ORIGINAL ARTICLE

# Nobiletin and its derivatives overcome multidrug resistance (MDR) in cancer: total synthesis and discovery of potent MDR reversal agents 

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#### Abstract

Our recent studies demonstrated that the natural product nobiletin (NOB) served as a promising multidrug resistance (MDR) reversal agent and improved the effectiveness of cancer chemotherapy in vitro. However, low aqueous solubility and difficulty in total synthesis limited its application as a therapeutic agent. To tackle these challenges, NOB was synthesized in a high yield by a concise route of six steps and fourteen derivatives were synthesized with remarkable solubility and efficacy. All the compounds showed improved sensitivity to paclitaxel (PTX) in P-glycoprotein (P-gp) overexpressing MDR cancer cells. Among them, compound 29d exhibited water solubility 280-fold higher than NOB. A drug-resistance A549/T xenograft model showed that 29d, at a dose of $50 \mathrm{mg} / \mathrm{kg}$ co-administered with PTX ( $15 \mathrm{mg} / \mathrm{kg}$ ), inhibited tumor growth more effective than NOB and remarkably increased PTX


[^0]Total synthesis
concentration in the tumors via P-gp inhibition. Moreover, Western blot experiments revealed that 29d inhibited expression of NRF2, phosphorylated ERK and AKT in MDR cancer cells, thus implying 29d of multiple mechanisms to reverse MDR in lung cancer.
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## 1. Introduction

As the second leading cause of death next to heart disease, cancer is a growing global threat with 18 million cases and 9.6 million deaths in 2018 around the world ${ }^{1}$. Chemotherapy is commonly used for treatment of many types of cancers, but multidrug resistance (MDR) often causes chemotherapy failure, leading to death of a majority of patients ${ }^{2}$. Overcoming drug resistance has far-reaching implications for cancer patients and society ${ }^{3}$. Cellular mechanisms of drug resistance have been classified into transportbased classical and nonclassical MDR phenotypes ${ }^{3}$. Overexpression of P-glycoprotein (P-gp, P-gp/ABCB1 encoded by the MDR1 gene), one of the ABC-family efflux transporters, is associated with poor prognosis in many types of cancers and is frequently observed in recurrent tumors in clinic ${ }^{4}$. In the past 20 years, three generations of MDR inhibitors have been developed to target efflux transporters, especially P-gp. Unfortunately, drawbacks of the targeted therapy occurred after treatment for long time ${ }^{4}$. Recently, studies on reversal of MDR by natural products and their synthesized analogues, such as secalonic acid D , epigallocatechin gallate derivative, WS-10, etc ${ }^{5-7}$. have received increasing attention. Among them, several fragments (Murcko, Murcko generic, RECAP and aromatic ring) are found to actively inhibit $\mathrm{ABCB} 1^{6}$, serving as templates for designing new potent reversal agents to overcome MDR of cancers.

As a polymethoxy flavonoid isolated from tangerines, nobiletin (NOB, $3^{\prime}, 4^{\prime}, 5,6,7,8$,-hexamethoxyflavone, Fig. 1) exhibits a good safety profile and a broad spectrum of pharmacological activities, such as anti-oxidative, anti-inflammatory ${ }^{8}$ and anti-tumor effects ${ }^{9}$, especially high potency in cocktail therapy for drug-resistant cancers ${ }^{10}$. However, NOB exists as an aglycone, has poor water solubility $(1-5 \mu \mathrm{~g} / \mathrm{mL})$ and low oral bioavailability $(<1 \%)$, which limit its application as a therapeutic agent ${ }^{11}$. In addition to the traditional methods of separation and extraction, NOB could be prepared by total synthesis of over eleven steps ${ }^{12}$. Our recent studies demonstrated that NOB worked as an inhibitor of the MDR-efflux proteins by competing with chemotherapy drugs for the same binding site of P-gp, thus leading to improvement of the effectiveness of cancer chemotherapy in vitro ${ }^{10}$. NOB showed the


Figure 1 Chemical structure of nobiletin (NOB).

MDR reversal ability to make the ovarian tumor cells of A2780/T, which resisted to paclitaxel (PTX), became $>400$ folds sensitive in comparison with its parent A2780 cells. Although there are lots of synthesized compounds or natural products demonstrated for their MDR reversal abilities, few of them have comparable reversal ability with NOB. In addition, our previous study revealed that NOB not only inhibited the P-gp function but also modulated the NRF2/PI3K/AKT pathways which are the important mechanisms of the MDR. These findings encourage further studies on producing NOB and its derivatives in a simpler and more effective synthetic way to find more potent and soluble products.

Herein, to overcome the disadvantages and expand the scope of application, NOB was newly generated in high yield by a concise route of six steps and fourteen derivatives with remarkable solubility were synthesized. Further, we analyzed the efficacy of the compounds as MDR reversal agents in P-gp-overexpressing MDR cancer cells in vitro and in an A549/T xenograft model in vivo. We also investigated the underlying mechanism for the MDR reversal effects of NOB and its derivatives.

## 2. Results and discussion

### 2.1. Synthesis of the designed compounds

The target compounds were prepared by the synthetic routes reported in Schemes $1-3$. First, commercially available 3,4,5trimethoxyphenol (1) was selected as the starting material and reacted with acetic anhydride to afford the acetylation product 2, followed by Fries rearrangement to provide 3, which was then halogenated with $N$-iodosuccinimide to obtain


Scheme 1 Syntheses of 5,6,7,8-tetramethoxy-2-phenyl-4H-chro-men-4-ones 7a-7c. Reagents and conditions: (a) $\mathrm{Ac}_{2} \mathrm{O}, \mathrm{AcOH}$, $110{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}$; (b) $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}, \mathrm{AcOH}, 70^{\circ} \mathrm{C}, 2 \mathrm{~h}$; (c) NIS, $p$-toluenesulfonic acid monohydrate, $\mathrm{CH}_{3} \mathrm{CN}$, rt, 2 h ; (d) $4 \mathrm{~mol} / \mathrm{L} \mathrm{MeONa}$ in $\mathrm{MeOH}, \mathrm{CuCl}, \mathrm{DMF}, 9{ }^{\circ} \mathrm{C}, 30 \mathrm{~min}$; (e) acyl chlorides, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM}$, $0{ }^{\circ} \mathrm{C}$ to $\mathrm{rt}, 2 \mathrm{~h}$; (f) triethylsilyl trifluoromethanesulfonate, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCE}$, reflux, 2 h .


Scheme 2 Syntheses of 2-(3,4-dimethoxyphenyl)-5,6,7-trimethoxy-4H-chromen-4-ones 13, $\mathbf{1 6}$ and 18. Reagents and conditions: (a) 3,4dimethoxybenzoyl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM}, 0^{\circ} \mathrm{C}$ to rt, 2 h ; (b) $\mathrm{Br}_{2}, \mathrm{AcOH}, \mathrm{AcONa}, 0^{\circ} \mathrm{C}$ to rt, 12 h ; (c) $\mathrm{HNO}_{3}, 0^{\circ} \mathrm{C}, 10 \mathrm{~min}$; (d) triethylsilyl trifluoromethanesulfonate, $\mathrm{Et}_{3} \mathrm{~N}$, DCE, reflux, 2 h ; (e) 3,4-dimethoxybenzoyl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM}, 0^{\circ} \mathrm{C}$ to $\mathrm{rt}, 2 \mathrm{~h}$; (f) 3,4-dimethoxybenzoyl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM}, 0{ }^{\circ} \mathrm{C}$ to $\mathrm{rt}, 2 \mathrm{~h}$; (g) triethylsilyl trifluoromethanesulfonate, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCE}$, reflux, 2 h ; (h) triethylsilyl trifluoromethanesulfonate, $\mathrm{Et}_{3} \mathrm{~N}$, DCE, reflux, 2 h ; (i) Pd/C, $\mathrm{H}_{2}, \mathrm{THF}, \mathrm{rt}, 12 \mathrm{~h}$.


Scheme 3 Syntheses of 2-(3,4-dimethoxyphenyl)-6,7,8-trimethoxy-4H-chromen-4-ones $\mathbf{2 3}$ and 29a-29d. Reagents and conditions: (a) $\mathrm{Ac}_{2} \mathrm{O}$, $\mathrm{AcOH}, 110{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}$; (b) $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}, \mathrm{AcOH}, 70{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}$; (c) 3,4-dimethoxybenzoyl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCE}, 0{ }^{\circ} \mathrm{C}$ to $\mathrm{rt}, 2 \mathrm{~h}$; (d) triethylsilyl trifluoromethanesulfonate, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCE}$, reflux, 2 h . (e) iodomethane, $\mathrm{K}_{2} \mathrm{CO}_{3}$, DMF, rt, 1 h ; (f) $\mathrm{HNO}_{3}, 0^{\circ} \mathrm{C}, 10 \mathrm{~min}$; (g) Pd/C, $\mathrm{H}_{2}$, THF, rt, 12 h ; (h) 3,4-dimethoxybenzoyl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM}, 0{ }^{\circ} \mathrm{C}$ to rt 2 h ; (i) $t$-BuOK, THF, reflux, 12 h ; (j) iodomethane, $\mathrm{K}_{2} \mathrm{CO}_{3}$, DMF, chloroalkanes, NaH , DMF, $90^{\circ} \mathrm{C}, 2.5 \mathrm{~h}$.
compound 4 (Scheme 1). The key intermediate 1-(2-hydroxy-3,4,5,6-tetramethoxyphenyl)ethan-1-one (5) was synthesized by the reaction of $\mathbf{4}$ with sodium methoxide in the presence of copper chloride ${ }^{13}$. Compounds $\mathbf{6 a - 6 c}$ were synthesized by acylation of $\mathbf{5}$ with the corresponding acyl chlorides ${ }^{14}$. The target compounds $\mathbf{7 a}-7 \mathbf{c}$ were obtained by cyclization of $\mathbf{6 a - 6 c}$ in the presence of triethylsilyl trifluoromethanesulfonate, respectively.

As outlined in Scheme 2, compounds $\mathbf{1 1}$ and $\mathbf{1 2}$ were obtained from $\mathbf{3}$ by treatment with bromine and nitric acid, respectively. Aryl
esters 10, 14 and 15 were synthesized by acylation of 1-(2-hydroxyphenyl)ethan-1-ones with the corresponding acyl chlorides, which were treated with triethylsilyl trifluoromethanesulfonate to provide the corresponding flavones 13,16 and 17 . Subsequently, compound 17 was reduced by catalytic hydrogenation with $\mathrm{Pd} / \mathrm{C}$ to afford 8-amino-2-(3,4-dimethoxyphenyl)-5,6,7-trimethoxy-4H-chro-men-4-one (18).

