

Pentaaqua(4,6-dihydroxybenzene-1,3-disulfonato- κ O¹)zinc pentahydrate

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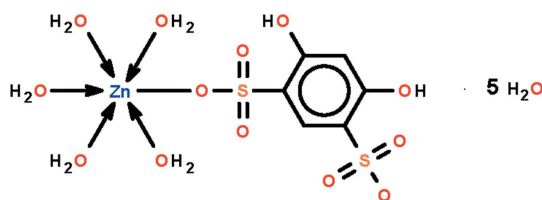
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.032; wR factor = 0.091; data-to-parameter ratio = 13.7.

The Zn^{II} atom in the title complex, $[\text{Zn}(\text{C}_6\text{H}_4\text{O}_8\text{S}_2)(\text{H}_2\text{O})_5] \cdot 5\text{H}_2\text{O}$, is coordinated by five water molecules and an O atom of a 4,6-dihydroxybenzene-1,3-disulfonate dianion. The coordination geometry is distorted octahedral, with the Zn—O_{sulfonate} bond relatively long compared to the Zn—O_{water} bonds. The coordinated and lattice water molecules interact with each other and with the hydroxy groups and sulfonate ligand through O—H...O hydrogen bonds, generating a tightly held three-dimensional network.

Related literature

For related structures, see: Xie *et al.* (2010); Bakirci *et al.* (2006).



Experimental

Crystal data

$[\text{Zn}(\text{C}_6\text{H}_4\text{O}_8\text{S}_2)(\text{H}_2\text{O})_5] \cdot 5\text{H}_2\text{O}$

$M_r = 513.74$

Triclinic, $P\bar{1}$

$a = 7.1479$ (3) Å

$b = 11.8929$ (5) Å

$c = 12.2044$ (6) Å

$\alpha = 109.368$ (1)°

$\beta = 104.690$ (1)°

$\gamma = 92.953$ (1)°

$V = 936.27$ (7) Å³

$Z = 2$

Mo $K\alpha$ radiation

$\mu = 1.62$ mm⁻¹

$T = 293$ K

0.21 × 0.17 × 0.14 mm

Data collection

Rigaku R-AXIS RAPID IP

diffractometer

Absorption correction: multi-scan

(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.747$, $T_{\max} = 0.820$

9270 measured reflections

4257 independent reflections

3883 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.032$

$wR(F^2) = 0.091$

$S = 1.06$

4257 reflections

310 parameters

32 restraints

H atoms treated by a mixture of independent and constrained refinement

$\Delta\rho_{\text{max}} = 0.67$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.35$ e Å⁻³

Table 1

Selected bond lengths (Å).

Zn—O1	2.5415 (15)	Zn—O3 _w	1.9671 (17)
Zn—O1 _w	1.9950 (15)	Zn—O4 _w	1.9588 (17)
Zn—O2 _w	1.9460 (18)	Zn—O5 _w	2.2355 (17)

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O7—H7...O9 _w ⁱ	0.83 (1)	1.79 (1)	2.612 (2)	170 (3)
O8—H8...O8 _w ⁱⁱ	0.83 (1)	1.84 (1)	2.664 (2)	172 (3)
O1 _w —H11...O2	0.84 (1)	2.09 (2)	2.852 (2)	151 (3)
O1 _w —H12...O7 ⁱⁱⁱ	0.83 (1)	2.09 (1)	2.923 (2)	175 (3)
O2 _w —H21...O6 ^{iv}	0.83 (1)	2.22 (1)	3.021 (3)	162 (3)
O2 _w —H22...O7 _w ⁱⁱⁱ	0.83 (1)	1.88 (1)	2.710 (3)	171 (4)
O3 _w —H31...O6 _w ^{iv}	0.84 (1)	2.01 (2)	2.765 (2)	149 (3)
O3 _w —H32...O10 _w ^v	0.83 (1)	1.79 (1)	2.618 (3)	178 (5)
O4 _w —H41...O6 _w ^{vi}	0.83 (1)	1.98 (1)	2.774 (2)	160 (3)
O4 _w —H42...O7 _w ^{vii}	0.84 (1)	1.99 (1)	2.802 (3)	164 (3)
O5 _w —H51...O4 ^v	0.83 (1)	2.00 (1)	2.833 (3)	176 (4)
O5 _w —H52...O6 ^{viii}	0.84 (1)	1.97 (1)	2.802 (3)	170 (3)
O6 _w —H62...O1	0.84 (1)	2.08 (1)	2.903 (2)	168 (3)
O6 _w —H61...O4 ^{iv}	0.83 (1)	1.99 (1)	2.821 (2)	176 (3)
O7 _w —H71...O2	0.84 (1)	2.05 (1)	2.874 (2)	170 (3)
O7 _w —H72...O5 ⁱⁱ	0.83 (1)	2.33 (2)	3.034 (2)	143 (3)
O8 _w —H81...O2	0.83 (1)	2.07 (2)	2.854 (2)	157 (3)
O8 _w —H82...O3 ^{vi}	0.84 (1)	1.96 (1)	2.799 (2)	176 (3)
O9 _w —H91...O5 ^{ix}	0.83 (1)	1.95 (1)	2.747 (2)	162 (3)
O9 _w —H92...O6	0.83 (1)	1.97 (1)	2.795 (2)	173 (3)
O10 _w —H101...O8 _w	0.83 (1)	2.07 (2)	2.887 (3)	167 (4)
O10 _w —H102...O9 _w ^{vi}	0.83 (1)	2.03 (1)	2.851 (3)	171 (5)

