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**Cancer Cells** 

# Article Structure Modification of Ginsenoside Rh<sub>2</sub> and Cytostatic Activity on

3', 6'-Dimaleimide caproyl ginsenoside Rh<sub>2</sub>

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Cite This: ACS Omega 2023, 8, 17245-17253 **Read Online** ACCESS III Metrics & More Article Recommendations SI Supporting Information **ABSTRACT:** Ginsenoside  $Rh_2$  ( $Rh_2$ ) is one of the most effective anticancer components extracted from red ginseng, but the poor % solubility limits its clinical application. In this paper, ginsenoside Viability Rh<sub>2</sub> was modified with maleimidocaproic acid or maleimidoundecanoic acid with functional groups at both ends. The structures of le l

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derivatives were determined by analysis of 1D and 2D nuclear magnetic resonance, Fourier transform infrared, and highresolution mass spectrometry. Antiproliferative cell experiments showed that  $Rh_2$  modified with maleimidocaproic acid (C-Rh<sub>2</sub>) displayed higher cytostatic activity against different tumor cells compared with Rh<sub>2</sub>, while Rh<sub>2</sub> modified with maleimidoundecanoic acid (U-Rh<sub>2</sub>) did not exhibit obvious cytotoxicity. The results suggest that the length of the spacer arm may play an important role in the cytostatic activity of the Rh<sub>2</sub> derivatives.

### INTRODUCTION

Ginseng has been used as a multipotent herbal medicine in East Asian countries (such as Korea, Japan, and China).<sup>1</sup> Many studies show that ginsenosides, especially triterpene saponins, are the main active ingredients in ginseng, which are composed of ginsengenin and sugar through an ether bond.<sup>2</sup> Among them, ginsenoside  $Rh_2$  ( $Rh_2$ ) (Figure 1) has been adequately



Figure 1. Structure of Rh<sub>2</sub>.

investigated and exhibits remarkable functions including antitumor, anti-inflammatory, and antiallergic effects and enhancement of immune activity.<sup>3-6</sup> It is widely used to resist liver cancer, leukemia, cervical cancer, prostate cancer, colon cancer, breast cancer, etc. $^{7-12}$  Several recognized anticancer mechanisms of  $Rh_2$  include (1) inducing apoptosis and autophagy of cancer cells;<sup>13</sup> (2) triggering cycle arrest of cancer cells;<sup>14</sup> (3) inhibiting invasion, migration, and meta-stasis of cancer cells;<sup>15,16</sup> and (4) suppressing angiogenesis and epithelial–mesenchymal transition (EMT).<sup>17,18</sup>

However, despite the advantages mentioned above, the clinical application of Rh<sub>2</sub> has been hampered due to its low solubility and difficulty in absorption.<sup>19</sup> Therefore, chemical structure modification (sulfation, esterification, acetylation, etc.) on ginsenosides has been carried out to improve the bioavailability and antitumor efficacy.<sup>20–23</sup> Maleimide aliphatic acids are a class of pharmaceutical intermediate with hydrophilic functional groups at both ends.<sup>24,25</sup> In our work, two maleimide fatty acids with different carbon chain lengths, namely, 6-maleimidocaproic acid and 11-maleimidoundecanoic acid, were attached to Rh<sub>2</sub>, respectively, through the reaction between carboxyl groups in maleimide fatty acid and hydroxy groups in Rh<sub>2</sub> to improve the disadvantages of Rh<sub>2</sub>. The junction sites and numbers were characterized using nuclear magnetic resonance (NMR), Fourier transform infrared (FTIR), and high-resolution mass spectrometry (HRMS). The effect of the length of the spacer arm on the antiproliferative activity of Rh<sub>2</sub> derivatives was preliminarily investigated.

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3', 6'-Dimaleimide undecanoyl ginsenoside Rh<sub>2</sub>

Scheme 1. Synthesis of Rh<sub>2</sub> Derivatives (C-Rh<sub>2</sub> and U-Rh<sub>2</sub>)



#### RESULTS AND DISCUSSION

Synthesis and Characterization of  $Rh_2$  Derivatives. 6-Maleimidocaproic acid-modified  $Rh_2$  (C- $Rh_2$ ) and 11-maleimidoundecanoic acid-modified  $Rh_2$  (U- $Rh_2$ ) were synthesized and purified as described in the Experimental Section (Scheme 1), and the structures were elucidated using a combination of FTIR, NMR, and HRMS.

C–Rh<sub>2</sub> was isolated as a white powder. The molecular formula of C–Rh<sub>2</sub> was deduced as  $C_{56}H_{84}N_2O_{14}$  on the basis of a protonated molecular ion peak at m/z 1009.5989 [M + H]<sup>+</sup> in HRESIMS (calcd 1009.5995, Figure S12, Supporting Information), suggesting that there were 2 equiv of 6-maleimidocaproic acid molecules connected to Rh<sub>2</sub>. The FTIR spectrum (Figure S13, Supporting Information) displayed absorption bands characterized as the hydroxyl group (3100–3500 cm<sup>-1</sup>), carbonyl group (1706 cm<sup>-1</sup>), and carbon–carbon double bond (1628 cm<sup>-1</sup>), whereas out-of-plane bending vibration ( $\gamma_{O-H}$ , 948 cm<sup>-1</sup>) of the hydroxyl group in the carboxyl group of 6-maleimidocaproic acid disappeared in C–Rh<sub>2</sub>, indicating that the carboxyl group was converted to the ester group after 6-maleimidocaproic acid was linked onto Rh<sub>2</sub>.

