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Design and synthesis of multi-target directed 1,2,3-triazole-dimethylaminoacryloyl-chromenone derivatives with potential use in Alzheimer's disease

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Abstract

To discover multifunctional agents for the treatment of Alzheimer's disease (AD), a new series of 1,2,3-triazole-chromenone derivatives were designed and synthesized based on the multi target-directed ligands approach. The in vitro biological activities included acetylcholinesterase (AChE) and butyrylcholinesterase (BuChE) inhibition as well as anti-A β aggregation, neuroprotective effects, and metal-chelating properties. The results indicated a highly selective BuChE inhibitory activity with an IC₅₀ value of 21.71 μ M for compound **10h** as the most potent compound. Besides, compound **10h** could inhibit self-induced A β ₁₋₄₂ aggregation and AChE-induced A β aggregation with 32.6% and 29.4% inhibition values, respectively. The Lineweaver–Burk plot and molecular modeling study showed that compound **10h** targeted both the catalytic active site (CAS) and peripheral anionic site (PAS) of BuChE. It should be noted that compound **10h** was able to chelate biometals. Thus, the designed scaffold could be considered as multifunctional agents in AD drug discovery developments.

Keywords: A β aggregation, Biometal chelator, BuChE inhibitor, Chromenones, Neuroprotectivity, 1,2,3-triazole

Introduction

Dementia is one of the noteworthy problems in the public health management as over 80% of dementia cases are suffering from Alzheimer's disease (AD). Currently, available therapies provide temporary symptomatic relief but do not target the distractive neuropathology. Therefore, a new treatment to delay or halt disease progression has remained as an urgent medical need.

The pathophysiological processes in AD have still remained unclear to this day. However, alongside its complexity, several neurodegenerative processes could

be identified which include (I) aggregation of insoluble amyloid beta (A β) plaques mostly trigger from sequential cleavage of amyloid precursor protein (APP) by the aspartyl protease β -site APP cleaving enzyme-1 (BACE1) and γ -secretase, (II) neurofibrillary tangles (NFTs) form through hyperphosphorylation of tau proteins, (III) biometals dysfunction, and (IV) oxidative stress which in return results in synapse loss and death of neuronal cells in the brain [1]. Also, different hallmarks have been recognized including the loss of cholinergic neurons, reduction of the neurotransmitter acetylcholine (ACh), and increased expression of inflammatory mediators [2–4].

Based on the approved theory for AD, the loss of cholinergic neurons causes reduction of ACh. As a result, inhibition of the acetylcholinesterase (AChE) raises the level of ACh and improves cognitive performance

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at the early stage of AD. The critical point is that the AChE level decreases with the progression of AD, subsequently, AChE inhibition seems to be ineffective during the progression of AD [5, 6]. Interestingly, the level of butyrylcholinesterase (BuChE) remains unchanged or even increases at the late stage of disease [7]. BuChE can hydrolyze ACh and thereby, compensates the reduction of AChE activity [8]. A recent experiment with AChE knockout mice supported this hypothesis [9]. Results from further studies were in accordance with the role of BuChE in AD brains and showed a positive correlation between selective BuChE inhibition and improved cognitive performance and memory [10, 11].

BuChE has 65% amino acid sequence similarity to AChE along with mostly similar functions [12–14]. AChE and BuChE active sites have five subsites, including catalytic active site (CAS), peripheral anionic site (PAS), acyl binding pocket, oxyanion hole, and anionic subsite [15]. One of the structural differences between AChE and BuChE can be associated with acyl pocket size. More specifically, smaller residue such as Leu286 and Val288 of BuChE acyl pocket provide a larger site in BuChE while aromatic and bulky Phe295 and Phe297 residues of AChE acyl pocket afford smaller space in AChE. Mentioned structural differences contribute to the design of selective inhibitors [5, 8, 16].

Moreover, produced A β peptides can aggregate into A β plaques which initiate pathogenic cascade leading to neuronal loss and dementia. Inhibition of the accumulation of A β peptide in the brain could be another therapeutic strategy against the development of AD [2]. The metal chelatory potential of compounds has also been demonstrated to exert beneficial effects via decreasing plaque aggregation [17, 18].

Results and discussion

Design

Because of the multifactorial and sporadic nature of AD, the modern approach “multi-target-one disease” could be efficient to develop effective agents to act simultaneously at different targets. Selective BuChE inhibitors could be a promising target for the treatment of AD at the moderate and advanced stages of the disease [19]. Closer looks at X-ray crystallography of BuChE depicted that it usually tolerates bigger scaffolds than AChE as the active site of BuChE is approximately 200 Å larger than AChE. Analysis of the potent BuChE inhibitors revealed that the N-containing ring could be effective for the interactions with ChE active site (Fig. 1). Molecular docking evaluation of compound A depicted that coumarin moiety interacted with Trp231 and Phe329 residues of CAS pocket and benzyl pyridinium moiety interacted with Trp82 of BuChE PAS pocket [20]. According to the

interaction mode of compound B, it can be understood that 1,2,3-triazole-aryl moiety led to the formation of hydrophobic interactions with amino acids of PAS and chromenone ring oriented towards CAS pocket [21]. In the case of compound C, diphenyl fit into the BuChE CAS pocket while the benzyl triazine pendant group showed H-bond interaction with the PAS of BuChE [22, 23]. As appeared in compounds B and C, increasing bulkiness and length of drug candidate could increase the selectivity of BuChE over AChE. As can be seen in compound D, the presence of relatively spacious moieties can affect the interaction of the designed compounds with that amino acid within the active site of BuChE. As a result, introducing a dimethylamino propenone entity into our system increased the length and volume of the designed scaffold which might be a good strategy to increase selectivity towards BuChE [24]. Besides, this part could also exhibit metal-chelating potential [25].

