

Organic Nanofibers Embedding Stimuli-Responsive Threaded Molecular Components

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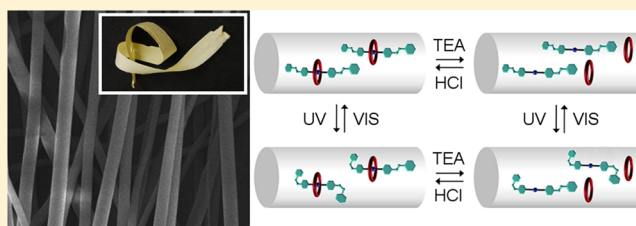
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S Supporting Information

ABSTRACT: While most of the studies on molecular machines have been performed in solution, interfacing these supramolecular systems with solid-state nanostructures and materials is very important in view of their utilization in sensing components working by chemical and photonic actuation. Host polymeric materials, and particularly polymer nanofibers, enable the manipulation of the functional molecules constituting molecular machines and provide a way to induce and control the supramolecular organization.

Here, we present electrospun nanocomposites embedding a self-assembling rotaxane-type system that is responsive to both optical (UV–vis light) and chemical (acid/base) stimuli. The system includes a molecular axle comprised of a dibenzylammonium recognition site and two azobenzene end groups and a dibenzo[24]crown-8 molecular ring. The dethreading and rethreading of the molecular components in nanofibers induced by exposure to base and acid vapors, as well as the photoisomerization of the azobenzene end groups, occur in a similar manner to what observed in solution. Importantly, however, the nanoscale mechanical function following external chemical stimuli induces a measurable variation of the macroscopic mechanical properties of nanofibers aligned in arrays, whose Young's modulus is significantly enhanced upon dethreading of the axles from the rings. These composite nanosystems show therefore great potential for application in chemical sensors, photonic actuators, and environmentally responsive materials.



1. INTRODUCTION

Molecular machines are multicomponent (supramolecular) assemblies able to perform a mechanical function following an external chemical, optical, magnetic, or electrical stimulus. Applications include their utilization as tweezers, catalysts, motors, switches, memories, logic gates, and valves of interest in materials science, information technology, and medicine.^{1–6} Threaded and interlocked molecular structures such as rotaxanes, catenanes, and related systems are most appealing platforms for the development of artificial nanoscale machines and motors.

To date, most of the studies on molecular machines have been performed in the liquid phase. Working in solution is convenient from the experimental viewpoint, allowing the molecular degrees of freedom at the base of the machine operations to be fully exploited. However, the random and variable distribution of the molecules prevents the addressing of individual molecules or given ensembles in a coherent way, which is an essential requirement to turn molecular functions into mesoscopic and macroscopic operations.^{6,7} Such functions would be enabled by the interaction of the molecular machines with the environment, triggered by external stimulation of

individual molecules or of a set of them and by suitable effects on the physicochemical properties that allow the actual molecular status to be sensed/read.⁸ Therefore, there is a growing need to interface molecular machines with solid supports, by integrating them with nanostructured materials and devices, and to study the properties of these hybrid materials.

In recent years, pioneering efforts have been devoted to the incorporation of molecular machines and motors in heterogeneous matrices and to the study of efficient strategies for delivering proper stimuli and detecting the resulting molecular changes. Examples include molecular machines deposited on surfaces as mono- or multilayers,^{9–12} embedded in solid crystals,¹³ liquid crystals,¹⁴ and gels,¹⁵ coupled with nanoparticles,^{16,17} integrated in electronic circuits,¹⁸ and incorporated in metal–organic frameworks.^{19–22} Only a few studies have been focused on polymeric host materials.^{23–26} In this kind of systems, polymer molding could enable the manipulation of the functional molecules, and provide a way

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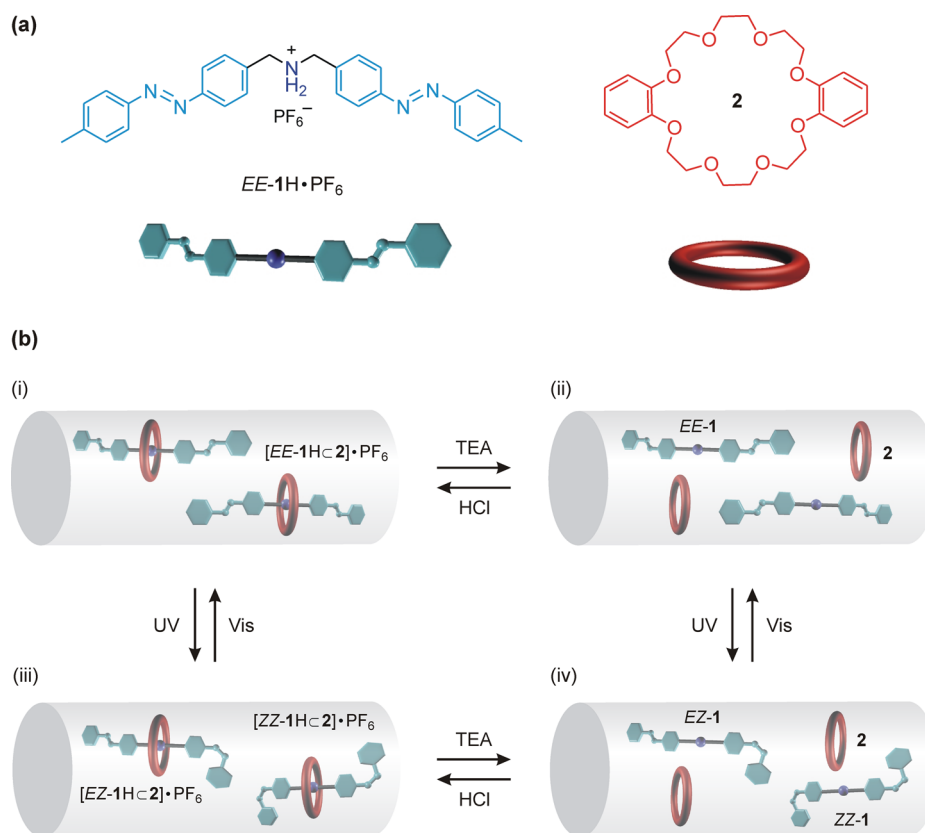


Figure 1. (a) Molecular structure of the axle ($EE-1H \cdot PF_6$) and ring (**2**). (b) Schematic representation of the photochemical ($E-Z$ isomerization of the azobenzene end units of the axle) and chemical (deprotonation/reprotonation-induced dethreading/rethreading) processes that the axle and ring molecular components can undergo inside the fibers.

to induce supramolecular organization or rearrangements of the incorporated molecular guests. Moreover, polymers can convey the proper stimuli to the embedded nanomachines, and they can be used for sensing the molecular state in real time by optical, electrical, or chemical signals. The large variety of polymers and nanocomposites with tailored physicochemical properties, together with the easy processability and low cost, are further elements of interest toward this approach. For instance, azobenzene-based photochromic systems have been integrated in brush-type polymers²⁷ and liquid-crystalline polymer springs,²⁸ an approach allowing reversible, macroscopic morphological changes of the composite material to be achieved by photoisomerization of the azobenzene units.

