

# Anthranilate phosphoribosyltransferase from the hyperthermophilic archaeon *Thermococcus kodakarensis* shows maximum activity with zinc and forms a unique dimeric structure

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#### Keywords

crystal structure; thermophilicity; tryptophan biosynthesis; zinc binding

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Anthranilate phosphoribosyltransferase (TrpD) is involved in tryptophan biosynthesis, catalyzing the transfer of a phosphoribosyl group to anthranilate, leading to the generation of phosphoribosyl anthranilate. TrpD belongs to the phosphoribosyltransferase (PRT) superfamily and is the only member of the structural class IV. X-ray structures of TrpD from seven species have been solved to date. Here, functional and structural characterization of a recombinant TrpD from hyperthermophilic archaeon Thermococcus kodakarensis KOD1 (TkTrpD) was carried out. Contrary to previously characterized Mg<sup>2+</sup>-dependent TrpD enzymes, TkTrpD was found to have a unique divalent cation dependency characterized by maximum activity in the presence of  $Zn^{2+}$  (1580 µmol·min<sup>-1</sup>·mg<sup>-1</sup>, the highest reported for any TrpD) followed by  $Ca^{2+}$  (948 µmol min<sup>-1</sup> mg<sup>-1</sup>) and  $Mg^{2+}$  (711 µmol·min<sup>-1</sup>·mg<sup>-1</sup>). *Tk*TrpD displayed an unusually low thermostability compared to other previously characterized proteins from T. kodakarensis KOD1. The crystal structure of TkTrpD was determined in free form and in the presence of  $Zn^{2+}$  to 1.9 and 2.4 Å resolutions, respectively. TkTrpD structure displayed the typical PRT fold similar to other class IV PRTs, with a small N-terminal α-helical domain and a larger C-terminal  $\alpha/\beta$  domain. Electron densities for  $Zn^{2+}$  were identified at the expected zinc-binding motif, DE(217-218), of the enzyme in each subunit of the dimer. Two additional  $Zn^{2+}$  were found at a new dimer interface formed in the presence of  $Zn^{2+}$ . A fifth  $Zn^{2+}$  was found bound to Glu118 at crystal lattice contacts and a sixth one was ligated with Glu235. Based on the  $TkTrpD-Zn^{2+}$  structure, it is suggested that the formation of a new dimer may be responsible for the higher enzyme activity of TkTrpD in the presence of  $Zn^{2+}$  ions.

#### Abbreviations

LB, Luria–Bertani; PRA, phosphoribosyl anthranilate; PRPP, phosphoribosyl pyrophosphate; PRTs, phosphoribosyltransferases; rmsd, root mean square deviation; TrpD, anthranilate phosphoribosyltransferase.

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Anthranilate phosphoribosyltransferase (TrpD, EC 2. 4.2.18) catalyzes the second step in tryptophan biosynthesis, which involves the transfer of a phosphoribosyl group to anthranilate to generate phosphoribosyl anthranilate (PRA), the basic skeleton of tryptophan (Fig. S1). TrpD belongs to the functional superfamily of phosphoribosyltransferases (PRTs) [1], which play important role in the metabolism of nucleotides and amino acids [2].

Phosphoribosyltransferases have been divided into four different classes on the basis of their tertiary structures [3,4]. Class I has a common  $\alpha/\beta$  fold and comprises uracil, orotate, and purine PRTs. Class II has an N-terminal  $\alpha/\beta$  sandwich domain and a Cterminal  $\alpha/\beta$  TIM barrel domain. This class includes the quinolinate and nicotinic acid PRTs. Class III has a unique domain structure and includes ATP-PRTase. Class IV PRTs are limited to TrpD [5] and exhibit a homodimeric structure and a novel PRT fold, consisting of a small N-terminal α-helical domain connected to a large C-terminal  $\alpha/\beta$  domain by a hinge region [6]. The X-ray structures of TrpD enzymes from Sulfolobus solfataricus (SsTrpD; PDB entry 2GVO) [7]. Pectobacterium carotovorum (PcTrpD; PDB entry 1KHD) [8], Mycobacterium tuberculosis (MtbTrpD; PDB entry 4X5B) [9], Ther*mus thermophilus* (*Tt*TrpD; PDB entry 1V8G; Shimizu and Kunishima, 2004, RIKEN Structural Genomics/Proteomics Initiative, unpublished), Acinetobacter sp. ADP1 (AsTrpD; PDB entry 4YI7; Evans et al., 2015, unpublished), Xanthomonas campestris (XcTrpD; PDB entry 4HKM; Ghosh et al., 2012; New York Structural Genomics Research Consortium, unpublished), and Nostoc sp. (NsTrpD; PDB entry 1VQU; Joint Center for Structural Genomics, 2005, unpublished) have been solved.

Most of PRTs have been shown to utilize  $Mg^{2+}$  as divalent cation for enzyme activity. However, *Salmonella typhimurium* and *P. carotovorum* TrpD enzymes have been found to utilize other metal ions, including  $Mn^{2+}$  and  $Co^{2+}$  in addition to  $Mg^{2+}$ , for enzyme activity [8,10]. These divalent cations have been implicated in phosphoribosyl pyrophosphate (PRPP) complexation, which induces prominent ordering of a conserved Gly-rich loop GTGGD in TrpD [7].

Here, we report the biochemical and structural characterization of TrpD (TkTrpD) from the hyperthermophilic archaeon *Thermococcus kodakarensis* KOD1, an obligate heterotroph that grows optimally at 85 °C and pH 6.5 [11]. The gene encoding TkTrpD was expressed in *Escherichia coli* and the recombinant gene product was purified, characterized, crystallized and its crystal structure was determined in free form as well as in the presence of  $Zn^{2+}$  to 1.9 and 2.4 Å resolutions, respectively. The results would provide a better understanding of the TrpD family of enzymes and help in biotechnological applications to synthesize compounds for use in biochemical assays [12,13]. Moreover, TrpD has also emerged as a potential candidate for biomedical applications. The importance of TrpD has been emphasized by a genome-wide transposon mutagenesis study in *M. tuberculosis* [14], which showed that the enzymes responsible for the biosynthesis of PRPP as well as biosynthetic enzymes that use PRPP, such as TrpD, are essential for mycobacterial growth [14,15].