The <sup>1</sup>H NMR and <sup>13</sup>C NMR data of C–Rh<sub>2</sub> were similar to those of Rh<sub>2</sub> (Table 1 and Figures S6 and S7, Supporting Information), except that the signals corresponding to maleimide fatty acyl appeared (Table 1). The <sup>1</sup>H NMR spectrum of Rh<sub>2</sub> displayed signals ranging from  $\delta_{\rm H}$  5.00 to 4.00 for protons in the glycogen moiety,  $\delta_{\rm H}$  5.30 typical for the double bond, and  $\delta_{\rm H}$  less than 4.00 for protons in the ginsengenin portion. In the <sup>13</sup>C NMR data, the carbon signals of  $\delta_{\rm C}$  ranging from 107.46 to 63.55 were mainly corresponding to glycogen (except  $\delta_{\rm C}$  89.26, 73.45, and 71.48 for C-3, C-20, and C-12 in ginsengenin, respectively), while the signals at  $\delta_{\rm C}$  131.25 (C-25), 126.81 (C-24), and  $\delta_{\rm C}$  less than 60 were diagnostic for ginsengenin.

Besides the signals mentioned above, the <sup>1</sup>H NMR spectrum of C–Rh<sub>2</sub> (Table 1 and Figure S14, Supporting Information) exhibited signals at  $\delta_{\rm H}$  6.82 (H-8" and H-9"), 3.50 (H-6"), 2.33 (H-2"), 1.59 (H-3"), 1.51 (H-5"), and 1.29 (H-4") diagnostic for maleimide caproyl groups. In the <sup>13</sup>C NMR spectrum of C–Rh<sub>2</sub> (Table 1 and Figure S15, Supporting Information), signals associated with maleimide caproyl groups were split into two adjacent groups, namely,  $\delta_{\rm C}$  173.85, 35.00, 25.26, 26.92, 28.96, 38.06, 171.75, and 134.87 (C-1"–8") and  $\delta_{\rm C}$ 

173.78, 34.66, 25.18, 26.78, 28.90, 38.02, 171.72, and 134.83 (another group of C-1" $\sim$ 8"). This just indicated that two 6-maleimidocaproic acid molecules had connected onto Rh<sub>2</sub>, which was consistent with the formula obtained by HRMS.

To figure out which two sites of Rh<sub>2</sub> the 6-maleimidocaproic acid molecules were connected to, the structure of C-Rh<sub>2</sub> was characterized by HSQC, <sup>1</sup>H-<sup>1</sup>H COSY, and HMBC twodimensional NMR. There were no significant changes in the  $\delta_{
m H}$ and  $\delta_{\rm C}$  of C-12, C-20, and neighboring atoms, while the main changes of chemical shifts were observed at the glucose moiety, so we speculated that the esterification occurred on the glycogen ring. Since there was no hydroxyl group on C-1' to participate in the reaction, so the chemical shifts of C-1' ( $\delta_{\rm C}$ 107.46) and H-1' ( $\delta_{
m H}$  4.97) were similar before and after esterification. The HSQC spectrum of C-Rh<sub>2</sub> (Figure S16, Supporting Information) showed that the  $\delta_{\rm C}$  value of C-6' was 64.56, while the  $\delta_{\rm H}$  values of H-6'<sub>a</sub> and H-6'<sub>b</sub> downfield shifted to 4.99 and 4.79, respectively. Meanwhile, the cross-peak signals of H-6'<sub>a</sub> ( $\delta_{\rm H}$  4.99) and H-6'<sub>b</sub> ( $\delta_{\rm H}$  4.79) to C-1" ( $\delta_{\rm C}$ 173.85 and 173.78) of ester carbonyl in maleimide caproyl groups were observed in the HMBC spectrum (Figure S17, Supporting Information), indicating that one of the 6maleimidocaproic acid molecules was attached to C-6' (Figure 2).

The coupling correlation between  $\delta_{
m H}$  4.07 and H-1'  $(\delta_{
m H}$ 4.89) in the <sup>1</sup>H-<sup>1</sup>H COSY spectrum of C-Rh<sub>2</sub> (Figure S18, Supporting Information) suggested that  $\delta_{\rm H}$  4.07 belonged to H-2', and the cross-peak of H-2' ( $\delta_{\rm H}$  4.07) to carbon  $\delta_{\rm C}$  73.81 in the HSQC spectrum implied that  $\delta_{\rm C}$  73.81 could be assigned to C-2'. The presence of a signal at H-6'  $_{\rm b}$  ( $\delta_{\rm H}$  4.79) and  $\delta_{\rm H}$  4.10 in the <sup>1</sup>H-<sup>1</sup>H COSY spectrum allowed the assignment of  $\delta_{\rm H}$  4.10 to H-5', and the correlation from H-5'  $(\delta_{\rm H} 4.10)$  to  $\delta_{\rm C}$  75.32 in HSQC further inferred that  $\delta_{\rm C}$  75.32 was due to C-5'. In the HMBC spectrum, besides the correlationship between H-6'<sub>b</sub> and C-5', the cross-peak of H- $6'_{\rm a}~(\delta_{\rm H}~4.99)$  to carbon  $\delta_{\rm C}~70.21$  supported the assignment of  $\delta_{\rm C}$  70.21 to C-4', and then the signal from  $\delta_{\rm H}$  4.12 to C-4' ( $\delta_{\rm C}$ 70.21) in the HSQC spectrum provided evidence that  $\delta_{\rm H}$  4.12 was due to H-4'. Finally, the HSQC cross-peak of the proton at  $\delta_{\rm H}$  5.80 and carbon  $\delta_{\rm C}$  79.43 should be assigned to H-3' and C-3', which was consistent with the H-3'  $(\delta_{\rm H}~5.80)$  and H-4'  $(\delta_{\rm H} 4.12)$  coupling observed in the <sup>1</sup>H-<sup>1</sup>H COSY spectrum and the correlation signal from H-3' ( $\delta_{
m H}$  5.80) to C-4' ( $\delta_{
m C}$ 70.21) and C-2' ( $\delta_{\rm C}$  73.81) in the HMBC spectrum.

