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A simple method to improve the performance of perovskite light-emitting diodes *via* layer-by-layer spin-coating CsPbBr₃ quantum dots†

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Recently, all-inorganic halide perovskite quantum dots have become a very promising material for light-emitting diodes. Herein, we demonstrate a facile method, namely, layer-by-layer spin-coating of CsPbBr₃ QDs to improve device performance. After optimization of the number of emissive layers, the maximum EQE can be increased from an initial value of 0.69% to 2.31%. Additionally, we inserted a CBP layer between PEDOT:PSS and CsPbBr₃ multilayers to balance charge transportation and recombination. As a result, a 37% improvement in EQE (up to 3.16%) and highest luminance of 2629 cd m⁻² are obtained.

Introduction

In the past ten years, halide perovskite materials AMX₃ (where A = CH₃NH₃⁺, HC(NH₂)₂⁺, Cs⁺; M = Pb²⁺, Sn²⁺; X = Cl⁻, Br⁻, I⁻) have attracted increasing attention due to their unique optoelectronic properties, such as a broad light-absorption spectrum, tunable optical band-gap, high color purity and tunability, good hole and electron-transport capability, low cost and easy preparation technology.^{1–6} Perovskite materials have been widely applied in solar cells, light-emitting diodes, lasers and photodetectors, *etc.*

In 2014, Friend *et al.* first reported the hybrid organic–inorganic CH₃NH₃PbBr₃ based perovskite light-emitting diode (PeLED) with maximum external quantum efficiency (EQE) of 0.76%.⁷ Since then, the performance of PeLED has been rapidly improved. Within only one year, the maximum EQE of the CH₃NH₃PbBr₃ based PeLED was drastically increased to 8.53%.⁸ Additionally, Wang *et al.* reported a red PeLED based on mixed organic cation multiple quantum well perovskites, which can maintain a high EQE of up to 12.7%.⁹ These results indicate that PeLEDs can be competitive with organic LEDs. However, organic–inorganic perovskites suffer chemical instabilities under conditions of moisture or oxygen, which limits further development of PeLEDs for commercial applications.

Comparing with organic cation-based lead-halide perovskites, all-inorganic perovskites, such as CsPbX₃ (X is usually Br,

Cl, I or mixed halide), exhibit improved chemical and thermal stability in ambient atmosphere. Moreover, all-inorganic perovskites usually have the merits of high photoluminescence quantum yield and narrow emissions.¹⁰ In 2015, Yantara *et al.* first applied CsPbBr₃ films in PeLED.¹¹ Concurrently, Zeng *et al.* reported CsPbX₃ quantum dots (QDs)-based PeLED with a maximum EQE of only 0.12%.¹² In the following three years, tremendous efforts have been focused on perovskite films or QDs optimization and device engineering in order to enhance the performance of all-inorganic PeLED. For example, Rogach *et al.* achieved an EQE of 0.35% by using an insulating material as a solution additive or with an additional hole-blocking layer in CsPbBr₃-based PeLED.¹³ Pan *et al.* employed a short ligand exchange strategy to grow CsPbBr₃ nanocrystals with the maximum EQE of the green PeLED reaching 3%. They also used similar method to fabricate a mixed-anion CsPbBr_x-Cl_{3-x}-based device and achieved a high EQE value of 1.9% for the blue-emitting PeLED.¹⁴ Subsequently, Zeng *et al.* improved the EQE value to 6.27% for solution-processed CsPbBr₃-based device, through balancing surface passivation and carrier injection *via* ligand density control.¹⁵ Recently, they further demonstrated a peak EQE of CsPbBr₃ based PeLED at 11.6% *via* room-temperature triple-ligand surface engineering.¹⁶ Additionally, Li *et al.* demonstrated a simple trimethylaluminum (TMA) crosslinking technique in red-emitting LED based on CsPbI₃ nanocrystals and obtained a remarkable EQE as high as 5.7%.¹⁷ Other methods to improve the performance of all-inorganic PeLEDs, such as polymer-assisted method,^{18,19} small molecule-assisted method,²⁰ device interfacial engineering and emitter-doped method,^{21–23} have also been reported, which boosted strategies for research.

