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Growth and Characterisation Studies of Eu_3O_4 Thin Films Grown on Si/SiO₂ and Graphene

Razan O. M. Aboljadayel^{1,2,*} , Adrian Ionescu¹, Oliver J. Burton³ , Gleb Cheglakov¹, Stephan Hofmann³ and Crispin H. W. Barnes¹

¹ Cavendish Laboratory, Physics Department, University of Cambridge, Cambridge CB3 0HE, UK; ai222@cam.ac.uk (A.I.); cheglakovg@yahoo.co.uk (G.C.); chw101@cam.ac.uk (C.H.W.B.)

² Diamond Light Source, Didcot OX11 0DE, UK

³ Department of Engineering, University of Cambridge, Cambridge CB3 0FA, UK; ob303@cam.ac.uk (O.J.B.); sh315@cam.ac.uk (S.H.)

* Correspondence: roma2@cam.ac.uk or razan.aboljadayel@diamond.ac.uk

Abstract: We report the growth, structural and magnetic properties of the less studied Eu-oxide phase, Eu_3O_4 , thin films grown on a Si/SiO₂ substrate and Si/SiO₂/graphene using molecular beam epitaxy. The X-ray diffraction scans show that highly textured crystalline $\text{Eu}_3\text{O}_4(001)$ films are grown on both substrates, whereas the film deposited on graphene has a better crystallinity than that grown on the Si/SiO₂ substrate. The SQUID measurements show that both films have a Curie temperature of $\sim 5.5 \pm 0.1$ K, with a magnetic moment of ~ 320 emu/cm³ at 2 K. The mixed valence of the Eu cations has been confirmed by the qualitative analysis of the depth-profile X-ray photoelectron spectroscopy measurements with the $\text{Eu}^{2+}:\text{Eu}^{3+}$ ratio of 28:72. However, surprisingly, our films show no metamagnetic behaviour as reported for the bulk and powder form. Furthermore, the microscopic optical images and Raman measurements show that the graphene underlayer remains largely intact after the growth of the Eu_3O_4 thin films.

Keywords: Eu_3O_4 ; graphene; thin film; heterostructure; metamagnetism; XPS



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1. Introduction

Mixed-valence or fluctuating valence behaviour are usually found in lanthanide-based compounds due to the intermixing of the $s-d$ band with the localised f band near the Fermi level. Therefore, they exhibit unique magnetic, thermal and electrical properties [1]. Eu cations in Eu-based compounds mostly occur in the 2⁺ valence. However, in trieuropium tetroxide (Eu_3O_4), Eu ions exhibit a mixed valence of Eu^{2+} and Eu^{3+} .

Eu_3O_4 crystallises into an orthorhombic structure (space group $Pnma$) similar to CaFe_2O_4 with the lattice parameters $a = 10.085$ Å, $b = 3.502$ Å and $c = 12.054$ Å [2,3]. Figure 1 shows the Eu_3O_4 structure, where the Eu^{2+} and Eu^{3+} ions occupy the Ca^{2+} and Fe^{3+} sites, respectively. The oxygen ions form a six- and eight-fold coordination around the Eu^{3+} and Eu^{2+} ions, respectively. The coordination is then completed with two oxygen ions lying at the corner of the crystal [2,4]. Therefore, Eu_3O_4 has two Eu^{3+} ions and one Eu^{2+} ion per unit formula [4].

The Eu_3O_4 compound has an antiferromagnetic arrangement below 5 K. Its bulk and powder forms show a metamagnetic behaviour below the Néel temperature (T_N), at a critical field of 2.4 kOe [4,5]. A large magnetocaloric effect has also been measured in Eu_3O_4 . Therefore, it is considered a potential material for magnetic refrigeration applications [3]. Although Eu_3O_4 is a mixed-valence compound, its magnetic ordering is mainly determined by the Eu^{2+} ions at low-temperature due to the high magnetic moment of Eu^{2+} (total angular momentum, $J = 7/2$) in comparison to the Eu^{3+} ions ($J = 0$) [3–5]. It has been proposed that the nearest neighbouring Eu^{2+} ions are strongly coupled by ferromagnetic interactions

at low temperature, whereas the distant ions are coupled by weaker antiferromagnetic coupling, resulting in the overall antiferromagnetic state of Eu_3O_4 [5].

