

4-[6,8-Dibromo-2-(2-chloro-5-nitrophenyl)-1,2,3,4-tetrahydroquinazolin-3-yl]cyclohexanol

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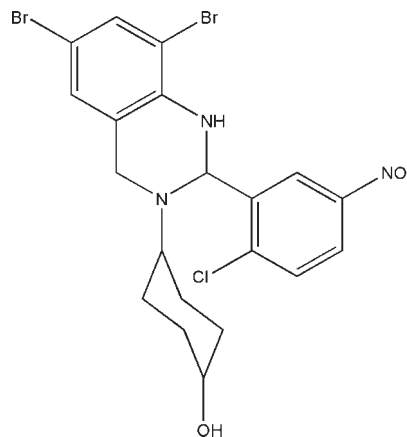
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.009$ Å; disorder in main residue; R factor = 0.055; wR factor = 0.154; data-to-parameter ratio = 14.7.

The title compound, $\text{C}_{20}\text{H}_{20}\text{Br}_2\text{ClN}_3\text{O}_3$, was synthesized by the condensation reaction of 2-chloro-5-nitrobenzaldehyde with 4-(2-amino-3,5-dibromobenzylamino)cyclohexanol in a methanol solution. There are two independent molecules in the asymmetric unit and in one molecule the atoms of the cyclohexane ring are disordered over two sets of sites with refined occupancies of 0.657 (12) and 0.343 (12). The dihedral angle between the two benzene rings is $89.5(2)^\circ$ in one molecule and $82.9(2)^\circ$ in the other. In the crystal structure, intermolecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into chains propagating along $[01\bar{1}]$.

Related literature

For details of the pharmaceutical uses of Ambroxol, systematic name 4-(2-amino-3,5-dibromobenzylamino)cyclohexanol, a compound closely related to the title compound see: Felix *et al.* (2008); Gaida *et al.* (2005); Lee *et al.* (2004). For a related structure, see: Wang *et al.* (2009). For standard bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{20}\text{Br}_2\text{ClN}_3\text{O}_3$
 $M_r = 545.66$
Triclinic, $P\bar{1}$
 $a = 10.2614(13)$ Å
 $b = 13.1418(17)$ Å
 $c = 16.931(2)$ Å
 $\alpha = 83.764(2)^\circ$
 $\beta = 73.309(2)^\circ$

$\gamma = 84.750(2)^\circ$
 $V = 2169.8(5)$ Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 3.89$ mm⁻¹
 $T = 298$ K
 $0.30 \times 0.28 \times 0.28$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.389$, $T_{\max} = 0.409$
11972 measured reflections
8470 independent reflections
4874 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.154$
 $S = 1.02$
8470 reflections
578 parameters

122 restraints
H-atom parameters constrained
 $\Delta\rho_{\max} = 1.56$ e Å⁻³
 $\Delta\rho_{\min} = -1.07$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|---|--------------|--------------------|-------------|----------------------|
| $\text{N3}-\text{H3N}\cdots\text{O4}^{\text{i}}$ | 0.86 | 2.33 | 2.942 (7) | 129 |
| $\text{O3}-\text{H3}\cdots\text{O6}^{\text{iii}}$ | 0.84 | 1.89 | 2.713 (12) | 166 |
| $\text{O6}-\text{H6}\cdots\text{O3}^{\text{iii}}$ | 0.84 | 2.04 | 2.877 (17) | 179 |
| $\text{O6}'-\text{H6}'\cdots\text{O3}^{\text{iii}}$ | 0.84 | 1.85 | 2.694 (10) | 179 |

Symmetry codes: (i) $-x + 1, -y, -z + 1$; (ii) $-x + 1, -y + 1, -z$; (iii) $x - 1, y, z$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5014).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Bruker (2002). *SAINT* and *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.

Felix, F. S., Brett, C. M. A. & Angnes, L. (2008). *Talanta*, **76**, 128–133.
Gaida, W., Klinder, K., Arndt, K. & Weiser, T. (2005). *Neuropharmacology*, **49**, 1220–1227.
Lee, H. J., Joung, S. K., Kim, Y. G., Yoo, J.-Y. & Han, S. B. (2004). *Pharm. Res.* **49**, 93–98.
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Wang, Z.-G., Wang, R., Zhang, Y., Zhi, F. & Yang, Y.-L. (2009). *Acta Cryst.* **E65**, o550.

supplementary materials

Acta Cryst. (2010). E66, o1205-o1206 [doi:10.1107/S1600536810015023]

4-[6,8-Dibromo-2-(2-chloro-5-nitrophenyl)-1,2,3,4-tetrahydroquinazolin-3-yl]cyclohexanol

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Comment

Ambroxol, 4-(2-amino-3,5-dibromobenzylamino)cyclohexanol, is an expectorant agent which leads to bronchial secretion due to its mucolytic properties (Felix *et al.*, 2008; Gaida *et al.*, 2005; Lee *et al.*, 2004). Recently, we reported the crystal structure of a derivative of Ambroxol (Wang *et al.*, 2009). In this paper, the crystal structure of the title compound, derived from the condensation reaction of 2-chloro-5-nitrobenzaldehyde with 4-(2-amino-3,5-dibromobenzylamino)cyclohexanol in methanol solution, is reported.

There are two independent molecules in the asymmetric unit of the title compound, Fig. 1. The dihedral angle between the two benzene rings is 89.5 (2)° in one molecule and 82.9 (2)° in the other. The cyclohexyl rings adopt chair configurations. All bond lengths are within normal ranges (Allen *et al.*, 1987).

Experimental

2-Chloro-5-nitrobenzaldehyde (1.0 mol, 185.6 mg) and 4-(2-amino-3,5-dibromobenzylamino)cyclohexanol (1.0 mmol, 378.1 mg) were dissolved in a methanol solution (30 ml). The mixture was stirred at room temperature to give a clear colorless solution. Crystals of the title compound were formed by gradual evaporation of the solvent for a week at room temperature.