## **Results and Discussion**

#### Production and purification of TkTrpD

The *Tk*TrpD gene (KEGG entry: TK0253) consists of an open reading frame (ORF) of 978 nucleotides, encoding for a polypeptide of 325 amino acid residues with a theoretical molecular mass of 34346.16 Da and pI of 4.9. *Tk*TrpD was produced in *E. coli* and purified to homogeneity using heat treatment and ion-exchange chromatography. Purified recombinant *Tk*TrpD exhibited a molecular weight of about 36 kDa (Fig. 1), matching the molecular weight calculated from the amino acid sequence. By gel filtration chromatography, the molecular mass of *Tk*TrpD was estimated to be 70 kDa, indicating that *Tk*TrpD is a homodimer in solution (Fig. S2).



**Fig. 1.** Purity of *Tk*TrpD in SDS/PAGE stained with Coomassie Brilliant Blue. Lane 1, Purified *Tk*TrpD eluted after ResQ column chromatography; Lane 2, molecular mass marker (Page Ruler prestained protein ladder # SM 0671, Fermentas).

## Effect of pH and temperature

The optimal pH for TkTrpD activity was found to be 8.5–9.0 (Fig. S3). The effect of temperature on TkTrpD activity was examined at optimal pH. TkTrpD exhibited highest activity at 55 °C (Fig. 2) although the optimal growth temperature of T. kodakarensis is 85 °C. This result is in contrast to most of the enzymes from hyperthermophiles but similar to ribose-5-phosphate pyrophosphokinases from T. kodakarensis [16] and Pvrobaculum calidifontis [12], and phosphoribosyl diphosphate synthase from S. solfataricus [17]. It should be noted that a protective mechanism of protein stabilization in hyperthermophiles has been suggested involving the secretion of small-molecule osmolytes in stressful conditions [18]. A similar mechanism may apply to TkTrpD to increase its stability and prevent unfolding at elevated temperatures.

#### **Cation dependency**

Anthranilate phosphoribosyltransferase enzymes from E. coli, S. typhimurium, Saccharomyces cerevisiae, S. solfataricus, and M. tuberculosis have been reported to be dependent on Mg<sup>2+</sup> for enzymatic activity [15,19,20]. Pectobacterium carotovorum and S. tvphimurium TrpDs have been reported to be activated by Mn<sup>2+</sup> [8,10]. Addition of EDTA completely inhibited the enzymatic activity of TkTrpD, indicating the dependency of the enzyme on metal cations. The effect of various cations on TkTrpD was therefore examined (Fig. 3). Surprisingly, addition of  $Zn^{2+}$  and  $Ca^{2+}$  led to higher specific activities than  $Mg^{2+}$ , whereas  $Cu^{2+}$ ,  $Ni^{2+}$ ,  $Co^{2+}$ , and  $Mn^{2+}$  showed lower activities. The decrease in enzyme activity in the presence of  $Co^{2+}$ 



**Fig. 2.** Optimal temperature for *Tk*TrpD enzymatic activity. The activity assays were conducted in triplicate at pH 8.5 and at various temperatures (35–85 °C).

and  $Mn^{2+}$  may be attributed to the slight precipitation of *Tk*TrpD in the presence of these metal ions.

#### Effect of cation concentration

*Tk*TrpD activity increased with the addition of  $Zn^{2+}$  until the  $Zn^{2+}$  concentration reached 100 μM. Higher concentrations of  $Zn^{2+}$  significantly inhibited the reaction (Fig. 4A). In the case of Mg<sup>2+</sup>, the activity was maximal at 200 μM; however, concentrations above 200 μM also decreased enzyme activity (Fig. 4B).

#### **Kinetic parameters**

The effect of substrate concentration on TkTrpD activity was investigated in the presence of  $Zn^{2+}$ . Anthranilate and PRPP were the two substrates used in the assays. The first substrate, PRPP, was kept constant at 1 mm during the measurement of the kinetic parameters toward anthranilate. Similarly, the second substrate, anthranilate, was kept constant at 4 µM when the kinetic parameters toward PRPP were mea-Anthranilate concentrations above 4 µM sured. resulted in reduced enzymatic activity, suggesting substrate inhibition by anthranilate as observed also in M. tuberculosis TrpD [21]. Apparent  $K_{\rm m}$  values for anthranilate and PRPP were 2.2 µM and 250 µM, respectively (Fig. 5). TkTrpD was highly active with specific activity of 1580 µmol·min<sup>-1</sup>·mg<sup>-1</sup>. To the best of our knowledge, this is the highest enzyme activity for any TrpD reported so far. A comparison of kinetic



Fig. 3. Effect of metal cations on *Tk*TrpD activity. Reactions were performed at pH 8.5 and temperature of 55 °C. Chloride salt of each cation and EDTA were used at 100  $\mu$ M concentration. Each measurement is the average value of three independent experiments.



**Fig. 4.** (A) Effect of  $ZnCl_2$  on TkTrpD activity. The activity assays were conducted with various concentrations of  $ZnCl_2$  at pH 8.5 and temperature of 55 °C in triplicate. (B) Effect of MgCl<sub>2</sub> on TkTrpD enzyme activity. The activity assays were conducted with various concentrations of MgCl<sub>2</sub> at pH 8.5 and temperature of 55 °C in triplicate.

parameters and specific activities of characterized TrpDs from various sources is shown in Table 1.

### Quality of the TkTrpD structure

*Tk*TrpD crystallizes with four molecules (A, B, C, and D) in the asymmetric unit that form two homodimers (A–C and B–D) (Fig. 6). The Matthews coefficient  $V_{\rm M}$  [22] for four molecules in the asymmetric unit is 2.4 Å<sup>3·</sup>Da<sup>-1</sup>, corresponding to a solvent content of ~48.5%. The refined structure shows a root mean square deviation (rmsd) of 0.012 Å and 1.15° from the ideal values of bond lengths and angles, respectively. The observed crystal form of *Tk*TrpD soaked with

ZnCl<sub>2</sub> has a dimer (A, B) in the asymmetric unit. The Matthews coefficient  $V_{\rm M}$  for two molecules in the asymmetric unit is 2.3 Å<sup>3</sup>·Da<sup>-1</sup>, corresponding to a solvent content of ~ 46.2%. As the crystals used for the Zn<sup>2+</sup> soaking had been grown in different conditions, unsoaked crystals were tested and found to have similar space group and cell dimensions as those of the free *Tk*TrpD, suggesting that soaking with Zn<sup>2+</sup> induced a rearrangement of the crystal packing. The refined structure shows an rmsd of 0.010 Å and 1.41° from the ideal values of bond lengths and bond angles, respectively. Detailed statistics of data collection and refinement for both structures are presented in Table 2.



