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Basic Quinolinonyl Diketo Acid Derivatives as Inhibitors of HIV Integrase and their Activity against RNase H Function of Reverse Transcriptase

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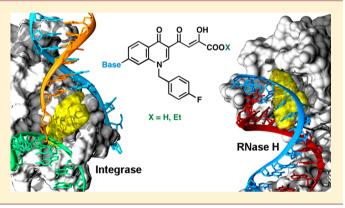
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Supporting Information

ABSTRACT: A series of antiviral basic quinolinonvl diketo acid derivatives were developed as inhibitors of HIV-1 IN. Compounds 12d,f,i inhibited HIV-1 IN with IC₅₀ values below 100 nM for strand transfer and showed a 2 order of magnitude selectivity over 3'-processing. These strand transfer selective inhibitors also inhibited HIV-1 RNase H with low micromolar potencies. Molecular modeling studies based on both the HIV-1 IN and RNase H catalytic core domains provided new structural insights for the future development of these compounds as dual HIV-1 IN and RNase H inhibitors.



INTRODUCTION

HIV/AIDS remains one of the most important global health challenges, especially in sub-Saharan Africa.^{1,2} Highly active antiretroviral therapy (HAART),³ the standard of care for HIV/ AIDS, comprises a multitarget regimen combining antiviral drugs with orthogonal mechanisms of action, thus increasing the genetic barrier against resistance selection when compared to monotherapy. Nevertheless, treatment adherence resides primarily on treatment tolerance and simplicity of administration, which remains a challenge with multipill HAART cocktails.⁴ A single compound capable of inhibiting simultaneously two viral targets could represent a therapeutic alternative. Multitarget inhibitors may alleviate dosing complexity, drug-drug interactions, and toxicities.⁵ In the field of medicinal chemistry, the design of active dual inhibitors against HIV reverse transcriptase (RT) and integrase (IN) is subject of great interest.⁶ These inhibitors act on the catalytic sites of the IN enzyme and the ribonuclease H (RNase H) domain of HIV

RT. IN contains three catalytic carboxylate residues, D64, D116, and E152, forming the DDE motif that coordinates two magnesium atoms of the IN catalytic site. Many HIV-1 IN inhibitors with metal-complexing properties have been reported.⁷ These inhibitors are referred to as strand transfer IN inhibitors (INSTIs). Three INSTIs, elvitegravir (EVG, 1), raltegravir (RAL, 2), and dolutegravir (DTG, 3) have already been approved by the Food and Drug Administration (Figure $1).^{8,9}$

RT is another important HIV-1 enzyme and the target of many anti-HIV drugs. This enzyme has RNA- and DNAdependent DNA polymerase, strand displacement, strand transfer, and RNase H activities.¹⁰ RNase H activity, which degrades RNA from RNA-DNA hybrid molecules, is required at several steps during reverse transcription and essential for

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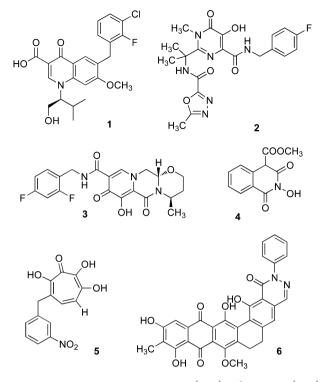


Figure 1. Anti HIV-1 agents targeting IN (1-3) and RNase H (4-6).

virus replication. The crystal and NMR structures of isolated HIV RNase H domain are similar to that of the RNase H domain in the context of the full-length HIV-RT protein.¹¹

These structures also showed that the folding of the HIV-1 RNase H catalytic core domain (CCD) is similar to that of HIV-1 IN and, consequently, the catalytic sites of the two enzymes share a similar geometry. Indeed, also RNase H features the DDE catalytic motif (comprising D443, E478, and D498 residues) chelating two magnesium ions, although a fourth carboxylate residue (D549) is required for catalysis.¹² Similar structural characteristics including three aspartate residues and two magnesium ions at a distance of 3.57 Å from each other were shown in the DNA polymerase active site of RT in complex with DNA primer template and an incoming nucleotide.¹²

Some diketo acid inhibitors of HIV-1 IN have shown activity on RNase H,^{13,14} whereas DNA aptamers used as inhibitors of RNase H have also been employed to inhibit HIV-1 IN.¹⁵ Tropolone (**5**),¹⁶ madurahydroxylactone (**6**),¹⁷ and 2-hydroxyquinoline-1,3(2H,4H)-dione (**4**)¹⁸ were recently described to be able to inhibit both enzymes (Figure 1).

Therefore, concurrent inhibition of HIV-1 RNase H and IN by compounds with metal-chelating activity could represent an interesting approach to alleviate some of the problems linked to multipill combination therapies.

The work of our research group commenced in the context of diketo acids (DKAs) compounds as HIV-1 IN inhibitors.¹⁹ The design of quinolinone derivatives characterized by one DKA moiety on position 3 of the quinolinone ring led to HIV-1 IN inhibitors with selectivity for strand transfer versus 3'-processing. Moreover, inserting a functional group with basic properties on position 7 of the quinolinone ring generated derivative 7 (Figure 2), which inhibited ST with nanomolar IC₅₀ value and submicromolar antiviral activity.^{19b} These results prompted us to investigate the position 7 of the quinolinone ring of newly designed DKAs, keeping unchanged the skeleton

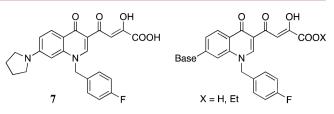


Figure 2. 1-Pyrrolidinyl substituted quinolinonyl DKA (7) and general structure of "base" substituted quinolinonyl DKAs.

of the quinolinonyl diketo acid inhibitors, including the *p*-fluoro benzyl group linked to the quinolone nitrogen.

Because of the similarities of the IN and RNase H catalytic sites and reports on dual activities of a number of compounds against IN/RNase H, we decide to test our newly designed quinolonyl DKAs against both enzymes.^{19b}

Herein we describe the synthesis and biological studies of a series of DKA derivatives that present a *p*-fluoro benzyl group on quinolone nitrogen and a variable "base-like" functional group in position 7 of the quinolone ring as inhibitors of HIV-1 IN. A comparative evaluation of the activity against RNase H function of the HIV RT of the newly synthesized compounds was also performed (Figure 2).

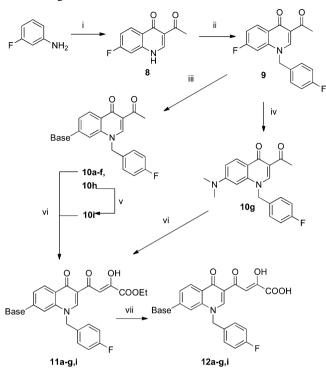
RESULTS AND DISCUSSION

Chemistry. Synthesis of derivatives **11a–g**,**i** and **12a–g**,**i** is outlined in Scheme 1. 3-Acetyl-4(1*H*)-quinolinone **8** were prepared by reaction of 3-fluoroaniline with ethyl orthoformate and ethyl acetoacetate, which were thermally condensed in the presence of an inert heating medium (dowtherm A) under argon atmosphere according to the Yoshizawa procedure.²⁰ Then **8** was alkylated with 4-fluorobenzyl bromide in alkaline medium (K₂CO₃) to give the N-1 substituted quinolinone **9**.