Nevertheless, all these studies focused on separating QDs to avoid aggregation quenching, for which precise control of ligand, doping or surface passivation were needed, thus increasing the

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complexity of the fabrication process. Therefore, it is necessary to explore strategies, alternative to the aforementioned attempts, to improve the efficiency of PeLEDs. It is known that the QDs can be uniformly dispersed and the phenomenon of aggregation can be greatly reduced if the concentration of dispersion is lowered. However, lower concentration would cause the decrease in the number of QDs in a single emissive layer, thus influencing the light-emitting efficiency. In view of this concern, we proposed a novel and facile approach for highly-efficient green-emitting PeLED *via* deposition of multilayers of low concentration of CsPbBr₃ QDs dispersion without specific chemical treatment, which can effectively simplify the process of device fabrication. Additionally, the device performance can be further improved by incorporating a 4,4'-bis(*N*-carbazolyl)-1,1'-biphenyl (CBP) layer between poly(3,4-ethylenedioxythiophene)-poly(styrenesulfonate) (PEDOT:PSS) and emitting layer. CBP can balance charge transportation and recombination. As a result, the optimized device with CsPbBr₃ multilayers exhibits luminance of 1923 cd m⁻² and maximum EQE of 2.31% (at 5.6 V), which are higher than the device with single layer of QDs. In addition, EQE can achieve about 37% improvement on the basis of the multilayers-based device on introducing a CBP layer between PEDOT:PSS and CsPbBr₃ layers.

Experimental section

Materials

Indium tin oxide (ITO) substrates (thickness of 185 nm, sheet resistance $\leq 6 \Omega$, transmittance $\geq 84\%$) were purchased from South China Science & Technology Company, Shenzhen. PEDOT:PSS (4083), CBP (purity > 99%), 1,3,5-tris(1-phenyl-1*H*-benzimidazol-2-yl)benzene (TPBI, purity > 99%) and lithium fluoride (LiF) were obtained from Xi'an Polymer Light Technology Company. All solvents were purchased from J&K Scientific, Ltd (China) unless otherwise indicated. All chemicals were used without further purification.

Device fabrication

ITO substrates were cleaned sequentially in ultrasonic solvent baths of deionized water, acetone and isopropyl alcohol for 15 min, in sequence. Then, the substrates were blow-dried with nitrogen and then treated with oxygen-plasma for 8 min. PEDOT:PSS was spin-coated onto the ITO substrates at 3000 rpm for 60 s and then baked at 130 °C for 20 min. Subsequently, ITO/PEDOT:PSS samples were transferred into a vacuum thermal evaporation chamber to deposit a CBP layer. The evaporation rate and thickness of the coating were 0.3 Å s⁻¹ and 15 nm, respectively. CsPbBr₃ QDs were synthesized as reported previously and provided by our collaborators.²⁴ For the perovskite layer, the as-prepared CsPbBr₃ QDs were first dispersed in tetrahydrofuran (THF) at a concentration of 2 mg mL⁻¹ under continuous stirring for 1 hour. Then, the solution was treated with static settlement over 48 hours in a glove box and consequently, distinct delamination was observed. We dipped the supernatant onto the treated substrates and spin-coated at 2000 rpm for 40 s. We repeated the above step several times in order to obtain samples with multilayers of

CsPbBr₃. Finally, the samples were loaded into a glovebox-integrated deposition chamber once again and pumped down to a pressure of $<10^{-4}$ Pa. TPBI (30 nm), LiF (1 nm) and Al (100 nm) layers were sequentially deposited by thermal evaporation through a shadow mask at rates of 0.3 Å s⁻¹, 0.2 Å s⁻¹ and 2 Å s⁻¹, respectively.

