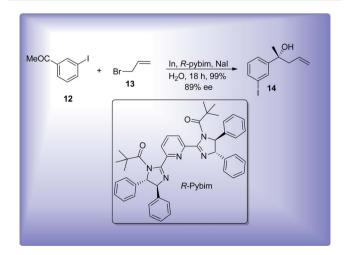
Furthermore, in order to achieve highly enantioselective Barbier product, chiral additives or ligands *i.e.*, (1S,2R)-2-amino-1,2-diphenylethanol,<sup>33</sup> R-pybim<sup>34</sup> and  $\beta$ -cyclodextrin<sup>35</sup> are significantly employed (Scheme 3).  $\gamma$ -Regioselectivity approach of Barbier reaction has the probability of leading towards two optically active centers in the target molecule, whose relative stereochemistry is defined by the ability of utilized metal towards the formation of cyclic transition state.

Utility of Barbier reaction has also been explored in polymer chemistry where benzoyl group and organic halide constituting monomers were added *via* one-pot A<sub>2</sub>, AB and B<sub>2</sub> Barbier polyaddition to attain a catalog of polymethanol substituted

polymers.<sup>36</sup> Several one-pot reactions are utilized to accomplish the synthesis of complex organic scaffolds.<sup>37–41</sup> There has been significant advancement towards the synthesis of several



Scheme 2 Layout for  $\gamma$ -regioselective and  $\alpha$ -regioselective Ti & Zn-mediated Barbier allylation reaction.



Scheme 3 Layout for an enantioselective In-mediated Barbier allylation reaction.

Fig. 1 Synthesis of some naturally occurring organic compounds by employing Barbier allylation reaction as a key step.

biologically active naturally occurring organic compounds.<sup>42-44</sup> Barbier reaction has also found its significance in this specific domain, as it has been found to involve as one of the main reaction steps leading towards the procurement of natural products.<sup>45,46</sup> Phorbin A,<sup>47</sup> Cochliomycin C<sup>48</sup> and (–)-8-*epi*grosheimin<sup>49</sup> are some of the naturally occurring compounds endowed with homolallylic alcohol functionality, introduced *via* Barbier allylation reaction (Fig. 1).

Considering the wide synthetic utility of Barbier reaction and its variants, significant research groups have contributed to summarize its scope, merits and remarkable applications. 50–56 However, very few reports have been published (till now) to emphasize the utilization of Barbier reaction towards the procurement of natural products. Recently, in 2022, Petrides and Georgiades reported a specific overview of stereocontrolled Barbier allylation and its applications towards the synthesis of natural products. The objective of our review is to highlight the latest applications of Barbier and Barbier-type reactions towards the synthesis of a variety of natural products, reported within 2020–2024.

#### 2 Literature review

#### 2.1. Synthesis of alkaloids-based natural products

The alkaloid (–)-205B is a medicinally active and lipid-soluble natural product, which was first came across by Daly and colleagues. The alkaloid (–)-205B is structurally composed of azaacenonaphthylene moiety with five chiral centers. The total synthesis of this naturally occurring alkaloid has been performed by several research groups *via* diverse synthetic and organic transformations. Taking into consideration the medicinal weightage and intricate structure of this alkaloid, Tripathy and Schneider attempted its total synthesis by involving various name reactions *i.e.*, Mukaiyama–Mannich reaction, aza-Prins annulation and zinc promoted Barbier allylation as key steps. Their developed briefest synthetic strategy initiated with vinylogous Mukaiyama–Mannich reaction between aldehyde moiety 18, tethered ester group 19 and amine 20 in the presence of compound 21 followed by the addition of

acetic acid to generate the lactam ring 22 in 73% yield with 98:2 enantiomeric ratio and more than 20:1 diastereoslectivity ratio. The  $\gamma$ -lactam 23 was easily derived from compound 22 in 81% yield over two steps. In the next step, the lactam 23 was initially reduced to the respective imine A on treatment with Cp<sub>2</sub>ZrHCl in tetrahydrofuran at -25 °C, which was then further subjected to zinc promoted Barbier allylation using methallylbromide 24 to afford indolizidinone 25 in 71% yield with 15:1 diastereomeric ratio, thereby resulting in substrate control optically active center. It was observed that indium promoted Barbier allylation reaction resulted in decreased yield and stereoselectivity of indolizidinone. The next step involved the introduction of methyl group at 6th position of indolizidinone on reaction with lithium diisopropyl amide and tetrahydrofuran to generate the corresponding enolate which was further made to react with methyl iodide in THF at -78 °C. As a result, diastereomeric mixture of methylated compounds 26a & 26b was attained in 1:3 diastereomeric ratio with 85% yield. The diastereomeric mixture was then subjected to aza-Prins annulation reaction in the presence of Vaska's complex IrCl(CO)(PPh<sub>3</sub>)<sub>2</sub> and TMDS followed by the treatment with p-TsOH·H<sub>2</sub>O to afford the target natural product 27a with its 6-epimer 27b (via elimination reaction) in 1:1 diastereomeric ratio. The chromatography technique was carried out to separate the resulting isomers, thereby resulting in the isolation of alkaloid (-)-205B 27a in 35% yield (overall 12% yield) (Scheme 4).

Indole alkaloids belong to the biologically active class of natural products, which are utilized in several organic transformations.61 Synthesis of indole alkaloids constituting azaquaternary optically active center i.e., subincanadine A, arborisidine and taberdivamine B, encounter severe challenges due to their convoluted structural framework. Arborsidine is composed of 5-membered cyclic structure with two chiral centers. It has been found to be potent against gastric cancer cell lines in coordination with pimelautide. Cancer is a deadly disease and researchers are continually making efforts to develop and synthesize anti-cancer agents. 62,63 Till now, two successful total synthesis of arborisidine have been reported.<sup>64,65</sup> In 2021, Wang and Jiao<sup>66</sup> established an efficient asymmetric total synthesis of arborisidine by employing Barbier-type addition reaction as a key step. In order to achieve the asymmetric synthesis of this natural product, Barbier type addition reaction was performed between harmalane 28 and allyl bromide 29 by involving chiral indium reagent in THF/ DMF at 35 °C followed by protection of nitrogen with Boc to afford compound ( $\pm$ )-30 in 51% yield. In the next step, compound (±)-30 was subjected to palladium mediated intramolecular dearomative asymmetric allylic alkylation (AAA) exploiting diaminocyclohexane (DACH)-phenyl Trost ligand (R,R)-L1 (via parallel kinetic resolution (PKR) regime), which resulted in asymmetric synthesis of compound ( $\pm$ )-31 & ( $\pm$ )-32. The compound  $(\pm)$ -31 was further treated with benzene sulfenylchloride followed by reaction with 2,4,6-coolidine/ tetrabutylammonium chloride involving elimination reaction to forge phenyl sulfide substituted compound ( $\pm$ )-33. The compound ( $\pm$ )-33 then underwent N-Boc deprotection with subsequent alkylation to afford corresponding alcohol (-)-34 in

Scheme 4 Synthesis of the alkaloid (-)-205B 27a.

34% overall yield (as a single isomer). The proceeding step involved the iodination of compound (-)-34 followed by 5-*exotrig* radical cyclization to result in compound (-)-35 in 74% yield. Later, the compound (-)-35 was transformed over few steps to assemble the naturally occurring (-)-arborisidine (-)-36 in fine yield with efficient diastereoselectivity (Scheme 5).

Casuarine is a natural product that is extracted from *Eugenia jambolana* Lam and *Casuarina equisetifolia* L.<sup>67</sup> Casuarine is composed of pyrrolizidine ring with 6 chiral centers, thus it exists in almost 64 stereoisomers. It also acts as an efficient isomaltase, maltase (rat instetinal) and α-glucosidase inhibitor (rice). Moreover, it has also been found to exhibit the properties of competitive inhibitor of *C. riparius* trehalase, NtMGAM (human N-terminal subunit of maltase-glucoamylase) and *A. niger* amyloglucosidase. In 2021, Li *et al.*<sup>68</sup> reported the synthesis of naturally occurring casuarine 41 and its natural stereoisomer 6-*epi*-casuarine 42 by applying Barbier conditions in one of the key steps. Their synthetic route initiated with the reaction of nitrone 37 (obtained from p-arabinose) with

allylbromide in the presence of zinc (Barbier-type reaction) in tetrahydrofuran and ammonium chloride followed by subsequent reaction with carboxybenzyl chloride (CBzCl) in the presence of sodium carbonate to afford compound 38 in 75% yield. In the following step, compound 38 was subjected to SeO<sub>2</sub> promoted allylic oxidation by employing t-BuOOH in dichloromethane to give corresponding alcohol 39 in 53% yield (as a single isomer). Here, the Barbier-type reaction involving pathway led to the relatively higher yield of alcoholic isomer 39 in comparison to Grignard reagent involving protocol. In the next step, -OH group of compound 39 was protected with MOM ether followed by diastereoselective dihydroxylation by employing osmium tetroxide and NMO in acetone to furnish the mixture of triols 40a and 40b in 84% yield. The isomers 40a and 40b were treated individually over few steps to accomplish the synthesis of casuarine 41 and 6-epi-casuarine 42 respectively (Scheme 6).

