



Article Computationally Assisted Structural Elucidation of Cembranoids from the Soft Coral Sarcophyton tortuosum

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Abstract: A persistent study on soft coral *Sarcophyton tortuosum* resulted in the characterization of two new cembranolides, tortuolides A and B (1 and 2), and a new related diterpene, *epi*-sarcophytonolide Q. Their structures were determined not only by extensive spectroscopic analysis but also by DFT calculations of ECD and NMR data, the latter of which was combined with statistical analysis methods, e.g., DP4+ and *J*-DP4 approaches. Anti-inflammatory and cytotoxicity activities were evaluated in this study.

Keywords: tortuolide A; tortuolide B; epi-sarcophytonolide; DP4+; J-DP4; Sarcophyton tortuosum

1. Introduction

Soft corals are known to produce a large variety of secondary metabolites [1]. In particular, soft coral of the genus *Sarcophyton* is a prolific source of promising bioactive cembranoids, some of which have exhibited antiviral [2], anti-inflammatory [3–7], and cytotoxic activities [4–6,8]. The flexibility of the macrocyclic ring in cembranoids makes the accurate determination of chemical structures particularly challenging. Despite the 2D NMR spectroscopic method being the most potent approach for structural elucidation, it suffers from inherently low accuracy for flexible structures, especially when there is no informative correlation in NMR spectrometry. Consequently, the computational approach and related statistical analysis methods, e.g., DP4+ and *J*-DP4 [9,10], have gradually became convenient and reliable alternatives.

Our pervious investigation on *Sarcophyton tortuosum* resulted in the isolation of several novel structures, including secotortuosenes A and B with a novel secoditerpenoidal skeleton, bistortuolide cyclobutane A with a novel cyclobutane biscembranoidal skeleton, and



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). tortuosenes A and B with a rare tricyclic diterpenoidal skeleton [7,11]. As part of our continuing effort to explore bioactive marine natural products from soft corals [3–8], the chemical constituents of *S. tortuosum* collected at Lanyu Island were investigated in this study, and three new cembranoids, namely tortuolides A and B (1 and 2) and *epi*-sarcophytonolide Q (3), were characterized (Figure 1). Structural elucidations were performed by a comprehensive 2D NMR spectroscopic analysis, as well as computational and statistical analysis methods. Their biological activities, including cytotoxicity and anti-inflammatory activities, were evaluated herein.



Figure 1. Structures of compounds 1–3.

2. Results and Discussion

Defrosted specimens (1.3 kg) of the soft coral *S. tortuosum* was freeze-dried, minced, and extracted with EtOAc to yield a crude extract (10.2 g), which was repeatedly chromatographed on silica gel and subsequently purified by high-performance liquid chromatography (HPLC) to obtain compounds **1** (8.9 mg), **2** (2.7 mg), and **3** (2.1 mg) (Figure 1).

Tortuolide A (1), obtained as a colorless oil, was found to have a molecular formula of $C_{23}H_{32}O_7$ based on the sodiated ion peak at m/z 443.2049 [M + Na]⁺ (calcd for $C_{23}H_{32}O_7Na$, 443.2040). The ¹H and ¹³C NMR data (Table 1) revealed evidence of two α,β -conjugated carboxylate systems [δ_C 136.9 (CH, C-3), 136.3 (C, C-4), and 167.0 (C, C-18); δ_H 6.69 (1H, d, J = 10.0 Hz, H-3); $\delta_{\rm C}$ 140.4 (CH, C-11), 132.2 (C, C-12), and 167.6 (C, C-20); $\delta_{\rm H}$ 6.33 (1H, br s, H-11)], an acetoxyl group [δ_C 169.4 (C), 21.1 (CH₃); δ_H 1.99 (s, CH₃)], an epoxy group $[\delta_{C} 68.4 (C, C-1), 59.2 (CH, C-2); \delta_{H} 3.28 (1H, d, J = 10.0 Hz, H-2)]$, an oxygenated methine $[\delta_{C} 69.8 (CH, C-7); \delta_{H} 4.93 (1H, dd, J = 8.5, 2.0 Hz, H-7)]$, a methoxyl group $[\delta_{C} 52.0 (CH_{3}, J_{2})]$ 18-OMe); $\delta_{\rm H}$ 3.73 (3H, s, 18-OMe)], and an oxygenated quaternary carbon [$\delta_{\rm C}$ 83.5 (C, C-8)]. Inspection of the NMR data revealed that the planar structures of 1 and emblide [11,12] were quite similar, with differences for the Δ^1 double bond in emblide replaced by an epoxy ring in 1, as indicated by the heteronuclear multiple bond correlation (HMBC) correlations from H₃-16, H₃-17, H₂-14, and H-2 to C-1, as well as the correlation spectroscopy (COSY) correlation between H-2 and H-3 (Figure 2). The cis geometry of the epoxide was assigned by the nuclear Overhauser effect (NOE) correlation of H-15/H-3, whereas the *E* double bond was deduced based on the NOE correlation of OMe/H-3 (Figure 3). The correlation of H-7/H-2 suggested both protons were pointed inside the macrocyclic ring. Furthermore, NOE correlations of H₃-19/H-6 ($\delta_{\rm H}$ 2.37), H-7/H-10 ($\delta_{\rm H}$ 2.28), and H-11/H₃-16 suggested H₃-19 and H-7 were oppositely oriented (Supplementary Materials, Figures S1–S8).

