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The effect of La $_2O_3$ in Tm $^{3+}$ -doped germanate-tellurite glasses for $\sim 2~\mu m$ emission

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A germanate-tellurite glass (GeO₂-TeO₂-K₂O-Nb₂O₅-La₂O₃) with thulium doping has been investigated for application as a laser material around 2.0 μ m regions. Under the 808 nm laser diode pumped, intense 1.8 μ m emission is obtained. Based on the absorption spectra, radiative properties are predicted using Judd-Ofelt theory. The maximum value of emission cross-section of Tm³⁺ around 1.8 μ m can reach 1.46 × 10⁻²⁰ cm², which indicated that the germanate-tellurite glass may provide high gain as a good medium for efficient 1.8 μ m laser system.

ecently, Tm³⁺ doped glass has been widely investigated for mid-infrared fiber lasers around 2.0 µm since the transitions of $Tm^{3+}:{}^{3}F_{4} \rightarrow {}^{3}H_{6}$. The mid-infrared fiber lasers in the 2.0 µm have generated great interest for its potential applications in several fields, such as medical surgery, remote sensing, eye-safe laser radar, military, and atmospheric pollution monitoring in the last decades¹⁻⁴. Rare-earth thulium ions are often chosen to produce laser radiation at 1.8 μ m through the transitions of Tm³⁺:³F₄ \rightarrow ³H₆. In this paper, Tm³⁺ has been selected as active ions because the ground state of Tm³⁺:³H₆ can be pumped directly by 808 nm commercial laser diodes. Laser emission at 1.8 µm strongly dependents on the efficiency of the energy transfer process among Tm³⁺ ions. In order to obtain high-power infrared emissions from Tm³⁺, the host glass plays an important role. Most researches in Tm³⁺ doped glass matrix for $\sim 2.0 \ \mu m$ emission have focused on silica⁵⁻⁸, silicate⁹⁻¹¹, fluoride¹²⁻¹⁴, fluorophosphates¹⁵⁻¹⁷, tellurite¹⁸⁻²¹, and germanate²²⁻²⁵ based glasses. Due to the energy cross-relaxation process, the host material should be chosen to have high solubility of the thulium ions. Compared with SiO₂based glass, GeO₂-TeO₂-based glass have the higher solubility for Tm³⁺ ions and the lower phonon energy (germanate-tellurite glass' is about $800-900 \text{ cm}^{-1}$), which can avoid very competitive nonradiative decay for Tm³⁺ ions. High phonon energy leads to faster multi-phonon relaxation and thus results in lower cross-relaxation rate. By contrast with germanate-tellurite glass, fluoride glass also has very low phonon energy (around 500 cm⁻¹) but the low mechanical strength and damage threshold limit its application in high power or energy fiber laser systems. Therefore, germanate-tellurite glass may have higher ions solubility and appropriate phonon energy for developing more efficient optical devices by comprehensive evaluation.

In this work, we report primarily a series of glass compositions in the Tm^{3+} -doped germanate-tellurite (GT) glasses to study mid-infrared spectroscopic properties. We induce La_2O_3 to substitute for Nb_2O_5 to improve the thermal stability and anti-crystallization of this glass system, by reasons of the ratio of La^{3+} ions (0.12 nm) and coordination number (8) are higher than Nb^{5+} ions (ratio = 0.07 nm/coordination number = 6). The absorption and emission spectra of Tm^{3+} -doped GT glasses were obtained. Results and analysis reveal that these germanate-tellurite glasses with good spectroscopic properties may provide a high gain as a good medium for high-power level laser.

Experiments and measurements

The gemanate-tellurite glass with molar composition of 70GeO_2 - 10TeO_2 - $10\text{K}_2\text{O}$ - $(9-x)\text{Nb}_2\text{O}_5$ - $x\text{La}_2\text{O}_3$ - $11\text{Tm}_2\text{O}_3$ (x = 1, 3, 5, and 7 mol. %) was investigated, which is hereafter denoted as GTLa-x glass. All the samples were prepared by traditional melt-quenching method with using high-purity of GeO₂, TeO₂, K₂CO₃, Nb₂O₅, La₂O₃, and Tm₂O₃ powder. Well-mixed 20 g batches of the samples were placed in a platinum crucible and heated with a SiC-resistance electric furnace at 1150°C for 30 min to melt. The melt was poured onto a preheated steel plate and



annealed for 2 hours in a furnace around the glass transition temperature (T_g), after which it was allowed to cool slowly in the furnace to room temperature. The annealed samples were fabricated and polished to $10 \times 10 \times 1$ mm for optical measurements.