Most of the biologically active natural products consist of spirocyclic frameworks.<sup>69</sup> In 2021, two diastereomeric naturally

occurring isoindolinone alkaloids *i.e.*, spirocollequins A and B have been extracted from *Colletotrichum boninense* AM-12-2 (an endophytic fungus).<sup>70</sup> They are composed of a furan ring

alongwith isoindolinone skeleton within their spirocyclic skeleton.<sup>71</sup> These isoindolinone alkaloids have the potency to exhibit anti-plasmodial activity. In 2022, Ichikawa and

Scheme 5 Synthesis of (-)-arborsidine (-)-36.

Scheme 6 Synthesis of casuarina 41 and 6-epi-casuarine 42.

coworkers72 carried out the first total synthesis of these alkaloids involving Barbier reaction as a key step. Their synthetic endeavor initiated with the reaction of phthalic anhydride 43 with 2-aminoethanol in the presence of triethylamine, which was proceeded by the protection of hydroxyl groups with tertbutyl silyl group to furnish compound 44 in 86% yield. In the next step, the compound 44 was subjected to magnesium induced Barbier allylation reaction with allylbromide, resulting in the synthesis of isoindolinone 45 in 82% yield, as a single product. The allylic functionality in compound 45 was further subjected to cross-metathesis reaction on treatment with compound 46, utilizing Grubb's 2nd generation catalyst with copper iodide in diethyl ether. This reaction was followed by osmium tetroxide NMO induced dihydroxylation followed by treatment with PPTS (pyridinium p-toluenesulfonate) to result in diastereomeric mixture of 47a and 47b (40% isolated yield). The compound 47a (34%) was purified via selective TBS protection of its hydroxyl functionality. The reaction was proceeded by the deprotection strategy followed by subsequent prenylation reaction to afford spirocolleguins A 49a in 89% yield. On the similar ground, 47b was converted to spirocollequins B 49b in 93% yield. As a result, the foremost total synthesis of (±)-spirocollequins B 49b and A 49a was achieved in 24% & 19% overall yields, correspondingly (Scheme 7).

#### 2.2. Synthesis of lactones-based natural products

Cephalosporolides are of huge significance owing to their unique structural and medicinal attributes. (–)-Cephalosporolide D is a naturally occurring eight membered lactone which

is extracted from Cephalosporium aphidicola (fermentation fungus).73 In 2000, Shiina and coworkers reported the first total synthesis of this naturally occurring lactone along with the ascertainment of its (absolute and relative) stereochemistry.74 Since then, there have been several reports concerning the synthesis of this bioactive natural product. Being persuaded by the gripping structural and pharmaceutical features of (-)-Cephalosporolide D, Kalavakuntla et al. 75 in 2022, carried out an efficient and compact route towards the its total synthesis by employing Barbier allylation reaction, Mitsunobu esterification and RCM (Ring Closing Metathesis) as key steps. Their synthetic approach commenced by treating optically active aldehyde 50 with allylbromide via zinc promoted Barbier allylation in tetrahydrofuran at 0 °C to afford carbinols (51a & **51b**) in a divisible mixture (syn and anti = 1:9) with 79% yield. The next step involved the protection of alcoholic functionality to ether 52 (in 88% yield) on treatment with p-methoxybenzyl bromide by utilizing sodium hydride as base in tetrahydrofuran. PMB ether 52 was then subjected to react with periodic acid followed by sodium borohydride mediated reduction to yield alcohol in 84% yield. The tosyl group was then installed on alcohol proceeded by its subsequent reaction with lithium aluminum hydride in dry tetrahydrofuran to access compound 53 in 77% yield. In the next step, alcoholic compound 54 was obtained (in 85% yield) as a result of deprotection of PMB ether by treating compound 53 with 2,3-dichloro-5,6-cyano-1,4benzoquinone (DDQ) in aqueous DCM. PMB ether 52 was also transformed to corresponding acid 55 over few steps. The resulting acid 55 then underwent Mitsunobu esterification with already prepared alcohol 54 followed by ring closing metathesis

Scheme 7 Synthesis of (±)-spirocolleguins A 49a & B 49b.

Scheme 8 Synthesis of Cephalosporolide D 57.

(RCM) reaction (exploiting G-II catalyst) to forge the eight membered heterocycle **56** in 66% yield. Finally, the compound **56** was subjected to Pd/C mediated hydrogenation to accomplish the synthesis of targeted compound **57** in 81% yield (Scheme 8).

The 5,6-dihydro-pyrone functionalities are copiously found in several medicinally active natural products as they illustrate efficient potency against various deadly diseases and a wide range of cancer cells. Obolactone and dihydroobolactone are natural products, whose structural formulas are incorporated with 5,6-dihydro-pyrone moieties. Cryptocaria obovate is the main source of obolactones, which are known to exhibit potent efficacy against Trypanosoma brucei 5.3 µM nasopharyngeal cancer cells with  $IC_{50}=3~\mu M.^{76,77}$  In 2021, Saini et al. 78 accomplished the total synthesis of these naturally occurring and pharmaceutically significant organic compounds. Their synthetic pathway initiated with the transformation of Laspartic acid 58 to alkyne 59 over few steps. In the next step, alkyne 59 was subjected to nucleophilic addition reaction to install the alkynol functionality 60 in 75% yield. Later, compound 60 underwent MnO<sub>2</sub> mediated oxidation followed by p-toluene sulfonic acid promoted rearrangement to forge the 2,3-dihydro-pyranone moiety 61 in 75% yield. The alcoholic functionality in compound 61 was subjected to Dess-Martin periodinane oxidation to transform it into aldehyde which further underwent treatment with allylbromide via zinc promoted Barbier allylation to result in the synthesis of compound 62 in 69% yield with dr = 1:1.3. The substrate directed Barbier approach was chosen to induce chirality within allylation reaction to afford the synthesis of natural products'

both diastereomers. The Barbier allylated product **62** was further transformed into naturally occurring (*R*,*R*)-obolactone **63a** (32%) and its separable isomer *i.e.*, 6-*epi*-obolactone **63b** (40%) *via* esterification reaction with ensuing G–II mediated ring closing metathesis (RCM). In the similar manner, alkyne **59** was also subjected to react with 3-phenylpropanal followed by the same series of reactions to afford the synthesis of naturally occurring 7′,8′-dihydroobolactone **64a** and its separable isomer *i.e.*, 6-*epi*-7′,8′-dihydroobolactone **64b** in 33% and 39% yield respectively (Scheme 9).

Marine bacteria (of Salinispora genus) are involved in the production of salinosporamides based natural products. They are structurally composed of bicyclic framework having γ-lactam-β-lactone functionality. Considering the unique characteristics of salinosporamides, Park et al. in 2022,79 unveiled the synthesis of salinosporamide B, which was initially extracted from the S. tropica. Their established protocol involved the use of Barbier allylation as one of the key steps. In the foremost step, β-ketoacid 65 was transformed to compound 66 as a result of reaction with (COCl)<sub>2</sub> followed by its condensation reaction with oxazolidinone 67 in acetone to generate required substrate 68 in 64% yield with E/Z = 1:1. Later, the compound 68 was treated over several steps to result in compound 69. In the next step, alcoholic functionality in compound 69 was subjected to Dess-Martin periodinane mediated oxidation to convert it into aldehyde intermediate 70. The aldehyde 70 was then further subjected to react with bromo cyclohexene 74 via indium catalyzed Barbier allylation reaction in the presence of ammonium chloride (additive) and tetrahydrofuran at room temperature to afford the addition product 75 in 70% yield with 10:1

Scheme 9 Synthesis of (R,R)-obolactone 63a, 6-epi-obolactone 63b, 7',8'-dihydroobolactone 64a and 6-epi-7',8'dihydroobolactone 64b.

