



How epigallocatechin gallate binds and assembles oligomeric forms of human alpha-synuclein

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The intrinsically disordered human protein α -synuclein (α SN) can self-associate into oligomers and amyloid fibrils. Several lines of evidence suggest that oligomeric α SN is cytotoxic, making it important to devise strategies to either prevent oligomer formation and/or inhibit the ensuing toxicity. (–)-epigallocatechin gallate (EGCG) has emerged as a molecular modulator of α SN self-assembly, as it reduces the flexibility of the C-terminal region of α SN in the oligomer and inhibits the oligomer's ability to perturb phospholipid membranes and induce cell death. However, a detailed structural and kinetic characterization of this interaction is still lacking. Here, we use liquid-state NMR spectroscopy to investigate how EGCG interacts with monomeric and oligomeric forms of α SN. We find that EGCG can bind to all parts of monomeric α SN but exhibits highest affinity for the N-terminal region. Monomeric α SN binds ~54 molecules of EGCG in total during oligomerization. Furthermore, kinetic data suggest that EGCG dimerization is coupled with the α SN association reaction. In contrast, preformed oligomers only bind ~7 EGCG molecules per protomer, in agreement with the more compact nature of the oligomer compared with the natively unfolded monomer. In previously conducted cell assays, as little as 0.36 EGCG per α SN reduce oligomer toxicity by 50%. Our study thus demonstrates that α SN cytotoxicity can be inhibited by small molecules at concentrations at least an order of magnitude below full binding capacity. We speculate this is due to cooperative binding of protein-stabilized EGCG dimers, which in turn implies synergy between protein association and EGCG dimerization.

The intrinsically disordered human protein α -synuclein (α SN) accumulates in the brains of patients with Parkinson's disease as intracellular deposits called Lewy bodies (1, 2). Self-assembly of α SN also occurs *in vitro*, where the protein is found to form oligomeric species and amyloid fibrils (3). The soluble oligomeric form that is formed by self-assembly is cytotoxic, in the sense that it binds and permeates cell

membranes (4–9). Recently, it was shown that monomeric α SN and iron-induced oligomeric forms of α SN interact in different ways with lipid membranes: whereas monomers cause membrane thinning and oligomers interact with and lead to changes within lipid raft-like domains (10). Different types of oligomers have been identified; some are thought to be obligate intermediates in fibril formation, whereas others are off pathway, forming independently and even inhibiting fibrillation (11–13). For example, an off-pathway oligomer can readily be formed by shaking α SN solutions at 37 °C. This oligomer consists of 30 protomers that form a compact β -sheet core surrounded by a diffuse corona (14, 15).

Given the demonstrated toxicity of oligomeric α SN, there is great interest in compounds that prevent their formation, or, once formed, reduce their toxicity. One such compound is (–)-epigallocatechin gallate (EGCG). Although EGCG typically shows low affinity for folded proteins (16), it binds numerous flexible proteins with no obvious sequence discrimination (17). Thus, EGCG binds monomeric α SN, affecting the entire amino acid sequence (16, 18), and also inhibits the fibrillation of several other proteins (19–24). EGCG reduces the toxicity of α SN aggregates by remodeling amyloids into smaller nontoxic oligomers (22), redirecting the aggregation of monomeric α SN into nontoxic oligomers (8, 16, 18) and inhibiting the toxicity of preformed toxic α SN and amyloid- β oligomers *in vitro* (25, 26). Both solid-state (8) and liquid-state (25) NMR spectroscopy have demonstrated that EGCG-induced oligomers have altered structures compared with “naked” oligomers. Whereas naked oligomers insert into and disrupt the membrane (possibly in association with interaction between protein and lipids and conformational changes to the oligomer (27, 28)), EGCG-induced structural changes lead to reduced affinity for and an inability to perturb membranes (8). To complicate matters, α SN can form multiple different species of coexisting oligomers (13, 29). This makes it difficult to study the effect of EGCG on the interaction between oligomeric and monomeric α SN. As a result, we still lack a detailed structural and kinetic characterization of the interaction between EGCG and α SN in the process of oligomer formation. Obtained under *in vitro* conditions to obtain maximal structural insight, such advances provide a foundation to better understand how these oligomers subsequently may interact with membranes and other components in the cell.

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Here, we use liquid-state NMR spectroscopy (30, 31) to show that EGCG binds to monomeric and oligomeric α SN. We find that monomeric and oligomeric α SN bind 54 and 7 EGCG molecules, respectively, and that EGCG slowly (over a 4-day period) binds to and immobilizes all protein residues. We suggest that these binding properties may explain the broad spectrum of activity by EGCG toward cytotoxic protein oligomers, and that related polyphenols act to bind together disordered proteins in a comparable way.

Results

EGCG redirects α SN monomer into oligomers in a dose-dependent manner

Determination of the stoichiometry of EGCG binding to α SN monomer by conventional titration is complicated by the fact that α SN oligomerizes in a time-dependent manner when EGCG is present (8, 16). Therefore, separate samples of α SN monomer were incubated with different amounts of EGCG, making sure to have identical incubation time (10 min) before recording 1D ^1H -NMR spectra. Changes in the NMR spectra of both EGCG and α SN are expected upon interaction (for EGCG NMR assignments, see Fig. S1). These changes should manifest themselves as line broadening or chemical shift

perturbation, with the exact outcome depending on binding kinetics (30). We hypothesized that our observations could be captured by a minimal model that considers (1) free and bound states for the protein as well as for EGCG and (2) that broadening of the NMR signals is caused by the modulation of the chemical (and magnetic) environment of the nuclei in the limiting states (free and bound), although more complex models can be envisaged (*e.g.*, intermediate exchange line broadening intrinsic to the bound form of EGCG and influence of EGCG on the self-association state of α SN). In addition, once oligomers of high molecular weight have formed, even a small population of the bound form in a weak-affinity interaction can bring about severe line broadening (32).

Figure 1A displays the aliphatic region of 1D ^1H -NMR spectra at different EGCG: α SN ratios. The region (2.5–0.75 ppm) shows signals from aliphatic side chains (mostly methyl groups) of α SN (EGCG does not show any resonances here, see Fig. S1) and is used to gauge the fate of the protein in solution. In contrast, the aromatic region (7.5–5.5 ppm) is dominated by EGCG (in red boxes in Fig. 1B), although there are weak signals from α SN's four Tyr and two Phe residues around 7.4 to 6.8 ppm (see lowest trace in Fig. 1B). Strikingly, using 10-min incubation, all protein signals fully disappeared when the [EGCG]:[α SN] ratio was 60:1

