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## Chiral auxiliary mediated 1,2-*cis* glycosylations for the solid supported synthesis of a biologically important branched $\alpha$ -glucan

Thomas J. Boltje, Jin-Hwan Kim, Jin Park, and Geert-Jan Boons\*

Complex Carbohydrate Research Center, University of Georgia, 315 Riverbend Road, Athens, Georgia 30602

### Abstract

Solid phase oligosaccharide synthesis (SPOS) offers the promise to provide libraries of oligosaccharides for glycomics research. A major stumbling block to SPOS has been a lack of general methods to stereoselectively install 1,2-*cis*-glycosides, and intractable mixtures of compounds will be obtained if several of such glycosides need to be installed. We have prepared on-resin a biologically important glucoside containing multiple 1,2-*cis*-glycosidic linkages with complete anomeric control by using glycosyl donors having a participating (*S*)-(phenylthiomethyl)benzyl chiral auxiliary at C-2. A branching point could be installed by employing 9-fluorenylmethyloxycarbonyl (Fmoc) and allyloxycarbonyl (Alloc) as a versatile set of orthogonal protecting groups. The synthetic strategy made it possible for partial on-resin deprotection of the completed oligosaccharide thereby increasing the overall efficiency of the synthesis. The combination of classical and auxiliary mediated neighboring group participation for controlling anomeric selectivity is bringing the promise of routine automated solid supported oligosaccharides synthesis closer.

As many as 50% of human proteins are *O*- or *N*-glycosylated and the carbohydrate moieties of these glycoproteins have been implicated as essential mediators of cellular processes such as protein folding, regulation of cell signaling, fertilization, embryogenesis, neuronal development and hormone activities.<sup>1</sup> However, carbohydrates are also important for pathogen recognition, modulation of innate immune responses, control of immune cell homeostasis, inflammation, and the development of autoimmune diseases and cancer.<sup>2–4</sup> The ability of cells to generate information rich glycans has created a new field of research termed "glycomics", which seeks to identify and understand the processes involved in the formation of cell type and developmental stage specific oligosaccharide patterns.<sup>5–8</sup> In this respect, collections of well-defined oligosaccharides are needed for the development of algorithms for the assignment of oligosaccharide MS spectra, for fabricating microarrays,

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\*gjboons@ccrc.uga.edu.

**Author contributions.** T.J.B, J-H.K and G.J.B conceived and designed the experiments. T.J.B carried out the solid supported synthesis of **22** and **26** and analyzed the results. J-H.K carried out the solution phase synthesis of a pentaglycoside. J.P carried out the solution synthesis of a galactoside containing trisaccharide. T.J.B and G.J.B wrote the paper.

for elucidating the biosynthetic pathways of glycoconjugate assembly, and as immunogens to produce monoclonal antibodies (MABs) for glycoprotein visualization and isolation by immunoprecipitation. In many cases, well-defined oligosaccharides can only be obtained by chemical or enzymatic approaches and although tremendous progress has been made, complex oligosaccharide synthesis is still very time consuming and it is not uncommon that the preparation of a single well-defined derivative can take as much as a year.<sup>9–13</sup> Solid phase oligosaccharide synthesis (SPOS) offers the promise to increase the speed of oligosaccharide assembly, primarily by eliminating intermediate purification steps and by automation.<sup>14</sup> However, SPOS requires that each glycosidic linkage is introduced with absolute stereoselectivity otherwise an intractable mixture of compounds will be obtained after several reaction cycles.

In this respect, 1,2-*trans* glycosides can be reliably introduced by exploiting neighboring group participation of a 2-*O*-acyl functionality (Figure 1a). The Seeberger group exploited this type of anomeric control for the automated solid phase synthesis of a phytoalexin elicitor  $\beta$ -glycan using a modified peptide synthesizer.<sup>15</sup> The introduction of 1,2-*cis* glycosidic linkages, such as  $\alpha$ -glucosides and  $\alpha$ -galactosides, requires glycosyl donors having a non-assisting functionality at C-2, and in general these glycosylations give mixtures of anomers (Figure 1b).<sup>16</sup> Not surprisingly, only a few examples of SPOS of oligosaccharides containing 1,2-*cis* glycosides have been reported, which rely on tedious separation of the anomers by HPLC or the preparation of a 1,2-*cis*-linked disaccharide in solution, which after purification to remove the unwanted 1,2-*trans*-anomer, can be employed in solid phase synthesis.<sup>17–19</sup> Thus, a major stumbling block in SPOS is the inability to reliably introduce 1,2-*cis* glycosides with complete stereoselectivity.

Herein, we report the solid phase synthesis of a well-defined biologically important branched 1,2-*cis*-linked penta-glucoside. Anomeric control was achieved by employing neighboring group participation by a (*S*)-(phenylthiomethyl)benzyl chiral auxiliary at C-2 of the glucosyl donors.<sup>20,21</sup> In this approach, neighboring group participation by the C-2 auxiliary results in the formation of an anomeric sulfonium ion as a *trans*-decalin system because the alternative *cis*-decalin system will place the phenyl-substituent in an axial position inducing unfavorable steric interactions. Displacement of the equatorial anomeric sulfonium ion by a sugar alcohol will then lead to the formation of a 1,2-*cis* glycoside (Figure 1c). Furthermore, a branching unit could easily be installed by using the 9-fluorenylmethoxycarbonyl (Fmoc) and allyloxycarbonyl (Alloc) as a versatile set of orthogonal protecting groups.<sup>22,23</sup> Also, the synthetic strategy made it possible for partial on-resin deprotection of the completed oligosaccharide thereby increasing the overall efficiency of the synthesis.