**Fig 5.** Effect of anthranilate and PRPP on *Tk*TrpD activity. (A) and (C) show relative activity, whereas (B) and (D) show Lineweaver–Burk plot of steady-state kinetic analysis. The kinetic parameters were examined at temperature of 55 °C and pH 8.5, in the presence of 100 μm ZnCl<sub>2</sub> in triplicate.

Table 1. Comparison of kinetic parameters of TrpD from various organisms. ND: no data available.

Organisms	K <sub>m</sub> anthranilate (µм)	K <sub>m</sub> PRPP (µм)	Specific activity (µmol <sup>·</sup> min <sup>-1·</sup> mg <sup>-1</sup> )	Reference
Thermococcus kodakarensis	2.2	250	1580	This study
Sulfolobus solfataricus	0.085	180	ND	[2]
Escherichia coli	0.28	50	ND	[39]
Salmonella typhimurium	5.9	3.8	1350	[40]
Pectobacterium carotovorum	ND	ND	24.5	[41]
Serratia marcescens	ND	ND	409	[42]
Aerobacter aerogenes	ND	ND	734	[43]
Neurospora crassa	ND	ND	16	[44]
Salmonella enterica subsp. enterica serovar typhimurium	ND	ND	1.54	[45]
Saccharomyces cerevisiae	1.6	22.4	1.58	[19]
Hansenula henricii	4.6	880	0.4	[46]
Corynebacterium glutamicum	ND	ND	0.049	[47]

## **Overall structure**

TkTrpD structure displays the PRT fold similar to other PRTs. Each TkTrpD molecule consists of 325

residues arranged in two domains (Fig. 7): a small N-terminal  $\alpha$ -helical domain containing four helices and a large C-terminal  $\alpha/\beta$  domain with a central sheet of



**Fig 6.** Ribbon diagram of TkTrpD tetramer in the asymmetric unit. Each subunit of the tetramer is shown in different color with the functional homodimers formed between A–C and B–D. Figure was created using UCSF Chimera [50].

seven  $\beta$ -strands (six parallel and one antiparallel) surrounded by eight  $\alpha$ -helices. A hinge region ( $\alpha$ 4- $\beta$ 1,  $\beta$ 3- $\alpha$ 8, and  $\alpha$ 9- $\beta$ 4) connects the two domains.

The N-terminal domain is involved in dimer formation in TrpD enzymes (*Ss*TrpD, *Mtb*TrpD, *Tt*TrpD, *As*TrpD, *Xc*TrpD, *Ns*TrpD, and *Pc*TrpD). Similarly, *Tk*TrpD subunits also associate with each other at their N-terminal ends through their small  $\alpha$ -helical domains (C-terminal end of  $\alpha 1$ ,  $\alpha 3$ , and  $\alpha 8$ ). In *Ss*TrpD, residues Ile36 and Met47 have been shown to be involved in dimerization. Mutations of these residues resulted in loss of dimeric form with decreased thermal stability [2]. Both of these residues are not conserved in TrpD family. The corresponding residues in *Tk*TrpD are Val31 and Thr42. Analysis of protein– protein interactions with PDBsum [23] shows that Ala35, Thr42 (located at the N and C termini of  $\alpha 3$ , respectively) and Leu162 (C-terminal end of  $\alpha 8$ ) are found to form highest number of inter-subunit interactions, showing that mostly hydrophobic residues are involved in inter-subunit interactions in TkTrpD dimer formation in agreement with dimer formation in other TrpD enzymes.

## **Structural comparison**

Structure-based sequence alignment of TkTrpD (Fig. 8) shows highest homology with SsTrpD (45%) followed by TtTrpD (43%), NsTrpD (39%), XcTrpD (39%), PcTrpD (37%), AsTrpD (36%), and MtbTrpD (34%). Several conserved sequences were found that play an important role in catalysis. When superimposed with SsTrpD, TkTrpD shows rmsd of 1.15 Å for 256 C $\alpha$ -atoms. At the C-terminal end of SsTrpD, there is a small helix of 10 residues which is absent in TkTrpD and all other reported TrpD structures.

#### Active site

Each monomer has an active site in a cleft found in the hinge region. In TkTrpD, substrate (anthranilate + PRPP) binding positions were found conserved as in other TrpDs. A conserved anthranilate binding motif (KHGN(101-104)) was found in  $\beta 2-\alpha 6$  loop. Lys101, in particular, is involved in anthranilate binding and HGN(102-104) in PRPP binding. This motif has been determined to be involved in catalysis in previously determined TrpD structures (e.g., SsTrpD, *Mtb*TrpD). Arg159 in *Tk*TrpD found on helix  $\alpha 8$  is also conserved and in previous structures [7,15] has been shown to be involved in anthranilate binding by forming hydrogen bond to it and is essential for catalytic function. The corresponding residues in MtbTrpD and SsTrpD are Arg193 and Arg164, respectively. A highly conserved Gly-rich sequence GTGGD(74–78) found in TkTrpD in  $\beta$ 1– $\alpha$ 5 loop is considered as a signature motif of TrpD family and is involved in PRPP binding. Identical sequences have been found in MtbTrpD (GTGGD(107-111)) and in SsTrpD (GTGGD(79-83)). The first Gly of this region, in particular, is known to interact with the PPi group of PRPP via its peptide amino group and also with the amino group of anthranilate.

#### Divalent ion binding sites

Metal ions bind to two sites in the TrpD family. The first metal ion binds to pyrophosphate and ribose oxygen atoms of PRPP, and this site is common in PRT superfamily. The second site is specific for TrpD family and involves a conserved DE motif whose residues

