

RESEARCH ARTICLE

Aceclofenac fast dispersible tablet formulations: Effect of different concentration levels of Avicel PH102 on the compactional, mechanical and drug release characteristics

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Abstract

The objective of this study was based on the formulation development of fast dispersible Aceclofenac tablets (100 mg) and to evaluate the influence of pharmaceutical mixtures of directly compressible Avicel PH102 with Mannitol and Ac-di-sol on the compressional, mechanical characteristics and drug release properties. Fast dispersible Aceclofenac formulations were developed by central composite design (CCD). Among them the best possible formulation was selected on the basis of micromeritic properties, appropriate tablet weight and disintegration time for further study. Tablets were directly compressed using manual hydraulic press with a compressional force ranging from 7.2 to 77.2 MN/m². Pre and post compression studies were performed and the compressed formulations (FA-FF) were assessed for different quality tests. The Heckel and Kawakita equations were applied for determination of compressional behavior of formulations. The quality attributes suggested that formulation (FB) containing avicel PH 102 (20%), mannitol (25%) and ac-di-sol (3%) as best optimized formulation showing better mechanical strength i.e. hardness $35.40 \pm 6.93\text{N}$, tensile strength 0.963 MN/m^2 , and friability 0.68%. Furthermore, compressional analysis of FB showed lowest P_V value 59.520 MN/m^2 and P_K value 1.040 MN/m^2 indicating plasticity of the material. Formulation FB disintegrated rapidly within 21 seconds and released 99.92% drug after 45 min in phosphate buffer pH 6.8. Results of drug release kinetics showed that all formulations followed Weibull and First-order models in three different dissolution media. Avicel PH102 based formulation mixture exhibit excellent compactional strength with rapid disintegration and quick drug release.

Introduction

Advancement in tablet manufacturing technology has offered viable dosage alternatives for those patients who are facing problem related to compliance with conventional dosage forms. One such alternative dosage form is fast dispersible tablets [1]. These tablets are of two types: first type is taken in mouth without water to disintegrate rapidly or disperse readily and the second type of tablets form dispersion or solution in water to be taken by patients [2, 3]. These tablets are usually developed by direct compression method. Aceclofenac is a Cyclooxygenase inhibitor having analgesic and anti-inflammatory activity. Due to its short half-life (4hr) and twice daily dose, it is considered a suitable candidate for fast dispersible tablets [4, 5].

Avicel[®], the first commercialized brand of microcrystalline cellulose (MCC), is introduced by FMC Corporation as a direct compression tableting ingredient. MCC is a partially depolymerized cellulose that is obtained as pulp by mineral acid treatment of alpha cellulose type lb of fibrous plant material. Cellulose is the most abundant natural polymer having linear chains of b-1, 4-D anhydroglucopyranosyl units. Pharmaceutical MCC is most commonly obtained from wood where cellulose chains are packed in layers held together by strong hydrogen bonds and lignin (cross-linking polymer). The primary particles of all MCC types (101, 102 and 200) are about 50µm but the difference in the larger 2%, 2% aggregated particle numbers. Type 102 has a median particle size of about 100µm indicating adequate flow properties for successful tableting. MCC deforms plastically and maximizes the interparticle bonding area during compression. It forms strong and cohesive compacts even under low compression pressure due to the formation of numerous hydrogen bonds. Tableting is further enhanced by mechanical interlocking of elongated and irregularly shaped particles [6].

Formulations designed for fast disintegration require sugar-based diluents to impart pleasant taste, water solubility, and ability to mask the bitterness of medicament. Mannitol was used in this study to improve the mouthfeel of fast dispersible tablets [7]. A combination of avicel PH102 with water soluble mannitol exhibits shorter disintegration time and increased water solubility of mannitol. It may lead to the formation of pores in the tablets matrix, causing capillary action for water permeation into the tablet matrix; resulting in fast disintegration [8].

Ac-Di-Sol (i.e. super disintegrant) exerts its action by wicking and swelling of the tablet. The porous nature of Ac-Di-Sol provides access for water diffusion in tablets, resulting in wicking and the faster disintegration of tablets. It is recommended to be used in 0.5–3% concentration for directly compressible tablets [9–12].

During tablet manufacturing, compression of powder shows a reduction in the volume of powder bed by the application of compressional pressure in a confined space. As a result, there is a formation of strong inter-particle bonds that produce a compact mass and built inherent strength in the compact to increase the overall mechanical strength of the dosage unit. This analysis is significant in understanding the behavior of poorly compactable powders during the manufacturing of the tablets [13]. Heckel and Kawakita equations have been largely employed to understand this relationship between applied compressional pressure and mechanical strength of the compact.

The aim of the present study was to prepare fast dispersible Aceclofenac tablets and to evaluate the compressional behavior of the newly developed tablets using Heckel and Kawakita analysis.

Formulations were designed by Central Composite Design (CCD) using varying concentrations of avicel PH102, mannitol, and ac-di-sol. All the formulations were developed by direct compression method.

Materials

Aceclofenac was gifted by Sami Pharmaceutical (Pvt.) Limited. Avicel PH-102, mannitol, ac-di-sol, aspartame, talc, and vanilla flavor were purchased from FMC Corporation, USA.

Methods

Experimental design

Twenty fast dispersible Aceclofenac tablet formulations (100mg) were designed with the help of Central Composite Design using Design Expert[®] 10.0 software (Stat-Ease, Inc, Minneapolis, MN 55413, USA). The selected independent variables were avicel PH102 (20–35%), mannitol (10–25%) and ac-di-sol (0.5–3%). Excipients such as aspartame, vanilla flavor and talc were used at a fixed concentration i.e. 2%, 1%, and 2% respectively. Disintegration time, percentage friability and hardness of formulations were selected as the dependent variables. The run type was arranged and recoded as F1 = FA, F2 = FB, F15 = FO. The composition of each formulation in % and mg is mentioned in Table 1. There was six center points out of 20 formulations (i.e. F3 = center point formulation) with the same composition. These six formulations showed almost the same responses (disintegration time, friability and hardness) therefore results of only one center point formulation (i.e. F3 = Centre point formulation) is mentioned in proceeding text.

Response surface methodology (RSM) was used to explore the interaction of avicel PH102, mannitol and ac-di-sol to establish the appropriate amount of excipient for optimized fast dispersible formulation. On the basis of the fit summary, ANOVA and multiple correlation coefficient appropriate model was selected.

Pre-compression studies

Assessment of powder densities and flow characteristics. Bulk density of all formulations was evaluated using a measuring cylinder. The cylinder's weight was tare to zero and then filled with certain amount of formulation blend and reweighed. The following formula was used to determine the bulk density (g/cm^3).

$$\rho_{\text{bulk}} = \text{Mass} / \text{bulk volume} \quad \text{Eq 1}$$

Where ρ_{bulk} is the bulk density, and bulk volume is the initial volume occupied by the formulation blend in the cylinder.