diastereoselective ratio. It was observed that the Barbier reaction streamlined the synthetic pathway, thereby affording the product 75 in efficient yield with higher diastereoelectivity. Addition product 75 was then subjected to tert-butyl ester hydrolysis via TFA proceeded by exposure with BOP-Cl to fulfil the synthesis of salinosporamide B 76 in 74% yield. Recently, in 2023, Park et al.80 also utilized indium promoted Barbier allylation approach to afford the first chiral synthesis of salinosporamide D 78 in eleven steps with 12% overall yield. To start with, D-serine based ester 71 was made to react with triethylamine and acetone followed by coupling with β-ketoacid and reaction with sodium ethoxide in ethanol to procure aldol compound 72 in 73% yield with more than 99% ee (as a single diastereomer). The synthesized aldol compound 72 was converted to corresponding aldehyde over several steps. The aldehyde 73 was then treated further via similar In-mediated Barbier allylation to afford compound 77 in 71% yield (as a single isomer). The resulting compound 77 was then made to react with BCl<sub>3</sub> to carry out the removal of benzyl and tert-butyl groups proceeded by exposure to BOPCl (bis(2-oxo-3oxazolidinyl)phosphinic chloride) to furnish the total synthesis of salinosporamide D 78 in 70% yield. Similarly, inpromoted Barbier reaction involving allylation approach was employed to synthesize salinosporamide I 79 (Scheme 10).

Almost 10% of natural products constitute γ-butyrolactone functionalities, indicating the importance of these organic moieties. The medicinal and pharmaceutical attributes of these butyrolactones have not been studied due to their extraction (from natural sources) in minute quantity. Thus, the need to synthesize these naturally occurring butyrolactones in efficient yields is enhancing continuously. In 2022, Gayke et al.81 adopted a stereoselective route towards the asymmetric synthesis of (4S,5S)-4-hydroxy- $\gamma$ -decalactone 85, which was first isolated alongwith its major isomer ((4S,5R)-4-hydroxy-γdecalactone) by Raspotnig group from the secretion of harvestmen Egaenus convexus ejection. In arachnids, these butyrolactones act as defensive agents by releasing exocrine secretions against hazardous chemicals and perpetrators. Gayke and coworkers initiated the synthesis by treating aldehyde **80** with allyl bromide *via* zinc promoted Barbier allylation reaction in tetrahydrofuran and water to generate homoallyl alcohol 81 in 89.4%. The hydroxyl group in compound 81 was protected on reaction with benzyl bromide using sodium hydride as base, followed by removal of acetonide on treatment with p-toluene sulfonic acid to forge respective diol 82 in 91% yield. The diol 82 then underwent oxidative cleavage by exploiting sodium periodate and dichloromethane followed by Grignard reaction with 1-bromoacetone using magnesium in

Scheme 10 Synthesis of salinosporamide B 76, salinosporamide D 78 and salinosporamide I 79.

diethyl ether to synthesize compound **83** in 82% yield. The compound **83** was treated further to result in lactone **84** (with 69% yield) on exposure with oxidizing agents (RuCl<sub>3</sub> and NaIO<sub>4</sub>) proceeded by acidic treatment. Finally, deprotection of hydroxyl group furnished the synthesis of naturally occurring minor isomer of (4S,5S)-4-hydroxy- $\gamma$ -decalactone **85** in 89% yield (Scheme 11).

Cryptorigidifoliol is a tetrahydrone pyrone constituting oxygenated natural product, which was obtained as a result of extraction from the *Cryptocarya rigidifolia* (in 2015).<sup>82</sup> Cryptorigidifoliol along with its other class members tend to showcase a broad spectrum of medicinal applications, thereby acting as efficient anti-fungal, anti-cancer, anti-inflammatory and anti-bacterial agents.<sup>83,84</sup> Besides its utilization as efficacious analgesics, this natural product has been found to demonstrate cytotoxic activity against specific human cancer cell lines and the antimalarial activity against *Plasmodium falciparum*. Till

now, a single report has been documented focusing the synthesis of cryptorigidifoliol I in 10.8% overall yield by Mohapatra and coworkers. Recently, in 2024, Jangid and Fernandes85 established a successful route to access this naturally occurring compound by employing Barbier allylation as one of the central steps. Their synthetic journey initiated with the procurement of δ-hydroxyalkynone 87 by subjecting L-aspartic acid 86 to a series of consecutive reactions. The δ-hydroxyalkynone intermediate 87 was then transformed into dihydropyranone 88 (in 91% yield) via p-TsOH mediated rearrangement reaction. The resulting dihydropyranone was then exposed to DMP promoted oxidation in continuation with zinc promoted Barbier allylation reaction with allyl bromide in tetrahydrofuran to afford diastereomeric mixture. The diastereomeric products were feasibly separated via column chromatography. The subsequent oxidation step resulted in the synthesis of diketone 89 (within 76% yield). Furthermore, the

Scheme 11 Synthesis of (4S,5S)-4-hydroxy- $\gamma$ -decalactone 85

Luche reduction of diketone 89 furnished diastereomeric mixture of bis-ethers 90a & 90b (dr about 2:1), alongwith reduction products 91a & 91b. The acidic hydrolysis of these ethers rendered diospongin analogues 92a and 92b respectively.

The diospongin analogue 92a was then made to react over several steps including sequence to accomplish the total synthesis of cryptorigidifoliol I 93 alongwith its epimer 94 (Scheme 12).

Scheme 12 Synthesis of cryptorigidifoliol | 93 and 2'-epi-cryptorigidifoliol | 94.

2.2.1. Synthesis of macrolides-based natural products. Thuggacins A–C are natural products, which were withdrawn from the myxobacterium *i.e.*, *Sorangium cellulosum*. <sup>86</sup> Their structural skeleton is composed of  $\alpha,\beta$ -unsaturated macrolide alongwith thiazole ring and conjugated diene framework. They exhibit potent activity against a diverse variety of microorganisms. They have been found to affect the respiratory chain of *Mycobacterium tuberculosis*, thereby acting as therapeutic agent against tuberculosis. Thuggacin cmc-A (which is composed of 17 members) was isolated by Jansen and colleagues from myxobacterium Chondromyces crocatus Cmc5. In 2021, Tsutsumi *et al.* <sup>87</sup> provided the first total synthesis of thuggacin cmc A **108** by applying indium-mediated Barbier propargylation as one of

the central steps. Initially, 1-malic acid 95 was treated over a number of steps to afford C1–C12 fragment of thuggacin cmc-A 96. In order to access C13–C25 fragment, optically active oxazolidinone 97 was treated with sodium hexamethyldisilazide followed by reaction with  $BrCH_2C$ =CTMS and sodium borohydride mediated reduction to attain corresponding alcohol 98 in 97% yield. The alcohol 98 was then transformed to terminal alkyne 99 over few steps. The reaction of terminal alkyne 99 with (tributylstannyl)methylmagnesium (which was generated within the reaction) utilizing copper cyanide as catalyst proceeded by the addition of methyl group and iodine afforded iodoalkene 100 in 56% yield. Iodoalkene 100 was further made to undergo  $Pd(PPh_3)_4$  catalyzed Suzuki–Miyaura coupling with

Scheme 13 Synthesis of thuggacin cmc-A 108.

Boronic acid 101 along with subsequent (TPAP-promoted) oxidation to furnish aldehyde 102 in 85% yield. The aldehyde 102 was next subjected to react with aminoalcohol 103 and allybromide Singaram's developed indium promoted Barbier propargylation to afford the required compound 104 in 91% yield. The Barbier reaction involving propargylation resulted in excellent yield (91%) of a single required isomer in comparison to the addition of Grignard reagent which afforded the product in unsatisfactorily diastereomeric ratio. The hydroxyl group in compound 104 was further protected with tert-butyl silyl group to generate fragment C13-C25 105 in 100% yield. The fragment C1-C12 96 and C13-C25 105 were further coupled via Sonogashira coupling in the presence of copper iodide, palladium chloride using triethylamine as base to achieve alkyne 106 in 90% yield. The removal of TBS group in alkyne 106 was carried out via tert-butylammonium fluoride to generate -OH substituted alkyne 107, which was then reacted over several steps to give rise to thuggacin cmc-A 108 (Scheme 13).