## Results and Discussion

$\alpha$ -Glucans (Figure 2) are structural elements of an immuno-modulatory polysaccharide isolated from *Aconitum carmichaeli* that has the potential to be developed as an adjuvant.<sup>24</sup> The polysaccharide is composed of an  $\alpha$ (1,6)-linked glucosyl backbone branched with  $\alpha$ (1,3)-linked glucoside moieties.  $\alpha$ -Glucans have also been isolated from various microbial sources such as *Pseudallescheria boydii*, which produced a polysaccharide that was essential

for conidial phagocytosis by macrophages and induction of innate immune responses in a TLR2 dependent manner.<sup>25</sup> Furthermore, it has been found that the *Streptococcus pneumoniae*  $\alpha$ -glucan metabolizing machinery is an important virulence factor.<sup>26</sup>  $\alpha$ -Glucans isolated from natural sources are heterogeneous in composition and well-defined derivatives are required to identify biologically active fragments.

It was envisioned that pentasaccharide **22**, which represents a significant synthetic challenge due to the presence of multiple  $\alpha$ -glucosides at primary and secondary sugar alcohols and branched architecture, could be assembled from the four strategically selected glycosyl donors **1–4** and polystyrene modified resin **7** (Figure 3). In addition, galactosides **5a/b** were expected to provide opportunities to prepare structural analogs. Glycosyl donors **2–5a/b** are modified at C-2 with a (*S*)-(phenylthiomethyl)benzyl chiral auxiliary, and this functionality will ensure the stereoselective formation of  $\alpha$ -glucosides.<sup>21</sup> Furthermore, glycosyl donor **1**, which is modified with an acetyl ester at C-2, will be coupled with the benzyl alcohol linker modified resin **7** to give resin bound  $\beta$ -linked glucoside **8**. After completion of the synthesis, the resulting oligosaccharide can be cleaved from the resin by transesterification and the remaining anomeric 4-hydroxymethyl-benzyl ether can be removed during the final hydrogenation step providing the anomeric lactol. Thus, the anomeric identity of the glycosidic linkage will be lost and therefore it was installed as a straightforward  $\beta$ -glucoside. The temporary Fmoc carbonate of **2** and **4** can be cleaved under mild conditions using piperidine/DMF (1/9, v/v) and the use of this protecting group will make it possible to form the  $\alpha$ (1,6)-linked backbone. Furthermore, monosaccharide building block **2** has a temporary Alloc carbonate at C-3, which can be removed with Pd(PPh<sub>3</sub>)<sub>4</sub> in a mixture of THF and AcOH without affecting the Fmoc protecting group or the anomeric linker and allows for installment of the branching unit.

Thus, linker-modified resin **6** was prepared by ester formation between the carboxylic acids of carboxypolystyrene resin (Advanced ChemTech™, 2.0 mmol g<sup>-1</sup>) and (4-trityloxymethyl-phenyl)-methanol using *N,N'*-diisopropylcarbodiimide (DIC) and *N,N*-dimethyl-4-aminopyridine (DMAP). The remaining carboxylic acids were capped as methyl esters by subsequent addition of methanol. The trityl ether was removed using trifluoroacetic acid (TFA) in CH<sub>2</sub>Cl<sub>2</sub> (1/9, v/v) and triethylsilane as the scavenger to afford **7** (loading 0.148 mmol g<sup>-1</sup>).<sup>27</sup> It is important to note that different resin loadings could be achieved by employing different molar quantities of (4-trityloxymethyl-phenyl)-methanol. However, it was found that a loading of approximately 0.15 mmol g<sup>-1</sup> resin gives optimal glycosylation results and further increases in loading led to decreases in coupling efficiencies.<sup>27,28</sup>

Next, the hydroxyl of resin **7** was coupled with glycosyl donor **1** (3.0 equivalents) in the presence of a catalytic amount of trimethylsilyl trifluoromethanesulfonate (TMSOTf) in CH<sub>2</sub>Cl<sub>2</sub> at -40°C for 30 min to afford **8**. The glycosylation was repeated to ensure completion of the reaction. Next, the Fmoc protecting group of **8** was removed by treatment with piperidine/DMF (1/9, v/v) to give resin bound acceptor **9**, which was coupled with auxiliary containing glucosyl donor **2** to install the first 1,2-*cis* linkage. Thus, for this coupling, glucosyl donor **2** was preactivated in a separate flask with a stoichiometric amount of TMSOTf at -40°C to form an intermediate sulfonium ion. The solution containing the sulfonium ion was added *via* cannula to a cooled (-40°C) suspension of resin **9** and 2,6-di-

*tert*-butyl-4-methylpyridine (DTBMP) in CH<sub>2</sub>Cl<sub>2</sub>. The reaction mixture was allowed to warm to rt over a period of 5 h and was shaken at ambient temperature for an additional 11 h. After this period of time, a small sample of resin (approximately 5.0 mg) was analyzed for product formation and the possible presence of starting material by treatment with methanolic sodium methoxide to cleave the ester linkage of the linker followed by analysis of the released product by TLC and MALDI-TOF. This study showed that mainly the disaccharide product had been formed but also the presence of a small amount of starting material (5%). Therefore, the glycosylation was repeated under identical conditions to ensure that a homogeneous product would be formed at the end of the synthetic sequence.

