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# Structural Aspects of Photopharmacology: Insight into the Binding of Photoswitchable and Photocaged Inhibitors to the Glutamate Transporter Homologue

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photocaged compound ONB-hydroxyaspartate. The essential role of glutamate transporters in the functioning of the central nervous system renders them potential therapeutic targets in the treatment of neurodegenerative diseases. The obtained structures provide a clear structural insight into the origins of photocontrol in photopharmacology and lay the foundation for application of photocontrolled ligands to study the transporter dynamics by using time-resolved X-ray crystallography.

# INTRODUCTION

Photopharmacology is an emerging field that aims to control the action of bioactive molecules by light, enabling spatiotemporal precision of pharmacological treatment, which can reduce undesired interactions and side effects and prevent the development of resistance.<sup>1-6</sup> Because the majority of drug targets are membrane proteins,<sup>7</sup> there is an increasing interest in the application of photoregulated compounds in membrane protein research, with recent examples including GPCRs,<sup>8–10</sup> ion channels,<sup>11</sup> receptors,<sup>12</sup> and receptor-linked enzymes.<sup>13</sup>

photoresponsive transport inhibitors—azobenzene derivative of TBOA (both in *trans* and *cis* configuration) and with the

Photopharmacology relies on the introduction of molecular photoswitches, such as azobenzene<sup>14</sup> (Figure 1, highlighted in



**Figure 1.** Structures of inhibitors and photoactive compounds. The azobenzene part of the photoswitch and the *o*-nitrobenzyl (ONB) part of the caged compound are highlighted in blue and red, respectively.

blue) and diarylethene molecules,<sup>15</sup> or photocages,<sup>16</sup> such as *o*nitrobenzyl (ONB, Figure 1, highlighted in red), coumarin, or BODIPY derivatives,<sup>17</sup> into the structure of bioactive compounds. The azobenzene scaffold has two forms: a thermally stable, linear *trans* isomer and a metastable, bent *cis* isomer. These forms, which can be reversibly interconverted (switched) by using light, differ in shape and properties, which translates to differences in biological activity of compounds to which the switches have been introduced. Importantly, the *cis*form is thermally unstable, and in time reverts to the stable *trans* form, with the half-life ranging from milliseconds to years. This dynamic behavior poses a major challenge for any structural investigation of photoswitchable azobenzene-based systems.

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Another photopharmacological approach uses temporary deactivation of a biologically active small molecule with a photocage (photoremovable protecting group, PPG), resulting

Received: October 29, 2020 Published: January 15, 2021

 $IC_{E0} = 2.5 \pm 0.4 \mu M$ 



in a so-called "caged" compound.<sup>18</sup> Irradiation breaks the bond between the PPG and the compound, releasing the ligand. This irreversible process triggers the biological response with a high spatiotemporal resolution and can be used for time-resolved structural studies on the uncaging event and the following protein response.

One of the key challenges of photopharmacology lies in the very limited understanding of the structural basis of the alterations in biological effects induced by photoswitching, which hampers our ability to optimize the potency and difference in activity between the two isomers. Achieving large differences in activity between isomers is crucial for accomplishing spatial-temporal control over drug activity, and the future success of photopharmacology relies on elucidation of the structural aspects for the photocontrol of biological activity. However, up to date, only a few structures of soluble proteins in complex with noncovalently bound photoswitchable inhibitors are described, 19-22 but none for membrane proteins, although the use of photoswitches is actively pursued in these proteins as seen in the recent study of TRPV6 channels.<sup>23</sup> Furthermore, to the best of our knowledge, there is not a single report on the structures of both isomers of an azobenzene-based small molecule photopharmacological agent bound to its target.

Here we report on crystallographic analysis of the binding mode of two photopharmacological agents, the photoswitchable *p*-OMe-azo-TBOA and photocaged ONB-hydroxyaspartate (Figure 1), to the membrane transporter  $Glt_{Tk}$ .

