



Development of a novel selective HPLC-chemiluminescence method by targeting degradation intermediates of nitroimidazoles and its application in animal food products

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ABSTRACT

Residues of nitro-containing drugs in food and the environment have aroused social concern because of their excessive use for the treatment of human and animal diseases. The disadvantages of complex and time-consuming pretreatments in existing methods make it imperative to develop novel methods. Herein, with the aid of recognition probe MCLA, an excellent selective method for the detection of nitroimidazoles (NDZs) was developed based on high-performance liquid chromatography combined with ultraviolet (UV) degradation and chemiluminescence (HPLC-UV-CL). The study proposed an innovative process for the formation of $\text{SO}_4^{\bullet-}$, $\bullet\text{OH}$, and $\bullet\text{NO}$ via advanced oxidation processes of NDZs. Six studied NDZs achieved baseline separation within 24 min with detection limits as low as 3 ng/mL ($S/N = 3$). The proposed method is efficient, cost-effective, and does not require complicated purification steps. This study presents a unique approach for the detection of nitro-containing contaminants and provides new insights into monitoring advanced oxidation processes.

1. Introduction

Economic development has increased the average healthy life expectancy and the demand for medicines (Jaunay et al., 2023). Concurrently, a large number of drugs have been developed for diseases treatment of animals and meat preservation (Ludwiczak et al., 2024). The nitro group is an important functional moiety in pharmaceutical chemistry, characterized by its strong electron-accepting capacity, which can create localized electron-deficient sites within the molecule.

Nitro-containing drugs can interact with bionucleophilic entities such as proteins, amino acids, nucleic acids and enzymes (Nepali et al., 2019). Consequently, nitro groups are commonly found in various anti-tumor agents, antibiotics, anti-tuberculosis medications, anti-parasitic drugs, insecticides, and herbicides (Dziduch et al., 2024). The use of nitro-containing drugs, including gastric medications such as ranitidine (Kim et al., 2024), antimicrobial agents such as metronidazole (Walsh et al., 2024), and antibiotics such as nitrofurantoin (Hussein et al., 2024), is common in clinical practice. Currently, these drugs are also widely used to treat various diseases in animals. Various pharmaceuticals may persist in the body or be released into the environment during manufacturing, use, and disposal (Rairat et al., 2024), which leads to a series of health risks and poses a threat to the global environment and

human health (Nordin et al., 2024; Wilkinson et al., 2022). Furthermore, these compounds are frequently associated with severe adverse reactions and toxicities, including carcinogenicity, hepatotoxicity, and mutagenicity (Wang et al., 2023). Therefore, monitoring nitro-containing drug residues is essential to ensure the safety of meat-based foods and safeguard human health (Shen et al., 2024; Yang et al., 2024).

Many methods for detecting and analyzing nitro-containing drug residues have been reported until now, for example, high performance liquid chromatography tandem mass spectrometry (HPLC-MS/MS) (Guo et al., 2017), electrochemical analysis (ECL) (Liu et al., 2021), enzyme-linked immunosorbent assays (ELISA) (Wang et al., 2011), and chemiluminescence (CL) (Chen et al., 2017). Among them, HPLC-MS/MS has gradually become one of the most commonly used methods (Melekhin et al., 2024). However, purification procedures such as solid-phase extraction (SPE) are inevitable when using HPLC-MS/MS, resulting in time consumption, complex pretreatment and high cost (Sasano et al., 2024). In order to address the challenges of global warming and reduce carbon emissions, the exploration of simpler, faster, safer, and more environmentally friendly detection methods is of importance (Gökmen, 2023; Jin et al., 2022). CL techniques have become popular in health inspection and drug analysis (Yu et al., 2024), and are active in the

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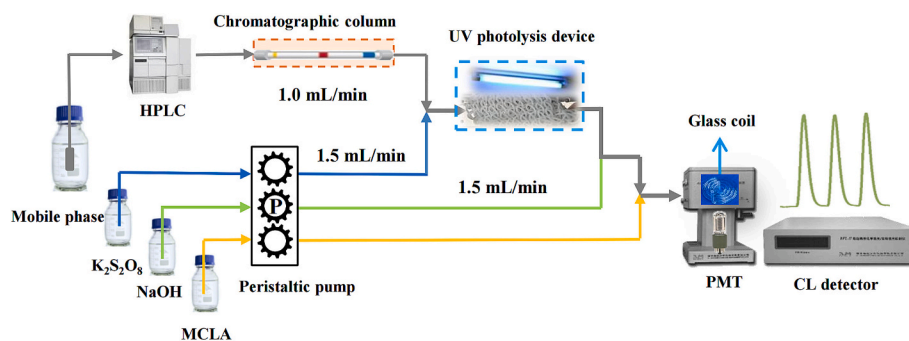


Fig. 1. Schematic diagram of the HPLC-UV-CL detection system.

multidisciplinary fields of chemistry, food, environment, life science, medicine, and materials (Cai et al., 2022; Shishavan & Amjadi, 2021). The combination of HPLC and CL (HPLC-CL) is helpful to solve the problem of matrix interference with the advantages of high sensitivity, high efficiency, and good selectivity (Wei, Cai, et al., 2024; Wei, Gan, et al., 2024).

