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6-Ferrocenoyl-7-(4-fluorophenyl)spiro-[hexahydropyrrolo[1,2-c][1,3]thiazole-5,11'-indeno[1,2-b]quinoxaline]

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Key indicators: single-crystal X-ray study; T = 293 K; mean σ (C–C) = 0.003 Å; R factor = 0.035; wR factor = 0.097; data-to-parameter ratio = 15.0.

In the title compound, $[Fe(C_5H_5)(C_{32}H_{23}FN_3OS)]$, both the thiazolidine ring and the pyrrolidine ring adopt a twist conformation on the N-C(H) bridging bond. Their mean planes are inclined to one another by 10.05 (10)°, and they make dihedral angles of 82.09 (10) and 89.67 (11)°, respectively, with the cyclopentane ring. The F atom deviates by -0.0238 (2) Å from the benzene ring to which it is attached. In the crystal, molecules are linked by a pair of C-H···O hydrogen bonds, forming inversion dimers.

Related literature

For the biological activity of ferrocene derivatives, see: Jaouen *et al.* (2004); Biot *et al.* (2004); Fouda *et al.* (2007). For a related structure, see: Vijayakumar *et al.* (2012).



22064 measured reflections

 $R_{\rm int} = 0.022$

5966 independent reflections

5126 reflections with $I > 2\sigma(I)$

Experimental

Crystal data

 $\begin{array}{ll} \left[\mathrm{Fe}(\mathrm{C}_{5}\mathrm{H}_{5})(\mathrm{C}_{32}\mathrm{H}_{23}\mathrm{FN}_{3}\mathrm{OS}) \right] & \gamma = 86.776 \ (1)^{\circ} \\ M_{r} = 637.53 & V = 1455.39 \ (6) \ \mathring{A}^{3} \\ \mathrm{Triclinic}, \ P\overline{1} & Z = 2 \\ a = 8.7097 \ (2) \ \mathring{A} & \mathrm{Mo} \ K\alpha \ \mathrm{radiation} \\ b = 12.6456 \ (3) \ \mathring{A} & \mu = 0.63 \ \mathrm{mm}^{-1} \\ c = 13.5477 \ (4) \ \mathring{A} & T = 293 \ \mathrm{K} \\ \alpha = 83.865 \ (1)^{\circ} & 0.30 \times 0.25 \times 0.20 \ \mathrm{mm} \\ \beta = 79.008 \ (1)^{\circ} \end{array}$

Data collection

Bruker SMART APEXII areadetector diffractometer Absorption correction: multi-scan (*SADABS*; Bruker, 2008) $T_{\rm min} = 0.833$, $T_{\rm max} = 0.884$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$	397 parameters
$wR(F^2) = 0.097$	H-atom parameters constrained
S = 1.06	$\Delta \rho_{\rm max} = 0.33 \ {\rm e} \ {\rm \AA}^{-3}$
5966 reflections	$\Delta \rho_{\rm min} = -0.34 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

 $D-H\cdots A$ D-H $H\cdots A$ $D\cdots A$ $D-H\cdots A$

 C25-H25\cdots O1ⁱ
 0.93
 2.54
 3.212 (2)
 129

Symmetry code: (i) -x + 1, -y + 1, -z + 1.

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2635).

References

- Biot, C., Dessolin, J., Richard, I. & Dive, D. (2004). J. Organomet. Chem. 689,4678–4682.
- Bruker (2008). APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (2012). J. Appl. Cryst. 45, 849-854.
- Fouda, M. F. R., Abd-Elzaher, M. M., Abdelsamaia, R. A. & Labib, A. A. (2007). Appl. Organomet. Chem. 21, 613–625.
- Jaouen, G., Top, S., Vessireres, A., Leclercq, G., Vaissermann, J. & McGlinchey, M. J. (2004). Curr. Med. Chem. 11, 2505–2517
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Spek, A. L. (2009). Acta Cryst. D65, 148-155.
- Vijayakumar, B., Gavaskar, D., Srinivasan, T., Raghunathan, R. & Velmurugan, D. (2012). Acta Cryst. E68, m1382–m1383.

supplementary materials

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6-Ferrocenoyl-7-(4-fluorophenyl)spiro[hexahydropyrrolo[1,2-c][1,3]thiazole-5,11'-indeno[1,2-b]quinoxaline]

Sivasubramanian Suhitha, Krishnaswamy Gunasekaran, Deivasigamani Gavaskar, Raghavachary Raghunathan and Devadasan Velmurugan

1. Comment

Ferrocene attached compounds are well known to have biological activities, such as antimalarial, antifungal (Biot *et al.*, 2004), antitumor (Jaouen *et al.*, 2004), and antibacterial (Fouda *et al.*, 2007). Against this background, and in order to gain information on the molecular conformations and crystal packing, we report herein on the synthesis and crystal structure of the title compound.

In the title compound, Fig. 1, both the thiazolidine ring and the pyrrolidine ring adopt a *twist* conformation on bond N3-C18. The thiazolidine ring (S1/N3/C18-C20) mean plane makes a dihedral angle of 10.05 (10)° with the pyrrolidine ring (N3/C15-C18) mean plane, it also makes a dihedral angle of 82.09 (10)° with the cyclopentane ring (C7-C9/C14/C15) which shows that they are almost perpendicular to each other. The pyrrolidine ring mean plane makes a dihedral angle of 89.67 (11)° with the cyclopentane ring which shows they too are almost orthogonal to each other. The fluorine atom F1 attached with the phenyl ring (C21-C26) deviates by -0.0238 (2)Å.

In the crystal, molecules are linked by a pair of C-H···O hydrogen bonds forming inversion dimers (Table 1 and Fig. 2).

2. Experimental

Ninhydrin (1 mmol) and 1, 2-phenylenediamine (1 mmol) were mixed and stirred with 10 ml of methanol for 10 min. To this mixture 1 mmol of thioproline and 1-ferrocenyl-3-(4-fluoro) phenyl prop-2-ene-1-one dipolarophile were added and the mixture was refluxed up to the end of the reaction as observed by TLC. The solvent content from the mixture was removed under reduced pressure and the crude product was obtained. Using column chromatography the crude extract was purified with a 4:1 ratio of petroleum ether and ethyl acetate. Finally, single crystals suitable for the X-ray diffraction were obtained by slow evaporation of the solvent at room temperature.

3. Refinement

Hydrogen atoms were placed in calculated positions with C—H ranging from 0.93 - 0.98 Å and refined using the riding model approximation with $U_{iso}(H) = 1.2U_{eq}(C)$.

Computing details

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT* (Bruker, 2008); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008) and *PLATON* (Spek, 2009).



