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# A New Paradigm Shift for the Green Synthesis of Antibacterial Silver Nanoparticles Utilizing Plant Extracts

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This review covers general information regarding the green synthesis of antibacterial silver nanoparticles. Owing to their antibacterial properties, silver nanoparticles are widely used in many areas, especially biomedical applications. In green synthesis practices, the chemical reducing agents are eliminated, and biological entities are utilized to convert silver ions to silver nanoparticles. Among the various biological entities, natural plant extracts have emerged as green reducing agents, providing eco-friendly routes for the preparation of silver nanomaterials. The most obvious merits of green synthesis are the increased biocompatibility of the resulting silver nanoparticles and the ease with which the reaction can be carried out. This review summarizes some of the plant extracts that are used to produce antibacterial silver nanoparticles. Additionally, background information regarding the green synthesis and antibacterial activity of silver nanoparticles is provided. Finally, the toxicological aspects of silver nanoparticles are briefly mentioned.

Key words: Plant extracts, Green synthesis, Silver nanoparticles, Antibacterial activity

### INTRODUCTION

As an important aspect of nanotechnology, nanoparticles (NPs, less than 100 nm in one dimension) have been developed for a variety of applications, especially in the area of nanomedicine (1). NPs possess characteristic properties that differ from those of their bulk counterparts. In the area of nanomedicine, NPs are commonly employed as drug delivery vehicles. Recently, NPs have been employed in therapeutic applications to target specific sites, such as lung tissue, as well as in cancer therapy and vaccinations (1). Additionally, the use of NPs continues to increase in microbial applications due to the potential of NPs to circumvent microbial resistance while satisfying the current need for novel antibiotics (2-4). The number of infections and outbreaks associated with multi-drug-resistant (MDR) bacteria has increased, threatening public health. The NPs used to overcome microbial resistance include nitric oxide-releasing NPs, chitosan NPs and metallic NPs (3). Among the metallic NPs, the medicinal potentials of gold NPs (AuNPs)

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and silver NPs (AgNPs) have been extensively discussed elsewhere (5). AgNPs possess excellent antibacterial, antiviral and antifungal activities (6). Metallic NPs are commonly synthesized by traditional chemical and physical methods (7). However, current sustainability issues have led to explorations of eco-friendly synthesis and green synthesis for the production of metallic NPs based on diverse biological entities (7). This review is not meant to be inclusive of all plant extracts utilized for the green synthesis of AgNPs. There are extensive reviews that address this area (8-12). Instead, this review focuses on the preparation of antibacterial AgNPs by using plant extracts as reducing agents.

## **GREEN SYNTHESIS OF AgNPs**

Advantages and applications. The chemical synthesis of AgNPs employs chemical reducing agents to convert Ag ions to AgNPs. One of the most widely used chemical reducing agents is sodium borohydride. This process involves the undesired use of hazardous chemicals, and the biocompatibility of the resulting AgNPs is too low for application in biological systems. Relative to this traditional chemical method, methods of green synthesis that use environmentally friendly or eco-friendly compounds as reducing agents are emerging (13-19). The biological entities used for green synthesis include bacteria, fungi, plants, pure compounds from plants, algae, carbohydrates, microorganisms, etc. The

process of green synthesis requires the use of water as an environmentally friendly solvent; of note is the fact that water is more biocompatible than organic solvents.

Green synthesis possesses the following advantages over traditional chemical methods. (1) green synthesis is simple and usually involves a one-pot reaction; (2) it is amenable to scale up; (3) the toxicity-associated hazardous chemicals are eliminated, increasing the biocompatibility of the resulting product with normal tissues for in vivo applications; and (4) green biological entities can be used as reducing agents and capping agents, providing AgNPs with enhanced colloidal stability. Colloidal stability is an important factor when making claims regarding the biological activity of AgNPs. (5) Finally, the process is cost-effective. These advantages are not limited to AgNPs. Other metallic NPs can be greensynthesized and have many uses in biomedical and industrial applications. There are extensive reviews elsewhere regarding the applications of green-synthesized metallic NPs (20,21). The applications of metallic NPs include (1) biomedical applications, such as antimicrobial applications, drug delivery vehicles, medical imaging and diagnostics; (2) environmental remediation applications, such as the catalytic degradation of pollutants; and (3) industrial applications, such as energy-related applications and catalysis in organic synthesis (20). Borase et al. have extensively reviewed AgNPs in a variety of applications, including anticancer, catalysis, biosensor, drug delivery, textiles and cosmetics, antituberculosis, antiviral, insect management and water purification (12).