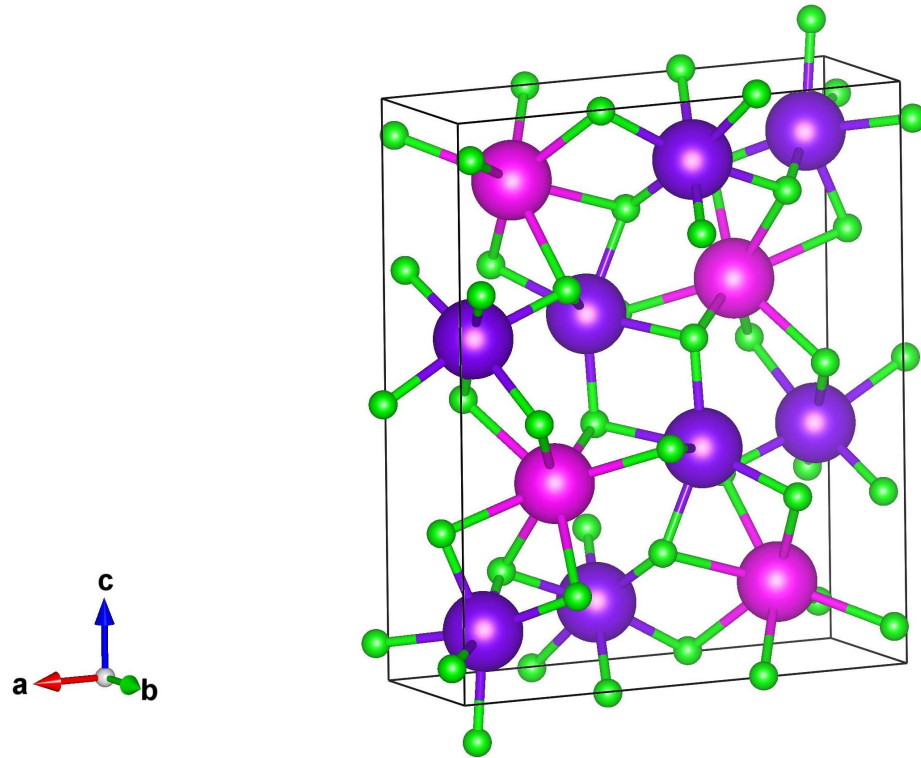


Figure 1. The orthorhombic structure of Eu_3O_4 , in which the purple, magenta and green spheres represent the Eu^{3+} , Eu^{2+} and O^{2-} ions, respectively. The black box represents one unit cell of the Eu_3O_4 crystal. The model was constructed using VESTATM software package [6].

Graphene is a potential material for many technological applications due to its attractive properties, such as its long spin-diffusion length, negligible spin-orbit coupling and high electron mobility [7–9]. So far, no study has been reported on the graphene/ Eu_3O_4 system, unlike the well-studied Eu-oxide phase (EuO) where many theoretical and experimental works have been published on the graphene/EuO heterostructure [10–17]. This may well be due to the difficulty of growing Eu_3O_4 , which is the unstable high-temperature phase of Eu-oxide, whereas EuO is grown at room temperature (RT). Eu-chalcogenides and Eu-oxides attracted great attention as promising materials for magneto-optical device applications [18–20]. Among these materials, a special interest was given to EuO due to its desirable properties for potential spintronics applications such as spin-filter [14,20–26]. On the other hand, only a few papers are found on the Eu_3O_4 compound [2–5,27–31], making it an interesting metamagnetic compound to be studied as there are still many aspects to be explored on this material.

In this study, 20 nm Eu_3O_4 thick films were grown on a Si/SiO₂ substrate and graphene sheet supported on Si/SiO₂ by molecular beam epitaxy (MBE) and subsequently capped with 5 nm of Au. Eu was deposited at high temperatures (300–600 °C) in an oxygen flux. The growth parameters such as the oxygen partial pressure, temperature and deposition rate were optimised to achieve a crystalline $\text{Eu}_3\text{O}_4(001)$ phase. The structural characterisation of the films was studied by X-ray diffraction (XRD) and reflection (XRR), where a superconducting quantum interference device magnetometer (SQUID) was used to study their magnetic properties. The results show a successful growth of crystalline, highly textured $\text{Eu}_3\text{O}_4(001)$ films with a Curie temperature (T_C) of ~ 5.5 K, which is in agreement with the values reported in References [3–5].

Depth-profile X-ray photon electron spectroscopy (XPS) scans were performed to prove the mixed valence of Eu cations in Eu_3O_4 . Furthermore, Raman spectroscopy mea-

measurements on the Si/SiO₂/graphene/Eu₃O₄ sample showed that although the growth of Eu₃O₄ film induces defects in the graphene sheet, the graphene retains its hexagonal lattice structure. Therefore, this study presents one of the first fundamental steps towards understanding the exchange coupling between Eu₃O₄ and graphene.

2. Materials and Methods

Firstly, 20 nm Eu₃O₄(001) films were deposited on cleaned Si/SiO₂ and commercially purchased Si/SiO₂/graphene (ACS Material, Pasadena, CA, USA) substrates by a bespoke MBE with a base pressure of 4×10^{-10} mbar. The substrates were heated to 400 °C, while the Eu was evaporated at a rate of 1.2 nm/min. Oxygen was then introduced into the growth chamber, resulting in a partial pressure of 1.1×10^{-8} mbar to deposit Eu₃O₄(001) at a rate of 1.11 nm/min. A 5 nm film of Au was grown subsequently on the Eu₃O₄(001) films to prevent them from oxidising to the most stable oxide phase of Eu (Eu₂O₃). The Au films were deposited at 45 °C, with a rate of 0.057 nm/min at a pressure of 1.9×10^{-10} mbar.