Derivatives 10a-f and 10h were obtained in few minutes with good yields by substituting the fluorine atom of 9 with the appropriate base in the presence of NEt₃ under microwave irradiation. Using the same microwave assisted technique, we obtained 10g from 9 using *N*,*N*-dimethylamine (aq soln 40%) as base. 10h was converted to the desired chlorinated derivative 10i by reaction with PPh₃ in CCl₄ and CH₂Cl₂ using a modified procedure of a patent of Otsuka Pharmaceutical.²¹ Derivatives 10a-f and 10i were then condensed with ethyl oxalate, and the resulting esters 11a-g,i were hydrolyzed to afford the required acids 12a-g,i (Scheme 1). Chemical, physical, and analytical data of intermediates 8, 9, and 10a-i, together with the spectroscopic data, are reported in Supporting Information. Data for final products 11a-g,i and 12a-g,i are listed in the Experimental Section.

Evaluation of Biological Activities. All newly synthesized compounds **11a**–**g**,**i** and **12a**–**g**,**i** were tested in vitro for their ability to inhibit RNase H activity of recombinant RT, and on recombinant IN to evaluate their potencies against its 3'-processing and strand transfer functions. IN activities were determined using gel-based assays and the corresponding IC₅₀ values were generated from the dose–response curve and are summarized in Table 1.

Anti-IN Activities. The newly synthesized DKAs inhibited IN with nano- to submicromolar potencies for strand transfer and IC₅₀ values ranging from 2.5 to of 333 μ M for 3'processing. All compounds exhibited some selectivity for strand transfer versus 3'-processing, with half of the compounds showing a selectivity of approximately 2 orders of magnitude or greater (Table 1). Carboxylic acid-containing compounds were Scheme 1. Synthetic Route to Quinolinonyl DKAs 11a-g,i and 12a-g,i^a



"Reagents and conditions: (i) triethyl orthoformate, ethyl acetoacetate, Dowtherm A, 95–254 °C, 8 h, 41% yield; (ii) 4-fluorobenzyl bromide, K_2CO_3 , DMF, 100 °C, 1 h, 29% yield; (iii) proper base, Et₃N, DMF, microwave, 100 W, 153 °C, 10 min, 14–78% yield; (iv) *N*,*N*dimethylamine aq soln 40%, 100 W, 175 °C, 10 min, 52% yield; (v) CCl₄, CH₂Cl₂, PPh₃, reflux, 24 h, 48% yield; (vi) diethyl oxalate, C₂H₃ONa, THF, room temp, 2 h, 76–99% yield; (vii) 1 N NaOH, THF/CH₃OH, room temp, 40 min, 30–100% yield.

always more potent than their corresponding ester counterparts.

Three compounds (12d,f,i) inhibited strand transfer with IC₅₀ values below 100 nM with a 100-fold selectivity over the 3'-processing step. Compounds $12d_{1}f$ showed an antiviral EC₅₀ of 17 and 26.3 μ M, respectively. Compound 12i was unable to block replication, possibly indicating reduced uptake. The least active acid-containing compound was 12a (IC₅₀ 2.0 μ M), in which the pyrrolidine of the lead was replaced by a piperazine moiety. This compound was the only one containing an unsubstituted heteroatom on the 7-substituent. In fact, when the quinolinone ring was substituted with piperazine bioisosters, like morpholine (12f) and thiomorpholine (12e), inhibition of strand transfer activity increased of 1-2 orders of magnitude (IC₅₀ 0.08 and 0.36 μ M, respectively). The same trend was observed when the piperazine hydrogen was replaced by alkyl or acyl chain. Among the 4-substituted piperazine derivatives, increased potency was achieved when the N-4 of the piperazine ring was substituted with acetyl (12d) or 3chloropropyl (12i) chains (IC₅₀ 0.08 and 0.05 μ M, respectively). Substitution with methyl (12b) or ethyl (12c) chains lead to submicromolar activity (IC₅₀ 0.10 and 0.21 μ M, respectively), indicating that these modifications were not optimal for strand transfer inhibition. Cumulatively, these results demonstrate that quinolinone derivatives with piperazine bioisoster substitutions are potent INSTIs with antiviral activity.

Inhibition of HIV-1 RNase H Activity. The newly synthesized quinolinone DKA also inhibited the RNase H function of HIV-1 RT. RNase H inhibition was first calculated as percentage of inhibition at a single concentration of 10 μ M, then for the most potent compounds the IC₅₀ values were determined (Table 1). Interestingly, 4-methylpiperazine (11b) and 4-ethylpiperazine (11c) ester derivatives were more active than the corresponding acids (12b and 12c, respectively) when tested against RNase H (41% and 63% inhibition (not shown) and IC₅₀ 18.5 and 26.2 μ M, respectively, versus 34% and 22% of the corresponding acids, Table 1). Five derivatives, including the lead compound 7b, inhibited RNase H at micromolar concentration showing IC₅₀s in the range 3.3–6.8 μ M: the 4acetyl-1-piperazine (12d), the morpholine (12f), the dimethylamine (12g), and the 4-(3-chloropropyl)-1-piperazine (12i) acid derivatives. As representative examples, compounds 7b, 12f, and 12i were also tested against the polymerase function of RT and were found inactive up to 40 μ M, suggesting that the observed inhibition is RNase H-mediated and not an indirect inhibition of the polymerase function of RT. RAL (2), EVG (1), and DTG (3, data not shown) were also tested for their ability to inhibit RNase H and found to be inactive, demonstrating that not all potent INSTIs can efficiently inhibit this enzyme in vitro (Table 1). These results demonstrate that in general the best INSTIs (7b and 12d,f,i) were also the most potent compounds to inhibit RNase H (Table 1).

Cell-Based Assays. All compounds including the previously reported 7a, 7b, RAL, and EVG were tested for antiviral activity and cytotoxicity using HeLa-CD4-LTR- β -gal cells infected by HIV-1(IIIB). EC₅₀ and CC₅₀ values for compounds **11a**–g,**i** and **12a**–g,**i** are reported in Table 1. In this assay, 7b exhibited an EC₅₀ value of 14 μ M, which is sensibly higher than the originally reported EC₅₀ value of 0.17 μ M in an HTLV-IIIB-based infection assay.^{19b} Among the tested compounds, derivatives **11a**–c,f,g,**i** and **12d**,f,g show EC₅₀ < 50 μ M. In particular, **11c** and **11g** are active in the micromolar range (EC₅₀ = 3.6 and 2.5 μ M, respectively), with selectivity indices (SI) of 14 and 20, respectively. All active compounds were characterized by low cytotoxicity against the same HeLa-CD4-LTR- β -gal cells, showing CC₅₀ values >50 μ M.

Molecular Modeling. To better rationalize the structure– activity relationship (SAR) data obtained for this new series of DKAs, molecular modeling studies were performed. Given the inhibitory activity of these compounds, docking studies were undertaken to detect the structural features responsible for the binding at the active sites of both the aforementioned enzymes.