**Procedures and characterization.** We utilized various biological entities, including plant extracts (*Leonurus japonicus*, *Artemisia capillaris*, *Polygala tenuifolia*, and *Caesalpinia sappan*), a pure compound from plants (chlorogenic

acid), polysaccharides (chondroitin sulfate and acharan sulfate), an oligosaccharide (sialyllactose) and invertebrate extracts (an African giant snail Achatina fulica and an earthworm Eisenia andrei), for the green synthesis of AgNPs (22-29). The green synthesis of AgNPs is a simple and facile approach that is carried out by mixing silver nitrate (the Ag ion source) with a biological entity (the reducing agent). A schematic representation of the synthesis process is depicted in Fig. 1. In our experiment, an external source of energy, for example, oven incubation, is used to facilitate the reaction. The applications that we evaluated were mainly antibacterial (minimum inhibitory concentration, MIC) and in vivo wound-healing activities. Remarkably, AgNPs synthesized with the extract of Leonurus japonicus exhibited enhanced antibacterial activities (approximately 127-fold increase) against Pseudomonas aeruginosa, Escherichia coli and Enterobacter cloacae (22). The aerial portion of *Leonurus japonicus* was utilized. In the case of AgNPs synthesized with the root extract of Polygala tenuifolia, the strain of Escherichia coli DC2 was shown to be the most effective when treated with AgNPs (approximately 16-fold increase) (24). The aerial portion of Artemisia capillaris was also used as a reducing agent for the synthesis of AgNPs and AgNPs enhance the antibacterial activity against Pseudomonas aeruginosa, Escherichia coli, Enterobacter cloacae, Klebsiella oxytoca and Klebsiella areogenes (23). In each case, MIC values are compared with values recorded for the original plant extract alone. Generally, AgNPs synthesized with the three plant extracts mentioned above show more effective antibacterial activities against Gram-negative than Gram-positive bacteria.

The formation of AgNPs is generally characterized with spectroscopic and microscopic methods, including UV-visible spectrophotometry, Fourier-transform infrared spectroscopy

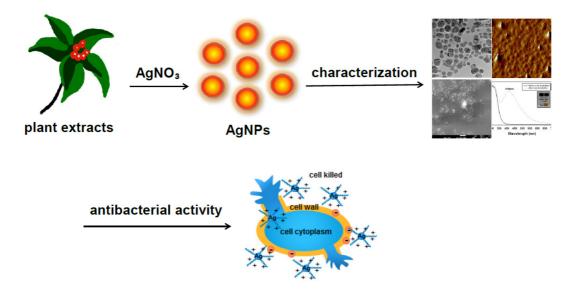


Fig. 1. A schematic of the procedure followed to synthesize AgNPs by using plant extracts as reducing agents.

(FT-IR), scanning electron microscopy (SEM), transmission electron microscopy (TEM), atomic force microscopy (AFM), X-ray diffraction (XRD) and zeta potential measurements. The formation of AgNPs is initially judged based on the color change of the solution, using the naked eye. The characteristic absorbance is observed in the range of 380~450 nm (yellow color), in which predominantly spherical AgNPs are synthesized. When the synthesis is completed, the size, shape and dispersion state of AgNPs are commonly visualized and measured by microscopic methods, such as TEM, AFM and SEM. Next, the crystalline nature of AgNPs is confirmed by XRD analysis. FT-IR spectra provide information regarding the functional groups of the reducing agents that are involved in the reduction of Ag<sup>+</sup> to AgNPs. Further, zeta potential measurements give information regarding the surface charges of AgNPs, which affect their colloidal stability. The reaction yield is measured by using both ultracentrifugation and inductively coupled plasma mass spectrometry (ICP-MS). After performing ultracentrifugation, AgNPs settle to the bottom of the tube. The colorless supernatant solution containing the unreacted Ag<sup>+</sup> is pooled and analyzed by ICP-MS. Then, the yield can be calculated by measuring the concentrations of both the total Ag in the solution of AgNPs and the unreacted Ag in the supernatant.

