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Research article

Comparative study on larvicidal activity of green synthesized silver nanoparticles and *Annona glabra* (Annonaceae) aqueous extract to control *Aedes aegypti* and *Aedes albopictus* (Diptera: Culicidae)



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L.D. Amarasinghe^{a,*}, P.A.S.R. Wickramarachchi^b, A.A.A.U. Aberathna^a, W.S. Sithara^b, C.R. De Silva^c

^a Department of Zoology & Environmental Management, Faculty of Science, University of Kelaniya, Dalugama, GQ 11600, Sri Lanka
^b Department of Chemistry, Faculty of Science, University of Kelaniya, Dalugama GQ 11600, Sri Lanka

Department of Chemistry, Faculty of Science, Oniversity of Relative, Datagana OQ 11000, Sh Lanka

^c Department of Chemistry & Physics, 213 Natural Sciences Building, Western Carolina University, USA

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ABSTRACT

The present study reports mosquito larvicidal potential of green synthesized silver nanoparticles by using *Annona glabra* leaves (An-AgNPs). Synthesized An-AgNPs were characterized by Ultraviolet-Visible spectroscopy (UV-VIS), Scanning Electron Microscopy (SEM), Dynamic Light Scattering (DLS) technique and Fourier transform infrared spectroscopy (FTIR). Colur change from pale yellow to brick red of the plant extract and AgNO₃ solution indicated the formation of An-AgNPs initially. Surface Plasmon Resonance (SPR) band at 435 nm in the UV-Vis confirmed the formation of An-AgNPs. SEM images showed that An-AgNPs were spherical in shape. FTIR proved that An-AgNPs were functionalized with biomolecules in *A. glabra* leaves. Based on DLS analysis the average size range of synthesized An-AgNPs was determine to be 10–100 nm and 100–1000 nm.

Third instar larvae of dengue vector mosquitoes, *Aedes aegypti* and *Aedes albopictus* were subjected to larvicidal bioassays in a range of concentrations of An-AgNPs and *A. glabra* crude aqueous leaf extract (2–10 mg/L). An-AgNPs exhibited very high larvicidal activity against dengue vector mosquito larvae; LC_{50} value for *Ae. aegypti* at 24 h exposure to An-AgNPs (Plant extract: AgNO₃ 1 : 10) 5.29 mg/L; An-AgNPs (Plant extract: AgNO₃ 2 : 10) 2.43 mg/L while LC_{50} value for *Ae. albopictus* at 24 h exposure to An-AgNPs (Plant extract: AgNO₃ 2 : 10) 2.43 mg/L while LC_{50} value for *Ae. albopictus* at 24 h exposure to An-AgNPs (Plant extract: AgNO₃ 1:10) 3.02 mg/L; An-AgNPs (Plant extract: AgNO₃ 2:10) 2.51 mg/L. LC_{50} values obtained for *A. glabra* leaf extract tested against *Ae. aegypti* and *Ae. albopictus* are 5.94 mg/L and 5.00 mg/L respectively at 24-hour exposure. This study further revealed that *Ae. albopictus* is more susceptible than to *Ae. aegypti* to a given concentration of An-AgNPs and to crude aqueous leaf extract of *A. glabra*. Larvicidal effect of An-AgNPs is superior to the crude aqueous leaf extract of *A. glabra*. An-AgNPs is a potent larvicide for dengue vector control.

1. Introduction

Aedes aegypti and Aedes albopictus are the two mosquito vectors of important arboviruses of the two genera of Flavivirus and Togavirus globally. Ae. aegypti is the main competent vector of flaviviruses such as ZIKA, dengue, chikungunya and yellow fever virus. Ae. albopictus is a vector for Flaviviruses such as yellow fever virus, Zika virus, dengue virus, Japanese encephalitis virus, and West Nile virus and Togaviruses such as Eastern equine virus and ross river virus (Paupy et al., 2009).