Furthermore, commercially available 2,3-dimethoxyphenol (19a) or 2,3,4-trimethoxyphenol (19b) were selected as the
starting materials, and the same procedures as in Schemes 1 and 2 were used to obtain compounds 23a-23d (Scheme 3). The intermediate 21b was treated with iodomethane and $\mathrm{K}_{2} \mathrm{CO}_{3}$ in the presence of $N, N$-dimethylformamide (DMF) to obtain 24, followed with nitric acid to afford compound 25. After 25 was reduced by catalytic hydrogenation with $\mathrm{Pd} / \mathrm{C}$ to yield compound 26, it was treated with 3,4-dimethoxybenzoyl chloride to obtain 27. Finally, 27 was treated with $t$-BuOK in a solution of tetrahydrofuran (THF) to produce 2-(3,4-dimethoxyphenyl)-5,6,7,8-tetramethoxyquinolin-4(1H)-one (28) ${ }^{15}$. Compounds 29a-29d were synthesized by the reaction of $\mathbf{2 8}$ and corresponding alkyl halides in the presence of sodium hydride.

To avoid the interference of false positive compounds in our subsequent study, Pan Assay Interference Compounds (PAINS) screening of the designed compounds was carried out via an online program "PAINS Remover" (http://www.cbligand.org/PAINS/ ), and all of the compounds passed the filter and were retained.

## 2.2. $N O B$ and its derivatives reverse $A B C B 1$-mediated $M D R$

The $\mathrm{IC}_{50}$ values of NOB and its derivatives against PTX-resistant HCT8/T cells (without adding $0.94 \mu \mathrm{~mol} / \mathrm{L}$ PTX to the culture medium) were listed in Table 1. These compounds showed antitumor effects against resistant human intestine and lung cancer cell lines, and their cytotoxicity are much lower than that of PTX ( $5.89 \mu \mathrm{~mol} / \mathrm{L}$ in HCT8/T).

The long-term colony formation assays further ascertained the enhanced anti-proliferative effect of NOB and its derivatives on A549/T and HCT8/T cells, as shown in Fig. 2A and B. We noticed that NOB, $\mathbf{1 3}$ and 29d could inhibit MDR cancer cell colony formation in a dose-dependent manner. However, there is no difference in the inhibition of the colony formation among 13, 29d and NOB at their lowest effect dosage, indicating the similar reversal effects of NOB and its derivatives after long-time treatment.

As shown in Table 1, NOB and its derivatives enhanced the anti-tumor effects of PTX by decreasing its $\mathrm{IC}_{50}$ against the resistant HCT8/T cells. Specifically, the $\mathrm{IC}_{50}$ values of PTX in the presence of NOB, 7c, 13 and $\mathbf{2 9 d}$ were reduced by 185-, 81-, 66and 87 -fold, respectively. However, NOB, 7c, 13 and 29d did not affect the $\mathrm{IC}_{50}$ of PTX for the sensitive HCT8 cells (data not shown). In view of the relatively stronger cytotoxicity ( $38.46 \mu \mathrm{~mol} / \mathrm{L}$ ) of $\mathbf{7 c}$ than that of NOB, only $\mathbf{1 3}, \mathbf{2 9 d}$ and NOB were selected for further studies.

### 2.3. Structure-activity relationships (SARs) of NOB and its derivatives

When a methoxy group was added at the $\mathrm{C}-5$ position (7c) relative to $\mathbf{7 b}$ (NOB), the affinity decreased since there was insufficient space for accommodation of another methoxy group. The presence of 1,3 -benzodioxole (7a) was unfavorable to affinity due to steric hindrance. For inhibitors 23a (one methoxy group), 23d (two methoxy groups), $\mathbf{1 3}$ (two methoxy groups) and 7b (three methoxy groups) with the same $\mathrm{R}_{4}$ group, we found that more methoxy groups on the benzene ring A was propitious to better inhibitory potency, whereas the opposite phenomenon was observed on the $\mathrm{R}_{4}$ position for $\mathbf{7 b}$ (two methoxy groups) and $\mathbf{7 c}$ (three methoxy groups). In addition, for inhibitors $\mathbf{1 3}, 16$ and 18, the bulky bromine atom of $\mathbf{1 6}$ or the $\mathrm{NH}_{2}$ group of $\mathbf{1 8}$ at the $\mathrm{R}_{1}$ group decreased the inhibitory affinity in comparison with the hydrogen atom of $\mathbf{1 3}^{16-18}$.

On the basis of these results, we decided to retain the methoxy groups on the A and B rings and change the side chain at the X position. The dimethylaminoethyl sidechain of 29d formed hydrophobic and van der Waals interactions with the residues of P-gp, which significantly differs from NOB. Therefore, other modifications such as replacement of methylethyldimethylamine with 2-methoxyethyl (29c), H (28), Me (29a) or $n$-propyl (29b), were failed to improve their binding affinities. Thus, NOB, 29d and $\mathbf{1 3}$ were selected for subsequent study.

Moreover, the solubility data of NOB, $\mathbf{1 3}$ and 29d were measured as $5.58 \pm 1.17,3.67 \pm 0.52$ and $1.58 \pm 0.13 \mathrm{mg} / \mathrm{mL}$ in sodium phosphate buffer ( $0.35 \mathrm{~mol} / \mathrm{L}, \mathrm{pH}=7.3$ ) at $25^{\circ} \mathrm{C}$, respectively, using the high-performance liquid chromatography (HPLC) approach. The presence of the dimethylaminoethyl sidechain of 29d possibly accounts for the notable improvement in the water solubility of $\mathbf{2 9 d}$ over NOB and $\mathbf{1 3}$ (280-fold higher), predicting that 29d possibly has better performance in vivo.

### 2.4. Anti-tumor effects in A549/T xenograft model

We next evaluated whether synthesized NOB and its derivatives could overcome PTX resistance in the A549/T xenograft model. Based on previous study, PTX was administered via intraperitoneal (i.p.) route at an effective dose ( $15 \mathrm{mg} / \mathrm{kg}$ ) for normal tumor ${ }^{19}$. As expected, PTX alone exhibited no effect on treating MDR tumors, as shown by similar tumor progression between the PTX treated group and vehicle control group, indicating resistance (Fig. 3). However, for the cocktail 29d and PTX administration, 29d at both $25 \mathrm{mg} / \mathrm{kg}$ and $50 \mathrm{mg} / \mathrm{kg}$ dosages significantly reduced the tumor volumes (Fig. 3B-D) compared with PTX alone, indicating considerable therapeutic efficacy. Moreover, there was no apparent weight loss observed in all nude mice (Fig. 3A), suggesting that the combination regimen did not bring additional toxicity. Notably, 29d ( $50 \mathrm{mg} / \mathrm{kg}$ ) with PTX greatly reduced the volumes of tumors by $57 \%$ in the A549/T xenograft model, which is more effective than NOB at the same dosage ( $P<0.05$ ). Similarly, NOB derivative $\mathbf{1 3}$ also enhanced the therapeutic efficacy of PTX in the A549/T xenograft model (Fig. 4). However, the tumor growth inhibition effect of $\mathbf{1 3}$ is similar to that of NOB.

### 2.5. Both 29d and $\mathbf{1 3}$ potentiate apoptosis induced by PTX

Consistent with the effects of NOB, 29d and $\mathbf{1 3}$ significantly potentiated the apoptosis induced by PTX, as shown in Fig. 5A and B, while PTX ( $5.89 \mu \mathrm{~mol} / \mathrm{L}), \mathbf{2 9 d}$ or $\mathbf{1 3}(9 \mu \mathrm{~mol} / \mathrm{L})$ alone did not induce cell apoptosis. Moreover, cell cycle progressions were evaluated by flow cytometry in asynchronously growing HCT8/T cells and sensitive HCT8 cells after treatment with PTX in the absence or presence of $\mathbf{2 9 d}$ or $\mathbf{1 3}$. When $0.94 \mu \mathrm{~mol} / \mathrm{L}$ PTX was added to the culture medium of HCT8/T for maintaining resistance, $71.9 \%$ of cells in G1 phase and $17.1 \%$ of cells in G2 phase after incubation were observed, which is similar to the phase ratios in the vehicle group, as shown in Fig. 5C and D. However, PTX $(0.94 \mu \mathrm{~mol} / \mathrm{L})$ treated with $\mathbf{2 9 d}$ or $\mathbf{1 3}(9 \mu \mathrm{~mol} / \mathrm{L})$ for 48 h resulted in significant G2M arrest ( $>80 \%$ ) in HCT8/T cells. Therefore, 29d or 13 significantly enhanced the cell growth inhibition, apoptosis and G2M cell cycle arrest induced by $0.94 \mu \mathrm{~mol} / \mathrm{L}$ PTX, although HCT8/T cells were remarkably resistant to $0.94 \mu \mathrm{~mol} / \mathrm{L}$ PTX.