Several X-ray structures of HIV-1 IN CCD have been reported,²³⁻³⁸ but unfortunately none of them were cocrystallized with nucleic acids. On the other hand, a number of crystal structures of the intasome (comprising an IN tetramer assembled on a pair of viral DNA ends) from the prototype foamy virus (PFV) have recently been resolved.³⁹⁻⁴² These pioneering studies have demonstrated that two \mbox{Mg}^{2+} ions are chelated by residues of the DDE motif (D64, D116, and E152 in HIV-1) inside the active site. Furthermore, both cations seem to be required for both IN catalytic activity and INSTI inhibition. Additionally, it has been proven that an induced-fit mechanism of inhibition for INSTIs occurs through displacement of the terminal 3'-adenosine of the viral DNA from the active site. Given these recent advances, we decided to revise our previous theoretical model³⁷ by constructing a new homology model of the HIV IN CCD/viral DNA complex, in its INSTI-inhibited form, starting from the structure of fullTable 1. Cytotoxicity, Antiviral, Anti-IN, and Anti-RNase H Activities of Compounds 11a-g,i and 12a-g,i

Activity in enzyme assays

Antiviral activity and cytotoxicity

	IC_{50}^{a} (µM)		RNase H		cytotoxicity				
Cpd	R	X	ST ^b	3'-P ^c	% in. at 10 μM ^d	IC ₅₀ ^e (µM)	EC ₅₀ f	CC ₅₀ g	SI ^h
11a	-N_NH	Et	24 ± 1	>333	18	nd ⁱ	20.9	>50	>2.4
11b	—N_N-Me	Et	0.6 ± 0.1	56 ± 8	41	18.5 ± 0.7	33.1	>50	>1.5
11c	-N_N-Et	Et	2.4 ± 0.3	>111	63	26.2 ± 0.9	3.6	>50	>13.8
11d		Et	0.17 ± 0.04	17 ± 1	30	nd	>50	nd	nd
11f		Et	2.1 ± 0.2	>111	0.7	nd	16.2	>50	>3.0
11g	_N Me Me	Et	2.9 ± 0.4	41 ± 2	26	nd	>50	nd	nd
11i		Et	1.3 ± 0.3	102 ± 17	35	nd	13.2	>50	>3.7
12a	-N_NH	Н	2.0 ± 0.3	65 ± 2	44	28.7 ± 0.7	>50	nd	nd
12b	—NN-Me	Н	0.10 ± 0.01	8.5 ± 0.9	34	nd	>50	nd	nd
12c	-N_N-Et	Н	0.21 ± 0.06	23 ± 2	22	nd	>50	nd	nd
12d		н	0.08 ± 0.01	8.5 ± 0.7	74	3.3 ± 0.1	17	>50	>2.9
12e	-N_S	н	0.36 ± 0.07	30 ± 4	nd	nd	>50	nd	nd
12f		Н	0.08 ± 0.1	13 ± 1	72	6.8 ± 0.1	26.3	>50	>1.9
12g	Me —N Me	н	0.15 ± 0.02	5.1 ± 0.5	66	6.6 ± 0.3	>50	nd	nd
12i		Н	0.05 ± 0.01	4.6 ± 0.5	78	5.7 ± 0.1	>50	nd	nd
7a ^j		Et	2.3	110	56	nd	0.83	>50	>60
7b ⁷		Н	0.028	14.9	71	5.1 ± 0.2	14.1	>50	>3.5
1			0.028 ± 0.006 ^k	8.1 ± 4.2 ^{<i>k</i>}	nd	91 ± 8	0.0142 +/- 0.0052	>50	>3521
2			$0.087 \pm 0.008^{\prime}$	12.8 ± 6^{k}	nd	>100	0.0236 +/- 0.0046	>50	>2118

^{*a*}Inhibitory concentration 50% (μ M) determined against rIN from gel-based assays and expressed as mean ± SD from at least three independent experiments. ^{*b*}ST: strand transfer. ^{*c*}3'-P: 3'-processing. ^{*d*}Percentage of inhibition of RT-associated RNase H activity at fixed concentration of inhibitor 10 μ M (%). ^{*e*}Inhibitory concentration 50% against HIV-1 RT-associated RNase H activity determined from dose response curves (μ M). ^{*f*}Effective concentration 50% in HIV-1 infected HeLa cells (μ M). ^{*g*}Cytotoxic concentration 50% in HeLa cells (μ M). ^{*h*}Selectivity index (CC₅₀/EC₅₀). ^{*i*}nd: not determined. ^{*j*}Reference 19b. ^{*k*}Reference 22.

length IN from PFV in complex with raltegravir (PDB code 3OYA).⁴¹ This model was constructed using Prime v3.0 (Schrodinger)⁴³ based on the sequence alignment suggested by

Tang and co-workers.⁴⁴ This alignment identified conserved regions and gave the best overlap between the catalytic DDE motives of the PFV and HIV enzymes. Moreover, original

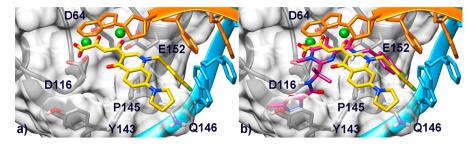


Figure 3. (a) Binding mode of compound 7b (yellow) within the HIV-1 IN/DNA model. The CCD is depicted as transparent light-gray surface and ribbons. Amino acid side chains involved in ligand binding are represented as sticks. The noncleaved (cyan) and processed (orange) viral DNA strands are shown as ribbon and sticks. Mg^{2+} metal ions are represented as green spheres. (b) Overlay of compound 7b and raltegravir (magenta) in the IN binding pocket.

nucleotides were mutated in order to obtain the HIV-1 retroviral DNA sequence. The model was then subjected to the Prime refinement protocol and finally used for docking studies. This new theoretical model was first used to perform docking simulations of raltegravir using Glide v. 5.7 (Schrödinger).45 Interestingly, this inhibitor adopts a binding mode similar to that shown in the parent crystal structure with its chelating oxygens oriented toward both active-site Mg2+ cations and its halobenzyl group fitting into the narrow pocket created by displacement of the terminal adenosine on the 3'-end of the viral DNA. These results prompted the docking calculations on our reference compound 7b, which is an efficient dual INSTI and RNase H inhibitor (Table 1). In the lowest energy conformation (GlideScore = -9.24), this compound occupies the DNA/IN interface with the DKA moiety chelating both the Mg⁺ ions (Figure 3a), similarly to the binding mode adopted by raltegravir (Figure 3b).

Interestingly, the predicted conformation of the DKA moiety strongly resembles that of a close analogue crystallized in its unbound conformation,⁴⁶ suggesting that for these set of compounds, the unbound conformation would also account for the bioactive one. Moreover, chelation by the carboxylate group should explain the higher inhibitory activity displayed by acidic compounds compared to the corresponding esters despite sharing the same predicted binding position in the HIV-1 IN binding site (compare docking pose of 7b in Figure 3a with that of 7a in Figure S1a in the Supporting Information).

In addition, the quinolinone ring is involved in a paralleldisplaced $\pi - \pi$ interaction with the terminal adenosine of the reactive DNA strand, further stabilizing the ligand in the IN active site. Moreover, the p-F-benzyl group is embedded in a hydrophobic cleft created by residues P145 and Q146, forming a well-oriented π -stacking interaction with the penultimate cytidine residue of the processed DNA strand. This interaction pattern (chelation of the Mg2+ cations and insertion in the DNA/enzyme cleft) is also established by raltegravir in the PFV IN catalytic site^{39,41} and in our HIV-1 IN model. Nevertheless, it is worth mentioning that raltegravir also places its terminal oxadiazole ring in a cleft, which cannot be explored by our current set of derivatives. In contrast, our INSTIs feature a basic substituent in position 7 of the quinolinone ring (-R)substituent, Table 1), which points outward the catalytic site, as further demonstrated by docking of compound 12i (Figure S1b in the Supporting Information), which is the most potent ST inhibitor among the newly synthesized derivatives.

