

Article

Briarenols I—K, New Anti-inflammatory 8,17-Epoxybriaranes from the Octocoral *Briareum excavatum* (Briareidae)

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Abstract: Five 8,17-epoxybriaranes, including three new compounds—briarenols I–K (1–3), along with two known analogues, briaexcavatolide P (4) and briaexcavatin P (5), were isolated from the octocoral *Briareum excavatum*. The structures of briaranes 1–3 were elucidated by spectroscopic methods, including 1D and 2D NMR studies and (+)-HRESIMS. Briarane 4 exerted inhibition effects on inducible nitric oxide synthase (iNOS) and cyclooxygenase-2 (COX-2) release from RAW 264.7.

Keywords: *Briareum excavatum*; briarenol; briarane; anti-inflammatory; iNOS; COX-2

1. Introduction

Octocorals of the genus *Briareum* (family Briareidae) [1–4] are proven to be the most important source to produce briarane-type diterpenoids [5]. The compounds of this type are only found in marine invertebrates, particularly in octocorals and demonstrated a wide spectrum of bioactivities, such as anti-inflammatory activity [6] and cytotoxicity [7]. In our continuing research into the chemical constituents of an octocoral *B. excavatum* (Nutting 1911), which was distributed extensively in the coral reefs of Taiwan, have resulted in isolation of three previously unreported 8,17-epoxybriaranes—briarenols I–K (1–3) along with two known analogues, briaexcavatolide P (4) [8] and briaexcavatin P (5) [9],

(Figure 1). In the current study, the comprehensive workflow of isolation, structure determination, and anti-inflammatory activity evaluation, was implemented on briaranes 1–5.

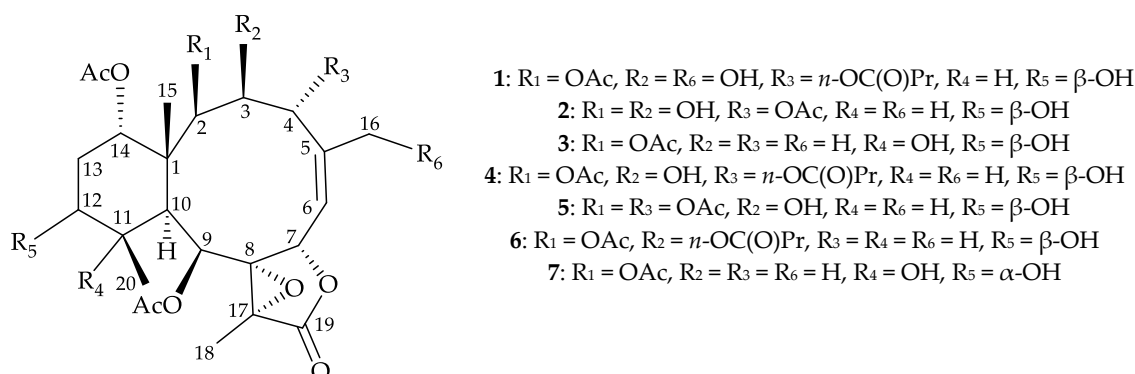


Figure 1. Structures of briarenols I–K (1–3), briaexcavatulide P (4), briaexcavatin P (5), excavatulide B (6), and briareolide B (7).