Resorcylic acid lactones are natural occurring compounds, constituting 14-membered macrolides in their structural framework.88 Ever since the extraction of radiciol, this particular class of natural products has been investigated keenly. It has been found to demonstrate the efficient potency as antibacterial, anti-fungal and anti-viral agents. There is continuous progress towards the development of efficient antimicrobial agents.89,90 In 2012, Chen and colleagues isolated naturally occurring resorcyclic acid lactones i.e., Paecilomycins A-F. 91 In propagation of their strives towards the total synthesis of various Paecilomycins, Gurram and coworkers<sup>92</sup> performed the synthesis of Pecilomycin F by employing Barbier-type allylation as a key step. Initially, they treated epoxide 109 over several steps involving synthetic pathway to forge the isopropylidine based ester 110. The synthesized ester 110 was then reduced via DIBAL-H in dichloromethane proceeded by zinc mediated Barbier allylation reaction with allylbromide in the presence of ammonium chloride and tetrahydrofuran to afford non-separable diastereomeric mixture of homo-allylated

alcohol 111 in 78% yield with 10:1 dr. The hydroxyl functionality in compound 111 was protected on treatment with MOMCl and DIPEA in dichloromethane with subsequent removal of silyl group by using tetrabutyl ammonium fluoride to furnish the alcoholic fragment 112 in 83% yield. Furthermore, tri-hydroxy substituted benzoic acid 113 was reacted over a number of steps to result in required acid fragment 114. In the next step, synthesized alcoholic and acid fragments (112 & 114) were joined via Steglich esterification to forge ester, which was subjected to Grubbs II catalyst promoted ring closing metathesis to accomplish the synthesis of Paecilomycin F 115 (85% yield) (Scheme 14).

2.2.2. Synthesis of sesquiterpene lactones-based natural **products.** Sesquiterpene lactones are incorporated with  $\alpha$ -exomethylene-γ-butyrolactone which exhibit promising biological activities i.e., deoxyelephantopin, helenalin and arglabin.93 Structural framework of paraconic acids is also endowed with this bioactive skeleton. The previously employed allylation routes to access bioactive skeleton resulted in anti-relationship within substituents, thus another route was required to access the syn-relationship. For this purpose, considering the utility of palladium metal in organic synthesis, 94,95 palladium promoted Barbier-type allylation was employed by involving dimethyl zinc as reducing agent. In 2021, Liu et al.96 developed Barbier-type allylation with subsequent translactonization reaction by using Pd(PhCN)<sub>2</sub>Cl<sub>2</sub> as catalyst with triphenylphosphine ligand in the presence of dimethyl zinc (reducing agent) in THF/toluene to afford the diastereoselective mixture of  $\beta$ ,  $\gamma$ -disubstituted  $\alpha$ -exomethylene-γ-butyrolactones. Dimethyl zinc was observed to play crucial role in both steps by acting as a nucleophile (for polarity inversion of  $\pi$ -allylpalladium) and Lewis acid respectively. This developed approach was then exploited further to achieve the synthesis of two paraconic acids and two 1,10-secoguaianolides. In this regard, market ready tulipalin A 116 was transformed to bromolactone 117 in 60%. In the next step, bromolactone 117 was subjected to treat with hexanal 118 via developed dimethylzinc involving Barbier-type allylation and

Scheme 14 Synthesis of Paecilomycin F 115.

Scheme 15 Synthesis of  $(\pm)$ -methylenolactocin 120 and (+)-C75 123.

translactonization sequence in tetrahydrofuran to attain the resulting addition product **119** in 74% yield. The addition product was subjected to Jones oxidation to accomplish the synthesis of  $(\pm)$ -methylenolactocin **120** in 80% yield. In another route, bromolactone **117** was made to react with nonanal **121** via palladium promoted Barbier allylation and subsequent lactonization to furnish compound **122** in 76% yield. In the final step, Jones oxidation of compound **122** resulted in the synthesis of second paraconic acid *i.e.*, (+)-C75 **123** in 60% yield (Scheme 15).

Further Barbier-type allylation sequence was harnessed for the synthesis of 1,10-seco-guaianolides 3-deshydroxy-iso-secotanapartholide 128 and 1,10-dioxo-1,10-deoxy-1,10-secogorgonolide 129 which are extracted from *Achillea ligustica* and *Artemisia gorgonum* respectively. Initially, alkene 124 was treated over few steps to synthesize corresponding aldehyde 125, which was further treated with bromolactone 117 *via* developed Barbier-type allylation and translactonization sequence to acquire addition product 126 in 74% yield. The compound 126 was subjected to reduction and oxidation

Scheme 16 Synthesis of 1,10-seco-quaianolides 128 and 129.

followed by proline mediated aldol condensation and acylation to afford conjugated ketone 127 in 45% yield. The conjugated ketone 127 further resulted in the synthesis of mixture of three compounds 128–130 as a result of Stryker's reagent induced reduction (of  $\alpha,\beta$ -unsaturated double bond) followed by allylic oxidation with the addition of  $CrO_2(Ot\text{-Bu})_2$ . In this way, two naturally occurring 1,10-seco-guaianolides 128 and 129 were obtained in 35% and 4% yield respectively. Other non-natural compound 130 (from the resulting mixture) was subjected to similar allylic oxidation to transform it into naturally occurring 1,10-seco-guaianolide 129 (Scheme 16).

Cynaropicrin is a naturally occurring sesquiterpene lactone, obtained from *Cynara scolymus* (artichoke). This particular natural product is composed of complex structural framework, consisting of three fused cyclic rings, four exo double bonds with six stereogenic centers. Cynaropicrin has been discovered to play the role of anti-trypanosomal, NF- $\kappa$ B activation inhibitors and anti-hepatitis agents. In 2021, Nakamura *et al.* <sup>97</sup> carried out the first total synthesis of (+)-cynaropicrin **140** by employing Favorskii rearrangement and indium mediated diastereoselective Barbier reaction as major steps. Initially, easily accessible natural precursor *i.e.*, (S)- $\alpha$ -pinene **131** was made to undergo several steps to generate corresponding lactone **132**. Lactone **132** then underwent Luche reduction proceeded by hydroxyl group protection with p-methoxy benzyl to furnish

compound 133 in 88% yield. The ester functionality in compound 133 was reduced by utilizing lithium aluminium hydride followed by Dess-Martin periodinane mediated oxidation to procure aldehyde 134 in 91% yield. The aldehyde 134 was then made to react with compound 135 via indium catalyzed diastereoselective Barbier allylation reaction tetrahydrofuran/water at room temperature, which gave diastereomeric mixture of compound 136 in 92:8. It was observed that indium promoted Barbier allylation resulted in better yield with higher diastereomeric ratio as compared to zinc mediated Barbier approach. Compound 136 was then subjected to translactonization in step sequence with Swern oxidation to attain aldehyde 137. The aldehyde 137 was further converted to tricyclic compound 138 (in 77% yield) via intramolecular ene reaction succeeded by DBU promoted elimination of methanol. In the next step, Mitsunobu-Mukaiyama inversion reaction between compound 138 and acid 139 installed the side chain to fused tricyclic ring skeleton followed by TBAF and DDQ induced deprotection of hydroxyl group to furnish the desired natural product 140 in 94% yield (Scheme 17).

Sesquiterpenes are naturally occurring organic compounds which have been observed to display a significant array of biological activities. As, 15-deoxygoyazensolide has unveiled itself as potent cytotoxic agents in comparison to parent *i.e.*, goyazensolide, which indicates the improvement in pharmaceutical

Scheme 17 Synthesis of 1,10-seco-quaianolides 138 and 140.

activity on slight structural modifications.98 In 2021, Liu et al.99 reported the synthesis of naturally occurring sesquiterpenes by carrying out zinc promoted Barbier-type allylation as a key step. To achieve the required task, they initiated the synthetic pathway by treating vinyl alcohol 141 over few steps involving route to procure alkyne 142. Then, zinc mediated Barbier-type allylation was carried out by reacting aldehyde 143 with allyl bromide 144 in tetrahydrofuran/water using ammonium chloride followed by lactonization to give the substituted iodoalkene 145 in 64% yield. The utilization of Barbier reaction within allylation approach led towards the installation of  $\alpha$ -exo-methylene- $\gamma$ -butyrolactone. The synthesized alkene 145 further experienced serium oxide (SeO<sub>2</sub>) allylic oxidation to furnish vinyl iodide 146 in diastereomeric ratio of dr = 1:8.1 with 42% vield. In the next step, vinyl iodide was made to react with initially generated alkyne 142 via Sonogashira coupling reaction proceeded by reaction with PBr<sub>3</sub> in diethyl ether, which resulted

in the attainment of compound 147. The compound 147 was further converted into two diastereomers 148a and 148b on reaction with CrCl<sub>2</sub> in DMF. The diastereomer 148b acted as an essential synthetic intermediate to afford the total synthesis of various naturally occurring sesquiterpenes *i.e.*, atripliciolide 149 and tagitinic C 151 (*via* several steps involving synthetic pathway). Exposure of acidic conditions to tagitanin C 151 led to the synthesis of tagitanin F 152. Similarly, diastereomer 148a acted as an essential precursor to attain 15-deoxygoyazensolide 150 *via* several steps sequence, which was then subjected to Stryker's reagent to furnish eremantholide C 153 in 40% yield (Scheme 18).

