

ORIGINAL RESEARCH

Potential Allylpyrocatechol Derivatives as Antibacterial Agent Against Oral Pathogen of S. sanguinis ATCC 10,556 and as Inhibitor of MurA Enzymes: in vitro and in silico Study

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Purpose: The study aims to identify antibacterial constituents of *P. betle* L. and evaluate their activities through two different methods including in vitro and in silico analysis.

Materials and Methods: The antibacterial agent was purified by bioactivity-guided isolation with combination chromatography methods and the chemical structure was determined by spectroscopic methods. The in vitro antibacterial activity was evaluated by disc diffusion and dilution methods while the in silico study of a compound binds on the MurA was determined using PyRx program.

Results: The antibacterial compound identified as allylpyrocatechol showed inhibitory activity against *S. sanguinis* with an inhibition zone of 11.85 mm at 1%, together with MIC and MBC values of 39.1 and 78.1 μ g/mL, respectively. Prediction for molecular inhibition mechanism of allylpyrocatechols against the MurA presented two allylpyrocatechol derivatives showing binding activity of -5.4, stronger than fosfomycin as a reference with the binding activity of -4.6.

Conclusion: Two allylpyrocatechol derivatives were predicted to have a good potency as a novel natural antibacterial agent against *S. sanguinis* through blocking MurA activity that causes disruption of bacterial cell wall.

Keywords: Piper betle L., allylpyrocatechol, Streptococcus sanguinis, MurA enzyme

Introduction

The low awareness of oral health can cause interference with the human health body. More than 750 bacteria are living in the oral environment and uncontrolled growth of bacteria to form the plaque, then it causes some oral diseases like caries,

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gingivitis, and periodontitis.² One of the original bacteria in the oral cavity of humans is *S. sanguinis* those important as key pioneering bacteria to form biofilms on the tooth surface called dental plaque.^{3–5} The bacteria will be receptor-anchoring salivary glycoprotein, which used to inhibit bacterial growth. The colonies of *S. sanguinis* can form a new environment to be localization other microorganisms, resulting in pathogenesis.^{6,7}

Currently, some antibiotics used as treatment stages to inhibit the bacteria, but it does not work completely, especially for pathogenic bacteria. Some antibiotics make an abnormal environment that causes stress conditions to bacteria, then induce cell death. The stressful condition influences evolution with genetic modification, known as resistance to antibiotics. 10

There has been a rising interest in the alternative for mechanical plaque removal with antibacterial agents, it can inhibit bacterial growth without genetic modification and become an effective and efficient treatment. One of the options is the disruption of the bacteria cell wall by inhibiting peptidoglycan biosynthesis. Many enzymes contributing to peptidoglycan biosynthesis include MurA (UDP-N-acetylglucosamine enolpyruvyl transferase). This enzyme catalyzes transfer enolpyruvyl to UDP-N-acetylglucosamine. ^{11,12} By blocking the action of MurA enzyme, bacteria cell wall will disrupt, and finally, bacteria will die automatically. This makes the MurA enzyme a promising target for antimicrobial agent. ^{13,14}

Nowadays, the research for drug discovery programs by many scientists used natural product sources to search lead compounds of new antibacterial agents. 15,16 Some antibacterial compounds against S. sanguinis have been found and reported from natural products, as kaurane diterpene from A. foliacea, 17 eugenol from E. carvophyllata, 18 berberine from B. vulgaris, 19 sabinene from C. japonica, 20 and β caryophyllene from R. officinalis. 21 Another edible medicinal plant that potency has antibacterial properties is a Plant from Piper genus which is Piper betle L.²² The ethyl acetate extract of this plant showed the highest antibacterial activity against Streptococcus pyogenes, Streptococcus aureus, Porphyromonas vulgaris, Escherichia and respectively.²³ Other paper reported antibacterial compounds were isolated from P. betle L. identified as phenolic compounds derivatives of eugenol, hydroxyl chavicol, and chavibetol showed active against S. marcescens, K. pneumonia, E. faecalis and B. subtilis, respectively, 24 while the active antibacterial constituents of P. betle L. against S. sanguinis have no reported.

