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Hysteresis of ligand binding in CNGA2 ion channels

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Tetrameric cyclic nucleotide-gated (CNG) channels mediate receptor potentials in olfaction and vision. The channels are activated by the binding of cyclic nucleotides to a binding domain embedded in the C terminus of each subunit. Here using a fluorescent cGMP derivative (fcGMP), we show for homotetrameric CNGA2 channels that ligand unbinding is ~50 times faster at saturating than at subsaturating fcGMP. Analysis with complex Markovian models reveals two pathways for ligand unbinding; the partially liganded open channel unbinds its ligands from closed states only, whereas the fully liganded channel reaches a different open state from which it unbinds all four ligands rapidly. Consequently, the transition pathways for ligand binding and activation of a fully liganded CNGA2 channel differ from that of ligand unbinding and deactivation, resulting in pronounced hysteresis of the gating mechanism. This concentration-dependent gating mechanism allows the channels to respond to changes in the cyclic nucleotide concentration with different kinetics.

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vclic nucleotide-gated (CNG) ion channels mediate phototransduction in photoreceptors and chemotransduction in olfactory neurons¹⁻⁴. Wild-type olfactory CNG channels are heterotetramers composed of two CNGA2 subunits, one CNGA4 subunit and one CNGB1b subunit^{5,6}, of which only CNGA2 subunits can form functional homotetrameric channels. When studying steady-state concentration-activation relationships in inside-out membrane patches of CNGA2 channels, CNGA2 channels consistently produced Hill coefficients of ~ 2 , indicating that at least two ligands are required for full channel activation^{2,7}. Subsequent analysis of homotetrameric single CNGA2 channels confirmed that at least two ligands must bind for channel activation⁸. Using flash photolysis-induced cGMP jumps we later showed by a global fit strategy of activation time courses that at least three ligands are involved in channel activation and that the binding of the second ligand generates the main opening effect⁹. Combining such measurements with measurements of the ligand binding and the activation gating, we later demonstrated that all four ligands are involved in the activation gating, thereby confirming that the binding of ligand no. 2 is the key event for channel activation¹⁰. Moreover, our analysis revealed that models assuming ligand binding exclusively in the closed channel are sufficient to interpret the data. Therefore, the activation process seems to be operated by a complex cooperative mechanism: The binding of the first ligand hinders the binding of the second ligand, the binding of the second ligand promotes the binding of the third ligand, and the binding of the fourth ligand is not essentially influenced by the binding of the third ligand. With respect to the microscopic binding affinity, this results in a cooperativity sequence of 'negative-positive-no'. A cooperativity sequence of comparable complexity has been identified for the ligand-induced activation of structurally related HCN2 channels pre-activated by voltage, although with different characteristics^{11–13}. These results are inconsistent with the predictions of all types of models favouring a monotonous increase of the binding affinity with each ligand binding, including the Monod-Wyman-Changeux (MWC) model¹⁴, which is widely used to interpret the action of allosteric proteins, including the cooperative oxygen binding to haemoglobin¹⁵.

Herein we used a fluorescence labelled cGMP (fcGMP)^{10,16} in combination with confocal patch-clamp fluorometry^{10,17} at elevated time resolution to further analyse ligand binding and activation gating in CNGA2 channels in terms of Markovian models. To determine not only equilibrium constants but also rate constants, we applied concentration jumps¹² and subjected multiple time courses of ligand binding and activation as well as ligand unbinding and deactivation to a global fit analysis. We show that, in comparison with the incompletely liganded or unliganded channel, the fully liganded channel reaches a different open state, which allows for a much faster successive unbinding of all four ligands, thereby giving rise to a pronounced hysteresis in the probability flow during gating.

Results

Time courses of ligand binding and channel gating. We measured first the steady-state concentration-binding and concentration-activation relationship (Fig. 1a). As reported previously, these relationships cross each other and cannot be described with a simple Hill function¹⁰. Herein, these relationships were used to identify the three most promising concentrations to cover a wide concentration range for activation on the one hand and to provide time courses with reasonable resolution on the other. At low concentrations, the fluorescence signals are limited by the detector noise in the confocal images,

whereas at high concentrations the limitations are set by the background fluorescence of the bath solution. For the kinetic studies the concentrations 0.5, 2.0 and $12.0 \,\mu\text{M}$ fcGMP were chosen (Fig. 1a, grey arrows).

The averaged time courses of ligand binding and unbinding as well as activation and deactivation obtained with jumps to the chosen fcGMP concentrations and back to zero are shown in Fig. 1b. These averaged time courses represent the respective individual time courses well, as shown for the example of the fastest ligand binding and unbinding with 12 µM fcGMP in Supplementary Fig. S1. Concerning ligand binding and activation gating, the results are the following: ligand binding is monotonously accelerated to the higher concentrations and the time course of activation closely follows the time course of ligand binding with an only short delay. Notably, none of the time courses for ligand binding and channel activation is sigmoid. This again rules out an independent gating of the subunits or a MWCtype mechanism¹⁴ but confirms previous results obtained with cGMP, which led to the conclusion that the second binding step critically determines the activation gating^{9,10}.

Concerning ligand unbinding and deactivation, the situation proved to be more complex. In case of independent binding sites, one would expect that neither the ligand unbinding nor the deactivation gating depends on the ligand concentration. Our traces however showed that the unbinding at 12.0 µM fcGMP is \sim 50 times faster than the unbinding at the two lower concentrations (Fig. 1c). This is surprising because the unbinding at 2.0 μ M fcGMP starts from 54% of the saturating signal (c.f. Fig. 1a) and this degree of binding should be passed when coming from a higher degree of ligand binding. A further remarkable result is that the time delay between ligand unbinding and deactivation is large at 12 µM fcGMP and much smaller at the lower concentrations (Fig. 1b). Superimposition of the deactivation time courses shows that they are only moderately different at the three fcGMP concentrations and that the time course at $2.0\,\mu\text{M}$ fcGMP is slightly slower than those at 0.5 and $12.0\,\mu\text{M}$ fcGMP (Fig. 1d). Though at the first glance these results are puzzling, they provide robust and stringent information that can be used to further unravel the molecular gating in CNGA2 channels.