Symmetry codes: (i) $-x + 2, -y + 1, -z + 2$; (ii) $-x + 1, -y + 1, -z + 2$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $-x + 1, -y, -z + 1$; (v) $-x, -y, -z + 1$; (vi) $x - 1, y, z$; (vii) $-x, -y + 1, -z + 1$; (viii) $x - 1, y, z - 1$; (ix) $x + 1, y, z$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalClear* (Rigaku/MSK and Rigaku, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5482).

References

- Bakirci, H., Koner, A. L., Dickman, M. H., Kortz, U. & Nau, W. M. (2006). *Angew. Chem. Int. Ed.* **45**, 7400–7404.
- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
- Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Rigaku (1998). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSC and Rigaku (2002). *CrystalClear*. Rigaku/MSC Inc., The Woodlands, Texas, USA.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.
- Xie, B.-Y., Huang, W., Zhang, Y., Yang, R.-Q. & Xie, Y.-R. (2010). *Acta Cryst.* **E66**, m341.

supplementary materials

Acta Cryst. (2012). E68, m492–m493 [doi:10.1107/S1600536812012317]

Pentaaqua(4,6-dihydroxybenzene-1,3-disulfonato- κ O¹)zinc pentahydrate**Zhi-Biao Zhu, Shan Gao, Seik Weng Ng and Edward R. T. Tiekink****Comment**

In earlier studies (Xie *et al.*, 2010), the crystal structure determination of $[\text{Zn}(\text{CH}_3\text{CN})(\text{H}_2\text{O})_5](\text{C}_6\text{H}_4\text{O}_8\text{S}_2)\cdot 3\text{H}_2\text{O}$ showed that the Zn atom is octahedrally coordinated by one acetonitrile-N atom and five water molecules. The 4,6-dihydroxybenzene-1,3-disulfonate anion did not interact directly with the metal atom, instead forming hydrogen bonds to the coordinated water molecules (Xie *et al.*, 2010). When the synthesis was repeated in the absence of acetonitrile, the title compound was obtained in which the 4,6-dihydroxybenzene-1,3-disulfonate anion is now bonded to the zinc atom, Fig. 1. The covalent Zn—O_{sulfonate} bond is relatively long compared to the Zn—O_{water} bonds, Table 1. The observed coordination geometry resembles that seen in a related pentaaquozinc/sulphonate structure (Bakirci *et al.*, 2006).

The coordinated and lattice water molecules interact with each other and with the sulphonate ligand through O—H \cdots O hydrogen bonds to generate a tightly-held three-dimensional network, Fig. 2 and Table 2.

Experimental

Zinc nitrate (1 mmol) and 2,4-dihydroxyl-1,5-benzenedisulfonic acid (1 mmol) were dissolved in water (10 ml). The solution was filtered and then set aside for the formation of crystals. Colourless crystals were obtained after a week.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H = 0.93 Å) and were included in the refinement in the riding model approximation, with $U(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The hydroxyl H and water H atoms were located in a difference Fourier map, and were refined with the distance restraints O—H = 0.84±0.01 Å and H \cdots H = 1.37±0.01 Å; $U(\text{H}) = 1.5U_{\text{eq}}(\text{O})$.

