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# Partial organic substitution for chemical fertilizer reduces N<sub>2</sub>O emissions but increases the risk of N loss through nitrification in Tibetan farmland

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The combination of organic fertilizers (OFs) and chemical fertilizers (CF) is a promising agricultural management strategy to improve soil fertility while mitigating N<sub>2</sub>O emissions in croplands. However, there is still lacking of in-depth understanding of the effects of different OF and CF blends on N<sub>2</sub>O emissions and the underlying drivers. To this end, we conducted a short-term soil incubation to address the influences of partial OF substitutions for CF, i.e., 40% substitution of compost (CP), Yak dung (YD), *Qingke* straw (QS), and sheep dung (SD) on the processes of nitrification and denitrification in sandy loam soils in the Lhasa Valley. We found that CP, QS, and SD reduced cumulative N<sub>2</sub>O emissions by 53.43%, 25.96% and 16.64%, respectively compared to pure chemical fertilizer (N), except YD caused a significant higher in total N<sub>2</sub>O emissions. Fertilization treatments primarily regulate potential N<sub>2</sub>O emissions by affecting denitrification processes. While ammonia-oxidizing archaea (AOA *amoA*) could be the main driver of nitrification, and *nirS* abundance explained most of the cumulative N<sub>2</sub>O emissions. In addition, NO<sub>3</sub><sup>-</sup>–N tends to accumulate in the farmland soils, indicating an increase in the risk of leaching and nutrient loss. Overall, soil N<sub>2</sub>O emission reduction was favored by applying partial organic fertilizer substitution especially after through compost. Co-composting of animal manure and crop residue has more impressive potential for mitigating farmland N<sub>2</sub>O emissions.

**Keywords** Partial organic substitution, Soil incubation, Microbial gene abundance, Nitrification, Denitrification, N<sub>2</sub>O emissions

 $N_2O$  is a potent greenhouse gas<sup>1,2</sup>, with a global warming potential nearly 298 times of carbon dioxide (CO<sub>2</sub>) over a 100-year period<sup>3</sup>. Due to extensive use of chemical fertilizers, agriculture as one of the major sources of  $N_2O$ , contributes to 25% of the total global  $N_2O$  emissions<sup>1,4,5</sup>. The intensity of  $N_2O$  emissions determines, to a certain extent, the overall greenhouse gas (GHG) balance of agricultural production systems<sup>6,7</sup>. Therefore, this makes it imperative to find effective and alternative fertilization solutions to mitigating  $N_2O$  emissions through reducing use of chemical fertilizers.

It has been documented that combined application of organic fertilizers (OFs) and chemical fertilizers (CFs) may be an effective strategy to reduce  $N_2O$  emissions<sup>8,9</sup>. The intensity of soil  $N_2O$  emissions depends on the production and reduction processes of  $N_2O$ , in which microbial nitrification and denitrification drive  $N_2O$  and NO release<sup>7,10</sup>. A number of studies has demonstrated that the combination of OFs and CF can reduced soil inorganic nitrogen surplus and significantly reduced  $N_2O$  emissions<sup>8,11</sup>. For example, Wu et al. has proven 45% organic substitution can meet the needs of both sustaining soil fertility and mitigating  $N_2O$  emissions. However, some have reported the increasing effect of organic fertilizers on  $N_2O$  emissions from low altitude croplands<sup>13,14</sup>. Cui et al. found that OFs supplied carbon source to enhance denitrification, thereby promoting  $N_2O$  emissions.

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Similarly, Lin et al. reported that fertilizer application led to increased  $N_2O$  emissions, primarily due to the rised abundance of bacterial *amoA* genes. These conflicting results suggest that the factors and processes controlling  $N_2O$  emissions are complex and context dependent in the fields. Current research indicates that various factors, including the C/N ratio of  $OFs^{17}$ , soil type<sup>18</sup>, and others environmental conditions further influence these processes<sup>19</sup>. As a matter of facts, OFs can stimulate microbial immobilization of fertilized nitrogen and, at the same time, act as a source of  $N_2O^{7,20,21}$ . To better understand nitrogen transformation and  $N_2O$  emissions after partial organic substitution, an in-depth mechanistic understanding is needed to disentangle the contribution of nitrification, denitrification to  $N_2O$  emissions.

The identification of nitrification and denitrification genes plays an important role in predicting N<sub>2</sub>O emissions<sup>5,18,22</sup>. Ammonia-oxidizing archaea (AOA amoA) and ammonia-oxidizing bacteria (AOB amoA) convert ammonia (NH<sub>4</sub>+) to nitrite (NO<sub>2</sub>-) through oxidation, which is the main step controlling the rate of nitrification<sup>23,24</sup>. Additionally, Comammox (e.g. *Nitrospira* branch) can oxidize NH<sub>4</sub> + to NO<sub>3</sub> -<sup>25-27</sup>. *nirS*, *nirK*, and nosZ are the functional genes involved in the transformation of denitrifying microorganisms<sup>28,29</sup>. And the composition and activity of these microbial communities may alter the balance between  $N_2O$  production and reduction through nitrogen mineralization and denitrification rates 30,31. Xu et al. found that a strong correlation between AOA amoA and nirS genes rendered a significant increase in N2O emission from OFs. While N2O emissions were reported to be suppressed by organic fertilizers replacing N fertilizers, which was attributed to the reduction of soil mineral nitrogen content and stimulation of denitrifying gene abundance9. In contrast, Other studies reported that co-addition of cow manure and CF decreased the abundance of nirS and nirK, but increased N<sub>2</sub>O emissions<sup>33</sup>. Related studies have focused on the link between N<sub>2</sub>O emissions and denitrifying bacterial community abundance, and these still present conflicting information 19,34-36. However, it remains poorly understood what types of metabolic functional genes are involved in the nitrogen transformation process, especially the ones for the increase or decrease in soil N<sub>2</sub>O emissions, by changing soil physicochemical properties, microbial composition after partial organic substation for CFs. Knowledge gaps remain regarding the main drivers of N<sub>2</sub>O emissions, including nitrification and denitrification genes and environmental factors. Therefore, more efforts are needed to explore multiple microbial functional genes related to the N<sub>2</sub>O production process after organic substitution, which can help understand the underlying mechanisms.