The C4L model is deficient to describe ligand unbinding. The C4L model (Fig. 2a) was determined in our previous work by evaluating steady-state binding and activation with fcGMP as well as activation kinetics with cGMP¹⁰. Our analysis was therefore started by testing whether this model is also adequate to describe the time courses of ligand binding in addition to the time courses of activation and the two steady-state relationships and to view the predictions for the time courses of ligand unbinding and deactivation (for fitting strategy see Supplementary Methods). Here and in all our fits we assumed equality of empty binding sites and equality of occupied binding sites, resulting in the stoichiometric factors 4, 3, 2, 1 and 1, 2, 3, 4 for the rates of binding and unbinding, respectively.

The result was that the C4L model fitted the time courses of ligand binding and activation together with the steady-state relationships well (Fig. 2b,c). Our fit also verified that two ligands suffice to fully activate the channel¹⁰, as shown by values of the equilibrium constants for closed–open isomerizations (E_0 , $E_1 \ll E_2$, E_3 , E_4 , Fig. 2a and Supplementary Table S1). When considering the time courses of ligand unbinding and deactivation, which were not fitted in this approach, it becomes obvious that at 12 μ M fcGMP the model does not match the fast ligand unbinding, whereas it approximately predicts the much slower deactivation time course (Fig. 2b).



Figure 1 | Experimental data of fcGMP binding and gating in CNGA2 channels. (a) Concentration-binding relationship and concentration-activation relationship under steady-state conditions over a wide concentration range between 30 nM and 29.1 μ M. Each data point is the mean of measurements obtained from 6 to 15 patches. Error bars denote s.e.m. The arrows indicate the concentrations used for the analyses of the time courses as shown in **b**. **(b)** Time courses of activation and deactivation as well as fcGMP binding and unbinding following a concentration pulse to either 0.5 μ M (13.0 s), 2.0 μ M (7.5 s), or 12.0 μ M (9.0 s) fcGMP and back to zero. The traces are averages of 7, 7, 10 currents, respectively, arising from different patches. For better comparison the traces were graphically cut after reaching a steady state. All traces were normalized with respect to the late current (I/I_{max}) or late fluorescence intensity (F/F_{max}). **(c)** Superimposition of the time courses of ligand unbinding. The colours correspond to **b**. The orange curves indicates the unbinding time course at 12.0 μ M fcGMP, which is ~ 50 times faster than the unbinding at the two lower concentrations. The obtained time courses are: 20.5 ms from 12 μ M fcGMP, 943.2 ms from 2.0 μ M fcGMP and 796.3 ms from 0.5 μ M fcGMP. **(d)** Superimposition of the deactivation time courses. The colors correspond to **b**.

To test whether at $12 \,\mu$ M fcGMP the model is able to fit also the time courses of ligand unbinding and deactivation when optimizing the parameters, we included also these time courses in the fit (Supplementary Fig. S2). The result was that the model still failed to describe the fast ligand unbinding and the slow channel deactivation at the same time. As this deficiency could not be overcome even when all rate constants were free parameters, the C4L model was classified as inadequate in its structure.

The C4L-*O4L model. The described inadequacies of the C4L model directed us to assume that pronounced ligand unbinding proceeds in the open channel. We therefore systematically extended the C4L model in several directions and fitted these models to the same data as used for the C4L model in Fig. 2. These models are shown in Supplementary Fig. S3.

In a first approach, we enabled ligand unbinding also from open states by inserting respective transitions in the C4L model. Starting from the fully liganded state O_4 , we successively increased the number of transitions between the open states from one to four (Supplementary Fig. S3, models 1–4). The result was that model 2 and 4 provided the best result. When removing the saturating ligand concentration of $12 \,\mu$ M fcGMP these models led to a rapid decrease of ligand binding to 50% of the amplitude but could not adequately describe the deactivation kinetics (not shown). Insertion of only one or three transitions (models 1 and 3) compromized the fit (see χ^2_r). We therefore suggested that other open states are passed during deactivation of the fully liganded and open channel than the ones populated during activation. To test this idea, we assumed that O4 qualitatively differs from O₀, O₁, O₂ and O₃ by being the starting point for an additional pathway mediating the fast ligand unbinding at 12 µM fcGMP, whereas the following small slow phase of unbinding was thought to become fed by emptying O₃ and/or O₂ along the respective C pathway. The O states in the extra pathway are termed in the following *O₄, *O₃, *O₂, *O₁, *O₀ (C4L-*O4L model; Fig. 3a). As the function of the *O-pathway was to enable a fast unbinding, we chose as simple approach to equalize both all forward rate constants and all backward rate constants. The result was a significant improvement of the fit compared with the C4L model (Fig. 3b,c). It should be noted that the binding constants along the *O-pathway were set to $6 \times$ 108 M⁻¹ s⁻¹, a reasonable value for a diffusion-limited binding¹⁸. As k_{C0^*O0} is very slow it could be suggested that activation to *O₄ proceeds exclusively along the C pathway, whereas deactivation from *O₄ proceeds to a significant degree along the *O-pathway. Moreover, inclusion of the unbinding and deactivation time courses allowed us to differentiate between