Computing details

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO* (Rigaku, 1998); data reduction: *CrystalClear* (Rigaku/MSM and Rigaku, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

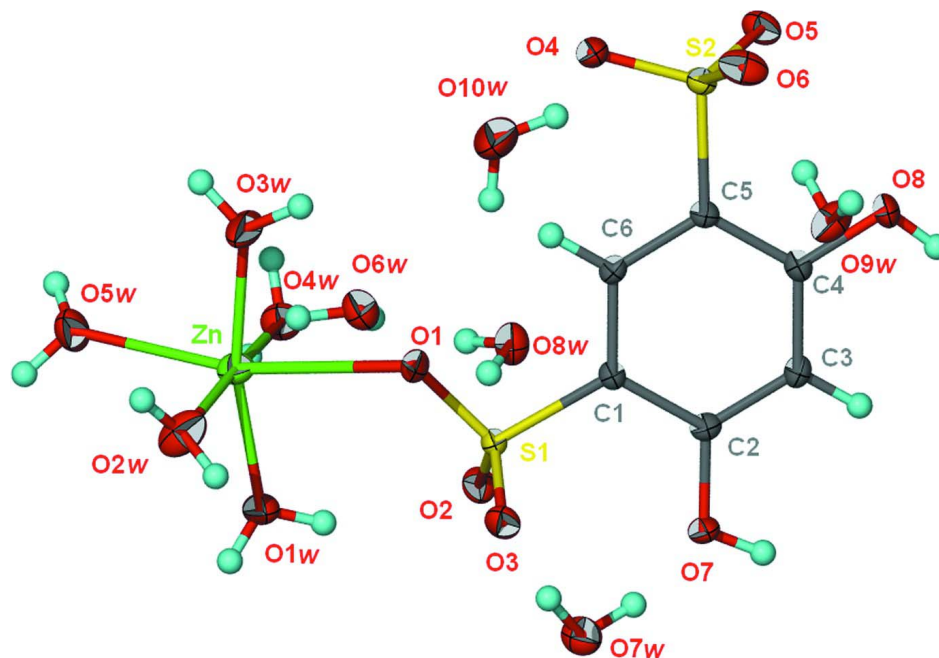
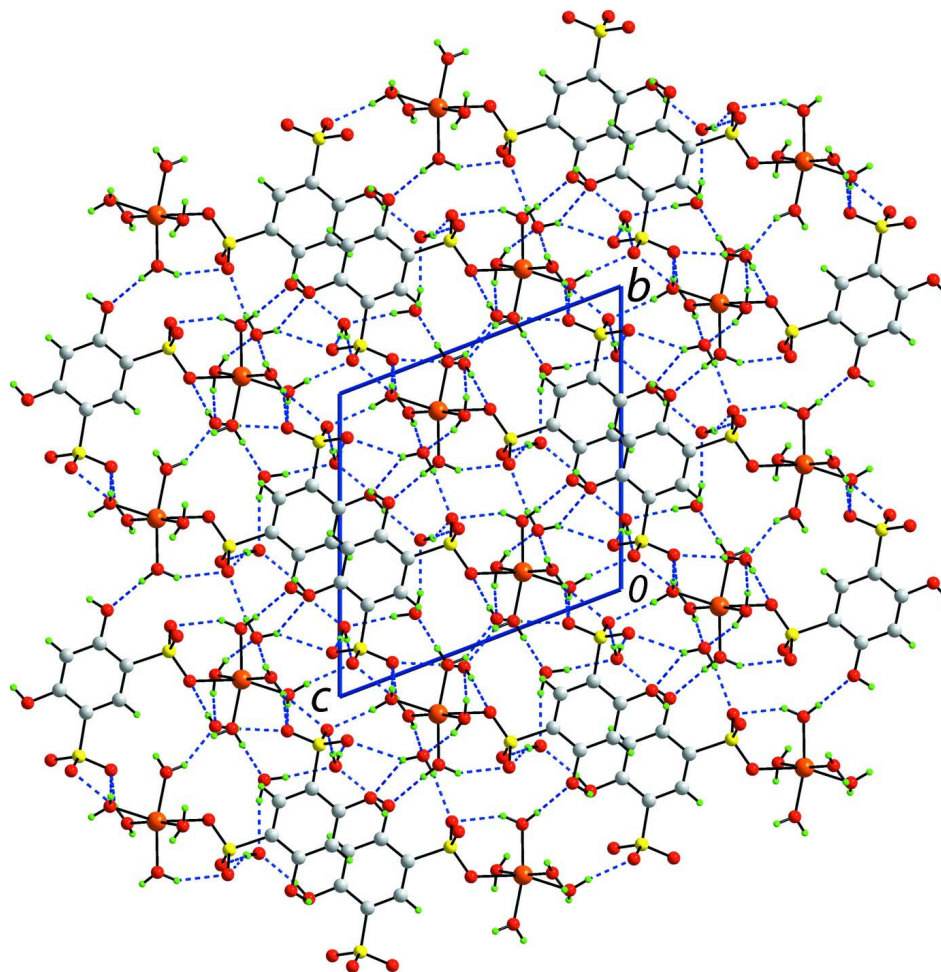