A germacranolide sesquiterpene based lactone *i.e.*, cnicin was initially extracted by Šorm and colleagues in 1959, from the *Cnicus benedictus's* leaves. <sup>100</sup> Cnicin is structurally composed of a diene constrained in a ten-membered ring joined with a lactone ring (5-membered). This naturally occurring

Scheme 18 Synthesis of naturally occurring sesquiterpene lactones 149–153.

sesquiterpene lactone exhibits allelopathic, anti-myeloma, cytostatic, anti-bacterial and anti-trypanosoma brucei activities.101 Its striking pharmacological potential urged the Kimura and Usuki<sup>102</sup> (in 2022) to develop an efficient route to access cnicin, exploiting Barbier allylation as a key step. Their synthetic pathway involved the procurement of aldehyde 155 by subjecting alcohol 154 to several steps involving sequence. The synthesized aldehyde 155 was then joined with Evans aldol substrate 156 via 1,2-addition reaction in the presence of (n-Bu)2OTf and diisopropyl ethylamine in dichloromethane in sequence with sodium borohydride mediated reduction to afford the inseparable mixture of compounds 157 and 158. However, treatment of compound 157 with ethyl vinyl ether and PPTS resulted in the synthesis of pure alcoholic fragment 158 in 69% yield (via three steps). The alcoholic functionality in compound 158 was oxidized using Dess-Martin Periodinane followed by indium promoted Barbier allylation with 3-bromo-2-methyl propene, thereby yielding diastereomeric mixture of coupled product 159 in 66% yield (dr = 4:1). The resulting product 159 was treated over a number of steps (which also involve RCM) to attain the synthesis of cnicin analogue i.e., (1Z)deacylcnicin 160. The inability of RCM to furnish the 1E-form as major product hindered the total synthesis of cnicin, thus paved pathway to result in cnicin analogue (Scheme 19).

γ-Butyrolactones are abundantly found in the structural skeleton of diverse naturally occurring compounds. <sup>103</sup> These scaffolds have been observed to display a significant array of potent biological activities which include cytotoxic, antibacterial, anti-viral, anti-inflammatory and anti-fungal activities. <sup>104</sup> Guaianolides *i.e.*, (+)-ligustrin, (+)-8-*epi*-ligustrin, (+)-grosheimin, (+)-8-*epi*-grosheimin & (–)-eupalinilide are naturally occurring sesquiterpene lactones whose structural

constituents involve tricyclic ring system framework with ybutyrolactone and hydroazulene moieties. These natural products demonstrate remarkable potencies as biological agents. In 2022, Fernandes and Ramakarishna<sup>105</sup> carried out the total synthesis of these sesquiterpene lactones by employing diastereoselective Barbier allylation and translactonization as key steps. The first step of synthetic journey involved the treatment of allyl substituted chiral bromolactone 162 with aldehyde 161 via diastereoselective Barbier allylation in the presence of zinc in dimethylformamide to afford the diastereomeric inseparable mixture of three-stereocenters involving compound 163a and **163b** (dr = 1:7.5), which was subjected to column chromatography to obtain dr = 8:92 of the resulting mixture. The compound 163b was further translactonized followed by two steps involving aldehyde-ene reaction, which ultimately furnished the (+)-ligustrin 164 in 76% yield. (+)-ligustrin 164 was treated over few steps involving synthetic route to furnish the synthesis of (-)-eupalinilide E 165. Furthermore, (+)-ligustrin 164 was transformed to (+)-8-epi-ligustrin 166 (in 75% yield) via Mitsunobu conditions. Next, epoxidation of (+)-8-epi-ligustrin 166 was proceeded by ring opening rearrangement with boron trifluoride diethyl etherate to accomplish the synthesis of (+)-grosheimin 167 in 67% yield (Scheme 20).

#### 2.3. Synthesis of terpenoids-based natural products

2.3.1. Synthesis of sesquiterpenoids-based natural products. Toxicodenane A is a sesquiterpenoid which consists of three fused cyclic rings anlogwith exocyclic double bond and bridged ether functionality. In 2013, it was first withdrawn from *Toxicodendron vernicifluum* by Cheng and coworkers. <sup>106</sup> Racemic isolation of toxicodenane A urged the researchers to carry out its enantioselective synthesis, to evaluate its biological

Scheme 19 Synthesis of (1Z)-deacylcnicin 160.

Scheme 20 Synthesis of naturally occurring sesquiterpene lactone 164–167.

applications. In 2021, Han and colleagues<sup>107</sup> demonstrated the first enantioselective synthesis of toxicodenane A by carrying out Lewis acid catalyzed transacetalation and Prins cascade reaction. In 2022, Nishikawa et al. 108 reported an efficient route to achieve the enantioselective synthesis of toxicodenane A followed by the biological assessment of both enantiomers against lipotoxicity. They accomplished the asymmetric synthesis of toxicodenane A by implementing SmI2 mediated Barbier-type cyclization, allylic oxidation and dehydration induced cyclization as key steps. The synthesis began with the vinylation of silyl ether 168 by involving Pd(TFA)2 and Bphen (bathophenanthroline) followed by Claisen rearrangement to afford required aldehyde 170 in 96% yield. The resulting aldehyde 170 was then subjected to react with cyclohexanedione 171 and compound 172 via reductive Knoevenagel condensation which resulted in diastereomeric mixture of 173 in 82% yield (with dr = 63:37). The silyloxy group in compound 173 was then transformed to sulfonyl group 174, which on further asymmetric desymmetrization and protection of hydroxyl group with methoxymethyl gave enantiopure compound 177 in 94% yield. In the next step, compound 177 was subjected to samarium iodide (SmI2-HMPA) promoted Barbier-type annulation in tetrahydrofuran at -78 °C to afford compound 178 stereo selectively in 99% yield with cis-configuration. The cyclized product 178 was then treated over several steps to forge (+)-toxicodenane A 179. Similarly, the synthesis of (-)-toxicodenane A ent-179 was procured by treating compound 174 with boronic acid 175 and compound ent-176 via similar mentioned (Barbier-type annulation involving) protocol. Both the enantiomers were found to protect the cells against lipotoxicity promoted inflammatory effects (Scheme 21).

Oxaspirolactone constituting terpenoids are known to exhibit remarkable biological role as they are efficiently employed against several deadly diseases. 108,109 In 2015, Kubo et al. first time isolated the oxaspirolactone based tetranorsesquiterpenoids i.e., lanceolactone A and B from the Illicium lanceolatum's leaves. 110 These naturally occurring sequiterpenoids were observed to illustrate the potent antimicrobial activity against teeth pathogen i.e., Porphyromonas gingivalis. Nanda's group, in 2018,111 carried out the initial report on the asymmetric total synthesis of lanceolactone in 16.2% overall yield by employing various name reactions as key steps. Being interested in the chemistry of oxaspirolactones, in 2022, Borade and Kontham<sup>112</sup> established a facile synthetic route to achieve the synthesis of lanceolactone A and its possible stereoisomers by exploiting chiral precursors without installation of any protecting group. They accomplished the synthesis of target natural products by carrying out ozonolysis, addition reaction, gold-promoted isomerization and dye-sensitized photooxidation as significant steps. In the first step, (S)-linalool 180 was made to undergo ozonolysis in the presence of pyridine and DCM to afford lactol 181 in 63% yield, without incorporating any protecting group. The synthesized lactol 181 was then subjected to indium powder promoted Barbier-type addition reaction with propargyl bromide 182 in MeOH:NH<sub>4</sub>Cl, which gave intermediate i.e., allene-diol 183 in 79% yield. The resulting allene-diol 183 was further subjected to Dess-Martin Periodinane mediated oxidation followed by treatment with AuCl in tetrahydrofuran to synthesize hydroxyalkyl-joined-furan 184 in 83% yield. Compound 184 was further exposed to MB (Mitsunobu-Vassilikogiannakis's) dye-sensitized photo-oxidation reaction, which

Scheme 21 Synthesis of (-)-toxicodenane A 179 and (-)-toxicodenane A ent-179.

afforded lanceolactone A 185a and its epimer 185b in 43% and 21% yield respectively (Scheme 22).

