



Crystal structure of methyl 2-hydroxy-5-[(4-oxo-4,5-dihydro-1,3-thiazol-2-yl)-amino]benzoate

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Received 26 March 2015; accepted 30 March 2015

Edited by P. C. Healy, Griffith University, Australia

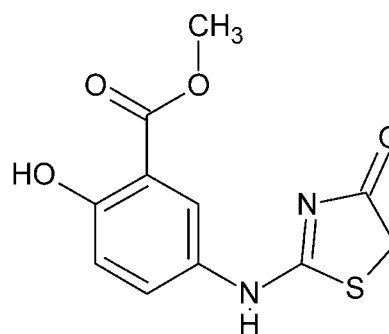
The title compound, C₁₁H₁₀N₂O₄S, crystallized with two independent molecules (*A* and *B*) in the asymmetric unit. They differ primarily in the rotational orientation of the five-membered heterocyclic ring. In molecule *A* this ring is inclined to the benzene ring by 48.17 (8)°, while in molecule *B* the same dihedral angle is 23.07 (8)°. In each molecule there is an intramolecular O—H...O hydrogen bond involving the adjacent hydroxyl group and the ester carbonyl O atom. In the crystal, the *A* molecules are linked *via* pairs of N—H...N hydrogen bonds, forming inversion dimers. These dimers are linked to the *B* molecules *via* N—H...O, C—H...O and C—H...S hydrogen bonds forming corrugated sheets lying parallel to (102).

Keywords: crystal structure; aminosalicic acid; thiazolidinones; hydrogen bonding.

CCDC reference: 1056711

1. Related literature

For pharmaceutical and chemotherapeutic properties of amino salicylic acid derivatives, see: Abdel-Alim *et al.* (2005); Abdu-Allah *et al.* (2005); Koelink *et al.* (2010). For general biological activities of thiazolidinone scaffold compounds, see: Tripathi *et al.* (2014).



2. Experimental

2.1. Crystal data

C ₁₁ H ₁₀ N ₂ O ₄ S	<i>V</i> = 2302.24 (10) Å ³
<i>M_r</i> = 266.27	<i>Z</i> = 8
Monoclinic, <i>P</i> 2 ₁ / <i>c</i>	Cu <i>K</i> α radiation
<i>a</i> = 4.7787 (1) Å	<i>μ</i> = 2.62 mm ⁻¹
<i>b</i> = 25.4128 (7) Å	<i>T</i> = 150 K
<i>c</i> = 18.9599 (5) Å	0.16 × 0.12 × 0.09 mm
<i>β</i> = 90.841 (1)°	

2.2. Data collection

Bruker D8 VENTURE PHOTON	17982 measured reflections
100 CMOS diffractometer	4582 independent reflections
Absorption correction: multi-scan	3972 reflections with <i>I</i> > 2σ(<i>I</i>)
(<i>SADABS</i> ; Bruker, 2014)	<i>R</i> _{int} = 0.031
<i>T</i> _{min} = 0.76, <i>T</i> _{max} = 0.80	

2.3. Refinement

<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.034	327 parameters
<i>wR</i> (<i>F</i> ²) = 0.089	H-atom parameters constrained
<i>S</i> = 1.04	Δ <i>ρ</i> _{max} = 0.29 e Å ⁻³
4582 reflections	Δ <i>ρ</i> _{min} = -0.26 e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1—H1A...O2	0.84	1.87	2.6287 (18)	150
O5—H5A...O6	0.84	1.92	2.6619 (18)	147
N1—H1N...N2 ⁱ	0.91	1.96	2.8624 (19)	175
N3—H3N...O4 ⁱⁱ	0.91	1.97	2.8703 (18)	170
C11—H11A...O8 ⁱⁱⁱ	0.99	2.39	3.371 (2)	171
C5—H5...S2 ^{iv}	0.95	2.71	3.5043 (18)	141

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $x + 1, y, z$; (iii) $x - 1, -y + \frac{1}{2}, z + \frac{1}{2}$; (iv) $-x + 2, -y + 1, -z + 1$.

Data collection: *APEX2* (Bruker, 2014); cell refinement: *SAINT* (Bruker, 2014); data reduction: *SAINT*; program(s) used to solve structure: *SHELXT* (Sheldrick, 2015a); program(s) used to refine structure: *SHELXL2014* (Sheldrick, 2015b); molecular graphics: *DIAMOND* (Brandenburg & Putz, 2012); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

Acknowledgements

The support of NSF–MRI grant No. 1228232 for the purchase of the diffractometer and Tulane University for support of the Tulane Crystallography Laboratory are gratefully acknowledged.