Figure 1

The molecular structure of the title molecule, with atom labelling. Displacement ellipsoids are drawn at the 30% probability level.



Figure 2

The crystal packing of the title compound viewed along the *a* axis. The C-H···O hydrogen bonds are shown as dashed lines (see Table 1 for details; H-atoms not involved in hydrogen bonding have been omitted for clarity).

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Crystal data	
$[Fe(C_5H_5)(C_{32}H_{23}FN_3OS)]$	Z = 2
$M_r = 637.53$	F(000) = 660
Triclinic, $P\overline{1}$	$D_{\rm x} = 1.455 {\rm ~Mg} {\rm ~m}^{-3}$
Hall symbol: -P 1	Mo <i>K</i> α radiation, $\lambda = 0.71073$ Å
a = 8.7097 (2) Å	Cell parameters from 5966 reflections
b = 12.6456 (3) Å	$\theta = 1.5 - 26.5^{\circ}$
c = 13.5477 (4) Å	$\mu = 0.63 \text{ mm}^{-1}$
$\alpha = 83.865 (1)^{\circ}$	T = 293 K
$\beta = 79.008 \ (1)^{\circ}$	Block, colourless
$\gamma = 86.776 \ (1)^{\circ}$	$0.30 \times 0.25 \times 0.20$ mm
V = 1455.39 (6) Å ³	
Data collection	
Bruker SMART APEXII area-detector	Absorption correction: multi-scan
diffractometer	(SADABS; Bruker, 2008)
Radiation source: fine-focus sealed tube	$T_{\min} = 0.833, \ T_{\max} = 0.884$
Graphite monochromator	22064 measured reflections
ω and φ scans	5966 independent reflections
	5126 reflections with $I > 2\sigma(I)$

$R_{\rm int} = 0.022$	$k = -15 \rightarrow 15$
$\theta_{\rm max} = 26.5^{\circ}, \theta_{\rm min} = 1.5^{\circ}$	$l = -16 \rightarrow 15$
$h = -10 \rightarrow 10$	

Kejinemeni	
Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.035$	Hydrogen site location: inferred from
$wR(F^2) = 0.097$	neighbouring sites
S = 1.06	H-atom parameters constrained
5966 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0495P)^2 + 0.4866P]$
397 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.001$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.33 \text{ e} \text{ Å}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.34 \text{ e} \text{ Å}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F², conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F² are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
C1	0.2262 (2)	-0.05617 (15)	0.90688 (16)	0.0455 (5)	
C2	0.1056 (3)	-0.12677 (18)	0.9492 (2)	0.0610 (6)	
H2	0.0586	-0.1263	1.0168	0.073*	
C3	0.0579 (3)	-0.19557 (19)	0.8913 (2)	0.0676 (7)	
H3	-0.0236	-0.2404	0.9195	0.081*	
C4	0.1296 (3)	-0.19979 (19)	0.7904 (2)	0.0645 (6)	
H4	0.0975	-0.2486	0.7525	0.077*	
C5	0.2470 (3)	-0.13261 (17)	0.74675 (18)	0.0545 (5)	
H5	0.2949	-0.1363	0.6797	0.065*	
C6	0.2949 (2)	-0.05802 (14)	0.80348 (15)	0.0422 (4)	
C7	0.4407 (2)	0.08107 (14)	0.81340 (13)	0.0356 (4)	
C8	0.3754 (2)	0.08071 (14)	0.91808 (14)	0.0390 (4)	
C9	0.4426 (2)	0.16713 (15)	0.95772 (14)	0.0402 (4)	
C10	0.4181 (3)	0.19841 (19)	1.05442 (16)	0.0550 (5)	
H10	0.3498	0.1625	1.1069	0.066*	
C11	0.4977 (3)	0.2844 (2)	1.07091 (18)	0.0645 (6)	
H11	0.4824	0.3069	1.1353	0.077*	
C12	0.5997 (3)	0.3376 (2)	0.99315 (18)	0.0595 (6)	
H12	0.6522	0.3951	1.0061	0.071*	
C13	0.6249 (2)	0.30632 (17)	0.89627 (16)	0.0476 (5)	
H13	0.6946	0.3417	0.8443	0.057*	
C14	0.5441 (2)	0.22119 (15)	0.87846 (14)	0.0376 (4)	
C15	0.5453 (2)	0.17673 (14)	0.77843 (13)	0.0347 (4)	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

C16	0.4710(2)	0.25222 (13)	0.69710 (13)	0.0323 (4)
H16	0.3694	0.2249	0.6934	0.039*
C17	0.5827 (2)	0.24183 (14)	0.59451 (13)	0.0346 (4)
H17	0.6462	0.3051	0.5782	0.042*
C18	0.6895 (2)	0.14625 (14)	0.61692 (14)	0.0368 (4)
H18	0.6387	0.0800	0.6128	0.044*
C19	0.8574 (2)	0.14255 (17)	0.55980 (16)	0.0468 (5)
H19A	0.8969	0.2137	0.5426	0.056*
H19B	0.8651	0.1082	0.4983	0.056*
C20	0.8016 (2)	0.06956 (18)	0.75419 (16)	0.0507 (5)
H20A	0.7472	0.0032	0.7670	0.061*
H20B	0.8373	0.0836	0.8150	0.061*
C21	0.4942 (2)	0.23492 (14)	0.50971 (13)	0.0355 (4)
C22	0.3981 (2)	0.14988 (15)	0.51023 (15)	0.0432 (4)
H22	0.3942	0.0939	0.5610	0.052*
C23	0.3088 (2)	0.14764 (16)	0.43653 (16)	0.0472 (5)
H23	0.2447	0.0911	0.4372	0.057*
C24	0.3172 (2)	0.23083 (17)	0.36242 (15)	0.0452 (5)
C25	0.4097 (2)	0.31543 (16)	0.35810 (15)	0.0443 (4)
H25	0.4131	0.3707	0.3067	0.053*
C26	0.4984 (2)	0.31637 (14)	0.43271 (14)	0.0394 (4)
H26	0.5624	0.3732	0.4309	0.047*
C27	0.4460 (2)	0.36722 (14)	0.72321 (14)	0.0350 (4)
C28	0.3001 (2)	0.39424 (14)	0.79037 (14)	0.0368 (4)
C29	0.1728 (2)	0.32643 (16)	0.83307 (14)	0.0435 (4)
H29	0.1722	0.2493	0.8302	0.052*
C30	0.0485 (3)	0.39089 (19)	0.88076 (15)	0.0517 (5)
H30	-0.0544	0.3662	0.9153	0.062*
C31	0.0960 (3)	0.49664 (19)	0.86815 (16)	0.0541 (5)
H31	0.0314	0.5580	0.8921	0.065*
C32	0.2502 (2)	0.50038 (16)	0.81311 (16)	0.0451 (5)
H32	0.3120	0.5642	0.7933	0.054*
C33	0.0960 (3)	0.3753 (2)	0.60022 (16)	0.0540 (5)
H33	0.1426	0.3063	0.5813	0.065*
C34	-0.0535 (3)	0.3921 (2)	0.65521 (18)	0.0628 (6)
H34	-0.1308	0.3376	0.6810	0.075*
C35	-0.0738 (3)	0.4998 (3)	0.6665 (2)	0.0795 (9)
H35	-0.1688	0.5351	0.7013	0.095*
C36	0.0650 (4)	0.5504 (2)	0.6174 (2)	0.0794 (9)
H36	0.0834	0.6268	0.6118	0.095*
C37	0.1690 (3)	0.4720 (2)	0.57671 (16)	0.0591 (6)
H37	0.2750	0.4831	0.5380	0.071*
N1	0.2691 (2)	0.01468 (13)	0.96630 (13)	0.0475 (4)
N2	0.40502 (19)	0.01406 (12)	0.75560 (12)	0.0413 (4)
N3	0.70264 (17)	0.15638 (12)	0.72110 (11)	0.0372 (3)
01	0.54339 (16)	0.43293 (10)	0.68911 (12)	0.0518 (4)
F1	0.22731 (18)	0.22959 (12)	0.29030 (10)	0.0684 (4)
Fe1	0.10914 (3)	0.44031 (2)	0.730212 (19)	0.03701 (9)
S1	0.96553 (7)	0.06529 (5)	0.64716 (5)	0.05986 (16)

