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Tracing the '9th Sulfur' of the Nitrogenase Cofactor via a Semi-Synthetic Approach

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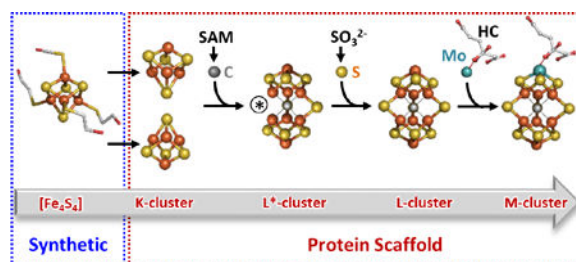
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Abstract

The M-cluster is the [(homocitrate)MoFe₇S₉C] active site of nitrogenase that is derived from an 8Fe core assembled via coupling and rearrangement of two [Fe₄S₄] clusters concomitant with the insertion of an interstitial carbon and a '9th sulfur'. Combining synthetic [Fe₄S₄] clusters with an assembly protein template, here we show that sulfite can give rise to the '9th sulfur' that is incorporated in the catalytically important belt region of the cofactor after the radical SAM-dependent carbide insertion and the concurrent 8Fe-core rearrangement have already taken place. Based on the differential reactivity of the formed cluster species, we also propose a new [Fe₈S₈C] cluster intermediate, the L*-cluster, that is similar to the [Fe₈S₉C] L-cluster but lacks the '9th S' from sulfite. This work provides a semi-synthetic tool for protein reconstitution that could be widely applicable for the functional analysis of other FeS systems.

Graphical abstract



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Author contributions

K.T. (Irvine), C.C.L., N.S.S., Y.H. and M.W.R. designed and analyzed experiments. K.T. (Irvine) performed experiments. K.T. (Nagoya) and Y.O. provided materials. Y.H. and M.W.R. analyzed experiments and wrote the manuscript with input from all authors.

Competing interests

The authors declare no competing financial and non-financial interests.

Nitrogenase plays a key role in the global nitrogen cycle, catalyzing the ambient reduction of nitrogen to ammonia at its active cofactor site.^{1,2} Designated the M-cluster (Fig. 1a, b), the cofactor of the Mo-nitrogenase consists of [MoFe₃S₃] and [Fe₄S₃] subclusters that are bridged by three μ_2 -‘belt’ sulfides (S²⁻) and one μ_6 -interstitial carbide (C⁴⁻), and it is further coordinated by homocitrate at its Mo end.³⁻⁶ Assembly of the M-cluster has attracted considerable attention because it is biologically important and chemically unprecedented. Previously, we have shown that the M-cluster is assembled through (i) formation of an L-cluster ([Fe₈S₉C]), an 8Fe core of the M-cluster, via coupling and rearrangement of a K-cluster (2x[Fe₄S₄]) concomitant with incorporation of an interstitial carbide and a ‘9th sulfur (S)’ on NifB, a radical S-adenosyl-L-methionine (SAM) enzyme; (ii) transformation of the L-cluster into a mature M-cluster via insertion of Mo and homocitrate on NifEN, an assembly scaffold; and (iii) transfer of the M-cluster to its binding site in NifDK, the catalytic component of the Mo nitrogenase (Fig. 1a, b).⁷⁻¹² Further, we have identified early steps along the carbide insertion pathway that involve attachment of the methyl group of SAM to a sulfide atom of the K-cluster, followed by abstraction of a hydrogen atom from this methyl group by a SAM-derived 5'-deoxyadenosyl radical (5'-dA•) and further processing of the resultant carbon-radical species into an interstitial carbide (Fig. 1a, b).⁷⁻⁹

While these studies have advanced our understanding of the nitrogenase cofactor assembly, certain crucial details of this process, such as the 4Fe-modular nature of the K-cluster, the origin of the ‘9th S’, and the sequence of events between C and S insertion, have remained unclear. Study of homologously expressed NifB from *Azotobacter vinelandii*, the organism most studied in nitrogenase research, has historically been hampered by an apparent instability and inactivity of the isolated enzyme, limiting its study to a few reports that used either the NifEN-B fusion protein⁷⁻⁹ or heterologous expression.^{13,14} Recently, we have established that the NifB protein from *Methanosarcina acetivorans* (designated *MaNifB*) is a functional homolog of its counterpart from *A. vinelandii* that converts the K-cluster to an L-cluster *in vitro* and subsequently donates the L-cluster to the heterologous biosynthetic machinery of *A. vinelandii* for further maturation into an M-cluster.¹³ Identification of a functional methanogen NifB homolog that can be expressed in *Escherichia coli* circumvents the difficulty of obtaining a stable, active form of the *A. vinelandii* NifB protein and presents a more simplified system than the *A. vinelandii* NifEN-B fusion protein.⁷⁻⁹ However, traditional reconstitution of FeS clusters (i.e., using FeCl₃ and Na₂S) for the as-isolated, cluster-incomplete *MaNifB* cannot directly facilitate the assessment of whether the K-cluster consists of two 4Fe modules; rather, this method often results in attachment of excess sulfide aggregates to the FeS clusters in the protein, making it impossible to trace the incorporation of the ‘9th S’ or uncouple this event from that of carbide insertion.

