Total Synthesis of Viniferifuran, Resveratrol-Piceatannol Hybrid, Anigopreissin A and Analogues – Investigation of Demethylation Strategies

Duc Duy Vo^a and Mikael Elofsson^{a,*}

^a Department of Chemistry, Umeå University, SE90187, Umeå, Sweden E-mail: Mikael.Elofsson@umu.se

Received: October 4, 2016; Published online: December 8, 2016

Supporting information for this article can be found under: http://dx.doi.org/10.1002/adsc.201601089.

© 2016 The Authors. Published by Wiley-VCH Verlag GmbH & Co. KGaA. This is an open access article under the terms of the Creative Commons Attribution-NonCommercial License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited and is not used for commercial purposes.

Abstract: Resveratrol-based natural products constitute a valuable source of unique compounds with diverse biological activities. In this report we investigate demethylation strategies to minimize formation of cyclized and dimerized products during the synthesis of viniferifuran and analogues. We found that boron trichloride/tetra-*n*-butylammonium iodide (BCl₃/TBAI) is typically more effective than boron tribromide (BBr₃). Based on these findings we car-

Introduction

Stilbenoids of plant origin constitute a rich source of polyphenolic compounds with intriguing structures and diverse biological activities.^[1] To date the structures and absolute stereochemistry for several hundred higher order stilbenoids have been determined. Most of these often very complex natural products are formed by oligomerization of the key building block resveratrol. The large number of reports describing the positive effects of stilbenoids on human health have instigated wide interest in the chemistry and biology of resveratrol and its oligomers. The mode of action on the molecular level is, however, known for only a few stilbenoids, for example, diptoindonesin G (13, Figure 1) that was recently identified as an inhibitor of CHIP E3 ubiquitin ligase.^[2] Due to scarce quantity of many higher order stilbenoids in nature, intense efforts have been made to develop synthetic methods employing both biomimetic and de novo strategies. Biomimetic strategies usually involve metal- or enzyme-catalyzed oxidative couried out the first syntheses of dehydro- δ -viniferin, resveratrol-piceatannol hybrid and anigopreissin A. In addition, we have developed a short and efficient route to viniferifuran that was obtained in 13% yield over six steps.

Keywords: demethylation; natural products; polyphenols; resveratrol oligomers; stilbenoids; total synthesis

plings of stilbenes but these protocols typically result in product mixtures and poor yields.^[1d,3] Recent improvements resulted in protocols for the preparation of racemic pallidol and quadrangularin by dimerization of protected resveratrol derivatives in excellent vields.^[4] However, biomimetic methods do not readily allow alteration of the core structures or substituent patterns, for example, to determine structure-activity relationships for bioactive stilbenoids. To address this Snyder and co-workers developed de novo methods allowing the preparation of a large number of resveratrol dimers and higher order stilbenoids.^[5] These impressive studies spawned a number of syntheses applying innovative approaches for the preparation of racemic compounds.^[5,6] The first enantiototal synthesis of a stilbene dimer was the case of δ -viniferin (3) where Shaw and co-workers elegantly applied a carbene-insertion chemistry using a rhodium catalyst to construct the dihydrobenzofuran core asymmetrically.^[7] Our interest in this intriguing family of natural products stems from the finding that the resveratrol tetramer (-)-hopeaphenol blocks type III secretion,





Figure 1. Selected natural and non-natural resveratrol dimers and the stilbenoid diptoindonesin G.

an indispensable virulence system, in the Gram-negative pathogens Yersinia pseudotuberculosis and Pseudomonas aeruginosa.^[8] While (-)-hopeaphenol can be obtained in gram quantities from natural sources^[9] we focused on the *de novo* synthesis of partial structures and recently published total syntheses of (\pm) -ampelopsin B (6)^[10] and (\pm) - ε -viniferin (1),^[10] the key intermediate to form almost, if not all, higher order of stilbene structures (Figure 1).^[1] To further advance our understanding of structural features that affect inhibition of type III secretion, we have continued to prepare resveratrol dimers. In this study we describe total syntheses of viniferifuran (2),^[11] its isomer anigopreissin A (11)^[12] and the recently reported resveratrol-piceatannol hybrid (12)^[13] (Figure 1). Viniferifuran is a competitive inhibitor of syk kinase (IC₅₀ = 62 nM)^[14] and this mechanism is believed to be key for the anti-inflammatory activity of viniferifuran. Anigopreissin A was recently discovered as an inhibitor of HIV-1 reverse transcriptase (IC₅₀ = $8 \mu M$) including two mutant enzymes resistant to the clinical drug nevirapine.^[12b] Viniferifuran (2) is a highly substituted 7-hydroxy-5-styryl-2,3-diarylbenzofuran structure that can be prepared in four steps starting with biomimetic dimerization of resveratrol to form (\pm) - ε viniferin (1).^[15] Subsequent transformations including a key oxidative dehydrogenation by DDQ furnished viniferifuran (2).^[11] In our hands viniferifuran could be obtained in gram quantity in 11% yield over four steps (Supporting Information, SI-1). In addition, we prepared a number of viniferifuran analogues and investigated conditions for demethylation of phenols, a critical transformation in the total synthesis of resveratrol oligomers.