<b>Table 2.</b> <i>Ik</i> rpD data collection and refinement statistic
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Data collection	Free <i>Tk</i> TrpD	<i>Tk</i> TrpD-Zn <sup>2+</sup>
Wavelength (Å)	0.96598	1.03320
Beamline	MASSIF-1, ESRF	P13, PETRA III
Detector	PILATUS 2M	PILATUS 6M
Temperature (K)	100	100
Space group	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	P22121
Unit cell, <i>a, b, c</i> (Å)	83.9, 85.6, 180.8	42.6, 81.3, 179.4
α, β, γ (°)	90.0, 90.0, 90.0	90.0, 90.0, 90.0
Mosaicity (°)	0.05	0.19
Resolution range (Å)	49.27-1.91 (1.98-1.91)	48.19-2.42 (2.50-2.42)
Total no. of measurements	453 345	172 488
No. of unique reflections	100 253	24 674
Completeness (%)	98.8 (95.9)	99.8 (99.0)
Multiplicity	4.5 (4.6)	7.0 (7.2)
/s	9.5 (1.5)	9.6 (1.4)
R <sub>meas</sub> [48] (%)	10.2 (113)	17.2 (152)
CC <sub>1/2</sub> [49]	0.997 (0.536)	0.997 (0.467)
Overall B factor from Wilson plot (Å <sup>2</sup> )	28.9	59.0
Refinement		
Resolution	49.27-1.91 (1.98-1.91)	48.19-2.42 (2.50-2.42)
No. of reflections (working/test)	95 240/4930	23 378/1230
R <sub>crvst</sub> /R <sub>free</sub> (%)	18.6/23.5	20.4/25.3
No. of protein atoms	9635	4835
No. of water molecules	971	127
No. of protein ligands	_	9 (6Zn <sup>2+</sup> , 2Na <sup>+</sup> , 1Cl <sup>-</sup> )
rmsd in bond lengths (Å)	0.012	0.010
rmsd in bond angles (°)	1.16	1.41
Residues in most favorable regions (%)	96.0	96.0
Residues in additionally allowed regions (%)	3.3	3.9
Average B factor (Å <sup>2</sup> )		
Protein	35.9	52.6
Water	41.9	52.4
Ligands	_	66.3
PDB id	5NOE	5NOF

<sup>a</sup>Values in parentheses are for the outer resolution shell.

are key to metal binding and are invariant in all TrpD enzymes structurally characterized until now (Table 3).

Soaking of TkTrpD with a ZnCl<sub>2</sub> solution resulted in the identification of a total of six  $Zn^{2+}$  ions in the dimer, while in other TrpDs only four  $Zn^{2+}$  ions per dimer are present. In each subunit of TkTrpD, one  $Zn^{2+}$  ion was found in the primary metal binding site, involving the conserved DE(217-218) motif and Asp78. As PRPP is not present in the structure, no additional Zn<sup>2+</sup> ion was found in the vicinity of the primary metal binding site. The structure of TkTrpD soaked with zinc has a dimer in the asymmetric unit. Gel filtration has also shown that in the presence of  $Zn^{2+}$ , *Tk*TrpD exists as a dimer in solution (Fig. S2). Structural comparison of the Zn<sup>2+</sup>-free and Zn<sup>2+</sup>bound structures at subunit level shows low rmsd between C $\alpha$  atoms (0.63 Å), suggesting no significant changes. Notable changes, however, were identified in

the position of helices  $\alpha 8$  and  $\alpha 9$  that move toward the active site in the Zn<sup>2+</sup>-bound structure. Most importantly, following superposition with the Zn<sup>2+</sup>-free TkTrpD (Fig. 9A), the structure solution of the  $Zn^{2+}$ bound TkTrpD revealed a different arrangement of the two subunits compared to the typical dimer found in other TrpD enzymes. Interestingly, two  $Zn^{2+}$  ions (V and VI) were found at the interface between Glu48 and Glu198 of subunit A and ED(295-296) of subunit B in the dimer. The two zinc ions are close to each other with a distance of 2.9 Å (Fig. 9B) and have similar B-factors (43.7 and 46.0  $\text{\AA}^2$ , respectively). These two additional  $Zn^{2+}$  binding sites in TkTrpD may therefore explain the effect of  $Zn^{2+}$  on TkTrpD by promoting a different dimer formation. At present, we cannot conclude whether this property is shared by this enzyme from other sources as well, or whether it is a unique property of TkTrpD. However, the

**Fig 7.** Ribbon diagram of X-ray crystal structure of *Tk*TrpD. Only one subunit of the homodimer is shown, with the amino acid chain colored from blue at the N terminus to red at the C terminus. Each subunit consists of a small  $\alpha$ -helical domain containing four helices ( $\alpha$ 1,  $\alpha$ 2,  $\alpha$ 3, and  $\alpha$ 4) and a larger C-terminal  $\alpha/\beta$  domain with a central  $\beta$ -sheet containing seven  $\beta$ -strands (six parallel and one antiparallel) surrounded by eight  $\alpha$ -helices. Figure was created using UCSF Chimera.

structure-based alignment (Fig. 8) shows that the ED (295–296) motif is not conserved, and therefore, other TrpDs may be unable to adopt the same dimer arrangement. Sequence variations are also evident for Glu48 and Glu198.

In conclusion, the biochemical and structural characterization of TkTrpD reported here may lead to new strategies to alter TrpD enzymatic activity. The new subunit–subunit interface may play a role in the increased activity of TkTrpD in the presence of  $Zn^{2+}$ . For example, Glu198 belongs to helix  $\alpha 10$  and slight alterations upon  $Zn^{2+}$  binding and dimer rearrangement could be traversed to the active site through the  $\alpha 8$  and  $\alpha 9$  helices. Alternatively, formation of the new dimer may affect the position of helix  $\alpha 8$ , which in the typical TrpD dimer is part of the conventional interface. In the new dimer, helix  $\alpha 8$  becomes free from any interactions with a neighboring subunit, and therefore, it may be able to adopt more favorable positions for substrate binding. Further studies, however, are needed to elucidate the precise role of the  $Zn^{2+}$ -binding sites and their potential direct and indirect effects on the active site of the enzyme.

# **Materials and methods**

Chemicals and materials used in this study were purchased from either Thermo-Fisher Scientific (Leicestershire, UK), Fluka (Buchs, Switzerland), Merck (Darmstadt, Germany), or Sigma-Aldrich (St. Louis, MO, USA). Gene-specific primers were commercially synthesized by Macrogen Inc (Seoul, Korea). *Escherichia coli* strains used were DH5a and BL21 Codon Plus (DE3)-RIL (Stratagene, La Jolla, CA, USA). Luria–Bertani (LB) medium was used for the cultivation of *E. coli* strains.

## Gene cloning

Gene encoding *Tk*TrpD was amplified from genomic DNA of *T. kodakarensis*, using sequence-specific forward F*Tk*-TrpD (CATATGAGCCTTCTTGCGAAGATCGTCGAT GG), which include a *Nde*I recognition site (shown in boldface) and reverse R*Tk*TrpD (TCAGCTTTTTGAGA GGCATGCTATCTCCTC) primers. PCR-amplified gene product was ligated to cloning vector pTZ57R/T. The resultant recombinant plasmid PTZ-*Tk*TrpD was digested with *Nde*I and *Hind*III to liberate *Tk*TrpD, which was cloned into the expression vector pET21a(+) using the same restriction sites. pET-*Tk*TrpD name was assigned to the resultant recombinant expression plasmid. Presence of *Tk*TrpD in the expression plasmid was subsequently confirmed by restriction analysis and DNA sequencing.