Atomic displacement parameters $(Å^2)$

	U^{11}	U ²²	U ³³	U^{12}	U ¹³	U ²³
C1	0.0461 (11)	0.0371 (10)	0.0505 (12)	-0.0021 (8)	-0.0049 (9)	0.0013 (8)
C2	0.0598 (14)	0.0514 (12)	0.0649 (15)	-0.0110 (11)	0.0037 (12)	0.0025 (11)
C3	0.0600 (14)	0.0517 (13)	0.088 (2)	-0.0202 (11)	-0.0078 (13)	0.0027 (13)
C4	0.0757 (16)	0.0484 (12)	0.0747 (17)	-0.0171 (11)	-0.0247 (14)	-0.0036 (11)
C5	0.0651 (14)	0.0446 (11)	0.0559 (13)	-0.0098 (10)	-0.0135 (11)	-0.0058 (10)
C6	0.0463 (10)	0.0338 (9)	0.0462 (11)	-0.0021 (8)	-0.0103 (9)	0.0014 (8)
C7	0.0385 (9)	0.0335 (9)	0.0338 (9)	0.0008 (7)	-0.0060 (7)	-0.0007 (7)
C8	0.0428 (10)	0.0370 (9)	0.0353 (10)	0.0016 (8)	-0.0051 (8)	-0.0006 (7)
C9	0.0431 (10)	0.0436 (10)	0.0335 (10)	0.0019 (8)	-0.0069 (8)	-0.0036 (8)
C10	0.0623 (13)	0.0670 (14)	0.0336 (11)	-0.0057 (11)	-0.0012 (10)	-0.0078 (10)
C11	0.0735 (16)	0.0840 (17)	0.0392 (12)	-0.0091 (13)	-0.0060 (11)	-0.0247 (12)
C12	0.0607 (14)	0.0718 (15)	0.0517 (13)	-0.0141 (11)	-0.0117 (11)	-0.0233 (11)
C13	0.0452 (11)	0.0581 (12)	0.0421 (11)	-0.0096 (9)	-0.0082 (9)	-0.0121 (9)
C14	0.0364 (9)	0.0425 (10)	0.0343 (9)	0.0014 (7)	-0.0073 (7)	-0.0064 (8)
C15	0.0368 (9)	0.0353 (9)	0.0316 (9)	-0.0027 (7)	-0.0046 (7)	-0.0047 (7)
C16	0.0341 (9)	0.0316 (8)	0.0320 (9)	-0.0037 (7)	-0.0065 (7)	-0.0047 (7)
C17	0.0384 (9)	0.0322 (8)	0.0328 (9)	-0.0046 (7)	-0.0040 (7)	-0.0037 (7)
C18	0.0396 (9)	0.0357 (9)	0.0344 (9)	-0.0013 (7)	-0.0034 (7)	-0.0062 (7)
C19	0.0440 (11)	0.0502 (11)	0.0438 (11)	0.0028 (9)	-0.0001 (9)	-0.0103 (9)
C20	0.0477 (12)	0.0574 (12)	0.0443 (12)	0.0095 (9)	-0.0073 (9)	-0.0013 (10)
C21	0.0388 (9)	0.0349 (9)	0.0319 (9)	-0.0011 (7)	-0.0021 (7)	-0.0078 (7)
C22	0.0546 (11)	0.0359 (9)	0.0390 (10)	-0.0070 (8)	-0.0077 (9)	-0.0031 (8)
C23	0.0526 (12)	0.0425 (10)	0.0488 (12)	-0.0091 (9)	-0.0085 (9)	-0.0135 (9)
C24	0.0493 (11)	0.0538 (11)	0.0356 (10)	0.0021 (9)	-0.0114 (8)	-0.0148 (9)
C25	0.0542 (12)	0.0428 (10)	0.0346 (10)	0.0006 (9)	-0.0071 (9)	-0.0018 (8)
C26	0.0447 (10)	0.0358 (9)	0.0366 (10)	-0.0048 (8)	-0.0031 (8)	-0.0048 (7)
C27	0.0351 (9)	0.0339 (9)	0.0383 (10)	-0.0012 (7)	-0.0113 (7)	-0.0057 (7)
C28	0.0397 (9)	0.0393 (9)	0.0336 (9)	-0.0001 (7)	-0.0112 (8)	-0.0069 (7)
C29	0.0433 (10)	0.0483 (11)	0.0363 (10)	-0.0001 (8)	-0.0041 (8)	0.0006 (8)
C30	0.0438 (11)	0.0721 (14)	0.0349 (11)	0.0040 (10)	-0.0002 (9)	-0.0019 (10)
C31	0.0577 (13)	0.0649 (14)	0.0405 (11)	0.0150 (11)	-0.0076 (10)	-0.0210 (10)
C32	0.0491 (11)	0.0451 (10)	0.0460 (11)	0.0046 (8)	-0.0153 (9)	-0.0186 (9)
C33	0.0558 (13)	0.0695 (14)	0.0418 (11)	-0.0021 (11)	-0.0155 (10)	-0.0170 (10)
C34	0.0495 (13)	0.0940 (19)	0.0493 (13)	-0.0165 (12)	-0.0147 (10)	-0.0095 (12)
C35	0.0613 (16)	0.124 (3)	0.0568 (16)	0.0405 (17)	-0.0259 (13)	-0.0215 (16)
C36	0.133 (3)	0.0550 (14)	0.0589 (16)	0.0013 (16)	-0.0489 (18)	0.0058 (12)
C37	0.0595 (13)	0.0843 (17)	0.0350 (11)	-0.0204 (12)	-0.0102 (10)	-0.0015 (11)
N1	0.0524 (10)	0.0433 (9)	0.0419 (9)	-0.0045 (7)	0.0020 (8)	0.0001 (7)
N2	0.0482 (9)	0.0367 (8)	0.0386 (9)	-0.0056 (7)	-0.0058 (7)	-0.0030 (7)
N3	0.0353 (8)	0.0419 (8)	0.0336 (8)	0.0018 (6)	-0.0052 (6)	-0.0039 (6)
01	0.0434 (8)	0.0372 (7)	0.0726 (10)	-0.0090 (6)	-0.0001 (7)	-0.0102 (7)
F1	0.0779 (9)	0.0831 (9)	0.0541 (8)	-0.0073 (7)	-0.0334 (7)	-0.0122 (7)
Fe1	0.03496 (15)	0.04285 (16)	0.03363 (15)	0.00073 (11)	-0.00654 (11)	-0.00640 (11)
S 1	0.0445 (3)	0.0693 (4)	0.0606 (4)	0.0151 (3)	-0.0026 (3)	-0.0046 (3)