Figure 1

The asymmetric unit of (I) showing the atom-labelling scheme and displacement ellipsoids at the 50% probability level.


Figure 2

A view in projection down the a axis of the unit-cell contents of (I). The O—H...O interactions are shown as orange dashed lines.

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Crystal data

$[\text{Zn}(\text{C}_6\text{H}_4\text{O}_8\text{S}_2)(\text{H}_2\text{O})_5] \cdot 5\text{H}_2\text{O}$

$M_r = 513.74$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 7.1479$ (3) Å

$b = 11.8929$ (5) Å

$c = 12.2044$ (6) Å

$\alpha = 109.368$ (1)°

$\beta = 104.690$ (1)°

$\gamma = 92.953$ (1)°

$V = 936.27$ (7) Å³

$Z = 2$

$F(000) = 532$

$D_x = 1.822$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 8162 reflections

$\theta = 3.0$ – 27.5 °

$\mu = 1.62$ mm⁻¹

$T = 293$ K

Prism, colourless

$0.21 \times 0.17 \times 0.14$ mm

Data collection

Rigaku R-AXIS RAPID IP diffractometer	9270 measured reflections
Radiation source: fine-focus sealed tube	4257 independent reflections
Graphite monochromator	3883 reflections with $I > 2\sigma(I)$
ω scan	$R_{\text{int}} = 0.019$
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 3.0^\circ$
$T_{\text{min}} = 0.747$, $T_{\text{max}} = 0.820$	$h = -8 \rightarrow 9$
	$k = -15 \rightarrow 15$
	$l = -15 \rightarrow 15$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.032$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.091$	$w = 1/[\sigma^2(F_o^2) + (0.0505P)^2 + 0.7681P]$
$S = 1.06$	where $P = (F_o^2 + 2F_c^2)/3$
4257 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
310 parameters	$\Delta\rho_{\text{max}} = 0.67 \text{ e } \text{\AA}^{-3}$
32 restraints	$\Delta\rho_{\text{min}} = -0.35 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Zn	0.13086 (4)	0.18054 (2)	0.34895 (2)	0.02620 (9)
S1	0.51381 (7)	0.36541 (4)	0.60727 (4)	0.01787 (11)
S2	0.58913 (7)	0.13241 (4)	0.92703 (4)	0.01968 (12)
O1	0.4285 (2)	0.24063 (13)	0.53364 (13)	0.0256 (3)
O2	0.3656 (2)	0.44586 (14)	0.60175 (14)	0.0260 (3)
O3	0.6847 (2)	0.40513 (15)	0.57826 (14)	0.0272 (3)
O4	0.4610 (3)	0.04682 (14)	0.81323 (15)	0.0335 (4)
O5	0.5033 (2)	0.15452 (15)	1.02630 (14)	0.0277 (3)
O6	0.7809 (2)	0.09405 (16)	0.95478 (17)	0.0331 (4)
O7	0.7208 (3)	0.57857 (13)	0.82673 (14)	0.0294 (4)
H7	0.777 (4)	0.637 (2)	0.8893 (18)	0.044*
O8	0.7840 (3)	0.37304 (15)	1.10942 (14)	0.0329 (4)
H8	0.828 (5)	0.4422 (15)	1.160 (2)	0.049*
O1w	0.1481 (3)	0.34947 (14)	0.35174 (15)	0.0289 (3)
H11	0.205 (4)	0.401 (2)	0.4201 (12)	0.043*
H12	0.178 (4)	0.369 (2)	0.2982 (17)	0.043*