Refinement

H atoms were included in calculated positions with, with C–H = 0.93–0.98 Å, N–H = 0.86 Å and O–H = 0.84 with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C,N})$ or $1.5U_{\text{eq}}(\text{O})$. The C29–C34 cyclohexyl ring is disordered over two distinct sites, with refined occupancies of 0.657 (12) and 0.343 (12). Bond length restraints were applied to the disorder model using the SADI command in SHELXL (Sheldrick, 2008).

Figures

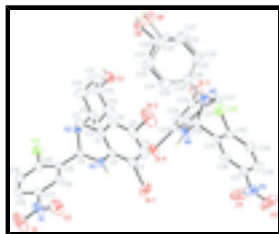


Fig. 1. The asymmetric unit of the title compound showing 30% probability ellipsoids. H atoms bonded to C atoms are not shown.

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Crystal data

| | |
|--------------------------------|---|
| $C_{20}H_{20}Br_2ClN_3O_3$ | $Z = 4$ |
| $M_r = 545.66$ | $F(000) = 1088$ |
| Triclinic, PT | $D_x = 1.670 \text{ Mg m}^{-3}$ |
| Hall symbol: $-P 1$ | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| $a = 10.2614 (13) \text{ \AA}$ | Cell parameters from 2646 reflections |
| $b = 13.1418 (17) \text{ \AA}$ | $\theta = 2.5\text{--}24.5^\circ$ |
| $c = 16.931 (2) \text{ \AA}$ | $\mu = 3.89 \text{ mm}^{-1}$ |
| $\alpha = 83.764 (2)^\circ$ | $T = 298 \text{ K}$ |
| $\beta = 73.309 (2)^\circ$ | Block, colorless |
| $\gamma = 84.750 (2)^\circ$ | $0.30 \times 0.28 \times 0.28 \text{ mm}$ |
| $V = 2169.8 (5) \text{ \AA}^3$ | |

Data collection

| | |
|---|--|
| Bruker SMART CCD area-detector diffractometer | 8470 independent reflections |
| Radiation source: fine-focus sealed tube graphite | 4874 reflections with $I > 2\sigma(I)$ |
| ω scans | $R_{\text{int}} = 0.022$ |
| Absorption correction: multi-scan (SADABS; Sheldrick, 1996) | $\theta_{\text{max}} = 26.2^\circ$, $\theta_{\text{min}} = 1.9^\circ$ |
| $T_{\text{min}} = 0.389$, $T_{\text{max}} = 0.409$ | $h = -12 \rightarrow 7$ |
| 11972 measured reflections | $k = -15 \rightarrow 16$ |
| | $l = -21 \rightarrow 20$ |

Refinement

| | |
|---------------------------------|--|
| Refinement on F^2 | Primary atom site location: structure-invariant direct methods |
| Least-squares matrix: full | Secondary atom site location: difference Fourier map |
| $R[F^2 > 2\sigma(F^2)] = 0.055$ | Hydrogen site location: inferred from neighbouring sites |
| $wR(F^2) = 0.154$ | H-atom parameters constrained |
| $S = 1.02$ | $w = 1/[\sigma^2(F_o^2) + (0.0631P)^2 + 2.7275P]$ |
| 8470 reflections | where $P = (F_o^2 + 2F_c^2)/3$ |
| 578 parameters | $(\Delta/\sigma)_{\text{max}} = 0.001$ |
| 122 restraints | $\Delta\rho_{\text{max}} = 1.56 \text{ e \AA}^{-3}$ |
| | $\Delta\rho_{\text{min}} = -1.06 \text{ e \AA}^{-3}$ |

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds

in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|------|---------------|---------------|--------------|----------------------------------|-----------|
| Br1 | 0.39058 (9) | -0.05133 (6) | 0.42693 (4) | 0.0888 (3) | |
| Br2 | -0.06018 (7) | 0.08773 (7) | 0.30819 (7) | 0.1156 (4) | |
| Br3 | 0.65212 (7) | 0.61194 (6) | 0.35191 (6) | 0.0919 (3) | |
| Br4 | 0.59092 (6) | 0.18627 (5) | 0.36046 (5) | 0.0747 (2) | |
| Cl1 | 0.87128 (15) | -0.09542 (13) | 0.01005 (10) | 0.0700 (4) | |
| Cl2 | -0.17164 (14) | 0.27481 (12) | 0.47738 (9) | 0.0628 (4) | |
| O1 | 0.4691 (6) | -0.4725 (4) | 0.1616 (4) | 0.133 (3) | |
| O2 | 0.3509 (5) | -0.3554 (4) | 0.2309 (4) | 0.0953 (16) | |
| O3 | 0.8184 (5) | 0.4244 (3) | 0.0419 (3) | 0.0816 (14) | |
| H3 | 0.8561 | 0.4573 | -0.0030 | 0.122* | |
| O4 | 0.2375 (7) | 0.1609 (5) | 0.7120 (4) | 0.119 (2) | |
| O5 | 0.0491 (7) | 0.1570 (6) | 0.8042 (4) | 0.145 (3) | |
| N1 | 0.4516 (6) | -0.3856 (4) | 0.1802 (4) | 0.0731 (15) | |
| N2 | 0.5854 (4) | 0.0384 (3) | 0.0978 (3) | 0.0435 (10) | |
| N3 | 0.5434 (4) | -0.0321 (3) | 0.2395 (2) | 0.0472 (11) | |
| H3N | 0.5793 | -0.0464 | 0.2797 | 0.057* | |
| N4 | 0.1168 (8) | 0.1719 (5) | 0.7331 (4) | 0.0849 (18) | |
| N5 | 0.1019 (4) | 0.3814 (3) | 0.4122 (2) | 0.0425 (10) | |
| N6 | 0.2870 (4) | 0.2572 (3) | 0.4237 (3) | 0.0557 (12) | |
| H6N | 0.3209 | 0.1949 | 0.4251 | 0.067* | |
| C1 | 0.3494 (5) | 0.0295 (4) | 0.1906 (3) | 0.0469 (13) | |
| C2 | 0.4071 (6) | -0.0031 (4) | 0.2565 (3) | 0.0482 (13) | |
| C3 | 0.3189 (7) | -0.0040 (4) | 0.3370 (3) | 0.0579 (15) | |
| C4 | 0.1815 (7) | 0.0235 (4) | 0.3527 (4) | 0.0704 (19) | |
| H4 | 0.1251 | 0.0217 | 0.4067 | 0.085* | |
| C5 | 0.1303 (6) | 0.0533 (5) | 0.2873 (5) | 0.0723 (19) | |
| C6 | 0.2116 (6) | 0.0578 (4) | 0.2078 (4) | 0.0598 (16) | |
| H6B | 0.1740 | 0.0801 | 0.1646 | 0.072* | |
| C7 | 0.4418 (5) | 0.0291 (4) | 0.1042 (3) | 0.0483 (13) | |
| H7A | 0.4333 | -0.0342 | 0.0819 | 0.058* | |
| H7B | 0.4119 | 0.0854 | 0.0702 | 0.058* | |
| C8 | 0.6281 (5) | -0.0394 (4) | 0.1557 (3) | 0.0407 (12) | |
| H8 | 0.7206 | -0.0254 | 0.1549 | 0.049* | |
| C9 | 0.6166 (6) | 0.1452 (4) | 0.1032 (4) | 0.0569 (15) | |
| H9 | 0.5423 | 0.1758 | 0.1462 | 0.068* | |
| C10 | 0.6231 (9) | 0.2061 (5) | 0.0216 (5) | 0.097 (3) | |
| H10A | 0.5357 | 0.2064 | 0.0102 | 0.116* | |
| H10B | 0.6912 | 0.1731 | -0.0221 | 0.116* | |

