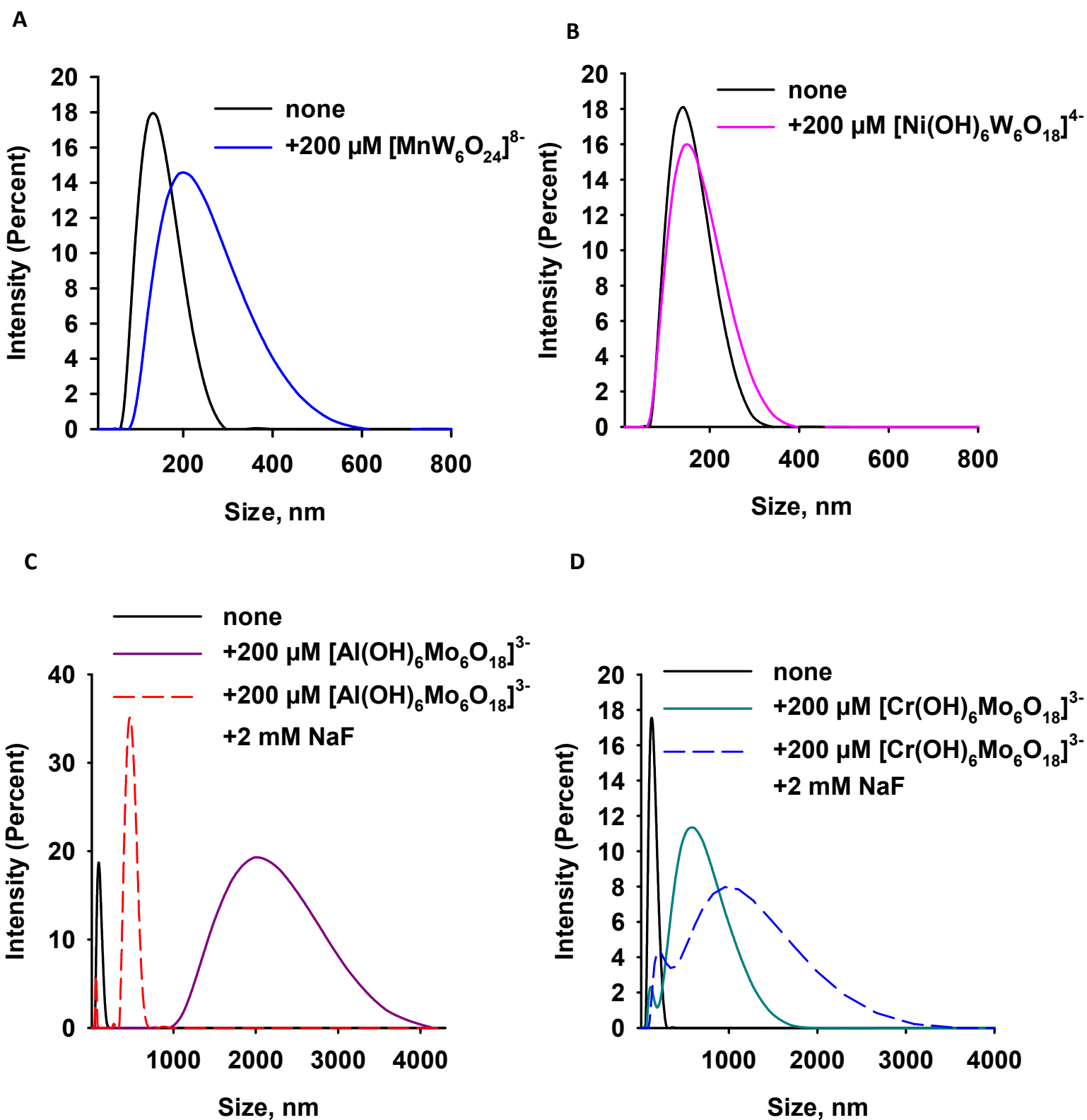


Molecular interactions at the interface: polyoxometalates of the Anderson-Evans type and lipid membranes

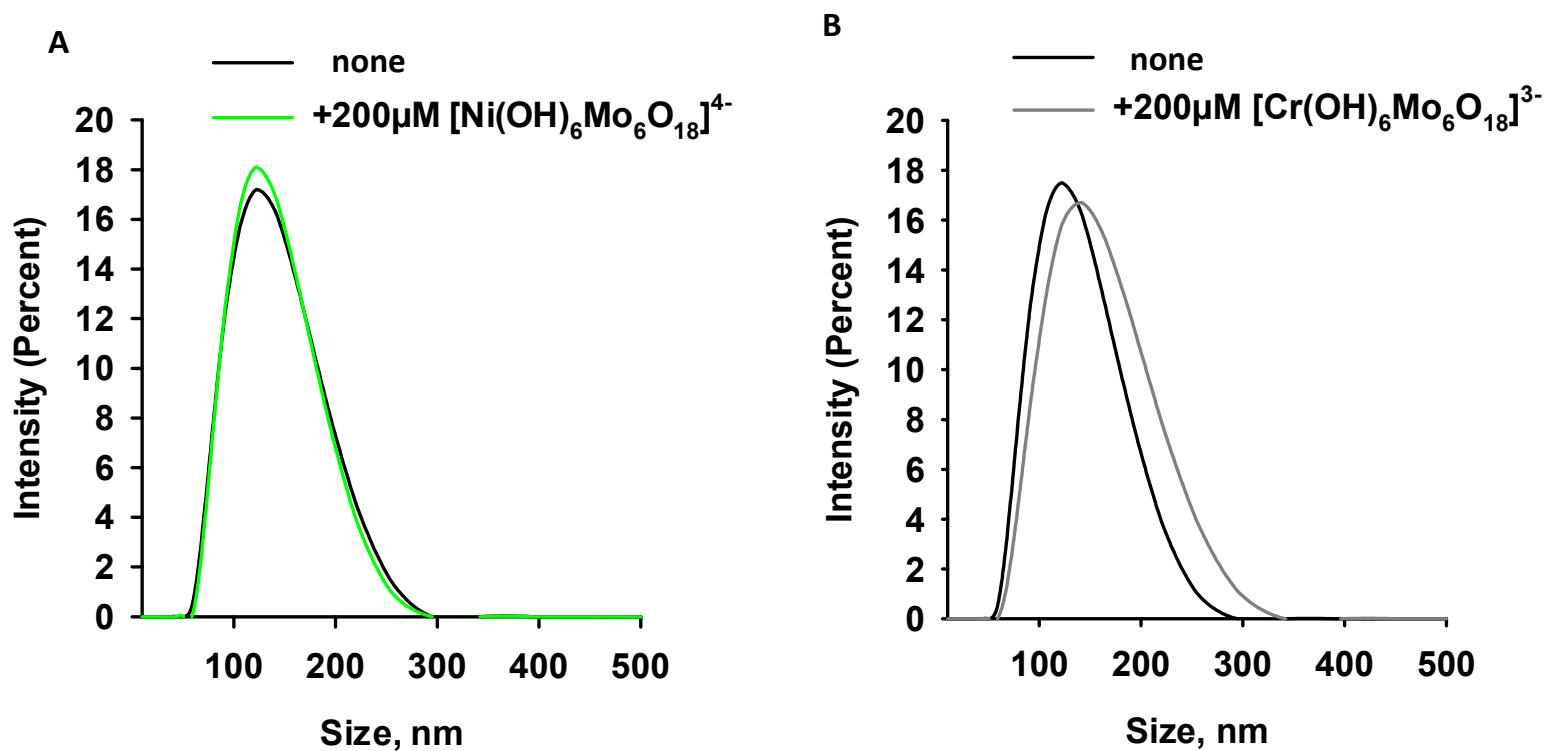
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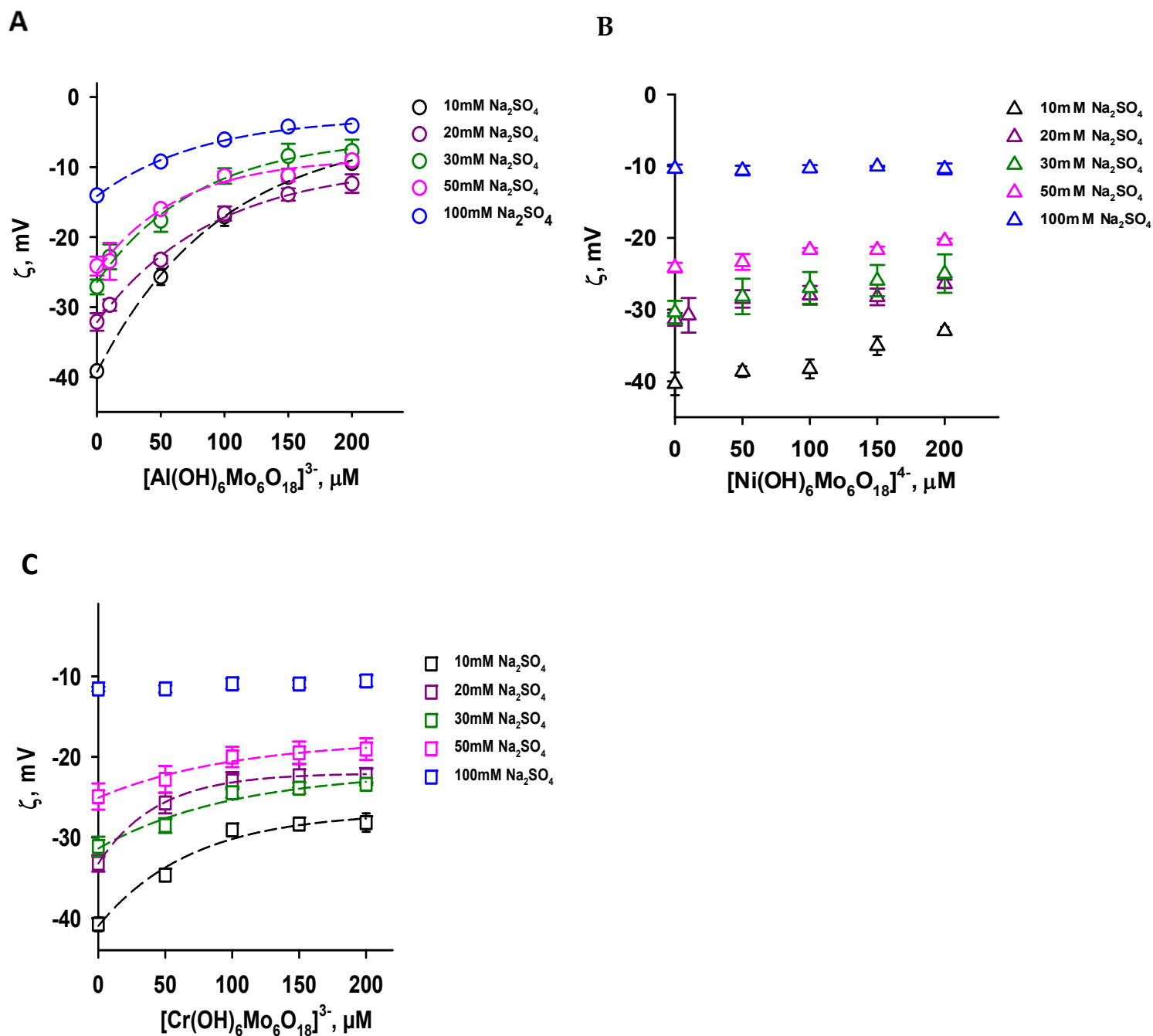
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Supplementary Figure 