On the Tibetan Plateau, crop residue and livestock manure are applied as OFs for agricultural production<sup>37</sup>. Mostly Tibetan sheep dung and partly air-dried Yak dung are the traditional source of nutrients for crops<sup>38,39</sup>. The straw of Tibetan hulless barley (*Hordeum vulgare* L. var. *nudum*), also known as *Qingke*<sup>40,41</sup>, is composted with animal manure through microbial action to make organic fertilizers<sup>42</sup>. These OFs application can affect soil nitrogen effectiveness directly by providing external N sources and indirectly by affecting soil nitrogen mineralization<sup>6,10,43</sup>. The organic carbon input from OFs can influence microbial activity, potentially affecting N<sub>2</sub>O emissions by modulating nitrification and denitrification processes<sup>44</sup>. Several studies have investigated N<sub>2</sub>O emissions from natural grasslands<sup>45–47</sup> and permafrost<sup>48,49</sup> on the Tibetan Plateau. However, the knowledge of N<sub>2</sub>O emissions after crop straw and animal manure substitution of chemical fertilizers is still lacking and the underlying drivers of nitrogen transformation and N<sub>2</sub>O emissions remain unclear.

In this study, we conducted a short-term incubation experiment to investigate the effect of partial OFs substitution for CF on soil  $N_2O$  emissions under equal nitrogen conditions. We analyzed the abundance of functional genes involved in the formation and reduction of  $N_2O$ , such as AOA amoA and AOB amoA, nirK, nirS, and nosZ as well as the community structure and composition of denitrification genes. Attempts were made to address the following questions: (1) What types of organic substitution can effectively reduce soil  $N_2O$  emissions? and (2) how do nitrogen transformation processes and associated nitrification and denitrification genes mediate  $N_2O$  emissions? We hypothesized that partial organic substitution for chemical fertilizer could mitigate  $N_2O$  emissions compared to CF through modulating soil properties and reducing denitrification gene abundance.

# Materials and methods Preparation of soil and fertilizer

The soils for incubation were collected from the top layer (0-20 cm) in August 2023 from the Lhasa Plateau Biological Research Base  $(29^{\circ}45'\text{N}, 90^{\circ}96'\text{E}, \text{altitude }3650 \text{ m})$ , in Caina Township, Lhasa City, Tibetan Autonomous Region. According to the soil system classification, the test soil was alluvial soil with a sandy loam soil texture. All soil samples were sieved through a 2 mm sieve to remove residual roots, litter, and stones. A portion of the samples were stored in a 4 °C refrigerator for laboratory incubation, while the remaining soils were air dried indoors for measuring soil properties. The basic soil properties were as follows: pH 7.24 (1:2.5, soil/water), total C content, 9.89 g kg  $^{-1}$ ; total N content, 0.63 g kg  $^{-1}$ ; bulk density, 1.23 g cm  $^{-3}$ , and natural moisture content, 7.22% in the field.

OFs include compost, yak dung, *Qingke* straw and Tibetan sheep manure. The raw components of compost were Yak dung, *Qingke* straw, and sheep dung, which were mixed thoroughly and then composted for 68 days using the aerobic composting technology of "air-flow membrane". All OFs materials were dried at 55 °C, grounded and subsequently sieved through a 2 mm mesh for partial organic fertilizer substitution. The chemical properties are shown in Table 1.

#### Laboratory incubation experiment

Soil incubation included six treatments: ambient soil without fertilization (CK) and five equal N fertilization regimes with 40% organic substitution, i.e. chemical fertilizer (urea) alone (N), compost plus urea (CP), yak dung plus urea (YD), Qingke straw plus urea (QS), and Tibetan sheep dung plus urea (SD). We set 12 replicates per treatment. The total nitrogen application rate for all fertilization was designed to be 0.25 mg g<sup>-1</sup> of dry soil. According to the optimal application ratio of OFs and CF in local practice and referenced strategy to improve

OFs type	pН	C (%)	N (%)	C: N	P (%)	K (%)
QS	7.6	41.53	0.71	58.49	0.09	0.74
YD	9.9	22.20	1.82	12.20	0.50	2.53
SD	9.6	23.51	1.73	13.59	0.46	2.59
СР	8.7	22.99	1.98	11.61	0.53	1.53

**Table 1**. The chemical properties of different OFs. CP (compost plus urea), YD (Yak dung plus urea), QS (*Qingke* straw plus urea), and SD (Tibetan sheep dung plus urea).

soil fertility while mitigating  $\rm N_2O$  emissions  $\rm ^{12}$ , the nitrogen ratio of CF to OFs were set at 6:4 in fertilization treatments, i.e., 40% organic substitution for CF. Before incubation, fresh soils were wetted to 60% water holding capacity and pre-incubated in dark conditions at 25 °C for 7 days. Then, 250 g (weight in dry soil) of each treatment was weighed into 500 mL culture bottles, and OF was added to the soil by thorough mixing. All bottles, covered with a thin film with small holes to allow gas exchange, were incubated at 25 °C in the dark for 85 days. Throughout the entire incubation period, soil moisture was kept constant by weighing the soil every other day and subsequently adding water to supplement the water loss through evaporation.

#### Gas and soil sampling and analyses

 $\rm N_2O$  gas samples were collected from culture bottles at 0, 1, 2, 3, 5, 7, 9, 14, 21, 28, 35, 45, 55, 70, and 85 days. At each gas sampling, the culture bottles were tightly sealed with a butyl rubber stopper and cap, and 50 ml of gas samples were taken with a syringe during 0 and 120 min, respectively, and analyzed using a gas chromatograph (Agilent Technologies, 7890B).  $\rm N_2O$  fluxes and cumulative  $\rm N_2O$  emissions were calculated by the following equations, respectively:

$$F = \frac{V}{m} \times \frac{M}{22.4} \times \frac{?c}{?t} \times \frac{273}{273 + T} \tag{1}$$

$$W = \sum_{i=1}^{n} F_i \times t \tag{2}$$

In which, F is the  $N_2O$  flux (mg kg<sup>-1</sup>day<sup>-1</sup>), V is the gas volume in the culture bottle (L), m is the weight of dry soil (g), M/22.4 is the mass density of standard gas (g L<sup>-1</sup>),  $\Delta c/\Delta t$  is the change in concentration per unit time (mg kg<sup>-1</sup>day<sup>-1</sup>), and T is the incubation temperature (°C). W is the cumulative  $N_2O$  (mg kg<sup>-1</sup>), n is the nth measurement, and t is the interval between two samples (day).

