

pubs.acs.org/OrgLett

# *trans*-Diastereoselective Ru(II)-Catalyzed Asymmetric Transfer Hydrogenation of $\alpha$ -Acetamido Benzocyclic Ketones via Dynamic Kinetic Resolution

Cite This: Org. Lett. 2019, 21, 3644–3648

Andrej Emanuel Cotman,<sup>†,‡,§</sup><sup>®</sup> Matic Lozinšek,<sup>‡,||</sup><sup>®</sup> Baifan Wang,<sup>†,⊥</sup> Michel Stephan,<sup>†, $\nabla$ </sup><sup>®</sup> and Barbara Mohar<sup>\*,†, $\nabla$ </sup><sup>®</sup>

<sup>†</sup>National Institute of Chemistry, Hajdrihova 19, SI-1000 Ljubljana, Slovenia

<sup>‡</sup>Faculty of Chemistry and Chemical Technology, University of Ljubljana, Večna pot 113, SI-1000 Ljubljana, Slovenia

**Supporting Information** 

**ABSTRACT:** A highly efficient enantio- and diastereoselective catalyzed asymmetric transfer hydrogenation via dynamic kinetic resolution (DKR–ATH) of  $\alpha$ , $\beta$ -dehydro- $\alpha$ -acetamido and  $\alpha$ -acetamido benzocyclic ketones to *ent-trans-\beta*-amido alcohols is disclosed employing a new *ansa*-Ru(II) complex of an enantiomerically pure *syn-N*,*N*-ligand, i.e. *ent-syn*-ULTAM-(CH<sub>2</sub>)<sub>2</sub>Ph. DFT calculations of the transition state structures

Organic

Letters



revealed an atypical two-pronged substrate attractive stabilization engaging the commonly encountered CH/ $\pi$  electrostatic interaction and a new additional O=S=O···HNAc H-bond hence favoring the *trans*-configured products.

A symmetric transformations comprising a dynamic kinetic resolution (DKR) step are practically very attractive since both involved enantiomers of the stereolabile racemic substrate would ideally converge to a diastereo- and enantiomerically pure product.<sup>1,2</sup> In particular, the DKR encountered in transition-metal-catalyzed reduction of ketones, such as in asymmetric transfer hydrogenation (ATH), is a powerful one-pot protocol for "deracemization" of substrates which possess a stereolabile  $\alpha$ -carbon by converting them into alcohols having two contiguous stereogenic centers.<sup>2</sup> A large number of these reductions are catalyzed by enantiomerically pure *ansa*-Ru(II)–[*ent-trans*-RSO<sub>2</sub>DPEN-( $\eta^{6}$ -arene)] complexes<sup>3</sup> (Figure 1) in the HCO<sub>2</sub>H/Et<sub>3</sub>N binary mixture, which acts as



**Figure 1.** Representative ATH-efficient *ansa*-Ru(II) complexes of tethered  $(R_1R)$ -RSO<sub>2</sub>DPEN and  $\eta^6$ -arene ligands.

both the H-source and an adequate "racemization medium" for the intermediate en route to the final product. The substrates scope for such DKR–ATH encompasses aryl,<sup>4</sup> perfluoroalkyl,<sup>5</sup> or acetylenic<sup>6</sup> ketones as well as  $\alpha$ -substituted benzocyclic ketones. Relevant to our present work are 2-Z-1-indanones and -tetralones wherein Z = alkyl, (het)aryl, F, Cl, CO<sub>2</sub>R', SO<sub>2</sub>Ph, C(O)Ph, SO<sub>2</sub>NHPh, and CH(OH)CF<sub>3</sub>, which furnish predominantly the corresponding enantiomeric  $\it cis$  configured products under these reaction conditions.  $^7$ 

Herein we present the exceptional highly *trans*-selective DKR–ATH of  $\alpha$ , $\beta$ -dehydro- $\alpha$ -acetamido and  $\alpha$ -acetamido benzocyclic ketones using a new chiral *ansa*-Ru(II) complex based on *ent-syn*-ULTAM *N*,*N*-ligand<sup>8</sup> (Scheme 1). Such a complex has enhanced thermal stability compared to its nontethered-type version.<sup>9</sup> In addition, the origin of the unexpected stereochemical outcome is investigated.