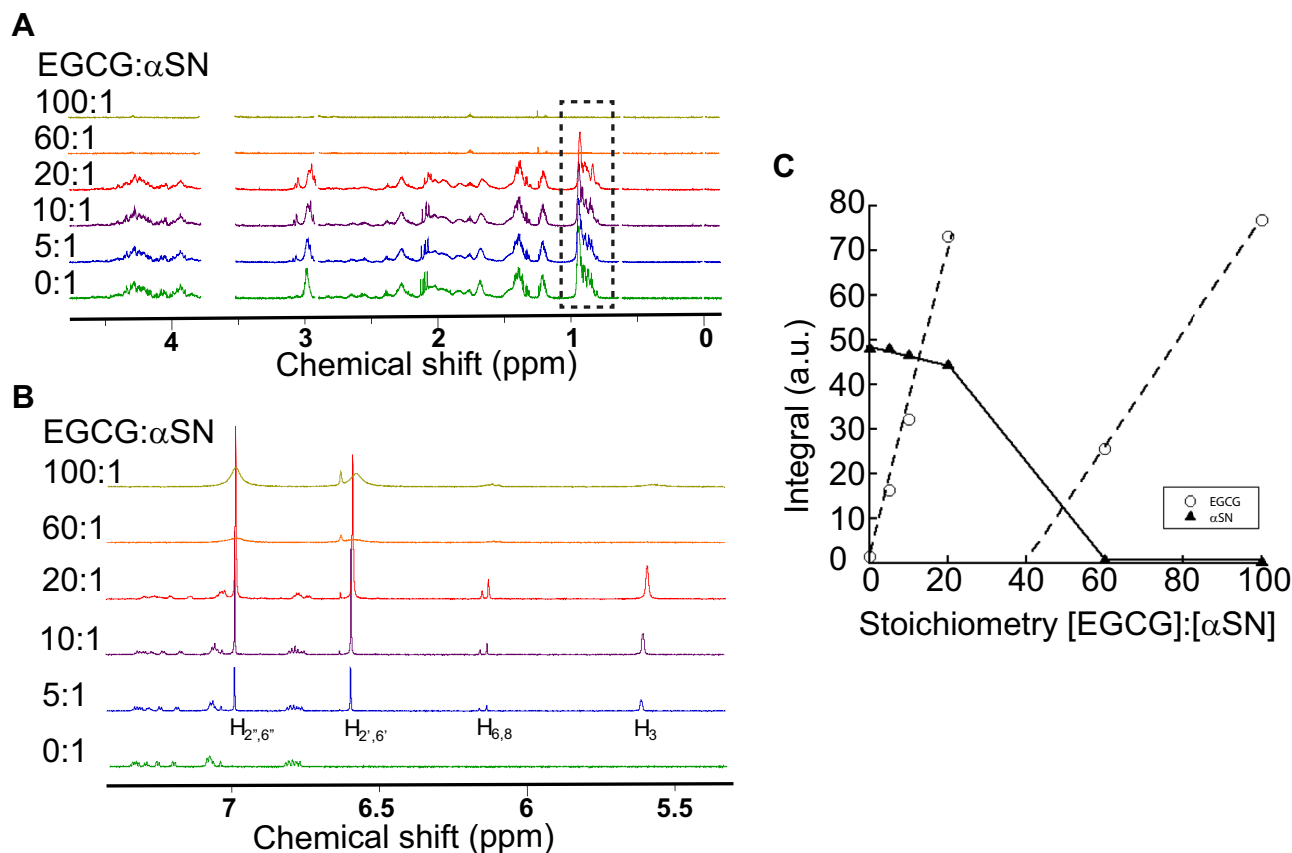


Figure 1. EGCG induces α SN oligomerization. A and B, 1D ^1H -NMR of α SN and EGCG at different stoichiometries. A, the methyl region from α SN (dotted box). Signals because of a glycerol impurity from protein concentration filter (3.6 ppm) and DSS (0.0, 0.625, and 2.9 ppm—but not the one at 1.75 ppm) have been removed to simplify the spectra (the unedited spectra can be seen in Fig. S2). B, the aromatic region (dominated by EGCG— $\text{H}_{2',6'}$, $\text{H}_{2,6'}$, $\text{H}_{6,8}$, and H_3 are assigned). C, integration of peak from EGCG $\text{H}_{2,6'}$ (at 6.6 ppm, see B) and integration of α SN methyl peaks (1.05–0.75 ppm). The dashed line and solid line in the figure are intended to guide the eye. α SN, α -synuclein; DSS, 2,2-dimethyl-2-silapentane-5-sulphonic acid; EGCG, (–)-epigallocatechin gallate.

(Fig. 1, A–C), whereas only a very modest decrease was observed for α SN signals when titrating in the first 20 equivalents of EGCG. At the same time, EGCG NMR signals initially increased linearly with concentration up to 20 EGCG per α SN, in proportion to the amount of EGCG added, but upon an increase from 20:1 to 60:1, also all the EGCG signals vanished (Fig. 1, A–C). Since we did not see any precipitation, this implies that at ligand-to-protein ratios in the range 20 to 60, all EGCGs become sequestered in large, soluble, or dispersed assemblies that simultaneously cause both α SN and EGCG signals to become NMR invisible. This is the kind of behavior expected for molecules in a slow tumbling complex. Negative-stain EM demonstrated that loss of the NMR spectrum coincided with the formation of oligomeric species of \sim 20 nm diameter (16).

Upon further addition of EGCG, α SN continues to remain NMR invisible, whereas a re-emergence of visible EGCG NMR signals is seen (Fig. 1C). The EGCG signals scale with the amount added, indicating accumulation of unbound EGCG. Consistent with this explanation, the line width of the EGCG peak for $H_{2,6}$ is very little affected up to $[EGCG]:[\alpha SN] = 20:1$. This indicates that, over the course of a 10-min incubation, most EGCG molecules exist in solution, and are only little affected by the presence of α SN, until a critical ratio is reached beyond 20 EGCGs per α SN. Once the NMR-invisible species have been fully formed, the signals of free ligand once more increase in intensity with further EGCG addition, but the peaks are very wide (Fig. 1, B and D). Broad peaks observed for small molecules typically point to the existence of a dynamic equilibrium between a rapidly tumbling free state and a slowly tumbling bound state (30). As 1H -NMR spectra of free EGCG consistently show sharp signals over the concentration range (0.15–3 mM) used in the α SN titration study (Fig. S3), we can rule out that broad signals result from self-association of EGCG (33, 34). This is in keeping with a reported EGCG self-association constant of 0.014 mM^{-1} (*i.e.*, 50% associated at 7 mM) at pH 6.0 (35). We conclude that a stoichiometrically well-defined EGCG: α SN complex is formed in the presence of a critical number of EGCG molecules.

Time-dependent oligomerization of α SN in the presence of EGCG

To obtain mechanistic insight into the self-assembly process, we next investigated the time-dependent structural changes that take place during coincubation. To address this, EGCG and α SN were mixed at EGCG: α SN molar ratios ranging from 8.7 to 51.9, and a series of NMR spectra were recorded over time. A marked decline was observed for all 1D 1H -NMR signals (Fig. 2A), whereas no such decline in signals was seen in a control experiment (*i.e.*, in the absence of EGCG; data not shown). Using the integrated intensity of α SN protein NMR signals (2.5–0.6 ppm), the signal loss could be fitted using an exponential decay function (assuming that all signal is eventually lost), giving a rate constant k , which increased in a power-law fashion with EGCG: α SN stoichiometry up to 34.6 (Fig. 2B) (at higher stoichiometries, the signal change was too small to provide a reliable estimate of the rate constant). The

order of a chemical reaction with regard to a given reactant can formally be determined by plotting the logarithm of the rate v (estimated from the initial slope in Fig. 2A) versus the logarithm of the concentration of that reactant. Whether using k or v , we obtain a slope of \sim 2, which indicates that two EGCG molecules associate per reaction step, suggesting an EGCG dimerization associated with the mechanism of interaction with α SN. We note that at neutral pH, EGCG dimerizes (on the minute–hour scale) because of oxidation (36, 37), and dimeric EGCG has been shown to possess a more potent disaggregating effect than monomeric EGCG (38). Dimerization is possible, but not conclusively demonstrated by these data, which provide a lower limit for cooperativity of the EGCG– α SN interaction.

Interestingly, this slow signal decay was preceded by a very rapid loss of signal within the dead time of measurement (6–18 min) corresponding to a burst phase. The magnitude of this burst phase increased linearly with EGCG stoichiometry from 25.9 to 51.9 EGCG, and this linear phase extrapolated to the upper limit of 1.0 (*i.e.*, complete disappearance of the α SN methyl signal) around 54 $[EGCG]:[\alpha SN]$ (Fig. 2C). Assuming that rapid binding of EGCG to α SN within the dead time is responsible for this signal disappearance, this suggests that α SN has a capacity to bind up to 54 EGCGs per α SN protomer in the formation of EGCG-induced oligomer. The appearance of EGCG-induced aggregates of α SN was monitored by transmission electron microscopy (TEM) (Fig. 2D), which revealed small round oligomers that resembled those observed in other studies (25, 39), either isolated or connected into chains. We have previously observed small amounts of oligomeric concatamers in the absence of EGCG, but not to the same extent as with EGCG, indicating that EGCG promotes formation of these high molecular weight assemblies.