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|------|--------------|-------------|------------|-------------|------------|
| C11 | 0.6590 (10) | 0.3179 (6) | 0.0215 (6) | 0.111 (3) | |
| H11A | 0.6649 | 0.3539 | -0.0324 | 0.133* | |
| H11B | 0.5873 | 0.3528 | 0.0622 | 0.133* | |
| C12 | 0.7856 (8) | 0.3205 (5) | 0.0402 (4) | 0.078 (2) | |
| H12 | 0.8577 | 0.2879 | -0.0027 | 0.093* | |
| C13 | 0.7829 (8) | 0.2642 (5) | 0.1224 (5) | 0.086 (2) | |
| H13A | 0.7152 | 0.2978 | 0.1659 | 0.103* | |
| H13B | 0.8711 | 0.2656 | 0.1325 | 0.103* | |
| C14 | 0.7484 (7) | 0.1518 (5) | 0.1241 (4) | 0.0737 (19) | |
| H14A | 0.8216 | 0.1165 | 0.0847 | 0.088* | |
| H14B | 0.7417 | 0.1176 | 0.1787 | 0.088* | |
| C15 | 0.6360 (5) | -0.1458 (4) | 0.1260 (3) | 0.0410 (12) | |
| C16 | 0.5412 (5) | -0.2159 (4) | 0.1637 (3) | 0.0465 (13) | |
| H16 | 0.4664 | -0.1973 | 0.2069 | 0.056* | |
| C17 | 0.5551 (6) | -0.3135 (4) | 0.1386 (3) | 0.0511 (13) | |
| C18 | 0.6655 (6) | -0.3457 (5) | 0.0763 (4) | 0.0662 (17) | |
| H18 | 0.6752 | -0.4123 | 0.0607 | 0.079* | |
| C19 | 0.7607 (6) | -0.2758 (5) | 0.0380 (4) | 0.0614 (16) | |
| H19 | 0.8361 | -0.2950 | -0.0045 | 0.074* | |
| C20 | 0.7458 (5) | -0.1784 (4) | 0.0618 (3) | 0.0487 (13) | |
| C21 | 0.3193 (5) | 0.4363 (4) | 0.4205 (3) | 0.0445 (12) | |
| C22 | 0.3731 (5) | 0.3365 (4) | 0.4085 (3) | 0.0443 (12) | |
| C23 | 0.5137 (5) | 0.3210 (4) | 0.3792 (3) | 0.0481 (13) | |
| C24 | 0.5984 (6) | 0.4014 (5) | 0.3633 (3) | 0.0586 (16) | |
| H24 | 0.6925 | 0.3895 | 0.3440 | 0.070* | |
| C25 | 0.5415 (6) | 0.4989 (5) | 0.3765 (4) | 0.0595 (15) | |
| C26 | 0.4024 (6) | 0.5159 (4) | 0.4057 (3) | 0.0539 (14) | |
| H26 | 0.3647 | 0.5820 | 0.4155 | 0.065* | |
| C27 | 0.1651 (5) | 0.4519 (4) | 0.4489 (3) | 0.0449 (12) | |
| H27A | 0.1338 | 0.4407 | 0.5087 | 0.054* | |
| H27B | 0.1378 | 0.5220 | 0.4331 | 0.054* | |
| C28 | 0.1395 (5) | 0.2766 (4) | 0.4373 (3) | 0.0463 (13) | |
| H28 | 0.1110 | 0.2319 | 0.4034 | 0.056* | |
| O6 | 0.0678 (19) | 0.4493 (13) | 0.0811 (8) | 0.057 (4) | 0.343 (12) |
| H6 | -0.0051 | 0.4425 | 0.0696 | 0.085* | 0.343 (12) |
| C29 | 0.1191 (7) | 0.4085 (5) | 0.3235 (3) | 0.079 (2) | 0.343 (12) |
| H29A | 0.1792 | 0.4634 | 0.3210 | 0.095* | 0.343 (12) |
| C30 | 0.0096 (19) | 0.4808 (16) | 0.3083 (8) | 0.052 (8) | 0.343 (12) |
| H30A | 0.0273 | 0.5481 | 0.3197 | 0.062* | 0.343 (12) |
| H30B | -0.0748 | 0.4616 | 0.3485 | 0.062* | 0.343 (12) |
| C31 | -0.0137 (17) | 0.4906 (19) | 0.2262 (8) | 0.067 (6) | 0.343 (12) |
| H31A | -0.0412 | 0.5617 | 0.2137 | 0.080* | 0.343 (12) |
| H31B | -0.0896 | 0.4502 | 0.2290 | 0.080* | 0.343 (12) |
| C32 | 0.1012 (18) | 0.4595 (13) | 0.1567 (7) | 0.065 (6) | 0.343 (12) |
| H32A | 0.1618 | 0.5162 | 0.1437 | 0.078* | 0.343 (12) |
| C33 | 0.1854 (18) | 0.3684 (13) | 0.1758 (7) | 0.053 (6) | 0.343 (12) |
| H33A | 0.1398 | 0.3080 | 0.1731 | 0.063* | 0.343 (12) |
| H33B | 0.2713 | 0.3671 | 0.1327 | 0.063* | 0.343 (12) |
| C34 | 0.2156 (15) | 0.3599 (13) | 0.2562 (6) | 0.057 (5) | 0.343 (12) |