Table 1 Nobiletin (NOB) and its derivatives reduce the $\mathrm{IC}_{50}$ of paclitaxel (PTX) for multidrug resistant HCT8/T cancer cells.


| Compd. | $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | $\mathrm{R}_{3}$ | $\mathrm{R}_{4}$ | X | Cytotoxicity $\mathrm{IC}_{50}(\mu \mathrm{~mol} / \mathrm{L})^{\mathrm{a}}$ | Cytotoxicity of combination $\mathrm{IC}_{50}(\mu \mathrm{~mol} / \mathrm{L})^{\mathrm{b}}$ | Reversal fold ${ }^{\text {c }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| PTX | - | - | - | - | - | $5.89 \pm 0.24$ | $5.89 \pm 0.24$ | 1 |
| 7a | OMe | OMe | OMe |  | O | >100 | $0.14 \pm 0.01$ | 43 |
| 7 b (NOB) | OMe | OMe | OMe |  | O | $44.67 \pm 3.83$ | $0.030 \pm 0.002$ | 185 |
| 7c | OMe | OMe | OMe |  | O | $38.46 \pm 3.30$ | $0.080 \pm 0.004$ | 81 |
| 13 | H | OMe | OMe |  | O | $74.99 \pm 4.04$ | $0.090 \pm 0.005$ | 66 |
| 16 | Br | OMe | OMe |  | O | >100 | $0.15 \pm 0.01$ | 40 |
| 18 | $\mathrm{NH}_{2}$ | OMe | OMe |  | O | $66.83 \pm 3.99$ | $0.29 \pm 0.05$ | 21 |
| 23a | OMe | H | H |  | O | $66.83 \pm 2.73$ | $0.29 \pm 0.01$ | 21 |
| 23b | OMe | H | H |  | O | $69.98 \pm 6.78$ | $0.44 \pm 0.04$ | 14 |
| 23c | OMe | H | H |  | O | $39.36 \pm 4.41$ | $0.36 \pm 0.02$ | 17 |
| 23d | OMe | OMe | H |  | O | $66.83 \pm 2.73$ | $0.13 \pm 0.01$ | 47 |
| 28 | OMe | OMe | OMe |  | NH | >100 | $0.23 \pm 0.02$ | 27 |
| 29a | OMe | OMe | OMe |  | NMe | $26.61 \pm 2.63$ | $0.11 \pm 0.01$ | 55 |


| Compd. | $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | $\mathrm{R}_{3}$ | $\mathrm{R}_{4}$ | X | Cytotoxicity $\mathrm{IC}_{50}(\mu \mathrm{~mol} / \mathrm{L})^{\mathrm{a}}$ | Cytotoxicity of combination $\mathrm{IC}_{50}(\mu \mathrm{~mol} / \mathrm{L})^{\mathrm{b}}$ | Reversal fold ${ }^{\text {c }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 29b | OMe | OMe | OMe |  |  | $21.13 \pm 1.77$ | $0.15 \pm 0.01$ | 40 |
| 29c | OMe | OMe | OMe |  |  | $59.57 \pm 3.62$ | $0.24 \pm 0.01$ | 26 |
| 29d | OMe | OMe | OMe |  |  | $>100$ | $0.070 \pm 0.008$ | 87 |
| -Not applicable. |  |  |  |  |  |  |  |  |
| ${ }^{\text {a }}$ Intrinsi <br> ${ }^{\mathrm{b}}$ Values <br> ${ }^{\mathrm{c}}$ The rev | toxicity <br> ${ }_{0}$ for cy <br> fold w | target oxicity alculat | mpoun | (P | 10 | TX) were as mea bitor $10 \mu \mathrm{~mol} / \mathrm{L}$ | SD of at least three independ PTX). | experimen |

## A

Colony formation assay: HCT8/T


B
Colony formation assay: A549T


Figure 2 NOB and its derivatives reverse the resistance of paclitaxel (PTX) to P-gp-overexpressing MDR cancer cells. 29d, 13 and NOB enhanced the anti-proliferative effects of PTX in two drug-resistant cancer cell lines, HCT8/T (A) and A549/T (B), in a dose-dependent manner according to the colony formation assay. The cells were treated with PTX $(0.24 \mu \mathrm{~mol} / \mathrm{L}$ for A549/T and $0.94 \mu \mathrm{~mol} / \mathrm{L}$ for $\mathrm{HCT} 8 / \mathrm{T}$ for maintaining the resistance of cancer cells) in the presence or absence of $\mathbf{2 9 d}, \mathbf{1 3}$ and NOB at the indicated concentrations.

### 2.6. 29d and 13 inhibit the $P$-gp function

We also evaluated the modulating effects of synthesized NOB and its derivatives on P-gp by evaluating the accumulation of P gp substrates, including doxorubicin (DOX) and flutax-2 (a fluorescent PTX derivative), using fluorescence microscopy and
flow cytometry analysis. Enhanced intracellular accumulations of DOX or flutax-2 by NOB and its derivatives were confirmed by flow cytometry analysis, as shown in Fig. 6A and C. Moreover, stronger fluorescence signals of DOX and flutax-2 were observed by adding NOB, 29d, 13 and quinidine (QND, positive control), as shown in Fig. 6B and D. The effects of 29d on the


Figure 3 NOB (N) and its derivative 29d enhanced the anticancer effect of PTX in the PTX-resistance A549/T cell nude mouse xenograft model. The changes in body weight (A) and tumor growth curves (B) were recorded after A549/T cell implantation. The excised tumors were photographed (C) and weighed (D) on day 30 after implantation. The data were shown as mean $\pm$ SD for each group $(n=6)$, ${ }^{*} P<0.05$, ${ }^{* *} P<0.01,{ }^{* * *} P<0.001 \mathrm{vs}$. PTX alone; ${ }^{\#} P<0.05 \mathrm{vs}$. administered PTX and NOB. CTR, the control group.


Figure $4 \mathrm{NOB}(\mathrm{N})$ and its derivative 13 enhanced the anticancer effect of PTX in the PTX resistance A549/T cell nude mouse xenograft model. The changes in body weight (A) and tumor growth curves (B) were recorded after A549/T cell implantation. The excised tumors were photographed (C) and weighed (D) on day 30 after implantation. The data were shown as mean $\pm$ SD for each group $(n=6),{ }^{*} P<0.05,{ }^{* *} P<0.01$, ${ }^{* * *} P<0.001$ vs. PTX alone.
intracellular accumulation were stronger than those of $\mathbf{1 3}$ and NOB.

LC-MS/MS was used to detect the contents of PTX, 29d and NOB in the tumors growing in the A549T xenograft model after treatment. Compared with the PTX alone group, NOB enhanced the content of PTX in tumors by 2.93 -fold, while 29d enhanced by 4.82-fold, which was approximately 1.65 -fold that of NOB
(Fig. 7A). Moreover, the concentration of 29d in tumor tissues was significantly increased by 1.72 -fold in the PTX $+\mathbf{2 9 d}(50 \mathrm{mg} / \mathrm{kg}$ ) group compared with the $\mathbf{2 9 d}$ ( $50 \mathrm{mg} / \mathrm{kg}$ ) group (Fig. 7B), indicating that 29d and PTX were competing for binding on P-gp. All of these results suggested that synthesized NOB and its derivatives could inhibit P-gp function in MDR cancer cells which led to increased intracellular chemotherapeutic agents to kill the MDR


Figure 5 The effects of 29d, $\mathbf{1 3}$ and $\mathrm{NOB}(\mathrm{N})$ on the apoptosis and cell cycle of HCT8/T induced by PTX. (A) 29d, $\mathbf{1 3}$ and NOB at $9 \mu \mathrm{~mol} / \mathrm{L}$ increased the proportion of apoptosis induced by $0.94 \mu \mathrm{~mol} / \mathrm{L}$ PTX. (B) Addition of $\mathbf{2 9 d}$, $\mathbf{1 3}$ or NOB to medium containing $0.94 \mu \mathrm{~mol} / \mathrm{L}$ PTX increased the percentage of cells in G2M phase, as exhibited by flow cytometry. The data were obtained from three different experiments and shown as mean $\pm \mathrm{SD}(n=3) .{ }^{* *} P<0.01,{ }^{* * *} P<0.001$ vs. absence of 29d, 13 or N .
cancer, and 29d exhibited a stronger inhibition effect on the function of P-gp than $\mathbf{1 3}$ and NOB.

### 2.7. Putative binding pattern of 29d to the pocket of $P-g p$

To show the putative binding pattern, 29d was docked by the Surflex-dock module embedded in Tripos Sybyl X 2.0 (St. Louis, USA) to the crystal structure of P-gp in complex with QZ59-RRR (PDB ID: 4M2S). The selected binding pose of 29d fitted the binding pocket well, with Surflex docking score (Total-Score) of 6.2274. In this binding modes (Fig. 8 and Supporting Information Section 1), several kinds of specific interactions between 29d and P-gp were observed. Inhibitor 29d formed two H-bonds with Tyr306 and Gln721, respectively. Strong $\pi-\pi$ and $\pi-\sigma$ interactions were also observed for 29d with residues Phe979, Phe339, Phe724 and Phe728. In addition to van der Waals interactions and hydrophobic interactions with residues of Phe299, Gln986, Ile302 and Ser725, the dimethylaminoethyl sidechain of 29d formed extra interactions with Leu335 and Phe331, which significantly differed from $\mathrm{NOB}^{10}$.

Subsequently, molecular dynamics (MD) simulations were applied for a more precise prediction of binding patterns using AMBER 14. As a result, the predicted binding free energies of the $\mathrm{P}-\mathrm{gp} / \mathbf{2 9 d}$ and $\mathrm{P}-\mathrm{gp} / \mathrm{NOB}$ complexes were $-43.99 \pm 3.12$ and $-39.47 \pm 2.62 \mathrm{kcal} / \mathrm{mol}$, respectively. From the MD simulation results, apparent hydrophobic interactions with residues Phe299, Phe724, Phe728, Phe979 and Leu336 in P-gp were observed by 29d and NOB. In addition, 29d and NOB formed hydrogen bonds with Gln721 and Tyr306, respectively. Nevertheless, dimethylaminoethyl chain of 29d formed extra van der Waal interactions with Phe339 and Ile302, which might account for its more negative binding free energy compared with NOB.

### 2.8. NOB and its derivatives overcome MDR via AKT/ERK/ NRF2 pathway

Multiple targets are characteristics of natural products and ultimately offer clinical benefits in comparison with the three generations of MDR inhibitors ${ }^{20}$. To explore the underlying


Figure 6 Effects of 29d, 13 and $\mathrm{NOB}(\mathrm{N})$ on intracellular accumulation of DOX and Flutax-2 in HCT8T cells. HCT8/T cells were treated with $5 \mu \mathrm{~mol} / \mathrm{L} \operatorname{DOX}(\mathrm{A})$ and (B) or $1 \mu \mathrm{~mol} / \mathrm{L}$ Flutax-2 (C) and (D) for 6 h in the absence or presence of $9 \mu \mathrm{~mol} / \mathrm{L} \mathbf{2 9 d}$, 13, NOB and $20 \mu \mathrm{~mol} / \mathrm{L}$ QND as indicated. Intracellular DOX or Flutax-2 were observed with a fluorescence microscope (B) and (D) and evaluated by measuring fluorescence with flow cytometry $(\mathrm{A})$ and $(\mathrm{C})$. The experiments were repeated 3 times, and representative images were presented.