To investigate the oligomerization process with higher structural resolution, time-dependent signal loss was monitored by 2D ^{15}N – 1H heteronuclear single quantum coherence (HSQC) NMR spectroscopy for a 20:1 ligand-to-protein ratio. We observe severe signal loss for all protein residues already after 20 min (Fig. 3A). Based on the observations made at higher EGCG ratio, this observation is consistent with a scenario in which most α SN is sequestered into NMR-invisible oligomeric species, removing all free EGCGs from solution in a cooperative process. As there is insufficient EGCG to aggregate all protein within the first rapid phase ($<10\%$ of the signal disappears in the burst phase at 20:1 EGCG: α SN, *cf.* Fig. 2C; this fraction is likely higher in the HSQC experiment where we have a fivefold increase in α SN and EGCG concentration), the remaining α SN will be subject to slower self-association processes. Plotting signal intensity versus residue number reveals a second and slower loss of the NMR-visible pool that is not evenly distributed over the sequence (Fig. 3B). Resonances belonging to the N-terminal portion of the protein (residues 1–70) are completely lost after 2 h, whereas signals from the C-terminal domain (residues 95–140) still remain detectable after multiple days. Previous 2D ^{15}N – 1H HSQC NMR studies have shown that the C-terminal \sim 40 residues remain highly mobile (and thus visible) in

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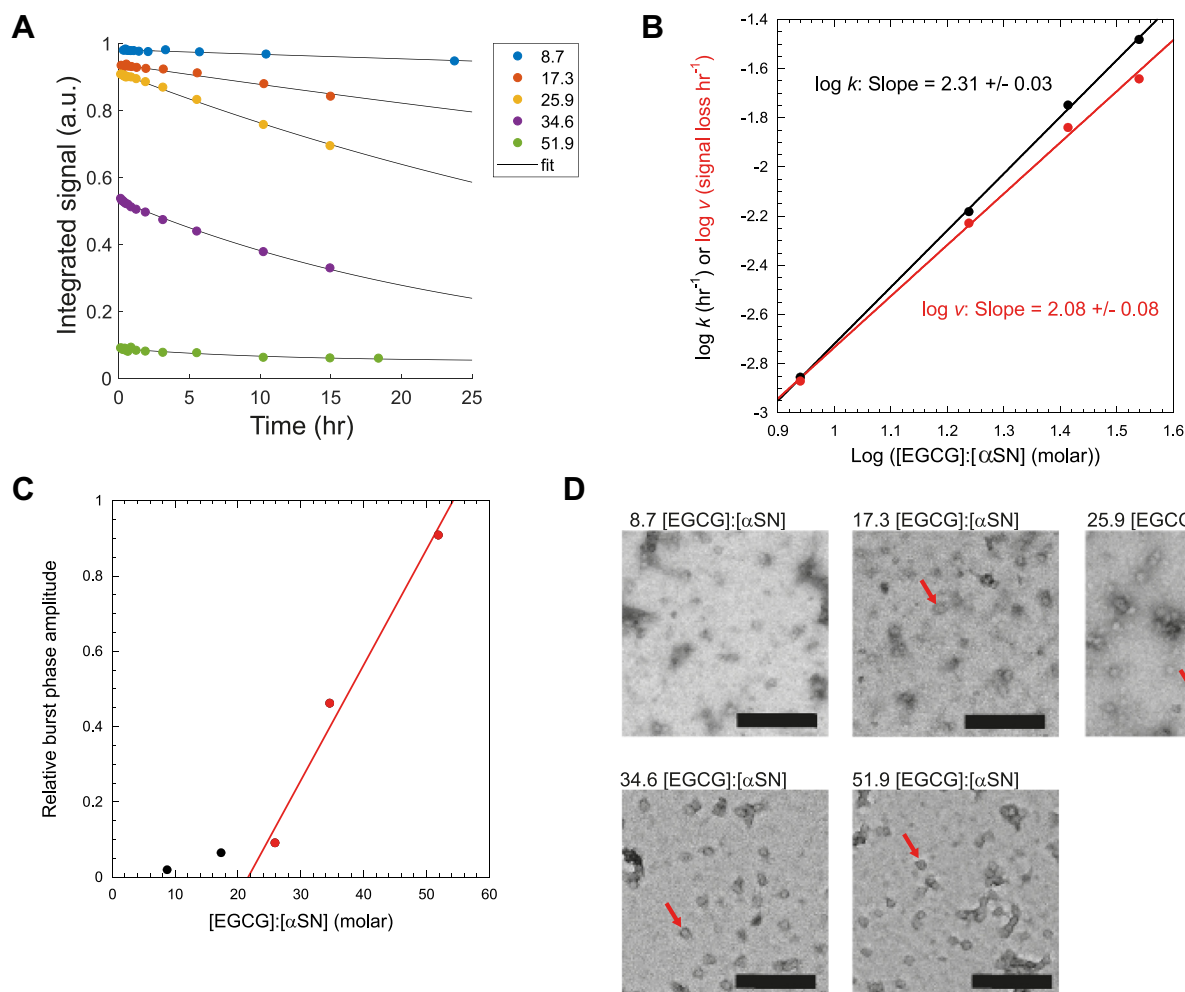


Figure 2. Oligomerization of α SN in the presence of EGCG over time. A, EGCG-induced oligomerization of α SN, followed by integration of α SN peaks (2.5–0.6 ppm) in 1D ^1H -NMR spectroscopy. Molar ratio [EGCG]:[α SN] and color code are provided alongside. B, growth rate from fit seen in A as a function of [EGCG]:[α SN] in a log–log plot. C, burst phase amplitude as a function of [EGCG]:[α SN] stoichiometry. Red line is a linear fit to the three data points at highest stoichiometry. The intercept at $y = 1.0$ occurs at a stoichiometry of ~ 54 . D, TEM images of α SN with EGCG taken after the NMR experiments. Red arrows are pointing at oligomers, and the black scale bar indicates 200 nm. α SN, α -synuclein; EGCG, (–)epigallocatechin gallate.

α SN oligomers that form in the absence of EGCG (25). In combination, these studies suggest the rapid (minutes) formation of EGCG-aggregated oligomers, followed by a slower (hours) phase producing oligomers in which only the C terminus remains flexible, and finally very slowly (days) forming oligomers with rigidified C termini. Although the kinetics differ from those seen in Figure 2 as the HSQCs were recorded at higher concentration, structural insight into oligomer formation was obtained in these experiments.

To analyze whether the α SN assemblies that formed during 4 days of incubation with EGCG at 20:1 were able to bind additional EGCG, we added EGCG to 50-fold and then to 100-fold excess in two steps and monitored the signal in the aromatic region, characteristic for EGCG (Fig. 3C). Only an extremely weak signal was obtained at 50-fold excess EGCG, consistent with the estimated ability of α SNs to bind around 54 EGCGs per protomer (Fig. 2C). At 100-fold EGCG, intense but broad peaks were observed, similar to the signals seen at 60-fold excess where EGCG was added to the α SN monomer in a single step (Fig. 1B). Thus, a fixed amount of EGCG is

sequestered during oligomer formation, and excess-free EGCG is present in a dynamic binding equilibrium with the oligomeric assemblies. It is important to bear in mind that despite their broad appearance, the EGCG peaks belong to free ligand. The line broadening here is not because of slow tumbling but arises from kinetic exchange with large molecular weight species (33).

EGCG binding to preformed oligomer

Finally, we investigated the binding of EGCG to preformed α SN assemblies. To this end, α SN oligomers were preformed by incubating monomeric α SN at high concentrations (8–10 mg/ml) under orbital shaking at 37 °C for 3 h (*i.e.*, in the absence of EGCG) and separated by size-exclusion chromatography (25). This type of oligomer consists of ~ 30 protomer units (11, 14). Subsequently, EGCG was titrated into NMR vials containing 50 μM (protomer concentration) oligomers, and 1D ^1H -NMR spectra were recorded at each titration step. EGCG signals (here the integrated intensity of the EGCG $\text{H}_{2''},6'$ peak) only appeared at higher EGCG: α SN ratios (Fig. 4A), and

