



Certain Environmental Conditions Maximize Ammonium Accumulation and Minimize Nitrogen Loss During Nitrate Reduction Process by *Pseudomonas putida* Y-9

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Realizing the smallest nitrogen loss is a challenge in the nitrate reduction process. Dissimilatory nitrate reduction to ammonium (DNRA) and nitrate assimilation play crucial roles in nitrogen retention. In this study, the effects of the carbon source, C/N ratio, pH, and dissolved oxygen on the multiple nitrate reduction pathways conducted by Pseudomonas putida Y-9 are explored. Strain Y-9 efficiently removed nitrate (up to 89.79%) with glucose as the sole carbon source, and the nitrogen loss in this system was 15.43%. The total nitrogen decrease and ammonium accumulation at a C/N ratio of 9 were lower than that at 12 and higher than that at 15, respectively (P < 0.05). Besides, neutral and alkaline conditions (pH 7-9) favored nitrate reduction. Largest nitrate removal (81.78%) and minimum nitrogen loss (10.63%) were observed at pH 7. The nitrate removal and ammonium production efficiencies of strain Y-9 increased due to an increased shaking speed. The expression patterns of nirBD (the gene that controls nitrate assimilation and DNRA) in strain Y-9 were similar to ammonium patterns of the tested incubation conditions. In summary, the following conditions facilitated nitrate assimilation and DNRA by strain Y-9, while reducing the denitrification: glucose as the carbon source, a C/N ratio of 9, a pH of 7, and a shaking speed of 150 rpm. Under these conditions, nitrate removal was substantial, and nitrogen loss from the system was minimal.

Keywords: *Pseudomonas putida* Y-9, nitrate reduction, carbon source, C/N ratios, pH, dissolved oxygen, *nirBD* expression

HIGHLIGHTS

The roles of DNRA and assimilatory reduction during NO_3^- removal and nitrogen conservation in soils have been insufficiently examined. Moreover, the effects of environmental factors on the NO_3^- reduction process when the three NO_3^- reduction pathways (denitrification, DNRA, and assimilation) coexist remain unclear. In this study, the effect of the carbon source, C/N ratio, pH, and dissolved oxygen on ammonium accumulation and the expression of *nirBD* in strain Y-9 are explored during the nitrate reduction processes. The following conditions facilitated nitrate assimilation and DNRA by strain Y-9 while simultaneously reducing denitrification: glucose as the carbon source, a C/N ratio of 9, a pH of 7, and a shaking speed of 150 rpm. Under these conditions, nitrate removal was substantial, and nitrogen loss from the system was

OPEN ACCESS

Edited by:

Qiaoyun Huang, Huazhong Agricultural University, China

Reviewed by:

Jin-tian Li, South China Normal University, China Ang Li, Harbin Institute of Technology, China

*Correspondence:

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Specialty section:

This article was submitted to Microbiotechnology, a section of the journal Frontiers in Microbiology

Received: 25 August 2021 Accepted: 11 November 2021 Published: 13 December 2021

Citation:

Huang X, Tie WZ, Xie DT, Jiang DH and Li ZL (2021) Certain Environmental Conditions Maximize Ammonium Accumulation and Minimize Nitrogen Loss During Nitrate Reduction Process by Pseudomonas putida Y-9. Front. Microbiol. 12:764241. doi: 10.3389/fmicb.2021.764241 minimal. These findings provide theoretical support for technical studies of nitrate removal and nitrogen retention in soils.

INTRODUCTION

Large amounts of industrial fertilizers are often applied to crops to increase crop yields. This leads to the considerable accumulation of nitrate (NO_3^-) in the soil (Kraft et al., 2014). NO₃⁻, a mobile anion, is prone to loss by denitrification or runoff into surface waters, and this not only decreases the efficiency of nitrogen fertilizers but also has various passive environmental impacts including water eutrophication and greenhouse gas (nitrous oxide, N₂O) emissions (Beeckman et al., 2018; Li et al., 2018; Sánchez and Minamisawa, 2019; Xia et al., 2020). The ammonium (NH_4^+) , produced by the dissimilatory reduction of NO_3^- to NH_4^+ (DNRA) via microorganismal respiration, can be adsorbed by soil colloids and then utilized by crops (Song et al., 2014; Zhang et al., 2015; Pandey et al., 2019). Similarly, the microbial assimilatory reduction of NO₃⁻ can reduce NO₃⁻ to NH₄⁺ via NO₂⁻ catalyzed by the relative reductase. Then, the NH_4^+ is incorporated into biomolecules and used by the bacterium. After death, the microorganisms release the NH₄⁺ via mineralization for plant use (Shao et al., 2011; Wang et al., 2020). It is clear that DNRA and NO₃⁻ assimilation ease the accumulation of NO3- in agricultural soils and improve the efficiency of nitrogen fertilizers. These processes reduce the risk of NO₃⁻ loss and mitigate the adverse effects of nitrogen fertilizer use. Several recent studies have investigated the role of DNRA in soil nitrogen conservation in farmlands (Shan et al., 2016; Friedl et al., 2018). Yet, the important role of assimilatory NO₃⁻ reduction in NO₃⁻ removal and nitrogen conservation in soils has been comparatively neglected.