The relative configuration of **1** was further secured by utilizing the computational NMR data coupled with a combined indirect *J*-DP4 (*iJ*-DP4) and direct *J*-DP4 (*dJ*-DP4) [10]. Four possible diastereomers, including $1R^*,2S^*,7S^*,8S^*-1$, $1R^*,2S^*,7R^*,8S^*-1$, $1R^*,2S^*,7S^*,8R^*-1$, and $1R^*,2S^*,7R^*,8R^*-1$, were subjected to conformational search using the Merck molecular force field (MMFF94) as implemented in the GMMX program. In this case, ${}^3J_{H,H}$ values of H-2/H-3 and H-7/H₂-6 were selected to restrict conformational sampling (Supplementary Materials, Tables S3–S6). Moreover, in order to reduce the computational cost, the strong NOE correlation of H₃-16/H-11 in compound **1** (Figure 3) was further selected as second restriction. After restrictions by ${}^3J_{H,H}$ values and NOE correlations, two of the diastere-

omers, $1R^*$, $2S^*$, $7S^*$, $8S^*$ -1 and $1R^*$, $2S^*$, $7R^*$, $8S^*$ -1, where the H₃-16 and H-11 are anti-oriented, were excluded, as this correlation (H₃-16/H-11) is not possible in these two candidates. On the other hand, the remaining two diastereomers without geometry optimization, $1R^*$, $2S^*$, $7S^*$, $8R^*$ -1 and $1R^*$, $2S^*$, $7R^*$, $8R^*$ -1, were subjected to gauge-invariant atomic orbital (GIAO) calculations of shielding tensors and coupling constants. The Boltzmann weighted computational data were analyzed utilizing the *J*-DP4 statistical method [10]. As a result, $1R^*$, $2S^*$, $7S^*$, $8R^*$ -1 was found to be the correct structure, with a high probability of 99.47% (Table 2). The absolute configuration of 1 was further determined by comparing the experimental and calculated electronic circular dichroism (ECD) spectra (Supplementary Materials, Table S1). The calculated ECD spectra (Figure 4) at the TDDFT/M06-2X/def2tzvp level of theory with integral equation formalism polarizable continuum model (IEFPCM) in MeOH suggested the absolute configuration of 1 to be 1R, 2S, 7S, and 8R.



Figure 2. COSY (bold) and selective HMBC (arrows) correlations of 1–3.



Figure 3. Selective NOE correlations of 1–3.