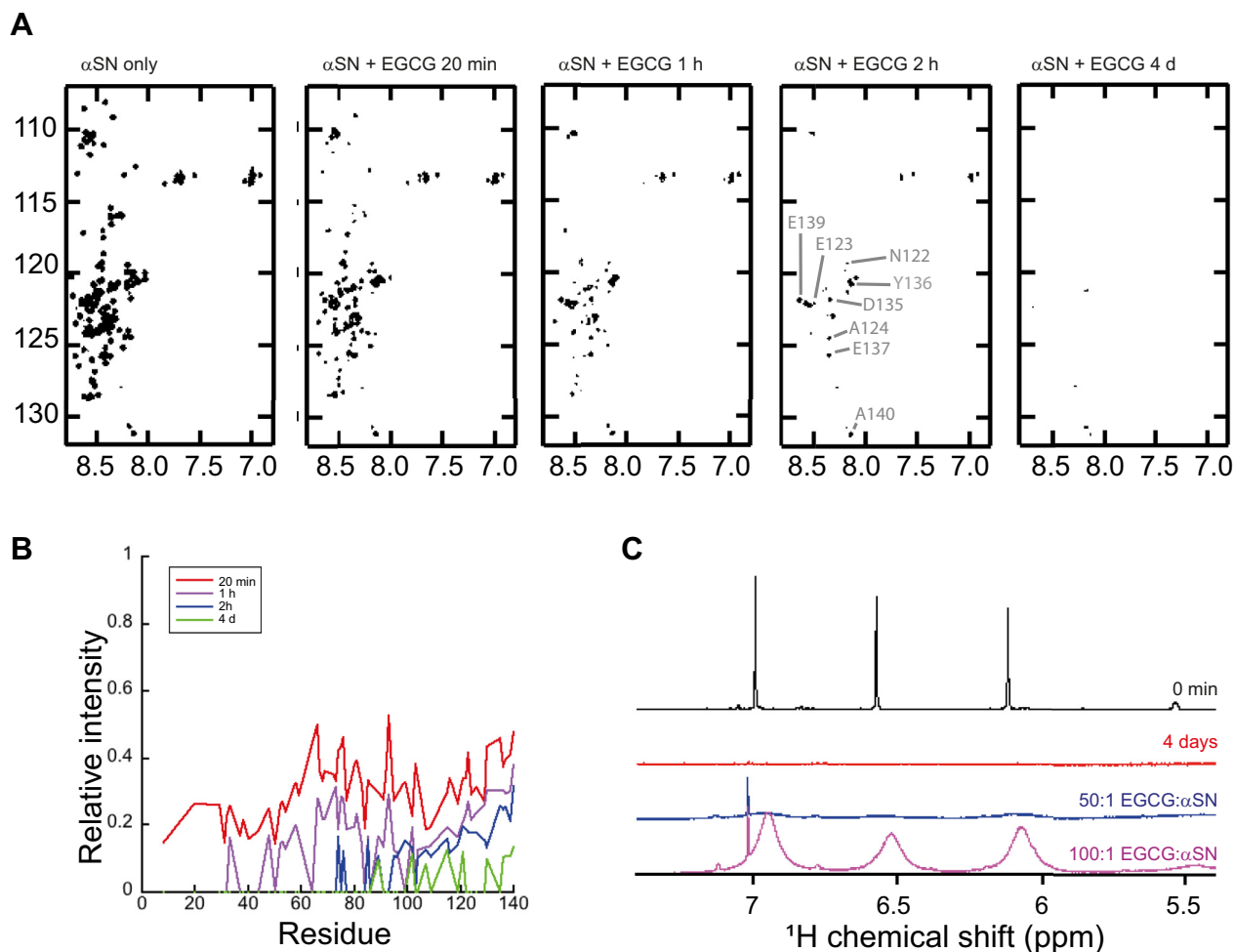


Figure 3. Oligomerization of α SN in the presence of EGCG involves mainly the first 80 residues. A, ^{15}N - ^1H HSQC of α SN upon incubation with EGCG ($[\alpha\text{SN}]:[\text{EGCG}] = 1:20$) over time. B, relative intensity of peaks in ^1H - ^{15}N HSQC spectra (A) of 0 min, 1 h, 2 h, and 4 days relative to α SN alone. C, 1D spectrum of 3 mM EGCG mixed with 150 μM α SN ($[\text{EGCG}]:[\alpha\text{SN}] = 20:1$) immediately after mixing (black) and after 4 days of incubation (red). 28 mM EGCG was stepwise added to give first an $[\text{EGCG}]:[\alpha\text{SN}]$ of 50:1 (blue) and then 100:1 (magenta). The sharp peak at 7.1 ppm stems from an impurity in the EGCG sample. The assigned ^{15}N - ^1H HSQC spectrum of α SN with high resolution is provided as Figure S4. α SN, α -synuclein; EGCG, (-)-epigallocatechin gallate; HSQC, heteronuclear single quantum coherence.

their integral increased linearly (Fig. 4B). By extrapolating the EGCG signal back to zero integral (free ligand concentration), we determine a high-affinity binding interaction where 7 ± 1 EGCG molecules are immobilized per α SN protomer in the oligomeric assembly. EGCG signals at molar ratios below 20 were very broad and could not be accurately integrated. These are therefore not included in the extrapolation. A plot of line width *versus* stoichiometry (Fig. 4B) shows that high-affinity binding is followed by dynamic ligand exchange as described earlier for the monomeric form, that is, broad peaks at low ligand ratios and a sharpening at higher ratios (*cf.*, Figs. 1B, 3C, and 4C). This reflects that addition of more ligand increases the fraction of free ligand. In addition, higher EGCG concentrations lead to an increased on-rate, which is reflected in faster kinetics. As a result, peaks sharpen when EGCG concentration is increased (Fig. 4C). A small gradual downfield change in chemical shift is seen for the EGCG peaks at higher concentrations, which we attribute to weak self-association, possibly leading to covalent dimerization as mentioned previously. This shift is also observed in binding experiments with

α SN monomers (Fig. 1B). Consistent with this, EGCG shows the same behavior on its own at higher concentrations (Fig. S3).

To confirm the dynamic interaction between free EGCG in solution with the oligomer directly, we turned to saturation transfer difference (STD) experiments. In these experiments, the oligomer magnetization is progressively saturated by a weak RF field applied to the protein methyl transitions. The slow tumbling of the oligomeric assembly makes the transfer of saturation through ^1H - ^1H dipolar couplings (known as cross-relaxation) very effective, such that all protein signals become quenched. When a small molecule transiently binds during the saturation time, cross-relaxation will lead to partial attenuation of the ligand signal, depending on the lifetime of complex formation and the distance between ligand and protein hydrogens. A difference spectrum directly displays all ligand signals that experience saturation transfer from binding. As the amount of transfer depends on the “contact time” of ligand to protein, a series of increasing saturation times shows progressive transfer of magnetization from protein to ligand.

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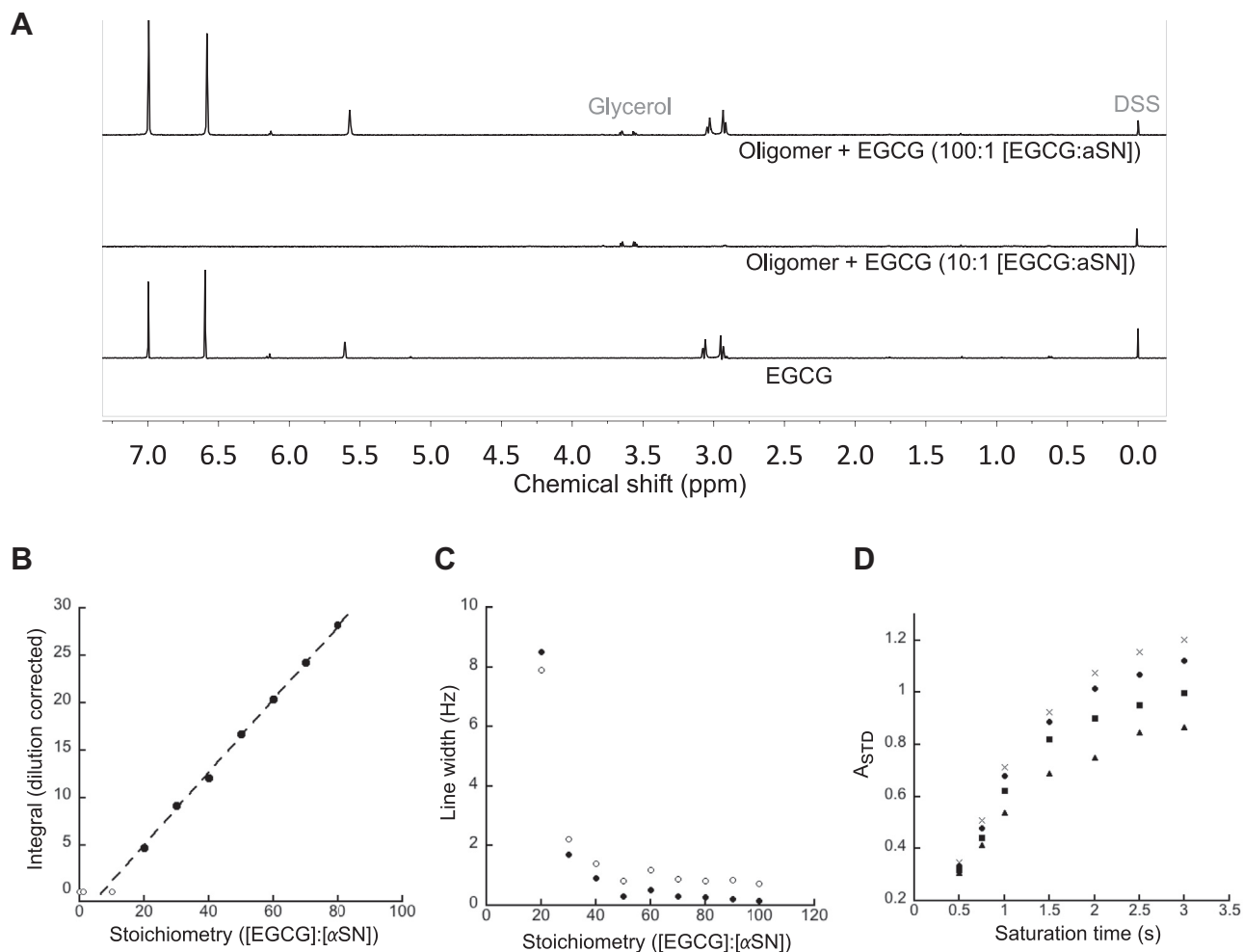