O2w	0.3028 (3)	0.15031 (17)	0.24487 (19)	0.0411 (4)
H21	0.300 (5)	0.0797 (13)	0.201 (3)	0.062*
H22	0.412 (3)	0.191 (2)	0.263 (3)	0.062*
O3w	0.1534 (3)	0.03068 (16)	0.38332 (19)	0.0500 (6)
H31	0.231 (5)	0.030 (3)	0.447 (2)	0.075*
H32	0.081 (5)	-0.0356 (19)	0.347 (3)	0.075*
O4w	-0.0565 (2)	0.22860 (16)	0.44222 (16)	0.0313 (4)
H41	-0.117 (4)	0.172 (2)	0.450 (3)	0.047*
H42	-0.136 (3)	0.263 (2)	0.406 (3)	0.047*
O5w	-0.1099 (3)	0.07189 (19)	0.18311 (16)	0.0389 (4)
H51	-0.214 (3)	0.041 (3)	0.186 (3)	0.058*
H52	-0.130 (4)	0.084 (3)	0.1172 (19)	0.058*
O6w	0.6621 (3)	0.05893 (17)	0.44204 (16)	0.0330 (4)
H61	0.623 (4)	0.025 (3)	0.3672 (10)	0.050*
H62	0.582 (4)	0.103 (3)	0.467 (2)	0.050*
O7w	0.3618 (3)	0.70069 (16)	0.70377 (16)	0.0320 (4)
H71	0.369 (5)	0.6284 (12)	0.668 (2)	0.048*
H72	0.365 (5)	0.712 (2)	0.7754 (12)	0.048*
O8w	0.0699 (3)	0.41514 (16)	0.71292 (15)	0.0320 (4)
H81	0.142 (3)	0.404 (3)	0.668 (2)	0.048*
H82	-0.0465 (17)	0.409 (3)	0.673 (2)	0.048*
O9w	1.1390 (3)	0.22383 (15)	0.98465 (17)	0.0345 (4)
H91	1.236 (3)	0.190 (2)	0.998 (3)	0.052*
H92	1.037 (2)	0.180 (2)	0.975 (3)	0.052*
O10w	0.0671 (5)	0.1796 (2)	0.7321 (2)	0.0718 (8)
H101	0.066 (8)	0.241 (2)	0.714 (4)	0.108*
H102	0.100 (8)	0.197 (4)	0.8071 (11)	0.108*
C1	0.5924 (3)	0.37148 (18)	0.75822 (17)	0.0186 (4)
C2	0.6887 (3)	0.47868 (18)	0.85336 (19)	0.0207 (4)
C3	0.7494 (3)	0.48013 (19)	0.97151 (18)	0.0231 (4)
H3	0.8104	0.5517	1.0346	0.028*
C4	0.7200 (3)	0.37540 (19)	0.99630 (18)	0.0218 (4)
C5	0.6239 (3)	0.26820 (18)	0.90129 (18)	0.0195 (4)
C6	0.5610 (3)	0.26795 (18)	0.78383 (18)	0.0200 (4)
H6	0.4965	0.1969	0.7210	0.024*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Zn	0.02933 (16)	0.02089 (14)	0.02952 (15)	0.00324 (11)	0.00921 (11)	0.00995 (11)
S1	0.0216 (2)	0.0159 (2)	0.0145 (2)	0.00053 (18)	0.00261 (17)	0.00571 (17)
S2	0.0248 (3)	0.0179 (2)	0.0185 (2)	0.00282 (19)	0.00699 (19)	0.00866 (18)
O1	0.0329 (8)	0.0185 (7)	0.0188 (7)	-0.0016 (6)	0.0006 (6)	0.0041 (6)
O2	0.0291 (8)	0.0240 (7)	0.0240 (8)	0.0080 (6)	0.0039 (6)	0.0096 (6)
O3	0.0283 (8)	0.0317 (8)	0.0239 (8)	-0.0009 (7)	0.0094 (6)	0.0123 (6)
O4	0.0508 (11)	0.0224 (8)	0.0213 (8)	-0.0085 (7)	0.0031 (7)	0.0076 (6)
O5	0.0330 (8)	0.0307 (8)	0.0260 (8)	0.0064 (7)	0.0148 (7)	0.0137 (7)
O6	0.0310 (9)	0.0364 (9)	0.0463 (10)	0.0153 (7)	0.0180 (8)	0.0262 (8)
O7	0.0471 (10)	0.0156 (7)	0.0209 (8)	-0.0056 (7)	0.0042 (7)	0.0064 (6)
O8	0.0491 (10)	0.0268 (8)	0.0151 (7)	-0.0062 (8)	-0.0021 (7)	0.0079 (6)

