



Article Insight into the Molecular Mechanism for the Discrepant Inhibition of Microcystins (MCLR, LA, LF, LW, LY) on Protein Phosphatase 2A

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Abstract: Microcystins (MCs) exhibit diversified inhibition effects on protein phosphatases (PPs) due to their structural differences. To fully evaluate the potential mechanism for the discrepant inhibition effects, the five most frequent MCs with varying residues at position Z⁴ were selected as the tested toxins. Their inhibition sequence on PP2A was detected as follows: MCLR > MCLW > MCLA > MCLF > MCLY. Combined with homology modeling and molecular docking technology, the major interaction parameters between the MCs and PP2A were obtained. The correlation analysis for the major interaction parameters and inhibition effects showed that the hydrophobicity of Z⁴ had an important influence on the interaction of the MCs to PP2A. The introduction of hydrophobic Z⁴ directly weakened hydrogen bonds Z⁴ → Pro₂₁₃ and Z⁴ ← Arg₂₁₄, indirectly weakened hydrogen bonds Adda⁵ ← Asn₁₁₇, Glu⁶ ← Arg₈₉, and MeAsp³ ← Arg₈₉, but indirectly enhanced ionic bonds Glu⁶ ← Arg₈₉, Glu⁶-Mn₁²⁺, and Glu⁶-Mn₂²⁺. In this way, the combination of the MCs with PP2A was blocked, and thus, the interactions between PP2A and the Mn²⁺ ions (in the catalytic center) were further affected; metal bonds Asp₈₅-Mn₁²⁺ and Asp₈₅-Mn₂²⁺ were weakened, while metal bond His₂₄₁-Mn₁²⁺ was enhanced. As a result, the interactions in the catalytic center were inhibited to varying degrees, resulting in the reduced toxicity of MCs.

Keywords: microcystins; protein phosphatase 2A; inhibition mechanism; homology modeling; molecule simulation

Key Contribution: This manuscript established a feasible approach to evaluate the mechanism for the discrepant inhibition of MCs on PP2A.

1. Introduction

Toxic microcystins (MCs), produced by blue-green cyanobacterial, pose a worldwide threat to humans and wildlife [1,2]. When orally ingested, MCs can be transported to the liver by organic anion transport proteins [3,4]. Within the hepatic cells, MCs are inclined to inhibit the activity of serine/threonine protein phosphatase 1 and 2A (PP1 and PP2A) [5,6]. In this way, the balance between protein phosphorylation and dephosphorylation is destroyed, and thus, the cytoskeleton of hepatic cells is disrupted [7,8].

MCs are a class of monocyclic heptapeptides that share a common structure of cyclo-(D-Ala¹-L-X²-D-isoAsp³-L-Z⁴-Adda⁵-D-isoGlu⁶-Mdha⁷) [9]. Due to the variable amino acids at positions 2 and 4, multiple variants have been identified [1,10]. Among MC variants, the most widespread and toxic congener MCLR was widely studied [11]. In consideration of its environmental risk, the World Health Organization recommended a provisional guideline value for MCLR (1 μ g/L) in drinking water [12]. Based on the crystal structure analysis for the complexes of MCLR-PP1 and MCLR-PP2A, a two-step inhibition process was certified: for the reversible step, the hydrophobic side chain of Adda⁵ is rapidly wrapped in the hydrophobic cage structures of PP1/PP2A; for the irreversible step, the C=C double bond



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). of Mdha⁷ undergoes an electrophilic addition reaction with the nucleophilic sites (Cys residue) in PP1/PP2A [13–15]. Both two steps together lead to decreased catalytic activity and cell necrosis.

Compared with MCLR, other MC variants also have cyclic peptide structures and identical Adda⁵/Mdha⁷ residues (Figure 1) [8,16]. The inhibition processes for other variants to PP1/PP2A also should include the reversible and irreversible steps. Due to the structural differences of MC variants, their inhibition effects on PP1/PP2A exhibit diversification [17,18]. As information about the crystal structures of MCs-PP1/PP2A is limited, it is difficult to elucidate the molecular mechanism for the discrepant inhibition of MCs on PP1/PP2A. Homology modeling could evaluate the interactions between structural analogues and macromolecules, and it has been successively used to evaluate the interactions between drugs/contaminants and proteins [19–21].



