Recent advances in organocatalytic asymmetric aza-Michael reactions of amines and amides

Pratibha Sharma, Raakhi Gupta and Raj K. Bansal*

Review

Address:

Department of Chemistry, The IIS (deemed to be University), Jaipur

302 020, India

Email:

Raj K. Bansal* - bansal56@gmail.com

* Corresponding author

Keywords:

asymmetric aza-Michael reaction; covalent bonding catalysis; nitrogen heterocycles; non-covalent bonding catalysis; organocatalysis

Beilstein J. Org. Chem. **2021**, *17*, 2585–2610. https://doi.org/10.3762/bjoc.17.173

Received: 12 July 2021 Accepted: 27 September 2021 Published: 18 October 2021

This article is part of the thematic issue "New advances in asymmetric

Open Access

organocatalysis".

Guest Editor: R. Šebesta

© 2021 Sharma et al.; licensee Beilstein-Institut. License and terms: see end of document.

Abstract

Nitrogen-containing scaffolds are ubiquitous in nature and constitute an important class of building blocks in organic synthesis. The asymmetric aza-Michael reaction (aza-MR) alone or in tandem with other organic reaction(s) is an important synthetic tool to form new C–N bond(s) leading to developing new libraries of diverse types of bioactive nitrogen compounds. The synthesis and application of a variety of organocatalysts for accomplishing highly useful organic syntheses without causing environmental pollution in compliance with 'Green Chemistry' has been a landmark development in the recent past. Application of many of these organocatalysts has been extended to asymmetric aza-MR during the last two decades. The present article overviews the literature published during the last 10 years concerning the asymmetric aza-MR of amines and amides catalysed by organocatalysts. Both types of the organocatalysts, i.e., those acting through non-covalent interactions and those working through covalent bond formation have been applied for the asymmetric aza-MR. Thus, the review includes the examples wherein cinchona alkaloids, squaramides, chiral amines, phase-transfer catalysts and chiral bifunctional thioureas have been used, which activate the substrates through hydrogen bond formation. Most of these reactions are accompanied by high yields and enantiomeric excesses. On the other hand, N-heterocyclic carbenes and chiral pyrrolidine derivatives acting through covalent bond formation such as the iminium ions with the substrates have also been included. Wherever possible, a comparison has been made between the efficacies of various organocatalysts in asymmetric aza-MR.

Introduction

The Michael reaction though discovered about 135 years ago [1,2] continues to attract attention of the chemists owing to its potential of making a vast variety of organic compounds partic-

ularly of pharmacological importance accessible. Over the years, its many versions known as aza-Michael, thio-Michael, oxa-Michael, phospha-Michael, etc. have been developed and

well exploited for their synthetic applications [3-7]. The reaction involving a nitrogen-based nucleophile as the Michael donor is known as the aza-Michael reaction (aza-MR). In view of its ability to introduce a nitrogen-containing functionality at the β-position of an activated alkenyl- or alkynyl-substrate, over the years, it has developed as an important synthetic strategy for the preparation of a large variety of β -amino carbonyl and similar motifs which are present in many bioactive natural products [8,9], antibiotics [10-12] and chiral auxiliaries [13-15]. However, the reaction of many nitrogen-nucleophiles, such as aromatic amines, amides, imides, etc. require the use of an appropriate catalyst to undergo a Michael addition with a suitable acceptor. In view of this, chemists endeavoured to develop different types of catalysts, particularly the chiral catalysts to accomplish asymmetric aza-MRs. The development of metal-free small organic molecules as catalysts has been a landmark advancement in organic synthesis in the recent past [16]. MacMillan and co-workers for the first time in the year 2000 termed these catalysts as 'Organocatalysts' [17]. It was followed by intense activity and phenomenal rise in the number of publications in this field. These organocatalysts have been found compatible with many aspects of 'Green Chemistry' on the one hand, and highly selective in many organic syntheses on the other hand [17]. It has an added advantage that a large number of enantiomerically pure organocatalysts can be accessed from the chiral pool. Both types of organocatalysts, namely those acting through non-covalent bonding as well as those working by making covalent bonding have been employed for accomplishing asymmetric aza-MRs.

There are several review articles available on organocatalytic asymmetric aza-MRs, each highlighting a certain aspect of the reaction. While Sánchez-Roselló et al. [18] classified these reactions on the basis of the nature of the substrates, Nayak et al. [19] and Bhanja et al. [20] focused on the stereoselective synthesis of nitrogen heterocycles via Michael cascade reactions. Recently, Vinogradov et al. [21] reviewed the synthesis of pharmacology-relevant nitrogen heterocycles via stereoselective aza-MRs. On the other hand, Enders et al. [22], Wang et al. [23] as well as Krishna et al. [24] highlighted the scope and catalytic performances of some organocatalysts in asymmetric aza-MRs. However, the last three review articles are almost 10 years old and they do not cover the application of many important organocatalysts, such as thioureas and nitrogen heterocyclic carbenes (NHCs) used for the asymmetric aza-MRs. Furthermore, in the last review article [24], the application of organocatalysts is included as a small part of a general review. In view of this, we considered it prudent to compile this mini review exclusively based on the application of all categories of the organocatalysts and highlighting their efficacies covering the literature of the last ten years.

Review

In the present review, the known stereoselective syntheses of pharmacology-oriented nitrogen containing heterocyclic scaffolds via non-covalent bonding and covalent bonding organocatalytic aza-MRs has been systematized. This classification is especially useful for researchers to understand both the noncovalent and covalent organocatalysis.

It is intended to overview the literature of the last 10 years, i.e., from 2011 through 2020 only. Nevertheless, wherever necessary, earlier references may also be cited to maintain coherence. Furthermore, nitrogen nucleophiles comprise a large variety of compounds; however, in order to comply with the requirements of a mini review, additions of amines and amides only will be included.

1. Non-covalent bonding organocatalytic aza-Michael reactions

Organocatalysts catalyzing aza-MRs through mainly hydrogen bonding include cinchona alkaloids, squaramide derivatives, phase-transfer catalysts and bifunctional thiourea derivatives.

1.1 Reactions catalyzed by chiral cinchona alkaloid derivatives

Cai et al. prepared and used a number of organocatalysts from Cinchona alkaloids for the aza-MR of aniline (1) with chalcone (2) to obtain the adducts 4 in poor to very good yields (24 to >99%) with poor to moderate ee (9 to 55%). A complete reversal of stereoselectivity was observed on introducing a benzoyl group in cinchonine and cinchonidine. It was demonstrated that racemization occurred in suitable solvents under mild conditions due to retro-MR of the initially formed Michael adduct (Scheme 1) [25]. The proposed catalytic cycle involved generation of the active complex through hydrogen bonding between catalyst and aniline followed by interaction with chalcone via π - π stacking of aromatic rings and hydrogen bonding leading to the Michael adduct.

Likewise, Lee et al. reported cinchona-based primary amine catalyzed cascade aza-Michael-aldol reaction of α,β -unsaturated ketones **6** with 2-(1*H*-pyrrol-2-yl)-2-oxoacetates **5** where triphenylacetic acid was used as an additive. This cascade reaction afforded highly functionalized chiral pyrrolizines **8** in good yields (70–91%) with excellent levels of stereocontrol (\approx 92% ee, >20:1 dr in all cases). The ketone group in the cascade product was reduced asymmetrically to a chiral secondary hydroxy group (Table 1) [26].