Furthermore, the features of polymer structures embedding molecular machines can be improved via nanostructuring, in order to enhance the surface-to-volume ratio, and make the interactions with the surrounding environment more efficient.^{26,29–31} In particular, polymer nanofibers show several advantages, such as their flexibility and the possibility to be easily integrated in optical or electronic devices. Moreover, the composition and structure of nanofibers made by electrospinning^{32–34} can be easily tailored to target specific applied stimuli (physical and/or chemical). The high stretching rate of electrified jets favors the orientation of macromolecules and of embedded dopants and nanoparticles as well.^{35–40}

Here, we present electrospun nanocomposites embedding a self-assembling rotaxane-type system that is responsive to both optical (UV–vis light) and chemical (acid/base) stimuli.⁴¹ The system is composed of a polyether molecular ring and a molecular axle containing a recognition site for the ring and

terminated with photoactive azobenzene units. Solution studies showed that the $E-Z$ photoisomerization of the terminal azobenzene units converts the system between thermodynamically stable (pseudorotaxane) and kinetically inert (rotaxane) forms.⁴¹ Here, the reversible photoisomerization of the azobenzene units in the nanofibers is investigated by absorption spectroscopy. Moreover, the reversible dethreading and rethreading of the axle and the ring in the electrospun fibers by chemical stimulation is also assessed, using photoluminescence (PL) spectroscopy. Interestingly, we find that the macroscopic mechanical properties of the nanofiber mats embedding the pseudorotaxane complexes are influenced by their dethreading and rethreading processes at the molecular scale.

2. RESULTS AND DISCUSSION

Nanofibers Embedding Pseudorotaxane Molecular Complexes. The system is composed of a molecular axle, $EE-1H \cdot PF_6$, comprising a dibenzylammonium recognition site and two azobenzene end groups and a dibenzo[24]crown-8 molecular ring **2** (Figure 1a).^{41,42} Poly(methyl methacrylate) (PMMA) is used as matrix due to its good plastic behavior, processability, and optical transparency (up to 93% in the visible spectral range), making it suitable for embedding photoactive compounds.^{25,43,44} In organic solution, $EE-1H \cdot PF_6$ and **2** self-assemble efficiently and rapidly to yield the pseudorotaxane $[EE-1H \cdot 2] \cdot PF_6$.

Our nanofibers, which in their pristine form embed the $[EE-1H \cdot 2] \cdot PF_6$ complex, can go through various photochemical or chemical processes as schematized in Figure 1b. In order to

induce the $E \rightarrow Z$ photoisomerization of the azobenzene end groups [processes (i) \rightarrow (iii) and (ii) \rightarrow (iv) in Figure 1b], the nanofibers are exposed to UV light for increasing time intervals. The photoisomerization of the azobenzene end groups is reversible, and the initial state can be restored upon $Z \rightarrow E$ conversion triggered by irradiation with blue light [processes (iii) \rightarrow (i) and (iv) \rightarrow (ii)]. Furthermore, on the basis of solution behavior, deprotonation of the ammonium recognition site with a base is expected to cause the disassembly of the complex, as schematized in processes (i) \rightarrow (ii) and (iii) \rightarrow (iv) in Figure 1b.^{41,42} Eventually, treatment with an acid should cause reprotonation of the amine center and trigger rethreading of the ring and axle components [processes (ii) \rightarrow (i) and (iv) \rightarrow (iii)]. The dethreading and rethreading of the supramolecular complexes in the fibers, induced by exposure to base and acid vapors, respectively, are here explored with the aim of evaluating the potential of such composite nanosystems as chemical sensors and environmentally responsive materials.

Figure 2a,b shows randomly oriented and nearly aligned fibers containing $[EE-1HC2] \cdot PF_6$, that feature an average

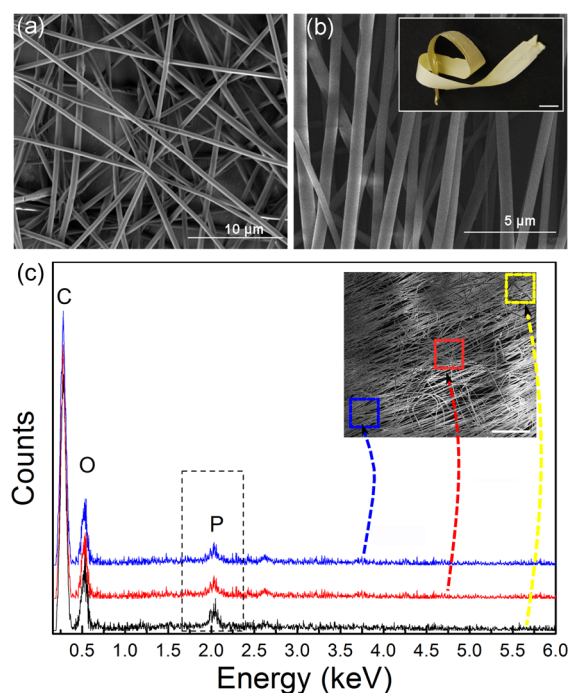


Figure 2. Scanning electron micrographs of randomly oriented (a) and prevalently aligned (b) PMMA/ $[EE-1HC2] \cdot PF_6$ fibers. In (b), about 70% of fibers have their longitudinal axis within 10° from the main alignment direction (see Figure S1). Inset: photograph of a uniaxially oriented, $[EE-1HC2] \cdot PF_6$ -based nanofiber mat (scale bar: 1 cm). (c) EDS spectra of $[EE-1HC2] \cdot PF_6$ -based fibers measured in three different areas of the sample. The spectra are vertically shifted for better clarity. Inset: SEM micrograph of the fibers showing the $150 \times 150 \mu m^2$ investigated regions. Scale bar: $200 \mu m$.

diameter of about 600 and 350 nm, respectively. Aligned fibers are relatively thinner and exhibit a narrower dispersion in diameter, because of the additional mechanical stretching exerted by the rotating collector on the polymeric jet.⁴⁵ The elemental analysis performed by energy dispersive X-ray spectroscopy (EDS) in different areas of the composite nanofibers (Figure 2c) indicates a uniform distribution of dopants throughout the sample mats.