Geometric parameters (Å, °)

C1—N1	1.377 (3)	С20—Н20А	0.9700
C1—C2	1.413 (3)	C20—H20B	0.9700
C1—C6	1.417 (3)	C21—C26	1.382 (3)
C2—C3	1.362 (4)	C21—C22	1.398 (3)
C2—H2	0.9300	C22—C23	1.381 (3)
C3—C4	1.397 (4)	С22—Н22	0.9300
С3—Н3	0.9300	C23—C24	1.369 (3)
C4—C5	1.369 (3)	С23—Н23	0.9300
C4—H4	0.9300	C24—F1	1.365 (2)
C5—C6	1.406 (3)	C24—C25	1.365 (3)
С5—Н5	0.9300	C25—C26	1.386 (3)
C6—N2	1.380(2)	С25—Н25	0.9300
C7—N2	1.301 (2)	C26—H26	0.9300
C7—C8	1.423 (3)	C27—O1	1.213 (2)
C7—C15	1.535 (2)	C27—C28	1.461 (3)
C8—N1	1.312 (2)	C28—C29	1.434 (3)
C8—C9	1.464 (3)	C28—C32	1.434 (3)
C9—C10	1.383 (3)	C28—Fe1	2.0191 (18)
C9—C14	1.397 (3)	C29—C30	1.414 (3)
C10—C11	1.382 (3)	C29—Fe1	2.025 (2)
C10—H10	0.9300	С29—Н29	0.9800
C11—C12	1.382 (3)	C30—C31	1.405 (3)
С11—Н11	0.9300	C30—Fe1	2.045 (2)
C12—C13	1.385 (3)	С30—Н30	0.9800
C12—H12	0.9300	C31—C32	1.407 (3)
C13—C14	1.385 (3)	C31—Fe1	2.051 (2)
С13—Н13	0.9300	C31—H31	0.9800
C14—C15	1.520 (2)	C32—Fe1	2.0398 (19)
C15—N3	1.464 (2)	C32—H32	0.9800
C15—C16	1.588 (2)	C33—C34	1.388 (3)
C16—C27	1.527 (2)	C33—C37	1.389 (3)
C16—C17	1.551 (2)	C33—Fe1	2.047 (2)
C16—H16	0.9800	C33—H33	0.9800
C17—C21	1.513 (2)	C34—C35	1.385 (4)
C17—C18	1.523 (3)	C34—Fe1	2.047 (2)
С17—Н17	0.9800	C34—H34	0.9800
C18—N3	1.457 (2)	C35—C36	1.415 (5)
C18—C19	1.520 (3)	C35—Fe1	2.025 (2)
C18—H18	0.9800	С35—Н35	0.9800
C19—S1	1.821 (2)	C36—C37	1.390 (4)
C19—H19A	0.9700	C36—Fe1	2.033 (2)
C19—H19B	0.9700	С36—Н36	0.9800
C20—N3	1.444 (2)	C37—Fe1	2.047 (2)
C20—S1	1.834 (2)	С37—Н37	0.9800
	- ()		
N1—C1—C2	119.1 (2)	C30—C29—C28	107.56 (18)
N1—C1—C6	122.18 (17)	C30—C29—Fe1	70.45 (12)
C2—C1—C6	118.7 (2)	C28—C29—Fe1	69.03 (10)