Figure 4. Binding of EGCG to preformed oligomer. *A*, titration of EGCG to α SN oligomer. At 1:100 = [α SN]:[EGCG], all EGCG peaks are broadened beyond observation. The peaks around 3.6 ppm are due to a small glycerol contamination in the oligomer sample. At excess EGCG (1:100 [α SN]:[EGCG]), peaks from EGCG reappear. *B*, integration of $H_{2'',6''}$ peak at different stoichiometries gives a binding stoichiometry of 7 ± 1 ([EGCG]: α SN) from the intercept at the x-axis. *C*, line width of $2'',6''$ (filled circles) and $2',6'$ peaks (open circles) of EGCG at the different stoichiometries. *D*, STD amplification factor. STD intensity as a function of saturation time at 1:100 ([α SN]:[EGCG]) (saturation field applied at -1 ppm) for $2'',6''$ (crosses), $2',6'$ (circles), 3 (squares), and $4_{Eq,4_{Ax}}$ (triangles). α SN, α -synuclein; EGCG, (-)-epigallocatechin gallate; STD, saturation transfer difference.

We applied the STD experiment to the sample containing the largest excess of EGCG to oligomer ratio (100:1), where rapid binding kinetics and low-affinity binding prevail (*i.e.*, bound state lifetime < 1 s; $K_d \sim \mu\text{M}$ – mM (40, 41)). Saturation curves for several hydrogen atoms of EGCG (Fig. 4D) directly demonstrate weak oligomer binding.

Discussion

EGCG drives α SN into oligomers of defined stoichiometry in a concerted manner

The small flavonoid polyphenol EGCG is able to promote the formation of nontoxic and unstructured α SN oligomers from both monomer α SN (16) and amyloid fibrils (22). Here, we have monitored EGCG-induced α SN oligomer formation by means of NMR signal loss of monomer over a fixed period. When monomeric α SN is incubated with EGCG at low stoichiometry ([EGCG]: α SN) = 8.7, no change in the relative NMR signal intensities of α SN is observed over 10 h, indicating that all added α SN is present in solution. However, increasing

the stoichiometry to 51.9:1 leads to the disappearance of all protein NMR signals together with those for EGCG. This suggests that, at short times and at a critical concentration, EGCG induces the rapid and comprehensive sequestration of α SN in a concerted manner. Incubation with more than 50-fold excess leads to the renewed observation of ligand signals, demonstrating that the oligomers comprise a well-defined amount of EGCG. Extrapolation leads to an [EGCG]: α SN stoichiometry of 54:1 for the saturated complex. Assuming that EGCG binds uniformly along the entire 140-residue polypeptide chain (our previous studies on the effect of EGCG on α SN peak intensities (18) indicated that all residues were affected, although two broad peak areas were identified in the N- and C-terminal regions, respectively), α SN sequesters one molecule of EGCG per 2.7 amino acids of disordered protein. Although the oligomeric state of α SN induced by EGCG is inaccessible to solution-state NMR spectroscopy, solid-state NMR and complimentary biophysical techniques indicate a disordered structure (8). We also show

that, over time, EGCG-induced oligomers lose their flexible C terminus when incubated with sufficient (20:1) EGCG. Previous studies have shown EGCG-induced dose-dependent NMR line broadening for α SN at 5:1 and 10:1 [EGCG]:[α SN], *that is*, at slightly lower stoichiometries than we explore here (8). Although the kinetics cannot be directly compared because of differences in concentration and temperature to the present study, the observation that oligomers formed at 10:1 [EGCG]:[α SN] are heterogeneous in size and can still bind antibodies that recognize the flexible C domain, most likely result from the coexistence of a mixture of EGCG-saturated and immobilized oligomers and monomeric species. The data in Figures 2 and 3 are also in line with the solid-state NMR data of Fusco *et al.* (8), where the immobilized regions of protomers contribute to dipolar-assisted rotational resonance spectra, whereas those that present a flexible C terminus contribute to both dipolar-assisted rotational resonance and insensitive nuclei enhancement by polarization transfer spectra for the rigid and flexible parts of the sequence, respectively. Unfortunately, these spectra are insufficiently quantitative to extract relative populations. Furthermore, solid-state NMR studies of oligomers formed in the absence and presence of EGCG reveal structural differences in the core of the oligomer, but both types of oligomers contain a flexible C terminus (8). The fact that the N-terminal (residues 1-60) and non-amyloid β (NAC; residues 61-95) regions remain immobilized in the EGCG-induced oligomer along with the fact that oligomers formed in the presence of EGCG are nontoxic suggest that the NAC region and N terminus are important determinants for the toxicity of α SN oligomers (5) and that the binding of EGCG to the N terminus is able to abrogate this toxicity.

EGCG-dependent oligomerization of α SN is graded and proceeds from N terminus to C terminus

In the presence of 20 equivalents of EGCG, α SN first undergoes a phase of oligomerization within few hours that is accompanied by a loss of intensity for all residues. This is followed by a second slower association phase that takes place over several days. We interpret the first phase to result from a cooperative EGCG-dependent formation of oligomers. At the same time, loss of signal at the N terminus is stronger, suggesting that parallel oligomer formation pathway exists. The species that are tethered at the N-terminal region approximately 50 residues slowly become NMR invisible over the following days. NMR signals of the C terminus eventually disappear over the course of several days, and this process is distinct from that observed for oligomers formed in the absence of EGCG, which are characterized by the persistence of disorder in the C-terminal region \sim 40 amino acids over days to weeks (12). A possible explanation for these observations is that EGCG-induced oligomers can self-associate to longer chains, as demonstrated by TEM.

The graded response, with signal loss in first N-terminal then NAC region and at last C-terminal region, is highly reminiscent of that seen for α SN interaction with 1-palmitoyl-2-oleoyl-sn-glycero-3-phospho-(1'-rac-glycerol) in

small unilamellar vesicles and nanodiscs (42, 43). However, α SN binding to negatively charged membranes forms alpha-helix structure, whereas oligomer formation proceeds without secondary structure formation.

The slope of two obtained from the power law relationship between the rate (and rate constant) of signal loss and EGCG stoichiometry (Fig. 2B) indicates that two EGCG molecules come together to react with α SN per encounter. Clearly, this does not limit the final stoichiometry of binding, which is more than an order of magnitude higher (54 EGCG: α SN). Rather, it implies that several EGCG molecules interact cooperatively when binding to α SN, perhaps by forming protein-stabilized dimers that increase affinity and thus the stability of the bound EGCG molecules. Other dimeric forms of EGCG bind to A β 40 and bovine insulin and inhibit fibrillation more effectively than EGCG, most likely because of their ability to make a larger number of hydrophobic interaction and hydrogen bonds with the polypeptide chain (44, 45).