Figure 7 The concentrations of PTX (A) and 29d (B) in MDR tumors after treatment with PTX alone or the combination of 29d and NOB (N). The concentrations of PTX or 29d in the tumor were evaluated with HPLC-MS, as described in contents. The data were shown as mean $\pm$ SD for each group $(n=6),{ }^{* *} P<0.01,{ }^{* * *} P<0.001 \mathrm{vs}$. PTX alone or 29d alone; ${ }^{\# \#} P<0.01 \mathrm{vs}$. PTX + NOB.
mechanisms for functions of NOB and its derivatives, we evaluated the signaling pathways (Fig. 9) of NRF2/PI3K/AKT and ERK, which had been shown to be closely associated with resistance to chemotherapy drugs ${ }^{16}$. Interestingly, we found that $N r f 2$, phosphorylated AKT and ERK were significantly reduced in HCT8/T cells after administration of NOB, 29d or 13, while the expression of total AKT and ERK remained the same (Fig. 9). These results indicated that the MDR reversal effects of 29d and 13 to chemotherapy agents were produced by the inhibition of $N r f 2 / \mathrm{PI} 3 \mathrm{~K} / \mathrm{AKT}$ pathways, the same as NOB.

Change of sensitivity of the MDR cancer cells to drugs could be due to reduction of ABCB 1 expression, competitive inhibition of the ABCB1 transporter, or both. Therefore, we evaluated the effects of synthesized NOB, 29d and $\mathbf{1 3}$ on the expression of


Figure 8 Binding patterns of 29d (A-D, yellow in C and D ) and NOB ( E and F , grey) with $\mathrm{P}-\mathrm{gp}$ (cyan) by docking and MD simulation methods. (A) Ribbon diagram of P-gp with docked 29d (green). (B) Two-dimensional interaction mode between docked 29d and P-gp. The interactions between 29d and P-gp (green sticks). Hydrogen bonds are depicted by dashed blue. Hydrogen bonds and $\pi-\pi / \sigma-\pi$ interactions are shown by green lines. Green and purple bubbles represent hydrophobic and polar amino acid residues, respectively. (C) and (E) refer to the binding patterns by docking method while (D) and (F) refer to the binding patterns by MD method. Therein, hydrogen bonds are depicted by dashed blue lines.

ABCB 1 . Interestingly, none of them at the reversal concentrations alter the protein level of ABCB 1 in HCT8/T cells (Fig. 9). In addition, NOB and its derivatives did not affect the expression of BCRP and MDR1 in A549/T cells (Fig. 10) which are two major efflux transporters related to the PTX resistance. Thus, synthesized NOB and its derivatives likely inhibited $A B C B 1$ transporter function but did not affect the expression of ABCB 1 .

Several metabolites of NOB including 3'-demethylnobiletin, $4^{\prime}$-demethylnobiletin, $\quad 3^{\prime}, 4^{\prime}$-didemethylnobiletin (DTF), 5-
demethylnobiletin, 5, $3^{\prime}$-didemethylnobiletin, 5,4'-didemethylnobiletin and 5, $3^{\prime}, 4^{\prime}$-tridemethylnobiletin have been identified as the major metabolites from the urine of mouse by the optimized HPLC method ${ }^{21-22}$. Anti-inflammatory effects of $4^{\prime}$-demethylnobiletin on TPA-induced mice ear inflammation have been shown through inhibition of PI3K/AKT/ERK phosphorylation ${ }^{22}$. Therefore, the metabolites of NOB derivatives possibly have inhibitory activity on P-gp function via the NRF2/PI3K/AKT pathways.


Figure 9 Effects of the combination treatment of PTX with 29d, 13 and NOB (N) on expressions of P-gp, AKT, ERK and NRF2 in HCT8/T cells. Cells were treated with $9 \mu \mathrm{~mol} / \mathrm{L} 29 \mathrm{~d}, \mathbf{1 3}$ and N for 48 h . Equal amounts of total lysate were used for Western blot. Combination treatment of PTX with 29d, $\mathbf{1 3}$ or NOB did not influence P-gp expression, but reduced the level of NRF2 as well as the phosphorylation of AKT/ERK. The experiments were repeated three times. The data were shown as mean $\pm \mathrm{SD} .{ }^{*} P<0.05$ and ${ }^{* *} P<0.01$.


Figure 10 The expression of efflux transporters P-gp, MRP2 and BCRP in A549/T and HCT8/T cells, and the effects of 29d and NOB (N) on their expression by comparing with sensitive A549 or HCT8 cells (red frame). Cells were treated with $9 \mu \mathrm{~mol} / \mathrm{L} \mathbf{2 9 d}$ or N for 48 h . Equal amounts of total lysate were loaded and detected by Western blot.

## 3. Conclusions

In short, NOB was produced with a high yield by total synthesis of six steps and fourteen derivatives were synthesized to discover potent compounds with remarkable solubility and efficacy as MDR reversal agents. Among them, 29d showed 280-fold higher water solubility than NOB and $\mathbf{1 3}$. As a result, 29d significantly increased PTX concentration in the tumor in vivo and showed a stronger tumor growth inhibition than NOB and 13 after coadministered with PTX. Moreover, our studies demonstrated that the activated NRF2/PI3K/AKT pathways in MDR cancer cells were remarkably inhibited by the conjunction of $\mathbf{2 9 d}$ and PTX.

## 4. Experimental

### 4.1. Chemicals and reagents

PTX, DOX, verapamil (Ver), QND, rhodamine 123 (Rho123), dimethyl sulfoxide (DMSO), propidium iodide (PI), trichloroacetic acid (TCA), crystal violet, sulforhodamine B (SRB) and
other chemicals were purchased from Sigma-Aldrich (St. Louis, USA). The flutax-2, RPMI 1640 medium, fetal bovine serum, penicillin and streptomycin were obtained from Life Technologies, Inc. (Grand Island, USA). P-gp antibodies were purchased from Calbiochem (cat. No. 517310). ERK1/2 and actin antibodies were purchased from Santa Cruz Biotechnology, USA. AKT, PAKT and P-ERK1/2 were purchased from Cell Signaling Technology, Inc. (Boston, USA).

All starting materials and reagents were purchased from commercial suppliers (Sigma-Aldrich, Adamas, Energy, Bide, ShuYa, J\&K and Meryer, Shanghai, China) and used directly without further purification. Chemical HG/T2354-92 silica gel (200-300 mesh, Haiyang ${ }^{\circledR}$, Qingdao, China) was used for chromatography, and silica gel plates with fluorescence F254 ( 0.25 mm , Huanghai ${ }^{\circledR}$, Qingdao, China) were used for thin-layer chromatography (TLC) analysis. Reactions requiring anhydrous conditions were performed under argon or with a calcium chloride tube. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded at room temperature (rt) on a Bruker AVANCE III 400 instrument (Germany) with tetramethylsilane (TMS) as an internal standard. The following abbreviations are used: s (singlet), br s (broad singlet), $d$ (doublet), dd (doublet of doublets), $t$ (triplet), $q$ (quartet) and $m$ (multiplet). Coupling constants were reported in Hz. Highresolution mass spectra (HRMS) were recorded on SHIMADZU LCMS-IT-TOF (Kyoto, Japan). The purities of compounds were determined by reverse-phase HPLC analysis, confirming purity to exceed $95 \%$. HPLC instrument: SHIMADZU LC-20AT (column: Hypersil BDS C18, $5.0 \mu \mathrm{~m}, 150 \mathrm{~mm} \times 4.6 \mathrm{~mm}$ (Elite); detector: SPD-20A UV/VIS detector, UV detection at 254 nm ; elution, MeOH in water $(80 \%, v / v) ; \mathrm{T}=25^{\circ} \mathrm{C}$; flow rate $=1.0 \mathrm{~mL} / \mathrm{min}$; Kyoto, Japan).

### 4.1.1. 3,4,5-Trimethoxyphenyl acetate (2)

A mixture of $\mathbf{1}(9.2 \mathrm{~g}, 50 \mathrm{mmol})$ and sodium acetate $(8.2 \mathrm{~g}$, $100 \mathrm{mmol})$ in acetic anhydride ( $47 \mathrm{~mL}, 500 \mathrm{mmol}$ ) was heated at $110{ }^{\circ} \mathrm{C}$ for 2 h . TLC analysis (petroleum ethe/ethyl acetate $=7: 3$ ) showed completion of the reaction. Then the reaction mixture was concentrated under reduced pressure, diluted with water and extracted with ethyl acetate thrice. The organic phase was washed
with brine, dried over anhydrous sodium sulfate. The organic layer was concentrated to afford the product $2(11.2 \mathrm{~g})$ as a white solid, which was used directly in the next step without further purification. Yield: $91 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 6.34(\mathrm{~s}, 2 \mathrm{H})$, $3.83(\mathrm{~s}, 9 \mathrm{H}), 2.29(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.2. 1-(6-Hydroxy-2, 3, 4-trimethoxyphenyl) ethanone (3)

Boron trifluoride etherate ( $c a .48 \% \mathrm{BF}_{3}, 250 \mathrm{~mL}$ ) was added to a solution of $2(11.2 \mathrm{~g}, 50 \mathrm{mmol})$ in glacial acetic acid ( 37.5 mL ). The reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 2 h . TLC analysis (petroleum ether/ethyl acetate $=7: 3$ ) showed completion of reaction. Then the reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=10: 1$ ) to afford product $\mathbf{3}(10.2 \mathrm{~g})$ as a yellow oil. Yield: $91 \%$. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right): \delta 13.44(\mathrm{~s}, 1 \mathrm{H}), 6.24(\mathrm{~s}, 1 \mathrm{H})$, $3.99(\mathrm{~s}, 3 \mathrm{H}), 3.89(\mathrm{~s}, 3 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 2.66(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.3. 1-(2-Hydroxy-3-iodo-4,5,6-trimethoxyphenyl) ethanone (4)

Adding NIS $(1.4 \mathrm{~g}, 6.0 \mathrm{mmol})$ and $p$-toluenesulfonic acid monohydrate ( $950 \mathrm{mg}, 5.0 \mathrm{mmol}$ ) to a stirred solution of $\mathbf{3}(1.1 \mathrm{~g}$, $5.0 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(50 \mathrm{~mL})$. The mixture was stirred at rt for 2 h until the start material disappeared as monitored by TLC. After evaporated under vacuum, the reaction mixture was extracted with ethyl acetate. The organic phase was washed with saturated aqueous sodium thiosulfate and brine, and dried over anhydrous sodium sulfate. The solvent was evaporated to afford product $\mathbf{4}$ as a yellow-green solid, which was used directly in the next step without further purification. Yield: $73 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 14.00(\mathrm{~s}, 1 \mathrm{H}), 4.02(\mathrm{~d}, J=4.6 \mathrm{~Hz}, 3 \mathrm{H}), 4.01$ $(\mathrm{s}, 3 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 2.70(\mathrm{~s}, 3 \mathrm{H})$.
4.1.4. 1-(2-Hydroxy-3,4,5,6-tetramethoxyphenyl) ethanone (5)

