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Metabolism of Paeoniae Radix Rubra and its 14 constituents in mice

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Objective: Paeoniae Radix Rubra (PRR) is a commonly used traditional Chinese medicine with the effects of clearing away heat, cooling the blood, and relieving blood stasis. To 1) elucidate the metabolites and metabolic pathways of PRR and its 14 main constituents in mice and 2) reveal the possible origins of the known effective forms of PRR and their isomers, the metabolism of PRR in mice was systematically studied for the first time.

Methods: PRR and its 14 constituents were administered to mice by gavage once a day for seven consecutive days, respectively. All urine and feces were collected during the 7 days of dosing, and blood was collected at 1 h after the last dose. Metabolites were detected and identified using high performance liquid chromatography with diode array detector and combined with electrospray ionization ion trap time-of-flight multistage mass spectrometry (HPLC-DAD-ESI-IT-TOF-MSⁿ).

Results: In total, 23, 16, 24, 17, 18, 30, 27, 17, 22, 17, 33, 3, 8, 24, and 31 metabolites of paeoniflorin, albiflorin, oxypaeoniflorin, benzoylpaeoniflorin, hydroxybenzoylpaeoniflorin, benzoyloxypaeoniflorin, galloylpaeoniflorin, lactiflorin, epicatechin gallate, catechin gallate, catechin, ellagic acid, 3,3'-di-O-methylellagic acid, methylgallate, and PRR were respectively identified in mice; after eliminating identical metabolites, a total of 195 metabolites remained, including 8, 11, 25, 17, 18, 30, 27, 17, 21, 17, 1, 2, 8, 20, and 20 newly identified metabolites, respectively. The metabolic reactions of PRR and its 14 main constituents in mice were primarily methylation, hydrogenation, hydrolysis, hydroxylation, glucuronidation, and sulfation.

Abbreviations: A, albiflorin; B, benzoylpaeoniflorin; BO, benzoyloxypaeoniflorin; BPCs, base peak chromatograms; C, catechin; CDL, curved desolvation line; CG, catechin gallate; CID, relative collision-induced dissociation; d, day; DEA, 3,3'-di-O-methylellagic acid; EA, ellagic acid; ECG, epicatechin gallate; EICs, extracted ion chromatograms; Err., error.; ESI, electrospray ionization; G, galloylpaeoniflorin; HPLC, high-performance liquid chromatography; HPLC-ESI-IT-TOF-MSⁿ, high performance liquid chromatography with diode array detector and combined with electrospray ionization ion trap time-of-flight multistage mass spectrometry; L, lactiflorin; Meas., measured; MG, methyl gallate; NLRP3, NOD-like receptor thermal protein domain associated protein 3; O, oxypaeoniflorin; OB, hydroxybenzoylpaeoniflorin; P, paeoniflorin; PRR. Paeoniae Radix Rubra; t_R, retention time.

Conclusion: We elucidated the metabolites and metabolic pathways of PRR and its 14 constituents (e.g., paeoniflorin, catechin, ellagic acid, and gallic acid) in mice and revealed the possible origins of the 10 known effective forms of PRR and their isomers. The findings are of great significance to studying the mechanism of action and quality control of PRR.

KEYWORDS

Paeoniae Radix Rubra, paeoniflorin, catechin, ellagic acid, methylgallate, *in vivo* metabolism

1 Introduction

The effective forms of traditional Chinese medicines can be the original constituents or the active metabolites produced *in vivo* (Xu et al., 2022). The metabolism of traditional Chinese medicines is the key link between their phytochemistry *in vitro* and their pharmacological activity *in vivo*. Therefore, studying the metabolism of traditional Chinese medicines is crucial to understanding the forms of the medicine that exist and are active *in vivo* along with the mechanisms of action of traditional Chinese medicines.

Paeoniae Radix Rubra (PRR), obtained from the dried roots of *Paeonia lactiflora* Pall. or *Paeonia veitchii* Lynch, is a commonly used traditional Chinese medicine with the effects of clearing away heat, cooling the blood, and relieving blood stasis (Zhao et al., 2021). PRR has many pharmacological effects, such as preventing liver fibrosis, curing jaundice, improving cholestasis in rats, relieving inflammation, and improving myocardial infarction, hypertrophy, and fibrosis (Tan et al., 2020).

The constituents of PRR have various structures, with the primary ones being monoterpene glycosides, tannins, flavonoids, and triterpenes (Yan et al., 2018). Monoterpene glycosides include paeoniflorin, albiflorin, oxypaeoniflorin, benzoylpaeoniflorin, hydroxybenzoylpaeoniflorin, benzoyloxypaeoniflorin, galloylpaeoniflorin, and lactiflorin. Tannins include ellagic acid and 3,3'-di-O-methylellagic acid. Catechins include catechin, catechin gallate, and epicatechin gallate. Gallic acids include methylgallate.

Among the 14 constituents mentioned above, 11 (all but hydroxybenzoylpaeoniflorin, benzoyloxypaeoniflorin, lactiflorin) exhibit various biological activities. For example, paeoniflorin (the main active constituent of PRR) shows good anti-inflammatory, immunomodulatory, and anti-tumor effects (Zhao et al., 2021). Albiflorin has the function of soothing the liver and relieving depression (Zhao et al., 2018). Oxypaeoniflorin can prevent acute lung injury caused by lipopolysaccharides in mice (Fan et al., 2021). Benzoylpaeoniflorin exhibits anti-allergic activity, making it a potential candidate drug for the treatment of allergic diseases (Zhong et al., 2021). Galloylpaeoniflorin can relieve osteoporosis following oophorectomy (Liu et al., 2021b). Ellagic acid reduces the toxicity of diclofenac in rat hepatocytes by enhancing the activity of antioxidant enzymes such as catalase (Hatefi-Hesari et al., 2021). 3,3'-Di-O-methylellagic acid significantly reduces retinal vasodilation caused by high glucose levels in juvenile zebrafish (Lee et al., 2018). Methylgallate improves potassium oxazinate-induced kidney damage in mice with hyperuricemia nephropathy by inhibiting the NOD-like receptor thermal protein domain associated protein 3 (NLRP3) pathway, thereby producing a renal protective effect (Liu et al., 2021a). Catechin protects rat cardiomyocytes against hypoxic damage (Fang et al., 2018) and also exerts an anti-inflammatory effect (Syed Hussein et al., 2015). Both catechin gallate and epicatechin gallate have significant anti-inflammatory and anti-proliferative activities (Kurbitz et al., 2011). Catechin gallate inhibits catechol methylation in rat hepatocyte cytoplasm and hepatocyte cultures by inhibiting the activity of catechol oxymethyltransferase (Kadowaki et al., 2005).