## Gene expression and protein purification

*Escherichia coli* BL21 CodonPlus (DE3)-RIL cells were transformed with recombinant pET-TkTrpD. Expression of gene was induced by 0.5 mM isopropyl- $\beta$ -D-thiogalactoside (IPTG). After induction for 6 h, cells grown in LB medium

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**Fig 8.** Structure-based sequence alignment of *Tk*TrpD. The enzymes used are *Ss*TrpD (PDB entry 2GVQ) [7], *Tt*TrpD (PDB entry 1V8G; Shimizu *et al.*, 2004, unpublished), *Ns*TrpD (PDB entry 1VQU; Joint Center for Structural Genomics, 2005, unpublished), *Xc*TrpD (PDB entry 4HKM; Ghosh *et al.*, 2012, unpublished), *Pc*TrpD (PDB entry 1KHD) [8], *As*TrpD (PDB entry 4YI7; Evans *et al.*, 2015, unpublished), and *Mtb*TrpD (PDB entry 4X5B) [9]. Figure was created using ESPRIPT 3.0 [51]. Conserved residues are indicated by white letters on a red background (strictly conserved) or red letters on a white background (global similarity score, 0.7) and framed in blue boxes. Markers indicate residues postulated to be involved in PRPP binding (magenta arrows), anthranilate binding (green stars), metal binding (blue ovals), and dimerization (red boxes). Residues involved in  $Zn^{2+}$  binding at the TrpD– $Zn^{2+}$  dimer interface are shown with brown ovals.

TkTrpD/1-325	α1 20000000	α2	α3 <u>00000000000000</u>	α4 <u>0000000000</u>
TkTrpD/1-325 SsTrpD/3-345 TtTrpD/1-329 NsTrpD/15-362 XcTrpD/1-342 PcTrpD/13-345 AsTrpD/23-369 MtbTrpD/27-378	MSLLAKIVDGKN INEILKKLINKSD MDAVKKAILGEV MYLLQQLIDGES TPQQALQRTIEHRE HQPILEKLFKSQS ICQALNHITKNIH WPQILGRLTDNRD	LSFEEAVELFNELK. LEINEAELAKAIIRGE LSESEAVEVMRALMAG SSGAAELNOGVLSE IFHDEMVDLMRQIMRG MTQEESHQLFAAIVRG LARGQAAWAMDQIMTGN	SD GVLIGAYLALQTKGY VPEILVSAILVALRMKGE MSPVRAAGLLVALSLRGE VVPPLSGAILTALNFKGV SVSDAMVSAILTGLRVKK SUSDAMVSAILTGLRVKK SUSDAMVSAILSMKMKGE ALEDSQLAAALISMKMKGE VATEAQIGALMMGLAMKGE	(TIGE BLAGLARA SKNEIVGFARA SRHEIAAMARA SADBLTGMAEV TIGEIAGAATV STIGEIAGAASA SIDEITAARV TIADEVGELAGV
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mbm	η2 η3	β4 η4	α10 β5	*
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Table 3. Divalent ion binding sites in TrpD family.

lons	Contacts	Reference
<i>Mtb</i> TrpD		
Mg <sup>2+</sup> -I	S119, E252, and PRPP	[15]
Mg <sup>2+</sup> -II	DE(251–252)	
<i>Ss</i> TrpD		
Mg <sup>2+</sup> -I	OH <sup>-</sup> groups of ribose and	[7]
	pyrophosphate oxygens of PRPP	
Mg <sup>2+</sup> -II	DE(223–224)	
<i>Pc</i> TrpD		
Mn <sup>2+</sup> -I	S103, E237, and PRPP	[8]
Mn <sup>2+</sup> -II	DE(236–237)	
<i>Tk</i> TrpD		
Zn <sup>2+</sup> -I	DE(217–218), D78 (subunit A)	This study
Zn <sup>2+</sup> -II	E118 (subunit A at crystal lattice	
	contacts with E118 from a	
	symmetry molecule)	
Zn <sup>2+</sup> -III	E235 (subunit B)	
Zn <sup>2+</sup> -IV	DE(217–218), D78 (subunit B)	
Zn <sup>2+</sup> -V	E198-A and E295-B	
Zn <sup>2+</sup> -VI	ED(295–296)-B, E48-A, E198-A	

were harvested and resuspended in 50 mM Tris/HCl pH 8.5 buffer containing 1 mM DTT, 1 mM PMSF, and 20% v/v glycerol. For purification, soluble portion obtained after sonication was heat-treated at 65 °C for 25 min and centrifuged (15 000 g for 15 min). ÄKTA Purifier chromatography system (GE Healthcare, Uppsala, Sweden) was used for further purification. Heat-treated supernatant was applied to anionexchange QFF (6 mL) column (GE Healthcare) and the recombinant TkTrpD was eluted with a linear gradient of 0-1 M NaCl. Fractions containing TkTrpD were desalted by dialysis against 50 mM Tris/HCl (pH 8.5) buffer containing 1 mM DTT, 1 mM PMSF, and 20% v/v glycerol. Dialyzed TkTrpD samples were applied to Resource Q (1 mL) column (GE Healthcare), and the protein was eluted with a linear gradient of 0-1 M NaCl. Analysis of the purified TkTrpD was performed by SDS/PAGE. Molecular weight and oligomeric nature of TkTrpD were determined by gel filtration chromatography column Superdex 75 10/300 GL attached to AKTA purifier (GE Healthcare). The standard curve was obtained with bovine pancreas chymotrypsinogen A (25 kDa), chicken egg white ovalbumin (48 kDa), and bovine serum albumin (63 kDa). Their gel-phase distribution coefficient  $(K_{av})$  values were calculated and plotted against the log of their molecular weight (Fig. S3). Protein concentration was determined spectrophotometrically at every step of purification by Bradford reagent [24].