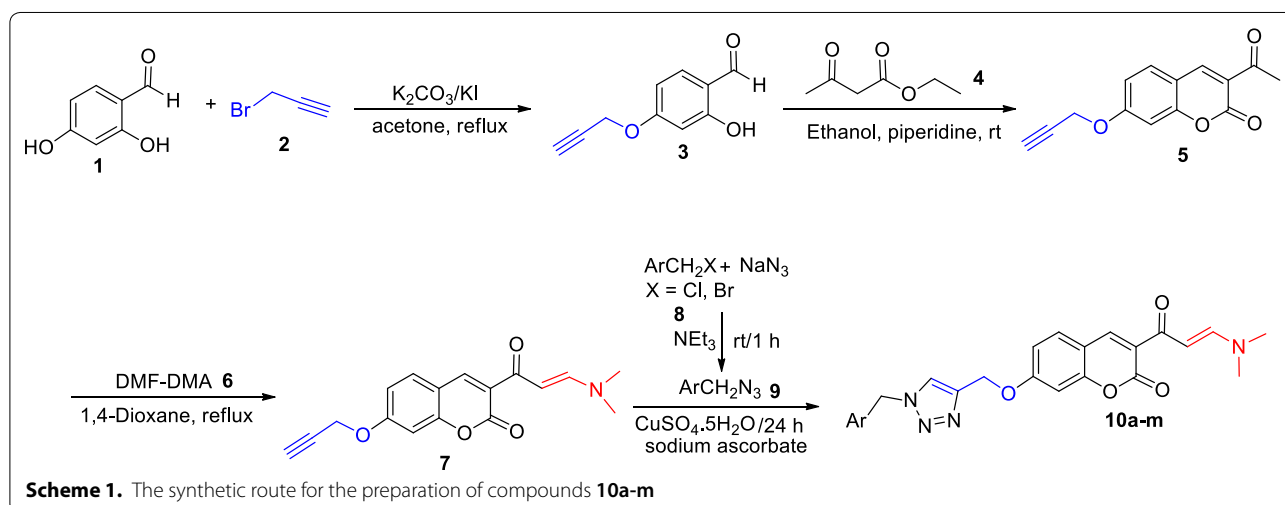
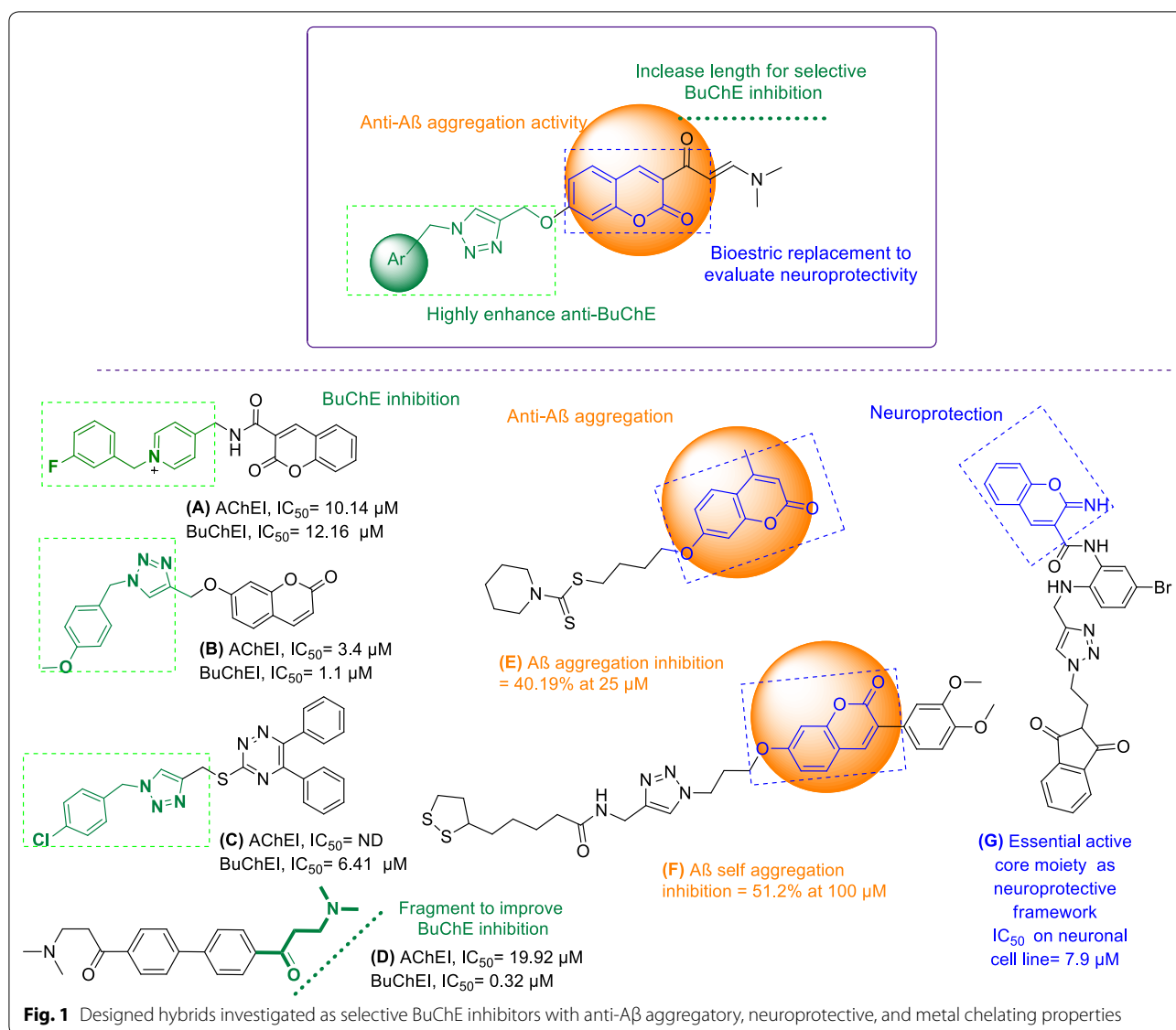
In the case of anti-A β plaques aggregation, it is important to keep the potential moiety in the structure to inhibit the aggregation of the toxic peptide.

Coumarin structures as active natural compounds may simultaneously possess anti-oxidative [2, 26], neuroprotective [27] anti-ChE [20] and anti-A β aggregative properties [28, 29]. Coumarins pharmacophore owing to the presence of polar elements in the structure (Fig. 1, compound E and F) might help to inhibit amyloid fibril formation through the interaction with the polar surface of A β [30–32]. Hence, coumarins could serve as a potent framework for the prevention of A β _{1–42} aggregation [28]. In addition to the inhibition of BuChE and A β plaques aggregation, an inhibitor that can tackle toxicity of A β peptide, ROS and RNS could be effective for a longer period of AD progression. Recently, iminochromene ring was characterized as a potent neuroprotective agent. In this project, the iminochromene group of compound G was bioisosterically replaced with chromenone moiety to evaluate the possible neuroprotectivity [33]. Hence, in the present work, a molecular hybridization and bioisosteric replacement approaches were used to design and synthesize multi-target agents with anti-BuChE, anti-A β aggregation, neuroprotective, and metal chelating properties.

Synthesis

of 1,2,3-triazole-dimethylaminoacryloyl-chromenones

Synthesis of the tilted compounds 10a–m was conducted according to the steps shown in Scheme 1. Desired starting material, 2-hydroxy-4-(prop-2-yn-1-yloxy)benzaldehyde (3) was exactly prepared according to the literature [34]. Then, the reaction of compound 3 and excess amounts of ethyl acetoacetate (4) in ethanol at room temperature overnight afforded



3-acetyl-7-(prop-2-yn-1-yloxy)-2*H*-chromen-2-one (5). Reaction of compound 5 and dimethylformamid-dimethylacetal (DMA-DMF) in 1,4-dioxane under reflux conditions for 6 h led to the formation of (*E*)-3-(3-(dimethylamino)acryloyl)-7-(prop-2-yn-1-yloxy)-2*H*-chromen-2-one (7). Finally, click reaction [35] of compound 7 and in situ prepared azides 9 in the presence of triethylamine and CuSO₄·5H₂O in the mixture of water and *tert*-butyl alcohol at room temperature for 24 h gave the corresponding products 10a-m.

AChE and BuChE inhibitory activity

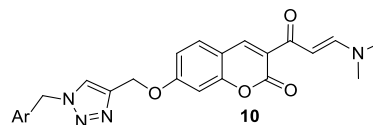
In vitro anti-AChE and anti-BuChE activity of synthesized compounds, 10a-m were performed based on the modified Ellman's method [36] comparing with donepezil as the reference drug. Compounds were initially screened in vitro against AChE, and none of them exhibited inhibitory properties against the AChE enzyme. Interestingly, half of the 1,2,3-triazole-chromenone derivatives showed remarkable and selective inhibitory potency towards BuChE, which exerted a more prominent role at later stages of the disease [37].

As can be seen in Table 1, the BuChEI activity directly depended on the electronic property of substituents and their positions on the benzyl moiety connected to 1,2,3-triazole ring. Results showed that compound 10h possessing 3,4-diF on the aryl ring induced the best BuChE inhibitory activity (IC₅₀ = 21.71 μM); however, the elimination of 3-F completely changed the activity in such a manner that compound 10g showed no inhibitory activity. *meta*-Fluorinated derivative 10f was found to be a moderate inhibitor as the calculated IC₅₀ value was 59.58 μM and *para*-fluorinated derivative 10e showed no inhibitory activity towards BuChE (IC₅₀ > 100 μM).

Considering the inhibitory activity of other halogenated derivatives 10i-m depicted that chlorinated compounds 10i and 10j showed no inhibitory activity (IC₅₀ > 100 μM). In the case of brominated derivatives 10k-m, compounds 10l possessing Br at 3-position of aryl ring showed moderate activity with IC₅₀ value of 65.96 μM.