Results and Discussion

Despite the number of methods available to construct the 2,3-diarylbenzofuran core,^[16] only a few attempts have been made towards the total synthesis of viniferifuran. Kraus and Gupta^[17] and Chen et al.^[18] applied strategies relying on cyclization and subsequent dehydration of keto benzyl ethers by heating at high temperature (170 °C) with excess of the strong hindered phosphazene base P₄-*t*-Bu^[17] or by using excess of the strong base LiTMP followed by dehydratation using pTsOH.^[18] Kim and Choi^[19] on the other hand employed a more versatile and milder procedure to construct a 3-arylbenzofuran by cyclization of the corresponding β -aryloxy ketone using Bi(OTf)₃, followed by introduction of a 2-aryl substituent by Pd(OAc)₂-catalyzed direct arylation.

After generation of an aldehyde, the styryl moiety was then introduced by a Wittig–Horner reaction.^[19] However, when reinvestigating the strategy by Kim and Choi (Supporting Information, **SI-2**) the Wittig– Horner reaction in our hands only resulted in very low conversion of aldehyde **16a** to the styryl moiety **17a**. Using NaH instead of *t*-BuOK^[20] under microwave irradiation at 120 °C during 30 min produced permethylated viniferifuran **17a** in 80% yield. We then modified the route described by Kim and Choi^[19] to conveniently prepare permethylated vini-





Scheme 1. Modified synthetic route to viniferifuran. *Reagents and conditions:* a) LiAlH₄ (3 equiv.), THF, 0°C, 10 min. b) Dess–Martin periodinane (1.2 equiv.), DCM, 0°C, 1 h, 80–90%, 2 steps. c) ArBr or ArI (1.5–2 equiv.), Pd(OAc)₂ (0.1 equiv.), PCy₃·HBF₄ (0.2 equiv.), K₂CO₃ (1.5 equiv.), PivOH (3 equiv.), 100°C, 20 h, 74%. d) ArBr (1.5–2 equiv.), PdCl(C₃H₅)dppb (0.05 equiv.), KOAc (2 equiv.), DMA, 150°C, 20 h, 60%. e) Phosphonate derivatives (1.5 equiv.), NaH (3 equiv.), THF, 120°C, MWI, 30 min, 80%. f) BCl₃, TBAI, DCM, 0°C to room temperature, 6 h, 23% (HPLC).

ferifuran 17a and analogues (Scheme 1). In this modified route, the ester moiety in 14a was transformed to the aldehyde 15a before harnessing the direct aryla $tion^{[10]}$ to prepare intermediate **16a** that is now ready for the Wittig-Horner olefination to give 17a. Interestingly, we could exploit the direct arylation of 15a with a low loading of 5% of $PdCl(C_3H_5)(dppb)$,^[21] conditions that were found to be unsuccessful with the ester substrate ${\bf 14a}.$ Kraus and ${\bf Gupta}^{[15]}$ reported the final demethylation of 17a to form viniferifuran (2) in 67% yield using 1 M BBr₃ in CH_2Cl_2 at 0°C. However, when applying these conditions we obtained the cyclized product dehydroampelopsin B (9, Figure 1) as major product (46% yield after purification by HPLC) with only minor amounts of viniferifuran (not isolated) and mixtures of its dimers that were detected by LC-MS (Supporting Information, SI-2). Alternative protocols for demethylation were investigated and we found that BCl₃/TBAI^[22] gave 2 and 9 in a 1:1 ratio without formation of dimerized products. HPLC purification furnished pure 2 and 9, each isolated in 23% yield (Supporting Information, SI-2, and Table 1). We realized that the final demethylation would be a challenge when synthesizing analogues with variation of substituents on the A-D rings with the goal to allow elucidation of structure-activity relationships. This prompted us to investigate if BCl₃/ TBAI could be used as a general demethylation reagent. Interestingly, we also found that the Tf₂O-catalyzed cyclization of viniferifuran (2) gave racemic dehydroampelopsin B (9) in 50% yield (Supporting Information, SI-1). We hypothesize that this compound might exist in nature since it is the core of shoreaketone (7) and malibatol A (8), both natural products (Figure 1).^[1]

We applied the modified route and prepared a number of permethylated compounds 17b-p with



Scheme 2. Modified synthetic route to viniferifuran. *Reagents and conditions:* a) ArBr or ArI (1.5–2 equiv.), $Pd(OAc)_2$ (0.1 equiv.), PCy_3 ·HBF₄ (0.2 equiv.), K_2CO_3 (1.5 equiv.), PivOH (3 equiv.), 100°C, 20 h, 35–74%. b) ArBr (1.5–2 equiv.), $PdCl(C_3H_5)(dppb)$ (0.05 equiv.), KOAc (2 equiv.), DMA, 150°C, 20 h, 47–60%. c) Phosphonate derivatives (1.5 equiv.), NaH (3 equiv.), THF, 120°C, MWI, 30 min, 52–82%. d) BCl₃, TBAI, DCM, 0°C to room temperature, 6 h.