Copper chloride ( $359 \mathrm{mg}, 3.6 \mathrm{mmol}$ ) was added to a solution of 4 ( $3.6 \mathrm{~g}, 10.2 \mathrm{mmol}$ ) in DMF ( 30 mL ) under an atmosphere of Ar at rt. Then freshly prepared $4.0 \mathrm{~mol} / \mathrm{L}$ solution of sodium methoxide in methanol ( $25 \mathrm{~mL}, 102 \mathrm{mmol}$ ) was added to the mixture at $90^{\circ} \mathrm{C}$ for 20 min . After cooling, the solution was poured into ice and acidified with $5.0 \mathrm{~mol} / \mathrm{L} \mathrm{HCl}$. The organic layer was extracted with ethyl acetate. The combined organic extracts were washed with brine and dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=10: 1)$ to afford product $5(1.5 \mathrm{~g})$ as a light yellow-green liquid. Yield: $59 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 13.16(\mathrm{~s}, 1 \mathrm{H})$, $4.08(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.86(\mathrm{~s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.68(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.5. General procedure for synthesis of compounds $\boldsymbol{\sigma} \boldsymbol{a}-\boldsymbol{\sigma} \boldsymbol{c}$

To a solution of 5 ( $128 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) in dichloromethane $(2.5 \mathrm{~mL})$ was added triethylamine ( $208 \mu \mathrm{~L}, 1.5 \mathrm{mmol}$ ) and acyl chlorides $(0.65 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The mixture was stirred at rt for 2 h . Then the reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford the product as white solids.
4.1.5.1. 2-Acetyl-3,4,5,6-tetramethoxyphenyl benzo[d][1,3] dioxole-5-carboxylate ( 6 a). White solid. Yield: $85 \%$. ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}, \mathrm{CDCl} 3) \delta 7.78(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{~s}, 1 \mathrm{H}), 6.91$
(d, $J=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.07(\mathrm{~s}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.90$ $(\mathrm{s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H})$.
4.1.5.2. 2-Acetyl-3,4,5,6-tetramethoxyphenyl 3,4dimethoxybenzoate ( $\boldsymbol{6} \boldsymbol{b}$ ). White solid. Yield: $80 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.82(\mathrm{dd}, J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.63(\mathrm{~d}$, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}$, $3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.47(\mathrm{~s}$, $3 \mathrm{H})$.
4.1.5.3. 2-Acetyl-3,4,5,6-tetramethoxyphenyl 3,4,5trimethoxybenzoate ( $\boldsymbol{\sigma c}$ ). White solid. Yield: $90 \% .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41(\mathrm{~s}, 1 \mathrm{H}), 7.39(\mathrm{~s}, 1 \mathrm{H}), 3.99(\mathrm{~s}, 4 \mathrm{H}), 3.96$ (s, 3H), 3.95 (s, 3H), 3.94 (s, 3H), 3.93 ( $\mathrm{s}, 6 \mathrm{H}$ ), 3.90 (s, 3H), 3.83 ( $\mathrm{s}, 3 \mathrm{H}$ ), 2.49 ( $\mathrm{s}, 3 \mathrm{H}$ ).

### 4.1.6. General procedure for synthesis of compounds $7 \boldsymbol{a}-7 \boldsymbol{c}$

 To a solution of $6(0.4 \mathrm{mmol})$ in dichloroethane $(2.4 \mathrm{~mL})$ was added triethylamine ( $166 \mu \mathrm{~L}, 1.2 \mathrm{mmol}$ ) and trimethylsilyl trifluoromethanesulfonate ( $434 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ). The mixture was stirred at $95^{\circ} \mathrm{C}$ for 2 h , then quenched with $\mathrm{CH}_{3} \mathrm{OH}$, and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=2: 1$ ) to afford the product.4.1.6.1. 2-(Benzo[d][1,3]dioxol-5-yl)-5,6,7,8-tetramethoxy-4H-chromen-4-one (7a). White solid. Yield: $75 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{MeOD}-d_{4}$ ) $\delta 7.62(\mathrm{dd}, J=8.3,1.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.47(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.66(\mathrm{~s}, 1 \mathrm{H})$, $6.11(\mathrm{~s}, 2 \mathrm{H}), 4.12(\mathrm{~s}, 3 \mathrm{H}), 4.03(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.33,160.86,151.46,150.51$, $148.50,148.38,147.67,144.13,138.07,125.54,121.15,114.86$, $108.85,107.12,106.12,101.90,62.28,62.07,61.8,61.68$. HRMS (ESI) $\mathrm{m} / \mathrm{z}$ Calcd. $\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{O}_{8}^{+} \quad[\mathrm{M}+\mathrm{H}]^{+}$387.1074, Found 387.1073.
4.1.6.2. 2-(3,4-Dimethoxyphenyl)-5,6,7,8-tetramethoxy-4Hchromenone (7b). White solid. Yield: $78 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.57(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.42$ (s, $1 \mathrm{H}), 7.00(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.62(\mathrm{~s}, 1 \mathrm{H}), 4.11(\mathrm{~s}, 3 \mathrm{H}), 4.03(\mathrm{~s}$, $3 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 177.35,161.04,151.92,151.43,149.28$, $148.42,147.72,144.08,138.01,124.00,119.62,114.85,111.22$, $108.53,106.88,62.27,61.98,61.84,61.69,56.09,55.97$. HRMS (ESI) m/z Calcd. $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{8}^{+}[\mathrm{M}+\mathrm{H}]^{+}$403.1389, Found 403.1389.
4.1.6.3. 5,6,7,8-Tetramethoxy-2-(3,4,5-trimethoxyphenyl)-4H-chromen-4-one (7c). White solid. Yield: $80 \%$. Purity: $96 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.17(\mathrm{~s}, 2 \mathrm{H}), 6.64(\mathrm{~s}, 1 \mathrm{H}), 4.11(\mathrm{~s}, 3 \mathrm{H})$, $4.03(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}, 12 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 177.31,160.80,153.59,151.55,148.44,147.70,144.16$, $141.07,137.98,126.69,114.82,107.63,103.67,103.39 \times 2$, $62.26,61.89,61.81,61.67,61.03,56.24 \times 2$. HR-MS (ESI) $m / z$ Calcd. $\mathrm{C}_{22} \mathrm{H}_{25} \mathrm{O}_{9}^{+}[\mathrm{M}+\mathrm{H}]^{+}$433.1493, Found 433.1483.

### 4.1.7. 2-Acetyl-3,4,5-trimethoxyphenyl 3,4-dimethoxybenzoate (10)

To a solution of $\mathbf{3}$ ( $904 \mathrm{mg}, 4.0 \mathrm{mmol}$ ) in dichloromethane $(16 \mathrm{~mL})$ was added triethylamine $(1.47 \mathrm{~mL}, 12 \mathrm{mmol})$ and $3,4-$
dimethoxybenzoyl chloride ( $1.04 \mathrm{~g}, 5.2 \mathrm{mmol}$ ) at $0{ }^{\circ} \mathrm{C}$. The mixture was stirred at rt for 2 h , then quenched with water, and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford product $\mathbf{1 0}$ as a white solid. Yield: $85 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.81$ (dd, $\left.J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 7.63(\mathrm{~d}$, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{~s}, 1 \mathrm{H}), 3.98(\mathrm{~s}$, $3 \mathrm{H}), 3.97$ ( $\mathrm{s}, 6 \mathrm{H}$ ), 3.91 ( $\mathrm{s}, 3 \mathrm{H}$ ), 3.90 ( $\mathrm{s}, 3 \mathrm{H}$ ), 2.51 ( $\mathrm{s}, 3 \mathrm{H})$.

### 4.1.8. 1-(3-Bromo-2-hydroxy-4,5,6-trimethoxyphenyl) ethanone (11)

Sodium acetate ( $1.77 \mathrm{~g}, 21.6 \mathrm{mmol}$ ) was added to a solution of $\mathbf{3}$ $(4.436 \mathrm{~g}, 19.6 \mathrm{mmol})$ in glacial acetic acid $(15 \mathrm{~mL})$. Then $\mathrm{Br}_{2}$ in glacial acetic acid ( $1.5 \mathrm{~mL}, 29.4 \mathrm{mmol}$ ) was added dropwise to the reaction mixture at $0{ }^{\circ} \mathrm{C}$. The mixture was stirred at rt for 12 h and poured into ice water $(150 \mathrm{~mL})$. The solid was filtered and washed with water to yield product $\mathbf{1 1}$ as a yellow solid, which was used directly in the next step without further purification. Yield: $81 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.75(\mathrm{~s}, 1 \mathrm{H}), 4.05(\mathrm{~s}, 3 \mathrm{H}), 4.04$ ( s , $3 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 2.72(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.9. 1-(2-Hydroxy-4,5,6-trimethoxy-3-nitrophenyl) ethanone

 (12)Nitric acid (ca. $70 \% \mathrm{HNO}_{3}, 621 \mu \mathrm{~L}, 14 \mathrm{mmol}$ ) was added dropwise to $3(1.58 \mathrm{~g}, 7 \mathrm{mmol})$ over 15 min at $0^{\circ} \mathrm{C}$. Then the reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford product $\mathbf{1 2}$ as a yellow solid. Yield: $79 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.56$ ( s , $1 \mathrm{H}), 4.10(\mathrm{~s}, 3 \mathrm{H}), 4.07(\mathrm{~s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 2.69(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.10. 2-(3,4-Dimethoxyphenyl)-5,6,7-trimethoxy-4H-chromen-4-one (13)

To a solution of $\mathbf{1 0}(1.17 \mathrm{~g}, 3.0 \mathrm{mmol})$ in dichloroethane ( 18 mL ) was added triethylamine ( $1.2 \mathrm{~mL}, 9.0 \mathrm{mmol}$ ) and trimethylsilyl trifluoromethanesulfonate ( $3.3 \mathrm{~mL}, 18 \mathrm{mmol}$ ). The mixture was then stirred at $95^{\circ} \mathrm{C}$ for 2 h , quenched with $\mathrm{CH}_{3} \mathrm{OH}$, and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=2: 1$ ) to afford product $\mathbf{1 3}(905 \mathrm{mg})$ as a brown solid. Yield: $81 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.51(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.33(\mathrm{~s}, 1 \mathrm{H}), 6.98(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~s}, 1 \mathrm{H}), 6.60(\mathrm{~s}, 1 \mathrm{H})$, $4.00(\mathrm{~s}, 6 \mathrm{H}), 3.99(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 177.28,161.17,157.68,154.51,152.57$, $151.79,149.25,140.35,124.08,119.59,112.85,111.10,108.60$, 107.37, 96.26, 62.21, 61.57, 56.33, 56.12, 56.08. HR-MS (ESI) ml $z$ Calcd. $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{O}_{7}^{+}[\mathrm{M}+\mathrm{H}]^{+}$373.1282, Found 373.1285.