Aculenes A-E are affiliated with bicyclic norsesquiterpene based L-proline associated keto esters or alcohols, which were extracted from Aspergillus aculeatus culture media and from Penicillium sp. XWS02F62 and SCS-KFD08.113 These norsesquiterpenes are fungal metabolites endowed with nordaucane framework. Among them, aculenes C, D and E perform the

Scheme 22 Synthesis of lanceolactone A 185a and 4-epi-lanceolactone 185b.

inhibition of bacterial communication (quorum sensing restrictive potency) opposed to Chromobacterium violaceum CV026 bacterium.114 Thus, being persuaded by the efficient pharmacological role of aculenes, Yokokawa et al. 115 in 2023, reported a significant pathway leading towards the synthesis of aculene B and D. Their synthetic journey involved the utilization of zinc promoted Barbier allylation as one of the essential steps. The synthesis commenced with the transformation of easily available precursor *i.e.*, carboxylic acid **186** into  $\alpha$ , $\beta$ -unsaturated ester 187 via several steps. In the next step, ketone functionality in compound 187 was subjected to acetalization reaction followed by the treatment with tetra ammonium fluoride, thereby giving acetal 189 in 82% yield. The obtained acetal was then subjected to undergo Swern oxidation to afford aldehyde 190. The aldehyde 190 was then made to undergo zinc-mediated Barbier allylation reaction with allylbromide 191, thereby giving the required diastereomer 192 in major quantity (66% yield) on subsequent removal of acetal group via acidic hydrolysis. The diastereomer 192 was made to react over a number of steps to furnish the naturally occurring aculene D 193. Aculene D 193 was treated further to acquire aculene B 195 in 73% yield as a result of its Steglich esterification with Boc protected Lproline 194 proceeded by the deprotection of amine (Scheme 23).

2.3.2. Synthesis of diterpenoids-based natural products. Havellockate belongs to the cembranoid family, which was first obtained from *Sinularia granosa*. Its structure is composed of fused tricyclic ring framework with  $\beta$ -hydroxybutanolide ring constituting overall eight chiral centers. Till now, no successful synthesis of havellockate has been presented. Considering the challenges towards its total synthesis, in 2022,

Hafeman et al. 117 performed the total synthesis of havellockate by utilizing Julia-Kocienski olefination, Diels-Alder cascade reaction, zinc catalyzed Barbier allylation alongwith copper mediated aerobic oxidation as major steps. The synthesis kickstarted with the synthesis of required aldehyde 197 and sulfone 199 individually by treating enone 196 and acyl oxazolidinone 198 respectively over few steps. The aldehyde 197 and sulfone 199 were then coupled via KHMDS mediated Julia-Kocienski olefination followed by desilylation to afford diol 200 in 57% yield with >20:1 E:Z. The resulting diol **200** was further made to underwent Steglich esterification with propiolic acid followed by [4 + 2] cycloaddition reaction to furnish Diels-Alder adduct 201 in 51% yield with >20:1 dr. The adduct 201 was later treated over several steps to synthesize enone 202, which was subjected to transesterification reaction to result in ester 203 in 70% yield. The ester 203 further underwent zinc promoted Barbier allylation reaction with substituted allyl bromide 204 in tetrahydrofuran followed by subsequent deacetylation to forge triol 205 in 25% with dr = 1:1. The triol compound 205 was later transformed to tetrol 206 in 34% yield as a result of iridium promoted anti-Markonikov hyrdrosilylation alongwith Tamao-Fleming oxidation. In the final step, synthesis of havellockate 207 was achieved in 80% by treating compound 206 via copper mediated (aerobic) oxidation/cyclization reaction (Scheme 24).

Synthesis of naturally occurring organic compounds have taken frontseat owing to their applicability as efficient drugs. Makassaric acid and fascioquinol B are meroditerpenoids obtained from sponges *i.e.*, *Acanthodendrilla* and *Fasciospongia* sp. respectively. Makassaric acid acts an efficient inhibitor of MAPKAP kinase 2 (MK2) enzyme, 118 thus acting as potent anti-inflammatory agent. Similarly, fascioquinol B has been known

Scheme 23 Synthesis of aculene D 193 aculene B TFA 195.

Scheme 24 Synthesis of havellockate 207.

to exhibit remarkable anti-bacterial potency. In 2022, Rosales  $et\ al.^{119}$  presented a synthetic layout for the accomplishment of these naturally occurring meroterpenoids. For the synthesis of Makassaric acid derivative, initially, Grignard derivative 208 was added to isoprene monoxide by exploiting the catalytic efficiency of CuBr DMS, to result in 92% of allylic alcohol 210. The resulting alcohol was then converted to aldehyde 211 via oxidation using Dess–Martin periodinane. The synthesized aldehyde 211 was further subjected to exposed to compound 212 via Cp<sub>2</sub>TiCl mediated Barbier-allylation reaction, which furnished polyene 213 in 51% yield. This catalytic strategy is essential for the formation of diversely substituted integrated polyenes. The hydroxyl group in ploynene 213 was acetylated to afford compound 214 in 99% yield, which was ultimately treated over several steps to procure the naturally occurring

makassaric acid's derivative **215**. The similar strategy was adopted by treating 2-bromo-1,4-dimethoxybenzene as a precursor, which followed the above synthetic protocol to furnish the synthesis of a close derivative of fascioquinol B **216** (Scheme 25).

The most enormous class of plants based secondary metabolites generally constitute diterpenoids (subclass of terpenoids). C20 diterpenoids have been structurally classified into same main categories which include *ent*-beyerane, *ent*-kaurane, *ent*-trachylobane, *ent*-gibberellane and *ent*-atisane *etc*. The main reason for the origin of these C8-ethano bridging diterpenoids is the series of cyclization reaction of geranylgeranyl pyrophosphate which gives rise to *ent*-beyeranyl cation I. This cation plays a key role in enzyme-mediated conversions to synthesize a variety of diterpenoids. Taking into consideration

Scheme 25 Synthesis of makassaric acid's derivative 215 & fascioquinol derivative 216

the medicinal applications and structural diversity of these diterpenoids, in 2023, Fan group122 unveiled a unified strategy by exploiting late stage conversions of ent-trachylobane and entkaurane scaffolds to achieve the synthesis of 9 ethano-bridged diterpenoids. The synthetic pathway commenced with the preparation of corresponding aldehyde 218 from optically active cyclohexenone 217. Meanwhile, Appel substitution reaction of γ-cyclogeraniol 219 afforded optically active allylic bromide 220 in 85% yield. The two resulting monoterpene fragments were then subjected to lithium-mediated Barbier allylation reaction by utilizing lithium/naphthalene in tetrahydrofuran followed by acidic exposure to furnish the diastereomeric mixture of 221a and 221b in about 1:1 diastereoselectivity ratio. Next, enonols 223 $\alpha$  and 223 $\beta$  were individually attained as a result of MnO<sub>2</sub> promoted allylic oxidation of 221a and 221b respectively. The earlier obtained diastereomeric mixture (221a and 221b) was also exposed to IBX and Dess-Martin periodinane mediated oxidation to procure 91% yield of keto-enone 222. Enonols 223α and  $223\beta$  were individually subjected to optimized hydrogen atom transfer cyclization (MHAT) to result in the generation of trans,trans-C2 224a and tran,trans-C1 224b along with cis,cis-C1 224c. These precursors were obtained in 36% yield with 75% brsm affording a diastereomeric ratio of 7:1. The synthesized trans,trans C2 224a was treated over several steps thereby furnishing the synthesis of atisirene 226, which was subjected

to Muakiyama hydration reaction condition to result in *ent*-atisan-16apha-ol 227 in 76% yield. Similarly, compound 224a was transformed into *epi*-canadol A 225a within 71% yield (*ent*-kaurane type diterpenoid) *via* Wolff–Kishner–Huang reaction conditions. Similarly, the *trans,trans*-C1 224b was transformed into sicanadiol 228 over several steps. This diastereomer 224b resulted in the synthesis of canadol A 225b (in 69% yield) *via* Wolff–Kishner–Huang reduction. The resulting canadol A 225b underwent cobalt mediated Mukaiyama hydration to accomplish the synthesis of powerol 229 in more than 20:1 diastereoselective ratio with 71% yield (Scheme 26).

