



Article

Rapid Evaluation of Chemical Consistency of Artificially Induced and Natural Resina Draconis Using Ultra-Performance Liquid Chromatography Quadrupole-Time-of-Flight Mass Spectrometry-Based Chemical Profiling

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Abstract: Resina Draconis is a highly valued traditional medicine widely used in Arabia since ancient times, and it has been commonly used as an antidiarrheic, antimicrobial, antiulcer, blood circulation promoter as well as an anti-inflammatory agent. The tree source from which this medicine orignates grows extremely slowly, producing a very low yield of Resina Draconis. To meet the increasing market demand, artificial methods for stimulating Resina Draconis formation have been developed and applied. However, the chemical differences between artificially induced Resina Draconis (AIRD) and natural Resina Draconis (NRD) have been rarely studied. The aim of this research was to explore and identify the chemical constituents of AIRD and NRD using ultra-performance liquid chromatography-quadrupole-time-of-flight mass spectrometry (UHPLC-QTOF-MS/MS) based chemical profiling. A total of 56 chromatographic peaks were detected in AIRD, of these, 44 peaks have had their structures tentatively characterized based on high-resolution mass spectra (HRMS) data, fragmentation ions information, reference standards data and literature review. In total, 40 peaks were found both in AIRD and NRD. The potential chemical transformation mechanisms active in Resina Draconis during formation were explored. To the best of our knowledge, this is the first evaluation of the chemical profiles of both AIRD and NRD. Furthermore, these findings are expected to provide a rational basis for the quality assessment of AIRD and the use of AIRD as a substitute for NRD.

Keywords: *Dracaena cochinchinensis* (Lour.) S. C. Chen; artificially induced; Resina Draconis; dragon's blood; UHPLC-QTOF-MS/MS

1. Introduction

Resina Draconis (also called "dragon's blood"), is a red resin derived from *Dracaena cochinchinensis* (Lour.) S. C. Chen and called "longxuejie" in China. It is a rare and precious traditional medicine that has been commonly used in China for the treatment of wounds, leucorrhea, fractures, diarrhea, as well as intestinal and stomach ulcers since ancient times [1]. Recent pharmacological research

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has shown that Resina Draconis has antithrombotic [2], antibacterial [3], anti-inflammatory [4,5], anti-diabetic [6], anti-Helicobacter pylori [7] bioactivity, and with the potential to be a therapeutic agent for neurodegenerative diseases [8]. Previous phytochemical studies of this resin have shown that it mainly contains phenolic compounds, including flavonoids, steroids and stilbenoids, which are considered to be the biologically active components of Resina Draconis [1]. Resina Draconis plants originate from four genera of Dracaena, Daemonorops, Croton and Pterocarpus, which are found all over the world. In China, the main source of Resina Draconis is from Dracaena cochinchinensis (Lour.) S. C. Chen and Dracaena cambodiana Pierre ex Gagnep [9]. In recent years, demand for Resina Draconis in the medicinal market has increased dramatically. However, the natural Dracaena tree grows extremely slowly, taking usually 30–50 years or more, and yields Resina Draconis with a very low efficiency. To meet the increasing demands for Resina Draconis, a considerable number of artificial methods for stimulating resin formation have been developed and applied [10,11]. However, a debate has continued since the emergence of artificially induced Resina Draconis (AIRD). The focus of this controversy is that the chemical components of AIRD are thought to perhaps differ from those of its natural form, which make their efficacy not equivalent. In fact, few of the chemical constituent of AIRD have been reported [10,12,13], however, until now, the chemical profile differences between AIRD and natural Resina Draconis (NRD) have been rarely studied. Whether the chemical constituents of AIRD consistent to that of NRD and how the secondary metabolites produced during Resina Draconis formation have are very important concerns for the efficacy and quality control of AIRD. Therefore, development of a generally reliable, sensitive, and confirmatory analytical method to examine the chemical constituents of both AIRD and NRD is desirable.

In recent years, tandem ultra-high-performance-liquid-chromatography with quadrupole time of flight mass spectrometry (UHPLC-QTOF-MS/MS) has been used as a rapid and effective technique to identify compounds in complex matrices. Compared with the low-resolution MS methods such as quadrupole, triple quadrupole and ion trap mass spectrometry, Q-TOF MS/MS has the ability to measure the exact mass for both precursor and fragment ions, which could be valuable for determination of structural conformation for non-target compounds in complex matrices when the reference compounds are unavailable. Thus, the present study aimed to investigate the chemical constituents in both AIRD and NRD using UHPLC-QTOF-MS/MS based chemical profiling. A salt solution containing 1% zinc sulfate and 2.0 g/L benzoic acid was used to treat the xylem of D. cochinchinensis to induce Resina Draconis formation by transpiration for 80 days. Then the chemical fingerprints of both AIRD and NRD were compared in both negative and positive ion modes by an improved UHPLC-QTOF-MS/MS analysis. The identities of all detected peaks, were confirmed by comparing the mass spectra and retention times with those of available reference compounds, and/or tentatively assigned by matching empirical molecular formula with those of published compounds, elucidating quasi-molecular ions and fragment ions referring to the available literature information. The potential chemical transformation mechanism of Resina Draconis during formation was further explored. The established approach was applied to rapidly identify the chemical profiles of AIRD and NRD, which offered a systematical and reliable approach for quality assessment and control of AIRD.