EGCG binds to oligomers that are formed in the absence of EGCG

We obtained a binding stoichiometry of seven EGCGs per protomer of preformed oligomers based on a linear extrapolation of the free ligand NMR signal integral. This number is much smaller than the \sim 54 EGCG molecules that associate with monomeric α SN during oligomer formation. The disparity indicates that multiple EGCG molecules are sequestered within the oligomer during its formation but cannot access this region once the oligomer is formed. Although the exact nature and sequence specificity of binding remain unclear (8), studies with other amyloidogenic proteins implicate aromatic sidechains (46, 47), binding of EGCG to α SN oligomers reduces the flexibility of the C terminus (which contains three Tyr and one Phe) (25), and EGCG has different binding modes dependent on the conformational state of the protein (17). As STD-NMR signals are typically obtained with macromolecule-ligand interactions with millimolar to micromolar binding constants (41), EGCG is expected to bind with weak affinity, in line with the fast-exchange behavior that is observed in titration experiments with monomeric α SN. Line broadening observed for excess EGCG in dynamic binding equilibrium is the hallmark of a ligand binding to large aggregates and also shows the expected narrowing when the excess ligand pool is increased, further corroborating the fluxional character of the interaction.

EGCG reduces cellular toxicity well below oligomer conversion stoichiometry

Since α SN toxicity may be mitigated by the formation of nontoxic oligomers, or by inhibition of membrane binding, EGCG may be an interesting lead for therapeutics aimed against Parkinson's disease. For example, EGCG has been shown to inhibit extracellular toxicity toward OLN93 cells by 50% at [EGCG]:[α SN] = 0.36:1, and *in vitro* calcein release was likewise inhibited at [EGCG]:[α SN] = 0.23:1 (25). The present study suggests that, under these conditions, EGCG acts at

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levels several orders of magnitude below those required for oligomer conversion. Possibly, the cooperative nature of EGCG-induced conversion, along with the cooperation between EGCG molecules in the binding reaction demonstrated here, may effect the accumulation of populations of oligomers sufficiently to generate the required response.

Experimental procedures

Preparation of recombinant α SN

Recombinant human α SN was produced in *Escherichia coli* using a pET11-D α SN construct induced *via* autoinduction as described (11, 48). The same procedure was followed to produce uniformly ^{15}N -labeled α SN, except that *E. coli* was grown in minimal medium using $^{15}\text{NH}_4\text{Cl}$ as the only nitrogen source.

EGCG solution

EGCG (Sigma; 458.4 Da) was weighed out and dissolved in PBS (heavy water) in an Eppendorf tube to a final concentration of 10 mM. The tube was wrapped in an aluminum foil to protect the sample from light and stored at $-20\text{ }^\circ\text{C}$ between experiments. The solution was used within 1 week and was always transparent, showing no sign of oxidation.

Oligomer preparation and purification

α SN oligomers were purified as described (9, 15) with the modification that the sample was incubated for 3 h under fibrillation conditions prior to purification. α SN was dissolved to 8 to 10 mg/ml and incubated at $37\text{ }^\circ\text{C}$, 900 rpm for 3 h. Insoluble material was removed by centrifugation at $4\text{ }^\circ\text{C}$ at 13,000 rpm for 10 min, and the supernatant was loaded on a Superose 6 prep grade XK 26/100 column (GE Healthcare) equilibrated in PBS buffer. Oligomer fractions were collected and stored at $4\text{ }^\circ\text{C}$ and concentrated using a 100 kDa spin filter that was washed with PBS to remove glycerol.

NMR spectroscopy

NMR experiments were performed on deuterated samples at 278 K (α SN incubation with EGCG followed over time with HSQC; Fig. 3) or 283 K on Bruker Avance NMR spectrometers at ^1H frequency of 950 MHz or 500 MHz. The spectra were processed using NMRPipe (49) and analyzed with Sparky (T. D. Goddard and D. G. Kneller; SPARKY 3, University of California, San Francisco), unless otherwise stated. Sodium salt of 2,2-dimethyl-2-silapentane-5-sulphonic acid (DSS) was included in all samples for chemical shift referencing (50).

Titration experiments

For titration of EGCG to monomer, separate samples were prepared. α SN was added to each sample right before the NMR measurement. The dead time from sample preparation to NMR experiment was 10 min. The samples contained 30 μM α SN monomer, 50 μM DSS, and 0, 0.15, 0.3, 0.6, 1.8, or 3 mM EGCG, in deuterated PBS. Reference spectra of EGCG were recorded for 0, 0.15, 0.3, 0.6, 1.0, 1.2, 2.5, 1.8, and 3 mM

EGCG. For titration of EGCG to oligomer, an oligomer sample of 30 μM α SN oligomer 50 μM DSS in deuterated PBS is used. Freshly prepared 10 mM EGCG stock solution was added to the samples and mixed giving 1, 10, 20, 30, 40, 50, 60, 70, 80, 90, and 100 [EGCG]:[α SN]. 1D ^1H -NMR employed water suppression using excitation sculpting with gradients and 32 scans for each experiment.

Time-dependent EGCG-induced α SN oligomerization followed by 1D ^1H -NMR

Samples contained 34.7 μM α SN, 50 μM DSS, and 300, 600, 900, 1200, and 1800 μM EGCG (*i.e.*, EGCG: α SN molar ratios of 8.7, 17.3, 25.9, 34.6, and 51.9) in PBS. ^1H -NMR employed water suppression using excitation sculpting with gradients with 128 scans for each experiment and variable number of dummy scans to change time between NMR experiments. The integrated methyl signal from α SN is normalized to the glycerol signal (contaminant in α SN sample) as an internal control. Exponential fit was prepared with MATLAB and varying only amplitude and velocity constant by setting the baseline to 0.05.

Time-dependent EGCG-induced α SN oligomerization followed by 2D ^{15}N - ^1H HSQC

Samples contained 150 μM ^{15}N - α SN, 80 μM DSS, and 3 mM EGCG in PBS. The matrix size of the HSQC was 128 (t_1) and 1024 (t_2) complex points, with a spectral width of 32 and 16 ppm for the ^{15}N and ^1H dimensions, respectively. A gradient and sensitivity-enhanced version of the HSQC experiment was used that is optimized to avoid mixed-phase artifacts (51).

1D STD NMR

About 35.7 μM of unlabeled oligomer incubated with 3.57 mM EGCG (1:100) was monitored by acquiring 1D STD NMR spectra (52). STD spectra were recorded with 64 scans, 32 K complex points, and a spectral width of 16.04 ppm. Selective saturation of α SN oligomers was achieved through methyl proton irradiation at -1 ppm using a Gaussian-shaped pulse train of 50 ms. A 30 ms spin lock (T_2 filter) was employed to suppress protein signals. Off-resonance (reference) experiment was recorded with saturation at 42 ppm. Spectra with 0.5, 0.75, 1, 1.5, 2, 2.5, and 3 s saturation times were recorded. For each saturation time, four spectra were recorded, alternating on and off resonance. The STD amplification factor A_{STD} was calculated for each saturation time. A_{STD} was calculated as $A_{\text{STD}} = \frac{I_0 - I_{\text{sat}}}{I_0} * \frac{[L]}{[P]}$, where I_0 is the intensity in the reference spectrum and I_{sat} is the intensity in the spectrum with irradiation at -1 ppm , [L] is the total EGCG concentration, and [P] is the α SN concentration. Signals of protons (2',6'), (2'',6''), 3, and 4 (axial and equatorial) were followed. See Fig. S1 for assignments.

TEM

Samples were taken from NMR samples after time-dependent experiment at the indicated stoichiometries and transferred to a 400-mesh carbon-coated copper grid (EM

resolution) that was glow discharged for 30 s. The oligomer sample was transferred to the grids and washed on two drops of doubly distilled water, stained with 1% uranyl formate, and blotted dry on filter paper. The samples were viewed at 120 kV. EM was done on a Tecnai G2 Spirit (FEI Company), and images were taken using a TemCam F416 camera (TVIPS).

Data availability

Data can be shared upon request to the corresponding author.

Supporting information—This article contains [supporting information](#).

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Abbreviations—The abbreviations used are: α SN, α -synuclein; DSS, 2,2-dimethyl-2-silapentane-5-sulphonic acid; EGCG, (-)-epigallocatechin gallate; HSQC, heteronuclear single quantum coherence; NAC, non-amyloid β ; STD, saturation transfer difference; TEM, transmission electron microscopy.