In a previous study, we found that only four of the 21 forms of PRR that can effectively treat toxic heat and blood stasis were the original constituents; the other 17 were metabolites (Xu et al., 2022). The four original constituents were paeoniflorin (C1), oxypaeoniflorin (C2), desbenzoylpaeoniflorin isomer (C3), and 3,7/8-dimethylellagic acid (C4), and the 17 metabolites were 3'-O-methyl (epi) catechin 5-O-glucuronide (C5), 3-hydroxy phenylpropionic acid sulfate (C6), 3-hydroxy-4-methoxyphenylpropionic acid sulfate (C7), 3/4-hydroxy benzoic acid sulfate (C8), C10H18O2 glucuronide (C9-C15), C10H18O4 glucuronide (C16), C10H14O3 glucuronide (C17), 3-methoxy-4-hydroxy-phenylpropionic acid sulfate (C18), C₈H₈O₃ glucuronide (C19, C20), and benzoyl glucuronide (C21). Of them, C9-C17, C19, C20, and M23 are derived from paeoniflorin; C5-C7, C18, and C21 are possibly derived from catechins; and C8 is possibly derived from catechins or gallic acids (Liang et al., 2013).

In this study, we systematically explored the metabolism of PRR and its 14 constituents with four structure types (paeoniflorins, catechins, gallic acids, and ellagic acids) in mice. The objectives of the study were to 1) elucidate the metabolites and metabolic pathways of PRR and its 14 main constituents in mice, 2) determine the possible origins of the metabolites of PRR, and 3) clarify whether these compounds can be converted into the recognized effective forms *in vivo*. The findings are helpful for further elucidating the forms of PRR that exist *in vivo* and identifying the effective forms of PRR. The

findings are also of great significance for studying the mechanisms of action and quality control of PRR.

2 Materials and methods

2.1 Medicinal materials and reagents

PRR was purchased from Baohua, a dealer of Chinese herbal medicines in Xunke County, Heilongjiang Province (Sample No. 6524). The medicine was identified as the dried roots of *Paeonia lactiflora* Pall. by Professor Shao-Qing Cai from Peking University School of Pharmaceutical Sciences. A voucher sample (No. 6524) was stored in the Herbarium of Peking University School of Pharmaceutical Sciences.

Paeoniflorin (Lot No. PS010957), albiflorin (Lot No. PS011455), No. oxypaeoniflorin (Lot PS010199), benzoylpaeoniflorin (Lot No. PS000157), hydroxyben zoylpaeoniflorin (Lot No. PS000693), benzoyloxypaeoniflorin (Lot No. PS012411), galloylpaeoniflorin (Lot No. PS010194), epicatechin gallate (Lot No. PS000163), catechin gallate (Lot No. PS010638) and 3,3'-di-O-methylellagic acid (Lot No. PS 22960025) were purchased from Chengdu Push Biotechnology Co., Ltd. (Chengdu, China). Catechin (Lot No. P21J11F 118,380) and methylgallate (Lot No. L19D5Y1) were obtained from Shanghai Yuanye Biotechnology Co., Ltd. (Shanghai, China). Lactiflorin (Lot No. MUST-21041202) and ellagic acid (Lot No. KN 960133) were purchased from Chengdu Must Biotechnology Co., Ltd. (Chengdu, China). The purity of all reference substances was greater than 95.0%, as indicated by high-performance liquid chromatography (HPLC; 254 nm). Formic acid (Lot No. 212271), acetonitrile (Lot No. F21LAL 202), and methanol (Lot No. 206409), all HPLC grade, were purchased from Thermo Fisher Scientific, Inc. (Waltham, MA, United States). HPLC-grade ethanol (Lot No. 32061) was purchased from Fuchen (Tianjin) Chemical Reagents Co., Ltd. (Tianjin, China). Sodium carboxymethyl cellulose (Lot No. A18105) was purchased from Sinopharm Chemical Reagents Co., Ltd. (Shanghai, China). Ultrapure water was prepared using a Milli-Q Integral 3 ultrapure water machine (Millipore, Billerica, MA, United States).

2.2 Preparation and qualitative analysis of the lyophilized powder of PRR decoction

The lyophilized powder of PRR decoction was prepared as described previously (Liang et al., 2013). Each Gram of the lyophilized powder was equivalent to 2.63 g of the crude drug. The constituents in the lyophilized powder of the PRR decoction were identified by high performance liquid chromatography with diode array detector and combined with electrospray ionization ion trap time-of-flight multistage mass spectrometry (HPLC-DAD-ESI-IT-TOF-MSⁿ) with reference

to the mass spectral data of the reference substances or the relevant literature (Supplementary Figure S1).

2.3 Animal and metabolic studies

Forty-eight ICR mice (male, 30 ± 2 g) were purchased from the Department of Laboratory Animal Sciences at the Peking University Health Science Center and randomized into one control group and 15 dosing groups (14 compound-dosed groups and one PRR decoction-dosed group) with three mice in each group. All mice were housed in mouse metabolism cages. The experiment lasted for 10 days. After adapting for the first 3 days, the mice were dosed by gavage once per day for the following 7 days. In the 14 compound-dosed groups, the dose was 40 mg/kg mouse weight, while the dose in the decoctiondosed group was 70 mg of lyophilized powder of PRR decoction per kg mouse weight (equivalent to 200 mg/kg of PRR crude drug). All compounds and the lyophilized powder of PRR decoction were suspended in 0.5% sodium carboxymethyl cellulose solution, and the dose volume was approximately 0.2 ml for each mouse. The control group was given the same volume of 0.5% sodium carboxymethyl cellulose solution. The mice were maintained in an environment at 22 ± 2°C (relative humidity 50 \pm 5%) and allowed to eat and drink *ad libitum*. All animal experiments were approved by the Animal Ethics Committee of Peking University Health Science Center (approval number: LA2019117).

2.4 Collection and preparation of samples

2.4.1 Collection of samples

All urine and feces were collected during the 7 days of dosing. One hour after the last dosing by gavage, the blood was collected into 1.5-ml heparin sodium-containing tubes by excising the eyeballs. All samples were stored in a refrigerator at -20° C prior to use.

2.4.2 Preparation of urine, feces, and plasma samples

All urine, feces, and plasma samples from the same group were pooled, resulting in 16 urine, 16 feces, and 16 plasma samples. Each urine sample was centrifuged at 8,000 rpm at 4°C for 15 min. The supernatant was harvested, concentrated, and dried at 55°C followed by the addition of 10 ml of methanol and ultrasonic extraction for 30 min. The extract was filtered, concentrated, and dried at 55°C. Each feces sample was dried at 50°C for 48 h and smashed followed by the addition of 30 ml of methanol and ultrasonic extraction for 30 min per round (three rounds of extraction). The extract was filtered, and the filtrates obtained from the three rounds of extraction were pooled, concentrated, and dried at 55°C

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followed by the addition of 10 ml of methanol, ultrasonic extraction for 30 min, and centrifugation at 8,000 rpm and 4°C for 15 min. The supernatant was concentrated and dried at 55°C. The urine and feces samples (0.5 g each) were separately dissolved in 1 ml methanol. Each plasma sample was centrifuged at 5,000 rpm and 4°C for 15 min, and approximately 0.9 ml supernatant was collected for each group. After adding 4.5 ml methanol, the mixture was centrifuged at 5,000 rpm and 4°C for 15 min. The supernatant was blown with nitrogen and dried at 40°C followed by the addition of 0.5 ml methanol for reconstitution. All samples were filtered through a 0.22- μ m membrane and stored at -20°C prior to further analysis.