## Table 1. <sup>13</sup>C (100 MHz) and <sup>1</sup>H NMR (400 MHz) Data for Rh<sub>2</sub>, C-Rh<sub>2</sub>, and U-Rh<sub>2</sub> in Pyridine-d<sub>5</sub>

		Rh <sub>2</sub>	C-Rh <sub>2</sub>		U-Rh <sub>2</sub>	
position	$\delta_{\rm C'}$ type	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{\mathrm{C}}$ , type	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{\mathrm{C}'}$ type	$\delta_{\mathrm{H}\prime}$ type
C-1	40.17, CH <sub>2</sub>	1.48, m	40.10, CH <sub>2</sub>	1.43, m	40.12, CH <sub>2</sub>	1.53, m
		0.78, s		0.84, s		0.82, s
C-2	27.34, CH <sub>2</sub>	2.24, dd (13.6, 3.5)	27.32, CH <sub>2</sub>	2.22, d (11.9)	27.34, CH <sub>2</sub>	2.24, d (9.7)
		1.83, m		1.90, m		1.92, m
C-3	89.26, CH	3.38, dd (11.5, 4.5)	89.90, CH	3.34, dd (11.5, 4.1)	89.93, CH	3.36, dd (11.6, 4.1)
C-4	40.49, C		40.50, C		40.53, C	
C-5	56.85, CH	0.74, d (11.5)	56.85, CH	0.79, d (10.8)	56.93, CH	0.82, d (10.8)
C-6	18.94, CH <sub>2</sub>	1.48, m	18.92, CH <sub>2</sub>	1.47, m	18.95, CH <sub>2</sub>	1.53, m
		1.36, m		1.33, m		1.41, m
C-7	35.64, CH <sub>2</sub>	1.48, m	35.60, CH <sub>2</sub>	1.51, m	35.64, CH <sub>2</sub>	1.53, m
		1.21, m		1.22, m		1.28, m
C-8	37.44, C		37.49, C		37.52, C	
C-9	50.86, CH	1.40, m	50.91, CH	1.48, m	50.96, CH	1.44, m
C-10	39.61, C		39.71, C		39.78, C	
C-11	32.55, CH <sub>2</sub>	2.05, m	32.52, CH <sub>2</sub>	2.10, m	32.54, CH <sub>2</sub>	2.10, m
_		1.51, m		1.52, m		1.53, m
C-12	71.48, CH	3.91, m	71.46, CH	3.92, m	71.46, CH	3.92, m
C-13	49.04 CH	2.02, m	49.06, CH	2.04, m	49.08, CH	2.07, m
C-14	52.19, C	1.54	52.19, C	1.51	52.19, C	1.55
C-15	31.83, CH <sub>2</sub>	1.54, m	$31.81, CH_2$	1.51, m	31.83, $CH_2$	1.55, m
C 16	27.21 CH	1.05, m	27.16 CH	1.06, m	27.10 CH	1.08, m
C-10	27.21, CH <sub>2</sub>	1.90, m	$2/.10, CH_2$	1.90, m	$2/.19, CH_2$	1.92, m
C 17	55 20 CH	1.42, m 2.25 $dd(10.5, 2.5)$	55 20 CH	1.43, m	55 20 CH	1.38, m
C-17	16 22 CH	2.55,  dd $(10.5, 5.5)$	16 21 CH	2.58, III	16 34 CH	2.57, III
C-10	16.86 CH	0.78 s	$16.91, CH_3$	0.96, 8	16.86 CH.	1.00, s
C-20	73.45 C	0.70, 3	73.44 C	0.04, 3	73.45 C	0.02, 3
C-21	27.57 CH	142 s	27.54 CH	143 s	27.57 CH	144 s
C-22	36.37, CH <sub>2</sub>	2.03. m	36.33, CH <sub>2</sub>	2.04, m	36.35, CH <sub>2</sub>	2.05. m
	00107, 0012	1.68, m	00000, 0002	1.69, m	00000, 0002	1.69. m
C-23	23.48, CH <sub>2</sub>	2.59, m	23.47, CH <sub>2</sub>	2.60, m	23.48, CH <sub>2</sub>	2.59, m
	, 2	2.27, m	, 2	2.28, m	, 2	2.27, m
C-24	126.81, CH	5.30, t (7.0)	126.79, CH	5.31, t (7.1)	126.80, CH	5.32, t (7.2)
C-25	131.25, C		131.25, C		131.24, C	
C-26	26.33, CH <sub>3</sub>	1.63, s	26.33, CH <sub>3</sub>	1.66, s	26.33, CH <sub>3</sub>	1.57, s
C-27	18.19, CH <sub>3</sub>	1.61, s	18.18, CH <sub>3</sub>	1.60, s	18.20, CH <sub>3</sub>	1.64, s
C-28	28.65, CH <sub>3</sub>	1.34, s	28.52, CH <sub>3</sub>	1.33, s	28.53, CH <sub>3</sub>	1.27, s
C-29	17.51, CH <sub>3</sub>	0.98, s	17.49, CH <sub>3</sub>	0.94, s	17.51, CH <sub>3</sub>	0.94, s
C-30	17.30, CH <sub>3</sub>	0.95, s	17.16, CH <sub>3</sub>	0.98, s	17.16, CH <sub>3</sub>	0.98, s
1'	107.46, CH	4.97, d (7.5)	107.18, CH	4.89, d (7.5)	107.24, CH	4.90, d (7.7)
2'	76.28, CH	4.06, m	73.81 CH	4.07, m	73.87, CH	4.08, m
3'	79.25, CH	4.25, m	79.43, CH	5.80, m	79.41, CH	5.86, m
4′	72.34, CH	4.23, m	70.21, CH	4.12, m	70.33, CH	4.14, m
5'	78.88, CH	4.03, m	75.32, CH	4.10, m	75.37, CH	4.10, m
6'	63.55, CH <sub>2</sub>	4.61, dd (7.5, 2.5)	64.56, CH <sub>2</sub>	4.99, d (11.4)	64.59, CH <sub>2</sub>	5.04, d (11.5)
		4.42, dd (11.5, 5.5)	_	4.79, d (6.4)	_	4.83, dd (11.6, 5.1)
1″			173.85, C		174.11, C	
2″			173.78, C	2 22 (77)	174.04, C	2.40.4(7.4)
2"			$35.00, CH_2$	2.33, t (7.7)	35.30 CH <sub>2</sub>	2.40, t (7.4)
2″			34.00, $CH_2$	1.50	34.95 CH <sub>2</sub>	1.67
3			25.20, $CH_2$	1.59, 111	25.80 CH	1.07, 111
4″			$25.13, CH_2$	1.29 m	30.19 CH.	1.27 m
т			26.78 CH	1.27, 111	29.97 CH	1.27, 111
5″			28.96. CH	1.51, m	29.87. CH	1.18–1.25. m
-			28.90, CH <sub>2</sub>		29.84, CH <sub>2</sub>	
6″			38.06, CH <sub>2</sub>	3.50, s	30.14, CH <sub>2</sub>	1.18–1.25, m
			38.02, CH <sub>2</sub>		30.10, CH <sub>2</sub>	,
7″			171.75, C		30.02, CH <sub>2</sub>	1.18–1.25, m