Here, we use synthetic [Fe₄S₄] clusters to reconstitute *MaNifB* and trace the origin of the ‘9th S’, which allows us to define the sequence of events between carbide- and sulfur-insertion during cofactor assembly while establishing a semi-synthetic approach that could be extended to functional analyses of other FeS systems.

Results & Discussion

In search of an alternative FeS reconstitution approach, we first examined whether a water-soluble, synthetic $[\text{Fe}_4\text{S}_4]$ compound ($[\text{Fe}_4\text{S}_4(\text{SCH}_2\text{CH}_2\text{OH})_4]^{2-}$, designated $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$; Fig. 1c) could be used to reconstitute NifH, the obligate reductase component of Mo-nitrogenase. The water-stable $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ contains an $[\text{Fe}_4\text{S}_4]$ core coordinated by ligands (*i.e.*, - $\text{SCH}_2\text{CH}_2\text{OH}$) that are exchangeable with protein-bound cysteines^{15,16} and, therefore, is a suitable agent for ‘clean’ FeS reconstitutions. Indeed, a dark-brown protein (designated NifH ^{$[\text{Fe}_4\text{S}_4]$}) could be re-isolated following incubation of the FeS-depleted apo-NifH (designated NifH^{apo}) with $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ clusters (Supplementary Fig. 1). The $[\text{Fe}_4\text{S}_4]$ clusters in NifH ^{$[\text{Fe}_4\text{S}_4]$} and the as-isolated holo-NifH (designated NifH^{holo}) were indistinguishable from each other, both displaying the same characteristic EPR features in the 0, +1 and +2 oxidation states and undergoing the same line-shape changes of EPR spectra upon nucleotide binding (Supplementary Fig. 2). Moreover, NifH ^{$[\text{Fe}_4\text{S}_4]$} and NifH^{holo} displayed nearly identical catalytic profiles when titrated against increasing amounts of the catalytic NifDK component and almost indistinguishable activities in substrate reductions (Supplementary Fig. 2).

Having established a proof-of-concept for the utility of $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ clusters in FeS reconstitution, we then generated a $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ -reconstituted form of *Ma*NifB (designated *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$}) by re-isolating the protein from an incubation mixture of apo-*Ma*NifB (designated *Ma*NifB^{apo}) and $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ clusters. Like the $\text{FeCl}_3/\text{Na}_2\text{S}$ -reconstituted *Ma*NifB,¹³ *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$} cleaved SAM into two products: (*i*) SAH, a product formed upon removal of the methyl group from SAM; and (*ii*) 5'-dAH, a product formed upon hydrogen abstraction from the SAM-derived methyl group by 5'-dA• (Fig. 2a). Moreover, consistent with the presence of multiple $[\text{Fe}_4\text{S}_4]$ clusters (*i.e.*, a SAM cluster and a K-cluster), *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$} displayed a composite, $[\text{Fe}_4\text{S}_4]^{1+}$ -specific $S=1/2$ EPR features in the dithionite (DT)-reduced state (Fig. 2b, *trace 1*) that became EPR-silent upon indigo disulfonate (IDS)-oxidation to the $[\text{Fe}_4\text{S}_4]^{2+}$ state (Fig. 2b, *trace 3*). In the presence of SAM, the composite $S=1/2$ signal of *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$} underwent changes in the DT-reduced state (Fig. 2b, *trace 2*) concomitant with the appearance of an 8Fe core-specific $g=1.92$ signal in the IDS-oxidized state (Fig. 2b, *trace 4*), which was indicative of transformation of a K-cluster ($2x[\text{Fe}_4\text{S}_4]$) into an L-cluster ($[\text{Fe}_8\text{S}_9\text{C}]$) via cluster coupling/rearrangement and carbon/sulfur insertion (Fig. 1a, b).^{7-9,13} Such a transformation was further confirmed by an *in vitro* cluster maturation assay, wherein the K-cluster on *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$} was converted into an L-cluster upon SAM treatment, followed by the L- to M-cluster maturation on NifEN and the subsequent transfer of the M-cluster to NifDK, resulting in a reconstituted NifDK protein that was active in substrate reduction (Fig. 2c). It should be noted that the maturation protein NifEN was homologously expressed in *A. vinelandii* and that the successful cluster transfer between this protein and the archaeal *Ma*NifB demonstrates an effective strategy to mix and match nitrogenase proteins from different kingdoms.