In the case of the electron-donating substituent (Me and OMe), relatively good results were obtained. The presence of methyl group at 2-position of aryl moiety (compound 10b) led to relatively good anti-BuChE activity (IC₅₀ = 35.73 μM); however, the presence of the same group at 4-position of compound 10c completely diminished inhibitory activity (IC₅₀ > 100 μM). Another point comes back to the derivative 10d containing OMe group at 3-position on the benzyl ring which led to higher activity (IC₅₀ = 23.44 μM). Finally, the absence of substituent on the aryl ring (compound 10a) also depicted good BuChE inhibitory activity (IC₅₀ = 34.41 μM).

Table 1 Anti-cholinesterase activity of 1,2,3-triazole-



dimethylaminoacryloyl-chromenone 10a-m

Entry	Ar	Product 10	AChEI IC ₅₀ (μM)	BuChEI IC ₅₀ (μM)
1	C ₆ H ₅	10a	>100	34.41 ± 0.23
2	2-Me-C ₆ H ₄	10b	>100	35.73 ± 0.21
3	4-Me-C ₆ H ₄	10c	>100	>100
4	3-MeO-C ₆ H ₄	10d	>100	23.44 ± 0.07
5	2-F-C ₆ H ₄	10e	>100	>100
6	3-F-C ₆ H ₄	10f	>100	59.58 ± 0.05
7	4-F-C ₆ H ₄	10g	>100	>100
8	3,4-diF-C ₆ H ₃	10h	>100	21.71 ± 0.57
9	2-Cl-C ₆ H ₄	10i	>100	>100
10	4-Cl-C ₆ H ₄	10j	>100	>100
11	2-Br-C ₆ H ₄	10k	>100	>100
12	3-Br-C ₆ H ₄	10l	>100	65.96 ± 0.004
13	4-Br-C ₆ H ₄	10m	>100	>100
14	donepezil		0.079 ± 0.002	5.19 ± 0.38

Data are expressed as mean ± SD (three independent experiments)

The in vitro anti-ChE assay showed that the un-substituted benzyl derivative (10a) along with the *meta*-substituted analogs (10d, 10f, and 10l) had significant anti-BuChE activity. Introduction of extra small-size halogen groups such as F (compound 10h) resulted in the most potent activity with an IC₅₀ value of 21.71 μM.

Kinetic study of BuChE inhibition

The kinetic study was performed to investigate the mechanism of inhibition by compound 10h against BuChE. Graphical analysis of the reciprocal Lineweaver-Burk plot of compound 10h described a mixed-type inhibition pattern (Fig. 2a) in which compound 10h may bound to the BuChE or it already bound to the substrate. Also, the *K_i* value was calculated using the secondary plot as 38.3 μM (Fig. 2b)

Inhibition of AChE-induced and self-induced Aβ aggregation

Aβ peptide is the major constituent of senile plaques in the brains of patients with AD. In this respect, the effect of the most potent compound 10h was assessed for the inhibition against Aβ₁₋₄₂ aggregation and AChE-induced Aβ₁₋₄₀ peptide aggregation using the Thioflavin T (ThT) assay. Comparing with donepezil and tacrine as the reference compounds, it can be understood that 10h was more potent than both controls in inhibiting

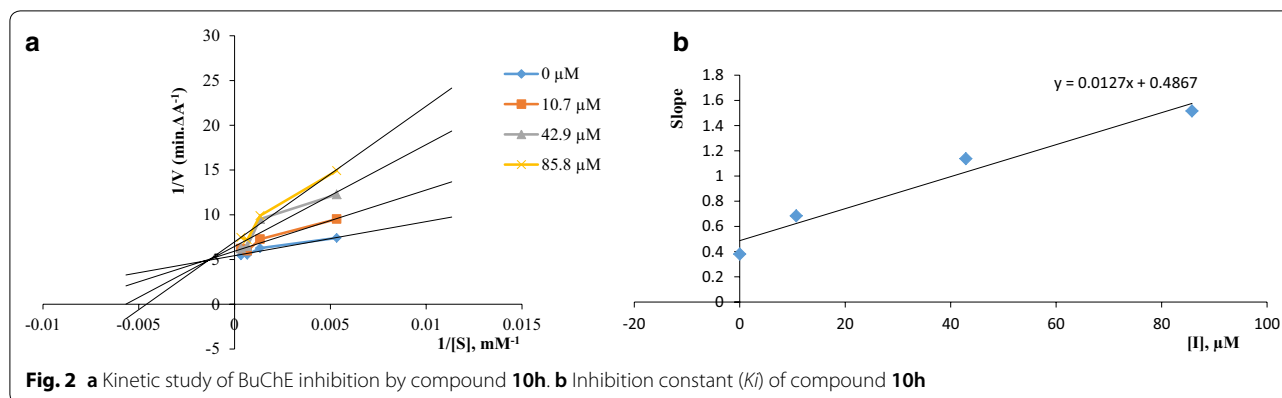


Table 2 Inhibitory activities of compounds **10h against $A\beta_{1-42}$ aggregation^a**

Samples	% Inhibition self-induced $A\beta_{1-42}$ aggregation	% Inhibition AChE-induced $A\beta$ aggregation
10h	32.6 ± 2.0 (10 μ M) ^b	29.4 ± 1.5 (100 μ M) ^c
Tacrine	7.6 ± 1.4 (10 μ M)	6.7 ± 0.9 (100 μ M)
Donepezil	18.1 ± 1.4 (10 μ M)	25.2 ± 1.7 (100 μ M)

^a Values are expressed as means ± SEM of three experiments. ^b Inhibition of self-induced $A\beta_{1-42}$ aggregation (25 mM) produced by the test compound at 10 μ M concentration. ^c Co-aggregation inhibition of $A\beta_{1-42}$ and AChE (2 μ M, ratio 100:1) by the test compound at 100 μ M

$A\beta_{1-42}$ self-aggregation, as depicted 32.6% inhibition at 10 μ M (Table 2). Furthermore, compound **10h** inhibited AChE-induced $A\beta$ aggregation by 29.4% at 100 μ M.

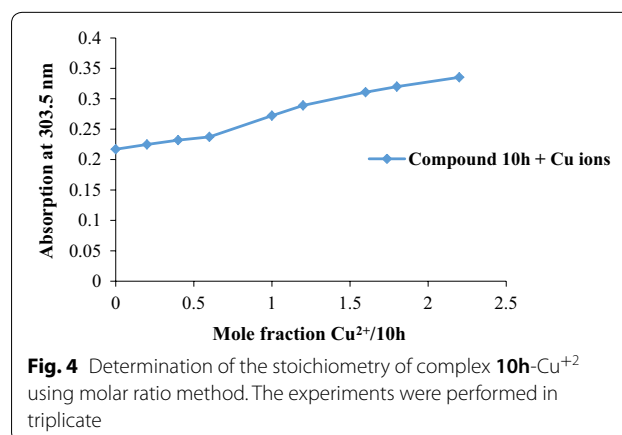
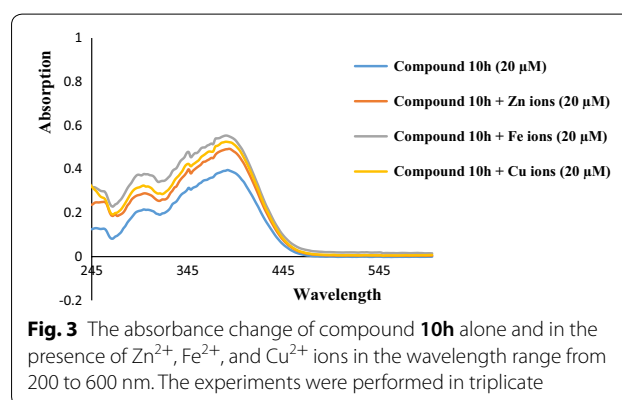
Neuroprotective studies on PC12 cell line

Compound **10h** was selected to study the neuroprotective ability using PC12 cell injury induced by $A\beta_{25-35}$ by MTT assay. This compound depicted no neuroprotective effect on $A\beta$ -induced on PC12 cells up to 50 μ M. It can be understood that bioisosteric replacement of iminochromene moiety (compound **F**, Fig. 1) with chromenone ring did not induce the desired neuroprotectivity.