An in-house built quartz crystal microbalance system was used during the deposition to monitor the growth rate and thus the thicknesses of the layers. RT XRD scans were used to study the crystallinity of the grown films. The XRR measurements were performed to confirm the results of the microbalance readings and deduce the density and roughness between the layers. These acquisitions were carried out using a Bruker D8 Discover HRXRD (Bruker, Billerica, MA, USA) with a Cu K α monochromatic beam with a voltage of 40 kV and a current of 40 mA. The magnetic properties of the Eu₃O₄ films were studied using a Quantum Design SQUID (Quantum Design, San Diego, CA, USA).

Depth-profile XPS scans were performed on the Si/SiO₂/Eu₃O₄/Au sample using a Thermo Fisher Scientific Escalab 250Xi XPS (Thermo Fisher Scientific, Waltham, MA, USA) with Al K α X-ray source (1486.68 eV, beam width of 500 μ m). This was done to study the homogeneity of the Eu₃O₄ film, determine the atomic ratio of Eu²⁺ and Eu³⁺ and confirm the mixed-valence character of the grown film. Furthermore, the effect of depositing Eu₃O₄ film on the graphene sheet was investigated by Raman spectroscopy measurements using a Renishaw InVia spectrometer (Renishaw plc, Wotton-under-Edge, UK) (100 \times objective, 10% laser power, spot size of \sim 1 μ m, 0.5 s exposure time, wavelength of 532 nm). However, the Au capping layer was selectively etched in KI/I₂ solution before the measurements to eliminate the Au interference with the Raman measurements. The Si/SiO₂/graphene/Eu₃O₄/Au sample was cleaved into \sim 2 \times 2 mm square and placed in the etchant solution for 5 minutes at RT, rinsed with DI water twice, then with IPA and dried with dry N₂. Raman scans were then taken at every 50 μ m in a grid pattern over an area of 1000 μ m \times 1000 μ m for the region coated with Eu₃O₄, and every 10 μ m over an area of 140 μ m \times 140 μ m for the bare graphene surface.

3. Results and Discussion

3.1. X-ray Diffraction and Reflection (XRD and XRR)

The RT XRD scans (from 10°–100°) of the Eu₃O₄(001) films grown on the Si/SiO₂ substrate and on graphene are shown in Figure 2a. The XRD scans show highly textured Eu₃O₄(002) films with no sign of other oxide-phase of Eu or unreacted Eu within the detection limit of the set-up. Additional Eu₃O₄(004) and (008) peaks are observed in the scan of the Si/SiO₂/graphene/Eu₃O₄/Au sample, indicating that the underlying graphene layer improves the crystallinity of the Eu₃O₄ film. This is also proven by the smaller full-width at half-maximum (FWHM) of the Eu₃O₄ peaks of the Si/SiO₂/graphene/Eu₃O₄/Au sample (0.633°) compared to the Si/SiO₂/Eu₃O₄/Au (1.13°) as can be seen in Figure 2c. Figure 2b shows the XRR scan and the corresponding fit for the Si/SiO₂/Eu₃O₄/Au sample, whereas the deduced values for the thickness, density and roughness of the layers are listed in the inset table.