To explain the strand transfer-selective inhibiting properties of our compounds, the predicted INSTI/HIV-1 IN CCD/viral DNA complex was superimposed with the crystal structure of the precatalytic target capture complex (TCC) from PFV IN (PDB code 3OS1, Figure 4).⁴⁰

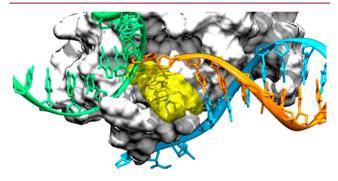


Figure 4. Superimposition between our model of the INSTI/HIV-1 IN CCD/viral DNA ternary complex and the precatalytic TCC from PFV (PDB code 3OS1). IN enzymes are represented as light-gray surfaces. The host DNA (green) and the nonprocessed (cyan) and processed (orange) viral DNA strands are shown as ribbon and sticks. Compound **7b** is depicted as yellow transparent surface and sticks.

This superimposition suggests that our ligands, similarly to raltegravir, would sterically hamper host DNA binding by complexing the Mg^{2+} cofactors and displacing the terminal 3'-adenosine of the viral DNA, thus accounting for the selective strand transfer integration step inhibition. On the other hand, the presence of the viral DNA in the inhibited complex might also explain why the 3'-processing step is not efficiently inhibited by the presented ligands.

To explain at molecular level the reasons for the RNase H inhibition, this enzyme was also considered in our theoretical studies. Several crystal structures of the full-length wild-type RT or of its isolated RNase H CCD from HIV-1 in complex with RNase H inhibitors have been reported.^{47–50} A superposition of all of these structures on $C\alpha$ atoms shows that folding, loops shape, and the side chain conformations of the majority of residues within the catalytic site are highly conserved (Figure S2 in the Supporting Information). On the basis of such an investigation, we selected the high resolution (2.09 Å) crystal structure of the full-length RT structure in complex with a RNase H pyrimidinol carboxylic acid inhibitor and nevirapine (PDB code 3QIP)⁴⁹ to carry out docking studies with Glide. The latter structure is made up by the RNase H domain in complex with an inhibitor, which is structurally related to our new series of DKAs. Because this structure is cocrystallized with Mn²⁺ ions, we replaced them with the most physiologically relevant Mg²⁺ cations, also used in the biochemical assays. Docking of compound 7b within the RNase H active site shows

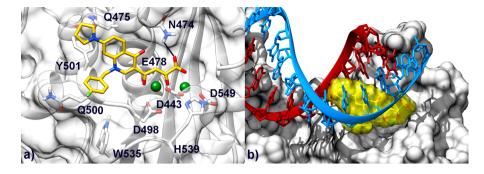


Figure 5. (a) Binding mode of compound 7b (yellow) in the HIV-1 RNase H active site. The active site is shown as transparent white surface and ribbons. Amino acid side chains important for ligand binding are represented as sticks. Mg^{2+} metal ions are depicted as green spheres. (b) Superimposition between our model of the INSTI/RNase H complex and the crystal structure of the full-length HIV-1 RT (p66 subunit) in complex with an RNA:DNA hybrid (PDB code 4B3O). RNase H residues are represented as white surfaces. The DNA (red) and RNA (light blue) strands are shown as ribbon and sticks. Compound 7b is depicted as yellow transparent surface and sticks.

that, as happened for IN, also in this case our reference compound places its DKA branch so as to chelate the two catalytic Mg²⁺ cations (Figure 5a). Remarkably, the DKA moiety again adopts an orientation similar to the unbound conformation of a closed analogue.⁴⁶ In addition to the chelation of the metal ions, the DKA branch H-bonds with the side chain of N474. Furthermore, compound 7b establishes favorable lipophilic interactions with the C β and C γ carbons of the Q475 through its quinolonyl nucleus. Additionally, the *p*-Fbenzyl group forms favorable contacts with the lipophilic residues Q500 and W535.

Finally, the pyrrolidine ring is able to take additional contacts with the Q475 and Y501 residues side chains. Docking of compound 7a (Figure S3a in the Supporting Information) shows that, as observed for IN, ester and acidic derivatives feature analogous predicted binding modes. Thus, also in the case of RNase H, the generally higher potency displayed by the acidic compounds might be ascribed to a more tight coordination of the metal ions by the DKA branch. Finally, also the newly synthesized derivative **12i**, which displays interesting RNase H inhibitory properties, is predicted to bind similarly to 7b (Figure S3b in the Supporting Information), thus suggesting that our INSTIs would all share the same binding position to the RNase H catalytic site.

To better explain the RNase H inhibitory activity of our compounds, we superimposed the predicted 7b/RNase H complex with the recently solved crystal structure of the full length HIV-1 RT in complex with a RNA:DNA hybrid⁵¹ (PDB code 4B3O) (Figure 5b). This superimposition suggests that binding of compound 7b would sterically hamper the hosting of the RNA:DNA hybrid at the RNase H catalytic site, thus preventing its hydrolysis.

Finally, to provide new hints for future lead optimization studies, we compared the binding pose of our reference compound 7b with that of two more potent inhibitors, namely a naphthyridinone ligand $(IC_{50} \approx 0.11 \,\mu M)^{48}$ and a pyrimidinol carboxylic acid $(IC_{50} = 0.23 \pm 0.01 \,\mu M)^{49}$ that were cocrystallized with the HIV-1 RT enzyme at the RNase H active site (PDB codes $3LP0^{48}$ and 3QIP,⁴⁹ respectively). From this comparison, it turned out that, similarly to 7b, these inhibitors chelate both the two Mg^{2+} cations, although adopting a slightly different coordination geometry (Figure S4 in the Supporting Information). On the other hand, due to size and structural differences, the considered compounds and our inhibitors establish different interaction patterns at the RNase H active site. In particular, the pyrimidinol carboxylic acid

derivative occupies through its 2-phenyl indole substituent a lipophilic pocket shaped by W535, P537, and A538 residues. This additional cleft might be thus explored by our nextgeneration DKAs to achieve an enhanced RNase H inhibitory profile to obtain dual IN/RNase H inhibitors.

CONCLUSIONS

We have designed, synthesized, and evaluated a series of novel quinolinonyl diketo acid derivatives endowed with a "base-like" moiety for their inhibition of HIV-1 IN. The compounds were also tested against the RNase H function of the RT. We identified nanomolar INSTIs that also exhibited low micromolar RNase H inhibition and antiviral properties. Our molecular modeling study also provided new insights for the future development of dual inhibitors of HIV-1 IN and RNase H.

EXPERIMENTAL SECTION

Chemistry. General. Melting points were determined with a Büchi 530 capillary apparatus and are uncorrected. Compounds purity were always >95% determined by high pressure liquid chromatography (HPLC). HPLC analysis were carried out with a Shimadzu LC-10AD VP CTO-10AC VP. Column used was generally Discovery Bio Wide Pore C18 (10 cm \times 4.6 mm, 3 μ m). Infrared (IR) spectra were recorded on a PerkinElmer Spectrum-One spectrophotometer. ¹H NMR spectra were recorded on a Bruker AC 400 spectrometer. Merck silica gel 60 F₂₅₄ plates were used for analytical TLC. Developed plates were visualized by UV light. Column chromatography was performed on silica gel (Merck; 70-230 mesh). Concentration of solution after reactions and extractions involved the use of a rotary evaporator operating at reduced pressure of approximately 20 Torr. Analytical results agreed to within $\pm 0.40\%$ of the theoretical values. Dimethyl sulfoxide-d₆ 99.9% (code 44,139-2) and deuterochloroform 98.8% (code 41,675-4) of isotopic purity (Aldrich) were used. Solvents were reagent grade and, when necessary, were purified and dried by standard methods. Organic solutions were dried over anhydrous sodium sulfate (Merck).