Device characterization

Transmission electron microscopy (TEM) image was recorded by using JSM-3100F FETEM. UV-vis spectra were obtained from Shimadzu UV-1750 UV-visible spectrophotometer. Photoluminescence (PL) measurements were performed using Ocean Optics Spectrometer (Maya 2000 PRO). Images of film surface morphology and device cross-section were acquired from scanning electron microscopy (MAIA3 model 2016, TESCAN). Thickness of the perovskite films was measured using an atomic force microscope (AFM, Bruker). Current-voltage and electroluminescence (EL) properties of PeLEDs were measured using a semiconductor characterization analyzer (Keithley 2634B) and Maya Spectrometer. Luminance-current density values were recorded by using a ST 86LA luminance meter. Results of EQE were obtained from the system of integrating sphere combined with Ocean Optics Spectrophotometer.

Results and discussion

Fig. 1(a) shows the TEM image of the nanocrystals. We can clearly see that these particles exhibit a cubic shape with an average size of about 20 nm. Fig. 1(b) shows a photograph of the QDs dispersion taken under UV light irradiation. Multilayers (1 to 5 layers) of CsPbBr₃ QDs films were fabricated by pipetting and spin-coating the supernatant. Then, we compared the

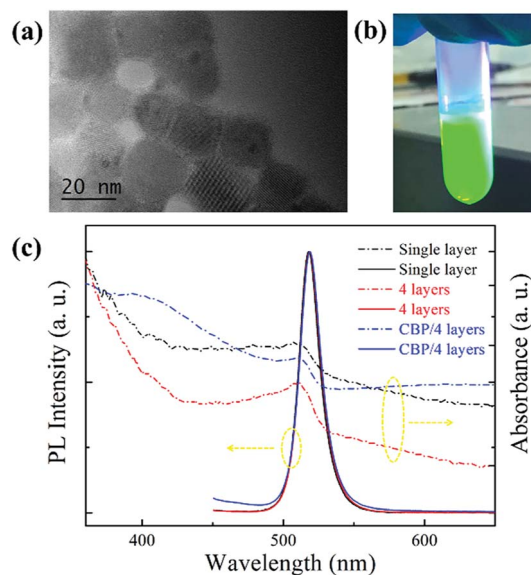


Fig. 1 (a) TEM image of CsPbBr₃ QDs. (b) Photograph of CsPbBr₃ solution with UV light irradiation. (c) UV-vis absorption and PL spectra of three types of films on glass substrates (single layer of CsPbBr₃, 4 layers of CsPbBr₃ and CBP/4 layers of CsPbBr₃).