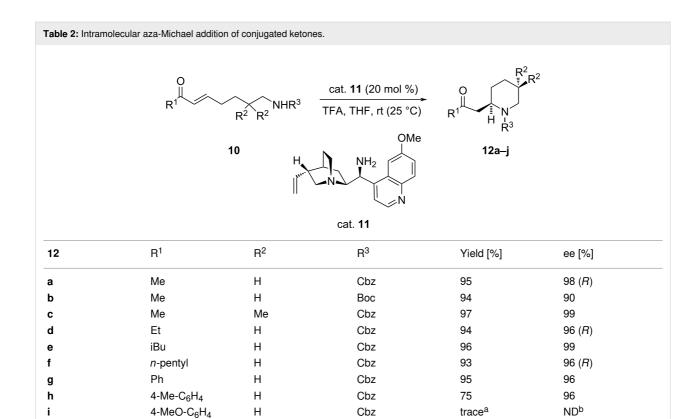
In this case, the role of Ph₃CCO₂H as additive is to furnish the conjugate base Ph₃CO₂⁻ anion which subsequently deproto-

Table 1: Asymmetric cascade aza-Michael–aldol reactions of α,β -unsaturated ketones with pyrroles. cat. 7 (20 mol %) OCH₃ Ph₃CCO₂H (40 mol %) toluene, 0 °C or rt 8a-f . X = Y = Br BH₃·SMe₂,THF, rt BH₃·SMe₂ (S)-Me-CBS THF, -40 °C cat. 7 9a 9a' yield 76%; dr 9:1 yield 79%; dr 2.9:1 8 Χ Υ ee [%] Yield [%] dr Br Br 86 91 >20:1 а b CI Br 91 90 >20:1 Br 75 92 >20:1 С 1 d Br CI 86 90 >20:1 CI 73 90 >20:1 е 70 f 92 >20:1

nates pyrrole to provide the stronger nucleophilic pyrrolide anion [27].

Similarly, Liu et al. accomplished an asymmetric intramolecular aza-Michael addition of various enone carbamates 10 using a chiral cinchona-based primary-tertiary diamine as catalyst to

obtain 2-substituted piperidines **12** in good yields (75–95%) with up to 99% ee. Several sulfonic acids and carboxylic acids were tested as co-catalysts and trifluoroacetic acid (TFA) was found to give the best results [28]. Here the role of the co-catalyst is to assist in the formation of the iminium intermediate (Table 2) [29]. It appears that in this case, both activation mech-



Cbz

anisms, namely through hydrogen bonding and iminium ion for-

mation are operating.

4-O₂N-C₆H₄

^aThe starting material was mainly recovered. ^bND = not determined.

Н

Using the same chiral cinchona-based primary-tertiary diamine as catalyst (cat. 11). Zhai et al. developed a highly efficient

intramolecular enantioselective aza-Michael addition of carbamates, sulfonamides and acetamides 13 bearing an α,β -unsaturated ketone to synthesize a series of 2-substituted five- and six-membered heterocycles in good yields (up to 99%) and excellent enantioselectivity (92–97.5% ee) (Table 3). As in an earlier

85

80

ble 3: Intra	molecular enantioselectiv	e aza-Michael additio	on.		
	0	W NUIDO —	cat. 11 (15 mol %)	x	0
		X NHPG —	DPP (15 mol %) PhMe, rt (25 °C), 40) h PG	
	13			14a–d	
4	Х	n	PG	Yield [%]	ee [%] ^a
	0	2	Вос	97	97
	0	1	Вос	96	94
)	S	2	Cbz	55	95
	S	1	Cbz	91	92

case [29], several acids were tested as co-catalysts and trifluoro-acetic acid and diphenyl hydrogenphosphate (DPP) were found to give the best results [30].

Cheng et al. reported an intramolecular 6-exo-trig aza-MR of hydroxylamine-derived enone **15** for the synthesis of chiral 3-substituted 1,2-oxazinanes **16**. The catalyst **11** was used in this case also and pentafluoropropionic acid (PFP) was used as a co-catalyst. In the presence of 1,4-dioxane solvent, products chiral 3-substituted 1,2-oxazinanes (**16**) were obtained in 99% yield with good ee of 96% (Scheme 2) [31].

Following a similar strategy, Ma et al. accomplished a highly enantioselective aza-Michael addition of 4-nitrophthalimide (17) with α,β -unsaturated ketones 18 using 9-*epi*-9-amino-9-deoxyquinine 19 as the catalyst, the corresponding Michael adducts being obtained in moderate to good yields (49–98%) with excellent ee (95–99%) (Table 4) [32].

Jakkampudi et al. [33] adopted a different approach for the use of cinchona-based organocatalysts. Instead of using the cinchona derivative alone, they employed a mixture of cinchona derivative and amino acid such as D-proline, termed as the

modularly designed organocatalyst (MDO) for the synthesis of bridged tetrahydroisoquinoline derivatives. It was perceived that the MDO self-assembled in situ from amino acids and cinchona alkaloid derivatives. For example, on reacting (*E*)-2-[2-(3-aryl-3-oxoprop-1-en-1-yl)phenyl]acetaldehydes **21** with ethyl or benzyl (*E*)-2-[(4-methoxyphenyl)imino]acetates **22** in the presence of the MDO **23/24** (quinidinethiourea + p-proline), instead of the expected domino Mannich/Michael product, the bridged tetrahydroisoquinoline product **25a** was obtained in high yield (90%) and excellent dr (94:6) and ee value (99%) (Table 5). The controlled reactions using **23** and **24** as the catalyst gave the product in very poor yield. It was concluded that the catalytic activity of the MDO was the result of the cooperative action of both constituents. Several examples of such

MDOs are included in the paper. The reported yield varies from 56-90% with excellent ee $\approx 99\%$ in all cases.

1.2. Reactions catalyzed by chiral squaramide derivatives

Squaramides are related to cinchona alkaloids but are much more effective organocatalysts than the latter due to the ability of dual hydrogen bonding besides a tertiary nitrogen atom of quinuclidine nucleus which may serve both as an H-bond acceptor and a base in asymmetric Michael addition reactions [34,35].

In 2015, Zhao et al. synthesized spiro[pyrrolidine-3,3'-oxindoles] **29** in single step by asymmetric cascade aza-Michael/

Table 5: Diastereoselective synthesis of bridged 1,2,3,4-tetrahydroisoquinoline derivatives using modularly designed organocatalyst. cat. 23/24 **PMP** toluene, rt. 3 h CO₂R³ 21 22 25a-o ОМе cat. 24 cat. 23 R^1 R^2 R^3 25 Yield [%] ee [%] dr C_6H_5 Н Εt 90 >99 94:6 а b 4-F-C₆H₄ Н Εt 81 >99 97:3 4-CI-C₆H₄ Н Εt 90 99 92:8 С 4-Br-C₆H₄ Н Et 80 >99 90:10 d 4-NC-C₆H₄ Н Εt 77 99 87:13 e Εt 76 99 f 4-Me-C₆H₄ Н 97:3 4-MeO-C₆H₄ Н Εt 79 97 96:4 q 3-CI-C₆H₄ Н Εt 72 >99 88:12 h ia Εt 2-F-C₆H₄ Н ja 2-CI-C₆H₄ Н Et F Ft 73 >99 90:10 C_6H_5 k I C_6H_5 MeO Ft 74 99 91:9 Me Н Et 56 92 86:14 m Н Bn 75 99 89:11 n C_6H_5 4-Br-C₆H₄ Н Bn 77 >99 93:7 0 ^aFormation of a complex mixture was observed.

Michael addition reaction between 4-tosylaminobut-2-enoates **27** and 3-ylideneoxindoles **26** catalyzed by a chiral bifunctional tertiary amine, squaramide (cat. **28**) which afforded the corresponding adducts in good yields ranging from 72–99% with excellent diastereoselectivity (up to >99:1 dr) and enantioselectivity (>99% ee) (Table 6) [36].

In another report, Yang et al. accomplished a highly asymmetric cascade aza-Michael/Michael addition reaction for the synthesis of tetrahydroquinolines and tetrahydrochromanoquinolines catalyzed by a squaramide catalyst. The corresponding adducts were obtained in excellent yields with excel-

lent diastereoselectivities and enantioselectivities (up to >99:1 dr, 99% ee) [37].

Following a similar strategy, Zhou et al. obtained a series of optically active tetrahydrobenzofuro[3,2-b]quinolines and tetrahydrobenzo[4,5]thieno[3,2-b]quinolines **33** in high yields ranging from 35-99% and excellent diastereo- (>20:1 dr), and enantioselectivities (up to $\approx 99\%$ ee) (Scheme 3) [38].