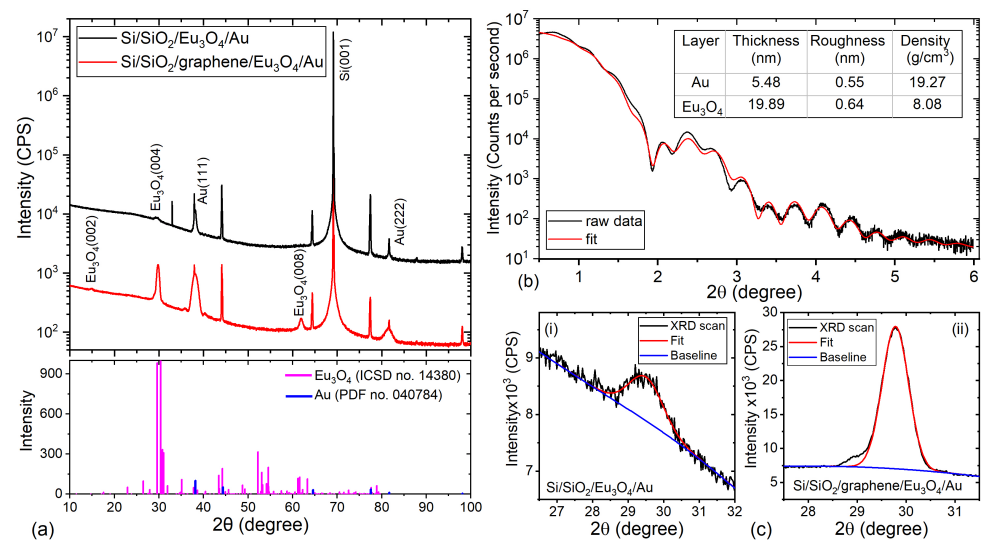


Figure 2. (a) The RT XRD scan of the Si/SiO₂/Eu₃O₄(001)/Au (black line) and Si/SiO₂/graphene/Eu₃O₄(001)/Au (red line) samples carried out between 10° and 100° using a monochromator and a 1D detector. The scan for the Si/SiO₂/graphene/Eu₃O₄(001)/Au sample is downshifted by a factor of five for ease of comparison. The bottom panel shows the reference patterns for Eu₃O₄ (magenta—ICSD no. 14380) and Au (blue—PDF no. 040784). (b) The XRR measurement of the Si/SiO₂/Eu₃O₄(001)/Au sample (black line) and the corresponding fit (red line) between 0° and 6°. The table lists the thickness, roughness and density of the deposited films as deduced from the fit. (c) The Gaussian fit for the Eu₃O₄(004) peak of the Si/SiO₂/Eu₃O₄(001)/Au (i) and the Si/SiO₂/graphene/Eu₃O₄(001)/Au sample (ii).

3.2. SQUID

Figure 3 shows the field-cooled (FC) and zero field-cooled (ZFC) measurements of the Si/SiO₂/Eu₃O₄(001)/Au and Si/SiO₂/graphene/Eu₃O₄(001)/Au samples, respectively. Both show a T_C of $\sim 5.5 \pm 0.1$ K, as can be deduced from the dM/dT vs. T (insets), which agrees with the values reported in the literature [3–5]. Therefore, care must be given to check for Eu₃O₄ impurities in EuO_{1-x} thin films, which sometimes show a pronounced double-dome feature at $T < 20$ K [32,33].

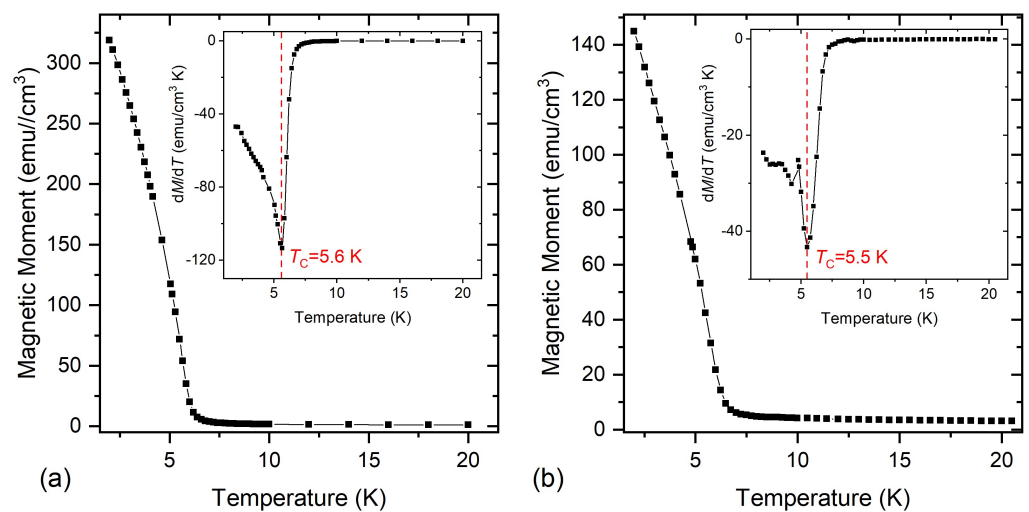


Figure 3. (a) The 20 Oe FC M vs. T measurement for Si/SiO₂/Eu₃O₄/Au. The inset presents the dM/dT vs. T plot used to determine the T_C . (b) The ZFC M vs. T measurement for the Si/SiO₂/graphene/Eu₃O₄/Au sample. The inset shows the dM/dT vs. T graph used to deduce the T_C of the Eu₃O₄.

The ZFC isothermal magnetisation measurements as a function of the applied magnetic field for the Si/SiO₂/graphene/Eu₃O₄/Au sample at 2 K, 5 K and 10 K are shown in Figure 4. The hysteresis curves show that the grown Eu₃O₄ films exhibit ferromagnetic behaviour with a coercive field of 22 Oe. The inset highlights the virgin magnetisation curves at these temperatures. Although the XRD scans (Figure 2a) and the M vs. T measurements (Figure 3) prove the growth of Eu₃O₄(001) thin films, surprisingly, the virgin M – H curves show no metamagnetic transition even with an applied in-plane magnetic field of 3 kOe as reported for crystal and powder Eu₃O₄ [3,4]. This could be attributed to the strain from the substrate which may be resolved by growing a thicker film.