Photoisomerization of $[EE-1HC2] \cdot PF_6$ -Based Fibers.

The occurrence of $E \rightarrow Z$ photoisomerization of the complexes induced by irradiation with UV laser light [process (i) \rightarrow (iii) in Figure 1b] is assessed by the change in the absorption spectrum of randomly oriented fibers (Figure 3a). The intensity of the peak at about 325 nm, characteristic of the EE -form of the axle component,⁴¹ decreases with increasing UV exposure time, as shown in Figure 3b. No more changes in the absorption intensity are observed for exposures longer than about 20 min, indicating that a photostationary state is reached, as commonly observed in azobenzene photoisomerization processes. On the basis of the solution absorption spectra of $EE-1H \cdot PF_6$ and $ZZ-1H \cdot PF_6$ it can be assumed that the absorbance of the Z -azobenzene unit at 325 nm is much smaller than that of the E -isomer and can therefore be neglected. Hence the absorption changes show that about 40% of the E -azobenzene units present in nanofibers are transformed to the Z -form. Since the two azobenzene units of $EE-1H \cdot PF_6$ are equivalent and independent from one another, the photoreacted units are statistically distributed among the axle molecules, and thus the photostationary state should contain about 44% EE -, 32% EZ -, and 24% ZZ -form of the axle. Exposure of the UV irradiated samples to a blue laser beam causes opposite spectral changes (Figure 3c), indicating that the Z -azobenzene units are converted to the E -isomer [process (iii) \rightarrow (i) in Figure 1b]. The initial value of absorption intensity at 325 nm is completely recovered after about 10 min (Figure 3d), related to the complete transformation of the molecules back to the original EE -form. This demonstrates both the expected reversibility of azobenzene photoisomerization in different environments^{46,47} and, more interestingly, the excellent photochemical properties of the axle which are retained in nanofibers. Figure 3e shows the change in the intensity of the 325 nm peak upon five consecutive cycles of alternated irradiation periods with UV and blue laser light, evidencing fatigue resistance in photochemical cycling.

In electrospun nanofibers, both the polymer and embedded guest molecules can be stretched because of the applied electric field.^{39,45,48–50} Polarized infrared spectroscopy can provide insightful information on the resulting molecular alignment. For instance, the Fourier-transform infrared (FTIR) absorption signal at 1730 cm^{-1} , arising from the $C=O$ stretching of PMMA,⁵¹ has a maximum for incident light polarized along the fiber axis, a result that evidences a partial alignment of the PMMA molecules (Figure 4a). For this mode we measure a dichroic ratio (R)³⁹ of 1.5, calculated as $R = A_{\parallel}/A_{\perp}$, where A_{\parallel} and A_{\perp} are the absorption intensities for light polarized parallel and perpendicular to the fiber axis of alignment, respectively. A similar analysis is performed for the absorption peak at 1592 cm^{-1} , attributed to vibration of the $C=C$ on the benzene ring⁵² (Figure 4a, inset) of the molecule 2 (see Figure S2). The found dichroic ratio of ~ -0.73 for this signal is indicative of a partial alignment of the $[EE-1HC2] \cdot PF_6$ complex along the fiber axis.

As the azobenzene photoisomerization could affect such a molecular arrangement, we analyzed the absorption intensity at 1592 cm^{-1} varying the polarization angle of the incident light (Figure 4b). The dependence of the transmitted intensity (I) of linearly polarized light upon passing through aligned fibers, on the angle (θ) formed by the directions of polarization of the incident light and the main axis of alignment, is roughly described by Malus' law, $I = I_0 \cos^2 \theta + I_1$, where I_1 indicates the intensity of the unpolarized background. Following UV