The density was tested by Archimedes' liquid-immersion method in distilled water and the refractive index of the glass samples was measured by prism minimum deviation method. The DSC curve is recorded by NETZSCH STA 409PC. Furthermore, the absorption spectra were recorded with a Perkin-Elmer-Lambda 900 UV/VIS/ NIR spectrophotometer in the range of 300–2000 nm, and the emission spectra were measured with a Triax 320 type spectrometer (Jobin-Yvon Co., France) upon excitation at 808 nm. For the fluorescence lifetime measurements, the instrument applied was FLSP 920 fluorescence spectrophotometer (Edinburgh Analytical Instruments Ltd, UK). All the measurements were carried out at room temperature.

Results and discussion

Thermal property. Figure 1 shows the thermal property of GTLa-x glass from 300°C to 900°C. From the DSC curve of present glasses system, we can find out that no crystallization peak is apparent, and the glass transition temperature T_g are 355°C, 459°C, 550°C, respectively. The T_g increase with the contents of La₂O₃ increase. It indicated the introduce La₂O₃ to substitute for Nb₂O₅ can improve the thermal stability and anti-crystallization ability of this glass system. We could conclude that thermal properties of the GTLa-x glass are good for fiber drawing from the analysis of DSC curve.

Absorption spectrum. The absorption spectra were obtained at room temperature over a wavelength region of 300-2000 nm. The absorption spectra of the GTLa-x samples at room temperature is shown in Fig. 2, and the absorption bands centered corresponding to the transitions from the ³H₆ ground state to the exited stated of ³F₄ (1694 nm), ${}^{3}\mathrm{H}_{5}$ (1208 nm), ${}^{3}\mathrm{H}_{4}$ (792 nm), ${}^{3}\mathrm{F}_{2}$ + ${}^{3}\mathrm{F}_{3}$ (686 nm) and ${}^{1}G_{4}$ (492 nm), respectively. The transitions to energy levels higher than ¹G₄ are not observed because of the intrinsic band gap absorption of host glass. The insert of the figure 2 is the transmittance curve of the GTLa-3 glass sample from 2.5 µm to 8 μ m. We can find that the OH⁻ concentration on present glass sample is lager and result in the lager absorption around 3 µm regions. As well known, OH- groups play an important role and the much OH⁻ groups on glass host would decrease the performance of the 1.8 µm emission. So, improving the OHconcentration on present glass system is a crucial issue to obtain efficiently 1.8 µm emission.

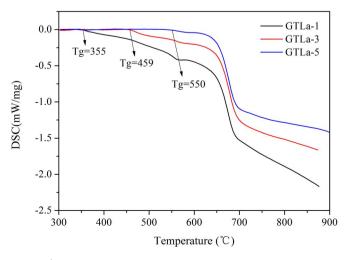


Figure 1 | DSC curve of GTLa-x glasses.

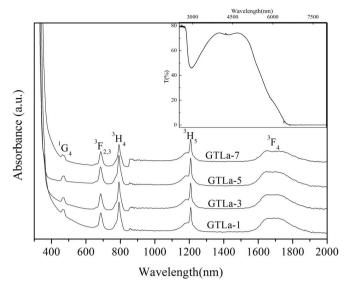


Figure 2 | **Absorption spectra of GT glasses.** The insert of figure is the transmittance curve of GTLa-3.

Judd-Ofelt analysis. Judd-Ofelt (J-O) theory^{26,27} has been commonly applied to determine the important spectroscopic and laser parameters of rare earth ions doped glasses by many researchers. According to the J-O theory, based on absorption spectra, the measured oscillator strengths and some spectroscopic parameters of many trivalent rare earths in solids, such as intensity parameters Ω_{λ} , radiative transition probability A_{rad} , and radiative lifetime τ_{rad} can be calculated. The theoretical (f_{theor}) and the measured (f_{exp}) oscillator strengths of the GTLa-1 glass are shown in Table 1.