C3—C2—C1	120.3 (2)	С30—С29—Н29	126.2
С3—С2—Н2	119.9	С28—С29—Н29	126.2
C1—C2—H2	119.9	Fe1—C29—H29	126.2
C2—C3—C4	120.9 (2)	C31—C30—C29	108.56 (19)
С2—С3—Н3	119.6	C31—C30—Fe1	70.15 (12)
С4—С3—Н3	119.6	C29—C30—Fe1	68.88 (11)
C5—C4—C3	120.5 (2)	С31—С30—Н30	125.7
C5—C4—H4	119.8	С29—С30—Н30	125.7
C3—C4—H4	119.8	Fe1—C30—H30	125.7
C4—C5—C6	120.0 (2)	C30—C31—C32	108.85 (19)
С4—С5—Н5	120.0	C30—C31—Fe1	69.72 (12)
С6—С5—Н5	120.0	C32—C31—Fe1	69.46 (11)
N2—C6—C5	118.71 (19)	C30—C31—H31	125.6
N2—C6—C1	121.64 (18)	C32—C31—H31	125.6
C5—C6—C1	119.62 (18)	Fe1—C31—H31	125.6
N2—C7—C8	123.79 (17)	C31—C32—C28	107.73 (19)
N2—C7—C15	125.82 (16)	C31—C32—Fe1	70.31 (12)
C8—C7—C15	110.17 (15)	C28—C32—Fe1	68.53 (10)
N1—C8—C7	123.68 (18)	C31—C32—H32	126.1
N1—C8—C9	127.97 (18)	C28—C32—H32	126.1
C7—C8—C9	108.30 (16)	Fe1—C32—H32	126.1
C10-C9-C14	121.14 (19)	C34—C33—C37	109.0 (2)
C10—C9—C8	130.35 (19)	C34—C33—Fe1	70.18 (13)
C14—C9—C8	108.51 (16)	C37—C33—Fe1	70.17 (13)
C11—C10—C9	118.2 (2)	С34—С33—Н33	125.5
C11—C10—H10	120.9	С37—С33—Н33	125.5
С9—С10—Н10	120.9	Fe1—C33—H33	125.5
C12—C11—C10	121.0 (2)	C35—C34—C33	107.6 (2)
C12—C11—H11	119.5	C35—C34—Fe1	69.27 (14)
C10—C11—H11	119.5	C33—C34—Fe1	70.16 (13)
C11—C12—C13	121.0 (2)	С35—С34—Н34	126.2
C11—C12—H12	119.5	С33—С34—Н34	126.2
C13—C12—H12	119.5	Fe1—C34—H34	126.2
C12—C13—C14	118.6 (2)	C34—C35—C36	108.3 (2)
C12—C13—H13	120.7	C34—C35—Fe1	70.97 (14)
C14—C13—H13	120.7	C36—C35—Fe1	69.87 (15)
C13—C14—C9	120.10 (18)	С34—С35—Н35	125.9
C13—C14—C15	128.08 (17)	С36—С35—Н35	125.9
C9—C14—C15	111.79 (16)	Fe1—C35—H35	125.9
N3—C15—C14	113.78 (14)	C37—C36—C35	107.2 (2)
N3—C15—C7	117.55 (15)	C37—C36—Fe1	70.63 (14)
C14—C15—C7	100.86 (14)	C35—C36—Fe1	69.30 (15)
N3—C15—C16	100.78 (13)	С37—С36—Н36	126.4
C14—C15—C16	116.14 (14)	С35—С36—Н36	126.4
C7—C15—C16	108.37 (14)	Fe1-C36-H36	126.4
C27—C16—C17	112.63 (14)	C33—C37—C36	107.9 (2)
C27—C16—C15	112.86 (14)	C33—C37—Fe1	70.15 (13)
C17—C16—C15	105.89 (14)	C36—C37—Fe1	69.52 (14)
С27—С16—Н16	108.4	С33—С37—Н37	126.0

C17—C16—H16	108.4	С36—С37—Н37	126.0
C15—C16—H16	108.4	Fe1—C37—H37	126.0
C21—C17—C18	116.13 (15)	C8—N1—C1	114.09 (17)
C21—C17—C16	112.03 (14)	C7—N2—C6	114.56 (16)
C18—C17—C16	103.55 (14)	C20—N3—C18	108.64 (15)
C21—C17—H17	108.3	C20—N3—C15	120.93 (15)
C18—C17—H17	108.3	C18—N3—C15	108.06 (14)
C16—C17—H17	108.3	C28—Fe1—C29	41.54 (7)
N3—C18—C19	104.67 (15)	C28—Fe1—C35	174.75 (12)
N3—C18—C17	101.91 (14)	C29—Fe1—C35	143.28 (12)
C19—C18—C17	119.10 (16)	C28—Fe1—C36	134.60 (12)
N3—C18—H18	110.2	C29—Fe1—C36	174.29 (12)
C19—C18—H18	110.2	C35—Fe1—C36	40.83 (13)
C17—C18—H18	110.2	C28—Fe1—C32	41.38 (7)
C18—C19—S1	104.45 (14)	C29—Fe1—C32	69.28 (8)
C18—C19—H19A	110.9	C35—Fe1—C32	134.48 (12)
S1—C19—H19A	110.9	C36—Fe1—C32	110.44 (11)
C18—C19—H19B	110.9	C28—Fe1—C30	68.86 (8)
S1—C19—H19B	110.9	C29—Fe1—C30	40.67 (8)
H19A—C19—H19B	108.9	C35—Fe1—C30	113.71 (11)
N3—C20—S1	103.11 (13)	C36—Fe1—C30	144.91 (12)
N3—C20—H20A	111.1	C32—Fe1—C30	68.10 (9)
S1—C20—H20A	111.1	C28—Fe1—C33	115.19 (8)
N3—C20—H20B	111.1	C29—Fe1—C33	110.01 (9)
S1—C20—H20B	111.1	C35—Fe1—C33	66.68 (11)
H20A—C20—H20B	109.1	C36—Fe1—C33	66.88 (11)
C26—C21—C22	118.07 (17)	C32—Fe1—C33	146.57 (9)
C26—C21—C17	120.58 (16)	C30—Fe1—C33	134.19 (10)
C22—C21—C17	121.22 (16)	C28—Fe1—C37	110.62 (9)
C23—C22—C21	121.04 (18)	C29—Fe1—C37	134.77 (10)
С23—С22—Н22	119.5	C35—Fe1—C37	67.38 (11)
C21—C22—H22	119.5	C36—Fe1—C37	39.85 (12)
C24—C23—C22	118.27 (18)	C32—Fe1—C37	116.09 (9)
С24—С23—Н23	120.9	C30—Fe1—C37	173.50 (10)
С22—С23—Н23	120.9	C33—Fe1—C37	39.68 (9)
F1—C24—C25	118.57 (19)	C28—Fe1—C34	144.67 (10)
F1—C24—C23	118.40 (19)	C29—Fe1—C34	113.39 (10)
C25—C24—C23	123.02 (19)	C35—Fe1—C34	39.76 (12)
C24—C25—C26	117.92 (18)	C36—Fe1—C34	67.59 (12)
C24—C25—H25	121.0	C32—Fe1—C34	173.03 (10)
C26—C25—H25	121.0	C30—Fe1—C34	109.37 (10)
C21—C26—C25	121.67 (17)	C33—Fe1—C34	39.65 (9)
C21—C26—H26	119.2	C37—Fe1—C34	67.06 (9)
С25—С26—Н26	119.2	C28—Fe1—C31	68.63 (8)
O1—C27—C28	121.81 (16)	C29—Fe1—C31	68.35 (9)
O1—C27—C16	121.04 (16)	C35—Fe1—C31	110.11 (10)
C28—C27—C16	117.15 (15)	C36—Fe1—C31	115.37 (11)
C29—C28—C32	107.30 (17)	C32—Fe1—C31	40.23 (9)
C29—C28—C27	127.91 (16)	C30—Fe1—C31	40.13 (9)

