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Data Article

Properties of particle phases for metal-matrix-composite design



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ABSTRACT

Successful metallurgical design of metal-matrix-composites relies on the knowledge of the intrinsic property profiles of the metal matrix and especially the compounds employed for particles, whiskers or fibres. In this work we compiled the key properties melting point, bulk modulus, shear modulus, Young's modulus, density, hardness, Poisson's ratio and structure/space group from the widespread literature data for the most relevant compound types, i.e. borides, carbo-borides, carbides, oxides, nitrides and intermetallic phases.

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Specifications Table

Subject area	<i>Physical metallurgy, Material Science, Engineering</i>
More specific subject area	<i>Metal-matrix-composites</i>
Type of data	<i>Table</i>
How data was acquired	<i>Literature survey</i>
Data format	<i>Raw, processed</i>
Experimental factors	–
Experimental features	–
Data source location	–
Data accessibility	<i>Data are accessible in this article</i>

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Value of the data

- The comprehensive data collection allows straightforward comparisons of individual properties, types and groups of compounds.
- Readily obtainable ratios of properties allow judging particles concerning their suitability for specific design goals (such as the material stiffness/density ratio).
- Specific effects of particles on the properties of the bulk composite can be estimated, for example regarding the co-deformation of particles and matrix influenced by their crystallographic coherency, or the amount of particles required for a specific gain in stiffness.

1. Data

Metal-matrix-composites allow overcoming the specific limitations of metallic and ceramic materials by blending their typically mutually exclusive property profiles. Knowledge based design of the composites requires, depending on the desired property profile and application, the choice of suitable metallic matrices and particles characterized by their intrinsic properties. In the following table the intrinsic properties (melting point, bulk modulus (B), shear modulus (G), Young's modulus (E), density, hardness, Poisson's ratio and structure/space group) of different types of phases (borides, carbo-borides, carbides, oxides, nitrides and intermetallics) are compiled from literature sources. The reference for each value or range of values is listed next to it on the right. Unless specified otherwise, values were assumed to have been determined experimentally as specifications are in most cases not given in the listed references. Densities determined by X-ray diffraction (XRD) are enclosed in curved brackets {}. Theoretically determined values are marked with a star *. Furthermore, the main selection criteria brittleness (expressed by the B/G ratio; B/G values below 1.75 are considered to represent 'more brittle' compounds [1]) and specific modulus (i.e. the $E/\text{density}$ ratio) have been derived. If more than one value is given for E and density of a compound, i.e. several values from one reference or diverging values from different references, the $E/\text{density}$ ratio is given as a range. In case of several values listed for B and G , the determined B/G ratio was chosen conservatively using the lowest B and highest G value, respectively (Table 1).

Table 1

Phase	Melting point / °C	Ref.	Bulk modulus / GPa	Shear modulus / GPa	Ref.	B / G	Young's modulus / GPa	Ref.	Density / g cm⁻³	Ref.	Specific modulus / GPa cm³ g⁻¹	Hardness / GPa	Ref.	Poisson's ratio	Ref.	Space group / structure	Ref.		
Borides																			
AlB₂	975	[2]	190*	95*	[3]	2.00	244.4*	[3]	{3.19} 2.9 2.955 2.7	[4] [5] [6] [7] [2]	76.6 - 90.5	23.6	[4]	0.29*	[3]	P6/mmm P6/mmm	[3] [2]		
BeB₂	> 1970	[2]	215*		[8]				2.32 - 2.48			31.2	[9]			P6/mmm	[2]		
BeB₆	~ 1700 ~ 2100	[6] [2]							2.35 / {2.33}	[2]		25.3	[9]			P43212	[2]		
CrB	2300 1515 / 1550 +50 ~ 2060	[6] [4] [2]							6.05	[4]		20.9	[4]			Cmcm	[2]		
CrB₂	2050	[6]							6.04 / {6.11}	[2]		19.2 - 22.9	[10]						
	2280 1960 / 1900 / 1850+-50 2200+-50 2100	[2] [4] [6] [7]	239.2*	139.9*	[3]	1.71	415.4*	[3] [4]	211 {5.6} 5.22 / {5.60}	[4] [6] [7]	34.0 - 79.6	17.7 20.3 - 22.5 20.6	[4] [10] [6]	0.26*	[3]	P6/mmm hex (c-32 type)	[2] [4]		
CrB₄ Cr₃B₂	1400 - 1600 1960 1960	[9] [4] [7]	265*	261*	[11]	1.02	312*	[12]	210.9 235.6 6.2	[9] [7] [7]		~ 38.0 [7]		48.0*	[11]	0.12*	[11]	orthorhombic	[4]
ErB₂			137.4*	119.7*	[3]	1.15	278.3*	[3]	-		41.5			0.16*	[3]	P6/mmm	[2]		
Fe₂B	1389 1389 ~ 1390 1410	[13] [2] [4] [9]	249.7* 194* 331* 253*	60.2* 67* 152.8* 177*	[14]	4.15 2.90 2.17 1.43	290 190* 184* 475.71*	[13] [14] [14] [15]	7.15 ~ 7.0	[13] [2]	25.7 - 41.4	13.1+-0.5	[9]			I4/mcm	[2]		
FeB₄			264.73*	197.97*	[15]							24.2*	[11]	0.2*	[15]				
GdB₂			131.2*	113.5*	[11]	1.16	264.3*	[3]						0.16*	[3]	P6/mmm	[2]		