Figure 2 | **Fit of ligand binding and activation with the C4L model.** (a) Scheme of the C4L model. C_x and O_x (n = 0...4) denote closed and open states, respectively. *L* is a ligand. The numbers are the values of the rate constants (s^{-1} or $s^{-1}M^{-1}$) determined by the fit. Rate constants set equal in the fit are indicated by using the same colours. The errors for each rate constant are given in Supplementary Table S1. (b) Time courses of binding and unbinding (green) as well as activation and deactivation (black) for three fcGMP concentrations. The time interval between the concentration jump from 0 to the respective fcGMP concentration and back to 0 is indicated for each trace by the black bar at the top. (c) Steady-state relationships for activation (black symbols) and ligand binding (green symbols). Each data point is the mean of measurements obtained from 6 to 15 patches. Error bars denote s.e.m. The red curves in **b** and **c** are the result of the best global fit of the time courses of ligand binding and activation as well as the steady-state relationships with the parameters shown in **a**. The reduced χ_r^2 for the activation/binding and steady-state was 5.0.

 K_{AC1} , K_{AC3} , and K_{AC4} , which could not be achieved in our previous study¹⁰. The values of the equilibrium association rate constants were $K_{\rm AC1} = 1.6 \times 10^5$, $K_{\rm AC2} = 8.2 \times 10^4$, $K_{\rm AC3} =$ 2.4×10^5 , and $K_{AC4} = 5.1 \times 10^5$. Before discussing the processes in the C4L-*O4L model in greater detail, it should be noted that multiple other models with an *O-pathway were considered that were however inferior to the C4L-*O4L model with respect to a minimum in complexity and number of parameters as well as to a maximum in the parameters' determinateness. A selection of models is shown in Supplementary Fig. S3, including models with ligand unbinding via C_x and *O_x starting from incompletely liganded states O_x and running to C_0 (model 5 and 6), models with ligand unbinding via C_x , O_x and $*O_x$ running to C_0 (model 7 and 8), models with ligand unbinding via Cx and *Ox starting from more than one O_x and running to C_0 (models 9–11), and models with ligand unbinding via C_x and C_x running to O_0 (models 12-14).

Time-dependent population of states in the C4L-*O4L model. Using the rate constants determined for the C4L-*O4L model (Fig. 3a; Supplementary Table S2) the probabilities of being in the individual open states ($P_{C,x}$, $P_{*O,x}$; Fig. 4a) and closed states ($P_{C,x}$; Fig. 4b) as well as the contributions of each state to the total ligand binding were computed as function of time (Fig. 4c).

When stepping to $12.0 \,\mu\text{M}$ fcGMP (Fig. 4, right), the rapid increase of *Po* is caused by the population of three states: *P*_{*0,4} is populated by a dominating initial fast phase followed by a slow phase to reach finally 0.8 (Fig. 4a, right). A minor portion of *Po* is

generated by $P_{O,3}$ and $P_{O,2}$. Hence, even at the saturating concentration of 12.0 μ M fcGMP $P_{O,3}$ and $P_{O,2}$ are significantly populated. The time courses of binding mirror this respectively (Fig. 4c, right). For the closed states the situation is simple, they are emptied rapidly (Fig. 4b, right). When removing fcGMP, there are two pathways of deactivation. *O₄ is emptied exclusively and with high speed along $*O_3 \rightarrow *O_2 \rightarrow *O_1 \rightarrow *O_0$. *O₀ is relatively stable and only the transition $*O_0 \rightarrow C_0$ causes slow deactivation (Fig. 4a,b, right). In contrast, O₃ empties via $C_3 \rightarrow C_2 \rightarrow C_1$ to C_0 , thereby dominating the small slow phase of unbinding. Similarly, O₂ empties via $C_2 \rightarrow C_1$ to C_0 . However, its contribution to the total unbinding is only minor. Hence, after removing the saturating fcGMP concentration of 12 μ M, there are two relevant but distinct pathways for both unbinding and deactivation.

At the intermediate fcGMP concentration of 2.0 μ M fcGMP, the occupancy of *O₄ reaches only <0.2, whereas O₃ and O₂ sum up to >0.6 (Fig. 4a, middle). As the latter can only deactivate along C₃ \rightarrow C₂ \rightarrow C₁ \rightarrow C₀ and C₂ \rightarrow C₁ \rightarrow C₀, respectively, unbinding is dominated by this slow process (Fig. 4c, middle). Concerning the closed states, C₁ and C₀ remain populated in the presence of the ligand in total occupancy 0.2, whereas the other closed states are not essentially populated (Fig. 4b, middle). Nearly all ligand binding comes from the open states O₂, O₃ and *O₄ (Fig. 4c, middle). In the logic of the previous, the total time course of unbinding is dominated by emptying O₃ and O₂ (Fig. 4c, middle), whereas for total deactivation also the reaction *O₀ \rightarrow C₀ is of relevance (Fig. 4a, middle).

At the low fcGMP concentration of $0.5 \,\mu$ M, there is only a moderate opening by populating O₂ (Fig. 4a, left). The other open



Figure 3 | **Fit with the C4L-*O4L model including ligand unbinding and deactivation.** (a) Scheme of the C4L-*O4L model. C_x and O_x (n = 0...4) denote closed and open states, respectively. * O_x denotes states of an extra pathway used only in deactivation. *L* is a ligand. Again, the numbers provide the values of the rate constants (s⁻¹ or s⁻¹M⁻¹) determined by the fit. Rate constants set equal in the fit are indicated by using the same colours. The errors for each rate constant are given in Supplementary Table S2. (b) The time courses of ligand binding and unbinding as well as activation and deactivation are the same as those in Fig. 2c. Each data point is the mean of measurements obtained from 6 to 15 patches. Error bars denote s.e.m. The red curves in **b** and **c** are the result of the best global fit with the C4L-*O4L model. The reduced χ^2_r was 6.0.

states do not play an essential role. Accordingly, also deactivation and ligand unbinding are dominated by emptying O₂ (Fig. 4a,c, left). In the presence of the ligand, the closed states C₀ and C₁ are populated by ~55% and ~20%, respectively (Fig. 4b, left). As *O₄ is not populated at this concentration, unbinding and deactivation along *O-pathway does not proceed. Hence, at this low concentration there is only one relevant pathway for deactivation, $O_2 \rightarrow C_2 \rightarrow C_1 \rightarrow C_0$.