2. Results and Discussion

Briarenol I (1) was isolated as an amorphous powder and displayed a sodiated adduct ion at m/z 649.24677 in the (+)-HRESIMS, which indicated its molecular formula was C₃₀H₄₂O₁₄ (calculated for C₃₀H₄₂O₁₄ + Na, 649.24668; unsaturation degrees = 10). The IR spectrum revealed absorptions for hydroxy (ν_{\max} 3524 cm⁻¹), γ -lactone (ν_{\max} 1783 cm⁻¹), and ester carbonyl (ν_{\max} 1736 cm⁻¹) moieties. Resonances in the ¹³C NMR of 1 at δ_C 172.9, 172.3, 170.5, 170.0, and 170.0 (5 × C) supported the presence of a γ -lactone and four esters (Table 1). Three of the esters were identified as acetates by the presence of three methyl singlet resonances in the ¹H NMR spectrum at δ_H 2.34, 2.15, and 2.08 (Table 2) and the remaining ester was found to be an *n*-butyryloxy group based on ¹H NMR studies, including a correlation spectroscopy (COSY) experiment, which revealed seven contiguous protons (δ_H 2.30, 2H, t, J = 7.2 Hz; 1.63, 2H, tq, J = 7.2 Hz; 0.95, 3H, t, J = 7.2 Hz). From the COSY spectrum (Figure 2), the proton sequences from H-6/H-7, H-9/H-10/H-11/H-12/H₂-13/H-14, and H-11/H₃-20 were established. The hydroxy proton signals at δ_H 4.30 (1H, d, J = 12.0 Hz), 1.49 (1H, d, J = 4.0 Hz), and 3.49 (1H, dd, J = 9.6, 4.4 Hz) were found to correlate with H-3 (δ_H 4.59, d, J = 12.0 Hz), H-12 (δ_H 4.10, m), and H₂-16 (δ_H 4.35, 1H, dd, J = 13.6, 4.4 Hz; 4.04, 1H, dd, J = 13.6, 9.6 Hz), respectively. Thus, the hydroxy groups should be positioned at C-3, C-12, and C-16, respectively. Olefinic resonances in the ¹³C NMR at δ_C 125.5 (CH-6) and 142.0 (C-5) indicated the presence of a trisubstituted carbon–carbon double bond. On the basis of these data and the heteronuclear multiple bond correlation (HMBC) experiment (Figure 2), the connectivity from C-1 to C-14 was established. A hydroxymethyl group at C-5 was revealed by the HMBC between C-16 oxymethylene protons to C-4, C-5, and C-6. The C-15 methyl group at C-1 was confirmed by the HMBC between H₃-15/C-1, C-2, C-10, C-14, and H-10/C-15. The *n*-butyrate positioned at C-4 was confirmed from the connectivity between H-4 (δ_H 6.14) and the carbonyl carbon of *n*-butyrate group (δ_C 172.3). HMBC from the oxymethine protons at δ_H 4.53 (H-2), 5.32 (H-9), and 4.88 (H-14) to the acetate carbonyls at δ_C 172.9, 170.0, and 170.0, placed the acetoxy groups on C-2, C-9, and C-14, respectively. Thirteen of the fourteen oxygen atoms in the molecular formula of 1 could be accounted for from the presence of a γ -lactone, four esters, and three hydroxy groups. The remaining oxygen atom had to be placed between C-8 and C-17 to form a tetrasubstituted epoxide based on the ¹³C NMR evidences at δ_C 70.8 (C-8) and 62.5 (C-17) and the ¹H NMR chemical shift of a tertiary methyl at δ_H 1.66 (3H, s, H₃-18).

Table 1. ^{13}C NMR (δ_{C} 100 MHz, CDCl_3) data for briaranes 1–3.

| Position | 1 | 2 | 3 |
|---------------------|-----------------------|-----------------------|-----------------------|
| 1 | 43.3, C ^a | 43.7, C | 47.7, C |
| 2 | 87.4, CH | 85.6, CH | 74.9, CH |
| 3 | 73.1, CH | 73.6, CH | 31.6, CH ₂ |
| 4 | 66.0, CH | 65.9, CH | 28.4, CH ₂ |
| 5 | 142.0, C | 139.3, C | 144.8, C |
| 6 | 125.5, CH | 124.3, CH | 118.4, CH |
| 7 | 74.1, CH | 74.0, CH | 74.9, CH |
| 8 | 70.8, C | 69.9, C | 70.8, C |
| 9 | 66.2, CH | 67.1, CH | 67.4, CH |
| 10 | 40.5, CH | 41.4, CH | 49.0, CH |
| 11 | 37.2, CH | 36.4, CH | 78.2, C |
| 12 | 66.6, CH | 67.0, CH | 73.4, CH |
| 13 | 30.2, CH ₂ | 30.4, CH ₂ | 30.2, CH ₂ |
| 14 | 80.5, CH | 80.1, CH | 74.8, CH |
| 15 | 18.6, CH ₃ | 19.1, CH ₃ | 14.3, CH ₃ |
| 16 | 62.5, CH ₂ | 16.8, CH ₃ | 27.2, CH ₃ |
| 17 | 62.5, C | 61.8, C | 66.5, C |
| 18 | 10.3, CH ₃ | 10.3, CH ₃ | 10.4, CH ₃ |
| 19 | 170.5, C | 170.9, C | 170.4, C |
| 20 | 8.9, CH ₃ | 8.7, CH ₃ | 16.9, CH ₃ |
| OAc-2 | 172.9, C | | 170.2, C |
| | 21.2, CH ₃ | | 21.2, CH ₃ |
| OAc-4 | | 169.5, C | |
| | | 21.0, CH ₃ | |
| OAc-9 | 170.0, C | 169.2, C | 168.1, C |
| | 21.5, CH ₃ | 21.1, CH ₃ | 21.5, CH ₃ |
| OAc-14 | 170.0, C | 170.0, C | 170.4, C |
| | 21.2, CH ₃ | 21.0, CH ₃ | 21.3, CH ₃ |
| <i>n</i> -OC(O)Pr-4 | 172.3, C | | |
| | 35.9, CH ₂ | | |
| | 18.2, CH ₂ | | |
| | 13.7, CH ₃ | | |