2.5 Instruments and Conditions

HPLC-DAD-ESI-IT-TOF-MSⁿ analysis was performed using an HPLC instrument and an IT-TOF mass spectrometer connected to two LC-20AD pumps, an SIL-20AC autosampler, a CTO-20A column heater, an SPD-M20A photo-diode array (PDA) detector, and a CBM-20A system controller (Shimadzu, Kyoto, Japan). Data analysis was performed using LCMS Solution v.3.60, Formula Predictor v.1.2, and Accurate Mass Calculator (Shimadzu, Kyoto, Japan).

The chromatographic conditions were as follows: chromatographic column, Phenomenex Gemini C18 (250 mm × 4.6 mm, 5 µm); guard column, Phenomenex Security Guard $(4 \text{ mm} \times 3.0 \text{ mm}, 5.0 \mu\text{m}; \text{Phenomenex}, \text{Torrance}, CA,$ United States); column temperature, 40°C; injection volume, 10 $\mu L;$ and flow rate, 1 ml/min. The mobile phase was 0.1% aqueous solution of formic acid (A) and acetonitrile (B). The mobile phase gradients were as follows: 0-12 min, 3% B; 12-33 min, 3%-8% B; 33-37.5 min, 8% B; 37.5-52.5 min, 8%-12% B; 52.5-82.5 min, 12%-25% B; 82.5-105 min, 25 %-60% B; 105-120 min, 60%-100% B; 120-130 min, 100% B. PDA detector: 200-700 nm. The mass spectrometry conditions were as follows: electrospray ionization (ESI), positive and negative ion mode; mass scan range, m/z 100-1,000 (MS), m/z 50-1,000 (MS² and MS³); relative collision-induced dissociation (CID) energy, 50%; heater block temperature, 200°C; curved desolvation line (CDL) temperature, 200°C; detection voltage, 1.70 kV; interface voltage (+) 4.5 kV, (-) 3.5 kV; drying gas, nitrogen; and drying gas flow rate, 1.5 L/min.

2.6 Identification of the forms of PRR present *in vivo* (original constituents and metabolites)

The *in vivo* existence forms of PRR were identified as previously described (Liang et al., 2013). First, the base peak chromatograms (BPCs) of the samples from the dosing and

control groups were compared to find the distinguishing peaks and tentatively determine the *in vivo* existence forms of PRR. Second, the extracted ion chromatograms (EICs) of the compounds in the dosing and control groups were compared to confirm the distinguishing peaks. The chromatographic peaks that appeared in the dosing groups but not in the control group were considered to represent the *in vivo* existence forms of PRR. Finally, the forms of PRR existing *in vivo* were analyzed structurally based on the obtained liquid chromatographyhigh resolution multi-stage mass spectrometry data combined with 1) the mass spectrometry data of reference substances, 2) the mass spectrometry fragmentation patterns, 3) the mass spectrometry fragmentation information reported in the literature, and 4) information obtained by searching the SciFinder database.

3 Results

In this study, 23, 16, 24, 17, 18, 30, 27, 17, 22, 17, 33, 3, 8, 24, and 31 metabolites of paeoniflorin (P), albiflorin (A), oxypaeoniflorin benzoylpaeoniflorin (O), (B), hydroxybenzoylpaeoniflorin (OB), benzoyloxypaeoniflorin (BO), galloylpaeoniflorin (G), lactiflorin (L), epicatechin gallate (ECG), catechin gallate (CG), catechin (C), ellagic 3,3'-di-O-methylellagic acid (EA), acid (DEA), methylgallate (MG), and PRR were respectively identified in mice. After identifying identical metabolites, a total of 195 metabolites remained (Table 1). The LC-MSⁿ data of the metabolites and the distributions of metabolites in urine, feces, and plasma of can be found in Supplementary Tables S1, S2.

In this study, the usual neutral losses in mass spectrometry were 30.01 Da (CH₂O), 14.01 Da (CH₂), 176.03 Da (C₆H₈O₆), 79.95 Da (SO₃), 81.97 Da (H₂SO₃), 43.99 Da (CO₂), 15.02 Da (CH₃•), 27.99 Da (CO), 42.01 Da (C₂H₂O), 46.01 Da (CH₂O₂), 104.01 Da (C₇H₄O), 122.04 Da (C₇H₆O₂), and 162.05 Da (C₆H₁₀O₅), indicating that a molecule which shows these neutral losses contains formaldehyde or methanyl, methyl, glucuronosyl, sulfonyl, sulfonyl, carboxy or lactone, methyl, carbonyl, acetyl, carboxy or lactone group, benzoyl, benzoyloxy, and hexosyl (more likely glucosyl) groups, respectively. In addition, the fragment ions at *m*/*z* 175.02 (C₆H₇O₆) and *m*/*z* 96.96 (SO₄H) indicate that the molecule contains glucuronosyl and sulfate groups.

3.1 Mass spectral features of the 14 reference substances

The mass spectral fragments and fragmentation pathways of the 14 reference substances are presented in Supplementary Table S3 and Supplementary Figure S2.