The C4L-*O4L model can indeed explain the surprising result of a 50 times faster ligand unbinding at saturating compared to non-saturating fcGMP by introducing an additional open state, *O, that can be reached only when all four binding sites are occupied and is emptied quickly when removing the ligand. Nevertheless, it remains remarkable that there is such a big kinetic difference between the two concentrations 2 and 12 µM fcGMP. To further approve the validity of our C4L-*O4L model, we first simulated time courses of ligand unbinding to identify a reasonably intermediate time course between 2 and 12 µM fcGMP and then tested whether the prediction is met by the experiment (Fig. 5). We chose $4.5 \,\mu\text{M}$ fcGMP. The predicted time course of unbinding shows an initial rapid phase, due to the unbinding along the *O-pathway, followed by a slow phase with an amplitude between that at 12 and 2 µM fcGMP, evoked by the unbinding from O_3 and O_2 , (blue curve in Fig. 5). As these basic features are indeed met by the experiment (green trace in Fig. 5)

with an only slight deviation, our C4L-*O4L model is considered to be adequate.

Transition pathways in the C4L-*O4L model. In order to visualize the qualitative differences in the gating at the saturating fcGMP concentration of 12 μ M and the low fcGMP concentration of 0.5 μ M, we considered the probability fluxes¹¹ in our C4L-*O4L model. Knowledge of the occupancy of each state at each time allowed us to compute for all individual transitions between two neighboured states X \leftrightarrow Y the total net probability flux, F_{XY} , that is, the net amount of probability moving between the two states either when fcGMP was applied or when it was removed (Supplementary Methods). Knowing all total net probability fluxes in our C4L-*O4L model, we computed the pathway net probability flux, $F_{p,XY}$, between two arbitrary, not necessarily neighboured, states X and Y within the model (equation S10 in Supplementary Methods).

As in the absence of fcGMP the channel is nearly completely in C_0 and at 12 µM fcGMP nearly completely in O_2 , O_3 , or * O_4 (Fig. 4a,b, right), for channel activation the pathway net probability fluxes $F_{p,C002}$, $F_{p,C003}$ and $F_{p,C0^*O4}$ were considered (red arrows in Fig. 6). In this context, C_0 is the flux producer and O_2 , O_3 , * O_4 are the flux collectors. It is evident that at 12 µM fcGMP $F_{p,C0^*O4}$ dominates channel activation by far (Fig. 6a),



Figure 4 | Time-dependent state population in the C4L-*O4L model. Using the parameters given in Fig. 3a, the population of all states in the C4L-*O4L model was computed for three fcGMP concentrations. (a) Upper time courses: open probability, *Po* (red curve, same curves as in Fig. 3b) and probability of being either in O_x , P_{O_x} (x = 0....3) or * O_x , P_{*O_x} (x = 0....4; coloured curves). The P_{O_x} and P_{*O_x} values sum up to *Po*. Lower time courses: part of the respective upper time courses at an expanded timescale. (b) Probability of being in the closed state x, P_{C_x} (x = 0....4). The $P_{O_x y}$ and $P_{C_{xy'}}$ values sum up to unity. (c) Ligand binding to the total channel (red curve) to the individual open ($x/4 \times P_{O_x y'}$ is $x/4 \times P_{*O_x y'}$ coloured continuous) and closed states ($x/4 \times P_{C_x y'}$ coloured dashed), given as fraction of unity (left ordinates). x/4 weighs the contribution of each liganded state to the total liganding. NL indicates the number of ligands bound to the channel and to the individual states under the assumption that maximally four ligands can bind to the channel (right ordinate).

whereas it becomes irrelevant at 0.5 μ M fcGMP, where the fractional activation is generated by $F_{p,C0O2}$ and $F_{p,C0O3}$ (Fig. 6c). At 2.0 μ M fcGMP, $F_{p,C0O2}$ is already larger and $F_{p,C0O3}$ in the same order as $F_{p,C0^*O4}$ (Fig. 6b).

For considering channel deactivation, the opposite pathway net probability fluxes were considered, $F_{p,O2C0}$, $F_{p,O3C0}$, and $F_{p,*O4C0}$ (blue arrows in Fig. 6). O_2 , O_3 , or * O_4 are now the flux producers and C_0 is the flux collector. O_2 and O_3 can only be emptied along the activation pathway to C_0 because the transition $C_3 \rightarrow C_4$ is blocked in the absence of a ligand. In contrast, practically all probability flux collected in * O_4 via the C pathway reaches C_0 along the *O pathway (Fig. 6a–c). The consequence is that channel activation and deactivation generated by populating * O_4 is associated with a nearly perfect hysteresis, whereas no hysteresis is associated with the population of O_2 and O_3 . Consequently, maximal liganding and opening of a CNGA2 channel produces an open state * O_4 that basically differs from $O_0....O_3$.