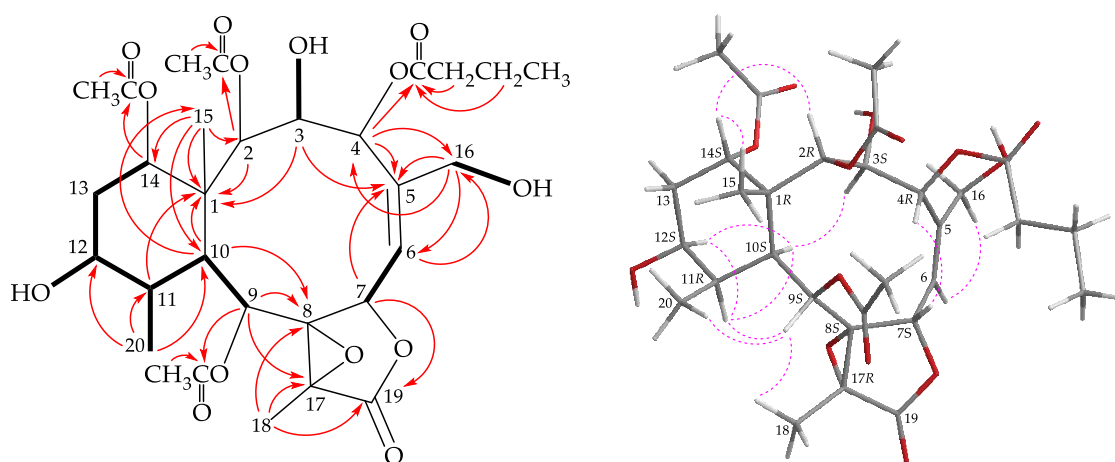
^a Multiplicity deduced by DEPT and HSQC spectra.**Figure 2.** The COSY (—) correlations, selective HMBC (↷), and protons with key NOESY correlations (⋯) of 1.

Table 2. ^1H NMR (δ_{H} , 400 MHz in CDCl_3) data (J in Hz) for briaranes 1–3.

| Position | 1 | 2 | 3 |
|-----------------------|---|--------------------|------------------------------------|
| 2 | 4.53 s | 3.45 d (10.4) | 5.13 d (8.4) |
| 3 α / β | 4.59 d (12.0) | 4.27 d (10.4) | 1.67 m; 2.60 ddd (16.0, 14.8, 6.0) |
| 4/4' | 6.14 s | 6.05 d (1.2) | 2.48 br d (16.0); 1.90 m |
| 6 | 5.53 d (6.0) | 5.29 dq (6.4, 1.6) | 5.19 s |
| 7 | 5.62 d (6.0) | 5.71 d (6.4) | 5.19 s |
| 9 | 5.32 d (8.8) | 5.26 d (9.2) | 5.78 d (1.2) |
| 10 | 2.64 dd (8.8, 4.8) | 2.55 dd (9.2, 5.2) | 2.14 br s |
| 11 | 2.41 m | 2.47 m | |
| 12 | 4.10 m | 4.05 m | 3.72 dd (12.4, 4.8) |
| 13 α / β | 1.75 m; 2.01 m | 1.69 m; 2.00 m | 1.67 m; 2.04 m |
| 14 | 4.88 dd (2.8, 2.8) | 4.92 dd (2.8, 2.8) | 4.79 dd (2.4, 2.0) |
| 15 | 0.99 s | 0.99 br s | 1.21 s |
| 16a/b | 4.35 dd (13.6, 4.4); 4.04 dd (13.6, 9.6) | 1.89 br s | 1.99 s |
| 18 | 1.66 s | 1.66 s | 1.77 s |
| 20 | 1.12 d (6.8) | 1.07 d (7.2) | 1.15 s |
| OH-2 | | 2.79 d (10.4) | |
| OH-3 | 4.30 d (12.0) | 2.87 d (10.4) | |
| OH-12 | 1.49 d (4.0) | 1.43 d (4.0) | - |
| OH-16 | 3.49 dd (9.6, 4.4) | | |
| OAc-2 | 2.08 s | | 2.00 s |
| OAc-4 | | 2.14 s | |
| OAc-9 | 2.34 s | 2.32 s | 2.22 s |
| OAc-14 | 2.15 s | 2.16 s | 2.03 s |
| <i>n</i> -OC(O)Pr-4 | 0.95 t (7.2) 1.63 tq (7.2) 2.30 t (7.2) | | |