The Alloc function of **10** was easily removed by treatment with Pd(PPh<sub>3</sub>)<sub>4</sub> in a mixture of THF and AcOH to give disaccharide acceptor **11**. Interestingly, attachment of **10** to the insoluble polymeric support did not appear to influence the outcome of the deprotection reaction. A number of attempts were made to couple the resulting glycosyl acceptor with glycosyl donor **3**. However, analysis of a small sample of resin indicated that no coupling had occurred. Subsequent model studies using solution phase chemistry indicated that the C-3' hydroxyl of **11** was sterically shielded due to the neighboring (*S*)-(phenylthiomethyl)benzyl ether. Thus, the auxiliary of **10** was converted into acetyl ester **12** by the treatment with acetic anhydride in the presence of BF<sub>3</sub>·OEt<sub>2</sub>. The Alloc of compound **12** could be removed under standard conditions to provide glycosyl acceptor **13**, which was coupled with preactivated **3** to give smooth formation of resin bound trisaccharide **14**. Thus, it was found that the auxiliary can be removed without affecting the Alloc and Fmoc protecting groups as well as the anomeric linker. The terminal Fmoc of **14** was cleaved by treatment with piperidine/DMF (1/9, v/v) and the resulting alcohol was glycosylated with preactivated **4** to afford tetrasaccharide **16**. The reaction sequence of Fmoc removal and glycosylation was repeated to afford the fully protected pentasaccharide **18**. Each glycosylation was performed twice with 2.0 equivalents of glycosyl donor to ensure complete conversion of the starting material. After each reaction step, a small sample of resin was treated with methanolic sodium methoxide and the resulting released material analyzed by TLC and MALDI-TOF MS. These studies showed clearly that each reaction step proceeded to completion with no or very little formation of side products. Furthermore, the pentaglycoside was also prepared in solution and careful examining of the products of each glycosylation confirmed complete α-anomeric selectivity (See supporting information).

Next, we explored whether pentasaccharide **18** could be partially deprotected when still attached to the resin. Thus, the auxiliaries of **18** were converted in the acetyl ester by treatment with acetic anhydride and BF<sub>3</sub>·OEt<sub>2</sub> in CH<sub>2</sub>Cl<sub>2</sub> to give **19**. Next, the Fmoc carbonate was removed with piperidine/DMF (1/9, v/v) to give **20**, which was released from the polymeric support with concomitant acetyl ester removal using methanolic sodium methoxide in CH<sub>2</sub>Cl<sub>2</sub>. The crude pentasaccharide was re-acetylated and then purified by size exclusion chromatography (LH-20). HPLC analysis of the resulting product showed **21** as the major products and the presence of a small amount to monodebenzylated material (~10%), which was probably formed during removal of the auxiliaries. Importantly, no anomeric isomers of **21** were detected (see supporting information S34). Further purification

by preparative HPLC gave pure **21** in an overall yield of 25%, which corresponds to a yield per step of 90% (thirteen on-resin steps).

The identity and purity of **21** was confirmed by  $^1\text{H}$  NMR and coupled HSQC experiments (Figure 4). The homonuclear anomeric coupling constants as well as the heteronuclear one bond  $\text{C}_1\text{--H}_1$  coupling constants unambiguously confirmed the presence of four 1,2-*cis* linkages and one 1,2-*trans* linkage. COSY, TOCSY and HSQC experiments were used for full spectral assignment (see supporting information S30, S31). Furthermore, an HMBC experiment confirmed the appropriate connectivity between the individual monosaccharides (see supporting information S32). Finally, **21** was converted into target compound **22** by removal of the acetyl esters using standard conditions followed by hydrogenation of the benzyl ethers using  $\text{Pd}(\text{OH})_2/\text{C}$  (20% wt) and  $\text{H}_2$  gas.

Next, attention was focused on the preparation of pentasaccharide **26** to demonstrate that the methodology can be extended to the stereoselective introduction of other types of monosaccharides and can be employed for the preparation of complex structural analogs (Figure 5). Compound **26** contains an  $\alpha$ -galactoside at the C-3 branching position, which is challenging to introduce due the relative low reactivity of the corresponding glycosyl acceptor. Furthermore, branching points of  $\alpha$ -glucans are expected to be critical for biological activity and hence compounds such as **26** provide an opportunity to explore the influence of subtle structural changes on biological activity.

Solution phase studies showed that glycosylations with galactosyl donors **5a,b** led to the exclusive formation of the corresponding  $\alpha$ -galactosides. It was found that the sulfonium ion of the tri-*O*-benzoyl protected derivative **5b** was stable at ambient temperature whereas acetylated derivative **5a** required glycosylation at  $-20^\circ\text{C}$  for optimal results. Thus, the more convenient to use galactosyl donor **5b** was activated with a stoichiometric amount of TfOH at  $-40^\circ\text{C}$  to form the intermediate sulfonium ion, which was added to a cooled ( $-40^\circ\text{C}$ ) suspension of resin bound **13** and 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP) in  $\text{CH}_2\text{Cl}_2$  (Figure 5). The reaction mixture was allowed to warm to rt over a period of 5 h and was shaken at ambient temperature for an additional 11 h. A repetition of the glycosylation led to clean formation of trisaccharide **23** as judged by TLC and MS analysis of a small sample of oligosaccharide released from the resin by NaOMe treatment. Removal of the Fmoc protecting group of **23** followed by glucosylation with **4** and a repetition of the reaction sequence led to the formation of polymer bound pentasaccharide, which was partially deprotected and released from the resin employing standard procedures. Purification as before afforded anomERICALLY pure **25** in an overall yield of 13%, which corresponds to a yield per step of 86% (thirteen on-resin steps). The same analytical procedures were used to confirm the purity and identity of **25** (see supporting information S36– S41). Finally, **25** was deprotected under the same conditions used before to afford **26**.

In conclusion, glycosyl donors having a (*S*)-(phenylthiomethyl)benzyl chiral auxiliary at C-2 have been successfully employed for the solid supported synthesis of complex branched oligosaccharides. To the best of our knowledge, this is the first example of a stereoselective solid supported synthesis of an oligosaccharide having multiple 1,2-*cis* glycosidic linkages. A particular interesting feature was that a relatively small excess of glycosyl donor was

required to drive the glycosylations to completion. Probably, the intermediate sulfonium ion is sufficiently stable to diffuse into the polymer support for glycosylation of the resin-bound sugar hydroxyls. Furthermore, it has been found that Fmoc and Alloc is an attractive set of orthogonal protecting groups for solid supported synthesis, which is compatible with the auxiliary based glycosylation methodology. Deprotection of the fully assembled oligosaccharide could partially be performed when still attached to the resin thereby further reducing the number of purification steps. The convenient protection of monosaccharides by a one-pot multi-step approach,<sup>29</sup> combined with classical and auxiliary mediated neighboring group participation for controlling anomeric selectivity, is bringing the promise of routine automated solid supported oligosaccharides synthesis closer. Such an approach can deliver libraries of well-defined oligosaccharides needed for glycomics research.

