# Fragment Optimization of Reversible Binding to the Switch II Pocket on KRAS Leads to a Potent, In Vivo Active KRAS ${ }^{G 12 C}$ Inhibitor 

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#### Abstract

Activating mutations in KRAS are the most frequent oncogenic alterations in cancer. The oncogenic hotspot position 12, located at the lip of the switch II pocket, offers a covalent attachment point for KRAS ${ }^{\text {G12C }}$ inhibitors. To date, KRAS ${ }^{\text {G12C }}$ inhibitors have been discovered by first covalently binding to the cysteine at position 12 and then optimizing pocket binding. We report on the discovery of the in vivo active KRAS $^{\mathrm{GinC}}$ inhibitor BI0474 using a different approach, in which small molecules that bind reversibly to the switch II pocket were identified and then optimized for non-covalent binding using structure-based design. Finally, the Michael acceptor containing warhead was attached. Our approach offers not only an alternative approach to discovering KRAS $^{\mathrm{G} 12 \mathrm{C}}$ inhibitors but also provides a starting point for the discovery of inhibitors against other oncogenic KRAS mutants.


## - INTRODUCTION

The KRAS oncoprotein is a GTPase acting as a key node in intracellular signaling pathways that are involved in cell growth and survival. ${ }^{1,2}$ In normal cells, KRAS functions as a molecular switch, alternating between an inactive GDP-bound and an active GTP-bound state. ${ }^{3}$ The transition between these states is modulated by guanine-nucleotide exchange factors such as SOS and GTP hydrolysis, which is catalyzed by GTPaseactivating proteins (GAPs) to inactivate KRAS. GTP-bound KRAS allows binding of effector proteins to trigger downstream signaling pathways, including the RAF-MEK-ERK (MAPK) pathway. ${ }^{4-7}$ Activating mutations in KRAS are the most frequent oncogenic driver events in cancer. ${ }^{8}$ Oncogenic KRAS mutations have been shown to shift the equilibrium between the two KRAS states toward the active GTP-bound form. However, the major oncogenic versions of KRAS still undergo GAP-mediated GTP/GDP exchange and reactivation. ${ }^{9}$

Targeting KRAS has been of interest for the last 30 years. However, the surface of KRAS is thought to be devoid of druggable pockets except for the binding location for GTP and GDP, which have picomolar affinity for KRAS. Together, this has given rise to the assumption that KRAS is an undruggable target.


In 2012, progress was made when the workers at Genentech identified small molecules that bind to the switch I/II pocket of KRAS. ${ }^{11}$ Simultaneously, we conducted a fragment-based screen and also found small molecules that bind to this site, which were subsequently optimized using structure-based design. ${ }^{12-14}$

Another pocket on KRAS, the switch II pocket, was identified by Shokat and colleagues. ${ }^{15}$ The current clinical KRAS ${ }^{\text {G12C }}$ inhibitors ${ }^{16}$ bind to inactive GDP-bound KRAS in the switch II pocket and form a covalent bond with the mutant cysteine of KRAS ${ }^{G 12 C}$. In doing so, the oncoprotein is trapped in its inactive state, preventing its reactivation by nucleotide exchange. Sotorasib has become the first approved KRAS ${ }^{\text {G12C }}$ drug, receiving accelerated approval by the US Food and Drug Administration in May 2021 for adult patients with advanced KRAS ${ }^{\text {G12C }}$-mutant NSCLC who received at least one prior systemic therapy.

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Figure 1. Schematic overview of the second site fragment screening approach: a KRAS ${ }^{G 12 \mathrm{~V}}$ construct with a S39C mutation was covalently modified with BIT to block the switch I/II pocket on KRAS (purple circle) and to stabilize the S II helix conformation (PDB 4 PZZ ). ${ }^{10}$ A HSQC NMR screen was performed with the blocked KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}, \mathrm{~S} 39 \mathrm{C}}$ protein, and hits in the switch II pocket (green circle) were identified (exemplified with the fragment co-crystal structure with PDB code 7 U 8 H ).

The reported KRAS ${ }^{\mathrm{G} 12 \mathrm{C}}$ inhibitors were discovered from hits or weakly binding analogues that contained an electrophilic warhead. The electrophilic nature of the compounds was retained throughout the entire optimization process until the final candidates were identified. Here, we describe a covalent KRAS ${ }^{G 12 \mathrm{C}}$ inhibitor showing in vivo efficacy that was discovered using a different approach. In search of compounds that bound to a second site, we conducted a fragment-based screen in which a first-site ligand that bound to switch I/II was covalently linked to a KRAS mutant (S39C) to hold it in place. ${ }^{10}$ The hits that were found did not bind near to the firstsite ligand as anticipated but instead bound reversibly to the switch II site. Starting from these fragments, we grew toward Cys12 and optimized reversible binding affinity. Finally, the resulting most promising binder was decorated with an acrylamide warhead to form a covalent bond to KRAS ${ }^{G 12 C}$ and boost its potency at the last stage of optimization.

## ■ RESULTS AND DISCUSSION

In reported fragment screens on KRAS, non-covalent fragments preferentially bind to the switch I/II pocket. Upon binding to KRAS, a conformational change occurs, which opens both the switch I/II pocket and a nearby subpocket. To identify fragments which bind to this subpocket, we attempted a fragment-based screen in the presence of high concentrations of a first-site ligand. However, the compounds in the second site screen simply displaced the weakly bound first-site ligand. As previously reported, ${ }^{10}$ we solved this problem by covalently blocking the first site with a small-molecule ligand. To block the site in this way, we initially evaluated cysteine mutations of six residues adjacent to the switch I/II binding site. We then evaluated covalent modification of each of these KRAS mutants with various electrophiles. Using both NMR and Xray crystallography, we verified that reaction of a KRAS ${ }^{G 12 V, S 39 C}$ mutant with ( 1 H -benzo[d]imidazol-2-yl)methanethiol (BIT) produced a properly folded, covalently modified KRAS ${ }^{G 12 \mathrm{~V}, S 39 \mathrm{C}-\mathrm{BIT}}$ protein with the switch I/II binding site occupied (Figure 1). ${ }^{10}$ This modified protein was then used to conduct a HSQC-based screen of $\sim 13,000$ fragments, revealing 20 fragment hits that displayed chemical shift perturbation patterns different than those observed with compounds that bind to switch I/II. ${ }^{10}$
Among these compounds was a cluster that included both aminocyanothiophenes and aminothiazoles (Table 1). NMR

Table 1. Aminocyanothiophene and Related Fragment Hits Measured on KRAS ${ }^{\text {G12V, }} \mathbf{3} 39 \mathrm{C}-\mathrm{BIT}$

| Compound | Structure | $\begin{gathered} \text { KRAS }^{\text {G12V,S39C-BIT }} \\ \text { HSQC K }_{\mathrm{D}}[\mu \mathrm{M}]^{a} \end{gathered}$ |
| :---: | :---: | :---: |
| 1 |  | $116 \pm 19$ |
| 2 |  | $193 \pm 29$ |
| 3 |  | $715 \pm 48$ |
| 4 |  | >10,000 |

${ }^{a} K_{\mathrm{D}}$ values are shown as mean $\pm \mathrm{SD}$ and were determined, as described in the Experimental Section.
titration of the hits using the modified KRAS construct revealed that 1 bound to the modified protein with an affinity of $116 \mu \mathrm{M}$, around two-fold more tightly than the corresponding five-membered ring analogue 2. Aminothiazole 3 also showed reduced affinity compared to $\mathbf{1}$, and its fivemembered ring analogue 4 displayed no shifts at all. Based upon these initial results from the screen, we chose to followup on fragment 1 with the goal of improving affinity and elucidating the binding mode.

Using a combination of targeted commercial purchasing and discrete compound synthesis, we first sought to identify the key features of the hit that govern the binding to KRAS (Table $2)$. The amine and the cyano groups were shown to be important, based on the loss of affinity observed for both 5 and 6. Furthermore, elimination of the alkyl ring resulted in loss of binding (7), and aromatization of the ring system (8) also reduced affinity compared with the initial fragment hit. Interestingly, 9 a ring-opened alkyl variant of the aminocyanothiophene, bound with an affinity nearly equal to the hit, and a seven-membered ring analogue, 10, bound with affinity comparable to the five-membered ring compound 2.

Table 2. Compounds from Fragment Optimization Measured on KRAS ${ }^{\text {G12V,S39C-BIT }}$
Compound
${ }^{a} K_{\mathrm{D}}$ values are shown as mean $\pm \mathrm{SD}$ and were determined, as described in the Experimental Section.

The crystal structure of $\mathbf{1}$ in complex with GDP. KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}, \mathrm{C} 1188,539 \mathrm{C}-\mathrm{BIT}}$ (PDB code 7U8H) revealed the binding mode of the fragment (Figure 2). The compound was found to indeed bind to the switch II pocket similar to that occupied by previously reported covalent inhibitors of KRAS ${ }^{G 12 \mathrm{C}} .{ }^{17,18}$ The cyclohexyl ring of the fragment hit occupies the lipophilic base defined by Val9, Met72, Phe78, Val103, and Ile100. The amine engages in hydrogen bond interactions with the side chain of Asp69 and the backbone carbonyl of Glu63, with distances of 2.7 and $2.8 \AA$, respectively. Furthermore, the cyano moiety interacts with both the amide NH of Glu63 (distance $2.9 \AA$ ) as well as nearby bound water. This network of interactions readily explains the rigid requirements for both of these functional groups for best affinity.
Analysis of the crystal structure revealed that the 4-position of the cyclohexyl ring was oriented directly toward the 12position residue of the protein $(\mathrm{G} 12 \mathrm{~V})$, which was approximately $11 \AA$ away. Furthermore, a mostly open channel lay between the hit and position 12 with only one major saltbridge between Glu62 and His95 keeping the switch II pocket in a stabilized conformation.
To confirm our ability to access the channel, we prepared and evaluated derivatives of 1 with methyl groups in the 4 -


Figure 2. Crystal structure of $\mathbf{1}$ in complex with GDP. KRAS ${ }^{\text {G12V,C118s, }, 399 \mathrm{C}-\mathrm{BIT}}$ (PDB code 7U8H, $1.7 \AA$ resolution; monomer B in yellow, color-coded by atom type, hydrogen bonds are shown as blue dotted lines). The gray protein surface illustrates the channel toward amino acid position 12 and is highlighted with a blue arrow indicating the distance between both ends.
position of the six-membered ring. The single 4-methyl of compound 11 conferred a five-fold increase in affinity, and the gem-dimethyl of analogue $\mathbf{1 2}$ produced a further gain. However, as NMR-derived $\mathrm{K}_{\mathrm{D}}$ determination below $10 \mu \mathrm{M}$ becomes inaccurate ${ }^{19}$ due to the high protein concentration, an exact value for this analogue could not be provided using KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}, \mathrm{~S} 39 \mathrm{C}-\mathrm{BIT}}$. We also evaluated this analogue using ${ }^{15} \mathrm{~N}$ labeled KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}}$ without the mutated Ser 39 and blocking benzimidazole (Table3). We obtained a slightly weaker affinity of $15 \mu \mathrm{M}$, confirming the ability of these fragments to bind to an unmodified oncogenic KRAS mutant.

The crystal structure of $\mathbf{1 2}$ in complex with KRAS ${ }^{\text {G12C }}$ (PDB code 8AFC) revealed that the binding mode of the compounds is preserved when the switch I/II site is not occupied. Furthermore, the addition of the gem-dimethyl did not affect the binding mode of the core scaffold and maintained all important polar interactions (Figures 2 and 3).

Together, these data suggested that it would be feasible to design a covalent KRAS ${ }^{\mathrm{Gi} 12 \mathrm{C}}$ inhibitor from the non-covalent binders discovered in this study. In contrast, a previously described fragment (GNE-2897, ${ }^{20}$ 13) and also the switch II pocket binding core groups of two reported KRAS ${ }^{\text {G12C }}$ inhibitors $\left(14,{ }^{17} \quad 15^{18}\right)$ did not display tight binding to KRAS without their covalent warheads. We hypothesized that the use of a KRAS binding core with native affinity to the protein should enable the discovery of a covalent inhibitor with greater selectivity against other cysteines and therefore lower toxicity as well as a higher second-order rate constant and improved KRAS ${ }^{\text {G12C }}$ inhibition, resulting in a stronger antiproliferative effect.

After the identification of a suitable exit vector, we turned our attention to growing the fragment toward Cys12. We hypothesized that growing the molecule into the channel leading to Cys 12 should lead to a significant improvement in KRAS binding affinity. To deal with the aforementioned limitations of the $\mathrm{K}_{\mathrm{D}}$ determination below $10 \mu \mathrm{M}$, $\mathrm{KRAS}^{\mathrm{G} 12 \mathrm{~V}}$ was used instead of KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}, \mathrm{~S} 39 \mathrm{C}-\mathrm{BIT}}$ for the NMR measurements during fragment growing toward amino acid position 12. As a first step, amides 16 and $17 a / b$ (Table 3) were prepared, but a significant drop in affinity was observed. To rigidify the structure, the corresponding oxadiazoles, accessible from a common acid intermediate 18 , which itself was in a similar $K_{\mathrm{D}}$

Table 3. Compounds from Fragment Growing Measured on KRAS ${ }^{\text {G12V }}$
Compound
${ }^{a} K_{\mathrm{D}}$ values are shown as mean $\pm \mathrm{SD}$ and were determined, as described in the Experimental Section.
range as the corresponding amides, were synthesized. The minimally decorated oxadiazole 19 still showed a slightly weaker affinity compared to the dimethyl analogue, but potency could be enhanced by further growing. Decorated aniline analogues 20a and 20b, synthesized from the (S)isomer 18a of acid 18, showed a significantly improved $K_{D}$, exceeding the binding affinity of the dimethyl analogue $\mathbf{1 2}$ and again reaching the limitation of accurate $K_{\mathrm{D}}$ determination (Table 3). To get an estimate on the binding selectivity for the inactive versus active state of KRAS, the affinity of 20a and 20b to KRAS ${ }^{\text {G12D }}$ loaded with a non-hydrolyzable GTP analogue was determined as well. Although an exact $K_{\mathrm{D}}$ determination was in this case not possible ( $K_{\mathrm{D}}>2 \mathrm{mM}$, as shown in Table S6), a clear preference for the inactive GDP-bound state of KRAS was observed.
The comparison of the KRAS co-crystal structures of $\mathbf{1 2}$ and 20a provided an explanation for the initial potency drop that needed to be compensated by further growing: to expand the pocket toward Cys12, the salt bridge between Glu62 and His95 observed in the structure of $\mathbf{1 2}$ bound to KRAS is


Figure 3. Comparison of the binding modes observed in the respective crystal structures. Crystal structure of 12 in complex with GDP•KRAS ${ }^{\text {G12C }}$ (PDB Code 8AFC, $2.4 \AA$ resolution; blue, colorcoded by atom type) and 20a in complex with GDP•KRAS ${ }^{\text {G12V,S39C-BIT }}$ (PDB code 8AFD, resolution $1.63 \AA$; monomer B in green, colorcoded by atom type). The lipophilic base with Val9, Met72, Phe78, Ile100, and Val103 is shown as gray transparent surface (water molecules are shown as red spheres, hydrogen bonds as blue dotted lines).
broken up by the aniline of 20a causing a reorientation of the Glu62 side chain. By installation of the oxadiazole moiety, affinity can be regained via an H-bond with His95 (Figure 3, PDB codes 8AFC and 8AFD).