Roy et al. accomplished an enantioselective intramolecular aza-Michael addition for the synthesis of dihydroisoquinoline and tetrahydropyridines from Michael reaction of *ortho*-homo-

Table 6: Synthesis of spiro[pyrrolidine-3,3'-oxindoles] via asymmetric cascade aza-Michael reaction catalyzed by squaramide. $R^{1}OC$ **TsHN** Вос Вос 26 27 29a-r OMe cat. 28 R^2 R^3 29 R^1 Yield [%] ee [%] dr Ph Н Me 72 98 96:4 а Ph b Н Ot-Bu 99 >99 93:7 Ph Н OBn 99 >99 88:12 С Н OEt 99 >99 65:35 d 4-FC₆H₄ Н >99 Ot-Bu 99 88:12 е f 4-CIC₆H₄ Н Ot-Bu 99 >99 89:11 Н 2-BrC₆H₄ Ot-Bu 91 >99 85:15 g 4-BrC₆H₄ Н Ot-Bu 99 >99 92:8 h 4-MeC₆H₄ Н Ot-Bu 95 >99 96:4 i j 3-MeOC₆H₄ Н Ot-Bu 88 >99 92:8 4-MeOC₆H₄ Н Ot-Bu 94 >99 89:11 k Ī 2-naphthyl Н Ot-Bu 92 >99 89:11 2-thienyl Н Ot-Bu >99 96 92:8 m F C_6H_5 Ot-Bu 99 >99 89:11 n CI 88:12 C_6H_5 Ot-Bu 99 >99 0 C_6H_5 Br Ot-Bu 99 >99 93:7 р q C_6H_5 Ме Ot-Bu 99 >99 97:3 OMe Ot-Bu 99 >99 99:1 r C_6H_5

$$R^{1}$$
 NO_{2} NO

Scheme 3: Asymmetric aza-Michael/Michael addition cascade reaction of 2-nitrobenzofurans and 2-nitrobenzothiophenes with 2-aminochalcones catalyzed by squaramide derivative.

formyl chalcone with various amines by using squaramide catalyst. The reaction occurred with good yields and excellent enantioselectivity [39].

Similarly, Li et al. reported an asymmetric cascade aza-Michael addition of 2-tosylaminoenones with unsaturated pyrazolones using squaramide as catalyst. The reaction proceeded smoothly under mild conditions to afford the corresponding spiro[pyrazolone-tetrahydroquinolines] in high yields (up to 99%) with excellent diastereoselectivities (up to >25:1 dr) and high enantioselectivities (up to 65–91%) [40].

Rajasekar et al. developed an efficient one-pot tandem rhodium(II)/chiral squaramide relay catalysis for the enantioselective construction of dihydro- β -carbolines 37 from the Michael reaction of suitably substituted indole derivatives 34 with *N*-sulfonyl-1,2,3-triazoles 35 in good yields (up to \approx 72%) and excellent enantioselectivity (up to 99% ee) (Table 7) [41].

In an interesting study, Wu et al. screened a number of cinchona derivatives and squaramides for their relative catalytic efficacies for the enantioselective aza-Michael additions between

Table 7: Asym	nmetric aza-Michael synthesis of dihydr	Ar.	36 (10 mol %)	Ar	
	N + CH ₃ Ph	N — Rh ₂ (0	DAc) ₄ (2 mol %) 100 °C, 5 h	N SO₂R O Ph	
	34	35	N H HN Ph	7a,b	
		С	at. 36		
37	Ar	R	Yield [%]	ee [%]	
a b	Ph Ph	Ph Me	71 72	99 80	

halogenated 2-hydroxypyridines (pyridin-2(1H)-ones) **38** and α , β -unsaturated 1,4-diketones or 1,4-ketoesters **39** in different solvents. The best results (yield 96%, ee >91%) were obtained on using squaramide catalyst in chloroform. However, for

others, the yields ranged from 50–98% with good to excellent enantioselectivity (47–98% ee). The observed results were rationalized with density functional theory calculations (Table 8) [42].

Table 8: Asymmetric aza-Michael synthesis of N-substituted 2-pyridones. cat. 40 (10 mol %) CHCl₃/H₂O 1:1 (v/v) rt, 48 h 38 39 Ar-NH $Ar = 3,5-(CF_3)_2C_6H_3$ cat. **40** 41 Χ R^1 R^2 Yield [%] ee [%] а 5-Br Ph Ph 98 98 b 5-I Ph Ph 78 91 5-F Ph Ph 60 74 С d Н Ph Ph 50 44 3-CI Ph Ph 88 93 е Ph 3-Br Ph 93 92 f 3-I Ph Ph 82 73 g 4-Br Ph Ph 60 47 h i 6-CI Ph Ph 0 5-CI 93 93 j p-F-C₆H₄ p-F-C₆H₄ 5-Br 95 >99 p-F-C₆H₄ p-F-C₆H₄ k ı 5-I p-F-C₆H₄ p-F-C₆H₄ 87 97 3-CI p-F-C₆H₄ p-F-C₆H₄ 82 99 m 88 97 3-Br p-F-C₆H₄ p-F-C₆H₄ n p-F-C₆H₄ 3-I p-F-C₆H₄ 90 99 0 5-CI p-NC-C₆H₄ p-NC-C₆H₄ 85 94 р 5-Br 73 97 p-NC-C₆H₄ p-NC-C₆H₄ q 5-CI 70 35 p-Me-C₆H₄ p-Me-C₆H₄ 76 82 5-Br s p-Me-C₆H₄ p-Me-C₆H₄ t 5-I p-Me-C₆H₄ p-Me-C₆H₄ 75 67 5-CI p-MeO-C₆H₄ p-MeO-C₆H₄ 0 u 78 5-CI OEt Ph 90 ν 5-Br OEt Ph 82 80 w 5-F OEt Ph 83 63 X 3-CI OEt Ph 70 90 У 3-Br OEt Ph 78 90 z 4-Br Ph OEt 73 60 aa ab 3-CI OEt p-F-C₆H₄ 90 80

1.3 Reactions catalyzed by chiral amines

He and co-workers developed heterogeneous synergistic catalysis using chiral amines SBA-15 (cat. 44), which promote aza-Michael-Henry cascade reactions between 2-aminobenz-aldehydes 42 and β -nitrostyrenes 43 to obtain chiral 3-nitro-1,2-dihydroquinolines 45 in good yields with up to 98% ee (Table 9) [43].

1.4 Reactions catalyzed by chiral phase-transfer catalysts

Chiral phase-transfer catalysts (PTC) have been recognized as versatile catalysts for the asymmetric aza-Michael addition reactions. Mahe et al. reported an effective, eco-friendly and cost-effective enantioselective synthesis of 3,5-diarylpyrazolines 49 by using phase-transfer methodology. They carried out a set of reactions between chalcones 46 and *N-tert*-butoxycarbonylhydrazine (47) in the presence of cesium carbonate and an *N*-benzylquininium salt as catalyst (cat. 48) (solid–liquid phase-transfer conditions) to give the corresponding adducts in 40–90% yields with excellent ee of up to 99% (Table 10) [44].

A different type of asymmetric aza-Michael addition was developed by Wang et al. They carried out asymmetric conjugate amination of *tert*-butylbenzyloxycarbamate (**50**) to β -nitro-

styrene **51** under neutral phase-transfer conditions in the presence of chiral bifunctional tetraalkylammonium bromide (cat. **52**) in water-rich biphasic solvent. The reaction proceeded with high ee values of up to 95% and very good yields (≈99%) in all cases (Table 11) [45].

Guo et al. synthesized a variety of benzoindolizidines (56) from α,β -unsaturated aminoketones 54 through intramolecular domino aza-Michael addition/alkylation reactions. The reactions were carried out in the presence of cinchona alkaloid-derived quaternary ammonium salts (cat. 55) as the phase-transfer catalyst. The products were obtained in high yields (53–93%) with high enantioselectivities (40–76% ee) (Table 12) [46].

Lebrun et al. developed a new method to synthesize optically active isoindolinones via asymmetric intramolecular aza-MR by using phase-transfer catalysts. Alkenylated benzamide was used as the substrate in this reaction. The resulting compounds were found to be useful intermediates for the synthesis and development of benzodiazepine-receptor agonists [47].

