

Article

Preparation and Characterization of Bioplastic-Based Green Renewable Composites from Tapioca with Acetyl Tributyl Citrate as a Plasticizer

Chi-Hui Tsou ^{1,2,3,*}, Maw-Cherng Suen ⁴, Wei-Hua Yao ³, Jen-Taut Yeh ⁵, Chin-San Wu ⁶, Chih-Yuan Tsou ¹, Shih-Hsuan Chiu ¹, Jui-Chin Chen ³, Ruo Yao Wang ³, Shang-Ming Lin ³, Wei-Song Hung ^{2,*}, Manuel De Guzman ², Chien-Chieh Hu ² and Kueir-Rarn Lee ²

- Department of Materials Science and Engineering, National Taiwan University of Science and Technology, No. 43, Sec. 4, Keelung Rd., Da'an Dist., Taipei 10607, Taiwan; E-Mails: keymykimo@gmail.com (C.-Y.T.); schiu@mail.ntust.edu.tw (S.-H.C.)
- Research and Development Center for Membrane Technology, Department of Chemical Engineering, Chung Yuan University, Chung-Li 32023, Taiwan; E-Mails: maweidegz@gmail.com (M.D.G.); cchu@cycu.edu.tw (C.-C.H.); krlee@cycu.edu.tw (K.-R.L.)
- Department of Materials and Textiles, Oriental Institute of Technology, Pan-Chiao 22064, Taiwan; E-Mails: angelayao@mail.oit.edu.tw (W.-H.Y.); fc011@mail.oit.edu.tw (J.-C.C.); steven5202001@yahoo.com.tw (R.Y.W.); fc013@mail.oit.edu.tw (S.-M.L.)
- ⁴ Department of Creative Fashion Design, Taoyuan Innovation Institute of Technology, Jhongli 32091, Taiwan; E-Mail: sun@tiit.edu.tw
- Department of Chemical and Biochemical Engineering, Kao Yuan University, Kaohsiung County 82151, Taiwan; E-Mail: jyeh@mail.ntust.edu.tw
- Faculty of Materials Science and Engineering, Hubei University, Wuhan 430062, China; E-Mail: cws1222@cc.kyu.edu.tw
- * Authors to whom correspondence should be addressed; E-Mails: mayko0301@hotmail.com (C.-H.T.); wshung@cycu.edu.tw (W.-S.H.); Tel.: +886-3-2654190 (C.-H.T. & W.-S.H.); Fax: +886-3-2654198 (C.-H.T. & W.-S.H.).

Received: 9 June 2014; in revised form: 25 July 2014 / Accepted: 29 July 2014 /

Published: 4 August 2014

Abstract: Granular tapioca was thermally blended with poly(lactic acid) (PLA). All blends were prepared using a plasti-corder and characterized for tensile properties, thermal properties and morphology. Scanning electron micrographs showed that phase separation occurred, leading to poor tensile properties. Therefore, methylenediphenyl diisocyanate (MDI) was used as an interfacial compatibilizer to improve the mechanical properties of

PLA/tapioca blends. The addition of MDI could improve the tensile strength of the blend with 60 wt% tapioca, from 19.8 to 42.6 MPa. In addition, because PLA lacked toughness, acetyl tributyl citrate (ATBC) was added as a plasticizer to improve the ductility of PLA. A significant decrease in the melting point and glass-transition temperature was observed on the basis of differential scanning calorimetry, which indicated that the PLA structure was not dense after ATBC was added. As such, the brittleness was improved, and the elongation at break was extended to several hundred percent. Therefore, mixing ATBC with PLA/tapioca/MDI blends did exhibit the effect of plasticization and biodegradation. The results also revealed that excessive plasticizer would cause the migration of ATBC and decrease the tensile properties.

Keywords: biodegradable; poly(lactic acid) (PLA); tapioca; methylenediphenyl diisocyanate (MDI); acetyl tributyl citrate (ATBC)