Several environmental factors, including the C/N ratio, oxygen concentration, carbon source, affect enzyme activity in microorganisms by controlling the expression of relevant genes (e.g., amoA, hao, narG, and nirK) and thus impacting nitrogen cycles (Szukics et al., 2010; Ke et al., 2013; Caranto and Lancaster, 2017; Yu et al., 2019). Thus, we speculated that soil $NO_3^$ removal could be maximized and soil nitrogen loss could be minimized by adjusting certain external environmental factors to enhance NO₃⁻ assimilation and DNRA while decreasing denitrification. Typically, higher C/N ratios favor DNRA over denitrification (Kraft et al., 2014; Yoon et al., 2015; Van den Berg et al., 2016; Putz et al., 2018). Some studies have reported that glucose addition improved the NO3⁻ assimilation capacity of the soil (Recous et al., 1990; Romero et al., 2015). However, the effects of environmental factors on the NO3⁻ reduction process when the three NO_3^- reduction pathways coexist are unclear.

Pseudomonas putida Y-9 performs NO_3^- assimilation, DNRA, and denitrification under aerobic conditions simultaneously. The gene *nirBD* has been shown to control the assimilation and DNRA process (Huang et al., 2020). In this study, we investigate the effects of the carbon source, C/N ratio, pH, and dissolved oxygen (DO) on the accumulation of ammonium in the medium and the expression of *nirBD* in strain Y-9 during

the nitrate reduction process. This study focuses on adjusting the environmental factor parameters to enhance the DNRA and NO_3^- assimilation of strain Y-9. The results will provide theoretical support for technical research on NO_3^- removal and nitrogen retention in soil.

MATERIALS AND METHODS

Microorganisms and the Culture Media

P. putida Y-9 (Genbank No. KP410740), which performs NO_3^- assimilation, denitrification, and DNRA under aerobic conditions simultaneously (Huang et al., 2020), was used in this study.

A denitrification medium (DM) was used to assess the nitrate reduction abilities of strain Y-9. The DM (per liter, pH = 7.2) contained 7.0 g K₂HPO₄, 3.0 g KH₂PO₄, 5.13 g CH₃COONa, 0.10 g MgSO₄ · 7H₂O, 0.72 g KNO₃, and 0.05 g FeSO₄ · 7H₂O. Luria-Bertani (LB) medium used for bacterial enrichment contained 10 g NaCl, 10 g tryptone, and 5 g yeast extract per liter (per liter, pH 7.0–7.2). All of the mediums were autoclaved for 30 min at 121°C.

Effects of the Different Factors on Nitrate Reduction

The preserved strain Y-9 bacteria were activated in the LB medium at 150 rpm and 15° C for 36 h. Cells in the logarithmic growth phase were inoculated into a DM medium to assess the effects of the carbon source, C/N ratio, pH, and DO on Y-9–driven NO₃⁻ reduction (Li et al., 2019; Yan et al., 2021).

In the carbon source experiments, one of the three carbon sources (sodium acetate, glucose, or sodium citrate) was added to 100 mL of DM medium. The C/N ratio, pH, and shaking speed were kept constant at 15, 7, and 150 rpm, respectively. In the C/N ratio experiments, 100 mL aliquots of the DM medium were amended with glucose to yield C/N ratios of 3, 6, 9, 12, or 15. The pH and shaking speed were kept constant at 7 and 150 rpm, respectively. In the pH experiments, the initial pH was adjusted using NaOH and HCl to 4, 6, 7, 8, or 9. The carbon source was glucose, and the C/N ratio and shaking speed were held constant at 9 and 150 rpm, respectively. To determine the effects of DO on NO_3^- reduction, the shaking speed was set to 0, 50, 100, 150, or 180 rpm according to previous studies (Ren et al., 2014; Lei et al., 2019; Chen et al., 2021; Yan et al., 2021). The carbon source was glucose, and the C/N ratio and pH were kept constant at 9 and 7, respectively. The cultures were incubated at 15°C for 4 d. All of the above experiments were performed in triplicate. Samples were taken every day from each system. The optical density at 600 nm (OD_{600}) , NH₄⁺, NO₃⁻, and total nitrogen (TN) were measured for each sample.