After synthesizing the first single-digit micromolar reversible binders and reaching the HSQC assay limit at $10 \mu \mathrm{M}$, we turned our attention toward adding acrylamide warheads to analogues 20a/b. A mass spectrometry (MS)-based assay to detect covalent modification of $\mathrm{KRAS}^{\mathrm{G12C}}$ and determine a rate constant readout by measuring the time course of Cys 12 modification was used as an initial driving assay for optimization. ${ }^{21}$ To extend the dynamic range of our driving assay for more potent analogues, an AlphaScreen assay measuring disruption of the GDP-KRAS ${ }^{\text {G12C }}::$ SOS1 interac$\operatorname{tion}^{22}$ was applied. Anilides 21a/b carrying the acrylamide warhead directly attached to the fragment showed measurable second-order rates in the MS-based assay but were not detectable in the AlphaScreen format. However, based on the co-crystal structure of 20a, we hypothesized that the introduction of a linker between the central phenyl core and the acrylamide should lead to a preferable positioning of the electrophile. Switching the exit vector from 21a to 21b and introducing a piperazine linker resulted in acrylamide $\mathbf{2 2}$ which did not lead to an improved second-order rate, but nonetheless showed an $\mathrm{IC}_{50}$ of $\sim 3 \mu \mathrm{M}$ in biochemical assay.

Because of this AlphaScreen data and the lower intrinsic reactivity of piperazine acrylamides compared to aryl acrylamides, ${ }^{25,26} 22$ was considered the more attractive analogue for further optimization, despite its lower MS-based activity compared to $\mathbf{2 1 b}$.
Based on analysis of the co-crystal structures, we hypothesized that both His95 and Glu62 $2^{27}$ could be addressed by appropriately occupying the subpocket formed by His 95 , Glu62, and Asp92. Changing the central phenyl ring to a pyridine improved the interaction with His95 and facilitated the introduction of a basic residue addressing the acidic environment. A (3S)-1,3-dimethylpiperazine substituent was

Table 4. Second-Order Rate Constants, Biochemical Activity, and Antiproliferative Activity of Acrylamide Analogues

${ }^{a} k_{\text {inact }} / K_{\mathrm{J}}, k_{\text {inact }}, K_{\mathrm{J}}, \mathrm{IC}_{50}$, and $\mathrm{EC}_{50}$ values are shown as geometric mean $\pm \mathrm{SD}(n \geq 3)$ and were determined, as described in the Experimental Section. ${ }^{b} 4895$ and $9900 \mathrm{M}^{-1} \mathrm{~s}^{-1}$ previously reported. ${ }^{18,23}{ }^{c} 6 \mathrm{nM}$ previously reported. ${ }^{24}$
found to be a suitable choice for this purpose. Together, this led to identification of BI-0474 which demonstrated around 500 -fold improvement in both the second-order rate constant and the GDP-KRAS::SOS1 protein-protein interaction assay (Table 4) and showed a tight interaction network in the pocket (Figure 4, PDB code 8AFB). Intriguingly, the improvement in potency for BI-0474 compared to 22 seemed to be attributable mainly to an increase in the $\mathrm{k}_{\text {inact }}$ despite the fact that no structural variation was made directly to the covalent warhead or the piperazine linker. Furthermore, BI-0474 showed potent


Figure 4. Structure of BI-0474 (23) in complex with GDP•KRAS ${ }^{\text {G12C }}$ (PDB Code $8 \mathrm{AFB}, 1.12 \AA$ resolution; cyan, color-coded by atom type), BI-0474 is covalently bound to Cys 12 of KRAS (water molecules are shown as red spheres, hydrogen bonds as blue dotted lines).
antiproliferative activity of 26 nM on NCI-H358 cells carrying a G12C mutation (Table 4 and Figure S9). However, an antiproliferative effect on cells carrying a KRAS ${ }^{\text {G12D }}$ mutation (GP2D or LS513 cells) was only observed above $4 \mu \mathrm{M}$ (data shown in Table S8). In comparison with AMG-510, BI-0474 already showed a slightly higher second-order rate and lower $K_{\mathrm{I}}$, comparable $k_{\text {inact }}$ and $\mathrm{IC}_{50}$ in the biochemical assay, and an only slightly weaker antiproliferative effect on NCI-H358 cells.

Intraperitoneal (ip) administration of BI-0474 at $40 \mathrm{mg} / \mathrm{kg}$ in NMRI nude mice resulted in plasma exposure expected to lead to $\mathrm{TO}>60 \%$ after multiple dosing based on a mouse PK model and on a model prediction in the KRAS ${ }^{G 12 \mathrm{C}}$ mutant NCI-H358 xenograft model (Figure 5a). ${ }^{21}$ Encouraged by these results, we assessed TO as well as pharmacodynamic (PD) biomarker modulation in the KRAS ${ }^{\mathrm{G} 12 \mathrm{C}}$ mutant NCIH358 model after a daily treatment with $40 \mathrm{mg} / \mathrm{kg}$ ip on 3 consecutive days. As shown in Figure 5b, MS-based TO analysis showed a strong and treatment-induced reduction of unmodified KRAS ${ }^{\mathrm{G} 12 \mathrm{C}}$ protein on the third day of consecutive treatment (five-fold and nine-fold reduction compared to respective vehicle controls at 2 and 6 h post last dose). This was confirmed by the analysis of RAS-GTP levels in the tumor samples (G-LISA, Figure 5c). Importantly, and in agreement with the TO and RAS-GTP data, p-ERK levels were also strongly reduced (five-fold and six-fold at 2 and 6 h post last dose, Figure 5d). Finally, we also measured cleaved caspase 3 levels from the tumor lysates (Figure 5e) and, while 2 h postdose induction of apoptosis was only detectable in one out of five treated samples, apoptosis induction was detectable in all five treated samples 6 h post-dose on day 3, indicating that KRAS ${ }^{\text {G12C }}$ inhibition leads to induction of programmed cell death in this xenograft model. Given these data, anti-tumor


Figure 5. BI-0474 shows efficacy and PD biomarker modulation in an NCI-H358 cell line-derived non-small cell lung cancer xenograft model. (a) Top graph: PK model of BI-0474 $40 \mathrm{mg} / \mathrm{kg}$ ip qdx3 including measured free plasma concentration of BI-0474 2, 6, and 24 h post last treatment ( $n$ $=5$, mean value $\pm$ SD); bottom graph: target occupancy (TO) model of BI- $047440 \mathrm{mg} / \mathrm{kg}$ ip qdx3 based on PK and MS-based second-order rate. (b-e) Treatment with BI-0474 leads to TO, reduction of RAS-GTP and pERK levels, as well as apoptosis induction in vivo. Five individual tumors per treatment group were analyzed, the individual values are plotted, as is the mean $\pm$ SD. (b) BI- 0474 treatment increases TO as measured by reduction of unmodified KRASG12C protein measured by MS, (c) reduces RAS-GTP levels as measured by RAS G-LISA, (d) reduces pERK levels as measured by a pERK1/2 immunoassay (MSD), and (e) induces cleaved caspase-3 levels measured by a caspase-3 immunoassay (MSD). (f) BI0474 shows efficacy in an NCI-H358 non-small cell lung cancer xenograft model. Left graph: mean tumor volume $\pm$ SD for vehicle ( $n=8$, black circles) compared to $40 \mathrm{mg} / \mathrm{kg}$ BI-0474 administered ip once a week ( q 7 d ) ( $n=8$, red squares) and $40 \mathrm{mg} / \mathrm{kg}$ BI- 474 administered ip twice a week on 2 consecutive days ( $q$ dx2 $)(n=8$, blue triangles). Tumor volumes for each individual mouse measured on day 19 are included in the Supporting Information (Figure S10). Right graph: mean body weight changes.
efficacy was also assessed in this KRAS ${ }^{\text {G12C }}$ mutant NCI-H358 model (Figure 5f). At this time in our KRAS-G12C program, bioavailability was not yet optimized; hence, we wanted to explore whether infrequent systemic administration of covalent KRAS ${ }^{\text {G12C }}$ inhibitors could be efficacious. BI-0474 was administered ip weekly or twice weekly on 2 consecutive days at $40 \mathrm{mg} / \mathrm{kg}$ and showed anti-tumor efficacy in both dose groups with $68 \%$ tumor growth inhibition (TGI) on day 19 post treatment start for the dose group receiving a weekly dose of $40 \mathrm{mg} / \mathrm{kg}$ ip and $98 \%$ TGI for the dose group receiving a weekly dose of $80 \mathrm{mg} / \mathrm{kg}$ ip $(=40 \mathrm{mg} / \mathrm{kg}$ ip on 2 consecutive days). These data confirm dose-dependent efficacy for BI0474 in the NCI-H358 model. Compared to animals receiving the vehicle only, both dose groups showed some body weight loss, which is less pronounced for the low dose group compared to the high dose group.

Chemistry. Compounds in Tables 1 and 2 were commercial or accessible via single-step procedures from commercial starting materials (Scheme 1). 12 was synthesized via the Gewald reaction using dimethyl cyclohexanone. 14 was obtained by dimethylation of the corresponding naphthylamine. Suzuki coupling of 2-chloro-3-fluoropyridine and the suitable phenylboronic acid yielded 15.

Scheme 1. Synthesis of Fragment Analogues 12, 14, and $15^{a}$



${ }^{a}$ Reagents and conditions: (a) $\mathrm{S}_{8}$, malono-nitrile, L-proline, DMF, BMIMCl, $12 \mathrm{~h}, 70^{\circ} \mathrm{C}, 61 \%$; (b) $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{CH}_{3} \mathrm{I}, \mathrm{DMF}, 18 \mathrm{~h}, 50^{\circ} \mathrm{C}$, $35 \%$; and (c) XPhos Pd G3, $\mathrm{Cs}_{2} \mathrm{CO}_{3}$, dioxane, $1 \mathrm{~h}, 80^{\circ} \mathrm{C}, 48 \%$.

As described in Scheme 2, amides 16 and $17 \mathbf{a} / \mathbf{b}$ as well as oxadiazole 19 could be obtained via the corresponding racemic

Scheme 2. Synthesis of Amides and Oxadiazole $19^{a}$

${ }^{a}$ Reagents and conditions: (a) $\mathrm{S}_{8}$, morpholine, malononitrile, EtOH, 1 $\mathrm{h}, 55^{\circ} \mathrm{C}, 55 \%$; (b) LiOH , dioxane/ $\mathrm{MeCN} /$ water, $48 \mathrm{~h} 50^{\circ} \mathrm{C}, 86 \%$; (c) HATU, DIPEA, $\mathrm{PhNH}_{2}$ or $\mathrm{CH}_{3} \mathrm{NH}_{3} \mathrm{Cl}$, DMF, $30 \mathrm{~min} ., 0^{\circ} \mathrm{C}-\mathrm{RT}$, $32 \%$; (d) $\mathrm{NaH}, \mathrm{CH}_{3} \mathrm{I}, \mathrm{DMF}, 12 \mathrm{~h}, 0{ }^{\circ} \mathrm{C}-\mathrm{RT}, 86 \%$; (e) $\mathrm{S}_{8}$, L-proline, malononitrile, $\mathrm{EtOH}, 48 \mathrm{~h}, 70^{\circ} \mathrm{C}, 26 \%$; (f) $\mathrm{KOH}, \mathrm{EtOH} /$ water, 12 h , $80{ }^{\circ} \mathrm{C}, 72 \% ;(\mathrm{g})(1)$ triethylamine, HATU, $N$-hydroxyacetamidine, DMSO, $1 \mathrm{~h}, \mathrm{RT}$; (2) $\mathrm{Bu}_{4} \mathrm{NOH}, \mathrm{THF} /$ water, 20 min., RT, $34 \%$.
carboxylic acids 30 and 18: the Gewald reaction on 28 or its methylated analogue 31 gave aminocyanothiophenes 29 and 32. ${ }^{28}$ Saponification resulted in the desired acid building blocks 30 and 18 which could be converted to the corresponding amides 16 and $17 \mathrm{a} / \mathrm{b}$ or to oxadiazole 19 via a two-step procedure by coupling the acid with $N$-hydroxyacetamidine followed by ring closure. Aryl-substituted oxadiazoles 20-22 were accessible by a common procedure (Scheme 3): reaction of a BOC-protected amino benzonitrile with hydroxylamine resulted in an amidoxime. Coupling with the enantiopure acid 18a, which was obtained by chiral SFC from the racemic material 18, followed by ring closure yielded the corresponding oxadiazole 37/38. After BOC-deprotection, the acrylamides were obtained by amidation using acryloyl chloride. ${ }^{29}$
For the synthesis of BI-0474 (Scheme 4), a suitably substituted nitrile (42) was obtained from its dichloro analogue 40 by regioselective nucleophilic aromatic substitutions followed by a Pd-catalyzed amination to introduce the second piperazine substituent. The desired acrylamide was then synthesized, as described above via reaction of the nitrile with hydroxylamine, coupling of the resulting amidoxime 43 with the enantiopure acid 18a followed by ring closure to the oxadiazole 45. Deprotection and amidation resulted in the acrylamide BI-0474.