1. Introduction

Poly(lactic acid) (PLA) resins are well-known biodegradable, linear aliphatic thermoplastics, which can be produced from renewable resources [1,2]. They are one of the most promising polymers of their family [3] and are highly accepted as biomedical materials, because of their biocompatibility and good mechanical properties [4,5]. However, their brittleness, slow crystallization and being easily hydrolyzed limit their usage in many applications. In fact, it is difficult to use them for film blowing or extrusion, unless their moisture content and processing conditions are carefully controlled. Moreover, their price competitiveness in the biodegradable plastic market is an essential attribute that cannot be ignored. The most effective approach to reduce the capital cost of PLA is to use fillers. Cost-effective reinforcements are organic renewable resources [6], flax [7–9], sisal [10], lyocell [11], short abaca [12], jute [13], bamboo [14], paper pulp [15,16], pineapple [17], Cordenka [18], microcrystalline cellulose [19], and kenaf [20]. Starch is attractive because of its low cost, renewability, biodegradability, low density and non-abrasiveness. A lot of studies on the blending of PLA/starch [21–30], such as wheat starch [21,24,30], corn starch [26,27,29,31] and cassava starch [23], have been researched. Tapioca was used as a filler in [32], because it is cheap, and fewer reports compared it with other starches. However, the poor interfacial adhesion between the filler and the polymer generally leads to composites with worse mechanical properties. Surface and bulk modifications of the filler and/or matrix are necessary to increase the interfacial compatibility between the hydrophilic filler and the hydrophobic PLA matrix. Some studies used methylenediphenyl diisocyanate (MDI) as a compatibilizer to improve the compatibility between PLA and starch [21,30] or between PLA and rice husk [33]. These biopolymers are successfully prepared with starch or rice husk blends using MDI as a coupling agent. Copolymerization or blending PLA with other polymers [34–43] or compounds (e.g., plasticizers) [44–49] was proven to be a feasible way to improve its processability in film products for extrusion and/or film blowing. In this study, tapioca was used as a filler to reduce the cost of PLA products; MDI was used as a coupling agent to enhance the interfacial compatibility between PLA and tapioca; and ATBC was used as a plasticizer to improve the processability, flexibility and ductility of glassy PLA/tapioca composites.

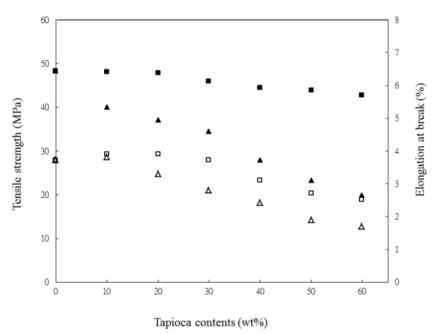
The result of this study could provide a database useful for the design and manufacture of biodegradable materials.

2. Results and Discussion

2.1. Tensile Property

The tensile strength (σ_f) and elongation at break (ϵ_f) of PLA_xTapioca_y and PLA₉₀Tapioca₁₀MDI are plotted in Figure 1. The σ_f of PLA is 48.3 MPa. After blending PLA with tapioca, PLA_xtapioca_y specimens revealed a substantial reduction in σ_f and ϵ_f . For example, the σ_f of PLA_xTapioca_y specimens was reduced from 48.3 to 17.5 MPa as the tapioca content increased from 0 to 60 wt%. When MDI was added, the σ_f of the PLA_xtapioca_y specimens was improved; the σ_f of PLA₄₀Tapioca₆₀MDI was 42.6 MPa, whereas that of PLA₄₀Tapioca₆₀ was 25.1 MPa. This improvement in σ_f is due to the increase in the PLA_xTapioca_y interfacial adhesion, as a result of the formation of urethane linkages between MDI and PLA, as well as those between MDI and tapioca, because MDI acts as a coupling agent [21]. The ϵ_f for all samples are between 1% and 4%; thus, their toughness is similar.

Figure 1. Tensile strength vs. tapioca content for (\blacktriangle) PLA/tapioca and (\blacksquare) PLA/tapiocaMDI_{0.5}; elongation at break vs. tapioca content for (\vartriangle) PLA/tapioca and (\Box) PLA/tapioca_yMDI_{0.5}.



The σ_f and ε_f of PLA and PLA_xtapioca_yMDI specimens as a function of ATBC are shown in Figures 2 and 3, respectively. With increasing plasticizer content, a common trend is shown by all series investigated: the σ_f decreases, whereas the ε_f increases. The σ_f of PLA_xtapioca_yMDI was reduced much more significantly after ATBC was added; the σ_f value of PLA₅₀tapioca₅₀MDI was reduced from 42.3 to 0.9 MPa, as the ATBC content was increased from 0 to 25 wt% (see Figure 2). This substantial reduction is due to the tapioca of PLA_xtapioca_yMDI that could not be plasticized by ATBC. In addition, the ε_f of PLA₉₀tapioca₁₀MDI, PLA₈₀tapioca₂₀MDI, PLA₇₀tapioca₃₀MDI, PLA₆₀tapioca₄₀MDI

and PLA₅₀tapioca₅₀MDI approaches the maximum value at 357.3%, 289.7%, 178.8%, 102.5% and 56.3%, respectively, as the ATBC content reaches an optimum value of 10 or 15 wt%. The ε_f of the PLA_xtapioca_yMDI specimens was reduced as the ATBC content increased from 10 or 15 wt%. These results clearly suggest that the relatively poor ductility of PLA_xtapioca_yMDI was improved after blending proper amounts of ATBC with the PLA_xTapioca_yMDI composites.