| | | | | | |
|------|-------------|-------------|------------|-------------|------------|
| H34A | 0.2257 | 0.2876 | 0.2737 | 0.068* | 0.343 (12) |
| H34B | 0.3032 | 0.3881 | 0.2476 | 0.068* | 0.343 (12) |
| O6' | 0.0198 (12) | 0.4810 (10) | 0.0975 (6) | 0.095 (4) | 0.657 (12) |
| H6' | -0.0436 | 0.4642 | 0.0802 | 0.142* | 0.657 (12) |
| C29' | 0.1191 (7) | 0.4085 (5) | 0.3235 (3) | 0.079 (2) | 0.657 (12) |
| H29B | 0.2138 | 0.4277 | 0.3026 | 0.095* | 0.657 (12) |
| C30' | 0.0385 (12) | 0.5058 (7) | 0.3093 (5) | 0.049 (3) | 0.657 (12) |
| H30C | 0.0738 | 0.5610 | 0.3292 | 0.059* | 0.657 (12) |
| H30D | -0.0552 | 0.4992 | 0.3425 | 0.059* | 0.657 (12) |
| C31' | 0.0389 (12) | 0.5356 (7) | 0.2214 (5) | 0.067 (3) | 0.657 (12) |
| H31C | -0.0314 | 0.5897 | 0.2198 | 0.081* | 0.657 (12) |
| H31D | 0.1262 | 0.5619 | 0.1908 | 0.081* | 0.657 (12) |
| C32' | 0.0143 (13) | 0.4461 (8) | 0.1814 (6) | 0.081 (4) | 0.657 (12) |
| H32B | -0.0764 | 0.4224 | 0.2101 | 0.098* | 0.657 (12) |
| C33' | 0.1193 (15) | 0.3608 (7) | 0.1855 (5) | 0.077 (4) | 0.657 (12) |
| H33C | 0.2091 | 0.3839 | 0.1565 | 0.092* | 0.657 (12) |
| H33D | 0.1037 | 0.3037 | 0.1583 | 0.092* | 0.657 (12) |
| C34' | 0.1142 (13) | 0.3259 (6) | 0.2735 (5) | 0.072 (3) | 0.657 (12) |
| H34C | 0.0309 | 0.2912 | 0.2989 | 0.087* | 0.657 (12) |
| H34D | 0.1902 | 0.2764 | 0.2739 | 0.087* | 0.657 (12) |
| C35 | 0.0602 (5) | 0.2507 (4) | 0.5269 (3) | 0.0420 (12) | |
| C36 | 0.1226 (6) | 0.2241 (4) | 0.5892 (4) | 0.0519 (14) | |
| H36 | 0.2172 | 0.2205 | 0.5769 | 0.062* | |
| C37 | 0.0451 (7) | 0.2028 (4) | 0.6695 (4) | 0.0570 (15) | |
| C38 | -0.0942 (7) | 0.2086 (4) | 0.6919 (4) | 0.0617 (16) | |
| H38 | -0.1444 | 0.1964 | 0.7469 | 0.074* | |
| C39 | -0.1569 (6) | 0.2328 (4) | 0.6310 (3) | 0.0516 (14) | |
| H39 | -0.2515 | 0.2360 | 0.6442 | 0.062* | |
| C40 | -0.0820 (5) | 0.2523 (4) | 0.5505 (3) | 0.0435 (12) | |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|------------|-------------|-------------|--------------|-------------|-------------|
| Br1 | 0.1211 (7) | 0.0938 (6) | 0.0491 (4) | -0.0380 (5) | -0.0139 (4) | 0.0028 (3) |
| Br2 | 0.0488 (4) | 0.0931 (6) | 0.1853 (10) | 0.0068 (4) | 0.0052 (5) | -0.0404 (6) |
| Br3 | 0.0649 (4) | 0.0829 (5) | 0.1281 (7) | -0.0302 (4) | -0.0261 (4) | 0.0094 (5) |
| Br4 | 0.0500 (4) | 0.0687 (4) | 0.0959 (5) | 0.0196 (3) | -0.0086 (3) | -0.0173 (4) |
| Cl1 | 0.0529 (9) | 0.0799 (11) | 0.0661 (10) | -0.0119 (8) | 0.0041 (8) | -0.0084 (8) |
| Cl2 | 0.0496 (8) | 0.0840 (11) | 0.0619 (9) | -0.0129 (7) | -0.0264 (7) | -0.0010 (8) |
| O1 | 0.120 (5) | 0.060 (3) | 0.193 (7) | -0.028 (3) | 0.019 (4) | -0.048 (4) |
| O2 | 0.073 (3) | 0.074 (3) | 0.120 (4) | -0.024 (3) | 0.013 (3) | -0.020 (3) |
| O3 | 0.106 (4) | 0.048 (3) | 0.083 (3) | -0.029 (2) | -0.010 (3) | 0.002 (2) |
| O4 | 0.096 (4) | 0.143 (5) | 0.134 (5) | -0.020 (4) | -0.076 (4) | 0.052 (4) |
| O5 | 0.147 (6) | 0.219 (8) | 0.072 (4) | -0.013 (5) | -0.056 (4) | 0.040 (4) |
| N1 | 0.073 (4) | 0.048 (3) | 0.094 (4) | -0.014 (3) | -0.011 (3) | -0.012 (3) |
| N2 | 0.046 (2) | 0.038 (2) | 0.048 (3) | -0.0040 (19) | -0.015 (2) | -0.004 (2) |
| N3 | 0.055 (3) | 0.051 (3) | 0.037 (2) | 0.004 (2) | -0.017 (2) | -0.006 (2) |
| N4 | 0.099 (5) | 0.081 (4) | 0.085 (5) | -0.012 (4) | -0.052 (4) | 0.022 (3) |

