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## **OPEN** Glaucumolides A and B, **Biscembranoids with New** Structural Type from a Cultured Soft Coral Sarcophyton glaucum

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Glaucumolides A (1) and B (2), novel biscembranes composed of an unprecedented  $\alpha_{\nu}\beta$ -unsaturated  $\epsilon$ -lactone, along with the known metabolites ximaolide A (3) and isosarcophytonolide D (4), were isolated from the cultured soft coral Sarcophyton glaucum. The structures of the new metabolites were determined by extensive spectroscopic analyses. Compounds 1 and 2 were shown to exhibit cytotoxicity against a limited panel of cancer cell lines. In anti-inflammation assay, compounds 1 and 2 displayed strong inhibition of superoxide anion generation and elastase release in human neutrophils stimulated by fMLP/CB. Furthermore, both 1 and 2 were shown to significantly inhibit the accumulation of the pro-inflammatory inducible nitric oxide synthase protein, and compounds 1-3 were found to effectively reduce the expression of cyclooxygenase-2 protein, in lipopolysaccharide-stimulated RAW264.7 macrophage cells.

Many cembrane-type diterpenes have been proven to exhibit cytotoxicity and anti-inflammatory activity<sup>1</sup>. More than 60 biscembranoids have been isolated from the soft corals of the genera Sarcophyton<sup>2-11</sup>, Lobophytum<sup>12</sup>, Sinularia<sup>13</sup>. In particular, those belonging to the genus Sarcophyton are the most prolific source of biscembranoids. A common structural feature among these biscembranes is biogenetically derived from Diels-Alder reaction of two different cembranoid units. In our continuing search for structurally unique bioactive metabolites from the genera Sarcophyton<sup>14</sup>, we carried out the chemical investigation of the cultured soft coral Sarcophyton glaucum (Fig. 1). This study has led to the discovery of two new Diels-Alder cyclized biscembranoids glaucumolides A and B (1 and 2), a known biscembranolide ximaolide A (3)<sup>8,11</sup>, along with isosarcophytonolide D (4)<sup>9</sup>, an expected biogenic dienophile precursor of 1 (Fig. 2). The structures of 1 and 2 were deduced by extensive spectroscopic analysis. Based on structural consideration, both 1 and 2 were classified as a new type of biscembranoids as they are biosynthesized using  $\varepsilon$ -lactonecembrane 6 as a new diene monomer which has not been discovered before the biosynthesis of biscembranoids formed by Diels-Alder reaction. The attracting biological activities of several known biscembranoids<sup>13,15</sup> further prompted us to evaluate the cytotoxic and anti-inflammatory activities of isolated metabolites 1-4. The results demonstrated that compounds 1

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#### Figure 1. The cultured soft coral Sarcophyton glaucum.





and **2**, possessing a  $\gamma$ - and an  $\varepsilon$ -lactone rings, exhibited inhibition against the growth of human cancer cell lines, promyelocytic leukemia (HL-60), leukemic lymphoblasts (CCRF-CEM), acute T lymphoblastic leukaemia (MOLT-4), and erythroleukemia (K-562), as well as anti-inflammatory activities by significantly reducing the superoxide anion generation and elastase release in human neutrophils stimulated by *N*-formyl-methionyl-leucyl-phenylalanine/cytochalasin B (fMLP/CB), and the expression of iNOS and COX-2 proteins in LPS-challenged RAW264.7 macrophage cells. In contrast, without the presence of the mentioned lactone rings, **3** only displayed weaker inhibition on COX-2 accumulation in the same macrophage cells.

#### Results

Glaucumolide A (1),  $[\alpha]_{D}^{25} - 207$  (*c* 0.007, CHCl<sub>3</sub>), was isolated as an amorphous solid. Its molecular formula,  $C_{42}H_{58}O_8$ , was established by HRESIMS (713.4022 *m/z*,  $[M + Na]^+$ ), implying 14 degrees of unsaturation. The presence of the hydroxyl group was suggested by an absorption band at 3499 cm<sup>-1</sup> in the IR spectrum. The <sup>13</sup>C NMR spectroscopic data of 1 (Table 1) showed the presence of 42 carbon atoms, including eight methyls, 11 methylenes, 11 methines, and 12 quaternary carbons. Its NMR spectrum showed the signals of four vinyl methyls ( $\delta_H 2.12$ , s; 1.81, s; 1.65, s; 1.62, s;  $\delta_C 20.2$ , 19.6, 17.9, 16.0),