## Methods

### General procedures

<sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian inova-300 (300/75 MHz), a Varian inova-500 (500/125 MHz) and a Varian inova-600 (600/150 MHz) spectrometer equipped with sun workstations. Chemical shifts are reported in parts per million (ppm) relative to tetramethylsilane (TMS) as the internal standard. NMR data is presented as follows: Chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, dd = doublet of doublet, m = multiplet and/or multiple resonances), coupling constant in Hertz (Hz), integration. All NMR signals were assigned on the basis of <sup>1</sup>H NMR, <sup>13</sup>C NMR, COSY and HSQC experiments. HPLC analysis was performed using an Agilent Technologies 1200 series HPLC system with UV detection at 250 nm. Optical rotations were measured using a Jasco P-1020 polarimeter. Mass spectra were recorded on an Applied Biosystems 4700 MALDI-TOF proteomics analyzer. The matrix used was 2,5-dihydroxy-benzoic acid (DHB) and ultamark 1621 as the internal standard. Column chromatography was performed on silica gel G60 (Silicycle, 60–200 μm, 60 Å). TLC-analysis was conducted on Silicagel 60 F<sub>254</sub> (EMD Chemicals inc.) with detection by UV-absorption (254nm) were applicable, and by spraying with 20% sulfuric acid in ethanol followed by charring at ~150°C or by spraying with a solution of (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>·H<sub>2</sub>O (25 g/L) in 10% sulfuric acid in ethanol followed by charring at ~150°C. CH<sub>2</sub>Cl<sub>2</sub> was freshly distilled from calcium hydride under nitrogen prior to use. Molecular sieves (4Å) were flame activated under vacuum prior to use. Solid phase reactions were shaken using an IKA labortechnik KS 125 shaker. All reactions were carried out under an argon atmosphere.

### General procedure for on-resin glycosylation with glycosyl donor 1

Glycosyl donor (0.3 mmol) was added to a suspension of the resin-bound glycosyl acceptor (0.670 g, 0.1 mmol, 0.148 mmol g<sup>-1</sup>) and activated 4Å molecular sieves in CH<sub>2</sub>Cl<sub>2</sub> (6 mL). The mixture was shaken for 15 min at rt before being cooled to -40°C. TMSOTf (8.15 μL, 0.045 mmol) was added at this temperature and the mixture was shaken for 30 min. The resin was decanted into a filter to remove the molecular sieves, washed with CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), MeOH (2 × 5 mL), CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), and MeOH (2 × 5 mL) followed by drying under vacuum in a desiccator for 16 h. This procedure was repeated to complete one coupling cycle.

### General procedure for on-resin glycosylation with (S)-(phenylthiomethyl)benzyl containing glycosyl donor 2, 3, 4 or 5b

Glycosyl donor (0.2 mmol) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (2 mL) and activated 4Å molecular sieves were added. The mixture was stirred for 15 min at rt before being cooled to -40°C. TMSOTf (36.2 µl, 0.2 mmol for **2,3** and **4**) or TfOH (17.6 µl, 0.2 mmol for **5b**) was added at this temperature and the mixture was stirred at -40°C for 15 min. The mixture containing the activated donor was transferred *via* cannula to a cooled (-40°C) flask containing the resin-bound glycosyl acceptor (0.670 g, 0.1 mmol, 0.148 mmol g<sup>-1</sup>), DTBMP (82.4 mg, 0.4 mmol), activated 4Å molecular sieves and CH<sub>2</sub>Cl<sub>2</sub> (6 mL). The mixture was shaken and allowed to slowly warm to rt during 5 h after which it was shaken for an additional 11 h at rt. The resin was decanted into a filter to remove the molecular sieves, washed with CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), MeOH (2 × 5 mL), CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), and MeOH (2 × 5 mL) followed by drying under vacuum in a desiccator for 16 h. This procedure was repeated to complete one coupling cycle.

### General procedure for Fmoc cleavage

The resin (0.670 g, 0.1 mmol, 0.148 mmol g<sup>-1</sup>) was allowed to swell in DMF (7.0 mL) for 5 min. Piperidine (0.7 mL) was added and the mixture was shaken for 5 min at rt. The resin was filtered, washed with CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), MeOH (2 × 5 mL), CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), and MeOH (2 × 5 mL) followed by drying under vacuum in a desiccator for 16 h.

### General procedure for (S)-(phenylthiomethyl)benzyl cleavage

The resin (0.670 g, 0.1 mmol, 0.148 mmol g<sup>-1</sup>) was allowed to swell in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) for 5 min at rt. Acetic anhydride (3 mL) was added and the mixture was cooled to 0°C. BF<sub>3</sub>·Et<sub>2</sub>O (50.0 µl 0.4 mmol) was added and the mixture was shaken at 0°C for 16 h. The resin was filtered, washed with CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), MeOH (2 × 5 mL), CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL), and MeOH (2 × 5 mL) followed by drying under vacuum in a desiccator for 16 h.

### General procedure for Alloc cleavage

The resin (0.670 g, 0.1 mmol, 0.148 mmol g<sup>-1</sup>) was allowed to swell in a mixture of THF (7 mL) and acetic acid (0.7 mL) for 5 min at rt. The solution was purged with argon gas for 2 min followed by the addition of Pd(PPh<sub>3</sub>)<sub>4</sub> (46.2 mg, 0.04 mmol). The mixture was shaken for 16 h at rt. The resin was filtered, washed with THF (2 × 5 mL), MeOH (2 × 5 mL), THF (2 × 5 mL), and MeOH (2 × 5 mL) followed by drying under vacuum in a desiccator for 16 h.