2. Results

2.1. Optimization of Chromatographic Conditions and TOFMS Method

In order to achieve a rapid and efficient analysis, the UHPLC analysis method was optimized. Different columns including Waters Acquity BEH C18 (100 mm \times 2.1 mm, 1.7 m), Waters Acquity HSS T3 (100 mm \times 2.1 mm, 1.8 m), Waters BEH HILIC (100 mm \times 2.1 mm, 1.7 m), and different temperatures (i.e., 25, 35, 40, and 50 °C) were tested. We found that ACQUITY BEH C18 column (100 mm \times 2.1 mm, 1.7 µm) at a temperature of 35 °C had the best baseline separation of the most of the constituents present in AIRD and NRD. Different kinds of mobile phases, including organic phase (MeOH, acetonitrile), variety of aqueous phase (water, water containing formic acid, water containing

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formic acid and ammonium) were tested. It was finally determined that formic acid added to the mobile phase not only improved the chromatographic peaks but also generated $[M - H + HCOOH]^{-}$ adduction, which was helpful for the confirmation of molecular ions $[M-H]^-$. Thus, a mixture system of acetonitrile-water (containing 0.1% formic acid) was finally selected to get the greatest separation, response value and better peak shapes. Different MS parameters (i.e., detection mode, capillary, cone voltage, desolvent gas flow rate, desolvent gas temperature, cone gas flow rate, and source temperature, induced dissociation energy) were investigated as well. Comparison of peak purity and a number of characteristic peaks were acquired under different ion modes. It was found that the negative-ion ESI-QTOF-MS produce more information and better signal-to-noise (S/N) ratios for active ingredients than that in positive-ion mode, so it was suitable for multi-ingredients characterization and fingerprint analysis of AIRD and NRD. Induced dissociation (CID) voltage ramping from 35 V to 50 V was selected to produce more precursor ions and fragments ions information by only one injecting sample under the MS^E mode. With these optimized chromatographic and MS conditions, a total 56 components could be detected in AIRD and NRD samples within 22 min. The representative chromatogram of AIRD (80-day inducing stage) and NRD obtained from the analysis in ESI negative mode is shown in Figure 1.

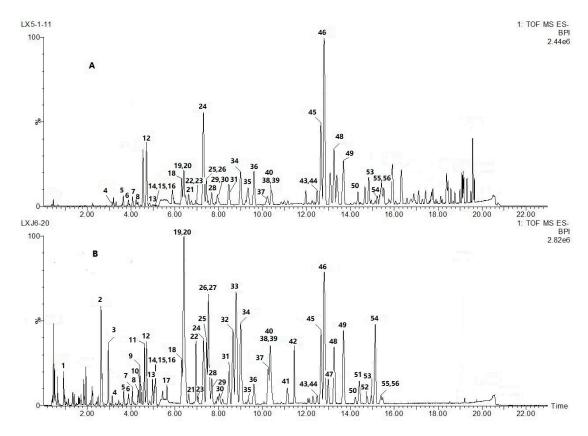


Figure 1. Representative basic peak ion (BPI) chromatograms of natural and artificial Resina Draconis in negative ion mode by ultra-high-performance-liquid-chromatography with quadrupole time of flight mass spectrometry (UHPLC-QTOF-MS/MS). (**A**) Natural Resina Draconis sample (*NRD*); (**B**) artificial induced Resina Draconis (*AIRD*) sample. The peak numbers are the same as displayed in Table 1.

2.2. Global Characterization of the Chemical Constituents in the Artificially Induced Resina Draconis by UHPLC-Q-TOF-MS/MS

UHPLC-TOF-MS/MS was performed to screen, identify, and further characterize the constituents of *AIRD* and *NRD* in both the positive and negative ionization modes. A total of 56 compounds

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were clearly detected in *AIRD* using the established method (Figure 1). Of these, 44 peaks were tentatively or unambiguously characterized by matching the molecular formulas, quasi-molecular ions, and fragmentations with those of published data for the first time. Among these, seven compounds, namely 7,4′-dihydroxyflavone, protogracillin, protodioscin, 7-hydroxyflavone, loureirin D, loureirin A and loureirin B (peaks **12**, **16**, **17**, **25**, **31**, **48**, **49**) were unequivocally identified by matching the retention times and mass spectra with those of reference compounds. A total of 40 peaks were found in both *AIRD* and *NRD*. The characterization of chemical constituents based on MS fragmentation behaviors, chromatographic retention times, and literature information are summarized in Table 1.

2.3. Characterization of Steroid Saponins in Artificially Induced Resina Draconis

Steroid saponins are one of the major bioactive compounds presented in Dracaena species. Usually, steroid saponin compounds provide accurate structural information on the glycan sequences and the aglycone structures by UHPLC–QTOF–MS/MS method. Deprotonated ions of saponins tend to produce signals with losses of several sugar moieties in the negative mode by the successive losses of hexose, pentose, or deoxyhexose (m/z 162, 132 or 146). In this work, rapid identification of 13 steroid saponins was achieved using UHPLC–QTOF-MS/MS by matching empirical molecular formulas with those of published compounds, and/or elucidating quasi-molecular ions and fragmentations referring to the fragmentation patterns as well as available information in the literature.