#### Enzyme assays

*Tk*TrpD activity was determined fluorometrically by measuring the decrease in the concentration of anthranilic acid. Anthranilic acid reacts with PRPP, leading to the production of PRA (Fig. S1). The initial rate of decrease in anthranilate was measured, as anthranilate is utilized by *Tk*TrpD to form PRA, resulting in a decrease in emission/ fluorescence at 390 nm. The activation and emission wavelengths for anthranilate were 315 and 390 nm, respectively. A standard curve was then used to convert fluorescent intensity to anthranilate concentration. The reaction mixture contained 4  $\mu$ M anthranilate, 1 mM PRPP, 100  $\mu$ M ZnCl<sub>2</sub>, 100 mM Tris/HCl buffer (pH 8.5), and 5  $\mu$ g of *Tk*TrpD. The reaction mixture without PRPP was incubated at 55 °C for 5 min. The reaction started by adding PRPP at 55 °C and continued for 2.5 min. Two control experiments were carried out: one without enzyme and one without PRPP. The half-life of PRPP is 56 min at 60 °C [20], suggesting that at 55 °C and for the time used for the reaction, no significant hydrolysis of PRPP is expected.

#### Effect of temperature, pH, and metal ions

For the measurement of optimal temperature, enzyme assays were performed at various temperatures ranging from 35 to 85 °C keeping the pH constant. For the estimation of optimal pH, assays were performed at various pH values keeping the temperature unchanged at 55 °C. The following buffers were used: Na-phosphate (pH: 6.0–7.0), Tris/HCl (pH: 7.0–9.0), and Na-bicarbonate (pH: 9.0–10.0). The effect of divalent metal ions on the enzyme activity was investigated in the presence of either 50 or 100  $\mu$ M of ZnCl<sub>2</sub>, MgCl<sub>2</sub>, CaCl<sub>2</sub>, MnCl<sub>2</sub>, NiCl<sub>2</sub>, CoCl<sub>2</sub>, and CuCl<sub>2</sub>. In case of EDTA, the final EDTA concentration was 100 and 2.5 mM. The effect of Zn<sup>2+</sup> and Mg<sup>2+</sup> concentration on the enzyme activity was measured in the range of 0–1 mM.

#### Crystallization

Purified TkTrpD was concentrated to 12 mg·mL<sup>-1</sup> in 10 mM Tris/HCl (pH 8.0) buffer containing 0.1 M NaCl and 0.002% (w/v) NaN<sub>3</sub>. PACT screen (Molecular Dimensions, Suffolk, UK) was performed in 96-well plate using the sitting drop vapor diffusion method. Promising crystals found in solution 79 (0.2 M sodium acetate (pH 7.5), 20% (w/v) PEG 3350) were optimized by the hanging-drop vapor diffusion method at 16 °C in Linbro 24-well cell culture plates. The reservoir solution consisted of 0.6 mL of condition 79 mixed with 0.2 mL of MilliQ water and the drops comprised 2 µL of 12 mg·mL<sup>-1</sup> TkTrpD mixed with 2 µL of reservoir solution. Crystals appeared after 1 day and were harvested after ~ 4 weeks for X-ray data collection. Crystals were transferred to a reservoir solution supplemented with 20% v/v glycerol and flash-cooled in liquid N<sub>2</sub>.

### ZnCl<sub>2</sub> crystal soaking

*Tk*TrpD crystals obtained from solution 89 of the PACT screen (0.2 M sodium nitrate, 0.1 M Bistris-propane buffer (pH 8.5), 20% (w/v) PEG 3350) were used after soaking in



**Fig 9.** (A) Superposition of TkTrpD dimers with and without Zn<sup>2+</sup>. Subunit (a) is shown in brown (with Zn<sup>2+</sup>) and cyan (without Zn<sup>2+</sup>). The Zn<sup>2+</sup>-binding sites are labeled with latin numbers as in the text. The second subunit of the dimer is colored in gray (without Zn<sup>2+</sup>, it corresponds to C as in Fig. 6) and in blue (with Zn<sup>2+</sup>) (B) The interface of TkTrpD with the bound Zn<sup>2+</sup> ions. The subunits are colored differently. Zn<sup>2+</sup> ions are shown as dark slate blue spheres. Distances are depicted. Figures were created using UCSF Chimera.

100 mM ZnCl<sub>2</sub> for  $\sim$  5–10 min. The crystals were subsequently transferred to a reservoir solution supplemented with 23% v/v glycerol and flash-cooled in liquid N<sub>2</sub>.

#### Data collection and structure determination

Diffraction data for the free TkTrpD were collected at ESRF (Grenoble) on the fully automatic high-throughput MASSIF-1 beamline [25] from a crystal that diffracted to 1.9 Å. xDS [26] was used to index and integrate the data

and AIMLESS [27] for merging and scaling. The crystal was found to belong to the  $P2_12_12_1$  space group. *Ss*TrpD (PDB entry 2GVQ) [7] was found to be the best matched search model by MOLREP [28] as implemented in MRBUMP [29] from CCP4 [30] and was used to obtain initial phases. After the solution was found, BUCCANEER [31] was employed for initial model building and automatic refinement with REFMAC5 [32]. Further refinement was carried out using PHENIX and water molecules were added with tools in PHENIX [33]. Manual rebuilding and structure visualization was performed by COOT [34]. The progress of refinement was monitored using the  $R_{\rm free}$  [35] with 5% of the reflections set aside.

## ZnCl<sub>2</sub> soaking

Data from a crystal soaked with  $ZnCl_2$  were collected at EMBL Hamburg (c/o DESY, Hamburg, Germany) on the P13 beamline at PETRA III and processed as before. Chain A of free *Tk*TrpD crystal structure was used as search model in PHASER for structure determination by molecular replacement. The best solution was found in the orthorhombic  $P22_{12}_1$  space group. Refinement was initially carried out using PHENIX and water molecules were added with tools in PHENIX. At the final stages of refinement, PDB\_REDO [36] was employed and REFMAC [32] was used. Manual rebuilding and structure visualization was performed by coot [34]. The progress of refinement was monitored with the  $R_{\rm free}$ .

#### **Structure analysis**

Interfaces were analyzed by PDBePISA [37]. Structural superpositions were performed with PDBeFold [38] as implemented in coot [34]. The superimposed structures were visually inspected using coot.

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# **Data Accessibility**

Structural data are available in the Protein Data Bank under the accession numbers 5NOE and 5NOF.

# Author contributions

SP performed experiments, analyzed data, and wrote the manuscript. NR analyzed data, planned experiments, and wrote the manuscript. XFT performed experiments and analyzed data. TI planned experiments and analyzed data. ACP performed and planned experiments, analyzed data, and wrote the manuscript.

