

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

17-Hydroxy-1,8-dimethyl-17-azapenta-cyclo[6.6.5.0^{2,7}.0^{9,14}.0^{15,19}]nonadeca-2,4,6,9(14),10,12-hexaene-16,18-dione

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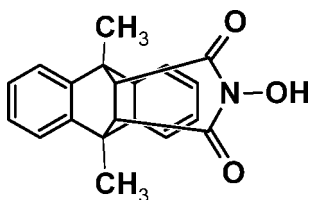
Received 27 July 2012; accepted 31 October 2012

 Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.036; wR factor = 0.096; data-to-parameter ratio = 12.7.

In the title compound, $\text{C}_{20}\text{H}_{17}\text{NO}_3$ (alternative name: *N*-hydroxy-9,10-dimethyl-9,10-ethanoanthracene-11,12-dicarboximide), the rigid ethanoanthracene-dicarboximide moiety has a roof-shaped geometry, the interplanar angle between the two terminal phenyl rings being $124.9(6)^\circ$. In the crystal, molecules are linked *via* $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds, forming chains along [010]. $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ interactions link adjacent chains, leading to the formation of a three-dimensional structure.

Related literature

For the synthesis of the title compound, see: Kossakowski & Jarocka (2000). For the biological activity of related compounds, see: Bova *et al.* (2009). For related structures, see: Atherton & Jones (2002); Smet *et al.* (2000); Su *et al.* (2011); Guo *et al.* (2010); Adams *et al.* (2006); He & Ng (2007); Weber *et al.* (1991, 1994); Yang & Swager (1998). The rigid ethanoanthracenedicarboximide moiety of the title compound shows the typical roof-shaped geometry (Weber *et al.*, 1991; Csöreghe *et al.*, 2003). For a description of the Cambridge Structural Database, see: Allen (2002).



Experimental

Crystal data

 $\text{C}_{20}\text{H}_{17}\text{NO}_3$
 $M_r = 319.36$

 Monoclinic, $P2_1/n$
 $a = 13.904(1)$ Å

 $b = 8.104(1)$ Å
 $c = 13.946(1)$ Å
 $\beta = 97.39(1)^\circ$
 $V = 1558.4(3)$ Å³
 $Z = 4$

 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 100$ K
 $0.40 \times 0.40 \times 0.30$ mm

Data collection

 Oxford Diffraction Xcalibur
 (Sapphire2) diffractometer
 5321 measured reflections

 2827 independent reflections
 2467 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.036$
 $wR(F^2) = 0.096$
 $S = 1.04$
 2827 reflections
 223 parameters

 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.30$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

 Cg1 is the centroid of the C6–C11 ring. Cg2 refers to the mid-point of the C15–C16 bond.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O3}-\text{H3A}\cdots\text{O1}^{\text{i}}$	0.96 (2)	1.69 (2)	2.630 (1)	167 (2)
$\text{C8}-\text{H8}\cdots\text{O1}^{\text{ii}}$	0.95	2.54	3.408 (2)	152
$\text{C15}-\text{H15}\cdots\text{O3}^{\text{iii}}$	0.95	2.46	3.273 (2)	143
$\text{C17}-\text{H17}\cdots\text{O2}^{\text{iv}}$	0.95	2.54	3.457 (2)	163
$\text{C4}-\text{H4}\cdots\text{Cg1}^{\text{v}}$	1.00	2.66	3.518 (3)	144
$\text{C10}-\text{H10}\cdots\text{Cg2}^{\text{vi}}$	0.95	2.77	3.668 (3)	158

 Symmetry codes: (i) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x - \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $x, y + 1, z$; (v) $-x, -y + 1, -z + 1$; (vi) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *publCIF*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BG2475).