#### Table 1. continued

	Rh <sub>2</sub>		C-Rh <sub>2</sub>		U-Rh <sub>2</sub>	
position	$\delta_{\rm C'}$ type	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{\mathrm{C'}}$ type	$\delta_{ m H}~(J~{ m in}~{ m Hz})$	$\delta_{\mathrm{C}'}$ type	$\delta_{\rm H^{\prime}}$ type
			171.72, C		30.00, CH <sub>2</sub>	
8″			134.87, CH	6.82, s	29.76, CH <sub>2</sub>	1.18–1.25, m
			134.83, CH			
9″			134.87, CH	6.82, s	27.49, CH <sub>2</sub>	1.18, m
			134.83, CH		27.43, CH <sub>2</sub>	
10″			171.75, C		29.34, CH <sub>2</sub>	1.57, m
			171.72, C		29.30, CH <sub>2</sub>	
11''					38.35, CH <sub>2</sub>	3.54, t (6.8)
					38.33, CH <sub>2</sub>	
12″					171.83, C	
13″					134.90, CH	6.87, s
14″					134.90, CH	6.87, s
15″					171.83, C	



Figure 2. Selected HMBC and COSY correlations of C-Rh<sub>2</sub> (A) and U-Rh<sub>2</sub> (B).

Furthermore, the cross-peak signal of H-3' ( $\delta_{\rm H}$  5.80) and C-1" ( $\delta_{\rm C}$  173.78) of the ester carbonyl in the maleimide caproyl group in the HMBC spectrum indicated that the second 6-maleimidocaproic acid molecule was attached to C-3' in the glucose ring (Figure 2).

According to the attribution of carbon and proton signals in 2D NMR discussed above, 6-maleimidocaproic acid was connected to C-3' and C-6' of the glucose moiety of Rh<sub>2</sub>; thus, C–Rh<sub>2</sub> finally elucidated as 3',6'-dimaleimide caproyl ginsenoside Rh<sub>2</sub>. The introduction of acyl groups onto the glucose moiety caused a significant downfield shift of the proton attached to acyl-linked carbon; therefore, the proton signals of H-3' and H-6' shifted about 1.55 and 0.38 ppm, respectively. Meanwhile, the carbon signals for C-2', C-4', and C-5' adjacent to the reaction site upfield shifted to  $\delta_{\rm C}$  73.81, 70.21, and 75.32, respectively.