Soil samples were destructively collected on days 0, 1, 45, and 85, with three replicates per time point. Day 0 samples were used to determine initial soil conditions to ensure equal nitrogen input across treatments but were not included in further analyses. Soil physicochemical properties were measured on days 1, 45, and 85, while the determination of functional genes at day 85 linked nitrification and denitrification to cumulative  $N_2O$  emissions, minimizing short-term disturbances. Soil pH was measured by the glass electrode method in deionized water at a soil-to-water ratio of 1:2.5. Microbial biomass nitrogen (MBN) was extracted by fumigation with chloroform (CHCl<sub>3</sub>) and potassium sulfate ( $K_2SO_4$ ). Soil  $NO_3^--N$  and  $NH_4^+-N$  were extracted with 1 mol  $L^{-1}$  KCl solution (soil-water ratio 1:10, shaking at 25 °C for 1 h), filtered, and then measured by continuous flow analyzer (AA3). The total phosphorus (TP) and total potassium (TK) were determined by molybdenum-antimony colorimetric method and flame photometer, respectively. Soil moisture content was determined by the ring knife method, and the elemental compositions of soil and organic fertilizer, such as C and N were determined by the Vario Macro CHNS elemental analyzer.

# DNA extraction and quantification of functional genes abundance

DNA was extracted from 0.5 g of incubated soil using the FastDNA\* Spin Kit (MP Biochemicals, Santa Ana, CA, USA). DNA concentration and mass picking were assayed with a Nanodrop 2000 (ThermoSecientific, Wilmington, DE, USA). DNA samples were diluted to 10 ng DNA  $\mu$ L<sup>-1</sup> with sterile water and used for quantitative PCR (qPCR) determination. The abundance of key nitrogen cycle microbial functional marker genes (AOA *amoA*, AOB *amoA*, *nirK*, *nirS*, *nosZ*) was quantified using real-time fluorescent qPCR on the Step One Plus Real-Time PCR System (ABI, USA), and all the primers and sequences used for qPCR experiments are shown in Table S1.

#### Functional gene sequencing

Diversity and community structure containing functional genes for denitrification (*nirK*, *nirS*, and *nosZ* genes) were analyzed by high-throughput sequencing. The library construction process included DNA fragmentation (enzymatic shearing), end repair, adapter ligation, purification and PCR amplification. The primers used for amplification were the same as those listed in Table \$1. Thermocycling conditions consisted of 3 min at 95 °C, PCR amplification was performed under the following thermocycling conditions: an initial denaturation at 95 °C for 3 min, followed by 9 cycles (20 s at 98 °C, 15 s at 60 °C and 30 s at 72 °C), and a final extension of 5 min at 72 °C until halted by user. PCR products were purified with Vazyme DNA Clean magnetic beads and quantified using Qubit 3.0 (Thermo Fisher Scientific, Waltham, MA, USA). Mixtures of amplification products were sequenced using the Illumina NovaSeq 6000 sequencing platform (Illumina Inc., San Diego, CA, USA).

#### Statistical analyses

Prior to conducting the One-way ANOVA, data normality was assessed using the Shapiro–Wilk test. One-way ANOVA and Tukey's HSD test were then performed using IBM SPSS 26.0 statistical package to address the effect of different fertilization treatments on soil properties, abundance of nitrification, and denitrification functional genes. Alpha diversity and Beta diversity analyses of AOA amoA, AOB amoA, nirK, nirS, and nosZ genes were performed using QIIME2 2020.6 software. Beta diversity of functional genes in different fertilization treatments was assessed by calculating weighted UniFrac distance matrix and Non MetricMulti Dimensional Scaling (NMDS). Line graphs, bar graphs, and box plots were plotted using Origin (2024b) software. Redundancy analysis (RDA) was performed using CANOCO 5.0 to determine the relationship between soil properties, abundance of microbial functional genes and cumulative  $N_2O$  emissions. We used the "rfPermute" package in R (version 4.0.3) for Random Forest (RF) analysis to determine the main predictors of soil and microbial properties on cumulative  $N_2O$  emissions. The "plspm" package was used to construct partial least squares structural equation modeling (PLS-SEM) to evaluate nitrogen mineralization processes and direct and indirect pathways affecting  $N_3O$  emissions.

#### Results

#### N<sub>2</sub>O emissions in the process of incubation

 $N_2^{7}O$  emissions showed similar variations among treatments, increasing rapidly within 10 days, then decreasing sharply and remaining stable after 40 days of incubation (Fig. 1a). CP had the lowest emissions, representing by both at peak and cumulative values, whereas YD was the highest, displaying 347.95% and 289.20% higher than CK and N treatment, respectively. As for the cumulative  $N_2O$  emissions during incubation, compared to N treatment, YD significantly increased by 1.15-fold, while CP, QS, and SD reduced by 53.43%, 25.96% and 16.64%, respectively (Fig. 1b).

# Changes in soil properties during incubation

Compared with CK, OFs and N significantly increased the levels of  $NO_3^--N$  and  $NH_4^+-N$  (p<0.05). The  $NO_3^--N$  levels tended to increase substantially, while  $NH_4^+-N$  dropped significantly throughout the incubation (Fig. 2a,b), N treatment released  $NH_4^+$  immediately, reaching the highest  $NH_4^+-N$  content in the first day. In addition, the partial organic substitution treatments significantly increased soil pH during the early incubation. Among them, N treatment had the highest pH content due to hydrolysis of urea. As incubation progress, soil pH gradually decreased and reached a relatively constant level in all OF substitution treatments. Among them, soil pH in the N treatment was the lowest (Fig. 2c). In the initial incubation, MBN and SOC were significantly higher in fertilized treatments than in CK, but had no difference between N and OFs treatments (Fig. 2d,e). At the end of incubation, the MBN contents were the highest in YD and N treatment (Fig. 2d).

#### The abundance of functional nitrification and denitrification genes

Compared to CK, OFs and N treatment had lower AOA *amoA*, but higher AOB *amoA* and denitrification genes (*nirK*, *nirS* and *nosZ*). In contrast to N treatment, the nitrification of OFs could be dominated by AOA *amoA* (Fig. 3). The copy numbers of *nirK* and *nirS* were higher in N treatment than in OFs except YD, especially, CP treatment had the lowest gene copy number (Fig. 3c,d). Conversely, *nosZ* gene abundance showed higher performance in OFs than in N treatment except for YD, among them the CP had the highest (Fig. 3e).