Our previous studies (Ding et al., 2023) have demonstrated that substances containing N-NO₂ structure would produce nitrogen oxygen free radicals (\cdot NO) after ultraviolet (UV) irradiation. Another interesting phenomenon was observed: NDZs reacted with 2-methyl-6-(4-methoxy-phenyl) 3,7 dihydroimidazo[1,2-a]-pyrazin-3-one (MCLA) to produce a strong CL after UV photolysis using K₂S₂O₈ as the photolysis accelerator, and the CL intensity was positively correlated with concentration. Therefore, a new HPLC-UV-CL method was developed in order to detect the NDZs. By optimizing the HPLC and CL conditions, we applied this method for to monitor NDZs residues in chicken gizzards, chicken breast, lean meat, pork liver, and pork kidneys, with the detection limit as low as ng/mL. Additionally, we conducted in-depth studies of the reaction mechanism using fluorescence (FL) spectroscopy, CL spectroscopy, free-radical quenching experiments, and electron paramagnetic resonance techniques (EPR). This study provides a novel approach for the detection of nitro-containing drugs.

2. Materials and methods

2.1. Materials and reagents

Unless otherwise specified, all the reagents were used directly without further purification. Acetonitrile (ACN) and methanol (MeOH) were purchased from Merck group (Darmstadt, Germany). Formic acid (FA, HPLC grade) was purchased from Shanghai Aladdin Biochemical Technology Co., Ltd. (Shanghai, China). MCLA was acquired from TCI (Shanghai) Development Co., Ltd. (Shanghai, China), and potassium persulfate (K₂S₂O₈) was purchased from Tianjin Kaitong Chemical Reagent Co., Ltd. (Tianjin, China). For the six NDZs, metronidazole (MNZ) and dimetridazole (DMZ) were purchased from Beijing Innochem Technology Co., Ltd. (Beijing, China), ronidazole (RNZ) was purchased from Merck group (Darmstadt, Germany), tinidazole (TNZ) and secnidazole (SNZ) were acquired from TCI (Shanghai) Development Co., Ltd. (Shanghai, China), and ornidazole (ONZ) was purchased from Shanghai Macklin Biochemical Technology Co., Ltd. (Shanghai, China). The standard product was dissolved in methanol and prepared into 0.5 mg/mL reserve liquids that were stored at -4°C .

The MCX solid phase extraction column utilized in this study was purchased from Waters Technologies (Milford, MA, USA). The required columns included the Dikma Diamonsil Plus C18-A (250×4.6 mm, 5 μm) (Beijing, China), Waters Xbridge BEH C18 (150×4.6 mm, 3.5 μm) (Milford, USA), and Waters SunFire C18 (250×4.6 mm, 5 μm) (Milford, USA).

2.2. Apparatus

The HPLC-UV-CL system consists of three main components: HPLC, UV photolysis, and CL detection. The HPLC system consisted of a Waters 2695 HPLC system (Waters, Milford, MA, USA) and thermostat column compartment. The UV photolysis device was purchased from Aura Industries Co., Ltd. (New York, NY, USA), and the wavelength of the UV lamp was 254 nm. The CL detection system was composed of a BT100-1F peristaltic pump (Baoding Longer Precision Pump, China), IFFS-A CL detector equipped with a glass coil, and RFL-I data analyzer (Xi'an Remex Analyze Instrument, China). A personal computer was used to obtain the experimental data. FL spectra were obtained using a FL spectrophotometer (F-320, Tianjin Gangdong Technology Co., Ltd., China), and CL spectra were measured using a FL spectrophotometer after blocking the xenon lamp emission window.

2.3. Condition of HPLC-UV-CL

A schematic representation of the HPLC-UV-CL detection system is shown in Fig. 1. All components were connected to polytetrafluoroethylene or peek tubes (0.8 mm id). Six NDZs were separated on a Waters SunFire C18 column using H₂O/ACN/1 % FA as the mobile phase at 25°C . The injected volume was 100 μL . K₂S₂O₈, NaOH, and MCLA were transported using a peristaltic pump. K₂S₂O₈ was first mixed with the post-column products and allowed to flow through the UV photolysis device. Subsequently, the mixture was then mixed with NaOH and finally with MCLA in a flow cell to generate CL. The detection of the six NDZs was based on the change in CL intensity, $\Delta I = I_s - I_b$, where I_s and I_b were the CL intensities in the presence and absence of NDZs, respectively.