The cylinder was tapped 100 times and reduction in formulation bed's volume was recorded as tapped volume and the following formula was used to calculate the tapped density (g/cm^3).

$$\rho_{\text{tap}} = \text{Mass} / \text{tapped volume} \quad \text{Eq 2}$$

Formulation blends were also assessed by Hausner's Ratio (HR) and angle of repose (Θ) using the following equations [14]:

$$\text{Hausner's Ratio} = \frac{\rho_{\text{tapped}}}{\rho_{\text{bulk}}} \quad \text{Eq 3}$$

$$\theta = \tan^{-1} \frac{2h}{D} \quad \text{Eq 4}$$

In the above equation θ is the angle of repose, 'h' is the height of powder heap and 'D' is the diameter of the heap formed. True density (ρ_t) of all powder blends were determined by *liquid*

Table 1. Composition of fast dispersible Aceclofenac formulations using central composite design.

Run	Formulation Codes	Ingredients amount in %							Ingredients amount in mg							
		Avicel PH102	Mannitol	Ac-di-sol	Aspartame	Talc	Flavor	Drug	Avicel PH102	Mannitol	Ac-di-sol	Aspartame	Talc	Flavor	Drug	Tablet weight
		X1 (%)	X2 (%)	X3 (%)	(%)	(%)	(%)	(%)	X1 (mg)	X2 (mg)	X3 (mg)	(mg)	(mg)	(mg)	(mg)	(mg/tab)
F1	FA	20.00	25.00	0.50	2	2	1	49.50	40.80	51.00	1.02	4.08	4.08	2.04	100	203.02
F2	FB	20.00	25.00	3.00	2	2	1	47.00	42.55	53.00	6.36	4.25	4.25	2.12	100	212.53
*F3	FC	27.50	17.50	1.75	2	2	1	48.25	57.20	36.45	3.64	4.16	4.16	2.08	100	207.69
F4	FD	27.50	17.50	3.39	2	2	1	46.61	58.91	37.45	7.10	4.28	4.28	2.14	100	214.16
F5	FE	35.00	10.00	0.50	2	2	1	49.50	71.42	20.40	1.02	4.08	4.08	2.04	100	203.04
F6	FF	35.00	10.00	3.00	2	2	1	47.00	74.20	21.20	6.36	4.25	4.25	2.12	100	212.38
F7	FG	20.00	10.00	0.50	2	2	1	64.50	31.24	15.62	0.78	3.12	3.12	1.56	100	155.44
F8	FH	20.00	10.00	3.00	2	2	1	62.00	32.20	16.10	4.83	3.22	3.22	1.61	100	161.18
F9	FI	27.50	07.63	1.75	2	2	1	58.12	47.52	13.18	3.02	3.45	3.45	1.72	100	172.34
F10	FJ	17.63	17.50	1.75	2	2	1	58.12	30.46	30.24	3.02	3.45	3.45	1.72	100	172.34
F11	FK	35.00	25.00	3.00	2	2	1	32.00	109.37	78.00	9.36	6.25	6.25	3.12	100	312.35
F12	FL	27.50	27.37	1.75	2	2	1	38.38	72.05	71.70	4.58	5.24	5.24	2.62	100	261.43
F13	FM	35.00	25.00	0.50	2	2	1	34.50	102.90	73.50	1.47	5.88	5.88	2.94	100	292.57
F14	FN	37.37	17.50	1.75	2	2	1	38.38	98.00	45.89	4.58	5.24	5.24	2.62	100	261.57
F15	FO	27.50	17.50	0.10	2	2	1	49.89	50.09	31.88	0.191	3.64	3.64	1.82	100	191.26

*F3 = Centre point formulation

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displacement method using xylene as displacement liquid with the help of a pycnometer. The difference between empty pycnometer (W) and xylene filled pycnometer (W_1) was calculated as the weight of xylene (W_2). Approximately 2 g sample of each formulation was weighed (W_3) and transferred to the pycnometer along with xylene and weighed again (W_4). True density (g/cm^3) was calculated using the following equation [15]:

Weight of empty pycnometer = W

Weight of pycnometer filled with xylene = W_1

Weight of xylene $W_2 = W_1 - W$

Weight of formulation blend = W_3

Weight of pycnometer filled with xylene and formulation blend = W_4

$$\rho_t = \frac{(W_2 \times W_3)}{50 \times (W_3 - W_4 + W_2 + W)} \tag{Eq 5}$$

Compression of powder blends. All the ingredients were weighed and passed through 20 mesh sieve separately and mixed for six minutes (optimized mixing time) by tumbling in a polybag. After mixing all the ingredients, talc was added and mixed for a further five minutes. Blends were compressed by manual filling of the die cavity. Tablets were compressed by applying varied compressional pressure 7.72, 23.16, 30.88, 38.0, 46.32, 54.04, 61.76, 69.48 and 77.2 MN/m^2 using a manual hydraulic tableting machine (locally manufactured) fitted with pressure gauge. Tablets were stored in a desiccator for 24 h over silica gel for elastic recovery and hardening.

Evaluation of physicochemical parameters of Aceclofenac formulations. Compressed tablets were weighed and their thickness and diameter were accurately measured with digital

vernier caliper (Digital Caliper: Seiko brand). Tablet hardness was determined by using hardness tester (OSK Fujiwara, Ogawa Seiki Co. Ltd., Tokyo, Japan). Tensile strength of round flat-faced tablets was calculated using the given equation [16]:

$$T = \frac{2F}{\pi DH} \quad \text{Eq 6}$$

Where, F (N) is the crushing load applied on tablets (i.e. hardness), 'H' is the thickness (mm) and 'D' is the diameter (mm) of tablets. Percentage friability was determined using Roche type Friabilator (H. Jurgens Gmbh H and Co- Bremen, D2800, Germany) with the help of following formula:

$$F = \frac{W_o - W_t}{W_o} \times 100 \quad \text{Eq 7}$$

Where W_o is the initial weight and W_t is the final weight of tablets after 100 rotations. For the disintegration test of fast dispersible tablets, 900 mL distilled water was maintained at 37°C. Tablets were disintegrated in seconds [17]. The assay assessment was carried out using UV-Spectrophotometer (UV-1800 Shimadzu Corporation Kyoto, Japan) at 274 nm absorbance [18, 19]. Similarly, the dissolution test was performed using 0.1N HCl, phosphate buffer pH 4.5 and pH 6.8.

Evaluation of compressional behavior. *Heckel analysis.* Mathematically Heckel equation is expressed as follows:

$$\ln[1/1 - \rho_r] = KP + A \quad \text{Eq 8}$$

This equation is used to relate the powder bed's relative density (ρ_r) to the applied pressure during compression which is the ratio of apparent density (ρ_A) of the tablet and the true density (ρ_T) of powder blend:

$$\rho_r = \rho_A / \rho_T \quad \text{Eq 9}$$

The apparent density ρ_A of tablet is calculated using the weight (g), radius (cm) and thickness (cm) of the tablet, whereas true density of the powder blend is determined by liquid displacement method (g/cm^3).