References

- Adams, H., Bawa, R. A. & Jones, S. (2006). *Org. Biomol. Chem.* **4**, 4206–4213.
 Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
 Atherton, J. C. C. & Jones, S. (2002). *J. Chem. Soc. Perkin Trans. 1*, pp. 2166–2173.
 Bova, S., Saponara, S., Rampa, A., Gobbi, S., Cima, L., Fusi, F., Sgaragli, G., Cavalli, M., de los Rios, C., Striessnig, J. & Bisi, A. (2009). *J. Med. Chem.* **52**, 1259–1262.
 Csöreghe, I., Finge, S. & Weber, E. (2003). *Struct. Chem.* **14**, 241–246.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Guo, J.-B., Xiang, J.-F. & Chen, C.-F. (2010). *Eur. J. Org. Chem.* pp. 5056–5062.
 He, L. & Ng, S. W. (2007). *Acta Cryst.* **E63**, o602–o603.
 Kossakowski, J. & Jarocka, M. (2000). *Acta Pol. Pharm.* **57**, 60–62.
 Oxford Diffraction (2009). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, England.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Smet, M., Corens, D., Van Meervelt, L. & Dehaen, W. (2000). *Molecules*, **5**, 179–188.
 Su, Y.-S., Liu, J.-W., Jiang, Y. & Chen, C.-F. (2011). *Chem. Eur. J.* **17**, 2435–2441.
 Weber, E., Finge, S. & Csöreghe, I. (1991). *J. Org. Chem.* **56**, 7281–7288.

Weber, E., Reutel, C., Foces-Foces, C. & Llamas-Saiz, A. L. (1994). *J. Chem. Soc. Perkin Trans. 2*, pp. 1455–1461.

Yang, J.-S. & Swager, T. M. (1998). *J. Am. Chem. Soc.* **120**, 11864–11873.

supplementary materials

Acta Cryst. (2012). E68, o3293–o3294 [doi:10.1107/S1600536812045151]

**17-Hydroxy-1,8-dimethyl-17-azapentacyclo-
[6.6.5.0^{2,7}.0^{9,14}.0^{15,19}]nonadeca-2,4,6,9(14),10,12-hexaene-16,18-dione**

Barbara Miroslaw, Anna E. Koziol, Magdalena Pakosinska-Parys and Marta Struga

Comment

Roof-shaped aromatic hydrocarbon derivatives have been used for inclusion of neutral compounds (Weber *et al.*, 1991; 1994) and as porous polymer films and sensors for dinitrotoluene with possible application as a land mine detectors (Yang & Swager, 1998). The *N*-substitution is the most common way of modification of these molecules (Weber *et al.*, 1994; Smet *et al.*, 2000). Some experiments were also conducted on the substitution in the aryl moiety (Atherton & Jones, 2002; Adams *et al.*, 2006; He & Ng, 2007). The search of the CSD (CSD v5.33 and updates; Allen, 2002) revealed 67 crystal structures of compounds with the rigid pentacyclic 9,10-ethanoanthracenedicarboximide skeleton. However, none of these derivatives has *N*-hydroxy substituent and only two molecules are symmetrically substituted at bridgehead C atoms (here C5, C12; see Fig. 1). In these crystals the polycyclic skeletons are combined with voluminous macrocyclic fragments. (Su *et al.*, 2011; Guo *et al.*, 2010). The rigid ethanoanthracenedicarboximide moiety of the title compound (Fig. 1) shows the typical roof-shaped geometry (Weber *et al.*, 1991; Csöregi *et al.*, 2003); the interplanar angle between the two terminal phenyl rings is 124.9°. The hydroxyl O atom interacts through the O3–H···O1 hydrogen bond (Fig. 2). Molecules form chains along the 2₁ screw axis. Between adjacent chains many C–H···O and C–H··· π interactions are observed (Figs. 2–4, Table 1).

