

Journal of Traditional and Complementary Medicine

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# Triterpenes from Kadsura coccinea

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## Abstract

In vitro anti-allergic screening of medicinal herbal extracts revealed that the 80% acetone extract of the rhizome of *Kadsura coccinea* inhibited nitric oxide (NO) production in a lipopolysaccharide (LPS) and recombinant mouse interferon- $\gamma$  (IFN- $\gamma$ ) activated murine macrophage like cell line, RAW264.7. Further fractionation of the EtOAc extract led to the isolation of one new 3,4-seco-lanostane type triterpene named kadsuracoccin acid A (1) together with two known triterpenes anwuweizonic acid (2) and neokadsuranic acid B (3). Compounds 1–3 did not exhibit any inhibitory activities for NO production and IFN- $\gamma$  activation.

Key words: Kadsura coccinea; Triterpene; NO production inhibitory activities

## Introduction

Schisandraceae family plants are known to be a rich source of lignanoids and triterpenoids with various biological activities (Zhonghuabencao Compilation Committee, 1999). The genus *Kadsura* (Schisandraceae) is closely related to *Schisandra*, and many of its species are extensively used as a substitute for *Schisandra* in Chinese Medicine in Taiwan, Japan and the mainland of China.

In our previous study of anti-inflammatory compounds from Kadsurae Coccineae Caulis et Folium (黑老虎 hēi lǎo hǔ; the stem and leaf of *K. coccinea*), a series of lignans were isolated from the CHCl3 extract (Li et al., 2006a; 2006b). To continue our search for anti-inflammatory compounds from *K. coccinea*, we studied EtOAc extract of this plant. Fractionation of this extract led to isolation of one new 3,4-secolanostane type triterpene named kadsuracoccin acid A (1) and two known triterpenes anwuweizonic acid (2) and neokadsuranic acid B (3). Herein, we describe the structure elucidation of compound 1 (Figure 1) and the biological evaluation of three compounds (1–3).

## Material and Methods

## General

Mp: Fisher-Johns melting point apparatus; uncorrected. IR Spectrum: Mattson Genesis II FT-IR spectrometer, in cm<sup>-1</sup>. The UV spectrum was obtained in MeOH on a Shimadzu UV-160 spectrophotometer. The NMR spectra were recorded on a Varian JMercury-300BB spectrometer, with TMS as an internal standard. The mass spectra (MS) were obtained on a Jeol ECA-600 spectrometer. Column chromatography was carried out with silica gel (Wako gel C-300, Wako Pure Chemical Industry Ltd.). Thin-layer chromatography (TLC) was performed on Merck TLC plates (0.25 mm thickness), with compounds visualized by spraying with 5% (v/v)  $H_2SO_4$  in ethanol solution and then heating on a hot plate. HPLC was performed on a JASCO PU-2089 apparatus equipped with a JASCO UV-2075. A COSMOSIL (10 × 250 mm i.d.) column, Fluofix 120N  $(10 \times 250 \text{ mm i.d.})$  and a Senshu Pak PEGASIL ODS 2 were used for preparative purposes.

#### **Plant material**

The dried rhizome of *K. coccinea* was collected in Guangxi Province, People's Republic of China, in April 2004 and was identified by Dr. Bao-Lin Guo, Peking Union Medical College, Beijing, People's Republic of China. Voucher specimens were deposited in the Department of Pharmacognosy, College of Pharmacy, Nihon University.