In 2018, Sallio et al. worked on the same reaction by using different PTCs in order to improve yield and diastereomeric excess. They incorporated PTC and chiral auxiliary and reacted

	CHO NH ₂ +	NO ₂ cat. 44 iPrOH, 40 °C 43	, 72 h R ¹ H	NO ₂
		Si OH OH OH SBA-15-AEP cat. 44	Yield [%] ^a	n, db
45	R ¹ in 42	R^2	rielu [76]	ee % ⁰
				ee [%] ^b
1	Н	2,3-(MeO) ₂ -C ₆ H ₃	67 (65)	98 (98)
a o	H H	2,3-(MeO) ₂ -C ₆ H ₃ 4-Me-C ₆ H ₄	67 (65) 60 (59)	98 (98) 90 (93)
a b	Н Н Н	2,3-(MeO) ₂ -C ₆ H ₃ 4-Me-C ₆ H ₄ 2,4-(Cl) ₂ -C ₆ H ₃	67 (65) 60 (59) 45 (40)	98 (98) 90 (93) 97 (96)
a o o	Н Н Н	2,3-(MeO) ₂ -C ₆ H ₃ 4-Me-C ₆ H ₄ 2,4-(Cl) ₂ -C ₆ H ₃ 3,4-(Cl) ₂ -C ₆ H ₃	67 (65) 60 (59) 45 (40) 42 (38)	98 (98) 90 (93) 97 (96) 95 (97)
a b c d	H H H H 3-MeO	2,3-(MeO) ₂ -C ₆ H ₃ 4-Me-C ₆ H ₄ 2,4-(Cl) ₂ -C ₆ H ₃ 3,4-(Cl) ₂ -C ₆ H ₃ Ph	67 (65) 60 (59) 45 (40) 42 (38) 52 (50)	98 (98) 90 (93) 97 (96) 95 (97) 95 (99)
45 a b c d d e f	Н Н Н	2,3-(MeO) ₂ -C ₆ H ₃ 4-Me-C ₆ H ₄ 2,4-(Cl) ₂ -C ₆ H ₃ 3,4-(Cl) ₂ -C ₆ H ₃	67 (65) 60 (59) 45 (40) 42 (38)	98 (98) 90 (93) 97 (96) 95 (97)

89 (+)

88 (-)

89 (-)

91 (-)

78 (-)

60

70

62

61

46

^aYield determined by NMR analysis of the crude reaction mixture using an internal standard.

4-MeOC₆H₄

4-CIC₆H₄

2-MeC₆H₄

2-thienyl

3-MeOC₆H₄

j

k

m

n

Ph

Ph

Ph

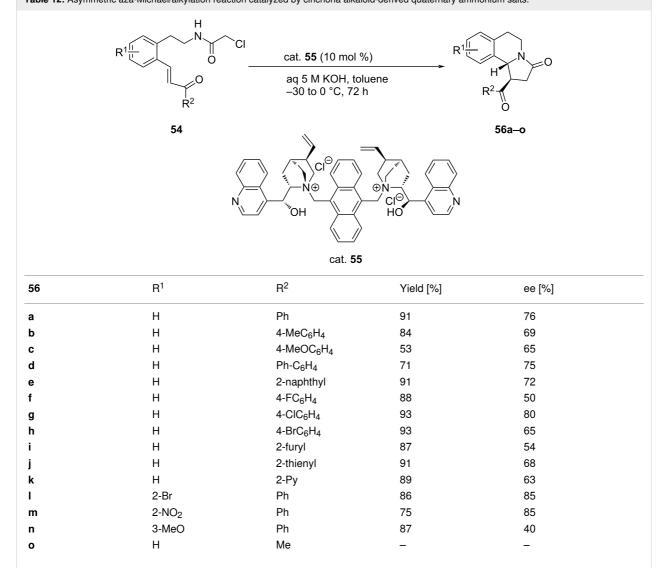
Ph

Ph

		Boc ,OBn H 50	+ R NO ₂ (S	C) cat. 52 (0.05–1 mol 9) H ₂ O/toluene 10:1 Ar Ar OH Br OH Ar Ar	R	NO_2	
			Ar =	3,5-[3,5-(CF ₃) ₂ C ₆ H ₃] ₂ C cat. 52	C ₆ H ₃		
				N/ 11 (0/)		[0/1	
53	Conditions ^a		R	Yield (%)		ee [%]	
53	Conditions ^a	В	К	A Yield (%)	В	ee [%]	В

b	Α	В	4-Me-C ₆ H ₄	93	90	90	91
С	Α	В	4-BrC ₆ H ₄	91	89	91	94
d	Α	В	3-FC ₆ H ₄	70	62	90	93
е	Α	В	4-TBSOC ₆ H ₄	94	92	92	95
f	Α	В	2-naphthyl	85	70	90	91
g	Α	В	2-thienyl	93	81	90	94
h	Α	В	2-furyl	90	82	90	93
i	Α	В	(CH ₃) ₂ CHCH ₂	95	99	77	82
j	Α	В	<i>t</i> -Bu	97	99	79	83

Table 12: Asymmetric aza-Michael/alkylation reaction catalyzed by cinchona alkaloid-derived quaternary ammonium salts.



a variety of chiral phthalimidines **57** to obtain isoindolinones **59** in good yields (\approx 85%) with excellent de ranging 48–96% (Table 13) [48].

1.5 Catalysis by chiral bifunctional thioureas

Thioureas constitute one of the most important class of organocatalysts [49].

	R ¹ R ²	O Me N Ar N Ar O 57	cat. 58 (10 mo	equiv) h 59 N Br [©]	O Me N Ar Y	
			cat. 58			
9	R ¹	R ²	Ar	Y	Yield [%]	de [%]
	Н	Н	Ph	-N	75	>96
	Н	Н	Ph	-NO	78	>96
	Н	Н	Ph	−N iPr iPr	79	56
	Н	Н	Ph	_N H _N Bn	80	44
	Н	Н	PMP ^a	-N	82	82
	Н	Н	PMP	-NO	80	98
	Н	Н	PMP	−N iPr iPr	85	98
	Н	Н	PMP	-NBn	83	48
	Н	Н	PMP	−N Cy	79	67
	MeO	MeO	PMP	N	78	98

Wang et al. reported a cascade aza-Michael/Michael reaction of anilines **60** to nitroolefin enoates **61** using chiral bifunctional thiourea as catalyst (cat. **62**). It provided a mild and efficient approach to the synthesis of three stereocentered polysubstituted chiral 4-aminobenzopyrans **63** in high yields (71–96%) with excellent stereoselectivities of up to >99% ee (Table 14) [50].

In an interesting report, five organocatalysts belonging to three categories, namely cinchona alkaloid bases, bifunctional squaramides and thioureas were screened for the enantioselective *N*-alkylation of isoxazolin-5-ones via a 1,6-aza-Michael addition of isoxazolin-5-ones **64** to *p*-quinone methides (*p*-QMs) **65** to give isoxazolin-5-ones **67** bearing a chiral diarylmethyl

moiety attached to the N atom. The best result in terms of enantioselectivity (85% ee) was obtained with quinine-derived thiourea in dichloroethane as the solvent. The scope of the reaction was also investigated vis-a-vis the effect of the substitution on the isoxazolinone ring and p-quinone methide (p-QM) partner. (Table 15) [51].

Takemoto and co-workers investigated three catalytic systems, namely arylboronic acid alone, its dual combination with chiral thiourea and integrated catalyst having boronic acid functionality in the chiral thiourea molecule. The dual combination of arylboronic acid with chiral thiourea was found as effective as arylboronic acid alone for the intermolecular asymmetric Michael addition of alk-2-enoic acids **68** with *O*-benzyl-

hydroxylamine (69) giving racemic mixture of the product in poor yield. However, the integrated catalyst having boronic acid functionality in the chiral thiourea molecule gave the desired β -benzyloxyamino acid as the single product in a satisfactory yield. Thus, a series of these catalysts was screened. The best results in term of the yield (83%) and ee (90%) were obtained while using the catalyst having a p-nitrophenyl group on the other side of thiourea moiety in CCl₄ in the presence of 4 Å molecular sieves (Table 16). The yields ranged 57–89% with ee 70–97% [52].

