

Discovery of Dihydrochalcone as Potential Lead for Alzheimer's Disease: *In Silico* and *In Vitro* Study

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Abstract

By the virtual screening method we have screened out Dihydrochalcone as a top-lead for the Alzheimer's disease using the database of about 32364 natural compounds. The binding affinity of this ligand to amyloid beta ($A\beta$) fibril has been thoroughly studied by computer simulation and experiment. Using the Thioflavin T (ThT) assay we have obtained the inhibition constant $IC_{50} \approx 2.46\mu M$. This result is in good agreement with the estimation of the binding free energy obtained by the molecular mechanic-Poisson Boltzmann surface area method and all-atom simulation with the force field CHARMM 27 and water model TIP3P. Cell viability assays indicated that Dihydrochalcone could effectively reduce the cytotoxicity induced by $A\beta$. Thus, both *in silico* and *in vitro* studies show that Dihydrochalcone is a potential drug for the Alzheimers disease.

Citation: Viet MH, Chen C-Y, Hu C-K, Chen Y-R, Li MS (2013) Discovery of Dihydrochalcone as Potential Lead for Alzheimer's Disease: *In Silico* and *In Vitro* Study. PLoS ONE 8(11): e79151. doi:10.1371/journal.pone.0079151

Editor: Eugene A. Permyakov, Russian Academy of Sciences, Institute for Biological Instrumentation, Russian Federation

Received: June 7, 2013; **Accepted:** September 18, 2013; **Published:** November 18, 2013

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Funding: The work was supported by Narodowe Centrum Nauki in Poland (grant No 2011/01/B/NZ1/01622) and Department of Science and Technology at Ho Chi Minh city, Vietnam. CKH is supported by National Science Council in Taiwan under grant number NSC-100-2923-M-001-003-MY3 and National Center for Theoretical Sciences in Taiwan. The funders had no role in study design, data collection and analysis, decision to publish, or preparation of the manuscript.

Competing Interests: The authors have declared that no competing interests exist.

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Introduction

Alzheimer's disease (AD) is the most common form of dementia among the senior population that is increasing substantially as populations age [1]. The patient with AD will lose memory, decay language, and experience problems with visual spatial search *etc.* AD may be pathologically characterized by progressive intracerebral accumulation of beta amyloid ($A\beta$) peptides [2] and tau protein [3]. However, genetic and pathological evidences strongly support the first hypothesis [4]. The $A\beta$ peptides are proteolytic by-products of the amyloid precursor protein and are most commonly composed of 40 ($A\beta_{1-40}$) and 42 ($A\beta_{1-42}$) amino acids. $A\beta$ peptides appear to be unstructured in monomer state but aggregate to form fibrils with an ordered cross- β -sheet pattern [5,6]. Increasing evidence from recent studies indicates that both soluble oligomers and mature fibrils are the toxic agents [2].

Presently, there is no cure or treatment for AD, and significant effort has, therefore, been made to find efficient drugs to cope with it. One of the promising approaches is to inhibit misfolding and reverse aggregation of amyloid peptides [7]. A large number of potential $A\beta$ fibrillogenesis inhibitors have been proposed including polyamines, metal chelators, carbohydrate-containing compounds, polyphenols, osmolytes, short peptides, and RNA aptamers *etc* [7,8]. One should mention a number of small molecule inhibitors such as poly-L-lysine [9], dopamine [10], L-dopa [11], melatonin [12], indole-3-propionic acids [13], apomorphine derivatives [14], salvianolic acids [15] and many others. Nutraceuticals, which are natural products or extracts therefrom, as shown by preclinical and certain clinical studies, might be of value as AD therapeutic agents [7].

In this paper we carry out the comprehensive study of binding affinity of compounds derived from Eastern herbs and plants to aggregates of $A\beta_{40}$ and $A\beta_{42}$. Having used the docking method we screened out 20 top-leads for $6A\beta_{9-40}$ and $5A\beta_{17-42}$ fibrils. However, we were able to purchase only compound Dihydrochalcone which is an extract from *Daemonorops draco* tree. Our experimental study shows that this compound is promising for AD having the inhibition constant $IC_{50} \sim 2.46\mu M$ and low toxicity. The experimental result of the inhibition constant has been confirmed by the molecular mechanic-Poisson Boltzmann surface area (MM-PBSA) method which is more sophisticated than the docking method. Thus for the first time by the experiment and simulation we predict that Dihydrochalcone is a good candidate for AD.