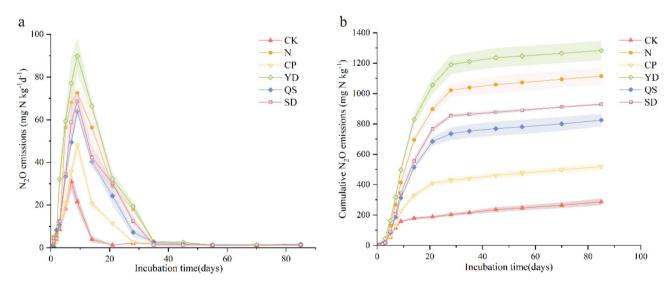
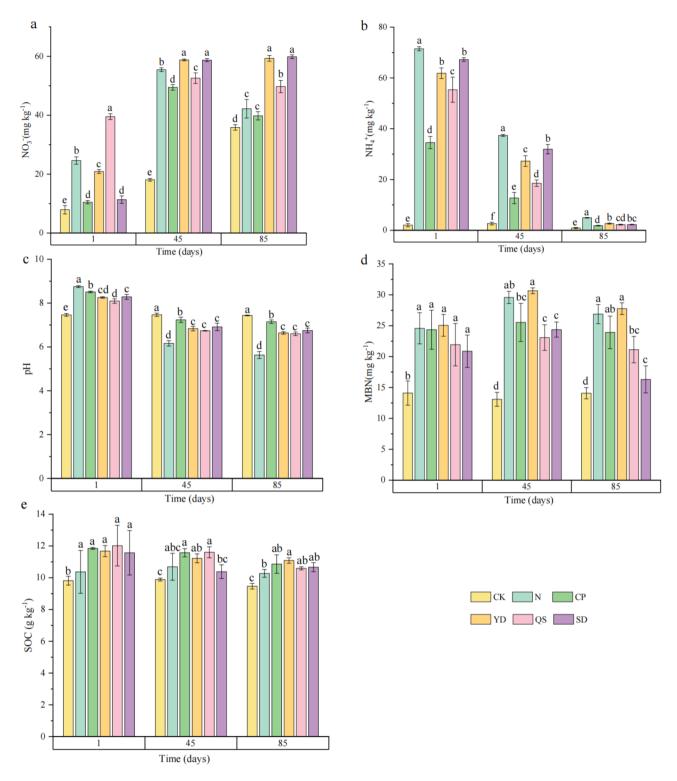


Fig. 1. Changes in  $N_2O$  emissions (a) and cumulative  $N_2O$  emissions (b) during incubation as affected by different fertilization treatments. CK (unfertilized control), N (chemical fertilizers alone), CP (compost and chemical fertilizers), YD (Yak dung and chemical fertilizers), QS (*Qingke* straw and chemical fertilizers), and SD (Tibetan sheep dung and chemical fertilizers). The shaded area represents the standard error of the mean.



**Fig. 2.** Changes in soil properties over time under fertilizer treatments. CK (unfertilized control), N (chemical fertilizers alone), CP (compost and chemical fertilizers), YD (Yak dung and chemical fertilizers), QS (*Qingke* straw and chemical fertilizers), and SD (Tibetan sheep dung and chemical fertilizers).

# Denitrification gene diversity

Fertilization increased the Chao1 index of denitrification genes (Table 2). Specifically, fertilization treatments increased *nirK*, *nirS*, and *nosZ* genes by 4.26–57.90%, 25.26–77.37%, and 6.02–82.23% compared with CK (Table 2). The denitrification richness (Chao1) was higher in YD than in the other fertilization treatments. However, there was no significant difference in *nirK* abundance between SD and CK treatments. The OFs and CF increased the enrichment of *nirK* by 1.01–18.18% compared to the CK treatment (Shannon). The *nirS* and *nosZ* 

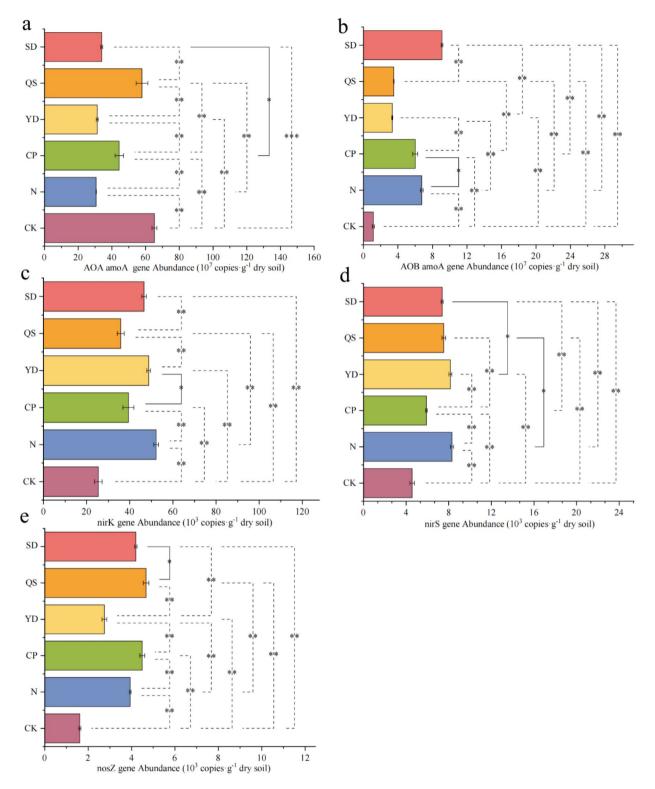
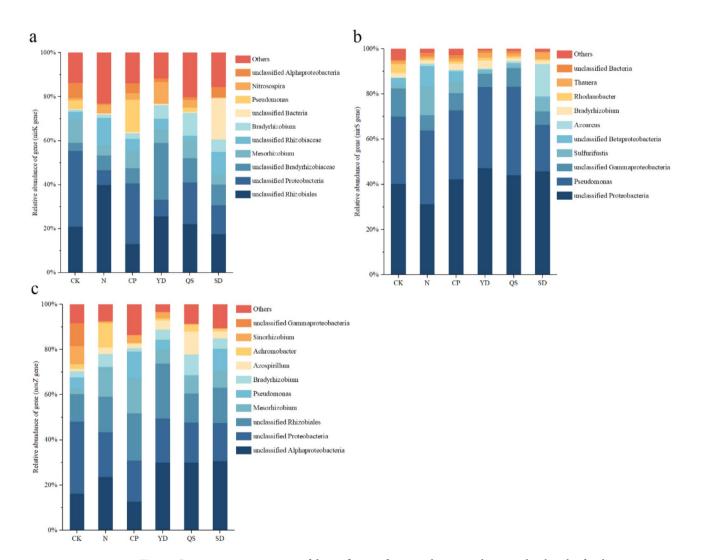