This protein belongs to the SLC1A family of transporters, which also includes seven human transporters: the glutamate or excitatory amino acid transporters EAAT1-5 and the neutral amino acid exchangers ASCT1 and ASCT2.<sup>24</sup> EAATs remove the neurotransmitter glutamate from the synaptic cleft to prevent overstimulation of glutamate receptors and related neurotoxic effects, and malfunctioning of these transporters is linked to several diseases.<sup>25–29</sup> Glutamate transport is strictly coupled to cotransport of three sodium ions and a proton and counter transport of a potassium ion. Hence, directionality is dictated by the electrochemical gradients of these cations across the membrane. Under ischemic conditions and epilepsy, when the gradients are disrupted, glutamate transporters can function in reverse filling up the synaptic cleft with an excess of glutamate, thus causing severe neurotoxicity.<sup>24</sup> This reverse transport is mainly ascribed to EAAT3 protein;<sup>30–32</sup> hence, the selective temporary blockage of these transporters could be helpful. Although in the majority of EAAT-linked diseases the most promising route is enhancing the transport function (e.g., via increased expression of EAATs), there are some indications that inhibition of glutamate transporters can be used in the treatment of chronic pain.<sup>33</sup> Furthermore, some disorders are linked to gain-of-function mutations of the chloride channel activity, e.g., P290R mutation in EAAT1 causing episodic ataxia.<sup>34</sup> Hence, these transporters are considered as potential drug targets. The archaeal homologues Glt<sub>Ph</sub> from Pyrococcus horikoshii and Glt<sub>Tk</sub> from Thermococcus kodakarensis are widely used as model systems to study the transport mechanism of the glutamate transporter family.<sup>35</sup>

Glutamate transporters are homotrimeric membrane proteins, where each protomer has a trimerization scaffold domain that anchors the protein in the membrane and a transport domain that binds the substrate and cations and transports them across the membrane via an elevator mechanism.<sup>37,40,44</sup> Over the years, numerous crystal and cryo-EM structures of pubs.acs.org/JACS

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archaeal homologues have been obtained, revealing several conformations that the transporters successively adopt during the transport cycle, including outward- and inward-facing states,<sup>35,37,40</sup> as well as providing insight in the cation– substrate coupling mechanism.<sup>38,39</sup> The transport activity of glutamate transporters can be blocked by different molecules,<sup>45–53</sup> including competitive nontransported inhibitors DLthreo- $\beta$ -benzyloxyaspartic acid (TBOA, Figure 1) and (3S)-3-[[3-[[4-(trifluoromethyl)benzoyl]amino]phenyl]methoxy]-Laspartic acid (TFB-TBOA, Figure 1). These inhibitors were also used for structural studies that provided details on the transport mechanism. When bound to the substrate-binding site of the glutamate transporter homologues  $Glt_{Ph}$  and  $Glt_{Tk}$ TBOA prevents closure of the HP2 loop that serves as an extracellular gate,<sup>36,40</sup> which was also shown in the crystal structure of EAAT1 complexed with TFB-TBOA.<sup>54</sup> A cryo-EM structure of a mutant of the neutral amino acid exchanger ASCT2 from *Homo sapiens* (50% sequence identity to EAAT1) in complex with TBOA revealed that the same HP2 loop also works as an intracellular gate.55

Azobenzene-based photoswitchable inhibitors of glutamate transporters showed successful inhibition of mammalian EAATs.<sup>56</sup> As most of structural studies of glutamate transporters have been done on their archaeal homologues, photoswitchable inhibitors tested on these proteins provide an important tool for understanding the biological effect of photoswitching. Recently, we have reported on the development of photoswitchable inhibitor *p*-OMe-azo-TBOA (Figures 1 and 2) based on the potent inhibitor TFB-TBOA (IC<sub>50</sub> = 0.4



**Figure 2.** Photoswitch isomerization. Structures of *p*-OMe-azo-TBOA in *trans* (thermal) and *cis* (irradiated) configurations.