For example, Compound 25 ($t_R = 7.44$ min) and compound 26 ($t_R = 7.53$ min) were identified as steroid saponins in Resina Draconis. Compound 25 readily yielded a strong [M + HCOO] $^-$ at m/z1093.5447 in negative mode with the molecular formula $C_{51}H_{84}O_{22}$. The main fragment ions at m/z $901.4797 [M - H - 146]^-$, $755.4218 [M - H - 146 - 146]^-$, $593.3641 [M - H - 146 - 146 - 162]^-$, 431.3161 [M - H - 146 - 146 - 162 - 162] corresponding to the loss of the glucosyl and rhamnosyl units (Table 1), suggesting that compound 25 contained two deoxyhexose and one hexose as a terminal unit and one hexose as an inner unit at the sugar moiety. The fragment ion m/z 413.3059 was ascribed to the aglycone ion of steroid aglycones. The postulated fragmentation pathway and MS/MS spectrum of compound 25 is shown in Figure 2. Compared with the reference and literature data, compound 25 was identified as protodioscin, which has been reported in *D. cochinchinensis* previously [14]. Compound **26** ($t_R = 7.53$ min) displayed a similar molecular ion at m/z 1093.5447 [M + HCOO]⁻ and a similar molecular formula, C₅₁H₈₄O₂₂, as compound **25** in the negative ionization mode. The most abundant ion observed at m/z 901.4796 corresponds to the neutral loss of one rhamnosyl residue [M - H - $[146]^{-}$, the other ions at m/z 755.4224 [M - H - 146 - 146] [-], m/z 737.4127 [M - H - 146 - 18] [-]m/z 593.3691 [M - H - 146 - 146 - 162]⁻, and m/z 575.3599 [M - H - 146 - 146 - 162]⁻ could be assigned to the subsequent loss of one and two glucosyl, rhamnosyl residues and H₂O respectively. Furthermore, the fragmentation ions at m/2 901, 755, 737, 689, 593, 575, 431 were observed both in compound 25 and 26. Thus compound 26 was characterized as an isomer of compound 25, and was tentatively identified as protoneodioscin, which has also been isolated from D. cochinchinensis during previous study [14]. By using a similar approach, another six steroid saponin compounds (10, 16, 22, 27, 42, 54) were tentatively identified through comparison with literature data combined with the structural data from other databases (Pubmed, Mass Bank, Chemspider, etc.)

Table 1. Components identified from artificial induced Resina Draconis using UHPLC-QTOF-MS/MS.

No.	t _R /min	[M – H] ⁻	Error/ppm	MS/MS Fragments Ions	Empirical Formula	Identification	Reference Literatures
1	0.9	351.1284	-0.2	296.9812, 289.1299, 207.0851, 161.0435, 125.0228, 113.0248	C ₁₄ H ₂₄ O ₁₀	unknown	
2	2.63	413.1445	-0.7	269.1028, 217.0434, 161.0453, 143.0343, 125.0249, 113.0246	C ₁₉ H ₂₆ O ₁₀	unknown	
3	2.96	413.1437	-2.7	269.1028, 217.0434, 161.0453, 143.0343, 125.0249,113.0246	$C_{19}H_{26}O_{10}$	unknown	
4 *	3.14	487.1818 (M + HCOO ⁻)	0.4	441.1693, 307.1003, 247.0826, 214.0602, 163.0604, 145.0508, 133.0649, 125.0239	$C_{21}H_{30}O_{10}$	unknown	
5 *	3.66	1129.4753	0.1	983.4068, 967.4224, 866.0862, 784.9418, 762.1688, 696.7462, 645.0305, 496.3572	$C_{46}H_{82}O_{31}$	saponins	
6 *	3.88	967.4213	-2.2	821.3612, 788.3539, 741.3986, 591.3541, 447.3194, 357.0483, 150.9695	$C_{40}H_{72}O_{26}$	saponins	
7*	4.05	817.422 (M + HCOO ⁻)	-2.2	663.2785, 261.0729, 168.0311, 173.0305, 165.0199	C ₃₉ H ₆₄ O ₁₅	Saponins (spirostanol monodesmosides)	
8 *	4.33	689.3203	4.4	671.3118, 527.2696, 411.1845, 369.1389, 327.1284, 233.0507, 161.1837, 150.4524	C ₃₆ H ₅₀ O ₁₃	unknown	
9	4.4	933.4703 (M + HCOO ⁻)	0.1	741.4053, 723.3961, 609.3652, 579.3525, 465.3005, 447.3121, 429.3012, 403.2820, 161.0430, 159.0315, 143.0380, 113.0242	C ₄₅ H ₇₄ O ₂₀	saponins	
10	4.49	1107.5237 (M + HCOO ⁻)	1.3	915.4579, 769.3976, 607.3532, 589.3440, 541.3199, 247.0806, 179.0542, 161.0447, 143.0329, 131.0348, 113.0233	$C_{51}H_{82}O_{23}$	Dracaenoside Q	[14]
11	4.6	787.4115 (M + HCOO ⁻)	-0.8	609.3638, 447.3108, 429.3003, 403.2857, 179.0560, 161.0455, 159.0297, 143.0339, 131.0338, 119.0346, 113.0239	$C_{38}H_{62}O_{14}$	Saponins	
12	4.7	253.0499	-0.8	235.0349, 224.0459, 208.0515, 196.0493, 180.0578, 135.0086, 117.0346	$C_{15}H_{10}O_4$	7,4'-dihydroxyflavone **	[15]
13 *	4.84	281.0812	-0.7	265.0502, 237.0547, 221.0622, 209.0598, 194.0729, 181.0645, 167.0500, 149.0234	C ₁₇ H ₁₄ O ₄	5,4'-dimethoxy-7-hydroxyflavylium #	[16]
14 *	4.97	299.0912	-2.3	193.0507, 178.0277, 149.0605, 139.0403, 117.0344	$C_{17}H_{16}O_5$	5,7,4'-trihydroxy-6-methyl-dihydrohomoisoflavone	[13]
15 *	5.07	283.0603	-1.1	268.0374, 240.0429, 211.0405, 196.0546, 183.0444, 135.0089	$C_{16}H_{12}O_5$	thevetiaflavone	[16]
16 *	5.11	1109.5398 (M + HCOO ⁻)	-0.7	917.4685, 901.4780, 755.4103, 593.3311, 247.0828, 199.0430, 179.0604, 163.0623, 119.0364, 113.0261	$C_{51}H_{84}O_{23}$	protogracillin **	[14]

 Table 1. Cont.