	1		2		
Position	<sup>1</sup> H <sup>a</sup>	<sup>13</sup> C <sup>b</sup>	<sup>1</sup> H <sup>a</sup>	<sup>13</sup> C <sup>b</sup>	
1		53.3 (C)		52.9 (C)	
2	2.33 m	41.5 (CH) <sup>d</sup>	1.87 m	41.4 (CH)	
3	3.95 dd (11.2, 6.0) <sup>c</sup>	84.4 (CH)	4.01 dd (11.2, 5.6)	85.5 (CH)	
4	1.50 d (10.8)	48.7 (CH)	1.85 m	47.3 (CH)	
5	4.82 dd (11.2, 3.2)	73.1 (CH)	4.82 dd (11.2, 3.2)	71.9 (CH)	
6	2.28 m; 2.24 m	41.0 (CH <sub>2</sub> )	2.26 m; 2.18 m	41.7 (CH <sub>2</sub> )	
7		132.5 (C)		131.7 (C)	
8	5.14 dd (6.4, 6.4)	127.2 (CH)	5.17 dd (7.6, 7.6)	127.9 (CH)	
9	2.32 m	24.7 (CH <sub>2</sub> )	2.29 m; 2.21 m	25.1 (CH <sub>2</sub> )	
10	2.31 m; 2.25 m	40.6 (CH <sub>2</sub> )	3.00 m	31.2 (CH <sub>2</sub> )	
11		161.1 (C)		162.7 (C)	
12	5.78 s	125.0 (CH)	5.98 s	124.6 (CH)	
13		197.4 (C)		198.1 (C)	
14	2.79 d (13.2); 2.46 d (13.2)	50.7 (CH <sub>2</sub> )	2.99 m; 2.37 m	52.4 (CH <sub>2</sub> )	
15	2.20 m	25.7 (CH)	2.17 m	25.6 (CH)	
16	1.08 d (6.8)	18.2 (CH <sub>3</sub> )	1.08 d (6.8)	18.4 (CH <sub>3</sub> )	
17	1.10 d (6.8)	25.1 (CH <sub>3</sub> )	1.17 d (6.8)	25.0 (CH <sub>3</sub> )	
18	1.65 s	17.9 (CH <sub>3</sub> )	1.65 s	17.1 (CH <sub>3</sub> )	
19	2.12 s	19.6 (CH <sub>3</sub> )	1.89 s	24.8 (CH <sub>3</sub> )	
20		177.9 (C)		179.1 (C)	
21	3.05 d (10.0)	45.5 (CH)	2.96 m	45.1 (CH)	
22	5.11 d (10.0)	123.0 (CH)	4.94 d (10.8)	122.8 (CH)	
23		138.7 (C)		138.3 (C)	
24	2.12 m; 2.08 m	36.2 (CH <sub>2</sub> )	2.14 m; 2.02 m	36.4 (CH <sub>2</sub> )	
25	1.82 m; 1.36 m	29.1 (CH <sub>2</sub> )	1.84 m; 1.31 m	29.2 (CH <sub>2</sub> )	
26	4.16 dd (7.6, 7.6)	68.3 (CH)	4.16 dd (7.6, 7.6)	67.8 (CH)	
27		82.9 (C)		82.6 (C)	
28	2.18 m; 2.04 m	34.8 (CH <sub>2</sub> )	2.20 m; 2.04 m	34.6 (CH <sub>2</sub> )	
29	2.46 m	26.2 (CH <sub>2</sub> )	2.47 m	26.6 (CH <sub>2</sub> )	
30	5.93 br s	135.1 (CH)	5.89 brs	135.4 (CH)	
31		132.8 (C)		NDe	
32	2.58 m; 2.18 m	30.0 (CH <sub>2</sub> )	2.65 m; 2.35 m	30.1 (CH <sub>2</sub> )	
33	2.66 m; 2.16 m	29.1 (CH <sub>2</sub> )	2.79 m; 2.07 m	29.3 (CH <sub>2</sub> )	
34		132.1 (C)		131.3 (C)	
35		127.4 (C)		127.4 (C)	
36	2.31 m; 2.04 m	34.8 (CH <sub>2</sub> )	2.31 m; 2.10 m	35.3 (CH <sub>2</sub> )	
37	1.81 s	20.2 (CH <sub>3</sub> )	1.81 s	20.3 (CH <sub>3</sub> )	
38	1.62 s	16.0 (CH <sub>3</sub> )	1.60 s	15.9 (CH <sub>3</sub> )	
39	1.32 s	22.0 (CH <sub>3</sub> )	1.31 s	21.8 (CH <sub>3</sub> )	
40		168.9 (C)		168.5 (C)	
41	2.13 s	21.5 (CH <sub>3</sub> )	2.11 s	21.5 (CH <sub>3</sub> )	
42		170.5 (C)		170.6 (C)	

**Table 1.** <sup>1</sup>H and <sup>13</sup>C NMR Data of 1 and 2. <sup>a</sup>Recorded at 400 MHz in CDCl<sub>3</sub> at 25 °C. <sup>b</sup>Recorded at100 MHz in CDCl<sub>3</sub> at 25 °C. <sup>c</sup>J values (Hz) in parentheses. <sup>d</sup>Attached protons were deduced by DEPTexperiment. <sup>e</sup>Designate signal not detected<sup>17</sup>.