 $\pm$  0.1  $\mu$ M for transport by Glt<sub>Tk</sub>).<sup>57</sup> We investigated the binding of *trans* and *cis* isomers of *p*-OMe-azo-TBOA to the Glt<sub>Tk</sub> transporter. *p*-OMe-azo-TBOA in its thermally stable *trans* form readily binds to Glt<sub>Tk</sub> with an affinity in the low micromolar range. It can be reversibly switched in solution with UV light to its *cis* form, which results in a 3.6-fold decrease in inhibitory activity toward Glt<sub>Tk</sub>.<sup>57</sup> In solution, this switching process is highly efficient and provides >95% of the less potent *cis* isomer, which has a long half-life of >10 h. Switching the compounds back to the *trans* isomer with white light leads to the recovery of the higher potency (Figure 2). Altogether, this compound has desirable characteristics for structural studies because of the high *trans-cis* switching efficiency and long *cis* form half-life, despite a moderate difference in potency found in transport assays.

To establish the structural basis for this behavior, we report here the crystal structures of glutamate transporter homologue  $Glt_{Tk}$  in complex with *p*-OMe-azo-TBOA in both nonirradiated and irradiated states. Additionally, we obtained a

structure of  $\text{Glt}_{\text{Tk}}$  with a photocaged compound ONBhydroxyaspartate, which serves as an inhibitor in the nonirradiated state and generates  $\beta$ -hydroxyaspartate after uncaging, which is a substrate of mammalian glutamate transporters<sup>58</sup> and its prokaryotic homologues.<sup>59,60</sup>

# RESULTS

 $Glt_{Tk}$  Crystal Structures in Complex with Photoswitches. The crystal structures of  $Glt_{Tk}$  in complex with the *trans* or *cis* isomer of *p*-OMe-azo-TBOA (Table S1) each showed an outward-facing open state of the transporter. The inhibitor is bound in such a way that its aspartate moiety resides in the substrate binding site of the protein, while the photoswitchable part sticks out of the pocket (Figure 3). In



**Figure 3.** Substrate-binding site of  $\operatorname{Glt}_{Tk}$  with *p*-OMe-azo-TBOA in (a) *trans* and (b) *cis* configurations.  $2F_o - F_c$  electron density map contoured at  $1\sigma$  is shown as a blue mesh;  $F_o - F_c$  map contoured at  $\pm 3\sigma$  is shown as a green-red mesh. The open HP2 loop is shown in red. Photoisomers are shown as black (*trans*) or green (*cis*) sticks. Possible hydrogen bonds (distances in Å) between the binding site residues and the ligand are represented by black dashed lines.

this binding mode, the photoswitchable part occupies the space where the HP2 loop would have to move for gate closure, which in turn is required for subsequent conformational changes leading to transport. Therefore, *p*-OMe-azo-TBOA inhibits transport by preventing closure of the extracellular gate, similar to what was observed in  $\operatorname{Glt}_{Ph}$  and  $\operatorname{Glt}_{Tk}$  with bound TBOA.<sup>36,40</sup> While the orientation of the TBOA moiety of the *p*-OMe-azo-TBOA is very similar for both isomers, the photoswitchable parts show differences. The *trans* configuration of the photoswitch is mostly elongated, while the *cis* form has a bend in the diazo bond, which causes a shift of the *p*-OMe-benzyl group toward the HP1 loop (Figure 3).

Comparison of the Glt<sub>Tk</sub>-p-OMe-azo-TBOA complex structures with the crystal structure of the Glt<sub>Ph</sub>-TBOA complex<sup>36</sup> and  $Glt_{Tk}$ -TBOA complex<sup>40</sup> shows that the presence of an additional azobenzene moiety in the ligand and its isomerization did not affect the HP2 conformation in its open state. The loop adopts an identical conformation in  $\operatorname{Glt}_{Tk}$  bound to either the cis or trans isomer and  $\operatorname{Glt}_{Ph}$  and  $Glt_{Tk}$  bound to TBOA, with the rmsd values below 0.4 Å (for  $C\alpha$  atoms, residues 355–361, Table S2). Therefore, more potent inhibition by the trans isomer in comparison with the cis isomer can be explained only by a higher affinity of the inhibitor in the trans state for the protein. Binding affinities of the compound to Glt<sub>Tk</sub>, determined by isothermal titration calorimetry, are 1.89  $\pm$  1.26 and 3.19  $\pm$  0.49  $\mu$ M for the *trans* and *cis* state, respectively.<sup>57</sup> The crystal structure reveals that the main-chain amide group of Gly360 in HP2 forms a