As well as discovering new lead active compounds is an important point, the structure–activity relationship (SAR) study is another critical aspect to determine the drugs candidates are selective, specific, and effective to a receptor of target diseases. To determine and evaluates the prospective molecular mechanism aspects of new active compounds can be conducted with in silico analysis by predicting ligand conformation and the site binding targets.²⁵ This step will complement the in vitro and in vivo assay data, thus speeding up the process of drug discovery.²⁶

In this study, the research is focused on determining active constituents of *P. betle* L. as lead antibacterial compounds, and then it will be evaluated for their activities as an antibacterial against pathogenic oral bacteria *S. sanguinis* together with a prediction of the mechanism of action of antibacterial compounds as an inhibitor of MurA enzyme will be predicted by in silico studies.

Materials and Methods Materials

The leaf of P. betle L. was cultivated in March 2019, from a local farmer in Bandung, West Java – Indonesia. The specimen was identified and deposit at the Laboratory of Taxonomy, Department of Biology, Universitas Padjadjaran, Indonesia. The chemicals for extraction and purification were used distilled organic solvent of methanol, n-hexane, ethyl acetate, acetone, ethanol, and distilled water, while the chemicals for spectroscopic analyses were used pro-analyzed (p.a) grades. The column chromatography separation was carried out on Silica G 60 (0.063-0.200 mm and 0.200-0.500 mm) and ODS RP-18 (0.040-0.063 mm), and on Silica G 60 F₂₅₄ and ODS RP-18 F₂₅₄S for TLC together with 10% H₂SO₄ (v/v) in ethanol for chemical identification analysis.

The bacteria *Streptococcus sanguinis* ATCC 10566 used for antibacterial test on Muller Hinton broth and Muller Hinton agar as a medium, chlorhexidine as a positive control, and anaerobic jar for antibacterial assay (purchased from Merck Co. Ltd. and Sigma Aldrich).

The material for in silico, the 3D structure of the MurA used in this study was obtained from the Protein Data Bank (PDB) ID: 1UAE.²⁷ The receptor used is a protein (enzyme) UDP-*N*-acetylglucosamine enolpyruvyl transferase (MurA) from UniProt knowledgebase (http://www.uniprot.org/): UniProtKB – P0A749. Structure 3D MurA obtained using the RSCB program (https://www.rcsb.org) with 1UAE format

PDB. The tested ligand is Allylpyrocatechol 1. (3-prop-2-enylbenzene-1,2-diol) and its stereoisomer of Allylpyrocatechol 2 (4-prop-2-enylbenzene-1,2-diol). Both compounds were retrieved from PubChem with ID compounds that are ID 70775 (1) and ID 292101 (2). Ligand as positive control is fosfomycin (ID 446987). All this data was taken from the PubChem compound database (https://www.ncbi.nlm.nih.gov/pccompound).

Instruments

The structure of active compounds was determined by spectroscopic methods of ultra-violet (UV) by 8452A Diode Array, infra-red (IR) with FTIR Shimadzu 8400, NMR (1 H-NMR, 13 C-NMR, DEPT 135°, HMQC, 1 H- 1 H COSY, HMBC) with JEOL type ECA 500 MHz, and mass spectrometry (MS) with Water Acquit UPLC type triquadrupole. The TLC plates visualized by UV detector lamps with wavelengths of λ_{max} at 254 and 365 nm. Antibacterial activity assay used microplate 96 well, micropipettes, microtubes, incubators, paper disks, and Biochrom microplate readers.

Isolation Compound from Extract of *P. betle* L

The freshly *P. betle* L. leaf (0.58 kg) was extracted with methanol and then was subsequently partitioned between *n*-hexane-water and water-ethyl acetate, to get extracts of methanol (32.82 g), *n*-hexane (14.39 g), and ethyl acetate (1.08 g) and water (10.03 g), respectively. All extract adjusted in a series concentration for antibacterial activity assay and the most active fraction was further separated.