# PLANT EXTRACTS AS REDUCING AGENTS FOR THE GREEN SYNTHESIS OF AgNPs

Following various sustainability initiatives, plant extracts are emerging as potential reducing agents for the green syn-

thesis of AgNPs. Various categories of plant extracts further act as stabilizing agents. Approximately 321,212 species of plants are known on earth, according to the Botanic Gardens Conservation International, indicating that researchers have access to a significant variety of natural plant material (12). The author extensively reviewed the green synthesis of AuNPs and AgNPs using polysaccharides and phytochemicals as reducing agents (13). Phytochemicals and plant-derived polysaccharides, such as cellulose, starch, dextran, and alginic acid, play dual roles as reducing and stabilizing agents. The oxidation of the hydroxyl groups of polysaccharides to carbonyl groups is most likely involved in the reduction of metal ions to produce metallic NPs (13).

Researchers have reported that four factors—pH, temperature, reaction time and the ratio of plant extracts to silver nitrate—affect the synthesis of AgNPs based on plant extracts (8,12). (1) By changing the pH, the charges of the biomolecules in extracts are altered. Because the Ag ion is a cation, the charge of the biomolecules also affects the synthesis of AgNPs. (2) Generally, increasing the reaction temperature leads to a higher reaction rate. In high-temperature reactions, thermo-stable compounds can play a major role in obtaining higher yields. (3) Additionally, by increasing the reaction time, the reaction rate generally increases. (4) Finally, the ratio of plant extract to silver nitrate is a very important factor for obtaining AgNPs with various sizes and shapes.

The plant examples given in Table 1 are listed with their intrinsic biological activities and the plant parts used for synthesis. Leaves are the most commonly employed plant parts. Other parts used include bark, aerial part, fruit (whole or peeled), flower, root, tuber, rhizome, latex, moss, bulb,

Table 1. Plant extracts used for the green synthesis of antibacterial AgNPs

Antibacterial AgNPs					
Scientific name	Intrinsic biological activities of plants	Plant part used for the green synthesis of AgNPs	Refs.		
Acalypha indica	<ol> <li>analgesic and antiinflammatory activities</li> <li>antituberculosis activity</li> <li>antibacterial activity</li> </ol>	leaves	(36)		
Allium cepa	1) antimicrobial activity	bulbs	(37)		
Argemone maxicana	healing of gastric ulceration     larvicidal and chemosterilant activities	leaves	(38)		
Artemisia capillaris	<ol> <li>antitumor activity</li> <li>antiobesity activity</li> <li>antiinflammatory activity</li> </ol>	aerial parts	(23)		
Artocarpus heterophyllus	1) antiproliferative activity 2) antibacterial activity 3) antimicrobial activity 4) inhibitory effect on human blood coagulation 5) antioxidant activity 6) hypoglycemic and hypolipidemic effects 7) wound healing activity 8) antiinflammatory activity	leaves	(39)		

Table 1. Continued

	Antibacterial AgNPs		
	1) antiinflammatory and antinociceptive activities		
Azadirachta indica	2) antioxidant and antiproliferative activities	leaves	(40)
	3) cytotoxic and melanogenesis-inhibitory activity		
Boswellia ovalifoliolata	1) antioxidant activity	stem barks	(41)
Caesalpinia sappan	1) antiinflammatory activity	heartwoods	(25)
Сиезагріній зарран	2) vasorelaxant activity	neartwoods	(23)
	1) antioxidant activity		
	2) blood anticoagulation and antiplatelet activities		
Camellia sinensis	3) neuroprotective effect	leaves	(42)
	4) antimicrobial activity		()
	5) hypoglycaemic and hypolipidemic activities		
	6) wound healing activity		
<i>a</i> :	1) antiplasmodial activity	C :4	(42)
Carica papaya	2) antioxidant activity	fruits	(43)
	3) anxiolytic and sedative effects		
	1) larvicidal efficacy		
Catharanthus roseus	antidiabetic and antioxidant activities     hypoglygomia activity.	lanyas	(44)
Camaraninus roseus	hypoglycemic activity     hypolipidemic activity	leaves	(44)
	5) AchE inhibitory activity		
Chrysanthemum morifolium	1) antiinflammatory activity		
	2) anti HIV activity	flower	(45)
Chrysaninemum morijoiium	3) antitumor activity	nower	(43)
	1) improves insulin sensitivity in the brain		
	2) antihypertension effect		
Cinnamon zeylanicum	3) effects neurodegenerative diseases and its potent antineuroinflammatory	bark	(46)
	capacity		
	1) antimicrobial potentials		
	2) analgesic and antiinflammatory activities		
Citrullus colocynthis	3) antibacterial activity	leaves	(47)
,	4) antioxidant activity		( )
	5) antitumor activity		
Citrus sinensis	1) antioxidant and antifungal activities	peel	(48)
Cochlospermum gossypium	none	gum	(49)
	1) wound healing activity		
Coleus aromaticus	2) diuretic activity	leaves	(50)
	1) wound healing activity		
C 1	2) anti HIV activity	4	(51)
Coleus forskohlii	3) diuretic activity	root	(51)
	4) antifungal activity		
<i>a</i> 1	1) antioxidant, antiinflammatory and chemosensitizer activities	tuber	(52)
Curcuma longa	2) hepatoprotective effect	tuber	(52)
Dagma diam tuiflamm	1) antioxidant and antiproliferative activities	whole plant	(52)
Desmodium triflorum	2) analgesic and antiinflammatory activities	whole plant	(53)
Dioscorea batatas	1) antiinflammatory activity	rhizome	(54)
Dioscorea vaiaias	2) antioxidant activity	mizome	(34)
Dioscorea bulbifera	1) antibacterial activity		
	2) antitumor activity	tuber	(55)
	3) antidiabetic activity	tubei	(33)
	4) analgesic and antiinflammatory activities		
Eucalyptus citriodora	1) antiinflammatory activity		
	2) analgesic activity		
	3) antiproliferative effect	leaves	(56)
	4) antituberculosis		
	5) antibacterial activity		