As is shown in Table 1, the root-mean-square error deviation of intensity parameters is 0.39×10^{-6} , which indicates the validity of the J-O theory for predicting the spectral intensities of Tm³⁺ and the reliable calculation. The J-O intensity parameters are important for investigating the local structure and bonding in the vicinity of RE ions. As known large Ω_2 parameter is related with the amount of the strong covalent bond between rare-earth ions and ligand anions, strongly depends on the asymmetry of the local environment at the Tm³⁺ ion site in the glass hosts, and a strong polarizability of the anion, while the Ω_6 parameter is related to the overlap integrals of 4f and 5d orbits^{28,29}. Values of Ω_4 and Ω_6 also provide some information on the rigidity and viscosity of hosts³⁰. In this work, the J-O intensity parameters Ω_{λ} are compared with those of various Tm³⁺ doped glasses presented in Table 2. It can be found that the value of Ω_2 becomes small when La₂O₃ increase but La₂O₃ < Nb₂O₅. When the contents of La₂O₃ approximate that of Nb₂O₅, the value of Ω_2 increases slightly, while $La_2O_3 > Nb_2O_5$, the value of Ω_2 drop down quickly. Similarly, Ω_6 also changed with the contents of La₂O₃. These proved the change of asymmetry of the local environment at the Tm³⁺ ion site and found a variation in the covalent and bond strength of Tm-O in glass hosts. In table 2, Ω_2 for GTLa-x glass is

Table 1 \mid Measured and calculated oscillator strengths of Tm^{3+} in GTLa-1 glass

						r strength) ⁻⁶)
Absorption	λ (nm)	$\left \left\langle U^{(2)}\right\rangle\right ^2$	$\left \left\langle U^{(4)}\right\rangle\right ^2$	$\left \left\langle U^{(6)}\right\rangle\right ^2$	Measured	Calculated
${}^{3}\text{H}_{6} \rightarrow {}^{3}\text{F}_{4}$		0.5470			3.531	3.529
${}^{3}\text{H}_{6} \rightarrow {}^{3}\text{H}_{4}$					3.836	3.844
${}^{3}\text{H}_{6} \rightarrow {}^{3}\text{F}_{3}$		0	0.3162	0.8408	2.796	2.819
${}^3H_6 \rightarrow {}^1G_4$	472	0.0006	0.0355	0.2082	1.287	0.889



Table 2 $\Omega_{\lambda}(\lambda$ = 2, 4, 6)(×10^{-21} cm²) of Tm³+-doped glass							
Glass	Ω_2	Ω_4	Ω_6	Reference			
GTLa-1 GTLa-3 GTLa-5 OTLa-5	5.97 5.64 5.87	1.10 0.97 1.03	1.27 1.26 1.39	This work			
GTLa-7 Silica Fluorophosphate Tellurite	3.93 3.7 3.01 3.20	0.5 2.3 2.56 2.01	1.1 0.6 1.54 1.83	[32] [16] [19]			

higher than fluoride glasses', because the electronegativity of oxide is smaller than that of fluorine. It means the asymmetry and covalent environment of the Tm-O band in GT glass are stronger than other host glasses'.

The radiative transition probabilities for $\text{Tm}^{3+}; {}^{3}\text{F}_{4} \rightarrow {}^{3}\text{H}_{6}$ level can be calculated by using J-O intensity parameters. The results of J-O analysis are shown in Table 3. The radiative transition probability A_{rad} for the transition in GTLa-1 glass is 372.1 s⁻¹, and this value is slightly higher than those in fluorophosphate glass¹⁶. Because radiative transition probability depends intensively on the refractive indices, obtaining of larger radiative transition probability is reasonable in GT glass with higher refractive index³¹, and larger values are beneficial in achieving intense near-infrared emission.

Infrared fluorescence spectra. The fluorescence spectra of the samples are measured as shown in figure 3 to investigate the nearinfrared emission characteristics of the prepared Tm³⁺-doped GT glasses as a function of the La2O3 contents. Besides, the inset of figure 3 is the energy level diagram and energy transfer sketch map of Tm³⁺ ions pumped at 808 nm. The cross relaxation (CR) and energy migration (EM) into the ³H₄ level are also indicated. The ${}^{3}F_{4} \rightarrow {}^{3}H_{6}$ fluorescence around 1.8 µm is obviously observed, and the intensity is much larger than that of 1.47 µm corresponding to ${}^{3}\text{H}_{4} \rightarrow {}^{3}\text{F}_{4}$ transition for samples under the 808 nm diode laser. The emission at 1.47 μ m due to the ${}^{3}H_{4} \rightarrow {}^{3}F_{4}$ transition is much lower in intensity relative to the 1.8 µm because the cross relaxation between Tm^{3+} ions is efficiency. Moreover, the intensity of 1.8 μm emission firstly decreased and then increased with the increment of La₂O₃ concentration. But the intensity of 1.8 µm emission slightly increased when the concentration of La2O3 more than the concentration of Nb₂O₅. The maximum fluorescence peak intensity for GT glass is observed around 1 mol. % La₂O₃.