Materials and Methods

Data Base of Compounds

We consider 32364 compounds derived from Eastern herbs and plants (see website: <http://tcm.cmu.edu.tw>). Applying Lipinski's rule of five [16] (see File S1) to this data base we obtained 3699 ligands that have drug-like properties and satisfy the Lipinski's rule. However, one should bear in mind that many natural products remain bioavailable despite violating the Rule of Five [17,18]. The virtual screening was applied to the reduced set of compounds.

Receptors

In order to study the binding affinity to mature fibrils we choose two typical structures of $6A\beta_{9-40}$ obtained from Prof. R. Tycko

[19] and 5A β_{17-42} (PDB ID: 2BEG [6]). Note that 8 and 16 disordered residues of the N-terminal of A β_{1-40} and A β_{1-42} are neglected from fibril constructions. In the case of A β_{40} , several NMR fibril structures are available, but to make a reasonable comparison between two types of fibrils, we choose 6A β_{9-40} because its structure is closest to 2BEG.

Docking Method

Both ligand-based and structure-based virtual screening methods have their advantages and drawbacks. However, since the docking methods are able to select more diverse actives than ligand-based methods like 2D similarity or substructure searching [20], we will use Autodock Vina method [21]. In our simulations the docking score is the binding energy and the best docking mode is the highest scoring pose or the conformation with the lowest binding energy. The details of this method are available in File S1.

Molecular Dynamics Simulation

In order to estimate the binding free energy of Dihydrochalcone to 6A β_{9-40} by the MM-PBSA method we have carried the molecular dynamics simulation using the force field CHARMM 27 [22] and water model TIP3P [23]. More details are described in File S1.

MM-PBSA Method

We have applied the MM-PBSA method to estimate the binding free energy of Dihydrochalcone to 6A β_{9-40} . The details of this method are given in our previous works [24]. Overall, in the MM-PBSA approach the binding free energy of ligand to receptor is defined as follows

$$\Delta G_{\text{bind}} = \Delta E_{\text{elec}} + \Delta E_{\text{vdw}} + \Delta G_{\text{sur}} + \Delta G_{\text{PB}} - T\Delta S, \quad (1)$$

where ΔE_{elec} and ΔE_{vdw} are contributions from electrostatic and vdW interactions, respectively. ΔG_{sur} and ΔG_{PB} are nonpolar and polar solvation energies. The entropic contribution $T\Delta S$ is estimated using the normal mode approximation. In order to calculate ΔG_{bind} , the molecular dynamics (MD) simulations have been carried out using the force field CHARMM 27 [22] and water model TIP3P [23]. The structures of receptor–ligand complex obtained in the best docking mode are used as starting configurations for simulations. For the 6A β_{9-40} –Dihydrochalcone complex four 20 ns MD trajectories were generated. Snapshots collected in equilibrium are used to compute the binding free energy given by Eq. 1.

A β Preparation

To prepare the A β stock, lyophilized A β 40 peptide, 0.5 mg, was freshly dissolved in 145 μ l Buffer A (10 mM Tris-HCl, pH = 7.4, and 150 mM NaCl) containing 8 M GdnHCl and refolded into Buffer A at a concentration of \sim 1 mg/ml. Then, the stock was centrifuged at 17,000 \times g, 4°C for 30 min. The supernatant was collected and quantified by absorbance at 280 nm ($\epsilon = 1,280 \text{ cm}^{-1}\text{M}^{-1}$) and used as a stock solution to prepare A β at 25 μ M for all experiments [25,26].

Dihydrochalcone Preparation

Dihydrochalcone was purchased from MP Biomedicals, Inc. (Fountain Pkwy, OH). Stock of Dihydrochalcone, 20 mM, was dissolved in 100% DMSO and diluted to different concentrations as indicated. The final DMSO concentration was constant in each condition.

