



## Article

# Improved Performance of NbO<sub>x</sub> Resistive Switching Memory by In-Situ N Doping

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**Abstract:** Valence change memory (VCM) attracts numerous attention in memory applications, due to its high stability and low energy consumption. However, owing to the low on/off ratio of VCM, increasing the difficulty of information identification hinders the development of memory applications. We prepared N-doped NbO<sub>x</sub>:N films (thickness = approximately 15 nm) by pulsed laser deposition at 200 °C. N-doping significantly improved the on/off ratio, retention time, and stability of the Pt/NbO<sub>x</sub>:N/Pt devices, thus improving the stability of data storage. The Pt/NbO<sub>x</sub>:N/Pt devices also achieved lower and centralized switching voltage distribution. The improved performance was mainly attributed to the formation of oxygen vacancy (V<sub>O</sub>) + 2N clusters, which greatly reduced the ionic conductivity and total energy of the system, thus increasing the on/off ratio and stability. Moreover, because of the presence of V<sub>O</sub> + 2N clusters, the conductive filaments grew in more localized directions, which led to a concentrated distribution of SET and RESET voltages. Thus, in situ N-doping is a novel and effective approach to optimize device performances for better information storage and logic circuit applications.

**Keywords:** NbO<sub>x</sub>; N-doping; memory device; resistance switching; oxygen vacancy



**Citation:** Xu, J.; Zhu, Y.; Liu, Y.; Wang, H.; Zou, Z.; Ma, H.; Wu, X.; Xiong, R. Improved Performance of NbO<sub>x</sub> Resistive Switching Memory by In-Situ N Doping. *Nanomaterials* **2022**, *12*, 1029. <https://doi.org/10.3390/nano12061029>

Academic Editor: Marc Cretin

Received: 29 January 2022

Accepted: 16 March 2022

Published: 21 March 2022

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## 1. Introduction

Resistive random access memory (RRAM) is highly popular, owing to its high write and erase speeds, high storage density, and multi-level storage; additionally, it is considered the most potential candidate for next-generation memory [1–7]. Several studies have reported the excellent performance of RRAM, such as MoS<sub>2</sub> filament transistors with high on/off ratio ( $2.6 \times 10^9$ ), power-efficient h-BN memories, and cellulose memories with a significant on/off ratio ( $10^6$ ) and low SET/RESET voltages (<0.5 V) [8–10]. However, the cross array structure of contemporary resistive memory devices has a serious crosstalk current problem. NbO<sub>x</sub>-based devices that exhibit either threshold or memory resistance switching (RS), according to the content of oxygen vacancies in the NbO<sub>x</sub> film [11], can self-assemble as 1Selector-1RRAM (1S1R) devices to suppress crosstalk current. Several studies have reported the threshold switching characteristics of NbO<sub>x</sub> devices [12–14], but few studies have investigated the optimization of NbO<sub>x</sub> memory RS characteristics [15,16]. The performance of NbO<sub>x</sub> memristor devices is insufficient for applications; thus, it is essential to optimize the performances of NbO<sub>x</sub> memristor devices. Research on optimizing memory RS characteristics of NbO<sub>x</sub> can provide insights for applications of NbO<sub>x</sub>-based 1S1R devices.

Doping, especially N-doping, in RS functional layers has been widely used as an effective method to improve RRAM performance. Park et al. [17] demonstrated that N-doping in the HfO<sub>x</sub>:N layers significantly restrained the randomness distribution of

Ag conductive filaments (CFs) and decreased the power consumption and AC response time of electrochemical change memory (ECM)-type Ag/HfO<sub>x</sub>:N/Pt/Ti devices. Sedghi et al. [18] reported that N-doping in Pt/Ti-Ta<sub>2</sub>O<sub>5</sub>:N/Pt effectively reduced the defect densities induced by oxygen vacancies and achieved multi-level switching. Yang et al. [19] reported that the formation of a hafnium-nitrogen complex near oxygen vacancies in a N-doped HfO<sub>2</sub> system, as determined by ab initio calculations, greatly reduced the total energy of the system and increased its stability. Several studies have reported the benefits of N-doping on ECM-type RRAM [17,18,20], mainly focusing on metal CFs. However, few studies have investigated the effect of N-doping on VCM-type, NbO<sub>x</sub>-based RRAM [18], that is, on CFs with oxygen vacancy. The low on/off ratio in VCM-type, NbO<sub>x</sub>-based RRAM devices limits their applications in memory and logic circuits. In this paper, the purpose of N doping is twofold: to increase the resistivity of the film and improve the on/off ratio of the device by replacing O with N; in addition, in situ N-doping can promote the uniform doping of N and uniform substitution of N for O in thin films. Research on the effect of in situ N-doping on VCM-type, NbO<sub>x</sub>-based RRAM devices can provide insights for their applications.

In situ N-doping can ensure uniform N-doping on thin films and does not require a subsequent annealing, simplifying the preparation process and reducing excess energy consumption. In this study, chemically inert Pt electrodes were used as the top and bottom electrodes, and the influence of in situ N-doping on VCM-type, NbO<sub>x</sub> RRAM devices was investigated. N was introduced via a nitrogen–oxygen mixture atmosphere during film deposition. Compared with Pt/NbO<sub>x</sub>/Pt device, the Pt/NbO<sub>x</sub>:N/Pt device had an improved on/off ratio, was more stable, and had longer retention time. Further, differences in the RS behaviors and conduction mechanisms of Pt/NbO<sub>x</sub>/Pt and Pt/NbO<sub>x</sub>:N/Pt memory devices were elucidated.

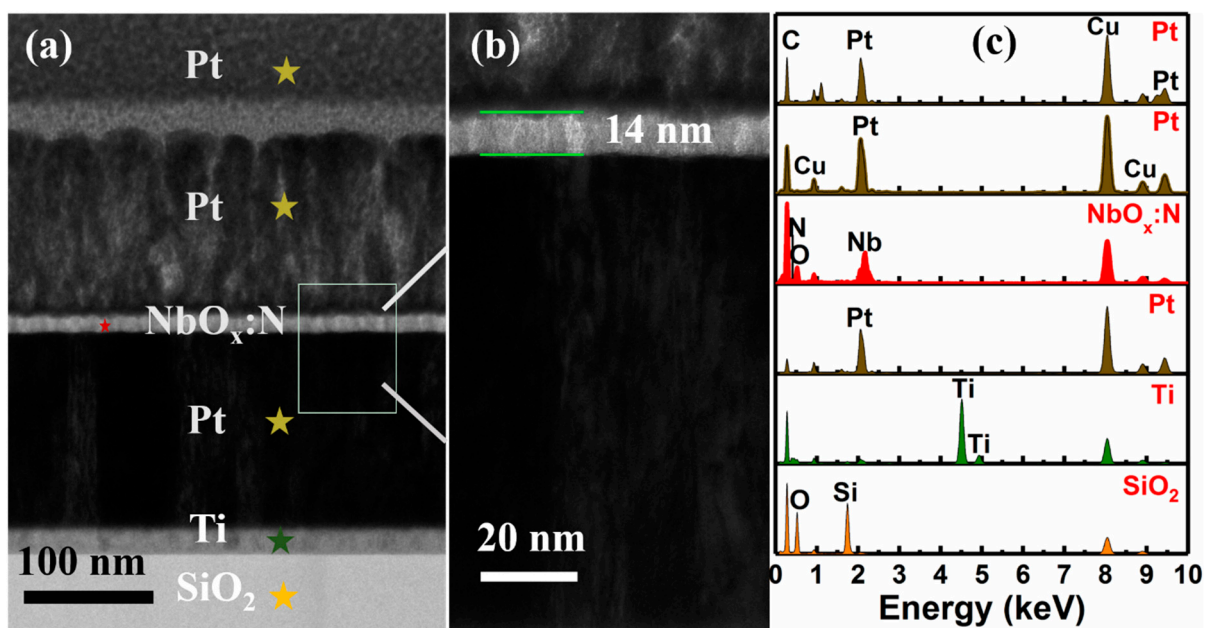
## 2. Experimental Section

Except the different deposition atmosphere on the Si/SiO<sub>2</sub>/Ti/Pt substrate at 200 °C, NbO<sub>x</sub> and NbO<sub>x</sub>:N RS films were deposited by pulsed laser deposition (KrF pulsed excimer laser, COMPEX PRO 201F, Coherent GmbH, Dieburg, Germany, wavelength = 248 nm), under the same deposition conditions (laser energy = 400 mJ, repetition frequency = 6 Hz, and bombardments with = 4000 counts). The NbO<sub>x</sub> film was deposited under a pure O<sub>2</sub> atmosphere, and the NbO<sub>x</sub>:N film was deposited under a mixed atmosphere of O<sub>2</sub>–N<sub>2</sub> (50:8, sccm). The background pressure of the chamber was  $8 \times 10^{-4}$  Pa, and it was maintained at 20 Pa during deposition. A Nb metallic target was used as the Nb source, and the distance between the Nb target and substrate was 4 cm. A circular pore Pt (thickness = 100 nm, diameter = 300 μm) was used as the top electrode.