#### **Extraction and isolation**

The dried rhizome of K. coccinea (1.75 kg) was extracted three times with 80% acetone. Evaporation of the solvent under reduced pressure from the combined extract gave the extract 82.5 g (NO inhibitory effect 100 µg/ml, 78.5%). The extract was dissolved and suspended in water (2 L) and partitioned with chloroform  $(3 \times 2 L)$ , ethyl acetate  $(3 \times 2 L)$ , and n-butanol (3  $\times$  2 L). The amounts extracted were 44.7 g (47.8%), 4.0 g (4.3%), and 14.2 g (15.4%), respectively, and the residual aqueous extract yielded 20.6 g (24.9%). The EtOAc fraction was subjected to CHP-20P column chromatography (40  $\times$  1000 mm, eluted with H<sub>2</sub>O and MeOH in reducing polarity, H2O-MeOH, 100:0  $\rightarrow$ 60:40). The column chromatographic fractions (100 mL each) were combined according to TLC monitoring into seven portions (toluene-EtOAc-acetic Acid, 70:33:3). Portion seven was further purified by HPLC (Fluofix-120N, 10 × 250 mm, H<sub>2</sub>O-CH<sub>3</sub>CN, 5:95, 3 mL/min, UV detector  $\lambda = 210$  nm) to give 1 (20.2 mg), 2 (10.1 mg) and 3 (3 mg).

Compound 1: Colorless prisms; mp 142-144 °C; UV (MeOH)  $\lambda$ max nm (log  $\varepsilon$ ): 202 (3.69); IR (KBr) cm<sup>-1</sup>: 3433, 3076, 1739, 1690, 1272; EI-MS *m/z* (rel. int. %): 482 [M]<sup>+</sup> (80), 383 (100), 275 (40), 95 (65); HR-EI-MS *m/z*: 482.3391 ([M]<sup>+</sup>, Calcd 482.3396 for C<sub>31</sub>H<sub>46</sub>O<sub>4</sub>); <sup>1</sup>H- and <sup>13</sup>C-NMR: see Table 1.

## Inhibitory Activity on NO Production from Activated Macrophages-Like Cell Line, RAW 264.7

The cells were seeded at  $1.2 \times 10^6$  cells/ml onto 96-wells flat bottom plate (Sumitomo Bakelite, #8096R, Tokyo) and then incubated at 37 °C for 2 h. Next, the test extract was added to the culture simultaneously with both Escherichia coil LPS (100 ng/mL) and recombinant mouse IFN-  $\gamma$  (0.33 ng/mL). Then cells were incubated at 37 °C for approximately 16 h and subsequently chilled on ice. One hundred micro liters of the culture supernatant was placed in duplicate in the wells of 96-well flat bottomed plates. A standard solution of NaNO<sub>2</sub> was placed in alternate wells on the same plate. To quantify nitrite, 50 µL of Griess reagent (1% sulfanilamide in 5% H<sub>3</sub>PO<sub>4</sub> and 0.1% N-1-naphthyletylenediamide dihydrochloride) was added to each well. After 10 min the reaction products were colorimetrically quantified at 550 nm using a Model 3550 Microplate Reader (BIO-RAD) and the background absorbance (630 nm) was subtracted. Cytotoxicity was measured by the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay method.

#### **Results and Discussion**

Kadsuracoccin acid A (1) was crystallized as colorless prisms and has the molecular formula of  $C_{31}H_{46}O_4$  by its HREIMS (found 482.3391, calcd 482.3396). Inspection of the <sup>13</sup>C-NMR spectrum revealed the presence of two carboxyl signals at  $\delta c$  175.1 and 172.6 (C-3 and C-26, respectively). The 1H-NMR spectrum indicated the three olefinic methyls at  $\delta H$  1.56, 1.74 and 1.90, one isopropenyl (*exo*-methylene protons at  $\delta_H$  4.85 and 4.69, H-28a and H-28b;  $\delta_H$  1.74, CH<sub>3</sub>-29), and three tertiary methyls at  $\delta_H$  0.70, 0.80 and 1.04, respectively.

Table 1. <sup>1</sup>H- and <sup>13</sup>C-NMR Data for Compound 1 (600 MHz, 150 MHz in CDCl3,  $\delta$  in ppm) <sup>a)</sup>