Kinetic Analysis of Nitrate Degradation

The modified Compertz model was used to describe the kinetics analysis of nitrate degradation by strain Y-9 (Chen et al., 2016). The kinetic equation was $y = y_0(1 - \exp(-\exp(\frac{eR_m}{y_0}(t_0 - t)1)))$, where y is the NO₃⁻ concentration at different incubation times (mg/L); y_0 is the initial concentration of NO₃⁻ (mg/L), R_m is the maximum conversion rate (mg/L/h), t_0 is the lag time (h), t is the reaction time (h), and e is the mathematical constant.

Expression of nirBD in Strain Y-9

Total RNA was extracted from strain Y-9 after 4 d of incubation under various conditions using а Trizol extraction United kit (Invitrogen, States), manufacturer's instructions. following the The specific primers B1/B2 (F: CGCAACCATCTGCTCGTGT; R: CTGGCGGGTGTAGGAAAAGT) were designed based on the nirBD gene sequence (GenBank, MK561362). These primers were used to amplify the *nirBD* gene from the isolates. The 16S rRNA gene was used as an internal standard, as structural rRNA is present in cells at reasonably constant levels under normal growth conditions (Edwards and Saunders, 2010). The 16S rRNA gene was amplified using the forward primer GAACGCTAATACCGCATACGTCC and the reverse primer ATCATCCTCTCAGACCAGTTAC. The total RNA was reversetranscribed using the RevertAid first-strand cDNA synthesis kit following the manufacturer's instructions. Real-time quantitative PCRs were performed using the SYBR® Premix Ex TaqTM II. Each real-time PCR was performed in triplicate. The PCR cycling conditions were as follows: initial denaturation at 95°C for 30 s; 38 cycles of 95°C for 15 s, 60°C for 30 s, and 72°C for 30 s; 1 cycle of 95°C for 15 s; and, finally, stepwise temperature increases from 55°C to 95°C to generate the melting curve. Standard curves were established using a dilution series of pMD19-T vectors containing the target gene.

Analytical Methods

The OD₆₀₀ was determined based on the absorbance at 600 nm, which was measured using a spectrophotometer. The contents of the different forms of nitrogen were determined as described by Huang et al. (2019). TN was measured in the suspension. The concentration of NH₄⁺, NO₃⁻, and NO₂⁻ was measured in the supernatant, which was obtained by centrifuging each sample at 8,000 rpm for 5 min. Three replicates were analyzed per sample, and the results are presented as means ± the standard deviation of the mean (SD). The TN and NO₃⁻ removal efficiencies were calculated as follows: $R_V = (T_1 - T_2)/T_1 \times 100$ %, where R_v



culture.

is the removal efficiency of TN or NO_3^- (%), and T_1 and T_2 are the initial and final concentrations of TN or NO_3^- in the system, respectively.

Statistical Analyses

One-way analyses of variance (ANOVAs), followed by Duncan's Multiple Range Tests were performed using SPSS 22, and the differences among means were considered statistically significant at P < 0.05. Graphs were drawn using Origin 8.6 and GraphPad Prism 6.

RESULTS AND DISCUSSION

Effects of the Carbon Source on Nitrate Reduction

A carbon source is typically essential for the growth of heterotrophic microorganisms, and it acts as an electron donor for nitrogen cycling (Sun et al., 2016). In this study, strain Y-9 grew vigorously and reached the stationary cell growth phase after 2 d when sodium acetate, glucose, or sodium citrate was used as the sole carbon source (Figure 1A). Moreover, sodium acetate, glucose, and sodium citrate were suitable carbon sources for NO3- removal, with removal efficiencies of 74.75, 89.79, and 100%, respectively, at 4 d (Figure 1B). These results were consistent with those of Guo et al. (2016), who reported that sodium acetate, glucose, and sodium citrate enhanced the NO₃⁻ removal capacity of Enterobacter cloacae strain HNR. Furthermore, the nitrate degradation rate followed the modified Compertz model ($R^2 > 0.90$), and the maximum NO₃⁻ conversion rates were 1.60, 4.92, and 44.35 mg/L/h in media containing sodium acetate, glucose, and sodium citrate, respectively (Table 1).