Table 1. Continued

	Antibacterial AgNPs		
	1) antimicrobial activity		
	2) mosquito larvicidal and pupicidal activities		
	3) antidiabetic and antioxidant potentials		
E 1 1 1 1	4) immunomodulatory activity	,	(55.50)
Euphorbia hirta	5) antianaphylactic effect	leaves	(57,58)
	6) burn-wound healing activity		
	7) antidiarrhoeic activity		
	8) analgesic, antipyretic and antiinflammatory properties		
	1) wound healing activity		
Euphorbia nivulia	2) haemostatic activity	latex	(59)
1	3) antibacterial activity		( )
	1) antimicrobial activity		
	2) antiinflammatory activity		
<b>=</b>	3) antidiabetic activity		(60)
Ficus benghalensis	4) antiatherogenic activity	leaves	(60)
	5) antioxidant activity		
	6) antiinflammatory and analgesic activities		
Fissidens minutus	none	mosses	(61)
	1) neuroprotective effect		(*-)
	2) antiinflammatory effect		
	3) antioxidant and antitumor activities		
	4) antimicrobial and antiprotozoal activities		
Garcinia mangostana	5) antiangiogenic effect	leaves	(62)
	6) antihyperglycemic activity		
	7) analgesic activity		
	8) antiviral activity		
Gliricidia sepium	1) larvicidal, ovicidal and pupicidal activities	leaves	(63)
эннский ѕерит	1) nephroprotective effect	icaves	(03)
	2) antimicrobial effect		
	3) hypolipidemic activity		
Hibiaana aab Janiffa		looves stoms	(64)
Hibiscus sabdariffa	4) wound healing activity 5) antidiabetic activity	leaves, stems	(64)
	6) diuretic effect		
	7) antioxidant activity 1) antioxidant activity		
T	2) anticancer activity	:-1	(22)
Leonurus japonicus	3) cardioprotective effect	aerial parts	(22)
	4) antibacterial activity		
	5) antiinflammatory activity		
	1) ameliorates diabetes and decrease in high density lipoprotein		
Mangifera indica	2) inhibitory effects of mango leaves against <i>S. typhi</i>	peel	(65)
87	3) chemopreventive activity	1	( )
	4) antiphotoaging activity in UVB-irradiated hairless mice		
	1) prevents chemotherapy-induced nausea and vomiting		
	2) antifungal activity		
Mentha piperita	3) antioxidant and free radical scavenging activities	leaves	(66)
менни ррени	4) a potential source of natural antimicrobial products		()
	5) analgesic effect		
	6) antispasmodic activity on rat trachea		
Mimusops elengi	1) antioxidant activity	fruits	(67)

heartwood, gum, callus and seed. Borase *et al.* have listed the active components of plants that are responsible for reducing silver ions to AgNPs; according to these authors, the list of active components includes proteins, flavonoids, terpenoids, polyphenols, phenol hydroxyl and carboxylic groups of arabinose and galactose, phenolic glycosides, reducing sugars, tannins, aliphatic amines, aliphatic alkenes of alkaloids, polysaccharides, aromatic amines, *sec*-alcohols,