Thioflavin T (ThT) Assay

A β (25 μ M) in Buffer A with different Dihydrochalcone concentrations and 25 μ M ThT were incubated in a 384-well ELISA plate and monitored by a microplate reader (SpectraMax M5; Molecule Devices) at 25°C. The samples were constantly rotated at 400 rpm during the incubation. The ThT fluorescence was measured at 485 nm where the excitation was at 442 nm. The final ThT intensity was plotted against the compound concentration and fitted to obtain IC50 using the equation, $Y = 100 / (1 + 10^{(X - \text{LogIC50})})$ in Prism 5 (GraphPad Software, Inc, Avenida de la Playa, CA).

TEM

The aggregated samples were placed on glow-discharged, 400-mesh Formvar carbon-coated copper grids (EMS Inc., Hatfield, PA, USA) for 3 min, rinsed, and negatively stained with 2% uranyl acetate. The samples were examined with a Hitachi H-7000 TEM (Hitachi Inc., Tokyo, Japan) with an accelerating voltage of 75 kV.

MTT Assay

The HEK293 cells were seeded into 96-well plates (100 μ l/well) one day prior to the experiment. The cells were treated with the end-point products and incubated at 37°C for 24 hr. MTT solution (Sigma) was then added to each well and incubated for another 4 hr. The medium was removed and 100 μ l of DMSO was added to dissolve the formazan crystals. The absorbance (A) was measured at 570 nm and the background signals caused by the samples without cells were subtracted. The data were normalized using the buffer control as 100%.

Results and Discussion

Theoretical Results

Top leads revealed by the virtual screening. The positions of 3699 ligands in the best docking mode for two targets are shown in Fig. S1 in File S1. In the case of 6A β_{9-40} all compounds are positioned inside the fibril and mainly near to the loop region. Most of them have contacts with peptides II – V. Only few ligands are located near terminals of peptides. The situation is very different in the case of 17A β_{17-42} , where binding sites are scattered not only inside but also outside of fibrils.

As follows from the distributions of binding energies obtained in the best docking mode (Fig. S2 in File S1), ligands show higher binding affinity toward 6A β_{9-40} compared to 5A β_{17-42} . This is presumably because they are mainly located outside 5A β_{17-42} . The most probable energies are about -6 and -8 kcal/mol for 5A β_{17-42} and 6A β_{9-40} , respectively.

We have made a ranking of ligands by their binding energies to two receptors. The 10 top leads are listed on Table S1 in File S1. Dia-aurantiamide acetate (ID: 30140) is a champion with $E_{\text{bind}} = -8.6$ kcal/mol to 5A β_{17-42} . In the case of 6A β_{9-40} Delavinone (ID: 32022) has the lowest binding energy. The common feature of 10 top leads is that they contain at least two rings which favor high binding affinity. Among them Dihydrochalcone is the lightest compound having weight of 200 Da. Moreover, the structure of this compound is similar (using the software SHAEP software (<http://users.abo.fi/mivainio/shaep/>) [27] one can show that the shape similarity between Dihydrochalcone and Curcumin is 67.12%) to that of curcumin undergoing the second phase of clinical trials. Both of them have two aromatic rings (Fig. 1A and Table S1 in File S1), which, as shown below, play a decisive role in binding affinity. From 10 top leads (Table S1 in File S1) we were able to purchase Dihydrochalcone, which is derived from *Daemonorops draco* tree