The stereochemistry of **1** was deduced from an NOESY experiment (Figure 2) and biogenetic considerations. The NOE correlations of H-10/H-11, H-10/H-12, and H-11/H-12 indicated that these protons were situated on the same face of the structure and were assigned as the α protons since the C-15 methyl is the β -substituent at C-1. The NOE correlation between H₃-15 and H-14 implied that H-14 had a β -orientation. H-3 exhibited a correlation with H-10, and, as well as a lack of coupling constants were detected between H-2/H-3 and H-3/H-4, indicating the dihedral angles between H-2/H-3 and H-3/H-4 were approximately 90° and the 2-acetoxy, 3-hydroxy, and 4-*n*-butyryloxy groups were β -, β -, and α -oriented, respectively. A correlation from H-4 to H-7, suggested that H-7 was β -oriented. The *Z*-configuration of C-5/6 double bond was confirmed based on the fact that the C-6 olefinic proton (δ_{H} 5.53) correlated to one of the C-16 hydroxymethyl protons (δ_{H} 4.04). H-9 was found to correlate with H-11, H₃-18, and H₃-20. From a consideration of molecular model, H-9 was found to be reasonably close to H-11, H₃-18, and H₃-20, thus, H-9 should be placed on the α face, and Me-18 was β -oriented in the γ -lactone moiety, and the 8,17-epoxy group should be α -oriented. It was found that the NMR signals of **1** were similar to those of a known briarane, briaexcavatolide P (**4**) (Figure 1) [8], except that the signals corresponding to the Me-16 vinyl methyl in **4** were replaced by signals for a hydroxymethyl group in **1**. Additionally, as briaranes **1**–**5** were isolated along with a known briarane, excavatolide B (**6**) [6,10,11] from the same target organism, *B. excavatum*, and the absolute configuration of **6** was determined by a single-crystal X-ray diffraction analysis [6,11]. Therefore, on biogenetic grounds to assume that briaranes **1**–**5** had the same absolute stereochemistry as that of **6**, tentatively, and the configurations of stereogenic carbons of **1** were determined as 1*R*,2*R*,3*S*,4*R*,7*S*,8*S*, 9*S*,10*S*,11*R*,12*S*,14*S*, and 17*R* (Supplementary Materials, Figures S1–S10).

Briarenol J (**2**) had a molecular formula $\text{C}_{26}\text{H}_{36}\text{O}_{12}$ by its (+)-HRESIMS at m/z 563.21007 (calculated for $\text{C}_{26}\text{H}_{36}\text{O}_{12} + \text{Na}$, 563.20990). The IR spectrum showed bands at 3483, 1779, and 1727 cm^{-1} , consistent with the presence of hydroxy, γ -lactone, and ester groups, respectively, in **2**. From the ^{13}C and DEPT data (Table 2), one trisubstituted double bond was deduced from the signals of two carbons at δ_{C} 139.3 (C-5) and 124.3 (CH-6). A methyl-containing tetrasubstituted epoxy group was confirmed from the signals of two oxygenated quaternary carbons at δ_{C} 69.9 (C-8) and 61.8 (C-17), and from the

chemical shift of a tertiary methyl (δ_{H} 1.66, 3H, s; δ_{C} 10.3, CH₃-18; Tables 1 and 2). Four carbonyl resonances at δ_{C} 170.9, 170.0, 169.5, and 169.2 in the ¹³C spectrum confirmed the presence of a γ -lactone and three esters. All the esters were identified as acetates by the presence of three methyl singlet resonances in the ¹H NMR spectrum at δ_{H} 2.32, 2.16, and 2.14, respectively.

Coupling constants information in the COSY spectrum of **2** enabled identification of H-6/H-7, H-9/H-10/H-11/H-12/H₂-13/H-14, H-11/H₃-20, and H-6/H₃-16 (by allylic coupling; Figure 3), these data, together with the HMBC experiment (Figure 3), the molecular framework of **2** could be established. The HMBC also indicated that the acetoxy groups should be attached at C-4, C-9, and C-14, respectively. Thus, the remaining hydroxy groups have to be positioned at C-2, C-3, and C-12, as indicated by the COSY correlations between H-2/OH-2, H-3/OH-3, and H-12/OH-12.