Adv. Synth. Catal. 2016, 358, 4085-4092



Substrate	\mathbb{R}^1	\mathbb{R}^2	R ³	Product	Product/cyclized form ^[d]	Yield [%] ^[a]
17a	3,5-diOMe	OMe	OMe	2	1:1	23 ^[b]
17b	3,5-diOMe	F	OMe	18 a	<3:7	$(< 10)^{[e]}$
17c	3,5-diOMe	COOH	OMe	18b	<3:7	$(< 10)^{[e]}$
17d	3,5-diOMe	OMe	Н	18c	>9:1	50
17e	3,5-diOMe	OMe	F	18d	1:0	$51^{[b]} (11)^{[c]}$
17f	3,5-diOMe	OMe	NO_2	18e	_	_
17g	3,5-diOMe	NO_2	Cl	18f	_	-
17h	3,5-diOMe	NH_2	Cl	18g	1:0	18 ^[b]
17i	2,5-diOMe	OMe	OMe	18h	1:0	46
17j	3-OMe	OMe	OMe	18i	7:3	24 ^[b]
17k	4-OMe	OMe	OMe	18j	1:0	19 ^[b]
171	$4-CF_3$	OMe	OMe	18k	1:0	54
17m	4-F	OMe	OMe	18 l	1:0	53 (16) ^[c]
17n	3-F	OMe	OMe	18m	1:0	55
170	3-Cl	OMe	OMe	18n	1:0	55
17p	Н	OMe	OMe	180	1:0	53

Table	1. Demethylation	on with BCl ₃ /	TBAI to aff	ford viniferifuran	analogues

^[a] Isolated yields.

^[b] Purified by HPLC.

^[c] Yield with BBr₃ as demethylation agent.

^[d] Ratio based on ¹H NMR of crude product.

^[e] Estimated yield, not isolated.

different substituents on rings B, C and D (Scheme 2, and Supporting Information, SI-3). The carboxylic esters 14a-i were transformed to aldehvdes 15a-i by a sequence of LiAlH₄ reduction followed by a Dess-Martin oxidation. Direct arylation of 15a-i was then carried out with a set of aryl bromides carrying different electron-withdrawing and electron-donating substituents. The methylated products 16b-m were obtained in 35–74% yield using 10 mol% (PdOAc)₂ or 47–60% using 5 mol% PdCl(C_3H_5)(dppb). Next, application of our improved conditions for the Wittig-Horner reaction to install the styryl moiety afforded the permethylated intermediates 17b-p in 52-82% yield. Finally, we investigated the impact of the different substituent patterns on demethylation and cyclization of 17b-p using BCl₃/TBAI in DCM at 0°C to room temperature for 6 h (Table 1).

Under these conditions all the starting materials were consumed and we anticipated that cyclization would be suppressed in compounds with less nucleophilic C rings. Indeed, **18h** (2,5-diOH) and **18i** (3-OH) were obtained in higher yields than viniferifuran (2) and minor amounts of the cyclized product was observed only with starting material carrying the 3-OMe group. This pattern was observed for other C-ring substituents (3-H, 3-F, 3-Cl, 4-F, 4-CF₃, 4-OMe) that furnished 18k-p in 19-55% yield. Modification of ring B with substituents such as 4-F or 4-COOH increased formation of the cyclized dehydroampelopsin B analogues 35 and 36, respectively (Supporting Information, SI-4) and 18a and b was observed as minor products (not isolated). We also found that removal of the 4-OMe group or replacement with a fluorine atom on ring D reduced the degree of cyclization and afforded **18c** and **18d** in 50 and 51% yields, respectively. Compounds **18e** and **18f** could not be isolated due to decomposition of the NO₂ group during the demethylation. However, **18g** could be obtained when NO₂ was replaced by NH₂. Several substrates were also demethylated with BBr₃ that typically resulted in lower yields than BCl₃/TBAI. In several cases, decomposition occurred or complex mixtures of dimerized products formed (Supporting Information, **SI-5**). Acid-catalyzed dimerization of stilbenes has been described using a variety of acids including BBr₃.^[3,23]

Our results indicate that BCl₃/TBAI is superior to BBr₃ for our purposes but for some compounds the yields are still low and in several cases HPLC was required to purify the compounds and remove excess TBAI. We therefore decided to explore alternative routes based on demethylation early in the synthetic sequence.