O1w	0.0389 (9)	0.0205 (7)	0.0278 (8)	0.0024 (7)	0.0081 (7)	0.0106 (6)
O2w	0.0415 (10)	0.0296 (9)	0.0476 (11)	0.0019 (8)	0.0255 (9)	-0.0010 (8)
O3w	0.0634 (14)	0.0237 (9)	0.0437 (11)	-0.0093 (9)	-0.0222 (10)	0.0183 (8)
O4w	0.0299 (8)	0.0332 (9)	0.0338 (9)	0.0037 (7)	0.0129 (7)	0.0133 (7)
O5w	0.0309 (9)	0.0531 (11)	0.0280 (9)	-0.0082 (8)	-0.0014 (7)	0.0181 (8)
O6w	0.0352 (9)	0.0363 (9)	0.0273 (8)	0.0055 (8)	0.0054 (7)	0.0137 (7)
O7w	0.0383 (9)	0.0302 (8)	0.0309 (9)	0.0087 (7)	0.0149 (7)	0.0109 (7)
O8w	0.0312 (9)	0.0354 (9)	0.0251 (8)	0.0032 (7)	0.0056 (7)	0.0073 (7)
O9w	0.0304 (9)	0.0253 (8)	0.0401 (10)	0.0006 (7)	0.0062 (8)	0.0053 (7)
O10w	0.103 (2)	0.0377 (12)	0.0521 (14)	-0.0216 (13)	-0.0148 (14)	0.0197 (11)
C1	0.0211 (9)	0.0187 (9)	0.0151 (9)	0.0011 (8)	0.0032 (7)	0.0065 (7)
C2	0.0241 (10)	0.0163 (9)	0.0209 (10)	0.0006 (8)	0.0053 (8)	0.0070 (8)
C3	0.0282 (11)	0.0184 (9)	0.0171 (9)	-0.0014 (8)	0.0024 (8)	0.0030 (7)
C4	0.0248 (10)	0.0231 (10)	0.0164 (9)	0.0014 (8)	0.0033 (8)	0.0079 (8)
C5	0.0233 (10)	0.0172 (9)	0.0178 (9)	0.0011 (8)	0.0045 (7)	0.0073 (7)
C6	0.0238 (10)	0.0153 (9)	0.0175 (9)	-0.0010 (8)	0.0025 (7)	0.0047 (7)

Geometric parameters (Å, °)