## CONCLUSIONS

Here, we report the discovery of an in vivo active KRAS ${ }^{\text {G12C }}$ inhibitor derived from a reversibly bound fragment hit. We identified fragments binding to KRAS using a second-site screening approach with a covalently blocked switch I/II pocket. Using structure-based design, we then optimized the initial hits into compounds that bind with single digit $\mu \mathrm{M}$ affinity to KRAS. Further optimization of these compounds via installation of an acrylamide warhead led to single digit nanomolar $\mathrm{IC}_{50}$ values in a biochemical assay. This strategy produced BI-0474 which shows in vivo biomarker modulation and efficacy in an NCI-H358 model after intermittent treatment via ip administration. In contrast to known KRAS ${ }^{\mathrm{G} 12 \mathrm{C}}$ inhibitors such as sotorasib or adagrasib, which are dosed orally, BI-0474 has only shown efficacy using ip dosing. However, a more advanced orally available analogue from this series has recently entered phase I clinical trials (NCT04973163)..$^{24,30,31}$ Based on its non-covalent binding features, the chemical matter identified via our approach has not only led to the discovery of KRAS ${ }^{\mathrm{G} 12 \mathrm{C}}$ inhibitors but also molecules that are highly attractive for further use in the discovery of inhibitors against other KRAS mutants.

## - EXPERIMENTAL SECTION

Chemistry. General Procedures. Commercial starting materials were used without further purification. Solvents used for reactions were of commercial "dry" or "extra-dry" or "analytical" grade. All other solvents used were of reagent grade. Unless specifically mentioned, all reactions were carried out in standard commercially available glassware using standard synthetic chemistry methods. Air-sensitive and moisture-sensitive reactions were performed under an atmosphere of dry nitrogen or argon with dried glassware. If not explicitly mentioned otherwise, the purity of all tested compounds was $\geq 95 \%$.

Preparative reversed-phase high-performance liquid chromatography (RP-HPLC) was carried out on Agilent or Gilson systems using columns from Waters (Sunfire C18 OBD, 5 or $10 \mu \mathrm{~m}, 20 \mathrm{~mm} \times 50$ $\mathrm{mm}, 30 \mathrm{~mm} \times 50 \mathrm{~mm}$, or $50 \mathrm{~mm} \times 150 \mathrm{~mm}$; X-Bridge C18OBD, 5 or $10 \mu \mathrm{~m}, 20 \mathrm{~mm} \times 50 \mathrm{~mm}, 30 \mathrm{~mm} \times 50 \mathrm{~mm}$, or $50 \mathrm{~mm} \times 150 \mathrm{~mm}$ ) or YMC (Triart C18, 5, or $10 \mu \mathrm{~m}, 20 \mathrm{~mm} \times 50 \mathrm{~mm}$, or $30 \mathrm{~mm} \times 50$ mm ). Unless specifically mentioned, compounds were eluted with $\mathrm{MeCN} /$ water gradients using either acidic $[0.2 \% \mathrm{HCOOH}$ or trifluoroacetic acid (TFA)] or basic water ( 5 mL of 2 M $\mathrm{NH}_{4} \mathrm{HCO}_{3}+2 \mathrm{~mL}$ of $\mathrm{NH}_{3}(32 \%)$ made up to 1 L with water).

NMR experiments were recorded on Bruker AVANCE HD 400, 500 , and 600 MHz spectrometers equipped with a TCI cryoprobe. Samples were dissolved in $600 \mu \mathrm{~L}$ of DMSO- $d_{6}$, and TMS was added as an internal standard. The temperature was set to 298 K. Processing of spectra was performed with Bruker Topspin 3.6 software. Spectra were analyzed with ACD/NMR Workbook 2019. Chemical shifts are reported in ppm on the $\delta$ scale. No zero filling was performed, and the spectra were manually integrated after automatic baseline correction. The $1 \mathrm{D}^{1} \mathrm{H}$ spectra were acquired with $30^{\circ}$ excitation pulses and an interpulse delay of 4.2 s with 64 k data points and a 20 ppm sweep width. The $1 \mathrm{D}{ }^{13} \mathrm{C}$ spectra were acquired with broadband composite pulse decoupling (WALTZ16) and an interpulse delay of 3.3 s with 64 k data points and a sweep width of 240 ppm . The 2D HSQC spectra were recorded on all samples to aid the interpretation of the data and to identify signals hidden underneath solvent peaks. The spectra were acquired with sweep widths obtained by automatic sweep width detection from the 1D reference spectra in the direct dimension with 1 k data points and with 210 ppm and 256 data points in the indirect dimension.

All samples were analyzed on an Agilent 1200 series LC system coupled with an Agilent 6140 mass spectrometer. Purity was determined via UV detection with a bandwidth of 170 nm in the range from 230 to 400 nm . LC parameters were as follows: Waters XBridge C18 column; $2.5 \mu \mathrm{~m}$ particle size; $2.1 \times 20 \mathrm{~mm}$; run time,

Scheme 3. Synthesis of Aryl-Substituted Oxadiazoles and Corresponding Acrylamides ${ }^{a}$

${ }^{a}$ Reagents and conditions: (a) $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}, \mathrm{NaHCO}_{3}$, $\mathrm{EtOH}, 16 \mathrm{~h}, 80^{\circ} \mathrm{C}, 57-65 \%$; (b) 18a, HATU, DIPEA, DMF, $1 \mathrm{~h}, 25^{\circ} \mathrm{C}$ then $12 \mathrm{~h}, 80^{\circ} \mathrm{C}$, $32-45 \%$; (c) TFA, DCM, $1 \mathrm{~h}, 0^{\circ} \mathrm{C}-\mathrm{RT}, 68-77 \%$; and (d) acryloyl chloride, $\mathrm{K}_{2} \mathrm{CO}_{3}$, acetone/water, $5 \mathrm{~h}, 0{ }^{\circ} \mathrm{C}, 32-63 \%$.

Scheme 4. Synthesis of BI-0474 (23) ${ }^{a}$

${ }^{a}$ Reagents and conditions: (a) BOC-piperazine, DIPEA, DMSO, $1 \mathrm{~h}, 60{ }^{\circ} \mathrm{C}, 69 \%$; (b) (S)-1,3-dimethylpiperazine, $\mathrm{Pd}_{2} \mathrm{dba}_{3}$, xantphos, $^{2} \mathrm{Cs}_{2} \mathrm{CO}_{3}$, dioxane, $16 \mathrm{~h}, 110^{\circ} \mathrm{C}, 19 \%$; (c) $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}, \mathrm{Na}_{2} \mathrm{CO}_{3}, \mathrm{EtOH}, 1 \mathrm{~h}, 85^{\circ} \mathrm{C}, 99 \%$; (d) 18a, HATU, DIPEA, DMF, $16 \mathrm{~h}, 25^{\circ} \mathrm{C}, 29 \%$; (e) DBU, 16 h , $90^{\circ} \mathrm{C}, 56 \%$; (f) HCl , dioxane, $1 \mathrm{~h}, 40^{\circ} \mathrm{C}, 67 \%$; and (g) acryloyl chloride, $\mathrm{K}_{2} \mathrm{CO}_{3}$, acetone/water, $15 \mathrm{~min}, \mathrm{RT}, 47 \%$.
2.1 min ; flow, $1 \mathrm{~mL} / \mathrm{min}$; column temperature, $60{ }^{\circ} \mathrm{C}$; $5 \mu \mathrm{~L}$ injections; solvent A ( $\left.20 \mathrm{mM} \mathrm{NH} 4 \mathrm{HCO}_{3} / \mathrm{NH}_{3}, \mathrm{pH} 9\right)$ and solvent B (MS-grade acetonitrile); start $10 \% \mathrm{~B}$, gradient 10 to $95 \%$ B from 0.0 to $1.5 \mathrm{~min}, 95 \%$ B from 1.5 to 2.0 min , gradient 95 to $10 \%$ B from 2.0 to 2.1 min .

HRMS data were recorded using an LTQ Orbitrap XL (Thermo Scientific) coupled with a Triversa Nanomate Nanospray ion source (ADVION Bioscience Inc.). The mass calibration was performed using the Pierce LTQ Velos ESI positive ion calibration solution from Thermo Scientific (product no. 88323). The scan window was set to $100-500 \mathrm{amu}$ with a maximum injection time of 500 ms and 1 microscan. Resolution of the Orbitrap was 60,000 with a mass accuracy of $\leq 5 \mathrm{ppm}$. The ion mode was set to positive with a capillary temperature of $200^{\circ} \mathrm{C}$ and a voltage of 60 eV . The tube lens potential was set to 110 eV . The NanoESI voltage was 1.45 kV , and the $\mathrm{N}_{2}$ gas pressure was set to 0.45 psi . The total sample volume was $5 \mu \mathrm{~L}$, and the acquisition time was 60 s , with 15 scans of averaging per spectrum. The sample dilution [ 10 mM dimethyl sulfoxide (DMSO) stock solution] was diluted $1: 200$ in $50 \% \mathrm{MeOH}+0.01 \%$ formic acid.

Synthesis of esters 29 and 32 has been previously described, ${ }^{28}$ and a detailed description of the methods used is provided in the Supporting Information.

2-Amino-4,4-dimethyl-4,5,6,7-tetrahydro-1-benzothiophene-3carbonitrile (12). To a solution of 2,2-dimethylcyclohexanone (24, $1.50 \mathrm{~g} ; 11.9 \mathrm{mmol}$ ) in dimethyl formamide (DMF) ( 15.0 mL ) was added BMIMCl ( $0.65 \mathrm{~g}, 2.4 \mathrm{mmol}$ ), sulfur ( $1.14 \mathrm{~g}, 35.7 \mathrm{mmol}$ ), malononitrile ( $1.73 \mathrm{~g}, 26.2 \mathrm{mmol}$ ), and l-proline ( $1.37 \mathrm{~g}, 11.9 \mathrm{mmol}$ ), and the reaction mixture was allowed to stir at $70{ }^{\circ} \mathrm{C}$ for 12 h . The reaction mixture was extracted with EtOAc , washed with water, dried, and concentrated in vacuo to obtain the crude product. This was purified by flash column chromatography (EtOAc/hexane) to afford 1.50 g of 12 (61\%).
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 6.86(\mathrm{~s}, 2 \mathrm{H}), 2.37(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=$ $6.1 \mathrm{~Hz}), 1.6-1.7(\mathrm{~m}, 2 \mathrm{H}), 1.5-1.6(\mathrm{~m}, 2 \mathrm{H}), 1.25(\mathrm{~s}, 6 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta 164.3,139.2,118.0,116.8$, 82.1, 39.4, 33.6, 28.7, 24.8, 20.3.

HRMS $(m / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{~S}, 207.09505$; found, 207.09516.

8-Chloro- $\mathrm{N}, \mathrm{N}$-dimethylnaphthalen-1-amine (14). 8-Chloronaph-thalen-1-amine $(25,50 \mathrm{mg}, 0.27 \mathrm{mmol})$ and potassium carbonate $(164 \mathrm{mg}, 1.19 \mathrm{mmol})$ were suspended in DMF $(0.5 \mathrm{~mL})$ and treated with $\mathrm{CH}_{3} \mathrm{I}(170 \mathrm{mg}, 0.594 \mathrm{mmol})$. The reaction mixture was heated to $50{ }^{\circ} \mathrm{C}$ and stirred overnight. Additional potassium carbonate ( 82 $\mathrm{mg}, 0.59 \mathrm{mmol})$ and $\mathrm{CH}_{3} \mathrm{I}(85 \mathrm{mg}, 0.59 \mathrm{mmol})$ were added, and stirring was continued at $50^{\circ} \mathrm{C}$ for additional 5 h until the reaction showed acceptable conversion. The reaction mixture was filtered, and the vial and remaining solids were washed with DMSO. The product was isolated from the filtrate via preparative RP HPLC (Agilent, Waters XBridge C18 $30 \times 50 \mathrm{~mm} 5 \mu \mathrm{~m}, 30-80 \% \mathrm{MeCN}$ in water + $0.1 \%$ formic acid) to afford 20 mg ( $35 \%$ ) of 14 .
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.86(\mathrm{dd}, 1 \mathrm{H}, J=0.9,8.3 \mathrm{~Hz})$, $7.6-7.6(\mathrm{~m}, 1 \mathrm{H}), 7.55(\mathrm{dd}, 1 \mathrm{H}, J=1.3,7.3 \mathrm{~Hz}), 7.46(\mathrm{t}, 1 \mathrm{H}, J=7.8$ $\mathrm{Hz}), 7.41(\mathrm{t}, 1 \mathrm{H}, J=7.8 \mathrm{~Hz}), 7.21(\mathrm{dd}, 1 \mathrm{H}, J=1.1,7.5 \mathrm{~Hz}), 2.72(\mathrm{~s}$, $6 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta 150.5,137.6,129.4,129.4$, 128.8, 127.1, 126.2, 125.0, 123.4, 116.5, 45.7.

HRMS $(\mathrm{m} / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{ClN}$, 206.07310; found, 206.07327.