Supporting information for this paper is available from the IUCr electronic archives (Reference: HG5436).

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supporting information

Acta Cryst. (2015). E71, o282–o283 [doi:10.1107/S2056989015006416]

Crystal structure of methyl 2-hydroxy-5-[(4-oxo-4,5-dihydro-1,3-thiazol-2-yl)amino]benzoate

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S1. Comment

5-Aminosalicylic acid (5-ASA) is a prototype drug that is commonly described for treatment of inflammatory bowel diseases (Abdel-Alim, *et al.*, 2005; Abdu-Allah, *et al.*, 2005). It was shown that 5-ASA has, also, chemopreventive and chemotherapeutic properties (Koelink, *et al.*, 2010). On the other hand 4-thiazolidinone derivatives have attracted continuing interest over the years because of their diverse biological activities, such as anti-inflammatory, anti-proliferative, antiviral, anticonvulsant, anti-diabetic, anti-hyperlipidemic, cardiovascular, anti-tubercular, antifungal, and antibacterial (Tripathi, *et al.*, 2014). Based in these findings, we were interested in the synthesis of hybrid molecules that combine both pharmacophores, therefore we report in this study the synthesis and crystal structure of the title compound.

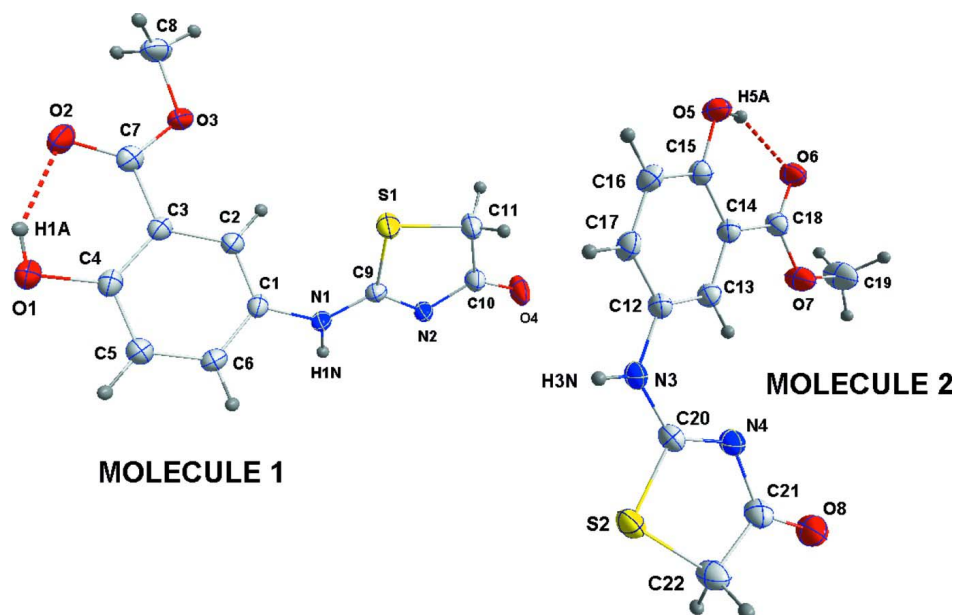
The title compound contains two independent molecules in the asymmetric unit which differ primarily in the rotational orientation of the 5-membered, heterocyclic ring (Fig. 1). Each molecule contains a strong, intramolecular hydrogen bond (Table 1 and Fig. 1) which determines the orientation of the ester group. The molecules pack in a zigzag fashion (Fig. 3) assembled by intermolecular N—H \cdots O, N—H \cdots N, C—H \cdots O and C—H \cdots S interactions (Table 1 and Fig. 2)