### 4.1.11. 2-Acetyl-6-bromo-3,4,5-trimethoxyphenyl 3,4dimethoxybenzoate (14)

To a solution of $\mathbf{1 1}(1.2 \mathrm{~g}, 4.0 \mathrm{mmol})$ in dichloromethane ( 20 mL ) was added triethylamine ( $1.4 \mathrm{~mL}, 12 \mathrm{mmol}$ ) and $3,4-$ dimethoxybenzoyl chloride ( $1.0 \mathrm{~g}, 5.2 \mathrm{mmol}$ ) at $0{ }^{\circ} \mathrm{C}$. The mixture was stirred at rt for 2 h , quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ )
to afford product 14 as a yellow solid. Yield: $75 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.85(\mathrm{dd}, J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.64(\mathrm{~d}$, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}$, $3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 2.49(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.12. 2-Acetyl-3,4,5-trimethoxy-6-nitrophenyl 3,4dimethoxybenzoate (15)

To a solution of $\mathbf{1 2}(1.2 \mathrm{~g}, 5.0 \mathrm{mmol})$ in dichloromethane ( 25 mL ) was added triethylamine ( $1.8 \mathrm{~mL}, 15 \mathrm{mmol}$ ) and 3,4dimethoxybenzoyl chloride ( $1.2 \mathrm{~g}, 6.0 \mathrm{mmol}$ ) at $0{ }^{\circ} \mathrm{C}$. The mixture was stirred at rt for 2 h , quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford product 15 as a beige solid. Yield: $90 \%$. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.73(\mathrm{dd}, J=8.5,2.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.52(\mathrm{~d}$, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.06(\mathrm{~s}, 3 \mathrm{H}), 4.02(\mathrm{~s}$, $3 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 2.52(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.13. 8-Bromo-2-(3,4-dimethoxyphenyl)-5,6,7-trimethoxy-4H-chromen-4-one (16)

To a solution of $\mathbf{1 4}(1.2 \mathrm{~g}, 2.6 \mathrm{mmol})$ in dichloroethane ( 15 mL ) was added triethylamine ( $1.0 \mathrm{~mL}, 7.5 \mathrm{mmol}$ ) and trimethylsilyl trifluoromethanesulfonate ( $2.7 \mathrm{~mL}, 15 \mathrm{mmol}$ ). The mixture was stirred at $95^{\circ} \mathrm{C}$ for 2 h , then quenched with $\mathrm{CH}_{3} \mathrm{OH}$ and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=1: 1$ ) to afford product $\mathbf{1 6}(900 \mathrm{mg})$ as a white solid. Yield: $79 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.64(\mathrm{dd}, J=8.5,2.1 \mathrm{~Hz}$, $1 \mathrm{H}), 7.53(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~s}$, $1 \mathrm{H}), 4.09(\mathrm{~s}, 3 \mathrm{H}), 4.00(\mathrm{~s}, 3 \mathrm{H}), 3.99(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}$, $3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.00,161.45,155.77$, 152.70, 152.07, 150.43, 149.28, 144.61, 123.50, 119.94, 115.81, 111.21, 108.78, 106.53, 101.43, 62.34, 61.84, 61.58, 56.10, 56.00. HR-MS (ESI) $m / z$ Calcd. $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{O}_{7} \mathrm{Br}^{+}[\mathrm{M}+\mathrm{H}]^{+} 451.0387$, Found 451.0392.

### 4.1.14. 2-(3,4-Dimethoxyphenyl)-5,6,7-trimethoxy-8-nitro-4H-chromen-4-one (17)

To a solution of $\mathbf{1 5}(1.9 \mathrm{~g}, 4.3 \mathrm{mmol})$ in dichloroethane ( 26 mL ) was added triethylamine ( $1.7 \mathrm{~mL}, 13 \mathrm{mmol}$ ) and trimethylsilyl trifluoromethanesulfonate ( $4.6 \mathrm{~mL}, 256 \mathrm{mmol}$ ). The mixture was stirred at $95^{\circ} \mathrm{C}$ for 2 h , then quenched with $\mathrm{CH}_{3} \mathrm{OH}$ and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=1: 1$ ) to afford product $17(1.5 \mathrm{~g})$ as a brown solid. Yield: $84 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.42$ (dd, $J=8.5,2.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.28(\mathrm{~d}$, $J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}), 4.17(\mathrm{~s}$, $3 \mathrm{H}), 4.05(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.15. 8-Amino-2-(3,4-dimethoxyphenyl)-5,6,7-trimethoxy-4H-chromen-4-one (18)

To a solution of 17 ( $140 \mathrm{mg}, 0.33 \mathrm{mmol}$ ) in tetrahydrofuran $(25 \mathrm{~mL})$ was added $\mathrm{Pd} / \mathrm{C}$ catalyst ( $21 \mathrm{mg}, 15 \%$ ) under an atmosphere of $\mathrm{H}_{2}$ at $40^{\circ} \mathrm{C}$ for 15 h . After cooling to rt , the mixture was filtered with diatomite and washed with ethyl acetate. The organic layer was evaporated under vacuum to afford a residue, which was purified by silica gel column chromatography (petroleum ether/
ethyl acetate $=2: 1)$ to yield product $18(121 \mathrm{mg})$ as a yellow solid. Yield: $95 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50(\mathrm{dd}, J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.98$ $(\mathrm{d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.57(\mathrm{~s}, 1 \mathrm{H}), 4.04(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.97$ (s, 3H), 3.96 (s, 3H), 3.91 (s, 3H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.80,160.99,151.86,149.31,144.13,143.75,143.40,141.82$, 126.01, 124.35, 119.56, 114.93, 111.17, 108.72, 107.33, 62.35, $61.61,60.91,56.10 \times 2$. HR-MS (ESI) $m / z$ Calcd. $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{NO}_{7}^{+}$ $[\mathrm{M}+\mathrm{H}]^{+}$388.1391, Found 388.1391.

### 4.1.16. 2,3,4-Trimethoxyphenyl acetate (20b)

A mixture of $\mathbf{1 9 b}(5.5 \mathrm{~g}, 30 \mathrm{mmol})$ and sodium acetate $(4.9 \mathrm{~g}$, $60 \mathrm{mmol})$ in acetic anhydride ( $28 \mathrm{~mL}, 300 \mathrm{mmol}$ ) was heated at $110{ }^{\circ} \mathrm{C}$ for 2 h . TLC analysis (petroleum ether/ethyl acetate $=9: 1$ ) showed completion of the reaction. Then the reaction mixture was concentrated under reduced pressure, diluted with water and extracted with ethyl acetate thrice. The organic phase was washed with brine, dried over anhydrous sodium sulfate. The organic layer was concentrated to afford product 20b $(6.4 \mathrm{~g})$ as a brown yellow oil, which was used directly in the next step without further purification. Yield: $94 \%$. ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 6.74(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.62(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.89$ $(\mathrm{s}, 3 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 2.31(\mathrm{~s}, 3 \mathrm{H})$.
4.1.17. 1-(2-Hydroxy-3,4-dimethoxyphenyl) ethan-1-one (21a) A mixture of $19 \mathrm{a}(1.5 \mathrm{~g}, 10 \mathrm{mmol})$ and sodium acetate $(1.6 \mathrm{~g}$, $20 \mathrm{mmol})$ in acetic anhydride ( $9.4 \mathrm{~mL}, 100 \mathrm{mmol}$ ) was heated at $110{ }^{\circ} \mathrm{C}$ for 2 h . TLC analysis (petroleum ether/ethyl acetate $=9: 1$ ) showed completion of the reaction. Then the reaction mixture was concentrated under reduced pressure, diluted with water and extracted with ethyl acetate thrice. The organic phase was washed with brine and dried over anhydrous sodium sulfate. The organic layer was concentrated to afford product 20a $(1.9 \mathrm{~g})$ as a red brown oil, which was used directly in the next step without further purification. Then boron trifluoride etherate ( $c a$. $48 \% \mathrm{BF}_{3}, 48.5 \mathrm{~mL}$ ) was added dropwise to a solution of compound $\mathbf{2 0 a}(1.9 \mathrm{~g}, 9.7 \mathrm{mmol})$ in glacial acetic acid $(7.5 \mathrm{~mL})$. The reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 2 h . TLC analysis (petroleum ether/ethyl acetate $=7: 3$ ) showed completion of reaction. Then the reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=10: 1$ ) to afford product 21a ( 1.8 g ) as a brown solid. Yield: $95 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 12.67(\mathrm{~s}, 1 \mathrm{H}), 7.09(\mathrm{~s}, 1 \mathrm{H}), 6.48(\mathrm{~s}, 1 \mathrm{H}), 3.94$ $(\mathrm{s}, 3 \mathrm{H}), 3.89(\mathrm{~s}, 3 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.18. 1-(2-Hydroxy-3,4,5-trimethoxyphenyl) ethan-1-one