3-Fluoro-2-(3-fluoropyridin-2-yl)phenol (15). 2-Chloro-3-fluoropyridine $(26,50 \mathrm{mg}, 0.37 \mathrm{mmol})$ was dissolved in dioxane $(1.0 \mathrm{~mL})$ and heated to $80^{\circ} \mathrm{C}$. 2- fluoro-6-hydroxyphenylboronic acid (27, 122 $\mathrm{mg}, 0.745 \mathrm{mmol}$ ) was dissolved in dioxane $(0.5 \mathrm{~mL})$ and added, followed by cesium carbonate $(2.0 \mathrm{M}$ aqueous solution, $0.47 \mathrm{~mL}, 0.93$ $\mathrm{mmol})$ and XPhos Pd G3 $(33.2 \mathrm{mg}, 0.037 \mathrm{mmol})$, and the reaction mixture was stirred for 1.5 h at $80^{\circ} \mathrm{C}$. The mixture was diluted with water and brine, extracted with dichloromethane (DCM), and the combined extracts were dried, filtered, and concentrated. The crude product was purified via preparative RP HPLC (Agilent, WatersSunfire C18 $30 \times 50 \mathrm{~mm} 5 \mu \mathrm{~m}, 10-70 \% \mathrm{MeCN}$ in water $+0.1 \%$ HCOOH ), yielding 37 mg ( $48 \%$ ) of $\mathbf{1 5}$ as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 10.09(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.52(\mathrm{td}, 1 \mathrm{H}$, $J=1.6,4.6 \mathrm{~Hz}), 7.80(\mathrm{ddd}, 1 \mathrm{H}, J=1.3,8.4,9.6 \mathrm{~Hz}), 7.52(\mathrm{td}, 1 \mathrm{H}, J=$ $4.4,8.5 \mathrm{~Hz}), 7.31(\mathrm{dt}, 1 \mathrm{H}, J=7.0,8.3 \mathrm{~Hz}), 6.81(\mathrm{~d}, 1 \mathrm{H}, J=8.3 \mathrm{~Hz})$, $6.7-6.8(\mathrm{~m}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta 160.9,157.4,157.8,145.8$, 140.7, 131.3, 125.4, 123.9, 112.1, 111.5, 106.1.

HRMS $(m / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~F}_{2} \mathrm{NO}, 208.05685$; found, 208.5674.

2-Amino-3-cyano-4,5,6,7-tetrahydro-1-benzothiophene-4-carboxylic Acid (30). Ethyl 2-amino-3-cyano-4,5,6,7-tetrahydro-1-benzo-thiophene-4-carboxylate $(29,1.0 \mathrm{~g}, 4.00 \mathrm{mmol})$ was dissolved in dioxane $/ \mathrm{MeCN} /$ water ( $1: 1: 5,7 \mathrm{~mL}$ ), lithium hydroxide ( 382 mg , 16.0 mmol ) was added, and the mixture was stirred for 48 h at $50^{\circ} \mathrm{C}$. $\mathrm{HCl}(6 \mathrm{M}$ in water) was added until pH 3 , and the product was extracted with EtOAc. The combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated, yielding 760 mg ( $86 \%$ ) of 30 as orange crystals which were used for the next step without purification.

2-Amino-3-cyano-N-phenyl-4,5,6,7-tetrahydro-1-benzothio-phene-4-carboxamide (16). 2-Amino-3-cyano-4,5,6,7-tetrahydro-1-benzothiophene-4-carboxylic acid (30, $100 \mathrm{mg}, 0.427 \mathrm{mmol}$ ) and $O$ -(7-azabenzotriazol-1-yl)- $N, N, N^{\prime}, N^{\prime}$-tetramethyluronium hexafluorophosphate (HATU) ( $23.5 \mathrm{mg}, 0.598 \mathrm{mmol}$ ) were dissolved in DMF ( 2.0 mL ) and cooled to $0^{\circ} \mathrm{C}$. Diisopropyl ethyl amine (DIPEA) $(166 \mathrm{mg}, 1.28 \mathrm{mmol})$ and aniline ( $52 \mathrm{mg}, 0.56 \mathrm{mmol}$ ) were added, and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min and then allowed to reach RT. The reaction mixture was filtered, and the product was isolated via preparative RP HPLC (Agilent, Waters XBridge C18 $30 \times$ $50 \mathrm{~mm} 5 \mu \mathrm{~m}, 5-70 \% \mathrm{MeCN}$ in water $+0.1 \%$ ammonium carbonate/ $\mathrm{NH}_{3}$ ), yielding 40 mg ( $32 \%$ ) of $\mathbf{1 6}$ as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 10.08(\mathrm{~s}, 1 \mathrm{H}), 7.59(\mathrm{dd}, 2 \mathrm{H}, \mathrm{J}$ $=0.9,8.6 \mathrm{~Hz}), 7.3-7.3(\mathrm{~m}, 2 \mathrm{H}), 7.0-7.1(\mathrm{~m}, 1 \mathrm{H}), 6.93(\mathrm{~s}, 2 \mathrm{H}), 3.59$ $(\mathrm{t}, 1 \mathrm{H}, J=6.0 \mathrm{~Hz}), 2.4-2.5(\mathrm{~m}, 2 \mathrm{H}), 2.0-2.0(\mathrm{~m}, 1 \mathrm{H}), 1.8-1.9(\mathrm{~m}$, $2 \mathrm{H}), 1.6-1.7(\mathrm{~m}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $d_{6}, 151 \mathrm{MHz}$ ): $\delta$ 169.8, 161.0, 137.6, 127.6, 126.9, 121.5, 118.0, 117.7, 114.2, 81.5, 40.9, 25.9, 21.9, 19.1.

HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{OS}$, 298.10086; found, 298.10105.

2-Amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothio-phene-4-carboxylic Acid (18). Ethyl 2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-4-carboxylate (32, 78 mg , $0.295 \mathrm{mmol})$ was dissolved in ethanol $(1.5 \mathrm{~mL})$, potassium hydroxide ( $370 \mu \mathrm{~L}, 4 \mathrm{M}$ in water, 1.48 mmol ) was added, and the mixture was stirred for 12 h at $80^{\circ} \mathrm{C}$. The mixture was diluted with water and $\mathrm{EtOAc}, \mathrm{KHSO}_{4}$ ( $10 \%$ in water) was added to adjust to pH 4 , and the mixture was extracted with EtOAc. The organic layers were combined and concentrated, and the crude product was purified via preparative RP HPLC (Gilson, $20-90 \% \mathrm{MeCN}$ in water $+0.1 \% \mathrm{HCOOH}$ ) to afford $50 \mathrm{mg}(72 \%)$ of 18 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 12.50(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.86(\mathrm{~s}, 2 \mathrm{H})$, $2.42(\mathrm{t}, 2 \mathrm{H}, J=6.1 \mathrm{~Hz}), 1.9-2.0(\mathrm{~m}, 1 \mathrm{H}), 1.73$ (quin, $2 \mathrm{H}, J=6.0$ $\mathrm{Hz}), 1.6-1.7(\mathrm{~m}, 1 \mathrm{H}), 1.42(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta 177.2,163.3,134.0,118.6$, 117.0, 83.4, 44.4, 35.9, 24.3, 24.3, 20.2.

HRMS $(m / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}, 237.06922$; found, 237.06961.

Chiral SFC of enantiomers of 18 was performed on a Thar SFC Pre80 using a Chiralpak AD-H $5 \mu \mathrm{~m}, 21 \mathrm{~mm} \times 250 \mathrm{~mm}$ column at a flow rate of $25 \mathrm{~g} / \mathrm{min}$ at $35^{\circ} \mathrm{C}$ and 120 bar with $75 \% \mathrm{CO}_{2}+25 \%$ MeOH ( $+0.5 \%$ isopropylamine) as the mobile phase to obtain 18a as the first eluting stereoisomer.

2-Amino-3-cyano-N,4-dimethyl-6,7-dihydro-5H-benzothio-phene-4-carboxamide (17a). The preparation was carried out as described for 16 using 2-amino-3-cyano-4-methyl-6,7-dihydro-5H-benzothiophene-4-carboxylic acid $(18,22 \mathrm{mg}, 0.093 \mathrm{mmol})$ and methyl ammonium chloride ( $7.5 \mathrm{mg}, 0.11 \mathrm{mmol}$ ) to afford 8 mg (32\%) of 17 a as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.11(\mathrm{q}, 1 \mathrm{H}, J=4.2 \mathrm{~Hz}), 6.94$ $(\mathrm{s}, 2 \mathrm{H}), 2.63(\mathrm{~d}, 3 \mathrm{H}, J=4.6 \mathrm{~Hz}), 2.48(\mathrm{t}, 2 \mathrm{H}, J=6.2 \mathrm{~Hz}), 1.9-2.0$ $(\mathrm{m}, 1 \mathrm{H}), 1.74$ (quin, $2 \mathrm{H}, J=6.0 \mathrm{~Hz}), 1.6-1.7(\mathrm{~m}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta$ 175.6, 163.8, 134.2, 119.7, 116.8, 83.0, 44.9, 36.7, 26.9, 24.5, 24.4, 20.2.

HRMS $(\mathrm{m} / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{OS}$, 250.10086; found, 250.10099.

2-Amino-3-cyano-4-methyl-N-phenyl-6,7-dihydro-5H-benzo-thiophene-4-carboxamide (17b). The preparation was carried out as described for 16 using 2-amino-3-cyano-4-methyl-6,7-dihydro-5H-benzothiophene-4-carboxylic acid (18, $100 \mathrm{mg}, 0.423 \mathrm{mmol}$ ) and aniline ( $51 \mathrm{mg}, 0.55 \mathrm{mmol}$ ) to afford $70 \mathrm{mg}(29 \%)$ of $\mathbf{1 7 b}$ as an offwhite solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 8.93(\mathrm{~s}, 1 \mathrm{H}), 7.5-7.6(\mathrm{~m}, 2 \mathrm{H})$, $7.2-7.3(\mathrm{~m}, 2 \mathrm{H}), 7.0-7.1(\mathrm{~m}, 1 \mathrm{H}), 6.90(\mathrm{~s}, 2 \mathrm{H}), 2.5-2.6(\mathrm{~m}, 1 \mathrm{H})$, $2.4-2.5(\mathrm{~m}, 1 \mathrm{H}), 2.04(\mathrm{ddd}, 1 \mathrm{H}, J=3.2,10.0,13.0 \mathrm{~Hz}), 1.7-1.8(\mathrm{~m}$, $3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $d_{6}, 151 \mathrm{MHz}$ ): $\delta$ 174.3, 163.6, 139.4, 133.9, 128.7, 124.0, 121.8, 119.9, 116.7, 82.9, 45.9, 35.9, 24.4, 24.2, 20.2.

HRMS $(\mathrm{m} / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{OS}$, 312.11651; found, 312.11654.

2-Amino-4-methyl-4-(3-methyl-1,2,4-oxadiazol-5-yl)-4,5,6,7-tet-rahydro-1-benzothiophene-3-carbonitrile (19). 2-Amino-3-cyano-4-methyl-6,7-dihydro-5H-benzothiophene-4-carboxylic acid (18, 120 $\mathrm{mg}, 0.498 \mathrm{mmol})$ and triethylamine $(143 \mu \mathrm{~L}, 0.995 \mathrm{mmol})$ were dissolved in DMSO ( 1.0 mL ). After addition of HATU ( $203 \mathrm{mg}, 0.23$ mmol ) and stirring at RT for $15 \mathrm{~min}, N$-hydroxyacetamidine ( 42 mg , 0.547 mmol ) was added, and the mixture was stirred for 1 h at RT. The reaction mixture was filtered, and the product was isolated via preparative RP HPLC (Agilent, YMC Triart C18 $19 \times 50 \mathrm{~mm} 5 \mu \mathrm{~m}$, MeCN in water $+0.1 \%$ formic acid), yielding a crude intermediate which was used directly for the next step. The intermediate was dissolved in tetrahydrofuran (THF) ( 1.0 mL ), tetrabutylammonium hydroxide ( $40 \%$ in water, $170 \mu \mathrm{~L}, 0.26 \mathrm{mmol}$ ) was added, and the mixture was stirred for 20 min at RT. The reaction mixture was concentrated, diluted with DMSO, and the product was isolated via preparative RP HPLC (Agilent, YMC Triart C18 $19 \times 50 \mathrm{~mm} 5 \mu \mathrm{~m}$, MeCN in water $+0.1 \%$ formic acid) to afford $46 \mathrm{mg}(34 \%)$ of 19 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.04(\mathrm{~s}, 2 \mathrm{H}), 2.32(\mathrm{~s}, 3 \mathrm{H}), 2.00$ (ddd, $1 \mathrm{H}, \mathrm{J}=2.9,9.5,13.0 \mathrm{~Hz}), 1.8-1.9(\mathrm{~m}, 2 \mathrm{H}), 1.7-1.8(\mathrm{~m}, 1 \mathrm{H})$, $1.72(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta$ 183.6, 167.2, 164.1, 132.1, 119.9, 116.3, 82.1, 38.9, 37.4, 25.1, 24.1, 19.9, 11.7.