hydrogen bond with the azobenzene nitrogen of the ligand when it is in a *trans* configuration (distance 2.8–3.1 Å). Isomerization of the diazo (N=N) bond from the *trans* to the *cis* configuration leads to a shift of the *p*-OMe-azo group away from the HP2 loop. The increased distance between the mainchain nitrogen atom of Gly360 and the azo-nitrogen atom of the compound (3.0–4.2 Å for the *cis* state, the range indicates that the compound is bound not uniformly to all three protomers) weakens or breaks this bond. Although in the *cis* configuration the *p*-OMe-azo group moves toward the HP1 loop, its oxygen is still too far away to form any additional polar contacts with the protein residues. Hence, the difference in affinity of the two photoisomers is likely explained by formation of an extra hydrogen bond for the *trans*-isomer in comparison with the *cis*-isomer.

Comparison of the inhibitor orientations in the different protomers of the homotrimeric protein using calculated omit maps (Figure S1) reveals that the photoswitchable moiety probably assumes a series of (slightly) different orientations in both the *cis* and *trans* configuration, reflecting its inherent flexibility. Alternatively, this might be a manifestation of an incomplete isomerization, similarly to what has been recently reported for reversibly switchable fluorescent proteins.<sup>61</sup>

To further investigate the structural basis for the lightinduced modulation of inhibitory potency, we additionally performed docking studies with the cocrystallized ligands and photoswitchable compounds that we studied earlier by transport assays.<sup>57</sup> The selected structures showed better docking score for the trans than the cis conformation for five out of seven ligands (Figure S2). We attempted to find a correlation between the differences in docking scores for cis and trans conformations of ligands and the differences in their inhibition potency according to Hoorens et al. For the ligand with the largest significant inhibition difference obtained experimentally, p-MeO-azo-TBOA, also used here for crystallization, the docking score difference is also the largest, supporting the previously obtained experimental results. Docking of other ligands also revealed the preference for trans conformation, but such differences were not so obvious in the experimental transport assays<sup>57</sup> (Figure S2). Interestingly the obtained poses of p-MeO-azo-TBOA by docking did not correspond well with the ligands' positions in the crystal structures (Figure S2), which might be either a result of shortcomings of our in silico calculations or the manifestation of site-specific radiation-induced damage of ligands. This was also observed by others, for example, in the crystal structure of ligand binding domain of a glutamate receptor GluK2 in complex with the photoswitchable compound Gluazo.<sup>22</sup> Alternatively, the influence of buffer composition (in this case 40% PEG 400 was used for cryoprotection) could have affected the ligand binding pose, which has been reported for other ligands.

Interestingly, according to our docking calculations, the difference in binding affinities can be attributed to an additional hydrogen bond formation in the *trans* state as well, albeit between different atoms, namely the *p*-OMe group of the inhibitor and main-chain nitrogens of Leu154 and Val153 (Figure S2).

 $Glt_{Tk}$  Crystal Structure in Complex with a Photocaged Inhibitor. The identical position of the HP2 loop in the presence of the *cis* and *trans* isomer of the inhibitor poses challenges to any future time-resolved measurements, as it could lead to ambiguity in the interpretation of the structural

data. To overcome this issue, we also crystallized  $Glt_{Tk}$  in complex with the photocaged compound ONB-hydroxyaspartate (Figures 1, 4, and 5), which does not suffer from potential



**Figure 4.** UV-vis absorption spectra of ONB-hydroxyaspartate in H<sub>2</sub>O (237  $\mu$ M) under UV-irradiation ( $t_{\rm irr}$  = 300 nm, photon flux = 3.25 × 10<sup>-9</sup> mol s<sup>-1</sup>).  $t_{\rm irr}$  = 480 s, T = 293 K. Right: before and after irradiation.