Figure 1. Chemical structures of MCs with varying amino acids at position Z⁴. Conditions: The blue and brown lines are connected to MeAsp³ and Adda⁵, respectively.

Based on the homology modeling strategy, the interactions between MCs and PP1/PP2A can be simulated, and the molecular mechanism for the discrepant inhibition effects of MCs can be evaluated more in-depth. The five most frequent MCs with the changed Z^4 residues (MCLR, LF, LA, LY, and LW) were selected as the typical variants, and their inhibition effects on PP2A were explored by a colorimetric protein phosphatase inhibition assay [22,23]. With the assistance of the molecular simulation, the models for MC-PP2A were constructed based on the crystal structure of the MCLR-PP2A complex. The major interaction parameters (such as combination energy changes, combination areas, related surface areas, and related chemical bonds) between typical MCs and PP2A were obtained by molecular docking. By analyzing the correlation between the inhibition data and the major interaction parameters, the key interactions and related sites were filtrated. Thus, the molecular mechanism for the discrepant inhibition of typical MCs on PP2A was clarified.

2. Results and Discussion

2.1. Evaluation of the Inhibition Effect of Typical MCs Targeted to PP2A

A traditional colorimetric PP inhibition assay was carried out to evaluate the inhibition effect of MCs on PP2A. As shown in Figure 2, all the MCs exhibited inhibition effects on PP2A and showed dose-effect relationships at 1 nM, 10 nM, and 100 nM. The inhibition sequence was MCLR > MCLW > MCLA > MCLF > MCLY. Among the variants, MCLR had a much higher inhibition effect than other MCs. By comparing the amino acid residues at position Z^4 , it was found that the electropositive Arginine (R) is more hydrophilic than the other four amino acids. When Arg^4 was replaced by $Trp^4/Tyr^4/Ala^4/Phe^4$ (with increased hydrophobicity), the corresponding inhibition effect generally decreased. The lower inhibition of MCLY might be attributed to the electronegative OH in the side-chain of Tyr⁴. Unfortunately, the crystal structures for most MC-PP2A complexes (except for MCLR-PP2A) have not been prepared and solved. It would be difficult to elucidate the relationship between structural differences (changed Z⁴) and the discrepant inhibition of MCS on PP2A.



Figure 2. The inhibition effect of routinely detected MCs on PP2A at 1 nM, 10 nM, and 100 nM with its standard deviation (n = 3).

2.2. Filtration of the Major Interaction Parameters between MCs and PP2A Based on Homology Modeling

The homology modeling strategy has been widely used to evaluate the interactions between structural analogues and proteins [24,25]. Based on this strategy, the interaction models for typical MCs and PP2A could be obtained by molecular simulation (Figure 3). The model for the MCLR-PP2A complex was revised from a crystal model in Protein Data Bank (PDB code 2IE3). The models for other MC-PP2A complexes were obtained by substituting the Arg⁴ residue in MCLR with Ala⁴, Phe⁴, Trp⁴, and Tyr⁴. With the help of molecular docking, the major interaction parameters for MC-PP2A complexes were obtained and are listed in Table S1. The major interaction parameters include the changed energies, combination areas, related surface areas, catalytic center exposure areas, the related hydrogen bonds, ionic bonds, and metal bonds. As PP2A was a type of metalloenzyme and regulated by two Mn²⁺ ions, the interaction parameters (logP and logS) for typical MCs were also calculated.



Figure 3. Illustration for the model construction of MC-PP2A complexes (with no PDB models) based on the homology modeling strategy.

2.3. Pearson Correlation Analysis for Inhibition Data and the Major Interaction Parameters

To evaluate the discrepant inhibition mechanism of MCs on PP2A, the correlation between inhibition data and the major interaction parameters was further evaluated by the Pearson correlation analysis. Regression analysis was not used to avoid deleting valid parameters associated with a few finite amino acid residues.