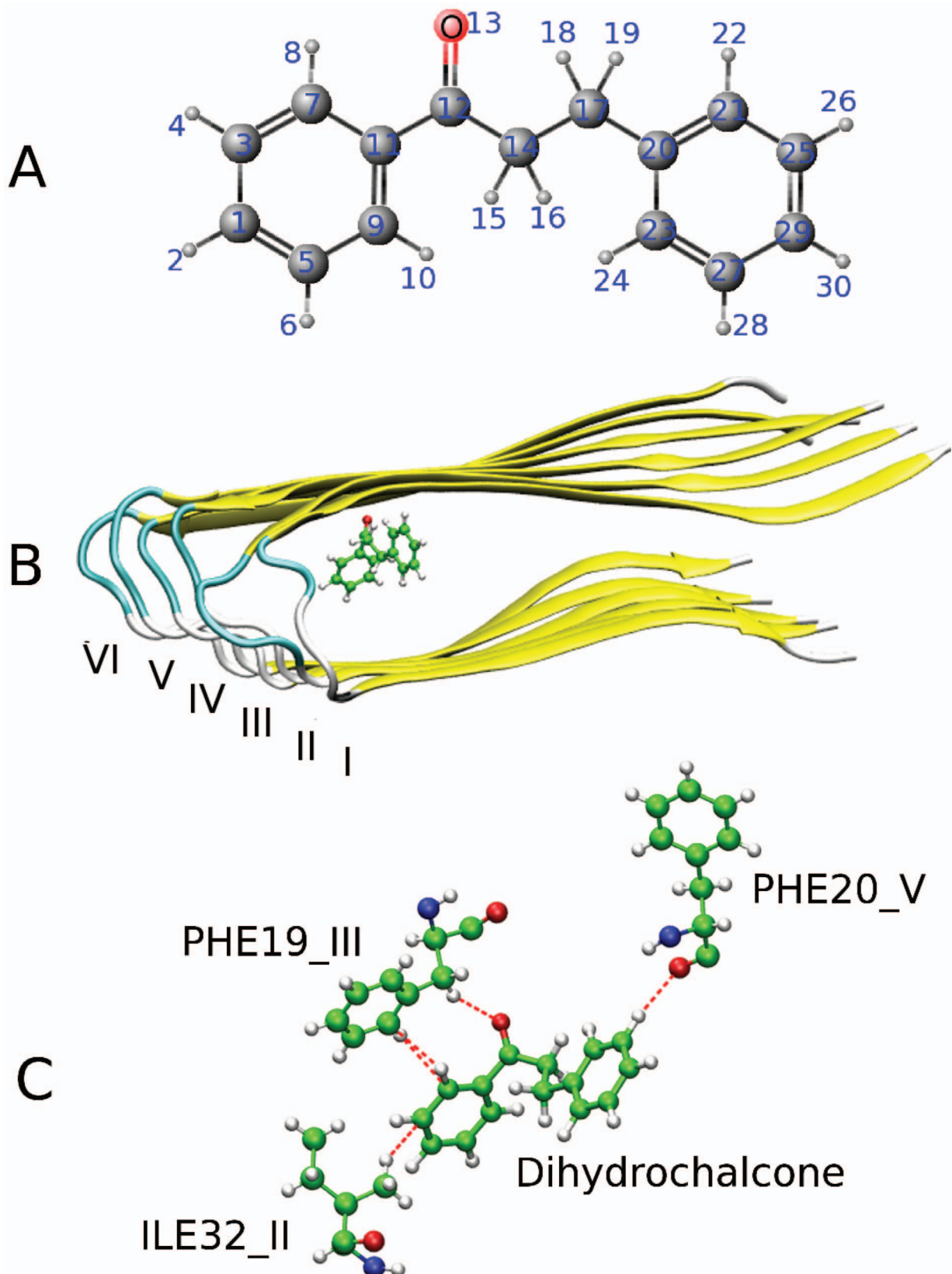


Figure 1. Chemical structure and the best docking pose of Dihydrochalcone. (A) Structure of Dihydrochalcone. (B) The best docking conformation of the $6A\beta_{9-40}$ -Dihydrochalcone complex. (C) Hydrogen bonds between Dihydrochalcone and fibril $6A\beta_{9-40}$ in the best docking mode. The ligand has 1, 3 and 1 hydrogen bonds with residues ILE32 of chain II, PHE19 of chain III and PHE20 of chain V, respectively.
doi:10.1371/journal.pone.0079151.g001