References

- Spillantini, M. G., Schmidt, M. L., Lee, V. M. Y., Trojanowski, J. Q., Jakes, R., and Goedert, M. (1997) α -Synuclein in Lewy bodies. *Nature* **388**, 839
- Spillantini, M. G., Crowther, R. A., Jakes, R., Hasegawa, M., and Goedert, M. (1998) α -Synuclein in filamentous inclusions of Lewy bodies from Parkinson's disease and dementia with Lewy bodies. *Proc. Natl. Acad. Sci. U. S. A.* **95**, 6469–6473
- Uversky, V. N., Li, J., and Fink, A. L. (2001) Evidence for a partially folded intermediate in α -synuclein fibril formation. *J. Biol. Chem.* **276**, 10737–10744
- Winner, B., Jappelli, R., Maji, S. K., Desplats, P. A., Boyer, L., Aigner, S., Hetzer, C., Lohr, T., Vilar, M., Campioni, S., Tzitzilonis, C., Soragni, A., Jessberger, S., Mira, H., Consiglio, A., et al. (2011) *In vivo* demonstration that α -synuclein oligomers are toxic. *Proc. Natl. Acad. Sci. U. S. A.* **108**, 4194–4199
- Lorenzen, N., Lemminger, L., Pedersen, J. N., Nielsen, S. B., and Otzen, D. E. (2014) The N-terminus of α -synuclein is essential for both monomeric and oligomeric interactions with membranes. *FEBS Lett.* **588**, 497–502
- Danzer, K. M., Haasen, D., Karow, A. R., Moussaud, S., Habeck, M., Giese, A., Kretschmar, H., Hengerer, B., and Kostka, M. (2007) Different species of α -synuclein oligomers induce calcium influx and seeding. *J. Neurosci.* **27**, 9220–9232
- Lashuel, H. A., Hartley, D., Petre, B. M., Walz, T., and Lansbury, P. T. (2002) Neurodegenerative disease - amyloid pores from pathogenic mutations. *Nature* **418**, 291
- Fusco, G., Chen, S. W., Williamson, P. T. F., Cascella, R., Perni, M., Jarvis, J. A., Cecchi, C., Vendruscolo, M., Chiti, F., Cremades, N., Ying, L. M., Dobson, C. M., and De Simone, A. (2017) Structural basis of membrane disruption and cellular toxicity by α -synuclein oligomers. *Science* **358**, 1440–1443
- Volles, M. J., Lee, S. J., Rochet, J. C., Shtilerman, M. D., Ding, T. T., Kessler, J. C., and Lansbury, P. T. (2001) Vesicle permeabilization by protofibrillar α -synuclein: Implications for the pathogenesis and treatment of Parkinson's disease. *Biochemistry* **40**, 7812–7819
- Perissinotto, F., Stani, C., De Cecco, E., Vaccari, L., Rondelli, V., Posocco, P., Parisse, P., Scaini, D., Legname, G., and Casalis, L. (2020) Iron-mediated interaction of α -synuclein with lipid raft model membranes. *Nanoscale* **12**, 7631–7640
- Lorenzen, N., Nielsen, S. B., Buell, A. K., Kaspersen, J. D., Arosio, P., Vad, B. S., Paslawski, W., Christiansen, G., Valnickova-Hansen, Z., Andreasen, M., Enghild, J. J., Pedersen, J. S., Dobson, C. M., Knowles, T. P. J., and Otzen, D. E. (2014) The role of stable α -synuclein oligomers in the molecular events underlying amyloid formation. *J. Am. Chem. Soc.* **136**, 3859–3868
- Paslawski, W., Andreasen, M., Nielsen, S. B., Lorenzen, N., Thomsen, K., Kaspersen, J. D., Pedersen, J. S., and Otzen, D. E. (2014) High stability and cooperative unfolding of cytotoxic α -synuclein oligomers. *Biochemistry* **53**, 6252–6263
- Paslawski, W., Mysling, S., Thomsen, K., Jørgensen, T. J. D., and Otzen, D. E. (2014) Co-existence of two different α -synuclein oligomers with different core structures determined by hydrogen/deuterium exchange mass spectrometry. *Angew. Chem. Int. Ed. Engl.* **53**, 7560–7563
- Zijlstra, N., Blum, C., Segers-Nolten, I. M. J., Claessens, M. M. A. E., and Subramaniam, V. (2012) Molecular composition of sub-stoichiometrically labeled α -synuclein oligomers determined by single-molecule photobleaching. *Angew. Chem. Int. Ed. Engl.* **51**, 8821–8824
- Giehm, L., Svergun, D. I., Otzen, D. E., and Vestergaard, B. (2011) Low resolution structure of a vesicle disrupting α -synuclein oligomer that accumulates during fibrillation. *Proc. Natl. Acad. Sci. U. S. A.* **108**, 3246–3251
- Ehrnhoefer, D. E., Bieschke, J., Boeddrich, A., Herbst, M., Masino, L., Lurz, R., Engemann, S., Pastore, A., and Wanker, E. E. (2008) EGCG redirects amyloidogenic polypeptides into unstructured, off-pathway oligomers. *Nat. Struct. Mol. Biol.* **15**, 558–566
- Fusco, G., Sanz-Hernandez, M., Ruggeri, F. S., Vendruscolo, M., Dobson, C. M., and De Simone, A. (2018) Molecular determinants of the interaction of EGCG with ordered and disordered proteins. *Biopolymers* **109**, e23117
- Kurnik, M., Sahin, C., Andersen, C. B., Lorenzen, N., Giehm, L., Mohammad-Beigi, H., Jessen, C. M., Pedersen, J. S., Mente, S., Christiansen, G., Pedersen, S. V., Staal, R., Krishnamurthy, G., Pitts, K., Reinhart, P. H., et al. (2018) Novel α -synuclein aggregation inhibitors, identified by HTS, mainly target the monomeric state. *Cell Chem. Biol.* **25**, 1389–1402.e9
- Ehrnhoefer, D. E., Duennwald, M., Markovic, P., Wacker, J. L., Engemann, S., Roark, M., Legleiter, J., Marsh, J. L., Thompson, L. M., Lindquist, S., Muchowski, P. J., and Wanker, E. E. (2006) Green tea (-)-epigallocatechin-gallate modulates early events in huntingtin misfolding and reduces toxicity in Huntington's disease models. *Hum. Mol. Genet.* **15**, 2743–2751
- Meng, F. L., Abedini, A., Plesner, A., Verchere, C. B., and Raleigh, D. P. (2010) The flavanol (-)-epigallocatechin 3-gallate inhibits amyloid formation by islet amyloid polypeptide, disaggregates amyloid fibrils, and protects cultured cells against IAPP-induced toxicity. *Biochemistry* **49**, 8127–8133
- Hudson, S. A., Ecroyd, H., Dehle, F. C., Musgrave, I. F., and Carver, J. A. (2009) (-)-Epigallocatechin-3-gallate (EGCG) maintains kappa-casein in its pre-fibrillar state without redirecting its aggregation pathway. *J. Mol. Biol.* **392**, 689–700
- Bieschke, J., Russ, J., Friedrich, R. P., Ehrnhoefer, D. E., Wobst, H., Neugebauer, K., and Wanker, E. E. (2010) EGCG remodels mature α -