Discussion

It is clear for long that the function of multimeric allosteric proteins is essentially defined by the cooperative interaction of the subunits. The general understanding of such kind of interaction however is only poor though there has been considerable progress in the past decades in both experimental approaches and computer simulations. Herein, we analysed the function of homotetrameric CNGA2 ion channels in terms of a complex Markovian model and quantified the cooperative interaction of the four equal channel subunits during a complete cycle of activation and deactivation in detail.

Methodologically, all tested Markovian models were globally fitted to a set of essentially different data, containing time courses of ligand binding and unbinding as well as channel activation and deactivation, evoked by jumps to three ligand concentrations and back to zero, respectively, in addition to the respective steadystate relationships. It is this manifold of experimental data, and in particular the duality of ligand binding and activation, which was



Figure 5 | Predicted and measured time course of ligand unbinding at 4.5 μ M fcGMP. The approximated time courses of ligand unbinding at 0.5 μ M, 2.0 μ M, 4.5 μ M and 12 μ M fcGMP are obtained by the global fit with the C4L-*O4L model (c.f. Fig. 3). The time course at 4.5 μ M (blue curve), simulated with C4L-*O4L model, approximately matches the experimentally determined time course at this concentration (green trace). The experimentally determined time course is the average from five individual time courses.

the prerequisite for getting the necessary constraints to identify an appropriate model and to determine a reasonable set of parameters. A similar approach we used previously to specify the subunit action in homotetrameric HCN2 channels pre-activated by a hyperpolarizing voltage¹².

The most important finding of this report is that the fully liganded and activated channel adopts an open conformation *O4 that qualitatively differs from the open conformations of the partially liganded channel reached during activation (O_x). From *O₄ ligand unbinding along the *O-pathway proceeds \sim 50 times faster than from the open states O1, O2, and O3 along the C-pathway passed after channel closure (Fig. 3a). This remarkable result is associated with a pronounced hysteresis for the activation of the fully liganded channel. Because at the saturating concentration of $12 \,\mu\text{M}$ fcGMP *O₄ dominates the open states by far (Fig. 4a, right), this indicates that for high ligand concentrations the hysteresis dominates the activationdeactivation cycle. The reason for the pronounced hysteresis is that during channel activation, starting from C₀, k_{C0^*O0} is undeterminably slow, leading to a block of the *O-pathway in the forward direction and passage of the C-pathway, whereas during channel deactivation, starting from $*O_4$, k_{*O4C4} is much slower than k_{*O4*O3} , k_{*O3*O2} , k_{*O2*O1} , k_{*O1*O0} (Supplementary Table S2), leading to a block of the C-pathway and passage of the *O-pathway. Hence, the *O-pathway plays the role of an unbinding bypass. To finally reach C_0 , the slow reaction * $O_0 \rightarrow C_0$ has to be passed (k_{*OOC0}). k_{*OOC0} is similarly slow as k_{*O4C4} (Supplementary Table S2).

Notably, apart from the *O-pathway, essential aspects of the C4L-*O4L model confirm our previous C4L model insofar that only two ligands suffice to fully open the channel ($E_0, E_1 \rightarrow 0; E_2, E_3, E_4 \rightarrow 1$) and that the microscopic affinity of the second binding step is critical. Solely these results, but still more the additional *O-pathway, show that the structure and the assumptions of the MWC model^{14,19} and related models, including the coupled dimer model²⁰, are oversimplifying to describe the molecular gating in CNGA2 channels.

In the context of ion channel gating, hysteresis was observed previously when an additional slow gating process superimposes with a fast gating process, thereby altering the gating parameters. Theoretically, this can only appear if both processes are coupled. Examples are the voltage gating in alamethicin channels²¹, HERG-channels²², NMDA receptors²³, TRPV3 channels²⁴ and HCN channels where the phenomenon was termed 'mode



Figure 6 | Transition pathways in the C4L-*O4L model. The net probability fluxes along specified transition pathways between state X and Y, $F_{p,XY}$, were computed by integrating the net probability flux densities and employing the principles of the transition pathway theory (see Supplementary Methods). The weight of the pathway net probability fluxes is semiquantitatively encoded by the thickness of the arrows (fluxes below 2×10^{-4} are indicated by dotted lines. The numbers besides the coloured graphs indicate the exact weights as fraction of unity. (a) 12 µM fcGMP. Pathway net probability fluxes along the pathways $C_0...^*O_4$, $C_0...O_3$, and $C_0...O_2$ after applying fcGMP (red) and after its removal (blue). (b,c) Respective pathway net probability fluxes at 2.0 µM and 0.5 µM fcGMP, respectively.

shift^{25,26}. We are not aware that hysteretic gating has been analysed and termed as such in ligand-gated channels. However, the situation is certainly related if activation of the channels is followed by desensitization and if both processes are coupled. In this context, desensitization means that the channel activity decays and that the response of the channel to the ligand conjointly decreases. For the nicotinic receptor desensitization has been studied extensively, thereby showing that desensitization is associated with an increase of the affinity for the ligand^{27–31}. Furthermore, it has been shown that the recovery rate from desensitization is much faster than desensitization itself. These results led to a cyclic mechanism describing the interplay of activation and desensitization that consists of four states (for review, refer to Sine *et al.*³²).

For the ligand gating in HCN2 channels we recently described a complex Markov model with pronounced hysteresis, in which ligand binding proceeds preferentially along the closed states, whereas ligand unbinding proceeds preferentially along the open states, and moreover, activation is facilitated only when one, three or four ligands are bound¹³. The hysteresis observed herein is similarly related to a complex Markov model, the C4L-*O4L model. In contrast to HCN2 channels, the hysteresis identified for the CNGA2 channels employs two types of open states, O_x and *O_x, of which the *O_x states are only reached after the fourth ligand has bound, and for which the binding affinity is essentially reduced. This is also different to the mentioned results in nicotinic receptors, where the complexity of the used models was much lower than in our C4L-*O4L model.