Microwave Irradiation Experiments. Microwave reactions were conducted using a CEM Discover system unit (CEM. Corp., Matthews, NC). The machine consists of a continuous focused microwave-power delivery system with operator selectable power output from 0 to 300 W. The temperature of the contents of the vessel was monitored using a calibrated infrared temperature control mounted under the reaction vessel. All experiments were performed using a stirring option whereby the contents of the vessel are stirred by means of a rotating magnetic plate located below the floor of the microwave cavity and a Teflon-coated magnetic stir bar in the vessel.

3-Acetyl-7-fluoroquinolin-4(1*H***)-one (8).** Triethyl orthoformate (4.0 g, 27 mmol), ethyl acetoacetate (3.5 g, 27 mmol), 3-fluoroaniline

(3.0 g, 27 mmol), and Dowtherm A (5.6 mL) were charged in a threenecked flask equipped with a condenser and a Dean-Stark apparatus. This mixture was stirred under argon atmosphere while the temperature was increased to 95 °C in 1 h then gradually to 162 °C in a further hour. Then the mixture was stirred at this temperature for 6 h. After this time, the resulting solution was added in portions during 3 h into 42 mL of Dowtherm A stirred in a three-necked flask equipped with thermometer and a Dean-Stark apparatus and heated at 253–254 °C. After addition, the mixture was heated at the same temperature for 2 h. Then the mixture was cooled at 90 °C, treated with 2-propanol (10 mL), cooled at 30 °C, filtered, and washed with 2propanol and light petroleum ether to give pure derivatives 8 (41% yield). Chemical and physical data of derivative 8 are reported in the Supporting Information. For spectroscopic data see ref 19b.

3-Acetyl-7-fluoro-1-[(4-fluorophenyl)methyl]quinolin-4(1*H***)one (9). A mixture of derivative 8 (0.23 g, 1.1 mmol), 4fluorophenylmethyl bromide (610 mg, 3.3 mmol), and anhydrous K_2CO_3 (210 mg, 1.5 mmol) in dry DMF (10 mL) was stirred at 100 °C for 1 h. After cooling, water was added (40 mL) and the precipitate that formed was filtered, washed with water and light petroleum ether in turn, and then dried under IR lamp to provide pure product 9 (29% yield). Chemical and physical data of derivatives 9 are reported in the Supporting Information. For spectroscopic data, see ref 19b.**

General Procedure for the Synthesis of 7-(Base)-4(1*H*)quinolinones (10a–f,h). A mixture of 9 (2.7 g, 8.5 mmol), proper amine (25.6 mmol), and TEA (0.8 g, 7.7 mmol) in dry DMF (40 mL) was irradiated with microwave at 153 °C for 10 min (applied potency 100 W) in an open vessel equipped with a condenser. After cooling, the reaction mixture was diluted with water (100 mL) and treated with 1 N HCl until pH 7. The solid that formed was collected by filtration. Purification of crude product was performed by column chromatography on silica gel (10:1 chloroform/ethanol as eluent) to give of pure 10a–f,h (14–78% yield). For spectroscopic, chemical, and physical data of derivatives 10a–f,h, see the Supporting Information.

3-Acetyl-7-(dimethylamino)-1-[(4-fluorophenyl)methyl]quinolin-4(1*H***)-one (10g). A mixture of dimethylamine in H_2O (40%, 1.3 mmol) and quinolinone 9 (0.45 mmol) was irradiated with microwave at 175 °C, 100 W, for 10 min. After cooling, the reaction mixture was diluted with water. The formed precipitate was collected by filtration thorugh a sintered glass septum and washed with water and diethyl ether. Purification of crude product was performed by column chromatography on silica gel (chloroform/ethanol 5:1 as eluent) to give pure product 10g (52% yield). For spectroscopic, chemical, and physical data of derivatives 10g, see the Supporting Information.**

3-Acetyl-7-[4-(3-chloropropyl)piperazin-1-yl]-1-[(4-fluorophenyl)methyl]quinolin-4(1*H***)-one (10i). A suspension of 10h** (500 mg, 1.14 mmol) in 3 mL of CCl_4 and 3 mL of CH_2Cl_2 was added triphenylphosphine (300 mg, 1.14 mmol). The mixture was heated at reflux for 24 h. The reaction was monitored with TLC (10:1 chloroform/methanol as eluent). After this time, the reaction was cooled at 25 °C, and the precipitate was filtered. The recovered organic solution was evaporated to afford 400 mg of crude product. Purification of crude product was performed by column chromatography on silica gel (10:1 chloroform/methanol as eluent) to give of pure **10i** (48% yield). For spectroscopic, chemical, and physical data of derivatives **10i**, see Supporting Information.

General Procedure for the Synthesis of Diketo Esters 11a– g,i. Freshly prepared sodium ethoxide (390 mg, 5.5 mmol) was added into a well stirred mixture of the appropriate acetyl derivative 10a-g,i (2.7 mmol) and diethyl oxalate (790 mg, 5.4 mmol) in anhydrous THF (2.7 mL) under nitrogen atmosphere. The mixture was stirred at room temperature for 2 h, and then was poured into *n*-hexane (50 mL). The collected precipitate was vigorously stirred for 30 min in 1 N HCl (50 mL). The yellow solid that formed was filtered, washed with water, and dried under IR lamp to afford the pure diketo esters 11a– g,i. Yield (%), melting point (°C), recrystallization solvent, IR, and ¹H NMR are reported for each of the following compounds.