Figure 3. Selected COSY ( $\blacksquare$ ) and HMBC ( $\rightarrow$ ) correlations of 1 and 2.

one methyl attached to oxygen-bearing quaternary carbon ( $\delta_{\rm H}$  1.32, s;  $\delta c$  22.0), one acetoxy group ( $\delta_{\rm H}$ 2.13, s;  $\delta c$  170.5, C; 21.5, CH<sub>3</sub>), two methyls of an isopropyl group ( $\delta_{\rm H}$  2.20, m; 1.10, d, J = 6.8 Hz; 1.08, d, J = 6.8 Hz;  $\delta c$  25.7, CH; 25.1, CH<sub>3</sub>; 18.2, CH<sub>3</sub>); four trisubstituted double bonds ( $\delta_{\rm H}$  5.93, br s; 5.78, s; 5.14, dd, *J* = 6.4, 6.4 Hz; 5.11, d, *J* = 10.0 Hz; & 161.1, C; 138.7, C; 135.1, CH; 132.8, C; 132.5, C; 127.2, CH; 125.0, CH; 123.0, CH); one tetrasubstituted double bond ( $\delta c$  132.1, C; 127.4, C); three oxygen-bearing methines ( $\delta_{\rm H}$  4.82, dd, J = 11.2, 3.2 Hz; 4.16, dd, J = 7.6, 7.6 Hz; 3.95, dd, J = 11.2, 6.0 Hz;  $\delta c$  84.4, CH; 73.1, CH; 68.3, CH); one oxygenated quaternary carbon (&c 82.9), and four carbonyl carbons (&c 197.4, 177.9, 170.5, 168.9). These evidences indicated the possible presence of a biscembranoid skeleton in compound 1. The COSY spectrum of 1 was used to identify seven different structural units from  $H_2$ -36 to isopropyl protons  $H_3$ -16 and  $H_3$ -17 via H-2, H-3 and H-4; H-5 to  $H_2$ -6; H-8 to  $H_2$ -10; H-21 to H-22; H<sub>2</sub>-24 to H-26; H<sub>2</sub>-28 to H-30; and H<sub>2</sub>-32 to H<sub>2</sub>-33, which were further assembled by HMBC correlations H-5 to C-3 and C-15; H<sub>2</sub>-14 to C-1, C-2, C-12 and C-13; H<sub>3</sub>-16 and H<sub>3</sub>-17 to C-4 and C-15; H<sub>3</sub>-18 to C-6, C-7 and C-8; H<sub>3</sub>-19 to C-10, C-11 and C-12; H-21 to C-1, C-2, C-33, C-34 and C-35; H-30 to C-31, C-32 and C-40; H<sub>3</sub>-37 to C-34, C-35 and C-36; H<sub>3</sub>-38 to C-22, C-23 and C-24; and H<sub>3</sub>-39 to C-26, C-27 and C-28 (Fig. 3). Moreover, the HMBC correlations of H-3 and H<sub>2</sub>-14 to C-20 clearly suggested the presence of an  $\alpha_{\beta}$ -unsaturated  $\gamma$ -lactone moiety at C-1–C-3 and C-20. Furthermore, the acetoxy group positioned at C-5 was confirmed from the HMBC correlations of H-5 ( $\delta$  4.82, dd, J=11.2, 3.2 Hz) and protons of an acetate methyl ( $\delta$  2.13, s) to the ester carbonyl carbon at  $\delta$  170.5. In addition, in the <sup>13</sup>C NMR spectrum of **1** the signal at  $\delta_c$  68.3, and in the <sup>1</sup>H NMR spectrum the signal at  $\delta_H$  4.16 (dd, J=7.6, 7.6 Hz) could be attributed to a hydroxyl-bearing methine at C-26. On the base of above results and by considering the degrees of unsaturation and molecular formula, C-27 and C-40 should be linked by an oxygen atom to form an  $\varepsilon$ -lactone ring. The gross structure of 1 was thus established. Metabolite 1 is the first Diels-Alder cyclized biscembranoid possessing not only a saturated  $\gamma$ -lactone but also a seven-membered  $\alpha_{\beta}$ -unsaturated  $\varepsilon$ -lactone in the molecule.