Fig. 3. Changes in nitrification and denitrification functional genes copy numbers under fertilization treatments.  $^*p \le 0.05$ ,  $^{**}p \le 0.01$ . CK (unfertilized control), N (chemical fertilizers alone), CP (compost and chemical fertilizers), YD (Yak dung and chemical fertilizers), QS (*Qingke* straw and chemical fertilizers), and SD (Tibetan sheep dung and chemical fertilizers).

diversity index of CP and QS treatments were significantly higher than that of the single N treatment (P<0.05). NMDS analysis revealed divergent patterns of denitrification genes in different fertilization treatments, except the close proximity clustering of nirS in CP, QS, and SD (Fig. S1). Overall, the combination of OFs and CF significantly increased the diversity of nirS.

	Chao1		Shannon			
	nirK	nirS	nosZ	nirK	nirS	nosZ
CK	680.62 ± 16.94e	470.68 ± 9.01e	847.09 ± 41.27d	5.94 ± 0.05e	$5.87 \pm 0.03d$	5.46 ± 0.14e
N	748.66 ± 20.78d	589.58 ± 8.42de	898.11 ± 10.25d	6.37 ± 0.09b	6.14 ± 0.03b	6.85 ± 0.02d
CP	921.55 ± 23.52b	771.89 ± 21.75b	1543.65 ± 33.51a	6.13 ± 0.03cd	6.54 ± 0.04a	8.04 ± 0.07a
YD	1074.72 ± 27.60a	834.86 ± 18.72a	967.06 ± 35.62c	7.02 ± 0.13a	$6.02 \pm 0.06c$	7.36 ± 0.01c
QS	783.25 ± 11.66c	626.16 ± 54.30cd	1072.28 ± 9.12b	6.27 ± 0.09bc	6.49 ± 0.05a	7.79 ± 0.02b
SD	709.59 ± 7.15e	658.18 ± 13.61c	985.00 ± 26.94c	6.00 ± 0.05de	6.15 ± 0.04b	6.78 ± 0.08d

**Table 2**. The alpha diversity index of *nirK*, *nirS*, and *NosZ* genes under different fertilization treatments. Values are means  $\pm$  standard error (n=3). Different letters indicate significant difference in a column (P<0.05).



**Fig. 4.** Community composition of denitrification functional genes at the genus level under fertilization treatment on day 85. CK (unfertilized control), N (chemical fertilizers alone), CP (compost and chemical fertilizers), YD (Yak dung and chemical fertilizers), QS (*Qingke* straw and chemical fertilizers), and SD (Tibetan sheep dung and chemical fertilizers).

# Denitrification gene community structure and composition

A total of 40,046–76,700 high-quality *nirK*, *nirS*, and *nosZ* sequences were obtained from all treatments, and clustering of Reads at 97.0% similarity level yielded 2901, 2037, and 3808 OTUs (Fig. S2). The Venn diagram showed that 178, 197, and 225 OTUs in *nirK*, *nirS* and *nosZ*, respectively, were common across treatments (Fig. S2). *Proteobacteria* were the dominant phylum of denitrifier genes (Fig. 4). At the genus level, *nirK* functional genes were present, with unclassified *Rhizobiales* being the most abundant taxa in the soil, having a relative abundance of 13.56-40.95%; followed by unclassified *Proteobacteria* (6.44–34.56%) (Fig. 4a). The relative

abundance of unclassified *Proteobacteria* in *nirS* and *nosZ* genes accounted for 30.26–46.75% and 16.66–33.68%, respectively (Fig. 4b,c). Specifically, partial organic substitution treatments in *nosZ* genes significantly reduced the abundance of unclassified *Proteobacteria* by 0.33–12.85% compared to the N treatment. In addition, the genera *Pseudomonas* and *Bradyrhizobium* coexisted in the *nirK*, *nirS* and *nosZ* genes. Among them, the *Pseudomonas* genus of *nirK* and *nosZ* genes had the highest proportion in CP treatment, accounting for 15.34% and 12.54%, respectively. It is noteworthy that *Mesorhizobium* genus in CP treatment was significantly higher in *nirK* (8.13%) and *nosZ* (16.45%) compared to other treatments. However, *Azospirillum* genus (2.03–9.59%) was found to be widely present in *nosZ* during fertilization treatment.

#### The relationship between functional gene abundance and soil properties

The result of RDA analysis showed that the explanations of axis 1 and axis 2 contributed 60.01% and 12.82%, respectively to the variation of nitrification and denitrification genes (Fig. 5a). Among them, cumulative N<sub>2</sub>O emission was the most influential indicator, followed by pH and NO<sub>3</sub><sup>-</sup>N, accounting for 49.01%, 9.30% and 8.90% of variations, respectively. AOB *amoA* and denitrification genes (*nirS* and *nirK*) were positively correlated with the MBN, SOC, NO<sub>3</sub><sup>-</sup>-N, NH<sub>4</sub><sup>+</sup>-N, and cumulative N<sub>2</sub>O emissions, while negatively associated with soil pH. CP, YD, QS, and SD treatments were largely separated from CK and N treatments along the RDA1 direction (Fig. 5a). RF model showed that *nirS* abundance, NH<sub>4</sub><sup>+</sup>-N, and soil pH were the main drivers explaining the cumulative N<sub>2</sub>O emissions (Fig. 5b). It is noteworthy that fertilizer treatments had significant positive effects on NO<sub>3</sub><sup>-</sup>-N (Fig. 6a). Furthermore, fertilization treatments affected soil pH and NO<sub>3</sub><sup>-</sup>-N directly or indirectly through NH<sub>4</sub><sup>+</sup>-N, thereby having positive and negative regulatory effects on soil nitrification and denitrification, respectively, and consequently affecting cumulative N<sub>2</sub>O emissions (Fig. 6a). In particular, denitrification functional genes, fertilization treatment and NH<sub>4</sub><sup>+</sup>-N had the highest total effect on N<sub>2</sub>O cumulative emissions, with 0.89, 0.63 and 0.45, respectively (Fig. 6b).