supplementary materials

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| N5 | 0.040 (2) | 0.045 (3) | 0.042 (2) | 0.0008 (19) | -0.0101 (19) | -0.0040 (19) |
| N6 | 0.042 (3) | 0.038 (3) | 0.079 (3) | 0.006 (2) | -0.005 (2) | -0.010 (2) |
| C1 | 0.047 (3) | 0.036 (3) | 0.057 (3) | -0.001 (2) | -0.014 (3) | -0.009 (3) |
| C2 | 0.059 (3) | 0.031 (3) | 0.053 (3) | -0.008 (2) | -0.008 (3) | -0.010 (2) |
| C3 | 0.076 (4) | 0.040 (3) | 0.050 (3) | -0.014 (3) | -0.002 (3) | -0.006 (3) |
| C4 | 0.068 (4) | 0.050 (4) | 0.072 (5) | -0.015 (3) | 0.023 (4) | -0.021 (3) |
| C5 | 0.049 (4) | 0.055 (4) | 0.101 (6) | 0.001 (3) | 0.003 (4) | -0.022 (4) |
| C6 | 0.052 (3) | 0.044 (3) | 0.084 (5) | 0.002 (3) | -0.015 (3) | -0.016 (3) |
| C7 | 0.049 (3) | 0.043 (3) | 0.056 (3) | 0.002 (2) | -0.021 (3) | -0.004 (3) |
| C8 | 0.040 (3) | 0.040 (3) | 0.044 (3) | -0.001 (2) | -0.016 (2) | -0.003 (2) |
| C9 | 0.063 (4) | 0.048 (3) | 0.063 (4) | -0.005 (3) | -0.023 (3) | -0.001 (3) |
| C10 | 0.142 (7) | 0.068 (5) | 0.106 (6) | -0.036 (5) | -0.080 (6) | 0.023 (4) |
| C11 | 0.148 (8) | 0.072 (5) | 0.141 (8) | -0.035 (5) | -0.094 (7) | 0.039 (5) |
| C12 | 0.099 (5) | 0.060 (4) | 0.070 (4) | -0.031 (4) | -0.012 (4) | 0.005 (3) |
| C13 | 0.110 (6) | 0.073 (5) | 0.088 (5) | -0.035 (4) | -0.045 (5) | 0.004 (4) |
| C14 | 0.079 (5) | 0.067 (4) | 0.082 (5) | -0.022 (4) | -0.037 (4) | 0.014 (4) |
| C15 | 0.043 (3) | 0.041 (3) | 0.040 (3) | 0.001 (2) | -0.014 (2) | -0.001 (2) |
| C16 | 0.047 (3) | 0.044 (3) | 0.048 (3) | 0.001 (2) | -0.013 (3) | -0.005 (2) |
| C17 | 0.052 (3) | 0.041 (3) | 0.059 (4) | -0.002 (3) | -0.014 (3) | -0.006 (3) |
| C18 | 0.070 (4) | 0.047 (4) | 0.078 (4) | 0.007 (3) | -0.012 (4) | -0.021 (3) |
| C19 | 0.056 (4) | 0.065 (4) | 0.052 (4) | 0.003 (3) | 0.004 (3) | -0.018 (3) |
| C20 | 0.043 (3) | 0.051 (3) | 0.048 (3) | 0.000 (2) | -0.008 (3) | -0.004 (3) |
| C21 | 0.044 (3) | 0.042 (3) | 0.046 (3) | 0.002 (2) | -0.014 (2) | 0.000 (2) |
| C22 | 0.045 (3) | 0.042 (3) | 0.044 (3) | 0.003 (2) | -0.013 (2) | 0.001 (2) |
| C23 | 0.041 (3) | 0.056 (3) | 0.044 (3) | 0.004 (3) | -0.010 (2) | -0.003 (3) |
| C24 | 0.036 (3) | 0.081 (5) | 0.058 (4) | -0.001 (3) | -0.016 (3) | 0.003 (3) |
| C25 | 0.049 (3) | 0.062 (4) | 0.066 (4) | -0.012 (3) | -0.016 (3) | 0.004 (3) |
| C26 | 0.053 (3) | 0.050 (3) | 0.058 (4) | -0.005 (3) | -0.014 (3) | -0.003 (3) |
| C27 | 0.041 (3) | 0.036 (3) | 0.052 (3) | -0.001 (2) | -0.006 (3) | -0.004 (2) |
| C28 | 0.042 (3) | 0.043 (3) | 0.054 (3) | -0.002 (2) | -0.013 (3) | -0.007 (3) |
| O6 | 0.080 (8) | 0.053 (7) | 0.038 (7) | 0.013 (6) | -0.021 (6) | -0.010 (5) |
| C29 | 0.095 (5) | 0.086 (5) | 0.056 (4) | 0.039 (4) | -0.028 (4) | -0.021 (4) |
| C30 | 0.055 (10) | 0.053 (10) | 0.045 (10) | 0.000 (8) | -0.011 (7) | -0.007 (7) |
| C31 | 0.072 (10) | 0.074 (10) | 0.056 (9) | 0.013 (8) | -0.027 (8) | -0.002 (8) |
| C32 | 0.099 (18) | 0.061 (13) | 0.038 (11) | 0.000 (12) | -0.025 (12) | -0.005 (9) |
| C33 | 0.040 (11) | 0.080 (14) | 0.038 (10) | -0.002 (9) | -0.010 (8) | -0.009 (9) |
| C34 | 0.045 (10) | 0.070 (12) | 0.053 (11) | 0.010 (9) | -0.013 (9) | -0.009 (9) |
| O6' | 0.114 (7) | 0.125 (8) | 0.053 (5) | -0.037 (6) | -0.030 (5) | 0.003 (5) |
| C29' | 0.095 (5) | 0.086 (5) | 0.056 (4) | 0.039 (4) | -0.028 (4) | -0.021 (4) |
| C30' | 0.060 (6) | 0.043 (6) | 0.050 (6) | -0.006 (5) | -0.021 (5) | -0.009 (4) |
| C31' | 0.066 (6) | 0.076 (6) | 0.063 (6) | -0.006 (5) | -0.027 (5) | 0.004 (5) |
| C32' | 0.079 (9) | 0.139 (12) | 0.029 (6) | -0.050 (9) | -0.015 (6) | 0.020 (6) |
| C33' | 0.103 (9) | 0.077 (7) | 0.047 (6) | -0.022 (6) | -0.009 (6) | -0.011 (5) |
| C34' | 0.092 (7) | 0.061 (6) | 0.067 (6) | -0.012 (5) | -0.027 (5) | 0.003 (5) |
| C35 | 0.042 (3) | 0.036 (3) | 0.050 (3) | 0.002 (2) | -0.017 (3) | -0.009 (2) |
| C36 | 0.046 (3) | 0.041 (3) | 0.069 (4) | 0.003 (2) | -0.022 (3) | 0.001 (3) |
| C37 | 0.075 (4) | 0.046 (3) | 0.058 (4) | -0.002 (3) | -0.036 (3) | 0.007 (3) |
| C38 | 0.075 (4) | 0.050 (4) | 0.055 (4) | -0.005 (3) | -0.013 (3) | 0.004 (3) |
| C39 | 0.049 (3) | 0.054 (3) | 0.050 (3) | -0.002 (3) | -0.014 (3) | 0.000 (3) |