We recently demonstrated for the HCN2 channel gating the principle of reciprocity, saying that if ligand binding promotes channel activation then activation promotes ligand binding¹⁷. We did this by quantifying the affinity increase for fcAMP, a fluorescent ligand analogue to fcGMP, when activating the channels by voltage. However, the principle of reciprocity also means that the ligand binding becomes more tight when the conformational change evoked by the ligand binding has taken place³³. Hence, in multimeric proteins with multiple binding sites, it is generally assumed that the channel affinity increases in proportion to the number of ligands bound. In HCN2 channels we showed that this is not necessarily the case: The third and fourth ligand are much less tightly bound than the first and the second ligand¹².

Herein we show a more severe derivation from this assumption: In the C4L-*O4L model the affinity depends, of course, on the respective values of K_{AC} but also on the closed-open isomerizations $C_2 \leftrightarrow O_2$ and $C_3 \leftrightarrow O_3$ that are strongly shifted to the respective open states. Progressive increase of the ligand concentration therefore increases *Po* because of an increasing recruitment of these closed-open isomerizations. The strong influence of the allosteric transition on the binding affinity has been insightfully demonstrated already for the simple del Castillo-Katz mechanism³³. Notably, the binding of the fourth ligand evokes opening to an additional open state *O₄, enabling an additional unbinding pathway, resulting in 50 times faster unbinding (Fig. 1c) along the states *O₃, *O₂, *O₁ and *O₀ (Fig. 3a).

This result indicates that completion of the binding by the fourth ligand evokes a wide conformational change in the tetrameric cyclic nucleotide-binding domain (CNBD) affecting all binding sites, possibly by generating a symmetric conformation. Conversely, when the channel is only double or triple liganded, a non-symmetric conformation of the CNBD would lead to a trapping of the ligands by populating O_2 and O_3 . For all these considerations upon hysteresis one should be aware that only during activation with the ligand present, the C4L-*O4L model can reach an equilibrium with a detailed balance, that is, the net flux in the loop of our model is zero. In contrast, after removal of the ligand during deactivation the model is far away from equilibrium and there is no detailed balance.

What could be the physiological meaning of a much faster cGMP unbinding from the fully than from the partially liganded open CNGA2 channel? When addressing this question one has to be aware that natural olfactory channels contain two CNGA2 subunits together with one CNGA4 and one CNGB1b subunit and it is *a priori* not clear whether the CNGA2 subunits work in these natural heterotetrameric channels similarly as in the homotetrameric channels considered herein. When assuming such a similarity, one has to consider that the olfactory channels can be activated by both cAMP and cGMP^{2,34}. In olfactory sensory neurons it is well established that cAMP essentially contributes to the transmembrane receptor potential by opening

CNG channels^{35–38}. However, cGMP is also available and should therefore interfere. cGMP has been reported to participate preferentially in long-term responses^{39–41}. The type of cooperation of the two ligands however is still elusive. Concerning our results, one may speculate that the rapid ligand unbinding in the fully activated channel might become relevant for enabling a rapid exchange of the two types of cyclic nucleotides at the fully but not the partially liganded channel. Another speculation is that the described fast ligand unbinding at high ligand concentrations leads to different cellular responses when the cyclic-nucleotide concentrations in the cell change. It seems to be an attractive idea to analyse in future heterotetrameric olfactory channels with a similar approach as used in the present study.

Methods

Oocyte Preparation and cRNA Injection. Oocytes were obtained surgically under anaesthesia (0.3% 3-aminobenzoic acid ethyl ester) from adult females of *Xenopus laevis*.

The procedures had approval from the authorized animal ethical committee of the Friedrich-Schiller University Jena (Germany).

The oocytes were treated with collagenase A (3 mg ml⁻¹, Roche, Grenzach-Wyhlen, Germany) for 105 min in Ca²⁺-free Barth's solution containing (in mM) 82.5 NaCl, 2 KCl, 1 MgCl₂, 5 HEPES (pH 7.5). After this procedure, oocytes of stage IV and V and were manually dissected and injected with ~50 ng of cRNA encoding WT CNGA2 channels. After injection with cRNA, the oocytes were cultured at 18 °C for 3–4 days in Barth's solution containing (in mM) 84 NaCl, 1 KCl, 2.4 NaHCO₃, 0.82 MgSO₄, 0.41 CaCl₂, 0.33 Ca(NO₃)₂, 7.5 TRIS, Cefuroxim, Penicillin/Streptomycin (pH 7.4).

Confocal patch-clamp fluorometry. The fluorescence and the ionic current were measured simultaneously by combining patch-clamp fluorometry⁴² with confocal microscopy^{10,17}. Macroscopic ion currents were recorded with the patch-clamp technique⁴³ in inside-out patches of *Xenopus laevis* oocytes expressing bovine homotetrameric CNGA2 channels (accession number: X55010). The patch pipettes were pulled from borosilicate glass tubing with an outer and inner diameter of 2.0 and 1.0 mm (Hilgenberg GmbH, Malsfeld, Germany). The pipette resistance was 0.7-1.2 MΩ. The bath and pipette solution contained (in mM): 150 KCl, 1 EGTA, 5 Hepes (pH 7.4 with KOH). Jumps of the ligand concentration were evoked by a theta-glass pipette mounted on a piezo-driven device⁴⁴ generating a laminar flow of a control bath solution and a bath solution including the ligand. The time course of solution exchange at the patch was 5.6 ± 0.6 ms (n = 8; 10–90% rise time; t_{10-90}), as determined from the increase of fluorescence close to the patch membrane when switching from zero to 3 µM fcGMP. This time was in the range of one to two sampling intervals for ligand binding and unbinding and therefore considered to be instantaneous. The duration of the solution pulses was 13.0 s, 7.5 s and 9.0 s at 0.5 µM, 2.0 µM and 12.0 µM fcGMP, respectively. Electrophysiology was controlled by the ISO3 hardware and software (MFK, Niedernhausen, Germany). The sampling rate for current recording was 5,000 Hz and the filter (4-pole Bessel) was set to 2 kHz. The recording voltage was generally set to 10 mV.