#### Discussion

This short-term microcosm incubation revealed overall decline in soil  $N_2O$  emissions after partial organic substitution for chemical fertilizer, which supported our hypothesis. Compared to CF, partial organic substitution reduced 16.6–53.4% emissions except yak substitution increased 1.15 fold. However, magnifying  $NO_3^--N$  production during the process of incubation implies great risk of N loss. nirS, and  $NH_4^+-N$  were among the most key factors to regulate  $N_2O$  emissions. Thus, based on this study, partial organic substitution is likely to mitigate  $N_3O$  emissions but increase the loss of nitrate.

# Partial organic substitution affected N<sub>2</sub>O emissions and soil properties

The N and OFs treatments increased the cumulative  $N_2O$  emissions during incubation compared to the CK treatment. This may be due to the fact that OFs provided more substrate for nitrification and denitrification through mineralization (Fig. 2). However, partial organic substitution treatments except YD reduced soil  $N_2O$ 

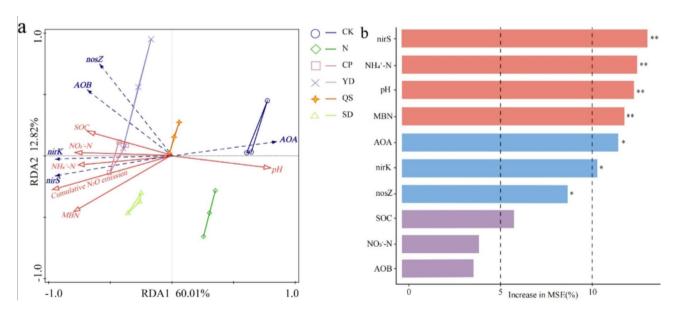
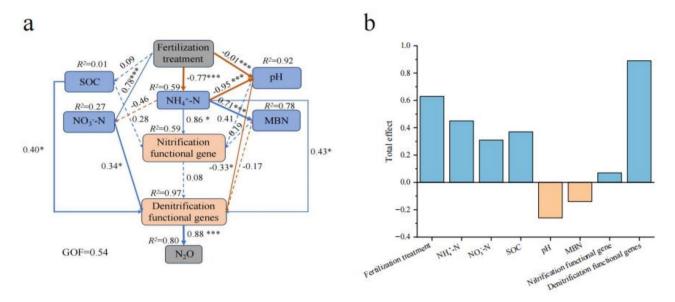


Fig. 5. The relationship between cumulative  $N_2O$  emissions, soil properties, and abundance of functional genes. (a) Principal component analysis of soil properties, cumulative  $N_2O$  emissions, and abundance of nitrification and denitrification genes; (b) RF analysis to determine the main explanatory variables for ranking cumulative  $N_2O$  emissions. MSE is the mean square error of the random forest model. The larger the MSE value, the greater the importance of the variable. CK (unfertilized control), N (chemical fertilizers alone), CP (compost and chemical fertilizers), YD (Yak dung and chemical fertilizers), QS (*Qingke* straw and chemical fertilizers), and SD (Tibetan sheep dung and chemical fertilizers).



**Fig. 6.** PLS-SEM was used to show how fertilizer application affects  $N_2O$  accumulation through key soil properties, nitrification and denitrification functional gene abundance in the main pathway (**a**) and total effect (**b**). The red and blue lines represent positive and negative effects, respectively, and the dashed lines represent non-significant effects.  $\mathbb{R}^2$  values associated with response variables indicate the proportion of variance explained by the relationship with other variables. Significance: \*Denotes p < 0.05, \*\*denotes p < 0.01, \*\*\*denotes p < 0.001. CK (unfertilized control), N (chemical fertilizers alone), CP (compost and chemical fertilizers), YD (Yak dung and chemical fertilizers), QS (*Qingke* straw and chemical fertilizers), and SD (Tibetan sheep dung and chemical fertilizers).

cumulative emissions compared with N treatment. The cumulative  $N_2O$  emissions in CP treatment were the lowest, which reduced by 58.00-147.17% compared to N, YD, QS and SD. The reason for this difference may be because minerlizable N containing substances had already been mineralized and the contents of available nitrogen were lower in CP compared other fertilization treatments during the composting process (Fig. 2a,b) and the compost components are more stable in structure and functionality. This would increase soil nitrogen immobilization and reduce  $N_2O$  production by decreasing the substrate for heterotrophic denitrification or ammonia oxidation  $N_2O$  emissions than other OFs (Fig.  $N_2O$ ). This phenomenon is in agreement with the observation of Shah et al., (2016).

The NH<sub>4</sub>+-N content, in consistent with N<sub>2</sub>O cumulative emissions<sup>52</sup>, was higher in the N treatment than in CP, QS, and SD during the whole incubation period. The possible reason could be that urea produced more  $NH_4^+$ -N in the soil by promoting the mineralization of organic N.  $NH_4^+$  stimulated the activities of nitrifiers and providing favorable conditions for nitrification in soils<sup>53–55</sup>. Alternatively, the mineralization of organic nitrogen is greater than the immobilization of mineral nitrogen in the N treatment 56,57. In other words, urea hydrolysis to NH<sub>4</sub><sup>+</sup>-N and nitrified NO<sub>3</sub><sup>-</sup>-N in the N treatment<sup>58-60</sup>, providing abundant nitrogen substrates for denitrification, which may contribute to N<sub>2</sub>O production (Fig. S3). Notably, NO<sub>3</sub><sup>-</sup>-N accumulated during the incubation process, with relatively high and stable levels in the later stages of incubation, especially after days 40 when N<sub>2</sub>O emissions kept very low (Fig. 2a). This may be attributed to the characteristics of this studied sandy loam soils in favor of nitrate accumulation <sup>61,62</sup>. Also, NH<sub>4</sub> +-N released from the decomposition of organic fertilizers can be converted to NO<sub>3</sub><sup>-</sup>-N (Fig. 6a). In addition, although partial organic substitution treatments reduced pH compared to CK, they increased pH compared to N treatment (Fig. 2c). These pH variations may result from the inherent properties of organic fertilizers, urea hydrolysis, and nitrogen mineralization processes that influence NH<sub>4</sub><sup>+</sup>-N concentrations. Notably, pH changes can regulate denitrification efficiency, thereby affecting N<sub>2</sub>O emissions<sup>56,63-65</sup>. Under favorable conditions, these processes may enhance microbial reduction of  $N_2O$  to  $N_2$ , contributing to the observed changes in  $N_2O$  emissions 15,66. Therefore, N and OF treatments potentially influence N2O emissions by altering soil properties and the interaction between nitrification and denitrification. Partial organic substitution, particularly the CP treatment, significantly reduced cumulative N,O emissions by enhancing nitrogen fixation and stabilizing compost composition.