HRMS $(m / z):[M+H]^{+}$calcd for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{OS}$, 275.09611; found, 275.09648.
tert-Butyl $N-\left\{4-\left[(Z)-N^{\prime}-H y d r o x y c a r b a m i m i d o y l\right] p h e n y l\right\}-$ carbamate (35a). To a stirred solution of tert-butyl $N$-(4cyanophenyl)carbamate ( $33 \mathrm{a}, 3.0 \mathrm{~g}, 13.8 \mathrm{mmol}$ ) in ethanol (30 mL ) was added hydroxylamine $\mathrm{HCl}(1.9 \mathrm{~g}, 27.5 \mathrm{mmol})$ and sodium bicarbonate ( $2.3 \mathrm{~g}, 27.45 \mathrm{mmol}$ ) at RT , and the reaction mixture was stirred for 16 h at $80^{\circ} \mathrm{C}$. The mixture was cooled to RT, filtered, and the precipitate was washed with EtOAc. The filtrate was concentrated, and the crude product was purified by NP chromatography (10-20\% EtOAc in PE) to yield 2.0 g of (57\%) 35a as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right): \delta 9.46(\mathrm{~s}, 1 \mathrm{H}), 9.43(\mathrm{~s}, 1 \mathrm{H}), 7.55$ $(\mathrm{d}, 2 \mathrm{H}, J=8.8 \mathrm{~Hz}), 7.43(\mathrm{~d}, 2 \mathrm{H}, J=8.8 \mathrm{~Hz}), 5.69(\mathrm{~s}, 2 \mathrm{H}), 1.48(\mathrm{~s}$, 9H).
tert-Butyl N -(4-\{5-[(4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetra-hydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}phenyl)carbamate (37a). To a stirred solution of (4S)-2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-4-carboxylic acid (18a, 329 mg 1.39 mmol ) in dry DMF $(7.0 \mathrm{~mL})$ was added DIPEA $(1.25 \mathrm{~mL}, 7.00 \mathrm{mmol})$ followed by HATU $(581 \mathrm{mg}, 1.53 \mathrm{mmol})$. The reaction mixture was stirred for 20 min at RT, then tert-butyl $N$ - $\{4$ -$\left[(Z)-N^{\prime}\right.$-hydroxycarbamimidoyl $]$ phenyl $\}$ carbamate (35a, 350 mg , 1.39 mmol ) was added, and the reaction mixture was stirred at 25 ${ }^{\circ} \mathrm{C}$ for 1 h and then heated to $80{ }^{\circ} \mathrm{C}$ for 12 h . The reaction was quenched with water, stirred for 30 min , extracted with EtOAc, and concentrated. The crude product was purified by NP chromatography ( $0-30 \% \mathrm{EtOAc}$ in PE ) to yield $209 \mathrm{mg}(41 \%)$ of 37 a as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right): \delta 9.71(\mathrm{~s}, 1 \mathrm{H}), 7.87(\mathrm{~d}, 2 \mathrm{H}, J=$ $8.8 \mathrm{~Hz}), 7.63(\mathrm{~d}, 2 \mathrm{H}, J=8.8 \mathrm{~Hz}), 7.06(\mathrm{~s}, 2 \mathrm{H}), 2.53(\mathrm{br} \mathrm{d}, 2 \mathrm{H}, J=6.4$ $\mathrm{Hz}), 2.0-2.1(\mathrm{~m}, 1 \mathrm{H}), 1.8-2.0(\mathrm{~m}, 3 \mathrm{H}), 1.78(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 9 \mathrm{H})$.
(4S)-2-Amino-4-[3-(4-aminophenyl)-1,2,4-oxadiazol-5-yl]-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile (20a). To a stirred solution of tert-butyl $N-(4-\{5-[(4 S)$-2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-1\}phenyl) carbamate ( $37 \mathrm{a}, 200 \mathrm{mg} 0.44 \mathrm{mmol}$ ) in dry DCM $(8.0 \mathrm{~mL})$ was added $15 \%$ TFA in DCM $(3.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$, and the reaction mixture was stirred for 1 h . The reaction was quenched with saturated $\mathrm{NaHCO}_{3}$ solution, the layers were separated, and the aqueous phase was extracted with DCM. The combined organic layers were dried and concentrated, and the crude product was purified by preparative RP HPLC to afford $120 \mathrm{mg}(77 \%)$ of 20 a as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.6-7.7(\mathrm{~m}, 2 \mathrm{H}), 7.04(\mathrm{~s}, 2 \mathrm{H})$, $6.6-6.7(\mathrm{~m}, 2 \mathrm{H}), 5.74(\mathrm{~s}, 2 \mathrm{H}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H}), 2.0-2.1(\mathrm{~m}, 1 \mathrm{H})$, $1.8-1.9(\mathrm{~m}, 3 \mathrm{H}), 1.76(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta 183.3,168.2,164.0,152.2$, 132.3, 128.8, 119.9, 116.3, 114.0, 113.2, 82.2, 39.0, 37.4, 25.3, 24.1, 20.0.

HRMS $(m / z):[M+H]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{OS}, 352.12266$; found, 352.12279.

N-(4-\{5-[(4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}phenyl)prop-2-enamide (21a). To a stirred solution of acryloyl chloride ( $60 \mathrm{mg}, 0.67 \mathrm{mmol}$ ) in acetone $(3.3 \mathrm{~mL})$ and water $(1.1 \mathrm{~mL}),(4 S)$-2-amino-4-[3-(4-aminophenyl)-1,2,4-oxadiazol-5-yl]-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile (20a, 55 mg 0.16 mmol ) in acetone and water $(2: 6,0.5 \mathrm{~mL})$ was slowly added at $0{ }^{\circ} \mathrm{C}$. Then potassium carbonate ( $43 \mathrm{mg}, 0.67 \mathrm{mmol}$ ) was added, and the mixture was stirred for 5 h at $0^{\circ} \mathrm{C}$. The mixture was concentrated, and the crude product was purified by preparative RP HPLC to yield 35 mg (55\%) of 21a as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 10.45(\mathrm{~s}, 1 \mathrm{H}), 7.9-8.0(\mathrm{~m}$, $2 \mathrm{H}), 7.8-7.9(\mathrm{~m}, 2 \mathrm{H}), 7.07(\mathrm{~s}, 2 \mathrm{H}), 6.47(\mathrm{dd}, 1 \mathrm{H}, J=10.2,17.0 \mathrm{~Hz})$, $6.31(\mathrm{dd}, 1 \mathrm{H}, J=1.8,16.9 \mathrm{~Hz}), 5.8-5.9(\mathrm{~m}, 1 \mathrm{H}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H})$, $2.1-2.1(\mathrm{~m}, 1 \mathrm{H}), 1.94(\mathrm{ddd}, 1 \mathrm{H}, J=2.7,7.8,13.3 \mathrm{~Hz}), 1.8-1.9(\mathrm{~m}$, $1 \mathrm{H}), 1.80(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta$ 184.2, 167.7, 164.1, 163.9, 142.3, 132.1, 128.4, 128.1, 121.4, 120.0, 120.0, 116.3, 82.0, 39.1, 37.3, 25.2, 24.1, 20.0.
$\operatorname{HRMS}(m / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}, 406.13322$; found, 406.13328.
tert-Butyl $N-\left\{3-\left[(Z)-N^{\prime}-H y d r o x y c a r b a m i m i d o y l\right] p h e n y l\right\}-$ carbamate (35b). The preparation was carried out as described for 35a using tert-butyl N -(3-cyanophenyl) carbamate (33b, $3.0 \mathrm{~g}, 13.8$ $\mathrm{mmol})$ to afford $2.1 \mathrm{~g}(58 \%)$ of 35 b as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right): \delta 9.58(\mathrm{~s}, 1 \mathrm{H}), 9.36(\mathrm{~s}, 1 \mathrm{H}), 7.82$ $(\mathrm{s}, 1 \mathrm{H}), 7.4-7.5(\mathrm{~m}, 1 \mathrm{H}), 7.2-7.3(\mathrm{~m}, 2 \mathrm{H}), 5.68(\mathrm{~s}, 2 \mathrm{H}), 1.47(\mathrm{~s}$, 9H).
tert-Butyl $N$-(3-\{5-[(4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetra-hydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}phenyl)carbamate (37b). The preparation was carried out as described for 37a using tert-butyl $N-\left\{3-\left[(Z)-N^{\prime}\right.\right.$-hydroxycarbamimidoyl $]$ phenyl $\}$ carbamate ( $35 b, 250 \mathrm{mg}, 0.995 \mathrm{mmol}$ ) to afford $200 \mathrm{mg}(45 \%)$ of 37 b as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right): \delta 9.60(\mathrm{~s}, 1 \mathrm{H}), 8.21(\mathrm{~s}, 1 \mathrm{H})$, $7.5-7.6(\mathrm{~m}, 2 \mathrm{H}), 7.4-7.5(\mathrm{~m}, 1 \mathrm{H}), 7.07(\mathrm{~s}, 2 \mathrm{H}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H})$, $2.0-2.1(\mathrm{~m}, 1 \mathrm{H}), 1.8-2.0(\mathrm{~m}, 3 \mathrm{H}), 1.79(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 9 \mathrm{H})$.
(4S)-2-Amino-4-[3-(3-aminophenyl)-1,2,4-oxadiazol-5-yl]-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile (20b). The preparation was carried out as described for 20a using tertbutyl $N$-(3-\{5-[(4S)-2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}phenyl)carbamate (37b, $200 \mathrm{mg}, 0.443 \mathrm{mmol}$ ) to afford $120 \mathrm{mg}(77 \%)$ of 20 b as an offwhite solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.24(\mathrm{t}, 1 \mathrm{H}, J=1.9 \mathrm{~Hz}), 7.1-$ $7.2(\mathrm{~m}, 1 \mathrm{H}), 7.1-7.1(\mathrm{~m}, 1 \mathrm{H}), 7.06(\mathrm{~s}, 2 \mathrm{H}), 6.7-6.8(\mathrm{~m}, 1 \mathrm{H}), 5.42$ $(\mathrm{s}, 2 \mathrm{H}), 2.55(\mathrm{dt}, 2 \mathrm{H}, J=2.8,6.0 \mathrm{~Hz}), 2.0-2.1(\mathrm{~m}, 1 \mathrm{H}), 1.93$ (ddd, $1 \mathrm{H}, J=2.7,7.8,13.3 \mathrm{~Hz}), 1.8-1.9(\mathrm{~m}, 2 \mathrm{H}), 1.78(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta$ 184.0, 168.5, 164.1, 149.8, 132.2, 130.1, 127.2, 119.9, 117.1, 116.2, 114.7, 112.5, 82.0, 39.1, 37.4, 25.2, 24.1, 20.0.

HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{OS}, 352.12266$; found, 352.12276.

N-(3-\{5-[(4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}pheyl) prop-2-enamide (21b). The preparation was carried out as described for 21a using (4S)-2-amino-4-[3-(3-aminophenyl)-1,2,4-oxadiazol-5-yl]-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile (20b, $55 \mathrm{mg}, 0.16$ $\mathrm{mmol})$ to afford $40 \mathrm{mg}(63 \%)$ of $\mathbf{2 1 b}$ as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 10.42(\mathrm{~s}, 1 \mathrm{H}), 8.30(\mathrm{~d}, 1 \mathrm{H}, J=$ $1.5 \mathrm{~Hz}), 7.98(\mathrm{dd}, 1 \mathrm{H}, J=0.8,8.2 \mathrm{~Hz}), 7.71(\mathrm{dd}, 1 \mathrm{H}, J=0.9,7.7 \mathrm{~Hz})$, $7.5-7.6(\mathrm{~m}, 1 \mathrm{H}), 7.08(\mathrm{~s}, 2 \mathrm{H}), 6.45(\mathrm{dd}, 1 \mathrm{H}, J=10.1,16.9 \mathrm{~Hz}), 6.3-$ $6.3(\mathrm{~m}, 1 \mathrm{H}), 5.80(\mathrm{dd}, 1 \mathrm{H}, J=1.7,10.2 \mathrm{~Hz}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H}), 2.10$ (ddd, 1H, $J=3.2,9.9,13.1 \mathrm{~Hz}), 1.9-2.0(\mathrm{~m}, 1 \mathrm{H}), 1.8-1.9(\mathrm{~m}, 2 \mathrm{H})$, 1.80 (s, 3H).
${ }^{13} \mathrm{C}$ NMR (DMSO- $\left.d_{6}, 151 \mathrm{MHz}\right): \delta$ 184.5, 167.9, 164.1, 163.9, $140.3,132.1,132.1,130.3,127.9,127.2,122.5,122.4,120.0,118.1$, 116.2, 81.9, 39.2, 37.3, 25.1, 24.1, 19.9.

HRMS $(m / z):[M+H]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}, 406.13322$; found, 406.13316.
tert-Butyl-4-\{3-[(Z)-N'-hydroxycabamimidoyl]phenyl\}piperazine1 -carboxylate (36). The preparation was carried out as described for 35a using tert-butyl 4-(3-cyanophenyl)piperazine-1-carboxylate (34, $3.0 \mathrm{~g}, 12.1 \mathrm{mmol})$ to afford $2.2 \mathrm{~g}(65 \%)$ of 36 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right): \delta 9.54(\mathrm{~s}, 1 \mathrm{H}), 7.2-7.2(\mathrm{~m}, 2 \mathrm{H})$, $7.1-7.2(\mathrm{~m}, 1 \mathrm{H}), 6.96(\mathrm{dd}, 1 \mathrm{H}, J=1.9,8.0 \mathrm{~Hz}), 5.76(\mathrm{br} \mathrm{d}, 2 \mathrm{H}, J=$ $1.6 \mathrm{~Hz}), 3.4-3.5(\mathrm{~m}, 4 \mathrm{H}), 3.1-3.1(\mathrm{~m}, 4 \mathrm{H}), 1.42(\mathrm{~s}, 9 \mathrm{H})$.
tert-Butyl 4-(3-\{5-[(4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetra-hydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}phenyl)-piperazine-1-carboxylate (38). The preparation was carried out as described for 37 a using tert-butyl-4-\{3-[(Z)- $N^{\prime}$ -hydroxycabamimidoyl]phenyl\}piperazine-1-carboxylate (36, 350 mg , $1.21 \mathrm{mmol})$ to afford $200 \mathrm{mg}(32 \%)$ of 38 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 400 \mathrm{MHz}\right): \delta 7.49(\mathrm{~s}, 1 \mathrm{H}), 7.4-7.4(\mathrm{~m}, 2 \mathrm{H})$, $7.19(\mathrm{td}, 1 \mathrm{H}, \mathrm{J}=2.2,7.4 \mathrm{~Hz}), 7.07(\mathrm{~s}, 2 \mathrm{H}), 3.48(\mathrm{br} \mathrm{d}, 4 \mathrm{H}, J=5.0$ $\mathrm{Hz}), 3.18($ br t, $4 \mathrm{H}, J=5.4 \mathrm{~Hz}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H}), 2.0-2.2(\mathrm{~m}, 1 \mathrm{H})$, $1.8-2.0(\mathrm{~m}, 3 \mathrm{H}), 1.79(\mathrm{~s}, 3 \mathrm{H}), 1.43$ ( $\mathrm{s}, 9 \mathrm{H})$.
(4S)-2-Amino-4-methyl-4-\{3-[3-(piperazin-1-yl)phenyl]-1,2,4-ox-adiazol-5-yl\}-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile (39). The preparation was carried out as described for 20a using tertbutyl 4-(3-\{5-[(4S)-2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}phenyl)piperazine-1-carboxylate $(38,200 \mathrm{mg}, 0.384 \mathrm{mmol})$ to afford $110 \mathrm{mg}(68 \%)$ of 39 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.38(\mathrm{~d}, 1 \mathrm{H}, J=1.7 \mathrm{~Hz}), 7.3-$ $7.3(\mathrm{~m}, 2 \mathrm{H}), 7.1-7.1(\mathrm{~m}, 1 \mathrm{H}), 7.00(\mathrm{~s}, 2 \mathrm{H}), 3.0-3.1(\mathrm{~m}, 4 \mathrm{H}), 2.8-$ $2.8(\mathrm{~m}, 4 \mathrm{H}), 2.5-2.5(\mathrm{~m}, 2 \mathrm{H}), 2.03(\mathrm{ddd}, 1 \mathrm{H}, J=2.9,9.8,13.2 \mathrm{~Hz})$, $1.87(\mathrm{ddd}, 1 \mathrm{H}, J=2.6,7.8,13.3 \mathrm{~Hz}), 1.7-1.8(\mathrm{~m}, 2 \mathrm{H}), 1.72(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO-d $\left.d_{6}, 151 \mathrm{MHz}\right): \delta 184.2,168.4,164.1,152.4$, 132.0, 130.3, 127.4, 120.0, 118.9, 117.9, 116.2, 113.3, 82.0, 49.3, 45.8, 39.1, 37.2, 25.3, 24.1, 20.0.