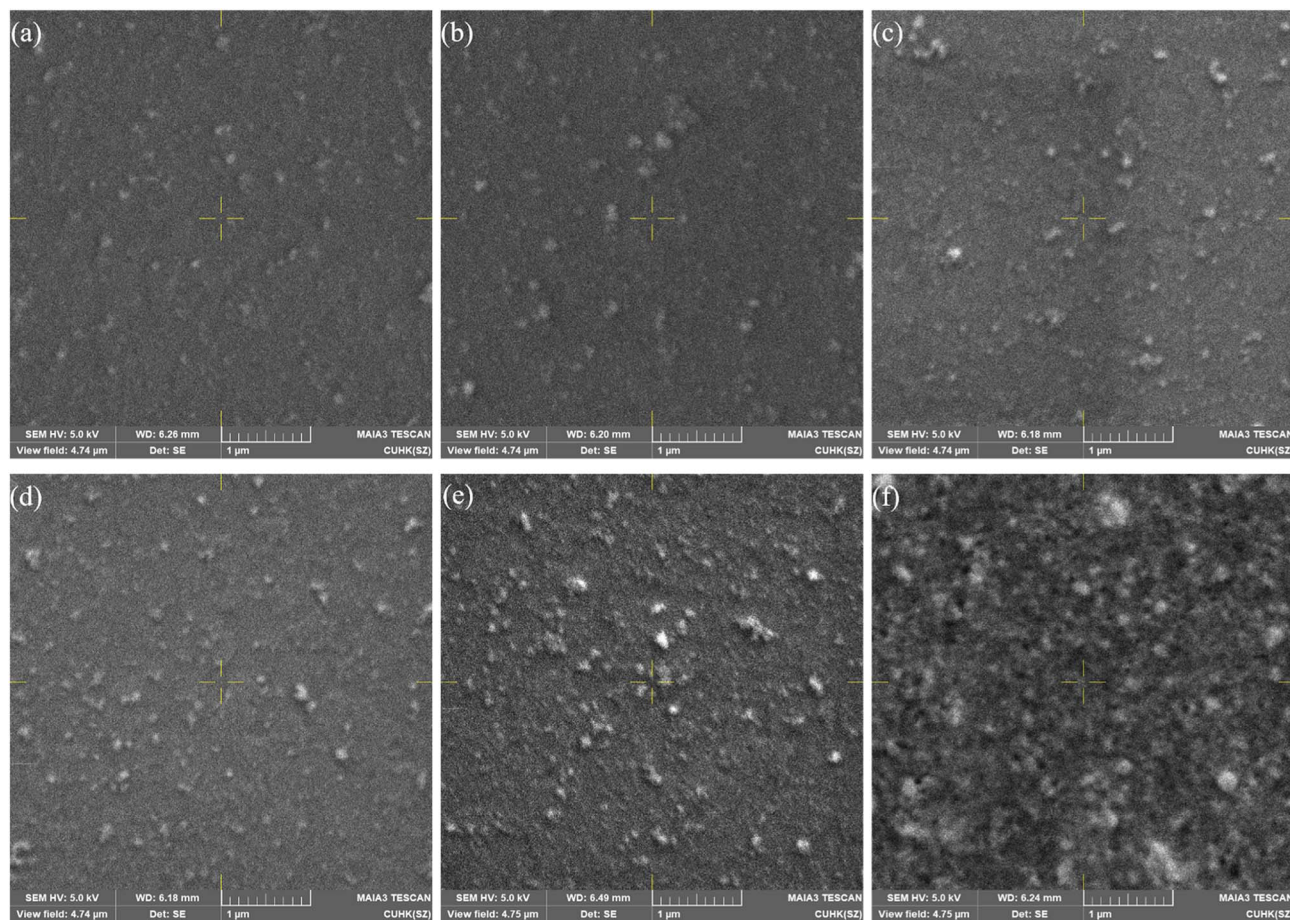


Fig. 2 SEM images of CsPbBr₃ perovskite films (low concentration of QDs, 2 mg mL⁻¹) for different layers: (a) single layer, (b) two layers, (c) three layers, (d) four layers and (e) five layers. (f) Surface morphology of single layer of perovskite film deposited from high concentration of CsPbBr₃ QDs (10 mg mL⁻¹). All perovskite films were prepared on ITO substrates.

optical characteristics of the films of single CsPbBr₃ layer, multiple CsPbBr₃ layers (4 layers) and CBP/CsPbBr₃ multilayers (4 layers). Normalized absorption and PL spectra of these three types of films are shown in Fig. 1(c), displaying similar absorption transitions. We also can see that their normalized PL spectra are overlapped. The PL emission peaks for all the films are located at around 520 nm and the full width at half maximum (FWHM) is only 18 nm. These results indicate that the optical performance stays unchanged with the increase in layers of CsPbBr₃ QDs. In addition, insertion of the CBP layer has no influence on the PL spectra.

In order to investigate the influence of concentration and number of spin-coated layers on the film surface morphology, we performed SEM measurements, and the images are exhibited in Fig. 2.

From Fig. 2(a–e), we can see that the surfaces of films with low concentration of QDs are relatively smooth and uniform when compared with the surface of the film spin-coated from high concentration of CsPbBr₃ QDs solution (Fig. 2(f)). In addition, QDs aggregation is very low in the single layer of perovskite film. However, it becomes severely agglomerated when the layers of QDs are continuously increased.

We also measured the thickness of the perovskite films with different layers using AFM, as shown in Fig. S1 (ESI[†]). All the films exhibit similar thickness in the range of 30–40 nm. Thus, we can conclude that the thickness is not dependent on the number of spin-coated layers.