	1 ^a		2 ^a		3 ^b	
Position	$\delta_{ m H}$ (J in Hz)	δ_{C} (Type)	$\delta_{ m H}$ (J in Hz)	δ_{C} (Type)	$\delta_{ m H}$ (J in Hz)	δ_{C} (Type)
1		68.4 (qC)		68.4 (qC)		154.5 (qC)
2	3.28 d (10.0)	59.2 (ĈH)	3.81 d (8.5)	57.5 (CH)	6.92 d (11.2)	120.0 (CH)
3	6.69 d (10.0)	136.9 (CH)	6.48 d (8.5)	143.0 (CH)	6.64 d (11.2)	138.5 (CH)
4		136.3 (qC)		135.5 (qC)		126.5 (qC)
5	2.59 m	23.6 (CH ₂)	2.64 m	25.6 (CH ₂)	2.29 m	29.4 (CH ₂)
			2.71 m		2.62 m	
6	1.74 m	28.8 (CH ₂)	1.71 m	31.8 (CH ₂)	1.53 m	30.3 (CH ₂)
	2.37 m		2.42 m		2.02 t (12.4)	
7	4.93 dd (8.5, 2.0)	69.8 (CH)	4.80 br s	73.9 (CH)	3.70 d (10.4)	70.3 (CH)
8		83.5 (qC)		81.8 (qC)		74.6 (qC)
9	2.11 m	35.6 (CH ₂)	1.99 m	34.9 (CH ₂)	1.58 m	37.1 (CH ₂)
			2.11 m		2.16 m	
10	2.28 m	26.8 (CH ₂)	2.40 m	27.1 (CH ₂)	2.16 m	25.0 (CH ₂)
	2.43 m				2.74 m	
11	6.33 br s	140.4 (CH)	6.44 br s	139.2 (CH)	6.08 dd (8.0, 7.6)	144.5 (CH)
12		132.2 (qC)		131.3 (qC)		132.4 (qC)
13	2.34 m	29.4 (CH ₂)	2.20 m	34.9 (CH ₂)	2.38 m	33.14 (CH ₂)
	2.64 m		3.02 br d (14.0)		2.75 m	
14	1.82 m	28.2 (CH ₂)	1.90 td (14.0, 6.0)	26.4 (CH ₂)	2.30 m	30.1 (CH ₂)
	2.16 m		2.05 m		2.57 m	
15	1.84 m	32.2 (CH)	2.22 m	27.6 (CH)	2.46 m	33.07 (CH)
16	1.04 d (7.0)	19.4 (CH ₃)	0.89 d (6.5)	16.1 (CH ₃)	1.05 d (7.2)	22.25 (CH ₃)
17	1.12 d (6.5)	17.8 (CH ₃)	1.07 d (7.0)	17.8 (CH ₃)	1.10 d (7.2)	20.9 (CH ₃)
18		167.0 (qC)		166.8 (qC)		168.0 (qC)
19	1.52 s	24.5 (CH ₃)	1.39 s	24.2 (CH ₃)	1.12 s	22.34 (CH ₃)
20		167.6 (qC)		165.0 (qC)		168.5 (qC)
OAc	1.99 s	21.1 (CH ₃)	2.09 s	21.2 (CH ₃)		-
		169.4 (qC)		170.3 (qC)		
18-OMe	3.73 s	52.0 (CH ₃)	3.78 s	52.0 (CH ₃)	3.75 s	51.2 (CH ₃)
20-OMe					3.62 s	51.6 (CH ₃)

Table 1. ¹H and ¹³C NMR spectroscopic data of 1–3 in CDCl₃.

^a Spectra were recorded at 500 (¹H NMR) and 125 MHz (¹³C NMR). ^b Spectra were recorded at 400 (¹H NMR) and 100 MHz (¹³C NMR).



Figure 4. Experimental and calculated ECD spectra of (1) and (2). Gaussian band shape (σ) with values of 0.20 (for 1) and 0.28 (for 2) eV.

Tortuolide B (**2**) was found to have the same molecular formula ($C_{23}H_{32}O_7$) as that of **1**, with the sodiated ion peak at m/z 443.2044 [M + Na]⁺ (calcd for $C_{23}H_{32}O_7$ Na, 443.2040). After careful analysis of the differences between compounds **1** and **2**, it was found that the two compounds shared the same planar structure, and **2** should be a configurational isomer of **1**. The HMBC and COSY correlations confirmed the above elucidation as shown