Experimental

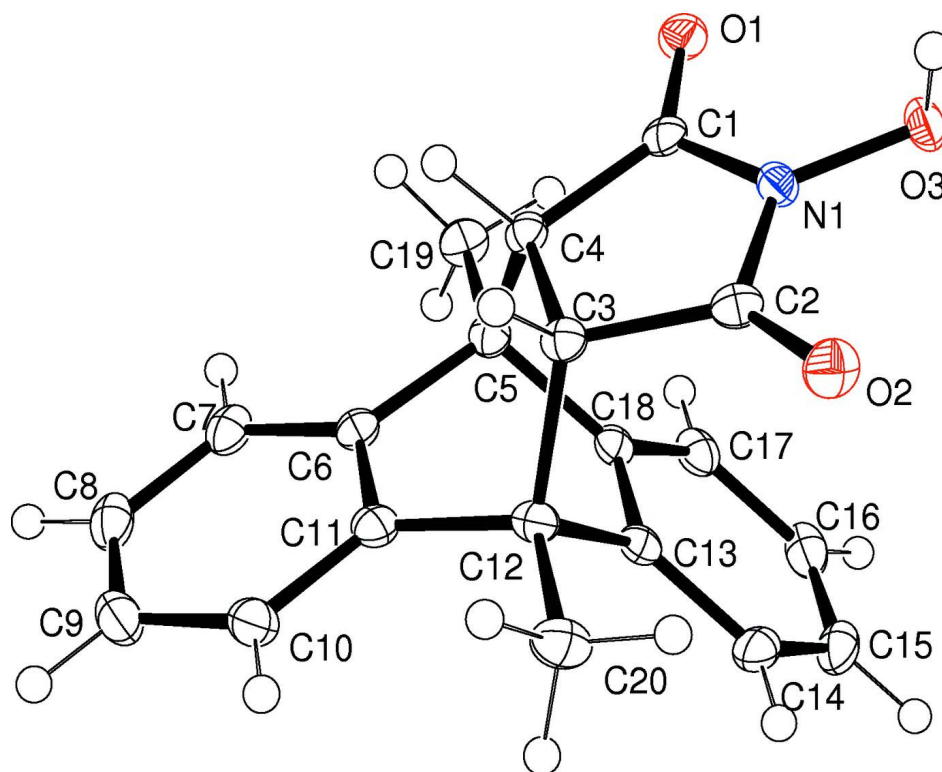
The title compound, UPAC name: 17-hydroxy-1,8-dimethyl-17-azapentacyclo-[6.6.5.0^{2,7}.0^{9,14}.0^{15,19}]nonadeca-2,4,6,9(14),10,12-hexaene-16,18-dione, was synthesized in the search of compounds with potential anxiolytic activity, as described previously (Kossakowski & Jarocka, 2000).

Refinement

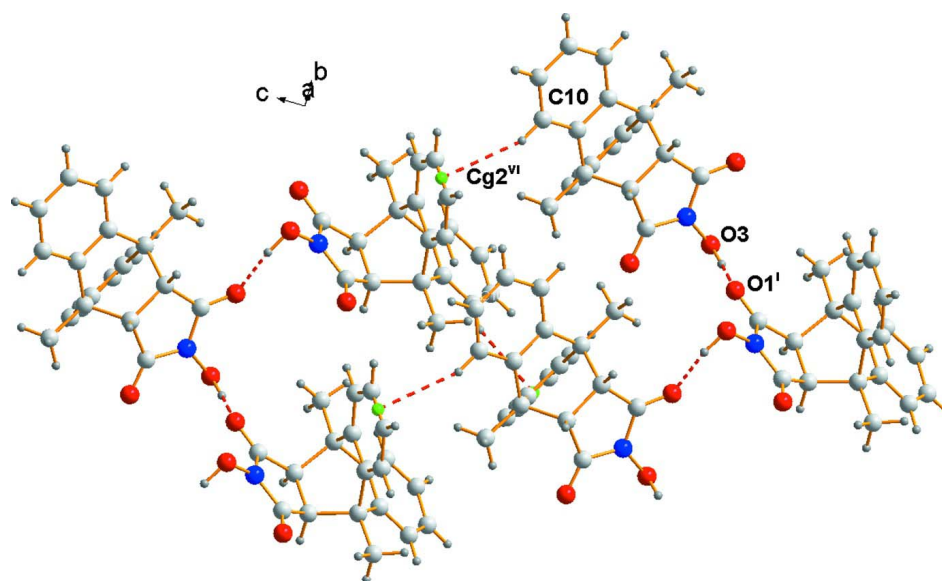
All C-bonded H atoms were positioned geometrically and allowed to ride on the attached atom with the C–H bond lengths of 0.95 Å for aromatic atoms, 1.00 Å for methine and 0.98 Å for methyl groups. $U_{\text{iso}}(\text{H})$ values were fixed to $1.2U_{\text{eq}}(\text{C})$ and $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$. The hydroxyl H atom was located in the difference electron density map and refined isotropically.