C40 0.044 (3) 0.040 (3) 0.051 (3) -0.001 (2) -0.020 (3) -0.005 (2)

Geometric parameters (Å, °)

| | | | |
|----------|-----------|-----------|------------|
| Br1—C3 | 1.897 (6) | C18—H18 | 0.9300 |
| Br2—C5 | 1.904 (6) | C19—C20 | 1.365 (8) |
| Br3—C25 | 1.890 (6) | C19—H19 | 0.9300 |
| Br4—C23 | 1.893 (5) | C21—C26 | 1.367 (7) |
| Cl1—C20 | 1.737 (5) | C21—C22 | 1.387 (7) |
| Cl2—C40 | 1.729 (5) | C21—C27 | 1.517 (7) |
| O1—N1 | 1.201 (6) | C22—C23 | 1.388 (7) |
| O2—N1 | 1.204 (7) | C23—C24 | 1.384 (8) |
| O3—C12 | 1.441 (7) | C24—C25 | 1.373 (8) |
| O3—H3 | 0.8401 | C24—H24 | 0.9300 |
| O4—N4 | 1.186 (8) | C25—C26 | 1.375 (8) |
| O5—N4 | 1.210 (8) | C26—H26 | 0.9300 |
| N1—C17 | 1.460 (7) | C27—H27A | 0.9700 |
| N2—C7 | 1.461 (6) | C27—H27B | 0.9700 |
| N2—C8 | 1.464 (6) | C28—C35 | 1.519 (7) |
| N2—C9 | 1.487 (7) | C28—H28 | 0.9800 |
| N3—C2 | 1.371 (7) | O6—C32 | 1.439 (10) |
| N3—C8 | 1.444 (6) | O6—H6 | 0.8401 |
| N3—H3N | 0.8600 | O6—H6' | 1.1466 |
| N4—C37 | 1.472 (8) | C29—C34 | 1.442 (8) |
| N5—C28 | 1.448 (6) | C29—C30 | 1.466 (8) |
| N5—C27 | 1.460 (6) | C29—H29A | 0.9800 |
| N5—C29 | 1.470 (7) | C30—C31 | 1.468 (9) |
| N6—C22 | 1.383 (6) | C30—H30A | 0.9700 |
| N6—C28 | 1.467 (6) | C30—H30B | 0.9700 |
| N6—H6N | 0.8600 | C31—C32 | 1.469 (9) |
| C1—C6 | 1.383 (7) | C31—H31A | 0.9700 |
| C1—C2 | 1.416 (7) | C31—H31B | 0.9700 |
| C1—C7 | 1.499 (7) | C32—C33 | 1.474 (9) |
| C2—C3 | 1.403 (8) | C32—H32A | 0.9800 |
| C3—C4 | 1.379 (9) | C33—C34 | 1.470 (9) |
| C4—C5 | 1.364 (9) | C33—H33A | 0.9700 |
| C4—H4 | 0.9300 | C33—H33B | 0.9700 |
| C5—C6 | 1.363 (9) | C34—H34A | 0.9700 |
| C6—H6B | 0.9300 | C34—H34B | 0.9700 |
| C7—H7A | 0.9700 | O6'—C32' | 1.432 (12) |
| C7—H7B | 0.9700 | O6'—H6 | 0.8290 |
| C8—C15 | 1.525 (7) | O6'—H6' | 0.8401 |
| C8—H8 | 0.9800 | C30'—C31' | 1.496 (7) |
| C9—C14 | 1.505 (8) | C30'—H30C | 0.9700 |
| C9—C10 | 1.507 (8) | C30'—H30D | 0.9700 |
| C9—H9 | 0.9800 | C31'—C32' | 1.494 (7) |
| C10—C11 | 1.548 (9) | C31'—H31C | 0.9700 |
| C10—H10A | 0.9700 | C31'—H31D | 0.9700 |
| C10—H10B | 0.9700 | C32'—C33' | 1.492 (8) |