The active ethyl acetate extract (0.62 g) was gradually chromatographed on Silica G 60 (0.063–0.200 mm) eluted with n-hexane, ethyl acetate, and methanol with gradient 5% (v/v) to yield 21 fractions. All fractions were tested by antibacterial assay to determine which fraction the most active is. The most active fraction is fraction 1 (0.2353 g) so it was further purified subsequently by chromatographed on ODS RP-18 eluted with H_2O -MeOH with a gradient of 5% (v/v) and on Silica G 60 eluted n-hexane-EtOAc of gradient 1% (v/v) to yielded compound 1 (18 mg).

Structure Determination of Active Compound I

The structure of active compounds was determined by comprehensive analysis data of spectroscopic methods including ultra-violet (UV) spectrum, infra-red (IR) spectrum, 1D and 2D-NMR spectra (¹H-NMR, ¹³C-NMR, DEPT 135°,

HMQC, ¹H-¹H COSY, HMBC) and mass spectrometry (MS) spectrum. The Original spectra of UV, FTIR, NMR and MS can be seen in the Supplement material section.

Evaluation for the Antibacterial Activity from the Extract and Active Compound of *P. betle* L. leaf

Antibacterial effects of betle (P. betle L.) extracts evaluated against Streptococcus sanguinis ATCC 10566 and were observed using Kirby-Bauer disk diffusion. The assay procedure of the sensitivity or resistance of S. sanguinis to compounds test according to CLSI protocols (CLSI, 2012).²⁸ The extracts of *P. betle* were adjusted in series concentrations of 2%, 4%, 6%, 8%, and 10% while compound 1 was diluted in various concentrations of 1-5% for antibacterial evaluations against S. sanguinis. Methanol and water were used as negative control while chlorhexidine 2% as a positive control. On the other hand, bacteria were prepared by growing 1 ose of bacteria in 5 mL of broth media. This solution was incubated for 24 h at 37°C. After incubation, the optical density of the solution was measured using a Microplate reader at 620 nm. This solution was diluted by broth media until making 0.5 Mc Farland of bacteria solution. This culture (100 µL) was swabbed onto the agar media. Then, paper discs (6 mm) were impregnated with 20 µL of each sample and then placed on the surface of the agar contained bacteria. Furthermore, the samples were incubated for 48 h at 37°C. These tests were performed in duplicate. Inhibitions zone value (in mm) was measured after the incubation.

The MIC and MBC evaluation of Allylpyrocatechol 1 against S. sanguinis ATCC 10566 were determined by the micro-dilution method in 96-well microplate.²⁹ The bacterial cells were pre-cultured in Muller Hinton broth at 37° C under aerobic conditions and incubated in the presence of compounds with the concentrations obtained by serial two-fold dilution at 37°C without shaking in the same broth for 24 h on microplate wells. The optical density of the solution in microplate was measured using a microplate at 620 nm. The MICs were estimated as the lowest concentrations where the bacterial cells were observed by OD value. Then, each solution in the well was spread on the surface of the agar and incubated for 24 h at 37°C. The minimum concentration of sample that is no bacteria growth under colony counter is determined as MBC value.

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In silico Characterization of the Allylpyrocatechol Compounds

Characteristics of allylpirocatechol were confirmed using an online program. The chemical structure of allylpyrocatechol was drawn using PubChem (https://pubchem.ncbi.nlm.nih.gov/compound/292101) to obtain Canonical SMILES (C = CCC1 = C (C (= CC = C1) O) O). Canonical SMILES is used to convert chemical structures into 3D using the OPEN BABEL 2.4.2 program in PDB format, and the Structure 3D of UDP-N-acetylglucosamine enolpyruvyl transferase (MurA) was obtained using the RSCB program (https://www.rcsb.org/structure/1UAE) PDB format.

Virtual screening and docking of ligand-protein used Autodock Vina in the open-source PyRx 0.8 software. Allylpyrocatechols 1 and 2 as a ligand were used to bind MurA as a protein target, ligand-free for blind docking. The selected conformation was the conformation with the lowest bond energy that has a bond energy score of less than 1.0Å in the mean square root deviation (RMSD). The results of docking are visualized using PYMOL and then analyzed by the online program proteins plus https://proteins.plus/. PYMOL program shows the docking position and ligand-residue interactions in the form of 3D molecules. The ligation position of each MurA-Allylpyrocatechol complex was compared with the 3D structure of MurA-Fosfomycin. This step was used to determine the similarity of the allylpirocatechols 1 and 2 ligation position with fosfomycin.