2.4. Sample preparation

Chicken gizzards, chicken brisket, lean meat, pork livers, and pork kidneys were purchased from a local market. After the chicken and pork were crushed, approximately 2 g of the sample was weighed into a 50 mL centrifuge tube, and the standard concentrations were set as follows: MNZ, DMZ, RNZ, and SNZ at 10 ng/mL, 20 ng/mL, 100 ng/mL; TNZ, and ONZ at 20 ng/mL, 40 ng/mL, and 200 ng/mL. The mixture was vortexed for 30 s, and 15 mL of ethyl acetate was added to the solution. The supernatant was then extracted by vortexing. The extraction steps were repeated twice, and the extraction solution was combined and dried with nitrogen at 25°C . A total of 5 mL of 0.1 mol/L HCl solution was added to the residue, which was vortexed for 1 min to ensure complete dissolution. Subsequently, 5 mL of n-hexane was added, oscillated, centrifuged, and the n-hexane layer was discarded. Next, 5 mL of n-hexane was added to the lower layer for repeated defatting and the n-hexane layer was discarded. After the extraction process was completed, there was no need to use solid-phase extraction column for purification, and all samples were filtered through a 0.22 μm membrane before testing.

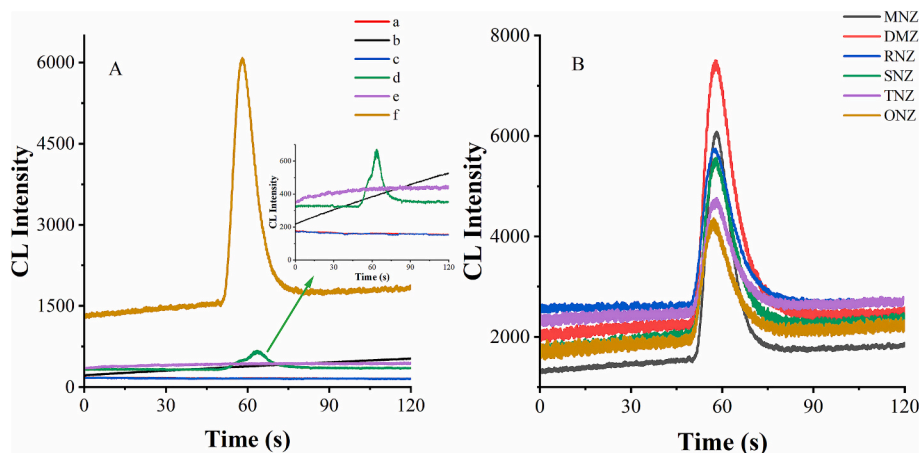


Fig. 2. Reaction kinetics curves. A: MCLA + K₂S₂O₈ - NO UV (a), MCLA + K₂S₂O₈ + MNZ - NO UV (b), MCLA + MNZ - NO UV (c), MCLA + K₂S₂O₈ - UV (d), MCLA + MNZ - UV (e), MCLA + K₂S₂O₈ + MNZ - UV (f); B: Kinetics curves of the six NDZs. Conditions: 0.5 μ mol/L MCLA, 0.5 mmol/L K₂S₂O₈, 0.1 μ g/mL NDZs.

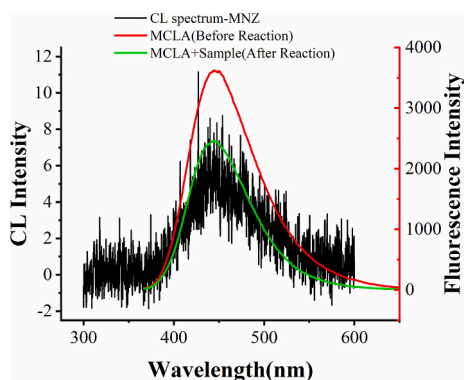


Fig. 3. CL spectra and FL spectra. Conditions: 0.2 μ M MCLA, 5 mM K₂S₂O₈, CL flow rate: 1.8 mL/min, 0.2 μ g/mL MNZ.

2.5. Statistical analysis

All experiments were performed in triplicate, and the results are presented as mean values. Data were analyzed using a paired sample *t*-test. *p* < 0.05 was considered significant.

3. Results and discussion

3.1. CL mechanism

The reaction kinetics curve of the CL system was obtained using an improved static injection method. The results are shown in Fig. 2: in the absence of UV irradiation, and with 5 mL of MCLA as the substrate, 150 μ L of K₂S₂O₈ (Fig. 2A, line a) and 150 μ L of MNZ (Fig. 2A, line c) were injected separately, and no CL was observed. When the substrate contained both MCLA and K₂S₂O₈, the baseline continued to rise. A total of 150 μ L MNZ (Fig. 2A, line b) was then injected, and still no CL signal was observed. After the UV photolysis device was introduced, 150 μ L of K₂S₂O₈ (Fig. 2A, line d) and 150 μ L of MNZ (Fig. 2A, line e) was injected with 5 mL of MCLA as a substrate. The results demonstrated that the injection of K₂S₂O₈ produced a CL signal with an intensity of approximately 350, indicating that the free radicals generated by K₂S₂O₈ activated via UV irradiation could react with MCLA to produce a weak CL. In Fig. 2A, lines e and f represent the CL produced by the injection of 150 μ L MNZ in the absence and presence of K₂S₂O₈, respectively. The results revealed that when there was no photolysis accelerator, almost no CL was present. After the mixed solution of NDZs/K₂S₂O₈ was degraded by UV light, it reacted with MCLA to produce a strong CL signal, reaching a maximum intensity at approximately 7.7 s, and indicating that there may be a new reaction in the system to produce a strong CL. Other NDZs exhibited similar CL phenomena (Fig. 2B).