Legend: *theoretical value; {} XRD;

Table 1 (continued)

Phase	Melting point / °C	Ref.	Bulk modulus / GPa	Shear modulus / GPa	Ref.	B / G	Young's modulus / GPa	Ref.	Density / g cm⁻³	Ref.	Specific modulus / GPa cm³ g⁻¹	Hardness / GPa	Ref.	Poisson's ratio	Ref.	Space group / structure	Ref.
NbB₂	2900	[4]	286.3*	210.4*	[3]	1.36	507*	[3]	6.4 / 6.5 / 6.6 / 6.97 / {7.21}	[4]	70.4 - 105.8	16.7 / 25.4	[4]	0.2*	[3]	hex (c-32 type)	[4]
	~ 3000	[2]					637.5	[9]	6.6	[2]			[9]			P6/mmm	[2]
	3000	[6]					676.8	[7]	6.97 / {7.00}	[9]		25.5	[9]				
NbB₄ NiB	3000	[7]							7.2	[7]							
	990	[2]	243*	194*	[11]	1.25										P4/mmbm	[2]
									7.13	[2]						Cmcm	[2]
NpB₂ PrB₆	> 2250	[2]	206.7*	169.8*	[3]	1.22	399.3*	[3]		[4]							
									6.5 / {7.13}	[6]							
									-							Pm3m	[2]
PuB₂	1825	[9]	207.4*	174.1*	[3]	1.19	408.1*	[3]	{12.674} / {11.1} / {12.81}	[9]	31.9 - 36.8	-		0.17*	[3]		
	< 1200	[2]							{3.667} / {3.67}	[2]						P6/mmm	[2]
ScB₂	2250	[2]	243.8*	256.6*	[3]	0.95	431*	[3]	3.65 / {3.667} / {3.67}	[9]	117.4 - 118.1	17.5+-2.7	[9]	0.11*	[3]		
	2250	[6]							{12.674} / {11.1} / {12.81}	[2]						P6/mmm	[2]
	3000 - 3150	[4]	128*	110.8*	[3]	1.16	258*	[3]	11.7 / {12.6}	[4,6]	20.4 - 40.4	24.9 / 25.6+-1.2 / 16.7	[4]	0.16*	[3]	hex (c-32 type)	[4]
TmB₂ TaB₂	3037	[9]					472.5*	[3]	12.38 / {12.62}	[9]		19 - 25	[17]			hex	[17]
	3040	[17]							12.54	[17]		24.5+-0.4	[9]			P6/mmm	[2]
TbB₂ TiB			131.3*	115.4*	[3]	1.14	267.8*	[3]	-	[9]		26.5- 27.5	[9]	0.16*	[3]	P6/mmm fcc	[2]
									5.09 / {4.565}							F43m	[2]

TiB₂	3225 2980 / 2900 +80 2900	[18] [4]	240 250.3* 240.4	255 260.7* [21]	[19] [3]	0.94 0.96 0.99	565 530 366	[19] [13]	4.52 4.5 4.5	[18] [19]	81.0 - 129.9	25.0 25 - 35	[19] [18]	0.108 0.11*	[19] [3]	hex P6/mmm hex	[19] [18]	
												33.3 / 33.0 / 26.6 / 25.3	[4]	0.109 / 0.11	[22]	hex (c-32 type)	[4]	
	2790	[9]					581*	[3]	4.5 / 4.52 / {4.52}	[4]		25 - 33	[17]			P6/mmm	[2]	
	3225	[17]					594* / 569	[22]	4.5	[20]		33.0+-0.6	[9]			hex	[17]	
	2800 2980	[2] [7]					370 529.6 551	[20] [9] [17]	4.52 4.38 4.5 / {4.52}	[17] [2] [6]						hex	[20]	
							529.6 535.5	[6] [7]	4.5	[7]				0.11	[17]			
TiB₄ TmB₂ UB₂		226* 137.5* 205.5* 209.5*	190* 120.5* [3]	[11]	1.19 1.14 0.98	279.7* 469.6	[3]			119 [7]	32.2*	[11]		0.16* 0.12*	[3] [3]	P6/mmm	[2]	
VB	~ 2250	[2]									36.9 - 37.0	13.6 14.8	[4] [9]				orthorhombic / Cmcm	[2,4]
VB₂	2040 - 2160	[4]	279.5*	240.9*	[3]	1.16	562.2*	[3]	4.61 / 5.28 / {5.10}	[4]	50.7 - 122.0	20.4	[4]	0.16*	[3]	hex (c-32 type)	[4]	
	~ 2400	[2]					267.7	[9]	5.06 - 5.28	[9]		27.5+-0.1	[9]			P6/mmm	[2]	
	2400+-50	[6]							4.61 / {5.10}	[2]								
	2100	[7]							5.28 / {5.10}	[6]								
									5.1	[7]								
VB₄ W₂B	2770+-80	[4]	241* 322.5*	237* 164.1*	[11] [3]	1.02 1.97	420.9*	[3]	16 / {10.72} 17.17 / 16 / 15.98 / {16.72}	[6] [4]	24.5 - 39.3	45.2* 23.5	[11] [6]	0.28*	[3]	tetragonal	[4]	
YB₂	2100	[2]	173.5*	145.3*	[3]	1.19	340.8*	[3]	{3.370} {2.91}	[9] [6]	101.1 - 117.1			0.17*	[3]	P6/mmm	[2]	
YbB₂ YbB₆	1538 +-33	[4]	153.7*	130.2*	[3]	1.18	304.6*	[3]	5.45 / {5.56}	[4]		25.5	[9]	0.17*	[3]	Pm3m	[2]	