Results

Isolation of Compound from the Extract of *P. betle* L. leaf

The form of an Isolated compound is a yellow oil that can be diluted in methanol. This pure compound that was isolated from *P. betle* has Rf value 0.45 in Thin-Layer Chromatography (TLC) on the ODS silica with methanol-water 3:7, v/v. Meanwhile, it has an Rf value of 0.4 in the 2D-TLC with a combination solvent *n*-hexane-ethyl acetate and *n*-hexane-acetone 4:1, v/v.

Structural Characterization of Compound

Spectral data of compound 1 are UV: 203.5 and 283.0 nm. IR: 3348, 1281, 3078, 1638, and 2925 cm $^{-1}$. MS (negative ion mode): (m/z): 149.37. 1 H-NMR (CD₃OD): $\delta_{\rm H}$ 3.26

(2H, d, 7.0 Hz, H-1), 6.79 (1H, d, 8.0 Hz, H-2), 5.07 (2H, ddd, 1.5; 8.0; 16.5 Hz, H-3), 6.71 (1H, d, 2.5 Hz, H-4), 6.62 (1H, dd, 2.0; 8.0 Hz, H-5), 5.89 and 5.94 (1H, pd, 7.0; 8.0; 16.5 Hz, H-7). 13 C-NMR (CD₃OD): $\delta_{\rm C}$ 39.6 (C-1), 115.6 (C-2), 115.7 (C-3), 115.9 (C-4), 121.2 (C-5), 133.4 (C-6), 137.7 (C-7), 141.7 (C-8), 143.5 (C-9).

Compound 1 was a yellow oil and the molecular weight (m/z) of 1 was 149.37 (C₉H₁₀O₂) based on the [M+H]⁻ peak in mass spectrometry ions. The IR spectra indicated the presence of hydroxyl and carbonyl group at absorptions of 3348 (O-H), 1281 (C-O), 3078 (C-H sp^2), 1638 (C = C), and 2925 cm⁻¹ (C-H sp^3), respectively.²⁴ Further supporting data from the UV-Vis spectrum showed absorption peaks at 203.5 nm correspond to the E band with an auxochrome substituent of a heteroatom and at 283.0 nm those suggested for an aromatic benzoic bond, respectively.³⁰

The 13 C-NMR spectrum of **1** revealed nine carbon signals including for one methylene carbon at $\delta_{\rm C}$ 39.6 ppm, two alkene carbons at $\delta_{\rm C}$ 137.7 and 115.7 ppm, four aromatic carbons at $\delta_{\rm C}$ 115.6, 115.9, 121.2, 133.4; ppm together with two oxygenated quaternary carbons at 141.8 and 143.5 ppm, respectively.

The ¹H-NMR spectrum showed three proton signals for an aromatic skeleton at $\delta_{\rm H}$ 6.71 (1H, d, J = 2.0 Hz); 6.62 (1H, dd, J = 2.0; 8.0 Hz); and 6.79 ppm (1H, d, J =8.0 Hz) which indicated the meta and ortho positions. The other proton signals at $\delta_{\rm H}$ 5.07 ppm (1H, dd, J=1.5; 8.0Hz dan 1H, dd, J = 1.5; 16.5 Hz) were considered as geminal proton,²⁴ while the double bond group was identified by proton signals at δ_H 5.89 and 5.94 ppm (1H, ddd, J = 7.0; 8.0; 16.5 Hz), respectively. From the J values of 8.0 and 16.5 Hz, the configuration cis and trans of the geminal protons to double bond were identified. Further analysis of HMBC analysis showed a correlated signal of H-1 to C2, C3, C5, C6, and C7 formed of the alleged group. Other correlated signals showed connectivity of H-2 to C5, C6 and C9, H-3 to C1 and C7, H-4 to C1, C5 and C8, H5 to C1, C4, and C8, and H-7 to C1 and C6, respectively. According to the spectroscopic data,