U–Rh<sub>2</sub> was obtained as an amorphous solid and assigned the molecular formula  $C_{66}H_{104}N_2O_{14}$  based on HREISMS ([M + H]<sup>+</sup> at m/z 1149.7555, calcd 1149.7560, Figure S19, Supporting Information). This molecular formula inferred two equivalent maleimide undecanoyl groups in U–Rh<sub>2</sub>. The FTIR absorption bands at 3100–3500, 1707, and 1629 cm<sup>-1</sup> indicated the presence of the hydroxyl group, carbonyl group, and carbon–carbon double bond. The disappearance of  $\gamma_{O-H}$  in carboxy (950 cm<sup>-1</sup>) confirmed the conversion of carboxyl to ester group after 11-maleimidoundecanoic acid was linked to Rh<sub>2</sub>.

The NMR spectroscopic data of U–Rh<sub>2</sub> resembled those of Rh<sub>2</sub> and C–Rh<sub>2</sub>, except for new signals associated with 11maleimidoundecanoic acid. The <sup>1</sup>H NMR spectrum (Figure S20, Supporting Information) exhibited signals at  $\delta_{\rm H}$  6.87 (H-13" and H-14"), 3.54 (H-11"), 2.40 (H-2"), 1.67 (H-3"), 1.57 (H-10"), 1.27 (H-4"), and 1.18 (H-9") due to maleimide



**Figure 3.** Inhibition of the proliferation of A549, MCF-7, and HeLa cells by Rh<sub>2</sub>, C–Rh<sub>2</sub>, and U–Rh<sub>2</sub> with different concentrations (5, 10, 20, 40, 60, 80, and 100  $\mu$ M) assessed using the MTT assay. All data are presented as means ± SD of three independent experiments. \**p* < 0.05, \*\**p* < 0.01, and \*\*\**p* < 0.001 vs Rh<sub>2</sub> group.

undecanoyl groups. In the <sup>13</sup>C NMR spectrum (Figure S21, Supporting Information), two sets of very close signals of  $\delta_{\rm C}$ 174.11, 35.30, 25.86, 30.19, 29.87, 30.14, 30.02, 29.76, 27.49, 29.34, 38.35, 171.83, and 134.90 (C-1″–13″) and  $\delta_{\rm C}$  174.04, 34.95, 25.80, 29.97, 29.84, 30.10, 30.00, 29.76, 27.43, 29.30, 38.33, 171.83, and 134.90 (another group of C-1″–13″) were observed, suggesting that two 11-maleimidoundecanoic acid molecules were grafted onto Rh<sub>2</sub>.

Similarly, in the <sup>1</sup>H NMR spectrum of U-Rh<sub>2</sub>, the proton signals downfield shifts of  $\delta_{\rm H}$  5.86 (H-3') and 5.04/4.83 (H- $6'_{a}/6'_{b}$ ) were observed. While in the <sup>13</sup>C NMR spectrum, the carbon signals at  $\delta_{\rm C}$  73.87 (C-2'), 70.33 (C-4'), and 75.37 (C-5') were shifted upfield compared with  $\delta_{\rm C}$  76.28, 72.34 and 78.88 in Rh<sub>2</sub>. These indicated that 11-maleimidoundecanoic acid molecules were also connected to C-3' and C-6' of Rh<sub>2</sub>. Further, this conclusion could be confirmed by the analysis of HSQC, <sup>1</sup>H-<sup>1</sup>H COSY, and HMBC spectra (Figures S22-24, Supporting Information). In brief, the correlation signals between H-6'<sub>a</sub>/6'<sub>b</sub> ( $\delta_{\rm H}$  5.04/4.83) on the glucose moiety and C-1" ( $\delta_{\rm C}$  174.04) in the maleimide undecanoyl group, as well as the signal between H-3' ( $\delta_{\rm H}$  5.86) and C-1" ( $\delta_{\rm C}$  174.11) in the HMBC spectrum, also provided evidence that U-Rh<sub>2</sub> could be elucidated as 3',6'-dimaleimide undecanoyl ginsenoside Rh<sub>2</sub>.

To evaluate the solubility property of Rh<sub>2</sub> modified with maleimide fatty acids, the oil-water partition coefficient and solubility were measured. The results showed that the values of oil-water partition coefficient log P were 3.48 for Rh<sub>2</sub>, 2.27 for C-Rh<sub>2</sub>, and 2.84 for U-Rh<sub>2</sub>. The solubility of Rh<sub>2</sub>, C-Rh<sub>2</sub>, and U-Rh<sub>2</sub> in water was 74.04, 292.06, and 162.69  $\mu$ g·mL<sup>-1</sup>, respectively. That is, the solubility of C-Rh<sub>2</sub> and U-Rh<sub>2</sub> increased to approximately 4 and 2 times, respectively, compared with Rh<sub>2</sub>. These indicate that the derivatives modified with maleimide fatty acids exhibit better solubility property compared with Rh<sub>2</sub> and are expected to show better bio-absorption performance. The stability test showed that there was no peak shift or other changes in the <sup>1</sup>H NMR spectra within 144 h in an alkaline medium, indicating that C-Rh<sub>2</sub> and U-Rh<sub>2</sub> were relatively stable (Figure S25, Supporting Information).