The Ru(II) complex C4 (CCDC 1905532) and its catalyst active form (active-C4) were prepared starting from enantiomerically pure *syn*-(3*R*,1'S)-ULTAM ligand following our improved procedure<sup>5a</sup> of the general one introduced by Wills for C1.<sup>10</sup> Accordingly, its selective mono-*N*-alkylation at rt led to the preligand 1 (79% yield), and the shelf-stable di- $\mu$ chlorido Ru(II) dimer 2 was readily formed (75% yield) by heating at 65 °C, in EtOH, the 1·HCl with RuCl<sub>3</sub> hydrate. Finally, the *ansa*-Ru(II)-Cl complex C4 was isolated in 43% yield by treatment of 2 with *i*-PrNEt<sub>2</sub> in CH<sub>2</sub>Cl<sub>2</sub> at rt; its structure was established by single-crystal X-ray diffraction. Conveniently, the catalyst *ansa*-Ru(II)-H active-C4 was generated *in situ* by stirring the Ru(II) dimer 2 at rt for 30 min in the HCO<sub>2</sub>H/Et<sub>3</sub>N medium. Alternatively, it can be generated similarly from C4.

We have initially surveyed the suitability of a selection of *ansa*-Ru(II) complexes (S/C = 500) for the DKR–ATH of the reference 2-acetamido-1-indenone substrate (S1) at 60 °C in a neat HCO<sub>2</sub>H/Et<sub>3</sub>N 3:2 mixture (Table1). S1 was prepared by

 Received:
 March 26, 2019

 Published:
 May 6, 2019



the Erlenmeyer azlactone synthesis followed by Friedel–Crafts intramolecular acylation.<sup>11</sup>

Table 1. Screening of *ansa*-Ru(II) Complexes in DKR–ATH of 2-Acetamido-1-indenone  $(S1)^a$ 

S1	O NHAC NHAC NHAC NHAC HCO2H/E cosolvent, 6 100% cor	II) cat. 500) t <sub>3</sub> N 3:2 t0 °C, 3 h iversion tran	OH NHAc ns-(1 <i>S</i> ,2 <i>S</i> )- <b>P1</b>	Reduction occurs via
entry	active Ru(II) cat.	cosolvent	trans/cis	ee (%) (trans)
1	C1	_	65:35	n.d.
2	C2	-	54:46	n.d.
3	C3	-	75:25	n.d.
4	C4	-	84:16	>99.9
5	C4	DMF	82:18	n.d.
6	C4	EtOH	84:16	n.d.
7	C4	EtOAc	87:13	n.d.
8	C4	$(CH_2Cl)_2$	90:10	>99.9
9	C4	toluene	91:9	>99.9
10	C4	PhCl	91:9 (>99) <sup>b</sup>	>99.9

<sup>*a*</sup>ATH of **S1** (187 mg, 1.0 mmol) was carried out at 60 °C using the Ru(II) cat. (S/C = 500, 2  $\mu$ mol) prepared *in situ* from the corresponding di- $\mu$ -chlorido Ru(II) dimer in HCO<sub>2</sub>H/Et<sub>3</sub>N 3:2 (1 mL); with a cosolvent (1 mL), less HCO<sub>2</sub>H/Et<sub>3</sub>N 3:2 (0.5 mL) was used. Conversion (100% within 3 h) and the *trans/cis* ratio were determined by <sup>1</sup>H NMR, and the ee of the *trans*-diastereomer was determined by chiral HPLC. n.d. = not determined. <sup>*b*</sup>After upgrading by trituration with MeCN of the crude (83% total yield).