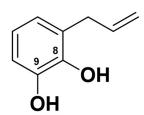


Figure I The structure of compound I.

compound 1 suggested as a phenolic derivative attached to two hydroxyl groups at C9 and C10, respectively. Based on the analysis of spectral data and compare data with published papers, the compound was identified as 1,2-dihydroxylallylbenzene (Allylpyrocatechol), as shown in Figure 1.^{31,32}

Determination of the Antibacterial Activity of Extract and Compound

The data of antibacterial extracts are reported in Table 1 presented that ethyl acetate and *n*-hexane extracts active inhibit the growth of *S. sanguinis* at 4% and 10% with inhibitions zone values of 22.45 and 21.75 mm, respectively, those similar to inhibition zones of 22.06 by 2% Chlorhexidine as a positive control. Sub-fraction of ethyl acetate extract (F1-F21) were tested antibacterial assay. The three most active fractions are F1-F3 with inhibition zone values of 14.11, 9.97, and 7.18 mm, respectively. Meanwhile, chlorhexidine as positive control showed inhibition zone values of 17.80 mm.

To evaluate the antibacterial activity of compound 1, the inhibition zone activity was measured by Kirby–Bauer. As shown in Table 2, compound 1 inhibits bacterial growth at 4% those equal with inhibition zones of chlorhexidine at 2% as a positive control. According to references of inhibition zones values, compound 1 at 1–5% with inhibitions zones of 11.-85–255 classified as a strong antibacterial agent (i.z \geq 6 mm: strong, 3–6 mm: moderate, \leq 3 mm: weak).

Further antibacterial activity evaluation, the minimum inhibition concentration (MIC) and minimum bactericidal concentration (MBC) were determined. The active compound 1 showed MIC and MBC values of 39.1 and 78.1 µg/mL, respectively. Based on the data in published papers, the MIC and MBC of 1 were categorized as good antibacterial activity.³⁴ On the other hand, fosfomycin as a control

Table 1 The Inhibition Zones of *P. betle* L. Extracts Against S. sanguinis

Samples	Inhibition Zone (mm)				
	2%	4%	6%	8%	10%
Methanol	9.6	12.05	15.4	16.75	19.05
n-Hexane	12.5	14.8	16.3	20.25	21.75
Ethyl acetate	14.8	22.45	23.15	23.45	27.9
Water-methanol	0	0	0	0	0
Methanol (-)	0	0	0	0	0
Water (-)	0	0	0	0	0
Chlorhexidine (+)	22.06	-	-	-	-

Table 2 The Inhibition Zones of Allylpyrocatechol I Against S. sanguinis

Allylpyrocatechol I	Inhibition Zone (mm)
1%	11.85
2%	16.60
3%	20.60
4%	22.90
5%	25.15
Methanol (-)	0
Chlorhexidine (+)	22.8

Table 3 Data of MIC and MBC of Allylpyrocatechols I and 2 Against S. sanguinis

Compound	Concentration (µg/mL)	
	MIC	МВС
Allylpyrocatechols I	39.1	78.2
Allylpyrocatechols 2	ND	ND
Chlorhexidine	3.12	6.25
Fosfomycin	62.5	None

showed the MIC value 62.5 μg/mL and does not have MBC value.³⁵ It means that fosfomycin just inhibits the bacteria growth but not kill the bacteria. Since compound **2** was found in a published paper, the MIC and MBC values were not reported, the structure will be used as a molecule model as isomer compound of **1** for in silico study. The assay data as shown in Table 3, suggested that the antibacterial activity of compound **1** is better than fosfomycin.