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No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	A	0	В	OB	BO	G	L	ECG	CG	С	EA	DEA	MG	PRR	Identification
P0 ^a	56.58	$C_{23}H_{28}O_{11}$	525.1590	-4.57	+	_	_	_	_	_	_	_	_	_	_	_	_	_	_	paeoniflorin
$A0^{a}$	51.95	$C_{23}H_{28}O_{11}$	525.1597	-3.24	-	+	-	-	-	-	-	-	-	-	-	-	-	-	-	albiflorin
$O0^{a}$	42.99	$C_{23}H_{28}O_{12}$	495.1501	-1.41	-	-	+	-	-	-	-	-	-	-	-	-	-	-	-	oxypaeoniflorin
OB0 ^a	84.65	$C_{30}H_{32}O_{13}$	599.1726	-7.34	-	-	-	-	+	-	-	-	-	-	-	_	-	-	-	hydroxybenzoylpaeoniflorin
$\mathrm{BO0}^{\mathrm{a}}$	87.12	$C_{30}H_{32}O_{13}$	599.1755	-2.5	-	-	-	-	-	+	-	-	-	-	-	-	-	-	-	benzoyloxypaeoniflorin
$G0^{a}$	70.17	$C_{30}H_{32}O_{15}$	631.1633	-5.55	-	-	-	-	-	-	+	-	-	-	-	-	-	-	-	galloylpaeoniflorin
$L0^{a}$	81.32	$C_{23}H_{26}O_{10}$	507.1508	0.00	-	-	-	-	-	-	-	+	-	-	-	_	-	-	-	lactiflorin
ECG0 ^a	67.80	$C_{22}H_{18}O_{10}$	441.0827	0.00	-	-	-	-	-	-	-	-	+	-	-	-	-	-	-	epicatechin gallate
$CG0^{a}$	70.36	$C_{22}H_{18}O_{10}$	441.0828	0.23	-	-	-	-	-	-	-	-	-	+	-	_	-	-	-	catechin gallate
DEA0 ^a	90.97	$C_{16}H_{10}O_8$	329.0315	3.65	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	3,3'-di-O-methylellagic acid
$M1^{a}$	56.58	$C_{23}H_{28}O_{11}$	525.1590	-4.57	-	-	+	+	+	-	+	-	-	-	-	-	-	-	-	paeoniflorin
M2 ^a	42.53	$C_{23}H_{28}O_{12}$	495.1502	-1.21	+	-	-	-	+	+	-	-	-	-	-	-	-	-	-	oxypaeoniflorin
M3	77.90	$C_{10}H_{16}O_4$	199.0987	5.52	-	-	-	-	-	+	+	-	-	-	-	-	-	-	-	paeonimetabolin II isomer 1
M4	78.72	$C_{10}H_{16}O_4$	199.0978	1.00	-	-	-	-	-	-	-	-	-	-	-	_	-	-	+	paeonimetabolin II isomer 2
M5	80.91	$C_{10}H_{16}O_4$	199.0980	2.01	-	-	-	-	-	-	-	-	-	-	-	-	-	-	+	paeonimetabolin II isomer 3
M6	80.24	$C_{10}H_{16}O_4$	199.0970	-3.01	-	-	-	-	-	+	+	-	-	-	-	_	-	-	-	paeonimetabolin II isomer 4
M7	23.69	$\mathrm{C_{10}H_{16}O_7S}$	279.0532	-4.30	-	+	-	-	-	-	-	-	-	-	-	-	-	-	-	paeonimetabolin II sulfate isomer 1
M8	31.89	$\mathrm{C_{10}H_{16}O_7S}$	279.0525	-6.81	-	-	-	-	+	-	-	-	-	-	-	-	-	-	-	paeonimetabolin II sulfate isomer 2
M9	43.84	$\mathrm{C_{10}H_{16}O_7S}$	279.0524	-7.17	-	+	-	-	-	-	-	-	-	-	-	-	-	-	-	paeonimetabolin II sulfate isomer 3
M10	49.22	$\mathrm{C_{10}H_{16}O_7S}$	279.0523	-7.53	-	+	-	-	-	-	+	-	-	-	-	-	-	-	-	paeonimetabolin II sulfate isomer 4
M11	8.77	$C_{16}H_{24}O_{10}$	421.1361	2.14	-	+	-	-	-	-	-	-	-	-	-	_	-	-	-	desbenzoyl albiflorin isomer 1
M12	31.28	$C_{16}H_{24}O_{10}$	375.1266	-8.26	-	+	-	-	-	-	-	-	-	-	-	_	-	-	-	desbenzoyl albiflorin isomer 2
M13	10.08	$C_{16}H_{24}O_{10}$	421.1332	-4.75	-	+	-	-	-	-	-	-	-	-	-	-	-	-	-	desbenzoyl albiflorin isomer 3
M14	51.34	$C_{16}H_{24}O_{10}\\$	421.1333	-4.51	+	+	+	+	+	-	-	-	-	-	-	-	-	-	-	desbenzoylpaeoniflorin isomer 1
M15	52.97	$C_{16}H_{24}O_{10}$	421.1346	-1.42	+	+	+	+	+	+	+	+	-	-	-	-	-	-	-	desbenzoylpaeoniflorin isomer 2
M16	29.71	$C_{17}H_{26}O_{10}$	435.1485	-5.29	+	-	-	-	-	-	-	-	-	-	-	-	-	-	-	methyl desbenzoylpaeoniflorin isomer 1
M17	26.15	$C_{17}H_{26}O_{10}$	435.1477	-7.12	-	-	-	-	-	-	-	-	-	-	-	-	-	-	+	methyl desbenzoylpaeoniflorin isomer 2
M18	32.07	$C_{16}H_{22}O_{10}$	373.1146	1.61	+	-	+	+	+	+	+	-	-	-	-	-	-	-	-	paeonimetabolin I glucuronide isomer 1
M19	33.01	$C_{16}H_{22}O_{10}$	373.1145	1.34	-	-	+	+	+	+	+	-	-	-	-	-	-	-	-	paeonimetabolin I glucuronide isomer 2
M20	29.78	$C_{16}H_{22}O_{10}$	373.1123	-4.56	-	-	-	-	-	-	-	+	-	-	-	-	-	-	-	paeonimetabolin I glucuronide isomer 3
M21	28.88	$C_{16}H_{22}O_{10}$	373.1135	-1.34	-	-	-	-	-	-	-	+	-	-	-	-	-	-	-	paeonimetabolin I glucuronide isomer 4
M22	55.09	$C_{16}H_{22}O_{10}$	373.1133	-1.88	-	-	+	-	-	-	-	-	-	-	-	-	-	-	-	paeonimetabolin I glucuronide isomer 5
M23	24.88	$C_{10}H_{14}O_6S$	261.0437	-0.38	+	-	_	+	+	+	+	-	_	_	-	_	_	_	_	C ₁₀ H ₁₄ O ₃ sulfate isomer 1

TABLE 1 Retention times (t_R) molecular formulae, and identities of 10 absorbed compounds and 195 metabolites of PRR and its 14 constituents in mice based on HPLC-ESI-IT-TOF-MSⁿ.

(Continued on following page)