S2. Experimental

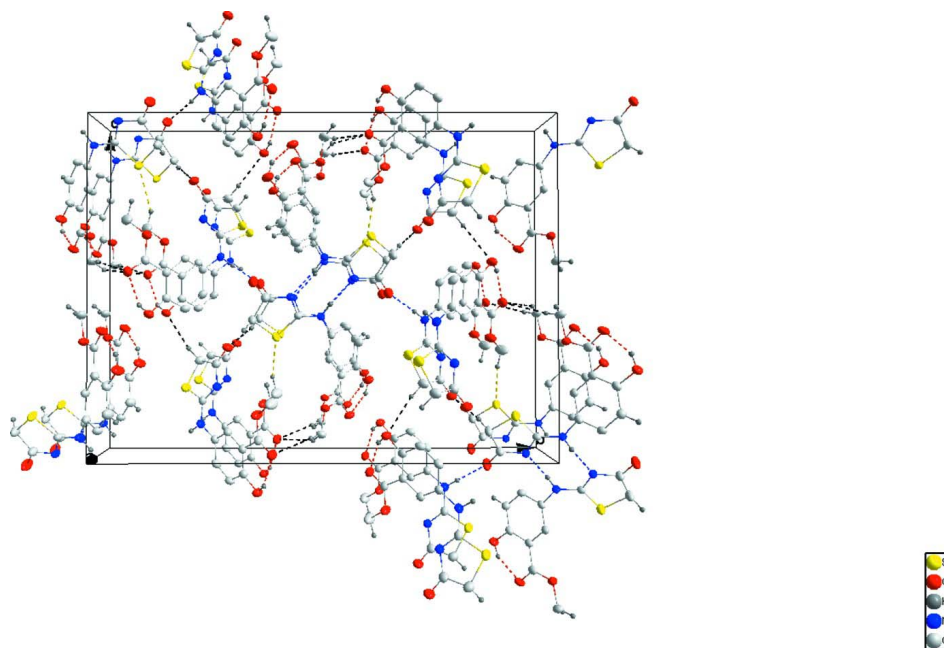
A solution of methyl 5-[(chloroacetyl)amino]-2-hydroxybenzoate (2.3 g, 9.5 mmol) and ammonium thiocyanate (1.5 g, 19.7 mmol) in 40 ml ethanol was refluxed for 3 h and allowed to stand overnight. The mixture was evaporated and the residue was washed with water and then recrystallized from ethanol/water to give the title compound (2.13 g, 85% yield); $R_f = 0.25$ (hexane:ethyl acetate, 2:1). Mp. 481–482 K.

S3. Refinement

H-atoms attached to carbon were placed in calculated positions (C—H = 0.95 - 0.98 Å) while those attached to nitrogen and oxygen were placed in locations derived from a difference map, refined initially to verify their presence and then their parameters adjusted to give N—H = 0.91 Å and O—H = 0.84 Å. All were included as riding contributions with isotropic displacement parameters 1.2 - 1.5 times those of the attached atoms.

**Figure 1**

The asymmetric unit for the title compound with labeling scheme and 50% probability ellipsoids. The intramolecular hydrogen bonds are shown as dotted lines.

**Figure 2**

Packing viewed down the *a* axis with O—H...O (red) N—H...O (blue), N—H...N (blue), C—H...O (black) and C—H...S (yellow) interactions shown as dotted lines.

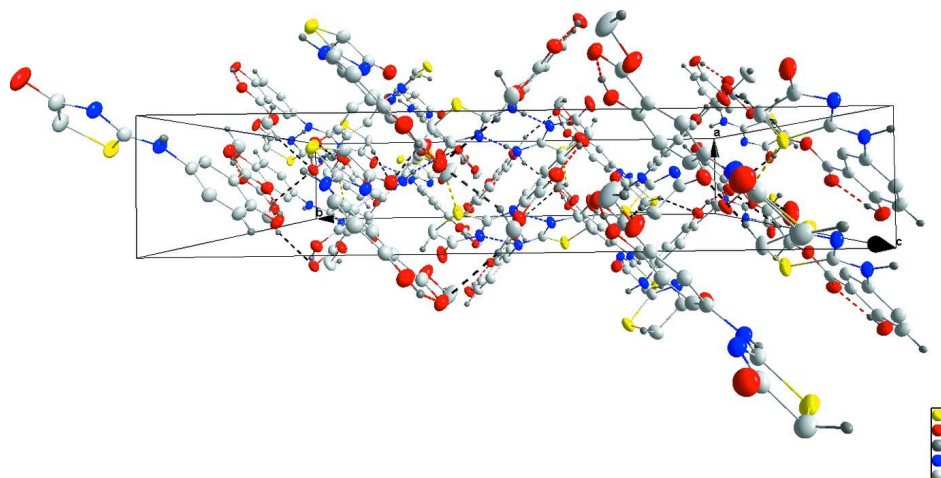


Figure 3

Packing viewed down the *c* axis. Key to dotted lines as for Figure 2.