Table 1. Continued

	1) antimicrobial, anticancer, antiinflammatory, antidiabetic, and antioxidant		
	effects		
	2) antidiabetic and antioxidant activities		
	3) antiinflammatory activity		
Moringa oleifera	4) hypoglycemia and hypolipidemia activities	leaves	(68)
moringa oteljera	5) antioxidant capacity and antimicrobial activities	icaves	(00)
	6) antifungal activity		
	7) recovery from hepatic damage		
	8) antispasmodic, antiinflammatory and diuretic activities		
	9) induces apoptosis		
Musa paradisiaca	1) antioxidant, antimutagenic, anticarcinogenic, and cytoprotective activities	peel	(69)
	1) antifungal activity		
Nerium indicum	2) antioxidant activity	leaves	(58)
Nertum indicum	3) anticancer effects	icaves	(36)
	4) antidiabetic activity		
Nicotiana tobaccum	none	leaves	(70)
Ocimum tenuiflorum	1) antihyperglycemic activity	leaves	(71)
	1) activity of the coagulants		
	2) antibacterial activity		
	3) inhibits the ulcerogenic activity		
Opuntia ficus-indica	4) wound healing activity	leaves	(72)
	<ol> <li>reduces hangover symptoms and inhibites the production of inflammatory mediators</li> </ol>		
	6) hepatoprotective effect		
	1) antiinflammatory activity		
Polygala tenuifolia	2) antitumor activity	roots	(24)
20 2	3) anxiolytic activity		. ,
	1) potential antimicrobial agent	1 11	(72)
Sesuvium portulacastrum	2) cholinesterase inhibitory activity	leaves, callus	(73)
Shorea tumbuggaia	1) decreases cholesterol and triglycerides	stem barks	(41)
Svensonia hyderabadensis	none	leaves	(74)
•	1) antitumor and antiangiogenic activities		. ,
	2) large gains in strength and lean muscle mass		
	3) inhibits oxidative stress		
Tribulus terrestris	4) reduces Cd load	fruits	(75)
	5) conventional analgesic drugs		. ,
	6) antihypertensive effect both systolic and diastolic		
	7) antidepressive effect		
12.	1) treats various inflammatory disorders	1	(7.0
Vitex negundo	2) analgesic and antiinflammatory activities	leaves	(76)

water-soluble heterocyclic components and saponins (12). Typically, plant extracts possess intrinsic biological activities, which may further manifest in the biological activities of AgNPs as a result of combining the two materials. Therefore, plant extracts can potentially be developed into novel nanomaterials with diverse biological activities.

### **ANTIBACTERIAL ACTIVITIES OF AGNPs**

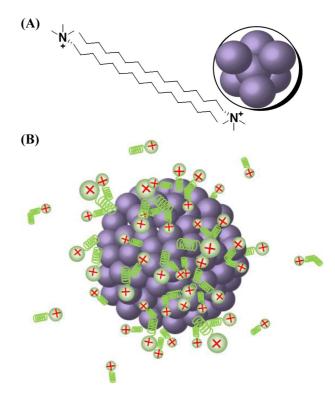
**MDR bacteria.** One of the most remarkable properties of AgNPs is their antibacterial activity. For 7,000 years, mankind has used Ag metal in coins, cutlery, textiles, cosmetics and medical implants. Ag has also been used as an

antibacterial agent in the form of ions, NPs and bulk metal (30). Currently, the increased use of AgNPs in medicine is closely associated with the antibacterial potential of AgNPs. The emergence of MDR bacteria requires novel antibiotics with improved antibacterial activities. In most respects, AgNPs have appeared as prospective alternatives for overcoming antibiotic resistance problems because AgNPs utilize multivalent or polyvalent mechanisms to exert their antibacterial activities. Additionally, the high surface-areato-volume ratio and the specific physical and chemical characteristics of AgNPs make them effective antibacterial agents against MDR bacteria, which includes methicillin-resistant *Staphylococcus aureus* (MRSA), methicillin-sensitive

Staphylococcus aureus (MSSA), methicillin-resistant Staphylococcus epidermis (MRSE), ampicillin-resistant Escherichia coli, vancomycin-resistant Staphylococcus aureus (VRSA), erythromycin-resistant Streptococcus pyogenes and multi-drug-resistant Pseudomonas aeruginosa (31).