Computing details

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO* (Oxford Diffraction, 2009); data reduction: *CrysAlis PRO* (Oxford Diffraction, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *pubCIF*.

**Figure 1**

Ortep view of the title compound with atom numbering scheme. Ellipsoids for non-hydrogen atoms were drawn at the 50 % probability level.

**Figure 2**

Linear association of molecules in crystal of 1 through the O-H...O hydrogen bonds and C-H... π interactions between adjacent chains. The Cg2 refers to the center of gravity between atoms C15/C16. Symmetry codes: (i) $-x+1/2, y-1/2, -z+3/2$; (vi) $-x+1/2, y-1/2, -z+1/2$.

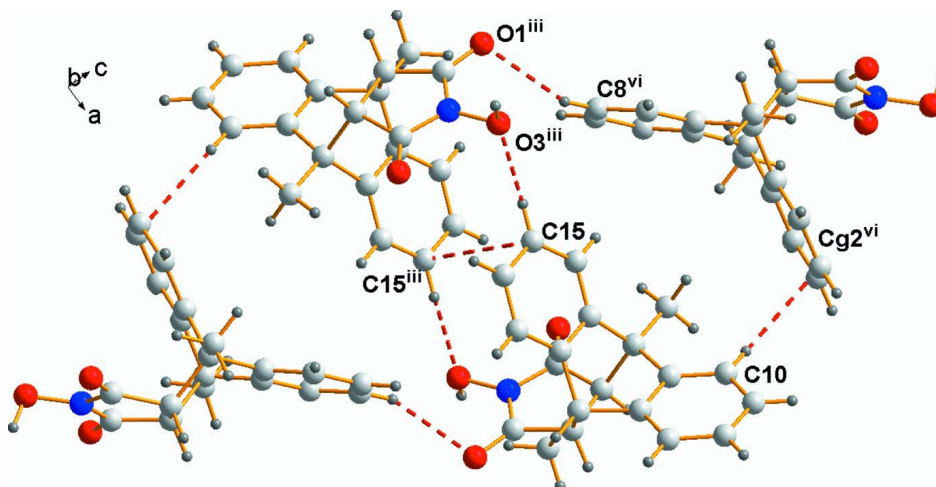


Figure 3

C–H...O and C–H... π interactions in crystal of 1 with short C15...C15ⁱⁱⁱ intermolecular contact (3.386 (2) Å). The Cg2 is the centroid of the C15/C16 bond. Symmetry codes: (iii) $-x+1, -y+1, -z+1$; (vi) $-x+1/2, y-1/2, -z+1/2$.