Methyl 2-hydroxy-5-[(4-oxo-4,5-dihydro-1,3-thiazol-2-yl)amino]benzoate

Crystal data

$C_{11}H_{10}N_2O_4S$

$M_r = 266.27$

Monoclinic, $P2_1/c$

$a = 4.7787$ (1) Å

$b = 25.4128$ (7) Å

$c = 18.9599$ (5) Å

$\beta = 90.841$ (1)°

$V = 2302.24$ (10) Å³

$Z = 8$

$F(000) = 1104$

$D_x = 1.536$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54178$ Å

Cell parameters from 9975 reflections

$\theta = 4.2\text{--}74.5^\circ$

$\mu = 2.62$ mm⁻¹

$T = 150$ K

Block, yellow-orange

$0.16 \times 0.12 \times 0.09$ mm

Data collection

Bruker D8 VENTURE PHOTON 100 CMOS
diffractometer

Radiation source: INCOATEC $I\mu$ S micro-focus
source

Mirror monochromator

Detector resolution: 10.4167 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2014)

$T_{\min} = 0.76$, $T_{\max} = 0.80$

17982 measured reflections

4582 independent reflections

3972 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.031$

$\theta_{\max} = 74.5^\circ$, $\theta_{\min} = 2.9^\circ$

$h = -5 \rightarrow 5$

$k = -31 \rightarrow 31$

$l = -21 \rightarrow 23$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.034$

$wR(F^2) = 0.089$

$S = 1.04$

4582 reflections

327 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: mixed

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0435P)^2 + 1.0566P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.29$ e Å⁻³

$\Delta\rho_{\min} = -0.26$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger. H-atoms attached to carbon were placed in calculated positions (C—H = 0.95 - 0.98 Å) while those attached to nitrogen and oxygen were placed in locations derived from a difference map, refined initially to verify their presence and then their parameters adjusted to give N—H = 0.91 Å and O—H = 0.84 Å. All were included as riding contributions with isotropic displacement parameters 1.2 - 1.5 times those of the attached atoms.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.20576 (11)	0.60092 (2)	0.64250 (2)	0.03308 (12)
O1	0.8378 (3)	0.39435 (5)	0.78252 (7)	0.0340 (3)
H1A	0.7774	0.4013	0.8228	0.041*
O2	0.5442 (3)	0.43997 (5)	0.88146 (6)	0.0313 (3)
O3	0.1993 (3)	0.49520 (5)	0.84894 (6)	0.0296 (3)
O4	-0.2571 (3)	0.64326 (5)	0.48484 (7)	0.0460 (4)
N1	0.1989 (3)	0.50385 (5)	0.58425 (7)	0.0244 (3)
H1N	0.1522	0.4827	0.5471	0.029*
N2	-0.0532 (3)	0.56801 (5)	0.52659 (7)	0.0263 (3)
C1	0.3659 (4)	0.48031 (6)	0.63838 (8)	0.0234 (3)
C2	0.3104 (3)	0.48582 (6)	0.70909 (8)	0.0235 (3)
H2	0.1641	0.5085	0.7237	0.028*
C3	0.4698 (3)	0.45798 (6)	0.75962 (8)	0.0224 (3)
C4	0.6808 (4)	0.42367 (6)	0.73737 (9)	0.0253 (3)
C5	0.7330 (4)	0.41851 (7)	0.66549 (9)	0.0295 (4)
H5	0.8772	0.3957	0.6501	0.035*
C6	0.5769 (4)	0.44630 (7)	0.61680 (9)	0.0264 (3)
H6	0.6131	0.4423	0.5680	0.032*
C7	0.4110 (4)	0.46286 (6)	0.83537 (9)	0.0244 (3)
C8	0.1382 (4)	0.50350 (7)	0.92288 (9)	0.0341 (4)
H8A	0.1074	0.4695	0.9458	0.051*
H8B	-0.0304	0.5252	0.9269	0.051*
H8C	0.2964	0.5215	0.9459	0.051*
C9	0.1095 (4)	0.55291 (6)	0.58058 (8)	0.0235 (3)
C10	-0.1141 (4)	0.62053 (7)	0.52919 (9)	0.0303 (4)
C11	0.0118 (4)	0.64938 (7)	0.59232 (10)	0.0337 (4)