The initially synthezsized keto-enone **222** was also subjected to MHAT cyclization to give rise to *trans,trans*-C3-CP **230**, which was treated over numerous steps to achieve naturally occurring trachinol **231a** and *epi*-trachinol **231b**. Similarly, Wolff–Kishner reduction of *trans,trans*-C3-CP **230** generated *ent*-beyerane **232** in 62% yield (Scheme 27).

#### 2.4. Synthesis of lignans-based natural products

Phrymarolin II is a 3,7-dioxabicyclooctane framework featuring furofuran lignan, which has been extracted from renowned chinese' traditional medicinal plant *i.e.*, *Phryma leptostachya* L.<sup>123</sup> This plant's roots have been determined to depict the angiogenesis, pesticidal, anti-bacterial and anti-inflammatory

Scheme 26 Synthesis of C20 diterpenoids 224–229.

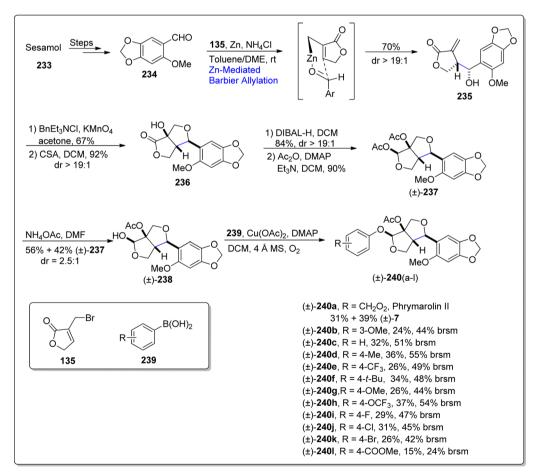
activities. There have been several reports concerning the total synthesis of phrymarolin II and its derivatives in literature, however they were observed to usually suffer from non-economic and low-yielding outcomes. Thus, in order to establish more convenient pathway for the synthesis of naturally occurring lignan, Chi and coworkers in 2021, 124 endeavored an efficient synthetic pathway to access phrymarolin II and its analogues. They utilized the market-ready precursor *i.e.*, sesamol to

accomplish the synthesis of this natural product (with overall 5.7% yield) in total nine steps, exploiting Barbier allylation reaction and anomeric O-arylation as significant steps. Initially, sesamol 233 was subjected to react over reported two steps to afford carbaldehyde 234. The resulting carbaldehyde was then made to react with 3-bromomethyl substituted furanone 135 via zinc promoted Barbier allylation reaction with the addition of ammonium chloride in toluene/dimethylether, thus procuring compound 235 in 70% yield with more than 19:1 diastereomeric ratio. The Barbier allylated product 235 was then exposed to dihydroxylation conditions proceeded by camphor sulfonic acid (CSA) mediated cyclization to forge tetrahydrofuran 236 in 92% yield. In the ongoing step, tetrahydrofuran was subjected to DIBAL-H promoted reduction in sequence with treatment of resulting lactol with acetic anhydride using dimethyl aminopyridine and triethylamine in dichloromethane, thus giving diacetate 237 in 90% yield. Further, deacetylation reaction by employing ammonium acetate resulted in diastereomeric mixture of compound 238 and starting diacetate 237 in 2.5:1 dr. Finally, compound 238 was further made to undergo copper acetate promoted Chan-Lam-Evan cross coupling reaction with substituted aryl boronic acids 239 to afford the total synthesis of phrymarolin II 240 and its various analogues. The in vivo studies interpret the good to excellent potential of these analogues against the activity of tobbaco mosaic virus, in comparison with ningnanmycin (Scheme 28).

Sacidmulignans A-D are naturally occurring compounds, which have been extracted from the Sarcostemma acidum (Roxb.) located in the China's Hainan island. 125 Efforts towards the total synthesis of sacidmulignans A-C (which constitute of 2,7' cyclolignans structurally) have gained significant momentum in synthetic organic chemistry. 126 Attempts towards the synthesis of sacidmulignan B have been reported by Ramana and Peng group individually, which resulted in 10% and 3% overall yield of this natural product respectively. 127,128 In 2022, Zhuang et al. 129 carried out the first racemic total synthesis of sacidmulignan B by exploiting zinc promoted Barbier-type allylation as a key step. Initially, diaryl ketone 241 was subjected to react with crotyl bromide 242 via zinc mediated Barbier-type allylation reaction in tetrahydrofuran to furnish the lactone 243 in 84% yield. The resulting lactone 243 experienced two reduction reactions consecutively followed by the selective protection of hydroxyl group to acquire monoalcohol 244 in 90% yield. The resulting tertiary alcohol 244 was subjected to Et<sub>3</sub>SiH and boron trifluoride diethyl ether assisted reduction conditions proceeded by the deprotection conditions leading to the removal of tert-butyldiphenylsilyl group, which afforded primary alcohol 245 (in 90% yield). The exposure of IBX promoted oxidation conditions to generated alcohol 245 gave the corresponding aldehyde 246 in 88% yield. The compound 246 was further subjected to hydrogenation to deprotect the benzyl protected alcoholic functionality proceeded by the p-TsOH mediated (Friedel-Crafts) cyclization, thus procuring sacidmulignan B 247 in 85% yield (Scheme 29).

Tobacco mosaic virus is one the highly afflicting disease of plants, which is responsible for the detrimental and hazardous effects on almost 125 plant species.<sup>130</sup> Considering the

Scheme 27 Synthesis of C20 diterpenoids 231a, 231b and 232.



Scheme 28 Synthesis of phrymarolin II 240 and its various analogues.

availability of very few anti-viral agents such as ningnanmycin, ribavirin and dufulin, there is a constant urge to discover and develop efficient anti-viral compounds. Naturally occurring

norlignan *i.e.*, nicotlactone A, which was extracted from *Nicotiana tabacum*, identified by Hu and colleagues.<sup>131</sup> The anti-viral potency of this naturally occurring product was unveiled to be

Scheme 29 Synthesis of sacidmulignan 247.

more efficient as compared to commercially utilized standard i.e., ningnanmycin. These results indicate the capability of nicotlactone A as efficacious lead compound against tobacco mosaic virus. In 2022, He  $et~al.^{132}$  procured the synthesis of natural product  $(\pm)$ -8-demethylnicotlactone A and its analogues by utilizing zinc mediated Barbier-type allylation reaction. The substituted benzaldehydes 248 were treated with methyl-2-(bromoethyl)acrylate via zinc promoted Barbier-type allylation reaction in tetrahydrofuran and ammonium chloride to afford

the corresponding lactones 249. The synthetic pathway was then proceeded by the  $Mn(dpm)_3$  promoted hydration of synthesized lactones to attain the diastereomeric mixture of compound 250a & 250b in 1:1 diastereomeric ratio (Scheme 30).

#### 2.5. Synthesis of miscellaneous natural products

Various natural products can be synthesized in a feasible manner by employing carbohydrates as precursors.

OHC 248 Methyl-2-(bromomethyl)acryate Zn, THF/NH<sub>4</sub>Cl Zn-Mediated Barbier-type Allylation 
$$\frac{\text{Mn}(\text{dpm})_3, \text{PhSiH}_3, \text{ O}_2}{\text{EtOH, DCM, dr} = 1:1} + \text{HO} + \text{P}^1$$

$$R^1, R^2 = \text{OCH}_2\text{O}, \text{OMe}$$

$$R^1 = \text{H, R}^2 = \text{OMe}$$

$$R^1 = \text{OMe, R}^2 = \text{H}$$

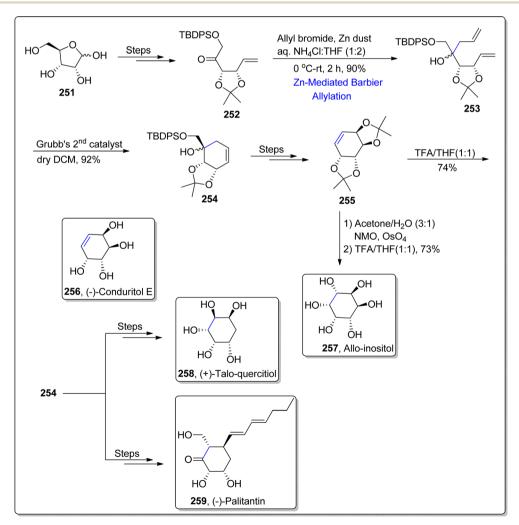
$$R^1 + \text{R}^2 = \text{OCH}_2\text{O}$$

Scheme 30 Synthesis of  $(\pm)$ -8-demethylnicotlactone 250a and its analogues.