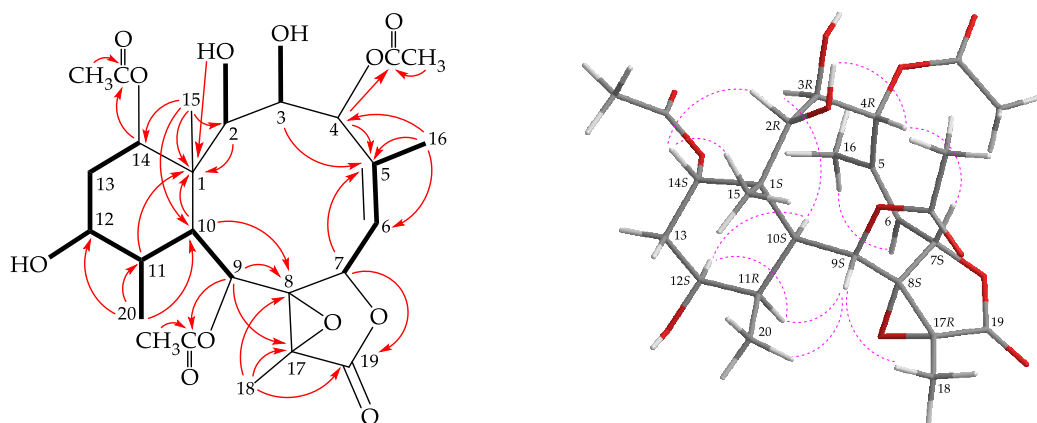


Figure 3. The COSY (—) correlations, selective HMBC (↪), and protons with key NOESY correlations (⋯) of **2**.

The stereochemistry of **2** was elucidated from the NOE interactions observed in a NOESY experiment (Figure 3) and by the vicinal ¹H–¹H coupling constant analysis. In the NOESY spectrum, correlations were observed between H-10 with H-3 and H-12; and H-12 correlated with H-11, indicating that these protons should be α -oriented. H-14 gave a correlation with H₃-15, confirming the β -orientation for this proton. H-2 showed a correlation with H-14, and a lack of coupling constant was detected between H-2/H-3, indicating the dihedral angle between H-2/H-3 is approximately 90° and the 2-hydroxy group was β -oriented. H-4 exhibited correlations with H-7 and 2-hydroxy proton, confirming the β -orientations for H-4 and H-7. H-9 was found to show correlations with H-11, H₃-18, and H₃-20, and from molecular models, H-9 and H₃-18 should be placed on the α - and β -face, respectively. The *Z*-configuration of C-5/C-6 double bond was elucidated by a correlation between H-6 and H₃-16. The NMR data of **2** were found to be similar to those of a known briarane, briarexavatin P (**5**) [9]. It was found that the 2-acetoxy substituent in **5** was replaced by a hydroxy group in **2**. By comparison of the proton and carbon chemical shifts, coupling constants, NOESY correlations, and rotation value of **2** with those of **5**, the stereochemistry of **2** was confirmed to be the same as that of **5**, and the configurations of the stereogenic centers of **2** were assigned as 1*S*,2*R*,3*R*,4*R*,7*S*,8*S*,9*S*,10*S*,11*R*,12*S*,14*S*, and 17*R* (Supplementary Materials, Figures S11–S20).

Briarane **3** (briarenol K) was found to have a molecular formula of C₂₆H₃₆O₁₁ based on its (+)-HRESIMS peak at *m/z* 547.21514 (calculated for C₂₆H₃₆O₁₁ + Na, 547.21498). Its absorption peaks in the IR spectrum showed ester carbonyl, γ -lactone, and broad OH stretching at 1739, 1780, and 3468 cm⁻¹, respectively. The ¹³C NMR spectrum indicated that three esters and a γ -lactone were present, as carbonyl resonances were observed at δ_{C} 168.1, 170.2, 170.4, and 170.4, respectively (Table 1). The ¹H NMR data also indicated that presence of three acetate methyls at δ_{H} 2.22, 2.03, and 2.00 (each 3H \times s; Table 2). It was found that the spectroscopic data of **3** were similar to those of a known briarane, briareolide B (**7**) [12]; however, by comparison of the ¹H and ¹³C NMR chemical shifts of CH-12 oxymethine (δ_{H} 3.72, 1H, dd, *J* = 12.4, 4.8 Hz; δ_{C} 73.4), CH₂-13 sp³ methylene (δ_{H} 1.67, 1H,