With the exception of yellow fever, for which an efficient vaccine has been available since the 1940s (Frierson, 2010), no vaccine is currently

commercially available against the viral diseases transmitted by *Ae. aegypti* and *Ae. albopictus*. Therefore, the prevention of these diseases is mainly achieved through mosquito population control (Bisset et al., 2006; WHO, 2009).

Larvicides are among the main tools in mosquito control programmes. The most widely used larvicides are organophosphates such as temephos, methoprene, growth inhibitors, and bacterial insecticides such as *Bacillus thuringiensis israelensis* (Bti) and *Bacillus sphaericus* (Bisset et al., 2006; De Silva and Mendes, 2007; Poopathi and Abidha, 2010; Anupam et al., 2012). Larvicides are applied to either natural or artificial bodies of water, as a result their effect to beneficial and other nontarget organisms,

* Corresponding author. E-mail address: deepika@kln.ac.lk (L.D. Amarasinghe).

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including humans must be harmless (Arjunan et al., 2012; Allgeier et al., 2018).

The persistent and in some cases indiscriminate use of synthetic chemical insecticides has resulted in a reduction of their efficacy due to the dramatic emergence of resistant insect populations during the last decades (Govindarajan et al., 2005). Increasing cost of novel insecticides and annual exportation expenditures, effect on non-target populations particularly on mammals, high toxicity to mammals, entering to food chains and food webs, bioaccumulation in non-target organisms' bodies, nonbiodegradable nature, ecological imbalance, the emergence of refractory vector behaviour and environmental pollution are occurred due to synthetic chemical insecticides (Anupam et al., 2012). The solution for these arising problems relies on searching for new and effective compounds that are easily degradable, environmental friendly, do not have any adverse effect on non-target populations, easily available and safe products at low cost.

Using botanical extracts is a sound option to avoid the adverse effects of synthetic chemical insecticides. Abd Kadir et al. (2013) reviewed the potential anti-dengue medicinal plants reported from 69 studies from 1997 to 2012. Plant derived insecticides usually contain a combination of several chemical compounds, unlike conventional insecticides which is mainly based on one specific active agent. These botanicals work synergistically and target different biological processes thus reducing the resistance development in their targets. An interesting advancement for the solely plant based insecticides is the development of plant-mediated nano based products combining the insecticidal activity of the botanicals and enhanced efficiency at the nano scale (1-100 nm) due to the high surface area to volume ratio of the nanoparticles. Combinatorial effect of these characteristics have enabled to achieve their insecticide efficacy at very low concentrations (≤30 mg/L) (Benelli et al., 2017). NPs can easily penetrate across the cell membrane of a living organism, due to their minute size thus avoid defense mechanisms. Afterwards, NPs migrate into the cell and reach organelles such as mitochondria, modifying the cell metabolism and leading to cell death. Therefore, NPs could be toxic to both vertebrates and invertebrates.

Recently, green fabrication of silver nanoparticles have acquired considerable attention owing to their mosquito larvicidal activity towards medically challenged pathogens and dreadful vectors (Priya and Santhi, 2014; Soni and Prakash, 2014; Benelli et al., 2017; Madanagopal et al., 2017; Morejón et al., 2018).

A comprehensive review on plant-mediated biosynthesis of nanoparticles as an emerging tool against mosquitoes has been published (Benelli, 2016). Majority of the plant-fabricated mosquitocidal nanoparticles reported to date are based on silver (Benelli, 2016). In the present study we attempt to use *Annona glabra* leaf extract to fabricate silver nanoparticles (An-AgNPs). *Annona glabra* (Pond apple) (Family Annonaceae) prefers wetter tropical and sub tropical habitats. *Annona glabra* is native for North, South and Central America and West Africa. It is regarded as an invasive weed in Australia due to its potential for spread, and negative impacts on environment and economy. Pond apple is also a problem in Sri Lanka and Fiji. *Annona glabra* is an aggressive and very hardy tree that forms dense thickets (Austin, 2004).

A. glabra is used in traditional medicine against several human ailments and disorders such as constipation, fever, ulcers, and tumour, including cancer because these plants consist with cytotoxic, antitumour, antifungal, antiparasitic, antibacterial, antispasmodic, antimicrobial, anticancer, antioxidant, and hepatoprotective repellent properties (Cave et al., 1997; Biba et al., 2014).