No.	t _R /min	[M – H] ⁻	Error/ppm	MS/MS Fragments Ions	Empirical Formula	Identification	Reference Literatures
17	5.64	287.0914	-1.7	271.0491, 177.0198, 151.0403, 147.0445, 134.0370, 124.0161, 120.0211, 108.0219	$C_{16}H_{16}O_5$	loureirin D **	[13]
18 *	6.31	269.0811	-1.1	253.0495, 225.0549, 211.0393, 161.0240, 136.0159, 117.0338	C ₁₆ H ₁₄ O ₄	4,4'- dihydroxy-2'-methoxyl-chalcone	[17]
19 *	6.41	269.0812	-0.7	253.0486, 229.4882, 209.0552, 163.0389, 148.0151, 135.0073, 119.0496, 109.0280	$C_{16}H_{14}O_4$	2,4'-dihydroxy-2'-methoxychalcone	[15]
20 *	6.48	269.0810	-1.5	253.0464, 237.0530, 201.4554, 163.0385, 148.0148, 135.0431, 119.0487, 109.0276	$C_{16}H_{14}O_4$	7,4'-dihydroxyhomoisoflavanone	[16]
21 *	6.62	269.0811	-1.1	253.0477, 225.0581, 211.0390, 163.0400, 148.0165, 135.0450, 119.0485, 109.0270	$C_{16}H_{14}O_4$	2',4'-dihydroxy-2-methoxylchalcone	[16]
22 *	6.96	947.484 (M + HCOO ⁻)	0.3	755.4211, 739.4275, 593.3695, 575.3596, 431.3154, 413.3112, 247.0826, 179.0577, 163.0603, 143.0346, 131.0364, 119.0330	$C_{45}H_{74}O_{18}$	trigofoenoside A $^{\#}$	
23 *	7.05	271.0967	-1.1	253.0498, 187.0408, 177.0198, 165.0199, 151.0041, 145.0299, 119.0504, 107.0138	$C_{16}H_{16}O_4$	7,3,4'-trihydroxy-8-methyl-flavane	[13]
24 *	7.30	271.0965	-1.8	253.0517, 243.0705, 225.0564, 215.0696, 197.0617, 185.0628, 161.0618, 151.0043, 135.0458, 120.0221	$C_{16}H_{16}O_4$	7,4'-dihydroxy-3'-methoxyl-flavane	[13]
25 *	7.44	1093.5447 (M + HCOO ⁻)	1.1	901.4803, 755.4205, 737.4112, 689.3853, 671.3862, 606.6843, 593.3641, 575.3621, 460.5794, 431.3161, 413.3059, 349.1158, 163.0628	$C_{51}H_{84}O_{22}$	protodioscin **	[14]
26 *	7.53	1093.5413 (M + HCOO ⁻)	1.2	901.4796, 755.4224, 737.4127, 689.3885, 593.3691, 575.3599, 431.3156, 179.0574, 163.0611, 143.0357, 119.0354, 113.0241	C ₅₁ H ₈₄ O ₂₂	protoneodioscin	[14]
27	7.58	1079.5287 (M + HCOO ⁻)	0.2	901.4780, 887.4639, 739.4208, 689.3886, 593.3665, 577.3621, 431.3163, 340.1174, 247.0835, 205.0723, 179.0560, 163.0625, 131.0351, 119.0346, 113.0249	$C_{50}H_{82}O_{22}$	desgalactotigonin #	
28 *	7.68	257.0809	-1.9	151.0398, 147.0452, 135.0078, 134.0374, 120.0215, 109.0283	C ₁₅ H ₁₄ O ₄	2,4,4'-trihydroxydihydrochalcone	[18]
29 *	7.94	301.1075	-0.3	285.0396, 268.0385, 257.0410, 241.0503, 213.0545, 199.0385, 185.0575, 164.0116, 151.0039, 136.0164	C ₁₇ H ₁₈ O ₅	2,4,4'-trihydroxy-3'-methoxyl-3-methyl-dihydroc	halcone [13]
30 *	8.02	301.0707	-1.7	285.0409, 268.0372, 258.0515, 242.0588, 174.0330, 164.0129, 151.0038, 136.0139	C ₁₆ H ₁₄ O ₆	3,2',3',4'-tetrahydroxy-4-methoxyl-chalcone	[13]
31 *	8.46	237.0546	-2.5	208.0531, 193.0663, 180.0579, 165.0709, 153.0710, 143.0498, 135.0087, 132.0213	$C_{15}H_{10}O_3$	7-hydroxyflavone **	[19]

 Table 1. Cont.