Figure 2. Tensile strength vs. ATBC content for (\blacksquare) PLA; (\circ) PLA₉₀tapioca₁₀MDI_{0.5}; (\Box) PLA₈₀tapioca₂₀MDI_{0.5}; (\Diamond) PLA₇₀tapioca₃₀MDI_{0.5}; (Δ) PLA₆₀tapioca₄₀MDI_{0.5}; and (\times) PLA₅₀tapioca₅₀MDI_{0.5}.

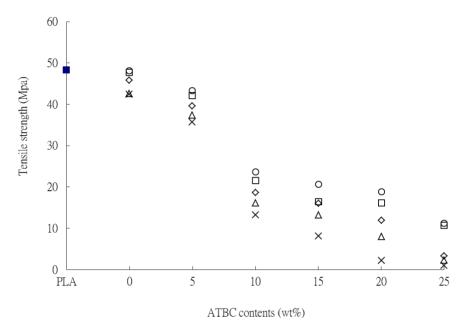
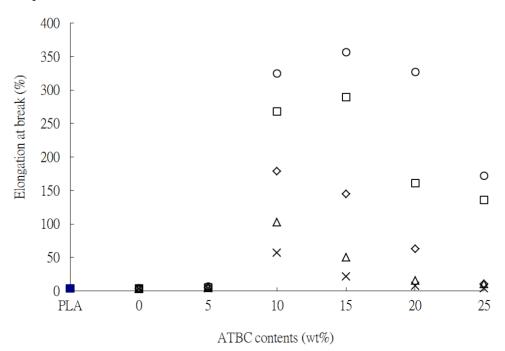


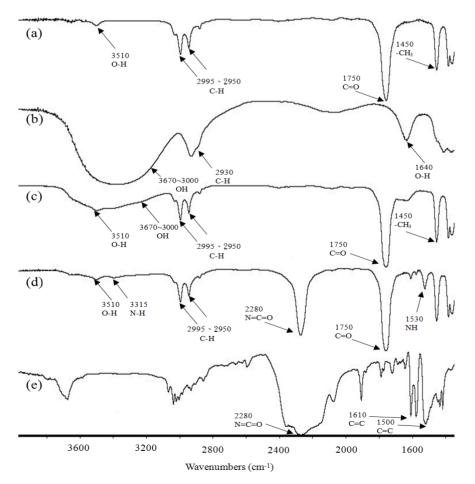
Figure 3. Elongation at break *vs.* ATBC content for (\blacksquare) PLA; (\circ) PLA₉₀tapioca₁₀MDI_{0.5}; (\Box) PLA₈₀tapioca₂₀MDI_{0.5}; (\Diamond) PLA₇₀tapioca₃₀MDI_{0.5}; (Δ) PLA₆₀tapioca₄₀MDI_{0.5}; and (\times) PLA₅₀tapioca₅₀MDI_{0.5}.



2.2. Fourier Transform Infra-Red Spectroscopy

Figure 4 illustrates typical FTIR spectra of PLA, tapioca, MDI, PLAxtapioca, and PLAxtapioca, MDI specimens. Four characteristic absorption bands centered at 1750, 2950, 2995 and 3510 cm⁻¹, corresponding to the motions of C=O bending, C-H aliphatic stretching, C-H aliphatic stretching (doublet) and C-O-O-H stretching vibrations, respectively, were found in the spectrum of PLA (see Figure 4a). The FTIR spectra of PLA_xtapioca_y specimens, indicated in Figure 4c, are very similar to those of PLA; the four main absorption bands centered at 1750, 2950, 2995 and 3510 cm⁻¹ were also found in the spectra of PLA_rtapioca_v specimens. The absorption bands around 3000 to 3670 cm⁻¹ were the O-H stretching vibration of tapioca. The FTIR spectra of PLA_xtapioca_y specimens are very similar to those of PLA_xtapioca_vMDI (see Figure 4d). However, the aforementioned 3000 to 3670 cm⁻¹ absorption bands originally shown in the FTIR spectra of PLA were gradually replaced by a newly developed absorption band centered at 3315 cm⁻¹, which corresponds to the motion of the amine (N-H) stretching vibration. The disappearance of the 3000 to 3670 cm⁻¹ bending absorption bands and the appearance of 3315 and 1550 cm⁻¹ (N-H) stretching absorption bands are attributed to the reaction of the hydroxyl (O-H) groups of tapioca molecules with the urethane (N=C=O) groups of MDI and/or to the reaction of the carboxylic acid (C-O-O-H) groups of PLA molecules with the urethane groups of MDI during the melt-blending of PLA_xtapioca_y specimens. The possible mechanism for PLA/tapioca/MDI composites is shown in Scheme 1.