Previous studies on insecticidal properties of *Annona glabra* leaf extracts against *Ae. albopictus* and *Ae. aegypti*, the two dengue vector mosquito species in Sri Lanka showed promising results (Amarasinghe and Ranasinghe, 2017). However, no studies have been reported on use of *A. glabra* mediated AgNPs to control vector mosquiotoes in Sri Lankan condition. The present study attempts to enhance the bioefficacy of *A. glabra* leaf extract by green fabricating AgNPs. To the authors knowledge

this is the first study on the larvicidal efficacy of An-AgNPs against Ae. albopictus and Ae. Aegypti.

2. Materials and methods

2.1. Plant material collection

Leaves of *Annona glabra* plant were collected in December 2018, from marshland in Hunupitiya (N 06° 58.904/, E 079° 54.281/), Kelaniya and kept in polythene bags and brought to the laboratory. Authentication of the plant was done with the help of the herbarium collection at the Department of Plant and Molecular Biology, Faculty of Science, University of Kelaniya.

2.2. Preparation of aqueous crude extract of Annona glabra leaves

Two hundred and fifty grams of fresh leaves were washed well with running tap water, rinsed with deionized water and decanted. The leaves were air dried and crushed well by using a pestle and mortar and placed in a conical flask (1000 mL) and 800 mL of deionized water was added. This was kept over night and the residue was filtered and discarded. Filtrate was concentrated under reduced pressure in a rotory evaporator at 70 °C to obtain the crude of aqueous leaf extract. Then the crude was freeze dried. Crude extract of 0.1 g was dissolved in 1000 mL of deionized water and 100 mg/L stock solution was prepared. The stock solution was used for the preparation of a concentration series of aqueous crude extract for bioassays.

2.3. Green synthesis of silver nanoparticles using Annona glabra leaf extract (An-AgNPs)

Fresh leaves weighing 160 g were washed well with running tap water, rinsed with deionized water and decanted. The leaves were crushed using a pestle and mortar and placed in a conical flask (1000 mL). Eight hundred mL of deionized water was added to the flask and covered with an aluminium foil. Then the sample was heated to 70 °C on a hot plate and left for 1 h with a magnetic stirrer to obtain the stock solution (200 g/L) of aqueous leaf extract. Eight hundred mL of the resultant leaf extract was filtered using a Whatman No.1 filter paper. AgNO₃ stock solution (1 g/L) was prepared by dissolving 1 g of AgNO₃ flakes (Sigma-Aldrich) in 1 L deionized water.

Annona glabra leaf extract and $AgNO_3$ stock solution were mixed to the ratios of leaf extract: $AgNO_3$ 1 : 10 and 2 : 10 v/v to obtain two final products. This was performed by mixing 10 mL and 20 mL respectively of Annona glabra aqueous leaf extract separately with 100 mL of $AgNO_3$ stock solution. Two mixtures were incubated at room temperature (27 $^\circ$ C \pm 1) for 3 h with continuous stirring on a stirrer mixer until the solution become brick red color.

Each sample was then centrifuged at 5000 rpm for 20 min using a centrifuge machine (HERMLEZ 206 A) and the supernatant was decanted. Five mL deionized water was added to the precipitate and centrifuged again at 5000 rpm for 20 min. This was repeated one more time to wash the NPs thoroughly. The final precipitate was dried in desiccator for two days and dried form was used for further confirmation.

2.4. Characterization of silver nanoparticles (AgNPs)

2.4.1. UV - vis spectroscopy

Formation of AgNPs was confirmed by observing the Surface Plasmon Resonance (SPR) band from 200 to 1000 nm using UV-Vis spectrophotometer (Thermo Scientific, 1800) at a resolution of 5 nm at room temperature (27 °C \pm 1) at the Department of Chemistry, Faculty of Science, University of Kelaniya.