A similar chiral multifunctional thiourea/boronic acid was used as an organocatalyst by Michigami et al. for the enantioselective synthesis of *N*-hydroxyaspartic acid derivatives **76** with

 Table 15: Enantioselective 1,6-aza-Michael addition of isoxazolin-5-ones to p-quinone methides.

$$R^{1}$$
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{4}
 R^{4

MeO S CF₃

cat. 66

67	R ¹	R ²	Ar	Yield [%]	ee [%] ^a
а	Me	Н	Ph	65	87
b	Et	Н	Ph	51	81
С	Pr	Н	Ph	50	81
d	Ph	Н	Ph	77	54
е	Pr	Н	Ph	78	89
f	Me	Me	Ph	66	62
g	Me	Н	p -MeC $_6$ H $_4$	62	88
h	Me	Н	p -MeOC $_6$ H $_4$	74	84
i	Me	Н	p -CIC $_6$ H $_4$	43	48
j	Me	Н	p-O ₂ NC ₆ H ₄	47	89
k	Me	Н	o-MeOC ₆ H₄	81	94
I	Me	Н	o-CIC ₆ H ₄	94	96
m	Me	Н	o-BrC ₆ H ₄	43	90
n	Me	Н	m-MeOC ₆ H ₄	20	25
0	Me	Н	m -CIC $_6$ H $_4$	36	81
p	Me	Н	m-O ₂ NC ₆ H ₄	56	77
q	Pr	Н	p -MeOC $_6$ H $_4$	75	79
r	Pr	Н	p -CIC $_6$ H $_4$	78	88
s	Pr	Н	p-O ₂ NC ₆ H ₄	80	86
t	Pr	Н	o-CIC ₆ H ₄	76	92
u	Pr	Н	m-MeOC ₆ H ₄	82	82
v	Pr	Н	m -CIC $_6$ H $_4$	80	88
w	Pr	Н	Ph	71	86

^aDetermined by HPLC using a chiral stationary phase.

perfect regioselectivity and high enantioselectivity (Table 17) [53].

Likewise, Miyaji et al. reported an efficient method for the synthesis of 2-substituted indolines **79** via intramolecular aza-Michael addition of α,β -unsaturated carboxylic acid derivatives **77** in the presence of bifunctional thiourea organocatalysts (cat.

78) (Table 18). The product was obtained in moderate to good yield of 53–99% with an ee of 74–93% [54].

Liu et al. accomplished a catalytic cascade aza-Michael-Henry-dehydration protocol for the preparation of chiral 3-nitro-1,2-dihydroquinolines **83** from the reaction of *N*-protected aminobenzaldehydes **80** with substituted nitroolefins **81** by using

0

Ph

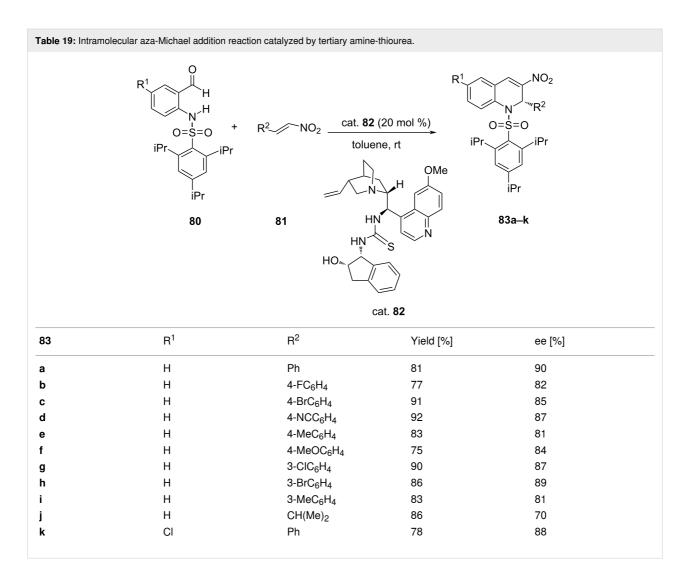
s

Table 17: Asymn urea/boronic acid		or the synthesis of N-	hydroxyaspartic acid derivatives ca	atalyzed by chiral multifunctional thio-
b	BnO	50	91	-
С	EtO	49	94	_
d	t-BuO ₂ C N—	40	-	67:33
е	t-BuO ₂ C N—	60	-	75:25
f	t-BuO₂C N—	66	63	-
g	t -BuO ₂ C $\stackrel{H}{\sim}$ t -BuO t -BuO	81	85	73:37

	R ² NHCbz 77	mesitylene, 25	R ² 5 °C, 24 h 79a	R ¹ O Cbz j
		N cat.		
79	R ¹	cat.		ee [%]
79 a	R ¹	cat.	78	ee [%] 87
		cat.	78 Yield [%]	
a	Ph	cat.	78 Yield [%]	87
a b	Ph 4-CH ₃ OC ₆ H ₄	cat. R ² H H	78 Yield [%] 99 73	87 84
a b c	Ph 4-CH ₃ OC ₆ H ₄ 4-CF ₃ C ₆ H ₄	cat. R ² H H H	78 Yield [%] 99 73 79	87 84 88
a b c d	Ph 4-CH ₃ OC ₆ H ₄ 4-CF ₃ C ₆ H ₄ 2-naphthyl	cat. R ² H H H H	78 Yield [%] 99 73 79 83	87 84 88 88
a b c d	Ph $4\text{-CH}_3\text{OC}_6\text{H}_4$ $4\text{-CF}_3\text{C}_6\text{H}_4$ 2-naphthyl $4\text{-BrC}_6\text{H}_4$	cat. R ² H H H H	78 Yield [%] 99 73 79 83 75	87 84 88 88 91
a b c d e f	Ph $4\text{-}CH_3OC_6H_4$ $4\text{-}CF_3C_6H_4$ $2\text{-}naphthyl$ $4\text{-}BrC_6H_4$ Ph	Cat. R ² H H H H CH ₃ O	78 Yield [%] 99 73 79 83 75 82	87 84 88 88 91 83
a b c	Ph $4\text{-CH}_3\text{OC}_6\text{H}_4$ $4\text{-CF}_3\text{C}_6\text{H}_4$ 2-naphthyl $4\text{-BrC}_6\text{H}_4$ Ph	Cat. R ² H H H H CH ₃ O F	78 Yield [%] 99 73 79 83 75 82 69	87 84 88 88 91 83

tertiary amine-thiourea catalyst (cat. **82**). This cascade reaction afforded aza-Michael adducts in 77–92% yields with high ee (up to 90%) (Table 19) [55].

Du et al. developed an enantioselective catalytic tandem aminolysis/aza-Michael addition for the asymmetric total synthesis of two natural *Apocynaceae* alkaloids, (+)-deethylibophyllidine



(88) and (+)-limaspermidine (89) from the reaction of *para*-dienone imide 84 with benzylamine (85) in the presence of bifunctional thiourea organocatalyst (Scheme 4) [56].

1.6 Reactions catalyzed by chiral binol-derived phosphoric acids

Binol-derived chiral phosphoric acids have been shown to catalyze the reactions via single or double hydrogen bonding [57,58].

Saito et al. accomplished the chiral phosphoric acid-catalyzed intramolecular aza-Michael addition reaction of *N*-unprotected 2-aminophenyl vinyl ketones **90** to obtain chiral 2-substituted 2,3-dihydro-4-quinolones **92** in very good yields (67–95%) with high ee (82–97%) (Table 20) [59].

Following a similar approach, Yang et al. reported asymmetric aza-Michael additions of anilines 94 to β -nitrostyrenes 93 using a chiral binol-derived phosphoric acid diester catalyst (cat. 95).

They succeeded in preparing β -nitroamines **96** in good yields (65–85%), but with only a moderate level of ee (19–70%) (Table 21) [60].

