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Theoretical Study on Tuning Band Gap and Electronic Properties of Atomically Thin Nanostructured MoS₂/Metal Cluster Heterostructures

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Abstructer: Nano-heterostructures have attracted inimense attention recently due to their remarkable interfacial properties determined by the heterointerface of different nanostructures. Here, using first-principles density functional theory (DFT) calculations, we examine what range the variable electronic properties such as the electronic band gap can be tuned by combining two dissimilar nanostructures consisting of atomically thin nanostructured MoS_2 clusters with small silver and gold nanoparticles (Ag/Au NPs). Most interestingly, our calculations show that the electronic band gap of the nanostructured MoS_2 cluster can be tuned from 2.48 to 1.58 and 1.61 eV, by the formation of heterostructures with silver and gold metal nanoclusters, respectively. This band gap is ideal for various applications ranging from flexible nanoelectronics to nanophotonics applica-



tions. Furthermore, the adsorption of H_2 molecules on both nano-heterostructures is investigated, and the computed binding energies are found to be within the desirable range. The reported theoretical results provide inspiration for engineering various optoelectronic applications for nanostructured MoS₂-based heterostructures.

1. INTRODUCTION

Among the various tunable band gap transition metal dichalcogenides (TMDs), monolayer and nanostructured molybdenum disulfides (MoS₂) are promising materials and attracted increasing attention recently because of its excellent physical, chemical, and optical properties.^{1–3} These advantageous properties have provided a solid basis for the fabrication of various MoS₂-based nanoscale electronics and optoelectronic applications.^{4–6} In contrast to multilayer MoS₂, monolayer or nanostructured MoS₂ has tunable band structures with band gaps about 1–2 eV.^{7–10} The modulation of the band gap of monolayer and nanostructured MoS₂ could be beneficial for a wide variety of applications. These include high catalytic activity,¹¹ water splitting reactions,¹² hydrogen evolution reaction catalysts,¹³ energy-efficient optoelectronic devices,¹⁴ photovoltaics,¹⁵ and heterogeneous catalysis.¹⁶

Recently, numerous studies have demonstrated that the electronic structure, the band gap, the energy levels, and the associated optoelectronic properties of monolayer and nanostructured MoS₂ can be well tuned (e.g., range from 1.0 to 2.0 eV) by a variety of physical and chemical parameters,¹⁷ e.g., via chemical doping,¹⁸ chemical modification,¹⁹ defect engineering,²⁰ foreigner species intercalation,²¹ strain engineering,²² and heterostructure construction.²³ Among the various approaches mentioned, nano-heterostructures have attracted great attention due to significant interest in fundamental research as well as many practical applications because of their unique chemical and physical properties.^{23,24} Combining different nanomaterials in intelligent and effective ways into a single structure results in nano-heterostructures. Further, more nano-heterostructures of diverse nanomaterials are specifically designed^{25,26} for smart optoelectronic device applications. In recent years, to tune the band gap and realign the energy levels in MoS₂-based nano-heterostructures and to enhance optoelectronic and photocatalytic device performances have been intensively investigated both experimentally and theoretically.^{27–30} The interfacial interactions in atomically thin nanostructured MoS₂-based heterostructures dominate the chemical, physical, and interfacial properties.³¹ However, such interfacial properties are not yet understood very well and

Received: October 28, 2020 Accepted: January 5, 2021 Published: March 5, 2021



need further theoretical and experimental study for designing new nano-heterostructures.

In the present study, with the goal of understanding the electronic and interfacial properties of nanostructured MoS₂ and the interfacial properties of MoS2-Ag/Au nano-heterostructures, we have carried out first-principles density functional theory (DFT) calculations. We reported the DFToptimized ground-state structures, electronic properties, the HOMO-LUMO (highest-occupied molecular orbital to lowestunoccupied molecular orbital) gap (E_g) , and binding energy of MoS₂-based respective heterostructures. Herein, we are particularly interested in exploring how the HOMO-LUMO gap changes with the formation of different nano-heterostructures, leading to applications in the tunable HOMO-LUMO band gap, which allows great flexibility and novel opportunities in design and optimization of nanoelectronic and optoelectronic properties beyond those of homogeneous materials. Most interestingly, our theoretical calculations indicated that the band gap and interfacial properties of the studied atomically thin nanostructured MoS2-based heterostructures can be well tuned by even 0.9 eV with small silver and gold nanoparticles. Subsequently, we have investigated H₂ adsorption on both nano-heterostructures in terms of adsorption energy. It is found that the calculated adsorption energies are close to the energies required for H₂ storage applications. The present results are beneficial to supply theoretical guidance on synthesizing efficient MoS₂-Ag/Au nano-heterostructures for the use of future various nanoscale optoelectronic applications.