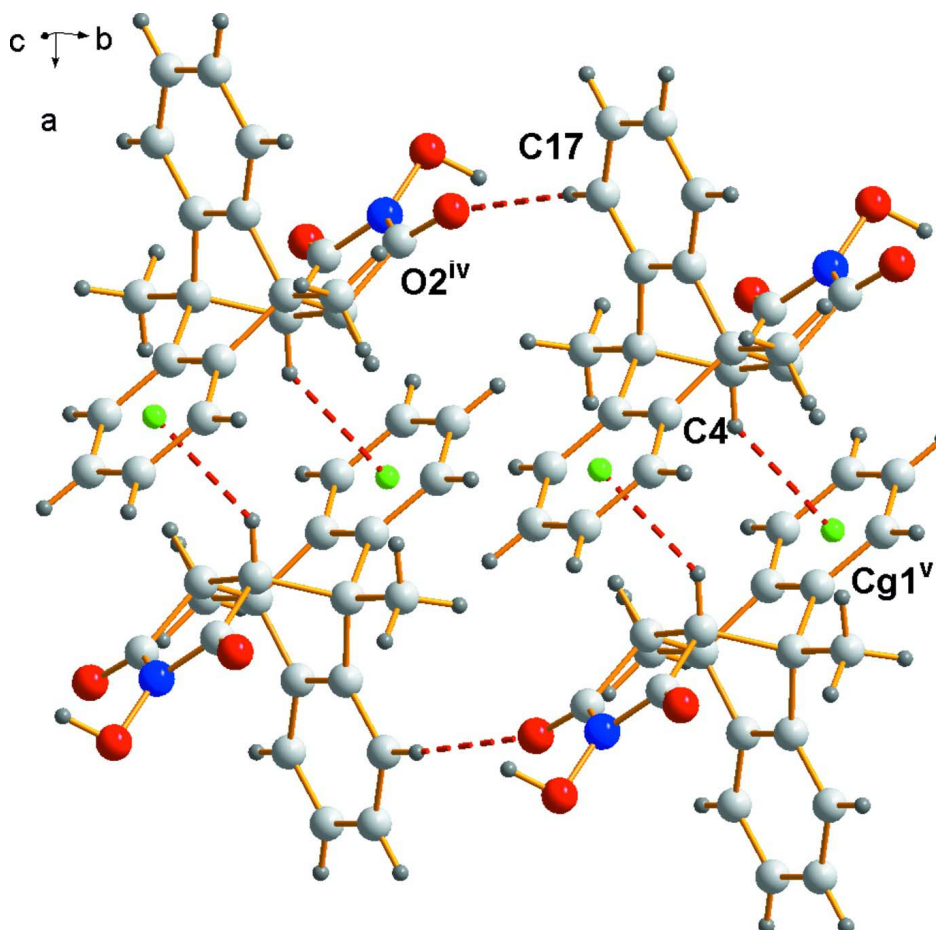


Figure 4

C–H...O hydrogen bonds and C–H... π interactions in crystal of 1. The Cg1 refers to the centroid of the C6–C11 ring. Symmetry codes: (iv) $x, y+1, z$; (v) $-x, -y+1, -z+1$.

17-Hydroxy-1,8-dimethyl-17-azapentacyclo[6.6.5.0^{2,7}.0^{9,14}.0^{15,19}]nonadeca-2,4,6,9(14),10,12-hexaene-16,18-dione

Crystal data

$C_{20}H_{17}NO_3$	$F(000) = 672$
$M_r = 319.36$	$D_x = 1.361 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: -P 2yn	Cell parameters from 3666 reflections
$a = 13.904 (1) \text{ \AA}$	$\theta = 2.9\text{--}29.8^\circ$
$b = 8.104 (1) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$c = 13.946 (1) \text{ \AA}$	$T = 100 \text{ K}$
$\beta = 97.39 (1)^\circ$	Prism, colourless
$V = 1558.4 (3) \text{ \AA}^3$	$0.40 \times 0.40 \times 0.30 \text{ mm}$
$Z = 4$	

Data collection

Oxford Diffraction Xcalibur (Sapphire2) diffractometer	2827 independent reflections
Radiation source: Enhance (Mo) X-ray Source	2467 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\text{int}} = 0.023$
Detector resolution: 8.4221 pixels mm^{-1}	$\theta_{\text{max}} = 25.2^\circ$, $\theta_{\text{min}} = 2.9^\circ$
ω scans	$h = -16 \rightarrow 16$
5321 measured reflections	$k = -9 \rightarrow 6$
	$l = -9 \rightarrow 16$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.036$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.096$	$w = 1/[\sigma^2(F_o^2) + (0.0465P)^2 + 0.4721P]$
$S = 1.04$	where $P = (F_o^2 + 2F_c^2)/3$
2827 reflections	$(\Delta/\sigma)_{\text{max}} = 0.009$
223 parameters	$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.28230 (8)	0.37858 (13)	0.63969 (8)	0.0154 (3)
O1	0.23524 (7)	0.61166 (12)	0.71225 (6)	0.0201 (2)
O2	0.30265 (7)	0.15262 (11)	0.54480 (7)	0.0217 (2)
O3	0.36001 (7)	0.34569 (12)	0.70937 (7)	0.0194 (2)
H3A	0.3338 (15)	0.258 (3)	0.7443 (15)	0.049 (6)*
C1	0.22513 (9)	0.51497 (16)	0.64443 (9)	0.0148 (3)
C2	0.25875 (10)	0.27735 (16)	0.55890 (9)	0.0161 (3)