In Vitro Antiproliferative Activity of the  $Rh_2$  Derivatives. To investigate the in vitro cytostatic activity of  $Rh_2$ derivatives, A549, HeLa, and MCF-7 cells were treated with  $Rh_2$ , C-Rh<sub>2</sub>, and U-Rh<sub>2</sub> at different concentrations (Figure 3). The MTT assay showed that the IC<sub>50</sub> values for Rh<sub>2</sub> on A549, MCF-7, and HeLa cells were 85.26, 73.58, and 67.95  $\mu$ M, respectively, while for C-Rh<sub>2</sub>, the IC<sub>50</sub> values were 54.82, 29.80, and 14.17  $\mu$ M. In other words, the cytotoxicity of C-Rh<sub>2</sub> against A549, MCF-7, and HeLa cell lines was approximately 1.6, 2.5, and 4.8 times that of Rh<sub>2</sub>, respectively. It can also be seen from Figure 3 that C-Rh<sub>2</sub> had the highest cytostatic activity against the HeLa cell line, which was also proved by the live/dead cell staining assay (Figure S26, Supporting Information). However, U-Rh<sub>2</sub> exhibited only slight cytotoxicity to the HeLa cell line and no cytotoxicity to the other two cells. The structural difference between  $C-Rh_2$ and U-Rh<sub>2</sub> is the number of carbon atoms on the acyl side chain, so it could be inferred that the length of the spacer arm might have a significant effect on the antiproliferative activity of Rh<sub>2</sub> derivatives. Further research would be carried out to illustrate this issue. Since C-Rh<sub>2</sub> showed better cytotoxicity to tumor cells, it would be discussed in more detail hereinafter.

The Annexin V/PI apoptosis assay by flow cytometry showed that the percentage of early and late apoptosis in the control group and Rh<sub>2</sub>-treated group was 6.2 and 5.3%, respectively, while it increased to 36.4% for the C-Rh<sub>2</sub> treated group (Figure 4A). These results suggested that C-Rh<sub>2</sub> demonstrated greater effectiveness in inducing apoptosis than Rh<sub>2</sub> in HeLa cells.

The JC-1 dye was used to examine the change of mitochondrial membrane potential. The JC-1 dye enters and accumulates in the energized and negatively charged mitochondria and spontaneously forms red fluorescent J-aggregates in healthy cells but retains its original green fluorescence in unhealthy or apoptotic cells.<sup>26</sup> Compared with the control group, the reduction of red fluorescence and the increase of green fluorescence in the C–Rh<sub>2</sub>-treated group were more remarkable than those in the Rh<sub>2</sub>-treated group (Figure 4B). This indicated that C–Rh<sub>2</sub> treatment induced severer damage to mitochondria than Rh<sub>2</sub> in HeLa cells.

The increase of intracellular ROS can result in mitochondrial damage and play a significant role in pro-apoptotic activities.<sup>27</sup> To evaluate whether  $C-Rh_2$  induced ROS accumulation in HeLa cells, DCFH-DA staining was conducted after treatment with  $Rh_2$  and  $C-Rh_2$ . It was found that the  $C-Rh_2$  treated group generated stronger DCF fluorescence intensity than the  $Rh_2$ -treated group at the same concentration (Figure 4C). Therefore,  $C-Rh_2$  administration could produce more ROS than  $Rh_2$  administration, suggesting that the increased ROS might be a critical reason for  $C-Rh_2$  to inhibit HeLa cell growth.



**Figure 4.** Analysis of the effects of C–Rh<sub>2</sub>-induced HeLa cell apoptosis. (A) HeLa cells were treated with Rh<sub>2</sub> ( $20 \mu$ M) and C–Rh<sub>2</sub> ( $20 \mu$ M) for 24 h. Then, the cells were stained with annexin V-FITC and propidium iodide (PI) and finally analyzed by flow cytometry. (B) Cultures were treated with compounds aforementioned for 24 h and stained subsequently with the JC-1 dye. (C) The cells were treated with compounds for 24 h and then stained with a fluorescent probe DCFH-DA. Images were observed under an inverted fluorescent microscope (scale bar = 50  $\mu$ m).

In conclusion, two derivatives of  $Rh_2$  with better solubility were prepared by esterification of maleimide fatty acid with different lengths of spacer arms. The results of in vitro antiproliferative activity test showed that  $C-Rh_2$  exhibited higher cytotoxicity against cancer cell lines compared with  $Rh_2$ , especially for HeLa cells, whereas  $U-Rh_2$  with longer carbon chains showed little cytotoxicity. These results suggested that the spacer arm length may have an important effect on the antiproliferative activity of  $Rh_2$  derivatives.

#### EXPERIMENTAL SECTION

**Materials.** Ginsenoside  $Rh_2$  ( $Rh_2$ ) was supplied by Nanjing Dilger Medical Technology Co., Ltd (Nanjing, China). 6-Maleimidocaproic acid and 11-maleimidoundecanoic acid were obtained from Macklin (Shanghai, China). 4-Dimethylaminopyridine (DMAP) was purchased from Aladdin (Shanghai, China).