The screening revealed the outperformance of the new Ru(II) catalyst active-C4 versus the ones based on the *trans*-(R,R)-RSO<sub>2</sub>DPEN-type *N*,*N*-ligands, leading to an increased *trans/cis* diastereomeric ratio (84:16) of the 2-acetamido-1-indanol product (P1) with a perfect enantioselectivity (>99.9% ee) for both diastereomers. A further improved diastereose-lectivity of up to 91:9 was attained in the presence of less polar cosolvents such as 1,2-dichloroethane, toluene, or chlorobenzene (Table 1, entries 8–10). Noteworthy, conducting this

two-step reduction process at 50 °C for 3 h led to the racemic 2-acetamido-1-indanone (*rac*-S1') (by chiral HPLC) in 90% yield (by <sup>1</sup>H NMR). Continued reduction at 60 °C afforded the stereoenriched  $\beta$ -acetamido alcohol P1, thus clearly validating the occurrence of a DKR during the keto function reduction step. Facile trituration with MeCN of the crude product gave the stereochemically pure *trans*-(1*S*,2*S*)-P1 (>99.9% ee) in 83% yield. The origin of this unusual and particularly high *trans*-selectivity of the Ru(II) catalyst active-C4 is addressed below. These results are interesting relative to the Rh(I)-catalyzed hydrogenation of S1 by the W. Zhang group.<sup>12</sup>

Following, the Ru(II) catalyst active-C4 was applied to the ATH of a diverse set of  $\alpha$ , $\beta$ -dehydro- $\alpha$ -acetamido and  $\alpha$ -acetamido benzocyclic ketones (S2–S9) (Figure 2). To our



Figure 2. Benzocyclic ketones  $S1{-}S10$  explored in Ru(II)-catalyzed DKR–ATH.

delight, the substituted 1-indenones **S2–S5** were reduced quantitatively within 3 h using an S/C = 500 under our standard optimized conditions (HCO<sub>2</sub>H/Et<sub>3</sub>N 3:2 in chlorobenzene at 60 °C) affording high *trans/cis* ratios (from 87:13 up to 92:8) with >99% ee. Trituration with MeCN of the crude yielded the enantio- and diastereochemically pure products *trans-*(1*S*,2*S*)-**P2–P5** (Figure 3). Furthermore, the never-before-reduced 2-acetamido-3-phenyl-1-indenone (**S6**) underwent DKR–ATH with 95% conversion delivering *trans,trans-***P6** in >99% ee as the major diastereomer (dr = 81:3:4:12). Interestingly, three contiguous stereogenic carbons



Figure 3. DKR–ATH products P2-P10 derived from the corresponding benzocyclic ketones S2-S10 of Figure 2. The enantioand diastereochemically pure products (>99% ee, dr >99, 44–83% yield) were obtained by trituration with MeCN or recrystallization from EtOAc.

were created in a one-pot procedure.<sup>7c</sup> Facile single recrystallization from EtOAc afforded the virtually enantioand diastereochemically pure *trans,trans-*(1*S,2S,3R*)-**P6** (>99% ee, dr >99); its absolute configuration was confirmed by singlecrystal X-ray diffraction (CCDC 1905533). Such an attained high level of *trans,trans*-selectivity is the result of the bias efficiency of the Ru(II) catalyst active-**C4** in the reduction step giving rise to a *trans*-configuration at C(1)–C(2), while the one at C(3) is thermodynamically driven.<sup>13</sup>

Next, the racemic 2-acetamido-1-acenaphthenone (*rac-S7*), prepared from acenaphthoquinone by Pd/C-catalyzed hydrogenation in  $Ac_2O$  of the mono-oxime, was subjected to DKR–ATH. This ketone gave the *trans*-diastereomer in >98% ee with a somewhat lower *trans/cis* ratio (84:16). Nonetheless, trituration with MeCN yielded stereochemically pure *trans*-(*S*,*S*)-P7.

When the focus was shifted to racemic 2-acetamido-1tetralones *rac*-**S8** and *rac*-**S9**, prepared from the corresponding  $\alpha$ -tetralone by treatment with isoamyl nitrite/KOt-Bu to form the 2-oxime and then Zn in AcOH/Ac<sub>2</sub>O reduction, their ATH resulted in >99% ee (*trans*) with 79:21 and 77:23 *trans/cis* ratios, respectively.<sup>14</sup> Gratifyingly here again, trituration with MeCN furnished the enantio- and diastereomerically pure *trans-(S<sub>1</sub>S)*-**P8** and -**P9** products.

