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Fluorescent monitoring of copper-occupancy in His-ended catalytic oligo-peptides

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ABSTRACT

Controlled generation of reactive oxygen species (ROS) is widely beneficial to various medical, environmental, and agricultural studies. As inspired by the functional motifs in natural proteins, our group has been engaged in development of catalytically active oligo-peptides as minimum-sized metalloenzymes for generation of superoxide anion, an active member of ROS. In such candidate molecules, catalytically active metal-binding minimal motif was determined to be X-X-H, where X can be most amino acids followed by His. Based on above knowledge, we have designed a series of minimal copper-binding peptides designated as $G_{0}H$ series peptides, which are composed of oligoglycyl chains ended with C-terminal His residue such as GGGGGH sequence (G5H). In order to further study the role of copper binding to the peptidic catalysts sharing the X-X-H motif such as G_5 H-conjugated peptides, we should be able to score the occupancy of the peptide population by copper ion in the reaction mixture. Here, model peptides with Cu-binding affinity which show intrinsic fluorescence due to tyrosyl residue (Y) in the UV region (excitation at ca. 230 and 280 nm, and emission at ca. 320 nm) were synthesized to score the effect of copper occupancy. Synthesized peptides include GFP-derived fluorophore sequence, TFSYGVQ (designated as Gfp), and Gfp sequence fused to C-terminal G₅H (Gfp-G₅H). In addition, two Y-containing tri-peptides derived from natural GFP fluorophores, namely, TYG and SYG were fused to the G_5H (TYG- G_5H and SYG- G_5H). Conjugation of metal-binding G₅H sequence to GFP-fluorophore peptide enhanced the action of Cu²⁺ on quenching of intrinsic fluorescence due to Y residue. Two other Y-containing peptides, TYG-G₅H and SYG-G₅H, also showed intrinsic fluorescence which is sensitive to addition of Cu²⁺. There was linear relationship between the loading of Cu^{2+} and the quenching of fluorescence in these peptide, suggesting that Cu^{2+} -dependent quenching of Y-reside-derived fluorescence could be a measure of copper occupancy in the peptides. Lastly, the fate of Y residue in the Cu-loaded peptides under oxidative condition in the presence of H_2O_2 was discussed based on the Cu/ H_2O_2 dependent changes in fluorescence spectra.

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Introduction

Controlled generation of reactive oxygen species (ROS) is widely beneficial to various medical, environmental engineering, and agricultural fields including clinically applied immunological modulations,^{2,7} degradation of polluting organic compounds,²³ inactivation of bacterial cells for the hygienic purpose,¹⁹ activated sludge process,²² and direct and indirect agricultural pest controls targeting pathogens and host plants, respectively.^{16,29}

In the last decade, our group has been engaged in development a novel classes of engineered biocatalysts including catalytic oligonucleotides (functional DNA sequences;¹⁰ and peptides,¹¹ designed to catalyze the production and/or removal superoxide anion radicals

 $(O_2^{\bullet-})$ through understanding and modification of natural catalytic proteins of animal and plant origins.

For example, we found that peptides derived from human and mammalian prion proteins $(PrPs)^{14,24}$ and plant stress-responsive peptides²⁵ have catalytic nature although they are not considered as enzymes at present. Actually, the kingdoms of plants and animals are rich in such small peptides with high affinity to metal ions which might aid in catalysis. By mimicking such natural peptide, novel series of minimal-sized oligo-peptidic artificial enzymes catalyzing the generation of $O_2^{\bullet-}$ in peroxidase-like manner requiring hydrogen peroxide (H_2O_2) and electron donating substrates such as phenolics or amines were developed.^{11,21}