No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	A	0	В	OB	BO	G	L	ECG	CG	С	EA	DEA	MG	PRR	Identification
M24	26.01	C10H14O6S	261.0438	0.00	+	_	_	+	+	+	+	_	_	_	_	_	_	_	_	C ₁₀ H ₁₄ O ₃ sulfate isomer 2
M25	28.75	$C_{10}H_{14}O_6S$	261.0423	-5.75	-	-	+	-	-	-	-	-	-	-	-	-	-	-	-	C ₁₀ H ₁₄ O ₃ sulfate isomer 3
M26	29.29	$\mathrm{C_{10}H_{14}O_6S}$	261.0423	-5.75	+	-	+	-	+	-	-	-	-	-	-	-	-	-	-	C ₁₀ H ₁₄ O ₃ sulfate isomer 4
M27	34.56	$C_{10}H_{14}O_6S$	261.0429	-3.45	+	-	-	+	+	-	-	-	-	-	-	-	-	-	-	C ₁₀ H ₁₄ O ₃ sulfate isomer 5
M28	37.82	$C_{10}H_{14}O_6S$	261.0432	-2.3	-	-	+	-	-	-	-	-	-	-	-	-	-	-	-	C ₁₀ H ₁₄ O ₃ sulfate isomer 6
M29	24.33	$C_{16}H_{26}O_{10}$	377.1475	5.83	-	-	+	-	+	+	-	-	-	-	-	-	-	-	-	C ₁₀ H ₁₈ O ₄ glucuronide isomer 1
M30	24.90	$C_{16}H_{26}O_{10}$	377.1443	-2.65	-	-	-	-	-	+	+	-	-	-	-	-	-	-	-	C ₁₀ H ₁₈ O ₄ glucuronide isomer 2
M31	26.71	$C_{16}H_{26}O_{10}$	377.1463	2.65	+	-	+	+	+	+	+	-	-	-	-	-	-	-	-	C ₁₀ H ₁₈ O ₄ glucuronide isomer 3
M32	30.81	$C_{16}H_{26}O_{10}$	377.1453	0.00	-	-	-	-	-	+	+	-	-	-	-	-	-	-	-	C ₁₀ H ₁₈ O ₄ glucuronide isomer 4
M33	31.97	$C_{16}H_{26}O_{10}$	377.1437	-4.24	+	-	+	+	-	+	+	-	-	-	-	-	-	-	-	C ₁₀ H ₁₈ O ₄ glucuronide isomer 5
M34	53.31	$C_{16}H_{26}O_{10}$	377.1443	-2.65	-	-	-	-	-	-	-	+	-	-	_	-	-	-	+	C ₁₀ H ₁₈ O ₄ glucuronide isomer 6
M35	56.90	$C_{16}H_{28}O_{10}$	379.1614	1.05	-	-	-	-	-	+	+	-	-	-	_	-	-	-	-	C ₁₀ H ₂₀ O ₄ glucuronide
M36	23.65	$C_{14}H_{16}O_9$	327.0731	2.75	-	+	-	-	-	-	-	-	-	-	_	-	-	-	-	C ₈ H ₈ O ₃ glucuronide isomer 1
M37	31.11	$C_{14}H_{16}O_9$	327.0712	-3.06	-	+	-	-	-	+	+	-	-	-	-	-	-	-	-	C ₈ H ₈ O ₃ glucuronide isomer 2
M38	32.32	$C_{14}H_{16}O_9$	327.0722	0.00	-	+	-	-	-	-	-	-	-	-	-	-	-	-	-	C ₈ H ₈ O ₃ glucuronide isomer 3
M39	35.98	$C_{14}H_{16}O_9$	327.0707	-4.59	-	+	-	-	-	-	-	-	-	-	_	-	-	-	-	C ₈ H ₈ O ₃ glucuronide isomer 4
M40	41.65	$C_{14}H_{16}O_9$	327.0696	-7.95	-	-	-	-	-	+	+	-	-	-	-	-	-	-	-	C ₈ H ₈ O ₃ glucuronide isomer 5
M41	34.25	$C_{10}H_{18}O_6S$	265.0730	-7.92	-	-	+	-	-	-	-	-	-	-	-	-	-	-	-	2,6-dihydroxycineol sulfate isomer 1
M42	47.90	$C_{10}H_{18}O_6S$	265.0732	-7.17	+	-	+	-	+	-	-	-	-	-	-	-	-	-	-	2,6-dihydroxycineol sulfate isomer 2
M43	51.20	$C_{10}H_{18}O_6S$	265.0736	-5.66	+	-	+	+	+	-	-	-	-	-	-	-	-	-	-	2,6-dihydroxycineol sulfate isomer 3
M44	54.17	$C_{10}H_{18}O_6S$	265.0745	-2.26	+	-	+	+	-	-	-	-	-	-	_	-	-	-	-	2,6-dihydroxycineol sulfate isomer 4
M45	63.27	$C_{10}H_{18}O_6S$	265.0729	-8.30	+	+	+	+	+	+	+	-	-	-	_	-	-	-	-	2,6-dihydroxycineol sulfate
M46	83.48	$C_{10}H_{18}O_6S$	265.0739	-4.53	+	-	+	-	-	-	-	-	-	-	_	-	-	-	-	2,6-dihydroxycineol sulfate isomer 6
M47	85.97	$C_{10}H_{18}O_6S$	265.0733	-6.79	+	+	+	+	-	+	+	-	-	-	_	-	-	-	-	2,6-dihydroxycineol sulfate isomer 7
M48	111.38	$C_{10}H_{18}O_6S$	265.0740	-4.15	-	-	-	-	-	-	-	-	-	-	-	_	-	_	+	2,6-dihydroxycineol sulfate isomer 8
M49	128.78	$C_{10}H_{18}O_6S$	265.0753	0.75	-	-	-	-	-	-	-	-	-	-	-	_	-	_	+	2,6-dihydroxycineol sulfate isomer 9
M50	69.08	$C_{16}H_{26}O_9$	361.1522	4.98	-	-	-	-	-	-	-	-	-	-	-	_	-	_	+	paeonimetabolin II glucoside isomer 1
M51	71.20	$C_{16}H_{26}O_9$	361.1486	-4.98	-	-	-	-	-	-	-	-	-	-	-	_	-	_	+	paeonimetabolin II glucoside isomer 2
M52	70.28	$C_{16}H_{26}O_9$	361.1520	4.43	+	-	-	-	-	-	-	-	-	-	_	-	-	-	-	paeonimetabolin II glucoside isomer 3
M53	72.81	C16H26O9	361.1508	1.11	_	-	_	-	-	-	_	-	-	_	_	_	_	_	+	paeonimetabolin II glucoside isomer 4
M54	66.15	C ₁₆ H ₂₆ O ₉	361.1489	-4.15	_	-	_	-	-	+	+	-	_	_	_	_	_	_	_	paeonimetabolin II glucoside
M55	61.87	$C_{10}H_{20}O_6S$	267.0936	0.00	+	+	+	-	+	+	+	-	_	_	_	_	_	_	_	hydrogenated 2,6-dihydroxycineol sulfate isomer 1
M56	57.88	C10H20O6S	267.0901	-2.62	_	-	_	-	-	-	_	-	-	_	_	_	_	_	+	hydrogenated 2,6-dihydroxycineol sulfate isomer 2

(Continued on following page)