HRMS $(\mathrm{m} / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{22} \mathrm{H}_{24} \mathrm{~N}_{6} \mathrm{OS}, 421.18051$; found, 421.18058.
(4S)-2-Amino-4-methyl-4-(3-\{3-[4-(prop-2-enoyl)piperazin-1-yl]-phenyl\}-1,2,4-oxadiazol-5-yl)-4,5,6,7-tetrahydro-1-benzothio-phene-3-carbonitrile (22). The preparation was carried out as described for 21a using (4S)-2-amino-4-methyl-4-\{3-[3-(piperazin-1yl) phenyl]-1,2,4-oxadiazol-5-yl\}-4,5,6,7-tetrahydro-1-benzothiophene3 -carbonitrile ( $39,110 \mathrm{mg}, 0.262 \mathrm{mmol}$ ) to afford $40 \mathrm{mg}(32 \%)$ of 22 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.5-7.5(\mathrm{~m}, 1 \mathrm{H}), 7.4-7.5(\mathrm{~m}$, $1 \mathrm{H}), 7.4-7.4(\mathrm{~m}, 1 \mathrm{H}), 7.2-7.2(\mathrm{~m}, 1 \mathrm{H}), 7.07(\mathrm{~s}, 2 \mathrm{H}), 6.86(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}$ $=10.5,16.7 \mathrm{~Hz}), 6.15(\mathrm{dd}, 1 \mathrm{H}, J=2.4,16.7 \mathrm{~Hz}), 5.7-5.8(\mathrm{~m}, 1 \mathrm{H})$, 3.73 (br d, $4 \mathrm{H}, J=25.9 \mathrm{~Hz}$ ), $3.24(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H}), 2.11$ (ddd, $1 \mathrm{H}, J=3.0,9.9,13.2 \mathrm{~Hz}$ ), 1.94 (ddd, $1 \mathrm{H}, J=2.7,7.7,13.3 \mathrm{~Hz}$ ), $1.8-1.9(\mathrm{~m}, 2 \mathrm{H}), 1.80(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (DMSO-d $\left.{ }_{6}, 151 \mathrm{MHz}\right): \delta 184.3,168.3,164.7,164.1$, $151.6,132.0,130.4,128.6,128.0,127.5,120.0,119.4,118.5,116.2$, 113.8, 82.0, 49.0, 48.5, 45.2, 41.7, 39.1, 37.2, 25.3, 24.1, 20.0.

HRMS $(m / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{25} \mathrm{H}_{26} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{~S}, 475.19107$; found, 475.19107.
tert-Butyl 4-(2-Chloro-6-cyanopyridin-4-yl)piperazine-1-carboxylate (41). To a stirred solution of 4,6-dichloropyridine-2-carbonitrile ( $40,5.00 \mathrm{~g}, 28.9 \mathrm{mmol}$ ) in DMSO ( 50 mL ) was added piperazine-1carboxylic acid tert-butyl ester ( $5.92 \mathrm{~g}, 31.8 \mathrm{mmol}$ ). Then, DIPEA $(11.2 \mathrm{~g}, 86.7 \mathrm{mmol})$ was added, and the reaction mixture was stirred at $60{ }^{\circ} \mathrm{C}$ for 1 h . The mixture was dissolved in EtOAc and washed with water. The organic phase was dried, filtered, and concentrated under reduced pressure. The crude product was purified via NP chromatography ( $90-100 \%$ EtOAc in hexane), yielding $6.44 \mathrm{~g}(69 \%)$ of 41 as a white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 500 \mathrm{MHz}\right): \delta 7.55(\mathrm{~d}, 1 \mathrm{H}, J=1.6 \mathrm{~Hz}), 7.12$ $(\mathrm{s}, 1 \mathrm{H}), 3.5-3.5(\mathrm{~m}, 4 \mathrm{H}), 3.4-3.4(\mathrm{~m}, 4 \mathrm{H}), 1.42(\mathrm{~s}, 9 \mathrm{H})$.
tert-Butyl 4-\{2-Cyano-6-[(2S)-2,4-dimethylpiperazin-1-yl]pyridin-4-yl\}piperazine-1-carboxylate (42). tert-Butyl 4-(2-chloro-6-cyano-pyridin-4-yl)piperazine-1-carboxylate ( $41,1.00 \mathrm{~g}, 3.10 \mathrm{mmol}$ ), ( $S$ )-1,3-dimethylpiperazine $(0.99 \mathrm{~g}, 8.7 \mathrm{mmol})$, tris(dibenzylideneacetone)dipalladium(0) ( $141.9 \mathrm{mg}, 0.154 \mathrm{mmol}$ ), xantphos $(184.8 \mathrm{mg}, 0.31 \mathrm{mmol})$, cesium carbonate $(2.02 \mathrm{~g}, 6.20$ $\mathrm{mmol})$, and dry dioxane $(8.0 \mathrm{~mL})$ were combined and stirred in a closed vessel under an argon atmosphere for 16 h at $110{ }^{\circ} \mathrm{C}$. Brine was added to the mixture, and the product was extracted with DCM. The combined organic phases were dried, filtered, and concentrated under reduced pressure. The crude product was purified via preparative RP HPLC (Gilson, Waters XBridge C18 $50 \times 150 \mathrm{~mm}$ $10 \mu \mathrm{~m}, 30$ to $98 \% \mathrm{MeCN}$ in basic water), yielding 240 mg (19\%) of 42 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $\left.d_{6}, 500 \mathrm{MHz}\right): \delta 2.76(\mathrm{td}, 1 \mathrm{H}, J=2.8,11.8 \mathrm{~Hz})$, $2.6-2.7(\mathrm{~m}, 2 \mathrm{H}), 2.5-2.6(\mathrm{~m}, 2 \mathrm{H}), 2.11(\mathrm{~s}, 3 \mathrm{H}), 1.77(\mathrm{dt}, 1 \mathrm{H}, \mathrm{J}=$ $3.2,11.0 \mathrm{~Hz}), 1.44(\mathrm{t}, 1 \mathrm{H}, J=10.1 \mathrm{~Hz}), 0.90(\mathrm{~d}, 3 \mathrm{H}, J=6.3 \mathrm{~Hz})$.
tert-Butyl 4-\{2-[(2S)-2,4-Dimethylpiperazin-1-yl]-6-[(Z)- $N^{\prime}-$ hydroxycarbamimidoyl]pyridin-4-yl\}piperazine-1-carboxylate (43). tert-Butyl 4-\{2-cyano-6-[(2S)-2,4-dimethylpiperazin-1-yl]pyridin-4$\mathrm{yl}\}$ piperazine-1-carboxylate ( $42,200 \mathrm{mg}, 0.499 \mathrm{mmol}$ ), hydroxylamine hydrochloride $(90 \mathrm{mg}, 1.3 \mathrm{mmol})$, and sodium carbonate $(68.8$ mg 0.649 mmol ) were dissolved in dry EtOH ( 2.0 mL ), and the mixture was stirred at $85{ }^{\circ} \mathrm{C}$ for 1 h . The mixture was diluted with water and extracted with $10 \% \mathrm{MeOH}$ in DCM. The combined organic layers were washed with brine, dried, filtered, and concentrated under reduced pressure to afford 215 mg ( $99 \%$ ) of 43 as a brown oil which was used for the next step without further purification.
tert-Butyl $4-\left\{2-\left[(Z)-N^{\prime}-[(Z)-(4 S)-2-A m i n o-3-c y a n o-4-m e t h y l-\right.\right.$ 4,5,6,7-tetrahydro-1-benzothiophene-4-carbonyloxy]-carbamimidoyl]-6-[(2S)-2,4-dimethylpiperazin-1-yl]pyridin-4-yl\}-piperazine-1-carboxylate (44). (4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-4-carboxylic acid (18a, 150 mg , $0.64 \mathrm{mmol})$ in DMF $(1.5 \mathrm{~mL})$ was treated with HATU $(241 \mathrm{mg}, 0.64$ $\mathrm{mmol})$ and DIPEA $(331 \mu \mathrm{~L}, 1.90 \mathrm{mmol})$, and the mixture was stirred for 20 min at RT. Then, tert-butyl 4-\{2-[(2S)-2,4-dimethylpiperazin-1-yl]-6-[(Z)- $N^{\prime}$-hydroxycarbamimidoyl $]$ pyridin-4-yl\}piperazine-1-carboxylate ( $43,215 \mathrm{mg}, 0.495 \mathrm{mmol}$ ) was added, and the mixture was stirred for 16 h at RT. The reaction mixture was diluted with water and extracted with EtOAc. The combined organic layers were washed with brine, dried, filtered, and concentrated to afford 120 mg (29\%) of 44 which was used directly for the next step.
tert-Butyl 4-(2-\{5-[(4S)-2-Amino-3-cyano-4-methyl-4,5,6,7-tetra-hydro-1-benzothiophen-4-yl]-1,2,4-oxadiazol-3-yl\}-6-[(2S)-2,4-di-methylpiperazin-1-yl]pyridin-4-yl)piperazine-1-carboxylate (45). A mixture of tert-butyl 4-\{2-[(Z)- $N^{\prime}-[(Z)-(4 S)$-2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-4-carbonyloxy]-carbamimi-doyl]-6-[(2S)-2,4-dimethylpiperazin-1-yl]pyridin-4-yl\}-piperazine-1-carboxylate $(44,120 \mathrm{mg}, 0.184 \mathrm{mmol})$ and DBU (140 $\mathrm{mg}, 0.92 \mathrm{mmol}$ ) was stirred at $90^{\circ} \mathrm{C}$ for 16 h . The reaction mixture was diluted with water, extracted with EtOAc, and the combined organic layers were washed with brine, dried, filtered, and concentrated. The crude product was purified via NP chromatography ( $\mathrm{MeOH} / \mathrm{DCM}$ ) followed by preparative RP HPLC (Gilson, Waters XBridge C18 $50 \times 150 \mathrm{~mm} 5 \mu \mathrm{~m}, 40-98 \% \mathrm{MeCN}$ in basic water), yielding $65 \mathrm{mg}(56 \%)$ of 45 as a light brown solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}, 400 \mathrm{MHz}$ ): $\delta 7.08$ ( $\mathrm{s}, 2 \mathrm{H}$ ), 6.89 ( $\mathrm{s}, 1 \mathrm{H}$ ), 6.18 $(\mathrm{s}, 1 \mathrm{H}), 4.5-4.6(\mathrm{~m}, 1 \mathrm{H}), 4.08(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, J=13.4 \mathrm{~Hz}), 3.45(\mathrm{br} \mathrm{s}$, $4 \mathrm{H}), 2.99(\mathrm{dt}, 1 \mathrm{H}, J=3.0,12.4 \mathrm{~Hz}), 2.83(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, J=10.1 \mathrm{~Hz})$, 2.7-2.7 (m, 1H), $2.19(\mathrm{~s}, 3 \mathrm{H}), 2.0-2.1(\mathrm{~m}, 2 \mathrm{H}), 1.8-2.0(\mathrm{~m}, 4 \mathrm{H})$, 1.78 (s, 3H), 1.42 ( $\mathrm{s}, 9 \mathrm{H}$ ), 1.13 (d, 3H, J = 6.6 Hz ).
(4S)-2-Amino-4-(3-[6-[(2S)-2,4-dimethylpiperazin-1-yl]-4-(piper-azin-1-yl)pyridin-2-y|\}-1,2,4-oxadiazol-5-yl)-4-methyl-4,5,6,7-tetra-hydro-1-benzothiophene-3-carbonitrile (46). tert-Butyl 4-(2-\{5-[(4S)-2-amino-3-cyano-4-methyl-4,5,6,7-tetrahydro-1-benzothiophen4 -yl]-1,2,4-oxadiazol-3-yl\}-6-[(2S)-2,4-dimethylpiperazin-1-yl]-pyridin-4-yl)piperazine-1-carboxylate ( $45,140 \mathrm{mg}, 0.208 \mathrm{mmol}$ ) was dissolved in 1,4-dioxane ( 5 mL ), $\mathrm{HCl}(4 \mathrm{M}$ in 1,4-dioxane, 0.52 mL , 2.08 mmol ) was added, and the mixture was stirred for 1 h at $40^{\circ} \mathrm{C}$. The mixture was concentrated, two drops of triethylamine were added, and the product was isolated via preparative RP HPLC (Agilent, Waters XBridge C18 $50 \times 150 \mathrm{~mm} 5 \mu \mathrm{~m}, 10-70 \% \mathrm{MeCN}$ in basic water), yielding 75 mg ( $67 \%$ ) of 46 as an off-white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}, 600 \mathrm{MHz}$ ): $\delta 7.07(\mathrm{~s}, 2 \mathrm{H}), 6.88(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $1.8 \mathrm{~Hz}), 6.14(\mathrm{~d}, 1 \mathrm{H}, J=1.8 \mathrm{~Hz}), 4.54(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.06(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, J=$ $12.5 \mathrm{~Hz}), 3.2-3.3(\mathrm{~m}, 4 \mathrm{H}), 2.99(\mathrm{dt}, 1 \mathrm{H}, J=3.1,12.5 \mathrm{~Hz}), 2.84(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}), 2.8-2.8(\mathrm{~m}, 4 \mathrm{H}), 2.69(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, \mathrm{J}=10.8 \mathrm{~Hz}), 2.5-2.6(\mathrm{~m}$, 2H), $2.19(\mathrm{~s}, 3 \mathrm{H}), 2.1-2.1(\mathrm{~m}, 2 \mathrm{H}), 1.8-2.0(\mathrm{~m}, 4 \mathrm{H}), 1.78(\mathrm{~s}, 3 \mathrm{H})$, $1.12(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.6 \mathrm{~Hz})$.
${ }^{13}$ C NMR (DMSO- $d_{6}, 151 \mathrm{MHz}$ ): $\delta$ 183.9, 168.9, 164.1, 160.1 , 158.0, 144.9, 132.1, 119.9, 116.2, 100.6, 91.7, 82.0, 60.4, 55.5, 47.7, 47.2, 46.6, 45.8, 39.6, 39.1, 37.3, 25.4, 24.1, 20.0, 14.3.