Antibacterial Activity Prediction of Allylpyrocatechols Through Molecular Interaction with UDP-N-Acetylglucosamine Enolpyruvyl Transferase (MurA)

Based analysis of in vitro antibacterial activity of allylpyrocatechol 1, the molecular mechanism prediction of allylpyrocatechols 1 and 2 as an inhibitor to inhibit MurA enzyme was evaluated by in silico study. The analysis of

Table 4 Prediction of Antibacterial Activity of MurA-Allylpyrocatechols **I** and **2**

Ligand	Binding Affinity
Allylpyrocatechols I	-5.4
Allylpyrocatechols 2	-5.4
Fosfomycin	-4.6

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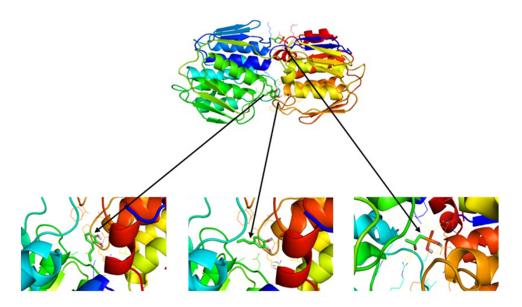


Figure 2 Ligation position of allylpyrocatechol on UDP-N-acetylglucosamine enolpyruvyl transferase (MurA).

the result as shown in Table 4 presented the molecular docking compounds 1 and 2 with MurA indicated the strength of the interaction between each compound and MurA that able to be seen from the binding energy.

As shown in Figure 2, the bond position of each complex is different. The Allylpyrocatechols 1 and 2 located in a similar cavity; meanwhile, fosfomycin occupies a different position. Residues that interact with the ligands as in Table 5. Compounds 1 and 2 bind the same amino acid residues that are Arg232A dan Ala297A. Although they ligate the same residues, the structural conformation of them is different. On the other hand, fosfomycin binds different and more residues. Beside notice what residues bound to MurA, interaction type that happens in the complex can be known through online program protein plus (Figure 3).

Discussion

Currently, discovery new prospective antibacterial agent those have selective, specific, effective, and appropriate to treat bacterial target is an interesting research focus to resolve some oral diseases caused by pathogenic bacteria. Bioactivity-guided purification of lead compounds as

Table 5 Hydrogen Bond in MurA-Allylpyrocatechols I and 2

Ligand	Residues Binding at Ligand-Protein Complex	
Allylpyrocatechols I Allylpyrocatechols 2	Arg232, Ala997 Arg232, Ala297	
Fosfomycin	Arg120A, Asn23A, Arg371A, Asp305A	

antibacterial constituents was subjected to some natural medicinal plants. Based on the ethnobotany and ethnopharmacology data, the edible herbal plant Sirih (*Piper betle L.*) was selected as sources to isolated new antibacterial agents against pathogenic oral bacteria. In the previous report, the extracts and active compounds of this plant were active as antibacterial, anti-fungal, anti-inflammation, anti-proliferative, anti-cancer, and anti-oxidant, anti-espectively.

Separation and purification chemical constituents of *P. betle* L. extract guided by bioactivity assay against *S. sanguinis* ATCC 10566 lead to isolating antibacterial compound, and the structure was identified as allylpyrocatechol (1,2-dihydroxylallylbenzene). The structure is shown in Figure 1. The published paper showed that the extract of *P. betle* L. was active against *S. mutans*, while the active constituent as an antibacterial agent against *S. sanguinis* is not reported, yet.³⁷

Allylpyrocatechol is an important natural small molecule because it used as a precursor compound to make other bioactive chemicals by chemicals synthesis reactions. Then, the isolation, characterization, and identification procedure of Allylpyrocatechol also was reported at some published papers. 46

Based on the antibacterial assay, Allylpyrocatechol (1) is weaker than chlorhexidine but it is stronger than fosfomycin. It can be suggested that Allylpyrocatechol has potential as a new candidate antibacterial agent for *S. sanguinis*. Some natural antibacterial against *S. sanguinis* were reported at some paper,^{34,47} but the antibacterial activity of

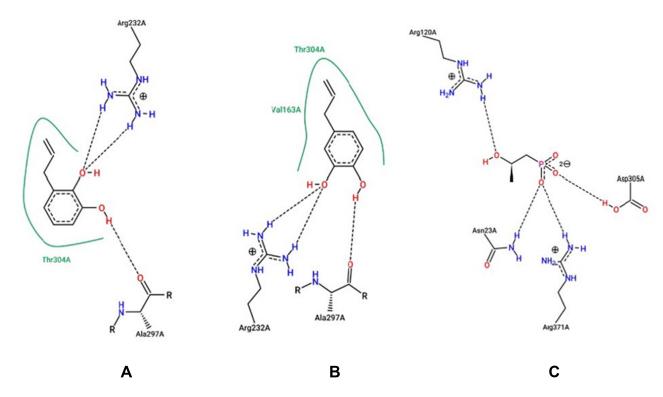


Figure 3 Interaction ligand-MurA with the ligand compound I (A) 2 (B) and fosfomycin (C).