Fig. 3(a) exhibits the device configuration of the three PeLED devices fabricated in this study. Device A represents PeLED with a single-layer of CsPbBr₃ QDs. Device B comprises multiple layers (4 layers) of CsPbBr₃ QDs. Device C is the CBP-modified structure with CsPbBr₃ multilayers (4 layers), and is composed

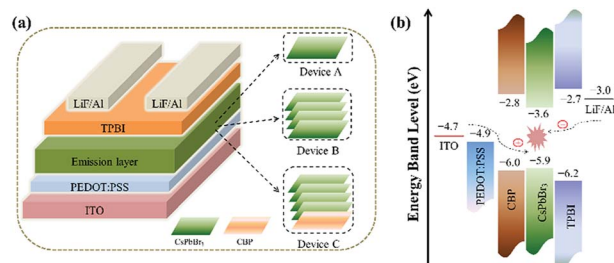


Fig. 3 (a) Schematics of the three types of PeLED devices. (b) Energy levels of various device components in the PeLED.

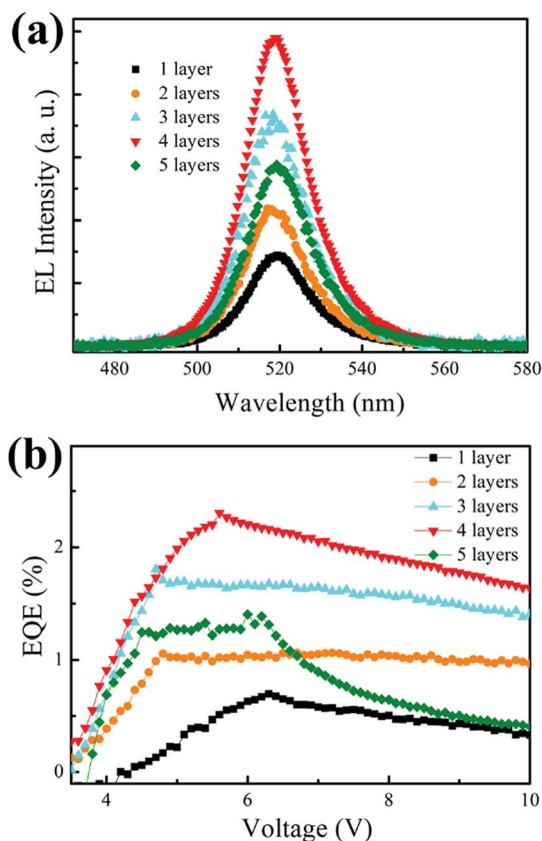


Fig. 4 (a) Electroluminescence spectra and (b) EQE–voltage characteristic curves of PeLED based on different layers of CsPbBr₃ QDs (1–5 layers).

of indium thin oxide (ITO), poly(3,4-ethylenedioxythiophene) polystyrene sulfonate (PEDOT:PSS), CBP, CsPbBr₃ QDs (4 layers), 2,2',2''-(1,3,5-benzinetriyl)-tris(1-phenyl-1*H*-benzimidazole) (TPBi), LiF and Al. Perovskite multilayers were formed by layer-by-layer deposition. Active area of these devices is 0.12 cm². Fig. S2 (ESI†) exhibits the cross-sectional SEM image of PeLED based on 4 layers of QDs. In addition, the energy level diagram of each layer of Device C, as a representative, is shown in Fig. 3(b). We inserted a 15 nm CBP layer, which can block electrons in the active layer, between PEDOT:PSS and CsPbBr₃. This balanced the charge transportation and improved efficiency of charge recombination in the device. Simultaneously, the hole injection barrier does not increase further, as shown in energy-level diagram, due to the well-matched HOMO level between CBP and CsPbBr₃ QDs.

To evaluate the influence of the number of layers of emissive material on the device performance, we investigated the devices based on 1 to 5 layers of CsPbBr₃ QDs; the results are recorded in Fig. 4 and Table 1. Fig. 4(a) exhibits the electroluminescence (EL) spectra of the different devices, which were electrically driven in constant current mode of 20 mA. We observe that all the emission peaks are located at around 520 nm and FWHM is as low as 18 nm, which indicates high color purity. EL intensity increased gradually when the number of layers was no more than four. Similar variation can be found in the maximum

Table 1 Summary of the performances of the PeLEDs. The driving voltages were measured at 0.5 mA