Table 1 shows the major interaction parameters that exhibited diversified correlations with the inhibition effect of the MCs. To filtrate the important interaction parameters, Venn diagrams were drawn (Figure 4). At the p < 0.01 level, the combination areas for Adda⁵ \rightarrow PP2A and Glu⁶ \rightarrow PP2A, the hydrogen bonds for Z⁴ \leftarrow Arg₂₁₄ and MeAsp³ \leftarrow Arg₈₉, as well as the ionic bond for $Glu^6 \leftarrow Arg_{89}$ were significantly correlated with the inhibition effect at the three test concentrations. The metal bond for $Asp_{85}-Mn_1^{2+}$ was significantly correlated with the inhibition effect at 1 nM and 10 nM. Total hydrogen bonds, the combination area for Mdha⁷ \rightarrow PP2A, and the positive accessible surface area for Adda⁵ \rightarrow PP2A were significantly correlated with the inhibition effect at 1 nM. LogS, the combination area for MeAsp³ \rightarrow PP2A, and the hydrogen bond for Glu⁶ \leftarrow Arg₈₉ were significantly correlated with the inhibition effect at 100 nM. At the p < 0.05 level, combination energy change, logP, the hydrophobic surface area for toxin \rightarrow PP2A, the positive accessible surface area for $Glu^6 \rightarrow PP2A$, the negative accessible surface area for Ala¹ $\rightarrow PP2A$, the hydrophobic surface areas for Adda⁵ \rightarrow PP2A/Mdha⁷ \rightarrow PP2A, the hydrogen bond for Adda⁵ \leftarrow Asn₁₁₇, the ionic bond for Glu⁶-Mn₁²⁺, the metal bonds for Asp₈₅-Mn₂²⁺, His₂₄₁-Mn₁²⁺, and combination energy change were significantly correlated with the inhibition effect at three test concentrations. LogS, the combination area for MeAsp³ \rightarrow PP2A, the negative accessible surface area for MeAsp³ \rightarrow PP2A, the hydrogen bond for Glu⁶ \leftarrow Arg₈₉, the catalytic center exposure area for $Asp_{85} + Mn_1^{2+}$, and the ionic bond for $Glu^6 - Mn_2^{2+}$ were significantly correlated with the inhibition effect at 1 nM and 10 nM. The combination area for Mdha⁷ \rightarrow PP2A and the positive accessible surface area for Adda⁵ \rightarrow PP2A were significantly correlated with the inhibition effect at 1 nM and 100 nM. Total hydrogen bonds were significantly correlated with the inhibition effect at 10 nM and 100 nM. The hydrogen bond for $Z^4 \rightarrow Pro_{213}$ and the catalytic center exposure area for $His_{241} + Mn_1^{2+}$ were significantly correlated with the inhibition effect at 1 nM. The metal bond for Asp₈₅-Mn₁²⁺ was significantly correlated with the inhibition effect at 100 nM. Obviously, the above interaction parameters (especially the parameters significantly correlated with the inhibition effect at two or three test concentrations) were important for the combination of the MCs to PP2A.