$$\rho_A = \text{Weight of tablet} / \pi r^2 h \quad \text{Eq 10}$$

By plotting the values of $\ln[1/(1 - \rho_r)]$ and applied pressure 'P', linear portion of the plot provides the values of slope and intercept (i.e. K and A). The value of K determines the mean yield pressure ($1/K = P_Y$). This mean yield pressure P_Y indicates the plasticity of material under compression. Smaller the value of $1/K$, higher the plasticity of material. Using the value of intercept 'A' densification of powder at different stages (i.e. D_o , D_A , and D_B) is determined [15, 20, 21]. The value of D_o (relative density = bulk density / true density) shows the powder densification at die filling stage. Since it measures the packing characteristics of powder, thus high value of D_o indicates high dense packing of powder blend [22, 23]. The D_B value indicates densification of powder when it is compressed and particles show movement and re-arrangement. Magnitude of rearrangement based on the theoretical point of densification at which deformation of particles started [24]. The value of D_B is calculated as follows:

$$D_B = D_A - D_o \quad \text{Eq 11}$$

The D_A value is the final compact densification ($D_A = \rho_r$) and it is calculated from intercept 'A' of the Heckel plot:

$$D_A = 1 - e^{-A} \tag{Eq 12}$$

Kawakita equation. Kawakita equation describes the compressional behavior of powders, either by tapping or from continuous compression experiments [25]. This equation explains the relationship between the degree of volume reduction 'C' upon the application of compressional pressure 'P'. The linear expression of the Kawakita equation is given below:

$$\frac{P}{C} = \frac{P}{a} + \frac{1}{ab}$$

or

$$C = \frac{(V_o - V_p)}{V_o} = \frac{abP}{(1 + bP)} \tag{Eq 13}$$

Where V_o is the initial bulk volume of the powder, V_p is the volume of the powder after compression, a is the total volume reduction for the powder bed or minimum porosity before compression and 'b' is the powder's plasticity [26]. A graph of $\frac{P}{C}$ versus P was plotted for all formulations and constants a and b were determined from slope and intercept of the plot. Reciprocal of slope produced the value of a (i.e. $a = 1/\text{slope}$) and reciprocal of the intercept yielded, ab (i.e. $ab = 1/\text{Intercept}$).

$$b = ab - a$$

Reciprocal of b is a pressure (P_K) which reduced the thickness of powder bed 50%.

Evaluation of in-vitro release behavior. Six tablets of each test (FA-FH) and immediate release marketed (reference) formulations were placed in the dissolution apparatus (USP Apparatus-II: Paddle Stirring Element) containing 900 mL of dissolution media at $37 \pm 0.5^\circ\text{C}$ at 50 rpm. Multiple points sampling was conducted in different dissolution media i.e. 0.1N HCl, phosphate buffer pH 4.5 and 6.8. 10 ml sample was withdrawn at 5, 10, 15, 20, 30, 45, 60, 90 and 120 min interval and substituted with fresh 10 ml of same solution. Each test and reference solutions were diluted, filtered and analyzed spectrophotometrically at 274nm [19].

Release kinetics. *Model- dependent method.* Various kinetic models were used for the evaluation of release patterns as mention in below equations [27, 28]:

First order kinetics. Explains that the drug release from systems is reliant on concentration, which represents time vs. log cumulative percentage drug remaining [27, 28].

$$\log Q_t = \log Q_0 + k_1 \frac{t}{2.303} \tag{Eq 14}$$

Where Q_t is the collective amount of drug release at time t , Q_0 is the initial concentration of drug, k_1 is the First—Order rate constant [29]

Weibull model. Weibull model expresses the fraction of drug release (m) in solution at time t , by following equation:

$$m = 1 - \exp \left[\frac{-(t - T_i)^b}{\alpha} \right] \tag{Eq 15}$$

or

$$\text{Log} [-\ln (1 - m)] = b \log (t - T_i) - \log \alpha$$

Where, T_i is the lag time, in various cases zero, α , is the time process and b is the shape parameter [30] ($b = 1$) illustrates the curve as exponential, ($b < 1$) parabolic with the elevated early slope and after that constant with the exponential, ($b > 1$) shows S-shaped with increasing curve followed by turning point.

Hixson-crowell model. Following equation represents his model:

$$\sqrt[3]{Q_0} - \sqrt[3]{Q_t} = k_{HC} t \tag{Eq 16}$$

Where Q_0 = initial concentration of drug, Q_t = drug concentration at time t and K_{HC} is the Hixson—Crowell rate constant [28].

Higuchi model. Based on diffusion process, the drug release can be described by Higuchi model Equation of Higuchi model is expressed as follows:

$$Q_t = k_H t^{1/2} \tag{Eq 17}$$

Where k is the Higuchi release rate constant and t is the time in hr [31].

Results and discussion

The purpose of this study was to prepare fast dispersible tablets of Aceclofenac and to evaluate the effect of different concentrations of avicel PH102 on the compressional behavior of newly developed formulations. The central composite design was used for formulation design. These formulations were arranged and assigned with specific codes for identification purposes and are presented in Table 1.

Flow properties

Powder blends of all formulations were evaluated for true, bulk and tapped densities and results were found in the range of 1.40–1.47, 0.44–0.52 and 0.51–0.63 g/cm³ respectively. Flow properties of all formulations were assessed by using Hausner’s ratio and angle of repose and their respective values were found to be 1.13–1.24, and 33.43–38.31° respectively, indicating better flow properties (Table 2). Formulation blends that failed to meet the acceptable limits of micromeritics characterization and excessive weight than the target formulation were excluded from study (i.e. F11, F12, and F13). Mannitol causes more adhesion problems during compression and this could be overcome by using it in combination with Avicel PH102 which facilitates the compression. After micromeritic evaluation remaining formulations were subjected to compression by direct compression method using manual hydraulic press.

Table 2. Micromeritic properties of selected Aceclofenac fast dispersible tablet formulations.

Form Codes	True density ρ_t (g/cm ³)	Bulk density ρ_b (g/cm ³)	Tapped density ρ_{tapp} (g/cm ³)	Hausner’s ratio	Angle of repose (θ)
FA	1.47	0.46	0.52	1.13	34.53
FB	1.40	0.44	0.51	1.17	33.43
FC	1.47	0.50	0.60	1.20	38.31
FD	1.44	0.49	0.61	1.24	37.09
FE	1.46	0.52	0.63	1.23	36.17
FF	1.47	0.51	0.59	1.15	34.67

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Disintegration time

All formulations except F13, and F15 showed acceptable disintegration time ranging from 18–45 sec. Formulations F13 and F15 contained 0.5% and 0.1% superdisintegrant respectively, therefore, failed to meet the disintegration time of fast dispersible tablets. Ac-di-sol in the concentration of 1.75% indicated acceptable disintegration time but the formulation F9 having the same amount of superdisintegrant presented disintegration time of 4min which is beyond the requirement of fast dispersible tablets. This difference might be due to the presence of lesser concentration of mannitol (7.63%) which has more water solubility and thus facilitate disintegration process.

On the basis of the weight of tablets closer to target formulation and disintegration time, six formulations were selected for further evaluation. These formulations were assessed by different quality tests and results were found to be inadequate limits. The quality attributes of compressed tablets are mentioned in Table 3.