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2.4.2. SEM analysis

The facility of the Scanning Electron Microscope (SEM) (EVOLS) was obtained to determine the shape of developed nanoparticles at the Faculty of Science, University of Peradeniya. Samples of AgNPs were deposited in dried form on a double conductive tape which was stick on a sample holder at room temperature (27 °C \pm 1). A thin layer of gold-platinum was coated to make samples conductive. Then the sample was imaged at 80 kV operation voltage.

2.4.3. Particle size measurements

Particle size was determined by dynamic light scattering technique (DLS) (CILAS NANO DS) at the Faculty of Science, University of Peradeniya, Sri Lanka

2.4.4. Fourier transformed infrared (FTIR) spectrum

FTIR spectrum of the sample was recorded by fourier transformed infrared spectrophotometer (Perkin Elmer) at the Department of Chemistry, University of Kelaniya, Sri Lanka. The FTIR spectrum was recorded from 4000 to 750 cm⁻¹ by placing the dried AgNP powder on the crystal.

2.5. Bioassay on larvicidal activity

Ae. aegypti and *Ae. albopictus* mosquito larvae were evaluated with the World Health Organization (2005) standard protocols.

2.5.1. Larvicidal activity of Annona glabra AgNano (An-AgNPs)

Healthy two day old third instar larvae of Ae. aegypti were transferred in batches of 20 in 50 mL of water into glass beakers of 150 mL (n = 24). The total volume of water in each beaker was increased to 90.0 mL by adding deionized water. 0.1 g of green synthesized AgNPs (Plant extract: AgNO3; 1:10) compound was dissolved by sonication in 1000 mL of deionized water to produce stock solutions of AgNPs (Plant extract: AgNO₃; 1:10). Concentrations of AgNPs (Plant extract: AgNO₃; 1:10) 0.002 g/L, 0.004 g/L, 0.006 g/L, 0.008 g/L and 0.01 g/L were prepared by micropipetting (Khader et al., 2017; Khader and Ramesh, 2018). One concentration was prepared by randomly selecting four beakers that each including 20 larvae and adding the required volume of AgNPs (Plant extract: AgNO₃; 1:10) solution into each beaker based on $C_1V_1 = C_2V_2$ equation. Then each solution was completed up to 100 mL with distilled water. This was continued for all the concentrations. A control treatment was run with 100 mL of deionized water. A fully randomized design was used (6×4). The number of dead mosquito larvae was counted after 24 and 48 h of exposure and the percentage mortality was calculated. This was continued for third instar larvae of Ae. albopictus. Same procedure was done for AgNPs (Plant extract: AgNO₃; 2:10).

2.5.2. Larvicidal activity of Annona glabra aqueous crude extract

Healthy third instar larvae *Ae. aegypti* were transferred in batches of 20 into glass beakers of 150 mL containing 75 mL of deionized water (n = 24). Concentrations of crude extract 2 mg/L, 4 mg/L, 6 mg/L, 8 mg/L and 10 mg/L were prepared One concentration was prepared by randomly selecting four beakers that each including 20 larvae and the addition of the required volume of crude extract stock solution into each beaker based on $C_1V_1 = C_2V_2$ equation. Then each solution was top up to 100 mL with distilled water. This was continued for all the concentrations. A control treatment was run with 100 mL of deionized water. A fully randomized design was used (6×4). The number of dead mosquito larvae was counted after 24 and 48 h of exposure and the percentage mortality was calculated. This was continued for third instar larvae of *Ae. albopictus.*

In both bioassays, moribund larvae were counted and added to dead larvae for calculating percentage mortality. Dead larvae are those that did not induce to move when they were probed with a needle in the siphon or the cervical region. Moribund larvae were those incapable of rising to the surface or not showing the characteristic diving reaction when the water was disturbed (WHO, 2005).

2.6. Statistical analysis

Anderson – Darling test was used to determine whether the mean percentage of mortality data are normally distributed. P > 0.05 was obtained for all tests. Probit Analysis was used to determine the LC_{50} for the AgNPs after 24 h and 48 h of exposure. One-Way Analysis of Variance (ANOVA) followed by Tukey's pairwise test was conducted to test if there was any significant difference among Arcsine transformed values of mean percentage mortality of mosquito larvae in each concentrations using MINITAB 14.