Absorption and stimulated emission cross section. According to the absorption spectra, the absorption cross-section (σ_{abs}) can be calculated by using Beer-Lambert equation

$$\sigma_{abs}(\lambda) = \frac{ln[I_0(\lambda)/I(\lambda)]}{Nl} \tag{1}$$

where N is the rare-earth ion concentration (ions/cm³), l is the sample thickness, $I_0(\lambda)$ is the incident optical intensity, and $I(\lambda)$ is the optical intensity throughout the sample.

Table 3 | The electric (S_{ed}) and magnetic (S_{md}) dipole line strengths, radiative transition probability (A_{rad}), branching ratio (β), and radiative lifetime (τ_{rad}) of Tm³⁺ in GT glass

			S_{ed}	S_{md}			
Glass	S,L,J	S′,L′,J′	(×10 ⁻²	¹ cm ²)	A_{rad} (s ⁻¹)	β (%)	τ _{rad} (ms)
GTLa-1	³ F ₄	${}^{3}H_{6}$	44.84	0	372.1	100	2.68
GTLa-3			42.0	0	348.5	100	2.86
GTLa-5			44.07	0	365.6	100	2.73
GTLa-7			28.53	0	236.7	100	4.22

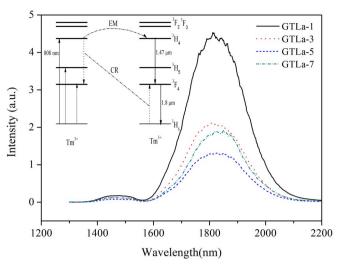


Figure 3 | Fluorescence spectra of the GT glasses with Tm^{3+} ions. The inset is the energy level diagram and energy transfer sketch map of Tm^{3+} when pumped at 808 nm.

The stimulated emission cross-section (σ_{em}) is calculated from the absorption cross-section by using McCumber formula equation³³

$$\sigma_{em}(\lambda) = \sigma_{abs}(\lambda) \frac{Z_l}{Z_u} exp\left[\frac{hc}{kT}\left(\frac{1}{\lambda_{ZL}} - \frac{1}{\lambda}\right)\right]$$
(2)

where Z_l and Z_u are the partition functions of the lower and upper states, respectively, *k* is the Boltzmann constant, T is the temperature, and λ_{ZL} is the wavelength for the transition between the lower Stark sublevels of the emitting multiplets and the lower Stark sublevels of the receiving multiplets (so called 'zero-phonon line'). In the following calculation, the ratio of the partition functions of the lower to the upper state Z_l and Z_u is equal to 13/9, for the reason that the values of the lower and upper states Z_l and Z_u simply become the degeneracy weightings of the two states in the high temperature limit. The zerophonon line is assumed to the peak wavelength of the absorption, $\lambda_{ZL} = \lambda_{peak-abs}$.

Figure 4 shows the σ_{abs} and σ_{em} for optical transitions involving the ground and first excited states of Tm³⁺ ions of GTLa-1 glass. The calculated maximum σ_{abs} and σ_{em} for Tm³⁺ are 4.17 × 10⁻²¹ cm² at 1700 nm and 14.6 × 10⁻²¹ cm² at 1806 nm, respectively. Moreover, the product of the stimulated emission cross-section and the radiative lifetime (τ_{rad}), $\sigma_{em} \times \tau_{rad}$ is another important parameter to

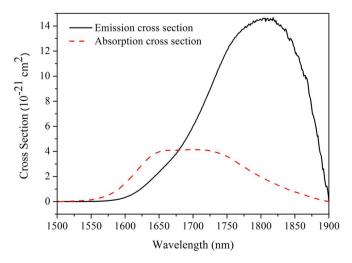
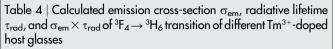


Figure 4 | The absorption and emission cross sections of GTLa-1 glass.



Glass	$\overset{\sigma_{\text{em}}}{(\times 10^{-21}\text{cm}^2)}$	τ _{rad} (ms)	$\substack{\sigma_{em}\times\\\tau_{rad}(\!\times10^{-21}cm^2ms)}$	Reference
GTLa-1	13.56	2.68	36.3	This work
GTLa-3	14.61	2.86	41.78	
GTLa-5	13.01	2.73	35.51	
GTLa-7	8.47	4.22	35.74	
Silica	4.3	7.0	30.1	[32]
Fluorophosphate	4.11	4.44	18.2	[16]
Tellurite	9.2	1.78	16.37	[21]
Germanate	9.3	2.48	23.1	[37]

characterize laser materials, which is the figure of merit (FOM) for amplifier gain^{28,34,35}.