in Figure 2. The NOE correlations of H-3/H-14 ($\delta_{\rm H}$ 1.90) and H-2/H-5 ($\delta_{\rm H}$ 2.71) suggested the geometries of trans epoxide and *E*-double bond, respectively (Figure 3). The NOE correlations of H-5 ($\delta_{\rm H}$ 2.71)/H-2 and H-5/H-7 indicated these protons were α -oriented. The H-11 in unsaturated ε -lactone was assigned to a point on the same face with the isopropyl group due to the observation of NOE correlations of H-13 ($\delta_{\rm H}$ 2.20)/H-11, H-13/H₃-16, and H-13/H₃-17. Moreover, correlations of H₂-9/H₃-19 and H-7/H-9 ($\delta_{\rm H}$ 2.11) indicated H₃-19 located opposite to H-7 (Supplementary Materials, Figures S9–S16). Similarly, the relative configurations were also confirmed by computational NMR calculation for the four possible diastereomers (1*R**,2*R**,7*R**,8*R**-2, 1*R**,2*R**,7*S**,8*R**-2, 1*R**,2*R**,7*R**,8*S**-2, and 1*R**,2*R**,7*S**,8*S**-**2**) and combined with subsequent *J*-DP4 analysis. The diastereomer with $1R^*$, $2R^*$, $7S^*$, $8S^*$ configuration was excluded using ${}^{3}J_{H,H}$ values (${}^{3}J_{H,H}$ values of H-2/H-3 and H-7/H₂-6) as conformational constrains (Supplementary Materials, Tables S7–S10). As a result, 1*R**,2*R**,7*S**,8*R**-**2** was suggested to be the correct structure, with a high *J*-DP4 probability of 100% (Table 3). Furthermore, the theoretical ECD spectra were calculated for 1R,2R,7S,8R-2 and its enantiomer, 1S,2S,7R,8S-2 (Supplementary Materials, Table S2), and the former showed good agreement with the calculated data (Figure 4).

Table 2. J-DP4 (PCM/B3LYP/6-31+G(d,p)) probabilities for compounds 1 and 2.

	J-DP4 (%)					
	1 <i>R*,2S*,7S*,8R*-</i> 1	1 <i>R</i> *,2 <i>S</i> *,7 <i>R</i> *,8 <i>R</i> *-1	1 <i>R</i> *,2 <i>R</i> *,7 <i>R</i> *,8 <i>R</i> *-2	1 <i>R</i> *,2 <i>R</i> *,7 <i>S</i> *,8 <i>R</i> *-2	1 <i>R</i> *,2 <i>R</i> *,7 <i>R</i> *,8 <i>S</i> *-2	
Н	0.06	99.94	3.38	96.62	0	
С	100.00	0	0.02	97.52	2.46	
H + C	99.16	0.84	0	100.00	0	
J	61.32	38.68	1.63	11.93	86.43	
all data	99.47	0.53	0	100.00	0	

epi-Sarcophytonolide Q (3), a white amorphous powder, was found to have a molecular formula of $C_{22}H_{34}O_6$, as determined by HRESIMS (m/z calcd 417.2248; found 417.2251, $[M + Na]^+$). The IR spectrum of 3 showed the presence of hydroxyl (ν_{max} 3474 cm⁻¹) and a conjugated carbonyl group (ν_{max} 1699 cm⁻¹). The latter was confirmed by the UV absorption maxima (λ_{max} 286, 216). The NMR spectroscopic data of **3** revealed the presence of two conjugated methyl esters [δ_C 168.0 (C, C-18), 51.2 (CH₃, 18-OMe); δ_H 3.75 s (18-OMe), and $\delta_{\rm C}$ 168.5 (C, C-20), 51.6 (CH₃, 20-OMe); $\delta_{\rm H}$ 3.62 s (20-OMe)], an oxymethine [$\delta_{\rm C}$ 70.3 (CH); $\delta_{\rm H}$ 3.70 d (*J* = 10.4 Hz, H-7)], and an oxygenated quaternary carbon [$\delta_{\rm C}$ 74.6 (C, C-8)] (Tables 1 and 2). The above data were very similar to those of sarcophytonolide Q [13]. Further COSY and HMBC correlations confirmed that they share the same planar structure (Figure 2). In the NOESY spectra of 3, H₃-19 showed an NOE correlation with H-6 ($\delta_{\rm H}$ 1.53) but not with H-7, which was reported in sarcophytonolide Q, revealing they were diastereomers (Supplementary Materials, Figures S17–S24). The above data inferred that they are structurally different at either C-7 or C-8; however, both compounds shared the same coupling constant at H-7 (10.4 Hz), revealing that the constraints for the J-DP4 method were unable to be applied in this case. Alternatively, the DP4+ method was performed for two possible diastereomers of **3** (7*S**,8*R**-**3** and 7*R**,8*R**-**3**) [9]. First, a conformational search was performed using GMMX, with subsequent geometry optimization at the B3LYP/6-31G(d) level of theory. Next, the NMR shielding tensors were calculated at the mPW1PW91/6-31+G(d,p) level with the polarizable continuum model (PCM) in CHCl₃, as recommended in the literature [9]. Finally, the Boltzmann-weighted data were subjected to DP4+ analysis. As a result, a relative $7S^*$, $8R^*$ configuration was suggested for **3** with 100% probability (Table 3).