Figure 5. Caged compound (ONB-hydroxyaspartate) binding to  $Glt_{Tk'}$  (a) Interactions in the binding site (coloring is the same as in Figure 3). (b) Radioactive aspartate uptake in the presence and absence of the caged compound.

problems related to *cis-trans* isomerization. In the nonirradiated form, this compound is a close structural analogue of TFB-TBOA (Figure 1), and we expected it to act as an inhibitor. Exposure to UV light cleaves the PPG and releases the hydroxyaspartate, which is a known substrate for transport.<sup>58-60</sup>

The ONB-hydroxyaspartate was prepared via a nine-step chemoenzymatic synthesis route by using an engineered variant of methylaspartate ammonia lyase (MAL-L384A) as a biocatalyst (for experimental details, see the Supporting Information).<sup>62,63</sup> Its UV–vis spectrum features a band at  $\lambda = 270$  nm (Figure 4), characteristic for the ONB system. Irradiation of ONB-hydroxyaspartate at  $\lambda = 312$  nm leads to the drop of absorbance of this band, indicative of the uncaging of the transport substrate. In water, this process (Figure 4) proceeds with a quantum yield of 28%, which translates to a high uncaging cross section (a product of quantum yield and molar attenuation coefficient  $\varepsilon_{300} = 2138$  M<sup>-1</sup> cm<sup>-1</sup>) of 599 M<sup>-1</sup> cm<sup>-1</sup>.

The obtained crystal structure (Figure 5) is essentially identical with the  $\text{Glt}_{\text{Tk}}$ -*p*-OMe-azo-TBOA structures (rmsd of 0.3 Å, all C $\alpha$  atoms). Similar to the structures with photoswitchable compound isomers, the density for the TBOA-equivalent part of the photocaged ligand is clearly observed in the omit map (Figure S1). This is indicative that the photocaged compound is likely a competitive inhibitor. ONB-hydroxyaspartate indeed inhibited uptake of radiolabeled aspartate into proteoliposomes reconstituted with purified  $\text{Glt}_{\text{Tk}}$  (Figure 5).

### DISCUSSION

The activity of glutamate transporters can be reversibly controlled by using photoswitchable TBOA derivatives,<sup>57,4</sup> with thermally adapted trans p-OMe-azo-TBOA inhibiting more potently than the cis isomer.<sup>57</sup> To obtain structural insight into the mode of action of the photoswitchable TBOA derivative p-OMe-azo-TBOA and a new photocaged compound ONB-hydroxyaspartate, we performed cocrystallization experiments with  $Glt_{Tk}$ . TBOA-like inhibitors are non-transportable blockers of glutamate transporters that inhibit transport by preventing closure of the gate formed by the HP2 loop.  $^{36,54,55}$  We solved two crystal structures of  $\text{Glt}_{\text{Tk}}$  in the complex with p-OMe-azo-TBOA, after thermal relaxation or irradiation of the crystals, producing the trans and cis isomers of the compound, respectively, and one structure of the protein bound to the photocaged ONB-hydroxyaspartate. In all three structures, Glt<sub>Tk</sub> was in an identical outward facing conformation with open HP2 gate (see Table S2 for rmsd values). The structures suggest that the conformation of the open gate reflects an intrinsically preferred state at least under our experimental conditions, regardless of the identity of the inhibitor present. Considering the absence of crystal contacts in the region of the gate, it is unlikely that its conformation is enforced by the crystal environment. In the crystal structure of  $Glt_{Ph}$  bound to TBOA,<sup>36</sup> as well as in the recent cryo-EM structure of Glt<sub>Tk</sub> with TBOA,<sup>40</sup> a similar conformation of HP2 was found, which further indicates that such an open gate conformation represents a defined low-energy state.