Zn—O1	2.5415 (15)	O3w—H32	0.833 (10)
Zn—O1w	1.9950 (15)	O4w—H41	0.833 (10)
Zn—O2w	1.9460 (18)	O4w—H42	0.838 (10)
Zn—O3w	1.9671 (17)	O5w—H51	0.831 (10)
Zn—O4w	1.9588 (17)	O5w—H52	0.840 (10)
Zn—O5w	2.2355 (17)	O6w—H61	0.831 (10)
S1—O3	1.4528 (16)	O6w—H62	0.836 (10)
S1—O1	1.4581 (15)	O7w—H71	0.837 (10)
S1—O2	1.4675 (16)	O7w—H72	0.834 (10)
S1—C1	1.760 (2)	O8w—H81	0.832 (10)
S2—O5	1.4475 (16)	O8w—H82	0.838 (10)
S2—O4	1.4577 (17)	O9w—H91	0.828 (10)
S2—O6	1.4607 (17)	O9w—H92	0.834 (10)
S2—C5	1.761 (2)	O10w—H101	0.832 (10)
O7—C2	1.355 (2)	O10w—H102	0.834 (10)
O7—H7	0.832 (10)	C1—C6	1.388 (3)
O8—C4	1.350 (2)	C1—C2	1.404 (3)
O8—H8	0.834 (10)	C2—C3	1.390 (3)
O1w—H11	0.836 (10)	C3—C4	1.394 (3)
O1w—H12	0.832 (10)	C3—H3	0.9300
O2w—H21	0.830 (10)	C4—C5	1.403 (3)
O2w—H22	0.834 (10)	C5—C6	1.388 (3)
O3w—H31	0.836 (10)	C6—H6	0.9300
O2w—Zn—O4w	171.28 (8)	Zn—O2w—H22	124 (2)
O2w—Zn—O3w	95.37 (10)	H21—O2w—H22	110.9 (17)
O4w—Zn—O3w	92.91 (10)	Zn—O3w—H31	121 (2)
O2w—Zn—O1w	87.93 (8)	Zn—O3w—H32	128 (2)
O4w—Zn—O1w	84.59 (7)	H31—O3w—H32	109.9 (17)
O3w—Zn—O1w	167.14 (8)	Zn—O4w—H41	114 (2)
O2w—Zn—O5w	86.55 (8)	Zn—O4w—H42	107 (2)