The relative configurations of the stereogenic centers in 1 were determined on the basis of NOE relationships and NMR spectroscopic data. It was found that NOESY correlations of H-3 ( $\delta$  3.95, dd, J = 11.2, 6.0 Hz) with H-2 ( $\delta$  2.33, m) and H-5 ( $\delta$  4.82, dd, J = 11.2, 3.2 Hz), H-4 ( $\delta$  1.50, d, J = 10.8 Hz) with H-2 and H-5, and H-2 with H-21 ( $\delta$  3.05, d, J = 10.0 Hz), and the upfield-shifted  $\delta_{\rm H}$  value of H-21 $\alpha$ relative to the H-21 $\beta$  of related biscembranoids in <sup>1</sup>H NMR<sup>10</sup>, indicated the  $\alpha$ -orientation for H-2, H-3, H-4, H-5, and H-21. This observation was also strongly supported by the similar <sup>1</sup>H NMR spectroscopic data of the above protons including chemical shifts and coupling constants with those of bislatumlide  $C^{10}$ . Furthermore, the NOE correlations of H-21 with H-33 $\alpha$  ( $\delta$  2.16, m) and H<sub>3</sub>-38, H<sub>3</sub>-38 with H-26  $(\delta 4.16, dd, J = 7.6, 7.6 Hz)$ , and H-26 with H-25 $\alpha$  ( $\delta 1.82, m$ ), H-29 $\alpha$  ( $\delta 2.46, m$ ) and H-30 ( $\delta 5.93, br$ s), reflected the  $\alpha$ -orientations of H-26 and the R\*-configuration at C-26. Above results and the NOE correlation between H<sub>3</sub>-39 ( $\delta$  1.32, s) and H-25 $\beta$  ( $\delta$  1.36, m) suggested the  $\beta$ -orientation of H<sub>3</sub>-39 and thus  $27S^*$  configuration of 1 as shown in Fig. 4. Further, the  $26R^*$  and  $27S^*$  configuration, not  $26R^*$  and  $27R^*$ , could be confirmed by comparing the  $\delta$  values of C-7 (69.2) and C-8 (82.9) of known compound sartrolide C<sup>16</sup> to the corresponding C-26 (68.3) and C-27 (82.9) of 1, while sartrolide A<sup>16</sup> with  $26R^*$  and  $27R^*$  configuration showed C-26 signal at  $\delta$  72.7 ppm. The chemical shift values of C-18 (17.9), C-19 (19.6) and C-38 (16.0) reflected the all trans geometry of the trisubstituted double bonds at C-7/C-8, C-11/C-12 and C-22/C-23 in the molecule of I. From the above observations, NOE correlation between H-30 and one of H<sub>2</sub>-33 ( $\delta$  2.66, m), and further analysis of other NOE interactions (Fig. 4), the relative configuration of 1 with rings A-E could be established.

Glaucumolide B (2),  $[\alpha]^{25}_{D} - 221$  (*c* 0.008, CHCl<sub>3</sub>), was showed the pseudomolecular ion peak  $[M + Na]^+$  at m/z 713.4023 in the HRESIMS, suggesting the molecular formula  $C_{42}H_{58}O_8$  and 14 degrees of unsaturation. The IR spectrum also suggested the presence of hydroxyl group in 2 ( $\nu_{max}$  3434 cm<sup>-1</sup>). The <sup>13</sup>C NMR spectroscopic data (Table 1) of 2 were found to be resembled to those of 1. Detailed analysis of 1D and 2D NMR spectra of 2 revealed the similar gross structure as that of 1. However, it was found that H-12 ( $\delta$  5.98, s) showed significant NOE interaction with H<sub>3</sub>-19 ( $\delta$  1.89, s), and the signal of the C-19 in 2 was remarkably downfield-shifted ( $\delta$  19.6 in 1, 24.8 in 2), indicating a Z geometry of



Figure 4. Selected NOE correlations for 1.