Analysis of the two  $\text{Glt}_{\text{Tk}}$  structures with bound *p*-OMe-azo-TBOA provides a possible explanation for the different inhibitory effect of the *trans* and *cis* isomers. An extra hydrogen bond between the HP2 loop and the inhibitor in the *trans* configuration, compared to the *cis* isomer, could explain the higher binding affinity and stronger inhibitory properties of the *trans-p*-OMe-azo-TBOA. Because the isomerization of the ligand did not affect the extent of opening of the external HP2 gate, it is likely that the different extent of inhibitory potency of the *cis-* and *trans-*isomer is entirely due to the difference in binding affinity.

The structures of nonirradiated and irradiated *p*-OMe-azo-TBOA bound to  $\text{Glt}_{\text{Tk}}$  showed configurations of the photoswitch in its final isomeric states. However, the place where the photoswitch isomerizes remains unknown, and the process could go via two possible scenarios: (i) photoisomerization could have taken place while the compound was bound to the protein, or (ii) the *trans*-form could have dissociated and then isomerized in the solvent followed by association of the *cis*form back to the protein. Future time-resolved X-ray crystallography studies may allow the distinction between these scenarios by testing whether isomerization takes place while the inhibitor is bound to the protein in the crystal.

The identical position of the HP2 loop in the presence of the *cis* and *trans* isomer of the inhibitor poses challenges to any future time-resolved measurements of protein dynamics. Therefore, we also solved the structure of  $Glt_{Tk}$  in complex with the photocaged compound ONB-hydroxyaspartate, which (i) readily binds to the transporter and (ii) can be irradiated to generate a transported substrate. Because gate closure is an essential conformational change for transport, uncaging of ONB-hydroxyaspartate can be a useful tool to study the dynamics of gate movement in SLC1A transporters with high

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time resolution since the caging allows precise and controlled photo-release of a substrate to trigger the reaction.

The design of photopharmacological agents is most often focused on the analysis of the bioactive compound and defining the molecular motifs that can be substituted with a photoswitch to enable control of activity with light.<sup>65</sup> Recently, computational molecular modeling also enables protein-based approaches in which the structure of the target is taken into consideration,<sup>10,19,66,67</sup> albeit mostly to explain the observed effects rather than to predict them. The structural information about differential binding of the two isomers of a photopharmacological agent, presented here, constitutes the next step in structural evaluation that shows for the first time how a photopharmacological agent binds in its two forms to the target and modulates its activity in a light-controlled manner. We believe that this insight will empower the future development of target-focused design principles for photopharmacology.

#### MATERIALS AND METHODS

**Protein Purification and Crystallization.** His<sub>8</sub>-tagged wild-type Glt<sub>Tk</sub> was purified in the *apo* state in 10 mM Hepes KOH, pH 8.0, 100 mM KCl, 0.15% DM as it is described in refs 39 and 40. Concentrated to ~7 mg mL<sup>-1</sup> (Vivaspin 500, 50000 MWCO, Vivaproducts), the protein sample was supplemented with 300 mM NaCl.

Dissolved in water, *p*-OMe-azo-TBOA was preheated at 60 °C for 10 min to generate the thermal (*trans*) state and added to a protein solution to 360  $\mu$ M concentration (with final protein concentration 120  $\mu$ M). Crystals of the protein with *p*-OMe-azo-TBOA were obtained in the dark in 23% PEG 400, 100 mM sodium acetate, pH 4.8, 50 mM NaCl, 20 mM MgCl<sub>2</sub> at 4 °C by using the hanging drop vapor diffusion technique. Crystallization conditions supplemented with 10% glycerol were used as a cryoprotectant. For the *cis*-state, crystals with the *p*-OMe-azo-TBOA in the *trans* state were irradiated with 365 nm UV-lamp for 1 min<sup>57</sup> and treated with cryoprotectant in red light to avoid further isomerization.