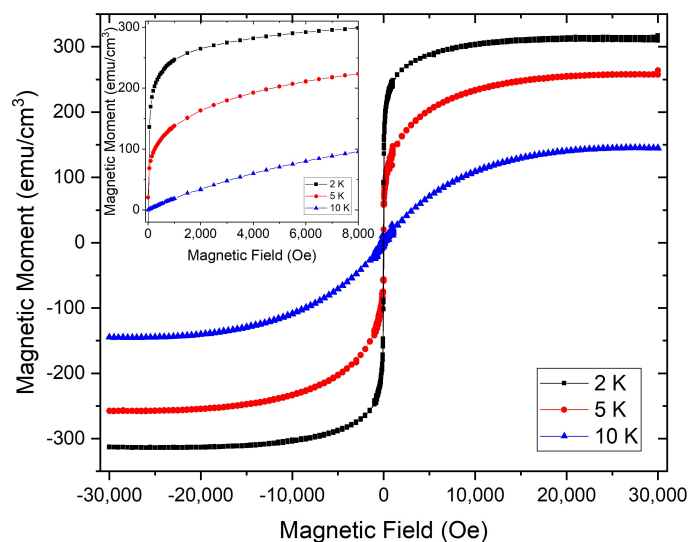


Figure 4. ZFC Isothermal magnetisation hysteresis loops of the Si/SiO₂/graphene/Eu₃O₄/Au sample measured as a function of the applied in-plane magnetic field at 2 K, 5 K and 10 K. The inset highlights the virgin magnetisation curves at these temperatures.

3.3. XPS

The existence of mixed-valence Eu cations was investigated by performing depth-profile XPS scans while measuring the Eu 3*d* and 4*d* spectra simultaneously after Ar⁺ plasma etching. Figure 5a shows the XPS etch profile of the sample, whereas the XPS survey collected at $t = 60$ s, 360 s and 510 s, highlighting the different detected elements are shown in Figure 5b. The 4*d* XPS spectra have a complicated structure (not shown) due to the strong unfilled 4*f*–4*d* hole interaction, whereas the 3*d* states have a weaker multiplet splitting and broader photoexcitation cross-section. Therefore, the latter is usually used to analyse the Eu XPS spectra and better estimate the Eu initial valence [34–37].

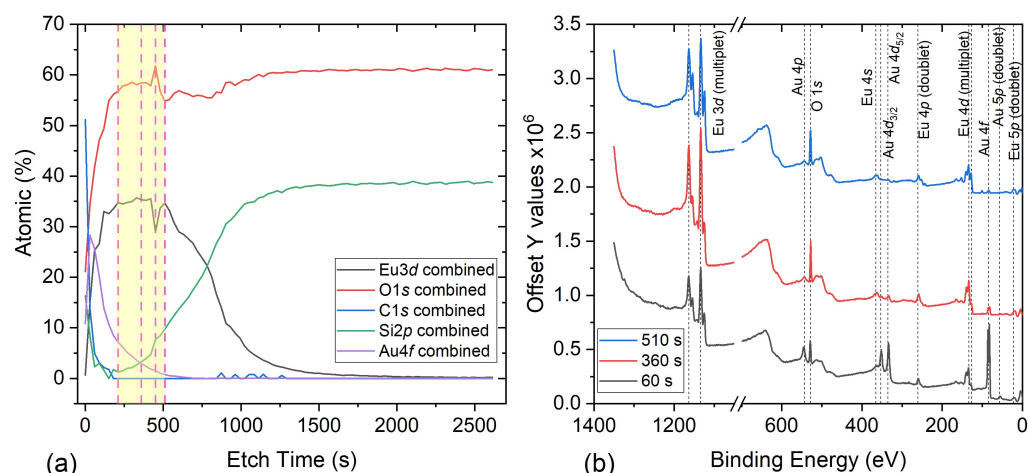


Figure 5. (a) XPS etch profile of the Si/SiO₂/Eu₃O₄/Au sample by Ar⁺ plasma etching. The yellow shaded region highlights the area considered for the analysis of Eu cations, whereas the dashed vertical lines indicate $t = 210$ s, 360 s, 450 s and 510 s. (b) The XPS survey spectra of the Si/SiO₂/Eu₃O₄/Au sample, collected at $t = 60$ s, $t = 360$ s and $t = 510$ s, highlighting the Au, Eu and O peaks.

Figure 6a–d shows the Eu 3d XPS spectra after subtracting an optimised Shirley background, measured at $t = 210$ s, 360 s, 450 s and 510 s. The peaks were deconvoluted using Gaussian–Lorentzian fitting, while the χ^2 value indicates the quality of the fit. Although the Eu₃O₄ layer was etched fully, only these scans were considered for the analysis of the Eu cation valency (the yellow shaded area of Figure 5a) to minimise the effect of interdiffusion at the SiO₂/Eu₃O₄ and Eu₃O₄/Au interfaces and increase the intensity of the Eu 3d and 4d peaks.