allylpyrocatechol against *S. sanguinis* is for the first time in this report. This study adds to the list of bacteria that can be inhibited by allylpyrocatechol which in previous studies mentioned that allylpyrocatechol is active against *Staphylococcus aureus* and *Candida albicans*. 38,48

In this study, an in silico test of compound 1 and its isomer, compound 2 (Figure 4) is carried out to determine the stereoselective and regioselective factors of compounds 1 and 2 of the same target receptor. The parameters measured are binding affinity and the residues binding the MurA. The bond energy or binding affinity (ΔG) is the free energy of a bond. The value of ΔG is getting smaller or more negative indicates the stability and strength of the best bond and the bond formed is getting stronger. Free energy of compounds 1, 2, and fosfomycin is -5.4, -5.4, and -4.6 Kcal/mol, respectively. Fosfomycin which has been known as an inhibitor of MurA enzyme has the weakest of the free energy of the other ligands. It showed that the tested ligands have better stability when interacting with the MurA so that the *P. betle* L. compound has good potential as a MurA inhibitor.

The protein residue of Asn23A, Arg120A, Asp305A, and Arg371A was ligating to fosfomycin. Those residues are part of residues that bond on the uridine diphosphate *N*-acetylglucosamine (UDP-GlcNAc) that has been known as a substrate of MurA. In the complex of UDP-GlcNAc-

MurA, Asn23A, and Asp305A residues are bound to the glucosamine group while the Arg120A residue is bound to a phosphate group.²⁷ Therefore, it can be stated that compounds **1** and **2** are competitive with each other but two of them does not competitive with fosfomycin and UDP-GlcNA.⁵⁰

Compounds 1 and 2 showed a similar hydrophilic interaction (interaction on Arg232A and Ala297A residues). Nevertheless, compound 2 has two hydrophobic interactions (Thr304A and Val163A residues) while compound 1 only has one hydrophobic interaction with Thr304A residue. Meanwhile, fosfomycin just has hydrophilic interaction (Asn23A, Arg120A, Asp305A, and Arg371A). By no hydrophilic interaction in the complex of MurA-fosfomycin, it

Figure 4 The structure of Allylpyrocatechol $\, I \,$ and $\, 2 \,$.

deduces that the reason why the free energy of fosfomycin is the weakest. According to the result of in silico studies, compounds 1 and 2 have similar characteristics so it can be concluded that the stereoselective and regioselective factors do not significantly affect the antibacterial properties, especially in inhibition of the MurA enzyme.

Data of free energy through in silico are in line with antibacterial activity provided in vitro assay which binding energy and MIC value of compounds 1 and 2 are lower than fosfomycin. It deduces that allylpyrocatechol able to be alternative potential inhibitor MurA substituted fosfomycin.

Conclusion

This study gives more information about the phenolic compound list from *P. betle* leaf. Based on the characterization data, the isolated compound that is 1,2-dihydroxylallylbenzene (Allylpyrocatechol). This compound is classified as a moderate active antibacterial agent against *S. sanguinis*. This result is in line with the in silico evaluation. Docking simulations show that Allylpyrocatechols **1** (4-prop-2-enylbenzene-1,2-diol) has better stability than fosfomycin when interacting with the MurA. It deduces Allylpyrocatechols **1** potencies as a non-competitive inhibitor of MurA. This research suggests that *P. betle* L. can be a source of the antibacterial agents. Even though further research including in vitro, in vivo and clinical studies, are needed, this research provides basic information in drug discovery, especially the screening of antibacterial compounds from *P. betle*.

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Disclosure

The authors report no conflicts of interest in this work.

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