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| C11—C12 | 1.427 (10) | C32'—H32B | 0.9800 |
| C11—H11A | 0.9700 | C33'—C34' | 1.497 (7) |
| C11—H11B | 0.9700 | C33'—H33C | 0.9700 |
| C12—C13 | 1.497 (9) | C33'—H33D | 0.9700 |
| C12—H12 | 0.9800 | C34'—H34C | 0.9700 |
| C13—C14 | 1.545 (9) | C34'—H34D | 0.9700 |
| C13—H13A | 0.9700 | C35—C36 | 1.382 (7) |
| C13—H13B | 0.9700 | C35—C40 | 1.397 (7) |
| C14—H14A | 0.9700 | C36—C37 | 1.377 (8) |
| C14—H14B | 0.9700 | C36—H36 | 0.9300 |
| C15—C16 | 1.372 (7) | C37—C38 | 1.367 (8) |
| C15—C20 | 1.394 (7) | C38—C39 | 1.360 (8) |
| C16—C17 | 1.376 (7) | C38—H38 | 0.9300 |
| C16—H16 | 0.9300 | C39—C40 | 1.369 (7) |
| C17—C18 | 1.378 (8) | C39—H39 | 0.9300 |
| C18—C19 | 1.375 (8) | | |
| C12—O3—H3 | 118.6 | C24—C23—C22 | 121.7 (5) |
| O1—N1—O2 | 122.5 (6) | C24—C23—Br4 | 119.4 (4) |
| O1—N1—C17 | 118.7 (6) | C22—C23—Br4 | 118.9 (4) |
| O2—N1—C17 | 118.8 (5) | C25—C24—C23 | 119.0 (5) |
| C7—N2—C8 | 109.3 (4) | C25—C24—H24 | 120.5 |
| C7—N2—C9 | 112.8 (4) | C23—C24—H24 | 120.5 |
| C8—N2—C9 | 115.5 (4) | C24—C25—C26 | 120.2 (5) |
| C2—N3—C8 | 121.6 (4) | C24—C25—Br3 | 120.6 (4) |
| C2—N3—H3N | 119.2 | C26—C25—Br3 | 119.1 (5) |
| C8—N3—H3N | 119.2 | C21—C26—C25 | 120.5 (5) |
| O4—N4—O5 | 123.2 (7) | C21—C26—H26 | 119.8 |
| O4—N4—C37 | 118.6 (7) | C25—C26—H26 | 119.8 |
| O5—N4—C37 | 118.1 (7) | N5—C27—C21 | 110.9 (4) |
| C28—N5—C27 | 109.7 (4) | N5—C27—H27A | 109.5 |
| C28—N5—C29 | 117.3 (4) | C21—C27—H27A | 109.5 |
| C27—N5—C29 | 112.7 (4) | N5—C27—H27B | 109.5 |
| C22—N6—C28 | 121.7 (4) | C21—C27—H27B | 109.5 |
| C22—N6—H6N | 119.2 | H27A—C27—H27B | 108.1 |
| C28—N6—H6N | 119.2 | N5—C28—N6 | 112.7 (4) |
| C6—C1—C2 | 119.6 (5) | N5—C28—C35 | 109.0 (4) |
| C6—C1—C7 | 122.8 (5) | N6—C28—C35 | 112.2 (4) |
| C2—C1—C7 | 117.6 (5) | N5—C28—H28 | 107.6 |
| N3—C2—C3 | 123.3 (5) | N6—C28—H28 | 107.6 |
| N3—C2—C1 | 119.7 (5) | C35—C28—H28 | 107.6 |
| C3—C2—C1 | 117.0 (5) | C32—O6—H6 | 134.0 |
| C4—C3—C2 | 122.4 (6) | C32—O6—H6' | 118.5 |
| C4—C3—Br1 | 119.0 (5) | H6—O6—H6' | 20.8 |
| C2—C3—Br1 | 118.5 (5) | C34—C29—C30 | 120.7 (6) |
| C5—C4—C3 | 118.5 (6) | C34—C29—N5 | 126.4 (6) |
| C5—C4—H4 | 120.8 | C30—C29—N5 | 112.1 (6) |
| C3—C4—H4 | 120.8 | C34—C29—H29A | 93.0 |
| C6—C5—C4 | 121.6 (6) | C30—C29—H29A | 93.0 |
| C6—C5—Br2 | 119.5 (6) | N5—C29—H29A | 93.0 |