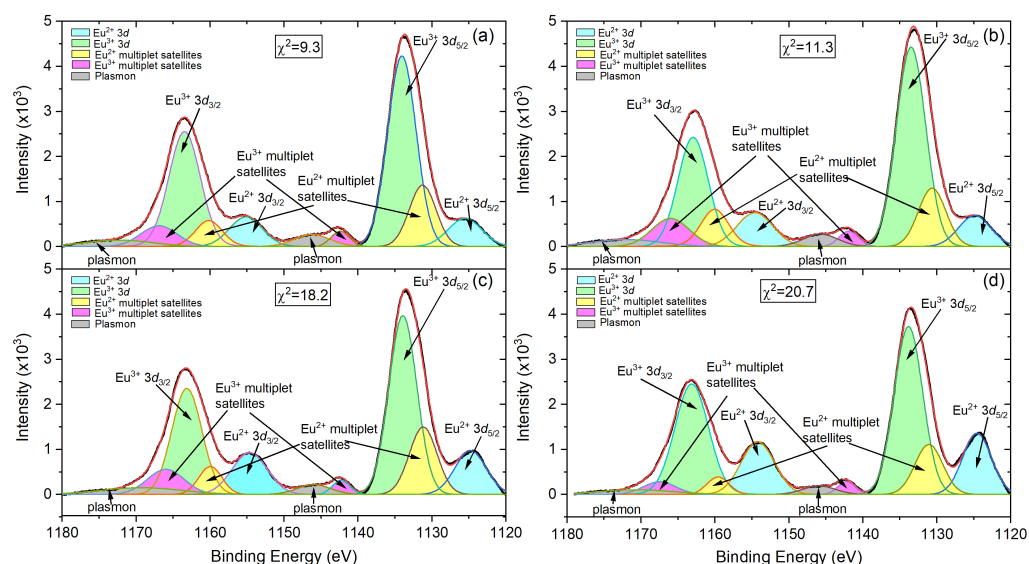


Figure 6. The deconvoluted 3d XPS spectra of the Eu₃O₄ film on Si/SiO₂ substrate using Al K α source (1486.68 eV) after the Shirley background subtraction, measured at etching time (a) $t = 210$ s, (b) $t = 360$ s, (c) $t = 450$ s and (d) $t = 510$ s. The raw data (black line), fitting curve (red line), Eu²⁺ 3d (blue shaded peaks), Eu²⁺ multiplet satellites (yellow shaded peaks), Eu³⁺ 3d (green shaded peaks), Eu³⁺ multiplet satellites (magenta shaded peaks) and plasmon excitations (grey shaded peaks).

All spectra in Figure 6 show the spin-orbit coupling (SOC) components, 3d_{5/2} and 3d_{3/2}, for Eu²⁺ and Eu³⁺ separated by $\Delta \sim 29.5$ eV, which agrees with previously reported values [34,36,38]. They also show additional peaks at slightly higher binding energy (BE) to

the SOC peaks for the Eu^{2+} and Eu^{3+} . These shake-up satellite peaks arise as a result of the multiplet structures of the $4f^7 - 3d$ hole in the final state [37]. Furthermore, the fast $3d$ photoelectrons create plasmon excitation structures observed as broad peaks at BE ~ 1146 eV and ~ 1170 eV [34]. The XPS spectra shown in Figure 6 prove the mixed valency of Eu cations as they agree well with previous work reported for Eu^{2+} and Eu^{3+} [34,36,39–41]. Moreover, the average atomic ratio of the Eu^{2+} to Eu^{3+} in the $3d_{5/2}$ and $3d_{3/2} \sim 28 : 72$ is consistent with the values reported in Eu-doped ZnO [42] and Eu-doped GaN nanowires [43]. Table 1 summarises the positions of the Eu^{2+} and Eu^{3+} $3d$ peaks, their FWHM, their corresponding multiplet satellites, the ratio of $\text{Eu}^{2+}/\text{Eu}^{3+}$ and the fits χ^2 values of the four spectra.

3.4. Raman Spectroscopy

Raman spectroscopy is a versatile and nondestructive technique widely used to study the graphene's structural and electronic properties [44,45]. A good-quality monolayer of graphene has two main characteristic Raman peaks; the G and 2D peaks at ~ 1582 cm^{-1} and ~ 2700 cm^{-1} , respectively. It can also possess other disorder-induced peaks, such as the D peak at ~ 1350 cm^{-1} [46–48]. Therefore, the presence or absence of these peaks and the ratio of the intensity of the D peak to the intensity of the G peak (I_D/I_G), which reflects the defect density in the graphene structure, were used to assess the quality of our graphene underlayer [49].

Table 1. The positions of the Eu^{2+} and Eu^{3+} $3d$ peaks, their FWHM and the position of their corresponding multiplet satellites. The atomic ratios of the Eu^{2+} to Eu^{3+} were deduced from the areas of the peaks. The table also lists the χ^2 value of the fittings.