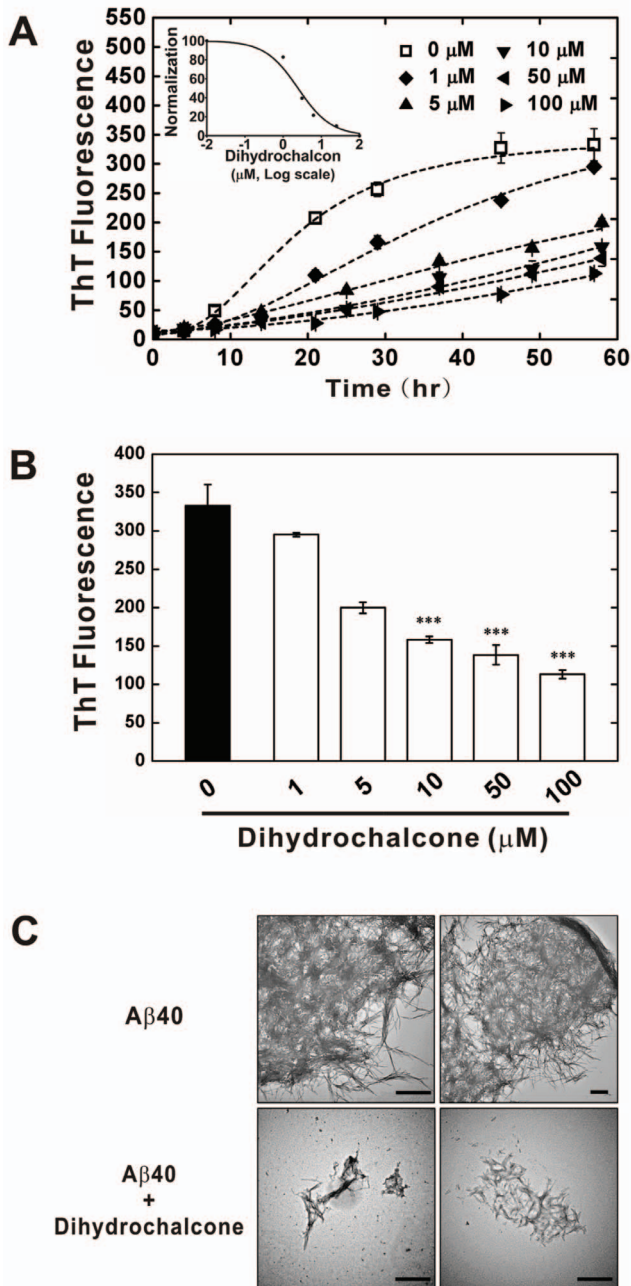


Figure 2. Dihydrochalcone suppresses Aβ fibrillization. (A) Aβ₄₀ (25 μM) was incubated at 25°C with different concentrations of Dihydrochalcone in the presence of ThT and the fibrillization was monitored by ThT fluorescence for 58 hr. Aβ in the absence (0 μM, □) and presence of various concentrations of Dihydrochalcone (1 μM, ◆; 5 μM, ▲; 10 μM, ▼; 50 μM, ◀; 100 μM, ▶). The final ThT intensity was plotted against Dihydrochalcone concentrations and shown in the inset (IC₅₀ = 2.46 μM) of panels (A) and (B). The mean value of the final ThT intensity and statistical significance was shown in panel (B). One-way ANOVAs, ***, P < 0.0001. (C) TEM images of the end-point products from ThT experiments. Scale bars = 500 nm. doi:10.1371/journal.pone.0079151.g002

(Fig. S3 in File S1), to perform *in vitro* study for its ability to prevent Aβ₄₀ aggregation. Therefore we consider this compound in more detail.

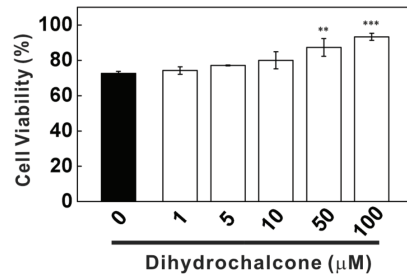


Figure 3. Cytotoxicity of the end-point products of Aβ fibrillization with and without Dihydrochalcone. HEK293 cells were treated with the end-point products of Aβ fibrillization with and without various concentrations of dihydro-chalcone as indicated in Figure 2 for 24 hr and subjected to MTT assay. Triplicate experiments were performed and the data were shown as mean ± standard deviation. The statistical significance was indicated by one-way ANOVAs (**, P < 0.005; ***, P < 0.0005). doi:10.1371/journal.pone.0079151.g003