General Experimental Procedures. FTIR spectra were recorded with a Nicolet 6700 FTIR spectrometer (Thermo, America). The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker Avance III 400 NMR spectrometer in pyridine-d<sub>5</sub> with tetramethylsilane (TMS) as a reference. Chemical shifts ( $\delta$ ) are expressed in parts per million (ppm), with the coupling constants (J) reported in hertz (Hz). Ultraviolet-visible (UVvis) absorption spectra were recorded using a UV-5500PC spectrophotometer (Metash, China). High-performance liquid chromatography (HPLC) measurements were carried out on a Shimadzu LC-20AT pump with a SIL-20A autosampler using a Kromasil 100-5-C18 column (250 mm × 4.6 mm i.d.). Detection was executed by a SPD-M20A photodiode array detector. HRMS data were obtained on an Orbitrap Fusion instrument in the ESI mode. Silica gel GF254 plates for thinlayer chromatography and silica gel (200-300 mesh) for column chromatography were produced by Qingdao Marine Chemical Factory.

Chemical Procedures for the Preparation of C–Rh<sub>2</sub> and U–Rh<sub>2</sub>. In order to prepare the derivatives of Rh<sub>2</sub>, 6maleimidocaproic chloride and 11-maleimidoundecanoyl chloride were first synthesized by the reaction of 6maleimidocaproic acid and 11-maleimidoundecanoic acid with SOCl<sub>2</sub> at 50 °C and refluxed for 3 h, respectively. Then, the obtained 6-maleimidocaproic chloride or 11maleimidoundecanoyl chloride was reacted with Rh<sub>2</sub> in the presence of triethylamine (TEA) as an acid binding agent and 4-dimethyl aminopyridine (DMAP) as the catalyst, as illustrated in Scheme 1.

3',6'-Dimaleimide Caproyl Ginsenoside Rh<sub>2</sub> (C-Rh<sub>2</sub>). Rh<sub>2</sub> (168.1 mg, 0.27 mmol) and TEA (185  $\mu$ L, 1.2 mmol) were dispersed in dry  $CH_2Cl_2$  (4 mL); then, 6-maleimidocaproic chloride (229.1 mg, 1.00 mmol, 3.70 equiv) was added dropwise. The reaction mixture was stirred at 0 °C for 30 min; subsequently, DMAP (1.8 mg, 0.015 mmol, 0.05 equiv) was added. After 6 h, the mixture was diluted by adding CH<sub>2</sub>Cl<sub>2</sub> (15 mL), and then a dilute hydrochloric acid solution (pH = 2,  $2 \times 20$  mL) was added. After that, the mixture was layered with a separating funnel, and the water layer was discarded. The organic layer was washed successively with saturated aqueous NaHCO<sub>3</sub> ( $2 \times 20$  mL) and saturated aqueous NaCl  $(2 \times 20 \text{ mL})$  and then dried by anhydrous Na<sub>2</sub>SO<sub>4</sub> and filtered through a funnel. Finally, the filtered solution was concentrated under vacuum and then purified by column chromatography (30 to 60% EtOAc in petroleum ether) to yield  $C-Rh_2$  as a white solid powder (107.9 mg, 0.11 mmol, 39.6%):  $R_f = 0.52$ 

(MeOH/CH<sub>2</sub>Cl<sub>2</sub> 1:17 v/v); IR (KBr)  $\nu_{max}$  3100–3500, 2934, 1706, 1628, 1402, 696 cm<sup>-1</sup>; <sup>1</sup>H NMR and <sup>13</sup>C NMR data, see Table 1; HRESIMS *m*/*z*: 1009.5989 [M + H]<sup>+</sup> (calcd for C<sub>56</sub>H<sub>84</sub>N<sub>2</sub>O<sub>14</sub>, 1009.5995).

3',6'-Dimaleimide Undecanoyl Ginsenoside Rh<sub>2</sub> (U-Rh<sub>2</sub>). 11-Maleimidoundecanoyl chloride (299.2 mg, 1 mmol, 3.70 equiv) was added dropwise to an anhydrous  $CH_2Cl_2$  (4 mL) solution of Rh<sub>2</sub> (168.1 mg, 0.27 mmol) and TEA (231  $\mu$ L, 1.5 mmol) with magnetic stirring, and then DMAP (1.8 mg, 0.015 mmol, 0.05 equiv) was added after 30 min. The reaction mixture was stirred at 0 °C to room temperature for 6 h before being diluted with CH<sub>2</sub>Cl<sub>2</sub> (15 mL). Next, dilute hydrochloric acid solution (20 mL) was added, resulting in phase separation. The collected organic phase was washed with saturated aqueous NaHCO<sub>3</sub> (20 mL) and saturated aqueous NaCl (20 mL) in turn. Following this, the organic phase was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and then evaporated under reduced pressure after filtering. Purification of the residue by column chromatography (20 to 50% EtOAc in petroleum ether) afforded U-Rh<sub>2</sub> as a pale yellow amorphous solid:  $R_f = 0.55$ (MeOH/CH<sub>2</sub>Cl<sub>2</sub> 1:17 v/v); IR (KRr)  $\nu_{max}$  3100–3500, 2929, 2856, 1707, 1629, 1402, 696 cm  $^{-1};\ ^1H$  NMR and  $^{13}C$  NMR data, see Table 1; HRESIMS m/z: 1149.7555 [M + H]<sup>+</sup> (calcd for  $C_{66}H_{104}N_2O_{14}$ , 1149.7560).