The observation that an 8Fe L-cluster could be generated from two $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ clusters on *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$} provided direct evidence that the K-cluster consists of a $[\text{Fe}_4\text{S}_4]$ cluster pair; furthermore, it suggested the possibility of using *Ma*NifB ^{$[\text{Fe}_4\text{S}_4]$} to investigate the origin of the ‘9th S’ without complications caused by excess sulfide in the $\text{FeCl}_3/\text{Na}_2\text{S}$ -reconstituted

*Ma*NifB. However, DT ($S_2O_4^{2-}$), the non-physiological reductant traditionally used in nitrogenase research, is known to break down into a number of S-based products such as sulfide (S^{2-}), sulfite (SO_3^{2-}) and sulfate (SO_4^{2-}),^{17,18} all of which could potentially donate the ‘9th S’ source. To prevent introduction of sulfur species by DT and its breakdown products, we used europium(II) ethyleneglycoltetraacetate (Eu^{II} -EGTA, $E^0 = -0.8$ V at pH 8)¹⁹ as a reductant in the cluster maturation assay to examine the ability of three physiologically relevant sulfur forms— S^{2-} , SO_3^{2-} , and SO_4^{2-} —to serve as the source of the ‘9th S’ for the K- to L-cluster conversion on *Ma*NifB^[Fe₄S₄]. Interestingly, only SO_3^{2-} supported cluster conversion on *Ma*NifB^[Fe₄S₄] in the presence of SAM, resulting in the formation of an L-cluster that could be further matured into an M-cluster to reconstitute NifDK (Supplementary Fig. 3). More excitingly, upon substitution of $^{35}SO_3^{2-}$ for unlabeled SO_3^{2-} in the same incubation mixture, the ^{35}S radiolabel could be detected in the re-purified *Ma*NifB^[Fe₄S₄] (Fig. 3a, left) and further traced to the L-cluster extracted from *Ma*NifB^[Fe₄S₄] (Fig. 3a, middle). These experiments provide the very first evidence that a DT breakdown product, SO_3^{2-} , can serve as the external ‘9th S’ source. Notably, when conducted in the presence of DT, the *in vitro* cluster maturation assays do not require supplementation of SO_3^{2-} (Fig 2c; Supplementary Fig. 3), which is consistent with an accumulation of SO_3^{2-} upon decomposition of DT in aqueous solutions.^{17,18} The extracted L-cluster could then be used in a maturation assay in which it was first converted to an M-cluster on the scaffold protein NifEN and subsequently used for the reconstitution of NifDK (Fig. 3a, right). In the absence of SAM, however, no significant retention of the ^{35}S radiolabel was observed when $^{35}SO_3^{2-}$ was supplied (Fig. 3a), suggesting that the insertion of the SAM-derived interstitial carbide likely precedes the incorporation of the ‘9th S’.

To further tackle the sequence of events between C and S insertion, we monitored cluster conversion on *Ma*NifB^[Fe₄S₄] in the presence of SAM alone or SAM plus SO_3^{2-} . When incubated with SAM, *Ma*NifB^[Fe₄S₄] displayed the same SAM cleavage pattern (Fig. 3b) and EPR spectroscopic changes (Fig. 3c, d) with or without SO_3^{2-} treatment. However, quenching those incubation mixtures with acid to release methanethiol—a technique previously used to detect the SAM-derived methyl group attached to a K-cluster S atom⁹—liberated substantially less methanethiol in the presence of SO_3^{2-} (Supplementary Fig. S4), suggesting an improved stability and, consequently, a reduced acid-susceptibility of the cluster species in the SO_3^{2-} -treated *Ma*NifB^[Fe₄S₄]. Additionally, the SO_3^{2-} -untreated *Ma*NifB^[Fe₄S₄] was inactive as an L-cluster donor, whereas the SO_3^{2-} -treated *Ma*NifB^[Fe₄S₄] was active in donating the L-cluster for further maturation into an M-cluster (Fig. 3e). The observation of the 8Fe-core-specific $g=1.92$ signal in SO_3^{2-} -untreated *Ma*NifB^[Fe₄S₄] after reaction with SAM (Fig. 3d, trace 2) is particularly intriguing, as it implies that the SAM-dependent carbon insertion and the concurrent 8Fe core rearrangement occur prior to the insertion of the ‘9th S’. Moreover, K-cluster fusion with C in the absence of SO_3^{2-} leads to the formation of a new intermediate species (designated the L*-cluster) that we tentatively assign as a [Fe₈S₈C] cluster that closely resembles the L-cluster in architecture but lacks the ‘9th S’ in the catalytically-important belt region of the cluster (Fig. 4). This so-called ‘vacant’ site of the L*-cluster may be occupied by a putative cysteine thiolate or H₂O ligand, and studies are underway to assess the exact L*-cluster composition and coordination.