C32—C28—C27	124.22 (17)	C33—Fe1—C31	172.76 (9)
C29—C28—Fe1	69.43 (11)	C37—Fe1—C31	146.21 (10)
C32—C28—Fe1	70.08 (11)	C34—Fe1—C31	133.81 (10)
C27—C28—Fe1	119.05 (13)	C19—S1—C20	93.34 (9)
N1—C1—C2—C3	178.2 (2)	C29—C28—Fe1—C30	37.83 (12)
C6—C1—C2—C3	0.5 (3)	C32—C28—Fe1—C30	-80.45 (13)
C1—C2—C3—C4	1.8 (4)	C27—C28—Fe1—C30	160.67 (16)
C2—C3—C4—C5	-1.7 (4)	C29—C28—Fe1—C33	-92.26 (13)
C3—C4—C5—C6	-0.6 (4)	C32—C28—Fe1—C33	149.46 (13)
C4—C5—C6—N2	-175.1 (2)	C27—C28—Fe1—C33	30.58 (17)
C4—C5—C6—C1	2.8 (3)	C29—C28—Fe1—C37	-135.23 (13)
N1—C1—C6—N2	-2.4(3)	C32—C28—Fe1—C37	106.49 (14)
C2-C1-C6-N2	175.1 (2)	C27—C28—Fe1—C37	-12.39(17)
N1—C1—C6—C5	179.65 (19)	C29—C28—Fe1—C34	-56.14 (19)
C2—C1—C6—C5	-2.8(3)	C32—C28—Fe1—C34	-174.42 (15)
N2—C7—C8—N1	-2.1(3)	C27—C28—Fe1—C34	66.7 (2)
C15—C7—C8—N1	172.84 (18)	C29—C28—Fe1—C31	81.04 (13)
N2—C7—C8—C9	-179.87(17)	C27—C28—Fe1—C31	-156.12 (17)
C15—C7—C8—C9	-4.9(2)	C30—C29—Fe1—C28	118.62 (17)
N1-C8-C9-C10	3.1 (4)	C30—C29—Fe1—C35	-58.1(2)
C7—C8—C9—C10	-179.2(2)	C28—C29—Fe1—C35	-176.72(16)
N1 - C8 - C9 - C14	-176.19(19)	C_{30} C_{29} F_{e1} C_{32}	80.12 (14)
C7-C8-C9-C14	15(2)	C_{28} C_{29} F_{e1} C_{32}	-3850(11)
C14—C9—C10—C11	-0.5(3)	C_{28} C_{29} F_{e1} C_{30}	-118.62(17)
C8-C9-C10-C11	-179.7(2)	C30—C29—Fe1—C33	-135.60(14)
C9-C10-C11-C12	-0.3(4)	C28—C29—Fe1—C33	105.78 (12)
C_{10} C_{11} C_{12} C_{13}	0.2(4)	C_{30} C_{29} F_{e1} C_{37}	-17320(13)
C_{11} $-C_{12}$ $-C_{13}$ $-C_{14}$	0.8(4)	C_{28} — C_{29} — F_{e1} — C_{37}	68.19 (16)
C_{12} C_{13} C_{14} C_{9}	-1.5(3)	C_{30} C_{29} F_{e1} C_{34}	-92.93(15)
C_{12} C_{13} C_{14} C_{15}	176.3 (2)	C_{28} C_{29} F_{e1} C_{34}	148.45 (12)
C10-C9-C14-C13	1.4(3)	C_{30} C_{29} F_{e1} C_{31}	36.85 (13)
C8-C9-C14-C13	-179.18(18)	C_{28} C_{29} F_{e1} C_{31}	-81.77(13)
C10-C9-C14-C15	-17675(19)	C_{34} C_{35} F_{e1} C_{29}	-550(2)
C8 - C9 - C14 - C15	2.6(2)	C_{36} C_{35} F_{e1} C_{29}	-17367(16)
C_{13} $-C_{14}$ $-C_{15}$ $-N_{3}$	50.0(3)	C34-C35-Fe1-C36	118.6 (2)
C9-C14-C15-N3	-132.05(16)	C34-C35-Fe1-C32	-174.18(14)
C_{13} C_{14} C_{15} C_{7}	176 76 (19)	C_{36} C_{35} F_{e1} C_{32}	67 2 (2)
C9-C14-C15-C7	-52(2)	C34-C35-Fe1-C30	-92.21(17)
C_{13} C_{14} C_{15} C_{16}	-664(3)	C_{36} C_{35} F_{e1} C_{30}	149 15 (16)
C9-C14-C15-C16	111 59 (18)	C34-C35-Fe1-C33	37 56 (15)
N2-C7-C15-N3	-54.8(2)	C_{36} C_{35} Fe_{1} C_{33}	-81.07(18)
C8-C7-C15-N3	130.35 (17)	C34—C35—Fe1—C37	80.82 (16)
N2-C7-C15-C14	-179.12(18)	C_{36} — C_{35} — F_{e1} — C_{37}	-37.81(16)
C8-C7-C15-C14	6.08 (19)	C36—C35—Fe1—C34	-118.6(2)
N2-C7-C15-C16	58.4 (2)	C34-C35-Fe1-C31	-13545(16)
C8-C7-C15-C16	-116.36(16)	C36-C35-Fe1-C31	105.91 (18)
N3-C15-C16-C27	-109.51(15)	C37-C36-Fe1-C28	65.81 (19)
C14—C15—C16—C27	13.9 (2)	C35—C36—Fe1—C28	-176.23 (15)
	\ /		- (-)