**Mechanisms.** The antibacterial mechanisms of AgNPs are not yet fully understood; however, the research community has put forward several hypotheses, which follow (31). (1) AgNPs penetrate the cell wall of Gram-negative bacteria, leading to increased cell permeability followed by cell death; (2) the formation of free radicals by AgNPs is attributed to membrane damage; and (3) the strong binding capacity of Ag+ with thiol groups and phosphorous-containing bases, such as vital enzymes and DNA bases, causes the inhibition of bacterial growth and death. Key factors, such as size and shape, affect the antibacterial activity of AgNPs (31). With decreasing size, the surface-area-to-volume ratio increases. The large surface area provides many opportunities for interactions with bacteria; thus, the small particle size is effective as an antibacterial agent. Pal et al. reported that triangular AgNPs are more active than spherical and rod-shaped AgNPs against Escherichia coli, suggesting that the shape of AgNPs should be considered to develop highly efficient antibacterial agents (32).

Stabilizers. For antibacterial applications of AgNPs, the colloidal stability is a major consideration. Thus, stabilizers are commonly employed to increase the stability of AgNPs. The most widely used stabilizers are anionic, cationic, and nonionic surfactants and polymers. Carmona-Ribeiro and de Melo Carrasco have reported that cationic compounds, especially compounds with quaternary ammonium groups, are promising candidates for the development of antimicrobial agents (33). One cationic surfactant, cetyltrimethylammonium bromide (CTAB), has been employed as a stabilizing agent during the synthesis of AgNPs. The mechanism of CTAB adsorption onto the AgNPs is not clearly understood; however, the formation of a CTAB bilayer is a possibility (34). As shown in Fig. 2A, the cationic head group of CTAB binds to the surface of AgNPs, with the hydrophilic group of the surfactant attaching to the AgNPs. Then, the outward hydrophobic tail binds to the hydrophobic tail of another CTAB molecule to form a bilayer structure. Consequently, due to the bilayer formation, the hydrophilic head group of CTAB is directed outward, contributing to the overall positive charge of CTAB-stabilized AgNPs (Fig. 2B). We performed a green synthesis reaction for AgNPs by using extract of Caesalpinia sappan in both the absence and presence of stabilizers. The stabilizers used were a nonionic polymer (PVP) and cationic (CTAB), anionic (SDS, NaDDSS) and nonionic (Tween 20 and 80) surfactants (25). Among the tested stabilizers, CTAB-stabilized AgNPs were found to be the most potent antibacterial agents against 19



**Fig. 2.** The binding of CTAB on the AgNP surface. (A) The bilayer formation of CTAB, and (B) CTAB-stabilized AgNPs with overall positive charges (34).

strains of MRSA (25). The bilayer shell of CTAB provides improved colloidal stability to the AgNPs, which is closely connected to their enhanced antibacterial activity against MRSA. The overall positive charge of the CTAB-stabilized AgNPs offers many opportunities for the NPs to interact with the negatively charged bacterial cell wall.

### **PERSPECTIVES**

Currently, AgNPs-based consumer products are widespread in medical devices, clothing, industry, household goods, healthcare products and cosmetics (35). The potent antimicrobial activity of AgNPs is a key feature being utilized in the development of nanosilver products. Additionally, as a result of the increasing interest in the use of AgNPs in medical applications, the number of scientific publications on this subject has risen 41-fold since the last decade (2000~ 2011) (35). Whereas there were only four related publications in 2000, the number of publication increased to 164 in 2011 when a PubMed database search was performed using the keyword 'nanosilver' (35). In spite of the broad range of AgNP therapeutic applications, the limited information has been reported regarding the toxicity of AgNPs. One mechanism of AgNPs-induced toxicity is cellular apoptosis, which decreases cell viability (6). Safety issues concerning the environment should also be considered because AgNPs

from consumer products enter ecosystems. However, understanding the toxicity and fate of AgNPs in ecosytems requires further investigation (6). The green synthetic strategy of AgNPs based on plant extracts, as reported in the current review, contributes to the protection of the ecosystem and our health by decreasing the use of toxicity-associated hazardous chemicals. Again, due to the increasing number of applications of AgNPs, the associated toxicity and safety issues should be investigated in depth to further the safe use of AgNPs.

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