 $\Delta^{11(12)}$  in **2**, in contrast to the 11*E* double bond in **1**. These results and other NMR data including NOE

correlations, established the structure of compound 2 to be the 11Z isomer of 1. The absolute configurations of 1 and 2 were further confirmed by comparison of the CD (circular dichroism) spectroscopic data with structurally related compound. As shown in Fig. 5, the CD spectrum of 2 showed a broad negative Cotton effect at 258 nm ( $\Delta \varepsilon = -7.1$ ) due to enone  $n-\pi^*$  transition absorption while the intense positive cotton effect at 209 nm ( $\Delta \varepsilon = +27.4$ ) resulted from the  $\pi-\pi^*$  transition of the two isolated  $\Delta^{22(23)}$  and  $\Delta^{34(35)}$  double bonds, very similar to that of bislatumlide C<sup>10</sup>. Thus, the absolute configurations for rings B and C should be the same for both bislatumlide C and 2. One the basis of above results and the fact that both 1 and 2 were isolated from the same organism, the structures and absolute configurations of both 1 and 2 were found to possess the 1*S*,2*R*,3*S*,4*S*,5*S*,21*S*,26*R*,27*S*.

A plausible biosynthetic pathway, involving a key Diels–Alder reaction, was postulated for the biosynthesis of compounds **1** and **2** in Figure 6. It is obvious that **4** is one of the two precursors of **1**. Another proposed precursor should be the yet to be discovered compound **6**. [4+2] Endo cycloaddition of the dienophile **4** and diene **6** occurs between 1,2-*Z* double bond of **4** and  $\Delta^{21(34)}$  and  $\Delta^{35(36)}$  conjugated double bonds diene of **6**. Although the possible involvement of a biosynthetic Diels-Alder reaction to afford biscembranoid has been mentioned frequently, no any other bicembranoid was found to be formed by using  $\varepsilon$ -lactone cembrane **6** or related cembranoidal  $\varepsilon$ -lactone as the diene precursor. In addition, almost all the previous cembrane dimers exhibited the reactive dienophile double bond at positions C-1 and C-14 (according to the numbering assigned to compound **4**), whereas the dienophile double bond in **4** was located at C-1 and C-2. The Diels–Alder addition which arises from supra–supra transition state explains the *trans* stereochemistry of H-2 and lactone as well as the *cis* geometry of lactone and H-21. Analogously, the formation of biscembrane **2** by Diels–Alder cyclization of the 11*Z* isomer of **4**, sarco-phytonolide A (**5**)<sup>6,9</sup>, with compound **6** could be hypothesized.

The cytotoxicity of compounds 1–4 against four human cancer cell lines, HL-60, CCRF-CEM, MOLT-4, and K-562 was investigated. The results (Table 2) demonstrated that compound 2 exhibited significant cytotoxicity against HL-60 and CCRF-CEM cancer cell lines with  $ED_{50}$  values of  $3.8 \pm 0.9$  and  $5.3 \pm 1.4 \,\mu$ g/mL, respectively. Also, compound 1 exhibited cytotoxicity against the above two cell lines with  $ED_{50}$  values of  $6.6 \pm 1.2$  and  $7.4 \pm 1.5 \,\mu$ g/mL, respectively. Further, compounds 1 and 2 displayed weaker activity against MOLT-4 and K-562 cell lines ( $ED_{50}$  11.0–19.2  $\mu$ g/mL). In contrast, compound 3 was inactive toward all the tested cell lines. Perhaps the enhanced cytotoxicity of compounds 1 and 2 relative to 3 is owing to the presence of a  $\alpha$ , $\beta$ -unsaturated  $\epsilon$ -lactone ring.

The anti-inflammatory activities of compounds 1–4 on neutrophil pro-inflammatory responses were evaluated by measuring their ability in suppressing fMLP/CB-induced superoxide anion  $(O_2^{-1})$ 

	Compounds (ED <sub>50</sub> , µg/mL)						
cancer cell line	1	2	3	4	5-Fluorouracil		
HL-60	$6.6\pm1.2$	$3.8\pm0.9$	_ <sup>a</sup>	$13.0\pm1.9$	$10.7\pm0.5$		
CCRF-CEM	$7.4\pm1.5$	$5.3\pm1.4$	-	$15.3\pm2.5$	2.3±0.6		
MOLT-4	$11.0\pm2.8$	$11.0\pm2.2$	-	$17.2\pm3.1$	$0.9\pm0.2$		
K-562	$19.2\pm2.3$	$12.6\pm0.7$	-	-	$4.3\pm1.2$		

**Table 2.** Cytotoxicity of compounds 1–4. The values are mean  $\pm$  SEM (n = 3). <sup>a</sup>—>20 µg/mL

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	Superoxide anion			Elastase release		
Compound	IC <sup>50</sup> (μM) <sup>b</sup>	Inh % <sup>a</sup>	1	IC <sup>50</sup> (μM)	Inh %	
1	$2.79\pm0.66$	$88.42\pm3.97$	***	$3.97\pm0.10$	$88.94\pm6.96$	***
2	$2.79\pm0.32$	$91.75\pm3.08$	***	$3.97\pm0.10$	$103.25\pm1.89$	***
3	>10	$15.33\pm4.15$	*	>10	$15.13\pm3.58$	*
4	>10	$12.40\pm2.56$	**	>10	$27.12\pm3.08$	***
Idelalisib	$0.07\pm0.01$	$102.81 \pm 2.21$	***	$0.28\pm0.09$	$99.56 \pm 4.19$	***