Table 1 (continued)

Phase	Melting point / °C	Ref.	B Bulk modulus / GPa	G Shear modulus / GPa	Ref.	B / G	E Young's modulus / GPa	Ref.	Density / g cm ⁻³	Ref.	Specific modulus / GPa cm³ g⁻¹	Hardness / GPa	Ref.	Legend: *theoretical value; {} XRD;			
														Poisson's ratio	Ref.	Space group / structure	Ref.
ZrB	> 2000 2922 / 2992 +50	[9] [4]							5.7 / {6.7}	[2,4,6]		69 - 72 HRA 35 - 36	[4]		cubic / Fm3m	[2,4]	
ZrB ₂	3000 2980 / 2990 +50 / 3040 +50	[18] [4]	220 245	225 243	[16] [16]	0.98 1.01	350 498 - 638	[18] [4]	6.1 {6.09} / {6.102}	[18] [4]	55.6 - 104.8	22 - 26 23+-0.9	[9] [18] [16]	0.13* 0.144	[3] [4]	hex hex (c-32 type)	[18] [4]
	2990	[20]	238.6*	231.4*	[3]	1.03	524.6*	[3]	6.1	[20]		15.3 / 21.6 / 87 - 89 HRA	[4]	0.14	[22]	hex	[17]
	3200	[9]	238* - 276*	239* - 260*	[16]	0.92 -	343.2	[9]	6.1	[17]		25.3 - 28.0	[17]	0.09 - 0.28	[22]	P6/mmm	[2]
	3245	[17]	207.6	192.2	[4]		554 / 502	[16]	6.17 / {6.09}	[2]		22.1	[6]	0.109 / 0.13	[16]		
	3040	[2]					520* - 555*	[16]	6.1	[7]		22.1+-0.2	[9]	0.137 - 0.144*	[16]		
	3060	[7]					350 500	[20] [17]						0.11	[17]		
							518.5	[7]				~ 85 [7]					
ZrB ₄			199*		[11]												
Carbo-Borides																	
Hf ₂ BC			207	150	[23]	1.38	362*	[23]									
Mo ₂ BC	2800	[2]	313 324	181 185 - 188	[23] [21]	1.73 1.72	466 - 473	[21]	8.71	[2]	53.5 - 54.3			0.26	[21]	Cmcm	[2]
							1.75										
Nb ₂ BC			259	163	[23]	1.59	404*	[23]									
Ta ₂ BC			286	168	[23]	1.70	421*	[23]									
Tl ₂ BC			208	158	[23]	1.32	378*	[23]									
V ₂ BC			260	178	[23]	1.46	435*	[23]									
W ₂ BC			350	184	[23]	1.90	468*	[23]									