First we applied an alternative route where a set of D ring moieties were attached by applying the Wittig–Horner reaction on the early aldehyde intermediate **15a** to afford **20a–c** in 60–80% yield (Scheme 3). Interestingly, BBr₃-mediated demethylation could effectively occur in these cases and after acetylation the protected derivatives **21a–c** were isolated in 38–62% yield. Bromination with NBS and subsequent deacetylation furnished the bromo polyphenolic substrates **22a–c**. As a final step substituted B rings were introduced by Suzuki couplings to give the target compounds **18d** and **18p--s** in 41–76% yield (Table 2). Next we exploited the strategy previously developed for the total synthesis of $(\pm)-\varepsilon$ -viniferin.^[10]





Scheme 3. Alternative routes for the synthesis of viniferifuran and analogues. *Reagents and conditions:* a) Phosphonate derivatives (1.5 equiv.), NaH (3 equiv.), THF, 120 °C, MWI, 30 min, 60–80%. b) BBr₃ (9–12 equiv.), DCM, -78 °C to room temperature, 6 h. c) Ac₂O, THF, Et₃N, room temperature, overnight, 38–62%, 2 steps. d) NBS (3 equiv.), pTsOH, room temperature, 24–48 h. e) KOH, MeOH, 0 °C, 30 min, 15–30%, 2 steps. f) PdCl₂(dppf) DCM (5% mol), boronic acid (1.5–2 equiv.), Na₂CO₃, DME/H₂O (1/1), 70 °C, MWI, 30 min, 41–76%. g) 4-Bromoanisole (2 equiv.), Pd(OAc)₂ (0.1 equiv.), PCy₃·HBF₄ (0.2 equiv.), K₂CO₃ (1.5 equiv.), PivOH (3 equiv.), 100 °C, 20 h, 25–36%. h) Styrene derivatives (1.2 equiv.), Pd(OAc)₂ (0.1 equiv.), Pd(OAc)₂ (0.

Table 2. Obtained viniferifuran analogues by final Suzuki or Heck couplings.

Compound	\mathbb{R}^2	R ³	Yield [%] ^[a]
18p	NHAc	ОН	76
18d	OH	F	50
18q	F	F	55
18r	OH	Cl	41
18s	F	Cl	41
18t	OH	3-NO ₂ , 4-OH	25 ^[b]
18u	OH	3-OH, 4-NO ₂	58
18v	OH	4-COOH	60
18w	OH	$4-NH_2$	64
18x	OH	4-EtoCOCH ₂ O	54
viniferifuran 2	OH	4-OH	64

^[a] Isolated yield.

^[b] Purified by HPLC.

Direct arylation of the bromo derivative **19**, prepared in three steps from 1-bromo-3,5-dimethoxybenzene in 59% yield,^[10] installed ring B to give **23** in 36% yield and subsequent demethylation by BBr₃ produced the bromo polyphenolic substrate **24** in 96% yield (Scheme 3). A set of D rings was introduced by Heck couplings of the corresponding styrene derivatives gave viniferifuran (**2**) in 64% yield and target compounds **18t-x** in 25–64% yield (Table 2). This synthesis of viniferifuran is the shortest achieved to date and provides the target compound in 13% yield over 6 steps from 1-bromo-3,5-dimethoxybenzene. Importantly this strategy avoids formation of dimers or cyclized products and also allows introduction of, for example, nitro and carboxylic acid substituents (*cf.* **18t**– **x**) that are not compatible with a final demethylation.

Having an understanding of the different reactivity of BBr₃ and BCl₃/TBAI, we turned our attention to the total synthesis of the benzofuryl stilbene dimers anigopreissin A (**11**),^[12] a resveratrol-piceatannol hybrid (**12**)^[13] and dehydro- δ -viniferin (**4**)^[24] (Figure 1). The resveratrol-piceatannol hybrid was isolated from a *Vitis* extract after a bioguided fractionation using a hepatitis C virus replication inhibition assay.^[13] Dehydro- δ -viniferin was recently synthesized by laccase-biocatalyzed dimerization of resveratrol to form δ -viniferin (**3**) followed by dehydrogenation, a strategy limited to 4-hydroxystilbene starting materials.^[24]

The final demethylations by BBr₃ and BCl₃/TBAI were studied on permethylated intermediates **29**, **31** and **34** (Scheme 4). In order to prepare these intermediates, we constructed 2-arylbenzofuran derivatives by a one pot two-step Sonogashira cyclization applied to *o*-iodophenol derivatives containing a free CHO group using a rapid and efficient protocol recently de-