No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	A	0	В	OB	BO	G	L	ECG	CG	С	EA	DEA	MG	PRR	Identification
M57	38.30	C ₁₆ H ₂₄ O ₉	359.1318	-8.35	+	_	+	+	_	+	+	_	_	_	_	_	_	_	_	dehydrogenated 2,6-dihydroxycineol glucuronide isomer 1
M58	39.01	$C_{16}H_{24}O_9$	359.1343	-1.39	+	-	+	+	-	+	+	-	-	-	-	-	-	-	-	dehydrogenated 2,6-dihydroxycineol glucuronide isomer 2
M59	39.61	$C_{16}H_{24}O_9$	359.1342	-1.67	-	-	-	-	-	+	-	-	-	-	-	-	-	-	-	dehydrogenated 2,6-dihydroxycineol glucuronide isomer 3
M60	41.87	$C_{16}H_{24}O_9$	359.1322	-7.24	+	-	+	-	-	+	+	-	-	-	-	-	-	-	-	dehydrogenated 2,6-dihydroxycineol glucuronide isomer 4
M61	48.65	$C_{16}H_{24}O_9$	359.1333	-4.18	+	-	+	+	+	+	+	-	-	-	-	-	-	-	-	dehydrogenated 2,6-dihydroxycineol glucuronide
M62	49.63	$C_{16}H_{24}O_9$	359.1353	1.39	-	-	-	-	-	+	-	-	-	-	-	-	-	-	-	dehydrogenated 2,6-dihydroxycineol glucuronide isomer 5
M63	32.99	$C_7H_6O_8S$	248.9691	5.62	-	-	-	-	-	-	+	-	-	-	-	-	_	-	-	gallic acid sulfate
M64	42.42	$C_7H_6O_8S$	248.9719	3.21	-	-	-	-	-	-	+	-	-	-	-	-	-	-	-	gallic acid sulfate
M65	37.17	$C_9H_9NO_3$	178.0516	3.37	-	+	-	-	-	+	+	-	-	-	-	-	-	-	+	hippuric acid
M66	27.64	$C_9H_9NO_4$	194.0469	5.15	-	-	-	-	-	+	-	-	-	-	-	-	-	-	-	hydroxyhippuric acid
M67	29.07	$C_9H_9NO_4$	194.0459	0.00	-	-	-	-	-	+	-	-	-	-	-	-	_	-	-	hydroxyhippuric acid
M68	28.31	$C_{13}H_{14}O_9$	313.0547	-5.75	-	-	-	-	-	+	-	-	-	-	-	-	_	-	-	salicylic acid glucuronide
M69	71.03	$C_{23}H_{28}O_{10}$	509.1654	-2.16	-	-	-	-	-	-	-	+	-	-	-	-	_	-	-	hydrogenated lactiflorin
M70	63.56	$C_{23}H_{28}O_{10}$	509.1652	-2.55	-	-	-	-	-	-	-	+	-	-	-	-	_	-	+	hydrogenated lactiflorin isomer
M71	71.64	$C_{23}H_{28}O_{11}$	525.1598	-3.05	-	-	-	-	-	-	-	+	-	-	-	-	-	-	-	hydrogenated hydroxylated lactiflorin
M72	35.33	$C_{23}H_{28}O_{11}$	525.1635	4.00	-	-	-	-	-	-	-	-	-	-	-	-	_	-	+	hydrogenated hydroxylated lactiflorin isomer 1
M73	33.67	$C_{23}H_{28}O_{11}$	525.1624	1.90	-	-	-	-	-	-	-	-	-	-	-	-	_	-	+	hydrogenated hydroxylated lactiflorin isomer 2
M74	37.32	$C_{23}H_{28}O_{11}$	525.1624	1.90	-	-	-	-	-	-	-	-	-	-	-	-	_	-	+	hydrogenated hydroxylated lactiflorin isomer 3
M75	85.18	$C_{17}H_{18}O_8S$	381.0612	-1.05	-	-	-	-	-	-	-	+	-	-	-	-	_	-	-	hydrogenated deglycosylated lactiflorin sulfate isomer 1
M76	66.53	$C_{17}H_{18}O_8S$	381.0625	2.36	-	-	-	-	-	-	-	+	-	-	-	-	_	-	-	hydrogenated deglycosylated lactiflorin sulfate isomer 2
M77	84.31	$C_{17}H_{18}O_8S$	381.0622	1.57	-	_	-	-	-	-	_	+	-	-	-	-	-	-	-	hydrogenated deglycosylated lactiflorin sulfate isomer 3
M78	100.37	$C_{17}H_{18}O_8S$	381.0629	3.41	-	_	-	-	-	-	_	+	-	-	-	-	-	-	-	hydrogenated deglycosylated lactiflorin sulfate isomer 4
M79 ^a	40.07	$\mathrm{C_{15}H_{14}O_6}$	289.0706	-4.15	-	-	-	-	-	-	-	-	-	+	-	-	_	-	-	catechin
M80	71.09	$C_{15}H_{14}O_9S$	369.0295	2.44	-	-	-	-	-	-	-	-	+	-	-	-	_	-	-	epicatechin sulfate
M81	66.12	$C_{15}H_{14}O_9S$	369.0285	-0.27	-	_	-	-	-	-	_	-	-	+	-	-	-	-	-	catechin 5/7-O-sulfate isomer 1
M82	67.48	$C_{15}H_{14}O_9S$	369.0268	-4.88	-	_	-	-	-	-	_	-	-	+	-	-	-	-	-	catechin sulfate isomer 2
M83	72.48	$C_{15}H_{14}O_9S$	369.0280	-1.63	-	_	-	-	-	-	_	-	-	+	-	-	-	-	-	catechin sulfate isomer 3
M84	61.79	$C_{15}H_{14}O_9S$	369.0283	-0.81	-	_	-	-	-	-	_	-	-	-	+	-	-	-	-	catechin 5/7-O-sulfate isomer 2
M85	68.11	$C_{15}H_{14}O_9S$	369.0283	-0.81	-	_	-	-	-	-	_	-	-	-	+	-	-	-	-	catechin 3'/4'-O-sulfate isomer
M86	37.64	$C_{21}H_{22}O_{12}$	465.1038	0.00	_	-	_	_	-	-	_	-	-	_	+	_	-	-	-	catechin glucuronide isomer 1
M87	36.14	$C_{21}H_{22}O_{12}$	465.1008	-6.45	_	-	_	_	-	-	_	-	-	_	+	_	-	-	-	catechin glucuronide isomer 2
M88	32.94	$C_{21}H_{22}O_{12}$	465.1053	3.23	-	-	-	-	-	-	_	-	-	-	+	-	-	-	-	catechin glucuronide isomer 3
M89	66.43	$C_{21}H_{22}O_{15}S$	545.0599	-1.47	-	-	-	-	-	-	-	-	-	+	-	-	-	_	-	catechin glucuronide sulfate isomer 1

(Continued on following page)