Zr₂BC		187	128	[23]	1.46	312*	[23]								
Carbides															
B₄C	2350	[13]	247*	200*	[11]	1.24	448	[13]	2.52	[18]	177.8 - 191.1	37 - 47	[18]	orthorhombic	[18]
	2450	[18]	175 (graph)		[24]		450	[18]	2.52	[20]		30 (graph)	[24]	orthorhombic	[20]
	2450	[20]					472*	[11]	2.51 / 2.484 / 2.47	[2]		31.7*	[11]	R3m	[2]
Cr₃C₂	2450	[2]					450	[20]							
	2420	[7]													
	1800	[13]					371	[13]	6.74	[13]	55.0 - 55.8	17.7	[9]	orthorhombic	[4]
	1830 - 1890	[4]					370 @ 449 °C	[25]	6.68 - 6.7	[4]				Pnma	[2]
	1895	[9]					372.7	[9]							
	~ 1900	[2]													
	1985	[7]													
Cr₄C	1510	[7]													
Cr₇C₃	1782	[2]	311.7* / 309*	143.9*	[14]	2.15	371 / 374	[14]	6.9	[2]	53.8 - 54.2	18.5	[9]	P31c	[2]
Fe₃C	1780	[7]													
HfC	3000–3900	[13]	259.2*	119.6*	[14]	2.17	177	[26]	7.4	[4]	23.9 - 24.8 - 23.8	24.8 - 31.4	0.26	[26]	[13]
							317	[13]	12.2	[13]					
	3890	[20]					400	[20]	12.3	[20]		26.0	[17]	cubic	[20]
	3900	[17]					352.1	[9]	12.76	[17]		29.0+-3.0	[27]	fcc	[17]
	3890	[2]					352	[17]						Fm3m	[2]
Mo₂C	2410	[28]					228	[13]	8.9	[13]	24.8 - 59.9	14.7+-1.3	[9]	P63/mmc	[2]
	2522	[2]					530 @ 390 °C	[25]	9.04 / {9.18}	[9]					
							533.5	[9]							
NbC	1900	[13]					338	[13]	7.6	[13]	43.2 - 72.7	23.0+-3.0	[27]	B1 fm3m	[29]
	3613	[9]					492.93 - 549.66	[30]	7.56 / {7.82}	[9]				Fm3m	[2]
							338.3	[9]	7.8	[7]					
	3600	[2]					540 @ 474 °C	[25]							
	3775	[7]					350 - 500	[31]							
							546	[7]			70.0 [7]				
Nb₂C	3100	[28]							7.86 / {7.85}	[9]				Pnma	[2]
	3035	[9]							6.7	[7]					
PKD	2675	[7]												cubic	[33]
							540*	[32]	0.82	[33]	3.51	[33]	70.0	[32]	[33]

Table 1 (*continued*)

V₂C	2750 2200 2187 2150	[7] [28] [9] [7]						{5.75} 5.8	[9] [7]								
WC	2800–2860	[13]	577	[38]		669	[13]	15.63	[13]	32.8 - 47.1	20 – 24	[18]	0.31 @ 347 °C	[25]	hex	[18]	
	2600 2867+–50 / 2870 / 2900 / 2777 / 2867 / 2627	[18] [4]				720 519 / 539.8 / 601.2 / 668.2 / 706.7	[18] [4]	15.7 15.60 / 15.63 / 15.7 / {15.8}	[18] [4]		21.6 17 / 23.5 / 18.3 / 18.4 / 92 HRA	[9] [4]		hex Fm3m	[4] [2]		
	2780 2785	[20] [2]				730 696.3	[20] [9]	15.7 15.5 – 15.7 / {15.77}	[20] [9]					hex	[20]		
	2777	[7]				700 @ 347 °C	[25]	15.7	[7]								
ZrC	3400	[13]	223.1	169.7	[36]	1.31	359	[13]	6.73	[13]	28.3 - 83.8	25.5 / 27.8 – 34.1 / 21 / 20.5 / 92.5 HRA	[4]	0.197	[36]	B1 fcc	[4]
	3532+–125 / 3532 / 3530 / 3550 / 3540 / 3175	[4]	214.2	124	[9]	1.31	406.2	[36]	6.606*	[36]		27.0	[17]	0.257	[9]	Fm3m	[2]
	3420	[2]					195.1 / 317.8 / 337.8 / 479.9	[4]	6.9 / {6.661. 6.73. 6.70. 6.44}	[4]		30+–3.0	[27]				
	3420 3400	[20] [17]					390 550 @ 505 °C	[20] [25]	6.6 6.56	[20] [17]				cubic fcc	[20] [17]		
	3525	[7]					348.1 348 408	[9] [17] [7]	6.8	[7]							
Oxides																	
Al₂O₃	2045	[13]	264.5	156.6	[39]	1.69	379 @ 1090 °C	[13]	3.98	[13]	99.2 - 104.1	20.7	[40]	0.27	[41]	hex	[18]
	2046.7+–8	[40]	136.67*	165	[34]	0.83	395.8	[39]	3.97	[40]		18 – 21	[18]		[40]	[40]	

Table 1 (continued)

Table 1 (continued)