### General procedure for product cleavage from the resin

The resin (0.670 g, 0.1 mmol, 0.148 mmol g<sup>-1</sup>) was allowed to swell in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) for 5 min at rt. MeOH (5 mL) and NaOMe (27.0 mg, 0.5 mmol) was added and the mixture was shaken for 3 h at rt. The resin was filtered and washed with a mixture of CH<sub>2</sub>Cl<sub>2</sub>/MeOH 1/1 (4 × 5 mL). This procedure was repeated three times to ensure complete product cleavage and the combined filtrates were neutralized using Dowex® 50W X8-200 H<sup>+</sup> resin. The resin was removed by filtration and the filtrate was concentrated under reduced pressure.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

## Acknowledgement

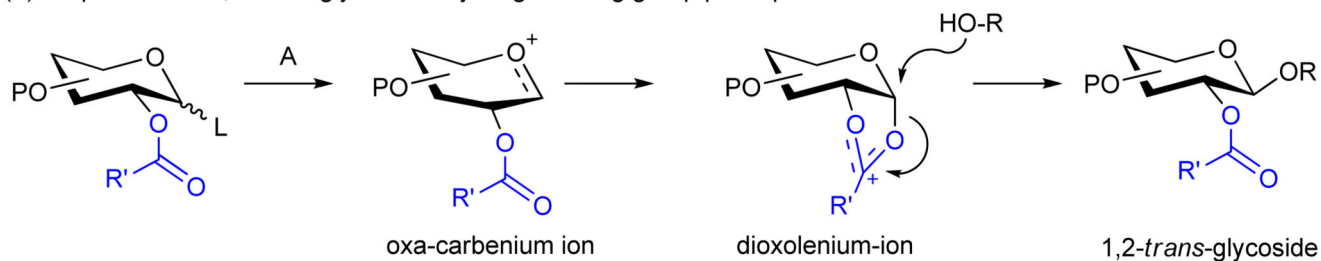
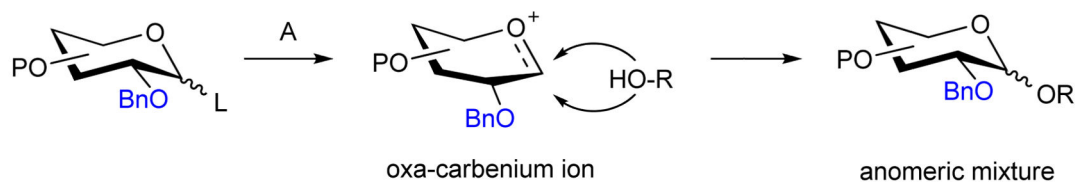
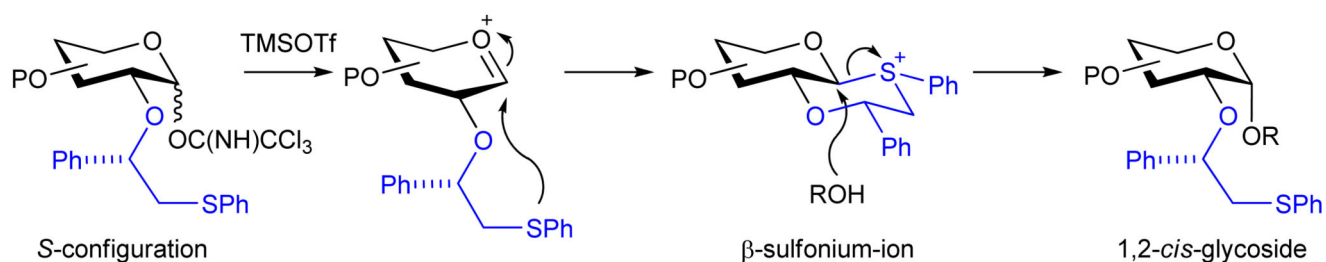
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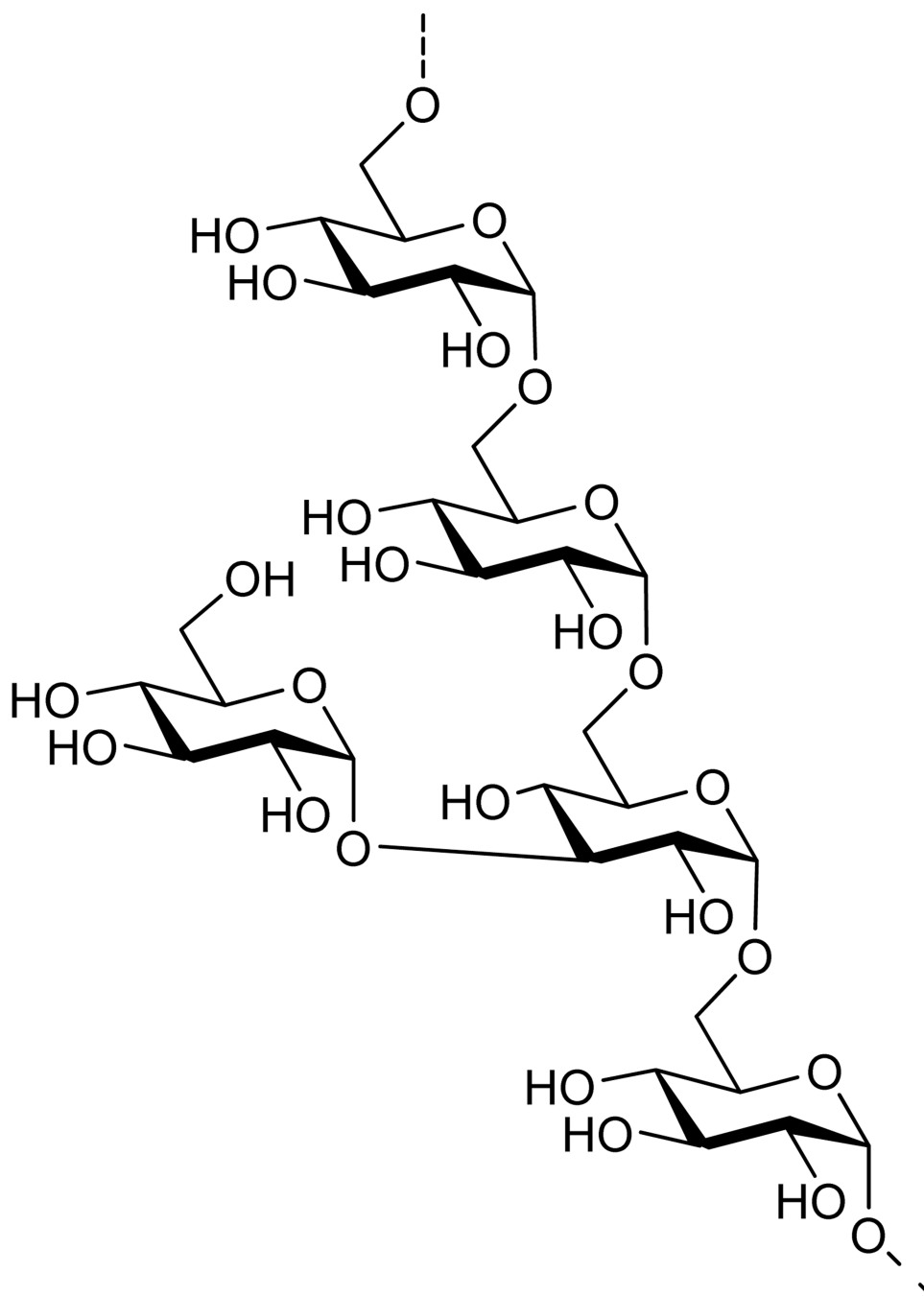
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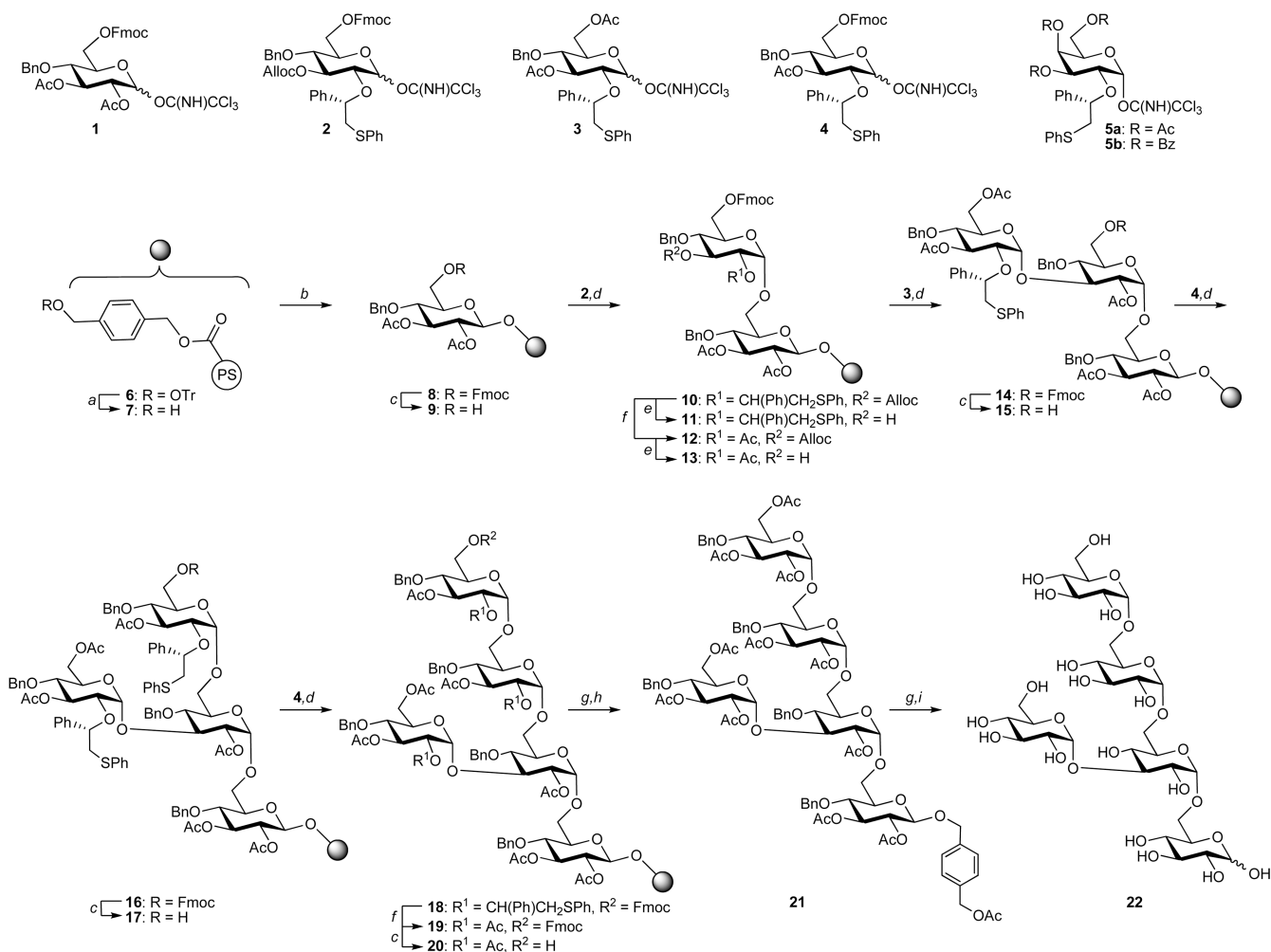
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(a) Preparation of 1,2-*trans* glycosides by neighbouring group participation(b) Preparation of 1,2-*cis* glycosides by employing non-participating C-2 protecting groups(c) Auxiliary controlled installation of 1,2-*cis*-glycosides**Figure 1.**