RSM plot and ANOVA summary

It is indicated in the RSM plot that disintegration time was increased with increased concentration of avicel PH102 as shown in Fig 1A and 1B. The ANOVA summary for the first response (disintegration) indicated that the model F value was 4.54 and the Probability value was less than 0.05 indicating that the quadratic model was significant. The “Adeq Precision” was 8.196 which indicated adequate signal-to-noise ratio and the design space could be navigated by the model. RSM plot Fig 1C and 1D indicated that friability was decreased with increased concentration of avicel PH 102 and mannitol. For the second response friability, F value, Probability and Adeq Precision were 10.18, < 0.05 and 11.265 respectively indicating quadratic model was valuable with the satisfactory signal. RSM plot Fig 1E and 1F presented that hardness of fast dispersible aceclofenac tablets was increased with higher concentration of avicel PH102 and mannitol. F value, Probability and Adeq Precision for third response hardness were 10.18, < 0.05 and 11.265 respectively showing model terms and linear model were acceptable with adequate signal. If A = Avicel PH102, B = Mannitol, and C = Ac-di-sol then the final equations in terms of coded factors for disintegration, friability and hardness are given below:

$$Disintegration = +14.01 + 5.39 * A + 3.76 * B - 2.64 * C + 5.75 * AB + 2.50 * AC - 2.50 * BC - 1.60 * A^2 + 3.59 * B^2 + 10.81 * C^2$$

$$Friability = +0.29 + 0.012 * A - 0.021 * B - 4.316E - 003 * C - 0.11 * AB - 0.069 * AC + 0.071 * BC + 0.016 * A^2 + 0.039 * B^2 + 0.086 * C^2$$

Table 3. Quality attributes of different Aceclofenac fast dispersible tablet formulations at different compressional pressures.

Test parameters	FA	FB	FC	FD	FE	FF
Weight (mg)	204.1 ± 1.912	212.1 ± 1.790	207.4 ± 1.955	214.2 ± 2.097	204.3 ± 2.311	212.2 ± 2.529
Diameter (mm)	8.468 ± 0.007	8.482 ± 0.010	8.49 ± 0.007	8.47 ± 0.006	8.47 ± 0.021	8.47 ± 0.014
Thickness (mm)	2.72 ± 0.055	2.76 ± 0.047	2.72 ± 0.081	2.77 ± 0.062	2.75 ± 0.087	2.73 ± 0.086
Hardness (N)	33.58 ± 8.00	35.40 ± 6.93	34.61 ± 10.46	40.68 ± 9.90	53.42 ± 11.80	43.99 ± 6.25
Tensile strength (MPa)	0.932 ± 0.239	0.9628 ± 0.203	0.958 ± 0.311	1.108 ± 0.292	1.468 ± 0.362	1.213 ± 0.207
Relative density (g/cm ³)	0.93 ± 0.01	0.95 ± 0.01	0.92 ± 0.02	0.93 ± 0.02	0.92 ± 0.03	0.92 ± 0.02
Friability (%)	0.50	0.68	0.30	0.32	0.34	0.29
Disintegration time (sec)	18	21	40	36	45	40
Assay (%)	99.95 ± 1.34	100.10 ± 0.68	99.63 ± 0.48	100.58 ± 1.68	98.63 ± 0.80	99.56 ± 1.23
Dissolution (%)	99.42 ± 0.52	100.38 ± 0.71	99.05 ± 0.90	100.35 ± 0.77	99.10 ± 1.04	100.10 ± 1.44

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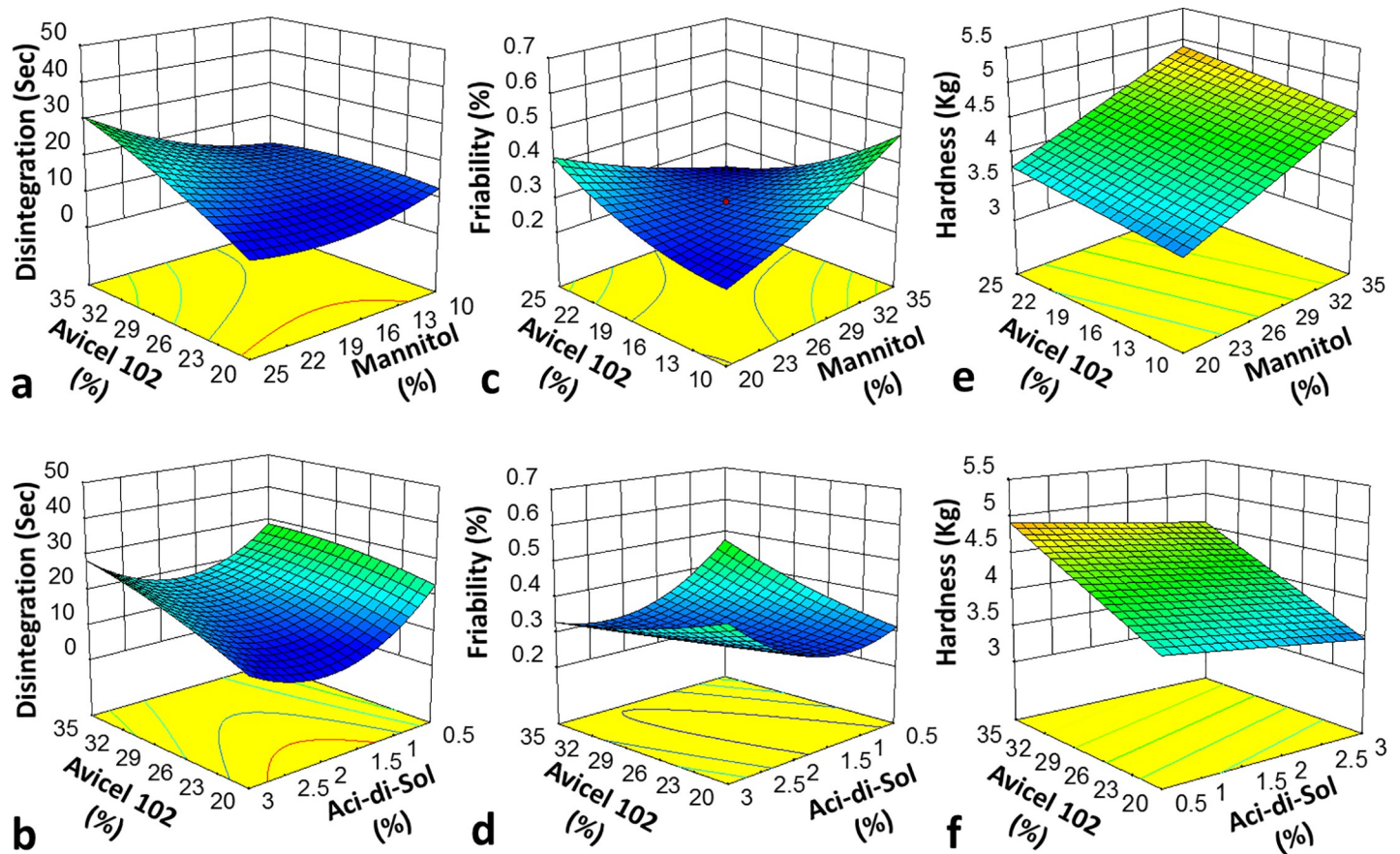


Fig 1. 3D Response surface plots of different fast dispersible Aceclofenac tablet formulations presenting effect of independent variables on (a & b) disintegration time, (c & d) Friability and (e & f) Hardness.