Scheme 4. Total synthesis of anigopreissin A, resveratrol-piceatannol hybrid and dehydro- δ -viniferin. *Reagents and conditions:* a) PdCl₂(PPh₃)₂ (5% mol), CuI (3% mol), ethynylanisole (1.2–1.5 equiv.), Et₃N/THF (2/1), MWI, 40 °C, 30 min then CH₃CN, 100 °C, MWI, 30 min, 74–90%. b) MeI, K₂CO₃, DMF, room temperature, overnight, 77%. c) NIS (1 equiv.), pTsOH cat., CH₃CN, room temperature, overnight, 59%. d) PdCl₂(dppf)·DCM (5% mol), 3,5-dimethoxyphenylboronic acid (1.5 equiv.), K₂CO₃, DME/H₂O (4/1), 80–100 °C, MWI, 30 min, 83–91%. e) Phosphonate derivatives (1.5 equiv.), NaH (3 equiv.), THF, 120 °C, MWI, 30 min, 58–84%. f) BBr₃, DCM, -78 °C to room temperature, 6 h. g) BCl₃, TBAI, DCM, 0 °C to room temperature, 6 h. h) NBS (1.1 equiv.), DCM, room temperature, overnight, 50%. i) PdCl₂(PPh₃)₂ (5% mol), CuI (3% mol), 4-ethynylanisole (1.2–1.5 equiv.), Et₃N/THF (2/1), MWI, 40 °C, 30 min then 1-iodo-3,5-dimethoxybenzene (1.5 equiv.), CH₃CN, 100 °C, MWI, 30 min, 25% **33**+60% **32**. j) 1-Bromo-3,5-dimethoxybenzene (1.5 equiv.), PdCl(C₃H₃)dppb (5% mol), KOAc (2 equiv.), DMA, 150 °C, 20 h, 60%.

scribed by Markina et al.^[25] This procedure was applied to **25** and **26** to afford **28** and **30** in 74 and 90% yields, respectively. In the case of **27**, addition of 1-iodo-3,5-dimethoxybenzene during the cyclization step partially resulted in a direct coupling to introduce the C-3 aryl on the benzofuran and 2,3-diaryl-benzofuran **33** and 2-arylbenzofuran **32** were isolated in 25% and 60% yields, respectively. Intermediates **28** and **30** were then transformed by halogenation, Suzuki coupling to install the C-3 aryl followed by the Wittig–Horner reaction to achieve the permethylated intermediates **29** and **31** in 31% (5 steps from **25**) and 27% (4 steps from **26**) yields, respectively. Interesting-ly, we could also apply direct arylation of **32** with 1-bromo-3,5-dimethoxybenzene in the presence of

5 mol% PdCl(C_3H_5)dppb to give **33** in 60% yield. This is the first reported C-3 arylation of a benzofuran when C-2 is blocked by an aryl group containing an electron-donating 4-methoxy substituent. In a recent study this type of arylation was unsuccessful using Pd(OAc)₂ and Sphos as ligand.^[26] A Wittig–Horner reaction was then applied to **33** to afford **34** in overall 43% yield (3 steps from **27**). During demethylation, we again observed different reactivity between BBr₃ and BCl₃/TBAI. Surprisingly, BBr₃-mediated demethylation of stilbenes **31** and **34** which both contain a 3,5-dimethoxystyryl moiety afforded the pure desired products **12** and **4** in 61% and 62% yields, respectively. The use of BCl₃/TBAI was less efficient and in both cases HPLC was required for purification



providing pure **12** and **4** in low yields. In contrast, BBr₃ failed to produce anigopreissin A while $BCl_3/TBAI$ gave the target compound in 23% yield after purification by HPLC.

Conclusions

We have prepared a number of permethylated natural and non-natural stilbenoid compounds and studied demethylations using BBr₃ and BCl₃/TBAI. The latter was found to be superior in demethylation of viniferifuran analogues and produced the product in higher yields and with less cyclized derivatives and higher order oligomers. In addition, strategies based on early demethylation were explored and viniferifuran (2) could be obtained in 13% yield over six steps, the shortest and most efficient synthesis reported so far. Finally, we achieved the first total syntheses of dehydro- δ -viniferin (4, 4 steps, 27%), resveratrol-piceatannol hybrid (12, 5 steps, 16%) and anigopreissin A (11, 6 steps, 4%). The synthesis of dehydro- δ -viniferin also included a successful benzofuran C-3 arylation with C-2 blocked by an aryl group containing an electrondonating substituent. The methods and strategies described have the potential be applicable in syntheses of other stilbenoid natural products and analogues to establish structure-activity relationships.