1. Size distribution of liposomes with and without 200 μM POMs. The liposomes were prepared from DOPC:DOPE (50:50%). The lipid concentration was 0.2 mg/ml. The buffer solution consisted of 20 mM Na_2SO_4 , 10 mM MES, 10 mM Tris at pH=7.34 and T=25°C.



Supplementary Figure 2. Size distribution of liposomes with and without 200 μ M POMos and without. The liposomes were prepared from DOPC:CL (90:10%). The lipid concentration was 0.2 mg/ml. The buffer solution consisted of 20 mM Na_2SO_4 , 10 mM MES, 10 mM Tris at pH=7.34 and T=25 $^\circ$ C.

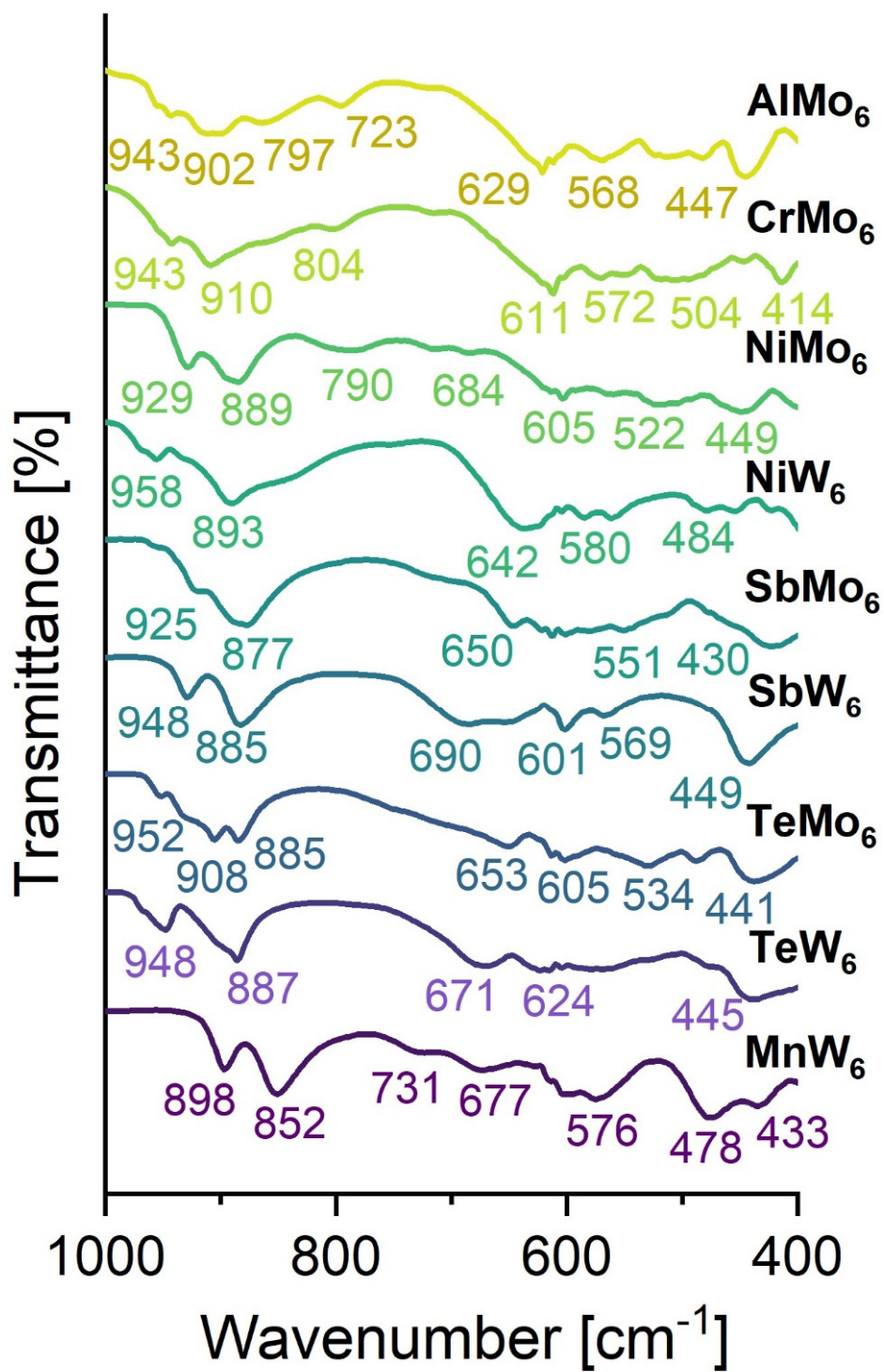


Supplementary Figure 3. Dependence of ζ potential of POMos with different central ions adsorbed on negatively charged liposomes:

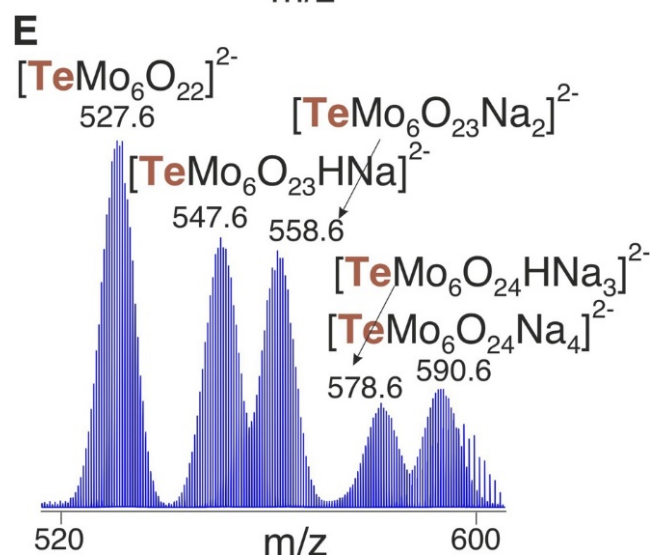
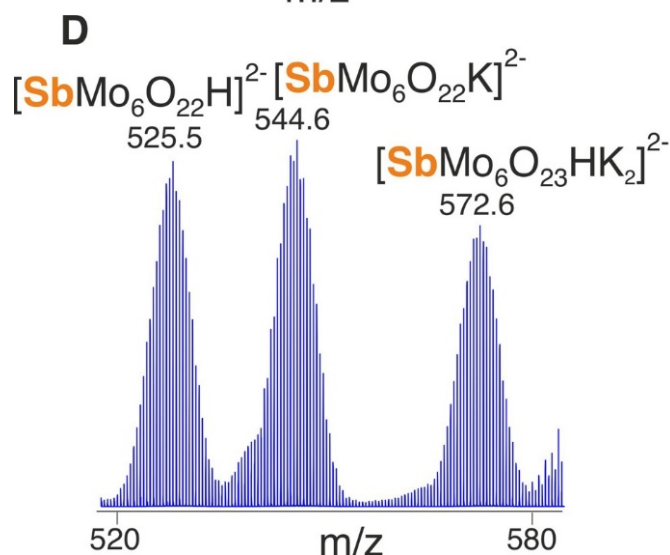
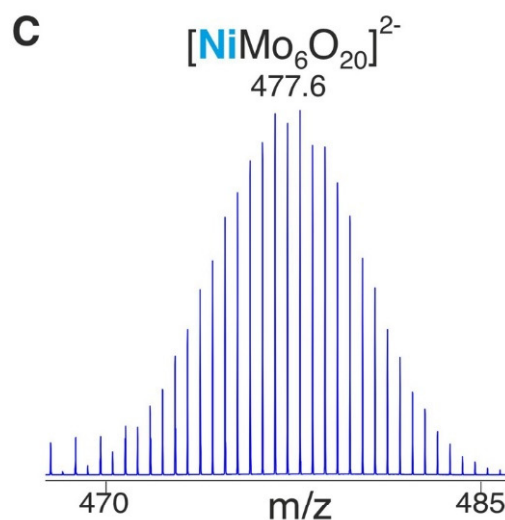
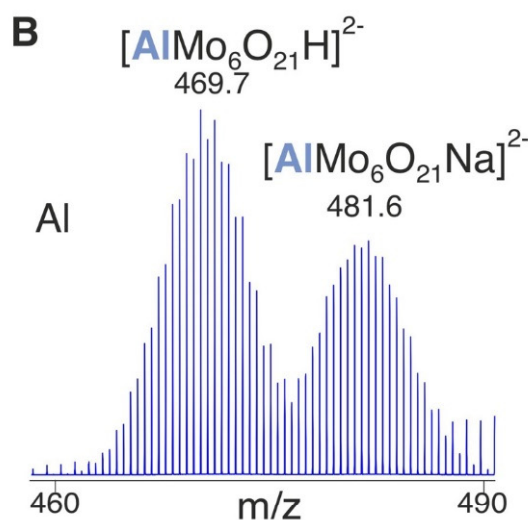
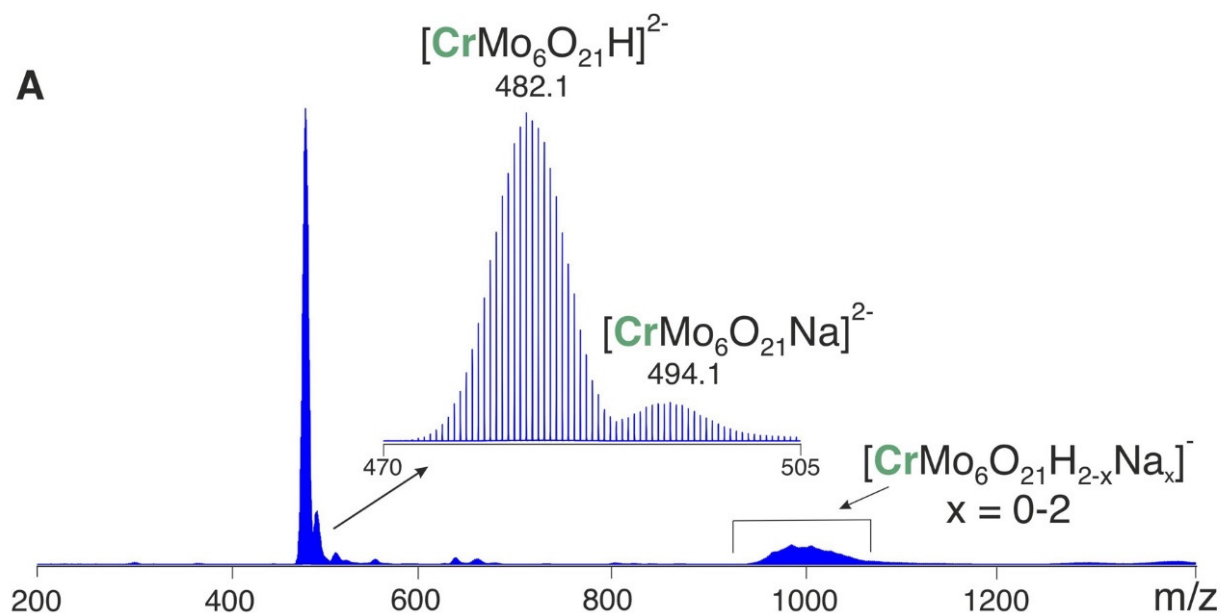
(A) $[\text{Al}(\text{OH})_6\text{Mo}_6\text{O}_{18}]^{3-}$, (B) $[\text{Ni}(\text{OH})_6\text{Mo}_6\text{O}_{18}]^{4-}$ and (C) $[\text{Cr}(\text{OH})_6\text{Mo}_6\text{O}_{18}]^{3-}$ at different ionic strength. The liposomes were prepared from DOPC:CL (90:10%) and the lipid concentration was 0.2 mg/ml. The buffer solutions consisted of 10, 20, 30, 50 and 100 mM Na_2SO_4 , 10 mM MES, 10 mM Tris at pH=7.34 and T=25 °C.

Supplementary Table 1. Characterization of the Anderson-Evans-type POMs.