Our proposal that the '9th S' is inserted when the inorganic core structure of the M-cluster is in place is consistent with the facile exchange of the belt S of the M-cluster with selenium (Se)²⁰ or replacement of a belt S with CO²¹ under turnover conditions (Supplementary Fig. 5). Interestingly, analogous to the utilization of an oxidized form of S (SO₃²⁻) for the incorporation of the '9th S', the belt S of the M-cluster can only exchange with the Se atom in selenocyanate (SeCN⁻) but not with the more reduced Se in selenide (Se²⁻) upon turnover.²⁰ It is possible, therefore, that insertion of the '9th S' involves donation of electrons from the electron-rich cluster to the sulfur source, which explains why a more oxidized S species (SO₃²⁻)—instead of the fully reduced S (S²⁻)—is used as a source of the '9th S'. In the case of SO₄²⁻, however, the S atom is unable to directly interact with the Fe atoms; additionally, the cluster may be poised at a certain redox potential that does not accommodate the redox change required for the conversion of SO₄²⁻ to a '9th S'.

Within the cell, sulfite is one of the central hubs of sulfur metabolism (Supplementary Fig. 6).²²⁻²⁴ A survey of the genomes of 13 nitrogen-fixing microbes,²⁵ including those of *A. vinelandii* and *M. acetivorans*, revealed the presence of genes encoding enzymes involved in various sulfite-generating routes; most notably, more than 10 of these organisms possess genes encoding the 3'-phosphoadenosine 5'-phosphosulfate sulfotransferase (PAPS transferase), which converts PAPS to sulfite as part of the assimilatory sulfate reduction pathway (Supplementary Fig. 6). This observation provides support for the physiological relevance of sulfite or related compounds to nitrogenase assembly. Although we cannot rule out the participation of reactive physiological sulfur sources other than SO₃²⁻ such as persulfides for the *in vivo* '9th S' delivery,²⁶ it should be noted that persulfides are not stable under reducing *in vitro* conditions. Nevertheless, our identification of shared sulfite-producing enzymes among multiple diazotrophs suggests that these pathways could be operative for supplying the '9th S' source to NifB. While details of these events are yet to be elucidated, identification of sulfite as the *in vitro* source of '9th S' that is inserted at a late stage of cofactor assembly suggests the possibility to specifically label this belt S for future mechanistic studies of nitrogenase (Supplementary Fig. 5).

Moreover, the methodology of using synthetic metalloclusters as reported herein could be applicable to a wide array of FeS-containing enzyme classes. The [Fe₄S₄]^{Syn} cluster could prove particularly useful for reconstituting other heterologously expressed proteins in the radical SAM superfamily²⁷ as well as other [Fe₄S₄]-cluster-containing enzymes associated with important processes such as DNA synthesis and gene regulation.²⁸⁻³⁰ Strategies to incorporate synthetic cofactors within apoproteins have been successfully employed in the hydrogenase field,³¹⁻³³ to gain mechanistic and biosynthetic insight. These approaches can also be used to generate artificial enzymes with altered reactivities.³⁴⁻³⁶

Conclusions

In summary, we have demonstrated the utility of a water stable and soluble [Fe₄S₄] cluster to reconstitute the nitrogenase protein NifB for the determination of the L-cluster '9th S' source *in vitro*. In the process, we have discovered that SO₃²⁻, and not S²⁻ or SO₄²⁻, can donate an S atom for this purpose, and that previous successful studies of cluster maturation on NifB appear to have benefited from the presence of the reductant DT, which can potentially

provide SO_3^{2-} to the system. More excitingly, we have identified for the first time a putative $[\text{Fe}_8\text{S}_8\text{C}]$ intermediate, the L^* -cluster, that can be matured to the L-cluster upon addition of SO_3^{2-} . Further investigation of the L^* -cluster on NifB is an ongoing pursuit.