3. Results

3.1. Characterization of AgNPs

Three hours of incubation of the mixture, *Annona glabra* leaf extract and AgNO₃ stock solution, to the ratios of leaf extract: AgNO₃ 1: 10 and 2: 10 v/v gave brick colour in the solution mixture.

Initial color and color changes after 3 h of incubation time of samples (Figure 1 a & b).

The formation of AgNPs was established by observing the Surface Plasmon Resonance (SPR) band at 435 nm using a UV-Vis spectrophotometer (Figure 2).

Image obtained using the Scanning Electron Microscope (SEM) (EVOLS) to determine the shape of developed nanoparticles is shown in Figure 3. This shows the spherical shape of the particles confirming that they are nano particles.

Particle size distribution analysis done using dynamic light scattering technique (DLS) (CILAS NANO DS) resulted in particle size ranges between 10-100 nm and 100–1000 nm (Figure 4 a & b) This confirms the products are nanoparticles.

FT-IR spectrum of the sample showed three prominent peaks at 1739 cm⁻¹, 1366 cm⁻¹ and 1217 cm⁻¹ indicating the characteristic functional groups of the phytochemicals (Figure 5). This confirms the AgNPs have uncapped the phytochemicals of *A. glabra* and the formation of An-AgNPs.

3.2. Larvicidal activity of An-AgNPs and determination of the 24 h and 48 h LC_{50} for AgNPs

The percentage mortality of *Ae. aegypti* and *Ae. albopictus* mosquito larvae exposed to varying concentrations of An-AgNPs (ratio 1:10 and 2:10) for 24 h and 48 h are shown in Figure 6 (a,b,c and d). They indicate that An-AgNPs demonstrated potential larvicidal activity even at lower concentrations of the treatment and larval mortalities increased with the increasing concentration of AgNPs and increasing exposure time from 24 h to 48 h. The highest mortalities were found in larvae of *Ae. albopictus* (LC₅₀ = 2.51 mg/L) and larvae of *Ae. aegypti* (LC₅₀ = 2.43 mg/L) exposued to An-AgNP (plant extract: AgNO₃ 2:10) (Table 1). The mortality was increased comparatively when their exposure time increased from 24 h to 48 h; *Ae. albopictus* (LC₅₀ = 2.10 mg/L);*Ae. aegypti* (LC₅₀ = 1.17 mg/L) (Table 1).

3.3. Larvicidal activity of Annona glabra leaf extract and determination of the 24 h and 48 h LC_{50} for Annona glabra leaf extract

Mosquito larvicidal effect of aqueous crude extract of *A. glabra* after 24 h and 48 hore exposure is shown in Figure 7 a & b. There is a significant reduction of laraval mortality in crude extracts compared to that of An-AgNP treatment (F = 3.64; p = 0.014) at 24 h (F = 119.72; p \leq 0.000) at 48 h exposure. The larval mortality was increased in increasing concentration with increased exposure time showing the LC₅₀ values of 5.00402 mg/L for *Ae. albopictus* and 5.94555 mg/L for *Ae. albopictus* and 3.5485 mg/L for *Ae. algopti* at 48 h exposure (Table 2; Figure 7a) whereas 2.73467 mg/L for *Ae. albopictus* and 3.5485 mg/L for *Ae. algopti* at 48 h exposure (Table 2; Figure 7b). This result also indicates that LC₅₀ values are always higher compared to



(a)

(b)

Figure 1. (a) Initial color and (b) color changes after 3 h of incubation time of samples containing Plant extract: AgNO₃ 1:10 (left) and Plant extract: AgNO₃ 2:10 (Right).



Figure 2. UV–vis spectrum shows a characteristics peak between 400 nm–500 nm.

the LC_{50} values of the larvae exposed to An-AgNP treatment and LC_{50} values obtained for *Ae. aegypti* are always higher than to those obtained for *Ae. albopictus*.