|    |                  | r r r r r r r r r r r r r r r r r r r  | · · · · · · · · · · · · · · · · · · |                  |  |
|----|------------------|--|-------------------------------------|------------------|--|
|    | $\delta_{\rm C}$ | $\delta_{\rm H}$ (mult; <i>J</i> , Hz) |                                     | $\delta_{\rm C}$ | $\delta_{\rm H}$ (mult; <i>J</i> , Hz) |
| 1  | 28.9             | 2.21, m; 2.47, m                       | 17                                  | 142.9            |  |
| 2  | 32.3             | 1.78, m; 2.04, m                       | 18                                  | 21.2             | 0.80, s                                |
| 3  | 175.1            |  | 19                                  | 26.7             | 1.04, s                                |
| 4  | 147.6            |  | 20                                  | 124.8            |  |
| 5  | 49.3             | 2.04, m                                | 21                                  | 19.6             | 1.56, s                                |
| 6  | 27.8             | 1.53, m; 1.78, m                       | 22                                  | 33.0             | 2.05, m; 2.27 , m                      |
| 7  | 26.7             | 1.23, m; 1.63, m                       | 23                                  | 29.3             | 2.47, m; 2.69, m                       |
| 8  | 41.5             | 2.09, m                                | 24                                  | 146.2            | 6.08, (t, 7.3)                         |
| 9  | 142.1            |  | 25                                  | 126.0            |  |
| 10 | 42.4             |  | 26                                  | 172.6            |  |
| 11 | 118.4            | 5.35 (d, 6)                            | 27                                  | 20.5             | 1.90, s                                |
| 12 | 36.5             | 2.24, m; 2.41, m                       | 28                                  | 113.7            | 4.69, (d, 1.7); 4.85, (d, 1.7)         |
| 13 | 47.2             |  | 29                                  | 23.2             | 1.74, s                                |
| 14 | 46.1             |  | 30                                  | 19.1             | 0.70 s                                 |
| 15 | 33.0             | 1.42, m; 2.06, m                       | -OCH <sub>3</sub>                   | 51.5             | 3.65, s                                |
| 16 | 28.8             | 2.20, m; 2.48, m                       |                                     |                  |  |
|    |                  |  |                                     |                  |  |

<sup>a)</sup> Data were recorded on Jeol ECA-600MHz spectrometer (<sup>1</sup>H, <sup>13</sup>C), assignments were confirmed by <sup>1</sup>H-<sup>1</sup>H COSY, HMQC and HMBC.



Figure 1. Structures of compounds 1-3.

Furthermore, two pair of non-equivalent methylene protons at  $\delta_{\rm H}$  2.21, 2.47 (H-1) and  $\delta_{\rm H}$  2.04, 1.78 (H-2) showed correlations in the COSY spectrum (Figure 2) with each other, and both of them showed long-range correlations with C-3 in the HMBC spectrum. An ester methyl signals at  $\delta_{\rm H}$  3.65 and  $\delta_{\rm C}$  51.5 were found in <sup>1</sup>H and <sup>13</sup>C-NMR spectrum. Additionally, the HMBC cross peak of ester methyl signal ( $\delta_{\rm H}$  3.65) with C-3 ( $\delta_{\rm C}$  175.1) confirmed the presence of a methyl ester.

The COSY spectrum of 1 also showed crosspeaks between  $\delta_{\rm H}$  1.90 (CH<sub>3</sub>-27) and 6.08 (t, J = 7.3Hz, H-24); H-24 with two non-equivalent methylene protons at  $\delta_{\rm H}$  2.47, 2.69 (H-23). The HMBC spectrum of 1 showed the long-rang correlations between CH<sub>3</sub>-27 and carbons at  $\delta_{\rm c}$  172.6 (C-26), 126.0 (C-25) and 146.2 (C-24); H-24 with C-26, C-25, C-27 ( $\delta_{\rm C}$  20.5), C-23 ( $\delta_{\rm C}$  29.3) and C-22 ( $\delta_{\rm C}$  33.0). The facts above indicated the presence of a side-chain with correlated protons: HOOC-C(CH3)=CH-CH<sub>2</sub>-CH<sub>2</sub>. Irradiation of CH<sub>3</sub>-27 resulted in NOE of H-24 showed the double-bond in C-24 has a Z-configuration.