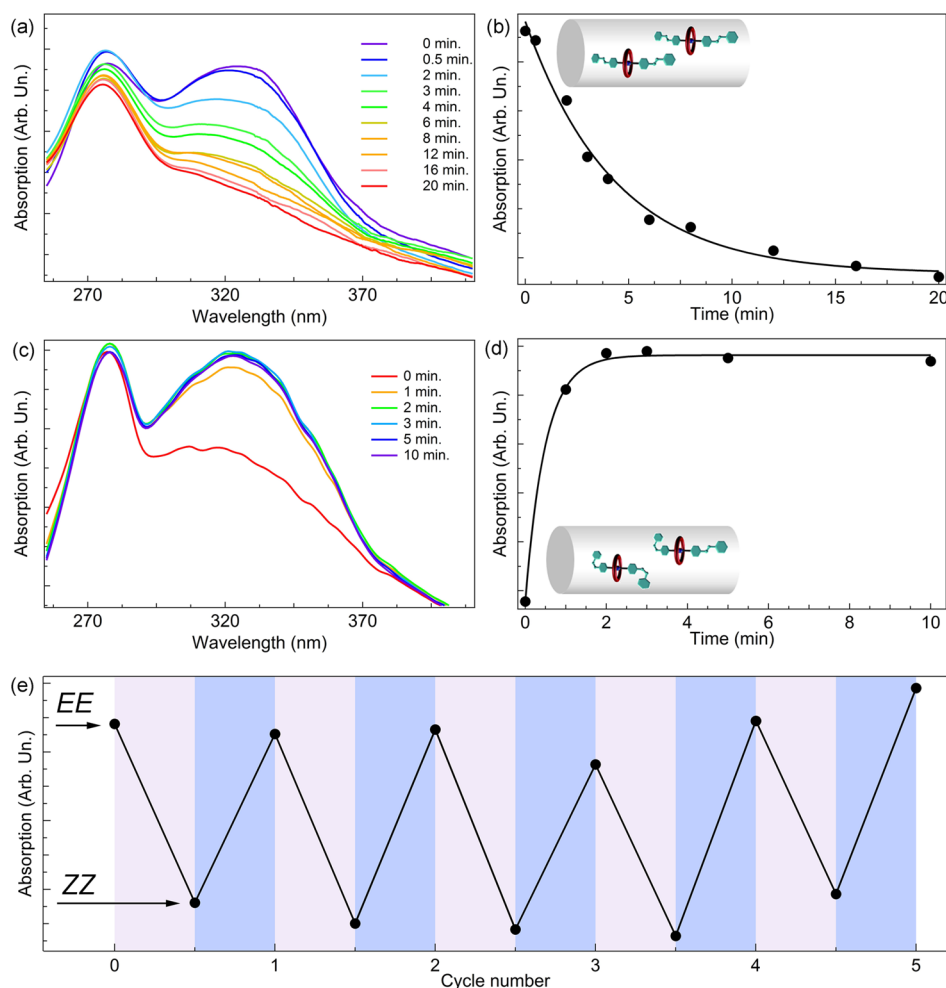


Figure 3. (a) Absorption spectra of $[EE-1HC2] \cdot PF_6$ -based nanofibers after different UV exposure intervals. Irradiation parameters: $\lambda = 355$ nm, pulsed incident fluence = $60 \mu J cm^{-2}$. (b) Absorption values ($\lambda = 325$ nm) vs UV exposure time, corresponding to the spectra shown in (a). The continuous line is a guide for the eye. (c) Absorption spectra of fibers, previously subjected to exhaustive UV irradiation, for increasing time intervals of blue laser light exposure. Irradiation parameters: $\lambda = 405$ nm, incident intensity = $1.4 mW cm^{-2}$. (d) Plot of the absorption values ($\lambda = 325$ nm) versus blue laser exposure time, corresponding to the spectra shown in (c). The continuous line is a guide for the eye. (e) Absorption changes at $\lambda = 325$ nm, for consecutive cycles of alternated UV and blue laser irradiation.

exposure, we do not find significant differences in the polarization curve. The relative positions of the azobenzene moieties with respect to the axle longitudinal direction are expected to change as a consequence of $E \rightarrow Z$ photoisomerization. Hence, our findings indicate that upon photoisomerization the transition moments related to the signal of **2** at $1592 cm^{-1}$ are not significantly tilted with respect to the fiber longitudinal axis. This result suggests that the geometrical changes associated with isomerization have a negligible effect on the ring-axle supramolecular arrangement.

Dethreading and Rethreading by Acid/Base Gas Exposure. In order to study the eventual dethreading and rethreading of the axle and ring in fibers, these are exposed to triethylamine (TEA) and HCl vapors for time intervals up to 4 h, and the molecular assembly is extensively analyzed by PL spectroscopy. Upon UV excitation, the ring has a characteristic fluorescence emission peaked at about 310 nm,^{53–55} whose intensity is completely quenched upon association with a molecular axle containing either *E*- or *Z*-azobenzene units (Figure S3). Conversely, the axle is not luminescent (Figure S3). We first investigate the fluorescence properties of the pseudorotaxane system in solution, following base/acid treat-

ment in order to induce the dethreading and rethreading of the ring and the axle.⁴¹ The results of such an analysis (Figure S4) show a quenching of the ring fluorescence upon formation of the pseudorotaxane complex, compared to a solution containing only the ring. The fluorescence intensity increases upon addition of TEA, suggesting that the deprotonated axle undergoes dethreading, and decreases again by adding HCl, in agreement with the rethreading of the protonated axle. The incomplete recovery of the initial emission intensity (Figure S4) is most likely related to the fact that, in $CHCl_3$, the strongly coordinating Cl^- anions compete with the ring for the ammonium site more efficiently than the PF_6^- counterions.^{56,57}

Therefore, the measurements of the ring PL provide a mean to investigate the dethreading and rethreading of the axle and ring. Here, this method is used to study such processes in solid-state fibers, since the spectral features of the axle and the ring are preserved upon electrospinning the molecular components with PMMA (Figure S5). Figure 5a shows the PL spectrum of the nanofibers embedding the $[EE-1HC2] \cdot PF_6$ complex, which exhibits an emission band at $\lambda_{max} = 305$ nm attributed to the uncomplexed ring molecules. The intensity of this emission band increases remarkably after exposure to TEA vapors

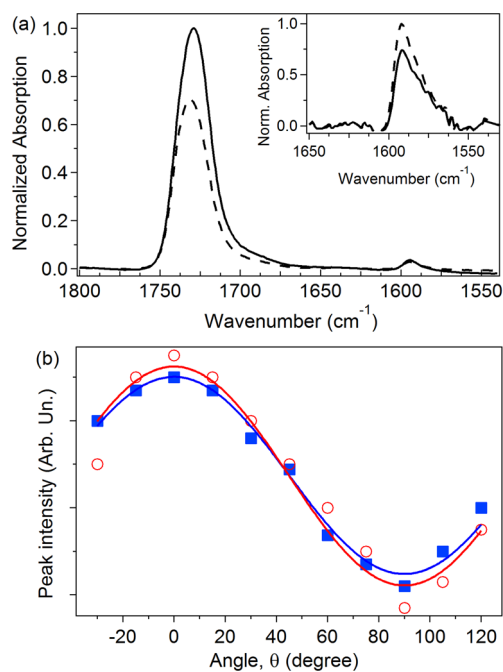


Figure 4. (a) FTIR spectra of $[EE-1HC2] \cdot PF_6$ -based fibers before UV exposure, taken at two angles of light polarization (0° and 90° with respect to fibers axis of alignment; continuous and dashed line, respectively). Inset: close-up of spectra around 1592 cm^{-1} . (b) Transmitted intensity of the peak at 1592 cm^{-1} vs angle (θ) formed by the polarization direction of incident light and the axis of alignment, before (full squares) and after (empty circles) UV exposure (pulsed incident fluence = $60\ \mu\text{J cm}^{-2}$). Continuous lines represent the data fitting according to Malus' law.