Finally, the reason for the high *trans*-selectivity obtained in the ATH with active-C4 was investigated. By now, it is well-established that the saturation of the keto function under  $[\text{Ru}(trans\text{-TsDPEN})(\eta^6\text{-arene})]$ -catalyzed ATH occurs via a six-membered pericyclic transition state (TS) involving the Ru(II)–H catalyst hydride and a proton of the *N*,*N*-ligand amino group, while the stereoselectivity is influenced by the attractive CH/ $\pi$  electrostatic interaction between the  $\eta^6$ -arene and the ketone aryl group.<sup>15</sup>

Considering the nonclassical Ru(II) catalyst active-C4 structure embedding a syn-configured N,N-ligand and the cosolvent effect observed when lowering the overall polarity of the reaction medium, we contemplated the existence of an additional attractive interaction with the substrate in the TS. Most likely, it consists of a H-bond between the AcN-H of the intermediate S1' and a proximal O atom of the sulfonamido function of active-C4. Thus, with the aim to validate this assumption, ATH using active-C4 of the racemic 2phthalimido-1-indanone (rac-S10) (an S1' close analog lacking the NHC(O) function) and of the simple basic  $\alpha$ acetamidoacetone (a linear nonaryl ketonic substrate) were carried out. Indeed, rac-S10 converted into the expected cisconfigured major product (1S,2R)-P10 with a 97:3 cis/trans ratio (94% ee for *cis*) (Scheme 2);<sup>16,17</sup> this is due to the outward orientation of the phthalimido group in the TS thereby minimizing the sterics as with 2-Z-1-indanones of ref 7. In the case of  $\alpha$ -acetamidoacetone, 1-acetamido-2-propanol was obtained in an er = 85:15. These findings support the presence of an additional favorable catalyst-substrate S1' interaction in the TS being a H-bond between O=S=O…H-NAc.

Moreover, the two most plausible TS geometries were located by applying DFT calculations in chlorobenzene: "*trans*-TS" whereby the H-transfer from active-C4 to enantiomeric intermediate (2*S*)-S1′ leads to *trans*-(1*S*,2*S*)-P1, and "*cis*-TS" whereby (2*R*)-S1′ is transformed into *cis*-(1*S*,2*R*)-P1 (Figure 4). These calculations predict the "*trans*-TS" to be favored over the "*cis*-TS" by  $\Delta = 8.8$  kcal mol<sup>-1</sup> ( $E_a = 16.0$  vs 24.8 kcal mol<sup>-1</sup>), which is in line with the experimentally observed high



Scheme 2. Strategy for *trans-* or *cis-\beta-*Amino-1-indanol Using Active-C4

Figure 4. Schematic orientations of S1' enantiomers with cat. Ru(II)–H active-C4 in their TS, and the corresponding energy level from DFT calculations in chlorobenzene.

*trans*-diastereoselectivity.<sup>18</sup> Thus, the "*trans*-TS" structure is stabilized by a two-pronged catalyst ligand—substrate attractive interaction: by the H-bond (2.12 Å) between O=S=O…H–NAc and by CH/ $\pi$  (T-shaped geometry) favoring the (2S)-configuration with an overall *trans*-selectivity. In the case of "*cis*-TS", the geometry with a H-bond of 2.06 Å but with the lack of a CH/ $\pi$  interaction is the lowest in energy.

The aforementioned DKR–ATH employing active-C4 can serve to selectively prepare either of enantiomerically pure *trans*- or *cis*- $\beta$ -amino-1-indanol following the deprotection of enantiomerically pure *trans*-(1*S*,2*S*)-**P1** or *cis*-(1*S*,2*R*)-**P10** (Scheme 2).