H11A	-0.1376	0.6650	0.6213	0.040*
H11B	0.1374	0.6779	0.5765	0.040*
S2	0.79895 (10)	0.68556 (2)	0.30469 (2)	0.03288 (12)
O5	-0.1163 (3)	0.86366 (5)	0.57459 (6)	0.0311 (3)
H5A	-0.2396	0.8788	0.5500	0.037*
O6	-0.3730 (3)	0.89682 (5)	0.45724 (7)	0.0315 (3)
O7	-0.2248 (3)	0.86409 (5)	0.35416 (6)	0.0359 (3)
O8	0.4930 (3)	0.78782 (5)	0.17658 (7)	0.0369 (3)
N3	0.4978 (3)	0.72813 (5)	0.40544 (7)	0.0266 (3)
H3N	0.5957	0.7034	0.4303	0.032*
N4	0.4467 (3)	0.76550 (5)	0.29297 (7)	0.0273 (3)
C12	0.3322 (3)	0.76357 (6)	0.44516 (9)	0.0244 (3)
C13	0.1336 (3)	0.79618 (6)	0.41508 (9)	0.0242 (3)
H13	0.1002	0.7952	0.3656	0.029*
C14	-0.0191 (3)	0.83065 (6)	0.45723 (8)	0.0232 (3)
C15	0.0258 (3)	0.83183 (6)	0.53042 (9)	0.0249 (3)
C16	0.2281 (4)	0.79875 (7)	0.55986 (9)	0.0284 (4)
H16	0.2617	0.7993	0.6094	0.034*
C17	0.3800 (4)	0.76519 (7)	0.51822 (9)	0.0276 (4)
H17	0.5181	0.7430	0.5391	0.033*
C18	-0.2245 (4)	0.86683 (6)	0.42429 (9)	0.0261 (3)
C19	-0.4239 (5)	0.89664 (8)	0.31670 (11)	0.0464 (5)
H19A	-0.6106	0.8913	0.3361	0.070*
H19B	-0.4260	0.8871	0.2666	0.070*
H19C	-0.3704	0.9337	0.3219	0.070*
C20	0.5553 (3)	0.73108 (6)	0.33708 (9)	0.0250 (3)
C21	0.5600 (4)	0.76067 (7)	0.22716 (9)	0.0283 (4)
C22	0.7842 (4)	0.71821 (7)	0.22087 (9)	0.0330 (4)
H22A	0.9675	0.7342	0.2100	0.040*
H22B	0.7338	0.6930	0.1829	0.040*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0490 (3)	0.0229 (2)	0.0269 (2)	0.00344 (17)	-0.01520 (19)	-0.00436 (15)
O1	0.0388 (7)	0.0389 (7)	0.0241 (6)	0.0151 (5)	-0.0018 (5)	0.0060 (5)
O2	0.0332 (7)	0.0394 (7)	0.0213 (6)	0.0014 (5)	-0.0020 (5)	0.0078 (5)
O3	0.0397 (7)	0.0296 (6)	0.0195 (6)	0.0054 (5)	0.0023 (5)	-0.0006 (5)
O4	0.0700 (10)	0.0286 (7)	0.0385 (8)	0.0144 (6)	-0.0250 (7)	0.0000 (6)
N1	0.0338 (8)	0.0220 (6)	0.0172 (6)	0.0042 (5)	-0.0053 (5)	-0.0024 (5)
N2	0.0373 (8)	0.0222 (6)	0.0193 (7)	0.0057 (6)	-0.0067 (6)	-0.0019 (5)
C1	0.0280 (9)	0.0206 (7)	0.0213 (8)	0.0010 (6)	-0.0042 (6)	0.0018 (6)
C2	0.0275 (9)	0.0209 (7)	0.0220 (8)	0.0015 (6)	-0.0007 (6)	-0.0007 (6)
C3	0.0254 (8)	0.0219 (7)	0.0200 (8)	-0.0018 (6)	-0.0013 (6)	0.0014 (6)
C4	0.0271 (9)	0.0251 (8)	0.0236 (8)	0.0019 (6)	-0.0029 (6)	0.0040 (6)
C5	0.0310 (9)	0.0315 (9)	0.0262 (9)	0.0087 (7)	0.0010 (7)	0.0005 (7)
C6	0.0317 (9)	0.0291 (8)	0.0185 (8)	0.0032 (7)	0.0010 (6)	-0.0006 (6)
C7	0.0271 (9)	0.0231 (7)	0.0231 (8)	-0.0051 (6)	-0.0005 (6)	0.0013 (6)