Methods

Methods other than protein reconstitution and ^{35}S -labeling experiments are included in Supplementary Information.

Protein Reconstitution

Purified NifH^{holo} or *Ma*NifB was treated with 20 mM bathophenanthroline disulfonate, an iron chelator, in a buffer containing 5 mM MgATP, 2 mM dithionite (DT; $\text{Na}_2\text{S}_2\text{O}_4$), 50 mM Tris-HCl (pH 8.0) and 500 mM NaCl, followed by incubation at room temperature for 1 h. The mixture was diluted with a buffer containing 50 mM Tris-HCl (pH 8.0) and loaded on a Q Sepharose column (GE Healthcare). Subsequently, the column was washed with a buffer containing 2 mM DT, 50 mM Tris-HCl (pH 8.0) and 100 mM NaCl prior to elution of apo-NifH (designated NifH^{apo}) or apo-*Ma*NifB (designated *Ma*NifB^{apo}) with a buffer containing 50 mM Tris-HCl (pH 8.0). Reconstitution of NifH^{apo} or *Ma*NifB^{apo} with $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ was carried out by adding a dimethylformamide (DMF) solution of $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ dropwise to NifH^{apo} or *Ma*NifB^{apo} at molar ratios of 1:1 and 5:1, respectively, in a buffer containing 2 mM DT, 20 mM BME, 50 mM Tris-HCl (pH 8.0) and 500 mM NaCl, with continuous stirring on ice. After incubation on ice for 1 h, the reaction mixture was diluted with a buffer containing 2 mM DT and 50 mM Tris-HCl (pH 8.0) and loaded on a Q Sepharose column. The column was then washed with a buffer containing 2 mM DT, 50 mM Tris-HCl and 100 mM NaCl prior to elution of the reconstituted NifH (designated NifH^[Fe₄S₄]) or *Ma*NifB (designated *Ma*NifB^[Fe₄S₄]) with a buffer containing 2 mM DT, 50 mM Tris-HCl (pH 8.0) and 500 mM NaCl. The DT-free reconstitution of *Ma*NifB^{apo} was carried out by passing *Ma*NifB^{apo} through a Sephadex G-25 desalting column (GE Healthcare) to remove DT, diluting the protein with a buffer containing 20 mM BME, 50 mM Tris-HCl (pH 8.0), and treating the protein solution with a DMF solution of $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ as described above. The mixture was directly loaded on a Q Sepharose column after incubation on ice for 30 min, and the column was washed with a buffer containing 0.5 mM Eu^{II} -EGTA, 50 mM Tris-HCl and 100 mM NaCl prior to elution of the reconstituted NifH^[Fe₄S₄] or *Ma*NifB^[Fe₄S₄] with a buffer containing 2 mM Eu^{II} -EGTA, 50 mM Tris-HCl (pH 8.0) and 500 mM NaCl. All solutions described above contained 10% (v/v) glycerol.

^{35}S -labeling Experiments—To monitor the event of ‘9th S’ insertion on *Ma*NifB^[Fe₄S₄] (see Fig. 3a, *left*), three different reactions were assembled, each containing, in a total volume of 40 μL , 2 mM Eu^{II} -EGTA, 2 mM $\text{Na}_2^{35}\text{SO}_3$, 10% (v/v) glycerol, 50 mM Tris-HCl (pH 8.0) and (i) 22 nmol DT-free *Ma*NifB^{apo}; (ii) 22 nmol DT-free *Ma*NifB^[Fe₄S₄]; and (iii) 22 nmol DT-free *Ma*NifB^[Fe₄S₄] and 5 mM SAM. All reactions were incubated for 30 min at room temperature with intermittent mixing and then run over Ni Sepharose resin (25 μL packed volume; GE Healthcare) pre-equilibrated with a buffer containing 2 mM Eu^{II} -EGTA, 500 mM NaCl, 10% (v/v) glycerol and 25 mM Tris-HCl (pH 8.0). The Ni Sepharose resin was washed with 100 μL of the same equilibration buffer before the bound protein was

eluted with 80 μ L buffer containing 2 mM Eu^{II} -EGTA, 250 mM imidazole, 500 mM NaCl, 10% (v/v) glycerol and 25 mM Tris-HCl (pH 8.0). The protein eluent was then diluted with 270 μ L equilibration buffer and mixed with Ni Sepharose resin (12.5 μ L packed volume; GE Healthcare). Subsequently, the protein-bound Ni Sepharose resin was re-suspended in the equilibration buffer and applied directly onto a Whatman Grade 1 qualitative filter paper (GE Healthcare). The blots were dried and exposed to a GE Healthcare Storage Phosphor Screen GP (20 \times 25 cm) for 16 h before imaging was performed on a GE Healthcare Typhoon Trio⁺ variable mode imager. To trace the ³⁵S radiolabel into the *Ma*NifB-bound cluster species (see Fig. 3a, *middle*), reactions *ii* and *iii* above were scaled up by 6-fold, followed by extraction of cluster species from these reaction mixtures using a protocol adapted from a previously described procedure.⁸ The extracted clusters were then blotted onto a GE Healthcare Whatman filter paper for imaging as described above.