In conclusion, we have introduced a new enantiomerically pure ansa-RuCl[syn-ULTAM-(CH<sub>2</sub>)<sub>3</sub>Ph] complex C4 and its Ru(II)-H active-C4. A diverse series of  $\alpha,\beta$ -dehydro- $\alpha$ acetamido and  $\alpha$ -acetamido benzocyclic ketones were reduced via DKR-ATH to the corresponding *trans-\beta*-amido alcohols P1-P9 with up to dr = 92:8 and excellent ee's. Facile trituration with MeCN yielded the enantio- and diastereochemically pure *trans*-products. DFT calculations relative to the diastereomeric TS structures revealed the existence of an atypical two-pronged attractive stabilization by CH/ $\pi$  interaction and by the O=S=O···H-NAc H-bond favoring the *trans*-products. These enantiomerically pure  $\beta$ -amido alcohols are valuable building blocks and amenable to further elaboration. In particular, either stereochemically pure *trans*or *cis-\beta*-amino-1-indanol (deprotected *trans*-P1 or deprotected *cis*-**P10**) can be selectively prepared using the same catalyst and procedure.

# ASSOCIATED CONTENT

#### **Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b01069.

Experimental data, chiral HPLC chromatograms, NMR spectra for prepared compounds, computational and SC-XRD details (PDF)

# **Accession Codes**

CCDC 1905532–1905533 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data\_request/cif, or by emailing data\_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

# AUTHOR INFORMATION

#### **Corresponding Author**

\*E-mail: moharbarbara@yahoo.com.

#### ORCID ®

Andrej Emanuel Cotman: 0000-0003-2528-396X Matic Lozinšek: 0000-0002-1864-4248 Michel Stephan: 0000-0002-3781-4011 Barbara Mohar: 0000-0001-5393-7030

#### **Present Addresses**

<sup>§</sup>Faculty of Pharmacy, University of Ljubljana, Aškerčeva 7, 1000 Ljubljana, Slovenia.

<sup>II</sup>Department of Inorganic Chemistry and Technology, Jožef Stefan Institute, Jamova cesta 39, 1000 Ljubljana, Slovenia, and Faculty of Chemistry and Chemical Technology, University of Ljubljana, Večna pot 113, 1000 Ljubljana, Slovenia.

<sup>1</sup>Department of Chemistry and Chemical Biology, Rutgers University, Piscataway, NJ 08854, USA.

<sup>V</sup>PhosPhoenix SARL, 115, rue de l'Abbé Groult, 75015 Paris, France.

#### Notes

The authors declare no competing financial interest.

# ACKNOWLEDGMENTS

This work was supported by the Slovenian Research Agency (Grant P1-0242). We are also grateful to Dr. Barbara Modec from the University of Ljubljana for collecting the SC-XRD data of (1S,2S,3R)-P6.

# **REFERENCES**

(1) (a) Huerta, F. F.; Minidis, A. B. E.; Bäckvall, J.-E. Chem. Soc. Rev. 2001, 30, 321–331. (b) Pellissier, H. Tetrahedron 2008, 64, 1563– 1601. (c) Blacker, J.; Headley, C. E. In Green Chemistry in the Pharmaceutical Industry; Dunn, P. J., Wells, A. S., Williams, M. T., Eds.; 2010; pp 269–288.

(2) For reviews on DKR in ATH, see: (a) Pellissier, H. *Tetrahedron* **2011**, 67, 3769–3802. (b) Echeverria, P.-G.; Ayad, T.; Phansavath, P.; Ratovelomanana-Vidal, V. *Synthesis* **2016**, *48*, 2523–2539.

(3) DPEN = *trans*-1,2-diphenylethylenediamine (available in both enantiomeric forms). For a review on *ansa*-Ru(II)-RSO<sub>2</sub>DPEN catalysts for ATH, see: (a) Nedden, H. G.; Zanotti-Gerosa, A.; Wills, M. *Chem. Rec.* **2016**, *16*, 2623–2643. Note, the *ansa*-Ru(II)-

Cl complex or the initial di- $\mu$ -chlorido Ru(II) dimer is the precursor of the corresponding Ru(II)-H active catalyst.