4. Discussion

Green approach consumes extracts from botanicals which act both as reducing and capping agents in nanoparticle synthesis (Landage and Wasif, 2012). Controlling vector mosquito larvae causing many diseases by using green synthesized AgNPs is an emerging technique (Priva and Santhi et al., 2014). The present study accomplished green synthesizing of AgNPs by using Annona glabra plant leaf extract for controlling Ae. Aegypti and Ae. albopictus as first attempt in Sri Lanka. A. glabra secondary metabolites and their synthetic derivatives in plants serve as a defense mechanism against insect attacks, provide an alternative sources as larvicides (Isman and Seffrin, 2014). Recent research has proved that the effectiveness of plant derived biologically important compounds, such as saponine, steroids, isoflavonoids, essential oils, alkaloids and tannins obtained from crude extracts of seeds, leaves, fruits, bark and twigs may act as larvicides, insecticides, repellents, antifeedants, moulting hormones, antimoulting hormones, oviposition deterrents, juvenile hormone mimics, growth inhibitors as well as attractants (Shad and Andrew, 2017).



Figure 3. SEM (Scanning electron microscope) image showing the spherical morphology and agglomeration of synthesized AgNPs.

In the current study, we focused on eco-friendly nano-synthetic approach to mosquitocidal AgNPs. Color changes, UV-VIS, FTIR spectroscopic measurements, SEM analysis, and particle size distribution analysis characterized biosynthesized An-AgNPs. Formation of AgNPs evidents from a remarkable change of solution color from pale to brown, reddish-brown and brick red (Madanagopal et al., 2017). The appearance of the SPR band at 435 nm in the UV-Vis absorption spectra confirmed the formation of An-AgNPs (Arjunan et al., 2012).

Distinct IR bands characteristics of O–H stretching (3457 cm^{-1}), CH₃/ CH₂ stretching ($2945 - 2973 \text{ cm}^{-1}$), C–N stretching (1217 cm^{-1}) and C=O stretching (1739 cm^{-1}) vibration were observed in the FT-IR spectrum of the An-AgNPs. These bands confirms the the presence of biomolecules on NPs as capping agents. Having these biomolecules such as flavones and reducing sugars in the plant extract make them to act as reducing agents in the synthesis of AgNPs.

The surface morphology of An-AgNPs was investigated using SEM and showed that most of the nanoparticles were roughly spherical in shape with smooth edges. Particle size distribution analysis in this study showed that the mixture of several sizes of nanoparticles was formed in both approaches. Particles of sizes between 1 and 100 nm are called nanoparticles (Elechiguerra et al., 2005). Although modern definitions



Figure 4. The graph of particle size distribution of AgNPs: (a) Plant extract: AgNO₃ 2:10; (b) Plant extract: AgNO₃ 1:10.



Figure 5. FT-IR spectrum shows different functional peaks.

say particulate dispersions or solid particles with a size range between 10 and 1000 nm are nanoparticles (Priya and Santhi, 2014). Particles of both 1–100 nm size range and 100–1000 nm size range resulted in two approaches of the present study. As the plant material concentration is increased; the size of nanoparticles is decreased because agglomeration is prevented by a high amount of available capping agents in the plant extract and this was evident from Figure 4 a,b.

In the present study, two approaches were tested to synthesize AgNPs using *Annona glabra* leaves (An-AgNPs); plant extract: AgNO₃ 1:10 mixture and plant extract: AgNO₃ 2:10 mixture. Both An-AgNPs products showed larvicidal activity to *Ae. aegypti* and *Ae. albopictus* mosquito larvae which is manifested by a high percentage of mortality in comparison to those in the control treatments. This reveals that the mortality of mosquito larvae is not due to the natural or external reasons, but instead, the mortality of mosquito larvae is due to the An-AgNPs (plant extract: AgNO₃ 2:10) are toxicity to *Ae. albopictus* and *Ae. aegypti* larvae (LC₅₀ = 2.51 mg/L and LC₅₀ = 2.43 mg/L respectively for 24 h exposure) superior to An-AgNPs (plant extract: AgNO₃ 1:10) (LC₅₀ = 3.02 g/L; 5.29 mg/L respectively). When the exposure time is increased; comparatively larval mortality is also increased (Araj et al., 2015).