The microstructure information of the prepared films was characterized using X-Ray diffraction (XRD, Rigaku SmartLab, Rigaku Corporation, Tokyo, Japan). The prepared films were cut using focused ion beam (FIB, GAIA3 XMH, TESCAN, Brno, Czech Republic). The film thickness and elemental composition were determined using transmission electron microscopy (TEM, JEM-F200, JEOL, Tokyo, Japan) with energy dispersive X-Ray spectroscopy (EDX, JEM-F200, JEOL, Tokyo, Japan). Surface morphologies and element mapping of the prepared films were characterized using scanning electron microscopy (SEM, Zeiss SIGMA, Carl Zeiss Microscopy Ltd., Cambridge, UK) with energy disperse spectroscopy (EDS, Aztec Energy, Oxford Instruments Nanoanalysis, Oxford, UK). The valence states of the elements and Fermi levels of the prepared films were measured using X-Ray photoelectron spectroscopy (XPS, ESCALAB 250Xi, Thermo Fisher Scientific, Waltham, MA, USA) and ultraviolet photoelectron spectroscopy (UPS, ESCALAB 250Xi, Thermo Fisher Scientific, Waltham, MA, USA). Electrical properties were measured using a semiconductor parameter analyzer (Keithley 4200-SCS, Keithley Instruments, Inc., Cleveland, OH, USA), with the DC mode at room temperature under an air atmosphere.

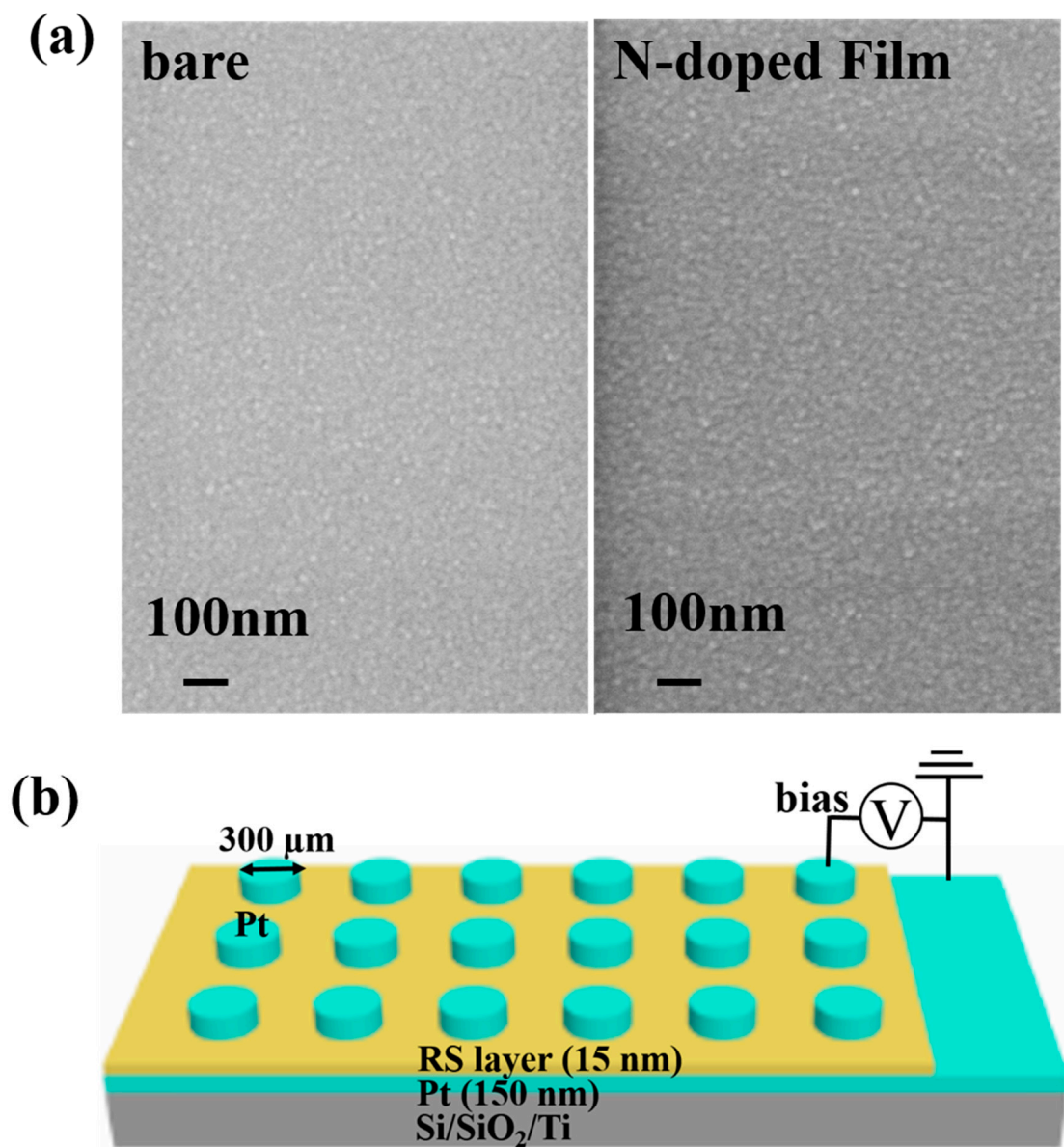
### 3. Results and Discussion

A multilayer substrate, Si/SiO<sub>2</sub> (300 nm)/Ti (20 nm)/Pt (150 nm), was used in this study. To determine the thickness of the prepared film, TEM was used to intuitively observe the thickness of each layer of the NbO<sub>x</sub>:N film. The thickness of each layer of the NbO<sub>x</sub>:N film is shown in Figure 1a. High-resolution TEM was conducted on a square area of the NbO<sub>x</sub>:N film of Figure 1a; as shown in Figure 1b, the thickness of NbO<sub>x</sub>:N film was 14 nm. Due to the similar preparation conditions of the NbO<sub>x</sub> and NbO<sub>x</sub>:N films, it can be inferred that the thickness of the NbO<sub>x</sub> film was also about approximately 15 nm. Figure 1c shows the EDX point scanning results of the corresponding layers in a pentagram region in Figure 1a. In the EDX scanning results of the NbO<sub>x</sub>:N layer, peaks corresponding to Nb, N, and O are observed; the appearance of an N peak indicates successful N-doping on the NbO<sub>x</sub> film. The observed film was placed on a Cu network with a C film, because of which, the peaks of Cu and C also appeared in the EDX scanning results.



**Figure 1.** (a) Cross-sectional TEM of Si/SiO<sub>2</sub>/Ti/Pt/NbO<sub>x</sub>:N multi-film; (b) enlarged view of the NbO<sub>x</sub>:N film in a square area of (a); (c) EDX point scanning of each layer in a pentagram region in (a).

The effect of N-doping on the surface morphologies of NbO<sub>x</sub> films was observed using SEM profiles. Figure 2a shows that bare as well as N-doped films were continuous and dense, and the particles forming the films were closely and neatly packed. A comparison of the two films shows that the surface of the NbO<sub>x</sub>:N film was more dense, size of the particles forming the film was more uniform, and particles were more closely arranged. Denser film means fewer defects in the NbO<sub>x</sub>:N film, which may be related to the passivation of oxygen vacancies by N doping. Fewer oxygen vacancies in NbO<sub>x</sub>:N film may result in an abrupt switching behavior in the RS device [16]. Due to the deposition temperature of 200 °C, which is lower than the crystallization temperature of the NbO<sub>x</sub> film, apart from the substrate peak, there was no crystalline peak of NbO<sub>x</sub> in the XRD results, as shown in Figure S1. The metal–insulator–metal structure schematic of the Si/SiO<sub>2</sub>/Ti/Pt/RS layer/Pt device is shown in Figure 2b. The top and bottom electrodes of the device were Pt electrodes; at the top electrode, a voltage was supplied, and the bottom electrode was grounded.