C8	0.0461 (11)	0.0347 (9)	0.0216 (9)	0.0007 (8)	0.0059 (8)	-0.0030 (7)
C9	0.0301 (9)	0.0228 (7)	0.0175 (7)	0.0012 (6)	-0.0018 (6)	-0.0013 (6)
C10	0.0413 (10)	0.0240 (8)	0.0253 (9)	0.0047 (7)	-0.0066 (7)	-0.0003 (7)
C11	0.0498 (12)	0.0216 (8)	0.0293 (9)	0.0050 (7)	-0.0116 (8)	-0.0010 (7)
S2	0.0378 (3)	0.0296 (2)	0.0314 (2)	0.01298 (17)	0.00235 (18)	0.00021 (17)
O5	0.0353 (7)	0.0344 (6)	0.0237 (6)	0.0065 (5)	0.0015 (5)	-0.0028 (5)
O6	0.0347 (7)	0.0282 (6)	0.0317 (7)	0.0084 (5)	0.0001 (5)	-0.0025 (5)
O7	0.0455 (8)	0.0370 (7)	0.0250 (6)	0.0180 (6)	-0.0089 (5)	-0.0021 (5)
O8	0.0466 (8)	0.0390 (7)	0.0250 (6)	0.0061 (6)	-0.0015 (5)	0.0053 (5)
N3	0.0309 (8)	0.0236 (7)	0.0253 (7)	0.0076 (6)	-0.0014 (6)	0.0039 (5)
N4	0.0316 (8)	0.0259 (7)	0.0242 (7)	0.0052 (6)	-0.0002 (6)	0.0011 (5)
C12	0.0265 (9)	0.0217 (7)	0.0250 (8)	0.0011 (6)	0.0018 (6)	0.0017 (6)
C13	0.0264 (9)	0.0244 (8)	0.0218 (8)	0.0001 (6)	-0.0022 (6)	0.0003 (6)
C14	0.0242 (8)	0.0220 (7)	0.0233 (8)	-0.0011 (6)	-0.0016 (6)	0.0008 (6)
C15	0.0262 (9)	0.0244 (8)	0.0241 (8)	-0.0024 (6)	0.0017 (6)	-0.0001 (6)
C16	0.0316 (9)	0.0330 (9)	0.0206 (8)	-0.0009 (7)	-0.0015 (7)	0.0027 (6)
C17	0.0280 (9)	0.0289 (8)	0.0257 (9)	0.0023 (7)	-0.0016 (7)	0.0070 (7)
C18	0.0288 (9)	0.0230 (8)	0.0265 (9)	0.0004 (6)	-0.0028 (7)	-0.0017 (6)
C19	0.0601 (14)	0.0448 (12)	0.0338 (11)	0.0248 (10)	-0.0156 (9)	0.0000 (8)
C20	0.0255 (8)	0.0212 (7)	0.0282 (9)	0.0025 (6)	-0.0003 (6)	-0.0017 (6)
C21	0.0317 (9)	0.0272 (8)	0.0260 (9)	-0.0015 (7)	-0.0026 (7)	-0.0026 (7)
C22	0.0344 (10)	0.0375 (10)	0.0272 (9)	0.0069 (8)	0.0002 (7)	-0.0032 (7)

Geometric parameters (Å, °)

S1—C9	1.7500 (16)	S2—C20	1.7585 (16)
S1—C11	1.8043 (17)	S2—C22	1.7933 (19)
O1—C4	1.353 (2)	O5—C15	1.354 (2)
O1—H1A	0.8400	O5—H5A	0.8400
O2—C7	1.221 (2)	O6—C18	1.220 (2)
O3—C7	1.331 (2)	O7—C18	1.331 (2)
O3—C8	1.452 (2)	O7—C19	1.440 (2)
O4—C10	1.221 (2)	O8—C21	1.220 (2)
N1—C9	1.319 (2)	N3—C20	1.331 (2)
N1—C1	1.422 (2)	N3—C12	1.422 (2)
N1—H1N	0.9099	N3—H3N	0.9098
N2—C9	1.333 (2)	N4—C20	1.312 (2)
N2—C10	1.367 (2)	N4—C21	1.373 (2)
C1—C2	1.378 (2)	C12—C13	1.377 (2)
C1—C6	1.394 (2)	C12—C17	1.401 (2)
C2—C3	1.406 (2)	C13—C14	1.398 (2)
C2—H2	0.9500	C13—H13	0.9500
C3—C4	1.403 (2)	C14—C15	1.401 (2)
C3—C7	1.473 (2)	C14—C18	1.477 (2)
C4—C5	1.395 (2)	C15—C16	1.392 (2)
C5—C6	1.374 (2)	C16—C17	1.377 (2)
C5—H5	0.9500	C16—H16	0.9500
C6—H6	0.9500	C17—H17	0.9500