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The first criterion for consisting such minimal peroxidase-like small peptides is the presence of His-containing motif(s) required for binding to metals (chiefly copper), and then, free-form and/or peptide-bound form of substrates fuels the reaction.²⁷ Our preliminary studies on PrP-derived peptides have pointed that His residues (at least single His) are required for anchoring of Cu onto PrP-derived peptides,^{13,14} and eventually, the catalytically active Cu-binding minimal motif was determined to be tri-peptidic sequence X-X-H, where X can be most amino acids followed by His^{12,27} An engineering example of a catalytic peptide sequence sharing XXH motif for development of biosensing and bioengineering materials was reported.²¹ Accordingly, one of X-X-H motif derivatives, Gly-Gly-His (G-G-H) sequence was introduced onto the glycidyl methacrylate grafted on the polyethylene platform (porous hollow fiber membrane). Chemiluminescence assay reveled that loading of Cu^{2+} on the peptideconjugated membrane conferred the catalytic activity to it, thus, catalyzing the generation of $O_2^{\bullet-}$ upon addition of a pair of substrates (H_2O_2 and tyramine).

In human PrP, His96 centered in G-G-G-T-H-S-Q-W-N sequences is considered as one of Cu-binding sites. Effect of His position on the catalytic activity in PrPderived peptides was examined by comparing the H-S-Q-W-N (His-started pentapeptide) and the G-G-G-T-H (His-ended pentapeptide).¹¹ While reaction with tyramine (given as a model phenolic substrate) and G-G-G-T-H peptide resulted in robust production of $O_2^{\bullet-}$, the H-S-Q-W-N peptide showed no catalytic activity, suggesting that G-T-H motif within the His-ended pentapeptide is one of X-X-H motif derivatives. As the catalytic activities among G-G-G-T-H and shorter derivatives (G-G-T-H and G-T-H) were compared, the importance of the N-terminal glycyl-chain elongation for maximal redox activity in C-terminal His anchored peptides was implied.¹¹ Furthermore, the likely common structure formed by Cu/X-X-H motif complex found in Cu-binding motifs in human PrP including octarepeat region⁹ helical region,²⁷ neurotoxic region¹² was proposed to be semi-planar shape resembling the structure of metal-centered heme in which metallic element is coordinated by planarly arranged four nitrogen atoms, the structure Ni/X-X-H by analogy to of metallopeptides.5,6

As inspired by the natural PrP-derived G-G-G-T-H sequence, we have designed a series of simplified model peptides designated as G_nH series peptides, which are composed of oligo-glycyl chains (G_n) ended with *C*-terminal His residue.¹¹ As expected, importance of the elongated *N*-terminal G_n chain with anchoring His was confirmed by comparing the G_nH series peptides (n = 2, 3, 4, 5 and 10) and G_n series peptides lacking His.

Notably, G_n series lacking the metal-binding motif showed no catalytic activity even in the presence of free Cu^{2+} . In G_nH series, G_3H tetrapeptide showed a detectable increase in production of $O_2^{\bullet-}$, and peptides with longer chain showed higher activity, confirming the importance of *N*-terminal G_n chain length. Data suggested that the requirement for the *N*-terminal G_n elongation is nearly fulfilled at between G_5 and G_{10} ,¹¹ and amazingly, unlikely to conventional enzymes, the catalytic activity in these metal-binding peptides survive the heating treatment such as autoclaving and the repeated freeze and thaw cycles.²⁸

Fluorometry often provides strong approaches for studying the molecular interaction.⁹ We have previously assessed (1) the quenching of Tb^{3+} fluorescence by PrPderived metal-binding peptides and (2) the Cu^{2+} -dependent quenching of intrinsic fluorescence in human PrP octarepeat peptide sequence. Quenching of Tb-fluorescence by interacting peptides implied the important role for His-ended peptidic sequence sharing X-X-H motif (in case of human PrP's octarepeat region, P-Q-H). On the other hand, quenching of intrinsic peptide fluorescence due to the presence of a tryptophan (W) residue by copper ion suggested that classically known H-G-G-G motif in PrP³ forms an active motif in metal binding. Taken together, in the mammalian PrP octarepeat regions, in which P-H-G-G-W-G-Q is repeated for four (human) to six (bovine) times, two distinct metal binding motifs, namely, X-X-H motif (in this case, Q-P-H tripeptide sequence) and H-G-G-G motif, could be overlaid by sharing common His residue (Q-P-H-G-G-G) and thus coexisted and synergically capturing the metals.⁹