Table 3. Inhibitory effects of compounds 1–4 on superoxide anion generation and elastase release by human neutrophils. <sup>a</sup>Percentage of inhibition (Inh %) at  $10 \mu$ M concentration. Results are presented as mean  $\pm$  S.E.M. (n = 3). \**P* < 0.05, \*\**P* < 0.01, \*\*\**P* < 0.001 compared with the control value. <sup>b</sup>Concentration necessary for 50% inhibition (IC<sup>50</sup>).



Figure 6. Plausible Diels-Alder reaction to derive compounds 1 and 2.

generation and elastase release in human neutrophils, and the results were shown in Table 3. From the results, 1 and 2 showed strong inhibitions ( $88.42\pm3.97$  and  $91.75\pm3.08\%$ , respectively.) toward super-oxide anion generation at  $10\mu$ M. Both of them also exhibited potent inhibitory activity against elastase release, with  $88.94\pm6.96$  and  $103.25\pm1.89\%$  inhibitions in the same fMLP/CB-stimulated cells at the same concentration. The IC<sub>50</sub> values of 1 and 2 in inhibiting the superoxide generation and elastase release were also measured. Although compound 4 did not exhibit strong activity in inhibiting superoxide anion generation, it was shown to display significant inhibitory activity in elastase release.

The *in vitro* anti-inflammatory activity of compounds 1–3 was also studied. In this assay, the up-regulation of the proinflammatory iNOS and COX-2 proteins of LPS-stimulated RAW264.7 macrophage cells was evaluated using immunoblot analysis. The results (Fig. 7) showed that at concentrations of 5, 10, and 20  $\mu$ M, compound 1 was found to significantly reduce the levels of iNOS and COX-2 to 59.4 ± 9.0 and 66.5 ± 4.4%; 31.3 ± 6.5 and 78.3 ± 5.0%; and -2.6 ± 2.7 and -0.5 ± 3.2%, respectively. At concentrations of 10 and 20  $\mu$ M, compound 2 was found to significantly reduce the levels of iNOS and COX-2 to 75.9 ± 3.5 and 64.3 ± 6.9%; and 43.4 ± 5.0 and 6.0 ± 3.6%, respectively. Moreover, at 20  $\mu$ M, 3 also reduced the level of COX-2 expression to 22.0 ± 6.5% in macrophage cells with LPS treatment. As they did not exhibit cytotoxicity to RAW264.7 cells, they might be promising anti-inflammatory agents. Also, 2 possessing promising cytotoxicity, could become a candidate for future anticancer drug development.



Figure 7. Effect of compounds 1–3 on iNOS and COX-2 protein expression of RAW264.7 macrophage cells by immunoblot analysis. (A) Immunoblots of iNOS and  $\beta$ -actin; (B) Immunoblots of COX-2 and  $\beta$ -actin. The values are mean  $\pm$  SEM (n=3). Relative intensity of the LPS alone stimulated group was taken as 100%. Under the same experimental condition CAPE (caffeic acid phenylethyl ester, 10  $\mu$ M) reduced the levels of the iNOS and COX-2 to 2.5  $\pm$  3.7% and 50.3  $\pm$  8.7%, respectively. \*Significantly different from LPS alone stimulated group (\*p < 0.05). aStimulated with LPS. bStimulated with LPS in the presence of 1–3 (20  $\mu$ M). cStimulated with LPS in the presence of 1–3 (5  $\mu$ M).

### Discussion

Compounds 1 and 2 are structurally novel as they belong to a new type of biscembranoids using the not yet isolated  $\varepsilon$ -lactonecembrane 6 as the first time discovered diene precursor for the biosynthesis of biscembranoids by Diels-Alder reaction. Metabolites 1 and 2, with the presence of a  $\alpha_{3}\beta$ -unsaturated  $\varepsilon$ -lactone ring, were shown to exhibit cytotoxicity against a limited panel of HL-60, CCRF-CEM, MOLT-4 and K-562 cancer cell lines. Compounds 1 and 2 also exhibited potent anti-inflammatory activity in inhibiting the superoxide generation and elastase release in fMLP/CB-induced human neutrophils. Furthermore, both 1 and 2 were shown to significantly inhibit the accumulation of the pro-inflammatory inducible nitric oxide synthase protein, and compounds 1–3 were found to effectively reduce the expression of cyclooxygenase-2 protein, in lipopolysaccharide-stimulated RAW264.7 macrophage cells.