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- synuclein and amyloid-beta fibrils and reduces cellular toxicity. *Proc. Natl. Acad. Sci. U. S. A.* **107**, 7710–7715
23. Lee, Y. H., Lin, Y., Cox, S. J., Kinoshita, M., Sahoo, B. R., Ivanova, M., and Ramamoorthy, A. (2019) Zinc boosts EGCG's hIAPP amyloid inhibition both in solution and membrane. *Biochim. Biophys. Acta Proteins Proteom.* **1867**, 529–536
 24. Hauber, L., Hohenberg, H., Holstermann, B., Hunstein, W., and Hauber, J. (2009) The main green tea polyphenol epigallocatechin-3-gallate counteracts semen-mediated enhancement of HIV infection. *Proc. Natl. Acad. Sci. U. S. A.* **106**, 9033–9038
 25. Lorenzen, N., Nielsen, S. B., Yoshimura, Y., Vad, B. S., Andersen, C. B., Betzer, C., Kaspersen, J. D., Christiansen, G., Pedersen, J. S., Jensen, P. H., Mulder, F. A. A., and Otzen, D. E. (2014) How epigallocatechin gallate can inhibit alpha-synuclein oligomer toxicity *in vitro*. *J. Mol. Biol.* **289**, 21299–21310
 26. Hyung, S. J., DeToma, A. S., Brender, J. R., Lee, S., Vivekanandan, S., Kochi, A., Choi, J. S., Ramamoorthy, A., Ruotolo, B. T., and Lim, M. H. (2013) Insights into anti-amyloidogenic properties of the green tea extract (-)-epigallocatechin-3-gallate toward metal-associated amyloid- β species. *Proc. Natl. Acad. Sci. U. S. A.* **110**, 3743–3748
 27. Sciacca, M. F., Lolicato, F., Tempra, C., Scollo, F., Sahoo, B. R., Watson, M. D., García-Viñuales, S., Milardi, D., Raudino, A., Lee, J. C., Ramamoorthy, A., and La Rosa, C. (2020) Lipid-chaperone hypothesis: A common molecular mechanism of membrane disruption by intrinsically disordered proteins. *ACS Chem. Neurosci.* **11**, 4336–4350
 28. Korshavn, K. J., Satriano, C., Lin, Y., Zhang, R., Dulchavsky, M., Bhunia, A., Ivanova, M. I., Lee, Y. H., La Rosa, C., Lim, M. H., and Ramamoorthy, A. (2017) Reduced lipid bilayer thickness regulates the aggregation and cytotoxicity of amyloid- β . *J. Biol. Chem.* **292**, 4638–4650
 29. Kaylor, J., Bodner, N., Edridge, S., Yamin, G., Hong, D.-P., and Fink, A. L. (2005) Characterization of oligomeric intermediates in α -synuclein fibrillation: FRET studies of Y125W/Y133F/Y136F α -synuclein. *J. Mol. Biol.* **353**, 357–372
 30. Ahmed, R., and Melacini, G. (2018) A solution NMR toolset to probe the molecular mechanisms of amyloid inhibitors. *Chem. Commun.* **54**, 4644–4652
 31. Cawood, E. E., Karamanos, T. K., Wilson, A. J., and Radford, S. E. (2021) Visualizing and trapping transient oligomers in amyloid assembly pathways. *Biophys. Chem.* **268**, 106505
 32. Fawzi, N. L., Ying, J., Ghirlando, R., Torchia, D. A., and Clore, G. M. (2011) Atomic-resolution dynamics on the surface of amyloid- β protofibrils probed by solution NMR. *Nature* **480**, 268–272
 33. Wróblewski, K., Muhandiram, R., Chakrabarty, A., and Bennick, A. (2001) The molecular interaction of human salivary histatins with polyphenolic compounds. *Eur. J. Biochem.* **268**, 4384–4397
 34. Eaton, J. D., and Williamson, M. P. (2017) Multi-site binding of epigallocatechin gallate to human serum albumin measured by NMR and isothermal titration calorimetry. *Biosci. Rep.* **37**, BSR20170209
 35. Charlton, A. J., Baxter, N. J., Khan, M. L., Moir, A. J., Haslam, E., Davies, A. P., and Williamson, M. P. (2002) Polyphenol/peptide binding and precipitation. *J. Agric. Food Chem.* **50**, 1593–1601
 36. Sang, S., Yang, I., Buckley, B., Ho, C.-T., and Yang, C. S. (2007) Autoxidative quinone formation *in vitro* and metabolite formation *in vivo* from tea polyphenol (-)-epigallocatechin-3-gallate: Studied by real-time mass spectrometry combined with tandem mass ion mapping. *Free Radic. Biol. Med.* **43**, 362–371
 37. Sang, S., Lee, M. J., Hou, Z., Ho, C. T., and Yang, C. S. (2005) Stability of tea polyphenol (-)-epigallocatechin-3-gallate and formation of dimers and epimers under common experimental conditions. *J. Agric. Food Chem.* **53**, 9478–9484
 38. Nie, R. Z., Zhu, W., Peng, J. M., Ge, Z. Z., and Li, C. M. (2017) Comparison of disaggregative effect of A-type EGCG dimer and EGCG monomer on the preformed bovine insulin amyloid fibrils. *Biophys. Chem.* **230**, 1–9
 39. Pieri, L., Mадiona, K., and Melki, R. (2016) Structural and functional properties of prefibrillar α -synuclein oligomers. *Sci. Rep.* **6**, 24526
 40. Venkitakrishnan, R. P., Benard, O., Max, M., Markley, J. L., and Assadi-Porter, F. M. (2012) Use of NMR saturation transfer difference spectroscopy to study ligand binding to membrane proteins. *Methods Mol. Biol.* **914**, 47–63
 41. Mayer, M., and Meyer, B. (2001) Group epitope mapping by saturation transfer difference NMR to identify segments of a ligand in direct contact with a protein receptor. *J. Am. Chem. Soc.* **123**, 6108–6117
 42. Viennet, T., Wördehoff, M. M., Uluca, B., Poojari, C., Shaykhalishahi, H., Willbold, D., Strodel, B., Heise, H., Buell, A. K., Hoyer, W., and Eitzkorn, M. (2018) Structural insights from lipid-bilayer nanodiscs link α -synuclein membrane-binding modes to amyloid fibril formation. *Commun. Biol.* **1**, 44
 43. Lokappa, S. B., Suk, J.-E., Balasubramanian, A., Samanta, S., Situ, A. J., and Ulmer, T. S. (2014) Sequence and membrane determinants of the random coil-helix transition of α -synuclein. *J. Mol. Biol.* **426**, 2130–2144
 44. Nie, R.-Z., Dang, M.-Z., Li, K.-K., Peng, J.-M., Du, J., Zhang, M.-Y., and Li, C.-M. (2019) A-type EGCG dimer, a new proanthocyanidins dimer from persimmon fruits, interacts with the amino acid residues of A β 40 which possessed high aggregation-propensity and strongly inhibits its amyloid fibrils formation. *J. Funct. Foods* **52**, 492–504
 45. Nie, R.-Z., Zhu, W., Peng, J.-M., Ge, Z.-Z., and Li, C.-M. (2016) A-type dimeric epigallocatechin-3-gallate (EGCG) is a more potent inhibitor against the formation of insulin amyloid fibril than EGCG monomer. *Biochimie* **125**, 204–212
 46. Xu, Z. X., Ma, G. L., Zhang, Q., Chen, C. H., He, Y. M., Xu, L. H., Zhou, G. R., Li, Z. H., Yang, H. J., and Zhou, P. (2017) Inhibitory mechanism of epigallocatechin gallate on fibrillation and aggregation of amidated human islet amyloid polypeptide. *Chemphyschem* **18**, 1611–1619
 47. Huang, R., Vivekanandan, S., Brender, J. R., Abe, Y., Naito, A., and Ramamoorthy, A. (2012) NMR characterization of monomeric and oligomeric conformations of human calcitonin and its interaction with EGCG. *J. Mol. Biol.* **416**, 108–120
 48. Paslawski, W., Lorenzen, N., and Otzen, D. E. (2016) Formation and characterization of α -synuclein oligomers. *Methods Mol. Biol.* **1345**, 133–150
 49. Delaglio, F., Grzesiek, S., Vuister, G. W., Zhu, G., Pfeifer, J., and Bax, A. (1995) NMRPipe: A multidimensional spectral processing system based on UNIX pipes. *J. Biomol. NMR* **6**, 277–293
 50. Markley, J. L., Bax, A., Arata, Y., Hilbers, C. W., Kaptein, R., Sykes, B. D., Wright, P. E., and Wuthrich, K. (1998) Recommendations for the presentation of NMR structures of proteins and nucleic acids. *J. Mol. Biol.* **280**, 933–952
 51. Mulder, F. A. A., Otten, R., and Scheek, R. M. (2011) Origin and removal of mixed-phase artifacts in gradient sensitivity enhanced heteronuclear single quantum correlation spectra. *J. Biomol. NMR* **51**, 199–207
 52. Mayer, M., and Meyer, B. (1999) Characterization of ligand binding by saturation transfer difference NMR spectroscopy. *Angew. Chem. Int. Ed. Engl.* **38**, 1784–1788