Biological synthesized stable silver nanoparticles using Annona squamosa leaf broth has been tested against Anopheles stephensi, Aedes

aegypti, and Culex quinquefasciatus showing lethal effect on fourth instars larvae and decline the longevity of adults (days) in male and female mosquitoes (Arjunan et al., 2012). Green synthesized nanoparticles have been successfully used to reduce mosquito young instar populations in the field with moderate larvicidal effects (Benelli & Govindarajan, 2016; Benelli et al., 2017) and the maximum efficacy (60.18%) was observed with the synthesized silver nanoparticles against the larvae of Cx. quinquefasciatus (Mondal et al., 2014; Rajasekharreddy and Rani, 2014). Gnanadesigan et al. (2011) reported that AgNPs synthesized using Rhizophora mucronata (Family: Rhizophoraceae) leaf extract given 0.585 mg/L (LC₅₀) and as 2.615 mg/L (LC₉₀) values for Ae. aegypti while 0.891 mg/L (LC₅₀) and 6.291 mg/L (LC₉₀) values for Cx. quinquefasciatus. Arjunan et al. (2012) reported that AgNPs synthesized using Annona squamosa leaf broth resulted a higher mortality of fourth instar larvae of Ae. aegypti, Cx. quinquefasciatus and Anopheles stephensi ($LC_{50} = 0.30$, 0.41, and 2.12 ppm respectively) and has declined the longevity of adult male and female mosquitoes. Sareen et al. (2012) reported that the larvicidal efficacy of AgNPs synthesized from aqueous leaf extract of Hibiscus rosasinensis (Family: Malvaceae) to control the larvae of Aedes albopictus. Patil et al. (2012) synthesized AgNPs using Pergularia daemia plant latex to use against Ae. aegypti and An. stephensi. Hence, green synthesized Ag-NPs could be considered to be used as possible alternative over physical and chemical methods.

Food consumption development is retarded in the animal exposed to AgNPS that influence their death. The biotoxicity against mosquito young instars related to the ability of nanoparticles to penetrate through the exoskeleton (Benelli, 2016). In the intracellular space, nanoparticles can bind to sulfur from proteins or to phosphorus from DNA, leading to the rapid denaturation of organelles and enzymes leads to death of mosquito larvae and pupae. Subsequently, the decrease in membrane permeability and disturbance in proton motive force may cause loss of cellular function and cell death (Benelli, 2016; Subramaniam et al., 2015). Generally, the active toxic ingredients of plant extracts are secondary metabolites that are evolved to protect from herbivores. The mode of action of phytochemicals in the target insect bodies could be of several types (Koul, 2008). The insects that feed on these secondary metabolites potentially encountering toxic substances with relatively non-specific effects on a wide range of molecular targets. The targets are range from proteins such as receptors, enzymes, signaling molecules,







(c)

(a)

(d)

Figure 6. Percentage mortality \pm SD of green synthesized AgNPs on *Ae. aegypti* and *Ae. albopictus* at different concentrations: (a) Plant extract: AgNO₃ 1:10 after 24 h exposure. (b) Plant extract: AgNO₃ 1:10 after 48 h exposure (c) Plant extract: AgNO₃ 2:10 after 24 h exposure (d) Plant extract: AgNO₃ 2:10 after 48 h exposure (There was no % mortality in 0 mg/L control treatment).

Product	Mosquito species	Exposure period (hours)	LC ₅₀ (mg/L)	95 % confidence interval for LC_{50}	
				LCL (mg/L)	UCL (mg/L)
(Plant extract: AgNO ₃ 1: 10)	Ae. aegypti	24	5.29	5.08	5.49
		48	1.51	1.34	1.65
	Ae. albopictus	24	3.02	2.86	3.17
		48	1.14	1.01	1.33
(Plant extract: AgNO ₃ 2: 10)	Ae. aegypti	24	2.43	2.19	2.45
		48	1.17	1.01	1.36
	Ae. albopictus	24	2.51	2.4	2.64
		48	2.10	2.01	2.18

ion-channels, and structural proteins; nucleic acids, bio membranes and other cellular components (Rattan, 2010).