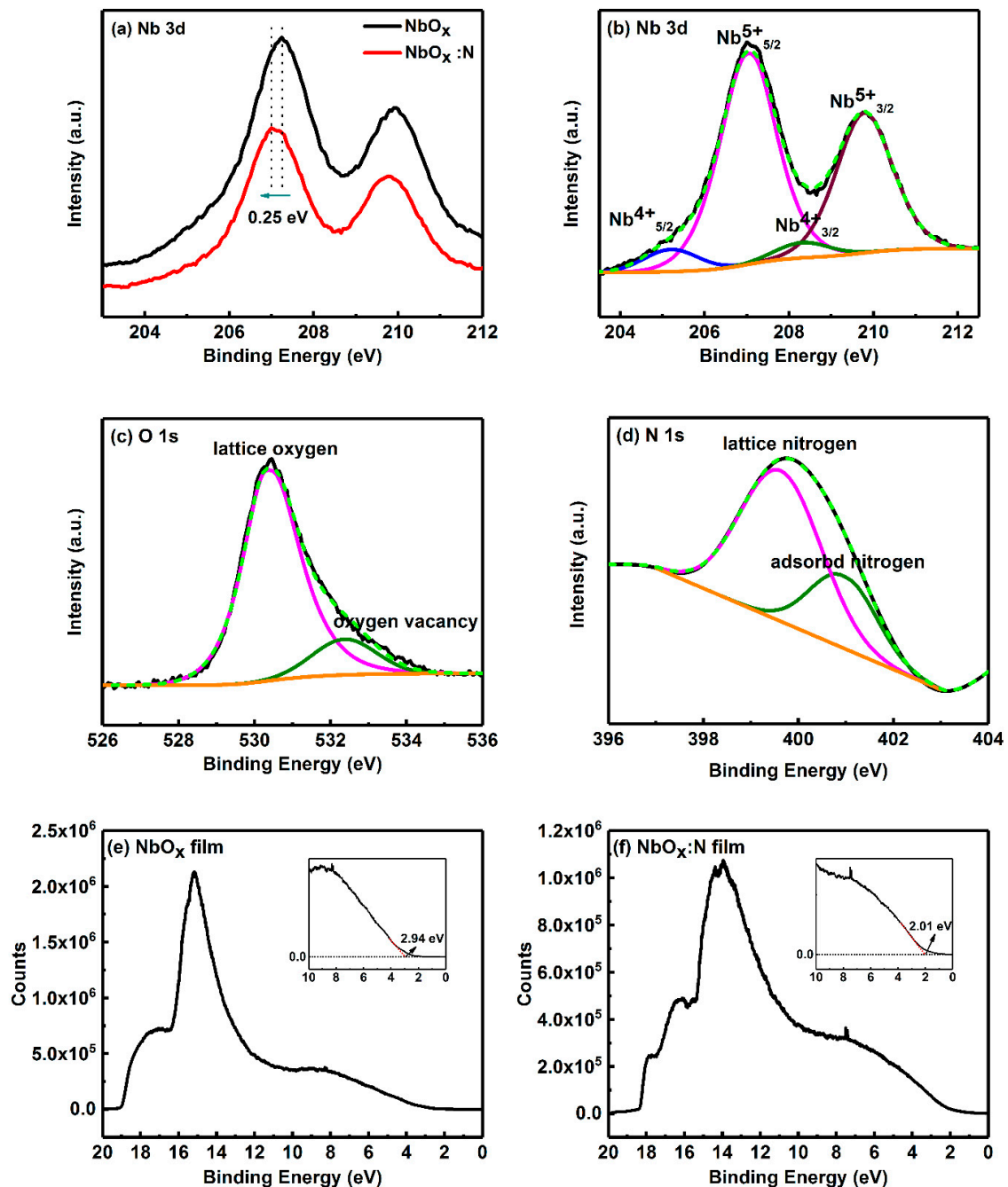


**Figure 2.** (a) Surface morphologies of bare NbO<sub>x</sub> and N-doped films. (b) Schematic diagram of the Pt/RS layer/Pt memory device (RS layer: NbO<sub>x</sub> or NbO<sub>x</sub>:N).

XPS profiles were used to confirm the successful doping of nitrogen into the lattice of NbO<sub>x</sub> film. The peak position of the binding energy for each element was calibrated by C 1s at 284.8 eV. The N 1s spectral lines were fitted after smoothing by Gaussian function. Figure 3a shows that the binding energy of Nb 3d shifted toward a lower binding energy of 0.25 eV after N incorporation, which is mainly because N atoms are less electronegative than O atoms. The result implied the presence of the Nb–O–N group, which is agreement with the findings of another similar report [17]. Figure 3b shows that two strong peaks at binding energy of 207.05 and 209.79 eV, implying that the chemical environment around Nb was the same as Nb<sub>2</sub>O<sub>5</sub> [21]. Two weak peaks appear at binding energy of 205.19 and 208.30 eV, which correspond to spin orbital splits of 3d<sub>5/2</sub> and 3d<sub>3/2</sub> of Nb<sup>4+</sup>, respectively. Figure 3c shows two peaks at binding energy of 530.36 and 532.38 eV, which correspond to lattice oxygen in most oxides and oxygen vacancies, respectively [22]. The existence of oxygen vacancies suggested that the RS mechanism was related to the oxygen vacancy CF in the Pt/NbO<sub>x</sub>:N/Pt memory device. Figure 3d shows two peaks at binding energies of 399.54 and 400.84 eV. Higher binding energies are typically ascribed to adsorbed N<sub>2</sub> in air,

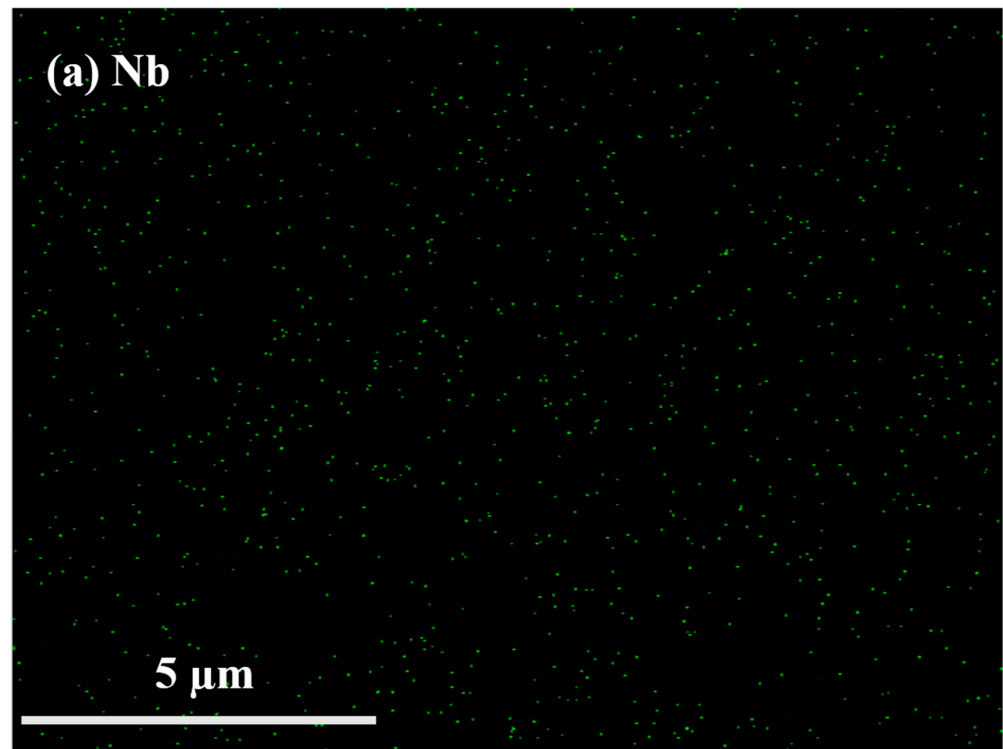


and a peak at binding energy of approximately 399.00 eV is ascribed to the Nb–O–N lattice that was formed by N atoms substituting oxygen atoms, which is in agreement with the results shown in Figure 2a [23–25]. XPS results suggested that N atoms were successfully incorporated and replaced the lattice O atoms in the NbO<sub>x</sub> film. The XPS spectra of the Nb 3d and O1s core-levels of NbO<sub>x</sub> films are shown in Figure S2. The UPS results of NbO<sub>x</sub> and NbO<sub>x</sub>:N film are shown in Figure 3e,f. The Fermi level of Au was defined as Fermi zero; thus, the Fermi level of NbO<sub>x</sub> film and NbO<sub>x</sub>:N film was 2.94 and 2.01 eV, respectively. The UPS results showed that the N-doping shifted downward the Fermi level of the NbO<sub>x</sub>:N film, resulting in the NbO<sub>x</sub>:N film with poor conductivity [26] and high resistance.

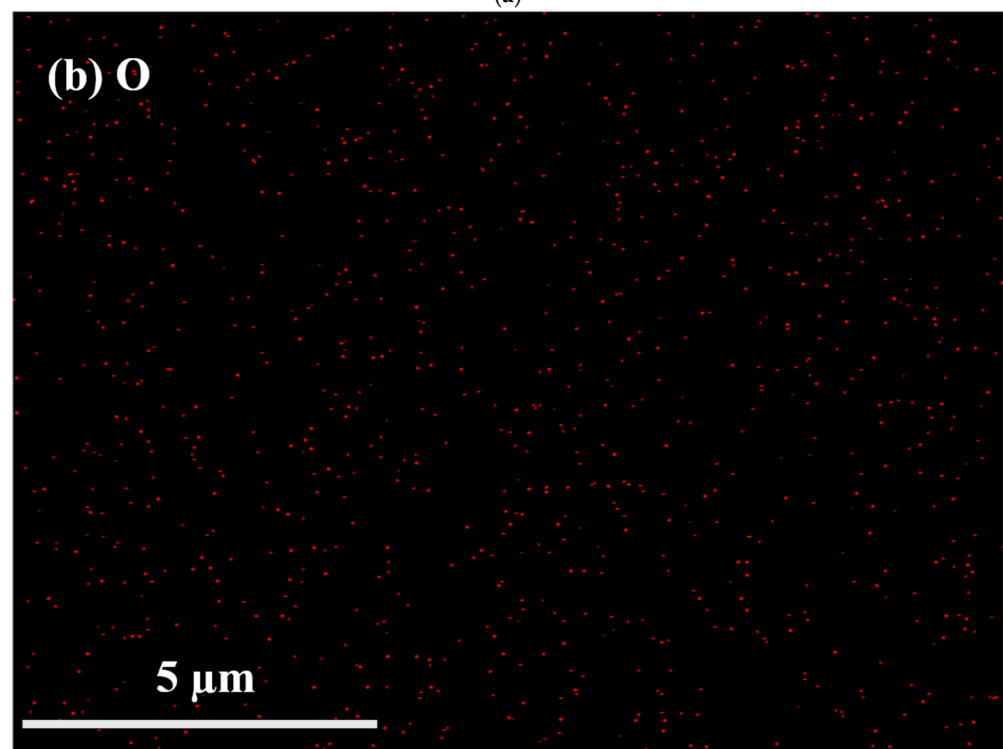


**Figure 3.** (a) Comparison of Nb 3d core level spectra in NbO<sub>x</sub> and NbO<sub>x</sub>:N films. (b) Nb 3d, (c) O 1s, and (d) N 1s core-level fitting spectra of NbO<sub>x</sub>:N film. UPS spectra of (e) NbO<sub>x</sub> and (f) NbO<sub>x</sub>:N film, measured with −6 V bias. The red dotted line represents the Fermi edge of each sample in subfigures (e,f).