# References

1 Lambrecht JA and Downs DM (2013) Anthranilate phosphoribosyl transferase (TrpD) generates phosphoribosylamine for thiamine synthesis from enamines and phosphoribosyl pyrophosphate. *ACS Chem Biol* **8**, 242–248.

- 2 Schwab T, Skegro D, Mayans O and Sterner R (2008) A rationally designed monomeric variant of anthranilate phosphoribosyltransferase from *Sulfolobus solfataricus* is as active as the dimeric wild-type enzyme but less thermostable. *J Mol Biol* **376**, 506–516.
- 3 Schramm VL and Grubmeyer C (2004) Phosphoribosyltransferase mechanisms and roles in nucleic acid metabolism. *Prog Nucleic Acid Res Mol Biol* 78, 261–304.
- 4 Hove-Jensen B, Andersen KR, Kilstrup M, Martinussen J, Switzer RL and Willemoës M (2017) Phosphoribosyl diphosphate (PRPP): biosynthesis, enzymology, utilization, and metabolic significance. *Microbiol Mol Biol Rev* **81**, e00040-16.
- 5 Castell A, Short FL, Evans GL, Cookson TVM, Bulloch EMM, Joseph DDA, Lee CE, Parker EJ, Baker EN and Lott JS (2013) The substrate capture mechanism of *Mycobacterium tuberculosis* anthranilate phosphoribosyltransferase provides a mode for inhibition. *Biochemistry* 52, 1776–1787.
- 6 Schlee S, Deuss M, Bruning M, Ivens A, Schwab T, Hellmann N, Mayans O and Sterner R (2009) Activation of anthranilate phosphoribosyltransferase from *Sulfolobus solfataricus* by removal of magnesium inhibition and acceleration of product release. *Biochemistry* 48, 5199–5209.
- 7 Marino M, Deuss M, Svergun DI, Konarev PV, Sterner R and Mayans O (2006) Structural and mutational analysis of substrate complexation by anthranilate phosphoribosyltransferase from *Sulfolobus solfataricus*. *J Biol Chem* **281**, 21410–21421.
- 8 Kim C, Xuong NH, Edwards S, Madhusudan Yee MC, Spraggon G and Mills SE (2002) The crystal structure of anthranilate phosphoribosyltransferase from the enterobacterium *Pectobacterium carotovorum*. *FEBS Lett* **523**, 239–246.
- 9 Cookson TVM, Evans GL, Castell A, Baker EN, Lott JS and Parker EJ (2015) Structures of *Mycobacterium tuberculosis* anthranilate phosphoribosyltransferase variants reveal the conformational changes that facilitate delivery of the substrate to the active site. *Biochemistry* **54**, 6082–6092.
- 10 Robison PD and Levy HR (1976) Metal ion requirement and tryptophan inhibition of normal and variant anthranilate synthase-anthranilate 5phosphoribosylpyrophosphate phosphoribosyltransferase complexes from *Salmonella typhimurium. Biochim Biophys Acta* 445, 475–485.
- 11 Atomi H, Fukui T, Kanai T, Morikawa M and Imanaka T (2004) Description of *Thermococcus kodakaraensis* sp. nov., a well-studied hyperthermophilic archaeon previously reported as *Pyrococcus sp.* KOD1. *Archaea* **1**, 263–267.
- 12 Bibi T, Perveen S, Aziz I, Bashir Q, Rashid N, Imanaka T and Akhtar M (2016) Pcal\_1127, a highly

stable and efficient ribose-5-phosphate pyrophosphokinase from *Pyrobaculum calidifontis*. *Extremophiles* **20**, 821–830.

- 13 Perveen S, Rashid N and Papageorgiou AC (2016) Crystal structure of a phosphoribosyl anthranilate isomerase from the hyperthermophilic archaeon *Thermococcus kodakaraensis. Acta Crystallogr F Struct Biol Commun* **72**, 804–812.
- 14 Sassetti CM, Boyd DH and Rubin EJ (2003) Genes required for mycobacterial growth defined by high density mutagenesis. *Mol Microbiol* 48, 77–84.
- 15 Lee CE, Goodfellow C, Javid-Majd F, Baker EN and Shaun Lott J (2006) The crystal structure of TrpD, a metabolic enzyme essential for lung colonization by *Mycobacterium tuberculosis*, in complex with its substrate phosphoribosylpyrophosphate. *J Mol Biol* 355, 784–797.
- 16 Rashid N, Morikawa M and Imanaka T (1997) Gene cloning and characterization of recombinant ribose phosphate pyrophosphokinase from a hyperthermophilic archaeon. *J Ferment Bioeng* 83, 412–418.
- 17 Andersen RW, Leggio Lo L, Hove-Jensen B and Kadziola A (2015) Structure of dimeric, recombinant *Sulfolobus solfataricus* phosphoribosyl diphosphate synthase: a bent dimer defining the adenine specificity of the substrate ATP. *Extremophiles* 19, 407–415.
- 18 Faria TQ, Lima JC, Bastos M, Maçanita AL and Santos H (2004) Protein stabilization by osmolytes from hyperthermophiles: effect of mannosylglycerate on the thermal unfolding of recombinant nuclease A from *Staphylococcus aureus* studied by picosecond timeresolved fluorescence and calorimetry. *J Biol Chem* 279, 48680–48691.
- 19 Hommel U, Lustig A and Kirschner K (1989) Purification and characterization of yeast anthranilate phosphoribosyltransferase. *FEBS J* 180, 33–40.
- 20 Ivens A, Mayans O, Szadkowski H, Wilmanns M and Kirschner K (2001) Purification, characterization and crystallization of thermostable anthranilate phosphoribosyltransferase from *Sulfolobus solfataricus*. *FEBS J* 268, 2246–2252.
- 21 Cookson TVM, Castell A, Bulloch EMM, Evans GL, Short FL, Baker EN, Lott JS and Parker EJ (2014) Alternative substrates reveal catalytic cycle and key binding events in the reaction catalysed by anthranilate phosphoribosyltransferase from *Mycobacterium tuberculosis. Biochem J* 461, 87–98.
- 22 Matthews BW (1968) Solvent content of protein crystals. J Mol Biol 33, 491–497.
- 23 Laskowski RA, Chistyakov VV and Thornton JM (2005) PDBsum more: new summaries and analyses of the known 3D structures of proteins and nucleic acids. *Nucleic Acids Res* 33, D266–D268.