In the case of the Family Annonaceae, the literature search indicated that only seven annonaceous plant species have been studied for their mosquito larvicidal properties with extracts from the genus *Annona* showing strong insecticidal activities. However, with about 150 plant species known from the genus *Annona* only 4 species; *A. crassiflora, A. glabra, A. muricata, A. squamosa* have been studied for mosquito larvicidal activities (Das et al., 2007). *A. crassiflora* showed larvicidal activity against *Ae. aegypti* (De Omena et al., 2007). *A. squamosa* have a larvicidal activity against *Ae. albopictus* and *Culex quinquefasciatus* (Das et al., 2007). Seed extract of *A. muricata* showed larvicidal activity against *Ae. aegypti* (Promisiri et al., 2006). De Omena et al. (2007) reported the larvicidal efficacy of ethanol stem bark extract of *A. glabra*

against *Ae. aegypti*. As reported by Amarasinghe and Ranasinghe (2017), *Annona glabra* leaves showed a mosquito larvicidal effect. Present study revealed that the An-AgNPs (Plant extract: AgNO₃ 1:10) has enhanced significantly the efficacy of *A. glabra* showing LC₅₀ values for *Ae. albopictus* and *Ae. aegypti* larvae as 3.02 mg/L and 5.29 mg/L respectively for 24 h exposure compared to the LC₅₀ values for *A. glabra* crude leaf aqueous extract (Figure 6 a-d; Table 2). Comparable results were reported by Khader et al. (2017) from a study conducted on *Annona squamosa* crude extractand its biosynthesized AgNPs against different species of mosquito larvae. Among the nano products tested in this study; An-AgNPs (Plant extract: AgNO₃ 2:10) is more suitable than An-AgNPs (Plant extract: AgNO₃ 1:10) for its further development as a mosquito larvicide.



Figure 7. Mean mortality percentage of Annona glabra leaf extract on Ae. aegypti and Ae. albopictus at different concentrations; (a) after 24 h and (b) after 48 h exposure.

	Table 2. Larvicidal activit	y of A. glabra leaf	extract against Ae.	aegypti and Ae. albopictus.
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Mosquito species	Exposure period (hours)	LC ₅₀ (mg/L)	95 % confidence interval for LC ₅₀	
			LCL (mg/L)	UCL (mg/L)
Ae. aegypti	24	5.94555	5.33	6.5585
	48	3.5485	3.06765	3.99527
Ae. albopictus	24	5.00402	4.43068	5.55518
	48	2.73467	2.31874	3.11563

LCL - Lower Confidence Limit UCL - Upper Confidence Limit.

5. Conclusions

Present study achieved a successful synthesis of AgNPs using *A. glabra* aqueous plant extract (An- AgNPs). There is a strong larvicidal activity of An-AgNPs aqueous leaf extract against dengue vector mosquitoes, *Ae. aegypti* and *Ae. albopictus*. An-AgNPs synthesized to the ratio of 2:10 (plant extract: AgNO₃) found to be more effective than that synthesized to the ratio 1:10. This further concludes that *A. glabra* leaf extract can be used as an effective capping agent as well as the reducing agent for the synthesis of silver nanoparticles as a larvicide.

Declarations

Author contribution statement

L.D. Amarasinghe, P.A.S.R. Wickramarachchi: Conceived and designed the experiments; Wrote the paper.

A.A.A.U. Aberathna: Performed the experiments; Analyzed and interpreted the data; Wrote the paper.

W. S. Sithara: Performed the experiments; Analyzed and interpreted the data.

C. R. De Silva: Analyzed and interpreted the data.

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Competing interest statement

The authors declare no conflict of interest.

Additional information

No additional information is available for this paper.

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