EDS mapping of  $\text{NbO}_x:\text{N}$  films was used to verify uniform N-doping in the  $\text{NbO}_x$  films. As shown in Figure 4a,b, large quantities of Nb and O are distributed evenly and densely on the substrate. Figure 4c shows that N content was relatively small, but uniformly distributed on the substrate, indicating uniform N-doping on the  $\text{NbO}_x$  films.

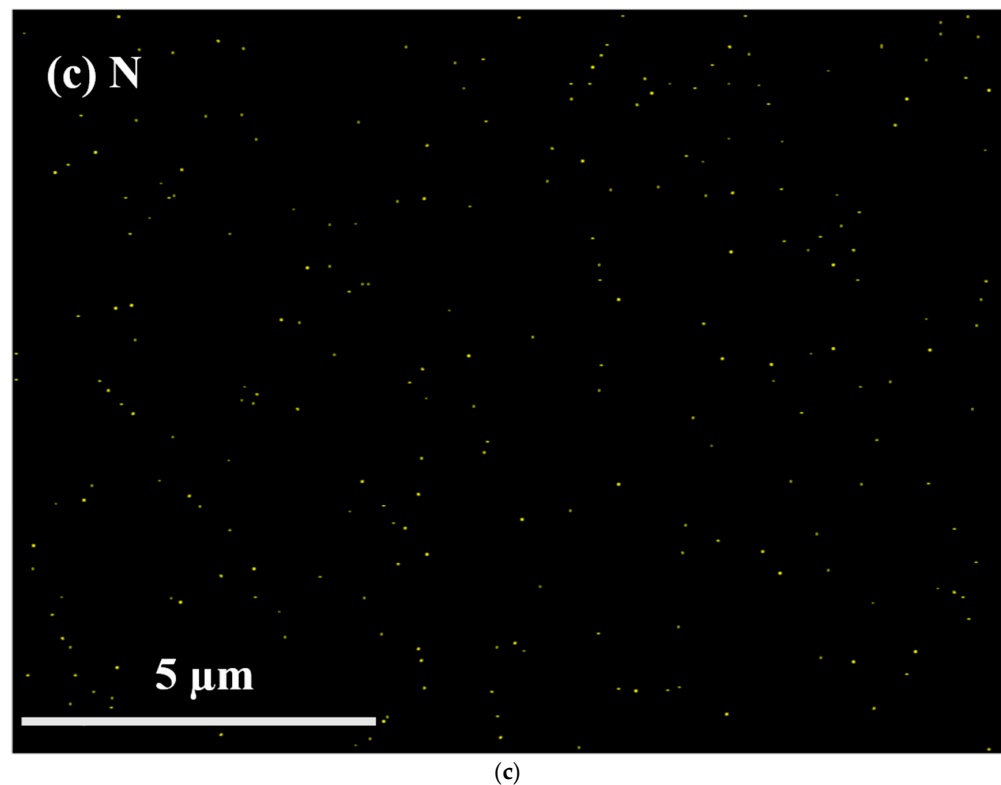


(a)



(b)

Figure 4. Cont.

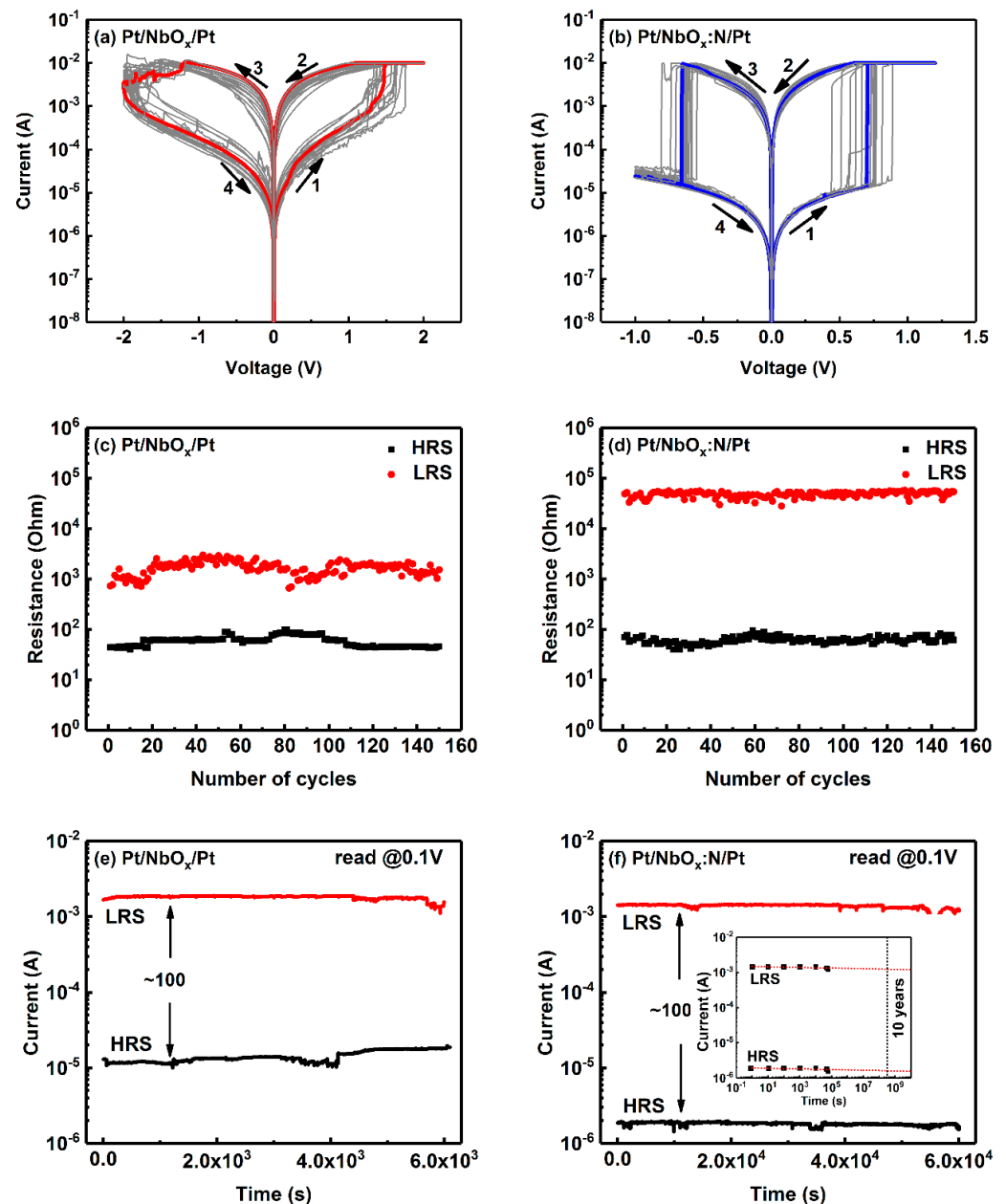


**Figure 4.** EDS mapping of (a) Nb, (b) O, and (c) N in NbO<sub>x</sub>:N films.

Electrical properties were measured with a 4200-SCS semiconductor parameter analyzer under an air atmosphere at room temperature. The sweeping voltage of the current–voltage ( $I$ - $V$ ) curves were in the order of  $0\text{ V} \rightarrow 2.5\text{ V} \rightarrow 0 \rightarrow -2.0\text{ V} \rightarrow 0\text{ V}$ .  $I$ - $V$  curves were plotted once every 5 cycles, and 20 cycles curve were plotted in total. The SET process is defined as a resistance state, switching from an HRS to a LRS, while the RESET process is defined as the opposite—that is, a resistance state switching from a LRS to an HRS. Both Pt/NbO<sub>x</sub>/Pt (named as M1) and Pt/NbO<sub>x</sub>:N/Pt (named as M2) memory devices showed bipolar RS characteristics (Figure 5a,b). Figure 5a shows that M1 exhibited a gradual RS process. Its resistance decreased, and then the reset process started at  $-1.1\text{ V}$ ; however, the whole reset process was not complete until the negative sweeping voltage reached  $-2.0\text{ V}$ . The SET process for M1 occurred at approximately  $1.7\text{ V}$ , and it was accompanied by an increase in the current, from  $0.002$  to  $0.01\text{ A}$ . For M2, the SET process occurred abruptly at approximately  $0.8\text{ V}$ , and it was accompanied by an increase in the current from  $1.17 \times 10^{-5}$  to  $0.01\text{ A}$ . When the negative sweeping voltage reached  $-0.6\text{ V}$ , the same steep and abrupt RESET process occurred. Compared with M1, that is, without N-doping, the performance of M2, that is, with N-doping, was obviously improved. The SET and RESET processes of M2 started abruptly and instantaneously, and the on/off ratio was obviously larger than that of M1; moreover, the  $I$ - $V$  cycle curves showed symmetrical distributions.