Table 4 shows the emission cross-section, radiative lifetime, and FOM gain ($\sigma_{em} \times \tau_{rad}$) of ${}^3F_4 \rightarrow {}^3H_6$ transition of Tm^{3+} -doped glasses. For laser glasses, it is generally desirable for the emission cross section to be as large as possible to provide a high gain^{36}. This peak emission cross section had reported as 4.3×10^{-21} cm² in silica glass by Turri^{32}, 4.11×10^{-21} cm² in fluorophosphate by Tian^{16}, and 9.20×10^{-21} cm² in tellurite by Balda^{21}. It can be seen that the σ_{em} in GTLa-1 sample has a maximum value which is higher than that of Tm^{3+} -doped other kinds of glasses. And the $\sigma_{em} \times \tau_{rad}$ of Tm^{3+} doped GT glasses is much larger than other Tm^{3+} -doped host glasses. The advantage of GT glass possessing larger emission cross section may prove it to be a promising laser glass with high gain to scale a high power level laser.

Fluorescence lifetime. The fluorescence lifetime of Tm³⁺:³F₄ becomes shorter with increasing of the La2O3 concentrations, and the lifetime becomes longer when doped with 7 mol% La₂O₃, as shown in table 5. However, it indicated that the measured lifetime is much shorter than the calculated lifetime. It is due to nonradiative quenching9. The nonradiative decay originated from several mechanisms, such as energy transfer processers between the Tm³⁺ ions themselves, multiphonon decay, etc. Energy transfer processes between the Tm³⁺ ions themselves are cross-relaxation and energy migration. The cross relaxation energy transfer process between Tm³⁺ ions described in the introduction does not quench fluorescence by itself, but energy migration is a resonant nonradiative mechanism, which increases the probability of luminescence quenching by impurities. Generally, the relatively longer radiation lifetime is beneficial to reduce the laser oscillation threshold³⁸. Therefore, this Tm³⁺-doped GT glass can be considered as an appropriate medium to achieve a 1.8 µm laser with high quality.

Summary

Spectroscopic properties of 1.8 μ m emission have been investigated in the Tm³⁺ -doped germanate-tellurite glasses (GeO₂-TeO₂-K₂O-Nb₂O₅-La₂O₃). We investigate the effect of the spectroscopic properties for different compositions with varying Nb⁵⁺/La³⁺ ions ratio. On the basis of our experimental results, Judd–Ofelt intensity parameter, spontaneous transition probability, radiative lifetime, and absorption cross-section, as well as stimulated emission cross-section

Table 5 | Measured fluorescence lifetime of the Tm: ${}^{3}H_{4}$ level and Tm: ${}^{3}F_{4}$ level at 295 K of different Tm ${}^{3+}$ -doped host alasses

$\label{eq:constraint} \begin{array}{ c c c c c c c } \hline Level & GTLa-1 & GTLa-3 & GTLa-5 & GTLa-7 \\ \hline $Tm^{3+}:^3F_4$ & $428.64\ \mu s$ & $284.40\ \mu s$ & $282.43\ \mu s$ & $482.13\ \mu s$ \\ \hline $Tm^{3+}:^3H_4$ & $121.42\ \mu s$ & $107.91\ \mu s$ & $110.56\ \mu s$ & $125.83\ \mu s$ \\ \hline \end{array}$	1111. 1 4 1010		aopoa nosi g	Jiaoooo	
	Level	GTLa-1	GTLa-3	GTLa-5	GTLa-7
			•		•

are calculated and discussed. For the Tm^{3+} -doped GT glass samples, the fluorescence measurement shows that the maximum value of 1.8 µm emission intensity occurs in GTLa-1 glass. The optimal concentration of La₂O₃ for a laser application has been determined, which is 1 mol.% La₂O₃. Results indicate that these germanate-tellurite glasses with good spectroscopic properties may provide a high gain as a good medium for mid-infrared laser.

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Author contributions

Y.P. wrote the main manuscript text and coauthor, X.Y. checked up. J.Z. and L.Z. are responsible for the experiment. All authors reviewed the manuscript.

Additional information

Competing financial interests: The authors declare no competing financial interests.

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