C3	0.17056 (9)	0.35563 (16)	0.49956 (9)	0.0151 (3)
H3	0.1129	0.2818	0.4999	0.018*
C4	0.15350 (9)	0.51968 (15)	0.55294 (9)	0.0147 (3)
H4	0.0859	0.5217	0.5702	0.018*
C5	0.16945 (10)	0.67321 (16)	0.48659 (9)	0.0157 (3)
C6	0.09511 (9)	0.64834 (16)	0.39688 (9)	0.0161 (3)
C7	0.02301 (10)	0.76017 (17)	0.36135 (10)	0.0199 (3)
H7	0.0172	0.8623	0.3936	0.024*
C8	-0.04079 (10)	0.72243 (19)	0.27836 (11)	0.0242 (3)
H8	-0.0894	0.7994	0.2538	0.029*
C9	-0.03310 (10)	0.5725 (2)	0.23190 (10)	0.0252 (3)
H9	-0.0771	0.5464	0.1761	0.030*
C10	0.03931 (10)	0.45948 (19)	0.26700 (10)	0.0216 (3)
H10	0.0442	0.3568	0.2351	0.026*
C11	0.10405 (9)	0.49792 (17)	0.34867 (9)	0.0172 (3)
C12	0.18898 (10)	0.38987 (16)	0.39219 (9)	0.0167 (3)
C13	0.27923 (10)	0.49978 (16)	0.40301 (9)	0.0159 (3)
C14	0.36774 (10)	0.45997 (18)	0.37202 (10)	0.0206 (3)
H14	0.3746	0.3601	0.3379	0.025*
C15	0.44605 (10)	0.56721 (19)	0.39128 (10)	0.0253 (3)
H15	0.5065	0.5400	0.3704	0.030*
C16	0.43639 (10)	0.71368 (19)	0.44081 (10)	0.0239 (3)
H16	0.4903	0.7861	0.4534	0.029*
C17	0.34813 (10)	0.75541 (17)	0.47223 (10)	0.0195 (3)
H17	0.3417	0.8559	0.5059	0.023*
C18	0.26956 (10)	0.64801 (16)	0.45364 (9)	0.0159 (3)
C19	0.15742 (11)	0.83661 (16)	0.53834 (10)	0.0207 (3)
H19A	0.0939	0.8391	0.5618	0.031*
H19B	0.1619	0.9282	0.4932	0.031*
H19C	0.2087	0.8474	0.5932	0.031*
C20	0.19871 (11)	0.22982 (17)	0.33594 (10)	0.0230 (3)
H20A	0.2102	0.2562	0.2698	0.034*
H20B	0.1389	0.1654	0.3342	0.034*
H20C	0.2533	0.1654	0.3678	0.034*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0145 (6)	0.0179 (6)	0.0133 (5)	-0.0008 (4)	0.0005 (4)	0.0025 (4)
O1	0.0195 (5)	0.0236 (5)	0.0173 (5)	-0.0011 (4)	0.0024 (4)	-0.0065 (4)
O2	0.0259 (6)	0.0150 (5)	0.0245 (5)	0.0039 (4)	0.0045 (4)	0.0008 (4)
O3	0.0143 (5)	0.0250 (5)	0.0176 (5)	-0.0010 (4)	-0.0024 (4)	0.0058 (4)
C1	0.0129 (6)	0.0164 (7)	0.0159 (7)	-0.0034 (5)	0.0058 (5)	0.0010 (5)
C2	0.0182 (7)	0.0144 (7)	0.0167 (7)	-0.0036 (5)	0.0062 (5)	0.0021 (5)
C3	0.0149 (6)	0.0152 (7)	0.0154 (7)	-0.0028 (5)	0.0032 (5)	-0.0006 (5)
C4	0.0130 (6)	0.0160 (6)	0.0155 (7)	-0.0015 (5)	0.0031 (5)	-0.0010 (5)
C5	0.0154 (7)	0.0153 (7)	0.0166 (7)	-0.0006 (5)	0.0026 (5)	0.0010 (5)
C6	0.0141 (7)	0.0190 (7)	0.0158 (7)	-0.0024 (5)	0.0049 (5)	0.0024 (5)
C7	0.0173 (7)	0.0193 (7)	0.0242 (8)	0.0006 (5)	0.0067 (6)	0.0055 (6)
C8	0.0158 (7)	0.0310 (8)	0.0261 (8)	0.0008 (6)	0.0034 (6)	0.0126 (6)