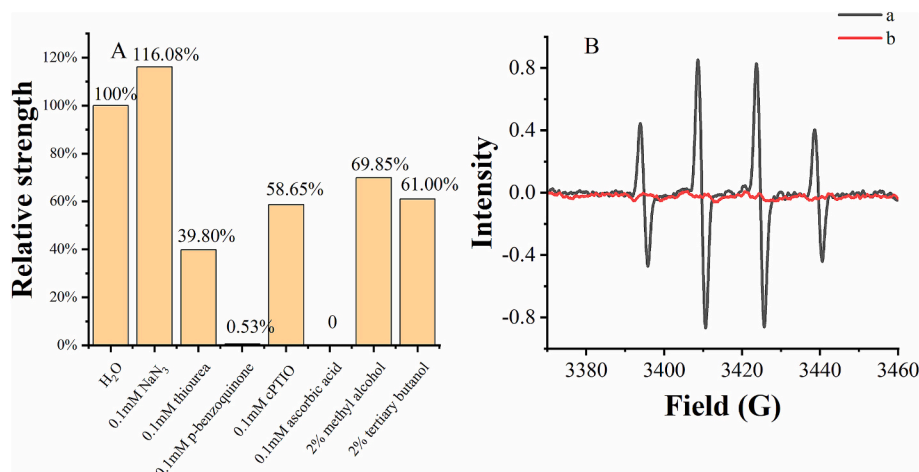


Fig. 4. Results of the free radical capture experiment (A) and electron paramagnetic resonance experiment (B). B-a: DMPO/•OH radical in K₂S₂O₈/UV; B-b: DMPO/•OH radical in K₂S₂O₈ + MNZ/UV.

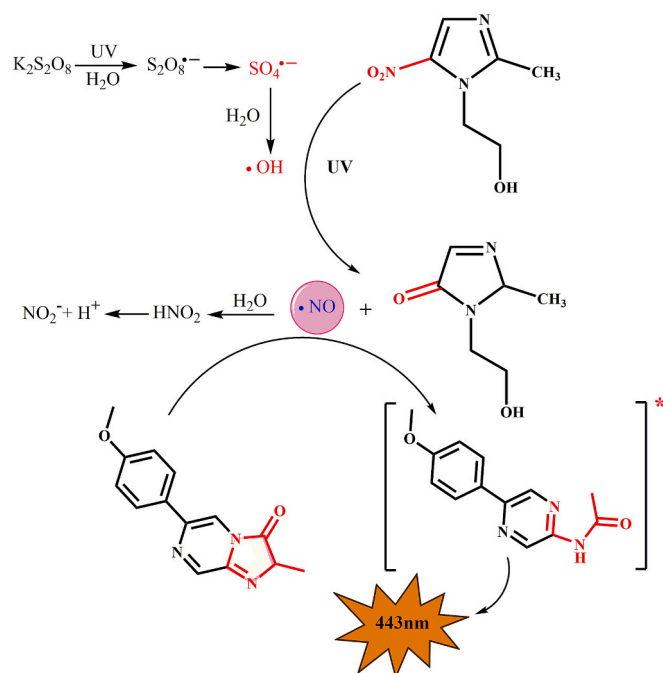


Fig. 5. Schematic diagram of the reaction mechanism of the NDZs / K₂S₂O₈ (UV) - MCLA system.

The FL and CL spectra of the system were measured to elucidate the mechanisms. The emission spectra of the MCLA before and after the reaction were obtained using a FL spectrophotometer. At the maximum excitation wavelength, the emission peaks before and after the reactions are located near 443 nm (Fig. 3). However, the intensity of the emission peak after the reaction was lower than that before the reaction, indicating that the structure of MCLA changed slightly after the reaction. An improved FL spectrophotometer was used to obtain CL spectra of the system. The wavelength of the CL spectrum was approximately 443 nm, which was consistent with the FL spectrum of MCLA after the reaction (Fig. 3), indicating that MCLA participated in the reaction and was the main contributor to the CL signal. Moreover, the wavelength of this CL system was 443 nm, which was different from the 460 nm generated by the reaction of ¹O₂ or O₂^{•-} with MCLA (Zeng et al., 2017), indicating that ¹O₂ and O₂^{•-} were not the key free radical.