Device	V_{driving} (V)	L_{max} (cd m ⁻²)	EQE (%)
CsPbBr ₃ (1 layer)	3.10	618	0.69
CsPbBr ₃ (2 layers)	3.13	1063	1.05
CsPbBr ₃ (3 layers)	3.25	1781	1.81
CsPbBr ₃ (4 layers)	3.87	1923	2.31
CsPbBr ₃ (5 layers)	4.16	1470	1.40
CBP/CsPbBr ₃ (4 layers)	4.23	2629	3.16

values of EQE, as shown in Fig. 4(b). The device based on 4 layers of CsPbBr₃ QDs exhibits the highest EQE of 2.31% (at 5.6 V), which is almost 3.4-fold higher than that of the single-layer based device (0.69% at 6.3 V). The maximum luminance improved to 1923 cd m⁻² from an initial value of 618 cd m⁻². However, the device performance decreased upon further increasing the number of QD layer (>4 layers). We speculate that the average increment of EQE with the increase in number of coating layers is mainly promoted by more excitons formation in the emissive layer. Maximum EQE and luminance appeared in the 4 layers-based device. Therefore, although the concentration of perovskite QDs is effectively increased, slight aggregation of QDs does not hamper the device performance. The decrease in device performance is mainly attributed to concentration-quenching or self-quenching.^{25,26} These results are consistent with the results of SEM surface morphology analysis. Moreover, the thicknesses of different numbers of QD layers, measured by AFM and profilometer, are closely matched even if we add more number of coating layers because the QDs from the upper layers may permeate into the former layers on the substrates, which may cause the overlapping of the layers. Then, it is reasonable to assume that concentration increase, rather than the emission zone enlargement, attributed to the decline in device efficiency. In addition, our method reduces the optimized concentration of the colloidal perovskite quantum dots as emissive layer, which is believed to be a key factor in PeLED devices. Additionally, we compared the current–voltage characteristics and the J – V curves, as shown in Fig. S3 (ESI†). The driving voltage increased from 3 V to 4.2 V (current density at 20 mA cm⁻²) upon increasing the number of CsPbBr₃ emissive layers, which is attributed to the increase in resistance in the device.

Finally, we further improved the device performance by introducing the CBP layer for the as-optimized PeLED with 4 layers of CsPbBr₃ QDs. The relevant results of these two types of devices are summarized in Fig. 5 and Table 1. EL spectra can be seen in Fig. 5(a) and the photograph of the uniform bright green emission of CBP-based device (captured by a digital camera) is displayed in the inset. The Commission International de l'Eclairage (CIE) color of the green emission is measured with coordinates of (0.1117, 0.7691), as shown in Fig. S4 (ESI†). The insertion of CBP in the device would also cause the increase in driving and turn-on voltages at 1 cd m⁻², which are 3.3 V and 3.5 V, respectively. However, the maximum EQE of the CBP-based PeLED was significantly enhanced from 2.31% to