Experimental Section

General

LC-MS analysis was carried out on a Waters LC system equipped with an Xterra MS C18 18.5 µm 4.6 × 50 mm column and an eluent system consisting of MeCN in water, both of which contained 0.2% formic acid. Detection was performed at 214 and 254 nm. Mass spectra were obtained by use of a Waters micromass ZG 2000, using both positive and negative electrospray ionization (ESI). ¹H NMR and ¹³C NMR spectra were recorded on a Bruker DRX-400 spectrometer in CDCl₃ solution [residual CHCl₃ ($\delta_{\rm H}$ = 7.26 ppm, $\delta_{\rm C}$ = 77.16 ppm) as internal standard] or in $SO(CD_3)_2$ solution [residual $SO(CD_3)(CD_2H)$ $(\delta_{\rm H} =$ 2.50 ppm, $\delta_{\rm C}$ =39.52 ppm) as internal standard] or in $CO(CD_3)_2$ solution [residual $CO(CD_3)(CD_2H)$] $(\delta_{\rm H} =$ 2.05 ppm, $\delta_{\rm C}$ =29.84 ppm) as internal standard] or in CD₃OD solution [residual CD₂HOH ($\delta_{\rm H}$ =3.31 ppm, $\delta_{\rm C}$ = 49.00 ppm) as internal standard]. A Biotage Initiator 400W was used for microwave heating. TLC analysis was carried out using TLC aluminum sheets from EMD/Merck KGaA (mfr. no. Merck, 1.05554.0001). Product purification was done using Biotage flash chromatography (FC) with cartridge or ultra cartridge of 10 g, 25 g, 50 g or 100 g of silica gel. HPLC purification was carried out on a Gilson system equipped with a Macherey-Nagel Nucleodur C18 HTEC 5 µm 21×250 mm column. All eluents contained 0.005% formic acid and the flow rate was set to 20 mL/minute for

CH₃CN and 15 mL/minute for MeOH (method A: $H_2O/MeOH 80/20$ to 0/100 over 30 min; method B: $H_2O/CH_3CN 80/20$ to 0/100 over 30 min).

Procedure A: Demethylation with BBr₃ [Exemplified with Dehydroampelopsin B (9)]

To a stirred solution of **17a** (25 mg, 0.048 mmol, 1 equiv.) at 0°C in DCM (3 mL) under a nitrogen atmosphere was added 1M BBr₃ solution in DCM (0.72 mL, 0.72 mmol, 15 equiv.). The mixture was allowed to warm up and stirred at room temperature for 6 h. The reaction was quenched with H₂O (1 mL) at 0°C. The mixture was diluted with EtOAc (25 mL), washed with H₂O (5 mL) and brine (5 mL). The organic phase was dried with Na₂SO₄ and concentrated under reduced pressure. FC (DCM:MeOH 9:1) afforded a mixture that was purified using HPLC method A to give dehydroampelopsin B **9** as yellow solid; yield: 10 mg (46%).

Procedure B: Demethylation with BCl₃/TBAI [Exemplified with Dehydroampelopsin B (9) and Viniferifuran (2)]

To a stirred solution of **17a** (25 mg, 0.048 mmol, 1 equiv.), TBAI (265 mg, 0.72 mmol, 15 equiv.) at 0°C in DCM (3 mL) under a nitrogen atmosphere was added 1 M BCl₃ solution in DCM (0.72 mL, 0.72 mmol, 15 equiv.). The mixture was stirred at room temperature for 6 h. The reaction was quenched with H₂O (1 mL) at 0°C. The mixture was diluted with EtOAc (25 mL), washed with 10% Na₂S₂O₃ solution (5 mL), H₂O (5 mL) and brine (5 mL). The organic phase was dried with Na₂SO₄ and concentrated under reduced pressure. FC (DCM:MeOH 9:1) afforded a 1:1 mixture of **2** and **9** that was purified using HPLC method A to give viniferifuran **2** (yield: 5 mg, 23%) and dehydroampelopsin B **9** (yield: 5 mg, 23%) as yellow solids.

Anigopreissin A (11): Application of procedure B for demethylation of **29** (20 mg, 0.038 mmol FC (DCM:MeOH 9:1) followed by HPLC method A gave **11** as yellow solid; yield: 4 mg (23%). ¹H NMR (400 MHz, CD₃OD): δ = 7.44–7.35 (m, 4H), 7.15 (d, *J*=0.9 Hz, 1H), 7.06 (d, *J*=16.2 Hz, 1H), 6.78 (d, *J*=6.0 Hz, 2H), 6.76 (d, *J*=0.9 Hz, 1H), 6.71 (d, *J*=8.8 Hz, 2H), 6.42 (d, *J*=2.2 Hz, 2H), 6.31 (t, *J*=2.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ =159.5, 158.7, 158.4, 157.2, 153.0, 151.2, 137.2, 136.8, 130.7, 129.3, 128.9, 128.8, 127.3, 123.7, 119.1, 116.8, 116.6, 116.2, 110.3, 107.2, 102.8, 101.7;: ESI-MS *m*/*z* = 451.02, calcd. for [*M*-H]⁻: 451.12. Analytical data were in agreement with those reported in the literature.^[12]