The detailed distribution of resistance values of M1 and M2 in HRS and LRS, along with the number of cycles, are shown in Figure 5c,d. The initial resistance of M1 was approximately  $50$  and  $700$ – $2000\ \Omega$  in LRS and HRS, respectively; HRS and LRS both slightly fluctuated during  $I$ - $V$  cycle measurements. The initial resistance of M2 was approximately  $70$  and  $30,000\ \Omega$  in LRS and HRS, respectively. The resistance values of HRS and LRS remained nearly constant during the  $I$ - $V$  cycle measurements. The enormous difference between the resistance values of M1 and M2 in HRS implied an improvement of nearly 10 times in the on/off ratio. Due to practical limitations, namely the complexity of manual measuring, owing to the measuring time and manual effort, only approximately 150 cycles were obtained. After 150 cycles, the device could still operate normally for several cycles, without changes in the on/off ratio and switching voltage. The retention time of LRS

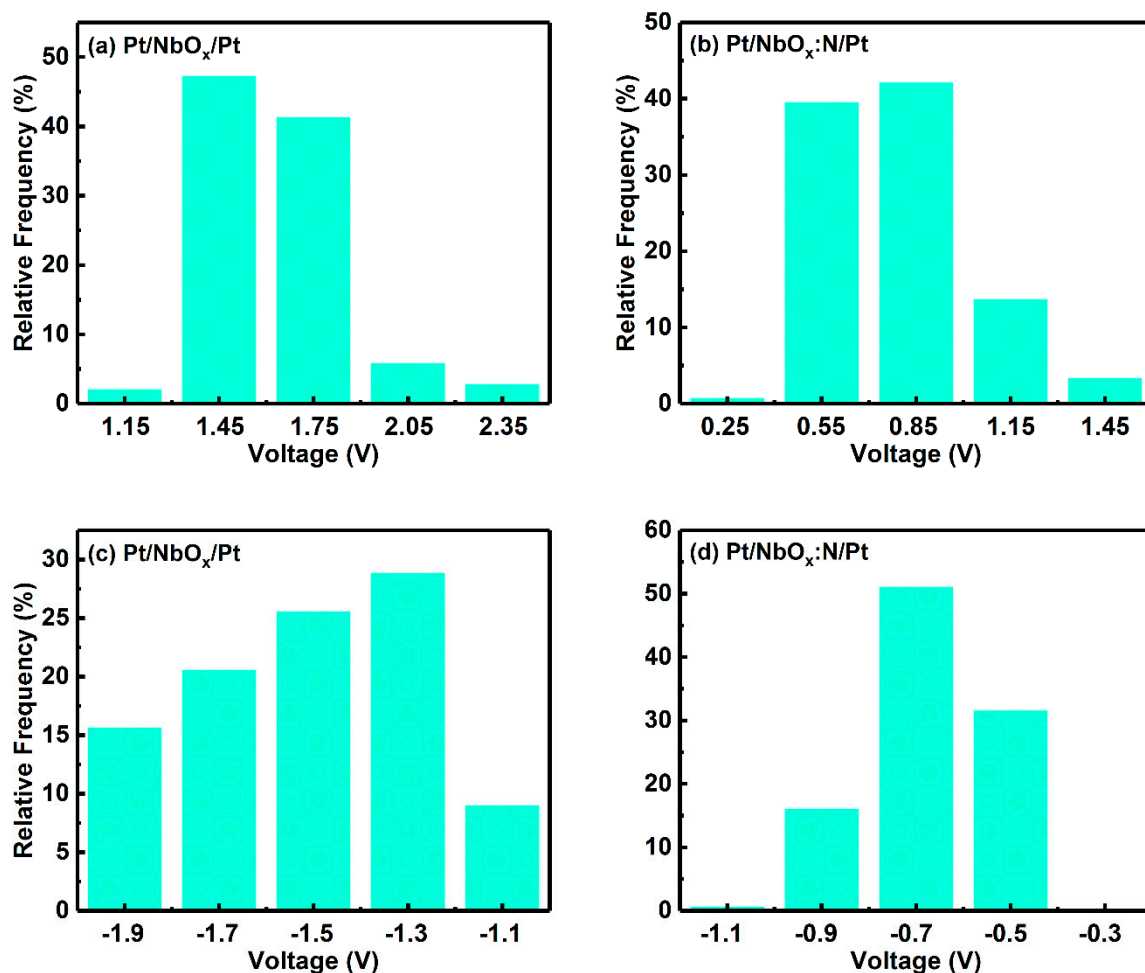
and HRS for M1 and M2 at a voltage of 0.1 V are shown in Figure 5e,f. Figure 5e shows that the HRS and LRS M1 current values were relatively stable in the first 4000 s, and the on/off ratio was approximately 100. However, after 4000 s, the current values of HRS and LRS increased and decreased, respectively. After 6000 s, the current values of LRS began to fluctuate. However, after *N*-doping, the on/off ratio was approximately 1000, and the resistance of HRS and LRS for M2 did not show any attenuation, even after testing for 60,000 s. Using linear extrapolation, at room temperature, the retention time of the Pt/NbO<sub>x</sub>:N/Pt memory device was determined to be more than 10 years [27]; that is, *N*-doping significantly improved the retention time of the device.



**Figure 5.** (a,b) Semi-logarithmic *I*-*V* curves with sweeping voltages between 1.5 V and −1.5 V; (c,d) resistance distribution with the number of cycles, retention time of low-resistance state (LRS) and high-resistance state (HRS) at a voltage of 0.1 V for the (e) Pt/NbO<sub>x</sub>/Pt and (f) Pt/NbO<sub>x</sub>:N/Pt memory devices, respectively. The inset graph in (f) is the linear extrapolation fitting of retention time for the Pt/NbO<sub>x</sub>:N/Pt memory device.



The differences in the SET and RESET voltage distributions of M1 and M2 were characterized as basis voltage distribution histograms, plotted in Figure 6. As seen in Figure 6a, the SET voltage of M1 device was distributed in the range 1.3–1.9 V, and the cumulative frequency of the two ranges was >85%; most of the SET voltage of the device was distributed in this range. As seen in Figure 6b, the SET voltage of M2 was concentrated in the range 0.4–1.3 V, cumulative frequency >85% was concentrated between 0.4 V and 1.0 V, and SET voltage was significantly lower than that of M1. As seen in Figure 6c, the RESET voltage distribution of M1 was relatively dispersed, and the voltage with a cumulative frequency of >85% was distributed between −1.2 V and −2.0 V. As seen in Figure 6d, the RESET voltage distribution of M2 was very concentrated, voltage with a cumulative frequency >85% was concentrated between −0.4 V and −0.8 V, and RESET voltage was significantly lower than that of M1. Therefore, compared with M1, the SET and RESET voltages of M2 were significantly lower, and the distribution of the RESET voltage was significantly more concentrated.



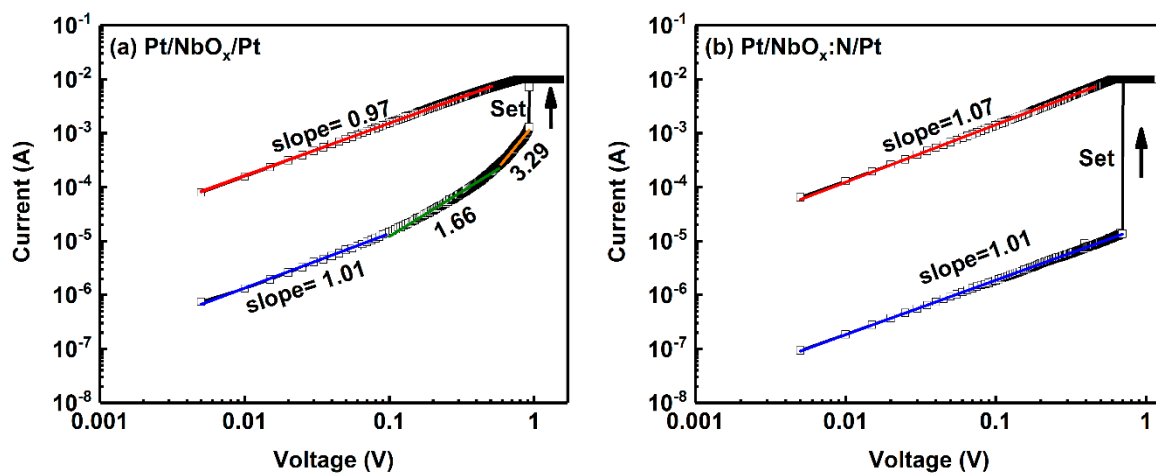
**Figure 6.** Distribution of SET voltage for (a) Pt/NbO<sub>x</sub>/Pt and (b) Pt/NbO<sub>x</sub>:N/Pt memory devices. Distribution of RESET voltage for (c) Pt/NbO<sub>x</sub>/Pt and (d) Pt/NbO<sub>x</sub>:N/Pt memory devices.