#### Partial organic substitution regulated the nitrification and denitrification functional genes

Fertilizer application can alter the potential for soil N<sub>2</sub>O emission by affecting the community structure and composition of nitrification and denitrification functional genes<sup>34,67</sup>. AOA *amoA* and AOB *amoA* can convert NH<sub>4</sub><sup>+</sup>-N to NH<sub>2</sub>OH and are key functional genes in nitrification (Fig. S3). However, these two groups of ammonia-oxidizing bacteria responded differently to the application of OFs. In this study, compared to N treatment, partial organic substitution treatments generally increased the copy number of AOA *amoA* while decreased the copy number of AOB *amoA* (Fig. 3a,b). This may be attributed to the presence of different ammonia oxidizing

bacteria host strains, with AOA *amoA* typically dominating in environment with low ammonium availability or high organic matter content, while AOB *amoA* thriving in conditions with higher ammonium concentrations and greater oxygen availability, reflecting their distinct ecological niches<sup>68–70</sup>. In addition, the copy number of AOA *amoA* was greater than that of AOB *amoA*, consistent with the copy numbers reported by other studies in most soils<sup>18,71–73</sup>. This suggested that AOA *amoA* could be the main driver of nitrification in the alluvial soil with sandy loam texture in this study.

Denitrifying bacteria (nirK, nirS, and nosZ) communities changed significantly under different fertilizer treatments 74,75. Specifically, both *nirK* and *nirS* genes were significantly lower in the partial organic substitution treatments than in N treatment (Fig. 3c, d), and the nirS copy number was more strongly correlated with N,O emissions than nirK (Fig. 5a,b). This may indicate that nirS functional genes are more sensitive to N<sub>2</sub>O emissions than nirK. Similarly, compared with nirK, nirS functional genes were clearly clustered in OFs (Fig. S1), indicating that nirS were more sensitive to fertilizer treatment response than nirK. More notably, partial organic substitution treatments except YD effectively increased the nosZ copy number, especially in CP treatment (Fig. 3e), reduced  $N_2O$  cumulative emissions (Fig. 1b). A possible explanation is that more  $N_2O$  was reduced to  $N_2$  during denitrification in CP treatment The alpha diversity indices (Chao1 and Shannon) of nosZ in CP treatment were significantly higher than those in other treatments, supporting this point (Table 2). By contrast, significantly lower nosZ in YD may explain its relative higher N<sub>2</sub>O emissions. Similarly, species richness (Chao1 index) of nirK and nirS was higher in YD treatment than other treatments, while nosZ was significantly lower (Table 2). It suggests that the Yak dung may have increased nitrite reduction and decreased N<sub>2</sub>O reduction capacity. This further explains the high cumulative N<sub>2</sub>O emissions from the YD treatment. In addition, previous studies have shown that genera of *Pseudomonas* and *Bradyrhizobium* influence denitrification processes 78,79. In this study, the higher presence of Pseudomonas and Bradyrhizobium in the nirK, nirS, and nosZ genes, and significantly higher of Pseudomonas in nirK and nosZ in CP treatment than in other organic substitution treatments, indicated that these genera may be the key bacteria to inhibit large amounts of N<sub>2</sub>O emissions from soils.

# Linkage among soil properties, functional genes and N<sub>2</sub>O emissions

Nitrification and denitrification, as biological processes, are influenced by both soil physicochemical and biological factors  $^{7,67,80}$ . In this study, the difference in nitrogen mineralization among partial organic substitution treatments with same nitrogen application rate was attributed to chemical composition (Table 1). Addition of OFs to soil causes changes in soil properties, profoundly affecting soil nitrification, denitrification, and  $N_2O$  emissions  $^{35,81,82}$ . Our results also verified that partial organic substitution has differentiated impacts on soil properties and functional gene abundance (Figs. 3 and 4).

The *nirS* and *nirK* denitrification functional genes can reduce  $NO_2^-$  to  $N_2O$ , while the *nosZ* type denitrifiers mediate the reduction of  $N_2O$  to  $N_2^-$  7,35,83. Although the *nirK* gene copy number was greater than *nirS* (Fig. 3), *nirS* was the most important factor to explain accumulative  $N_2O$  emissions (Fig. 5b), consistent with previously reported results  $^{15,33,84}$ . However, there are some studies showing that the increase in *nirK* gene abundance is greater than *nirS* with organic fertilizer application  $^{85}$ . The reason for the difference between the two results may be due to the difference in the concentration of available substrates in OFs-amended soils  $^{35,86}$ . The contribution of *nirK* to nitrite reduction and *nosZ*-mediated reduction of  $N_2O$  to  $N_2$  cannot be ignored. In this study, *Azospirillum*, involved in assimilation and non-differentiation of  $NO_3^-$  reduction, was found in *nosZ*<sup>87</sup>, while *Mesohizobium* is widely present in *nirK* and *nosZ*, especially in CP treatment (Fig. 4). The nitrogen generated by partial organic substitution can be immobilized by *Mesorhizobium* and *Azospirillum* $^{10,33,86}$ , thereby reducing  $N_2O$  emissions. This is supported by the fact in this study that the CP treatment had the lowest cumulative  $N_2O$  missions (Fig. 1b).