To monitor the ligand binding, we used 8-DY547-AET-cGMP (fcGMP) as fluorescent ligand^{10,16}. fcGMP consists of a cGMP to which the fluorescent dye DY547 (Dyomics Jena, Germany) is linked by an aminoethylthio-spacer to position 8 of the guanosine moiety. Recording was performed with the confocal microscope LSM 5 Live (Zeiss, Jena, Germany). The microscope was triggered by the ISO3 software (MFK, Niedernhausen, Germany). The recording rate of the images was 277 Hz.

To distinguish in an LSM image the fluorescence of the non-bound fcGMP from that of the bound fcGMP, a second dye, DY647 (ref. 10), was added to the bath solution. fcGMP and DY647 were excited with the 532 nm and 635 nm line of a solid-state laser via the AchroGate beam splitter, respectively. The emitted light was split into two channels by a NFT635 dichroic mirror and further filtered by BP550–615 and LP 650 emission filter (Zeiss, Jena, Germany), respectively. The fcGMP molecules bound to the channels in the patch dome were obtained from the fluorescence intensity in the patch dome from which the fluorescence arising from the unbound molecules was subtracted. The procedure has been described previously¹⁰. This fluorescence, *F*, was normalized in each patch with respect to the fluorescence r_{max} at the saturating fcGMP concentration of 29.1 μ M.

Global fit of kinetic models. The rate and/or equilibrium constants for a given model were obtained by a global fit to averaged normalized time courses of binding/unbinding and of activation/deactivation, each at 0.5, 2.0 and 12.0 μ M fcGMP, as well as to steady-state binding and activation over a wide concentration range. The strategy is described in Supplementary Methods.

Statistical analysis. Data are given as mean ± s.e.m.

References

- Zagotta, W. N. & Siegelbaum, S. A. Structure and function of cyclic nucleotidegated channels. Annu. Rev. Neurosci. 19, 235–263 (1996).
- Kaupp, U. B. & Seifert, R. Cyclic nucleotide-gated ion channels. *Physiol. Rev.* 82, 769–824 (2002).
- Craven, K. & Zagotta, W. CNG and HCN channels: two peas, one pod. Annu. Rev. Physiol. 68, 375–401 (2006).
- Matulef, K. & Zagotta, W. N. Cyclic nucleotide-gated ion channels. Annu. Rev. Cell Dev. Biol. 19, 23–44 (2003).
- Bönigk, W. et al. The native rat olfactory cyclic nucleotide-gated channel is composed of three distincts subunits. J. Neurosci. 19, 5332–5347 (1999).
- Zheng, J. & Zagotta, W. N. Stoichiometry and assembly of olfactory cyclic nucleotide-gated channels. *Neuron* 42, 411–421 (2004).
- Dhallan, R. S., Yau, K. W., Schrader, K. A. & Reed, R. R. Primary structure and functional expression of a cyclic nucleotide-activated channel from olfactory neurons. *Nature* 347, 184–187 (1990).
- Li, J. & Lester, H. A. Single-channel kinetics of the rat olfactory cyclic nucleotide-gated channel expressed in xenopus oocytes. *Mol. Pharmacol.* 55, 883–893 (1999).
- Nache, V. et al. Activation of olfactory-type cyclic nucleotide-gated channels is highly cooperative. J. Physiol. 569, 91–102 (2005).
- Biskup, C. et al. Relating ligand binding to activation gating in CNGA2 channels. Nature 446, 440–443 (2007).
- 11. Benndorf, K., Thon, S. & Schulz, E. Unraveling subunit cooperativity in homotetrameric HCN2 channels. *Biophys. J.* **103**, 1860–1869 (2012).
- Kusch, J. et al. How subunits cooperate in cAMP-induced activation of homotetrameric HCN2 channels. Nat. Chem. Biol. 8, 162–169 (2012).
- Benndorf, K., Kusch, J. & Schulz, E. Probability fluxes and transition paths in a markovian model describing complex subunit cooperativity in HCN2 channels. *PLoS Comput. Biol.* 8, e1002721 (2012).
- 14. Monod, J., Wyman, J. & Changeux, J. P. On the nature of allosteric transitions: a plausible model. J. Mol. Biol. 12, 88-118 (1965).
- Perutz, M. F., Wilkinson, A. J., Paoli, M. & Dodson, G. G. The stereochemical mechanism of the cooperative effects in hemoglobin revisited. *Annu. Rev. Biophys. Biomol. Struct.* 27, 1–34 (1998).
- Nache, V. *et al.* Differential regulation by cyclic nucleotides of the CNGA4 and CNGB1b subunits in olfactory cyclic nucleotide-gated channels. *Sci. Signal.* 5, ra48 (2012).
- 17. Kusch, J. *et al.* Interdependence of receptor activation and ligand binding in HCN2 pacemaker channels. *Neuron* **67**, 75–85 (2010).
- van Holde, K. E. A hypothesis concerning diffusion-limited protein-ligand interactions. *Biophys. Chem.* 101-102, 249–254 (2002).
- Goulding, E. H., Tibbs, G. R. & Siegelbaum, S. A. Molecular mechanism of cyclic-nucleotide-gated channel activation. *Nature* 372, 369–374 (1994).
- Liu, D. T., Tibbs, G. R., Paoletti, P. & Siegelbaum, S. A. Constraining ligandbinding site stoichiometry suggests that a cyclic nucleotide-gated channel is composed of two functional dimers. *Neuron* 21, 235–248 (1998).
- Duclohier, H., Kociolek, K., Stasiak, M., Leplawy, M. T. & Marshall, G. R. C-terminally shortened alamethicin on templates: influence of the linkers on conductances. *Biochim. Biophys. Acta* 1420, 14–22 (1999).
- Pennefather, P. S., Zhou, W. & DeCoursey, T. E. Idiosyncratic gating of HERGlike K + channels in microglia. J. Gen. Physiol. 111, 795–805 (1998).
- Nowak, L. M. & Wright, J. M. Slow voltage-dependent changes in channel open-state probability underlie hysteresis of NMDA responses in Mg(2 +)-free solutions. *Neuron* 8, 181–187 (1992).
- Liu, B., Yao, J., Zhu, M. X. & Qin, F. Hysteresis of gating underlines sensitization of TRPV3 channels. J. Gen. Physiol. 138, 509–520 (2011).
- Männikkö, R., Pandey, S., Larsson, H. P. & Elinder, F. Hysteresis in the voltage dependence of HCN channels: conversion between two modes affects pacemaker properties. J. Gen. Physiol. 125, 305–326 (2005).
- Bruening-Wright, A. & Larsson, H. P. Slow conformational changes of the voltage sensor during the mode shift in hyperpolarization-activated cyclicnucleotide-gated channels. *J. Neurosci.* 27, 270–278 (2007).
- Quast, U. et al. Ligand-induced conformation changes in Torpedo californica membrane-bound acetylcholine receptor. *Biochemistry* 17, 2405–2414 (1978).
- Weber, M. & Changeux, J. P. Binding of Naja nigricollis (3H)alpha-toxin to membrane fragments from Electrophorus and Torpedo electric organs. 3. Effects of local anaesthetics on the binding of the tritiated alpha-neurotoxin. *Mol. Pharmacol.* 10, 35–40 (1974).
- Weiland, G., Georgia, B., Lappi, S., Chignell, C. F. & Taylor, P. Kinetics of agonist-mediated transitions in state of the cholinergic receptor. *J. Biol. Chem.* 252, 7648–7656 (1977).