Previous results have shown that strain Y-9 performs DNRA and nitrate assimilation under aerobic conditions (Huang et al., 2020). Based on the duration of cultivation (4 d), the detectable $\rm NH_4^+$ in the supernatant might have resulted from DNRA and nitrate assimilation followed by mineralization (**Figure 1C**). It is worth noting that after 3 d of cultivation, when all the cells were in the stationary phase, the detectable $\rm NH_4^+$ began to decrease in the glucose-containing system. However, it continued to increase in the media containing sodium acetate and sodium citrate. These results demonstrated that $\rm NO_3^-$ reduction by strain Y-9 differed when glucose was the sole carbon source compared to the other two carbon sources.

The TN decreases in our system were due to the denitrification activities of strain Y-9, and *nirBD* in strain Y-9 controls DNRA and nitrate assimilation (Huang et al., 2020). The TN in the media supplemented with different carbon sources tended to decrease (**Figure 1D**). The maximum TN decrease (60.27 mg/L) was found in the sodium citrate-containing medium, and the minimum TN decrease occurred (22.77 mg/L) in the glucose medium. This was in accordance with data from Yang et al. (2012) who reported that *Pseudomonas stutzeri* D6 most effectively removed TN when sodium citrate was the carbon source. Moreover, the *nirBD* expression level in strain Y-9 peaked when glucose was the carbon source (**Figure 2**). These findings demonstrated that glucose

addition promoted DNRA and nitrate assimilation, effectively removing most of the NO_3^- from the system (up to 89.79%) while inhibiting denitrification (i.e., the total nitrogen lost from the system was 22.77 mg/L).

Effects of the C/N Ratio on Nitrate Reduction

The effects of the C/N ratio on the nitrate reduction conducted by strain Y-9 were further studied. Strain Y-9 growth improved as

 TABLE 1 | Kinetic parameters and final removal efficiency for the degradation of nitrate by strain Y-9 under different environmental conditions.

	Environmental factor	R _m (mg/L/h)	t ₀ (h)	R ²	The last nitrate removal efficiency (%)
Carbon sources	Sodium acetate	1.6	20.97	0.94	74.14
	Glucose	4.92	23.99	0.97	89.79
	Sodium citrate	44.35	23.82	1	100
C/N	3	0.44	9.35	0.69	30.46
	6	0.58	-21.1	0.64	54
	9	1.91	5.17	0.9	81.78
	12	5.44	16.99	0.99	100
	15	3	11.68	0.97	89.79
рН	4	-1.63	-3177.13	3-3.21	4.99
	6	0.8	-7.7	0.74	58.27
	7	2.05	4.87	0.90	81.78
	8	1.91	10.99	0.87	80.57
	9	1.5	6.04	0.87	79.14
Shaking speed	0	1.33	38.7	0.97	69.9
	50	1.03	6.756	0.93	71.47
	100	1.41	3.4	0.83	71.18
	150	1.90	3.17	0.89	81.78
	180	1.68	26.96	0.93	76.17

In C/N, pH and shaking speed, glucose was chosen as carbon source.





the C/N ratio increased (**Figure 3A**). This result was consistent with previous studies (Kim et al., 2008; Liu et al., 2016), that the growth of *P. putida* AD-21 and *Marinobacter* strain NNA5 increased as the relative proportion of carbon increased in the medium. This might have been because electron transfer slowed when carbon concentrations were low, providing insufficient energy for microbial growth (Kim et al., 2008; Zhao et al., 2018). Greater than 80% of the NO₃⁻ was removed at C/N ratios of 9–15. However, the removal efficiency of NO₃⁻ did not exceed 30.46 and 54.00% when the C/N ratio was 3 and 6, respectively (**Figure 3B**). Furthermore, nitrate degradation rates at C/N ratios of 9–15 were consistent with the predictions of the modified Compertz model ($R^2 > 0.90$), and the NO₃⁻ conversion rate was maximum at a C/N ratio of 12 (**Table 1**).