2. COMPUTATIONAL METHODS

In common practices, theoretical modeling of the homogeneous nanostructures involves constructing a super cell and performing periodic boundary condition (PBC) calculations with LDA or GGA DFT functionals, which can reproduce the results of experimental analysis with reasonable accuracy. However, nano-heterostructures, which consist of multiple materials, result in non periodic structures often, and hence, performing PBC calculations became a difficult task. Herein, the theoretical analysis of geometries and energies of atomically thin nanostructured MoS₂ clusters and both nanoheterostructures MoS₂-Au and MoS₂-Ag has been performed in the framework of ab initio density functional theory with the PBE0^{32,33} (Perdew-Burke-Ernzerhof hybrid) functional and the LANL2DZ basis set. It is generally accepted that the PBE functional exhibits better performance for nanostructured semiconducting clusters and solids.³⁴ For all DFT calculations, we used the Gaussian 09 software package.³⁵ We performed full geometry optimization followed by frequency calculations done for all nanostructured systems to confirm that these geometries are minima on their respective potential energy surface. Single-point energy calculations on PBE0-optimized geometries with the HSEH1PBE functional were also performed at the LANL2DZ level. To reduce the computational complexity involved in the DFT calculations, we have reduced the size of the system to the following: for nanostructured MoS_2, Mo_8S_{12} is used; for the silver/MoS_2 heterostructure, Ag₄/Mo₈S₁₂; and for the gold/MoS₂ heterostructure, Au₄/Mo₈S₁₂, respectively. The geometric, electronic, and interfacial properties were computed on the optimized structures.

3. RESULTS AND DISCUSSION

In this study, our focus will be on the design of novel nanoheterostructures with tuned optoelectronic properties, formed by the atomically thin nanostructured MoS₂ cluster with silver and gold nanoclusters. To shed light on the molecular, electronic, and interfacial properties of MoS2-Ag/Au nanoheterostructures, we have performed ab initio DFT calculations using the PBE0 functional and the LANL2DZ basis set. Initially, we have optimized the nanostructured Mo₈S₁₂ cluster and calculated the optoelectronic properties. The Mo-S bond length in the central zones of our optimized structure predicted a value of 2.51 Å, which is in good agreement with the experimental values^{36,37} of 2.41 \pm 0.06 and 2.41 Å, respectively. The calculated Mo-Mo bond length value is 3.13 Å, which showed an excellent agreement with the recent estimation of 3.16 Å, obtained from the grazing incidence extended X-ray fine structure value of MoS₂ films.⁵

To compute the stability of the nano-heterostructures, we calculated the binding energy for the MoS_2 -Ag/Au heterostructures defined by the following equation

$$E_{\text{bind}} = E_{\text{MoS}_2 - \text{Ag/Au}} - (E_{\text{MoS}_2} + E_{\text{Ag/Au}}) + E_{\text{BSSE}}$$
(1)

where $E_{\rm bind}$ is the total binding energy of MoS₂-Ag/Au nanoheterostructures with BSSE³⁸ correction. In eq 1, $E_{MoS_2 - Ag/Au}$ E_{MoS_2} , and $E_{Ag/Au}$ are the total energies of the respective optimized structures. The binding energies of Ag₄/Mo₈S₁₂ and Au_4/Mo_8S_{12} heterostructures were computed to be -4.63 and -3.90 eV, respectively. These negative binding energies indicate that the optimized nano-heterostructures are stable and energy-favorable and also indicate the weak covalent interactions present in the systems. The observed weak covalent interactions between the nanostructures are consistent with that reported in the recent studies,^{26,39} where the TMD-based nanostructures can effectively functionalize through covalent interactions. Moreover, the adsorption of silver and gold clusters on the MoS₂ system leads to the stable nano-heterostructures, and results are in line with previous MoS_2 -based studies.⁴⁰⁻⁴³ In the nano-heterostructures, the computed nearest Ag-Mo distance in Ag₄/Mo₈S₁₂ is found to be 3.27 Å, and Au–Mo distance in Au_4/Mo_8S_{12} is 3.1 Å; the calculated average Ag-S and Au-S bond lengths are 2.64 and 2.62 Å, respectively. Our calculations reveal that the binding energy of a Ag-MoS₂ heterostructure is higher than the binding energy of a Au-MoS₂ heterostructure.