These data above were closely related to those previously reported kadsuric acid (Yamada et al., 1976), but clearly differed in the signals corresponding to the CH<sub>3</sub>-21, C-17 and C-22. Both proton signals of methine on CH-17 and CH-20 disappeared. And the most apparently, the three-proton doublet (CH<sub>3</sub>-21) disappeared and downshifted to  $\delta_{\rm H}$  1.56 as a single signal. The facts above indicated the presence of a double bond between C-17 and C-20. The orientation of H-8 was determined on the basis of NOSEY correlations of CH<sub>3</sub>-19/H-8 and CH<sub>3</sub>-18/H-8 (Figure 2). In addition, the correlations of CH<sub>3</sub>-30/H-5 suggested



Figure 2. Key COSY, HMBC correlations and NOEs of compound 1.

that H-5 has an  $\alpha$ -orientation; H-16 at  $\delta_{\rm H}$  2.20 showed cross peaks with CH<sub>3</sub>-21 and CH<sub>3</sub>-18 at  $\delta_{\rm H}$  0.80 showed cross peaks with non-equivalent methylene protons at  $\delta_{\rm H}$  2.05, 2.27 (CH<sub>2</sub>-22) indicating the double bond in C-17 has a Z-configuration. Thus, the structure of 1 was assigned as shown in Figure 1. Thus 1 was suggested to be kadsuracoccinic acid A ester (Li et al., 2008). Therefore, the structure of 1 was suggested to be 3,4-*seco*-lanosta-4(28),9(11),17(20),17(Z),24(Z)-tetraene-26-oic acid-3-oic acid methyl ester.

Compounds 2 and 3 (Ito et al., 2001; Li et al., 1989) were known compounds, whose structures were elucidated by comparison with those in the literature.

Compounds 1–3 were tested for their ability to inhibit nitric oxide (NO) production in a LPS and IFN- $\gamma$  activated murine macrophage like cell line RAW 264.7.Compounds 1–3 did not exhibit any inhibitory activity.

#### Acknowledgements

Project 30901853 supported by NSFC; The Projectsponsored by SRF for ROCS, SEM; the project 09KJB350002 supported by University Science Research Project of Jiangsu Province; the research work was supported by the PAPD (A Project Funded by the Priority Academic Program Development of Jiangsu Higher Education Institutions).

#### References

- Ito, J., Chang, F.R., Wang, H.K., Park, Y.K., Ikegaki, M., Kilgore, N., Lee, K.S., 2001. Anti-AIDS Agents .Anti-HIV Activity of Moronic Acid Derivatives and the New Melliferone-Related Triterpenoid Isolated from Brazilian Propolis. J. Nat. Prod. 64, 1278-1281.
- Li, H.R., Feng, Y.L., Yang, Z.G., Wang, J., Daikonya, A., Susumu, K., Xu, L.Z., Yang, S.L., 2006a. New lignans from Kadsura coccinea and their nitric oxide inhibitory activities. Chem. Pharm. Bull. 54, 1022-1025.

- Li, H.R., Feng, Y.L., Yang, Z.G., Wang, J., Daikonya, A., Susumu, K., Xu, L.Z., Yang, S.L., 2006b. New lignans from Kadsura coccinea and their nitric oxide inhibitory activities. Heterocycles 68, 1259-1265.
- Li, H.R., Wang, L.Y., Miyata, S., Kitanaka, S., 2008. Kadsuracoccinic Acids A-C, Ring-A seco-Lanostane Triterpenes from kadsura coccinea and their effects on embryonic cell division of Xenopus laevis. J. Nat. Prod. 71, 739-741.
- Li, L.N., Xue, H., Kangouri, K., Kawashima, A., Omura, S.,1989. Triterpenoid acids from Kadsura heteroclita; III. Isolation and structure elucidation of 12β- acetoxycoccinic acid, 12β-hydroxycoccinic acid, 12α-acetoxycoccinic acid, and 12α-hydroxycoccinic acid. Planta Med. 55, 548-550.
- Yamada, Y., Hsu, C.S., Iguchi, K., Suzuki, S., Hsu, H.Y., Chen, Y.P.,1976. Structure of kadsuric acid. A new seco-triterpenoid from Kadsura japonica Dunal. Chem. Lett. 12, 1307-1310.
- Zhonghuabencao Compilation Committee, 1999. Zhonghuabencao. Shanghai Science and Technologic Publisher, Shanghai, pp.895.