Metal chelating

Compound **10h** was tested for its metal chelating ability towards Fe^{2+} , Cu^{2+} , and Zn^{2+} ions (Fig. 3) The UV spectrum of methanolic solution (20 μ M) of that compound showed two characteristic absorption peaks at 309.9 and 386.7 nm. After the interaction of compound **10h** with the above mentioned ions for 30 min, small shifts as well as absorption intensity changes were observed in the spectra confirming biometal-ligand interactions.

Interaction of compound **10h** with Zn^{2+} ions demonstrated two absorption peaks at 299.2 and 388.7 nm. Similar changes were observed in the case of Fe^{2+} ions and



those absorptions were observed at 303.5 and 390.9 nm. When compound **10h** was treated with Cu^{2+} ions, two peaks were observed at 303.5 and 382.4 nm. The stoichiometry of complex **10h**- Cu^{2+} was also studied (Fig. 4) due to important role of copper ions in AD [38]. The concentration of the test compound **10h** was 20 μ M and the final concentration of Cu^{2+} ranged from 0–44 μ M with 4 μ M intervals at 303.5 nm.

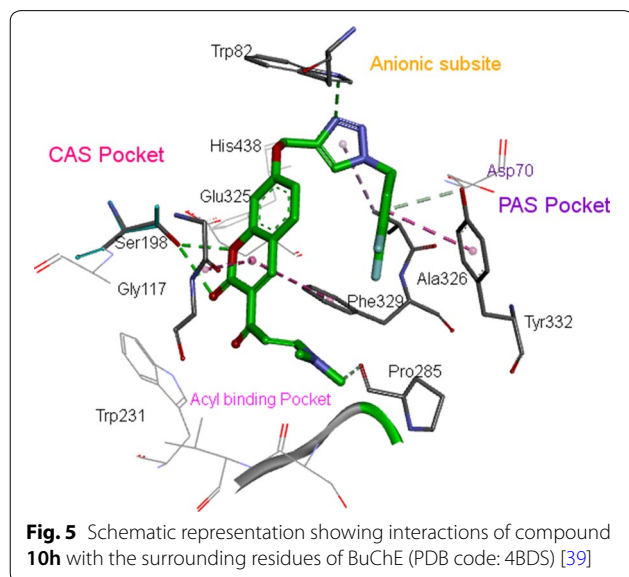
The plot was obtained by the corresponding absorption against the mole fraction of Cu^{2+} to ligand **10h**. According to the plot, the ratio 1:1 complexation ration of **10h**- Cu^{2+} can be seen at the fracture point of the plot with the mole fraction of 0.6.

Docking study of BuChE

As discussed in the introduction and designing sections, the volume of the BuChE active site is considerably bigger than the one found in the AChE, so BuChE can accommodate bulkier inhibitors, and this may constitute the basis for the selectivity of these derivatives. An overlay of the best pose for **10h** with BuChE was depicted in Fig. 5. Chromenone core is mostly surrounded by residues of CAS pocket, while 3,4-difluoro benzyl moiety oriented towards PAS pocket. More specifically, the carbonyl group of the chromenone ring form a hydrogen bond with the oxygen of Ser198 of the CAS while the another hydrogen bond interaction was seen between the oxygen of chromenone pendant group and Ser198. Also, the 1,2,3-triazole ring formed a third hydrogen bond with Trp82 of the anionic subsite.

The nitrogen of the dimethylamino propenone interacted with Pro285 via Van der Waals interaction. The Van der Waals interaction was also constructed between 1,2,3-triazole moiety and Ala326 of the PAS residue. The 3,4-difluoro benzyl ring also showed π - π and van der Waals interactions with Tyr332. These results strongly supported the high potency of compound **10h** against BuChE.

It is concluded that not only dimethylamino propenone attached to the chromenone ring is an essential factor for designing an active and selective BuChE inhibitor,



but also the nature of substituents on the aryl moiety connected to 1,2,3-triazole is the necessary element to afford a higher BuChE inhibitory effect. Further computer-aided lead optimization to improve anti-BuChE can be performed via replacing of the dimethylamino propenone pendant group with a cyclic amine to evaluate the size and bulkiness for the selective BuChE inhibition. Besides, the 1,2,3-triazole moiety can be substituted by different aliphatic spacer containing CO, NHCO, NH to improve the capability of H-bonding interaction with the active site.

In silico ADME evaluation

In silico ADME/T studies of the synthesized compounds was performed using <https://lmmd.ecust.edu.cn/admet/sar2/> and <https://preadmet.bmdrc.kr>, to evaluate pharmacokinetic properties of possible and potential candidates/drug molecules which can be helpful for future anti-AD drug developments [40]. As shown in Table 3, most of the compounds showed drug-like characteristics based on Lipinski's rule of five ($\text{MW} < 500$, $\text{cLogP} < 5$, $\text{HB donor} \leq 5$, $\text{HB acceptor} \leq 10$). Our results indicated that lipophilicity and solubility of the derivatives were drug-like. Furthermore, molecular weight, cLogP , and blood-brain barrier were well within the standard ranges.

Materials and methods

Instrumental methods

Melting points of synthesized compounds were determined on a Kofler hot stage apparatus. ^1H and ^{13}C NMR spectra were determined on a Varian FT-500, using TMS as an internal standard. IR spectra were recorded using KBr disks on a Bruker Tensor 27 FTIR spectrophotometer. Elemental analysis was carried out with an Elemental Analyzer system GmbH VarioEL CHN mode.

Synthesis of 2-hydroxy-4-(prop-2-yn-1-yloxy)benzaldehyde (3)

Compound **3** was prepared from the reaction of 2,4-dihydroxybenzaldehyde **1** and propargyl bromide **2** in the presence of potassium carbonate (K_2CO_3) and potassium iodide (KI) in acetone at 50°C , according to the literature [34].

Synthesis of 3-acetyl-7-(prop-2-yn-1-yloxy)-2H-chromen-2-one (5)

A few drops of piperidine were added to the mixture of compound **3** (1 mmol) and ethyl acetoacetate (2.5 mmol) **4** in ethanol (10 mL) and it was stirred overnight at room temperature to obtain yellow precipitates. After completion of the reaction (checked by TLC), they were filtered off and used for the next step with no further purification.

Table 3 Calculated molecular profile for synthesized compounds 10a-m

Compound	Descriptor						
	Mw	cLogP	H-bond acceptor	H-Bond Donor	BBB	Human intestinal absorption	Caco-2 permeability
10a	430.46	3.27	8	0	+0.9757	+0.9298	-0.8029
10b	444.49	3.58	8	0	+0.9754	+0.9370	-0.7985
10c	444.49	3.58	8	0	+0.9757	+0.9370	-0.7908
10d	460.49	3.28	9	0	+0.9757	+0.9298	-0.7587
10e	448.45	3.41	8	0	+0.9757	+0.9294	-0.8157
10f	448.45	3.41	8	0	+0.9757	+0.9294	-0.7970
10g	448.45	3.41	8	0	+0.9757	+0.9294	-0.8152
10h	466.44	3.55	8	0	+0.9757	+0.9294	-0.8211
10i	464.91	3.92	8	0	+0.9746	+0.9319	-0.8289
10j	464.91	3.92	8	0	+0.9746	+0.9319	-0.8211
10k	509.36	4.03	8	0	+0.9751	+0.9131	-0.8299
10l	509.36	4.03	8	0	+0.9751	+0.9131	-0.8115
10m	509.36	4.03	8	0	+0.9751	+0.9131	-0.8306

It was also completely characterized and compared with reported in the literature [41].