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$$\text{Hardness} = +4.19 + 0.55 * A + 0.16 * B - 0.17 * C$$

Tensile strength and hardness

Mechanical strength is another important parameter that also represents the inter-particulate bonding. For tablet dosage form it is recommended to determine the mechanical properties of dominant ingredient so as to predict the overall compressional behavior of tablets [32].

Mechanical properties were estimated by measuring the tensile strength which was in the range of 0.932 ± 0.239 – 1.468 ± 0.362 MN/m². Tablet hardness of selected formulations was found to be in the range from 33.58 ± 8.00 – 53.42 ± 11.80 N. Results of tensile strength and hardness indicated sufficient mechanical strength of all formulations due to the development of strong inter-particle bonds of powder blends and % friability of the tablets was 0.29–0.68%.

Compressional behavior analysis

For evaluating compressional behavior, ten tablets from each formulation were prepared by applying different compressional pressure 7.72, 23.16, 30.88, 38.0, 46.32, 54.04, 61.76, 69.48 and 77.2 MN/m². Heckel and Kawakita equations were employed for estimating the

Table 4. Compressional parameters obtained from heckel and kawakita equations of formulation blends.

Formulation Codes	Heckel Parameters				Kawakita Parameters	
	D_o	D_A	D_B	P_Y (MN/m ²)	$DI = 1-a$	$P_K = 1/b$ (MN/m ²)
FA	0.313	0.892	0.579	86.950	0.325	1.712
FB	0.314	0.922	0.608	59.520	0.319	1.040
FC	0.343	0.870	0.527	75.750	0.357	4.180
FD	0.340	0.883	0.543	64.930	0.349	4.350
FE	0.355	0.869	0.514	66.660	0.367	6.560
FF	0.339	0.863	0.524	67.560	0.350	7.690

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compressional behavior of fast dispersible Aceclofenac tablets. Previously it was reported that both these compaction equations are suitable for describing the compression process of powder materials based on Avicel PH 102 (24). Different parameters of Heckel and Kawakita equations were derived from these plots i.e. D_o , D_A , D_B , P_Y , DI and P_K and reported in Table 4. Heckel and Kawakita plots are presented in Figs 2 and 3 respectively.

It was reported that D_o values for different aceclofenac formulations were increased as the concentration of binder was increased. This phenomenon indicates that at die filling the initial packing of the formulations increases with the higher concentration of binder [32]. In formulations FC, FD, FE and FF, the amount of avicel PH 102 was high (27.5–35%) which resulted in increased hardness, tensile strength and less percentage friability (Table 3). Scientists also

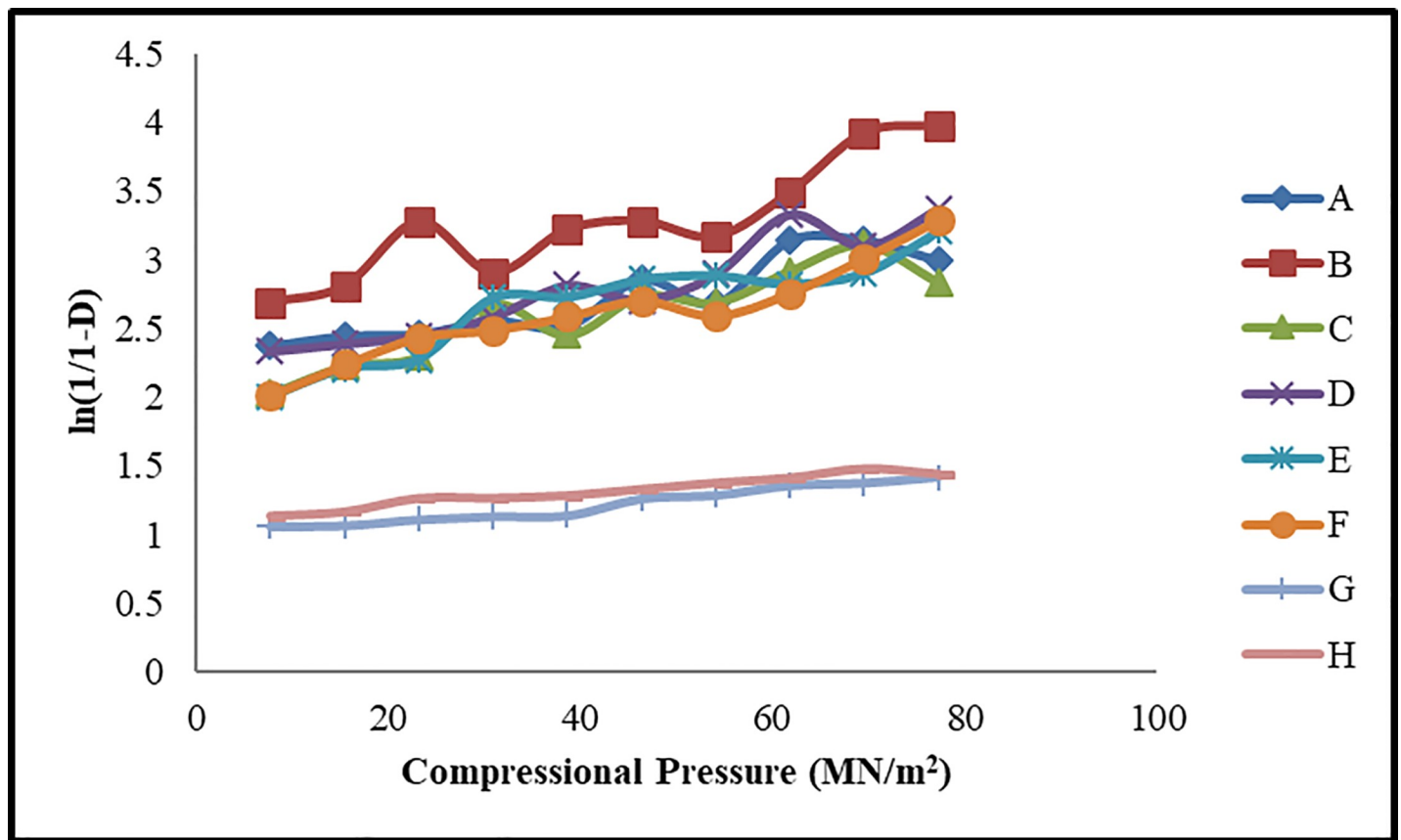


Fig 2. Heckel plots of fast dispersible Aceclofenac tablets.