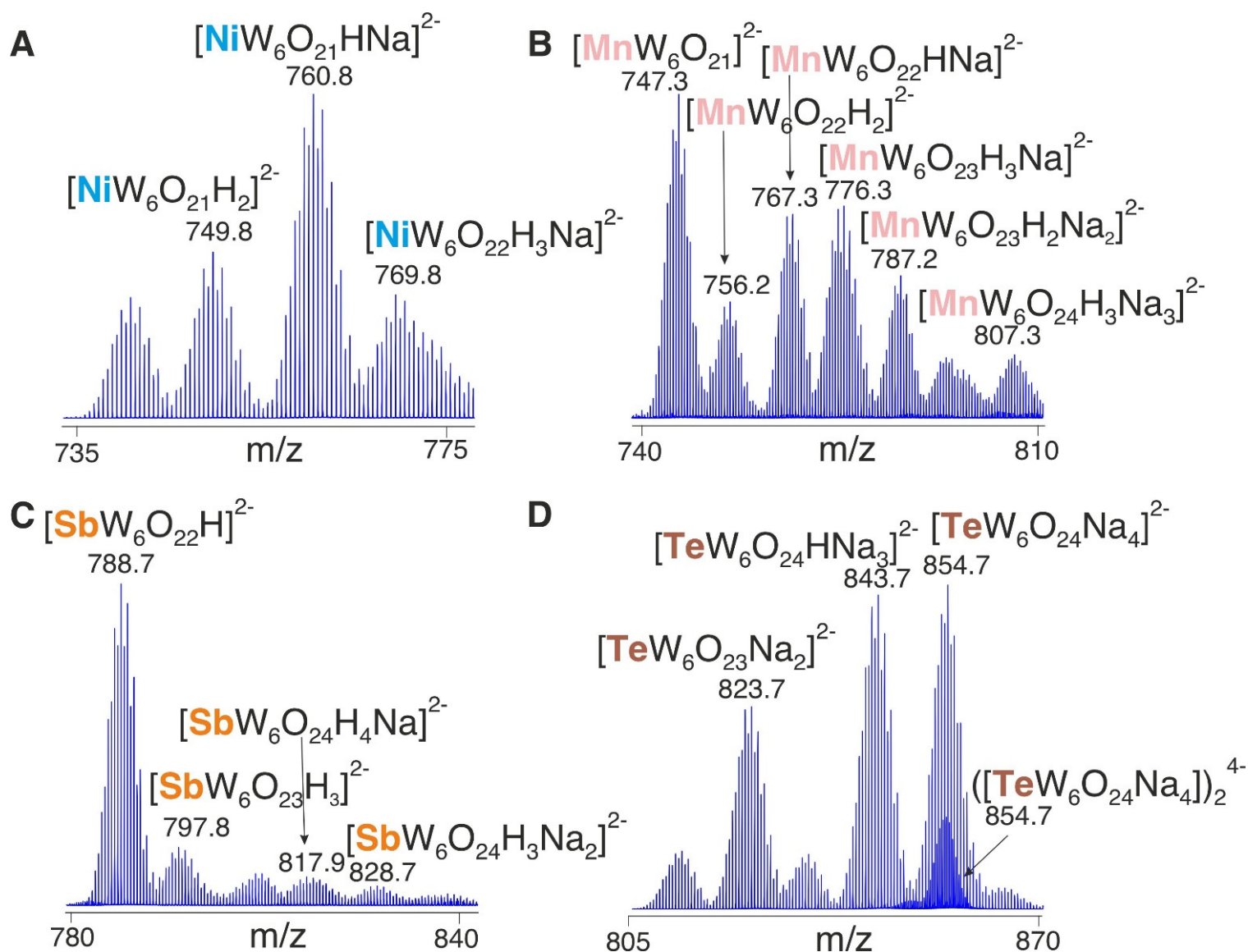
Formula	IR bands in the region 1000-400 cm^{-1} , (cm^{-1})	IR bands according to literature, (cm^{-1})	SXRD lattice constant measured at room temperature, ($\text{\AA}$, $^\circ$)	Reference
$\text{Na}_3[\text{Al}(\text{OH})_6\text{Mo}_6\text{O}_{18}]\cdot 8\text{H}_2\text{O}$	943, 902, 797, 723, 629, 568, 447	947, 845, 650, 574, 530, 447	$a = 6.4$, $b = 10.8$, $c = 10.9$; $\alpha = 110$, $\beta = 95$, $\gamma = 110$	Manikumari, Shivaiah et al. 2002
$\text{Na}_3[\text{Cr}(\text{OH})_6\text{Mo}_6\text{O}_{18}]\cdot 8\text{H}_2\text{O}$	943, 910, 804, 611, 572, 504, 414	not reported	$a = 10.9$, $b = 10.9$, $c = 6.5$, $\alpha = 108$, $\beta = 83$, $\gamma = 114$	Perloff 1970
$\text{Na}_4[\text{Ni}(\text{OH})_6\text{Mo}_6\text{O}_{18}]\cdot 16\text{H}_2\text{O}$	929, 889, 790, 684, 605, 522, 449	931, 882, 824, 706, 635, 583, 467	$a = 8.1$, $b = 10.6$, $c = 12.3$, $\alpha = 70$, $\beta = 72$, $\gamma = 83$	Gumerova, Melnik et al. 2015
$\text{Na}_4[\text{Ni}(\text{OH})_6\text{W}_6\text{O}_{18}]\cdot 16\text{H}_2\text{O}$	958, 893, 642, 580, 484	958, 863, 659, 585, 499	$a = 8.1$, $b = 10.7$, $c = 12.3$, $\alpha = 71$, $\beta = 72$, $\gamma = 85$	Rozantsev, Radio et al. 2009
$\text{K}_5[\text{H}_2\text{SbMo}_6\text{O}_{24}]\cdot 6\text{H}_2\text{O}$	925, 877, 650, 551, 430	927, 878, 650, 533, 426	$a = 21$, $b = 10.5$, $c = 15.3$, $\alpha = 90$, $\beta = 113$, $\gamma = 90$	Ogawa, Yamato et al. 1988