(Figure 5b) and decreases again to smaller values after subsequent exposure to HCl vapors (Figure 5c). These results are consistent with a dethreading of the ring from the axle upon deprotonation of the ammonium site, induced by the TEA vapors [Figure 1b, process (i) \rightarrow (ii)], and with successive rethreading caused by acid treatment [Figure 1b, process (ii) \rightarrow (i)]. The observed increase of the ring PL intensity suggests that the TEA-induced dethreading is quantitative (Figure 5a,b). In addition, the initial PL intensity is almost exactly restored upon HCl exposure (Figure 5a–c), in contrast to what observed in solution (Figure S4a–c). Such a finding suggests that rethreading is more efficient in the fibers with respect to the organic solvent, possibly because the association of the chloride anions with the ammonium center, which hinders rethreading in organic solution, does not take place in the PMMA matrix.

In control experiments performed by exposing either (i) bare PMMA fibers or PMMA fibers embedding only the ring component to TEA or (ii) PMMA/ $[EE-1HC2] \cdot PF_6$ fibers to N_2 (used as a carrier gas for TEA treatments), the samples do not show any variation of the intensity of the PL band attributed to the ring (Figure S6). These results evidence that the change of the ring PL signal cannot arise from interactions of either the fiber matrix or the fluorescent component with the TEA molecules. Overall, our findings indicate that the exposure of $[EE-1HC2] \cdot PF_6$ fibers to TEA vapors causes the disassembly of the threaded complex, which is reversibly reassembled by treatment with HCl.

In order to have a more in-depth understanding of the dethreading/rethreading kinetics, $[EE-1HC2] \cdot PF_6$ -based fibers

are exposed to TEA vapors and, successively, to HCl vapors for variable exposure times in a range comprised between 15 min and 4 h. The results, shown in Figure 5d,e, highlight that most of the PL intensity increase/decrease occurs within about 120 min. Such time-dependent changes can be attributed to the barrier associated with the dethreading/rethreading processes in the polymer matrix, combined with gas diffusion in electrospun fibers, which is known to occur on time scales of a few tens of minutes.⁴⁴ Furthermore, $[EE-1HC2] \cdot PF_6$ fibers exposed first to UV laser light and then to TEA/HCl do not show significant differences (Figure 5f–h) compared to samples directly treated with TEA/HCl (Figure 5a–c), indicating that the dethreading/rethreading [processes (iii) \rightarrow (iv) and (iv) \rightarrow (iii) in Figure 1b] of the ring and axle components in the fibers can occur even after the photoisomerization of the azobenzene units of the axle, as reported in solution.^{41,42} Similarly, the *E* \rightarrow *Z* photoisomerization of the azobenzene units of the axle in the fibers is not significantly affected by the TEA-induced dethreading, as shown in Figure 6 wherein processes (ii) \rightarrow (iv) and (iv) \rightarrow (ii) of Figure 1b are examined.

Mechanical Properties. Finally, we investigate the possible corresponding variations of the *macroscopic* mechanical properties of the fibers due to the embedded pseudorotaxane components. The results reported in Figure 7 clearly show that, while the mechanical properties of pristine PMMA nanofibers are in good agreement with previous reports,⁵⁸ the presence and, more interestingly, the dethreading of the axle and ring affect the mechanical properties of nanofibers. In particular, while the maximum strain (Figure 7b) and ultimate tensile strength (Figure 7c) show minor changes, large differences are found for the Young's modulus. The average Young's modulus of pure PMMA nanofibers aligned in arrays is here measured to be around (50 ± 10) MPa (not varying significantly following exposure to TEA), whereas it doubles in consequence of the embedment of either $[EE-1HC2] \cdot PF_6$ (Figure 7a) or its axle component alone (Figure S7a). Upon TEA exposure, the Young's modulus of $[EE-1HC2] \cdot PF_6$ -doped fibers further increases up to (160 ± 20) MPa, indicating a significantly enhanced stiffness of the nanofibrous mat upon ring-axle dethreading. Instead, control experiments carried out on nanofibers with only the axle component do not lead to a similar increase of the Young's modulus (Figure S7), thus ruling out a significant influence of the bare axle deprotonation on the mechanical properties at macroscale.

In addition, we find that the photoisomerization of the azobenzene end groups by exposure to UV light [namely, performing the (ii) \rightarrow (iv) and (iv) \rightarrow (ii) processes in Figure 1b] does not affect significantly the fiber mechanical properties. Indeed, as shown in Figure 8 the strain, the tensile strength and the modulus of $[EE-1HC2] \cdot PF_6$ -based fibers (both taken in their pristine state and following TEA treatment) remain almost unaltered after UV irradiation. An interpretation of these observations on the basis of the properties of the pseudorotaxane and its molecular components is not straightforward. It can be noticed, however, that the axle component as well as the resulting ring-threaded species are considerably more rigid than the PMMA molecules. It may therefore be argued that fibers in which these dopants are partially aligned with the host macromolecules could exhibit an increased stiffness, as observed in this work. Following TEA treatment and ring-axle dethreading, one also observes that the corresponding stress–strain curve becomes sublinear (upward