In order to further study the role of copper binding to the biocatalysts sharing the X-X-H motif such as G_nH series catalytic peptides, we should be able to score the occupancy of the peptide population by copper ion in the reaction mixture. In the present study, we attempted to monitor the binding of copper to G_nH catalytic peptides by designing the chimeric molecule fusing fluorescent oligo-peptide sequence derived from green fluorescence protein (GFP) and G_5H sequence.

Materials and methods

Peptides and chemicals

Model peptides with Cu-binding affinity which show intrinsic fluorescence in the UV region were synthesized to score the effect of copper occupancy. The peptides were obtained from the custom peptide service department of Sigma Genosys Japan, Ishikari, Hokkaido, Japan. The amino acid sequences of the peptides chemically synthesized were purified on high pressure liquid chromatography prior to the experimental use. Other chemicals such as $CuSO_4$ and salts for buffer used in this study were of reagent grade purchased from Wako Pure Chemical Industries Ltd. (Osaka, Japan).

Synthesized peptides include GFP-derived fluorophore sequence, TFSYGVQ (designated as Gfp), and Gfp sequence fused to GGGGGH, thus, designated as Gfp- G_5H . Note that the synthesized peptide sequences corresponding to GFP fluorophore do not show green fluorescence without post-translational process for developing the molecular rigidity in living cells. Instead, intrinsic fluorescence due to presence of the tyrosyl residue (Y) can be expected. Therefore, two Y-containing tri-peptide sequences found in natural GFP fluorophores, namely, TYG and SYG were fused to the Cu-binding GGGGGH sequence (thus, designated as TYG-G₅H and SYG-G₅H, respectively).

Fluorometric analysis

Intrinsic fluorescence from the 30 μ M peptides with and without loading of copper ions were detected in potassium phosphate buffer (50 mM, pH 7.0) using a fluorescence spectrophotometer (F-4500 Hitachi High-Technol. Co., Tokyo). The three-dimensional (3D) spectral measurement of fluorescence was carried out at the excitation wavelength between 200 and 700 nm with 5 nm intervals and emission wavelength between 200 and 700 nm with 5 nm intervals.

To perform full occupancy of peptide with metals, the concentrations of metals of interest tend to be higher than the peptide concentration as previously observed in fluorescence measurements with prion-derived oligopeptides.^{9,13} High concentration of metals in phosphatebased reaction mixture often cause a slight turbidity which sensitively interferes with fluorescence measurement by developing an intensive diagonal band of scattered light over the fluorescence contour plot. To avoid such situation, peptides needs to be maintained at around 30 μ M (so that metal concentration attaining full occupancy can be lowered) as described in our previous reports.^{9,13}

Results and discussion

Enhanced Cu²⁺-dependent changes in intrinsic fluorescence spectra of a GFP-fluorophore peptide conjugated with metal binding sequence

GFP fluorophore-derived peptides tested here showed non-green intrinsic fluorescence at UV region (Fig. 1). Upon excitation at 230 nm (peak a) and 280 nm (peak b), both Gfp-G₅H (TFSYGVQ-GGGGGGH) and Gfp (TFSYGVQ) showed fluorescence emission at around 320 nm. Fluorescence signals at the peak a (230 nm excitation/320 nm emission) by 30 μ M Gfp-G₅H and Gfp showed tendency to be quenched in the presence of 30 μ M Cu²⁺ by 61.6 % and 32.0 %, respectively. Under the same conditions, the fluorescence signals at the peak b (280 nm excitation/320 nm emission) by Gfp-G₅H and Gfp showed Cu²⁺-dependent quench by 48.5 % and 22.7 %, respectively. These data suggested that presence of Cu may partially quench the fluorescence of Gfp-heptapeptide lacking the Cu-binding motif due to non-specific interaction. Addition of C-terminal G₅H sequence doubled the extent of fluorescence quenching induced by CuSO₄, thus suggesting the impact of selective binding of copper onto the G5H sequence conjugated to the sequence with fluorophore. Therefore, we can expect that Cu-dependent quenching of UVC-excited UVA fluorescence by Gfp-G₅H can be used as a measure of Cu-occupancy in G₅H domain.