Correlation Analysis Data ^a		Combination Energy (KJ/Mol)	Combination Area (Å ²)								
			Total	Ala ¹ \rightarrow PP2A	Leu ² → PP2A	MeAsp ³ \rightarrow PP2A	$Z^4 \rightarrow PP2A$	$Adda^5 \rightarrow PP2A$	$Glu^6 \rightarrow PP2A$	$Mdha^7 \rightarrow PP2A$	
1 nM ^b	R (p)	-0.892 * (0.042)	0.346 (0.569)	-0.392 (0.514)	-0.341 (0.574)	0.902 * (0.037)	0.282 (0.646)	0.978 ** (0.004)	0.979 ** (0.004)	0.978 ** (0.004)	
10 nM ^b	R (p)	-0.890 * (0.043)	0.464 (0. 431)	-0.438 (0. 461)	-0.254 (0.680)	0.951 * (0.013)	0.413 (0.490)	0.994 ** (0.001)	0.996 ** (0.000)	0.956 * (0.011)	
100 nM ^b	R (p)	-0.888 * (0.044)	0.523 (0.365)	-0.538 (0.350)	-0.172 (0.782)	0.982 ** (0.003)	0.492 (0.400)	0.980 ** (0.003)	0.979 ** (0.004)	0.910 * (0.032)	
Correlation Analysis Data ^a		logP (o/w)	Positive accessible surface area (Å ²)								
			Total	Ala ¹ \rightarrow PP2A	Leu ² →PP2A	MeAsp ³ →PP2A	$Z^4 \rightarrow PP2A$	Adda ⁵ \rightarrow PP2A	$Glu^6 \rightarrow PP2A$	Mdha ⁷ →PP2A	
1 nM	R (p)	-0.929 * (0.023)	0.456 (0.440)	-0.765 (0.132)	-0.652 (0.233)	0.125 (0.841)	0.177 (0.775)	0.888 ** (0.044)	0.910 * (0.032)	-0.386 (0.521)	
10 nM	R (p)	-0.936 * (0.019)	0.581 (0.305)	-0.712 (0.177)	-0.560 (0.326)	0.245 (0.691)	0.321 (0.598)	0.905 * (0.034)	0.922 * (0.026)	-0.410 (0.494)	
100 nM	R (p)	-0.888 * (0.044)	0.660 (0.226)	-0.593 (0.292)	-0.419 (0.483)	0.345 (0.570)	0.424 (0.477)	0.900 * (0.037)	0.886 * (0.046)	-0.345 (0.569)	
Correlation Analysis Data ^a		logS	Negative accessible surface area (Å ²)								
			Total	Ala ¹ \rightarrow PP2A	Leu ² →PP2A	$MeAsp^3 \rightarrow PP2A$	$Z^4 \rightarrow PP2A$	$Adda^5 \rightarrow PP2A$	$Glu^6 \rightarrow PP2A$	Mdha ⁷ →PP2A	
1 nM	R (p)	0.884 * (0.046)	0.388 (0.519)	0.882 * (0.048)	-0.548 (0.339)	0.394 (0.512)	0.466 (0.429)	0.378 (0.530)	0.474 (0.419)	-0.140 (0.822)	
10 nM	R (p)	0.925 * (0.025)	0.512 (0. 378)	0.892 * (0.042)	-0.607 (0.278)	0.324 (0.595)	0.560 (0.326)	0.390 (0.516)	0.378 (0.531)	-0.225 (0.716)	
100 nM	R (p)	0.966 ** (0.008)	0.585 (0.300)	0.915 * (0.029)	-0.707 (0.182)	0.179 (0.773)	0.583 (0.302)	0.429 (0.471)	0.303 (0.620)	-0.365 (0.546)	
Correlation Analysis Data ^a			Hydrophobic surface area (Å ²)								
		Total	Ala ¹ \rightarrow PP2A	$Leu^2 \rightarrow PP2A$	MeAsp ³ →PP2A	$Z^4 \rightarrow PP2A$	$Adda^5 \rightarrow PP2A$	$Glu^6 \rightarrow PP2A$	$Mdha^7 \rightarrow PP2A$		
1 nM	R (p)	-0.929 * (0.022)	-0.087 (0.889)	0.590 (0.295)	-0.763 (0.134)	-0.900 * (0.038)	0.924 * (0.025)	0.564 (0.322)	0.934 * (0.020)		
10 nM	R (p)	-0.920 * (0.027)	-0.033 (0.958)	0.607 (0.277)	-0.734 (0.158)	-0.938 * (0.018)	0.953 * (0.012)	0.645 (0.240)	0.923 * (0.026)		
100 nM	R (p)	-0.889 * (0.044)	0.063 (0.920)	0.546 (0.341)	-0.730 (0.161)	-0.932 * (0.021)	0.953 * (0.012)	0.757 (0.138)	0.891 * (0.042)		
Correlation Analysis Data ^a		Hydrogen bonds (KJ/Mol)									
		Total	$Z^4 \rightarrow Pro_{213}$	Ala ¹ \leftarrow Arg ₂₆₈	Leu ² ←Arg ₈₉	MeAsp ³ ←Arg ₈₉	$Z^4 {\leftarrow} Arg_{214}$	$Adda^5 {\leftarrow} His_{118}$	Glu ⁶ ←Arg ₈₉	Mdha ⁷ \leftarrow Arg ₂₆₈	
1 nM	R (p)	-0.980 ** (0.003)	-0.893 * (0.041)	-0.244 (0.693)	-0.780 (0.119)	-0.977 ** (0.004)	-0.995 ** (0.000)	0.216 (0.727)	-0.880 * (0.049)	-0.020 (0.975)	