Figure 4. FTIR of: (a) PLA; (b) tapioca; (c) PLA/tapioca; (d) PLA/tapioca/MDI; and (e) MDI.



Scheme 1. Possible reactions for PLA/tapioca/MDI composites.

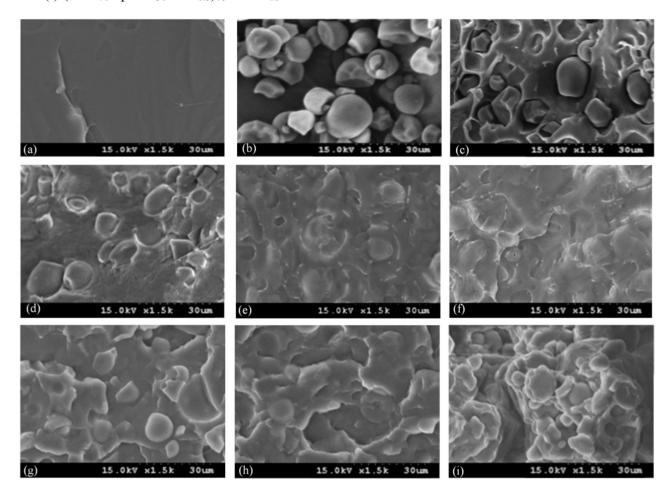
O=C=N

$$CH_2$$
 $N=C=O$
 $N=C=$

2.3. Morphology Analysis

Typical SEM micrographs of (PLA₇₀tapioca₃₀MDI)_aATBC_b specimens are shown in Figure 5. As shown in Figure 5a, a relatively brittle and smooth surface morphology was found for PLA. Tapioca is shown in Figure 5b. After blending PLA with tapioca, intervals between PLA and tapioca granules appeared (see Figure 5c) These results are similar to some reports on PLA/wheat starch [21] and PLA/corn starch blends [27,29]. This morphology is typical of incompatible blends, resulting in a poor tensile property, which is consistent with the result in Figure 1. After MDI was added, the compatibility of the PLA₇₀tapioca₃₀ specimen was improved (see Figure 5d). It shows excellent compatible morphologies, without the interval and voids associated with poor interfacial adhesion. The better compatibility of PLA₇₀tapioca₃₀MDI was due to the reaction of the hydroxyl groups of tapioca with the urethane groups of MDI and the reaction of the carboxylic acid groups of PLA with the urethane groups of MDI. The SEM micrographs of PLA₇₀tapioca₃₀MDI as a function of the increasing ATBC content (from 5 to 25 wt%) are shown in Figure 5e-h. The surface of PLA₇₀tapioca₃₀MDI was still smooth and without intervals when the ATBC content was 5 wt%. Furthermore, two phases can be seen after the ATBC content reaches 10 wt%, which is the threshold limit value of high ε_f for (PLA₇₀tapioca₃₀MDI)_aATBC_b specimens (see Figure 3). With increasing ATBC content, more demarcated plastic-deformed PLA debris or fibrils were found on the surface of PLA₇₀Tapioca₃₀MDI (see Figure 5f-i). This was attributed to the increasing distance between PLA molecules, or the exudation of ATBC, and also to the deterioration of the interfacial adhesion between tapioca and PLA. Therefore, the σ_f of PLA_xtapioca_vMDI decreased significantly with increasing ATBC.

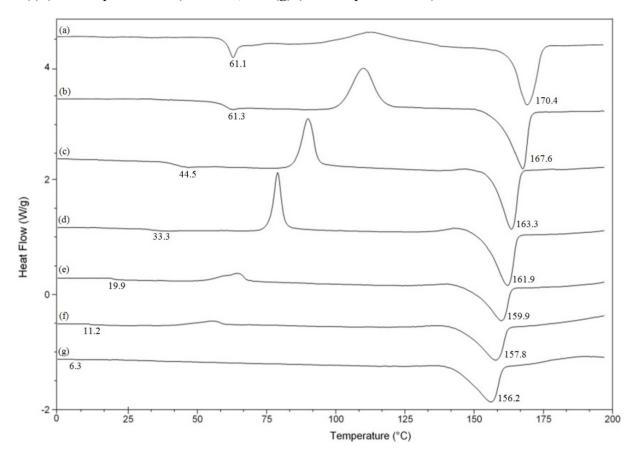
Figure 5. SEM for: (a) PLA; (b) tapioca; (c) PLA₇₀tapioca₃₀; (d) PLA₇₀Tapioca₃₀MDI_{0.5}; (e) (PLA₇₀tapioca₃₀MDI_{0.5})₉₅ATBC₅; (f) (PLA₇₀tapioca₃₀MDI_{0.5})₉₀ATBC₁₀; (g) (PLA₇₀tapioca₃₀MDI_{0.5})₈₅ATBC₁₅; (h) (PLA₇₀tapioca₃₀MDI_{0.5})₈₀ATBC₂₀; and (i) (PLA₇₀tapioca₃₀MDI_{0.5})₇₅ATBC₂₅.