Spectrum	$3d_{5/2}$	FWHM	$3d_{5/2}$ Satellites	$3d_{3/2}$	FWHM	$3d_{3/2}$ Satellites
Eu^{2+} (eV)						
$t = 210$ s	1125.54	5.39	1131.26	1155.14	5.29	1160.16
$t = 360$ s	1124.88	5.17	1130.64	1154.75	5.76	1160.04
$t = 450$ s	1124.78	5.01	1131.19	1154.65	5.64	1159.96
$t = 410$ s	1124.56	4.58	1131.07	1154.21	5.36	1159.58
Eu^{3+} (eV)						
$t = 210$ s	1134.07	4.56	1142.39	1163.46	5.03	1166.92
$t = 360$ s	1133.49	4.69	1141.90	1162.96	4.85	1166.00
$t = 450$ s	1133.93	4.46	1142.29	1163.15	4.80	1166.00
$t = 410$ s	1133.83	4.56	1142.23	1163.15	5.38	1167.48
	Atomic ratio $3d_{5/2}$			Atomic ratio $3d_{3/2}$		χ^2
	Eu^{2+}	Eu^{3+}		Eu^{2+}	Eu^{3+}	
$t = 210$ s	49.81	50.19		21.39	78.61	9.30
$t = 360$ s	14.63	85.37		26.84	73.16	11.30
$t = 450$ s	21.58	78.42		31.61	68.39	18.20
$t = 410$ s	26.80	73.20		31.96	68.04	20.70
Average	28.20	71.80		27.95	72.05	

Figure 7a,b show the microscopic optical images of the sample highlighting three different regions of the Si/SiO₂/graphene/Eu₃O₄/Au sample taken before etching the Au capping layer. The zoomed-in image collected with a $\times 100$ objective lens (Figure 7b) shows that the graphene layer consists of discontinuous mixture of mono- and multilayer graphene domains rather than a continuous homogeneous monolayer, which could be either a result of the growth of the Eu₃O₄ film or the quality of the commercial graphene. Therefore, one would expect the presence of defect-induced peaks in the Raman scans [49].

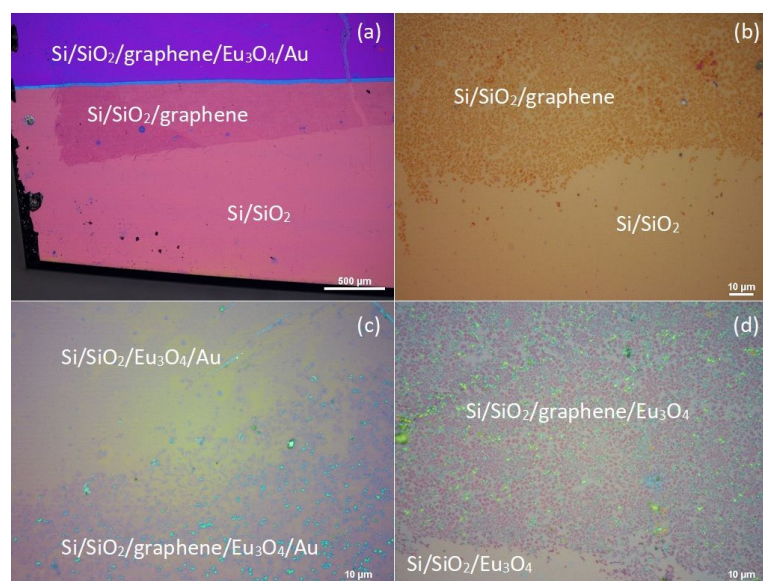


Figure 7. The microscopic optical images of the Si/SiO₂/graphene/Eu₃O₄/Au sample highlighting the different regions of the surface of the sample. (a) the bare and coated area of the Si/SiO₂ substrate before the etching process, (b) a zoomed-in view of the surface using a 100X objective lens showing the mixed domain structures of the graphene underlayer, (c) the graphene under the Eu₃O₄ layer before and (d) after etching the Au capping layer.

Figure 7c,d show the microscopic optical images of the graphene edge under the Eu₃O₄ film before and after removing the Au layer, respectively. No significant change in the contrast is observed between the two images, suggesting that the etching process did not remove the graphene underlayer.

Four random Raman scans taking on two different areas of the sample's surface, bare graphene and the graphene/Eu₃O₄ region after removing the Au layer, are shown in Figure 8a,b, respectively. The emergence of the additional defect-induced peaks in the spectra of the graphene/Eu₃O₄ area and the increase in their intensities (Figure 8b) compared to the scans of the bare graphene (Figure 8a) indicates that the growth of Eu₃O₄ film increased the defect density in the graphene structure. This is also seen by the shift in the I_D/I_G ratio of the graphene under the Eu₃O₄ film towards the higher values (Figure 9b), compared to the bare graphene sheet (Figure 9a). This is because I_D/I_G is known to be small for low-defect-density graphene [48,50]. However, the optical images (Figure 7) and Raman measurements (Figure 8) show that the underlayer graphene is largely maintained after the growth of the Eu₃O₄ film. Moreover, the data suggest the growth parameters can be optimised further to reduce the effect of the deposition of Eu₃O₄ and improve the Eu₃O₄/graphene interface.