Carbohydrates endowed with several hydroxyl group functionalities bear significant weightage owing to their enhanced water solubility and bio-availability. Synthesis of several natural products by employing a common intermediate is an efficient divergent route, exploited widely in past. On the similar ground, Sivakarishna et al. 133 in 2022, carried out the total synthesis of various polyhydroxylated natural products by employing an efficient divergent synthetic route. They demonstrated the synthesis of common intermediate by implementing Barbier allylation and ring closing metathesis as major steps. At first, p-Ribose was transformed to ketone 251 over several steps involving synthetic pathway. In the next step, the resulting ketone 252 was made to react with allylbromide via zinc dust promoted Barbier allylation in ammonium chloride:tetrahydrofuran (1:2) at 0 °C to room temperature, thereby furnishing the allylation product i.e., diene 253 in 90% yield. The zinc metal was exploited within Babier reaction considering its inexpensive feature. The diene 253 was further subjected to ring closing metathesis (RCM) in the presence of Grubb's 2nd generation catalyst in dichloromethane to afford cyclohexenyl derivative 254 in (inseparable) diastereomeric mixture (92%).

The cyclohexenyl derivative **254** was then utilized as a common intermediate to accomplish the synthesis of various natural products. This common intermediate **254** was treated over several steps to generate alkene **255**. The alkene **255** was made to undergo *syn*-dihydroxylation *via* NMO and osmium tetroxide followed by subsequent trifluoroacetic acid mediated acid hydrolysis, which ultimately resulted in the synthesis of alloinositol **257** in 73% yield. On the other hand, conduritol E **256** was easily accessed in 74% yield (overall 10% yield) *via* acid hydrolysis of alkene **255**. The cyclohexenyl derivative **254** was also utilized to attain the asymmetric synthesis of (—)-palitantin **259** and (+)-talo-quercitol **258** (Scheme 31).

Azimic acid is a trisubstituted piperidine constituting natural product, extracted from the *Azima tetrucuntha* Lam. This hydrolyzed form of azimine alkaloid has been observed to manifest promising medicinal applications. In 2023, Gharpure *et al.*<sup>134</sup> developed a novel approach by subjecting enynyl amines to Lewis acid *i.e.*, TMSOTf (5/6-*endo*-dig) reductive hydroamination in the absence of metal to access piperidne and pyrrolidine based natural products. They conducted the formal synthesis of naturally occurring (+)-azimic acid by employing



Scheme 31 Synthesis of (-)-conduritol E 256, allo-inositol 257, (-)-palitantin 259 and (+)-talo-quercitol 258.

Scheme 32 Synthesis of (+)-azimic acid 265.

the beneficial aspects of the developed 5/6-endo-dig reductive hydroamination and zinc mediated Barbier reaction as key steps. In the first step, aldehyde 259 was made to undergo zinc promoted Barbier propargylation reaction with alkynyl bromide 260 in the presence of ammonium chloride and tetrahydrofuran:water to give alkynylamine 261 in 80% yield (dr = 1:1). The alcoholic functionality in compound 261 was then acylated with acetic anhydride followed by the Sonogashira coupling of resulting alkyne with vinyliodide 262 to give enynylamine 263 in 60% yield (syn isomer = 20%). In the next step, generated enynylamine 263 was made to undergo developed reductive hydroamination reaction by using TMSOTf, Et<sub>3</sub>SiH in DCM to furnish diastereomeric mixture (dr = 1.5:1) of piperidine derivative in 63% yield. Enynylamine 263 was subjected to react with TMSOTf and Et<sub>3</sub>SiH in dichloromethane followed by potassium hydroxide involving hydrolysis in methanol to furnish piperidine derivative 264. The removal of tosyl group from piperidine derivative 264 was carried out in the presence of magnesium in methanol to result in azimic acid 265 (Scheme 32).

Ambergis is a naturally occurring compound, generally produced within the digestive cavity of sperms whales. It is known to constitute (–)-cis- $\alpha$ -ambrinol, which is a naturally occurring product and is highly acclaimed in perfumes

industry. In 2023, Martinez *et al.*<sup>135</sup> reported the two pathways for the facile generation of (–)-cis- $\alpha$ -ambrinol **269a** (in 35% over all yield) from  $\alpha$ -ionone/ $\alpha$ -dihydroionone. In pathway A, marketready  $\alpha$ -ionone **266** was transformed to dihydroionone *via* reduction over raney-nickle, followed by *m*-CPBA mediated oxidation to afford the epoxide containing oxidized compound **267**. Epoxide ring opening reactions result in the synthesis of diverse and useful organic compounds. <sup>136–138</sup> The epoxide ring in compound **267** was treated with acid followed by the formation of carbonate on reaction with ethyl chloroformate. The synthesized carbonate **268** was subjected to Cp<sub>2</sub>TiCl<sub>2</sub> mediated intramolecular Barbier allylation reaction to result in *cis-* $\alpha$ -ambrinol **269a** alongwith *trans-* $\alpha$ -ambrinol **269b** and compound **270** (Scheme 33).

The second synthetic strategy for the synthesis of cis- $\alpha$ -ambrinol **269a** involved the chlorination of  $\alpha$ -dihydroionone **271** with the addition of sodium perchlorate and phosphoric acid, thereby yielding allylic chloride **272** in 96% yield. The synthesized allylic chloride **272** was then exposed to CpTiCl<sub>2</sub> (generated within the reaction as a result of reduction of CpTiCl<sub>3</sub> with manganese) promoted intramolecular Barbier allylation in the presence of trimethyl silylbromide to afford the separable diastereomeric mixture of cis- $\alpha$ -ambrinol **269a** and trans- $\alpha$ -ambrinol **269b** in 2.3:1. (46% overall yield) (Scheme 34).

Scheme 33 Synthesis of (-)-cis- $\alpha$ -Ambrinol 269a.

Scheme 34 Synthesis of (–)-cis- $\alpha$ -ambrinol 269a.

Scheme 35 Synthesis of diospongin B analogue 277.

Diospongin B, a disubstituted tetrahydropyran (THP) based natural product, was isolated from Dioscorea spongiosa's rhizomes by Kadota and colleagues.139 It has been known to demonstrate anti-osteoporotic activity. Recently, in 2024, Fernandes et al.140 attempted the synthesis of Diospongin B analogue by utilizing Barbier reaction and Luche reduction as significant steps. Their synthetic route initiated with the preparation of alkynol 274 by treating alkenol 273 over few steps. The synthesized alkynol 274 was then added to 4-chlorobenzaldeyde with subsequent oxidation reaction and deprotection conditions to yield 2,3-dihydro-4H-pyran-4-one 275 in 85% yield. The hydroxyl group in tetrahydropyan compound 275 was subjected to oxidation proceeded by zinc mediated Barbier allylation reaction with allylbromide in tetrahydrofuran to afford diastereomeric mixture of corresponding alcohols. The Barbier product was further oxidized to afford diketone 276 in 74% yield. The diketone 276 was then transformed into diospongin B analogue 277 (in 90% yield) via Luche reduction in sequence with acidic hydrolysis (Scheme 35).

#### 3 Conclusion

To conclude, a summary of recent applications of Barbier reaction and its variants to achieve a variety of natural products have been provided in this review. Barbier reaction is one of the renowned C–C bond forming reactions, which possibly results in the formation of optically active center, leading to the

stereoselectivity in organic synthesis. This reaction has found several applications in synthetic organic chemistry, most significantly towards the total synthesis of a variety of natural products. In this regard, synthetic layouts of several alkaloids, lactones, terpenoids and lignans *etc.*, harnessing Barbier reaction as one the significant steps, have been overviewed here, documented within 2020–2024. For the synthesis of each natural product, utility of corresponding Barbier methodology has also been highlighted. We envision that this communication will stimulate the synthetic chemists to further delve within the Barbier reaction to unveil its potential beneficiary roles in organic synthesis.

# Data availability

No new data has been generated and all data is contained in the manuscript.

#### Author contributions

A. M. wrote the manuscript. M. N. A., S. G. K., N. A. and U. N. did literature survey and data analysis. A. M., H. A., A. R. C. and A. I. did data analysis and acquired funding. A. F. Z. supervised the project. All authors reviewed and edited the manuscript.

#### Conflicts of interest

There are no conflicts to declare.

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