Synthesis of (E)-3-(3-(dimethylamino)acryloyl)-7-(prop-2-y n-1-yloxy)-2H-chromen-2-one (7)

The mixture of compound **5** (1 mmol) and DMA-DMF **6** (2 mmol) in 1,4-dioxane (10 mL) was heated at reflux for 6 h. Then, the solvent was evaporated under vacuum and the residue was purified using plate chromatography with ethyl acetate as eluent.

Yield: 55%; M.p. 98–100 °C. IR (KBr): 2925, 2850, 2150, 1715, 1642, 1598 cm^{-1} . ^1H NMR (CDCl_3 , 500 MHz): δ = 8.58 (s, 1H, H4), 7.93 (d, J = 12.4 Hz, 1H, CH), 7.54 (d, J = 8.3 Hz, 1H, H5), 6.93–6.91 (m, 2H, H6, H8), 6.34 (d, J = 12.4 Hz, 1H, CH), 4.77 (s, 2H, CH_2), 3.17 (s, 3H, CH_3), 2.97 (s, 3H, CH_3), 2.60 (1H, CH) ppm. ^{13}C NMR (CDCl_3 , 125 MHz): 182.3, 162.1, 160.1, 156.4, 155.0, 145.9, 131.1, 123.3, 113.5, 113.4, 101.3, 95.2, 77.2, 77.0, 56.3, 45.2, 37.6 ppm.

Synthesis of 1,2,3-triazole-dimethylaminoacryloyl-chromenone hybrids 10a-m

The final step was performed by the click reaction of compound **7** and in situ prepared azides **9**. For this purpose, a solution of benzyl chloride/bromide derivative **8** (1.1 mmol), sodium azide (0.06 g, 0.9 mmol), and triethylamine (0.13 g, 1.3 mmol) in water (4 mL) and *tert*-butyl alcohol (4 mL) was stirred at room temperature for 30 min. Then, compound **7** (0.5 mmol) and $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ (7 mol%) were added to the mixture and it was continued for 24 h. Upon completion of the reaction checked TLC, the mixture was diluted with water, extracted with

chloroform, and dried over anhydrous Na_2SO_4 . After evaporation of the solvent, the residue was recrystallized from ethyl acetate and petroleum ether to give pure product **10**. In the case of some compounds, they were purified using plate chromatography with ethyl acetate as eluent.

(E)-7-((1-Benzyl-1H-1,2,3-triazol-4-yl)methoxy)-3-(3-(dimethylamino)acryloyl)-2H-chromen-2-one (10a)

Yield: 54%; M.p. 186–188 °C. IR (KBr): 2920, 2853, 1715, 1640, 1597, 1558 cm^{-1} . ^1H NMR (CDCl_3 , 500 MHz): δ = 8.55 (s, 1H, triazole), 7.91 (d, J = 12.5 Hz, 1H, CH), 7.60 (s, 1H, H4), 7.50 (d, 1H, J = 8.5 Hz, H5), 7.38–7.29 (m, 5H, H2', H3', H4', H5', H6'), 6.92–6.90 (m, 2H, H6, H8), 6.33 (d, J = 12.5 Hz, 1H, CH), 5.55 (s, 2H, CH_2), 5.24 (s, 2H, CH_2), 3.17 (s, 3H, CH_3), 2.96 (s, 3H, CH_3) ppm. ^{13}C NMR (CDCl_3 , 125 MHz): δ = 182.4, 171.7, 162.9, 156.2, 151.4, 145.6, 144.0, 134.7, 133.3, 131.1, 129.8, 128.9, 128.2, 122.6, 121.5, 113.3, 101.2, 92.2, 63.4, 57.2, 47.6, 38.1 ppm. Anal. calcd. for $\text{C}_{24}\text{H}_{22}\text{N}_4\text{O}_4$: C, 66.97; H, 5.15; N, 13.02. Found: C, 66.71; H, 5.38; N, 12.86.

(E)-3-(3-(Dimethylamino)acryloyl)-7-((1-(2-methylbenzyl)-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one (10b)

Yield: 59%; M.p. 203–205 °C. IR (KBr): 2924, 2855, 1712, 1640, 1596, 1558 cm^{-1} . ^1H NMR (CDCl_3 , 500 MHz): δ = 8.58 (s, 1H, triazole), 7.92 (d, J = 12.3 Hz, 1H, CH), 7.52 (d, 1H, J = 8.3 Hz, H5), 7.46 (s, 1H, H4), 7.32–7.18 (m, 4H, H3', H4', H5', H6'), 6.93–6.91 (m, 2H, H6, H8), 6.35 (d, J = 12.3 Hz, 1H, CH), 5.57 (s, 2H, CH_2), 5.24 (s, 2H, CH_2), 3.18 (s, 3H, CH_3), 2.98 (s, 3H, CH_3), 2.29 (s,

3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): δ = 182.4, 162.4, 160.2, 156.5, 154.9, 145.9, 143.0, 137.0, 132.0, 131.1, 130.7, 129.5, 129.3, 126.7, 123.5, 122.7, 113.3, 113.0, 101.3, 95.2, 62.4, 52.5, 45.3, 37.6, 19.0 ppm. Anal. calcd. for C₂₅H₂₄N₄O₄: C, 67.55; H, 5.44; N, 12.60. Found: C, 67.31; H, 5.70; N, 12.44.

(E)-3-(3-(Dimethylamino)

acryloyl)-7-((1-(4-methylbenzyl)-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one (10c)

Yield: 55%; M.p. 199–201 °C. IR (KBr): 2923, 2855, 1715, 1640, 1596, 1558 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): δ = 8.57 (s, 1H, triazole), 7.92 (d, *J* = 12.3 Hz, 1H, CH), 7.55 (s, 1H, H4), 7.51 (d, 1H, *J* = 8.3 Hz, H5), 7.19–7.17 (m, 4H, H2', H3', H5', H6'), 6.94–6.91 (m, 2H, H6, H8), 6.34 (d, *J* = 12.3 Hz, 1H, CH), 5.51 (s, 2H, CH₂), 5.24 (s, 2H, CH₂), 3.17 (s, 3H, CH₃), 2.97 (s, 3H, CH₃), 2.36 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): δ = 182.5, 162.4, 160.2, 156.6, 154.9, 145.9, 143.1, 138.9, 131.2, 130.7, 129.8, 128.2, 123.3, 122.8, 113.3, 133.2, 101.2, 95.2, 62.4, 54.6, 45.6, 37.6, 21.2 ppm. Anal. calcd. for C₂₅H₂₄N₄O₄: C, 67.55; H, 5.44; N, 12.60. Found: C, 67.40; H, 5.26; N, 12.71.