m; 2.04, 1H, m; δ_C 30.2), C-11 oxygenated quaternary carbon (δ_C 78.2), and Me-20 tertiary methyl (δ_H 1.15, 3H, s; δ_C 16.9) of **3** with those of **7** (δ_H 3.56, 1H, m; δ_C 73.9, CH-12; δ_H 2.03, 1H, m; 2.12, 1H, m; δ_C 27.6, CH₂-13; δ_C 74.7, C-11; δ_H 1.16, 3H, s; δ_C 22.5, Me-20) [12] showed that the hydroxy group at C-12 in **3** was β -oriented. The locations of the functional groups were further confirmed by other HMBC and COSY correlations (Figure 4), and hence briarenol K was assigned the structure of **3**. The NOESY spectrum exhibited a correlation from H-10 to H-12, further supporting that H-12 was α -oriented and the stereogenic centers of **3** were assigned as 1*S*,2*S*,7*S*,8*S*,9*S*,10*S*,11*S*,12*S*,14*S*, and 17*R*, by the correlations observed in a NOESY spectrum (Figure 4) and this compound was found to be the 12-epimer of briareolide B (**7**) [12] (Supplementary Materials, Figures S21–S30).

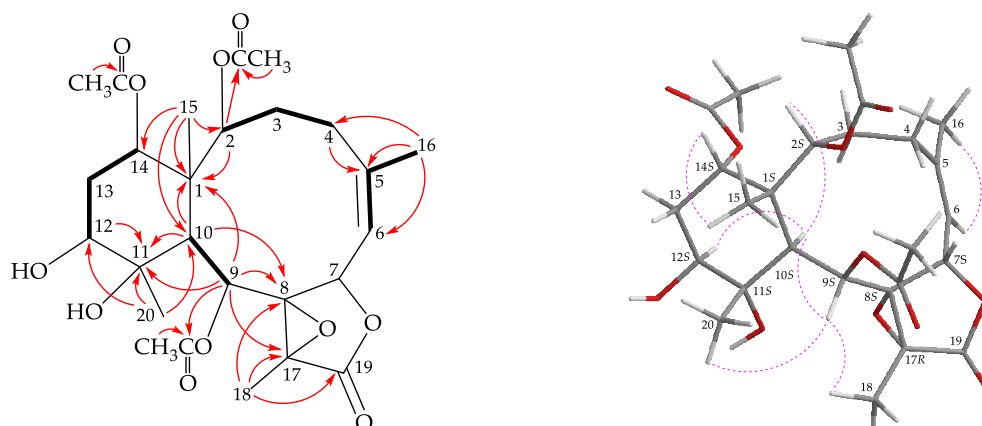


Figure 4. The COSY (—) correlations, selective HMBC (↷), and protons with key NOESY correlations (⋯) of **3**.

The inhibition effects of briaranes **1–5** on the release of inducible nitric oxide synthase (iNOS) and cyclooxygenase-2 (COX-2) protein from lipopolysaccharides (LPS)-stimulated RAW 264.7 were assessed. The results showed that briarane **4** reduced the release of iNOS and COX-2 to 35.37% and 54.61% at a concentration of 10 μ M, respectively (Figure 5 and Table 3). Briarane **1** was found to be weaker than those of **4** in term of reducing the expression of iNOS and COX-2, indicating that the hydroxy group at C-16 in **1** reduced the activity.

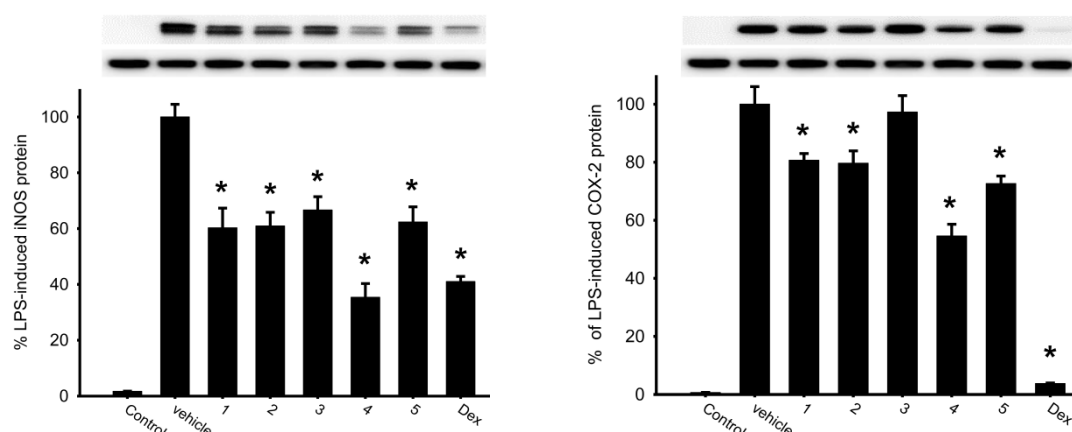


Figure 5. Western blotting showed that briarane **4** downregulated the expression of iNOS and COX-2. Data were normalized to the cells treated with LPS only, and cells treated with dexamethasone (Dex; 10 μ M) were used as a positive control. Data are expressed as the mean \pm SEM ($n = 2-4$). * Significantly different from cells treated with LPS ($p < 0.05$).