#### Conclusion

The unusual structural framework with  $\alpha$ , $\beta$ -unsaturated  $\epsilon$ -lactone system were reported here for glaucumolides A and B (1 and 2), along with a known biscembranolide ximaolide A (3), and isosarcophytonolide D (4) from the cultured soft coral *S. glaucum*. From the results of biological activities, it appears that compounds 1 and 2 might be useful for future biomedical applications. The discovery of glaucumolides with a novel carbon scaffold provides additional evidence that cultured soft corals might be a promising source of structurally novel bioactive natural products which could be used for further pharmacological investigation.

#### Methods

**General Experimental Procedures.** Optical rotations were measured on a Horiba High Sensitivity Polarimeter SEPA-300. Ultraviolet spectra were recorded on a JASCO V-650 spectrophotometer. IR spectra were recorded on a JASCO FT/IR-4100 infrared spectrophotometer. CD spectra were recorded on a JASCO J-815 CD spectrophotometer. NMR spectra were recorded on a Varian 400MR FT-NMR instrument at 400 MHz for <sup>1</sup>H and 100 MHz for <sup>13</sup>C in CDCl<sub>3</sub>. LRMS and HRMS were obtained by ESI on a Bruker APEX II mass spectrometer. Silica gel (Merck, 230–400 mesh) was used for column chromatography. Precoated silica gel plates (Merck, Kieselgel 60 F-254, 0.2 mm) were used for analytical TLC. High-performance liquid chromatography was performed on a Hitachi L-2455 HPLC apparatus with a Supelco C18 column ( $250 \times 21.2$  mm, 5  $\mu$ m).

**Animal Material.** The cultured soft coral *Sarcophyton glaucum* used in this study was originally collected from the wild and cultured for five years in an 80-ton cultivation tank (height 1.6 m) located in the National Museum of Marine Biology and Aquarium, Taiwan. The tank was a semiclosed recirculating aquaculture system and did not require deliberate feeding. To the best of our knowledge, this is the first farming system for *S. glaucum* in the world. The specimens were then collected by hand in January 2010 and were stored in a -20 °C freezer. The soft coral was identified by one of the authors (C.-F.D.). A voucher specimen (specimen no. 201001C3) was deposited in the Department of Marine Biotechnology and Resources, National Sun Yat-sen University.

**Extraction and Isolation.** The frozen bodies of *S. glaucum* (0.6 kg, wet wt) were minced and extracted exhaustively with  $CH_2Cl_2$  and MeOH (1:1, 0.5 L × 6). The  $CH_2Cl_2$  and MeOH extract of the soft coral *S. glaucum* was partitioned between EtOAc and  $H_2O$  to afford the EtOAc-soluble fraction. The EtOAc extract (4.5 g) was chromatographed over silica gel by column chromatography and eluted with EtOAc in *n*-hexane (0–100%, stepwise) and then with MeOH in EtOAc (5–50%, stepwise) to yield 24 fractions. Fraction 15 (23.3 mg), eluting with *n*-hexane–EtOAc (5:1), was further purified by reversed-phase HPLC using MeOH–H<sub>2</sub>O (3:1) to afford **4** (2.6 mg). Fraction 20 (69.7 mg), eluting with *n*-hexane–EtOAc (1:1), was further purified over silica gel using *n*-hexane–acetone (2:1) to afford six subfractions (A1–A6). Subfraction A3 (13.5 mg) was further purified by reversed-phase HPLC using MeOH–H<sub>2</sub>O (5:2) to afford **3** (5.2 mg). Subfraction A4 (20.0 mg) was further purified by reversed-phase HPLC using MeOH–H<sub>2</sub>O (2:1) to afford **1** (4.4 mg) and **2** (2.8 mg).

**Glaucumolide A (1).** white amorphous powder;  $[\alpha]^{25}_{D} - 207$  (*c* 0.007, CHCl<sub>3</sub>); IR (neat)  $\nu_{max}$  3499, 2939, 2876, 1734, 1716, 1699, 1375, 1239, 1024, and 755 cm<sup>-1</sup>; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 205 (3.4) and 236 (3.2) nm; CD ( $1.9 \times 10^{-4}$  M, MeOH)  $\lambda_{max}$  ( $\Delta \varepsilon$ ) 244 (-23.7), and 215 (+36.0) nm; <sup>13</sup>C and <sup>1</sup>H NMR data, see Table 1; ESIMS *m*/*z* 713 [M + Na]<sup>+</sup>; HRESIMS *m*/*z* 713.4022 [M + Na]<sup>+</sup> (calcd for C<sub>42</sub>H<sub>58</sub>O<sub>8</sub>Na, 713.4024).