Table 1. Pearson correlation analysis of the inhibition data and the major interaction parameters.

Table 1. Cont.

Correlation Analysis Data ^a		Combination Energy (KJ/Mol)	Combination Area (Å ²)								
			Total	Ala ¹ → PP2A	Leu ² →PP2A	MeAsp ³ → PP2A	$Z^4 \rightarrow PP2A$	$Adda^5 \rightarrow PP2A$	Glu ⁶ →PP2A	Mdha ⁷ →PP2A	
10 nM	R (p)	-0.953 * (0.012)	-0.841 (0.074)	-0.372 (0.537)	-0.798 (0.106)	-0.993 ** (0.001)	-0.995 ** (0.000)	0.284 (0.643)	-0.938 * (0.018)	0.128 (0.837)	
100 nM	R (p)	-0.901 * (0.037)	-0.764 (0.133)	-0.521 (0.368)	-0.777 (0.122)	-0.991 ** (0.001)	-0.971 ** (0.006)	0.319 (0.600)	-0.984 ** (0.002)	0.296 (0.628)	
Correlation Analysis Data ^a		Hydrogen bond	Metal bonds (KJ/Mol)								
		$Adda^5 \leftarrow Asn_{117}$	Total	$Glu^6-Mn_1^{2+}$	$Glu^6-Mn_2^{2+}$	$Asp_{57}-Mn_1^{2+}$	Asp ₅₇ -Mn ₂ ²⁺	Asp ₈₅ -Mn ₁ ²⁺	$Asp_{85}-Mn_2^{2+}$	His ₂₄₁ -Mn ₁ ²⁺	
1 nM	R (p)	-0.904 * (0.035)	-0.519 (0.371)	0.517 (0.372)	0.731 (0.161)	0.742 (0.151)	-0.850 (0.068)	-0.985 ** (0.002)	-0.935 * (0.020)	0.926 * (0.024)	
10 nM	R (p)	-0.946 * (0.015)	-0.571 (0.314)	0.400 (0.505)	0.621 (0.264)	0.792 (0.110)	-0.796 (0.107)	-0.964 ** (0.008)	-0.937 * (0.019)	0.956 * (0.011)	
100 nM	R (p)	-0.959 * (0.001)	-0.669 (0.217)	0.287 (0.640)	-0.480 (0.414)	0.773 (0.126)	-0.688 (0.199)	-0.904 * (0.035)	-0.894 * (0.041)	0.936 * (0.019)	
Correlation Analysis Data ^a		Ionic bonds (KJ/Mol)									
		Total	Leu ² ←Arg ₈₉	MeAsp ³ ←Arg ₈₉	Glu ⁶ ←Arg ₈₉	$Glu^6-Mn_1^{2+}$	$Glu^{6}-Mn_{2}^{2+(c)}$	Asp ₅₇ -Mn ₁ ²⁺	$Asp_{57}-Mn_2^{2+}$	$Asp_{85}-Mn_1^{2+}$	
1 nM	R (p)	0.794 (0.108)	0.301 (0.623)	-0.171 (0.784)	0.989 ** (0.001)	0.914 * (0.030)	0.955 * (0.012)	0.665 (0.221)	0.622 (0.262)	0.445 (0.452)	
10 nM	R (p)	0.735 (0.157)	0.293 (0.633)	-0.017 (0.978)	0.994 ** (0.001)	0.928 * (0.023)	0.904 * (0.035)	0.773 (0.125)	0.549 (0.338)	0.325 (0.593)	
100 nM	R (p)	0.660 (0.225)	0.246 (0.690)	0.146 (0.815)	0.964 ** (0.008)	0.886 * (0.046)	0.817 (0.091)	0.858 (0.063)	0.470 (0.424)	0.177 (0.776)	
Correlation Analysis Data ^a		Ionic bond	Active center exposure $(Å^2)$								
		Asp ₈₅ -Mn ₂ ²⁺	Asn ₁₁₇ + Mn ₁ ²⁺	Asp ₈₅ + Mn ₁ ²⁺	Asp ₅₇ + Mn ₁ ²⁺	His ₂₄₁ + Mn ₁ ²⁺	Asp ₈₅ + Mn ₂ ²⁺	Asp ₅₇ + Mn ₂ ²⁺			
1 nM	R (p)	-0.171 (0.784)	0.874 (0.053)	-0.955 * (0.011)	-0.787 (0.114)	-0.885 * (0.046)	-0.735 (0.157)	-0.628 (0.257)			
10 nM	R (p)	-0.017 (0.978)	0.819 (0.090)	-0.957 * (0.010)	-0.786 (0.115)	-0.878 (0.050)	-0.759 (0.137)	-0.658 (0.228)			
100 nM	R (p)	0.146 (0.815)	-0.739 (0.154)	-0.809 (0.097)	-0.724 (0.166)	-0.629 (0.255)	-0.724 (0.166)	-0.629 (0.255)			