(4) Selected articles: (a) Seashore-Ludlow, B.; Saint-Dizier, F.; Somfai, P. Org. Lett. **2012**, *14*, 6334–6337. (b) Xu, F.; Zacuto, M. J.; Kohmura, Y.; Rosen, J.; Gibb, A.; Alam, M.; Scott, J.; Tschaen, D. Org. Lett. **2014**, *16*, 5422–5425. (c) Son, S.-M.; Lee, H.-K. J. Org. Chem. **2014**, *79*, 2666–2681. (d) Monnereau, L.; Cartigny, D.; Scalone, M.; Ayad, T.; Ratovelomanana-Vidal, V. Chem. - Eur. J. **2015**, *21*, 11799– 11806. (e) Sun, G.; Zhou, Z.; Luo, Z.; Wang, H.; Chen, L.; Xu, Y.; Li, S.; Jian, W.; Zeng, J.; Hu, B.; Han, X.; Lin, Y.; Wang, Z. Org. Lett. **2017**, *19*, 4339–4342. (f) Xiong, Z.; Pei, C.; Xue, P.; Lv, H.; Zhang, X. Chem. Commun. **2018**, *54*, 3883–3886.

(5) (a) Šterk, D.; Stephan, M.; Mohar, B. *Org. Lett.* **2006**, *8*, 5935–5938. (b) Mohar, B.; Stephan, M.; Urleb, U. *Tetrahedron* **2010**, *66*, 4144–4149.

(6) Fang, Z.; Wills, M. J. Org. Chem. 2013, 78, 8594-8605.

(7) For DKR-ATH of 2-Z-1-indanones and -tetralones wherein Z = alkyl, see: (a) Kišić, A.; Stephan, M.; Mohar, B. Adv. Synth. Catal. 2015, 357, 2540-2546. (b) Morris, J. C.; McErlean, C. S. P. Org. Biomol. Chem. 2016, 14, 1236-1238. (c) Cotman, A. E.; Modec, B.; Mohar, B. Org. Lett. 2018, 20, 2921-2924. For Z = het(aryl), see: (d) Peach, P.; Cross, D. J.; Kenny, J. A.; Mann, I.; Houson, I.; Campbell, L.; Walsgrove, T.; Wills, M. Tetrahedron 2006, 62, 1864-1876. (e) Vyas, V. K.; Bhanage, B. M. Org. Lett. 2016, 18, 6436-6439. For Z = F or Cl, see: (f) Ros, A.; Magriz, A.; Dietrich, H.; Fernández, R.; Alvarez, E.; Lassaletta, J. M. Org. Lett. 2006, 8, 127-130. For Z = CO<sub>2</sub>Me, see: (g) Šterk, D.; Stephan, M. S.; Mohar, B. Tetrahedron: Asymmetry 2002, 13, 2605-2608. (h) Rast, S.; Modec, B.; Stephan, M.; Mohar, B. Org. Biomol. Chem. 2016, 14, 2112-2120. For Z = CO2Et, see: (i) Ros, A.; Magriz, A.; Dietrich, H.; Lassaletta, J. M.; Fernández, R. Tetrahedron 2007, 63, 7532-7537. For Z = SO<sub>2</sub>Ph, see: (j) Ding, Z.; Yang, J.; Wang, T.; Shen, Z.; Zhang, Y. Chem. Commun. 2009, 5, 571-573. For Z = SO<sub>2</sub>NHPh, see: (k) Geng, Z.; Wu, Y.; Miao, S.; Shen, Z.; Zhang, Y. Tetrahedron Lett. 2011, 52, 907–909. For Z = C(O)Ph, see: (1) Wu, Y.; Geng, Z.; Bai, J.; Zhang, Y. Chin. J. Chem. 2011, 29, 1467–1472. For Z = CH(OH)CF<sub>3</sub>, see: (m) Cotman, A. E.; Cahard, D.; Mohar, B. Angew. Chem., Int. Ed. 2016, 55, 5294-5298.

(8) syn-ULTAM = syn-3-( $\alpha$ -aminobenzyl)-benzo- $\gamma$ -sultam (accessible in both enantiomeric forms). For its preparation, see: (a) Reference 7h. (b) Jeran, M.; Cotman, A. E.; Stephan, M.; Mohar, B. Org. Lett. **2017**, 19, 2042–2045.