(E)-3-(3-(Dimethylamino)

acryloyl)-7-((1-(3-methoxybenzyl)-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one (10d)

Yield: 52%; M.p. 209–211 °C. IR (KBr): 2923, 2855, 1712, 1640, 1595, 1557 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): δ = 8.57 (s, 1H, triazole), 7.92 (d, *J* = 12.4 Hz, 1H, CH), 7.59 (s, 1H, H4), 7.52 (d, 1H, *J* = 8.4 Hz, H5), 7.30 (t, *J* = 7.9 Hz, 1H, H5'), 6.94–6.81 (m, 5H, H6, H8, H2', H4', H6'), 6.35 (d, *J* = 12.4 Hz, 1H, CH), 5.52 (s, 2H, CH₂), 5.25 (s, 2H, CH₂), 3.79 (s, 3H, OCH₃), 3.17 (s, 3H, CH₃), 2.97 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): δ = 182.4, 162.4, 160.2, 156.6, 154.9, 145.9, 144.0, 143.2, 135.7, 130.7, 130.3, 123.4, 122.9, 119.3, 114.3, 113.8, 133.3, 112.5, 101.2, 95.2, 62.4, 55.3, 54.3, 45.2, 37.6 ppm. Anal. calcd. for C₂₅H₂₄N₄O₅: C, 65.21; H, 5.25; N, 12.17. Found: C, 65.41; H, 5.10; N, 12.32.

(E)-3-(3-(Dimethylamino)

acryloyl)-7-((1-(2-fluorobenzyl)-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one (10e)

Yield: 55%; M.p. 180–183 °C. IR (KBr): 2924, 2852, 1712, 1640, 1597, 1558 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): δ = 8.57 (s, 1H, triazole), 7.91 (d, *J* = 12.4 Hz, 1H, CH), 7.67 (s, 1H, H4), 7.51 (d, 1H, *J* = 8.0 Hz, H5), 7.39–7.38 (m, 1H, H4'), 7.30 (td, *J* = 7.6, 1.5 Hz, 1H, H3'), 7.17–7.11 (m, 2H, H5', H6'), 6.94–6.92 (m, 2H, H6, H8), 6.34 (d, *J* = 12.4 Hz, 1H, CH), 5.61 (s, 2H, CH₂), 5.25 (s, 2H, CH₂), 3.17 (s, 3H, CH₃), 2.97 (s, 3H, CH₃) ppm. ¹³C

NMR (CDCl₃, 125 MHz): δ = 182.3, 162.4, 160.6 (d, *J*_{C-F} = 251.2 Hz), 160.1, 158.7, 156.6, 154.8, 145.9, 143.2, 131.1 (d, *J*_{C-F} = 8.1 Hz), 130.7 (d, *J*_{C-F} = 3.3 Hz), 124.9 (d, *J*_{C-F} = 3.4 Hz), 123.3, 121.3, 121.6 (d, *J*_{C-F} = 14.5 Hz), 115.9 (d, *J*_{C-F} = 21.03 Hz), 113.3, 113.2, 101.2, 95.2, 62.4, 47.9, 45.2, 37.5 ppm. Anal. calcd. for C₂₄H₂₁FN₄O₄: C, 64.28; H, 4.72; N, 12.49. Found: C, 64.36; H, 4.57; N, 12.55.

(E)-3-(3-(Dimethylamino)

acryloyl)-7-((1-(3-fluorobenzyl)-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one (10f)

Yield: 60%; M.p. 183–185 °C. IR (KBr): 2922, 2850, 1713, 1640, 1597 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): δ = 8.57 (s, 1H, triazole), 7.91 (d, *J* = 12.3 Hz, 1H, CH), 7.61 (s, 1H, H4), 7.51 (d, 1H, *J* = 8.3 Hz, H5), 7.36 (td, 1H, *J* = 7.5, 2.2 Hz, H5'), 7.08–7.04 (m, 2H, H4', H6'), 6.97 (d, 1H, *J* = 9.2 Hz, H2'), 6.94–6.91 (m, 2H, H6, H8), 6.34 (d, *J* = 12.3 Hz, 1H, CH), 5.55 (s, 2H, CH₂), 5.26 (s, 2H, CH₂), 3.17 (s, 3H, CH₃), 2.96 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): δ = 180.0, 162.3, 160.7 (d, *J*_{C-F} = 262.2 Hz), 160.1, 156.7, 154.9, 145.8, 143.5, 136.6 (d, *J*_{C-F} = 19.4 Hz), 130.8 (d, *J*_{C-F} = 8.1 Hz), 130.7, 123.6, 123.3, 122.9, 115.9 (d, *J*_{C-F} = 20.9 Hz), 115.1 (d, *J*_{C-F} = 22.0 Hz), 113.3, 113.2, 101.2, 95.2, 62.4, 53.7, 45.0, 37.6 ppm. Anal. calcd. for C₂₄H₂₁FN₄O₄: C, 64.28; H, 4.72; N, 12.49. Found: C, 64.47; H, 4.84; N, 12.60.

(E)-3-(3-(Dimethylamino)

acryloyl)-7-((1-(4-fluorobenzyl)-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one (10g)

Yield: 62%; M.p. 183–185 °C. IR (KBr): 2923, 2855, 1714, 1640, 1598 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): δ = 8.56 (s, 1H, triazole), 7.92 (d, *J* = 12.4 Hz, 1H, CH), 7.58 (s, 1H, H4), 7.51 (d, 1H, *J* = 8.4 Hz, H5), 7.29 (dd, *J* = 8.7, 5.1 Hz, 2H, H2', H6'), 7.07 (t, *J* = 8.7 Hz, 2H, H3', H5'), 6.93–6.91 (m, 2H, H6, H8), 6.33 (d, *J* = 12.4 Hz, 1H, CH), 5.52 (s, 2H, CH₂), 5.24 (s, 2H, CH₂), 3.16 (s, 3H, CH₃), 2.96 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): δ = 180.1, 162.4, 160.5 (d, *J*_{C-F} = 260.5 Hz), 160.1, 156.5, 154.8, 145.1, 143.4, 134.5 (d, *J*_{C-F} = 19.5 Hz), 130.7, 130.1, 123.3, 122.8, 116.2 (d, *J*_{C-F} = 21.0 Hz), 113.3, 113.2, 101.2, 95.1, 62.1, 53.5, 45.0, 37.5 ppm. Anal. calcd. for C₂₄H₂₁FN₄O₄: C, 64.28; H, 4.72; N, 12.49. Found: C, 64.44; H, 4.60; N, 12.27.

(E)-7-((1-(3,4-Difluorobenzyl)-1H-1,2,3-triazol-4-yl)

methoxy)-3-(3-(dimethylamino)

acryloyl)-2H-chromen-2-one (10h)

Yield: 59%; M.p. 188–190 °C. IR (KBr): 2922, 2850, 1718, 1640, 1594 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): δ = 8.57 (s, 1H, triazole), 7.91 (d, *J* = 12.3 Hz, 1H, CH), 7.63 (s, 1H, H4), 7.52 (d, 1H, *J* = 8.4 Hz, H5), 7.21–7.11 (m, 2H, H5', H6'), 7.06–7.04 (m, 1H, H2'), 6.93–6.91 (m, 2H, H6, H8),

6.33 (d, $J=12.3$ Hz, 1H, CH), 5.52 (s, 2H, CH₂), 5.27 (s, 2H, CH₂), 3.18 (s, 3H, CH₃), 2.97 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): $\delta=182.3, 160.1, 156.6, 154.9, 150.5$ (d, $J_{C-F}=243.1$ Hz), 149.6, (d, $J_{C-F}=240.0$ Hz), 145.8, 144.0, 143.6, 132.8, 131.2, 130.7, 124.3, 123.4, 122.9, 118.1 (d, $J_{C-F}=17.2$), 117.3 (d, $J_{C-F}=17.6$), 113.3, 101.2, 95.2, 62.4, 53.2, 45.2, 37.6 ppm. Anal. calcd. for C₂₄H₂₀F₂N₄O₄: C, 61.80; H, 4.32; N, 12.01. Found: C, 61.63; H, 4.17; N, 11.84.