^a: Sample size n=15; ^b: The inhibition effect of MCs at three test concentrations; ^c: $Glu^6-Mn_2^{2+}$ is the interactions between carbonyl O/hydroxyl O of Glu^6 and Mn_2^{2+} , respectively; logP (o/w) is the Log octanol/water partition coefficient; logS is the water solubility parameter; R is the Pearson correlation between inhibition data and the major interaction parameters; p is the 2-tailed significance of the related data; ** means significant at the 0.01 level; * means significant at the 0.05 level.



Figure 4. Venn diagrams of the important interaction parameters at the p < 0.01 level (**a**) and the p < 0.05 level (**b**). Different colors represent different kinds of interaction parameters. Condition: Toxin is different kinds of integral parameters.

2.4. Molecular Mechanism Analysis of the Discrepant Inhibition of MCs on PP2A

According to the Pearson correlation analysis, the integral parameters such as total hydrogen bonds (|R| = 0.945), logS (|R| = 0.925), logP (|R| = 0.918), ASA-H for toxin \rightarrow PP2A (|R| = 0.913), and combination energy change (|R| = 0.890) were highly correlated with the inhibition effect of the MCs. Obviously, logP, logS, and hydrophobic surface area (ASA-H) were associated with the hydrophobicity of MCs. When the hydrophilic Arg^4 was substituted with Trp⁴/Ala⁴/Phe⁴/Tyr⁴, logP and hydrophobic surface areas (ASA-H) for MCs gradually increased, while logS showed a downward trend. In view of this, the hydrophobicity should have an important influence on the combination of MCs to PP2A; the increased hydrophobicity of Z⁴ would hinder the combination of MCs to PP2A by weakening the total hydrogen bonds (negative correlated with inhibition effect) and by acting on specific important interactions. Statistical frequency analysis of the key sites associated with the important interaction parameters (Figure 5A) showed that Glu⁶, Mn₁²⁺, Adda⁵, Z⁴, Ala¹, Mn₁²⁺, MeAsp³, and Mdha⁷ participated in the interactions between MCs and PP2A in varying degrees. Combined with the statistical analysis for the total |R| values related to the above sites (Figure 5B), it could be found that the influence of Glu^6 , Mn_1^{2+} , and Adda⁵ was more significant than that of Z^4 and other sites. Obviously, the hydrophobicity of Z^4 mainly influenced the combination of MCs to PP2A in an indirect way.