(9) Noteworthy, ATH of benchmark substrates 1-indanone and  $\alpha$ -tetralone using our new tethered Ru(II)–[*syn*-(3*R*,1'S)-ULTAM-(CH<sub>2</sub>)<sub>3</sub>Ph] complex (active-C4) with an S/C = 1000 at 60 °C was complete within 4.5 h furnishing 98% ee (*R*) for both (see Supporting Information), while with the nontethered complex [Ru(*syn*-(3*R*,1'S)-ULTAM)(*p*-cymene)] with an S/C = 200 was required for full conversion within 4 h at 40 °C giving 99% ee (*R*) for 1-indanol or 1-tetralol.<sup>7h</sup> Note, this nontethered-type version has a short lifetime at 60 °C which is unsuitable for the quantitative C=C and C=O reductions of S1.

(10) Jolley, K. E.; Zanotti-Gerosa, A.; Hancock, F.; Dyke, A.; Grainger, D. M.; Medlock, J. A.; Nedden, H. G.; Le Paih, J. J. M.; Roseblade, S. J.; Seger, A.; Sivakumar, V.; Prokes, I.; Morris, D. J.; Wills, M. Adv. Synth. Catal. **2012**, 354, 2545–2555.

(11) Zhang, Z.; Hu, Q.; Wang, Y.; Chen, J.; Zhang, W. Org. Lett. 2015, 17, 5380-5383.

(12) By contrast, asymmetric hydrogenation of **S1** catalyzed by 1% Rh(I)-(3*H*-QuinoxP\*) in dioxane proceeded stereoselectively in the absence of DKR with reduction of only the C=C bond at 3 atm of H<sub>2</sub> (92% ee for **S1**'), or both C=C and C=O under 20 atm of H<sub>2</sub> (*trans/cis* = 93:7, 94% ee for *trans*-**P1**). For this, see: Hu, Q.; Chen, J.; Zhang, Z.; Liu, Y.; Zhang, W. Org. Lett. **2016**, *18*, 1290–1293.

(13) Noteworthy, under DKR–ATH (using Ru(II) cat. active-C4) of 2-methoxycarbonyl-3-phenyl-1-indanone (keto/enol = 88:12) and 2,3-diphenyl-1-indenone leading to dr = 57:43:0:0 (quant, 97% ee for major, >99% ee for minor) and 34:41:25:0 (quant), respectively, very

# **Organic Letters**

slow epimerization at C(3) occurred. For this, see: Cotman, A. E. PhD Dissertation, University of Ljubljana, 2018.

(14) In the literature, stereochemically pure *trans*-(R,R)-**P8** was prepared through a non-racemizing intramolecular Friedel–Crafts acylation of (R)-homotyrosine followed by diastereoselective reduction with Red-Al. For this, see: Melillo, D. G.; Larsen, R. D.; Mathre, D. J.; Shukis, W. F.; Wood, A. W.; Colleluori, J. R. *J. Org. Chem.* **1987**, *52*, 5143–5150.

(15) (a) Yamakawa, M.; Yamada, I.; Noyori, R. Angew. Chem., Int. Ed. 2001, 40, 2818–2821. (b) Noyori, R.; Yamakawa, M.; Hashiguchi, S. J. Org. Chem. 2001, 66, 7931–7944.

(16) In the literature, *trans*-(1*R*,2*R*)-**P10** (>99% ee) was prepared via a *trans*-selective reduction of *rac*-**S10** with NaCNBH<sub>3</sub>/AcOH followed by kinetic resolution with *Arthrobacter sp.* lipase. For this, see: Rouf, A.; Gupta, P.; Aga, M. A.; Kumar, B.; Parshad, R.; Taneja, S. C. *Tetrahedron: Asymmetry* **2011**, *22*, 2134–2143.

(17) Noteworthy, an identical *cis/trans* ratio but with a higher 98% ee was obtained employing the  $PipSO_2DPEN$ -based catalyst active-C3.

(18) Gas-phase DFT calculations similarly predicted the "trans-TS" to be favored over the "cis-TS" by  $\Delta = 7.3$  kcal mol<sup>-1</sup> ( $E_a = 11.1$  vs 18.4 kcal mol<sup>-1</sup>).