No.	t _R /min	[M – H] ⁻	Error/ppm	MS/MS Fragments Ions	Empirical Formula	Identification	Reference Literatures
32	8.64	299.0915	-1.3	281.0819, 266.0580, 178.0275, 163.0040, 136.016, 108.0216	$C_{17}H_{16}O_5$	7,3'-dihydroxy-4'-methoxyl-dihydrohomoisoflavone	[13]
33	8.78	285.0758	-1.8	270.1771, 268.9290, 252.2793, 246.3228, 179.0353, 151.0036, 135.0444, 122.0376	$C_{16}H_{14}O_5$	3,2',4'-trihydroxy-4-methoxyl-chalcone	[20]
34 *	9.00	285.0758	-1.8	267.0646, 243.0655, 215.0707, 199.0771, 187.0404, 165.0198, 119.0507, 121.0301	$C_{16}H_{14}O_5$	7,4'-dihydroxy-5-methoxyflavanone	[16]
35 *	9.34	283.0601	-1.8	265.0530, 250.0248, 239.0720, 224.0503, 215.0722, 163.0407, 135.0460, 121.0299	$C_{16}H_{12}O_5$	3',7-dihydroxy-4'-methoxylflavone	[21]
36 *	9.60	315.0862	-2.2	299.0552, 191.0345, 178.0281, 165.191, 150.0320, 134.0373, 121.0303, 108.0222	$C_{17}H_{16}O_6$	3,7,4'-trihydroxy-5-methoxy homoisoflavanone	[16]
37 *	10.26	301.1077	0.3	271.1297, 207.0655, 177.0197, 164.0488, 153.0560, 147.0453, 134.0457,120.0220	$C_{17}H_{18}O_5$	4,4'-dihydroxy -2,6-dimethoxydihydrochalcone	[16]
38 *	10.32	285.0759	-1.4	269.0446, 241.0507, 200.0438, 197.0602, 177.0190, 165.0181, 119.0482	$C_{16}H_{14}O_5$	7,4'-dihydroxy-3'-methoxyflavanone	[16]
39 *	10.34	315.0865	-1.3	297.0767, 282.0535, 189.0201, 178.9994, 152.0123, 124.0174	$C_{17}H_{16}O_6$	5,7-dihydroxy-2',4'-dimethoxyl-isoflavanone	[13]
40 *	10.42	239.0709	0.2	221.0603, 211.0737, 197.0606, 169.0661, 148.0173, 135.0084, 120.0208, 109.0279	$C_{15}H_{12}O_3$	7-hydroxyflavanone	[19,22]
41	11.13	281.0808	-2.5	251.0500, 237.0548, 221.0609, 209.0602, 193.0660, 160.0160, 153.0180, 135.0083	C ₁₇ H ₁₄ O ₄	5,7-dimethoxyflavone	[23]
42	11.44	1135.5547 (M + HCOO ⁻)	-0.2	1047.5375, 1029.5267, 943.4902, 901.4797, 883.4693, 755.4202, 737.4052, 689.3869, 673.3993, 593.3668, 575.3519, 497.2161, 431.3281, 413.3172, 247.0783, 205.0778, 179.0491, 163.0622	$C_{53}H_{86}O_{23}$	spongioside B [#]	[24]
43 *	12.48	253.0405	1.6	237.0549, 209.0597, 193.0652, 161.0239, 136.0158, 120.0211	$C_{15}H_{10}O_4$	5,7-dihydroxyflavone	[16]
44 *	12.48	283.0965	-1.8	268.0368, 241.0054, 197.0655, 161.0236, 146.0345, 134.0363, 120.0215, 106.0423	$C_{17}H_{16}O_4$	5,7-dimethoxyflavanone	[16]
45 *	12.66	253.0860	-2.0	237.0554, 209.0608, 193.0658, 161.0243, 136.0166, 120.0214, 108.021	C ₁₆ H ₁₄ O ₃	7-methoxydihydroflavone	[25]
46 *	12.80	255.0656	-0.4	213.0561, 185.0613, 171.0456, 164.0122, 151.0041, 145.0662, 136.0166, 107.0139	C ₁₅ H ₁₂ O ₄	7,4'-dihydroxyflavanone	[13]
47	12.99	329.1019	-1.8	311.0912, 296.0675, 208.0363, 193.0130, 190.0261, 166.0257, 138.0305, 121.0276	C ₁₈ H ₁₈ O ₆	cambodianol	[13]
48 *	13.26	285.1123	-1.4	270.0906, 257.0449, 242.0938, 229.0582, 147.0449, 134.0369, 120.0212, 106.0419	C ₁₇ H ₁₈ O ₄	loureirin A **	[13,26]

 Table 1. Cont.

No.	t _R /min	[M – H] ⁻	Error/ppm	MS/MS Fragments Ions	Empirical Formula	Identification	Reference Literatures
49 *	13.69	315.1229	-1.0	299.0902, 283.0934, 253.0513, 191.0337, 165.0559, 147.0450, 134.0370, 120.0212	$C_{18}H_{20}O_5$	loureirin B **	[13,26]
50 *	14.22	315.1232	0.0	300.0907, 283.0933, 206.0605, 191.0368, 165.0552, 147.0471, 134.0449, 120.0205	$C_{18}H_{20}O_5$	unknown	
51	14.41	297.0763	0.0	281.0450, 265.0530, 253.0509, 237.0547, 225.0542, 209.0599, 195.0456, 176.0114, 151.0037, 130.0424, 125.0247, 107.0136	C ₁₇ H ₁₄ O ₅	7,4'-dihydroxy-5-mehtoxy-8-metllylflavone	[23]
52	14.76	299.0915	-1.0	284.0689, 255.0996, 193.0474, 178.0253, 150.0305, 122.0355	C ₁₇ H ₁₆ O ₅	7-methoxy-5,4'-dihydroxy-8-methylflavanone	[16]
53 *	14.95	585.2125 (M + HCOO ⁻)	0.7	429.1734, 405.1335, 387.1245, 285.1133, 281.0823, 253.0856, 163.0389, 151.0422, 147.0450, 133.0667	$C_{33}H_{32}O_{7}$	flavonoid dimers	
54 *	15.13	929.4763	0.0	737.4115, 591.3526, 525.3197, 429.3015, 349.1093, 247.0842, 205.0735, 163.0618	$C_{45}H_{72}O_{17}$	cambodianoside C	[27]
55 *	15.38	239.0706	-0.8	221.0590, 211.0755, 197.0598, 169.0648, 148.0149, 135.0071, 120.0195, 109.0279	C ₁₅ H ₁₂ O ₃	6-hydroxyflavanone	[16]
56 *	15.47	269.0807	-2.6	251.0717, 241.0869, 227.0703, 165.0177, 155.0852, 150.0308, 137.0219, 121.0277	$C_{16}H_{14}O_4$	cardamomin	[16]

Note: Compounds marked with * were found in both *AIRD* and *NRD*, compounds marked with # were identified in Resina Draconis for the first time and compounds marked with ** were unambiguously identified with reference.