Data Availability

The authors declare that all data supporting the findings of this study are available within the article and its Supplementary Information files and from the corresponding authors upon reasonable request.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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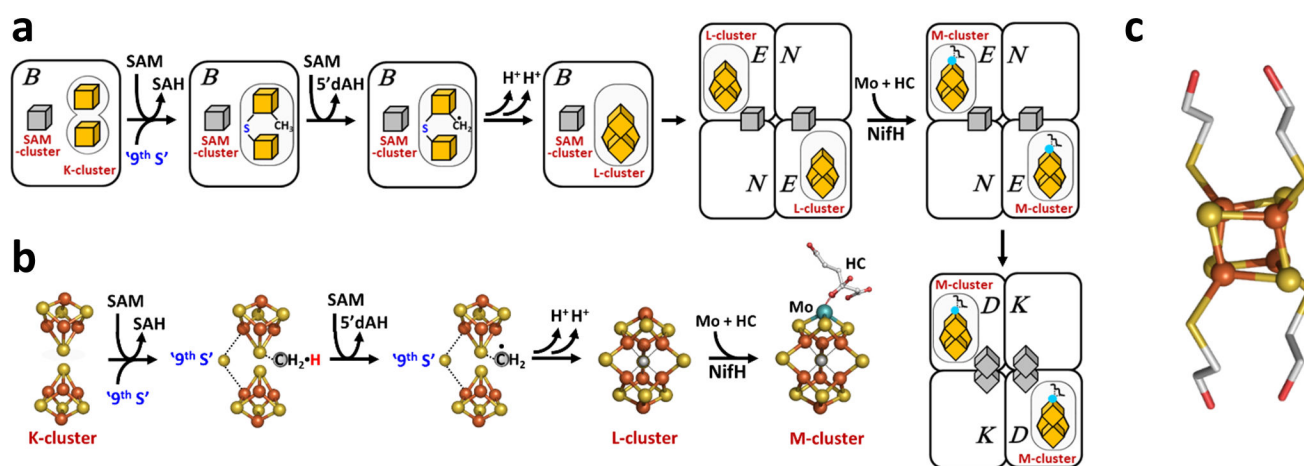


Figure 1. Assembly of the M-cluster

(a) The assembly of M-cluster involves SAM-dependent conversion of K-cluster to L-cluster on NifB, transfer of L-cluster to NifEN, maturation of L-cluster on NifEN upon insertion of Mo and homocitrate (HC) by NifH, and transfer of the resultant M-cluster to NifDK. (b) Coupling of the 4Fe units of K-cluster ($2 \times [\text{Fe}_4\text{S}_4]$) into an 8Fe L-cluster ($[\text{Fe}_8\text{S}_9\text{C}]$) concomitant with insertion of an interstitial carbide and a '9th sulfur', followed by conversion of the L-cluster to a mature M-cluster ($[\text{MoFe}_7\text{S}_9\text{C}]$) via insertion of Mo and HC. Carbide insertion begins with methyl transfer from SAM to the K-cluster and hydrogen abstraction from the SAM-derived methyl group by 5'-dA• and continues with further deprotonation/dehydrogenation of the carbon-radical until a carbide appears in the center of the L-cluster. (c) The $[\text{Fe}_4\text{S}_4]^{\text{Syn}}$ cluster ($[\text{Fe}_4\text{S}_4(\text{SCH}_2\text{CH}_2\text{OH})_4]^{2-}$) with ligands that are exchangeable for protein-bound cysteines. The hydrogen atoms are omitted for clarity. Color code: Fe, orange; S, yellow; Mo, cyan; C, gray; N, blue; O, red.