(E)-7-((1-(2-Chlorobenzyl)-1H-1,2,3-triazol-4-yl) methoxy)-3-(3-(dimethylamino) acryloyl)-2H-chromen-2-one (10i)

Yield: 58%; M.p. 178–180 °C. IR (KBr): 2923, 2852, 1713, 1640, 1597 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): $\delta=8.60$ (s, 1H, triazole), 7.91 (d, $J=11.0$ Hz, 1H, CH), 7.68 (s, 1H, H4), 7.51 (d, 1H, $J=8.1$ Hz, H5), 7.43 (d, 1H, $J=7.9$ Hz, H3'), 7.32–7.24 (m, 3H, H4', H5', H6'), 6.93–6.87 (m, 2H, H6, H8), 6.33 (d, $J=11.0$ Hz, 1H, CH), 5.69 (s, 2H, CH₂), 5.29 (s, 2H, CH₂), 3.16 (s, 3H, CH₃), 2.96 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): $\delta=182.7, 171.5, 163.8, 158.5, 156.3, 151.7, 144.5, 133.6, 131.7, 130.7, 130.5, 130.4, 130.0, 127.7, 123.4, 120.1, 113.3, 112.8, 100.6, 96.6, 61.1, 52.3, 46.4, 40.4$ ppm. Anal. calcd. for C₂₄H₂₁ClN₄O₄: C, 62.00; H, 4.55; N, 12.05. Found: C, 61.81; H, 4.38; N, 11.90.

(E)-7-((1-(4-Chlorobenzyl)-1H-1,2,3-triazol-4-yl) methoxy)-3-(3-(dimethylamino) acryloyl)-2H-chromen-2-one (10j)

Yield: 61%; M.p. 178–180 °C. IR (KBr): 2891, 2850, 1715, 1640, 1597, 1558 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): $\delta=8.54$ (s, 1H, triazole), 7.90 (d, $J=12.4$ Hz, 1H, CH), 7.61 (s, 1H, H4), 7.50 (d, $J=8.5$ Hz, 1H, H5), 7.34 (d, $J=8.1$ Hz, 2H, H3', H5'), 7.22 (d, $J=8.1$ Hz, 2H, H2', H6'), 6.92–6.89 (m, 2H, H6, H8), 6.34 (d, $J=12.4$ Hz, 1H, CH), 5.52 (s, 2H, CH₂), 5.19 (s, 2H, CH₂), 3.11 (s, 3H, CH₃), 2.95 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): $\delta=182.9, 162.4, 160.1, 156.5, 154.9, 145.8, 144.2, 143.4, 134.9, 130.7, 129.5, 129.4, 123.3, 122.9, 113.3, 133.2, 101.2, 95.2, 62.4, 53.5, 45.2, 37.6$ ppm. Anal. calcd. for C₂₄H₂₁ClN₄O₄: C, 62.00; H, 4.55; N, 12.05. Found: C, 62.11; H, 4.40; N, 12.28.

(E)-7-((1-(2-Bromobenzyl)-1H-1,2,3-triazol-4-yl) methoxy)-3-(3-(dimethylamino) acryloyl)-2H-chromen-2-one (10k)

Yield: 58%; M.p. 202–205 °C. IR (KBr): 2924, 2850, 1709, 1641, 1598, 1559 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): $\delta=8.57$ (s, 1H, triazole), 7.92 (d, $J=12.3$ Hz, 1H, CH), 7.70 (s, 1H, H4), 7.63 (d, 1H, $J=7.5$ Hz, H3'), 7.52 (d,

1H, $J=8.3$ Hz, H5), 7.33 (t, 1H, $J=7.5$ Hz, H5'), 7.25–7.21 (m, 2H, H4', H6'), 6.95–6.93 (m, 2H, H6, H8), 6.34 (d, $J=12.3$ Hz, 1H, CH), 5.69 (s, 2H, CH₂), 5.27 (s, 2H, CH₂), 3.21 (s, 3H, CH₃), 2.97 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): $\delta=182.5, 162.4, 160.1, 156.6, 154.9, 145.9, 143.1, 133.8, 133.4, 133.3, 130.7, 130.6, 130.5, 128.3, 123.6, 123.3, 113.4, 113.3, 100.4, 95.2, 62.0, 54.0, 44.8, 37.2$ ppm. Anal. calcd. for C₂₄H₂₁BrN₄O₄: C, 56.59; H, 4.16; N, 11.00. Found: C, 56.37; H, 4.30; N, 11.21.

(E)-7-((1-(3-Bromobenzyl)-1H-1,2,3-triazol-4-yl) methoxy)-3-(3-(dimethylamino) acryloyl)-2H-chromen-2-one (10l)

Yield: 64%; M.p. 218–220 °C. IR (KBr): 2925, 2856, 1710, 1640, 1595 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): $\delta=8.59$ (s, 1H, triazole), 7.93 (d, $J=12.4$ Hz, 1H, CH), 7.61 (s, 1H, H4), 7.55–7.51 (m, 3H, H5, H2', H4'), 7.26–7.18 (m, 2H, H5', H6'), 6.96–6.94 (m, 2H, H6, H8), 6.35 (d, $J=12.4$ Hz, 1H, CH), 5.51 (s, 2H, CH₂), 5.28 (s, 2H, CH₂), 3.19 (s, 3H, CH₃), 2.98 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): $\delta=180.1, 162.3, 160.2, 156.5, 154.9, 145.9, 143.5, 136.4, 133.7, 132.1, 131.1, 130.9, 130.8, 126.6, 122.9, 122.2, 113.4, 113.3, 101.1, 95.0, 62.8, 53.6, 44.8, 38.8$ ppm. Anal. calcd. for C₂₄H₂₁BrN₄O₄: C, 56.59; H, 4.16; N, 11.00. Found: C, 56.31; H, 4.24; N, 11.18.

(E)-7-((1-(4-Bromobenzyl)-1H-1,2,3-triazol-4-yl) methoxy)-3-(3-(dimethylamino) acryloyl)-2H-chromen-2-one (10m)

Yield: 64%; M.p. 173–175 °C. IR (KBr): 2925, 2850, 1715, 1640, 1597 cm⁻¹. ¹H NMR (CDCl₃, 500 MHz): $\delta=8.57$ (s, 1H, triazole), 7.91 (d, $J=12.3$ Hz, 1H, CH), 7.60 (s, 1H, H4), 7.52–7.51 (m, 3H, H5, H3', H5'), 7.17 (d, $J=8.4$ Hz, 2H, H2', H6'), 6.93–6.91 (m, 2H, H6, H8), 6.34 (d, $J=12.3$ Hz, 1H, CH), 5.51 (s, 2H, CH₂), 5.22 (s, 2H, CH₂), 3.17 (s, 3H, CH₃), 2.96 (s, 3H, CH₃) ppm. ¹³C NMR (CDCl₃, 125 MHz): $\delta=182.4, 162.9, 160.1, 156.6, 154.9, 146.6, 144.4, 143.5, 133.3, 132.4, 130.7, 129.8, 123.3, 123.1, 122.9, 112.7, 100.8, 95.2, 62.4, 53.6, 45.2, 37.6$ ppm. Anal. calcd. for C₂₄H₂₁BrN₄O₄: C, 56.59; H, 4.16; N, 11.00. Found: C, 56.68; H, 4.38; N, 10.87 (Additional file 1).

Inhibitory activities against AChE and BuChE

All enzymes and reagents required for the assay were obtained from Aldrich. The in vitro anticholinesterase activity of all synthesized compounds **10a–m** was assayed using modified Ellman's method using a 96-well plate reader (BioTek ELx808) according to the literature [36, 42]. Initially, the stock solutions of compounds **10** were prepared by dissolving the test compound (1 mg) in DMSO (1 mL) and then, diluted solutions at final concentrations of 1, 10, 20, and 40 μ g/mL were prepared

using methanol. Each well contained 50 μL potassium phosphate buffer ($\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$, 0.1 M, pH 8), 25 μL sample solution, and 25 μL enzyme (final concentration 0.22 U/mL in buffer). Control experiments were also performed under the same conditions without enzyme. After incubation at room temperature for 15 min, 125 μL DTNB (3 mM in buffer) was added and the characterization of enzymatic reaction was spectrometrically performed at 405 nm followed by the addition of substrate (ATCI 3 mM in water) after 5–10 min. The IC_{50} values were determined graphically from inhibition curves (log inhibitor concentration vs. percent of inhibition). Also, the same method was used for the BuChE inhibition assay.