- Heidmann, T., Bernhardt, J., Neumann, E. & Changeux, J. P. Rapid kinetics of agonist binding and permeability response analyzed in parallel on acetylcholine receptor rich membranes from Torpedo marmorata. *Biochemistry* 22, 5452–5459 (1983).
- Sine, S. M. & Taylor, P. The relationship between agonist occupation and the permeability response of the cholinergic receptor revealed by bound cobra alpha-toxin. J. Biol. Chem. 255, 10144–10156 (1980).
- 32. Sine, S. M. End-plate acetylcholine receptor: structure, mechanism, pharmacology, and disease. *Physiol. Rev.* **92**, 1189–1234 (2012).
- Colquhoun, D. Binding, gating, affinity and efficacy: The interpretation of structure-activity relationships for agonists and of the effects of mutating receptors. Br. J. Pharmacol. 125, 923–947 (1998).
- Dhallan, R. S., Yau, K. W., Schrader, K. A. & Reed, R. R. Primary structure and functional expression of a cyclic nucleotide-activated channel from olfactory neurons. *Nature* 347, 184–187 (1990).
- Kurahashi, T. & Yau, K. W. Co-existence of cationic and chloride components in odorant-induced current of vertebrate olfactory receptor cells. *Nature* 363, 71–74 (1993).
- Schild, D. & Restrepo, D. Transduction mechanisms in vertebrate olfactory receptor cells. *Physiol. Rev.* 78, 429–466 (1998).
- Gold, G. H. Controversial issues in vertebrate olfactory transduction. Annu. Rev. Physiol. 61, 857–871 (1999).
- Firestein, S. How the olfactory system makes sense of scents. Nature 413, 211–218 (2001).
- Kroner, C., Boekhoff, I., Lohmann, S. M., Genieser, H. G. & Breer, H. Regulation of olfactory signalling via cGMP-dependent protein kinase. *Eur. J. Biochem.* 236, 632–637 (1996).
- Leinders-Zufall, T. *et al.* Contribution of the receptor guanylyl cyclase GC-D to chemosensory function in the olfactory epithelium. *Proc. Natl Acad. Sci. USA* 104, 14507–14512 (2007).
- 41. Pietrobon, M. *et al.* Interplay among cGMP, cAMP, and Ca2 + in living olfactory sensory neurons *in vitro* and *in vivo. J. Neurosci.* **31**, 8395–8405 (2011).
- Zheng, J. & Zagotta, W. N. Patch-clamp fluorometry recording of conformational rearrangements of ion channels. Sci. STKE 176, 1–13 (2003).
- Hamill, O. P., Marty, A., Neher, E., Sakmann, B. & Sigworth, F. J. Improved patch-clamp techniques for high-resolution current recording from cells and cell-free membrane patches. *Pflugers Arch.* 391, 85–100 (1981).
- 44. Jonas, P. High-speed solution switching using piezo-based micropositioning stages. *In Single-Channel Recording* (eds Sakmann, B. & Neher, E.) 231–243 (Plenum Press, 1995).

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Author contributions

V.N. carried out the combined electrophysiological and optical measurements, analysed the experimental data, apart from the global fit, and prepared the figures. T.E. and E.S. performed the global fits. R.S. contributed to the experimental design. K.B. planned the project, designed experiments and wrote the manuscript.

Additional information

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