Evaluations for inhibitory effect toward the superoxide anion generation and elastase release in fMLF/CB-induced human neutrophils were performed on compounds 1–3. The result showed that compounds 1 and 2 exhibited weak inhibitory activity of 13.64 \pm 2.27 and 14.15 \pm 3.57%, respectively, on elastase release at a concentration of 10 μ M. Compounds

1–3 were further screened for cytotoxicity toward murine leukemia (P388), human chronic myelogenous leukemia (K562), human colon carcinoma (HT-29), human lung adenocarcinoma (A-549), and lymphoblastic leukemia (Molt-4); unfortunately, the tested compounds were also found to be inactive against the above cell lines, with IC_{50s} over 40 μ M.

Table 3. DP4+ (PCM/mPW1PW91/6-31+G(d,p)) probabilities for compound 3.

		DP4 + (%)	
	Н	С	All Data
7 <i>S</i> *,8 <i>R</i> *-3	99.75	100.00	100.00
7R*,8R*-3	0.25	0	0

3. Materials and Methods

3.1. General Experimental Procedures

Specific optical rotations were measured in CHCl₃ using a JASCO P-1020 digital polarimeter (JASCO Corporation, Tokyo, Japan). IR spectra were recorded on a JASCO FT/IR-4100 spectrometer and an FT/IR-4100 infrared spectrophotometer (JASCO Corporation, Tokyo, Japan). The NMR experiments were performed in CDCl₃ on a Varian 400MR FT-NMR instrument and a Varian Unity INOVA 500 FT-NMR spectrometer (Varian Inc., Palo Alto, CA, USA). LR- and HR-ESIMS were measured with a Bruker APEX II mass spectrometer and a Bruker Apex-Qe 9.4T mass spectrometer (Bruker, Bremen, Germany), respectively. Before column chromatography using Si gel or C18 gel (40–63 μ m, Merck, Darmstadt, Germany), TLC analysis was performed on aluminum plates coated with Si gel or C18 gel (Kieselgel 60 F₂₅₄, 0.25 mm, Merck, Darmstadt, Germany). HPLC was performed on a Hitachi L-2455 apparatus equipped with a Supelco C18 column (ODS-3, 5 μ m, 250 \times 20 mm; Sciences Inc., Tokyo, Japan).

3.2. Animal Material

The animal material, *S. tortuosum*, was manually collected by an underwater diver from the coral reef of Lanyu Island of Taiwan in August 2008. The specimen was identified by Prof. C.-F. Dai. A voucher specimen (specimen no. sheuCYJ-001) was deposited with the Department of Marine Biotechnology and Resources, National Sun Yat-sen University, Kaohsiung 804, Taiwan.

3.3. Extraction and Separation

The defrosted *S. tortuosum* organism was weighed and subsequently freeze-dried, minced, and extracted repeatedly (1L) with ethyl acetate (EtOAc) to obtain a crude product (10.2 g), which was fractionated to obtain 25 fractions (F1-F25) as described previously [11]. Fractions 18 and 19, showing similar compositions on TLC plates, were combined and fractionated by chromatography on Si gel using *n*-hexane-EtOAc (3:1) and then *n*-hexane-acetone (6:1 and 3:1) as eluents to yield a crude residue, which was purified by normal-phase HPLC (*n*-hexane-EtOAc, 5:2) to give compound **3** (2.1 mg).

A subfraction purified by an Si gel open column (*n*-hexanes-EtOAc, 5:1) from fraction 16 was further separated by semipreparative NP-HPLC eluting with *n*-hexane-EtOAc (5:1) to afford a crude residue, which was further purified by reverse-phase HPLC (CH₃CN-H₂O, 3:2) to give compounds **1** (8.9 mg) and **2** (2.7 mg).