2.4. Differential Scanning Calorimetry

The thermal behavior of PLA and $(PLA_{70}tapioca_{30}MDI)_aATBC_b$ specimens was investigated. Figure 6 shows the DSC curves of PLA, the $(PLA_{70}tapioca_{30}MDI)_aATBC_b$ specimens and the plasticizer. The crystallization and melting enthalpies are identical, showing that PLA is totally amorphous after melt quenching. The DSC curves of $PLA_{70}tapioca_{30}MDI$ show a single glass transition that decreases with increasing ATBC concentration (from 0 to 25 wt%). This result agrees with the data by Yeh *et al.* [50] and Baiardo *et al.* [51], who analyzed PLA/triacetin (TAc) and PLA/ATBC composites, respectively, over a more limited composition range (0%-30% TAc and ATBC). A decreasing T_g trend with increasing ATBC content is also shown in Figure 6, which is related to the cold crystallization and melting phenomena.

Figure 6. DSC for: (a) PLA; (b) PLA₇₀tapioca₃₀MDI; (c) (PLA₇₀tapioca₃₀MD)₉₅ATBC₅; (d) (PLA₇₀tapioca₃₀MDI)₉₀ATBC₁₀; (e) (PLA₇₀tapioca₃₀)₈₅ATBC₁₅; (f) (PLA₇₀tapioca₃₀MDI)_aATBC_b; and (g) (PLA₇₀tapioca₃₀MDI)₇₅ATBC₂₅.

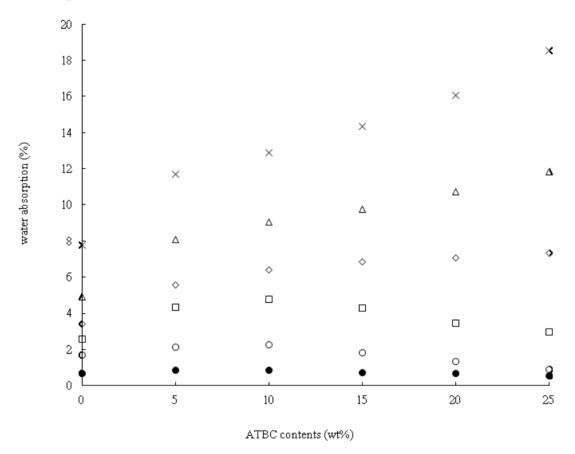


2.5. Water Absorption

Figure 7 presents data on the water absorption of the PLA_xATBC_y specimens as a function of varying tapioca content. It shows a rising water absorption rate with increasing tapioca content at the same ATBC content. Water absorption rates were 0.58% to 18.57% when the tapioca content was from 0% to 50% at 25% ATBC content. This is due to the hydrophilic property of tapioca, causing the percentage of water absorption to increase. Furthermore, water absorption rates increased when ATBC was added; the water absorption rate of PLA₆₀tapioca₄₀ was 4.91% to 11.85% when the ATBC content was from 0% to 25%. ATBC is hydrophobic, but shows an interesting phenomenon of increasing moisture content. The trend of increasing water absorption is attributed to ATBC, which could enhance the free volume in PLA. This is evidenced by DSC analysis (see Figure 6): T_g decreased with increasing ATBC content. Therefore, the water molecule could be absorbed by PLA_xtapioca_y easily when the ATBC content increased the mobility of PLA molecules. The water adsorption for the lower tapioca content (10% to 30%) was first increasing and then decreasing. This might be due to slight exudation when the ATBC content is from 10% to 25%. It could be evidenced from the morphology analysis. The phase separation may be because of the exudation of ATBC; the phase separation worsens when the amount of ATBC was higher than 10% (See Figure 5f-i). However, when the tapioca content approached 40% and 50%, the ATBC might be absorbed by tapioca. Therefore, the exudation effect might not be apparent. On the basis of the results from the tensile property,

morphology analysis and water absorption, the miscibility limits of ATBC content are suggested to be 10% for PLA_xtapioca_yMDI_{0.5} composites.