According to the above analysis of the performances of M1 and M2, M2 with N-doping exhibited better performance: increased on/off ratio, significantly improved device stability, significantly lower SET and RESET voltage, and more concentrated distribution of RESET voltage. Table 1 lists the performance parameters of NbO<sub>x</sub>-based memory RS devices, prepared in this paper and compared with previous reports.

**Table 1.** Performance of NbO<sub>x</sub>-based memory RS devices, prepared in this paper and compared with previous reports.

Structure	Thickness (nm)	Deposition Method	I-V Characteristic	SET/RESET Voltage (V)	On/Off Ratio	Retention Time (s)	Ref
Au/Nb <sub>2</sub> O <sub>5</sub> /Pt	60	RF-sputtering	Bipolar	0.96/−1.3	10	-	[14]
Cu/Nb <sub>2</sub> O <sub>5</sub> /Pt	11–14	PLD	Bipolar	1.5/ −0.5−−1	~200	~5 × 10 <sup>3</sup>	[16]
Pt/Nb <sub>2</sub> O <sub>5</sub> /Pt	26	E-beam evaporation	Bipolar	0.8/−0.8	20	3.6 × 10 <sup>4</sup>	[28]
W/Nb <sub>2</sub> O <sub>5</sub> /NbO <sub>x</sub> /Ru	-	DC-sputtering	Bipolar	1.1/ −1.5−−1.8	~50	-	[29]
Al/Ti/SiO <sub>2</sub> :Nb <sub>2</sub> O <sub>5</sub> /TiN/Si	18–22	ALD	Bipolar	1/−1	<10	-	[30]
Pt/Nb <sub>2</sub> O <sub>5</sub> /Pt/TiN/SiO <sub>2</sub> /Si	50	DC-sputtering	Unipolar	1.7/0.8	100	1.67 × 10 <sup>4</sup>	[31]
Pt/NbO <sub>x</sub> :N/Pt	~15	PLD	Bipolar	0.4–1.0/ −0.4−−0.8	~10 <sup>3</sup>	6 × 10 <sup>4</sup>	This work

To investigate the reason for the improved performance of the Pt/NbO<sub>x</sub>:N/Pt memory device, the conduction mechanisms of the Pt/NbO<sub>x</sub>/Pt and Pt/NbO<sub>x</sub>:N/Pt memory devices were analyzed. The complex nonlinear relationship between the current density and electric field can be characterized as a simple linear relationship by selecting the double-logarithmic coordinate axis; this will allow the conduction mechanism to be judged by fitting the slope of the double-logarithmic coordinate axis. Figure 7a,b shows the double-logarithmic *I-V* curves. Experimental data are represented by black hollow boxes, and different color fitting lines are plotted in different regions. For the Pt/NbO<sub>x</sub>/Pt memory device (Figure 7a), the linear fitting of the logarithmic current vs logarithmic voltage, with slope = 0.97, indicates  $I \propto V^{0.97}$ , corresponding to ohmic conductive behavior, which manifested the CF mechanism in LRS [32,33]. The fitting of HRS was divided into three different regions:  $I \propto V^{1.01}$  in the low electric field region,  $I \propto V^{1.66}$  in the medium electric field region, and  $I \propto V^{3.29}$  in the high electric field region. The increasing fitting slopes of the three regions correspond to the space-charge-limited current (SCLC) mechanism in HRS [34,35]. The different conduction mechanism in LRS and HRS manifested the local Vo CF conduction mechanism in the Pt/NbO<sub>x</sub>/Pt memory device. The RS processes of HRS and LRS were mainly attributed to the formation and rupture of CFs.



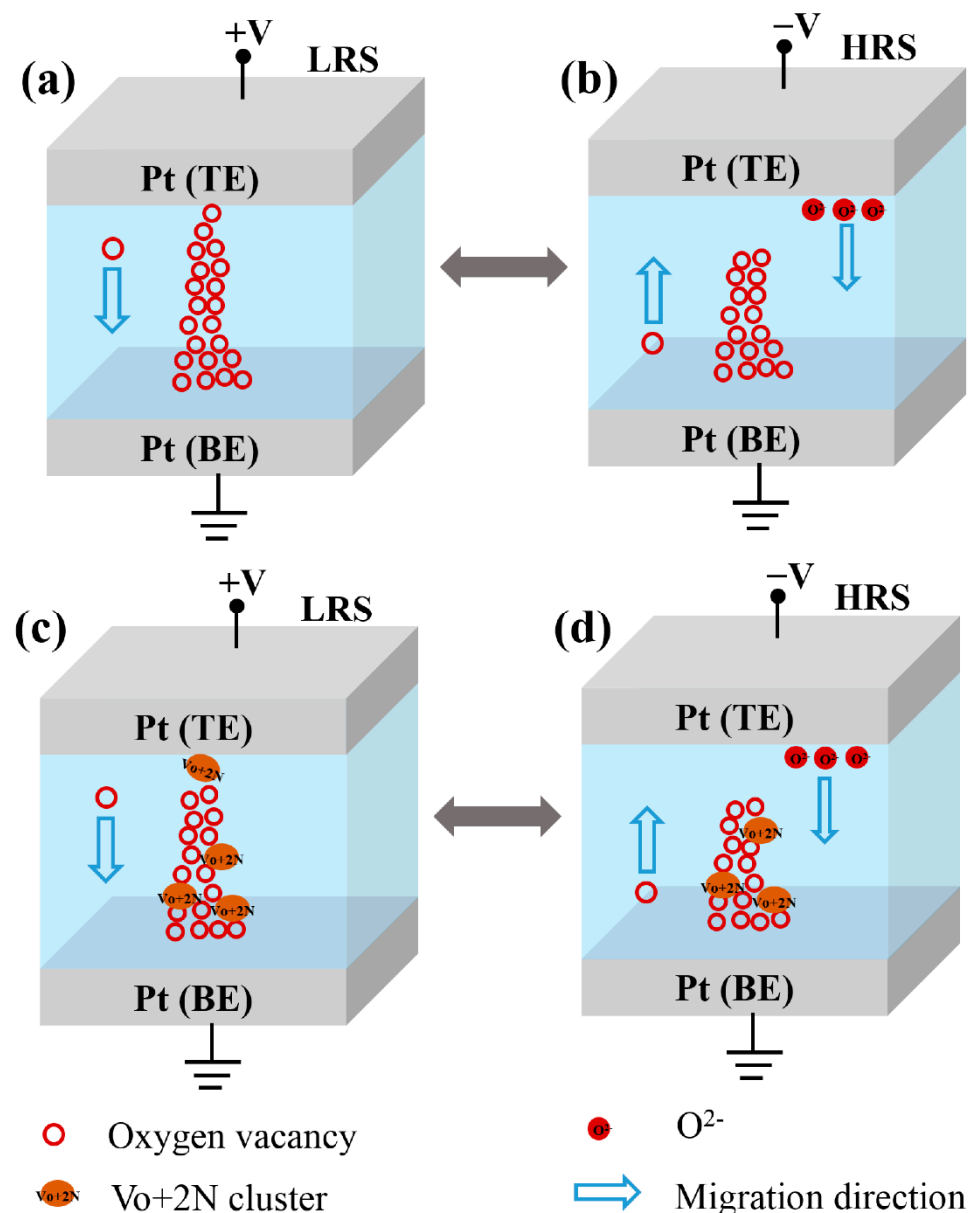
**Figure 7.** Double-logarithmic *I-V* curves of (a) Pt/NbO<sub>x</sub>/Pt and (b) Pt/NbO<sub>x</sub>:N/Pt memory devices. Standard errors of all fitted parameters were <0.01. Standard error values of fitting parameters and the overall fitting degree  $R^2$  of the equation are detailed in Tables S1 and S2.

In Figure 7b, the similar linear fitting slope of approximately 1 in LRS for the Pt/NbO<sub>x</sub>:N/Pt memory device indicates the same ohmic conduction mechanism in LRS. Thus, *N*-doping did not change the conduction mechanism of ohmic conduction in LRS. The linear fitting slope of HRS in different electric field regions was 1.07, which did not satisfy the theory that the slope of the SCLC mechanism increased gradually in the high electric field region. Therefore, the conduction mechanism of the Pt/NbO<sub>x</sub>:N/Pt device in HRS cannot be explained by the SCLC mechanism. *N*-doping passivates oxygen vacancies in oxides [36]; that is, oxygen vacancies decrease significantly after *N*-doping. The leakage current through the dielectric layer is reduced; as a result, compared to the Pt/NbO<sub>x</sub>/Pt device, the off-current (in the HRS) is one order of magnitude lower after *N*-doping (Figure 5c,d). Thus, there is improvement in the on/off ratio of the Pt/NbO<sub>x</sub>:N/Pt devices. The breakdown phenomenon is followed by the Ohmic region, without a Child's law regime (Figure 7b).