Feng et al. accomplished an asymmetric intramolecular aza-Michael addition of activated α,β -unsaturated ketones **97** by using chiral *N*-triflylphosphoramide as catalyst (cat. **98**). The products, namely 2-aryl-2,3-dihydro-4-quinolones **99** were obtained in good yields of up to 95% and good ee (58–72%) (Table 22) [61].

2. Covalent-bonding organocatalysis of aza-Michael reactions

This category of organocatalysts includes N-heterocyclic carbenes and pyrrolidine derivatives.

2.1 Catalysis by N-heterocyclic carbenes (NHC)

In recent years, NHCs have been used as organocatalysts for a wide variety of reactions [62].

Table 20: Intramolecul	ar aza-Michael addition reaction catalyze	d by chiral phosphoric acid. (contir	nued)	
h	3-MeOC ₆ H ₄	95	92	
i	4-FC ₆ H ₄	quant	87	
j	4-CIC ₆ H ₄	67	82	
k	4-MeC ₆ H ₄	quant	97	
I	2-naphthyl	82	81	
m	<i>t</i> -Bu	64	88	

Wang et al. investigated the use of several 1,2,4-triazolo-annelated chiral NHCs as organocatalysts to catalyze enantiose-lective aza-MR between primary amines (100) and β -trifluoro-methyl- β -arylnitroolefins 101 and the best results (yield 99%, ee 91%) were obtained in the reaction of benzylamine (R¹ = Ph) on using the NHC precursor as shown below in the presence of hexafluoroisopropanol (HFIP) as additive along with molecular sieves (4 Å) (Table 23) [63]. The role of HFIP is to act as proton shuttle, i.e., to assist in 1,3-prototropic shift.

2.2 Catalysis by chiral pyrrolidine derivatives

Chiral pyrrolidine derivatives, such as (S)-proline are widely used as organocatalysts [54,64].

Lee et al. synthesized bromopyrrole alkaloids **107** via aza-Michael addition of 4,5-dibromo-1*H*-pyrrole-2-carbonitrile **104** to Bz-protected (*E*)-4-hydroxybut-2-enal **105** in the presence of (S)- α , α -bis[3,5-bis(trifluoromethyl)phenyl]-2-pyrrolidinemethanol trimethylsilyl ether as the organocatalyst (cat.

76 (+)

76 (-)

1-naphthyl

2-naphthyl

h

i

81

98

_			
h	C ₇ H ₁₅ CH ₂ -	87	95
i	C ₆ H ₅ (CH ₂) ₃ -	82	93
j	(CH ₃) ₂ CH(CH ₂) ₂ -	79	89
k	cyclopropyl-	79	97
I	cyclobutyl-	78	94
m	2-pyridylethyl-	87	93
n	BocHN(CH ₂) ₂	79	91
0	MeO(CH ₂) ₃ -	87	87
p	$(Me)_2N(CH_2)_2$ -	99	91
q	2-thienyl(CH ₂) ₂ -	98	93

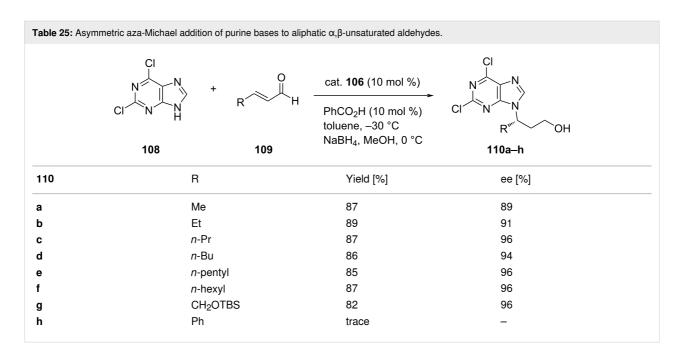
106) and using PhCO₂H as the acid additive. Desired products were obtained in good yields ≈78% with excellent enantioselectivities of up to 93% (Table 24) [65]. The role of the additive is to assist in the formation of the iminium intermediate from the reaction of pyrrolidine with the aldehyde group.

Following a similar approach, Guo et al. accomplished the first organocatalytic asymmetric aza-Michael addition of purine bases 108 to aliphatic α,β -unsaturated aldehydes 109 and synthesized biologically active acyclonucleoside 110 via an iminium-ion activation mechanism. The initially formed prod-

uct was reduced in situ to afford the final product in 82–89% yield and 89–96% ee (Table 25) [66].

In a similar method, Joie et al. accomplished an asymmetric organocatalytic quadruple cascade reaction of various α -ketoamides 111 with aromatic α,β -unsaturated aldehydes 112 to obtain tetraaryl-substituted 2-azabicyclo[3.3.0]octadienones 114 in good yields (34–71%) with excellent diastereo- and enantioselectivities (84–97%). The reaction occurred via an aza-Michael/aldol condensation/vinylogous Michael addition/aldol condensation sequence (Table 26) [67].

Table 24: Asymmetric aza-Michael additions of pyrroles to protected (*E*)-4-hydroxybut-2-enals. i) cat. 106 (20 mol %) PhCO₂H (40 mol %) toluene, -20 °C, 18 h ii) NaBH₄ (110 mol %) EtOH (0.1 mol %), -20 °C 104 105 5 h $Ar = 3,5-(CF_3)_2C_6H_3$ cat. 106 R^2 107 R^1 Yield [%] ee [%] CN Вz 70 76 а n.d.a CO₂CH₃ Βz b CN **TBS** 78 80 C 87 d CN **TBDPS** 77 е CN **TBDPS** 76 91 CN **TBDPS** 76 93 f aNot determined.



	R! _N H	$R^2 + R^3$	Cat. 113 (20 mol %) NaOAc (20 mol %) CH ₂ Cl ₂ , rt, 5 d R ³	N R ²	
	1	11 112	Ph Ph OTMS	114a–i	
114	R ¹	R ²	R ³	Yield [%]	ee [%] ^a
а	Ph	Ph	Ph	63	97
b	Ph	Ph	4-MeOC ₆ H ₄	51	89 (91)
С	Ph	Ph	4-CIC ₆ H ₄	34	85 (95)
d	Ph	Ph	2,3-(OCH ₂ O)C ₆ H ₃	56	84 (87)
е	4-MeOC ₆ H ₄	Ph	Ph	66	92 (91)
_	3-CIC ₆ H ₄	Ph	Ph	69	91 (95)
f		Ph	Ph	58	95
f g	4-O ₂ NC ₆ H ₄				
-	4-О ₂ NО ₆ п ₄ Ph	4-MeC ₆ H ₄	Ph	70	88

Recently, the synthesis of axially chiral 4-naphthylquinoline-3-carbaldehydes **117** has been reported via Michael/Aldol cascade reaction of alkynals **116** with *N*-(2-(1-naphthoyl)phenyl)benzenesulfonamides **115** using the same pyrrolidine

catalyst 113. The products were obtained in excellent yields and enantioselectivities (Table 27) [68]. In this context, the presence of a strong electron-withdrawing sulfonyl group was found to be essential. On comparing efficacies of different sulfonyl

R ¹ NHS	Ph HCI	$\begin{array}{c} & \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ & \\ \\ \end{array}$			
115	116				117a–s
117	R ¹	R ²	R	Yield [%]	ee [%]
a	Н	Н	4-CIC ₆ H ₄	97	93
b	Н	Н	Ph	94	95
С	Н	Н	4-FC ₆ H ₄	86	94
d	Н	Н	4-BrC ₆ H ₄	95	94
e	Н	Н	4-MeC ₆ H ₄	81	94
f	Н	Н	4-MeOC ₆ H ₄	86	95
g	Н	Н	3-CIC ₆ H ₄	92	91
h	Н	Н	3-MeC ₆ H ₄	88	93
i	Н	Н	3-MeOC ₆ H ₄	83	93
İ	Н	Н	<i>n</i> -C ₅ H ₁₁	85	87
k	4-Me	Н	4-CIC ₆ H ₄	82	92
I	4-MeO	Н	4-CIC ₆ H ₄	91	90
m	4-F	Н	4-CIC ₆ H ₄	83	94
n	Н	6-CI	4-CIC ₆ H ₄	90	94
0	Н	7-CI	4-CIC ₆ H ₄	91	91
р	Н	7-CI	Ph	95	96
q	Н	7-CI	4-BrC ₆ H ₄	95	94
r	Н	7-CI	3-MeC ₆ H ₄	90	95
s	Н	7-Cl	n-C ₅ H ₁₁	88	90

groups, benzenesulfonyl moieties with electron-donating groups were found to be most effective. Furthermore, the utility of the newly developed method was demonstrated by preparing useful chiral 4-naphthylquinolines from the resulting products [68].