Tortuolide A (1): colorless oil; $[\alpha]^{26} {}_{\rm D}$ + 24 (c 0.74, CHCl₃); UV (MeOH) $\lambda_{\rm max}$ (log ε) 282 (3.2), 228 (4.0) nm; IR (KBr) $\nu_{\rm max}$ 2961, 2927, 1740, 1713, 1693, 1260, 1212 cm⁻¹; ¹H and ¹³C NMR data, see Table 1; positive ESIMS m/z 443 [M + Na]⁺; positive HRESIMS m/z 443.2049 [M + Na]⁺ (calcd for 443.2040 for C₂₃H₃₂O₇Na).

Tortuolide B (2): colorless oil; $[\alpha]^{26}$ _D +111 (c 0.77, CHCl₃); UV (MeOH) λ_{max} (log ε) 228 (4.2) nm; IR (KBr) ν_{max} 2928, 2857, 1719, 1704, 1253, 1208 cm⁻¹; ¹H and ¹³C NMR data, see Table 1; positive ESIMS m/z 443 [M + Na]⁺; positive HRESIMS m/z 443.2044 [M + Na]⁺ (calcd for 443.2040 for C₂₃H₃₂O₇Na).

epi-Sarcophytonolide Q (3): white amorphous solid; $[\alpha]^{25}_{D}$ +214 (c 1.43, CHCl₃); UV (MeOH) λ_{max} (log ε) 286 (4.1), 216 (4.0) nm; IR (KBr) ν_{max} 3474, 3016, 2955, 2873, 1698, 1629, 1439, 1378 cm⁻¹; ¹H and ¹³C NMR data, see Table 1; positive ESIMS *m*/*z* 417 [M + Na]⁺; positive HRESIMS *m*/*z* 417.2251 [M + Na]⁺ (calcd for 417.2248 for C₂₂H₃₄O₆Na).

3.4. Computational Method

A conformational search was performed at the MMFF94 force field using GMMX package implemented in Gaussian 16 software [14]. The resulted conformers, within a 6 kcal/mol window, were subjected for further NMR and ECD calculations.

For iJ/dJ-DP4, the shielding tensors and Fermi contacts were calculated at the PCM/ B3LYP/6-31+G(d,p)//MMFF94 level. The resulting data were weighted based on Boltzmann population using energies calculated at the same level of theory. *J*-DP4 probabilities were generated using the Excel sheet provided by Zanardi et al [10]. For DP4+, the conformers were subjected to geometry optimization at B3LYP/6-31G(d) [9]. The NMR chemical shifts were computed at the PCM/mPW1PW91/6-31G+(d,p)//B3LYP/6-31G(d) level in chloroform with the Boltzmann population refined in the solvation model based on density (SMD) for CHCl₃ at a new level (M06-2X/6-31G+(d,p)) [15]. The DP4+ probability was determined using the Excel sheet provided in the literature [9].

The conformers resulting from MMFF94 calculations were subjected to geometry optimizations and frequency calculations at the M06-2X/def2svp level using IEFPCM in MeOH. The generated ECD spectra calculated at TDDFT/M06-2X/def2TZVP with IEF-PCM in MeOH were weighted based on the Boltzmann population using Gibbs free energy, obtained by the sum of single-point energy at M06-2X/def2TZVP and the thermal correction at M06-2X/def2svp. The calculated ECD spectra were generated using GaussView 6 by applying a Gaussian band shape with 0.20 and 0.28 eV width for **1** and **2**, respectively. It should be noted that Grimme's dispersion (D3 version) was used for empirical dispersion correction in ECD calculation, and the *g09defaults* keywords were applied for NMR calculation.

3.5. Cytotoxicity Assay

The assay was implemented using the published Alamar Blue assay according to the published protocols [16,17]. Concisely, cancer cells, including P388, K562, HT-29, A549, and MOLT-4, were purchased from the American Type Culture Collection and individually seeded into a 96-well microtiter plate and incubated following the previously published procedure. The tested compounds were dissolved in DMSO and added to each well of cancer cells. After three days of culture, attached cells were treated with Alamar Blue for 4 h and subsequently measured at 595 nm using a microplate reader.