C9	0.0193 (7)	0.0376 (9)	0.0177 (7)	-0.0065 (6)	-0.0011 (6)	0.0059 (6)
C10	0.0202 (7)	0.0280 (8)	0.0168 (7)	-0.0038 (6)	0.0026 (5)	-0.0007 (6)
C11	0.0167 (7)	0.0207 (7)	0.0149 (7)	-0.0026 (5)	0.0048 (5)	0.0019 (5)
C12	0.0187 (7)	0.0187 (7)	0.0130 (6)	-0.0008 (5)	0.0028 (5)	-0.0005 (5)
C13	0.0172 (7)	0.0194 (7)	0.0110 (6)	0.0013 (5)	0.0015 (5)	0.0048 (5)
C14	0.0218 (7)	0.0233 (7)	0.0179 (7)	0.0056 (6)	0.0072 (6)	0.0054 (6)
C15	0.0170 (7)	0.0350 (9)	0.0251 (8)	0.0061 (6)	0.0079 (6)	0.0138 (7)
C16	0.0159 (7)	0.0310 (8)	0.0242 (8)	-0.0048 (6)	0.0005 (6)	0.0118 (6)
C17	0.0198 (7)	0.0206 (7)	0.0175 (7)	-0.0030 (6)	-0.0003 (5)	0.0055 (6)
C18	0.0158 (7)	0.0189 (7)	0.0127 (6)	0.0011 (5)	0.0012 (5)	0.0051 (5)
C19	0.0229 (7)	0.0174 (7)	0.0223 (7)	0.0001 (6)	0.0046 (6)	-0.0011 (6)
C20	0.0292 (8)	0.0219 (7)	0.0181 (7)	0.0003 (6)	0.0039 (6)	-0.0034 (6)

Geometric parameters (Å, °)

N1—C1	1.3679 (17)	C9—H9	0.9500
N1—O3	1.3831 (14)	C10—C11	1.3933 (19)
N1—C2	1.3981 (17)	C10—H10	0.9500
O1—C1	1.2223 (16)	C11—C12	1.5314 (19)
O2—C2	1.2099 (16)	C12—C13	1.5304 (19)
O3—H3A	0.96 (2)	C12—C20	1.5308 (19)
C1—C4	1.5149 (18)	C13—C14	1.3936 (19)
C2—C3	1.5264 (18)	C13—C18	1.4085 (18)
C3—C4	1.5567 (17)	C14—C15	1.392 (2)
C3—C12	1.5758 (18)	C14—H14	0.9500
C3—H3	1.0000	C15—C16	1.388 (2)
C4—C5	1.5831 (18)	C15—H15	0.9500
C4—H4	1.0000	C16—C17	1.397 (2)
C5—C19	1.5275 (18)	C16—H16	0.9500
C5—C6	1.5303 (18)	C17—C18	1.3948 (19)
C5—C18	1.5345 (18)	C17—H17	0.9500
C6—C7	1.3944 (19)	C19—H19A	0.9800
C6—C11	1.4053 (19)	C19—H19B	0.9800
C7—C8	1.399 (2)	C19—H19C	0.9800
C7—H7	0.9500	C20—H20A	0.9800
C8—C9	1.387 (2)	C20—H20B	0.9800
C8—H8	0.9500	C20—H20C	0.9800
C9—C10	1.402 (2)		
C1—N1—O3	121.87 (10)	C11—C10—H10	120.0
C1—N1—C2	115.83 (11)	C9—C10—H10	120.0
O3—N1—C2	122.27 (11)	C10—C11—C6	119.83 (12)
N1—O3—H3A	100.7 (12)	C10—C11—C12	125.48 (12)
O1—C1—N1	123.07 (12)	C6—C11—C12	114.68 (11)
O1—C1—C4	129.30 (12)	C13—C12—C20	114.73 (11)
N1—C1—C4	107.62 (11)	C13—C12—C11	106.67 (11)
O2—C2—N1	123.39 (12)	C20—C12—C11	113.32 (11)
O2—C2—C3	130.24 (12)	C13—C12—C3	103.89 (10)
N1—C2—C3	106.36 (11)	C20—C12—C3	111.94 (11)
C2—C3—C4	104.88 (10)	C11—C12—C3	105.43 (10)