The electronic properties, chemical bonding nature, and the electronic band gap tunability of MoS2-Ag/Au nano-heterostructures consisting of atomically thin nanostructured MoS₂ clusters with small silver and gold nanoparticles depend on the electronic energy level alignment between the different nanoclusters. Thus, in such nano-heterostructure interactions, the properties mainly depend on the energetic offsets between the HOMO and LUMO orbitals of the different nanoheterostructures. It is helpful to understand what the HOMO and LUMO levels represent. In individual molecules or nanoclusters, the HOMO energy (minus) corresponds exactly to the amount of energy required to remove an electron from the molecule or nanocluster (vertical ionization potential), while the LUMO energy (minus) level corresponds to the amount of energy required to add an electron to the molecule/ cluster (vertical electron affinity). However, in solid-state materials, the large number of involved orbitals leads to the concept of bands, the valence band (VB) and conduction bands (CB). In general, the formation of solid-state bands (VB or CB) is closely related to HOMO and LUMO levels. Moreover, it was found that both HOMO and LUMO have significant contributions to the formation of the conduction band and valence bands. Thus, heterostructure interactions and properties are closely related to molecular orbital distribution, HOMO/LUMO energy levels, and band gaps $(E_{o}s, difference in energy between HOMO and LUMO).$ However, the band gaps of TMD-based nanomaterials as obtained from the Kohn-Sham HOMO-LUMO gaps (within the density functional framework) are significantly smaller with respect to experimental fundamental band gaps. This is known as the well-known band gap problem. Over the past decades, there were a lot of approaches to solve the fundamental band gap problem,44,45 within the DFT framework, but they are beyond the scope of the present study.

For new nano-heterostructured materials, it is important to study their electronic structures. Here, we study how the structural energies and HOMO-LUMO gaps of MoS_2 clusters are altered by new heterostructures made of silver and gold. Table 1 shows the computed HOMO and LUMO level

Table 1. The Computed HOMO and LUMO Level Energies and the Energy Gap (in eV) for the Studied Systems: MoS_2 , Ag-MoS₂, and Au-MoS₂^d

studied systems	HOMO (eV)	LUMO (eV)	$E_{\rm g}~({\rm eV})$
Mo_8S_{12}	-6.30	-3.82	$2.48, 1.1-2.7^{a}, 1.88^{b}, 2.48^{c}$
Ag_4/Mo_8S_{12}	-5.11	-3.53	1.58
Au_4/Mo_8S_{12}	-5.37	-3.76	1.61

^{*a*}Ref 46, the range of energy gap values obtained with the different DFT functionals. ^{*b*}Ref 47, obtained from experimental photoconductivity spectra. ^{*c*}Ref 12, G_0W_0 calculation. ^{*d*}Energy values are obtained by the DFT-PBE0/LANL2DZ level of theory.

energies and the energy gap, for the studied systems: MoS_2 , MoS_2 -Ag, and MoS_2 -Au. Figure 1a,b presents the optimized structure and HOMO-LUMO plots of nanostructured MoS_2 . For the cluster Mo_8S_{12} , we have obtained a HOMO-LUMO gap value of 2.48 eV from DFT-PBE0/LANL2DZ calculation (a value of 1.74 eV obtained via single-point calculation with the HSEH1PBE/LANL2DZ level of theory). This result is in



Figure 1. (a) DFT-PBE0/LANL2DZ-optimized model structure of monolayer MoS_2 and (b) energies, electronic band gap (E_g), and HOMO and LUMO orbitals of the MoS_2 cluster calculated at the DFT-PBE0/LANL2DZ level of theory.