$\text{K}_5[\text{H}_2\text{SbW}_6\text{O}_{24}]\cdot 6\text{H}_2\text{O}$	948, 885, 690, 601, 559, 449	927, 850, 703, 632, 563, 420	$a = 13.8$, $b = 13.7$, $c = 12.9$, $\alpha = 90$, $\beta = 90$, $\gamma = 120$	Naruke and Yamase 1992
$\text{Na}_6[\text{TeMo}_6\text{O}_{24}]\cdot 22\text{H}_2\text{O}$	952, 908, 885, 653, 605, 534, 441	not reported	$a = 10.4$, $b = 10.7$, $c = 11.1$, $\alpha = 90$, $\beta = 116$, $\gamma = 105$	Robl and Frost 1993
$\text{Na}_6[\text{TeW}_6\text{O}_{24}]\cdot 22\text{H}_2\text{O}$	948, 887, 671, 624, 445	not reported	$a = 10.3$, $b = 10.7$, $c = 11.2$, $\alpha = 91$, $\beta = 115$, $\gamma = 106$	Schmidt, Schrobilgen et al. 1986
$\text{Na}_2\text{K}_6[\text{MnW}_6\text{O}_{24}]\cdot 12\text{H}_2\text{O}$	898, 852, 731, 677, 576, 478, 433	912, 863, 700	$a = 13.2$, $b = 13.2$, $c = 18.4$, $\alpha = 90$, $\beta = 90$, $\gamma = 120$	Nolan, Burns et al. 2000



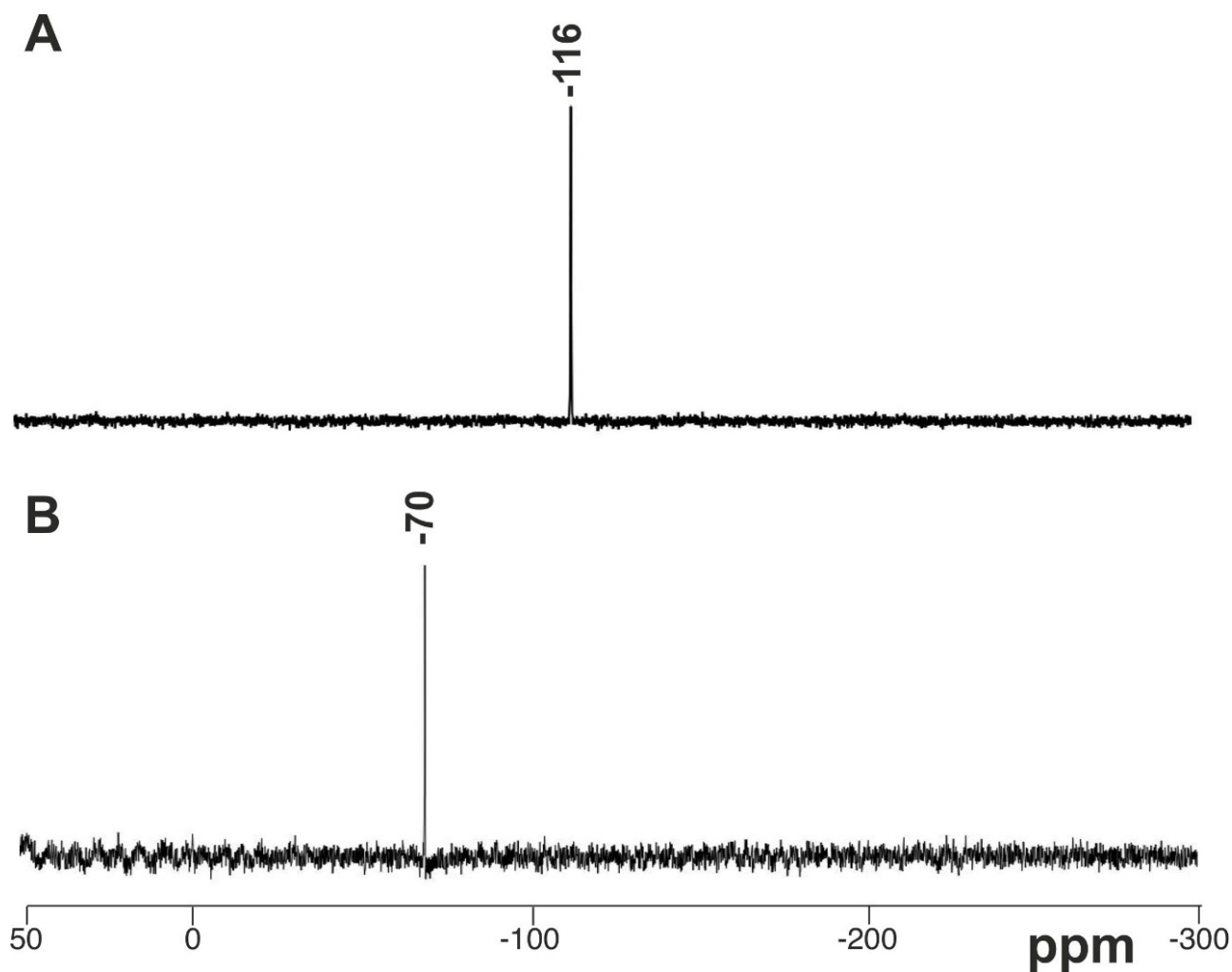
Supplementary Figure 4. IR spectra of Anderson POMs in the fingerprint region 1000 – 400 cm^{-1} .



Supplementary Figure 5. ESI mass spectra of 100 μM aqueous solutions of A) $\text{Na}_3[\text{Cr}(\text{OH})_6\text{Mo}_6\text{O}_{18}] \cdot 8\text{H}_2\text{O}$; B) $\text{Na}_3[\text{Al}(\text{OH})_6\text{Mo}_6\text{O}_{18}] \cdot 8\text{H}_2\text{O}$; C) $\text{Na}_4[\text{Ni}(\text{OH})_6\text{Mo}_6\text{O}_{18}] \cdot 16\text{H}_2\text{O}$; D) $\text{K}_5[\text{H}_2\text{SbMo}_6\text{O}_{24}] \cdot 7\text{H}_2\text{O}$; E) $\text{Na}_6[\text{TeMo}_6\text{O}_{24}] \cdot 22\text{H}_2\text{O}$. The spectra were recorded in negative ion mode within the m/z