Hydrogen network of dihydrochalcone. The 6Aβ₉₋₄₀-Dihydrochalcone complex in the best docking mode ($E_{\text{bind}} = -8$ kcal/mol) is shown in Fig. 1B. In this configuration Dihydrochalcone forms 1, 3 and 1 hydrogen bonds (HB) with residues ILE32 of chain II, PHE19 of chain III and PHE20 of chain V, respectively (Fig. 1C). Note that among all of five non-standard HBs one has three C-H...C and two C-H...O bonds that may be important for the interaction of small molecules with other molecules [28,29]. Fig. S4 (File S1) shows the dependence of binding energies of 3699 ligands to target 6Aβ₉₋₄₀ on the number of HBs. Since the correlation between two these quantities is very low one can conclude that the HB network is not strong enough to play the key role in the binding affinity to Aβ₄₀ fibrils. This conclusion is also valid for Aβ₄₂ target (results not shown).

Estimation of binding free energy of dihydrochalcone to 6Aβ₉₋₄₀ by MM-PBSA method. It is well known that the docking method is not accurate enough due to omission of ligand dynamics and a limited number of trial positions of ligand. Therefore we will use the MM-PBSA method which is more reliable in estimating the binding free energy ΔG_{bind} (Eq. (1)).

We use the conformation obtained in the best docking mode (Fig. 1B) as a starting conformation for MD simulation. Four independent 20 ns MD trajectories have been generated with different random seed numbers that are needed to create different pools for initial velocities of atoms. The time dependence of C_α root mean square displacement (RMSD) of the receptor 6Aβ₉₋₄₀ shows that the systems reaches equilibrium (curves reach saturation) after about 10 ns (Fig. S5 in File S1).

We have stored snapshots every 10 ps during last 10 ns for MM-PBSA calculation using Eq. (1). Although ΔG_{bind} is sensitive to MD runs (Table S2 in File S1) we have the clear trend that the van der Waals interaction dominates over the electrostatic interaction. The entropic and nonpolar (ΔG_{sur}) contributions are almost homogeneous over for all trajectories. Averaging over four MD runs we obtain $\Delta G_{\text{bind}} = -9.39$ kcal/mol which corresponds to the inhibition constant IC₅₀ ~ 1 μM. This result is in reasonable agreement with our experimental data (see below).

We have also studied the binding 3 ligands Delavinone, Sisalagenin and Sipeimine that can cross the blood brain barrier easily (log(BB) > 0, see below) using the MM-PBSA method. As in the case of Dihydrochalcone the systems reach equilibrium after about 10 ns (Fig. S5 in File S1). Within the error bars they have the binding free energy compatible to Dihydrochalcone (Table S2 in File S1). Therefore, these compounds also deserve further *in vitro* and *in vivo* studies.

Aromatic rings play the key role in ligand binding. We have considered the contributions of individual atoms of Dihydrochalcone to the electrostatic and van der Waals interactions with the receptor (Fig. S6 in File S1). The results have been obtained as averages over four MD trajectories. Except atom 11 (Fig. 1A) all Carbon atoms have the repulsive interaction with the receptor, while the Coulomb interaction with hydrogen atoms is attractive. The electrostatic interactions of aromatic rings are almost compensated (Fig. S6 in File S1). The electrostatic interaction between Oxygen atom with the receptor is the strongest one. The contribution of Carbon atoms 12 and 17 from the middle part (Fig. 1A) is more important than Carbon atom 14. Similar to Curcumin case [24], two aromatic rings are very important in binding affinity of Dihydrochalcone to fibril.

Ligand binding slows down the fibril growth process. In this section we discuss the impact of ligand binding on fibril assembly and disassembly at the qualitative level. As evident from Fig. 1, the ligand is located inside fibril leading to its stabilization. The fact that the ligand binding can slow down fibril assembly may be qualitatively understood as follows. Suppose that the fibril growth proceeds by addition of a nascent monomer to the preformed template [30,31]. Then the fluctuations of template would facilitate this process [31]. From this point of view the stabilization by ligand binding makes the template more rigid reducing the fibril formation rates as observed in our ThT fluorescence experiments.

The question about the influence of ligand binding on fibril disassembly looks more complicated. In our opinion, there is a possibility that ligands locate not only inside but also outside fibrils. Then outside ligands may destabilize fibrils, but this question requires further investigation.