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Figure 2. The hypothesized fragmentation pathway and MS/MS spectrum of compound 25.

2.4. Characterization of Flavonoid Compounds in Artificially Induced Resina Draconis

The flavonoid compounds observed in AIRD were mainly categorized into several subtypes, such as flavones, flavanones, chalcones, dihydrochalcones, homoisoflavanes. From the results of the QTOF-MS/MS mass spectra, we can see that most of the flavonoids compounds had strong responses under negative ion scan modes and easily gave quasi-molecular ion peaks of $[M - H]^-$, $[M + HCOO]^$ and $[2M - H]^-$. The fragmentation rules of flavonoids facilitate their structure identification, such as the neutral losses of CO (28), due to the contraction of the ring with carbonyl group, the loss of H_2O (18) implies the presence of a hydroxyl group, and the losses of CH₃ (15), OCH₃ (31) explain the presence of a methoxyl group. The proposed fragmentation pathways of flavonoid compounds presented in AIRD and NRD are summarized in Figure 3. Usually, flavanes and flavanones easily utilize the Retro-Diels-Alder (RDA) fragmentation pathway. Fragment ions derived from RDA fragments are more abundant than the loss of other radical ions, for example, the loss of CH_3 , CO, OH or H_2O . On the other hand, chalcones, dihydrochalcones, flavanes as well as homoisoflavanes have very weak RDA fragmentation. Their main fragmentation pathway is the breakage of C₃–C₉ bond, which lead to the loss of the B-ring. The structures of homoisoflavanoids are special because of an additional methylene linking between B- and C-rings. The cleavages of C₃-C₉ and/or C₉-C₁ leading to loss of B-ring, were regarded as diagnostic fragments for this kind of flavonoids. For chalcones, cleavage beside the carbonyl group usually occurs because there is a C=C double bond at the α - β position, while dihydrochalcones, with a saturated C-C backbone, always encounter the breakage of the C-C bone at the α - β position.

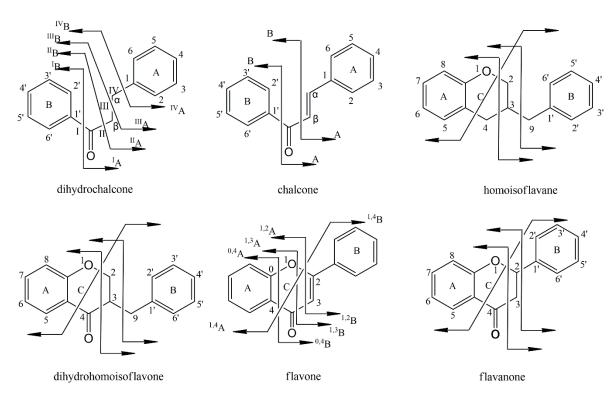


Figure 3. Fragmentation pathway of flavonoid compounds in Resina Draconis.

The prominent fragmentation pathways proposed here were expected to facilitate the characterization of flavonoid compounds in AIRD and NRD. Thus, by using the method described above, a total of 36 flavonoid compounds in AIRD were identified. For example, as a typical dihydrochalcone, the deprotonated molecularion of compound **49** ($t_R = 13.69$ min) was observed at m/z 315.1229 [M – H]⁻ under negative ionization mode. The product ions of m/z 300.0998 [M – H – CH₃]⁻ and 285.0801 [M – H – 2CH₃]⁻ corresponded to the radical cleavage of the methyl

group, and the fragmentations at m/z 147.0450, 134.0370 and 120.0212 contributed to the successive neutral loss of the B ring at the II, III and IV bond positions, respectively. The characteristic fragments, primary cleavages, and MS/MS spectrum of compound 49 are elucidated in Figure 4. Compared with reference and literature data [13,26], compound 49 was identified to be loureirin B. It is interesting to note that there were common fragment ions observed at most of the dihydrochalcone compounds at m/z 147.05, 134.04 and 120.02, which could due to the successive neutral loss of the B ring at the positions of II, III and IV bond. For example, compounds 17 ($t_R = 5.64 \text{ min}$), 28 ($t_R = 7.68 \text{ min}$), 37 $(t_{\rm R} = 10.26 \text{ min})$ and 48 $(t_{\rm R} = 13.69 \text{ min})$ displayed molecular ions at m/z 287.0914 [M – H]⁻, 257.0809 $[M-H]^-$, 301.1077 $[M-H]^-$ and 285.1123 $[M-H]^-$, respectively in the negative mode. With the assistance of formula predictor software, the elemental composition of the sequasi-molecular ions was calculated to be $C_{16}H_{16}O_5$, $C_{15}H_{14}O_4$, $C_{17}H_{18}O_5$ and $C_{17}H_{18}O_4$. From the MS/MS spectrum (Table 1), characteristic fragments were observed at m/z 147.05, 34.04 and 120.02 in compounds 17, 28, 37, and 48, respectively, suggesting these compounds are derivatives of typical dihydrochalcones. Thus, by comparing these fragmentation behaviors with those of the reference compounds, MS/MS fragmentation data and literatures, compounds 17, 28, 37 and 48 were tentatively identified as loureirin D [13], 2,4,4'-trihydroxydihydrochalcone [18], 4,4'-dihydroxy-2,6-dimethoxydihydrochalcone [16] and loureirin A [13, 26], respectively.