range of 100 to 1500, and the spectrometer was calibrated using the standard tune-mix to ensure an accuracy of approximately 5 ppm in the m/z region of 100–1500. For $\text{Na}_3[\text{Cr}(\text{OH})_6\text{Mo}_6\text{O}_{18}] \cdot 8\text{H}_2\text{O}$ the whole range is shown (A), for four other POMos just the region where the envelope for XMo_6 is visible.



Supplementary Figure 6. ESI mass spectra of 100 μM aqueous solutions of A) $\text{Na}_4[\text{Ni}(\text{OH})_6\text{W}_6\text{O}_{18}] \cdot 16\text{H}_2\text{O}$; B) $\text{Na}_2\text{K}_6[\text{MnW}_6\text{O}_{24}] \cdot 12\text{H}_2\text{O}$ B) $\text{K}_5[\text{H}_2\text{SbW}_6\text{O}_{24}] \cdot 6\text{H}_2\text{O}$; C) $\text{Na}_6[\text{TeW}_6\text{O}_{24}] \cdot 22\text{H}_2\text{O}$. The spectra were recorded in negative ion mode within the m/z range of 100 to 1500, and the spectrometer was calibrated using the standard tune-mix to ensure an accuracy of approximately 5 ppm in the m/z region of 100–1500. For all POTs, the regions displaying the envelope for the XW_6 species are shown.



Supplementary Figure 7. ^{183}W NMR spectra of 10 mM solutions of A) $\text{K}_5[\text{H}_2\text{SbW}_6\text{O}_{24}] \cdot 6\text{H}_2\text{O}$ and B) $\text{Na}_6[\text{TeW}_6\text{O}_{24}] \cdot 22\text{H}_2\text{O}$ in 20 mM Na_2SO_4 , 10 mM Tris, 10 mM MES, pH 7.34. The one signal at -70 ppm for SbW_6 and -116 ppm for TeW_6 corresponds to one type of W coordination in the Anderson structure and is in accordance with the literature (Chen, Gang et al., 2004).