Free-radical quenching experiments are a key step in exploring the types of free radicals involved in reaction (Zhang et al., 2019). Therefore, 0.1 mmol/L ascorbic acid, thiourea, sodium azide, and p-benzoquinone were used as free radical quenchers to explore the role of free radicals in the CL system (Fig. 4A). Owing to its strong reducibility, ascorbic acid could react with all free radicals (Brewer, 2011). In this experiment, ascorbic acid simultaneously quenched the CL of the samples and the background CL. These results indicated that free radicals played a key role in the reaction of K₂S₂O₈ with MCLA and in the luminescence effect of NDZs with MCLA. Furthermore, thiourea, sodium azide, and p-benzoquinone reacted directly with K₂S₂O₈ in this system. Therefore, they could not reflect the role of free radicals. The addition of cPTIO reduced CL intensity to 58.65 %, indicating that •NO may play a role in the system. Methanol and tert-butanol were used as trapping agents of •OH/SO₄^{•-} and •OH, respectively (Cai et al., 2022). When 2 % methanol and tert-butanol were added to the system, the CL signal decreased by approximately 30–40 %, indicating that •OH and SO₄^{•-} were likely to exist in the system.

Electron paramagnetic resonance (EPR) was employed in order to determine the type of free radicals. The compound 5, 5-dimethyl-1-pyrroline N-oxide (DMPO) is a commonly used •OH trapping agent that can capture •OH and produce a characteristic peak at 1:2:2:1 (Liu et al., 2024). Fig. 4B presents the results of free-radical detection using EPR.

Curve (a) indicates the characteristic peak of 1:2:2:1 produced by •OH captured by DMPO. Curve (b) indicates the EPR results after the addition of the NDZs. The characteristic peaks of •OH disappeared, and no characteristic peaks of other free radicals were observed. This indicated that K₂S₂O₈ could produce •OH after UV irradiation and that •OH was completely reacted in the system.

The effect of dissolved oxygen on this system was also investigated. All reaction solutions, including K₂S₂O₈, MCLA, and the NDZs, were fed into O₂ and high-purity nitrogen for five minutes to compare the effects of dissolved oxygen. The baseline and CL signals of the NDZs/K₂S₂O₈(UV)-MCLA system exhibited little change after aeration and deaeration, indicating that the CL intensity of the NDZs/K₂S₂O₈ (UV)-MCLA system was independent of dissolved oxygen.

A possible reaction mechanism for the NDZs/K₂S₂O₈ (UV)-MCLA system was deduced based on the reaction kinetics curves, FL spectra, CL spectra, free-radical quenching experiments, and EPR experiments. As presented in Fig. 5, under the action of UV light, K₂S₂O₈ was first activated to produce SO₄^{•-}, and SO₄^{•-} further reacted with water to produce •OH (Zeng et al., 2024), that then reacted with the nitro group of the NDZs to produce •NO and the corresponding carbonyl products via UV photolysis. •NO subsequently interacted with MCLA, making the MCLA transition to the excited state MCLA*. In the process of its return to the ground state, energy was released and CL signals were generated.

Previous research has demonstrated that the CL intensity is affected by many factors, including column type, mobile phase, column temperature, flow rate, and the type and concentration of the promoting media. These conditions were optimized in order to increase the S/N and reduce the detection limit. Under optimal conditions, the low detection limits, linear ranges, regression equations, correlation coefficients, and other indices were obtained.

3.2. Optimization of the HPLC conditions

Based on literature search, combined with the results of previous experiments, we compared the combined effects of various chromatographic columns with organic phases, acids, and buffer solutions (El-Maghrabey et al., 2021). We compared the separation results of the Diamonsil Plus C18-A, Xbridge BEH C18 and SunFire C18 columns and observed that the best separation results were obtained using the SunFire C18 column. MeOH, one of the most commonly used organic phases, was used in this study (Atapattu, 2023). The experimental results demonstrated that the CL signal was significantly inhibited even at a low concentration of MeOH, and this was related to the free-radical quenching effect of MeOH (Liu et al., 2019). The introduction of ACN also inhibited the CL signal in the sample. However, when a certain proportion of FA was introduced, the CL signal was significantly enhanced, and the baseline was stable. Based on this phenomenon, ACN and FA were selected as mobile phases for NDZs separation, and the elution procedure was optimized according to the separated substance. When the ratio of ACN was low, due to the strong interaction between the separated substance and C18 filler, the mobile phase elution ability was insufficient, resulting in serious peak broadening. This condition could not achieve separation of the six substances, and the analysis time was long. With increasing ACN ratio, the elution ability of the mobile phase was enhanced, the overall elution time of the separated substance was shortened, and the degree of separation increased. Concurrently, both the column temperature and column flow rate were optimized, and the column temperature exerted no obvious effect on the separation effect. Therefore 25 °C was chosen as the optimal temperature for separation. By optimizing the flow rate at 0.5–1.2 mL/min, we observed that although the separation of the six NDZs could be achieved when the flow rate was lower than 1.0 mL/min, the retention time was long, and a large amount of organic solvents was consumed. When the flow rate of the chromatographic pump continued to increase, TNZ could not be separated from the slope peak caused by the gradient change, and the S/N ratio decreased (Fig. S1). Therefore, when considering the separation

Table 1

Comparison of the performance on the detection of the NDZs with the different methods.