Intrinsic fluorescence in tyrosine-containing peptides showed sensitivity to Cu

The intrinsic fluorescence signals (the peaks a and b) from the three G_5H -conjugated GFP-derived peptides (30 μ M of Gfp- G_5H , TYG- G_5H , and SYG- G_5H) plotted on 3-dimentional contour graphs were compared. Data showed that the intrinsic fluorescence by all peptides can be quenched in the presence of CuSO₄ (10, 25 and 100 μ M; Fig. 2). Among three peptides examined, SYG- G_5H was most sensitive to lower range of CuSO₄ concentration (10 μ M, molar ratio to peptide: *ca*. 0.17–0.33). The fluorescent signals from TYG- G_5H also showed higher sensitivity to Cu as compared to the signals from Gfp- G_5H .

*Linear relationship between loading of Cu*²⁺ *and quenching of fluorescence*

Among three peptides examined, only Gfp-G₅H showed linear decrease in fluorescent signal along with occupancy with copper in the range between 0.17 and 1.33 of molar ratios of Cu²⁺ over peptide (Figs. 3 and 4). Note that within this range of copper concentration, the squared correlation coefficients (r²) for Cu-dependent quenching of fluorescence signals at the peaks a and b in Gfp-G₅H were 0.982 and 0.989, respectively (Figs. 3 and 4, insets). On the other hand, r² for regressions corresponding to Cu-dependent quenching of fluorescence signals by TYG-G₅H and SYG-G₅H ranged at relatively low scores between 0.700 and 0.906 (Figs. 3 and 4, insets), suggesting that linear relationship between the quenching of TYG- and SYG-conjugated molecules are



Figure 1. Different quenching action of copper ion against intrinsic tyrosine fluorescence in GFP-derived fluorophore sequence with and without fusing to copper binging sequence hexapeptide. Peptides (30 μ M) used were Gfp-G₅H (TFSYGVQ-GGGGGH) and Gfp (TFSYGVQ). Peaks of tyrosine fluorescence (emission at *ca*. 320 nm) were observed with excitation at 230 nm (a) and 280 nm (b). Quenching of fluorescence in the presence of 30 μ M CuSO₄ was assessed.



Figure 2. Effect of copper ion on quenching of intrinsic fluorescence signals by three GFP fluorophore-derived oligo-peptides conjugated with copper-binding hexapeptide motif. Peptides (30 μ M) used were Gfp-G₅H (TFSYGVQ-GGGGGGH), TYG-G₅H (TYG-GGGGGGH), and SYG-G₅H (SYG-GGGGGGH). In the absence of CuSO₄, two typical peaks of tyrosine fluorescence (emission at *ca*. 320 nm) were observed with excitation at 230 nm (a) and 280 nm (b). Quenching of fluorescence in the presence of 10, 25 and 100 μ M CuSO₄ was assessed.



Figure 3. Effect of copper concentration on quenching of 230 nm excitation/3;20 nm emission signals by three G₅H-conjugated GFP fluorophore-derived oligo-peptides. Peptides (30 μ M) used were as in Figure 2. Quenching of fluorescence was performed with 5–100 and 100 μ M CuSO₄ was assessed. Three different symbols represent the data points obtained. Curves were merely approximation of the response (note that they are not regression curves). In the inset, linear relationships between the remitted range of Cu concentration (up to 40 μ M) and the decrease in peptidic fluorescent signals are shown.