4-[1-[(4-Fluorophenyl)methyl]-7-(piperāzin-1-yl)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11a). Yield 98%; 150–152 °C; washed with 2-propanol; IR ν 3347 (OH enol), 1711 (C=O ester), 1666 (C=O ketone) cm^{-1.} ¹H NMR (DMSO d_6) δ 1.31 (t, 3H, CH₃CH₂), 3.45–3.55 (m, 4H, piperazine H), 3.25– 3.35 (m, 4H, piperazine H), 4.31 (q, 4H, CH₂CH₃), 5.71 (s, 2H, CH₂ benzyl), 6.86 (s, 1H, quinolinone C8-H), 7.22–7.31 (m, 3H, benzene H and quinolinone C6-H), 7.43–7.52 (m, 2H, benzene H), 8.00 (s, 1H, CH enol), 8.10 (d, 1H, quinolinone C5-H), 9.00 (s, 1H, quinolinone C2-H). Anal. (C₂₆H₂₆FN₃O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl)/methyl]-7-(N-methylpiperazin-1-yl)-4-(1H)quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11b). Yield 99%; 96–99 °C; washed with 2-propanol; IR ν 3381 (OH enol), 1713 (C=O ester), 1605 (ketone) cm⁻¹. ¹H NMR (DMSO- d_6) δ 1.02 (s, 3H, CH₃ piperazine), 1.22 (t, 3H, CH₃CH₂), 2.25–2.53 (m, 4H, piperazine H), 3.23–3.31 (m, 4H, piperazine H), 4.10 (q, 4H, CH₂CH₃), 5.54 (s, 2H, CH₂ benzyl), 6.72 (s, 1H, quinolinone C8-H), 7.05 (d, 1H, quinolinone C6-H), 7.11–7.20 (m, 2H, benzene H), 7.23–7.31 (m, 2H, benzene H), 8.07 (d, 1H, quinolinone C5-H), 8.55 (s, 1H, CH enol), 8.61 (s, 1H, quinolinone C2-H). Anal. (C₂₇H₂₈FN₃O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl)methyl]-7-(N-ethylpiperazin-1-yl)-4-(1H)quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11c). Yield 99%; 200 °C (dec); washed with 2-propanol; IR ν 3388 (OH enol), 1711 (C=O ester), 1606 (ketone) cm⁻¹. ¹H NMR (DMSO-d₆) δ 1.17 (t, 3H, *J* = 5.5 Hz, *CH*₃CH₂ piperazine), 1.21 (t, 3H, *CH*₃CH₂), 2.19 (q, 3H, *J* = 5.5 Hz, *CH*₂CH₃ piperazine), 2.30–2.40 (m, 4H, piperazine H), 3.15–3.30 (m, 4H, piperazine H), 4.11 (q, 4H, *CH*₂CH₃), 5.55 (s, 2H, CH₂ benzyl), 6.72 (s, 1H, quinolinone C8-H), 7.01–7.07 (m, 2H, CH enol and quinolinone C6-H), 7.18–7.20 (m, 2H, benzene H), 7.22–7.30 (m, 2H, benzene H), 8.07 (d, 1H, quinolinone C5-H), 8.54 (s, 1H, quinolinone C2-H). Anal. (C₂₈H₃₀FN₃O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl)methyl]-7-(N-acetylpiperazin-1-yl)-4-(1H)quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11d). Yield 91%; 178–180 °C; washed with 2-propanol; IR ν 3409 (OH enol), 1716 (C=O ester), 1604 (ketone) cm⁻¹. ¹H NMR (DMSO-d₆) δ 1.24 (t, 3H, CH₃CH₂), 2.03 (s, 3H, CH₃ acetylpiperazine), 3.16–3.35 (m, 4H, piperazine H), 3.45–3.60 (m, 4H, piperazine H), 4.11 (q, 4H, CH₂CH₃), 5.56 (s, 2H, CH₂ benzyl), 6.75 (s, 1H, quinolinone C8-H), 7.01 (s, 1H, CH enol), 7.05 (d, 1H, quinolinone C6-H), 7.18–7.22 (m, 2H, benzene H), 7.28–7.32 (m, 2H, benzene H), 8.09 (d, 1H, quinolinone C5-H), 8.53 (s, 1H, quinolinone C2-H). Anal. (C₂₈H₂₈FN₃O₆) C, H, F, N.

4-[1-[(4-Fluorophenyl))methyl]-7-(thiomorpholin-1-yl)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11e). Yield 99%; 170 °C; washed with 2-propanol; IR ν 3388 (OH enol), 1711 (C=O ester), 1606 (ketone) cm⁻¹. ¹H NMR (DMSO- d_6) δ 1.21 (t, 3H, CH₃CH₂), 3.16–3.30 (m, 4H, thiomorpholine H), 3.65– 3.82 (m, 4H, thiomorpholine H), 4.11 (q, 4H, CH₂CH₃), 5.56 (s, 2H, CH₂ benzyl), 6.77 (s, 1H, quinolinone C8-H), 7.08–7.37 (m, 6H, CH enol, quinolinone C6-H and benzene H), 8.10 (d, 1H, quinolinone C5-H), 8.60 (s, 1H, quinolinone C2-H). Anal. (C₂₆H₂₅FN₂O₅S) C, H, F, N, S.

4-[1-[(4-Fluorophenyl))methyl]-7-(morpholin-4-yl)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11f). Yield 76%; 175–177 °C; trituration on diethyl ether; IR ν 3388 (OH enol), 1711 (C=O ester), 1606 (ketone) cm⁻¹. ¹H NMR (DMSO- d_6) δ 1.21 (t, 3H, CH₃CH₂), 3.14–3.32 (m, 4H, morpholine H), 3.62–3.80 (m, 4H, morpholine H), 4.11 (q, 4H, CH₂CH₃), 5.56 (s, 2H, CH₂ benzyl), 6.77 (s, 1H, quinolinone C8-H), 7.08–7.37 (m, 6H, CH enol, quinolinone C6-H and benzene H), 8.10 (d, 1H, quinolinone, C5-H), 8.60 (s, 1H, quinolinone C2-H). Anal. (C₂₆H₂₅FN₂O₆) C, H, F, N.

4-[1-[(4-Fluorophenyl))methyl]-7-(dimethylamino)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11g). Yield 98%; 150 °C (dec); washed with 2-propanol; IR. ν 3345 (OH enol), 1710 (C=O ester), 1636 (ketone) cm⁻¹. ¹H NMR (DMF- d_7) δ 1.30 (t, 3H, CH₃CH₂), 3.00 (s, 6H, CH₃ dimethylamine), 4.19 (q, 4H, CH₂CH₃), 5.65 (s, 2H, CH₂ benzyl), 6.52 (s, 1H, quinolinone C8-H), 6.88 (s, 1H, quinolinone C6-H), 7.18 (s, 1H, CH enol), 7.21–7.25 (m, 2H, benzene H), 7.45–7.48 (m, 2H, benzene H), 8.19 (d, 1H, quinolinone C5-H), 8.70 (s, 1H, quinolinone C2-H). Anal. (C₂₄H₂₃FN₂O₅) C, H, F, N.

4-[1-(4-[(4-Fluorophenyl)methyl]-7-[N-(3-chloroprop-1-yl)-piperazin-1-yl]-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid Ethyl Ester (11i). 77%; 175–177 °C; washed with acetone; IR ν 3488 (OH enol), 1723 (C=O ester), 1617 (ketone) cm⁻¹. ¹H NMR (DMF- d_7) δ 1.37 (t, 3H, CH₃CH₂), 1.98 (q, 2H, J = 5.5 Hz, ClCH₂CH₂CH₂N piperazine), 2.48–2.54 (m, 6H, piperazine H and ClCH₂CH₂CH₂N piperazine), 3.38–3.39 (m, 4H, piperazine H, 3.73–3.76 (q, 4H, CH₂CH₃), 4.35–4.39 (q, 2H, ClCH₂CH₂CH₂CH₂N piperazine), 5.87 (s, 2H, CH₂ benzyl), 6.98 (s, 1H, quinolinone C8-H), 7.22–7.29 (m, 3H, quinolinone C6-H and benzene H), 7.58–7.62 (m, 2H, benzene H), 8.10 (m, 2H, CH enol and quinolinone C5-H), 9.14 (s, 1H, quinolinone C2-H). Anal. (C₂₉H₃₁ClFN₃O₅) C, H, Cl, F, N.

General Procedure for the Synthesis of Diketo Acids 12a– g,i. A mixture of 1 N NaOH (6.5 mL) and the appropriate ester 11a– g,i (1.3 mmol) in 1:1 THF/methanol (12 mL) was stirred at room temperature for 40 min and then poured onto crushed ice. The aqueous layer was separated and treated with 1 N HCl until pH 3 was reached, and the yellow solid that formed was collected by filtration, then washed with water, hot dry ethanol, and light petroleum ether to afford pure acids 12a-g,i. Yield (%), melting point (°C), recrystallization solvent, IR, ¹H NMR are reported for each of the following compounds.