The decrease of TN in this system generally mirrored the change in NO_3^- (Figure 3D). The reduction in TN at the extremely high C/N ratio of 15 was lower than the reduction in TN at a C/N ratio of 12, suggesting that a C/N ratio

of 12 was optimal for denitrification. Our results indicated that the influences of C/N ratios on Y-9-driven denitrification agreed with many previous studies. They showed that extremely low or high carbon concentrations suppressed microorganismal denitrification (Kim et al., 2008; Guo et al., 2016; Zhao et al., 2018). NH₄⁺ concentration in the supernatant initially increased and then decreased during NO₃⁻ reduction (**Figure 3C**), consistent with our carbon source analysis (**Figure 1C**). When the C/N ratio was 9, strain Y-9 removed most of the NO₃⁻ (removal efficiency 81.78%) via DNRA and NO₃⁻ assimilation (*nirBD* in strain Y-9 was most strongly expressed (**Figure 4**)). Notably, the denitrification performance of strain Y-9 at a C/N ratio of 9 was significantly weaker than that at a C/N ratio of 12 (P < 0.05).

Effects of the Initial pH on Nitrate Reduction

The impacts of the initial pH on the nitrate reduction performance of strain Y-9 are shown in Figure 5. At an



culture with glucose as the sole carbon source.



initial pH of 4, the bacterial density did not noticeably increase, and NO_3^- reduction was minimal throughout the experiment (**Figures 5A,B**), suggesting that an overly acidic

environment was detrimental to these bacteria. However, in the pH range 7–9, strain growth and NO₃⁻ removal were significantly improved (P < 0.05) (**Figures 5A,B**). These results are in agreement with the general finding that neutral or alkaline environments are beneficial for bacteria growth and bacterium-driven NO₃⁻ removal (Li et al., 2017; Rout et al., 2017). The NO₃⁻ removal efficiency was significantly positively correlated with the growth of strain Y-9 (P < 0.01) (**Figure 5**), indicating that pH might control the NO₃⁻ removal efficiency by influencing the growth of strain Y-9. However, this possibility requires further study. The TN concentration in the suspension decreased as the initial pH increased, and the TN decreased by 35.01 mg/L at pH 9 (**Figure 5D**). This indicated that alkaline environments favored the denitrification in strain Y-9 under aerobic conditions.

After 3 d of culture, a negligible amount of NH_4^+ was detected at pH 4. However, the accumulation of NH_4^+ at pH 7–9 was higher than 5.0 mg/L (**Figure 5C**). At the end of the experiment, the *nirBD* expression level in strain Y-9 at pH 7–9 was better than at pH 4 or 6 (**Figure 6**). These results showed that the initial pH affected the expression of *nirBD* in strain Y-9, and this might influence NH_4^+ production from DNRA and NO_3^- assimilation as well



FIGURE 5 | Effects of pH on the OD₆₀₀ (A), NO₃⁻ concentration (B), NH₄⁺ concentration (C), and TN concentration (D) in the *Pseudomonas putida* Y-9 culture with glucose as the carbon source.



as its subsequent mineralization (Huang et al., 2020). The results of previous studies on the effects of pH on NO_3^- reduction by soil microorganisms are widely contradictory

(Nägele and Conrad, 1990; Stevens et al., 1998). Here, strain Y-9 effectively performed NO_3^- assimilation, DNRA, and denitrification at pH 7–9. This finding was inconsistent with a previous study (Yoon et al., 2015) that suggested that a low pH was more favorable for denitrification, while a high pH promoted the production of NH_4^+ via DNRA. This discrepancy indicated that the effects of pH on the microbial nitrogen cycle were complex and required further study. Our results suggested that a neutral pH was most favorable for NO_3^- removal and nitrogen retention.

Effects of Dissolved Oxygen on Nitrate Reduction

Strain Y-9 growth and NO₃⁻ reduction increased gradually as the shaking speed increased (**Figures 7A,B**). Changes in the NO₃⁻ degradation rates at various rotating speeds were consistent with the predictions of the modified Compertz model ($R^2 > 0.80$), and the NO₃⁻ conversion rate achieved its maximum at 150 rpm (**Table 1**). These results suggested that increasingly aerobic conditions improved strain growth and NO₃⁻ reduction. TN decreased gradually throughout the incubation process, irrespective of the DO concentration.



FIGURE 7 | Effects of shaking speed on the OD₆₀₀ (A), NO₃⁻ concentration (B), NH₄⁺ concentration (C), and TN concentration (D) in the *Pseudomonas putida* Y-9 culture with glucose as the sole carbon source.