Methods	NDZs	Pretreatment	ILOD ($\mu\text{g/kg}$)	MLOD ($\mu\text{g/kg}$)	Sample	References
HPLC-MS/MS	Seven NDZs	MISPE	0.5–2.5	0.1–0.5	Honey	(Guo et al., 2017)
HPLC-MS/MS	DMZ	SPE	5.0	0.5	Egg and chicken tissue	(Mo et al., 2024)
HPLC-UV	DMZ/RNZ/SNZ/ MNZ/ONZ	SPE	0.2–0.4, 4–10 5–7	0.02–0.04, 0.4–1.0 0.5–0.7	Water, honey and chicken breast	(J. Wang et al., 2023)
Differential-pulse voltammetry technique (DPV)	RNZ	None	107	–	Bovine meat	(Diniz et al., 2020)
ECL	MNZ	None	8.6	–	Raw milk	(Liu et al., 2021)
HPLC-UV-CL	MNZ/DMZ/RNZ/ SNZ/TNZ/ONZ	None	3–6	0.5–1.0	chicken gizzards, chicken breast, lean meat, pork liver, and pork kidneys	This work

time and effect, we determined the optimal gradient elution procedure, as presented in Table S1.

3.3. Optimization of the CL conditions

The CL conditions of the system were optimized by considering the S/N ratio and baseline stability as dependent variables. Then we took six NDZs mixtures of 0.1 $\mu\text{g/mL}$ MNZ, DMZ, RNZ, and SNZ and 0.2 $\mu\text{g/mL}$ TNZ and ONZ as examples to observe the CL intensity under different conditions and construct the optimal reaction system.

Different mixing methods reflect different reaction orders, that exert an important influence on the CL intensity. Initially, we compared the effects of different mixing methods. The experimental results indicated that $\text{K}_2\text{S}_2\text{O}_8$, which functioned as a photolysis accelerator, could only generate a strong CL when subjected to UV photolysis in conjunction with NDZs. The introduction of an alkaline solution enhanced the CL signal; however, it must be added after UV photolysis in order to avoid a significant baseline drift. If the alkaline solution was pre-mixed with MCLA, the structure of MCLA would change, resulting in substantial background absorption. Consequently, when the NDZs were introduced into the system, no CL signals were observed. We determined that the best method was to first mix the post-column separation solution with $\text{K}_2\text{S}_2\text{O}_8$, which was then degraded under UV light and mixed with an alkaline solution. Finally, the mixture was mixed with MCLA in the coil, and the CL signals were recorded.

The flow rate of the peristaltic pump had a great influence on whether we could obtain the maximum S/N. When the flow rate was low, CL reaction did not occur in the detection cell. If the flow rate was too high, the solution would flow out before the CL reaction was completed. After studying the flow rate in the range of 1.0–2.2 mL/min, we observed that the maximum S/N ratio could be obtained when the flow rate was 1.5 mL/min, therefore, we determined that the optimal flow rate was 1.5 mL/min, as shown in Fig. S2.

The effects of different concentrations of NaOH, Na_2CO_3 and NaHCO_3 on the reaction system are compared, and the results are shown in Fig. S3. The introduction of an alkaline solution to neutralize FA in the mobile phase could increase the S/N ratio of the reaction system; however, if the concentration of the alkaline solution was too high, the CL intensity of MCLA would be strongly inhibited (Zeng et al., 2017). We determined that 0.01 mol/L NaOH was the best reaction solution.

MCLA, the only CL reagent used in this system, played an essential role in this study. The concentrations of MCLA ranging from 0.01 to 1.0 $\mu\text{mol/L}$ were examined. The results were shown in Fig. S4 (A), which indicated that the S/N increased with increasing concentration in the range of 0.01–0.5 $\mu\text{mol/L}$. Then the CL intensity decreased when the concentration exceeded 0.5 $\mu\text{mol/L}$. This may be because the excess MCLA reacted with other substances to increase the noise, while the signal of the sample did not change significantly, and the S/N ratio decreased. Finally, 0.5 $\mu\text{mol/L}$ MCLA was selected as the optimal reaction concentration.