4-[1-[(4-Fluorophenyl))methyl]-7-(piperazin-1-yl)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12a). Yield 98%; 224 °C; washed with 2-propanol; IR ν 3500–2000 (OH), 1700 (C=O acid), 1649 (C=O keton) cm⁻¹. ¹H NMR (DMSO- d_6) δ 3.30–3.37 (m, 4H, piperazine H), 3.41–3.50 (m, 4H, piperazine H), 5.64 (s, 2H, CH₂ benzyl), 6.82 (s, 1H, quinolinone C8-H), 7.12–7.20 (m, 3H, benzene H and quinolinone C6-H), 7.37–7.41 (m, 2H, benzene H), 8.08 (d, 1H, quinolinone C5-H), 8.42 (s, 1H, CH enol), 8.75 (s, 1H, C2-H quinolinone), 15.00 (br s, 2H, OH). Anal. (C₂₄H₂₂FN₃O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl))methyl]-7-(N-methylpiperazin-1-yl)-4-(1H)quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12b). Yield 99%; 224 °C; washed with 2-propanol; IR ν 3500–2000 (OH), 1700 (C= O acid), 1599 (C=O ketone) cm⁻¹. ¹H NMR (DMSO- d_6) δ 2.38 (s, 3H, CH₃ piperazine), 2.61–2.75 (m, 4H, piperazine H), 3.21–3.35 (m, 4H, piperazine H), 5.67 (s, 2H, CH₂ benzyl), 6.81 (s, 1H, quinolinone C8-H), 7.29–7.08 (m, 3H, benzene H, quinolinone C6-H), 7.31–7.48 (m, 2H, benzene H); 7.69 (br s, 1H, CH enol), 8.07 (d, 1H, quinolinone C5-H), 8.93 (s, 1H, C2-H quinolinone), 15.00 (br s, 2H, OH). Anal. (C₂₅H₂₄FN₃O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl))methyl]-7-(N-ethylpiperazin-1-yl)-4-(1H)quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12c). Yield 99%; > 240 °C; washed with 2-propanol; IR ν 3500–2000 (OH), 1700 (C=O acid), 1578 (C=O ketone) cm^{-1.} ¹H NMR (DMSO-d₆) δ 1.06 (t, 3H, J = 5.5 Hz, CH₃CH₂ piperazine), 2.22 (q, 3H, J = 5.5 Hz, CH₂CH₃ piperazine), 2.28–2.50 (m, 4H, piperazine H), 2.70–3.85 (m, 4H, piperazine H), 5.65 (s, 2H, CH₂ benzyl), 6.78 (s, 1H, quinolinone C8-H), 7.12 (s, 1H, quinolinone C6-H), 7.10–7.20 (m, 2H, benzene H), 7.37 (m, 2H, benzene H), 7.69 (br s, 1H, CH enol), 8.06 (d, 1H, quinolinone C5-H), 8.90 (s, 1H, quinolinone C2-H), 15.00 (br s, 2H, OH). Anal. (C₂₆H₂₆FN₃O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl))methyl]-7-(N-acetylpiperazin-1-yl)-4-(1H)quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12d). Yield 90%; 195–197 °C; washed with acetonitrile; IR ν 3392–2000 (OH), 1716 (C=O acid), 1599 (C=O ketone) cm^{-1.} ¹H NMR (DMSO- d_6) δ 2.05 (s, 3H, CH₃ acetylpiperazine), 3.25–3.42 (m, 4H, piperazine H), 3.50–3.69 (m, 4H, piperazine H), 5.71 (s, 2H, CH₂ benzyl), 6.81 (s, 1H, quinolinone C8-H), 7.17–7.28 (m, 3H, benzene H and quinolinone C6-H), 7.30–7.50 (m, 2H, benzene H), 7.97 (br s, 1H, CH enol), 8.10 (d, 1H, quinolinone C5-H), 9.01 (s, 1H, quinolinone C2-H), 15.00 (br s, 2H, OH). Anal. (C₂₆H₂₄FN₃O₆) C, H, F, N.

4-[1-[(4-Fluorophenyl)methyl]-7-(thiomorpholin-1-yl)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12e). Yield 30%; 190 °C; washed with 2-propanol; IR ν 3500–2000 (OH), 1711 (C= O acid), 1606 (C=O ketone) cm⁻¹. ¹H NMR (DMSO-d₆) δ 2.43– 2.60 (m, 4H, thiomorpholine H), 3.56–3.75 (m, 4H, thiomorpholine H), 5.58 (s, 2H, CH₂ benzyl), 6.66 (s, 1H, quinolinone C8-H), 7.10– 7.16 (m, 4H, CH enol, quinolinone C6-H and benzene H), 7.32–7.35 (m, 2H, benzene H), 7.22 (d, 1H, quinolinone C5-H), 8.78 (s, 1H, quinolinone C2-H), 15.00 (br s, 2H, OH). Anal. $(C_{24}H_{21}FN_2O_5S)$ C, H, F, N, S.

4-[1-[(4-Fluorophenyl)methyl]-7-(morpholin-4-yl)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12f). Yield 99%; 187– 189 °C; washed with 2-propanol; IR ν 3500–2000 (OH), 1726 (C= O acid), 1600 (C=O ketone) cm^{-1.} ¹H NMR (DMSO-d₆) δ 3.17– 3.34 (m, 4H, morpholine H), 3.61–3.80 (m, 4H, morpholine H), 5.70 (s, 2H, CH₂ benzyl), 6.83 (s, 1H, quinolinone C8-H), 7.10–7.29 (m, 3H, quinolinone C6-H and benzene H), 7.32–7.48 (m, 2H, benzene H), 7.99 (br s, 1H, CH enol), 8.10 (d, 1H, quinolinone C5-H), 9.02 (s, 1H, quinolinone C2-H), 15.00 (br s, 2H, OH). Anal. (C₂₄H₂₁FN₂O₆) C, H, F, N.

4-[1-[(4-Fluorophenyl))methyl]-7-(dimethylamino)-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (**12g**). Yield 98%; 203–205 °C; washed with 2-propanol; IR ν 3500–2000 (OH), 1735 (C= O acid), 1620 (C=O ketone) cm⁻¹. ¹H NMR (DMF- d_7) δ 3.22 (s, 6H, CH₃ dimethylamine), 6.00 (s, 2H, CH₂ benzyl), 6.82 (s, 1H, quinolinone C8-H), 7.14 (s, 1H, quinolinone C6-H), 7.42–7.46 (m, 2H, benzene H), 7.70–7.85 (m, 2H, benzene H), 8.36 (d, 1H, quinolinone C5-H), 8.41 (s, 1H, CH enol), 9.27 (s, 1H, quinolinone C2-H), 15.00 (br s, 2H, OH). Anal. (C₂₂H₁₉FN₂O₅) C, H, F, N.

4-[1-[(4-Fluorophenyl)methyl]-7-[N-(3-chloroprop-1-yl)piperazin-1-yl]-4-(1H)-quinolinon-3-yl]-2-hydroxy-4-oxo-2-butenoic Acid (12i). Yield 100%; 175–177 °C; washed with THF; IR ν 3410– 2000 (OH), 1717 (C=O ester), 1655 (ketone) cm⁻¹. ¹H NMR (DMF- d_7) δ 2.18 (q, 2H, J = 6.0 Hz, ClCH₂CH₂CH₂N piperazine), 2.73–2.79 (m, 6H, piperazine H and ClCH₂CH₂CH₂N piperazine), 3.50–3.65 (m, 4H, piperazine H), 3.92 (q, 2H, J = 6.0 Hz, ClCH₂CH₂CH₂N piperazine), 6.03 (s, 2H, CH₂ benzyl), 7.16 (s, 1H, quinolinone C8-H), 7.38–7.45 (m, 3H, quinolinone C6-H and benzene H), 7.74–7.78 (m, 2H, benzene H), 8.37–8.39 (m, 2H, CH enol and quinolinone C5-H), 9.27 (s, 1H, quinolinone C2-H), 15.00 (br s, 2H, OH). Anal. (C₂₇H₂₇ClFN₃O₅) C, H, Cl, F, N.