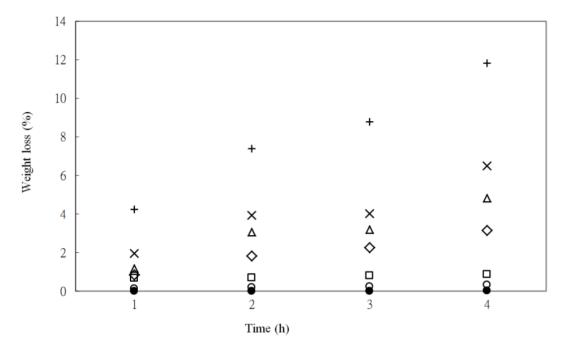
Figure 7. Water absorption *vs.* ATBC content for (\bullet) PLA; (\circ) PLA₉₀tapioca₁₀MDI_{0.5}; (\Box) PLA₈₀tapioca₂₀MDI_{0.5}; (\Diamond) PLA₇₀tapioca₃₀MDI_{0.5}; (Δ) PLA₆₀tapioca₄₀MDI_{0.5}; and (\times) PLA₅₀tapioca₅₀MDI_{0.5}.



2.6. Enzymatic Hydrolysis

The weight loss of PLA₇₀tapioca₃₀MDI and (PLA₇₀tapioca₃₀MDI)_aATBC_b at varying enzymatic hydrolysis time is indicated in Figure 8. It shows a common result: the increasing percentage of weight loss for all series, as the hydrolysis time increases. The weight loss of PLA₇₀tapioca₃₀MDI increased significantly from 0.33% to 11.82%, as the ATBC content increased from 0 to 25 wt% after 120 h of hydrolysis time. Enzymes could attack the molecules of PLA easily after ATBC was blended with PLA₇₀tapioca₃₀MDI. Furthermore, the interfacial adhesion between PLA and tapioca might deteriorate when ATBC was added. The weight loss increased significantly when the ATBC content was higher than 5%. The migration of ATBC occurred from 10% content. This conjecture of exudation was evidenced by the morphological analysis and water absorption.

8. Weight for PLA; **Figure** loss **(•)** (o) PLA₇₀tapioca₃₀MDI_{0.5}; (PLA₇₀tapioca₃₀MDI_{0.5})₉₅ATBC₅; (PLA₇₀tapioca₃₀MDI_{0.5})₉₀ATBC₁₀; (□) (\diamondsuit) (PLA₇₀tapioca₃₀MDI_{0.5})₈₅ATBC₁₅; (PLA₇₀tapioca₃₀MDI_{0.5})₈₀ATBC₂₀; (×) (Δ) (+) (PLA₇₀tapioca₃₀MDI_{0.5})₇₅ATBC₂₅.



3. Experimental Section

3.1. Materials and Preparation

PLA resin, with a trade name of Nature Green 4032D, was obtained from Cargill-Dow. Tapioca was purchased from United Global Agencies (Bangkok, Thailand). Before melt-blending, PLA and tapioca were dried in a vacuum oven at 80 °C for 8 h to remove the residual water. Acetyl tributyl citrate (ATBC-food grade) was supplied by Chou Feng Enterprise Co., Ltd. (Shulin, Taiwan). Dried components of PLA/starch at varying weight ratios were melt-blended using a Brabender. Three compounds were evaluated: PLA_xTapioca_y, PLA_xTapioca_yMDI and (PLA_xTapioca_yMDI)_aATBC_b. During each compounding process, the Brabender was operated at a temperature of 190 °C and a screw speed of 120 rpm for 3 min for all samples without ATBC and for an additional 2 min after adding ATBC. All prepared series were then hot-pressed at 190 °C and 10 MPa for 2 min and then cooled in air at about 25 °C. The compositions of all specimens are summarized in Table 1. Before hot-pressing, the specimens were dried in a vacuum oven at 80 °C for 12 h.

Table 1. Compositions of PLA, PLA_xtapioca_y, PLA_xtapioca_yMDI and $(PLA_x tapioca_y MDI)_a ATBC_b$ specimens.

Sample	PLA (%)	Tapioca (%)	ATBC (%)	MDI (phr)
PLA	100	0	0	0
PLA ₉₀ tapioca ₁₀	90	10	0	0
PLA ₈₀ tapioca ₂₀	80	20	0	0
PLA ₇₀ tapioca ₃₀	70	30	0	0

Table 1. Cont.