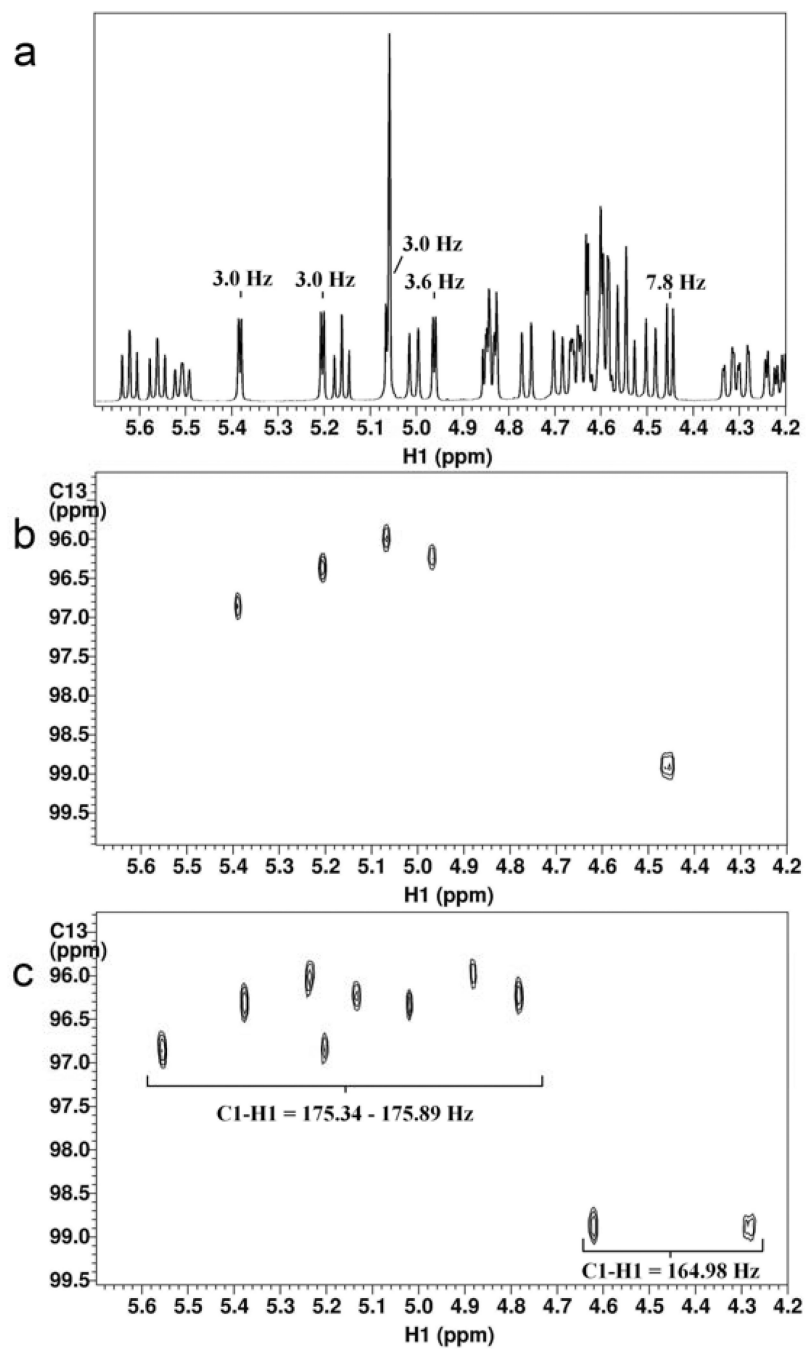
Stereoselective introduction of glycosidic linkages. a, Preparation of 1,2-*trans*-glycosides by neighboring group participation of a C-2 ester. Activation of an anomeric-leaving group by a promoter results in its departure and the formation of an oxacarbenium ion. Subsequent, neighboring group participation by the 2-*O*-acyl protecting group will give a more stable five-membered dioxolenium-ion, which can only be formed as a 1,2-*cis* fused ring system. An alcohol can attack the anomeric center of the dioxolenium -ion from only one face providing a 1,2-*trans*-glycoside. b, The use of a non-participating protecting groups at C-2 of a glycosyl donor generally results in the formation of mixture of anomers although the axial glycoside often predominates. c, Chiral auxiliary controlled installation of 1,2-*cis*-glycosides. An intermediate  $\beta$ -sulfonium ion forces the incoming alcohol to attack the  $\alpha$ -face resulting in the stereoselective introduction of a 1,2-*cis*-glycoside.