$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$106.61 (15) \\ -135.43 (17) \\ -172.70 (15) \\ -54.7 (3) \\ -37.41 (14) \\ 80.55 (18) \\ 118.0 (2) \\ -80.57 (16) \\ 37.39 (16) \\ 150.14 (14) \\ -91.90 (18) \\ -119.23 (18) \\ -80.59 (14) \\ 38.64 (11) \\ 65.5 (2) \\ -175.29 (15) \\ 125.5 (17) \\ 125.5 $
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} -135.43 (17) \\ -172.70 (15) \\ -54.7 (3) \\ -37.41 (14) \\ 80.55 (18) \\ 118.0 (2) \\ -80.57 (16) \\ 37.39 (16) \\ 150.14 (14) \\ -91.90 (18) \\ -119.23 (18) \\ -80.59 (14) \\ 38.64 (11) \\ 65.5 (2) \\ -175.29 (15) \\ 125.5 (2) \end{array}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-172.70(15) -54.7(3) -37.41(14) 80.55(18) 118.0(2) -80.57(16) 37.39(16) 150.14(14) -91.90(18) -119.23(18) -80.59(14) 38.64(11) 65.5(2) -175.29(15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} -54.7 (3) \\ -37.41 (14) \\ 80.55 (18) \\ 118.0 (2) \\ -80.57 (16) \\ 37.39 (16) \\ 150.14 (14) \\ -91.90 (18) \\ -119.23 (18) \\ -80.59 (14) \\ 38.64 (11) \\ 65.5 (2) \\ -175.29 (15) \\ 105.5 (217) \end{array}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-37.41 (14) 80.55 (18) 118.0 (2) -80.57 (16) 37.39 (16) 150.14 (14) -91.90 (18) -119.23 (18) -80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15) 125.50 (17)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	80.55 (18) $118.0 (2)$ $-80.57 (16)$ $37.39 (16)$ $150.14 (14)$ $-91.90 (18)$ $-119.23 (18)$ $-80.59 (14)$ $38.64 (11)$ $65.5 (2)$ $-175.29 (15)$ $125 (50 (17))$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	118.0 (2) -80.57 (16) -80.57 (16) -150.14 (14) -91.90 (18) -119.23 (18) -80.59 (14) -80.59 (14) -80.59 (14) -80.59 (14) -80.55 (2) -175.29 (15) -1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-80.57 (16) 37.39 (16) 150.14 (14) -91.90 (18) -119.23 (18) -80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	37.39 (16) 150.14 (14) -91.90 (18) -119.23 (18) -80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	150.14 (14) -91.90 (18) -119.23 (18) -80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-91.90 (18) -119.23 (18) -80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15)
N3—C18—C19—S137.60 (16)C31—C32—Fe1—C28C17—C18—C19—S1150.55 (14)C31—C32—Fe1—C29C18—C17—C21—C26 $-128.77 (18)$ C28—C32—Fe1—C29C16—C17—C21—C26112.55 (18)C31—C32—Fe1—C35C16—C17—C21—C2255.4 (2)C28—C32—Fe1—C35C16—C17—C21—C22 $-63.2 (2)$ C31—C32—Fe1—C36C16—C17—C21—C22 $-63.2 (2)$ C31—C32—Fe1—C36C16—C17—C21—C22—C23 $-0.6 (3)$ C28—C32—Fe1—C36C17—C21—C22—C23 $175.32 (18)$ C31—C32—Fe1—C30C21—C22—C23—C24 $0.3 (3)$ C28—C32—Fe1—C30C22—C23—C24—F1 $-179.06 (18)$ C31—C32—Fe1—C33C22—C23—C24 $0.1 (2)$ C28—C32—Fe1—C33	-119.23 (18) -80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-80.59 (14) 38.64 (11) 65.5 (2) -175.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	38.64 (11) 65.5 (2) -175.29 (15)
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$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-175.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	105 50 (17)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	105.50(1/)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-135.27 (15)
C21-C22-C23-C24 $0.3 (3)$ $C28-C32-Fe1-C30$ 8 $C22-C23-C24-F1$ $-179.06 (18)$ $C31-C32-Fe1-C33$ $-1720-C33-C24-C33$ $C22-C23-C24-F1$ $-179.06 (18)$ $C31-C32-Fe1-C33$ $-1720-C33-C32-Fe1-C33$	-36.80 (13)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	82.42 (13)
(2) (2) (2) (2) (2) (1) (2) (2) (2) (2) (2) (2)	-175.80 (16)
$U_{22} - U_{23} - U_{24} - U_{23} = U_{11} + U_{13} = U_{23} - U_{23} - U_{24} - U_{23} = U_{23} - U$	-56.6 (2)
F1-C24-C25-C26 179.02 (17) C31-C32-Fe1-C37 1	148.64 (15)
C_{23} — C_{24} — C_{25} — C_{26} -0.2 (3) C_{28} — C_{32} —Fe1—C37 $-$	-92.14 (14)
C22-C21-C26-C25 0.5 (3) $C28-C32-Fe1-C31$ 1	119.23 (18)
$C_{17}-C_{21}-C_{26}-C_{25}$ -175.38 (17) $C_{31}-C_{30}-F_{e1}-C_{28}$ 8	81.52 (14)
C24-C25-C26-C21 -0.2 (3) $C29-C30-Fe1-C28$ -	-38.62 (12)
C17—C16—C27—O1 -25.5 (2) C31—C30—Fe1—C29 1	120.15 (19)
C15-C16-C27-O1 94.4 (2) C31-C30-Fe1-C35 -	-93.52 (17)
C17-C16-C27-C28 154.34 (15) C29-C30-Fe1-C35 1	146.33 (16)
C15-C16-C27-C28 -85.81 (19) C31-C30-Fe1-C36 -	-57.9 (2)
01-C27-C28-C29 177.49 (19) C29-C30-Fe1-C36 -	-178.00 (18)
C_{16} C_{27} C_{28} C_{29} -2.3 (3) C_{31} C_{30} F_{e1} C_{32} 3	36.89 (13)
01-C27-C28-C32 7.3 (3) C29-C30-Fe1-C32 -	-83.26 (13)
C16—C27—C28—C32 -172.54 (17) C31—C30—Fe1—C33 -	-173.38 (13)
01-C27-C28-Fe1 91.9 (2) C29-C30-Fe1-C33 6	66.47 (17)
C16—C27—C28—Fe1 -87.91 (17) C31—C30—Fe1—C34 -	-136.17 (14)
C32-C28-C29-C30 -0.1 (2) C29-C30-Fe1-C34 1	103.68 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19)
C32-C28-C29-C30 -0.1 (2) $C29-C30-Fe1-C34$ 1 $C27-C28-C29-C30$ -171.60 (18) $C29-C30-Fe1-C31$ -171.60 (18) $Fe1-C28-C29-C30$ -60.19 (14) $C34-C33-Fe1-C28$ 1	103.68 (15) -120.15 (19) 147.72 (15)
C32-C28-C29-C30 -0.1 (2) $C29-C30-Fe1-C34$ 1 $C27-C28-C29-C30$ -171.60 (18) $C29-C30-Fe1-C31$ -171.60 (18) $Fe1-C28-C29-C30$ -60.19 (14) $C34-C33-Fe1-C28$ 1 $C32-C28-C29-Fe1$ 60.13 (13) $C37-C33-Fe1-C28$ $-100-C28-C28-C28$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16) 102.87 (16)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16) 102.87 (16) -137.29 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16) 102.87 (16) -137.29 (15) -37.66 (18)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16) 102.87 (16) -137.29 (15) -37.66 (18) 82.17 (19)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16) 102.87 (16) -137.29 (15) -37.66 (18) 82.17 (19) -82.27 (19)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	103.68 (15) -120.15 (19) 147.72 (15) -92.45 (16) 102.87 (16) -137.29 (15) -37.66 (18) 82.17 (19) -82.27 (19) 37.56 (17)