Sample	PLA (%)	Tapioca (%)	ATBC (%)	MDI (phr)
PLA ₆₀ tapioca ₄₀	60	40	0	0
PLA ₅₀ tapioca ₅₀	50	50	0	0
PLA ₄₀ tapioca ₆₀	0	60	0	0
PLA ₉₀ tapioca ₁₀ MDI	90	10	0	0.5
PLA ₈₀ tapioca ₂₀ MDI	80	20	0	0.5
PLA ₇₀ tapioca ₃₀ MDI	70	30	0	0.5
PLA ₆₀ tapioca ₄₀ MDI	60	40	0	0.5
PLA ₅₀ tapioca ₅₀ MDI	50	50	0	0.5
PLA ₆₀ tapioca ₄₀ MDI	40	60	0	0.5
(PLA ₉₀ tapioca ₁₀ MDI) ₉₅ ATBC ₅	85.5	9.5	5	0.5
(PLA ₈₀ tapioca ₂₀ MDI) ₉₅ ATBC ₅	76	19	5	0.5
(PLA ₇₀ tapioca ₃₀ MDI) ₉₅ ATBC ₅	66.5	28.5	5	0.5
(PLA ₆₀ tapioca ₄₀ MDI) ₉₅ ATBC ₅	57	38	5	0.5
(PLA ₅₀ tapioca ₅₀ MDI) ₉₅ ATBC ₅	47.5	47.5	5	0.5
(PLA ₉₀ tapioca ₁₀ MDI) ₉₀ ATBC ₁₀	81	9	10	0.5
(PLA ₈₀ tapioca ₂₀ MDI) ₉₀ ATBC ₁₀	72	18	10	0.5
(PLA ₇₀ tapioca ₃₀ MDI) ₉₀ ATBC ₁₀	63	27	10	0.5
(PLA ₆₀ tapioca ₄₀ MDI) ₉₀ ATBC ₁₀	54	36	10	0.5
(PLA ₅₀ tapioca ₅₀ MDI) ₉₀ ATBC ₁₀	45	45	10	0.5
(PLA ₉₀ tapioca ₁₀ MDI) ₈₅ ATBC ₁₅	76.5	8.5	15	0.5
(PLA ₈₀ tapioca ₂₀ MDI) ₈₅ ATBC ₁₅	68	17	15	0.5
(PLA ₇₀ tapioca ₃₀ MDI) ₈₅ ATBC ₁₅	59.5	25.5	15	0.5
(PLA ₆₀ tapioca ₄₀ MDI) ₈₅ ATBC ₁₅	51	34	15	0.5
(PLA ₅₀ tapioca ₅₀ MDI) ₈₅ ATBC ₁₅	42.5	42.5	15	0.5
(PLA ₉₀ tapioca ₁₀ MDI) ₈₀ ATBC ₂₀	72	8	20	0.5
(PLA ₈₀ tapioca ₂₀ MDI) ₈₀ ATBC ₂₀	64	16	20	0.5
(PLA ₇₀ tapioca ₃₀ MDI) ₈₀ ATBC ₂₀	56	24	20	0.5
(PLA ₆₀ tapioca ₄₀ MDI) ₈₀ ATBC ₂₀	48	32	20	0.5
(PLA ₅₀ tapioca ₅₀ MDI) ₈₀ ATBC ₂₀	40	40	20	0.5
(PLA ₉₀ tapioca ₁₀ MDI) ₇₅ ATBC ₂₅	67.5	7.5	25	0.5
(PLA ₈₀ tapioca ₂₀ MDI) ₇₅ ATBC ₂₅	60	15	25	0.5
(PLA ₇₀ tapioca ₃₀ MDI) ₇₅ ATBC ₂₅	52.2	22.5	25	0.5
(PLA ₆₀ tapioca ₄₀ MDI) ₇₅ ATBC ₂₅	45	30	25	0.5
(PLA ₅₀ tapioca ₅₀ MDI) ₇₅ ATBC ₂₅	37.5	37.5	25	0.5

3.2. Tensile Property

The tensile properties of the hot-pressed PLA, PLA_xTapioca_y and (PLA_xTapioca_y)/ATBC specimens at 25 °C were determined using a tensile testing machine (model AG-10KNA, Shimadzu Corporation, Kyoto, Japan) with a crosshead speed of 50 mm/min. A 35-mm gauge length was used for each tensile experiment. Dog-bone-shaped specimens were prepared according to the ASTM D638 Type IV standard [52]. On the basis of the average tensile results of at least five tensile specimens, the values of tensile strength and elongation at break were obtained.

3.3. Fourier Transform Infra-Red Spectroscopy

FTIR measurements were performed on a PerkinElmer spectrometer (model Spectrum One, PerkinElmer Inc., Waltham, MA, USA). The spectra of the samples were obtained by averaging 15 scans, with a wavenumber range of 4000 to 650 cm⁻¹ and a resolution of 2 cm⁻¹.

3.4. Morphology Analysis

The morphology of specimens, before and after the hydrolytic degradation, was observed by using a scanning electron microscope (model SU1510, Hitachi High-Technologies Corporation, Tokyo, Japan). Specimens of a 2×2 cm² area were fixed on a sample holder using a conductive adhesive tape. They were coated with a thin layer of gold at 15 keV for 15 s to improve the image resolution and were then photographed at 3.00 K magnification and a low voltage of 2.1 kV.