C8—H8A	0.9800	C19—H19A	0.9800
C8—H8B	0.9800	C19—H19B	0.9800
C8—H8C	0.9800	C19—H19C	0.9800
C10—C11	1.520 (2)	C21—C22	1.526 (2)
C11—H11A	0.9900	C22—H22A	0.9900
C11—H11B	0.9900	C22—H22B	0.9900
C9—S1—C11	89.67 (8)	C20—S2—C22	89.28 (8)
C4—O1—H1A	105.4	C15—O5—H5A	106.6
C7—O3—C8	116.14 (13)	C18—O7—C19	116.99 (14)
C9—N1—C1	127.82 (14)	C20—N3—C12	127.20 (14)
C9—N1—H1N	116.1	C20—N3—H3N	115.5
C1—N1—H1N	116.0	C12—N3—H3N	116.7
C9—N2—C10	112.07 (14)	C20—N4—C21	111.24 (14)
C2—C1—C6	119.94 (15)	C13—C12—C17	119.54 (15)
C2—C1—N1	123.06 (15)	C13—C12—N3	123.18 (15)
C6—C1—N1	116.70 (14)	C17—C12—N3	117.27 (14)
C1—C2—C3	120.12 (15)	C12—C13—C14	120.16 (15)
C1—C2—H2	119.9	C12—C13—H13	119.9
C3—C2—H2	119.9	C14—C13—H13	119.9
C4—C3—C2	119.52 (15)	C13—C14—C15	120.37 (15)
C4—C3—C7	119.59 (14)	C13—C14—C18	119.84 (15)
C2—C3—C7	120.85 (15)	C15—C14—C18	119.78 (15)
O1—C4—C5	117.43 (15)	O5—C15—C16	117.68 (15)
O1—C4—C3	123.12 (15)	O5—C15—C14	123.66 (15)
C5—C4—C3	119.44 (15)	C16—C15—C14	118.66 (15)
C6—C5—C4	120.33 (16)	C17—C16—C15	120.89 (16)
C6—C5—H5	119.8	C17—C16—H16	119.6
C4—C5—H5	119.8	C15—C16—H16	119.6
C5—C6—C1	120.63 (15)	C16—C17—C12	120.36 (16)
C5—C6—H6	119.7	C16—C17—H17	119.8
C1—C6—H6	119.7	C12—C17—H17	119.8
O2—C7—O3	123.05 (15)	O6—C18—O7	123.49 (15)
O2—C7—C3	123.57 (16)	O6—C18—C14	124.10 (15)
O3—C7—C3	113.38 (14)	O7—C18—C14	112.39 (14)
O3—C8—H8A	109.5	O7—C19—H19A	109.5
O3—C8—H8B	109.5	O7—C19—H19B	109.5
H8A—C8—H8B	109.5	H19A—C19—H19B	109.5
O3—C8—H8C	109.5	O7—C19—H19C	109.5
H8A—C8—H8C	109.5	H19A—C19—H19C	109.5
H8B—C8—H8C	109.5	H19B—C19—H19C	109.5
N1—C9—N2	119.84 (14)	N4—C20—N3	124.96 (15)
N1—C9—S1	122.80 (12)	N4—C20—S2	118.27 (13)
N2—C9—S1	117.33 (12)	N3—C20—S2	116.77 (12)
O4—C10—N2	123.68 (16)	O8—C21—N4	124.13 (17)
O4—C10—C11	121.66 (15)	O8—C21—C22	120.97 (16)
N2—C10—C11	114.66 (14)	N4—C21—C22	114.89 (15)
C10—C11—S1	106.27 (11)	C21—C22—S2	106.04 (12)