HRMS $(\mathrm{m} / \mathrm{z}):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{27} \mathrm{H}_{35} \mathrm{~N}_{9} \mathrm{OS}$, 534.27580; found, 534.27579.
(4S)-2-Amino-4-(3-\{6-[(2S)-2,4-dimethylpiperazin-1-yl]-4-[4-(prop-2-enoyl)piperazin-1-yl]pyridin-2-y/\}-1,2,4-oxadiazol-5-yl)-4-methyl-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile (23). Acryloyl chloride ( 1 M in acetone, $419 \mu \mathrm{~L}, 0.419 \mathrm{mmol}$ ) was added to a mixture of potassium carbonate ( $57.8 \mathrm{mg}, 0.419 \mathrm{mmol}$ ), acetone ( 3.7 mL ), and water $(0.74 \mathrm{~mL})$. After $10 \mathrm{~min},(4 S)-2$-amino-4-(3-\{6-[(2S)-2,4-dimethylpiperazin-1-yl]-4-(piperazin-1-yl)pyridin2 -yl\}-1,2,4-oxadiazol-5-yl)-4-methyl-4,5,6,7-tetrahydro-1-benzothio-phene-3-carbonitrile ( $46,74.5 \mathrm{mg}, 0.14 \mathrm{mmol}$ ) in acetone/water ( $5: 1$, 0.5 mL ) was added, and the mixture was stirred for 15 min at RT. The mixture was concentrated, and the product was isolated via preparative RP HPLC (Agilent, Waters XBridge C18 $30 \times 50 \mathrm{~mm}$ $5 \mu \mathrm{~m}, 10-80 \% \mathrm{MeCN}$ in basic water), yielding $39 \mathrm{mg}(47 \%)$ of 23 as a white solid.
${ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.d_{6}, 600 \mathrm{MHz}\right): \delta 7.00(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, 1 \mathrm{H}, J=$ 1.8 Hz ), 6.76 (dd, $1 \mathrm{H}, J=10.5,16.8 \mathrm{~Hz}$ ), $6.12(\mathrm{~d}, 1 \mathrm{H}, J=1.7 \mathrm{~Hz}$ ), 6.08 (dd, $1 \mathrm{H}, \mathrm{J}=2.4,16.7 \mathrm{~Hz}$ ), $5.6-5.7(\mathrm{~m}, 1 \mathrm{H}), 4.49(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 4.01 (br d, $1 \mathrm{H}, J=12.7 \mathrm{~Hz}$ ), $3.62(\mathrm{br} \mathrm{d}, 4 \mathrm{H}, J=28.1 \mathrm{~Hz}), 3.32(\mathrm{br} \mathrm{s}$, $4 \mathrm{H}), 2.92(\mathrm{dt}, 1 \mathrm{H}, J=3.1,12.5 \mathrm{~Hz}), 2.76(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, J=10.8 \mathrm{~Hz})$, $2.63(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, J=11.0 \mathrm{~Hz}), 2.5-2.5(\mathrm{~m}, 2 \mathrm{H}), 2.12(\mathrm{~s}, 3 \mathrm{H}), 2.0-2.1$ $(\mathrm{m}, 2 \mathrm{H}), 1.7-1.9(\mathrm{~m}, 4 \mathrm{H}), 1.71(\mathrm{~s}, 3 \mathrm{H}), 1.06(\mathrm{~d}, 3 \mathrm{H}, J=6.6 \mathrm{~Hz})$.
${ }^{13} \mathrm{C}$ NMR (DMSO- $d_{6}$, 151 MHz ): $\delta$ 184.0, 168.9, 164.8, 164.1, $160.0,157.3,145.1,132.1,128.6,128.0,119.9,116.2,100.6,92.1$, 81.9, 60.4, 55.5, 47.2, 46.8, 46.6, 46.2, 44.9, 41.4, 39.6, 39.1, 37.3, 25.4, 24.1, 20.0, 14.3.

HRMS $(m / z):[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{30} \mathrm{H}_{37} \mathrm{~N}_{9} \mathrm{O}_{2} \mathrm{~S}$, 588.28637; found, 588.28629.
Protein Purification. Proteins of KRAS ${ }^{\text {G12C }}$, KRAS $^{\text {G12V }}$, KRAS ${ }^{\text {G12V-S39C }}$, and KRAS ${ }^{\text {G12V-C118S-S33C-BIT }}$ (uniprot reference ID P01116) were cloned, expressed, and purified, as previously described. ${ }^{13}$ Protein batches of BIT-modified proteins (indicated by -BIT) were obtained by addition of 1 H -benzimidazol-2-ylmethanethiol hydrochloride (BIT) to the unmodified protein at a concentration of $2 \mathrm{mg} / \mathrm{mL}$ in 50 mM Tris-HCL ( pH 8.5 ), 100 mM $\mathrm{NaCl}, 2 \mathrm{mM}$ GDP, and $5 \mathrm{mM} . \mathrm{MgCl}_{2}$. After overnight incubation at 4 ${ }^{\circ} \mathrm{C}$, the protein was transferred to a HiPrep Desalting column 26/10, washed with the BIT free buffer, and concentrated to $25-40 \mathrm{mg} / \mathrm{mL}$ for further usage.
Crystallization and Structure Determination. Crystals of KRAS ${ }^{\text {G12V,C118S,S33C-BIT }}$ in complex with (1) were obtained by cocrystallization mixing $0.2 \mu \mathrm{~L}$ of protein solution $[24.6 \mathrm{mg} / \mathrm{mL}$ protein in 20 mM 4 -(2-hydroxyethyl)-1-piperazineethanesulfonic acid
(HEPES) $\mathrm{pH} 7.5,150 \mathrm{mM} \mathrm{NaCl}, 5 \mathrm{mM} \mathrm{MgCl}, 2 \mathrm{mM}$ GDP, and 5 mM (1)] and $0.2 \mu \mathrm{~L}$ of reservoir solution (30\% PEG3350, 0.1 M bicine pH 8.5 , and 0.2 M ammonium sulfate) at $4^{\circ} \mathrm{C}$. Crystals of KRAS ${ }^{\mathrm{G112C}}$ in complex with 12 were obtained by co-crystallization using $0.2 \mu \mathrm{~L}$ of protein solution [ $40 \mathrm{mg} / \mathrm{mL}$ in 20 mM HEPES pH $7.5,150 \mathrm{mM} \mathrm{NaCl}, 3.2 \mathrm{mM}$ (12), and $3.2 \%$ DMSO] and $0.2 \mu \mathrm{~L}$ of reservoir solution [ 100 mM NaAc buffer $\mathrm{pH} 5,1.5 \mathrm{M}\left(\mathrm{NH}_{4}\right)_{2} \mathrm{SO}_{4}$ ] at $4^{\circ} \mathrm{C}$. The structure of KRAS ${ }^{\mathrm{G12V}, S 39 \mathrm{C}-\mathrm{BIT}}$ in complex with 20a was obtained using a soaking procedure. Crystals of KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}, \mathrm{~S} 39 \mathrm{C}-\mathrm{BIT}}$ were obtained using the hanging drop method by mixing $1 \mu \mathrm{~L}$ of protein solution ( $29 \mathrm{mg} / \mathrm{mL}$ in 50 mM Tris-HCL pH 8.5, 100 mM $\mathrm{NaCl}, 2 \mathrm{mM}$ GDP, and $5 \mathrm{mM} \mathrm{MgCl}{ }_{2}$ ) with $1 \mu \mathrm{~L}$ of reservoir solution ( 100 mM PCB buffer $\mathrm{pH} 5.9-6.2$ and $23-26 \%$ PEG1500, $2 \%$ at 20 ${ }^{\circ} \mathrm{C}$ ). Cubic crystals appeared within a week and were soaked overnight with 50 mM 20 a . Crystals of $\mathrm{KRAS}^{\mathrm{G} 12 \mathrm{C}}$ in complex with 23 were obtained by co-crystallization using $0.2 \mu \mathrm{~L}$ of protein solution ( $40 \mathrm{mg} / \mathrm{mL}$ in 20 mM HEPES $\mathrm{pH} 7.5,150 \mathrm{mM} \mathrm{NaCl}, 1 \mathrm{mM} 23$, and $1 \%$ DMSO) and $0.2 \mu \mathrm{~L}$ of reservoir solution ( $10 \%$ PEG8000, $10 \%$ PEG1000) at $4{ }^{\circ} \mathrm{C}$. All crystals were flesh frozen with $10-25 \%$ ethylene glycol before measurements. Details of data collection, data processing, and refinement (using standard procedures described in the Supporting Information) and the respective statistics are shown in Tables S1-S4. Final models were analyzed with MolProbity revealing 97.1/2.70/0.2 (1), 96.5/3.5/0/0 (12), 96.2/3.8/0 (20a), and 96.3/ 3.1/0 (23) of the residues in Ramachandran favored, allowed, and disallowed regions, respectively. The refined 2Fo - Fc electron density maps are shown in Figure S1. Structures are deposited in the Protein Data Bank (PDB, http://www.rcsb.org) with the following accession numbers (1) 7U8H, (12) 8AFC, (20a) 8AFD, and (23) 8 AFB .
$K_{\mathrm{D}}$ Determination by NMR Spectroscopy. Dissociation constants of the fragments were determined, as described previously ${ }^{14}$ by titrating compounds from a 50 mM DMSO- $d_{6}$ stock solution to a $70 \mu \mathrm{M}$ solution of KRAS. $4-5$ titration points were performed with individual NMR samples for every concentration and careful correction of the total DMSO- $d_{6}$ content. A minimum of three resonances of the protein were monitored, and the SD was calculated from the $K_{\mathrm{D}} \mathrm{s}$ of the individually monitored resonances (Figures S6 and S7). $K_{\mathrm{D}}$ values are indicated as $<10 \mu \mathrm{M}$ when the CSPs could no longer be reliably fitted because all CSPs were already close to saturation at the first titration point of $125 \mu \mathrm{M}$. This limitation is due to the relatively high protein concentration required in the NMR assay of $70 \mu \mathrm{M}$. $K_{\mathrm{D}}$ values are indicated as $>2000 \mu \mathrm{M}$ if we observed dose-responsive CSPs but they could not be fitted at the highest concentration used of $1000 \mu \mathrm{M} . K_{\mathrm{D}}$ values are indicated as $>10,000$ $\mu \mathrm{M}$ if we observed no CSPs at a ligand concentration of $500 \mu \mathrm{M}$.
$\boldsymbol{k}_{\text {inact }} / K_{\text {l }}$ Determination by Mass Spectrometry. Samples were prepared at a protein concentration of $1 \mu \mathrm{M}$ in HBS-N buffer, pH 7.4 , $1 \%$ DMSO. Compounds were added in concentrations between 0.78 and $100 \mu \mathrm{M}$, and incubation was performed at room temperature with incubation times between 30 s and 24 h . Samples were quenched by adding an equal volume of $4 \%$ formic acid, diluted with $0.1 \%$ formic acid to a final protein concentration of $0.2 \mu \mathrm{M}$, and immediately frozen until they were analyzed by liquid chromatography/MS (LC/ MS).

LC/MS analysis was performed on a Waters Xevo XS instrument equipped with an M-Class LC system. A Waters BEH C4 column ( 50 $\mathrm{mm}, 5 \mu \mathrm{~m}, 300 \AA$ ) was used for desalting ( $T=60^{\circ} \mathrm{C}$, flow rate: 40 $\mu \mathrm{L} / \mathrm{min}$ ). $5 \mu \mathrm{~L}$ of the sample was injected onto the column. Proteins were eluted with a 1.5 min gradient of $5-95 \%$ acetonitrile in $0.1 \%$ formic acid. The mass spectrometer was calibrated with direct infusion of sodium iodide solution ( 200 femtomoles $/ \mu \mathrm{L}$, SigmaAldrich) at a flow rate of $20 \mu \mathrm{~L} / \mathrm{min}$. A low-flow ESI probe was used, and instrument settings were as follows: 1 scan $/ \mathrm{s}$, mass range: $400-$ $3500 \mathrm{~m} / \mathrm{z}$, sensitivity mode, capillary voltage 3 kV , sampling cone: 70 V , source offset: 80 V , source temperature: $120^{\circ} \mathrm{C}$, and desolvation temperature: $400^{\circ} \mathrm{C}$.