Kinetic characterization of BuChE inhibitory activity

The kinetic study for the inhibition of BuChE by compound **10h** was carried out according to Ellman's method used for the inhibition assay using four different concentrations of inhibitor (0, 10.7, 42.9, and 85.8 μM). The Lineweaver–Burk reciprocal plot was provided by plotting $1/V$ against $1/[S]$ at variable concentrations of butyrylthiocholine as the substrate (187.5, 750, 1500, 3000 μM). The inhibition constant K_i was achieved by the plot of slopes versus the corresponding concentrations of compound **10h** [43, 44].

Inhibition of $\text{A}\beta_{1-42}$ aggregation and disaggregation of aggregated $\text{A}\beta_{1-40}$ induced by AChE

Inhibition of $\text{A}\beta_{1-42}$ self-aggregation was measured by ThT fluorescence assay. The details of the method were reported in our previous study [45]. To study $\text{A}\beta_{42}$ aggregation inhibition, a reported method, based on the fluorescence emission of ThT was followed. Briefly, the mixtures of $\text{A}\beta_{1-40}$ peptide (Bachem company, Switzerland) and AChE (Sigma, *Electrophorus electricus*), in the presence or absence of the test inhibitor were incubated for 24 h at room temperature. The final concentrations of $\text{A}\beta$ (dissolved in DMSO and diluted 0.215 M sodium phosphate buffer, pH 8), AChE (dissolved in 0.215 M sodium phosphate buffer, pH 8.0), and the test compound are 200 μM , 2 μM and 100 μM , respectively. After co-incubation, 20 μL of the mixed solution was diluted to a final volume of 2 mL with ThT (1.5 μM in 50 mM glycine–NaOH buffer, pH 8.5) and the absorbance was measured with a multi-mode plate reader at the excitation and emission wavelength of $\lambda_{\text{ex}} = 450$ nm and $\lambda_{\text{em}} = 485$ nm, respectively [45].

Neuroprotection assay against $\text{A}\beta$ -induced damage

MTT reduction assay was used to evaluate the neuroprotective effect of compound **10h** on neuronal PC12 cell damage induced by $\text{A}\beta_{25-35}$. The cells were grown

in monolayer culture on collagen-coated plates at 37 $^\circ\text{C}$ in a humidified atmosphere of 5% CO_2 . Neuronal PC12 cells were plated at a density of 5×10^5 cells/well on 96-well plates. The cells were pre-incubated with compound **10h** for 3 h before human $\text{A}\beta_{25-35}$ (final concentration of 5 μM) was added. After 24 h, 90 μL the medium was taken out and 20 μL of MTT (0.5 mg/ml dissolved in RPMI containing phenol red) was added and incubated for an additional 2 h at 37 $^\circ\text{C}$. The absorbance (A_{570} nm) was measured using a Bio-Rad microplate reader (Model 680, Bio-Rad). The details were reported in our previous work [33, 46].

Metal chelation studies

To study the metal chelating ability, the solutions of compound **10h** and Fe^{2+} , Cu^{2+} , and Zn^{2+} ions (from FeSO_4 , $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$, and ZnCl_2) were prepared in methanol. The mixture of compound **10h** (1 mL) and the test ion solutions (1 mL) with the same final concentration of 20 μM in a 1 cm quartz cuvette was incubated at room temperature for 30 min. Then, the absorption spectra were recorded in the range of 200–600 nm.

The stoichiometry of complex **10h**– Cu^{2+} was also studied using the molar ratio method [47, 48]. The concentration of compound **10h** was 20 μM and the final concentration of Cu^{2+} ranged from 0–44 μM with 4 μM intervals at 303.5 nm. The plot was obtained by the corresponding absorption versus mole fraction of Cu^{2+} to ligand **10h**. All experiments were performed in triplicate.

Molecular docking

The molecular docking studies of the most potential ligand was performed on BuChE (PDB code: 4BDS) [39] to observe the binding orientation and consensual binding interactions using AutoDock 4.2. The X-ray crystal structure of the receptor was downloaded from the PDB database. All water and ligand molecules were removed from the structure, and the protein was prepared for docking. The co-crystallized ligand within the PDB structures was defined as a center of the binding site. All ligands were created using Chem3D Ultra software, and energy minimizations were done by the semiempirical MM^+ [49]. The compounds were docked into the active site of proteins using default parameters for each ligand with 100 runs and 27,000 as the maximum number of generations. The grid boxes were set with 60, 60, and 60 points in the x, y, and z directions, respectively. All other options were set as default. The calculated geometries were ranked in terms of free energy of binding and the best pose was selected for further analysis. Molecular

visualizations were performed by Discovery Studio 4.0 client software [20].

Prediction of ADME descriptors

ADME-Tox properties of the synthesized compounds were performed by using online servers especially <https://lmm.d.ecust.edu.cn:8000/predict/> and <https://preadmet.bmdrc.kr>.

Conclusion

In summary, a new series of 1,2,3-triazole-dimethyl-aminoacryloyl-chromenone derivatives were designed and synthesized as multifunctional anti-Alzheimer's agents. All the target compounds were synthesized and screened as AChE/BuChE inhibitors. The most active compound was further evaluated by the multiple biological activities including $A\beta_{1-42}$ aggregation inhibition, metal-chelating properties, and neuroprotective effects against $A\beta_{25-35}$ -induced PC12 cell injury. Our results showed that these compounds had a high inhibitory potency and selectivity towards BuChE with an IC_{50} value of 21.71 μ M for **10h** as the most potent BuChE inhibitor. The inhibition kinetic analysis revealed a mixed-type inhibition pattern for this compound. The molecular modeling study of the most potent compound **10h** with BuChE indicated that it bound to both CAS and PAS of the BuChE. Moreover, this compound had a significant anti- $A\beta$ aggregation capacity and served as a metal chelator. These results indicated that this hybridization approach could be a successful strategy for the further developments of potential multifunctional candidates against AD.

Supplementary information

Supplementary information accompanies this paper at <https://doi.org/10.1186/s13065-020-00715-0>.

Additional file 1. The supplementary file include copies of NMR spectra and elemental analysis report.

Abbreviations

AD: Alzheimer's disease; AChE: Acetylcholinesterase; BuChE: Butyrylcholinesterase; CAS: Catalytic active site; PAS: Peripheral anionic site; $A\beta$: Amyloid β ; APP: Amyloid precursor protein; BACE1: β -Site APP cleaving enzyme-1; NFTs: Neurofibrillary tangles; IC_{50} : The half maximal inhibitory concentration; 1H NMR: Proton nuclear magnetic resonance; ^{13}C NMR: Carbon-13 nuclear magnetic resonance.

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This paper is dedicated to the memory of our unique teacher in Chemistry and Medicinal Chemistry, Professor Abbas Shafiee (1937–2016)

Authors' contributions

MS contributed to the design and characterization of compounds as well as preparation of the manuscript. HKA synthesized compounds. AI performed docking study and contributed to the preparation of the manuscript. AR performed the biological assay. SNAB and OF supervised the biological tests. TA supervised all phases of the study. All authors read and approved the final manuscript.

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Availability of data and materials

The datasets used and analyzed during the current study are available from the corresponding author on reasonable request.

Ethics approval and consent to participate

The manuscript does not contain studies with animal subjects.

Consent for publication

Not applicable.

Competing interests

The authors declare that they have no competing interests.

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