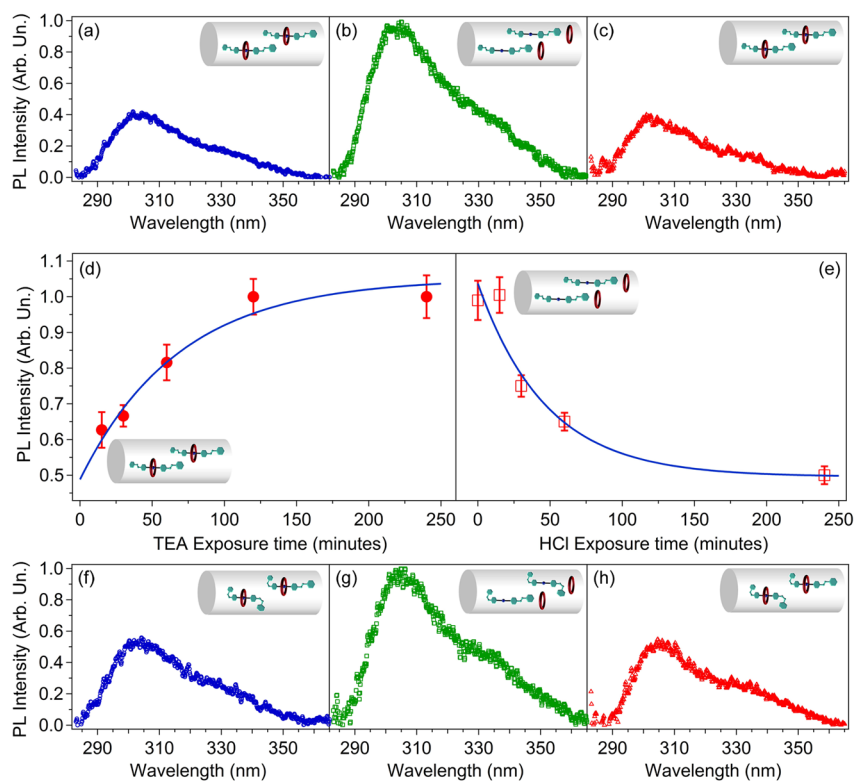


Figure 5. (a–c) PL spectra of $[EE-1HC2] \cdot PF_6$ -based fibers, before (a) and after (b) exposure to TEA vapors for 4 h. (c) PL spectra of the samples shown in (a) and (b) after an additional exposure to HCl vapors for 4 h. (d) and (e) PL intensity of $[EE-1HC2] \cdot PF_6$ -based fibers versus TEA (d) and HCl (e) exposure times. The samples used in (e) are first exposed to TEA vapors for 4 h. Continuous lines are guides for the eye. (f–h) PL spectra of $[ZZ-1HC2] \cdot PF_6$ -based fibers PMMA, before (f) and after exposure to TEA (g) and HCl (h) vapors, respectively. In all cases, excitation was performed at 266 nm.

triangles in Figure 7d), which suggests that the $[EE-1HC2] \cdot PF_6$ -doped fibers undergo microscopic rearrangements and eventually plastic flow in their strain-hardening regime. This is especially interesting since the so obtained Young's modulus $[(160 \pm 20) \text{ MPa}]$ is significantly higher than the value found for fibers embedding the free axle $[(100 \pm 10) \text{ MPa}]$, Figure S7]. Taken together, these results indicate that a specific increase of stiffness is associated with fibers embedding dethreaded molecular components. Issues related to the tensile properties of nanofibers and nanocomposites have been largely investigated for polymers embedding WS_2 ⁵⁸ or carbon^{59–61} nanotubes, whereas the behavior of molecular blends and their nanostructures is rarely studied. In general, interfacial shear stresses between different components are at the base of load transfer to rigid fillers from the plastic matrix, thus determining improved mechanical properties in nanocomposites. Various mechanisms can improve load transfer, such as micro-mechanical interlocking between the host components and the fillers, chemical bonding, or van der Waals interactions,⁵⁹ and achieving a uniform dispersion of dopants in the polymer is also needed, as in our case. In particular, for the fibers studied in this work, dethreaded species can similarly promote load transfer from the plastic matrix due to van der Waals effects and correspondingly improved interfacial interactions between the polymer phase or increasing the volume fraction of the dispersed component in the composite material, an effect which seems to take advantage of the presence of the separated axle and ring components. A full rationalization of the molecular details which lead to the observed mechanical properties will certainly deserve further investigations.

In general, the dependence of the mechanical properties of polymeric fibers on the switching of the supramolecular ring-axle interactions^{62–64} is a significant step forward toward the exploitation of molecular-scale phenomena, including molecular movements, to bring about effects at the macroscopic level. This problem is indeed of the highest importance for the real world application of molecular devices and machines.^{6–8} Materials like those described here are appealing for the development of sensing components working by chemical and photonic actuation.

3. SUMMARY

In summary, nanofibers embedding self-assembled rotaxane-type supramolecular complexes have been produced via electrospinning. These species are interesting because they are photoactive (azobenzene *E*–*Z* photoisomerization) and responsive to base–acid stimuli (chemically driven dethreading–rethreading). The possibility to reversibly interconvert the azobenzene units of the axle between their *E*- and *Z*-forms in the nanofibers has been assessed as well as the reversible switching features brought about by dual (optical and chemical) stimulation of the embedded supramolecular system. Importantly, we demonstrated the influence of the ring and axle molecular components and of their dethreading on the macroscopic mechanical properties of fibers. These results are a step forward toward the incorporation of functional molecular machines into solid polymeric nanostructures and their exploitation for the development of innovative devices and materials for photonics, (bio)chemical sensing, molecular release, and mechanical actuation.