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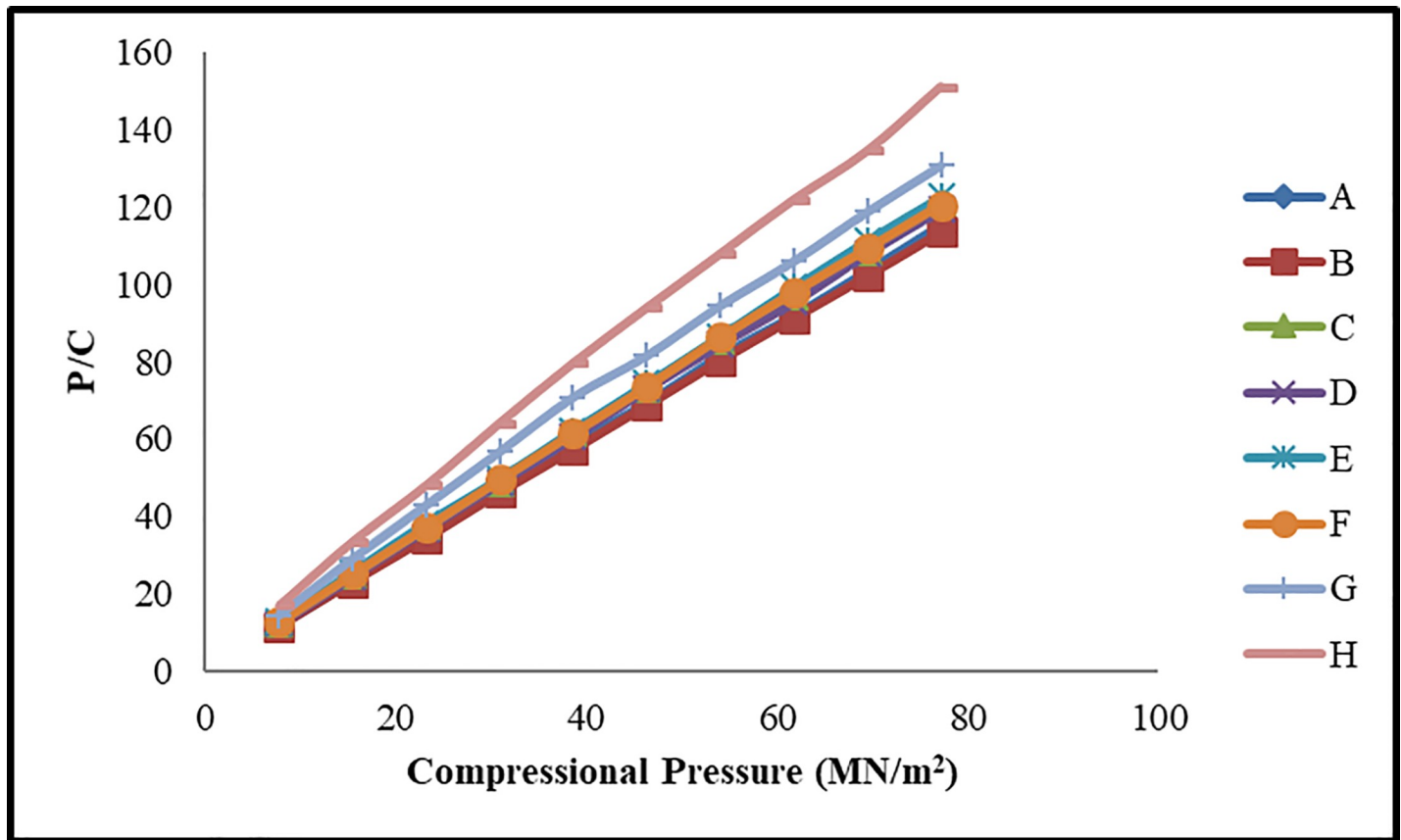


Fig 3. Kawakita plots of fast dispersible Aceclofenac tablets.

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found that compaction behavior of powder bed mostly dependent upon the deformation properties of the blends and the applied processing methods [32].

Heckel analysis

From Heckel plots, it was found that the initial packing of powder blends (D_o) in FE was found to be high i.e. $D_o = 0.355$ as FE contained increased concentration of avicel PH102 (35%). The D_A values at zero and low pressures presented the total degree of packing. Generally formulations with increased concentration of avicel PH102 exhibited the lowest values of D_A . The D_A value which showed the degree of packing at low compressional pressure was found lowest in FE (0.869) and FF (0.863). Values of D_A for different formulations observed in the presented sequence: $FF < FE < FC < FD < FA < FB$.

The D_B values presented the densification of powder bed at low pressure which shows the rearrangement of the particles by applying compressional pressure leading to the particle fragmentation. This fragmentation could be plastic or elastic. It was observed that formulations containing low amount of avicel PH102 i.e. FA (0.568) and FB (0.579) showed decreased values of D_B . The sequence is $FE < FF < FC < FD < FA < FB$. Amin *et al.*, 2012 stated that each powder has its own compressional characteristics. In the form of powder blend these characteristics get significantly changed which affects tablet stability [32]. It was observed that all formulations yielded low values of D_o than D_B due to particle fragmentation and particle

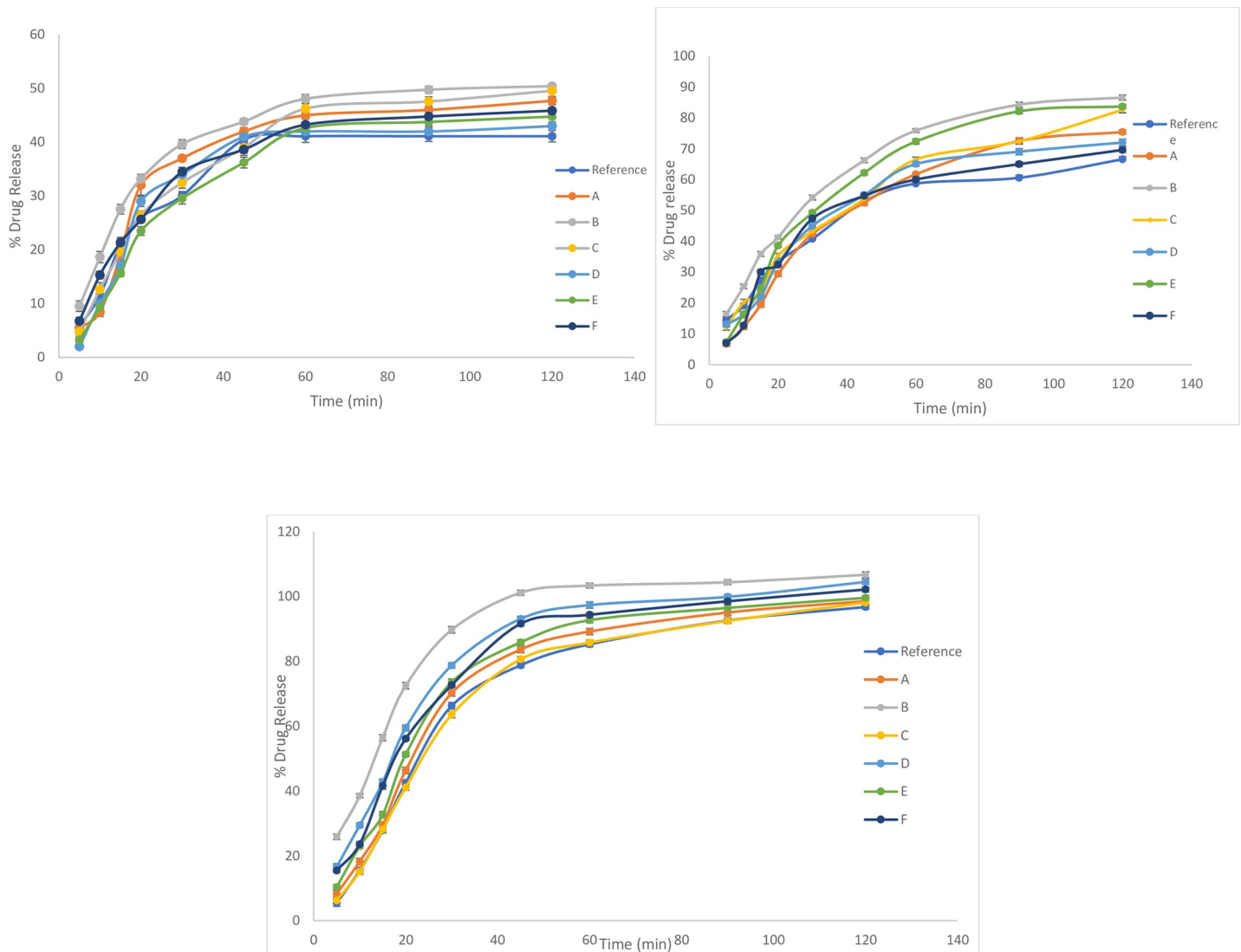


Fig 4. Drug release kinetics of Aceclofenac fast dispersible tablets in 900 ml of (a) pH 1.2 (b) phosphate buffer pH 4.5 (c) phosphate buffer pH 6.8.