Figure 5. Pie charts for the statistical frequency (**A**) and the total $|\mathbf{R}|$ values (**B**) related to the key interaction sites. Conditions: $\mathbf{\bar{R}}$ is the average of Pearson correlations at three toxin concentrations.

A two-dimensional ligand-receptor interaction diagram for the combination of MCs with PP2A illustrated the key interactions, including hydrogen bonds MeAsp³ \leftarrow Arg₈₉,

 $Z^4 \leftarrow Arg_{214}, Z^4 \rightarrow Pro_{213}, Adda^5 \leftarrow Asn_{117}, and Glu^6 \leftarrow Arg_{89}, ionic bonds Glu^6 \leftarrow Arg_{89}, Glu^6 - Arg_{117}, and Glu^6 \leftarrow Arg_{117}, and Glu^6$ Mn_1^{2+} , and $Glu^6-Mn_2^{2+}$, and metal bonds $Asp_{85}-Mn_1^{2+}$, $Asp_{85}-Mn_2^{2+}$, and $His_{241}-Mn_1^{2+}$ (Figure 6). The Pearson correlation analysis showed the hydrophobicity of Z^4 had the most important influence on the interactions of the MCs to PP2A. According to the hydrophobic surface area analysis, with the increased hydrophobicity of Z^4 , the hydrophobic surface area between Z^4 and PP2A increased. The hydrophobic combination of Z^4 with PP2A directly weakened hydrogen bonds $Z^4 \leftarrow Arg_{214}$ and $Z^4 \rightarrow Pro_{213}$ to varying degrees. The hydrophobic combination of Z^4 with PP2A could intervene the interactions between other residues of MCs and PP2A by weakening the hydrogen bonds Adda⁵←Asn₁₁₇ (hydrophobic competition), $Glu^6 \leftarrow Arg_{89}$, $MeAsp^3 \leftarrow Arg_{89}$, and enhancing ionic bonds $Glu^6 \leftarrow Arg_{89}$, Glu⁶-Mn₁²⁺, and Glu⁶-Mn₂²⁺. Correspondingly, the combination areas of MeAsp³, Adda⁵, and Glu⁶ with PP2A all decreased. As a result, the combination of the MCs with PP2A was blocked to certain degrees. Subsequently, the interactions between PP2A and the Mn²⁺ ions in the catalytic center were further affected; the metal bonds Asp₈₅-Mn₁²⁺ and Asp₈₅-Mn₂²⁺ were weakened, while the metal bond His_{241} - Mn_1^{2+} was enhanced. As a result, there was an increase in the exposure area of the catalytic center (especially for Mn_1^{2+} ion), resulting in lower inhibition effects of the MCs.



Figure 6. The 2D ligand-receptor interaction diagram for the combination of the MCs with PP2A.

3. Conclusions

To obtain a better understanding of the discrepant inhibition effect of the MCs on PP2A, the five most frequent MCs with the changed Z⁴ residues (MCLR, LF, LA, LY, and LW) were selected as the typical variants. A protein phosphatase inhibition assay showed that their inhibition effects on PP2A were in the sequence of MCLR > MCLW > MCLA > MCLF > MCLY. With the assistance of molecular modeling, the interaction models for MC-PP2A were constructed by homology modeling, and the major interaction parameters between MCs and PP2A were obtained by molecular docking. The Pearson correlation analysis for the major interaction parameters and inhibition data verified the important influence of the hydrophobicity of Z⁴ on MC toxicity. The increased hydrophobicity of Z⁴ directly promoted the hydrophobic combination of Z⁴ to PP2A and weakened the hydrogen bonds Z⁴ — Arg₂₁₄ and Z⁴ → Pro₂₁₃. The hydrophobic combination of Z⁴ with PP2A could

intervene with the interactions between other residues of the MCs and PP2A by weakening the hydrogen bonds MeAsp³ \leftarrow Arg₈₉, Adda⁵ \leftarrow Asn₁₁₇, and Glu⁶ \leftarrow Arg₈₉ and by enhancing the ionic bonds Glu⁶ \leftarrow Arg₈₉, Glu⁶-Mn₁²⁺ and Glu⁶-Mn₂²⁺. The above key interactions between the MCs and PP2A further influenced the interactions between PP2A and the Mn²⁺ ions (the metal bonds Asp₈₅-Mn₁²⁺ and Asp₈₅-Mn₂²⁺ were weakened but the metal bond His₂₄₁-Mn₁²⁺ was enhanced). The eventual result was the increased exposure area of the catalytic center (especially for the Mn₁²⁺ ion) and the lower inhibition effect on PP2A.