**Glaucumolide B (2).** white amorphous powder;  $[\alpha]^{25}_{D} - 221$  (*c* 0.008, CHCl<sub>3</sub>); IR (neat)  $\nu_{max}$  3434, 2940, 2878, 1734, 1716, 1698, 1376, 1239, 1024, and 754 cm<sup>-1</sup>; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 205 (3.3) and 236 (3.2) nm; CD ( $1.2 \times 10^{-4}$  M, MeOH)  $\lambda_{max}$  ( $\Delta \varepsilon$ ) 258 (-7.1), and 209 (+27.4) nm; <sup>13</sup>C and <sup>1</sup>H NMR data, see Table 1; ESIMS *m*/*z* 713 [M + Na]<sup>+</sup>; HRESIMS *m*/*z* 713.4022 [M + Na]<sup>+</sup> (calcd for C<sub>42</sub>H<sub>58</sub>O<sub>8</sub>Na, 713.4023).

**Cytotoxicity Testing.** Cell lines were purchased from the American Type Culture Collection (ATCC). Cytotoxicity assays of compounds 1–4 were performed using the Alamar Blue assay<sup>18,19</sup>. To measure the cytotoxicity activities of tested compounds, three concentrations in DMSO with three replications were performed on each cell line. 5-Fluorouracil and DMSO were used as positive and negative controls, respectively in this assay.

**Preparation of Human Neutrophils.** Human neutrophils obtained from peripheral blood of healthy adult volunteers (20–30 years old) were enriched using a standard method of dextran sedimentation, Ficoll-Hypaque centrifugation, and hypotonic lysis<sup>20,21</sup>. Purified neutrophils were resuspended in a Ca<sup>2+</sup> -free HBSS buffer (pH 7.4) at 4 °C prior to use.

**Measurement of O2<sup>-•</sup> Generation.** The O<sub>2</sub><sup>-•</sup> production was assayed based on the superoxide oxide dismutase inhibitable reduction of ferricytochrome  $c^{22-24}$ . Briefly, neutrophils (6 × 10<sup>5</sup> cells/mL) incubated with ferricytochrome c (0.5 mg/mL) and Ca<sup>2+</sup> (1 mM) were equilibrated at 37 °C for 2 min and then treated with DMSO as control or different concentrations of compounds for 5 min. Neutrophils were activated by 100 nM fMLP for 10 min in the pretreatment of cytochalasin B (CB, 1µg/mL) for 3 min (fMLP/CB).

**Measurement of Elastase Release.** The elastase release was assayed using MeO-Suc-Ala-Ala-Pro-Val-*p*-nitroanilide as the enzyme substrate<sup>23,24</sup>. Briefly, neutrophils ( $6 \times 10^5$  cells/mL) incubated with MeO-Suc-Ala-Ala-Pro-Val-*p*-nitroanilide ( $100 \mu$ M) were equilibrated at 37 °C for 2 min and treated with compounds for 5 min. Neutrophils were then activated with fMLP (100 nM)/CB ( $0.5 \mu$ g/mL) for 10 min.

**Statistical Analysis.** Results are expressed as the mean  $\pm$  SEM, and comparisons were made using Student's t-test. A probability value of 0.05 or less was considered significant. The software SigmaPlot was used for the statistical analysis.

*In Vitro* Anti-Inflammatory Assay. Macrophage (RAW264.7) cells were purchased from ATCC. *In vitro* anti-inflammatory activities of compounds 1–3 were measured by examining the inhibition of lipopolysaccharide (LPS) induced upregulation of iNOS (inducible nitric oxide synthetase) and COX-2 (cyclooxygenase-2) proteins in macrophages cells using Western blotting analysis<sup>25</sup>. For statistical analysis, all of the data were analyzed by a one-way analysis of variance (ANOVA), followed by the Student-Newman-Keuls *post hoc* test for multiple comparisons. A significant difference was defined as a *p* value of <0.05.

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#### **Author Contributions**

J.-H.S. designed the whole experiment and contributed to manuscript preparation. C.-Y.H., P.-J.S. and C.U. carried out the experiment and wrote the manuscript. C.-Y.H., J.-H.S. and S.-L.W. performed the isolation of compounds and the analysis of NMR and MS data. C.-Y.H., M.-C.L. and T.-L.H. performed and analyzed the bioassay. C.-F.D. identified the soft coral. All authors reviewed the manuscript.

#### **Additional Information**

Competing financial interests: The authors declare no competing financial interests.

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