C30—C31—C32—C28	0.2 (2)	C37—C33—Fe1—C32	-54.9 (2)
Fe1—C31—C32—C28	-58.50 (13)	C34—C33—Fe1—C30	63.4 (2)
C30-C31-C32-Fe1	58.73 (15)	C37—C33—Fe1—C30	-176.78 (14)
C29—C28—C32—C31	-0.1 (2)	C34—C33—Fe1—C37	-119.8 (2)
C27—C28—C32—C31	171.83 (17)	C37—C33—Fe1—C34	119.8 (2)
Fe1—C28—C32—C31	59.62 (14)	C33—C37—Fe1—C28	104.99 (14)
C29—C28—C32—Fe1	-59.72 (13)	C36—C37—Fe1—C28	-136.05 (17)
C27—C28—C32—Fe1	112.22 (18)	C33—C37—Fe1—C29	63.85 (18)
C37—C33—C34—C35	-0.3 (3)	C36—C37—Fe1—C29	-177.19 (16)
Fe1—C33—C34—C35	59.40 (17)	C33—C37—Fe1—C35	-80.23 (18)
C37—C33—C34—Fe1	-59.65 (16)	C36—C37—Fe1—C35	38.72 (18)
C33—C34—C35—C36	0.2 (3)	C33—C37—Fe1—C36	-119.0(2)
Fe1—C34—C35—C36	60.20(18)	C_{33} C_{37} Fe_{1} C_{32}	149.89(14)
C_{33} — C_{34} — C_{35} —Fe1	-59.96 (16)	C_{36} C_{37} F_{e1} C_{32}	-91.16(18)
C_{34} C_{35} C_{36} C_{37}	-0.1(3)	$C_{36} - C_{37} - Fe_{1} - C_{33}$	1190(2)
Fe1-C35-C36-C37	60 75 (17)	C_{33} C_{37} Fe_{1} C_{34}	-36.95(15)
C_{34} C_{35} C_{36} E_{e1}	-60.88(18)	C_{36} C_{37} E_{e1} C_{34}	82 01 (19)
$C_{34} = C_{33} = C_{30} = C_{30}$	0.2(3)	C_{33} C_{37} F_{e1} C_{31}	-172.93(16)
$E_{2} = \frac{1}{2} \frac{1}$	-59.49(16)	$C_{35} = C_{37} = 1 C_{10} = C_{31}$	-54.0(2)
$C_{24} = C_{23} = C_{27} = C_{20}$	59.49 (10)	$C_{30} = C_{37} = P_{C1} = C_{31}$	-175 30 (16)
$C_{34} = C_{35} = C_{37} = C_{37}$	39.00(10)	$C_{33} = C_{34} = C_{12} = C_{28}$	-567(2)
$E_{33} = C_{30} = C_{37} = C_{33}$	50.88(16)	$C_{35} = C_{34} = 1 = C_{28}$	30.7(2)
$C_{25} = C_{26} = C_{27} = C_{25}$	-50.00(17)	$C_{33} = C_{34} = F_{c1} = C_{29}$	-02.57(16)
C_{3} C_{3} C_{3} C_{3} C_{3} C_{3} C_{3}	-39.90(17)	$C_{22} = C_{24} = Fe_1 = C_{25}$	-93.37(10)
C = C = N = C	0.7(3)	$C_{33} = C_{34} = Fe_1 = C_{33}$	110.7(2)
$C_{2} = C_{3} = N_{1} = C_{1}$	1/8.01(18) 17(14(10))	$C_{33} = C_{34} = FeI = C_{30}$	-38.37(18)
$C_2 = C_1 = N_1 = C_8$	-1/6.14(19)	$C_{33} = C_{34} = Fe1 = C_{30}$	80.32 (18)
C_{0} C_{1} N_{1} C_{8}	1.4 (3)	C35—C34—FeI—C30	104.12 (18)
C8 - C7 - N2 - C6	1.1 (3)	C33—C34—FeI—C30	-13/.19(15)
C15 - C7 - N2 - C6	-1/3.0/(1/)	C35—C34—FeI—C33	-118.7(2)
C_{5} — C_{6} — N_{2} — C_{7}	1/8.99 (18)	C35—C34—FeI—C37	-81./1(18)
C1 - C6 - N2 - C7	1.1 (3)	C33—C34—FeI—C37	36.98 (16)
SI-C20-N3-C18	44.82 (17)	C35—C34—FeI—C31	65.9 (2)
S1—C20—N3—C15	170.52 (13)	C33—C34—Fe1—C31	-175.40 (14)
C19—C18—N3—C20	-55.58 (19)	C30—C31—Fe1—C28	-82.14 (13)
C17—C18—N3—C20	179.74 (15)	C32—C31—Fe1—C28	38.28 (12)
C19—C18—N3—C15	171.53 (14)	C30—C31—Fe1—C29	-37.32 (13)
C17—C18—N3—C15	46.85 (17)	C32—C31—Fe1—C29	83.10 (13)
C14—C15—N3—C20	71.3 (2)	C30—C31—Fe1—C35	103.31 (17)
C7—C15—N3—C20	-46.3 (2)	C32—C31—Fe1—C35	-136.27 (16)
C16—C15—N3—C20	-163.71 (16)	C30—C31—Fe1—C36	147.40 (16)
C14—C15—N3—C18	-162.77 (14)	C32—C31—Fe1—C36	-92.18 (17)
C7—C15—N3—C18	79.71 (18)	C30—C31—Fe1—C32	-120.42 (19)
C16—C15—N3—C18	-37.75 (16)	C32—C31—Fe1—C30	120.42 (19)
C32—C28—Fe1—C29	-118.28 (16)	C30—C31—Fe1—C37	-177.60 (16)
C27—C28—Fe1—C29	122.84 (18)	C32—C31—Fe1—C37	-57.2 (2)
C29—C28—Fe1—C36	-173.88 (14)	C30—C31—Fe1—C34	64.86 (19)
C32—C28—Fe1—C36	67.84 (18)	C32—C31—Fe1—C34	-174.72 (14)
C27—C28—Fe1—C36	-51.0 (2)	C18—C19—S1—C20	-11.28 (15)
C29—C28—Fe1—C32	118.28 (16)	N3-C20-S1-C19	-17.86 (15)

C27—C28—Fe1—C32 -118.88 (19)

Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H····A	D····A	<i>D</i> —H…A
C25—H25…O1 ⁱ	0.93	2.54	3.212 (2)	129

Symmetry code: (i) -x+1, -y+1, -z+1.