- 24 Bradford MM (1976) A rapid and sensitive method for the quantitation of microgram quantities of protein utilizing the principle of protein-dye binding. *Anal Biochem* 72, 248–254.
- 25 Svensson O, Malbet-Monaco S, Popov A, Nurizzo D and Bowler MW (2015) Fully automatic characterization and data collection from crystals of biological macromolecules. *Acta Crystallogr D Biol Crystallogr* 71, 1757–1767.
- 26 Kabsch W (2010) XDS. Acta Crystallogr D Biol Crystallogr 66, 125–132.
- 27 Evans PR and Murshudov GN (2013) How good are my data and what is the resolution? *Acta Crystallogr D Biol Crystallogr* 69, 1204–1214.
- 28 Vagin A and Teplyakov A (1997) MOLREP: an automated program for molecular replacement. J Appl Crystallogr 30, 1022–1025.
- 29 Keegan RM and Winn MD (2008) MrBUMP: an automated pipeline for molecular replacement. Acta Crystallogr D Biol Crystallogr 64, 119–124.
- 30 Winn MD, Ballard CC, Cowtan KD, Dodson EJ, Emsley P, Evans PR, Keegan RM, Krissinel EB, Leslie AGW and McCoy A (2011) Overview of the CCP4 suite and current developments. *Acta Crystallogr D Biol Crystallogr* 67, 235–242.
- 31 Cowtan K (2006) The Buccaneer software for automated model building. 1. Tracing protein chains. *Acta Crystallogr D Biol Crystallogr* 62, 1002–1011.
- 32 Murshudov GN, Skubak P, Lebedev AA, Pannu NS, Steiner RA, Nicholls RA, Winn MD, Long F and Vagin AA (2011) REFMAC5 for the refinement of macromolecular crystal structures. *Acta Crystallogr D Biol Crystallogr* 67, 355–367.
- 33 Adams PD, Afonine PV, Bunkóczi G, Chen VB, Davis IW, Echols N, Headd JJ, Hung L-W, Kapral GJ, Grosse-Kunstleve RW *et al.* (2010) PHENIX: a comprehensive Python-based system for macromolecular structure solution. *Acta Crystallogr D Biol Crystallogr* 66, 213–221.
- 34 Emsley P and Cowtan K (2004) Coot: model-building tools for molecular graphics. *Acta Crystallogr D Biol Crystallogr* 60, 2126–2132.
- 35 Brünger AT (1992) Free R value: a novel statistical quantity for assessing the accuracy of crystal structures. *Nature* 355, 472–475.
- 36 Joosten RP, Long F, Murshudov GN and Perrakis A (2014) The PDB\_REDO server for macromolecular structure model optimization. *IUCrJ* 1, 213–220.
- 37 Krissinel E and Henrick K (2007) Inference of macromolecular assemblies from crystalline state. J Mol Biol 372, 774–797.
- 38 Krissinel E and Henrick K (2004) Secondary-structure matching (SSM), a new tool for fast protein structure alignment in three dimensions. *Acta Crystallogr D Biol Crystallogr* 60, 2256–2268.

- 39 Gonzalez JE and Somerville RL (1986) The anthranilate aggregate of *Escherichia coli*: kinetics of inhibition by tryptophan of phosphoribosyltransferase. *Biochem Cell Biol* 64, 681–691.
- 40 Henderson EJ, Zalkin H and Hwang LH (1970) The anthranilate synthetase-anthranilate 5phosphoribosylpyrophosphate phosphoribosyltransferase aggregate. Catalytic and regulatory properties of aggregated and unaggregated forms of anthranilate 5phosphoribosylpyrophosphate phosphoribosyltransferase. *J Biol Chem* **245**, 1424–1431.
- 41 Largen M, Mills SE, Rowe J and Yanofsky C (1978) Purification and properties of a third form of anthranilate-5-phosphoribosylpyrophosphate phosphoribosyltransferase from the Enterobacteriaceae. *J Biol Chem* **253**, 409–412.
- 42 Largen M, Mills SE, Rowe J and Yanofsky C (1976) Purification, subunit structure and partial amino-acid sequence of anthranilate-5-phosphoribosylpyrophosphate phosphoribosyltransferase from the enteric bacterium *Serratia marcescens. Eur J Biochem* **67**, 31–36.
- 43 Egan AF and Gibson F (1972) Anthranilate synthaseanthranilate 5-phosphoribosyl 1-pyrophosphate phosphoribosyltransferase from *Aerobacter aerogenes*. *Biochem J* 130, 847–859.
- 44 Wegman J and DeMoss JA (1965) The enzymatic conversion of anthranilate to indolylglycerol phosphate in *Neurospora crassa. J Biol Chem* 240, 3781–3788.
- 45 Marcus SL and Balbinder E (1972) Purification of anthranilate 5-phosphoribosylpyrophosphate phosphoribosyltransferase from *Salmonella typhimurium* using affinity chromatography: resolution of monomeric and dimeric forms. *Biochem Biophys Res Commun* **47**, 438–444.

- 46 Bode R and Birnbaum D (1979) Enzymes of the aromatic amino acid biosynthesis in *Hansenula henricii*: determination and characterization of the pretyrosine pathway enzymes. Z Allg Mikrobiol 19, 83–88.
- 47 O'Gara JP and Dunican LK (1995) Mutations in the trpD gene of *Corynebacterium glutamicum* confer 5-methyltryptophan resistance by encoding a feedback-resistant anthranilate phosphoribosyltransferase. *Appl Environ Microbiol* **61**, 4477–4479.
- 48 Weiss MS (2001) Global indicators of X-ray data quality. *J Appl Crystallogr* **34**, 130–135.
- 49 Diederichs K and Karplus PA (2013) Better models by discarding data? *Acta Crystallogr D Biol Crystallogr* **69**, 1215–1222.
- 50 Pettersen EF, Goddard TD, Huang CC, Couch GS, Greenblatt DM, Meng EC and Ferrin TE (2004) UCSF Chimera–a visualization system for exploratory research and analysis. *J Comput Chem* **25**, 1605–1612.
- 51 Robert X and Gouet P (2014) Deciphering key features in protein structures with the new ENDscript server. *Nucleic Acids Res* **42**, W320–W324.

# **Supporting information**

Additional Supporting Information may be found online in the supporting information tab for this article:

Fig. S1. Reaction catalyzed by TrpD.

Fig. S2. Gel filtration elution profile of TkTrpD with and without Zn<sup>2+</sup>.

**Fig. S3**. Determination of optimal pH for *Tk*TrpD enzymatic activity.