good agreement with the recent theoretical estimate of Wang et al.,⁴⁶ who have obtained a range of HOMO-LUMO gap values, 1.1–2.7 eV, with the different DFT functional (B3LYP and PW91) calculations for various nanostructured MoS₂-based clusters. Similarly, our result is in reasonable agreement with the band gap value of 1.88 eV, derived from photo-conductivity spectra reported for a large MoS₂ nanostructured sheet by Mak et al.⁴⁷ Our result is also in good agreement with the theoretical band gap prediction of 2.48 eV from a G₀W₀ calculation⁴⁸ and 2.02 eV from an HSE theoretical estimate.¹²

Figure 2a,b presents the minimum energy structure and HOMO-LUMO plots of the MoS₂-Ag heterostructure. Figure



Figure 2. (a) DFT-PBE0/LANL2DZ-optimized model structure of the Ag-MoS₂ nanostructure and (b) frontier orbitals, the orbital energies, and electronic band gap (E_e) of the corresponding structure.



Figure 3. (a) DFT-PBE0/LANL2DZ-optimized model structure of the Au-MoS₂ nanostructure and (b) orbital energies, electronic band gap (E_g), and HOMO and LUMO orbitals of the Au-MoS₂ calculated at the DFT-PBE0/LANL2DZ level of theory.

3a,b presents the optimized structure and HOMO-LUMO plots of the MoS_2 -Au heterostructure. Our calculations show that the electronic band gap highly depends on the HOMO-LUMO gap of the heterostructure system. For example, the electronic band gap of the Mo_8S_{12} nanostructured cluster can be changed from 2.48 eV to even a low value of 1.58 eV by Ag_4/Mo_8S_{12} and 1.61 eV by Au_4/Mo_8S_{12} hetero-nanostructures, respectively. Values of 0.9 and 0.92 eV have also been obtained via single-point calculations with the HSEH1PBE/

LANL2DZ level of theory. The formation of weak covalent interactions between the nanostructures reduces the molecular HOMO-LUMO gap significantly. The range of tunable band gap values obtained in this study is 0.9 and 0.87 eV, respectively. Thus, the tunable electronic band gap results show an excellent agreement with an experimental⁴⁹ estimate of 0.85 eV by Huang et al., from an unexpected band gap tunability obtained with distance against the single layer MoS_2 grain boundary. The density of state (DOS) calculations for nanostructured MoS_2 , the silver/ MoS_2 heterostructure, and the gold/ MoS_2 heterostructure are given in the Supporting Information.

For the application prospect, we have used density functional theory to investigate the adsorption of molecular hydrogen on both the nano-heterostructures (based on the optimized nano-heterostructures) within the framework of the DFT-PBE0/LANL2DZ method. Herein, we are particularly interested in understanding the binding energy of the hydrogen adsorption process (however, in this preliminary study, other important parameters like orientation, different adsorption sites, etc. are not included). The adsorption energy (E_{ads}) is calculated by the following equation

$$E_{\rm ads} = E_{\rm (heterostructure + H_2)} - E_{\rm heterostructure} - E_{\rm H_2}$$
(2)

The calculations give the adsorption values of MoS_2 -Au and MoS_2 -Ag nano-heterostructures, which are -0.1 and -0.11 eV, respectively, and are in good agreement with the most acceptable energy limits of 0.1-0.2 eV for the hydrogen adsorption values predicted previously.^{50,51} It can be seen that the adsorption energies (E_{ads}) obtained are negative; this indicates stable adsorption (a positive value indicates instability) and predicted a lower adsorption energy theoretically. It is found that hydrogen molecule adsorption on the heterostructures originates from the vdW interactions. Based on DFT calculations, we found that Ag and Au NPs bind strongly with nanostructured MoS_2 and further serve as a binding center for H₂ molecules with a good adsorption energy lying between the accepted energy range reported previously.^{50,51}

Thus, the results suggest that the MoS_2 -Ag and MoS_2 -Au nano-heterostructures are beneficial to improve the application of nanostructured MoS_2 and make it interesting for different optoelectronic device applications through the tunable band gap.^{1-3,52,53} Moreover, our first-principles density functional theory calculation results can provide guidance for discovering and designing new nanostructured dichalcogenide-based heterostructures.