Chang-Jiang et al. developed a catalytic strategy by using a combination of prolinol silyl ether (cat. 120) and benzoic acid (A1) catalysts to bring about reaction between 3-formyl-substituted indoles or pyrroles 118 and diverse electrophiles, including carbonyls, imines and other Michael acceptors (Scheme 5)

[69]. The reaction with secondary amines occurred via the formation of HOMO raised dearomative aza-dienamine-type intermediates, which undergo direct aza-Michael addition to β -trifluoromethyl enones to afford *N*-functionalized heteroarenes **121** efficiently in moderate to excellent yields, albeit with low to fair enantioselectivity. However, asymmetric aza-Michael additions of these heteroarenes with crotonaldehyde yielded the adducts in moderate to good enantioselectivity under dual catalysis of chiral amines (Scheme 5) [69].

Conclusion

The asymmetric aza-Michael reaction being a useful synthetic strategy for constructing C-N bonds to make a variety of nitrogen-containing chiral scaffolds of wide applications in the fields of pharmaceuticals, organic synthesis building blocks and accessible catalysis continues to attract attention of the chemists. During the last two decades, many new chiral organocatalysts have been developed for accomplishing these reactions with the nitrogen nucleophiles, such as aromatic amines and amides which are otherwise averse to reacting. The organocatalysts have emerged as catalysts of choice due to various reasons, such as their compatibility with the 'Green Chemistry' and possibility of tailoring them according to the requirements. Efforts are directed towards enhancing not only the yields of the products but also enantio- and diastereoselectivities of the aza-Michael reactions. New strategies have been adopted while making optimum utilization of the efficacies of the catalysts. Of these strategies, cascade reactions of the Michael addition in conjunction with one or more reactions leading to overall very high yields and ee are noteworthy. Another strategy of interest appears to be the generation of organocatalysts of enhanced efficacy in situ by mixing squaramides with amino acids again giving >99% ee. It may be perceived that in the coming years, more sophisticated methodologies will be developed with the advent of new organocatalysts to accomplish asymmetric aza-Michael reactions of even the so far unexplored and obstinate amines and amides substrates.

Funding

The authors express gratitude to the authorities of the IIS (deemed to be University), Jaipur, India for providing the necessary research facilities.

ORCID® iDs

Raakhi Gupta - https://orcid.org/0000-0003-0732-5134 Raj K. Bansal - https://orcid.org/0000-0002-8154-9817

References

 Michael, A. J. Prakt. Chem. 1887, 35, 349–356. doi:10.1002/prac.18870350136

- 2. Michael, A. *J. Prakt. Chem.* **1894**, *49*, 20–25. doi:10.1002/prac.18940490103
- Krause, N.; Hoffmann-Röder, A. Synthesis 2001, 171–196. doi:10.1055/s-2001-10803
- Sibi, M. P.; Manyem, S. Tetrahedron 2000, 56, 8033–8061. doi:10.1016/s0040-4020(00)00618-9
- Leonard, J.; DíezBarra, E.; Merino, S. Eur. J. Org. Chem. 1998, 2051–2061. doi:10.1002/(sici)1099-0690(199810)1998:10<2051::aid-ejoc2051>3.0. co:2-t
- Rossiter, B. E.; Swingle, N. M. Chem. Rev. 1992, 92, 771–806. doi:10.1021/cr00013a002
- d'Angelo, J.; Desmaële, D.; Dumas, F.; Guingant, A. Tetrahedron: Asymmetry 1992, 3, 459–505. doi:10.1016/s0957-4166(00)80251-7
- Bartoli, G.; Cimarelli, C.; Marcantoni, E.; Palmieri, G.; Petrini, M. J. Org. Chem. 1994, 59, 5328–5335. doi:10.1021/jo00097a039
- Hecht, S. M. Acc. Chem. Res. 1986, 19, 383–391. doi:10.1021/ar00132a002
- Wang, Y.-F.; Izawa, T.; Kobayashi, S.; Ohno, M. J. Am. Chem. Soc. 1982, 104, 6465–6466. doi:10.1021/ja00387a060
- Hashiguchi, S.; Kawada, A.; Natsugari, H.
 J. Chem. Soc., Perkin Trans. 1 1991, 2435–2444.
 doi:10.1039/p19910002435
- Cardillo, G.; Tomasini, C. Chem. Soc. Rev. 1996, 25, 117–128. doi:10.1039/cs9962500117
- Eliel, E. L.; He, X.-C. J. Org. Chem. 1990, 55, 2114–2119. doi:10.1021/jo00294a026
- Hayashi, Y.; Rohde, J. J.; Corey, E. J. J. Am. Chem. Soc. 1996, 118, 5502–5503. doi:10.1021/ja960766s
- Genov, M.; Dimitrov, V.; Ivanova, V. Tetrahedron: Asymmetry 1997, 8, 3703–3706. doi:10.1016/s0957-4166(97)00534-x
- Thematic issue on organocatalysis. Chem. Rev. 2007, 107, 5413–5883. doi:10.1021/cr078412e
- Ahrendt, K. A.; Borths, C. J.; MacMillan, D. W. C. J. Am. Chem. Soc. 2000, 122, 4243–4244. doi:10.1021/ja000092s
- Sánchez-Roselló, M.; Aceña, J. L.; Simón-Fuentes, A.; del Pozo, C. Chem. Soc. Rev. 2014, 43, 7430–7453. doi:10.1039/c4cs00156g
- Nayak, S.; Panda, P.; Bhakta, S.; Mishra, S. K.; Mohapatra, S.
 RSC Adv. 2016, 6, 96154–96175. doi:10.1039/c6ra21191g
- Bhanja, C.; Jena, S.; Nayak, S.; Mohapatra, S. Beilstein J. Org. Chem.
 8, 1668–1694. doi:10.3762/bjoc.8.191
- Vinogradov, M. G.; Turova, O. V.; Zlotin, S. G. Org. Biomol. Chem.
 2019, 17, 3670–3708. doi:10.1039/c8ob03034k
- Enders, D.; Wang, C.; Liebich, J. X. Chem. Eur. J. 2009, 15, 11058–11076. doi:10.1002/chem.200902236
- Wang, J.; Li, P.; Choy, P. Y.; Chan, A. S. C.; Kwong, F. Y.
 ChemCatChem 2012, 4, 917–925. doi:10.1002/cctc.201200135
- 24. Krishna, P. R.; Sreeshailam, A.; Srinivas, R. *Tetrahedron* **2009**, *65*, 9657–9672. doi:10.1016/j.tet.2009.08.021
- 25. Cai, Y.-F.; Li, L.; Luo, M.-X.; Yang, K.-F.; Lai, G.-Q.; Jiang, J.-X.; Xu, L.-W. *Chirality* **2011**, *23*, 397–403. doi:10.1002/chir.20940
- Lee, H.-J.; Cho, C.-W. J. Org. Chem. 2013, 78, 3306–3312. doi:10.1021/jo4001614
- Bae, J.-Y.; Lee, H.-J.; Youn, S.-H.; Kwon, S.-H.; Cho, C.-W. Org. Lett.
 2010, 12, 4352–4355. doi:10.1021/ol101811c
- Erkkilä, A.; Majander, I.; Pihko, P. M. Chem. Rev. 2007, 107, 5416–5470. doi:10.1021/cr068388p