Figure 4. Effect of copper concentration on quenching of 280 nm excitation/320 nm emission signals by three G₅H-conjugated GFP fluorophore-derived oligo-peptides. Peptides (30 μ M) used were as in Figure 2. Quenching of fluorescence was performed as in Figure 3. In the inset, linear relationships between the remitted range of Cu concentration (up to 40 μ M) and the decrease in peptidic fluorescent signals are shown.

less significant compared to the Gfp-conjugated molecule. Therefore, we can conclude that the fluorometric kinetics reported by Gfp-conjugated G_5H peptide is most proportional to the copper occupancy in G_5H sequence.

Previously, catalytic nature of G₅H hexapeptide was assessed.¹¹ It has been shown that catalytic activity in G₅H peptide requires the binding of copper to it. Furthermore, Michaelis constant ($K_{\rm m}$) for $O_2^{\bullet-}$ production using tyramine as a model substrate for Cu/G₅H complex (0.15 mM) was determined to be 0.24 mM. Then, V_{max} at molar basis and weight basis were determined to be 52.91 mmol $(O_2^{\bullet-})$ mmol (peptide)⁻¹ min⁻¹ and 0.12 mmol $(O_2^{\bullet-})$ mg (peptide) $^{-1}$ min $^{-1}$, respectively.¹¹ With molar-basis comparison, the catalytic activity looks weak, however, due to its low molecular weight characteristics, weightbasis comparison of the catalytic activity reaches applicable range which is almost 1/6 of purified horseradish peroxidase.¹¹ It is obvious that this type of approach for creating heat-stable biocatalysts require further innovation. Therefore, we expect that the use of Gfp-fused G₅H peptide for quantification of Cu-binding to G₅H motif may contribute to further engineering of the G_n H-based catalytic peptides.

Fate of tyrosine residue after oxidative reaction

Up to here we mostly discussed the role of His-ended metal binding motif in novel class of catalytic peptides with aid by fluorescent signal which could be attributed to the presence of Tyr residue. We view that Tyr residue has an additional important role in designing the peptidic $O_2^{\bullet-}$ -generating catalysts.¹⁵

Reportedly, supplementation of structurally similar free catecholamine-related chemicals (tyramine or phenylethylamine), a free amino acid Tyr (Y), or Tyr-rich oligopeptides (such as tyrosyl-tyrosyl-arginine, YYR) as model substrates (instead of typical peroxidase substrates such as phenolics or amines) to the reaction mixture containing Cu-bound peptides sharing X-X-H motif such as Cu/VNITKQHTVTTTT (helical Cu-binding motif in mammalian PrP;24 Cu/G;GGTH (short Cubinding motif in human PrP;¹⁴ Cu/G;GGGGH (artificial catalyst;¹¹ Cu/GGGFGH,²⁵ or Cu/NPGFPH²⁶ resulted in H₂O₂-dependent O₂^{•-}-generation. Notably, Y residuecontaining peptides with X-X-H motif including Cu/ GGGYGH (plant ozone-inducible peptide sequence;²⁵ Cu/NPGYPH (chicken PrP hexa-repeat sequence;²⁶ and Cu/FLTEYVA-GGGGGH (Erk1/Erk2 MAP kinase substrate sequence fused with metal binding sequence designated as ErkG₅H;¹⁵ showed catalytic activity for H₂O₂dependent $O_2^{\bullet-}$ generation without supplementation of any free phenolic substrates. These knowledges suggest that Y-residue mimics the role for free phenolics in H_2O_2 -dependent $O_2^{\bullet-}$ generating reactions.

The above view was supported by Y-to-F mutation in chicken PrP sequence²⁴ and plant ozone-inducible peptide sequence²⁵ by which Tyr residues on catalytic peptides were replaced with Phe residues. The difference was merely the presence and absence of the OH group on the aromatic ring. Moreover, masking of Y-residue (at OH group) in ErkG₅H through tyrosyl phosphorylation in Erk1/Erk2 MAP kinase substrate moiety of ErkG₅H peptide was performed, and catalytic activity for H₂O₂-dependent O₂^{•–} generation was largely lost in the resultant Y-phosphorylated peptide.¹⁵ To date, ErkG₅H is the only artificial catalyst which can be attenuated by phosphorylation event.