Table 3. Effects of briaranes 1–5 on LPS-induced pro-inflammatory iNOS and COX-2 protein expression in macrophages.

| | iNOS | COX-2 | β -Actin | <i>n</i> |
|------------------|-----------------------|-----------------------|-----------------------|----------|
| | Expression (% of LPS) | Expression (% of LPS) | Expression (% of LPS) | |
| Negative Control | 1.71 \pm 0.13 | 0.62 \pm 0.09 | 120.48 \pm 1.28 | 2 |
| LPS | 100.00 \pm 4.53 | 100.00 \pm 6.05 | 100.00 \pm 3.09 | 4 |
| 1 | 60.27 \pm 7.05 | 80.63 \pm 2.32 | 100.29 \pm 2.46 | 4 |
| 2 | 60.94 \pm 4.89 | 79.65 \pm 4.27 | 98.29 \pm 3.35 | 4 |
| 3 | 66.64 \pm 4.79 | 97.28 \pm 5.66 | 100.49 \pm 6.44 | 4 |
| 4 | 35.37 \pm 4.94 | 54.61 \pm 4.03 | 104.56 \pm 2.83 | 4 |
| 5 | 62.36 \pm 5.42 | 72.63 \pm 2.6 | 104.79 \pm 2.76 | 4 |
| Dexamethasone | 41.00 \pm 2.63 | 3.73 \pm 0.35 | 104.24 \pm 5.82 | 2 |

Data were normalized to those of cells treated with LPS alone, and cells treated with dexamethasone were used as a positive control. Data are expressed as the mean \pm SEM ($n = 2$ –4).

3. Materials and Methods

3.1. General Experimental Procedures

Optical rotation values were measured using a Jasco P-1010 digital polarimeter (Japan Spectroscopic, Tokyo, Japan). IR spectra were measured on a Thermo Scientific Nicolet iS5 FT-IR spectrophotometer (Waltham, MA, USA). NMR spectra were taken on a Jeol Resonance ECZ 400 S NMR spectrometer (Tokyo, Japan), using the residual CHCl_3 signal (δ_{H} 7.26 ppm) and CDCl_3 (δ_{C} 77.1 ppm) as the internal standard for ^1H and ^{13}C NMR, respectively; coupling constants (J) are presented in Hz. ESIMS and HRESIMS were recorded using a Bruker 7 Tesla solarix FTMS system. Column chromatography was carried out with silica gel (230–400 mesh, Merck, Darmstadt, Germany). TLC was performed on plates precoated with Kieselgel 60 F₂₅₄ (0.25-mm-thick, Merck, Darmstadt, Germany), then sprayed with 10% H_2SO_4 solution followed by heating to visualize the spots. Normal-phase HPLC (NP-HPLC) was performed using a system comprised of a Hitachi L-7100 pump (Tokyo, Japan) and a Rheodyne 7725i injection port (Rohnert Park, CA, USA). Reverse-phase HPLC (RP-HPLC) was performed using a system comprised of a Hitachi L-2130 pump (Tokyo, Japan), a Hitachi L-2455 photodiode array detector (Tokyo, Japan), and a Rheodyne 7725i injection port (Rohnert Park, CA, USA). A semipreparative normal-phase column (YMC-Pack SIL, S-5 μm , 250 mm \times 20 mm, Sigma-Aldrich, St. Louis, MO, USA) was used for NP-HPLC. A semipreparative reverse-phase column (Luna, 5 μm , C18(2) 100 Å, AXIA Packed, 250 mm \times 21.2 mm; Phenomenex, Torrance, CA, USA) was used for RP-HPLC.

3.2. Animal Material

Specimens of *B. excavatum* were collected in June 2017 by hand with self-contained underwater breathing apparatus (SCUBA) divers off the coast of Lanyu Island (Orchid Island), Taiwan. The samples were then stored in a -20°C freezer until extraction. A voucher specimen was deposited in the National Museum of Marine Biology and Aquarium, Taiwan (NMMBA-TW-SC-2017-418). Identification of the species of this organism was performed by comparison as described in previous publications [1–4].