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rearrangement in the die at reduced pressure. Generally, increased porosity of powder blend at zero pressure yields low values of D_o .

Another parameter ‘ P_Y ’ (mean yield pressure) is the measure of the plasticity of the material. Plastic material is desirable for compression while elastic material creates problems during compression due to elastic recovery. Formulations having greater tendency to deform plastically usually have low values of P_Y . In this study FB (59.52 MN/m²) and FD (64.93 MN/m²) showed the lowest ‘ P_Y ’ value and the highest plasticity (Table 4). It means that as the amount of avicel PH 102 affects the plasticity of the formulation and facilitates the manufacturing process.

Kawakita analysis

The Kawakita plot was also constructed to evaluate the compressional behavior of formulations (Fig 2). Kawakita plots presented a linear relationship at different compressional

Table 5. Release kinetics of fast dispersible Aceclofenac (100 mg) tablets at different pH.

Formulation Codes	First Order		Higuchi		Hixson-Crowell		Weibull model		
	r^2	K (hr ⁻¹)	r^2	K _H (hr ^{-1/2})	r^2	K _{HC} (hr ^{-1/3})	r^2	α	B
0.1N HCl									
Reference	0.374	0.007	0.765	4.622	0.645	0.001	0.893	9.695	0.378
FA	0.353	0.009	0.768	5.046	0.664	0.001	0.963	8.065	0.367
FB	0.408	0.009	0.796	5.107	0.696	0.002	0.939	8.181	0.379
FC	0.323	0.009	0.756	5.043	0.653	0.001	0.925	7.764	0.359
FD	0.349	0.009	0.760	5.106	0.657	0.001	0.925	7.812	0.364
FE	0.324	0.009	0.751	5.031	0.646	0.001	0.919	7.803	0.359
FF	0.330	0.009	0.760	5.089	0.659	0.001	0.925	7.745	0.362
pH 4.5									
Reference	0.681	0.041	0.868	6.480	0.809	0.002	0.950	9.179	0.483
FA	0.948	0.021	0.943	8.357	0.944	0.004	0.985	17.192	0.733
FB	0.946	0.022	0.945	8.418	0.950	0.004	0.988	17.079	0.736
FC	0.950	0.022	0.950	8.391	0.951	0.004	0.991	16.625	0.726
FD	0.954	0.022	0.946	8.474	0.950	0.004	0.989	17.511	0.744
FE	0.964	0.022	0.953	8.527	0.958	0.005	0.987	20.903	0.785
FF	0.950	0.021	0.943	8.359	0.941	0.004	0.982	17.076	0.730
pH 6.8									
Reference	0.948	0.044	0.781	11.553	0.969	0.012	0.996	41.658	1.267
FA	0.946	0.051	0.736	11.438	0.950	0.012	0.991	27.069	1.223
FB	0.954	0.052	0.733	11.339	0.951	0.012	0.992	21.981	1.144
FC	0.948	0.051	0.713	11.351	0.944	0.012	0.996	29.015	1.248
FD	0.952	0.051	0.741	11.398	0.953	0.012	0.993	21.695	1.111
FE	0.951	0.050	0.728	11.240	0.948	0.012	0.992	20.114	1.080
FF	0.949	0.050	0.735	11.374	0.950	0.012	0.989	24.857	1.165

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pressures with a correlation coefficient above 99% for formulations FA-FF. From the slope and intercept of Kawakita plots the values of *a* and *ab* were determined respectively. Values of 1-*a* indicated initial relative density (*D_I*) of the formulations. By using reciprocal of *b* values, the inverse measurement of plasticity (*P_K*) was estimated as given in Table 4. It was observed that initial relative density of formulations (*D_I*) was decreased in formulations containing low concentration of avicel PH102. The initial relative density (*D_I*) of all formulations was in the range of 0.319–0.367. Results indicated that higher *D_I* values were observed as compared to the corresponding values of *D_O*, which has been previously reported by other researches [32]. Furthermore the *P_K* values of these formulations were also higher.

It was observed in the present study that as the concentration of avicel PH102 was increased the *P_K* value was also increased (1.04–7.69 MN/m²) which is indicative of elasticity of material. Formulations FE and FF had the highest *P_K* values and FA and FB had the lowest *P_K* values. It was found from Heckel and Kawakita parameters analysis that FE and FF showed the fastest onset of plastic deformation whereas FA and FB showed maximum plastic deformation in combination with aceclofenac and other excipients. However no clear cut variation pattern of Heckel and Kawakita parameters was observed.

In vitro dissolution

In the present study *in vitro* drug release profiles of newly developed and optimized aceclofenac formulations were determined in different dissolution media as shown in Fig 4A–4C. All

formulations demonstrated maximum drug release in phosphate buffer pH 6.8. Different kinetic models were used to analyze the release behavior of formulations FA-FF. Results indicated that all formulations followed First-order and Weibull model in different dissolution media with highest r^2 values found in phosphate buffer pH 6.8 i.e. 0.946–0.954 and 0.989–0.996 respectively as shown in [Table 5](#).

Limitation of the study

This study based on the results generated on manual tablet press and not on the large scale rotary compression machine.

Conclusion

In this study, fast dispersible aceclofenac tablets were prepared and the effect of avicel PH102 was examined on compressional, mechanical and release properties of fast dispersible aceclofenac formulations. Using the Heckel and Kawakita equations, the compressional behavior was observed. The concentration of avicel PH102 exhibited a significant impact on the compressional, mechanical and release properties of the Aceclofenac fast dispersible formulations. Formulation FB having avicel PH102 (20%), mannitol (25%) and ac-di-sol (3%) exhibited excellent compaction strength with rapid disintegration and quick drug release. Hence a suitable selection of excipient with appropriate concentration is important at formulation development stage to ensure stable, elegant, and palatable dosage form for the patient.

Supporting information

S1 Data.

(XLSX)

S2 Data.

(XLSX)

Author Contributions

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Visualization: Farya Zafar.