Involvement of free Y or Y residue in the Cu/peptide complex-catalyzed H_2O_2 -dependent generation of $O_2^{\bullet-}$ suggests that a phenoxy radical derived from Y (tyrosyl radical, Y[•]) can be formed in aid of single electron reduction of molecular oxygen, by analogy to plant enzymes H_2O_2 -dependently generating $O_2^{\bullet-}$ by coupling to oxidation of phenolics (such as salicylic acid) to form phenoxy radical.^{17,18}

There would be two distinct models for the fate of Y residue in model Cu/peptide complex. One likely model is that Y-residue is simply consumed and oxidized by the Cu-bound catalytic center in the presence of H_2O_2 as observed for various free phenolics.¹¹ Another likely model is that Y residue lasts longer by repeatedly participating the reaction as shuttle for transferring electron, by analogy to the putative intra-molecular substrate-like roles for Y residues within ribonucleotide reductases.¹ and cyclooxygenase-2,²⁰ in which corresponding reactions proceed *via* transient formation of Y[•] and recycling of Y.

To obtain a clue to this view, we examined the fate of Y-dependent fluorescence in Gfp-G₅H, TYG-G₅H, and SYG-G₅H peptides after addition of Cu and H₂O₂ (Fig. 5). Ratio of H₂O₂ concentration (1 mM) over Cu/ peptide concentration (30 μ M) was set at excess level since higher range of H₂O₂ concentration has been employed in the previous studies using G₅H-based catalysts.

Addition of copper to three peptides largely lowered the fluorescence signals as described earlier in this report. Addition of H_2O_2 to SYG-G₅H lowered the fluorescence signals. Contrary, addition of H_2O_2 enhanced the fluorescence at both peaks a and b in TYG-G₅H and the peak a in Gfp-G₅H. The reason why two peaks of Y fluorescence in different peptides showed different sensitivity to H_2O_2 should be attributed to the fact that even a monomer of phenolic compound often possesses multiple fluorophores within the molecule despite its simple structure as in the case of ferulic acid.^{4,8}



Figure 5. Changes in UV-excited fluorescence contour spectra in three peptides after addition of copper and/or excess hydrogen peroxide. Peptides used were as in Figure 2, namely Gfp-G₅H, TYG-G₅H and SYG-G₅H. Each peptide (30 μ M) was treated with none, either or both of CuSO₄ (30 μ M) or/and H₂O₂ (1 mM). Numbers after (a) and (b) shown with each spectrum represent the relative changes in fluorescence intensities at peaks a (230 nm excitation/320 nm emission) and b (280 nm excitation/320 nm emission), respectively.

To the combination of Cu and H₂O₂, three peptides responded differently. Response to the Cu/H₂O₂ co-treatment in Gfp-G₅H was almost identical to the response to Cu alone. Changes in SYG-G₅H were less obvious. The fluorescence intensities at 230 nm excitation/320 nm emission and 280 nm excitation/320 nm emission corresponding to the peaks a and b in Cu/H₂O₂ co-treated TYG-G₅H were seemed to be maintained at higher level compared to control. Note that the peak excitation wavelength at peak b fluorescence slightly shifted from 280 to 290 nm, therefore the product of peptide-catalyzed redox reaction challenging Y-residue under Cu/H2O2 co-treatment must be no-longer intact Y residue. The case in TYG-G₅H suggests that after possible formation of Y[•] via Cu/peptide-catalyzed H₂O₂-dependent reaction, recycling of Y did not sufficiently occur thus a spectral change (shift in the excitation peak) was observed.

Disclosure of potential conflicts of interest

No potential conflicts of interest were disclosed.

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