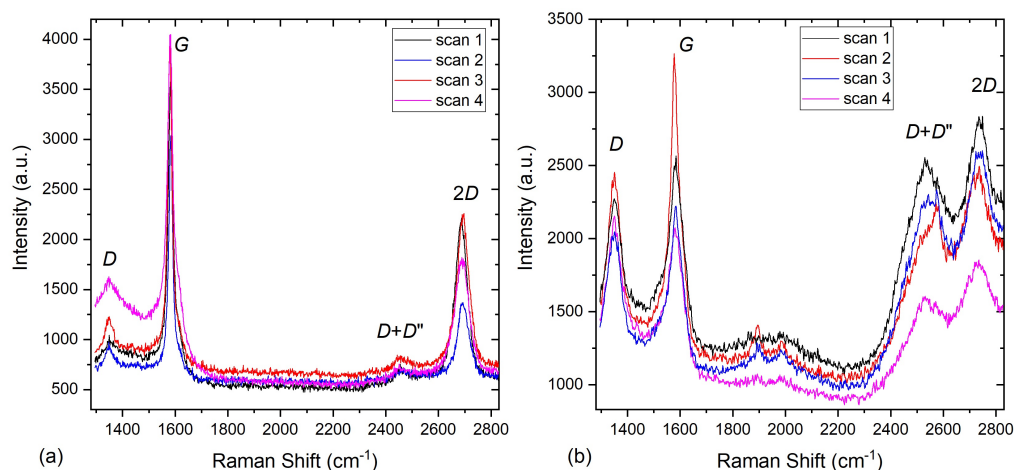


Figure 8. RT Raman spectroscopy measurements for (a) the bare graphene sheet on the Si/SiO₂ substrate and (b) graphene under the Eu₃O₄ film after etching the Au capping layer.

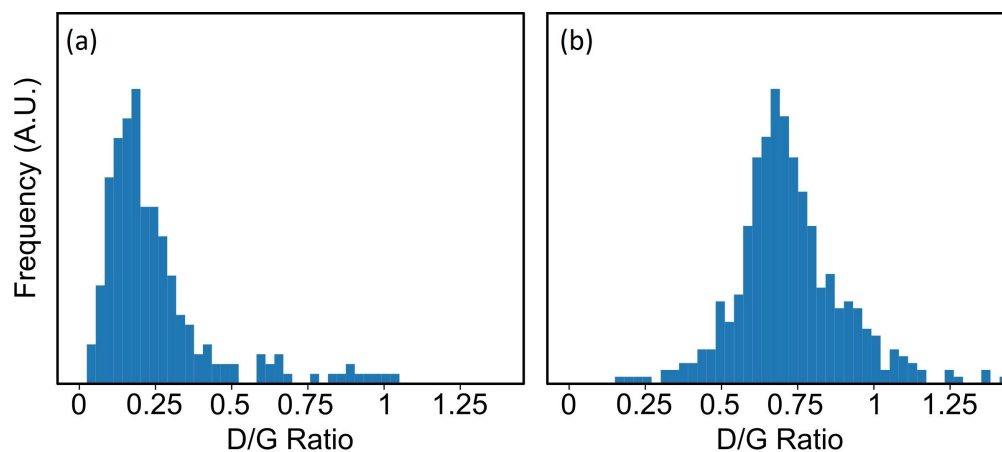


Figure 9. The frequency distribution of the ratio of the intensity of the D peak to that of the G peak of (a) the bare graphene and (b) the graphene under the Eu₃O₄ film. Raman spectra of the regions with no graphene signal were removed from the statistics.

4. Conclusions

In summary, we discussed the experimental work carried out to study the growth of Eu₃O₄ thin film on Si/SiO₂ and on a graphene sheet by MBE at 400°C. The structural and magnetic characterisations show successful deposition of crystalline, highly textured Eu₃O₄(001) films with a T_C of $\sim 5.5 \pm 0.1$ K and a magnetic moment of ~ 320 emu/cm³ at 2 K. However, the films show no metamagnetic behaviour which could be attributed to the strain from the substrate. Furthermore, a qualitative analysis of the depth-profile XPS scans confirms the mixed valency of the Eu cation with the Eu²⁺ : Eu³⁺ ratio of 28 : 72, which agrees with the values reported for other Eu-doped systems. Optical microscopy and Raman measurements show that the graphene layer remains largely intact after the Eu₃O₄ growth. Therefore, this study represents the first successful step towards integrating a Eu₃O₄ thin film in two widely used electronic substrates for future spintronics applications.

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Conflicts of Interest: The authors declare no conflict of interest.

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