The reaction between the NDZs degraded by UV irradiation and

MCLA exhibited no CL signal and low efficiency, and the introduction of certain photolysis accelerators significantly improved the CL intensity. The effects of different oxidants, organic acids, and inorganic acids on this system were compared. The results revealed that KIO_4 had no significant effect on the CL of the system, whereas KHSO_5 , acetic acid, HCl, H_2SO_4 and $\text{H}_2\text{C}_2\text{O}_4$ could enhance the CL of the system. Among these, KHSO_5 and $\text{K}_2\text{S}_2\text{O}_8$ yielded the best results. However, when KHSO_5 was introduced, it reacted directly with the MCLA to produce a strong background CL. $\text{K}_2\text{S}_2\text{O}_8$ provided a more stable baseline and signal; based on this, $\text{K}_2\text{S}_2\text{O}_8$ was introduced into the system as the best photolysis accelerator. The concentration of $\text{K}_2\text{S}_2\text{O}_8$ was optimized in the range of 0.5–8.0 mmol/L, and the effect is presented in Fig. S4 (B). When the concentration of $\text{K}_2\text{S}_2\text{O}_8$ was low, fewer free radicals were generated by $\text{K}_2\text{S}_2\text{O}_8$ after UV irradiation, and the photolysis-promoting effect on the NDZs was weak. With an increase in concentration, the promoting effect became increasingly evident. When the concentration reached 5.0 mmol/L, the promoting effect reached a maximum and the S/N ratio was the highest. As the concentration continued to increase, the S/N ratio decreased. Therefore, 5.0 mmol/L was selected as the optimal reaction concentration.

Finally, we optimized the detector voltage and compared the effects of voltage in the range of 300–600 V on the CL intensity. As shown in Fig. S5, as the voltage increased, the CL intensities of the six analytes gradually increased, and the noise increased. The maximum S/N was obtained at 400 V, and the continuous increase in the voltage exceeded the detection range of the instrument. Therefore, 400 V was chosen as the optimal voltage.

3.4. Method validation

This method was validated according to the 2002/657/EC Decision (European Commission, 2002). Under the aforementioned conditions, the standard chromatograms of the six NDZs were obtained using HPLC-UV-CL method (Fig. S6). The retention times for MNZ, DMZ, RNZ, SNZ, TNZ, and ONZ were 7.46, 9.49, 11.66, 13.20, 17.98, and 22.38 min, respectively, with a successful baseline separation within 24 min. Linearity, correlation coefficient (r), and limits of detection (LODs) were used to verify the accuracy of the method, and the relative standard deviation (RSD) was used to verify the precision of the method. The results are summarized in Table S2. The linear ranges of MNZ and SNZ were 20–1000 ng/mL and 20–2000 ng/mL, respectively. And the linear ranges of DMZ, RNZ, TNZ, ONZ were 10–1000 ng/mL. The correlation coefficients were all greater than 0.99 with LODs as low as 3 ng/mL ($S/N = 3$).

In Table 1, the proposed HPLC-UV-CL method is listed along with previously reported analytical techniques. In comparison, the detection limit of this method can reach a lower level. Moreover, for the pre-treatment of complex substrates, this method does not require additional purification steps, and greatly reduces the sample processing time and reagent consumption. In addition, compared with mass spectrometry detectors, the use of CL detectors not only reduces costs, but also greatly

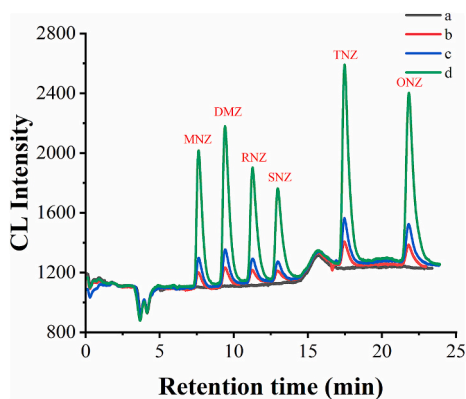


Fig. 6. Representative chromatogram of six NDZs in the samples of chicken gizzards unlabeled (a) and labeled (b, c, d) by HPLC-UV-CL. Conditions: 0.5 μ M MCLA, 0.01 M NaOH, 5 mM $K_2S_2O_8$, CL flow rate: 1.5 mL/min, voltage: 400 V.

reduces carbon emissions.

The intraday RSDs were 3.44–5.97 % ($n = 6$) and the inter-day RSDs of the six analytes was 8.58–9.48 % over five days. Additionally, the sample recovery rate was in the range of 66.67–116.16 %. It was proven that the method had high sensitivity, a wide linear range, low detection limit, and good reproducibility, providing a new method for the detection of NDZs residues in chicken and pork.

3.5. Application

To evaluate the applicability of this method in real samples, NDZs were detected in chicken gizzards, chicken breast, lean meat, pork liver, and pork kidneys. The samples were extracted using different

pretreatment methods and analyzed under optimal conditions. Each sample was repeatedly processed into three groups, and each group of samples was injected thrice in parallel. The spiked concentrations of MNZ, DMZ, RNZ, and SNZ were 10 ng/mL (Fig. 6, line b), 20 ng/mL (Fig. 6, line c), 100 ng/mL (Fig. 6, line d), and the spiked concentrations of TNZ and ONZ were 20, 40, and 200 ng/mL. The CL intensity was used to reflect the content of each component, and the spiking recovery rates were calculated. The standard recovery rates for chicken and pork were listed in Table 2. NDZs residues were not detected in any of the tested samples. Fig. 6 presents the HPLC-UV-CL chromatographic separation of the six NDZs in chicken gizzards, demonstrating that the method exhibits good selectivity and separation. (See Table 2.)