Deconvolution of mass spectra was performed in Unifi using MaxEnt (batch mode). Ratios between unmodified and modified proteins were extracted from the deconvoluted spectra in Unifi and
fitted in GraphPad Prism (one-phase decay model). The obtained $k_{\text {obs }}$ values were used as input into the GraphPad Prism MichaelisMenten model and $k_{\text {inact }}$ and $K_{\mathrm{I}}$ values obtained. Mean values are reported with standard deviation ( $n=3$ independent experiments).

NCI-H358 Proliferation Assay. NCI-H358 (ATCC \#CRL-5807) cells were seeded in flat bottom 96 well microtiter plates (white PerkinElmer CulturePlate 96, cat. no. 6005689) in $90 \mu \mathrm{~L}$ RPMI/10\% fetal calf serum (FCS), $1 \%$ l-Glu, $1 \times$ NEAA, and $1 \times \mathrm{Na}$-Pyrovat at a density of 2000 cells/well. On day $2,10 \mu \mathrm{~L}$ of dilutions of the test compounds covering a concentration range between app. 1 and $10,000 \mathrm{nM}$ were added to the cells. Cells were incubated for 3 days in a humidified, $\mathrm{CO}_{2}$ controlled incubator at $37^{\circ} \mathrm{C} .100 \mu \mathrm{~L}$ of the cell titer glow reagent (CellTiter Glo Luminescent cat. no. G7571, Promega) was added to each well and incubated for additional 10 min at RT (with agitation). Luminescence was measured on a Wallac Victor using standard luminescence read out. $\mathrm{IC}_{50}$ values were calculated using standard Levenburg Marquard algorithms (GraphPad Prism).
GP2D and LS513 Proliferation Assay. Cells (GP2D or LS513 cells) were plated in 96 -well round bottom plates (Costar 7007 ULA Round Bottom 96 well) in $180 \mu \mathrm{~L}$ of medium (Dulbecco's modified Eagle's medium for GP2D; RPMI for LS513) containing 2\% FCS ( 1000 cells per well) and incubated overnight at $37^{\circ} \mathrm{C}$. The next day, $20 \mu \mathrm{~L}$ of appropriate serial compound dilutions in medium was added, spanning a concentration range of $10 \mu \mathrm{M}$ down to sub nanomolar concentrations. Cells were incubated for 5 days at $37^{\circ} \mathrm{C}$ in a $\mathrm{CO}_{2}$ incubator. After addition of $20 \mu \mathrm{~L}$ of the CTG reagent (Promega) and 10 to 30 min agitation, luminescence was measured. Data analysis (dose-response curves) was carried out using GraphPad Prism-based software.

Cell Line-Derived Efficacy and Biomarker Study in the NCIH358 Model. To establish the non-small cell lung cancer cell linederived NCI-H358 xenograft model, $7-8$ weeks old female NOG mice (NOD.Cg-Prkdcscid Il2rgtm1Sug/JicTac) with a bodyweight of 20 g from Taconic, Denmark were engrafted subcutaneously with 5 million NCI-H358 cells (ATCC \#CRL-5807) suspended in growth factor reduced, phenol red-free Matrigel (Corning). Mice were grouphoused (eight mice per cage) under pathogen-free and controlled environmental conditions ( $21 \pm 1.5{ }^{\circ} \mathrm{C}$ temperature, $55 \pm 10 \%$ humidity, and a 12 h light-dark cycle) and handled according to the internal institutional, Austrian governmental and European Union guidelines (Austrian Animal Protection Laws, ETS-123). All animal studies were approved by the internal ethics and the local governmental committee. For the biomarker study, 40 mice were injected subcutaneously with tumor cells into the right flank, and 20 of these were randomized to vehicle versus treatment arm based on tumor size once the tumors reached a mean size of $180 \mathrm{~mm}^{3}$ with $n=$ 5 mice per group and sampling timepoint ( $2,6 \mathrm{~h}$ ). Tumor-bearing mice were treated for 3 consecutive days with either vehicle or BI$047440 \mathrm{mg} / \mathrm{kg}$, and tumors were explanted and frozen to analyze biomarker modulation 2 and 6 h hours post last dose on the third day. For the efficacy study, 50 mice were injected subcutaneously with tumor cells into the right flank, and 24 mice were randomized based on tumor size into three groups of $n=8$ mice once the tumors reached a mean size of $160 \mathrm{~mm}^{3}$. In the efficacy study, a vehicle control arm was compared with two treatment arms, testing two dose schedules [BI-0474 $40 \mathrm{mg} / \mathrm{kg}$ once a week ( q 7 d ) and BI-0474 40 $\mathrm{mg} / \mathrm{kg}$ twice a week on 2 consecutive days ( qdx 2 )]. Tumor size was measured with an electronic caliper three times per week, and the results were converted to tumor volume $\left(\mathrm{mm}^{3}\right)$ by the formula: length $\times$ width $^{2} \times \pi / 6$. Body weight changes were assessed on a daily basis as a measure of tolerability. For statistical evaluation, a one-sided nonparametric Mann-Whitney-Wilcoxon $U$-test was used to compare treatment groups with the vehicle control. BI-0474 was formulated in $25 \%$ HP- $\gamma$-CD, and the vehicle control group was treated with $25 \%$ $\mathrm{HP}-\gamma$-CD once a week in the efficacy study and daily for 3 consecutive days in the biomarker study. All dose groups were treated intraperitonally (ip) using an application volume of $10 \mathrm{~mL} / \mathrm{kg}$. The studies described were not blinded.

Cell Lysis and Protein Lysate Generation. Approximately 5 $\mathrm{mm}^{3}$ pieces/tumors were excised from the bulk tumors and placed in lysis buffer. The tumor pieces were ground carefully before centrifugation. The protein concentration was measured according to manufacturer's recommendations (BRADFORD).

TO Measurements. KRAS ${ }^{\text {G12C }}$ TO was addressed by measuring the disappearance of the peptide mapping the Ras G12C locus upon drug binding in a targeted proteomics approach using Ras G12 wild-type- and G12C mutant-mapping peptides LVVVGAGGVGK and LVVVGAGCVGK and their stable isotope-labeled (SIL) variants as internal standards. Protein from tumor lysates was precipitated using the methanol-chloroform method. Air-dried pellets were reduced, alkylated, and digested with 5 mM tris(2-carboxyethyl) phosphine hydrochloride, 10 mM 2 -chloroacetamide, $1 \%$ sodium deoxycholate, SIL peptides ( $\mathrm{K}+8$, C alkylated, SpikeTide TQL, JPT Technologies), and trypsin in 50 mM triethylammonium bicarbonate pH 8 at $37^{\circ} \mathrm{C}$. Digests were acidified with TFA and supernatants purified by solidphase extraction (SepPak tC18, Waters). Eluted peptides were dried to completion, dissolved in $0.1 \%$ TFA, and injected into the LC/MS system. LC/MS analyses were performed on a nanoflow HPLC system (EASY-nLC II) interfaced to a triple quadrupole mass spectrometer (Quantiva) through a nano-ESI source (EASYspray). Nano-LC separation was performed with a gradient from 2 to $40 \%$ solvent B ( 80 : 20 acetonitrile: water, $0.1 \%$ formic acid) in solvent A ( $0.1 \%$ formic acid in water) in 120 min over a pre-column (Acclaim PepMap 100 C18, $5 \mu \mathrm{~m}$ particle size, $200 \mu \mathrm{~m} \times 2 \mathrm{~cm}$ ) connected in line with an analytical column (EASYspray C18, $2 \mu \mathrm{~m}$ particle size, 75 $\mu \mathrm{m} \times 50 \mathrm{~cm}$ ) in a vented column setup at $50^{\circ} \mathrm{C}$ analytical column temperature. MS data were acquired by multiple reaction monitoring of Ras G12-mapping peptides with Q 1 set to $[\mathrm{M}+2 \mathrm{H}]^{2+}$ and Q 3 to $\mathrm{y}^{7+}$ and $\mathrm{y}^{8+}$ for Ras G12 wild-type peptides and $\mathrm{y}^{7+}$, $\mathrm{y}^{8+}$, and $\mathrm{y}^{9+}$ for Ras G12C mutant peptides. All LC and MS instruments and LC columns were purchased from Thermo Fisher Scientific. Peak integration was performed with Skyline (University of Washington). Analytical process variance mapped by both SIL peptides was $10 \%$ CV across all samples; to account for additional inter-sample variance, endogenous/standard peak area ratios of the Ras G12C-mutant peptides were normalized to those of the Ras G12 wild-type peptide for each individual animal. Assuming no target engagement in the control animal groups, TO in the treated animals groups was then calculated as 1 - median(G12C/G12 ratio $)_{\text {treated animals }} /$ median(G12C/G12 ratio) control animals $($ Table S4).

RAS-GTP Measurements. The protein concentration was adjusted to $1 \mu \mathrm{~g} / \mu \mathrm{L}$ right before the assay performance (G-LISA Ras Activation Assay Biochem Kit, Cytoskeletion Inc.). Measurements were taken according to the manufacturer's recommendations. $25 \mu \mathrm{~g}$ of the sample was added per well of RAS GTP G-LISA plates. Absorbance was measured at 490 nm on a PerkinElmer Enspire Multilabel Reader.

Phospho/Total ERK1/2 Measurements. Measurements were performed according the manufacturers recommendations (Phospho/ Total ERK1/2 Whole Cell Lysate Kit, MSD). $20 \mu \mathrm{~g}$ of samples was added per well of the MSD plate. The plate was analyzed on an MSD SECTOR S600 Imager. Results were saved as the text file and converted to an Excel file. Averages of all pERK and total ERK ratios (two measurements per sample) were calculated for all samples.

Cleaved Caspase-3 Measurements. Measurements were performed according to the manufacturer's recommendations (Apoptosis Whole Cell Lysate Kit, MSD). $20 \mu \mathrm{~g}$ of samples were added per well of the MSD plate. The plate was analyzed on an MSD SECTOR S600 Imager.

## - ASSOCIATED CONTENT

## (s) Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jmedchem.2c01120.

Synthetic procedures for intermediates previously described in the literature; crystallographic data collection and refinement statistics; electron density
maps of ligands with the respective compounds from the crystal structures; raw data from in vivo TO measurements; HPLC traces for acrylamides used in biochemical and in vivo profiling; examples of HSQC-NMR spectra, titration curves, and experimental details on $K_{\mathrm{D}}$ determinations; $K_{\mathrm{D}}$ values for 20a and 20b measured on GDP•KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}}$, GDP•KRAS ${ }^{\mathrm{G} 12 \mathrm{~V}, \mathrm{~S} 39 \mathrm{C}-\mathrm{BIT}}$, GDP. $\mathrm{KRAS}^{\mathrm{G} 12 \mathrm{D}, \mathrm{S} 39 \mathrm{C}-\mathrm{BIT}}$, and $\mathrm{GppCp} \cdot \mathrm{KRAS}^{\mathrm{G} 12 \mathrm{D}}$; MS assay results of single assay runs and examples of hyperbolic fit curves for the determination of $k_{\text {inact }}$ and $K_{\mathrm{I}}$; GP2D and LS513 proliferation data for BI-0474 and AMG-510 and titration curves for NCI-H358 proliferation assay; single tumor volumes of individual animals from the NCI-H358 in vivo efficacy study (PDF)
Molecular formula strings (CSV)

## Accession Codes

The atomic coordinates of the KRAS crystal data have been deposited in the PDB and will be released upon article publication (PDB IDs: $7 \mathrm{U} 8 \mathrm{H}, 8 \mathrm{AFB}, 8 \mathrm{AFC}, 8 \mathrm{AFD}$ ).

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## Author Contributions

J.B. and A.G.W. contributed equally. J.B. supervised the BI medicinal chemistry team, designed compounds and synthetic strategies, and wrote the manuscript. A.G.W. supervised the VU medicinal chemistry team, designed and selected compounds, contributed to synthetic strategies, and wrote the manuscript. C.S. designed compounds and synthetic strategies and contributed to the manuscript. D.K. managed the BI/VU collaboration, performed crystallography, and contributed to the manuscript. J.B. performed crystallography and contributed to the manuscript. M.M. performed analytics, NMR spectroscopy, and contributed to the manuscript. G.G. performed analytics and NMR spectroscopy. J.P. performed crystallography and contributed to the manuscript. A.L. and J.R.A. designed compounds and synthetic strategies. Q.S. performed NMR spectroscopy and crystallography. M.G. supervised the BI biology team, performed in vitro assays, and contributed to the manuscript. D.R. performed in vivo studies and contributed to the manuscript. H.A. performed TO modeling and contributed to the manuscript. K.R. performed $k_{\text {inact }} / K_{\mathrm{I}}$ measurements and contributed to the manuscript. F.S. performed PD studies and contributed to the manuscript. T.G. performed AlphaScreen measurements and contributed to the manuscript. N.M. performed in vivo TO measurements and contributed to the manuscript. M.T. designed compounds and synthetic strategies and contributed to the manuscript. L.H., T.W., A.G., H.W., and A.M. contributed to the manuscript.
O.K. supervised the BI medicinal chemistry team during hit finding and contributed to the finding strategy and compound design. D.B.M. was responsible for the medicinal chemistry strategy and reviewed the manuscript. S.W.F. was responsible for the finding and fragment optimization strategy and reviewed the manuscript.

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## Notes

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${ }^{\S}$ This author is recently deceased.

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## - ABBREVIATIONS

BMIMCl, 1-butyl-3-methyl-imidazolium-chloride; DBU, 1,8-diazabicyclo[5.4.0]undec-7-ene; DCM, dichloromethane; DIPEA, diisopropyl ethyl amine; DMF, dimethyl formamide; EtOAc, ethyl acetate; HATU, O-(7-azabenzotriazol-1-yl)$N, N, N^{\prime}, N^{\prime}$-tetramethyluronium hexafluorophosphate; ip, intraperitoneal; NP, normal phase; PD, pharmacodynamic; PE, petroleum ether; RP, reversed phase; RT, room temperature; TGI, tumor growth inhibition; TO, target occupancy

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