4. CONCLUSIONS

In summary, we have performed ab initio DFT calculation to study the structural, electronic, and interfacial properties of MoS_2 -Ag/Au nano-heterostructures. The higher binding energy calculated at the optimized nano-heterostructures and the average bond lengths (Ag–S and Au–S) indicate that both the nano-heterostructures are stable. The electronic band gap of atomically thin nanostructured MoS_2 clusters shows tunability with forming heterostructures with silver and gold. Most interestingly, our DFT calculations show that the band gap of nanostructured MoS_2 clusters can be tuned from 2.48 to 1.58 and 1.61 eV with silver and gold heterostructures, respectively. Moreover, our findings showed that H_2 molecules can be adsorbed on nano-heterostructures through an appropriate range of adsorption energy. In this work, our results based on DFT calculations and the electronic structures of nanostructured MoS_2 clusters as well as MoS_2 -Ag/Au nanoheterostructures serve as models for nanostructured MoS_2 and MoS_2 -based heterostructures and provide a new way for designing and applications in future MoS_2 -based nanodevices with controllable band gaps.

ASSOCIATED CONTENT

③ Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.0c05274.

Density of state (DOS) calculations of the nanostructured structures: (Figure S1) MoS_2 , (Figure S2) silver/MoS₂ heterostructure, and (Figure S3) gold/ MoS_2 heterostructure (PDF)

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Author Contributions

The manuscript was written through contributions of all authors. All the authors have given approval to the final version of the manuscript.

Funding

The present work is carried out thhrough the Mahatma Gandhi University (MGU) Research Grant (216/2019/MGU-ADMINISTRATION).

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

The authors would like to acknowledge the research computing facility at the International and Inter University Centre for Nanoscience and Nanotechnology (IIUCNN), MGU for providing computational and storage resources.

REFERENCES

(1) Abrams, B. L.; Wilcoxon, J. P. Nanosize Semiconductors for Photooxidation. *Crit. Rev. Solid State Mater. Sci.* 2005, 30, 153–182.

(2) Barua, S.; Dutta, H. S.; Gogoi, S.; Devi, R.; Khan, R. Nanostructured MoS_2 -Based Advanced Biosensors: A Review. ACS Appl. Nano Mater. **2018**, 1, 2–25.

(3) Yang, X.; Li, B. Monolayer MoS_2 for nanoscale photonics. NANO 2020, 9, 1557–1577.

(4) Jaramillo, T. F.; Jorgensen, K. P.; Bonde, J.; Nielsen, J. H.; Horch, S.; Chorkendorff, I. Identification of Active Edge Sites for Electrochemical H_2 Evolution from MoS₂ Nanocatalysts. *Science* **2007**, 317, 100–102.

(5) Conley, H. J.; Wang, B.; Ziegler, J. I.; Haglund, R. F., Jr.; Pantelides, S. T.; Bolotin, K. I. Bandgap Engineering of Strained Monolayer and Bilayer MoS₂. *Nano Lett.* **2013**, *13*, 3626–3630.

(6) Voiry, D.; Goswami, A.; Kappera, R.; Silva, C. d. C. C. e.; Kaplan, D.; Fujita, T.; Chen, M.; Asefa, T.; Chhowalla, M. Covalent functionalization of monolayered transition metal dichalcogenides by phase engineering. *Nat. Chem.* **2015**, *7*, 45–49.

(7) Laursen, A. B.; Kegnæs, S.; Dahl, S.; Chorkendorff, I. Molybdenum sulfides-efficient and viable materials for electro-and photoelectrocatalytic hydrogen evolution. *Energy Environ. Sci.* **2012**, *5*, 5577–5591.

(8) Gong, Y.; Liu, Z.; Lupini, A. R.; Shi, G.; Lin, J.; Najmaei, S.; Lin, Z.; Elías, A. L.; Berkdemir, A.; You, G.; Terrones, H.; Terrones, M.; Vajtai, R.; Pantelides, S. T.; Pennycook, S. J.; Lou, J.; Zhou, W.; Ajayan, P. M. Band Gap Engineering and Layer-by-Layer Mapping of Selenium-Doped Molybdenum Disulfide. *Nano Lett.* **2014**, *14*, 442–449.

(9) Zhao, J.; Liu, H.; Yu, Z.; Quhe, R.; Zhou, S.; Wang, Y.; Liu, C. C.; Zhong, H.; Han, N.; Lu, J.; Yao, Y.; Wu, K. Rise of silicene: A competitive 2D material. *Prog. Mater. Sci.* **2016**, *83*, 24–151.