3.3. Extraction and Isolation

The freeze-dried and sliced bodies (wet/dry weight = 1344/568 g) of the specimen were extracted with supercritical CO_2 to give 58.9 g of extract. Partial extract (36.4 g) was then applied on silica gel column and eluted with gradients of *n*-hexane/EtOAc to furnish fractions A–K. Fraction F was purified by NP-HPLC using a mixture of *n*-hexane/acetone (4:1) to yield fractions F1–F13. Fraction F6 was repurified by RP-HPLC, using a mixture of MeOH/ H_2O (60:40; at a flow rate = 4 mL/min) to afford 4 (6.7 mg). Fraction G was separated by NP-HPLC, using a mixture of *n*-hexane/acetone (3:1) to yield fractions G1–G12. Fractions G6 and G7 were repurified by RP-HPLC using a mixture of MeOH/ H_2O

(60:40; at a flow rate = 4.0 mL/min) to afford **5** (1.3 mg) and **3** (1.0 mg), respectively. Fraction H was separated by NP-HPLC using a mixture of *n*-hexane and acetone (3:1) to yield fractions H1–H18. Fractions H12 and H15 were repurified by RP-HPLC, using a mixture of MeOH/ H₂O (60:40; at a flow rate = 4.0 mL/min) to afford **2** (2.1 mg) and **1** (0.6 mg), respectively.

Briarenol I (**1**): Amorphous powder; $[\alpha]_D^{22} + 207$ (c 0.03, CHCl₃), IR (ATR) ν_{\max} 3524, 1783, 1736, 1222, 891 cm⁻¹; ¹³C (100 MHz, CDCl₃) and ¹H (400 MHz, CDCl₃) NMR data (see Tables 1 and 2); ESIMS: *m/z* 649 [M + Na]⁺; HRESIMS *m/z* 649.24677 (calculated for C₃₀H₄₂O₁₄ + Na, 649.24668).

Briarenol J (**2**): Amorphous powder; $[\alpha]_D^{26} + 140$ (c 0.08, CHCl₃), IR (ATR) ν_{\max} 3483, 1779, 1727, 1220, 890 cm⁻¹; ¹³C (100 MHz, CDCl₃) and ¹H (400 MHz, CDCl₃) NMR data (see Tables 1 and 2); ESIMS: *m/z* 563 [M + Na]⁺; HRESIMS *m/z* 563.21007 (calculated for C₂₆H₃₆O₁₂ + Na, 563.20990).

Briarenol K (**3**): Amorphous powder; $[\alpha]_D^{23} + 37$ (c 0.06, CHCl₃), IR (ATR) ν_{\max} 3468, 1780, 1739, 1255, 892 cm⁻¹; ¹³C (100 MHz, CDCl₃) and ¹H (400 MHz, CDCl₃) NMR data (see Tables 1 and 2); ESIMS: *m/z* 547 [M + Na]⁺; HRESIMS *m/z* 547.21514 (calculated for C₂₆H₃₆O₁₁ + Na, 547.21498).

Briaexcavatulide P (**4**): Amorphous powder; $[\alpha]_D^{24} + 182$ (c 0.3, CHCl₃) (ref. [8], $[\alpha]_D^{27} + 167$ (c 1.0, CHCl₃)), IR (ATR) ν_{\max} 3513, 1783, 1731, 1218, 889 cm⁻¹; ¹H and ¹³C NMR data were found to be in agreement with previous study [8]; ESIMS: *m/z* 633 [M + Na]⁺.

Briaexcavatin P (**5**): Amorphous powder; $[\alpha]_D^{23} + 134$ (c 0.05, CHCl₃) (ref. [9], $[\alpha]_D^{25} + 198$ (c 0.08, CHCl₃)), IR (ATR) ν_{\max} 3503, 1785, 1735, 1240, 889 cm⁻¹; ¹H and ¹³C NMR data were found to be in agreement with previous study [9]; ESIMS: *m/z* 605 [M + Na]⁺.

3.4. In Vitro Anti-inflammatory Assay

The proinflammatory suppression assay was employed to assess the activities of the isolated compounds **1–5** against the release of iNOS and COX-2 from macrophage cells as the literature reported [13–15].

4. Conclusions

B. excavatum was demonstrated to have a wide structural diversity of briarane-type diterpenoids that possessed various pharmacological properties, especially in anti-inflammatory activity. In our continued study on *B. excavatum*, three previously unreported briaranes, briarenols I–K (**1–3**), along with the known analogues, briaexcavatulide P (**4**) and briaexcavatin P (**5**), were isolated. In the present study, the anti-inflammatory activity of **1–5** was assessed using inhibition of pro-inflammatory iNOS and COX-2 release from macrophages. The results indicated that briaexcavatulide P (**4**) showed the most potent suppressive effect on iNOS release.

Supplementary Materials: The Supplementary Materials are available online. ESIMS, HRESIMS, IR, 1D and 2D NMR spectra of new compounds **1–3**.

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