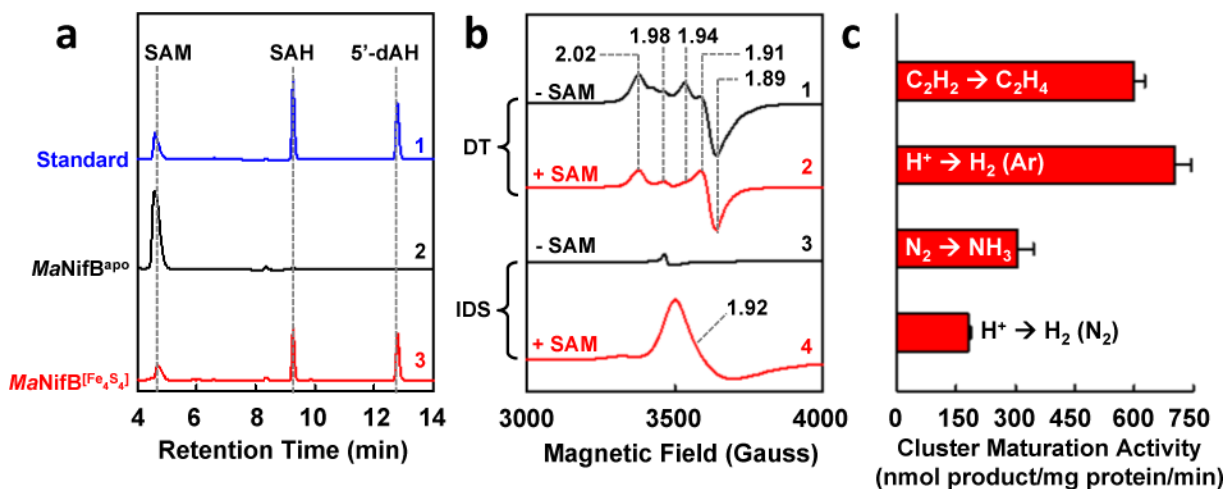


Figure 2. Dithionite (DT)-dependent cluster maturation

(a) HPLC elution profiles of (1) SAM, SAH and 5'-dAH standards, (2) SAM incubated with *MaNifB*^{apo} and DT, and (3) SAM incubated with *MaNifB*^[Fe₄S₄] and DT, showing that contrary to *MaNifB*^{apo}, *MaNifB*^[Fe₄S₄] was capable of methyltransfer from one SAM molecule and hydrogen abstraction from this methyl group via a 5'-dA• radical derived from another SAM molecule, generating SAH and 5'-dAH as the respective products of these reactions. (b) EPR spectra of DT-reduced (1, 2) or IDS-oxidized (3, 4) *MaNifB*^[Fe₄S₄] incubated without (black) or with (red) SAM, demonstrating the formation of an L-cluster on *MaNifB*^[Fe₄S₄] through a partial disappearance of the composite *S*=1/2 signal in the DT-reduced state (3) and the concomitant appearance of an L-cluster-specific, *g*=1.92 signal in the IDS-oxidized state (4). (c) Activities of *MaNifB*^[Fe₄S₄] upon maturation and transfer of its cluster species to NifDK, showing the competence of *MaNifB*^[Fe₄S₄] as an M-cluster donor for NifDK upon cluster maturation. Activity data were obtained from three independent experiments (*n*=6) and presented as mean±s.d. (c).