- Liu, J.-D.; Chen, Y.-C.; Zhang, G.-B.; Li, Z.-Q.; Chen, P.; Du, J.-Y.;
 Tu, Y.-Q.; Fan, C.-A. Adv. Synth. Catal. 2011, 353, 2721–2730.
 doi:10.1002/adsc.201100282
- 30. Zhai, X.-D.; Yang, Z.-D.; Luo, Z.; Xu, H.-T. *Chin. Chem. Lett.* **2017**, *28*, 1793–1797. doi:10.1016/j.cclet.2017.04.017
- Cheng, S.; Yu, S. Org. Biomol. Chem. 2014, 12, 8607–8610. doi:10.1039/c4ob01646g
- 32. Ma, S.; Wu, L.; Liu, M.; Xu, X.; Huang, Y.; Wang, Y. RSC Adv. **2013**, *3*, 11498–11501. doi:10.1039/c3ra41179f
- Jakkampudi, S.; Konda, S.; Arman, H.; Zhao, J. C.-G. *Adv. Synth. Catal.* 2020, 362, 2419–2426. doi:10.1002/adsc.202000227
- Malerich, J. P.; Hagihara, K.; Rawal, V. H. J. Am. Chem. Soc. 2008, 130, 14416–14417. doi:10.1021/ja805693p
- Ullah, M. S.; Itsuno, S. ACS Omega 2018, 3, 4573–4582. doi:10.1021/acsomega.8b00398
- Zhao, B.-L.; Du, D.-M. Asian J. Org. Chem. 2015, 4, 1120–1126. doi:10.1002/ajoc.201500306
- Yang, W.; He, H.-X.; Gao, Y.; Du, D.-M. Adv. Synth. Catal. 2013, 355, 3670–3678. doi:10.1002/adsc.201300670
- Zhou, X.-J.; Zhao, J.-Q.; Chen, X.-M.; Zhuo, J.-R.; Zhang, Y.-P.;
 Chen, Y.-Z.; Zhang, X.-M.; Xu, X.-Y.; Yuan, W.-C. *J. Org. Chem.* 2019, 84, 4381–4391. doi:10.1021/acs.joc.9b00401
- Roy, T. K.; Parhi, B.; Ghorai, P. Angew. Chem., Int. Ed. 2018, 57, 9397–9401. doi:10.1002/anie.201805020
- Li, J.-H.; Du, D.-M. Chem. Asian J. 2014, 9, 3278–3286. doi:10.1002/asia.201402706
- Rajasekar, S.; Anbarasan, P. Org. Lett. 2019, 21, 3067–3071. doi:10.1021/acs.orglett.9b00652
- Wu, Y.-C.; Jhong, Y.; Lin, H.-J.; Swain, S. P.; Tsai, H.-H. G.; Hou, D.-R. *Adv. Synth. Catal.* 2019, 361, 4966–4982. doi:10.1002/adsc.201900997
- An, Z.; Chen, L.; Jiang, Y.; He, J. Catalysts 2019, 9, 713–724. doi:10.3390/catal9090713
- Mahe, O.; Dez, I.; Levacher, V.; Briere, J.-F. Angew. Chem., Int. Ed. 2010, 49, 7072–7075. doi:10.1002/anie.201002485
- Wang, L.; Shirakawa, S.; Maruoka, K. Angew. Chem. 2011, 123, 5439–5442. doi:10.1002/ange.201101307
 Angew. Chem. Int. Ed., 2011, 50, 5327-5330.
- 46. Guo, J.; Yu, S. *Org. Biomol. Chem.* **2015**, *13*, 1179–1186. doi:10.1039/c4ob02227k
- Lebrun, S.; Sallio, R.; Dubois, M.; Agbossou-Niedercorn, F.;
 Deniau, E.; Michon, C. Eur. J. Org. Chem. 2015, 1995–2004.
 doi:10.1002/ejoc.201403573
- Sallio, R.; Lebrun, S.; Capet, F.; Agbossou-Niedercorn, F.; Michon, C.; Deniau, E. Beilstein J. Org. Chem. 2018, 14, 593–602. doi:10.3762/bjoc.14.46
- Yu, X. H.; Wang, W. Chem. Asian J. 2008, 3, 516–532. doi:10.1002/asia.200700415
- 50. Wang, X.-F.; An, J.; Zhang, X.-X.; Tan, F.; Chen, J.-R.; Xiao, W.-J. Org. Lett. **2011**, *13*, 808–811. doi:10.1021/ol1031188
- Torán, R.; Vila, C.; Sanz-Marco, A.; Muñoz, M. C.; Pedro, J. R.; Blay, G. *Eur. J. Org. Chem.* **2020**, 627–630. doi:10.1002/ejoc.201901907
- Hayama, N.; Azuma, T.; Kobayashi, Y.; Takemoto, Y.
 Chem. Pharm. Bull. 2016, 64, 704–717. doi:10.1248/cpb.c15-00983
- Michigami, K.; Murakami, H.; Nakamura, T.; Hayama, N.; Takemoto, Y.
 Org. Biomol. Chem. 2019, 17, 2331–2335. doi:10.1039/c9ob00045c
- Miyaji, R.; Asano, K.; Matsubara, S. Org. Lett. 2013, 15, 3658–3661. doi:10.1021/ol401538b

- Liu, X.; Lu, Y. Org. Biomol. Chem. 2010, 8, 4063–4065. doi:10.1039/c0ob00223b
- Du, J.-Y.; Zeng, C.; Han, X.-J.; Qu, H.; Zhao, X.-H.; An, X.-T.;
 Fan, C.-A. J. Am. Chem. Soc. 2015, 137, 4267–4273.
 doi:10.1021/jacs.5b01926
- Sunoj, R. B. Acc. Chem. Res. 2016, 49, 1019–1028. doi:10.1021/acs.accounts.6b00053
- Rueping, M.; Kuenkel, A.; Atodiresei, I. Chem. Soc. Rev. 2011, 40, 4539–4549. doi:10.1039/c1cs15087a
- Saito, K.; Moriya, Y.; Akiyama, T. Org. Lett. 2015, 17, 3202–3205. doi:10.1021/acs.orglett.5b01654
- Yang, L.; Xia, C.; Huang, H. Chin. J. Catal. 2011, 32, 1573–1576. doi:10.1016/s1872-2067(10)60261-6
- Feng, Z.; Xu, Q.-L.; Dai, L.-X.; You, S. L. Heterocycles 2010, 80, 765–771. doi:10.3987/com-09-s(s)66
- 62. Ojha, M.; Bansal, R. K. *Curr. Catal.* **2021**, in press. doi:10.2174/2211544709999201201120617
- Wang, L.; Chen, J.; Huang, Y. Angew. Chem., Int. Ed. 2015, 54, 15414–15418. doi:10.1002/anie.201508371
- Mukherjee, S.; Yang, J. W.; Hoffmann, S.; List, B. Chem. Rev. 2007, 107, 5471–5569. doi:10.1021/cr0684016
- Lee, S.-J.; Youn, S.-H.; Cho, C.-W. Org. Biomol. Chem. 2011, 9, 7734–7741. doi:10.1039/c1ob06078c
- 66. Guo, H.-M.; Yuan, T.-F.; Niu, H.-Y.; Liu, J.-Y.; Mao, R.-Z.; Li, D.-Y.; Qu, G.-R. Chem. – Eur. J. 2011, 17, 4095–4098. doi:10.1002/chem.201100435
- Joie, C.; Deckers, K.; Raabe, G.; Enders, D. Synthesis 2014, 1539–1546. doi:10.1055/s-0033-1340982
- Zhang, J.; Xu, Y.; Wang, Z.; Zhong, R.; Wang, Y. J. Org. Chem. 2021, 86, 4262–4273. doi:10.1021/acs.joc.1c00163
- Xu, C.-J.; Du, W.; Albrecht, Ł.; Chen, Y.-C. Synthesis 2020, 2650–2661. doi:10.1055/s-0040-1707176

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (https://creativecommons.org/licenses/by/4.0). Please note that the reuse, redistribution and reproduction in particular requires that the author(s) and source are credited and that individual graphics may be subject to special legal provisions.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions:

(https://www.beilstein-journals.org/bjoc/terms)

The definitive version of this article is the electronic one which can be found at:

https://doi.org/10.3762/bjoc.17.173