O4w—Zn—O5w	91.25 (7)	H41—O4w—H42	109.4 (16)
O3w—Zn—O5w	86.37 (8)	Zn—O5w—H51	123 (2)
O1w—Zn—O5w	106.26 (7)	Zn—O5w—H52	123 (2)
O2w—Zn—O1	89.53 (8)	H51—O5w—H52	109.2 (16)
O4w—Zn—O1	95.00 (7)	H61—O6w—H62	110.4 (16)
O3w—Zn—O1	77.08 (6)	H71—O7w—H72	109.6 (16)
O1w—Zn—O1	90.55 (6)	H81—O8w—H82	110.1 (16)
O5w—Zn—O1	162.56 (7)	H91—O9w—H92	110.9 (17)
O3—S1—O1	113.06 (10)	H101—O10w—H102	111.0 (18)
O3—S1—O2	112.15 (10)	C6—C1—C2	119.17 (18)
O1—S1—O2	110.90 (9)	C6—C1—S1	119.49 (15)
O3—S1—C1	107.07 (9)	C2—C1—S1	121.33 (15)
O1—S1—C1	106.27 (9)	O7—C2—C3	121.69 (18)
O2—S1—C1	106.94 (9)	O7—C2—C1	118.50 (18)
O5—S2—O4	112.66 (10)	C3—C2—C1	119.81 (18)
O5—S2—O6	112.17 (10)	C2—C3—C4	120.69 (19)
O4—S2—O6	110.50 (11)	C2—C3—H3	119.7
O5—S2—C5	108.45 (10)	C4—C3—H3	119.7
O4—S2—C5	105.67 (10)	O8—C4—C3	121.99 (19)
O6—S2—C5	107.00 (10)	O8—C4—C5	118.46 (18)
S1—O1—Zn	123.46 (9)	C3—C4—C5	119.53 (18)
C2—O7—H7	111 (2)	C6—C5—C4	119.40 (18)
C4—O8—H8	111 (2)	C6—C5—S2	119.21 (15)
Zn—O1w—H11	114 (2)	C4—C5—S2	121.34 (15)
Zn—O1w—H12	123 (2)	C1—C6—C5	121.38 (18)
H11—O1w—H12	110.0 (16)	C1—C6—H6	119.3
Zn—O2w—H21	118 (2)	C5—C6—H6	119.3
O3—S1—O1—Zn	-99.48 (12)	O7—C2—C3—C4	-178.1 (2)
O2—S1—O1—Zn	27.48 (13)	C1—C2—C3—C4	1.6 (3)
C1—S1—O1—Zn	143.35 (10)	C2—C3—C4—O8	177.4 (2)
O2w—Zn—O1—S1	99.04 (12)	C2—C3—C4—C5	-1.5 (3)
O4w—Zn—O1—S1	-73.50 (12)	O8—C4—C5—C6	-178.5 (2)
O3w—Zn—O1—S1	-165.34 (14)	C3—C4—C5—C6	0.5 (3)
O1w—Zn—O1—S1	11.11 (12)	O8—C4—C5—S2	-0.8 (3)
O5w—Zn—O1—S1	175.93 (17)	C3—C4—C5—S2	178.15 (17)
O3—S1—C1—C6	-122.53 (17)	O5—S2—C5—C6	-131.59 (17)
O1—S1—C1—C6	-1.4 (2)	O4—S2—C5—C6	-10.6 (2)
O2—S1—C1—C6	117.08 (18)	O6—S2—C5—C6	107.21 (18)
O3—S1—C1—C2	56.62 (19)	O5—S2—C5—C4	50.8 (2)
O1—S1—C1—C2	177.71 (17)	O4—S2—C5—C4	171.81 (18)
O2—S1—C1—C2	-63.77 (19)	O6—S2—C5—C4	-70.4 (2)
C6—C1—C2—O7	179.0 (2)	C2—C1—C6—C5	-0.4 (3)
S1—C1—C2—O7	-0.1 (3)	S1—C1—C6—C5	178.79 (16)
C6—C1—C2—C3	-0.6 (3)	C4—C5—C6—C1	0.4 (3)
S1—C1—C2—C3	-179.75 (17)	S2—C5—C6—C1	-177.26 (16)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O7—H7···O9 ^w ⁱ	0.83 (1)	1.79 (1)	2.612 (2)	170 (3)
O8—H8···O8 ^w ⁱⁱ	0.83 (1)	1.84 (1)	2.664 (2)	172 (3)
O1 ^w —H11···O2	0.84 (1)	2.09 (2)	2.852 (2)	151 (3)
O1 ^w —H12···O7 ⁱⁱⁱ	0.83 (1)	2.09 (1)	2.923 (2)	175 (3)
O2 ^w —H21···O6 ^{iv}	0.83 (1)	2.22 (1)	3.021 (3)	162 (3)
O2 ^w —H22···O7 ^w ⁱⁱⁱ	0.83 (1)	1.88 (1)	2.710 (3)	171 (4)
O3 ^w —H31···O6 ^w ^{iv}	0.84 (1)	2.01 (2)	2.765 (2)	149 (3)
O3 ^w —H32···O10 ^w ^v	0.83 (1)	1.79 (1)	2.618 (3)	178 (5)
O4 ^w —H41···O6 ^w ^{vi}	0.83 (1)	1.98 (1)	2.774 (2)	160 (3)
O4 ^w —H42···O7 ^w ^{vii}	0.84 (1)	1.99 (1)	2.802 (3)	164 (3)
O5 ^w —H51···O4 ^v	0.83 (1)	2.00 (1)	2.833 (3)	176 (4)
O5 ^w —H52···O6 ^{viii}	0.84 (1)	1.97 (1)	2.802 (3)	170 (3)
O6 ^w —H62···O1	0.84 (1)	2.08 (1)	2.903 (2)	168 (3)
O6 ^w —H61···O4 ^{iv}	0.83 (1)	1.99 (1)	2.821 (2)	176 (3)
O7 ^w —H71···O2	0.84 (1)	2.05 (1)	2.874 (2)	170 (3)
O7 ^w —H72···O5 ⁱⁱ	0.83 (1)	2.33 (2)	3.034 (2)	143 (3)
O8 ^w —H81···O2	0.83 (1)	2.07 (2)	2.854 (2)	157 (3)
O8 ^w —H82···O3 ^{vi}	0.84 (1)	1.96 (1)	2.799 (2)	176 (3)
O9 ^w —H91···O5 ^{ix}	0.83 (1)	1.95 (1)	2.747 (2)	162 (3)
O9 ^w —H92···O6	0.83 (1)	1.97 (1)	2.795 (2)	173 (3)
O10 ^w —H101···O8 ^w	0.83 (1)	2.07 (2)	2.887 (3)	167 (4)
O10 ^w —H102···O9 ^w ^{vi}	0.83 (1)	2.03 (1)	2.851 (3)	171 (5)

Symmetry codes: (i) $-x+2, -y+1, -z+2$; (ii) $-x+1, -y+1, -z+2$; (iii) $-x+1, -y+1, -z+1$; (iv) $-x+1, -y, -z+1$; (v) $-x, -y, -z+1$; (vi) $x-1, y, z$; (vii) $-x, -y+1, -z+1$; (viii) $x-1, y, z-1$; (ix) $x+1, y, z$.