Resveratrol-piceatannol hybrid (12): Application of procedure A for demethylation of **31** (50 mg, 0.09 mmol). FC (DCM:MeOH 9:1) gave **12** as yellow solid; yield: 26 mg (61%). Application of procedure B for demethylation of **31** (25 mg, 0.045 mmol). FC (DCM:MeOH 9:1) followed by HPLC method A gave **12** as yellow solid; yield: 3 mg (14%). ¹H NMR (600 MHz, acetone- d_6): δ =9.12–8.04 (br, OHs), 7.56 (d, J=8.9 Hz, 2H), 7.43 (d, J=8.3 Hz, 1H), 7.08 (d, J=16.2 Hz, 1H), 6.83 (d, J=8.3 Hz, 1H), 6.82 (d, J=8.9 Hz, 2H), 6.76 (d, J=16.2 Hz, 1H), 6.58 (t, J=2.2 Hz, 1H), 6.49 (d, J=2.2 Hz, 2H), 6.22 (d, J=2.1 Hz, 2H).6.20 (t, J=2.1 Hz, 1H); ¹³C NMR (150 MHz, acetone- d_6): δ = 161.3, 160.3, 159.6, 152.5, 143.8, 143.6, 142.1, 138.3, 131.6,



129.8, 128.2, 126.7, 125.0, 124.1, 121.8, 118.6, 117.2, 113.0, 110.9, 106.8, 104.7, 103.4; ESI.MS: m/z = 467.04, calcd. for [M-H]: 467.11. Analytical data for the resveratrol-piceatannol have not been reported in the literature.^[13]

Dehydro-\delta-viniferin (4): Application of procedure A for demethylation of 34 (15 mg, 0.029 mmol). FC (DCM:MeOH 9:1) gave **4** as yellow solid; yield: 8 mg (62%). Application of procedure B for demethylation of 34 (15 mg, 0.029 mmol). FC (DCM:MeOH 9:1) gave a mixture of 4 and partially deprotected products+excess of TBAI. ¹H NMR (400 MHz, acetone- d_6): $\delta = 8.76$ (s, 1 H, OH), 8.42 (s, 2H, OHs), 8.22 (s, 2H, OHs), 7.66-7.50 (m, 5H), 7.23 (d, J = 16.3 Hz, 1 H), 7.06 (d, J = 16.3 Hz, 1 H), 6.60 (d, J =2.1 Hz, 2H), 6.51 (d, J=2.2 Hz, 2H), 6.45 (t, J=2.2 Hz, 1 H), 6.30 (t, J = 2.1 Hz, 1 H); ¹³C NMR (100 MHz, acetone d_6): $\delta = 161.1, 160.6, 159.9, 155.3, 153.4, 141.6, 136.4, 134.8,$ 132.6, 130.6, 130.5, 129.7, 124.9, 123.8, 119.5, 117.6, 117.3, 112.8, 109.9, 106.9, 104.1, 104.0; ESI-MS: m/z = 451.02, calcd, for $[M-H]^{-}$: 451.12. Analytical data are in agreement with those reported in the literature.^[24]

Acknowledgements

This work was supported by the Swedish Research Council (grant 2014-4670). We are grateful to Dr. Anders Lindgren for helpful comments on the manuscript.

References

- a) T. Shen, X.-N. Wang, H.-X. Lou, *Nat. Prod. Rep.* 2009, 26, 916; b) S. Quideau, D. Deffieux, C. Douart-Cassassus, L. Pouysegu, *Angew. Chem.* 2011, 123, 610; *Angew. Chem. Int. Ed.* 2011, 50, 586; c) C. Riviere, A. D. Pawlus, J.-M. Merillon, *Nat. Prod. Rep.* 2012, 29, 1317; d) M. H. Keylor, B. S. Matsuura, C. R. J. Stephenson, *Chem. Rev.* 2015, 115, 8976.
- [2] Z. Zhao, L. Wang, T. James, Y. Jung, I. Kim, R. Tan, F. M. Hoffmann, W. Xu, *Chem. Biol.* 2015, 22, 1608.
- [3] S. S. Velu, N. F. Thomas, J.-F. F. Weber, *Curr. Org. Chem.* 2012, 16, 605, and references cited therein.
- [4] a) W. Li, H. Li, Y. Li, Z. Hou, Angew. Chem. 2006, 118, 7609; Angew. Chem. Int. Ed. 2006, 45, 7609; b) B. S. Matsuura, M. H. Keylor, B. Li, Y. X. Lin, S. Allison, D. A. Pratt, C. R. J. Stephenson, Angew. Chem. 2015, 127, 3825; Angew. Chem. Int. Ed. 2015, 54, 3754.
- [5] a) S. A. Snyder, A. L. Zografos, Y. Lin, Angew. Chem. 2007, 119, 8334; Angew. Chem. Int. Ed. 2007, 46, 8186;
 b) S. A. Snyder, S. P. Breazzano, A. G. Ross, Y. Lin, A. L. Zografos, J. Am. Chem. Soc. 2009, 131, 1753;
 c) S. A. Snyder, N. E. Wright, J. J. Pflueger, S. P. Breazzano, Angew. Chem. 2011, 123, 8788; Angew. Chem. Int. Ed. 2011, 50, 8629; d) S. A. Snyder, A. Gollner, M. I. Chiriac, Nature 2011, 474, 461; e) N. E. Wright, S. A. Snyder, Angew. Chem. 2014, 126, 3477; Angew. Chem. Int. Ed. 2014, 53, 3409; f) T. H. Jepsen, S. B. Thomas, Y. Lin, C. I. Stathakis, I. de Miguel, S. A. Snyder, Angew. Chem. 2014, 126, 6865; Angew. Chem. Int. Ed. 2014, 53, 6747.