(10) Tan, C.; Cao, X.; Wu, X.-J.; He, Q.; Yang, J.; Zhang, X.; Chen, J.; Zhao, W.; Han, S.; Nam, G.-H.; Sindoro, M.; Zhang, H. Recent Advances in Ultrathin Two-Dimensional Nanomaterials. *Chem. Rev.* **2017**, *117*, 6225–6331.

(11) Yu, Y.; Nam, G.-H.; He, Q.; Wu, X.-J.; Zhang, K.; Yang, Z.; Chen, J.; Ma, Q.; Zhao, M.; Liu, Z.; Ran, F.-R.; Wang, X.; Li, H.; Huang, X.; Li, B.; Xiong, Q.; Zhang, Q.; Liu, Z.; Gu, L.; Du, Y.; Huang, W.; Zhang, H. High phase-purity 1T'- MoS_2 - and 1T-' $MoSe_2$ -layered crystals. *Nat. Chem.* **2018**, *10*, 638–643.

(12) Kang, J.; Tongay, S.; Zhou, J.; Li, J.; Wu, J. Band offsets and heterostructures of two-dimensional semiconductors. *Appl. Phys. Lett.* **2013**, *102*, No. 012111.

(13) Li, Y.; Wang, H.; Xie, L.; Liang, Y.; Hong, G.; Dai, H. MoS_2 Nanoparticles Grown on Graphene: An Advanced Catalyst for the Hydrogen Evolution Reaction. *J. Am. Chem. Soc.* **2011**, *133*, 7296– 7299.

(14) Lopez-Sanchez, O.; Lembke, D.; Kayci, M.; Radenovic, A.; Kis, A. Ultrasensitive photodetectors based on monolayer MoS₂. *Nat. Nanotechnol.* **2013**, *8*, 497–501.

(15) Eda, G.; Yamaguchi, H.; Voiry, D.; Fujita, T.; Chen, M.; Chhowalla, M. Photoluminescence from Chemically Exfoliated MoS₂. *Nano Lett.* **2011**, *11*, 5111–5116.

(16) Rawal, T. B.; Le, D.; Rahman, T. S. Effect of Single-Layer MoS_2 on the Geometry, Electronic Structure, and Reactivity of Transition Metal Nanoparticles. *J. Phys. Chem. C* **2017**, *121*, 7282–7293.

(17) Komsa, H.-P.; Krasheninnikov, A. V. Two-Dimensional Transition Metal Dichalcogenide Alloys: Stability and Electronic Properties. *J. Phys. Chem. Lett.* **2012**, *3*, 3652–3656.

(18) Mouri, S.; Miyauchi, Y.; Matsuda, K. Tunable Photoluminescence of Monolayer MoS_2 via Chemical Doping. *Nano Lett.* **2013**, *13*, 5944–5948.

(19) Zou, H.; Zeng, Q.; Peng, M.; Zhou, W.; Dai, X.; Ouyang, F. Electronic structures and optical properties of P and Cl atoms

adsorbed/substitutionally doped monolayer MoS₂. Solid State Commun. 2018, 280, 6–12.

(20) Xiao, P.; Mao, J.; Ding, K.; Luo, W.; Hu, W.; Zhang, X.; Zhang, X.; Jie, J. Solution-Processed 3D RGO- MoS₂/Pyramid Si Heterojunction for Ultrahigh Detectivity and Ultra-Broadband Photodetection. *Adv. Mater.* **2018**, *30*, 1801729.

(21) Zhang, J.; Yang, A.; Wu, X.; van de Groep, J.; Tang, P.; Li, S.; Liu, B.; Shi, F.; Wan, J.; Li, Q.; Sun, Y.; Lu, Z.; Zheng, X.; Zhou, G.; Wu, C.-L.; Zhang, S.-C.; Brongersma, M. L.; Li, J.; Cui, Y. Reversible and selective ion intercalation through the top surface of few-layer MoS₂. *Nat. Commun.* **2018**, *9*, 5289.

(22) Li, A.; Pan, J.; Yang, Z.; Zhou, L.; Xiong, X.; Ouyang, F. Charge and strain induced magnetism in monolayer MoS_2 with S vacancy. *J. Magn. Mater.* **2018**, 451, 520–525.