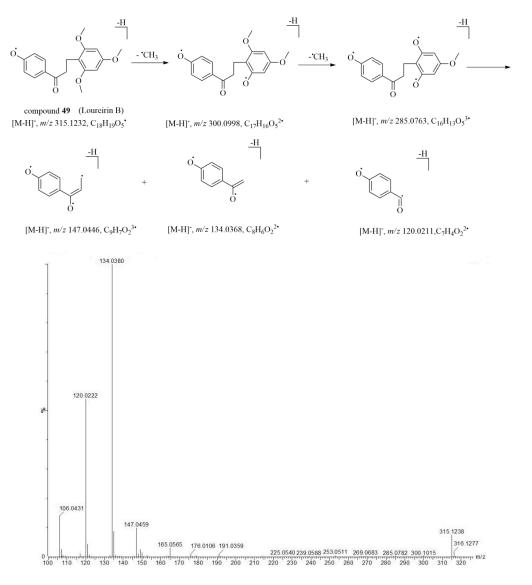


Figure 4. The hypothesized fragmentation pathway and MS/MS spectrum of compound 49.

2.5. Chemical Transformation of Artificially Induced Resina Draconis and Natural Resina Draconis

Qualitative and semi-quantitative comparisons of chemical constituents between AIRD and NRD samples were carried out using our newly established UHPLC-QTOF-MS/MS method. The relative peak height of steroid saponins 22, 25, 26, 27, 42 and 54 detected in AIRD were significantly higher than those in NRD. It has been reported that the chemical constituents of fresh stems from D. cochinchinensis trees were quite different from the constituents of their xylem resins [10]. More than 60% of compounds isolated from fresh stems were steroid saponins, and the flavonoids compounds were undetectable in fresh stem xylem. However, flavonoid content increased dramatically after wounding and/or fungal infection of these plants [10]. Conversely, steroid saponins decreased considerably along with infection time. It was hypothesized that the wound-activated defense response of trees result in Resina Draconis formation; this phenomenon is thought to be a defense response of plants against attack by foreign bodies. Wounding triggers the biosynthesis of flavonoids components through the phenylpropanoid pathway during the defensive reaction process [10]. Flavonoid content was positively correlated with infection time while the steroid saponin content was negatively correlated. All of the above results are in agreement with reports from the literature. Furthermore, it was found that, the number and intensity of higher polarity components in AIRD, eluted before 6 min at the BPI chromatograms (Figure 1), were more abundant than that those of NRD. While the lower polarity components, eluted after 16 min, were increased significantly in NRD but decreased markedly or even not detected in AIRD. These lower polarity compounds, identified as flavonoid dimers according to their molecular formulas, and related literatures, had been reported in NRD in a previously published report [28]. Interestingly, the composition of the phenolic substances in AIRD was dramatically different from that of NRD. The relative peak heights of flavonoids compounds (19, 20, 32, 33, 34, 38, 39, 40, 41) detected in AIRD were higher than those of NRD. This phenomenon could be attributed to the formation and conversion of the simple flavonoids with endogenous enzymes and/or the metabolism by exogenous microorganisms during the production of the resin [28,29]. It has been reported that dihydrochalcone compounds are one of the main components of Resina Draconis, and could be polymerized further with the other flavones components to form dimer, trimer, and tetramer compounds [28,29], such content may reach 20% of the total mass accompanying the formation of Resina Draconis. In this study, the accumulation of Resina Draconis only took 80 days and during that time a considerable amount of flavonoids were synthesized, however, no flavonoid oligomers were detected by MS. It is likely that the induction time of Resina Draconis was too short to oxidize the dihydrochalcone and flavones further to form flavonoid oligomers.

3. Materials and Methods

3.1. Materials and Reagents

Wild *D. cochinchinensis* samples were collected from authenticated locations in Ningming county, Guangxi province, China, the indigenous wild region of this plant. These voucher specimens, identified by Professor Baoyou Huang, have been deposited in the Herbarium Centre, Guangxi Botanical Garden of Medicinal Plants, Nanning, China. Five year-old *D. cochinchinensis* saplings were approximately 1.5 m in height, with a diameter of 4.0–6.0 cm at 10 cm above ground. The reference compounds 7,4′-dihydroxyflavone, protogracillin, protodioscin, 7-hydroxyflavone, loureirin D, loureirin A and loureirin B were purchased from the National Institutes for Food and Drug Control (Beijing, China). All of the chemicals were obtained with a purity of 98% or above, as confirmed by normalization of the peak area detected by UHPLC-QTOF-MS (Waters Corporation, Milford, MA, USA). HPLC-grade acetonitrile, methanol, and formic acid (purity: 98%) were purchased from E. Merck (Darmstadt, Germany). Ultra-pure water was obtained from a Milli-Q water purification system (Millipore, Bedford, MA, USA). Other solvents and chemicals were of analytical grade and purchased from the Guangdong Guanghua Sci-Tech Co., Ltd. (Shantou, China).

3.2. Induction of D. cochinchinensis Saplings to Produce Resina Draconis

A small hole (0.4 cm in diameter, 10 cm above ground) was drilled into the xylem stem of a wild *D. cochinchinensis* sapling, and an injector containing a salt solution (1% zinc sulfateand 2.0 g/L benzoic acid) was inserted into the hole. The salt solution was left to flow freely into the xylem via transpiration. The observed flow rate was approximately 20 mL/h. A transfusion of double distilled water was used as a control. After treating the tree for 80 days, a tissue core about 1.5 cm in diameter of the inoculated sapling xylem was collected by an electric perforator (Figure 5).

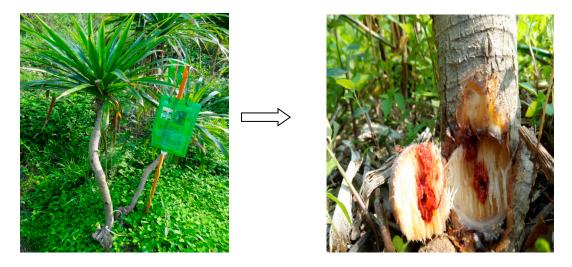


Figure 5. Resina Draconis produced in the stem xylem of *D. cochinchinensis* trees after chemical solution induction for 80 days.