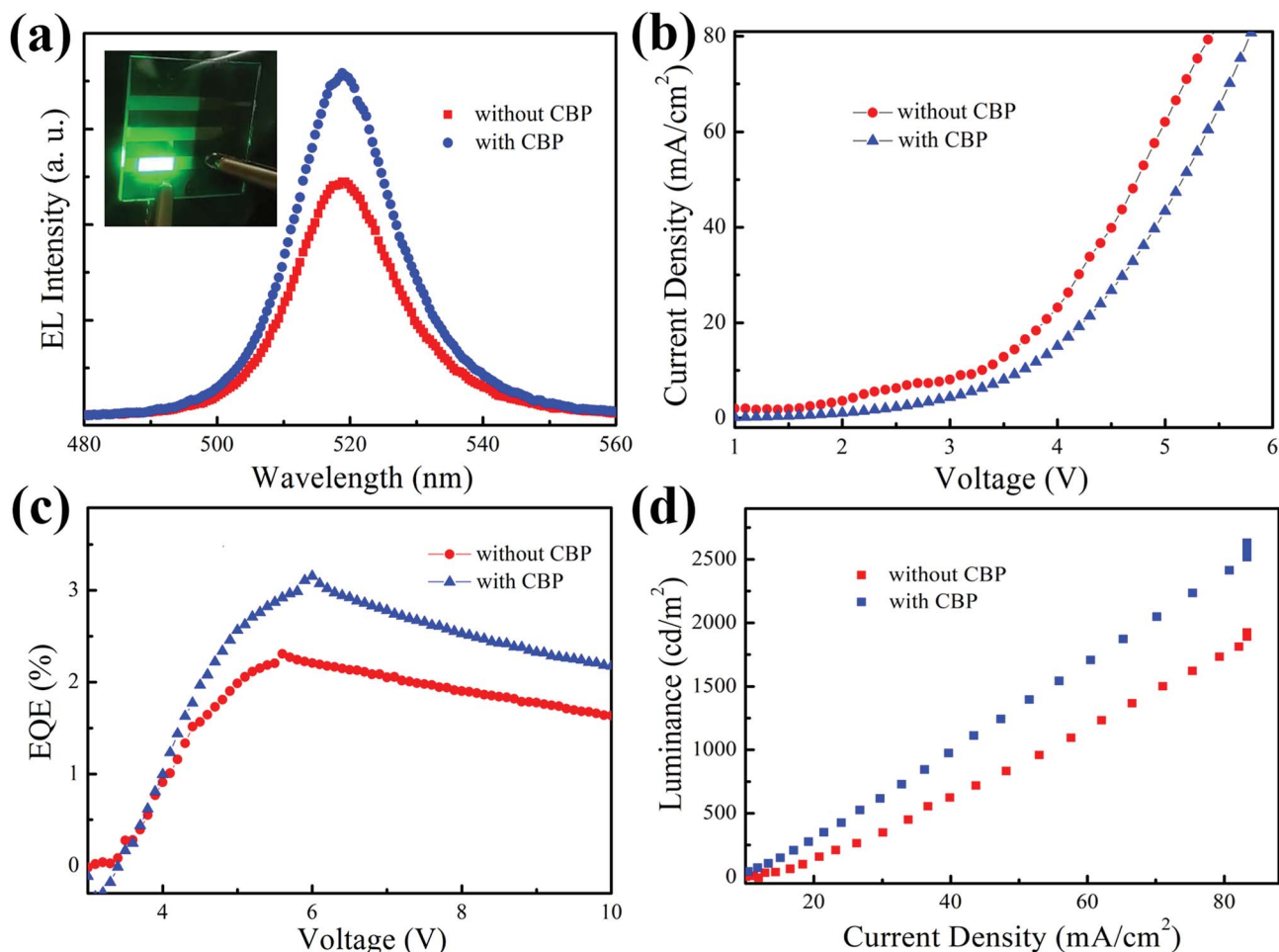


Fig. 5 Light-emitting characteristics of PeLED devices with and without CBP: (a) EL spectra, (b) current density–voltage curves, (c) EQE–voltage curves and (d) luminance–current density curves.

3.16%. Additionally, fast degradation occurred when current density was over 80 mA cm^{-2} , while maximum luminance of 2629 cd m^{-2} was obtained from the device with the CBP layer.

All of the above results indicate that the device with CBP layer has a wide band gap, which can effectively confine the electrons in the emitting layer and balance the charge transportation, thus resulting in highly-efficient charge recombination in the perovskite emissive layer.

Conclusions

In conclusion, we report a facile method for the preparation of highly-efficient perovskite LED by successively spin-coating multilayers of CsPbBr_3 QDs in this study. Maximum EQE increased up to 2.31% from an initial value of 0.69% upon gradually increasing the number of layers of CsPbBr_3 QDs from 1 to 4. However, the device performance declined upon further increasing the number of perovskite QDs layers, which was due to concentration quenching. Additionally, we introduced an energy level-matched CBP layer between PEDOT:PSS and CsPbBr_3 layers in the optimized 4 multilayers based-device to balance the charge transportation and

recombination. The device with the CBP layer can achieve maximum EQE of 3.16% and maximum luminance of 2629 cd m^{-2} . Therefore, the strategy reported in this paper is promising for improving the light-emitting performance of PeLED.

Conflicts of interest

There are no conflicts to declare.

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