(23) Liao, W.; Huang, Y.; Wang, H.; Zhang, H. Van der Waals heterostructures for optoelectronics: Progress and prospects. *Appl. Mater. Today.* **2019**, *16*, 435–455.

(24) Li, C.; Cao, Q.; Wang, F.; Xiao, Y.; Li, Y.; Delaunay, J.-J.; Zhu, H. Engineering graphene and TMDs based van der Waals heterostructures for photovoltaic and photoelectrochemical solar energy conversion. *Chem. Soc. Rev.* **2018**, *47*, 4981–5037.

(25) Deng, Y.; Luo, Z.; Conrad, N. J.; Liu, H.; Gong, Y.; Najmaei, S.; Ajayan, P. M.; Lou, J.; Xu, X.; Ye, P. D. Black Phosphorus-Monolayer MoS₂ van der Waals Heterojunction p-n Diode. *ACS Nano* **2014**, *8*, 8292–8299.

(26) Guan, G.; Han, M.-Y. Functionalized Hybridization of 2D Nanomaterials. *Adv. Sci.* 2019, *6*, 1901837.

(27) Wu, Y. J.; Wu, P. H.; Jadczak, J.; Huang, Y. S.; Ho, C. H.; Hsu, H. P.; Tiong, K. K. Piezoreflectance study of near band edge excitonic-transitions of mixed-layered crystal $Mo(S_xSe_{1-x})_2$ solid solutions. J. Appl. Phys. **2014**, 115, 223508.

(28) Zhang, Z.; Qian, Q.; Li, B.; Chen, K. J. Interface Engineering of Monolayer MoS_2 /GaN Hybrid Heterostructure: Modified Band Alignment for Photocatalytic Water Splitting Application by Nitridation Treatment. ACS Appl. Mater. Interfaces **2018**, 10, 17419–17426.

(29) Sun, X.; Zhang, B.; Li, Y.; Luo, X.; Li, G.; Chen, Y.; Zhang, C.; He, J. Tunable Ultrafast Nonlinear Optical Properties of Graphene/ MoS₂ van der Waals Heterostructures and Their Application in Solid-State Bulk Lasers. *ACS Nano* **2018**, *12*, 11376–11385.

(30) Zhang, K.; Jariwala, B.; Li, J.; Briggs, N. C.; Wang, B.; Ruzmetov, D.; Burke, R. A.; Lerach, J. O.; Ivanov, T. G.; Haque, M.; Feenstra, R. M.; Robinson, J. A. Large scale 2D/3D hybrids based on gallium nitride and transition metal dichalcogenides. *Nanoscale* **2018**, *10*, 336–341.

(31) Jiang, H. Electronic Band Structures of Molybdenum and Tungsten Dichalcogenides by the GW Approach. J. Phys. Chem. C 2012, 116, 7664–7671.

(32) Ernzerhof, M.; Scuseria, G. E. Assessment of the Perdew-Burke-Ernzerhof exchange-correlation functional. *J. Chem. Phys.* **1999**, *110*, 5029.

(33) Adamo, C.; Barone, V. Toward reliable density functional methods without adjustable parameters: The PBE0 model. *J. Chem. Phys.* **1999**, *110*, 6158.

(34) Kurth, S.; Perdew, J. P.; Blaha, P. Molecular and solid-state tests of density functional approximations: LSD, GGAs, and meta-GGAs. *Int. J. Quantum Chem.* **1999**, *75*, 889–909.

(35) Frisch, M. J. *Gaussian 09;* revision A.1, Gaussian, Inc.: Wallingford, CT, 2009.

(36) Dickinson, R. G.; Pauling, L. THE CRYSTAL STRUCTURE OF MOLYBDENITE. J. Am. Chem. Soc. **1923**, 45, 1466–1471.

(37) Wu, L.; Longo, A.; Dzade, N. Y.; Sharma, A.; Hendrix, M. M. R. M.; Bol, A. A.; de Leeuw, N. H.; Hensen, E. J. M.; Hofmann, J. P. The Origin of High Activity of Amorphous MoS_2 in the Hydrogen Evolution Reaction. *ChemSusChem* **2019**, *12*, 4383–4389.