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|---------------|-----------|----------------|------------|
| C4—C5—Br2 | 118.8 (5) | C29—C30—C31 | 118.7 (7) |
| C5—C6—C1 | 120.8 (6) | C29—C30—H30A | 107.6 |
| C5—C6—H6B | 119.6 | C31—C30—H30A | 107.6 |
| C1—C6—H6B | 119.6 | C29—C30—H30B | 107.6 |
| N2—C7—C1 | 114.2 (4) | C31—C30—H30B | 107.6 |
| N2—C7—H7A | 108.7 | H30A—C30—H30B | 107.1 |
| C1—C7—H7A | 108.7 | C30—C31—C32 | 117.0 (8) |
| N2—C7—H7B | 108.7 | C30—C31—H31A | 108.0 |
| C1—C7—H7B | 108.7 | C32—C31—H31A | 108.1 |
| H7A—C7—H7B | 107.6 | C30—C31—H31B | 108.0 |
| N3—C8—N2 | 111.8 (4) | C32—C31—H31B | 108.0 |
| N3—C8—C15 | 113.2 (4) | H31A—C31—H31B | 107.3 |
| N2—C8—C15 | 110.3 (4) | O6—C32—C31 | 115.5 (14) |
| N3—C8—H8 | 107.1 | O6—C32—C33 | 109.5 (12) |
| N2—C8—H8 | 107.1 | C31—C32—C33 | 115.2 (9) |
| C15—C8—H8 | 107.1 | O6—C32—H32A | 105.1 |
| N2—C9—C14 | 113.4 (5) | C31—C32—H32A | 105.1 |
| N2—C9—C10 | 108.7 (5) | C33—C32—H32A | 105.1 |
| C14—C9—C10 | 109.1 (5) | C34—C33—C32 | 117.0 (8) |
| N2—C9—H9 | 108.5 | C34—C33—H33A | 108.1 |
| C14—C9—H9 | 108.5 | C32—C33—H33A | 108.1 |
| C10—C9—H9 | 108.5 | C34—C33—H33B | 108.1 |
| C9—C10—C11 | 111.9 (6) | C32—C33—H33B | 108.1 |
| C9—C10—H10A | 109.2 | H33A—C33—H33B | 107.3 |
| C11—C10—H10A | 109.2 | C29—C34—C33 | 118.0 (7) |
| C9—C10—H10B | 109.2 | C29—C34—H34A | 107.8 |
| C11—C10—H10B | 109.2 | C33—C34—H34A | 107.8 |
| H10A—C10—H10B | 107.9 | C29—C34—H34B | 107.8 |
| C12—C11—C10 | 110.9 (6) | C33—C34—H34B | 107.8 |
| C12—C11—H11A | 109.5 | H34A—C34—H34B | 107.1 |
| C10—C11—H11A | 109.5 | C32'—O6'—H6 | 118.2 |
| C12—C11—H11B | 109.5 | C32'—O6'—H6' | 116.1 |
| C10—C11—H11B | 109.5 | H6—O6'—H6' | 32.6 |
| H11A—C11—H11B | 108.1 | C31'—C30'—H30C | 108.4 |
| C11—C12—O3 | 111.2 (6) | C31'—C30'—H30D | 108.4 |
| C11—C12—C13 | 111.6 (6) | H30C—C30'—H30D | 107.5 |
| O3—C12—C13 | 108.1 (5) | C32'—C31'—C30' | 111.1 (7) |
| C11—C12—H12 | 108.6 | C32'—C31'—H31C | 109.4 |
| O3—C12—H12 | 108.6 | C30'—C31'—H31C | 109.4 |
| C13—C12—H12 | 108.6 | C32'—C31'—H31D | 109.4 |
| C12—C13—C14 | 110.3 (5) | C30'—C31'—H31D | 109.4 |
| C12—C13—H13A | 109.6 | H31C—C31'—H31D | 108.0 |
| C14—C13—H13A | 109.6 | O6'—C32'—C33' | 111.3 (9) |
| C12—C13—H13B | 109.6 | O6'—C32'—C31' | 107.2 (9) |
| C14—C13—H13B | 109.6 | C33'—C32'—C31' | 110.1 (7) |
| H13A—C13—H13B | 108.1 | O6'—C32'—H32B | 109.4 |
| C9—C14—C13 | 111.8 (5) | C33'—C32'—H32B | 109.4 |
| C9—C14—H14A | 109.3 | C31'—C32'—H32B | 109.4 |
| C13—C14—H14A | 109.3 | C32'—C33'—C34' | 110.9 (7) |

supplementary materials

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| C9—C14—H14B | 109.3 | C32'—C33'—H33C | 109.5 |
| C13—C14—H14B | 109.3 | C34'—C33'—H33C | 109.5 |
| H14A—C14—H14B | 107.9 | C32'—C33'—H33D | 109.5 |
| C16—C15—C20 | 116.8 (5) | C34'—C33'—H33D | 109.5 |
| C16—C15—C8 | 122.2 (5) | H33C—C33'—H33D | 108.1 |
| C20—C15—C8 | 120.8 (5) | C33'—C34'—H34C | 108.7 |
| C15—C16—C17 | 120.9 (5) | C33'—C34'—H34D | 108.7 |
| C15—C16—H16 | 119.5 | H34C—C34'—H34D | 107.6 |
| C17—C16—H16 | 119.5 | C36—C35—C40 | 116.3 (5) |
| C16—C17—C18 | 121.8 (5) | C36—C35—C28 | 122.9 (5) |
| C16—C17—N1 | 119.2 (5) | C40—C35—C28 | 120.9 (4) |
| C18—C17—N1 | 119.0 (5) | C37—C36—C35 | 120.2 (5) |
| C19—C18—C17 | 117.6 (5) | C37—C36—H36 | 119.9 |
| C19—C18—H18 | 121.2 | C35—C36—H36 | 119.9 |
| C17—C18—H18 | 121.2 | C38—C37—C36 | 122.7 (5) |
| C20—C19—C18 | 120.7 (5) | C38—C37—N4 | 119.3 (6) |
| C20—C19—H19 | 119.7 | C36—C37—N4 | 118.0 (6) |
| C18—C19—H19 | 119.7 | C39—C38—C37 | 117.7 (6) |
| C19—C20—C15 | 122.1 (5) | C39—C38—H38 | 121.1 |
| C19—C20—Cl1 | 117.5 (4) | C37—C38—H38 | 121.1 |
| C15—C20—Cl1 | 120.4 (4) | C38—C39—C40 | 120.7 (5) |
| C26—C21—C22 | 121.0 (5) | C38—C39—H39 | 119.7 |
| C26—C21—C27 | 122.4 (5) | C40—C39—H39 | 119.7 |
| C22—C21—C27 | 116.7 (5) | C39—C40—C35 | 122.4 (5) |
| N6—C22—C21 | 120.0 (5) | C39—C40—Cl2 | 116.8 (4) |
| N6—C22—C23 | 122.3 (5) | C35—C40—Cl2 | 120.8 (4) |
| C21—C22—C23 | 117.7 (5) | | |

Hydrogen-bond geometry (\AA , $^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------------|-------|-------------|-------------|---------------|
| N3—H3N \cdots O4 ⁱ | 0.86 | 2.33 | 2.942 (7) | 129 |
| O3—H3 \cdots O6 ⁱⁱ | 0.84 | 1.89 | 2.713 (12) | 166 |
| O6—H6 \cdots O3 ⁱⁱⁱ | 0.84 | 2.04 | 2.877 (17) | 179 |
| O6'—H6' \cdots O3 ⁱⁱⁱ | 0.84 | 1.85 | 2.694 (10) | 179 |

Symmetry codes: (i) $-x+1, -y, -z+1$; (ii) $-x+1, -y+1, -z$; (iii) $x-1, y, z$.

Fig. 1