Phase	Melting point / °C	Ref.	Bulk modulus / GPa	Shear modulus / GPa	Ref.	B / G	Young's modulus / GPa	Ref.	Density / g cm⁻³	Ref.	Specific modulus / GPa cm³ g⁻¹	Hardness / GPa	Ref.	Poisson's ratio	Ref.	Space group structure	Ref.
VO₂	2800	[20]	162	74.1	[39]	2.19	162.8 - 245.18	[40]	10.37	[39]	17.6 - 18.6						
	2875	[7]							10.97	[20]							
	1545	[40]							4.4	[4]						monoclinic	[40]
	2376	[40]							4.87	[4]						hex	[40]
	1470	[40]							7.16	[4]						tetra.	[40]
Y₂O₃	1470	[7]														monoclinic	
	2450	[20]	148.9+-3	69.2+-2	[51]	2.15	179.8 +-4.8	[51]	5.02	[51]	34.1 - 41.0	6.8	[40]	0.299	[51]	cubic	[40]
			135.7	66.5	[39]	2.04	171.5	[39]	5.03	[39]				0.298	[39]		
			141.5*		[39]		180	[20]	4.5	[20]							
ZnO	1975	[40]	134	44	[39]	3.05	199+-2							0.351	[39]	tet - mono @ 1170 °C	[52]
			143.6	45.5	[42]	3.16	119	[39]								hex	[40]
ZrO₂	2690	[40]	137.68*	75	[34]	1.84	190	[34]	5.56	[40]	30.3 - 35.4	16.6	[40]	0.27	[34]	monoclinic	[40]
	2700	[7]		77.9	[39]		197	[39]	6.27	[40]						cubic	[40]
									5.75	[7]							
Nitrides																	
AlN	2300	[20]	159.9 - 207	126.4	[53]	1.64	350	[20]	3.25	[20]	90.5 - 110					hex	[20]
	2375	[7]					350	[9]	3.2	[54]							
BC₂N							294.2 - 323.6 / 343.7	[7]	3.2	[7]							
							352										
	2730	[7]	408	445	[11]	0.92	980	[24]									
	2973		400	405	[11]	0.99	909	[24]	2.1	[7]				0.096	[24]		[24]
BN									3.45 - 3.48	[33]	261.2 - 263.5	48*	[11]	0.121	[55]	cub (hex)	[24]
														0.119	[15]	cubic [F 4 3 m]	[33]

															61 (graph)
		376	383.67	[21]	0.98 [21]	921	[15]								
		415 (graph)	405	[24]	1.02										
hex-BN															
CrN															
HfN	3385	[17]													
NbN	3225	[7]													
	2300	[9]													
	2330	[7]													
Si₃N₄	1900	[13]	290	120	[13]	2.42	295	[34]	3.18	[13]	66.7 - 92.8	15.5	[18]	0.29	[34]
	1900	[24]		115	[34]		220	[20]	3.2	[7]					
	1900	[7]							3.2-3.3	[54]					
	1900	[20]								[20]					
TaN	2700	[17]													
	3075	[7]													
TiN	2930	[13]	295	212.23*	[21]	1.39 [21]	600	[13]	5.44	[13]	46.2 - 115.4	21.0+-3.0	[27]		
	2950	[20]	320*		[37]		260	[20]	5.4	[20]					
	2950	[9]					251.1	[9]	5.43 / {5.44}	[9]					
	2950	[17]							5.39	[17]					
VN	2900	[7]													
	2050	[9]													
ZrN	2200	[7]													
	2980	[20]													
	2980	[9]													
	3950	[17]													
	2950	[7]													
Intermetallics															
Be₁₂Ti	1593	[2]	117	128	[56]	0.91	282	[56]	2.3	[2]	122.6			0.099	[56]
CoAl	1635	[56]	162	114	[56]	1.42	278	[56]						D2b tl26 P6/mmm	[56] [2]
CoSi₂	1326	[2]	210.1	67.5	[56]	3.11	182.9	[56]	4.94 / {4.95}	[9]	36.9 - 37.0	5.2	[80Sam] [9]	0.214	[56]
														B2cP2 Fm3m	[56] [2]

Table 1 (continued)