Biological Methods. *HIV-1 IN Inhibition.* HIV-1 IN gel-based assays were carried out as previously published.⁵²

HIV-1 RT RNase H Inhibition. IC₅₀ values were determined as previously reported53 using an 18-nt 3'-fluorescein-labeled RNA annealed to a complementary 18-nt 5'-dabsyl-labeled DNA. To a 96-well plate was added 1 μ L of each inhibitor (in DMSO), followed by 10 μ L of the appropriate RT (15–80 ng/mL) in reaction buffer. Hydrolysis was initiated by adding 10 μ L of RNA/DNA hybrid (2.5 μ M). Final assay conditions were 50 mM Tris-HCl, pH 8.0, 60 mM KCl, 10 mM MgCl₂, 1% DMSO, 150-800 ng of RT, 250 nM substrate, and increasing concentrations of inhibitor. Wells containing only DMSO were used as negative control. Plates were incubated at 37 °C in a Spectramax Gemini EM fluorescence spectrometer for 10 min, and fluorescence (λ_{ex} = 485 nm; λ_{em} = 520 nm) was measured at 1 min intervals such that linear initial rates could be measured in the presence (v_i) and absence (v_0) of inhibitor. Percent inhibition was calculated as $100(v_0 - v_i)/v_{0}$, and plotted against log[I]. IC₅₀ values were determined using Prism5 (GraphPad Software). All assays were performed in triplicate.

HIV-1 RT Polymerase Inhibition. HIV-1 RT polymerase assays were carried out as previously reported.^{13b}

HIV-1 Replication Inhibition. Compounds antiviral activity was determined in a cell-based assay according to the procedure described previously⁵⁴ and modified as follows. HeLa-CD4-LTR- β -gal cells were maintained in DMEM with 10% serum and 0.5 mg/mL G418. The day prior experimentation, 96-well plates were prepared to contain 10000 cells per well in 100 µL of DMEM medium complemented with 10% serum. On day one, each drug was serial diluted directly on cells following a 3-fold dilution over 6 points and each well was then filled to 200 µL with either fresh medium or concentrated viral supernatant (HIV-1(IIIB), Advanced Biotechnologies Inc.). The highest compound concentration tested was 50 µM. On day two, cells were washed three times with PBS before adding 200 µL of a solution containing 50 mM Tris-HCl pH 7.5, 100 mM β-mercaptoethanol, 0.05% Triton X100, and 5 mM 4-methyl-umbelliferyl-β-D-galactopyranoside (4-MUG, Sigma). On day three, sealed plates were read in a

SpectraMax GEMINI-XS (Molecular Devices) with $\lambda_{ex/em} = 360/460$ nm.

Cellular Toxicity. Similarly to the antiviral assays, plates were prepared with 10,000 HeLa-CD4-LTR- β -gal cells per well and a serial dilution of compounds in 100 μ L. After 24 h of culture, 100 μ L of ATPlite reagent (PerkinElmer) was added to each well. After 5 min at room temperature, plates' luminescence was quantified using an EnVision multilabel reader (PerkinElmer) according to manufacturer's instructions.

Homology Modeling. A novel homology model of HIV-1 IN CCD/viral DNA complex was built using Schrodinger Prime (version 3.0)⁴³ software accessible through the Maestro interface.⁵⁵ To construct the model, the crystal structure of full-length IN from PFV in complex with raltegravir⁴¹ (PDB code 3OYA) was used as template. The sequence alignment used for the homology modeling was based on the secondary structure alignment of the template and the HIV-1 IN CCD (PDB code 1QS4)²⁶ on the $C\alpha$ positions of the active-site residues Asp64, Asp116, and Glu152 (DDE motif) to identify the structure conserved regions according to the scheme suggested by Tang and co-workers.⁴⁴ Prior to model building, bases of the 19-mer PFV DNA were mutated in order to obtain the HIV-1 retroviral DNA sequence (GenBank: AAC37875.1). The newly built model was the subjected to Prime refinement protocol in order to optimize the active site loop and amino acidic side chains conformation.

Ligands Setup. The 3D structures of all of the compounds were generated with the Maestro Build Panel⁵⁵ and were then submitted to Polak–Ribiere conjugate gradient minimization (0.0005 kJ/(Å mol) convergence) using MacroModel (version 9.9).⁵⁶ DKAs were modeled in their enolic tautomeric form because it has been clearly established that such compounds mainly exist in this form in solution and also given the influence of the two metal cations in the binding site.⁵⁷

Proteins Setup. The HIV-1 CCD/DNA model and the RNase H crystal structure (PDB code 3QIP) were prepared for docking calculations using the "Protein Preparation Wizard" panel of Schrödinger 2010 molecular modeling package.⁵⁵ In particular, using the "preprocess and analyze structure" tool, the bond orders and disulfide bonds were assigned, all the hydrogen atoms were added, and all the water molecules in a distance greater than 5 Å from any heterogroup were deleted; in addition, both Mn²⁺ ions in the RNase H crystal structure were replaced with Mg²⁺ cations. Using Epik 2.0, a prediction of the side chains heterogroups ionization and tautomeric states was performed.⁵⁸ An optimization of the hydrogen-bonding network was performed using the "H-bond assignment" tool. Finally, using the "impref utility", the positions of the hydrogen atoms were optimized by keeping all the heavy atoms in place.

Docking Calculations. Docking studies at both the HIV-1 IN and RNase H active sites were carried out with Glide v. 5.7 (Schrödinger).⁴⁵ Glide is a grid-based ligand docking with energetics approach and searches for favorable interactions between ligands and receptors. The shape and properties of the receptor are represented on a grid by different sets of fields that provide progressively more accurate scoring of the ligand pose. These fields are generated as preprocessing steps in the calculation and hence need to be computed only once for each receptor. For the grid generation of both the receptors, a box centered on the Mg2+ cations was created. This box gives a more precise measure of the effective size of the search space. However, ligands can move outside this box during grid minimization. The Cartesian coordinates of the outer box, X, Y, and Z length were set to 25 Å. The conformational space of the ligand is defined by Glide by several lowest-energy poses that are subjected to a Monte Carlo procedure that examines nearby torsional minima. This procedure is needed in some cases to properly orient peripheral groups and occasionally alters internal torsion angles. The default value (1.00) for the van der Waals radii scaling factor was chosen, which means no scaling for the nonpolar atoms was performed (no flexibility was simulated for the receptor). In the present study, the standard precision (SP) mode of GlideScore function was used to score the obtained binding poses. The force field used for the docking was the OPLS-2005.59

All of the pictures were rendered with the UCSF Chimera package from the Resource for Biocomputing, Visualization, and Informatics at the University of California, San Francisco. 60

ASSOCIATED CONTENT

S Supporting Information

Analyses of compounds 8, 9, 10a-i, 11a-g,i, 12a-g,i and molecular modeling. This material is available free of charge via the Internet at http://pubs.acs.org.

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The manuscript was written through contribution of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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ABBREVIATIONS USED

HAART, highly active antiretroviral therapy; IN, integrase; RT, reverse transcriptase; RNase H, ribonuclease H; INSTI, strand transfer IN inhibitor; DKA, diketo acid; SI, selectivity index; SAR, structure–activity relationship; CCD, catalytic core domain; PFV, prototype foamy virus; TCC, target capture complex; IR, infrared

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