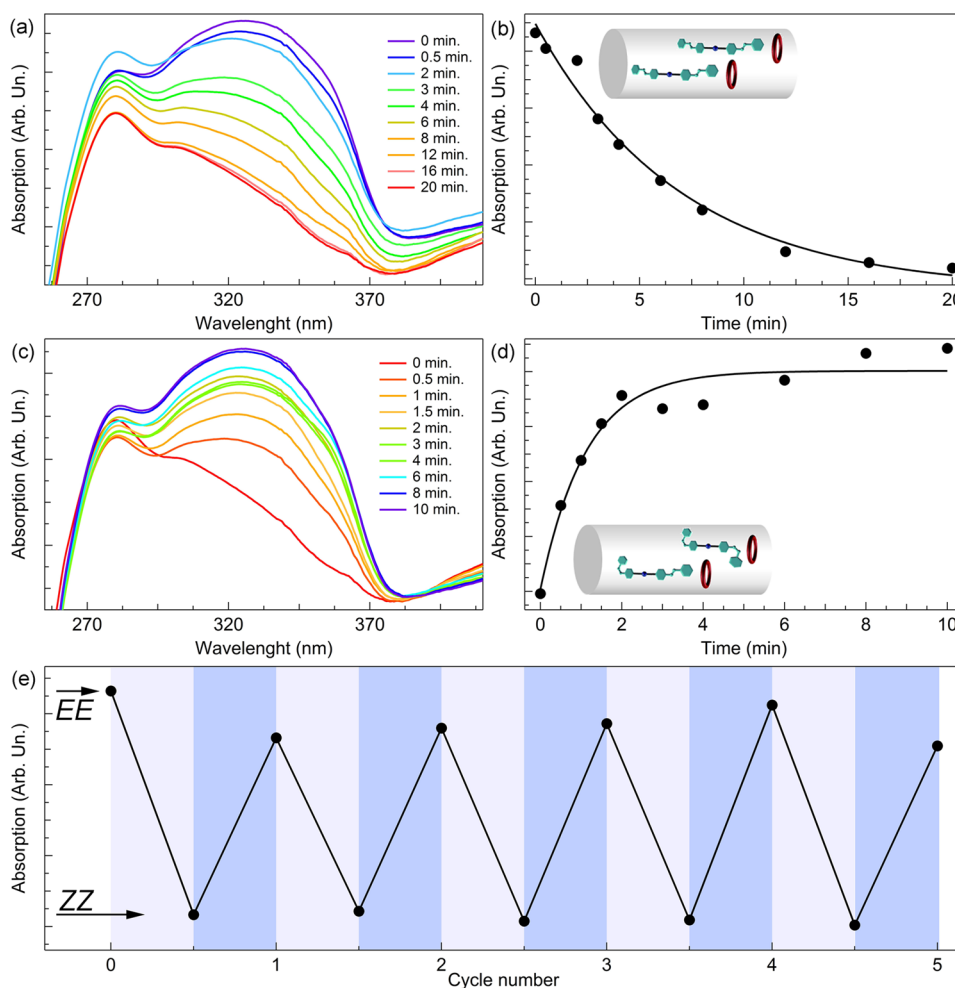


Figure 6. (a) Absorption spectra of $[EE-1HC2] \cdot PF_6$ -based nanofibers after exposure to TEA for 2 h and subsequent, different UV exposure intervals. Irradiation parameters: $\lambda = 355$ nm, pulsed incident fluence = $60 \mu J cm^{-2}$. (b) Corresponding absorption values ($\lambda = 325$ nm) as a function of UV exposure times. The continuous line is a guide for the eye. (c) Absorption spectra of nanofibers, previously underwent UV irradiation, for increasing time intervals of blue laser light exposure. Irradiation parameters: $\lambda = 405$ nm, incident fluence = $1.4 mW cm^{-2}$. (d) Corresponding absorption values ($\lambda = 325$ nm) as a function of exposure time. The continuous line is a guide for the eye. (e) Plot of the absorption changes at $\lambda = 325$ nm, for consecutive cycles of alternated UV and blue laser irradiation.

4. EXPERIMENTAL SECTION

[EE-1HC2]·PF₆-Based Nanofibers. Nanofibers are produced by electrospinning a 10% w/w PMMA (120 kDa) solution in chloroform (Sigma-Aldrich). The two pseudorotaxane components are added at a total concentration of 42% w/w compared to PMMA (i.e., 15 mg of *EE-1H*·PF₆ and 48 mg of **2** in 150 mg of PMMA). Under these conditions the threaded complex $[EE-1HC2] \cdot PF_6$ is quantitatively afforded, as its stability constant in chloroform is of the order of $10^6 M^{-1}$.⁴¹ $[EE-1H] \cdot PF_6$ -based fibers are produced adding 15 mg of *EE-1H*·PF₆ in PMMA for control mechanical experiments. The electrospinning apparatus comprises a syringe (1 mL, Hamilton) and a 27 gauge stainless steel needle connected to a syringe pump (Harvard Apparatus) and a high-voltage supply (EL60R0.6–22, Glassman High Voltage Inc.). A positive voltage (5 kV) is applied to the needle. The solution is supplied at a constant flow rate of $10 \mu L min^{-1}$, and fibers are collected on Al foils or on quartz coverslips. A copper plate biased at –6 kV and placed at 10 cm away from the needle is used as collector. Arrays of uniaxially aligned fibers are produced by a metallic collector rotating at 2500 rpm. Electrospinning is performed at ambient conditions (room temperature and relative humidity of about 50%). Fibers are stored at room temperature before analysis. The morphology and the elemental composition of the pseudorotaxane-embedding fibers are investigated by a scanning electron microscopy (SEM) system (Nova NanoSEM 450, FEI) equipped with an energy

dispersive X-ray spectrometer (Quantax, Bruker). SEM images and elemental data are collected with an accelerating voltage of 1–5 kV and of 30 kV and with an aperture size of 30 and 100 μm , respectively.

Photoisomerization. The *E* → *Z* photoisomerization of the azobenzene end groups is induced by exposure to UV light using the third harmonic of a pulsed Nd:YAG laser (Spectra Physics $\lambda = 355$ nm, repetition rate = 10 Hz, pulse duration = 10 ns) for 0–20 min. The reversibility of the azobenzene photoisomerization (i.e., the *Z* → *E* conversion) is investigated by exposing samples to blue laser light (Micro Laser System, Inc.), with incident intensity $1.4 mW/cm^2$ and $\lambda = 405$ nm for different time intervals (0–10 min). Absorption spectra of the fibers in the spectral interval 250–400 nm are collected by a UV–vis spectrophotometer (Cary 300 Scan, Varian Inc.). Polarized FTIR spectroscopy is carried out using a spectrometer (Spectrum 100, PerkinElmer Inc.) equipped with an IR grid polarizer (Specac Limited, U.K.), consisting of 0.12 μm -wide Al strips. To this aim, freestanding samples of aligned fibers are analyzed in transmission mode.

Dethreading/Rethreading Experiments. In order to induce the dethreading and rethreading of the ring and axle components, freestanding nanofibrous mats are exposed to a continuous nitrogen flow carrying TEA and HCl vapors, respectively. The used system is composed of two glass flasks connected through pipes. The first vial, containing 10 mL of TEA (or alternatively HCl), is connected to a controlled nitrogen flow (0.05 L per minute), whereas the second bowl contains the sample. This is kept in the TEA (HCl) saturated