3.5. Differential Scanning Calorimetry

The thermal properties of PLA composite resins were determined using a TA Q100 differential scanning calorimetry (DSC). All DSC scans were performed at a heating rate of 10 °C/min and under flowing nitrogen with a flow rate of 50 mL/min. The DSC was calibrated using pure indium. For $T_{\rm g}$ and $T_{\rm m}$ determination, samples weighing approximately 0.5 mg were placed in standard aluminum-sample pans.

3.6. Water Absorption

Five specimens (10 mm \times 10 mm \times 0.5 mm) were used for the water absorption test. After conditioning in desiccators for three weeks, specimens were weighed. They were immersed in distilled water at room temperature for 24 h. Then, they were dabbed with tissue paper to remove the water from the surface. Water absorption was calculated using the following Equation (1):

$$WA(\%) = \frac{W_{1-}W_0}{W_0} \times 100 \tag{1}$$

where W_0 is the weight of the dry sample and W_1 is the weight of the sample immersed in distilled water for 24 h.

3.7. Enzymatic Hydrolysis

The degradation of the enzymatic hydrolysis of specimens was evaluated at 27 °C using 50 mg starch enzyme in $(0.025 \text{ mol Na}_2\text{HPO}_4 + 0.025 \text{ mol KH}_2\text{PO}_4)$ aqueous solution. Specimens with a dimension of 5×5 cm² were tested for various days, washed with distilled water and dried completely in a vacuum oven at 70 °C for 8 h. On the basis of weight loss, the degree of degradation was determined using Equation (2):

weight loss =
$$\frac{W_{1-}W_0}{W_0} \times 100\% \tag{2}$$

where W_0 is the dry weight before degradation and W_t is the dry weight at time t.

4. Conclusions

The σ_f of PLA_rtapioca_vMDI specimens was significantly higher than that of PLA_rtapioca_v specimens. The ε_f of PLA and PLA_rtapioca_vMDI specimens approached the maximum value, as the ATBC content reached an optimum value of 10 and 15 wt%, respectively. The threshold limits of the ε_f were high when the ATBC content was 10 wt%. FTIR demonstrated the disappearance of the 3000 to 3670 cm⁻¹ bending absorption band and the appearance of the 3315 and 1550 cm⁻¹ NH stretching absorption band, which were attributed to the reaction of the OH groups of tapioca molecules with the N=C=O groups of MDI and/or to the reaction of the C-O-O-H groups of PLA molecules with the urethane groups of MDI during the melt-blending of PLA_xtapioca_y specimens. SEM micrographs revealed the intervals between PLA and tapioca. Voids from the matrix of PLA_xtapioca_v were significantly improved after MDI was added. Furthermore, two phases can be seen after the ATBC content reached 10 wt%. This is due to the exudation of ATBC; with increasing ATBC content, more demarcated plastic deformation was found on the surface of PLA₇₀Tapioca₃₀MDI. DSC curves of the PLA₇₀tapioca₃₀MDI specimen showed a single glass transition and cold crystallization that decreased as the ATBC content increased from 0 to 25 wt%. The increasing trend of water absorption with increasing ATBC content was attributed to the increasing free volume in PLA, causing the water molecules to be easily absorbed in the PLA_xtapioca_y specimens. Enzymatic hydrolysis tests indicated that the weight loss of PLA₇₀tapioca₃₀MDI increased significantly as the ATBC content increased.

Acknowledgments

The authors express their appreciation to Grabio Greentech Corporation, Ching-Yung Trading Co., Ltd., Fabric King Textile Co., Ltd., Ministry of Economic Affairs (100-EC-17-A-10-S1-186) and the National Science Council (NSC 102-2221-E-161-011, NSC 100-3113-E-033-001 and NSC 102-2221-E-003-065) for support of this work.

Author Contributions

All authors contributed to this study. Chi-Hui Tsou and Wei-Song Hung designed the research and wrote this paper. Chi-Hui Tsou, Maw-Cherng Suen, and Chih-Yuan Tsou edited the paper and gave final approval of the version to be submitted. Chih-Yuan Tsou and Ruo Yao Wang did the analysis. Manuel De Guzman, Chien-Chieh Hu, and Kueir-Rarn Lee also contributed in analyzing data and in rewriting the revised manuscript. Jen-Taut Yeh, Wei-Hua Yao, Chin-San Wu, Shih-Hsuan Chiu, Jui-Chin Chen, Shang-Ming Lin, and Manuel De Guzman supervised the conduct of experiments and performed the theoretical analysis of the data.

Conflicts of Interest

The authors declare no conflict of interest.

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