- [6] For selected total syntheses of resveratrol-based oligomers from other groups, see: a) K. C. Nicolaou, T. R. Wu, Q. Kang, D. Y.-K. Chen, Angew. Chem. 2009, 121, 3492; Angew. Chem. Int. Ed. 2009, 48, 3440; b) K. C. Nicolaou, Q. Kang, T. R. Wu, C. S. Lim, D. Y.-K. Chen, J. Am. Chem. Soc. 2010, 132, 7540; c) B. H. Lee, Y. L. Choi, S. Shin, J.-N. Heo, J. Org. Chem. 2011, 76, 6611; d) F. Klotter, A. Studer, Angew. Chem. 2014, 126, 2505; Angew. Chem. Int. Ed. 2014, 53, 2473; e) G. Sudhakar, K. Satish, Chem. Eur. J. 2015, 21, 6475; and refs cited in ref.^[1d]
- [7] C. Soldi, K. N. Lamb, R. A. Squitieri, M. Gonzalez-Lopez, M. J. Di Maso, J. T. Shaw, J. Am. Chem. Soc. 2014, 136, 15142.
- [8] C. E. Zetterström, J. Hasselgren, O. Salin, R. A. Davis, R. J. Quinn, C. Sundin, M. Elofsson, *PLoS ONE* 2013, 8, e81969.
- [9] R. A. Davis, K. D. Beattie, M. Xu, X. Yang, S. Yin, H. Holla, P. C. Healy, M. Sykes, T. Shelper, V. M. Avery, M. Elofsson, C. Sundin, R. J. Quinn, *J. Nat. Prod.* 2014, 77, 2633.
- [10] A. E. G. Lindgren, C. T. Öberg, J. M. Hillgren, M. Elofsson, *Eur. J. Org. Chem.* 2016, *3*, 426.
- [11] a) J. Ito, Y. Takaya, Y. Oshima, M. Niwa, *Tetrahedron* 1999, 55, 2529; b) K. S. Huang, M. Lin, Y. H. Wang, *Chin. Chem. Let.* 1999, 10, 817.
- [12] a) D. Hoelscher, B. Schneider, *Phytochemistry* 1996, 43, 471; b) C. Tancharoen, *Master Thesis*, Kasetsart University, 2012; c) R. Brkljaca, J. M. White, S. Urban, *J. Nat. Prod.* 2015, 78, 1600.
- [13] C. H. Lee, K. D. Yoon, T.-H. Heo, *PCT Int. Appl.* WO 2015126129 A2 20150827, **2015**.
- [14] M. Jiang, R. Liu, Y. Chen, Q. Zheng, S. Fan, P. Liu, Int. J. Mol. Sci. 2014, 15, 17188.
- [15] G.-W. Wang, H.-L. Wang, D. A. Capretto, Q. Han, R.-B. Hu, S.-D. Yang, *Tetrahedron* **2012**, *68*, 5216.
- [16] Y. Jia, T. Li, C. Yu, B. Jiang, C. Yao, Org. Biomol. Chem. 2016, 14, 1982 and references cited therein.
- [17] G. A. Kraus, V. Gupta, *Tetrahedron Lett.* 2009, 50, 7180.
- [18] D. Y.-K. Chen, Q. Kang, R. T. Wu, *Molecules* 2010, 15, 5909.
- [19] I. Kim, J. Choi, Org. Biomol. Chem. 2009, 7, 2788.
- [20] J. R. Hwu, K.-S. Chuang, S. H. Chuang, S.-C. Tsay, Org. Lett. 2005, 7, 1545.
- [21] For application of PdCl(C₃H₅)dppp in direct arylation of heterocycles, see: a) L. Loukotova, K. Yuan, H. Doucet, *ChemCatChem* 2014, 6, 1303; b) C. B. Bheeter, L. Chen, J.-F. Soule, H. Doucet, *Catal. Sci. Technol.* 2016, 6, 2005, and references cited therein.
- [22] P. R. Brooks, M. C. Wirtz, M. G. Vetelino, D. M. Rescek, G. F. Woodworth, B. P. Morgan, J. W. Coe, J. Org. Chem. 1999, 64, 9719.
- [23] X.-C. Li, D. Ferrreira, Tetrahedron 2003, 59, 1501.
- [24] E. Beneventi, S. Conte, M. R. Cramarossa, S. Riva, L. Forti, *Tetrahedron* 2015, 71, 3052.
- [25] N. A. Markina, Y. Chen, R. C. Larock, *Tetrahedron* 2013, 69, 2071.
- [26] T. Dao-Huy, M. Haider, F. Glatz, M. Schnurch, M. D. Mihovilovic, *Eur. J. Org. Chem.* 2014, 8119.