In *N*-incorporated oxides, Vo preferably combines with two neutral *N* atoms to form a Vo + 2*N* complex [19,20,37]. In *N*-doped NbO<sub>x</sub> films, in the presence of excess Vo, these complexes may form Vo + 2*N* clusters. First, principle studies have shown that the formation of clusters greatly decreases the total energy and ionic conductivity of the system and makes it more stable [17,38]; moreover, CFs in the device are more likely to grow along with clusters [39], as CFs grow in more localized directions. Therefore, in the

Pt/NbO<sub>x</sub>:N/Pt device, the formed CFs mainly comprised of V<sub>O</sub> and V<sub>O</sub> + 2N clusters, which greatly decreased the ionic conductivity, resulting in the increased resistivity in the HRS of the device [18]; thus, when the LRS resistance was almost unchanged, the on/off ratio increased. The localized CFs also concentrated the distribution of SET and RESET voltages. The occurrence of V<sub>O</sub> + 2N clusters decreased the total energy of the system and made the device more stable. Therefore, N-doping can significantly improve the performance of Pt/NbO<sub>x</sub>:N/Pt devices, such as the durability, stability, and on/off ratio, as well as achieve lower switching voltage and centralized voltage distribution.

Figure 8 shows the schematic diagram of the CF formation and ruptures, both before and after N-doping. For the Pt/NbO<sub>x</sub>/Pt memory device, as shown in Figure 8a, after positive bias was applied to the top electrode (TE), under the influence of the electric field, the V<sub>O</sub> in the NbO<sub>x</sub> film moved from the TE to the bottom electrode (BE). The V<sub>O</sub> accumulated at the BE, developed into nuclei for the CFs, acting as extensions of the cathode, and the CFs then grew toward the anode. Finally, the BE and TE were connected via the CFs; meanwhile, the device changed from HRS to LRS. When a negative bias voltage was applied to TE, as shown in Figure 8b, most of the Joule heat accumulated in the thinnest part of the CF; at the same time, the O<sup>2-</sup> moved from the TE to the BE under the influence of the electric field. The recombination of O<sup>2-</sup> and V<sub>O</sub> in the RS layer resulted in the rupture of the CFs, and the device changed to HRS [2]. For the Pt/NbO<sub>x</sub>:N/Pt devices, as shown in Figure 8c, when positive bias was applied to the TE, V<sub>O</sub> in the NbO<sub>x</sub>:N film moved from the TE to the BE, under the influence of the electric field. At this time, V<sub>O</sub> and N atoms combined to form V<sub>O</sub> + 2N clusters. These V<sub>O</sub> + 2N clusters gradually accumulated and extended and, eventually, formed the CFs connecting the BE and TE. When negative bias was applied to the TE, as shown in Figure 8d, O<sup>2-</sup> similarly moved from the TE to the BE, under the influence of the electric field. The combination of O<sup>2-</sup> and V<sub>O</sub> in the RS layer led to the rupture of the CF; thus, the device returned to HRS.



**Figure 8.** Schematic diagrams of CF formation and ruptures in LRS and HRS: (a,b) Pt/NbO<sub>x</sub>/Pt memory device; (c,d) Pt/NbO<sub>x</sub>:N/Pt memory device.

#### 4. Conclusions

In situ N-doping was used to prepare uniformly doped NbO<sub>x</sub> films. The on/off ratio, stability and retention time of the Pt/NbO<sub>x</sub>:N/Pt device were better than that of the Pt/NbO<sub>x</sub>/Pt device; the SET and RESET voltages were significantly reduced, and the RESET voltage distribution was more concentrated. After N-doping, the formed CFs mainly comprised of V<sub>O</sub> and V<sub>O</sub> + 2N clusters, which greatly reduced the ionic conductivity and total energy, increasing the resistivity in the HRS of the Pt/NbO<sub>x</sub>:N/Pt device. Fitting the *I-V* curves of the device revealed that both devices had an ohmic conduction mechanism in LRS. In situ N-doping is an effective technique to optimize memory device performance for information storage, logic circuit, and neuromorphic computing applications.

**Supplementary Materials:** The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/nano12061029/s1>. Figure S1: XRD profiles of NbO<sub>x</sub> and NbO<sub>x</sub>:N films; Figure S2: (a) Nb 3d and (b) O1s core-level fitting spectra of NbO<sub>x</sub> films; Table S1: Standard



error values of fitting parameters and the fitting degree  $R^2$  of the equation in Figure 6a; Table S2: Standard error values of fitting parameters and the fitting degree  $R^2$  of the equation in Figure 6b.

**Author Contributions:** Conceptualization, J.X. and Y.L.; methodology, J.X.; software, Y.Z. and H.W.; validation, Y.Z.; formal analysis, J.X. and H.M.; investigation, J.X.; resources, Z.Z.; data curation, X.W.; writing—original draft preparation, J.X. and Y.L.; writing—review and editing, J.X., H.W. and Y.Z.; visualization, J.X.; supervision, R.X. and Y.L.; project administration, R.X.; funding acquisition, Y.L., H.W. and Y.Z. All authors have read and agreed to the published version of the manuscript.

**Funding:** This work was supported by National Natural Science Foundation of China (grant numbers: 12074291, 11804211, and 11905119).

**Institutional Review Board Statement:** Not applicable.

**Informed Consent Statement:** Not applicable.

**Data Availability Statement:** The data that support the findings of this study are available within this article and its Supplementary Materials.

**Conflicts of Interest:** The authors declare no conflict of interest.

## References

1. Zidan, M.A.; Strachan, J.P.; Lu, W.D. The future of electronics based on memristive systems. *Nat. Electron.* **2018**, *1*, 22–29. [[CrossRef](#)]
2. Banerjee, W.; Zhang, X.; Luo, Q.; Lv, H.; Liu, Q.; Long, S.; Liu, M. Design of CMOS Compatible, High-Speed, Highly-Stable Complementary Switching with Multilevel Operation in 3D Vertically Stacked Novel  $\text{HfO}_2/\text{Al}_2\text{O}_3/\text{TiO}_x$  (HAT) RRAM. *Adv. Electron. Mater.* **2018**, *4*, 1700561. [[CrossRef](#)]
3. Gao, S.; Yi, X.; Shang, J.; Liu, G.; Li, R.W. Organic and hybrid resistive switching materials and devices. *Chem. Soc. Rev.* **2019**, *48*, 1531–1565. [[CrossRef](#)] [[PubMed](#)]
4. Wang, Z.; Wu, H.; Burr, G.W.; Hwang, C.S.; Wang, K.L.; Xia, Q.; Yang, J.J. Resistive switching materials for information processing. *Nat. Rev. Mater.* **2020**, *5*, 173–195. [[CrossRef](#)]
5. Zhang, Z.; Wang, Z.; Shi, T.; Bi, C.; Rao, F.; Cai, Y.; Liu, Q.; Wu, H.; Zhou, P. Memory materials and devices: From concept to application. *InfoMat* **2020**, *2*, 261–290. [[CrossRef](#)]
6. Sebastian, A.; le Gallo, M.; Khaddam-Aljameh, R.; Eleftheriou, E. Memory devices and applications for in-memory computing. *Nat. Nanotechnol.* **2020**, *15*, 529–544. [[CrossRef](#)]
7. Paul, F.; Paul, S. To Be or Not to Be—Review of electrical bistability mechanisms in polymer memory devices. *Small* **2022**, *2106442*, 2106442. [[CrossRef](#)]
8. Wang, X.F.; Tian, H.; Liu, Y.; Shen, S.; Yan, Z.; Deng, N.; Yang, Y.; Ren, T.L. Two-mode  $\text{MoS}_2$  filament transistor with extremely low subthreshold swing and record high on/off ratio. *ACS Nano*. **2019**, *13*, 2205–2212. [[CrossRef](#)]
9. Sun, L.; Hwang, G.; Choi, W.; Han, G.; Zhang, Y.; Jiang, J.; Zheng, S.; Watanabe, K.; Taniguchi, T.; Zhao, M.; et al. Ultralow switching voltage slope based on two-dimensional materials for integrated memory and neuromorphic applications. *Nano Energy* **2020**, *69*, 104472. [[CrossRef](#)]
10. Hussain, T.; Abbas, H.; Youn, C.; Lee, H.; Boynazarov, T.; Ku, B.; Jeon, Y.R.; Han, H.; Lee, J.H.; Choi, C.; et al. Cellulose nanocrystal based bio-memristor as a green artificial synaptic device for neuromorphic computing applications. *Adv. Mater. Technol.* **2022**, *7*, 2100744. [[CrossRef](#)]
11. Bae, J.; Hwang, I.; Jeong, Y.; Kang, S.O.; Hong, S.; Son, J.; Choi, J.; Kim, J.; Park, J.; Seong, M.J.; et al. Coexistence of bi-stable memory and mono-stable threshold resistance switching phenomena in amorphous  $\text{NbO}_x$  films. *Appl. Phys. Lett.* **2012**, *100*, 062902. [[CrossRef](#)]
12. Herzig, M.; Weiher, M.; Ascoli, A.; Tetzlaff, R.; Mikolajick, T.; Slesazeck, S. Improvement of  $\text{NbO}_x$ -based threshold switching devices by implementing multilayer stacks. *Semicond. Sci. Technol.* **2019**, *34*, 075005. [[CrossRef](#)]
13. Fang, Y.; Wang, Z.; Cheng, C.; Yu, Z.; Zhang, T.; Yang, Y.; Cai, Y.; Huang, R. Investigation of  $\text{NbO}_x$ -based volatile switching device with self-rectifying characteristics. *Sci. China Inf. Sci.* **2019**, *62*, 229401. [[CrossRef](#)]
14. Aziz, J.; Kim, H.; Rehman, S.; Khan, M.F.; Kim, D.K. Chemical nature of electrode and the switching response of rf-sputtered  $\text{NbO}_x$  films. *Nanomaterials* **2020**, *10*, 2164. [[CrossRef](#)]
15. Aziz, J.; Kim, H.; Rehman, S.; Kadam, K.D.; Patil, H.; Aftab, S.; Khan, M.F.; Kim, D.K. Discrete memristive levels and logic gate applications of  $\text{Nb}_2\text{O}_5$  devices. *J. Alloys Compd.* **2021**, *879*, 160385. [[CrossRef](#)]
16. Xu, J.; Wang, H.; Zhu, Y.; Liu, Y.; Zou, Z.; Li, G.; Xiong, R. Tunable digital-to-analog switching in  $\text{Nb}_2\text{O}_5$ -based resistance switching devices by oxygen vacancy engineering. *Appl. Surf. Sci.* **2022**, *579*, 152114. [[CrossRef](#)]
17. Park, J.H.; Kim, S.H.; Kim, S.G.; Heo, K.; Yu, H.Y. Nitrogen-induced filament confinement technique for a highly reliable hafnium-based electrochemical metallization threshold switch and its application to flexible logic circuits. *ACS Appl. Mater. Interfaces* **2019**, *11*, 9182–9189. [[CrossRef](#)]