 (21b)Boron trifluoride etherate ( $c a .48 \% \mathrm{BF}_{3}, 140 \mathrm{~mL}$ ) was added to a solution of $\mathbf{2 0 b}(6.4 \mathrm{~g}, 28 \mathrm{mmol})$ in glacial acetic acid ( 21 mL ). The reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 2 h . TLC analysis (petroleum ether/ethyl acetate $=7: 3$ ) showed completion of reaction. Then the reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=10: 1)$ to afford product $\mathbf{2 1 b}(6.0 \mathrm{~g})$ as a yellow solid. Yield: $95 \%$. ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 12.44$ (s, 1H), 6.93 (s, 1 H ), $4.04(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H})$.
4.1.19. General procedure for synthesis of compounds 22a-22d To a solution of compound 21a-21d ( $196 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) in dichloromethane ( 5 mL ) was added triethylamine ( $418 \mu \mathrm{~L}$, $3.0 \mathrm{mmol})$ and acyl chloride $(1.5 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The mixture was stirred at rt for 2 h , then quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford the products 22a-22d as white solids.
4.1.19.1. 6-Acetyl-2,3-dimethoxyphenyl 3,4-dimethoxybenzoate (22a). Yield: $78 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.90$ (dd, $J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.69(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{~s}, 1 \mathrm{H}), 6.99$ (d, $J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~s}, 1 \mathrm{H}), 3.99(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93$ $(\mathrm{s}, 3 \mathrm{H}), 2.51(\mathrm{~s}, 3 \mathrm{H})$.
4.1.19.2. 6-Acetyl-2,3-dimethoxyphenyl 3-methoxybenzoate (22b). Yield: $82 \% .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.83$ (d, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~s}, 1 \mathrm{H}), 7.48-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{dd}$, $J=8.3,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.68(\mathrm{~s}, 1 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.89$ $(\mathrm{s}, 3 \mathrm{H}), 2.51(\mathrm{~s}, 3 \mathrm{H})$.
4.1.19.3. 6-Acetyl-2,3-dimethoxyphenyl 4-methoxybenzoate (22c). Yield: $82 \% .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.18$ (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{~s}, 1 \mathrm{H}), 7.02(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.67(\mathrm{~s}$, $1 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 2.50(\mathrm{~s}, 3 \mathrm{H})$.
4.1.19.4. 6-Acetyl-2,3,4-trimethoxyphenyl 3,4-dimethoxybenzoate (22d). Yield: $82 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.93$ (dd, $J=8.4,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{~s}, 1 \mathrm{H}), 7.00$ $(\mathrm{d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.01(\mathrm{~s}, 3 \mathrm{H}), 3.99(\mathrm{~s}, 3 \mathrm{H}), 3.99(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{~s}$, $3 \mathrm{H}), 3.87$ (s, 3H), $2.53(\mathrm{~s}, 3 \mathrm{H})$.
4.1.20. General procedure for synthesis of compounds 23a-23d To a solution of $22(0.8 \mathrm{mmol})$ in dichloroethane $(4.8 \mathrm{~mL})$ was added triethylamine ( $332 \mu \mathrm{~L}, 2.4 \mathrm{mmol}$ ) and trimethylsilyl trifluoromethanesulfonate ( $868 \mu \mathrm{~L}, 4.8 \mathrm{mmol}$ ). The mixture was stirred at $95^{\circ} \mathrm{C}$ for 2 h , then quenched with $\mathrm{CH}_{3} \mathrm{OH}$ and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=2: 1$ ) to afford the products 23a-23d.
4.1.20.1. 2-(3,4-Dimethoxyphenyl)-7,8-dimethoxy-4H-chromen-4-one (23a). Light yellow solid. Yield: $82 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.57(\mathrm{~s}, 1 \mathrm{H}), 7.54(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}$, $1 \mathrm{H}), 7.37(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{~d}, J=9.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.73(\mathrm{~s}$, $1 \mathrm{H}), 4.03(\mathrm{~s}, 3 \mathrm{H}), 3.99(\mathrm{~s}, 6 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 177.63,162.83,154.37,152.21,151.85,149.29,147.61$, $124.49,119.72,117.29,111.16,108.73,106.07,104.45,99.79$, 56.51, 56.38, 56.12, 56.09. HR-MS (ESI) $m / z$ Calcd. $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{O}_{6}^{+}$ $[\mathrm{M}+\mathrm{H}]^{+}$343.1176, Found 343.1166.
4.1.20.2. 7,8-Dimethoxy-2-(3-methoxyphenyl)-4H-chromen-4one (23b). Breen solid. Yield: $82 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.58(\mathrm{~s}, 1 \mathrm{H}), 7.50(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H})$, $7.44(\mathrm{dd}, J=9.0,6.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.08(\mathrm{dd}, J=8.0,2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $7.01(\mathrm{~s}, 1 \mathrm{H}), 6.79(\mathrm{~s}, 1 \mathrm{H}), 4.04(\mathrm{~s}, 3 \mathrm{H}), 4.01(\mathrm{~s}, 3 \mathrm{H}), 3.91$ $(\mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 177.66, 162.57, 160.01, 154.52, 152.30, 147.71, 133.31, 130.10, 118.53, $117.37,116.88,111.59,107.32,104.38,99.81,56.51,56.39$,
55.49. HR-MS (ESI) $m / z$ Calcd. $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{O}_{5}^{+}[\mathrm{M}+\mathrm{H}]^{+}$313.1071, Found 313.1072.
4.1.20.3. 7,8-Dimethoxy-2-(4-methoxyphenyl)-4H-chromen-4one (23c). Brick red solid. Yield: $82 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.86(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.56(\mathrm{~s}, 1 \mathrm{H}), 7.02(\mathrm{~d}$, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.99(\mathrm{~s}, 1 \mathrm{H}), 6.71(\mathrm{~s}, 1 \mathrm{H}), 4.02(\mathrm{~s}, 3 \mathrm{H}), 3.99(\mathrm{~s}$, $3 \mathrm{H}), 3.89(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.68,162.85$, $162.18,154.31,152.20,147.55,127.76 \times 2,124.27,117.28$, $114.43 \times 2,105.76,104.45,99.76,56.48,56.39,55.51$. HR-MS (ESI) $\mathrm{m} / \mathrm{z}$ Calcd. $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{O}_{5}^{+}[\mathrm{M}+\mathrm{H}]^{+}$313.1071, Found 313.1067.
4.1.20.4. 2-(3,4-Dimethoxyphenyl)-6,7,8-trimethoxy-4H-chro-men-4-one (23d). White solid. Yield: $83 \%$. Purity: $96 \%$. ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{MeOD}-d_{4}\right) \delta 7.68(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{~s}$, $1 \mathrm{H}), 7.36(\mathrm{~s}, 1 \mathrm{H}), 7.13(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~s}, 1 \mathrm{H}), 4.12(\mathrm{~s}$, 3 H ), $4.04(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 6 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 177.72,162.82,152.00,151.16,149.32,147.38,145.75$, $142.03,124.40,119.79 \times 2,111.25,108.69,105.64,100.08$, 62.02, 61.52, 56.30, 56.11, 55.99. HR-MS (ESI) $\mathrm{m} / \mathrm{z}$ Calcd. $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{O}_{7}^{+}[\mathrm{M}+\mathrm{H}]^{+}$373.1282, Found 373.1275.

### 4.1.21. 1-(2,3,4,5-Tetramethoxyphenyl)-ethan-1-one (24)

To a solution of $\mathbf{2 1 b}(2.3 \mathrm{~g}, 10 \mathrm{mmol})$ in DMF was added potassium carbonate ( $2.1 \mathrm{~g}, 15 \mathrm{mmol}$ ) and iodomethane ( $679 \mu \mathrm{~L}$, $11 \mathrm{mmol})$. The reaction mixture was stirred at rt for 1 h and then extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and concentrated to afford product $24(2.3 \mathrm{~g})$ as a brown yellow oil, which was used directly in the next step without further purification. Yield: $96 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.07(\mathrm{~s}, 1 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}$, $3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.86(\mathrm{~s}, 3 \mathrm{H}), 2.64(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.22. 1-(2,3,4,5-Tetramethoxy-6-nitrophenyl) ethan-1-one

 (25)Nitric acid (ca. $70 \% \mathrm{HNO}_{3}, 798 \mu \mathrm{~L}, 18 \mathrm{mmol}$ ) was added dropwise to $24(2.2 \mathrm{~g}, 9.0 \mathrm{mmol})$ over 15 min at $0^{\circ} \mathrm{C}$. Then the reaction mixture was quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford product $\mathbf{2 5}(1.9 \mathrm{~g})$ as a light yellow solid. Yield: $\mathbf{7 4 \%}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.00(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.88$ $(\mathrm{s}, 3 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.23. 1-(2-Amino-3,4,5,6-tetramethoxyphenyl) ethan-1-one

 (26)To a solution of $\mathbf{2 5}(1.9 \mathrm{~g}, 6.7 \mathrm{mmol})$ in tetrahydrofuran $(134 \mathrm{~mL})$ was added $\mathrm{Pd} / \mathrm{C}$ catalyst ( $285 \mathrm{mg}, 15 \%$ ) under an atmosphere of $\mathrm{H}_{2}$ at $40^{\circ} \mathrm{C}$ for 15 h . After cooling to rt , the mixture was filtered with diatomite and washed with ethyl acetate. The organic layer was evaporated under vacuum to afford a residue, which was purified by silica gel column chromatography (petroleum ether/ ethyl acetate $=2: 1$ ) to yield product $26(1.6 \mathrm{~g})$ as a brown yellow oil. Yield: $95 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.12(\mathrm{~s}, 3 \mathrm{H}), 4.03$ $(\mathrm{s}, 3 \mathrm{H}), 4.02(\mathrm{~s}, 3 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 2.83(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.24. $N$-(2-Acetyl-3,4,5,6-tetramethoxyphenyl)-3,4dimethoxybenzamide (27)

To a solution of $\mathbf{2 6}(570 \mathrm{mg}, 2.2 \mathrm{mmol})$ in dichloromethane ( 11 mL ) was added triethylamine ( $915 \mu \mathrm{~L}, 6.6 \mathrm{mmol}$ ) and $3,4-$
dimethoxybenzoyl chloride ( $883 \mathrm{mg}, 4.4 \mathrm{mmol}$ ) at $0{ }^{\circ} \mathrm{C}$. The mixture was stirred at rt for 2 h , then quenched with water and extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and purified by silica gel column chromatography (petroleum ether/ethyl acetate $=5: 1$ ) to afford product $27(738 \mathrm{mg})$ as a white solid. Yield: $80 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.24(\mathrm{~s}, 1 \mathrm{H}), 7.46(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{dd}$, $J=8.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~s}$, $3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}), 2.65(\mathrm{~s}, 3 \mathrm{H})$.