Phase	Melting point / °C	Ref.	B Bulk modulus / GPa	G Shear modulus / GPa	Ref.	B / G	E Young's modulus / GPa	Ref.	Density / g cm⁻³	Ref.	Specific modulus / GPa cm³ g⁻¹	Hardness / GPa	Ref.	Legend: *theoretical value; {} XRD;			
														Poisson's ratio	Ref.	Space group / structure	Ref.
CrSi₂	1277	[6]	172	153.3	[56]	1.12	354.6	[56]	4.91 4.91 / {4.978}	[2] [9]	71.1 - 72.1	6.9	[9]	0.156	[56]	P6222	[2]
	1475	[2]			[6]				6.648)	[9]							
Fe₃Al FeAl	2050	[56]	209.7	191.1	[56]	1.10	261 384	[56]	5.9 - 6.3 / {6.24}	[9]	46.7	12.95 - 15.2	[6]	0.151	[56]	B2cP2 tet	[56]
		[20]			[6]				5.585 6.2	[6]	60.3 - 71.5		[9]				[18,56]
MoSi₂	2030	[20]	173	77.3	[56]	2.24	201.9	[56]	{7.293}	[9]							[20]
	2030	[6]			[56]	2.11	421.7	[20] [6]		[20]		6.9	[9]				[2]
Ni₃Al Ni₃Fe NiAl	1638	[56]	180.6	85.5	[56]	2.37	221.4	[56]	5.85 9.1 9.1 8.83 / {9.1}	[13] [2] [9]	31.5	9.8 - 11.8	[9]	0.305 0.296 0.315	[56]	I4/mmm	[56]
		[56]			[56]		184.1	[56]			37.1 - 38.3		[9]				[56]
TaSi₂			166	70	[56]		338	[13]									P6222
Ti₃Sn TiAl			97.5	41.9	[56]	2.33	110	[56]	3.84 / {3.63}	[9]	45.1 - 47.7	1.8	[9]	0.312 0.234	[56]	L10 tP4	[56]
			110	70	[56]	1.57	173	[56]		[9]			[9]				[56]
TiAl₃	1375	[56]	105.6	93	[56]	1.14	215.7	[56]	3.31 / {3.371}	[9]	64.0 - 65.2	6.7	[9]	0.16	[56]		[56,57]
TiCr₂ TiSi₂	1550	[56]	159	71	[56]	2.24	184	[56]		[6]	59.0 - 69.1	6.78	[9]	0.31 0.189	[56]	C14. hP12 Fddd	[56]
		[56]	148.9	116.7	[56]	1.28	277.8	[56]	4.39 / {4.13}	[9]			[9]				[6]
Ti₅Si₃ VSi₂	2150	[7]	166 - 167.2	142 - 147.9	[56]	1.12	331 - 342.6	[56]	4.02 / {4.043} 4.2 4.34 / {4.627}	[7] [9]	71.5 - 78.9	8.7 - 9.4	[9]	0.158	[56]		[56]
		[7]			[56]					[7] [9]			[9]				
V₅Si₃ WSi₂	2180	[7]	222.4	203.6	[56]	1.09	467.9	[56]	5.1 9.25 / {9.857}	[7] [9]	47.5-50.6	10.5	[9]	0.149	[56]		[56]
		[7]			[56]					[7] [9]			[9]				
YAl₂ YFe₂	1645	[56]	89.2	65.5	[56]	1.36	158	[56]	{3.933}	[9]	40.2			0.205	[56]	C15 cF24	[56]
		[56]			[56]	1.25	127	[56]		[9]						C15 cF24	[56]
ZrAl₂	1645	[56]	97	49.6	[56]		222	[56]		[9]				0.184	[56]	C14. HP12	[56]