4. Materials and Methods

4.1. Materials

MCLR, MCLF, MCLA, MCLY, and MCLW were purchased from Sigma (Saint-Quentin Fallavier, France). PP2A was purchased from New England Biolabs Inc (Beverly, MA, USA). Bovine serum albumin, dithiothreitol, MnCl₂, P-nitrobenzene disodium orthophosphate, sodium thiosulfate, tris(hydroxymethyl) aminomethane, and other reagents were purchased from Sinopharm (Shanghai, China).

4.2. PP2A Inhibition Assay

The biological toxicity of the typical MCs was evaluated by a colorimetric protein phosphatase inhibition assay modified by Zong et al. [22,23]. First, PP2A was diluted to 5 U/mL with a buffer solution (50 mM Tris-HCl pH 7.4, 2.0 mM dithiothreitol, 1 mM MnCl₂) and 1.0 g/L bovine serum albumin. Then, 10 µL of PP2A and 100 µL of test samples were mixed in a 96-well polystyrene microplate. With gentle shaking, the microtiter plates were kept at 25.0 °C for 15 min, and 90 µL p-nitrophenyl disodium orthophosphate (5 mM) was added. After 1 h, the absorbances at 405 nm were measured in a microplate reader. The inhibition of the test samples on PP2A was calculated by the formula of I_{PP2A} (%) = (A_{control} – A_{sample})/A_{control} × 100%. A_{control} and A_{sample} were the absorbance of the reference sample (without PP2A) and the test sample at 405 nm, respectively.

4.3. Molecular Docking for the Interactions between MCs and PP2A

The molecular docking simulation was performed with MOE software (version 20.09, Cloud Scientific, Shanghai, China). The original model for the MCLR-PP2A complex was obtained from the Protein Data Bank (PDB code 2IE3). Hydrogen atoms and charges were supplemented to PP2A to obtain the revised model. Based on the homology modeling strategy, models for MCLA-PP2A, MCLF-PP2A, MCLY-PP2A, and MCLW-PP2A complexes were obtained by substituting Arg⁴ (in the revised model of MCLR-PP2A) with Ala⁴, Phe⁴, Trp⁴, and Tyr⁴, respectively [24,26]. All the models for MC-PP2A complexes were minimized for energy optimization. To ensure the comparability between MCs, "Template dock" mode was used (the options for "Placement" and "Refinement" were set to no change). The specific docking parameters were set as follows: amber 10 EHT, solvation r-field, temperature 25.0°C, pH 7.4, salinity 0.05 M. The major interaction parameters (combination energies, combination areas, related surface areas, catalytic center exposure areas, the hydrogen bonds/ionic bonds/metal bonds for major interaction sites) and basic property parameters (logP/logS) for the MCs were obtained to evaluate the interactions between the MCs and PP2A.

4.4. Statistical Analysis

The Pearson correlation analysis was used to analyze the correlation between the inhibition data and the major interaction parameters by IBM SPSS statistics (version 26.0, Chicago, IL, USA).

Supplementary Materials: The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/toxins14060390/s1. All data generated or analyzed during this study are included in this published article and its supplementary information files, Table S1: Main interaction parameters for the complexes of MCs and PP2A. **Author Contributions:** Conceptualization, Y.X. and W.Z.; methodology, Y.X. and W.Z.; supervision, H.Y.; writing—original draft, Y.X. and J.C.; writing—review and editing, Y.X. and W.Z. All authors have read and agreed to the published version of the manuscript.

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