3.6. Anti-Inflammatory Assay

Freshly isolated human neutrophils from blood using dextran sedimentation were incubated according to the published procedure [18,19]. The incubated neutrophils (6×10^5 cells mL⁻¹) were treated with compounds 1–3 in DMSO for 5 min. After the neutrophils were activated with fMLF (100 nM)/CB for 10 min, the amounts of superoxide generation and elastase release were measured at 550 nm and 405 nm, respectively, using a UV spectrometer apparatus.

4. Conclusions

Two cembranolides, namely tortuolides A and B (1 and 2), and a related cembranoid, namely *epi*-sarcophytonolide Q (3), were characterized from the persistent study of the soft coral *Sarcophyton tortuosum*. Compounds 1 and 2 are structurally related to emblide [12,20], featuring a C-8–C-20 α , β -unsaturated ε -lactone ring, and represented the first emblide-related cembranolide with a 1,2-epoxy functionality. The flexible nature of macrocyclic compounds, e.g. cembranoids, make the unambiguous assignment of chemical structures particularly challenging. In the present study, we showed the successful application of DFT

calculations combined with statistical analysis methods, e.g. DP4+ and *J*-DP4, as well as the conventional NOESY approach.

Supplementary Materials: The following supporting information can be downloaded at: https://www.action.com/actionals //www.mdpi.com/article/10.3390/md20050297/s1, Figure S1: LR- and HR-ESIMS spectra of 1, Figure S2: ¹H NMR spectrum of 1 in CDCl₃, Figure S3: ¹³C NMR spectrum of 1 in CDCl₃, Figure S4: DEPT and ¹³C NMR spectra of 1 in CDCl₃, Figure S5: HSQC spectrum of 1 in CDCl₃, Figure S6: ¹H–¹H COSY spectrum of **1** in CDCl₃, Figure S7: HMBC spectrum of **1** in CDCl₃, Figure S8: NOESY spectrum of 1 in CDCl₃, Figure S9: LR- and HR-ESIMS spectra of 2, Figure S10: ¹H NMR spectrum of 2 in CDCl₃, Figure S11: ¹³C NMR spectrum of 2 in CDCl₃, Figure S12: DEPT and ¹³C NMR spectra of 2 in CDCl₃, Figure S13: HSQC spectrum of 2 in CDCl₃, Figure S14: ¹H–¹H COSY spectrum of 2 in CDCl₃, Figure S15: HMBC spectrum of 2 in CDCl₃, Figure S16: NOESY spectrum of 2 in CDCl₃, Figure S17: LR- and HR-ESIMS spectra of **3**, Figure S18: ¹H NMR spectrum of **3** in CDCl₃, Figure S19: ¹³C NMR spectrum of 3 in CDCl₃, Figure S20: DEPT and ¹³C NMR spectra of 3 in CDCl₃, Figure S21: HSQC spectrum of **3** in CDCl₃, Figure S22: ¹H–¹H COSY spectrum of **3** in CDCl₃, Figure S23: HMBC spectrum of 3 in CDCl₃, Figure S24: NOESY spectrum of 3 in CDCl₃, Table S1. Low-energy conformers of 1R,2S,7S,8R-1 for ECD calculations, Table S2: Low-energy conformers of 2 for ECD calculations, Table S3: Conformers of 1R*,2S*,7S*,8S*-1 for J-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S4: Conformers of 1R*,2S*,7R*,8S*-1 for J-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S5: Conformers of 1*R**,2*S**,7*S**,8*R**-1 for *J*-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S6: Conformers of 1*R**,2*S**,7*R**,8*R**-1 for *J*-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S7: Conformers of 1R*,2R*,7R*,8R*-2 for J-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S8: Conformers of 1R*,2R*,7S*,8R*-2 for J-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S9: Conformers of 1*R**,2*R**,7*R**,8*S**-2 for J-DP4 calculation and the dihedral angles of selected protons at MMFF94 level, Table S10: Conformers of 1R*,2R*,7S*,8S*-2 for J-DP4 calculation and the dihedral angles of selected protons at MMFF94 level. Tables S3-S10 [21].

Author Contributions: J.-H.S. conceptualized, designed, and guided the whole experiment and contributed to manuscript preparation. C.-H.C. and K.-H.L. elucidated the structure of the work and prepared the manuscript. C.-Y.H. and T.-L.H. performed bioassays. H.-C.H. and C.-H.C. contributed technical support for computational software and methodology, respectively. C.-F.D. identified the soft coral. All authors have read and agreed to the published version of the manuscript.

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