C10—C11—H11A	110.5	C21—C22—H22A	110.5
S1—C11—H11A	110.5	S2—C22—H22A	110.5
C10—C11—H11B	110.5	C21—C22—H22B	110.5
S1—C11—H11B	110.5	S2—C22—H22B	110.5
H11A—C11—H11B	108.7	H22A—C22—H22B	108.7
C9—N1—C1—C2	47.7 (3)	C20—N3—C12—C13	22.3 (3)
C9—N1—C1—C6	-138.67 (18)	C20—N3—C12—C17	-156.84 (17)
C6—C1—C2—C3	1.2 (2)	C17—C12—C13—C14	0.0 (2)
N1—C1—C2—C3	174.71 (15)	N3—C12—C13—C14	-179.07 (15)
C1—C2—C3—C4	-1.5 (2)	C12—C13—C14—C15	-0.9 (2)
C1—C2—C3—C7	-179.19 (15)	C12—C13—C14—C18	177.82 (15)
C2—C3—C4—O1	-178.04 (15)	C13—C14—C15—O5	-179.17 (15)
C7—C3—C4—O1	-0.3 (2)	C18—C14—C15—O5	2.1 (2)
C2—C3—C4—C5	1.3 (2)	C13—C14—C15—C16	1.1 (2)
C7—C3—C4—C5	178.98 (15)	C18—C14—C15—C16	-177.59 (15)
O1—C4—C5—C6	178.58 (16)	O5—C15—C16—C17	179.77 (15)
C3—C4—C5—C6	-0.8 (3)	C14—C15—C16—C17	-0.5 (3)
C4—C5—C6—C1	0.5 (3)	C15—C16—C17—C12	-0.4 (3)
C2—C1—C6—C5	-0.7 (3)	C13—C12—C17—C16	0.6 (3)
N1—C1—C6—C5	-174.61 (16)	N3—C12—C17—C16	179.74 (15)
C8—O3—C7—O2	1.9 (2)	C19—O7—C18—O6	-3.5 (3)
C8—O3—C7—C3	-177.46 (14)	C19—O7—C18—C14	178.15 (16)
C4—C3—C7—O2	3.0 (2)	C13—C14—C18—O6	177.39 (16)
C2—C3—C7—O2	-179.34 (16)	C15—C14—C18—O6	-3.9 (3)
C4—C3—C7—O3	-177.66 (14)	C13—C14—C18—O7	-4.2 (2)
C2—C3—C7—O3	0.0 (2)	C15—C14—C18—O7	174.46 (15)
C1—N1—C9—N2	-178.03 (16)	C21—N4—C20—N3	176.22 (16)
C1—N1—C9—S1	4.4 (3)	C21—N4—C20—S2	-3.0 (2)
C10—N2—C9—N1	-176.80 (16)	C12—N3—C20—N4	-6.2 (3)
C10—N2—C9—S1	0.9 (2)	C12—N3—C20—S2	172.97 (13)
C11—S1—C9—N1	176.98 (16)	C22—S2—C20—N4	4.75 (15)
C11—S1—C9—N2	-0.60 (15)	C22—S2—C20—N3	-174.50 (14)
C9—N2—C10—O4	179.11 (19)	C20—N4—C21—O8	179.46 (17)
C9—N2—C10—C11	-0.7 (2)	C20—N4—C21—C22	-1.1 (2)
O4—C10—C11—S1	-179.56 (17)	O8—C21—C22—S2	-176.26 (15)
N2—C10—C11—S1	0.3 (2)	N4—C21—C22—S2	4.23 (19)
C9—S1—C11—C10	0.17 (14)	C20—S2—C22—C21	-4.54 (13)

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O1—H1A \cdots O2	0.84	1.87	2.6287 (18)	150
O5—H5A \cdots O6	0.84	1.92	2.6619 (18)	147
N1—H1N \cdots N2 ⁱ	0.91	1.96	2.8624 (19)	175
N3—H3N \cdots O4 ⁱⁱ	0.91	1.97	2.8703 (18)	170

C11—H11A...O8 ⁱⁱⁱ	0.99	2.39	3.371 (2)	171
C5—H5...S2 ^{iv}	0.95	2.71	3.5043 (18)	141

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $x+1, y, z$; (iii) $x-1, -y+3/2, z+1/2$; (iv) $-x+2, -y+1, -z+1$.