However, TN decreased at low shaking speeds (≤50 rpm) was significantly greater than those at high shaking speeds $(\geq 100 \text{ rpm})$ (P < 0.05). The decrease in the TN at the end of the experiment reached the maximum (34.58 mg/L) at a shaking speed of 50 rpm (Figure 7D). These results suggested that the denitrification performance of strain Y-9 first increased and then decreased as the DO concentration increased. This finding was consistent with the results of Zhao et al. (2018); Rout et al. (2017), and Huang and Tseng (2001). Previous studies demonstrated that the denitrification performance remained stable as long as the DO concentration remained within a fixed range. Nevertheless, the denitrification enzyme activity levels improved noticeably when the DO concentration decreased below a threshold value (Song et al., 2011). For strain Y-9, 50 rpm might be the threshold DO value that affects denitrification enzymes, although this possibility requires further testing.

The quantitative PCR amplification results indicated that the expression of nirBD increased with an increase in the shaking speed (Figure 8). These results, in conjunction with the NO₃⁻ reduction performance of strain Y-9 (Huang et al., 2020), indicated that high DO concentrations stimulated the expression of *nirBD* in strain Y-9, promoting NO₃⁻ assimilation as well as DNRA, and thus releasing more NH_4^+ into the supernatant (Figure 7C). Consistent with this, Yang et al. (2012) and Zhao et al. (2018) found that NH₄⁺ production increased with the DO content during NO₃ reduction by P. stutzeri D6 and P. stutzeri strain XL-2. Variance analyses indicated that the amounts of $NO_3^$ removal from the culture media and NH4+ accumulated in the culture media at high shaking speeds (>100 rpm) differed obviously from those at low rotation speeds (\leq 50 rpm) (P < 0.05). These results indicated that good aeration effectively promoted NO₃⁻ removal and NH₄⁺ production by strain Y-9.





The heavy application of chemical nitrogen fertilizers leads to an accumulation of highly mobile nitrate in upland soils and significantly increases the risk of nitrogen loss (Lin et al., 2020; Vidal et al., 2020). Therefore, it is essential to control NO₃⁻ concentrations in soil. The nitrogen cycle conducted by microorganisms plays a critical role in regulating nitrate concentrations in soil, compared to artificially limiting the application of ammonium and nitrate fertilizers (Shao et al., 2011; Song et al., 2014; Zhang et al., 2015; Pandey et al., 2019; Wang et al., 2020). Denitrification effectively removes excess NO3from soil systems but leads to nitrogen losses in the form of nitrogen gas or the greenhouse gas N₂O (Stein and Klotz, 2016). For example, Putz et al. (2018) showed that approximately 70-78% of all N₂O originated from denitrification in annual cereal soils. Both the DNRA and NO3⁻ assimilation processes can decrease soil NO3⁻ concentration and facilitate soil nitrogen conservation by reducing NO_3^- to NH_4^+ via NO_2^- (Kuypers et al., 2018; Wang et al., 2020). Thus, to pursue the minimum loss of nitrogen and maximize nitrogen fertilizer efficiency, strategies that strengthen DNRA as well as NO3⁻ assimilation while weakening denitrification in surface soils should be pursued. Previously, we found that strain Y-9 performs simultaneous nitrate assimilation, DNRA, and denitrification under aerobic conditions. It has also been clarified that the gene, nirBD, controls NO3- assimilation and DNRA process in strain Y-9 (Huang et al., 2020). In this study, we further explored the environmental factors that affect the nitrate removal pathways of strain Y-9. Our results provide a theoretical reference for technical studies of nitrate removal and nitrogen conservation in farmland soils.

CONCLUSION

Four common external environmental conditions (carbon source, C/N ratio, pH, and dissolved oxygen) affected the nitrate reduction performance of strain Y-9. A high initial pH enhanced nitrate assimilation, denitrification, and the DNRA of strain Y-9.

The optimal conditions for the nitrate assimilation and the DNRA of strain Y-9 were glucose as the carbon source, C/N 9, pH 7.0, and 150 rpm. Under these conditions, the nitrogen loss from the system was the smallest.

DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/**Supplementary Material**, further inquiries can be directed to the corresponding authors.

AUTHOR CONTRIBUTIONS

XH and ZL: conceptualization. XH: methodology, data curation, visualization, supervision, and writing—original draft preparation. WT: software, formal analysis, and investigation.

XH, ZL, and DX: validation. ZL: resources and project administration. XH, ZL, DJ, and DX: writing—review and editing. XH and ZL: funding acquisition. All authors have read and agreed to the published version of the manuscript.

FUNDING

This work was supported financially by the National Natural Science Fund of China (42107333 and 42077217).

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ACKNOWLEDGMENTS

We thank LetPub for its linguistic assistance during the preparation of this manuscript.

SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fmicb. 2021.764241/full#supplementary-material

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