18. Sedghi, N.; Li, H.; Brunell, I.F.; Dawson, K.; Potter, R.J.; Guo, Y.; Gibbon, J.T.; Dhanak, V.R.; Zhang, W.D.; Zhang, J.F.; et al. The role of nitrogen doping in ALD Ta<sub>2</sub>O<sub>5</sub> and its influence on multilevel cell switching in RRAM. *Appl. Phys. Lett.* **2017**, *110*, 102902. [[CrossRef](#)]
19. Yang, M.; Bae, J.H.; Yang, C.W.; Benayad, A.; Baik, H. Formation of an oxygen vacancy-dinitrogen complex in nitrogen-doped hafnium oxide. *J. Anal. At. Spectrom.* **2013**, *28*, 482–487. [[CrossRef](#)]
20. Kim, T.; Kim, S.H.; Park, J.H.; Park, J.; Park, E.; Kim, S.G.; Yu, H.Y. An artificial neuron using a bipolar electrochemical metallization switch and its enhanced spiking properties through filament confinement. *Adv. Electron. Mater.* **2021**, *7*, 2000410. [[CrossRef](#)]
21. Wagner, C.; Riggs, W.M.; Ddaavis, L.E.; Moulder, J.F.; Muilenberg, G.E. *Handbook of X-ray Photoelectron Spectroscopy*; Perkin-Elmer Corporation: Waltham, MA, USA, 1979.
22. Ge, J.; Chaker, M. Oxygen vacancies control transition of resistive switching mode in single-crystal TiO<sub>2</sub> memory device. *ACS Appl. Mater. Interfaces* **2017**, *9*, 16327–16334. [[CrossRef](#)] [[PubMed](#)]
23. Fang, J.; Wang, F.; Qian, K.; Bao, H.; Jiang, Z.; Huang, W. Bifunctional N-doped mesoporous TiO<sub>2</sub> Photocatalysts. *J. Phys. Chem.* **2008**, *112*, 18150–18156. [[CrossRef](#)]
24. Asahi, R.; Morikawa, T.; Ohwaki, T.; Aoki, K.; Taga, Y. Visible-light photocatalysis in nitrogen-doped titanium oxides. *Science* **2001**, *293*, 269–271. [[CrossRef](#)] [[PubMed](#)]
25. Zhou, Y.; Wen, T.; Kong, W.; Yang, B.; Wang, Y. The impact of nitrogen doping and reduced-niobium self-doping on the photocatalytic activity of ultra-thin Nb<sub>3</sub>O<sub>8</sub>- nanosheets. *Dalt. Trans.* **2017**, *46*, 13854–13861. [[CrossRef](#)] [[PubMed](#)]
26. Xu, Z.; Wu, J.; Wu, T.; Bao, Q.; He, X.; Lan, Z.; Lin, J.; Huang, M.; Huang, Y.; Fan, L. Tuning the fermi level of TiO<sub>2</sub> electron transport layer through Europium doping for highly efficient perovskite solar cells. *Energy Technol.* **2017**, *5*, 1820–1826. [[CrossRef](#)]
27. Wang, Y.; Liu, Q.; Long, S.; Wang, W.; Wang, Q.; Zhang, M.; Zhang, S.; Li, Y.; Zuo, Q.; Yang, J.; et al. Investigation of resistive switching in Cu-doped HfO<sub>2</sub> thin film for multilevel non-volatile memory applications. *Nanotechnology* **2010**, *21*, 045202. [[CrossRef](#)]
28. Sahoo, S. Conduction and switching behavior of e-beam deposited polycrystalline Nb<sub>2</sub>O<sub>5</sub> based nano-ionic memristor for non-volatile memory applications. *J. Alloys Compd.* **2021**, *866*, 158394. [[CrossRef](#)]
29. Zhou, Z.; Yang, M.; Fu, Z.; Wang, H.; Ma, X.; Gao, H. Electrode-induced polarity conversion in Nb<sub>2</sub>O<sub>5</sub>/NbO<sub>x</sub> resistive switching devices. *Appl. Phys. Lett.* **2020**, *117*, 243502. [[CrossRef](#)]
30. Ossorio, G.; Vinuesa, G.; García, H.; Sahelices, B.; Dueñas, S.; Castán, H.; Ritala, M.; Leskela, M.; Kemell, M.; Kukli, K. Analysis of the performance of Nb<sub>2</sub>O<sub>5</sub>-doped SiO<sub>2</sub>-based MIM devices for memory and neural computation applications. *Solid. State. Electron.* **2021**, *186*, 2–5. [[CrossRef](#)]
31. Lee, K.; Kim, J.; Mok, I.S.; Na, H.; Ko, D.H.; Sohn, H.; Lee, S.; Sinclair, R. RESET-first unipolar resistance switching behaviour in annealed Nb<sub>2</sub>O<sub>5</sub> films. *Thin Solid Films* **2014**, *558*, 423–429. [[CrossRef](#)]
32. Roy, S.; Niu, G.; Wang, Q.; Wang, Y.; Zhang, Y.; Wu, H.; Zhai, S.; Shi, P.; Song, S.; Song, Z.; et al. Toward a reliable synaptic simulation using Al-doped HfO<sub>2</sub> RRAM. *ACS Appl. Mater. Interfaces* **2020**, *12*, 10648–10656. [[CrossRef](#)] [[PubMed](#)]
33. Wang, Y.; Liu, H.; Wang, X.; Zhao, L. Impacts of Cu-doping on the performance of La-based RRAM devices. *Nanoscale Res. Lett.* **2019**, *14*, 224. [[CrossRef](#)] [[PubMed](#)]
34. Chen, Z.; Zhang, Y.; Yu, Y.; Cao, M.; Che, Y.; Jin, L.; Li, Y.; Li, Q.; Li, T.; Dai, H.; et al. Light assisted multilevel resistive switching memory devices based on all-inorganic perovskite quantum dots. *Appl. Phys. Lett.* **2019**, *114*, 181103. [[CrossRef](#)]
35. Kim, S.; Jeong, H.Y.; Choi, S.Y.; Choi, Y.K. Comprehensive modeling of resistive switching in the Al/TiO<sub>x</sub>/TiO<sub>2</sub>/Al heterostructure based on space-charge-limited conduction. *Appl. Phys. Lett.* **2010**, *97*, 033508. [[CrossRef](#)]
36. Xiong, K.; Robertson, J.; Clark, S.J. Passivation of oxygen vacancy states in HfO<sub>2</sub> by nitrogen. *J. Appl. Phys.* **2006**, *99*, 044105. [[CrossRef](#)]
37. Umezawa, N.; Shiraishi, K.; Ohno, T.; Watanabe, H.; Chikyow, T.; Torii, K.; Yamabe, K.; Yamada, K.; Kitajima, H.; Arikado, T. First-principles studies of the intrinsic effect of nitrogen atoms on reduction in gate leakage current through Hf-based high-k dielectrics. *Appl. Phys. Lett.* **2005**, *86*, 143507. [[CrossRef](#)]
38. Schmitt, R.; Spring, J.; Korobko, R.; Rupp, J.L.M. Design of oxygen vacancy configuration for memristive systems. *ACS Nano.* **2017**, *11*, 8881–8891. [[CrossRef](#)]
39. Yuan, F.Y.; Deng, N.; Shih, C.C.; Tseng, Y.T.; Chang, T.C.; Chang, K.C.; Wang, M.H.; Chen, W.C.; Zheng, H.X.; Wu, H.; et al. Conduction mechanism and improved endurance in HfO<sub>2</sub>-Based RRAM with nitridation treatment. *Nanoscale Res. Lett.* **2017**, *12*, 3–8. [[CrossRef](#)]