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No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	A	0	В	OB	BO	G	L	ECG	CG	С	EA	DEA	MG	PRR	Identification
M90	61.43	C ₂₁ H ₂₂ O ₁₅ S	545.0576	-5.69	_	_	_	_	_	_	_	_	_	_	+	_	_	_	_	catechin glucuronide sulfate isomer 2
M91	54.85	$C_{21}H_{22}O_{15}S$	545.0621	2.57	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	catechin glucuronide sulfate isomer 3
M92	83.80	$C_{15}H_{16}O_8S$	355.0485	-2.25	-	-	_	-	-	-	-	-	+	-	-	-	-	-	-	3-HPP-2-ol sulfate isomer 1
M93	85.03	$C_{15}H_{16}O_8S$	355.0475	-5.07	-	-	-	-	-	-	-	-	+	-	-	-	-	-	-	3-HPP-2-ol sulfate isomer 2
M94	79.52	$C_{15}H_{16}O_8S$	355.0473	-5.63	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	3-HPP-2-ol sulfate isomer 3
M95	40.82	$C_{12}H_{14}O_{12}S$	381.0138	1.31	-	-	-	-	-	-	-	-	+	+	-	-	-	-	-	pyrogallol-O-glucuronide sulfate isomer 1
M96	41.84	$C_{12}H_{14}O_{12}S$	381.0149	4.20	-	-	_	-	-	-	-	-	-	+	-	-	-	-	-	pyrogallol-O-glucuronide sulfate isomer 2
M97	50.13	$C_{11}H_{14}O_4$	209.0812	-3.35	-	-	_	-	-	-	-	-	+	-	-	-	-	-	-	5-(3,4-dihydroxyphenyl)-valeric acid
M98	69.82	$C_{11}H_{14}O_7S$	289.0386	-0.35	-	-	_	-	-	-	-	-	+	-	-	-	-	-	-	5-(3,4-dihydroxyphenyl)-valeric acid sulfate isomer 1
M99	125.97	$C_{11}H_{14}O_7S$	289.0400	4.50	-	-	_	-	-	-	-	-	-	-	+	-	-	-	-	5-(3,4-dihydroxyphenyl)-valeric acid sulfate isomer 2
M100	66.80	$\mathrm{C}_{11}\mathrm{H}_{14}\mathrm{O}_{7}\mathrm{S}$	289.0368	-6.57	-	-	-	-	-	-	-	-	-	_	+	-	-	_	-	5-(3,4-dihydroxyphenyl)-valeric acid sulfate isomer 3
M101	64.15	$\mathrm{C}_{11}\mathrm{H}_{14}\mathrm{O}_8\mathrm{S}$	305.0348	3.61	-	-	-	-	-	-	-	-	+	_	-	-	-	_	-	trihydroxy benzenepentanoic acid sulfate isomer 1
M102	60.56	$\mathrm{C}_{11}\mathrm{H}_{14}\mathrm{O}_8\mathrm{S}$	305.0316	-6.88	-	-	-	-	-	-	-	-	-	_	+	-	-	_	-	trihydroxy benzenepentanoic acid sulfate isomer 2
M103	79.45	$C_{11}H_{12}O_7S$	287.0251	2.44	_	-	_	-	-	-	_	_	+	+	_	-	-	_	-	5-(3,4-dihydroxyphenyl)-γ-valerolactone sulfate isomer 1
M104	72.89	$C_{11}H_{12}O_7S$	287.0243	4.18	_	-	_	-	-	-	_	_	+	-	_	-	-	_	-	5-(3,4-dihydroxyphenyl)-γ-valerolactone sulfate isomer 2
M105	77.34	$C_{11}H_{12}O_7S$	287.0227	-1.39	-	-	-	-	-	-	-	-	-	+	+	-	-	_	-	5-(3,4-dihydroxyphenyl)-γ-valerolactone sulfate isomer 3
M106	76.08	$C_{11}H_{12}O_7S$	287.0219	-4.18	_	-	_	-	-	-	_	_	-	-	+	-	-	_	-	5-(3,4-dihydroxyphenyl)-γ-valerolactone sulfate isomer 4
M107	46.82	$C_{17}H_{20}O_{10}$	383.1002	4.70	-	-	-	-	-	-	-	-	+	-	-	-	-	-	-	5-(3,4-dihydroxyphenyl)-γ-valerolactone glucuronide isomer 1
M108	47.38	$C_{17}H_{20}O_{10}$	383.1005	5.48	-	-	-	-	-	-	-	-	-	+	-	-	-	-	-	5-(3,4-dihydroxyphenyl)-γ-valerolactone glucuronide isomer 2
M109	76.97	C10H10O7S	273.0072	-0.73	-	-	_	-	-	-	-	-	+	+	-	-	-	-	-	ferulic acid sulfate
M110	87.39	$C_{11}H_{12}O_6S$	271.0286	1.48	-	-	_	-	-	_	_	-	+	_	-	_	-	_	_	5-(3-hydroxyphenyl)-γ-valerolactone sulfate isomer 1
M111	82.81	$C_{11}H_{12}O_6S$	271.0284	0.74	-	-	_	-	-	_	_	-	-	_	+	_	-	_	_	5-(3-hydroxyphenyl)-γ-valerolactone sulfate isomer 2
M112	78.53	C23H20O10	455.0985	0.22	-	-	_	-	-	_	_	-	-	+	-	_	-	_	_	methyl catechin gallate
M113	99.94	$C_{15}H_{16}O_{6}$	291.0891	5.84	-	-	-	-	_	_	-	-	_	_	+	_	_	_	_	3,4-diHPP-2-ol
M114	95.61	$C_{15}H_{16}O_{6}$	291.0866	-2.75	-	-	-	-	-	-	-	-	-	_	-	-	-	_	+	3,4-diHPP-2-ol isomer 1
M115	98.62	$C_{15}H_{16}O_{6}$	291.0875	0.34	-	-	_	-	-	_	_	-	-	_	-	_	-	_	+	3,4-diHPP-2-ol isomer 2
M116	102.43	$C_{15}H_{16}O_{6}$	291.0863	-3.78	-	-	_	-	-	_	_	-	-	_	-	_	-	_	+	3,4-diHPP-2-ol isomer 3
M117	108.16	$C_{15}H_{16}O_{6}$	291.0885	3.78	-	-	_	-	-	_	_	-	-	_	-	_	-	_	+	3,4-diHPP-2-ol isomer 4
M118	110.64	$C_{15}H_{16}O_{6}$	291.0880	2.06	-	-	_	-	-	_	_	-	-	_	-	_	-	_	+	3,4-diHPP-2-ol isomer 5
M119	125.68	$C_{15}H_{16}O_{6}$	291.0882	2.75	_	_	_	_	-	_	_	_	-	-	_	_	-	_	+	3,4-diHPP-2-ol isomer 6
M120	95.25	$C_{21}H_{24}O_{15}S$	547.1477	3.66	_	-	_	-	_	_	_	-	_	+	-	_	_	_	_	3,4-diHPP-2-ol glucuronide sulfate
M121	67.91	$C_{21}H_{24}O_{14}S$	531.0796	-3.39	_	_	_	_	-	-	_	_	-	_	+	_	-	_	_	3-HPP-2-ol glucuronide sulfate