Blood-Brain Barrier (BBB). The BBB is a physical barrier in the circulatory system that a compound should cross in order to travel into the central nervous area [32]. Thus the requirement of passing this barrier is necessary for any AD drug candidate. The crossing ability is measured by $\log(\text{BB})$ which is the logarithm base 10 of the ratio of the compound concentration in the brain to that in the blood. Using the quantitative structure-activity relationship (QSAR) implemented in the PreADME prediction software [33] we obtained $\log(\text{BB}) = 0.18$ for Dihydrochalcone. This implies that the crossing ability of Dihydrochalcone is much better than Curcumin which has $\log(\text{BB}) = -1.04$.

Experimental Results

To validate the potency of the natural compounds virtually screened from the data base, we performed the $A\beta_{40}$ fibrillization experiment with Dihydrochalcone that is commercially available. We first employed Thioflavin T (ThT) assay to detect $A\beta$ fibrillization in the presence of different concentrations of Dihydrochalcone. Five concentrations ranging from 1 to 100 μM of Dihydrochalcone were used to examine the inhibitory effect. In the absence of Dihydrochalcone, $A\beta$ rapidly increased in 7 hr and reached a steady state after 30 hr monitored by ThT assay (Fig. 2A). After addition of 1 μM Dihydrochalcone, the elongation phase of $A\beta$ fibrillization was significantly decreased. Upon increasing Dihydrochalcone, the ThT intensity at the steady

state decreased in a dose dependent manner. In the presence of 100 μM Dihydrochalcone, ThT intensity decreased to $\sim 66\%$ of that in the absence of Dihydrochalcone suggesting reduction of $A\beta$ fibrils. We plotted the intensity versus Dihydrochalcone concentration and calculated its IC_{50} to be $\sim 2.46 \mu\text{M}$ (Figure 2A, inset), that is comparable with our theoretical estimation and with that of curcumin [34]. The averaged final ThT intensity was plotted (Figure 2B). The data showed the inhibitive effect of Dihydrochalcone was statistically significant. We also compared the inhibitive effect of Dihydrochalcone with the reported $A\beta$ fibril inhibitors including curcumin, rosmarinic acid, and resveratrol (Figure S7 and S8 in File S1) by ThT assay. The IC_{50} of Dihydrochalcone was similar to those of curcumin and resveratrol but better than rosmarinic acid. To confirm the fibril quality and morphology, we used transmission electron microscopy (TEM) to examine the end-point products of the ThT experiments (Fig. 2C). In the absence of Dihydrochalcone, $A\beta$ fibrillized to clusters of thread-like fibrils in high density. With higher concentration of Dihydrochalcone, we observed much fewer amyloid fibrils by TEM and the clusters were much reduced. We further examined the cytotoxicity of the end-point products by MTT assay using HEK293 cells (Fig. 3). In the absence of Dihydrochalcone, $A\beta$ treated cells have 72.5% viability than the buffer control. The cytotoxicity was significantly reduced with higher Dihydrochalcone concentrations. The cell viability was increased to 93.2% in the presence of 100 μM Dihydrochalcone comparing to the buffer control. The result clearly showed that Dihydrochalcone can inhibit $A\beta$ aggregation and toxicity.

Conclusion

By virtual screening we have sorted out the most potent candidates for AD from the large data base of natural products. *In silico* and *in vitro* studies clearly show that the extract from *Daemonorops draco* tree Dihydrochalcone satisfies requirements for a AD drug such as the binding affinity, BBB crossing ability and non-toxicity. Therefore we recommend it for further *in vivo* study and possible clinical trials.

Supporting Information

File S1 Lipinski's rule of five; docking method; molecular dynamics simulation and supporting tables and figures.

(PDF)

Acknowledgments

We are very thankful to R. Tycko for providing the atomic structure of $6A\beta_{9-40}$ and P.D.Q. Huy for useful discussions. Allocation of CPU time at the supercomputer center TASK in Gdansk (Poland) is highly appreciated.

Author Contributions

Conceived and designed the experiments: MSL YRC. Performed the experiments: MHV CYC MSL YRC. Analyzed the data: MHV CYC CKH MSL YRC. Contributed reagents/materials/analysis tools: CKH MSL YRC. Wrote the paper: MSL YRC.

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