### 4.1.25. 2-(3,4-Dimethoxyphenyl)-5,6,7,8-tetramethoxyquinolin-4(1H)-one (28)

To a solution of $27(738 \mathrm{mg}, 1.8 \mathrm{mmol})$ in anhydrous tetrahydrofuran ( 60 mL ) was added potassium tert-butoxide ( 606 mg , 5.4 mmol ) under an atmosphere of Ar at $75^{\circ} \mathrm{C}$ for 18 h . After cooled to rt, the solution was poured into water and acidified by addition of $2.0 \mathrm{~mol} / \mathrm{L}$ aqueous HCl . The resulting solution was extracted with portion of ethyl acetate. The combined organic extracts were dried over anhydrous sodium sulfate and concentrated to give a crude, which was purified by silica gel column chromatography (petroleum ether/ethyl acetate $=3: 1$ ) to yield product $\mathbf{2 8}(592 \mathrm{mg})$ as a white solid. Yield: $82 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{~d}$, $J=8.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.72(\mathrm{~s}, 1 \mathrm{H}), 4.12(\mathrm{~s}, 3 \mathrm{H}), 4.09(\mathrm{~s}, 3 \mathrm{H}), 4.03(\mathrm{~s}$, $3 \mathrm{H}), 4.00(\mathrm{~s}, 6 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz , MeOD- $d_{4}$ ) $\delta 176.98,151.67,150.93,149.56,149.38,147.53,143.88,137.77$, 132.72, 125.99, 120.52, 114.51, 111.48, 110.73, 107.50, 61.46, 60.91, 60.88, 60.61, 55.25, 55.14. HR-MS (ESI) m/z Calcd. $\mathrm{C}_{21} \mathrm{H}_{24} \mathrm{NO}_{7}^{+}[\mathrm{M}+\mathrm{H}]^{+}$402.1547, Found 402.1542.

### 4.1.26. 2-(3,4-Dimethoxyphenyl)-5,6,7,8-tetramethoxy-1-methylquinolin-4(1H)-one (29a)

To a solution of $28(80 \mathrm{mg}, 0.2 \mathrm{mmol})$ in DMF was added potassium carbonate ( $41 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) and iodomethane ( $14 \mu \mathrm{~L}$, $0.22 \mathrm{mmol})$. The reaction mixture was stirred at rt for 1 h and then extracted with ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulfate, and concentrated to give a crude, which was purified by silica gel column chromatography (petroleum ether/ethyl acetate $=3: 1$ ) to yield product 29a ( 62 mg ) as a light yellow oil. Yield: $75 \%$. Purity: $99 \%$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.94(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.65(\mathrm{dd}$, $J=8.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{~s}, 1 \mathrm{H}), 6.98(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.20$ $(\mathrm{s}, 3 \mathrm{H}), 4.13(\mathrm{~s}, 3 \mathrm{H}), 4.11(\mathrm{~s}, 3 \mathrm{H}), 4.04(\mathrm{~s}, 3 \mathrm{H}), 4.01(\mathrm{~s}, 3 \mathrm{H}), 3.96$ $(\mathrm{s}, 3 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 163.70, $155.98,150.38,149.21,148.14,145.20,144.52,144.45,142.28$, $132.78,119.75,112.38,110.83,110.45,97.27,62.33,62.21$, $61.76 \times 2,56.01,55.96 \times 2$. HR-MS $(E S I) \mathrm{m} / \mathrm{z}$ Calcd. $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{NO}_{7}^{+}[\mathrm{M}+\mathrm{H}]^{+}$416.1704, Found 416.1710.
4.1.27. General procedure for synthesis of compounds 29b-29d To a solution of $\mathbf{2 8}(0.2 \mathrm{mmol})$ in anhydrous DMF $(2.0 \mathrm{~mL})$ was added sodium hydride ( $48 \mathrm{mg}, 2.0 \mathrm{mmol}$ ). The mixture was stirred at $90^{\circ} \mathrm{C}$ for 30 min and chloralkanes $(0.6 \mathrm{mmol})$ was added. Then the reaction mixture was further stirred at $90{ }^{\circ} \mathrm{C}$ for 2 h . After cooled to rt, the solution was poured into water and acidified by the addition of $2.0 \mathrm{~mol} / \mathrm{L}$ aqueous HCl . The resulting solution was extracted with portion of ethyl acetate. The combined organic extracts were washed with brine, dried over anhydrous sodium sulfate, and concentrated to give crude, which was purified by
silica gel column chromatography (petroleum ether/ethyl acetate $=3: 1$ ) to yield the oily products.
4.1.27.1. 2-(3,4-Dimethoxyphenyl)-5,6,7,8-tetramethoxy-1-propylquinolin- $4(1 \mathrm{H}$ )-one (29b). A light yellow oil. Yield: $62 \%$. Purity: $99 \%{ }^{1} \mathrm{H}$ NMR ( 400 MHz , MeOD- $d_{4}$ ) $\delta 7.89$ (d, $J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.68(\mathrm{dd}, J=8.4,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~s}, 1 \mathrm{H}), 7.06$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.25(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.11(\mathrm{~s}, 3 \mathrm{H}), 4.06$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $3.97(\mathrm{~s}, 3 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 2.01$ (dd, $J=13.9,6.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.19(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 167.39,160.83,154.56,153.11,152.03$, $149.02,148.70,148.09,145.85,136.38,124.05,116.32,115.07$, $114.73,102.11,74.21,65.30,65.23,64.67 \times 2,59.04,58.99$, 26.17, 13.65. HR-MS (ESI) $m / z$ Calcd. $\mathrm{C}_{24} \mathrm{H}_{30} \mathrm{NO}_{7}^{+}[\mathrm{M}+\mathrm{H}]^{+}$ 444.2017, Found 444.2019.
4.1.27.2. 2-(3,4-Dimethoxyphenyl)-5,6,7,8-tetramethoxy-1-(2-methoxyethyl)-quinolin-4(1H)-one (29c). A light yellow oil. Yield: $70 \%$. Purity: $99 \%$. ${ }^{1}$ H NMR ( $400 \mathrm{MHz}, \mathrm{MeOD}-d_{4}$ ) $\delta 7.90$ (d, $J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.71(\mathrm{dd}, J=8.4,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.28(\mathrm{~s}, 1 \mathrm{H}), 7.07$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{t}, J=4.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.11(\mathrm{~s}, 3 \mathrm{H}), 4.07(\mathrm{~s}$, $3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{t}, J=4.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H})$, $3.89(\mathrm{~s}, 3 \mathrm{H}), 3.49(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.05$, 160.80, 154.60, 153.12, 152.10, 149.12, 148.63, 148.11, 145.86, $136.30,124.09,116.27,115.08,114.74,102.22,74.50,71.91$, 65.37, 65.32, 64.69, 64.67, 61.80, 59.05, 59.00. HR-MS (ESI) $\mathrm{m} / \mathrm{z}$ Calcd. $\mathrm{C}_{24} \mathrm{H}_{30} \mathrm{NO}_{8}^{+}[\mathrm{M}+\mathrm{H}]^{+} 460.1966$, Found 460.1966 .
4.1.27.3. 2-(3,4-Dimethoxyphenyl)-1-(2-(dimethylamino)-ethyl)-5,6,7,8-tetramethoxyquinolin-4(1H)-one (29d). A brown oil. Yield: $71 \%$. Purity: $97.1 \%$. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{MeOD}-d_{4}\right)$ $\delta 7.94(\mathrm{~s}, 1 \mathrm{H}), 7.74(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{~s}, 1 \mathrm{H}), 7.08(\mathrm{~d}$, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.44(\mathrm{t}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.13(\mathrm{~s}, 3 \mathrm{H}), 4.08(\mathrm{~s}$, $3 \mathrm{H}), 3.99(\mathrm{~s}, 6 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.03(\mathrm{t}, J=5.5 \mathrm{~Hz}$, $2 \mathrm{H}), 2.48$ ( $\mathrm{s}, 6 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR ( 126 MHz , MeOD- $d_{4}$ ) $\delta 162.92$, $156.80,150.69,149.19,148.18,145.14,144.69,144.14,141.93$, $132.32,120.17,112.25,111.13,110.81,98.38$, 66.81, 61.39, $61.31,60.73,60.71,57.46,55.12,55.06,44.81 \times 2$. HR-MS (ESI) $\mathrm{m} / \mathrm{z}$ Calcd. $\mathrm{C}_{25} \mathrm{H}_{33} \mathrm{~N}_{2} \mathrm{O}_{7}^{+}[\mathrm{M}+\mathrm{H}]^{+}$473.2282, Found 473.2286.

### 4.2. Molecular docking and simulation for $P-g p$ and $29 d$

The detailed methods for molecular docking and simulation are shown in Supporting Information Section 1.

### 4.3. Cell viability assay, colony formation assay and animal models

The detailed methods for cell viability assay, colony formation assay and animal models are shown in Supporting Information Section 2.

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## Author contributions

Deyan Wu, Ying Xie, and Hai-Bin Luo designed the research. Senling Feng and Huifang Zhou carried out the experiments and performed data analysis. Deyan Wu, Dechong Zheng, Biao Qu, Ruiming Liu, Chen Zhang, and Zhe Li participated part of the experiments. Huifang Zhou, Ying Xie, and Hai-Bin Luo wrote the manuscript. Ying Xie and Hai-Bin Luo revised the manuscript. All of the authors have read and approved the final manuscript.

## Conflicts of interest

The authors have no conflicts of interest to declare.

## Appendix A. Supporting information

Supporting data to this article can be found online at https://doi. org/10.1016/j.apsb.2019.07.007.

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[^0]:    Abbreviations: $\mathrm{Ac}_{2} \mathrm{O}$, acetic anhydride; AcOH , acetic acid; AcONa , sodium acetate; $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$, boron trifluoride diethyl etherate; DCM , dichloromethane; DCE, dichloroethane; DMF, $N, N$-dimethylformamide; DMSO, dimethyl sulfoxide; DOX, doxorubicin; Et ${ }_{3} \mathrm{~N}$, triethylamine; Flutax-2, a fluorescent taxol derivative; MDR, multidrug resistance; NOB, nobiletin; NIS, $N$-iodosuccinimide; P-gp, P-glycoprotein; PI, propidium iodide; PTX, paclitaxel; QND, quinidine; Rho123, rhodamine 123; $t$-BuOK, potassium tert-butylate; TCA, trichloroacetic acid; THF, tetrahydrofuran; TLC, thin-layer chromatography; SRB, sulforhodamine B; Ver, verapamil.
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