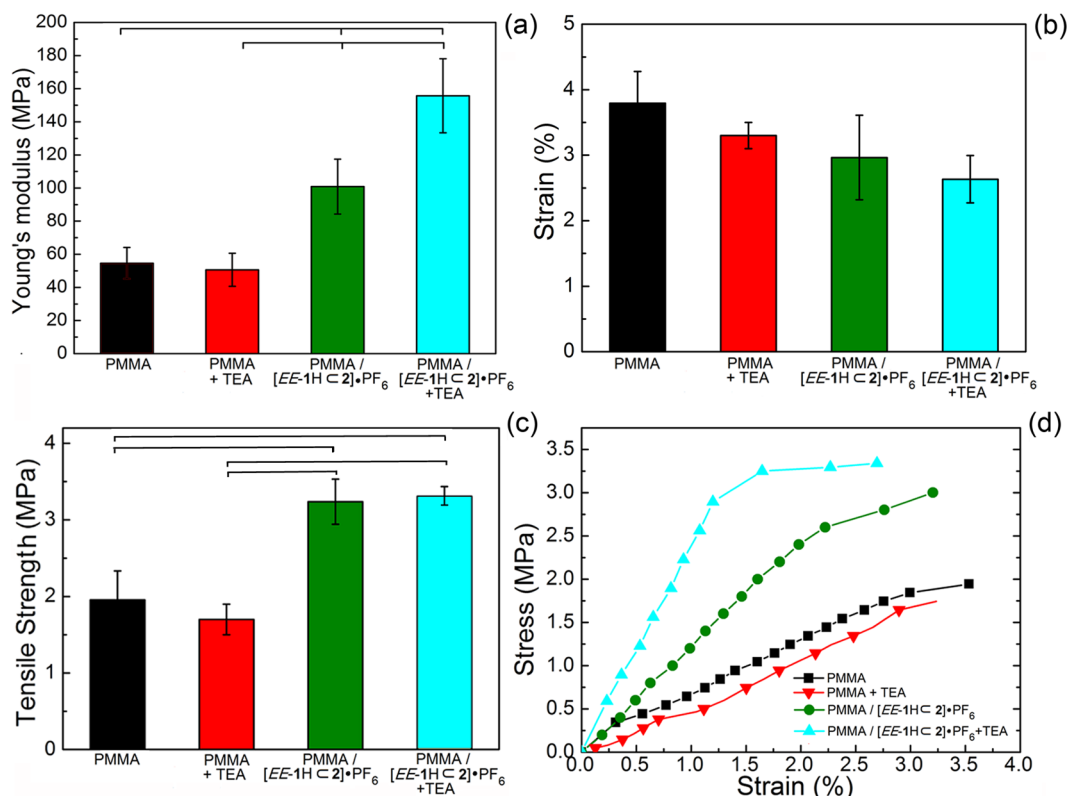


Figure 7. Mechanical properties of PMMA and PMMA/[EE-1HC2]·PF₆ nanofibers before and after TEA exposure. (a) Young's modulus; (b) strain; and (c) tensile strength. (d) Representative stress/strain curves. Results are expressed as mean ± standard deviation. Bars show statistically significant differences ($P < 0.05$).

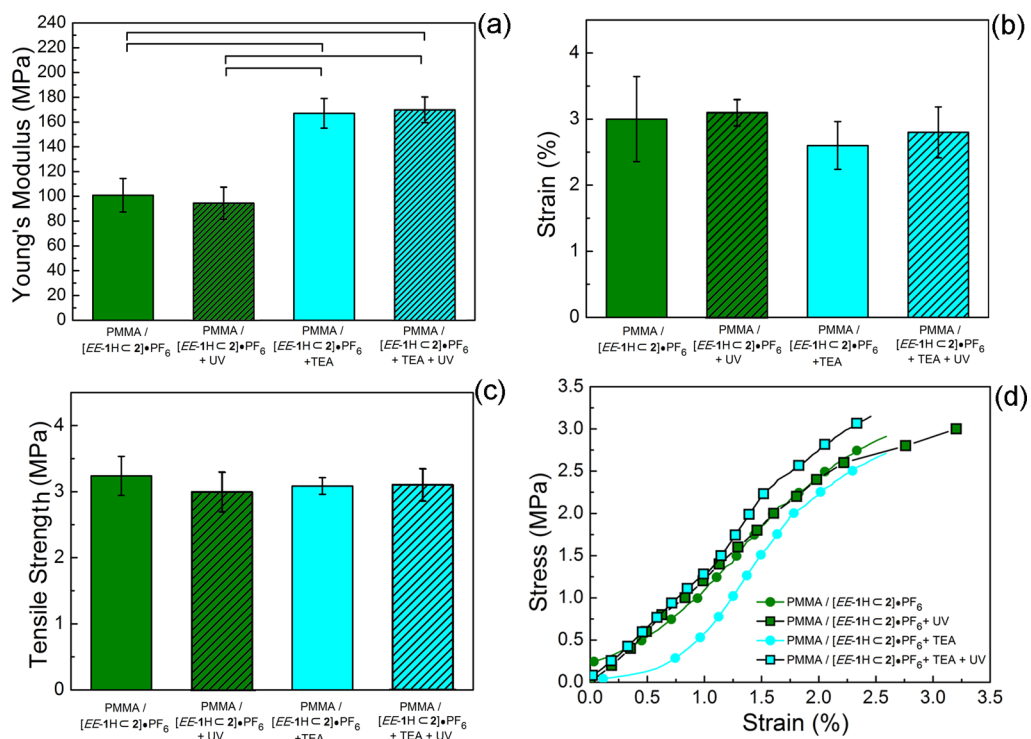


Figure 8. Mechanical properties of PMMA/[EE-1HC2]·PF₆ and PMMA/[EE-1HC2]·PF₆ exposed to TEA vapors for 2 h, before and after UV exposure. (a) Young's modulus; (b) strain; and (c) tensile strength. (d) Representative stress–strain curves. Results are expressed as mean ± standard deviation. Bars show statistically significant differences ($P < 0.05$).

nitrogen flux for variable exposure times in the range 15 min–4 h and investigated before and after TEA/HCl treatments by PL spectroscopy.

To this aim, the samples are excited by the fourth harmonic of the Nd:YAG pulsed laser ($\lambda = 266$ nm), and the emission is collected by a

quartz optical fiber, coupled to a monochromator (iHR320, Jobin Yvon) and a charged coupled device (CCD) detector (Symphony, Jobin Yvon). Since part of the incident light is diffused by the sample and collected by the optical fiber, a background spectrum is subtracted from each measured one (see Figure S8). Dethreading and rethreading experiments are also performed in solution. To this aim, a solution containing *EE*-1H-PF₆ and **2** in 1.5 mL of chloroform is used, and dethreading and rethreading of the pseudorotaxane components are induced by adding 250 μ L of TEA and HCl, respectively.

Mechanical Measurements. Mechanical properties are determined using a dynamic mechanical analyzer (DMA Q800, TA Instruments, New Castle, DE). Each sample made of aligned nanofibers ($n = 5$ specimens) is cut into a rectangular shape (about $10 \times 8 \text{ mm}^2$) before testing, and its thickness is measured using a digital micrometer (0.04–0.06 mm). Stress–strain curves are recorded with a ramp/rate of 1 N min^{-1} (up to 18 N).

■ ASSOCIATED CONTENT

Supporting Information

Additional technical details on nanofiber morphological properties and on spectral and mechanical characterization. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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