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References

- [1] S.F. Pugh, XCII. Relations between the elastic moduli and the plastic properties of polycrystalline pure metals, *Lond. Edinb. Dubl. Philos. Mag.* 45 (367) (1954) 823–843.
- [2] P. Eckerlin, H. Kandler, *Structure Data of Elements and Intermetallic Phases*, Springer Berlin Heidelberg, 1971.
- [3] Y.H. Duan, Y. Sun, Z.Z. Guo, M.J. Peng, P.X. Zhu, J.H. He, Elastic constants of AlB₂-type compounds from first-principles calculations, *Comput. Mater. Sci.* 51 (1) (2012) 112–116.
- [4] P.T.B. Shaffer, *Plenum Press Handbooks of High-Temperature Materials - No. 1 Materials Index*, Springer, US, 1963.
- [5] U. Burkhardt, V. Gurin, F. Haarmann, H. Borrmann, W. Schnelle, A. Yaresko, Y. Grin, On the electronic and structural properties of aluminum diboride Al_{0.9}B₂, *J. Solid State Chem.* 177 (2) (2004) 389–394.
- [6] G.V. Samsonov, *No.2 Properties Index*, Springer, US, 1964.
- [7] N. El Mahallawy, M.A. Taha Hanafi, Reinforcement considerations for high temperature metal matrix composites, *Key Eng. Mater.* 79–80 (1993) 1–14.
- [8] A.K.M.A. Islam, F.N. Islam, M.S. Iqbal, A.F. Jalbout, L. Adamowicz, Elastic and electronic properties of BeB₂ in comparison to superconducting MgB₂ and NbB₂, *Solid State Commun.* 139 (6) (2006) 315–320.
- [9] G.V. Samsonov, I.M. Vinitskii, *Handbook of Refractory Compounds*, Springer, US, 1980.
- [10] S. Okada, T. Atoda, I. Higashi, Y. Takahashi, Preparation of single crystals of a new boride Cr₂B₃ by the aluminium-flux technique and some of its properties, *J. Less Common Met.* 113 (2) (1985) 331–339.
- [11] H.Y. Niu, J.Q. Wang, X.Q. Chen, D.Z. Li, Y.Y. Li, P. Lazar, R. Podloucky, A.N. Kolmogorov, Structure, bonding, and possible superhardness of CrB₄, *Phys. Rev. B* 85 (14) (2012) 144116.
- [12] Y. Pan, Y.H. Lin, M. Wen, Q.N. Meng, Correlation between hardness and pressure of CrB₄, *RSC Adv.* 4 (109) (2014) 63891–63896.
- [13] Y. Feng, Strengthening of steels by ceramic phases, *Fak. Georesourcen und Materialtechnik* (2013) (RWTH Aachen).
- [14] B. Xiao, J.D. Xing, J. Feng, C.T. Zhou, Y.F. Li, W. Su, X.J. Xie, Y.H. Cheng, A comparative study of Cr₇C₃, Fe₃C and Fe₂B in cast iron both from ab initio calculations and experiments, *J. Phys. D: Appl. Phys.* 42 (11) (2009) 115415.
- [15] M. Zhang, M. Lu, Y. Du, L. Gao, C. Lu, H. Liu, Hardness of FeB₄: Density functional theory investigation, *J. Chem. Phys.* 140 (17) (2014) 174505.
- [16] X.H. Zhang, X.G. Luo, J.C. Han, J.P. Li, W.B. Han, Electronic structure, elasticity and hardness of diborides of zirconium and hafnium: First principles calculations, *Comp. Mater. Sci.* 44 (2) (2008) 411–421.
- [17] P.B. Narottam, *Handbook of Ceramic Composites*, Kluwer Academic Publishers, Springer, USA, 2005.
- [18] B. Basu, G.B. Raju, A.K. Suri, Processing and properties of monolithic TiB₂ based materials, *Int. Mater. Rev.* 51 (6) (2006) 352–374.
- [19] R.G. Munro, Material Properties of Titanium Diboride, *J. Res Natl. Inst. Stand.* 105 (5) (2000) 709–720.
- [20] H. Salmann, H. Scholze, *Keramik – Teil 2: Keramische Werkstoffe*, Springer, Berlin, Heidelberg, New York, 1983.
- [21] J. Emmerlich, D. Music, M. Braun, P. Fayek, F. Munnik, J.M. Schneider, A proposal for an unusually stiff and moderately ductile hard coating material: Mo₂BC, *J. Phys. D: Appl. Phys.* 42 (18) (2009) 185406.
- [22] H.Y. Wang, F.Y. Xue, N.H. Zhao, D.J. Li, First-principles calculation of elastic properties of TiB₂ and ZrB₂, *Adv. Mater. Res.* 150–151 (2010) 40–43.
- [23] H. Bolvardi, J. Emmerlich, M. to Baben, D. Music, J. von Appen, R. Dronskowski, J.M. Schneider, Systematic study on the electronic structure and mechanical properties of X₂BC (X = Mo, Ti, V, Zr, Nb, Hf, Ta and W), *J. Phys. : Condens. Matter* 25 (4) (2013) 045501.
- [24] V.L. Solozhenko, S.N. Dub, N.V. Novikov, Mechanical properties of cubic BC₂N, a new superhard phase, *Diam. Relat. Mater.* 10 (12) (2001) 2228–2231.
- [25] I.N. Frantsevich, A.B. Lyashenko, Young's moduli of the carbides of some transition metals, *Sov. Powder Metall. Met. Ceram.* 5 (7) (1966) 573–574.
- [26] H. Mizubayashi, S.J. Li, H. Yumoto, M. Shimotomai, Young's modulus of single phase cementite, *Scr. Mater.* 40 (7) (1999) 773–777.
- [27] O.H. Krikorian, Thermal expansivity correlations for refractory materials with the NaCl-type structure, *High Temp. - High. Press.* 20, 169–175.
- [28] E. Hornbogen, G. Eggeler, E. Werner, *Werkstoffe: und Eigenschaften von Keramik-, Metall-, Polymer- und Verbundwerkstoffen*, Springer Berlin Heidelberg, Berlin, Heidelberg, 2006.
- [29] K.B. Joshi, U. Paliwal, First-principles study of structural and bonding properties of vanadium carbide and niobium carbide, *Phys. Scr.* 80 (5) (2009) 055601.