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References

1. Fukami J, Yonemochi E, Yoshihashi Y, Terada K. Evaluation of rapidly disintegrating tablets containing glycine and carboxymethylcellulose. *International journal of pharmaceuticals*. 2006; 310(1–2):101–9. <https://doi.org/10.1016/j.ijpharm.2005.11.041> PMID: 16434157
2. Schiermeier S, Schmidt PC. Fast dispersible ibuprofen tablets. *European journal of pharmaceutical sciences*. 2002; 15(3):295–305. [https://doi.org/10.1016/s0928-0987\(02\)00011-8](https://doi.org/10.1016/s0928-0987(02)00011-8) PMID: 11923062
3. Martin TP, Hayes P, Collins DM. Tablet dispersion as an alternative to formulation of liquid dosage forms. *Australian Journal of Hospital Pharmacy*. 1993; 23(6):378–86.
4. Setty CM, Prasad D, Gupta V, Sa B. Development of fast dispersible aceclofenac tablets: effect of functionality of superdisintegrants. *Indian journal of pharmaceutical sciences*. 2008; 70(2):180. <https://doi.org/10.4103/0250-474X.41452> PMID: 20046709
5. Parfitt K, Martindale. *Martindale: The Complete Drug Reference*. 36th ed. Massachusetts, London, UK, 2009: RPS Publishing Britian; 2009.
6. Thoorens G, Krier F, Leclercq B, Carlin B, Evrard B. Microcrystalline cellulose, a direct compression binder in a quality by design environment—A review. *International journal of pharmaceuticals*. 2014; 473(1):64–72. <https://doi.org/10.1016/j.ijpharm.2014.06.055>.
7. Chang R-K, Guo X, Burnside BA, Couch RA. Fast-dissolving tablets. *Pharmaceutical technology*. 2000; 24(6):52–.
8. Bala R, Khanna S, Pawar PK. Formulation and optimization of fast dissolving intraoral drug delivery system for clobazam using response surface methodology. *Journal of advanced pharmaceutical technology & research*. 2013; 4(3):151.
9. Panigrahi R, Behera SP, Panda CS. A review on fast dissolving tablets. 2010.
10. Mor J, Chauhan P, Jalwal P. Development and Evaluation of Oral Fast Dissolving Tablets of Lornoxicam using Superdisintegrants-A Comparative Study. *Cellulose*. 2016; 100(100):100.
11. Kumar V. Formulation and Evaluation of Meclizine Hcl Orally Dispersible Tablets by using Natural Super Disintegrants. *Int J Pharm Sci & Scient Res*. 2016; 2:1–53.
12. Mor J, Chauhan P, Jalwal P. Development and evaluation of oral fast dissolving tablets of Lornoxicam using superdisintegrants-A comparative study. *The Pharma Innovation*. 2016; 5(7, Part A):1.
13. Nicklasson F, Alderborn G. Analysis of the compression mechanics of pharmaceutical agglomerates of different porosity and composition using the Adams and Kawakita equations. *Pharmaceutical research*. 2000; 17(8):949–54. <https://doi.org/10.1023/a:1007575120817> PMID: 11028940
14. Lin C-W, Cham T-M. Compression behavior and tensile strength of heat-treated polyethylene glycols. *International journal of pharmaceuticals*. 1995; 118(2):169–79.
15. Odeku O, Awe O, Popoola B, Odeniyi M, Itiola O. Compression and mechanical properties of tablet formulations: containing corn, sweet potato, and cocoyam starches as binders. *Pharmaceutical technology*. 2005; 29(4):82–90.
16. Fell J, Newton J. Determination of tablet strength by the diametral-compression test. *Journal of pharmaceutical sciences*. 1970; 59(5):688–91. <https://doi.org/10.1002/jps.2600590523> PMID: 5446428
17. Eur.Pharm. *European Pharmacopoeia*. In: European Department for the Quality of Medicines SS, editor. 3rd ed2001.
18. Bhardwaj S, Jain V, Jat R, Mangal A, Jain S. Formulation and evaluation of fast dissolving tablet of aceclofenac. *International journal of drug delivery*. 2010; 2(1).
19. Sharma AJ, A.; Purohit A.; Jatav R.; Sheorey R. Formulation and evaluation of aceclofenac fast dissolving tablets. *International Journal Of Pharmacy & Life Sciences*. 2011; 4(2):681–6.
20. Mustapha MA, Igwilo CI, Silva BO. Influence of Concentration of Modified Maize Starch on Compaction Characteristics and Mechanical Properties of Paracetamol Tablet Formulations. *Medical Journal of Islamic World Academy of Sciences*. 2013; 109(893):1–7.
21. Mohammed B, Isah A, Ibrahim M. Influence of compaction pressures on modified cassava starch as a binder in paracetamol tablet formulations. *Nigerian Journal of Pharmaceutical Sciences*. 2009; 8(1):80–8.
22. Mahmoodi F, Alderborn G, Frenning G. An experimental evaluation of an effective medium based compaction equation. *European Journal of Pharmaceutical Sciences*. 2012; 46(1–2):49–55. <https://doi.org/10.1016/j.ejps.2012.02.006> PMID: 22366112
23. Denny P. Compaction equations: a comparison of the Heckel and Kawakita equations. *Powder Technology*. 2002; 127(2):162–72.

24. Adetunji OA, Odeniyi MA, Itiola OA. Compression, mechanical and release properties of chloroquine phosphate tablets containing corn and trifoliolate yam starches as binders. *Tropical journal of pharmaceutical research*. 2006; 5(2):589–96.
25. Kawakita K, Lüdde K-H. Some considerations on powder compression equations. *Powder technology*. 1971; 4(2):61–8.
26. Adetunji OA, Odeniyi MA, Itiola OA. Compression, mechanical and release properties of chloroquine phosphate tablets containing corn and trifoliolate yam starches as binders. *Tropical journal of pharmaceutical research*. 2007; 5(2):589–96.
27. Costa P, Lobo JMS. Modeling and comparison of dissolution profiles. *European journal of pharmaceutical sciences*. 2001; 13(2):123–33. [https://doi.org/10.1016/s0928-0987\(01\)00095-1](https://doi.org/10.1016/s0928-0987(01)00095-1) PMID: 11297896
28. Vudathala GK, Rogers JA. Dissolution of fludrocortisone from phospholipid coprecipitates. *Journal of pharmaceutical sciences*. 1992; 81(3):282–6. <https://doi.org/10.1002/jps.2600810318> PMID: 1640368
29. Dash S, Murthy PN, Nath L, Chowdhury P. Kinetic modeling on drug release from controlled drug delivery systems. *Acta Pol Pharm*. 2010; 67(3):217–23. PMID: 20524422
30. Shoaib MH, Siddiqi SAS, Yousuf RI, Zaheer K, Hanif M, Rehana S, et al. Development and evaluation of hydrophilic colloid matrix of famotidine tablets. *Aaps Pharmscitech*. 2010; 11(2):708–18. <https://doi.org/10.1208/s12249-010-9427-7> PMID: 20422332
31. Higuchi T. Rate of release of medicaments from ointment bases containing drugs in suspension. *Journal of pharmaceutical sciences*. 1961; 50(10):874–5.
32. Amin MCIM, Albawani SM, Amjad MW. A comparative study of the compaction properties of binary and bilayer tablets of direct compression excipients. *Tropical Journal of Pharmaceutical Research*. 2012; 11(4):585–94.