3.3. Preparation of Sample and Standard Solutions

The preparation of the sample solutions for the chemical profile analyses was performed as follows: six batches of red resin collected from induced D. cochinchinensis and nature D. cochinchinensis were combined respectively to get enough amounts of AIRD and NRD sample. Then, the samples were dried in an oven at $60\,^{\circ}\text{C}$ for 2 h. The dried AIRD and NRD samples were cut into small sections and ground to a fine powder in a mill. Samples were then accurately weighed powder $(0.2\,\text{g})$ was suspended in 20 mL of 80% methanol (v/v) in a 25 mL volumetric flask and sonicated (Kun Shan Ultrasonic Instruments Co., Ltd., Kun Shan, China) for 30 min at room temperature. Then, the final volume was made up to 25 mL with 80% methanol (v/v). The supernatant of the extracts was filtered through a $0.22\,\mu\text{m}$ PTFE syringe filter before UHPLC-QTOF-MS/MS analysis. Three replicates were prepared for each sample.

Stock solutions: A certain amount of 7,4′-dihydroxyflavone, protogracillin, protodioscin, 7-hydroxyflavone, loureirin D, loureirin A and loureirin B dissolved in 80% methanol-water (v/v), to create reference compound stock solutions (about 1.0 mg/mL), which was stored under 4 °C. Both solutions were filtered by a 0.22 μ m PTFE syringe filter before being subjected to UHPLC-QTOF-MS analysis.

3.4. Qualitative Analysis by UHPLC-QTOF-MS/MS

Detection was performed using a Waters ACQUITY UHPLC TM system (Waters Corporation, Milford, MA, USA). Chromatography separation was performed with an ACQUITY BEH C_{18} column (100 mm \times 2.1 mm, 1.7 μ m; Waters, Milford, MA, USA), coupled with a C_{18} pre-column (2.1 mm \times 5 mm, 1.7 μ m, Van-Guard TM BEH, Waters, Milford, MA, USA.) and a column temperature of 35 °C. The mobile phase was a mixture of 0.1% formic acid-water (A) and acetonitrile (B), with an optimized linear gradient elution as follows: 0–0.5 min, 10% B; 0.5–1.0 min, 10–18% B; 1.0–4.5 min,

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18-27% B; 4.5-10 min, 27-32% B; 10-11.0 min, 38-45% B; 11.0-16.0 min, 38-45% B; 16.0-18.0 min, 45-55% B; 18.0-20.0 min, 55-85% B; 20.0-20.1 min, 85-10% B; 20.1-23.0 min, 10-10% B. The injection volume was 2 μ L and the flow rate was set at 0.50 mL/min.

Mass spectrometry was performed on a definition accurate mass quadrupole time-of-flight (Q-TOF) Xevo G2-S mass spectrometer (Waters MS Technologies, Manchester, UK) equipped with electrospray ionization (ESI) source. The ESI source of the MS was operated in both positive and negative modes. All MS data were produced using LockSprayTM to ensure mass accuracy and reproducibility. The Leucine-enkephalin ions, $[M-H]^-$ and $[M+H]^+$ at mass to charge ratio m/z 554.2615 and m/z 556.2771 were used as the lock mass in negative and positive electrospray ionization modes, respectively. Mass spectrawere acquired in the negative ion mode by scanning from 100 to 1500 Da with a 0.20 s scan time and a 0.01 s inter-scan delay over a 22 min analysis time. MS analysis was conducted under the following operating parameters: The desolvation gas flow rate was 800 L/h at a temperature of 400 °C, and the cone gas was 20 L/h and the source temperature was 100 °C; the capillary voltage and cone voltage were set at 2400 V and 40 V, respectively. The energies for collision induced dissociation (CID) were 6 V for the precursor ion and 35–55 V for fragmentation information.

UHPLC data analysis and accurate mass as well as elemental composition were analyzed with the MassLynx 4.1 software (Waters Co., Mil-ford, CT, USA). Structural elucidation was performed by the MassFragment tool provided by MassLynx 4.1.

4. Conclusions

In the present study, a rapid and sensitive UHPLC-QTOF-MS/MS method was carried out to compare and identify the chemical constituents of the *AIRD* and *NRD* for the first time. A total of 44 compounds were unequivocally or tentatively identified and 40 peaks were found both in *AIRD* and *NRD*. The chemical profiles of *AIRD* and *NRD* were further explored to find the differences in secondary metabolites. Dihydrochalcone was the main bioactive component in both *AIRD* and *NRD*, which was shown to form dimer, trimer, and tetramer compounds with the other flavones through oxidation along with the formation of Resina Draconis. Oligomer of dihydrochalcone could be found in *NRD* but not in *AIRD*, which may due to an insufficient induction time of Resina Draconis (only 80 days). The association between the chemical transformation of Resina Draconis and induction time as well as the effect on the bioactivities of *AIRD* and *NRD* needs to be further explored. In short, this study clearly demonstrated the chemical constituents of *AIRD* and *NRD*, thus providing a rational basis for *AIRD* for use as a substitute for *NRD* to meet the market demand of Resina Draconis. Additionally, these results should be very useful for the quality assessment of this valuable medicine.

Author Contributions: Q.C. and L.H. conceived and designed the experiments; Q.C., C.M. and Z.Z. performed the experiments; Q.C., H.L. and X.G. analyzed the data; Q.C., L.H. and Y.W. wrote the paper. All authors read and approved the final manuscript.

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Conflicts of Interest: The authors declare no conflict of interest.

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Sample Availability: Samples of the compounds 7,4'-dihydroxyflavone, protogracillin, protodioscin, 7-hydroxyflavone, loureirin D, loureirin A and loureirin B are available from the authors.



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