A paired sample *t*-test was conducted on the sample detection results before and after purification with the solid-phase extraction column (Table S3) and the results revealed that the *p*-values for all six NDZs exceeded 0.05, indicating that there was no difference between the two pretreatment methods and that satisfactory recovery could be achieved without purification.

4. Conclusion

Based on the targeted recognition effect of MCLA on the NDZs UV photolysis intermediates, a new HPLC-UV-CL method was developed to detect six NDZs residues in chicken and pork. These findings revealed that the system exhibited high accuracy and selectivity for the detection of NDZs in chicken and pork and was essentially unaffected by matrix effects. This method was used to detect NDZs, and six samples were separated within 24 min, with a detection limit as low as 3 ng/mL ($S/N = 3$). Furthermore, this method does not require complex purification steps during the pretreatment process, thereby improving the efficiency of the analysis and reducing detection costs. A mechanism study revealed the formation process of $SO_4^{\bullet-}$, $\bullet OH$, and $\bullet NO$ and also revealed

Table 2
Determination of the NDZs in the different samples with the HPLC-UV-CL method.

NDZs	Add(ng/mL)	Chicken breast		Chicken gizzards		Lean meat		Pork liver		Pig kidneys	
		Found (ng/mL)	Recovery (%)	Found (ng/mL)	Recovery (%)	Found (ng/mL)	Recovery (%)	Found (ng/mL)	Recovery (%)	Found (ng/mL)	Recovery (%)
MNZ	0	ND	/	ND	/	ND	/	ND	/	ND	/
	10	9.38	93.80	8.60	86.05	8.16	81.63	11.02	110.20	8.98	89.80
	20	18.15	90.76	16.97	84.87	19.80	99.00	17.00	85.00	20.40	102.00
	100	87.96	87.96	97.63	97.63	95.70	95.70	93.15	93.15	80.57	80.57
DMZ	0	ND	/	ND	/	ND	/	ND	/	ND	/
	10	10.42	104.20	9.51	95.10	6.67	66.67	10.53	105.26	8.25	82.46
	20	18.18	90.88	17.88	89.42	16.31	81.55	12.62	63.11	20.97	104.85
	100	82.55	82.55	90.63	90.63	94.09	94.09	95.51	95.51	111.82	111.82
RNZ	0	ND	/	ND	/	ND	/	ND	/	ND	/
	10	8.40	83.96	8.11	81.13	9.38	93.75	7.50	75.00	9.22	92.19
	20	17.86	89.29	16.33	81.63	20.15	100.74	13.82	69.12	19.41	97.06
	100	82.62	82.62	92.53	92.53	100.29	100.29	85.27	85.27	84.54	84.54
SNZ	0	ND	/	ND	/	ND	/	ND	/	ND	/
	10	9.08	90.80	8.05	80.46	9.77	97.67	11.16	111.63	10.23	102.33
	20	18.34	91.72	16.56	82.80	18.54	92.68	16.34	81.71	21.71	108.54
	100	83.23	83.23	92.92	92.92	104.62	104.62	106.59	106.59	87.69	87.69
TNZ	0	ND	/	ND	/	ND	/	ND	/	ND	/

Table 2
Determination of the NDZs in the different samples with the HPLC-UV-CL method (continued).

	20	15.24	76.22	16.54	82.70	19.22	96.08	22.55	112.75	22.55	112.75
	40	33.47	83.67	34.52	86.30	34.66	86.64	28.50	71.26	38.38	95.95
	200	166.53	83.27	180.54	90.27	203.98	101.99	210.86	105.43	181.36	90.68
	ONZ	0	ND	/	ND	/	ND	ND	/	ND	/
	20	17.52	87.58	17.39	86.93	16.36	81.82	23.23	116.16	20.81	104.04
	40	33.19	82.99	34.86	87.15	39.32	98.31	35.25	88.14	43.84	109.60
	200	169.80	84.90	180.73	90.37	202.84	101.42	208.52	104.26	180.73	90.37

ND: Not detected.

the key roles of $\text{SO}_4^{\cdot-}$ and $\bullet\text{OH}$ in the degradation of NDZs. Additionally, this combined method provides a new idea for designing drug residue detection methods for food products.

CRedit authorship contribution statement

Mengdie Cai: Writing – review & editing, Writing – original draft. **Xinyu Wang:** Methodology. **Kun Lu:** Methodology. **Bixiao Zhou:** Validation. **Lijun Wei:** Supervision, Methodology. **Xianglei Cheng:** Writing – review & editing, Funding acquisition, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.fochx.2025.102205>.

Data availability

No data was used for the research described in the article.

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