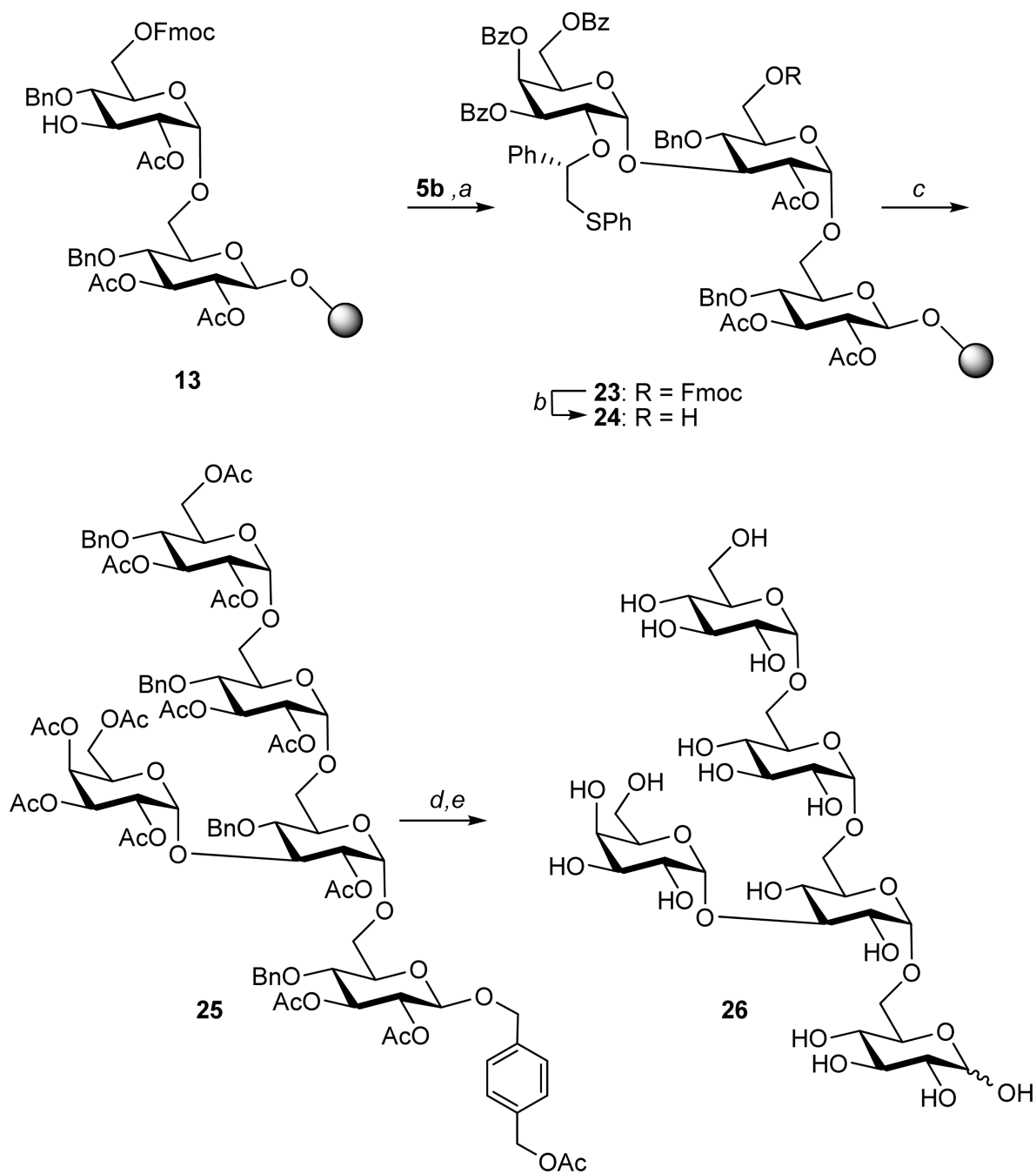
**Figure 2.** Structure of an  $\alpha$ -glucan pentasaccharide repeating unit found in *Aconitum carmichaeli*. This adjuvant candidate is a significant synthetic challenge for SPOS due to the multiple  $\alpha$ -1,6-linked glucosides in the backbone and the  $\alpha$ -1,3-glycosidic branching point.

**Figure 3.**

The stereoselective solid supported synthesis of pentasaccharide **22** using monosaccharide building blocks **1–5** and linker modified resin **7**. Reagents and conditions: a) TFA/CH<sub>2</sub>Cl<sub>2</sub> (1/9, v/v), Et<sub>3</sub>SiH, 10 min, rt. b) **1**, TMSOTf, CH<sub>2</sub>Cl<sub>2</sub>, MS4Å, 30 min, –40°C, double coupling. c) piperidine/DMF (1/9, v/v), 5 min, rt. d) **2**, **3** or **4**, TMSOTf, CH<sub>2</sub>Cl<sub>2</sub>, MS4Å, 15 min, –40°C then added to **9**, **13**, **15** or **17**, DTBMP, CH<sub>2</sub>Cl<sub>2</sub>, MS4Å, 16 h –40°C → rt, double coupling. e) Pd(PPh<sub>3</sub>)<sub>4</sub> (40 mol%), THF/AcOH (10/1, v/v), 16 h, rt. f) BF<sub>3</sub>·Et<sub>2</sub>O, Ac<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub> (1/2, v/v), 16 h, 0°C. g) NaOMe, MeOH/CH<sub>2</sub>Cl<sub>2</sub> (1/1, v/v), 9 h for **21**, 16 h for **22**, rt. h) Ac<sub>2</sub>O/pyridine (1/3, v/v), 12 h, rt. i) Pd(OH)<sub>2</sub>/C (20 wt%), H<sub>2</sub>, EtOH/H<sub>2</sub>O (1/1, v/v), 16 h, rt. MS = molecular sieves, PS = polystyrene, rt = room temperature, Tr = Trityl.



**Figure 4.** NMR data of pentasaccharide **21**: a)  $^1\text{H}$  NMR spectrum b) decoupled HSQC c) coupled HSQC. The homonuclear and heteronuclear coupling constants confirm the correct anomeric configuration of the product.



**Figure 5.**

The stereoselective solid supported synthesis of galactoside containing analogue **26**. The successful preparation of this derivative demonstrates that the methodology can be employed for a variety of glycosyl donors. Reagents and conditions: a) **5b**, TfOH, CH<sub>2</sub>Cl<sub>2</sub>, MS4Å, 15 min, -40°C then added to **13**, DTBMP, CH<sub>2</sub>Cl<sub>2</sub>, MS4Å, 16 h -40°C → rt, double coupling. b) piperidine/DMF, (1/9 v/v), 5 min, rt. c) same sequence and reagents as used for the conversion of **15** to **20**. d) NaOMe, MeOH/CH<sub>2</sub>Cl<sub>2</sub> (1/1, v/v), 8 h, rt. e)

$\text{Pd}(\text{OH})_2/\text{C}$  (20 wt%),  $\text{H}_2$ , EtOH/ $\text{H}_2\text{O}$  (1/1, v/v), 16 h, rt. MS = molecular sieves, rt = room temperature.

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