Determination of the Oil-Water Partition Coefficient. The oil-water partition coefficient was determined by the shake flask method. Rh<sub>2</sub>, C-Rh<sub>2</sub>, and U-Rh<sub>2</sub> were added into the pre-prepared water-saturated n-octanol solution, respectively, and dissolved ultrasonically. 1 mL of the solution was taken out and centrifuged at 10,000 rpm for 10 min. Then, the supernatant was diluted to measure its absorbance at 217 nm, and the concentration in the oil phase  $(C_0)$  was calculated using the standard curve method. Another 1 mL of the solution was added into 1 mL of water solution saturated with noctanol. The mixture was oscillated at 37 °C for 24 h (100 rpm) to achieve dissolution equilibrium in the two phases. Subsequently, the mixture was centrifuged at 10,000 rpm for 10 min, and then the absorbance of the water phase (the lower layer) was measured, and the concentration in the water phase  $(C_w)$  was calculated. The oil-water partition coefficient was calculated according to the formula

$$P = (C_0 - C_w)/C_w$$

**Solubility Measurement.** Excess  $Rh_2$ ,  $C-Rh_2$ , and  $U-Rh_2$  were added to 2 mL of distilled water, respectively, and oscillated for 24 h at 37 °C (100 rpm). The mixture was centrifuged at 10,000 rpm for 10 min, then the supernatant was taken, and the absorbance was measured at 217 nm. The solubility was calculated using the standard curve method.

**Stability Test.**  $C-Rh_2$  and  $U-Rh_2$  were dissolved, respectively, in pyridine- $d_5$  (0.5 mL). <sup>1</sup>H NMR scans were conducted at 0, 12, 24, 48, 72, and 144 h.

**Cell Culture.** All cells were cultured in DMEM (Gibco Life Technology Co., Ltd., USA) containing 10% (v/v) fetal bovine serum (Procell Life Science & Technology Co., Ltd.) and 1% (w/v) penicillin–streptomycin (Gibco Life Technology Co., Ltd.) at 37 °C in a humidified chamber supplemented with 5%  $CO_2$ . The culture media were replaced every 48 h, and the cultured cells were digested with 0.25% trypsin-EDTA solution when reached 80–90% confluency.

**Cell Proliferation Activity Assay.** For the evaluation of the cytostatic activities of  $Rh_2$ ,  $C-Rh_2$ , and  $U-Rh_2$ , the cells were seeded at a density of  $5 \times 10^3$  cells/well in 96-well flat-

bottomed plates. After incubation for 24 h, the cells were treated with the test compounds mentioned above (final concentration 5–100  $\mu$ M) and simultaneously treated with 1% DMSO as a control group. Thereafter, the effect of the test compounds on the proliferation of tumor cells was assessed according to the 3-(4,5-dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT) method. The results were obtained on a multi-detection microplate reader (Tecan, Infinite M1000 PRO) using a wavelength of 490 nm. Data were presented as cell viability, and IC<sub>50</sub> values were calculated using GraphPad Prism software (GraphPad Software, CA). In each MTT assay, every sample was tested in three to five replicates, and percent cell survival was calculated using the following equation

Cell viability (%) = absorbance of the experimental

group/absorbance of the control group

#### × 100%

**Flow Cytometry Analysis of Apoptosis.** To evaluate the effect of C–Rh<sub>2</sub> on cell apoptosis, we performed a flow cytometry analysis. Briefly, HeLa cells were placed at a density of  $5 \times 10^5$  cells/well in six-well plates with 3 mL of DMEM containing 10% (v/v) fetal bovine serum and 1% (w/v) penicillin–streptomycin and then incubated for 24 h. Next, the supernatants were replaced by fresh culture medium with Rh<sub>2</sub> (20  $\mu$ M) and C–Rh<sub>2</sub> (20  $\mu$ M) and incubated for 24 h. The cells were harvested and resuspended in a binding buffer after treatment. Annexin V-FITC and PI (Solarbio Science Co., Ltd.) were then added according to the manufacturer's instructions. Finally, the cell apoptosis rate was analyzed by a flow cytometer (FACSAria, Becton Dickinson, USA).

Mitochondrial Membrane Potential Assessment by JC-1 Staining. HeLa cells were seeded in six-well plates and treated with compounds described previously for 24 h. Then, the cells were stained with JC-1 according to the manufacturer's protocol. Last, the stained cells were captured using an inverted fluorescence microscope (Axio Observer A1, Germany).

**Reactive Oxygen Species Measurement by DCFH-DA Staining.** The cell culture and treatment processes were the same as described above. After that, the cells were stained with a ROS assay kit (containing DCFH-DA as the indicator) following the manufacturer's instructions. The green fluorescence was observed under a blue light with a fluorescence microscope.

**Statistical Analysis.** The results were shown as the mean  $\pm$  standard deviation (SD) of at least three independently performed experimental measurements to avoid possible variation in cell cultures. Student's *t*-test was used, and p < 0.05 was considered to be statistically significant. Statistical significances were indicated with asterisks (\*p < 0.05, \*\*p < 0.01, and \*\*\*p < 0.001).

#### ASSOCIATED CONTENT

#### **Supporting Information**

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.3c01665.

NMR, HRMS, and FTIR spectra for compounds (PDF)

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The authors declare no competing financial interest.

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