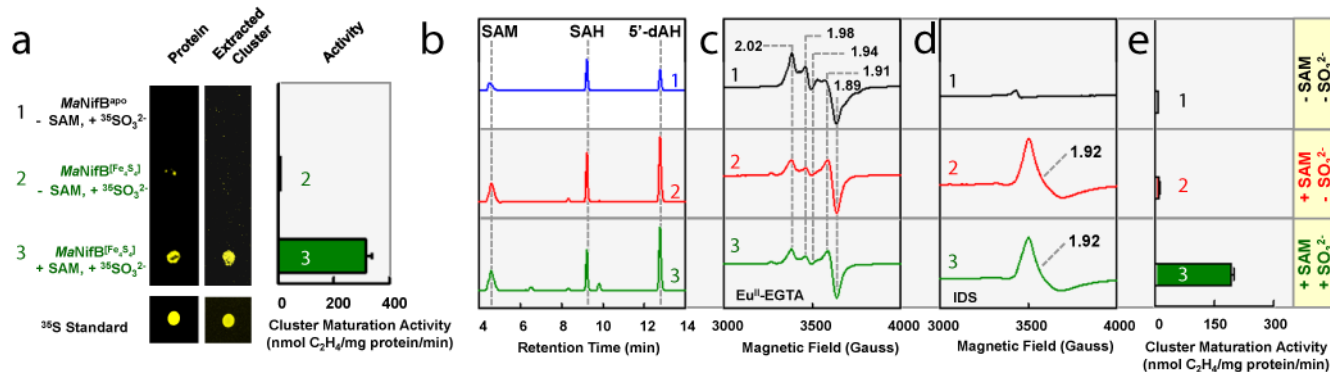


Figure 3. Dithionite (DT)-free cluster maturation

(a) Proteins captured on Ni Sepharose resin after incubation of (1) His-tagged *MaNifB*^{apo} with ³⁵SO₃²⁻, (2) His-tagged *MaNifB*^[Fe₄S₄] with ³⁵SO₃²⁻, and (3) His-tagged *MaNifB*^[Fe₄S₄] with SAM and ³⁵SO₃²⁻ (left); clusters extracted from samples a, 2 and a, 3 (middle); and activities of C₂H₂-reduction by clusters extracted from samples a, 2 and a, 3, after maturation and transfer of clusters to NifDK (right). The ³⁵S label and activity were only detected in SAM/SO₃²⁻-treated *MaNifB*^[Fe₄S₄], suggesting incorporation of S from SO₃²⁻ into a fully-assembled L-cluster. (b) HPLC elution profiles of (1) SAM, SAH and 5'-dAH standards, (2) SAM incubated with *MaNifB*^[Fe₄S₄], and (3) SAM incubated with *MaNifB*^[Fe₄S₄] and SO₃²⁻, showing the competence of SAM-treated *MaNifB*^[Fe₄S₄] in methyltransfer and hydrogen abstraction with or without SO₃²⁻-treatment. (c, d) EPR spectra of Eu^{II}-EGTA-reduced (c) or IDS-oxidized (d) *MaNifB*^[Fe₄S₄] incubated without SAM (black), with SAM (red), or with SAM and SO₃²⁻ (green), suggesting formation of an 8Fe core of the L-cluster in SAM-treated *MaNifB*^[Fe₄S₄] with or without SO₃²⁻. (e) Activities of *MaNifB*^[Fe₄S₄] treated with (1) no additive, (2) SAM, and (3) SAM and SO₃²⁻ upon maturation and transfer of clusters to NifDK, showing the sole competence of SAM/SO₃²⁻-treated *MaNifB*^[Fe₄S₄] as an M-cluster donor. Activity data were obtained from three independent experiments (*n*=6) and presented as mean±s.d. (e).

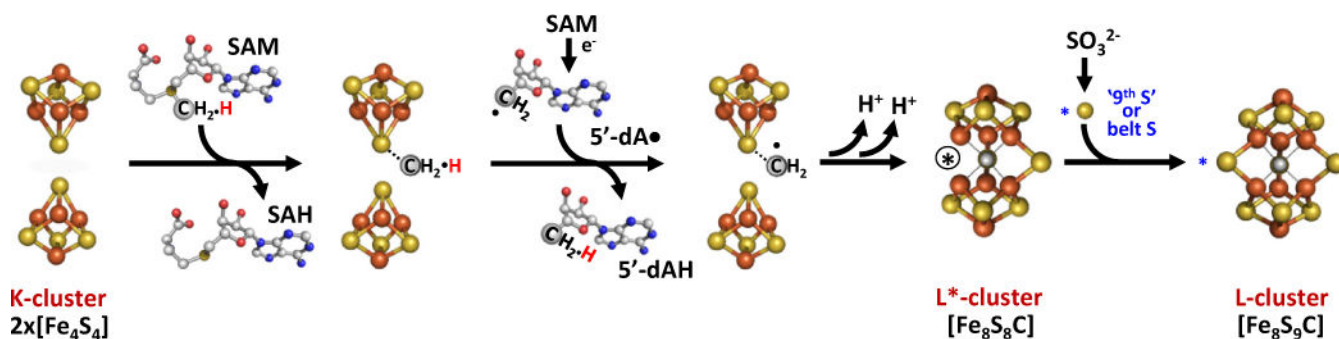


Figure 4. Refined model of L-cluster assembly on NifB

The radical SAM-dependent carbide insertion and the concurrent 8Fe core rearrangement precede the incorporation of the ‘9th S’ at the catalytically important belt region of the nitrogenase cofactor, resulting in an L*-cluster ($[\text{Fe}_8\text{S}_8\text{C}]$) that is nearly indistinguishable from the L-cluster ($[\text{Fe}_8\text{S}_9\text{C}]$) except for the missing ‘9th S’. The encircled black asterisk represents the ‘vacant’ site of the L*-cluster, which may be occupied by a cysteine thiolate or water molecule. Color code: Fe, orange; Mo, cyan; S, yellow; C, gray; N, blue; O, red.