C2—C3—C12	111.78 (11)	C14—C13—C18	119.86 (13)
C4—C3—C12	110.94 (10)	C14—C13—C12	125.47 (12)
C2—C3—H3	109.7	C18—C13—C12	114.58 (11)
C4—C3—H3	109.7	C15—C14—C13	119.71 (13)
C12—C3—H3	109.7	C15—C14—H14	120.1
C1—C4—C3	104.90 (10)	C13—C14—H14	120.1
C1—C4—C5	112.69 (10)	C16—C15—C14	120.43 (13)
C3—C4—C5	110.51 (10)	C16—C15—H15	119.8
C1—C4—H4	109.5	C14—C15—H15	119.8
C3—C4—H4	109.5	C15—C16—C17	120.54 (13)
C5—C4—H4	109.5	C15—C16—H16	119.7
C19—C5—C6	113.39 (11)	C17—C16—H16	119.7
C19—C5—C18	114.55 (11)	C18—C17—C16	119.30 (13)
C6—C5—C18	106.32 (10)	C18—C17—H17	120.3
C19—C5—C4	111.91 (10)	C16—C17—H17	120.3
C6—C5—C4	104.10 (10)	C17—C18—C13	120.16 (12)
C18—C5—C4	105.72 (10)	C17—C18—C5	125.47 (12)
C7—C6—C11	119.84 (12)	C13—C18—C5	114.35 (11)
C7—C6—C5	125.74 (12)	C5—C19—H19A	109.5
C11—C6—C5	114.42 (11)	C5—C19—H19B	109.5
C6—C7—C8	120.17 (13)	H19A—C19—H19B	109.5
C6—C7—H7	119.9	C5—C19—H19C	109.5
C8—C7—H7	119.9	H19A—C19—H19C	109.5
C9—C8—C7	119.95 (13)	H19B—C19—H19C	109.5
C9—C8—H8	120.0	C12—C20—H20A	109.5
C7—C8—H8	120.0	C12—C20—H20B	109.5
C8—C9—C10	120.27 (13)	H20A—C20—H20B	109.5
C8—C9—H9	119.9	C12—C20—H20C	109.5
C10—C9—H9	119.9	H20A—C20—H20C	109.5
C11—C10—C9	119.92 (14)	H20B—C20—H20C	109.5

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C6—C11 ring. Cg2 refers to the mid-point of the C15—C16 bond.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O3—H3A \cdots O1 ⁱ	0.96 (2)	1.69 (2)	2.630 (1)	167 (2)
C8—H8 \cdots O1 ⁱⁱ	0.95	2.54	3.408 (2)	152
C15—H15 \cdots O3 ⁱⁱⁱ	0.95	2.46	3.273 (2)	143
C17—H17 \cdots O2 ^{iv}	0.95	2.54	3.457 (2)	163
C4—H4 \cdots Cg1 ^v	1.00	2.66	3.518 (3)	144
C10—H10 \cdots Cg2 ^{vi}	0.95	2.77	3.668 (3)	158

Symmetry codes: (i) $-x+1/2, y-1/2, -z+3/2$; (ii) $x-1/2, -y+3/2, z-1/2$; (iii) $-x+1, -y+1, -z+1$; (iv) $x, y+1, z$; (v) $-x, -y+1, -z+1$; (vi) $-x+1/2, y-1/2, -z+1/2$.