Soil  $NH_4^{+-}N$  and  $NO_2^{--}N$  concentrations are key substrates for nitrification and denitrification  $^{7,10}$ . Fertilization reduced  $NH_4^{+-}N$  and increased  $NO_3^{--}N$  contents (Fig. 2a,b), with  $NH_4^{+-}N$  exerting a stronger influence on cumulative  $N_2O$  emissions than  $NO_3^{--}N$  (Fig. 5), consistent with Huang et al. (2020). This may be due to the dual role of  $NH_4^{+-}N$  in  $N_2O$  production through nitrification (conversion of  $NH_4^{+-}$  to  $NO_2^{--}$ ) and denitrification (conversion of  $NO_2^{--}$  to  $N_2O$ ). While  $NO_3^{--}N$  acts only through denitrification  $NO_2^{--}N$  is Furthermore,  $NH_4^{+-}N$  affects soil pH and MBN content, which in turn regulates denitrifier community composition and  $N_2O$  emission potential (Fig. 6).

Soil pH has been recognized as a key soil variable mediating  $N_2O$  emissions  $^{64,82,88,89}$ . Our results of this study showed that the CP treatment had a neutral soil pH and low cumulative  $N_2O$  emissions (Figs. 1 and 2). which might be attributed to the fact that the *nosZ* gene was more predominant in soil environments with a neutral pH, facilitating  $N_2O$  conversion to  $N_2^{65}$ . SOC and MBN was also important factors affecting  $N_2O$  in our study (Figs. 5 and 6). This could potentially be attributed to the fact that they can impact  $N_2O$  emissions via multiple pathways  $^{90,91}$ , including serving as a nitrogen substrate affecting nitrification and denitrification processes  $^{92,93}$ , as well as influencing microbial metabolic activities and community structure associated with  $N_2O$  production  $^{5,94}$ .

In our study, denitrification appears to be the primary pathway for N<sub>2</sub>O emission. This can be attributed to two aspects. First, nitrification-related functional genes contributed less to N<sub>2</sub>O emission (Figs. 5b and 6b). Although AOA *amoA* is likely to be the main driver of nitrification, previous studies have shown that the nitrification-denitrification process is mainly mediated by AOB *amoA* <sup>95,96</sup>, which implies that AOA *amoA* may not contribute significantly to N<sub>2</sub>O production through nitrification. Instead, in this study, denitrification-related functional genes played a key role in regulating cumulative N<sub>2</sub>O emissions (Fig. 6b), especially *nirS*, which was identified as the most important predictor (Fig. 5b). Second, sufficient organic carbon can stimulate denitrification by providing electron donors for NO<sub>3</sub><sup>-</sup> reduction, thereby promoting N<sub>2</sub>O emission, especially under oxygen-limited conditions <sup>97,98</sup>. In addition, other pathways may contribute to N<sub>2</sub>O production. For example, abiotic decomposition of nitrite may also contribute <sup>99</sup>. Therefore, there are many uncertainties in exploring the relationship between N<sub>2</sub>O emissions, soil properties, and nitrification-denitrification genes, especially under

different fertilization treatments. Future studies are needed to quantitatively assess the contribution of different pathways to potential  $N_2O$  emissions.

#### Implications for organic fertilizer application

This study is of great significance for understanding and management of  $N_2O$  emissions from different organic fertilizer substitutions for chemical fertilizers in agricultural systems, especially in the alluvial soils in the Lhasa River valleys. We found that partial organic substitution for chemical fertilizer, especially the compost substitution had impressive potential to mitigate  $N_2O$  emissions. In contrast, direct yak manure substitution would cause increase in  $N_2O$  emissions. This result implies that proper composting of organic fertilizers, rather than direct application of untreated yak dung, may be a more environmentally friendly pathway for green agriculture. Partial organic substitution, especially after compost, enables  $N_2O$  emission reduction while soil fertility maintenance. This study may corroborate the rationale of traditional manure use habit of Tibetan people, collecting yak dung for domestic fuel and mostly use sheep manure for fertilizer in croplands. In addition, we found that  $NO_3^--N$  accumulated in this incubation experiment even when  $N_2O$  emissions were very low, suggesting partial organic substitution also has risk of leading to nitrate leaching, nutrient loss and waterbody contamination in the Lhasa River Valleys. We recommend co-composting animal manure, crop residue for partial organic substitution of chemical fertilizer and using nitrification inhibitors to achieve the goals of both mitigating  $N_2O$  emissions and maintaining soil fertility.

Given that denitrification plays a crucial role in determining whether accumulated nitrate is converted to  $N_2O$  or further reduced to  $N_2$ , our study focused on denitrification functional genes (*nirK*, *nirS*, *nosZ*). However, future studies should include AOA *amoA* and AOB *amoA* communities for a comprehensive understanding of the interactions between nitrification, denitrification, and  $N_2O$  emissions under different organic replacement strategies. Moreover, freeze-thaw effects are recognized as major contributors to  $N_2O$  emissions in high-altitude ecosystems due to their potential to physically disrupt soil microbial communities and alter denitrification processes  $^{100-103}$ . Although no significant freeze-thaw events occurred in this farmland study area, it is undeniable that such processes can influence nitrogen cycling in high-elevation ecosystems. Therefore, future research should integrate field experiments to explore additional environmental factors, such as freeze-thaw effects and nitrate leaching, to comprehensively evaluate their impact on nitrogen cycling and greenhouse gas emissions in ecosystems.

#### **Conclusions**

This short-term microcosm incubation provided the evidence that application of organic and chemical fertilizer blends generally reduced  $\rm N_2O$  fluxes under equal nitrogen addition compared to chemical fertilizer alone. The relatively higher  $\rm N_2O$  emissions from yak dung substitution can be evidently reduced through co-composting animal manure and crop residue. Fertilization treatments may modulate potential  $\rm N_2O$  emissions primarily by influencing denitrification processes. Denitrification functional genes, especially the *nirS* abundance,  $\rm NH_4^{+}-N$  content and pH were among the predominant factors to determining cumulative  $\rm N_2O$  emissions. However,  $\rm NO_3^{-}-N$  was prone to accumulation when applying partial organic substitution, which could render increasing risk of nitrogen leaching and nutrient loss in the alluvial soils. Taken together, soil  $\rm N_2O$  emission reduction is favored by applying partial organic fertilizer substitution especially after through co-composting.

#### Data availability

The data that support the findings of this study are available from the corresponding author upon reasonable request. The raw high-throughput sequencing data have been deposited in the National Center for Biotechnology Information (NCBI) under accession number PRJNA1242584.

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#### **Declarations**

# Competing interests

The authors declare no competing interests.

#### Additional information

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