(38) Boys, S. F.; Bernardi, F. The calculation of small molecular interactions by the differences of separate total energies. Some procedures with reduced errors. *Mol. Phys.* **1970**, *19*, 553–566.

(39) Huang, Y. L.; Zheng, Y. J.; Song, Z.; Chi, D.; Wee, A. T. S.; Quek, S. Y. The organic-2D transition metal dichalcogenide heterointerface. *Chem. Soc. Rev.* **2018**, *47*, 3241.

(40) You, B.; Wang, X.; Zheng, Z.; Mi, W. Black phosphorene/ monolayer transition-metal dichalcogenides as two dimensional van der Waals heterostructures: a first-principles study. *Phys. Chem. Chem. Phys.* **2016**, *18*, 7381–7388.

(41) Wang, C.; He, Q.; Halim, U.; Liu, Y.; Zhu, E.; Lin, Z.; Xiao, H.; Duan, X.; Feng, Z.; Cheng, R.; Weiss, N. O.; Ye, G.; Huang, Y.-C.; Wu, H.; Cheng, H.-C.; Shakir, I.; Liao, L.; Chen, X.; Goddard, W. A., III; Huang, Y.; Duan, X. Monolayer atomic crystal molecular superlattices. *Nature* **2018**, *555*, 231–236.

(42) Liu, Y.; Huang, Y.; Duan, X. Van der Waals integration before and beyond two-dimensional materials. *Nature* **2019**, *567*, 323–333.

(43) Li, B.; Yin, J.; Liu, X.; Wu, H.; Li, J.; Li, X.; Guo, W. Probing van der Waals interactions at two-dimensional heterointerfaces. *Nat. Nanotechnol.* **2019**, *14*, 567–572.

(44) Baer, R.; Livshits, E.; Salzner, U. Tuned Range-Separated Hybrids in Density Functional Theory. *Annu. Rev. Phys. Chem.* **2010**, *61*, 85–109.

(45) Cohen, A. J.; Mori-Sánchez, P.; Yang, W. Insights into Current Limitations of Density Functional Theory. *Science* **2008**, *321*, 792–794.

(46) Wang, Y.-Y.; Deng, J.-J.; Wang, X.; Che, J.-T.; Ding, X.-L. Small stoichiometric $(MoS_2)_n$ clusters with the 1T phase. *Phys. Chem. Chem. Phys.* **2018**, *20*, 6365–6373.

(47) Mak, K. F.; Lee, C.; Hone, J.; Shan, J.; Heinz, T. F. Atomically Thin MoS_2 : A New Direct-Gap Semiconductor. *Phys. Rev. Lett.* **2010**, 105, 136805.

(48) Rasmussen, F. A.; Thygesen, K. S. Computational 2D Materials Database: Electronic Structure of Transition-Metal Dichalcogenides and Oxides. *J. Phys. Chem. C* 2015, *119*, 13169–13183.

(49) Huang, Y. L.; Chen, Y.; Zhang, W.; Quek, S. Y.; Chen, C.-H.; Li, L.-J.; Hsu, W.-T.; Chang, W.-H.; Zheng, Y. J.; Chen, W.; Wee, A. T. S. Bandgap tunability at single-layer molybdenum disulphide grain boundaries. *Nat. Commun.* **2015**, *6*, 6298.

(50) Bhatia, S. K.; Myers, A. L. Optimum Conditions for Adsorptive Storage. *Langmuir* **2006**, *22*, 1688.

(51) Putungan, D. B.; Lin, S.-H.; Wei, C.-M.; Kuo, J.-L. Li adsorption, hydrogen storage and dissociation using monolayer MoS_{2} : an *ab initio* random structure searching approach. *Phys. Chem. Chem. Phys.* **2015**, *17*, 11367–11374.

(52) Wilcoxon, J. P.; Newcomer, P. P.; Samara, G. A. Synthesis and optical properties of MoS_2 and isomorphous nanoclusters in the quantum confinement regime. *J. Appl. Phys.* **1997**, *81*, 7934–7944.

(53) Thurston, T. R.; Wilcoxon, J. P. Photooxidation of Organic Chemicals Catalyzed by Nanoscale MoS₂. J. Phys. Chem. B **1999**, 103, 11–17.

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