Crystals of Glt<sub>Tk</sub> with the caged compound were obtained in the dark in 20% PEG 400, 250 mM glycine, pH 9.2, 50 mM MgCl<sub>2</sub> at 4 °C. The protein and ligand concentrations were 115 and 430  $\mu$ M, respectively. For cryoprotection, crystals were soaked in crystallization conditions containing 40% PEG 400 under the red light.

Ligand Docking. For each model, a single chain for docking studies was selected based on  $2F_{o} - F_{c}$  maps quality around ligand and surrounding residues (in both cases chain A). Four models (chain A with and without sodium ions for both *trans* and *cis* structures) were generated to assess whether they can reflect Glt<sub>Tk</sub> inhibition data.<sup>57</sup> The molecular docking was performed by using ICM Molsoft Pro (ver. 3.9-1b). The cocrystallized ligands were removed, and structures were converted into ICM format with default settings, which includes building of any missing side chains, addition of hydrogens, and energy-based Gln/Asn/His conformation optimization. A docking box for all four models was selected around the crystallographic ligand position in the cis conformation. We repeated docking studies five times with effort (ligand sampling depth), 35 each time saving three best conformations for each ligand. Docking scores were exported by using rdkit (ver. 2020.03.6) and analyzed by using pandas (ver. 1.1.2) and seaborn (0.11.0). Among all models, cis a Na (crystal structure with ligand bound in cis conformation, A chain, Na<sup>+</sup> ions present) showed a better docking score for the trans conformation of the ligand for five out of seven ligands (less for others), and poses generated by docking in this model were analyzed visually.

**Data Collection and Structure Determination.** Crystals were flash-frozen in liquid nitrogen, and data were collected at the PETRA III beamlines P13 and P14 (EMBL/DESY, Hamburg). Data were processed with XDS,<sup>68</sup> and the structures were solved by Molecular

Replacement with Phaser<sup>69</sup> by using  $\text{Glt}_{\text{Tk}}$  model PDB ID 5E9S. Manual modification of the structure was performed in  $\text{COOT}^{70}$  and refinement in Phenix refine.<sup>71</sup> Refined models were deposited into PDB under accession codes 6ZLH, 6ZL4, and 6ZGB for  $\text{Glt}_{\text{Tk}}$  with *p*-OMe-azo-TBOA *trans*- and *cis*-state and the caged compound, respectively. The final refinement statistics of the models are summarized in Table S1.

**Transport Assay.** Reconstitution of purified  $Glt_{Tk}$  into proteoliposomes and activity assays were performed as described previously.<sup>57</sup>

#### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacs.0c11336.

Tables S1 and S2; Figures S1–S4, and information about synthesis (PDF)

#### **Accession Codes**

The structural models and associated structure factors are available at PDB data bank under the following accession codes: 6ZLH, 6ZL4, and 6ZGB.

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#### Funding

The work was supported by NWO ECHO Grant 711.017.012 to W.S. and D.J.S. and Grant NWO OCENW.KLEIN.141 to A.G.

#### Notes

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

H.F. acknowledges funding from the China Scholarship Council; E.M. and A.G. acknowledge support from RFBR (19-29-12022).

# ABBREVIATIONS

Glt<sub>Tk</sub>, glutamate transporter homologue from *Thermococcus kodakarensis*; Glt<sub>Ph</sub>, glutamate transporter homologue from *Phyrococcus horikoshii*; SLC, solute carrier; GPCR, G-protein coupled receptor; ONB, *o*-nitrobenzyl; BODIPY, boron dipyrromethene; PPG, photoremovable protecting group; TRPV, transient receptor potential vanilloid channel; EAAT, excitatory amino acid transporter; ASCT, alanine-serinecystein transporter; TBOA, DL-threo- $\beta$ -benzyloxyaspartic acid; TFB-TBOA, (3S)-3-[[3-[[4-(trifluoromethyl)benzoyl]amino]phenyl]methoxy]-L-aspartic acid.

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