- [30] C. Kral, W. Lengauer, D. Rafaja, P. Ettmayer, Critical review on the elastic properties of transition metal carbides, nitrides and carbonitrides, *J. Alloy. Compd.* 265 (1–2) (1998) 215–233.
- [31] U. Bohnenkamp, R. Sandstrom, Evaluation of the elastic modulus of steels, *Steel Res.* 71 (3) (2000) 94–99.
- [32] M. Hebbache, First-principles calculations of the bulk modulus of diamond, *Solid State Commun.* 110 (10) (1999) 559–564.
- [33] G.E. Spriggs, Properties of diamond and cubic boron nitride; Figs. 13.1 - 13.91, in: P. Beiss, R. Ruthardt, H. Warlimont (Eds.), *Materials - Powder Metallurgy Data, Refractory, Hard and Intermetallic Materials, Landolt-Börnstein - Group VIII Advanced Materials and Technologies*, Springer, Berlin, Heidelberg, 2002.
- [34] H. Salmang, H. Scholze, *Keramik - Teil 1: Allgemeine Grundlagen und wichtige Eigenschaften*, Springer Berlin, Heidelberg, 1982.
- [35] C.W. Nan, D.R. Clarke, The influence of particle size and particle fracture on the elastic/plastic deformation of metal matrix composites, *Acta Mater.* 44 (9) (1996) 3801–3811.
- [36] R. Chang, L.J. Graham, Low-Temperature Elastic Properties of ZrC and TiC, *J. Appl. Phys.* 37 (10) (1966) 3778.
- [37] R. Ahuja, O. Eriksson, J.M. Wills, B. Johansson, Structural, elastic, and high-pressure properties of cubic TiC, TiN, and TiO, *Phys. Rev. B* 53 (6) (1996) 3072–3079.
- [38] V.P. Zhukov, V.A. Gubanov, Energy band structure and thermo-mechanical properties of tungsten and tungsten carbides as studied by the LMTO-ASA method, *Solid State Commun.* 56 (1) (1985) 51–55.
- [39] R.G. Munro, *Elastic Moduli Data for Polycrystalline Ceramics*, National Institute of Standards and Technology, Gaithersburg, Maryland (2002) 20899.
- [40] G.V. Samsonov, *The Oxide Handbook*, 1973.
- [41] C. Industry, *Material Properties Charts*, 2013.
- [42] O.L. Anderson, J.E. Nafe, The bulk modulus-volume relationship for oxide compounds and related geophysical problems, *J. Geophys. Res.* 70 (16) (1965) 3951–3963.
- [43] Calcium Oxide (CaO) Young's, shear and bulk modulus, Poisson's ratio, in: O. Madelung, U. Rössler, M. Schulz (Eds.), II-VI and I-VII Compounds; Semimagnetic Compounds, Springer Materials.
- [44] Z.-W. Niu, Y. Cheng, X.-R. Chen, K. Xu, G.-F. Ji, Elastic and electronic properties of Ce₂O₃ from first principles, *J. Alloy. Compd.* 551 (0) (2013) 672–676.
- [45] S. Rekhi, L.S. Dubrovinsky, R. Ahuja, S.K. Saxena, B. Johansson, Experimental and theoretical investigations on eskolaite (Cr₂O₃) at high pressures, *J. Alloy. Compd.* 302 (1–2) (2000) 16–20.
- [46] H.G. Drickamer, R.W. Lynch, R.L. Clendenen, E.A. Perez-Albueene, X-Ray Diffraction Studies of the Lattice Parameters of Solids under Very High Pressure, in: S. Frederick, T. David (Eds.), *Solid State Physics*, 1967, pp. 135–228, Academic Press.
- [47] B.R.K. Gupta, R.P. Goyal, Static and thermophysical properties of chalcogenide crystals with NaCl structure, *Solid State Commun.* 49 (6) (1984) 559–562.
- [48] M. Catti, G. Valerio, R. Dovesi, Theoretical study of electronic, magnetic, and structural properties of α -Fe₂O₃ (hematite), *Phys. Rev. B* 51 (12) (1995) 7441–7450.
- [49] M. Magnesiumoxid, (<http://www.korth.de/index.php/material-detailansicht/items/22.html>)(Accessed 28 February 2017), 2017.
- [50] O.B. Shcherbina, M.N. Palatnikov, V.V. Efremov, Mechanical properties of Nb₂O₅ and Ta₂O₅ prepared by different procedures, *Inorg. Mater.* 48 (4) (2012) 433–438.
- [51] O. Yeheskel, O. Tevet, Elastic moduli of transparent yttria, *J. Am. Ceram. Soc.* 82 (1) (1999) 136–144.
- [52] J. Eichler, U. Eisele, J. Rödel, Mechanical Properties of Monoclinic Zirconia, *J. Am. Ceram. Soc.* 87 (7) (2004) 1401–1403.
- [53] Aluminum nitride (AlN) bulk modulus, Young's and shear modulus, in: O. Madelung, U. Rössler, M. Schulz (Eds.), Group IV Elements, IV-IV and III-V Compounds. Part a - Lattice Properties, Springer Materials.
- [54] M.G. Nicholas, D.A. Mortimer, L.M. Jones, R.M. Crispin, Some observations on the wetting and bonding of nitride ceramics, *J. Mater. Sci.* 25 (6) (1990) 2679–2689.
- [55] T. Taniguchi M.P. D'Evelyn, Elastic properties of translucent polycrystalline cubic boron nitride as characterized by the dynamic resonance method, *Diam. Relat. Mater.* 8 (8–9) (1999) 1522–1526.
- [56] J.H. Westbrook, R.L. Fleischer, *Intermetallic Compounds - Volume 1, Principles*, 1995.
- [57] T. Hong, T.J. Watson-Yang, A.J. Freeman, T. Oguchi, J. Xu, Crystal structure, phase stability, and electronic structure of Ti-Al intermetallics: TiAl₃, *Phys. Rev. B: Condens. Matter* 41 (18) (1990) 12462–12467.