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No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	A	0	В	OB	BO	G	L	ECG	CG	С	EA	DEA	MG	PRR	Identification
M122	85.09	C ₁₆ H ₁₆ O ₉ S	383.0449	1.83	_	_	_	_	_	-	_	_	+	-	_	_	_	_	_	methyl catechin sulfate isomer 1
M123	82.21	$C_{16}H_{16}O_9S$	383.0456	3.65	-	-	-	-	-	-	-	-	-	+	-	-	-	-	-	methyl catechin sulfate isomer 2
M124	81.83	$C_{16}H_{16}O_9S$	383.0454	3.13	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin sulfate isomer 3
M125	75.73	$C_{16}H_{16}O_9S$	383.0458	4.18	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin sulfate isomer 4
M126	77.29	$C_{16}H_{16}O_9S$	383.0424	-4.70	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin sulfate isomer 5
M127	42.29	$C_{22}H_{24}O_{12}$	479.1165	-6.26	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin glucuronide isomer 1
M128	51.77	$C_{22}H_{24}O_{12} \\$	479.1171	-5.01	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin glucuronide isomer 2
M129	40.51	$C_{22}H_{24}O_{12} \\$	479.1198	0.63	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin glucuronide isomer 3
M130	55.14	$C_{22}H_{24}O_{15}S$	559.0763	0.00	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin glucuronide sulfate isomer 1
M131	53.57	$C_{22}H_{24}O_{15}S$	559.0735	-5.01	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin glucuronide sulfate isomer 2
M132	65.48	$C_{22}H_{24}O_{15}S$	559.0752	-1.97	-	-	-	-	-	-	-	-	-	-	+	-	-	-	-	methyl catechin glucuronide sulfate isomer 3
M133	81.08	$C_9H_8O_6S$	242.9965	-1.65	-	-	-	-	-	-	-	-	+	-	-	-	-	-	-	m-coumaric acid sulfate
M134	56.30	$C_7H_6O_6S$	216.9820	3.69	-	-	-	-	-	-	-	-	+	-	-	-	-	-	-	3/4-hydroxy benzonic acid sulfate isomer 1
M135	40.62	$C_7H_6O_6S$	216.9809	-1.38	-	-	-	-	-	-	-	-	-	-	-	-	-	-	+	3/4-hydroxy benzonic acid sulfate isomer 2
M136	126.14	$C_{15}H_8O_{11}S$	394.9707	-2.03	-	-	-	-	-	-	-	-	-	-	-	+	-	-	-	methyl ellagic acid sulfate isomer 1
M137	127.57	$C_{15}H_8O_{11}S$	394.9710	-1.27	-	-	-	-	-	-	-	-	-	-	-	+	-	-	-	methyl ellagic acid sulfate isomer 2
M138	127.83	$\mathrm{C_{13}H_8O_7S}$	306.9927	2.93	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	urolithin A sulfate
M139	125.56	$\mathrm{C_{13}H_8O_6S}$	290.9960	-3.09	-	-	-	-	-	-	-	-	-	-	-	+	+	-	-	urolithin B sulfate isomer 1
M140	129.13	$\mathrm{C_{13}H_8O_6S}$	290.9959	-3.44	-	-	-	-	-	-	-	-	-	-	-	-	+	-	+	urolithin B sulfate isomer 2
M141	116.13	$\mathrm{C_{13}H_8O_6S}$	290.9947	-7.65	-	-	-	-	-	-	-	-	-	-	-	-	-	-	+	urolithin B sulfate isomer 3
M142 ^a	126.77	$C_{16}H_{10}O_8$	329.0300	-0.91	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	3,3'-di-O-methylellagic acid isomer
M143	128.12	$C_{16}H_{10}O_{11}S$	408.9866	-1.22	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	3,3'-di-O-methylellagic acid sulfate isomer 1
M144	126.52	$C_{16}H_{10}O_{11}S$	408.9867	-0.98	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	3,3'-di-O-methylellagic acid sulfate isomer 2
M145	73.88	$C_{22}H_{18}O_{14}$	505.0627	-2.18	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	$3{,}3^{\prime}{\rm -di-}O{\rm -methylellagic}$ acid glucuronide isomer 1
M146	74.73	$C_{22}H_{18}O_{14}$	505.0634	1.98	-	-	-	-	-	-	-	-	-	-	-	-	+	-	-	$3,3^\prime\text{-di-}O\text{-methylellagic}$ acid glucuronide isomer 2
M147	43.79	$C_8H_8O_8S$	262.9867	0.00	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate
M148	45.96	$C_8H_8O_8S$	262.9872	1.90	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate isomer 1
M149	59.29	$C_8H_8O_8S$	262.9850	-6.46	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate isomer 2
M150	61.06	$C_8H_8O_8S$	262.9867	0.00	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate isomer 3
M151	64.01	$C_8H_8O_8S$	262.9852	-5.70	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate
M152	84.88	$C_8H_8O_8S$	262.9862	-1.90	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate isomer 4
M153	88.43	$C_8H_8O_8S$	262.9854	-4.94	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate isomer 5
M154	125.33	$C_8H_8O_8S$	262.9864	-1.14	-	-	-	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate sulfate isomer 6

(Continued on following page)

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No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	Α	0	В	OB	BO	G	L	ECG	CG	С	EA	DEA	MG	PRR	Identification
M155	127.18	C ₈ H ₈ O ₈ S	262.9887	7.60	_	-	_	-	_	_	-	-	_	_	-	_	_	+	_	methyl gallate sulfate isomer 7
M156	36.25	$C_{14}H_{16}O_{11}$	359.0622	0.56	-	-	_	-	-	-	-	-	-	-	-	-	-	+	-	methyl gallate glucuronide isomer 1
M157	37.21	$C_{14}H_{16}O_{11}$	359.0620	0.00	-	-	-	-	-	-	-	-	-	_	-	_	-	+	_	methyl gallate glucuronide
M158	50.09	$C_{14}H_{16}O_{11}$	359.0597	-6.41	_	_	_	_	-	-	_	_	-	-	_	_	-	+	-	methyl gallate glucuronide isomer 2
M159	26.75	$C_{14}H_{16}O_{11}$	359.0618	-0.56	-	-	-	-	-	-	-	-	-	_	-	_	-	+	_	methyl gallate glucuronide isomer 3
M160	40.24	$C_{20}H_{24}O_{17}$	535.0949	1.50	-	-	-	-	-	-	-	-	-	_	-	_	-	+	_	methyl gallate diglucuronide isomer 1
M161	38.77	C20H24O17	535.0943	0.37	-	-	_	-	-	_	-	-	-	-	_	_	_	+	-	methyl gallate diglucuronide isomer 2
M162	58.58	$C_{14}H_{16}O_{14}S$	439.0199	2.51	-	-	_	-	-	_	-	-	-	-	_	_	_	+	-	methyl gallate sulfate glucuronide
M163	42.90	$C_{15}H_{18}O_{11}$	373.0765	-2.95	-	-	_	-	-	_	-	-	-	-	_	_	_	+	-	methylated methyl gallate glucuronide isomer 1
M164	44.57	$C_{15}H_{18}O_{11}$	373.0773	-0.80	-	-	_	-	-	_	-	-	-	-	_	_	_	+	-	methylated methyl gallate glucuronide isomer 2
M165	43.71	$C_{15}H_{18}O_{11}$	373.0763	-3.48	_	_	_	_	-	-	_	_	-	-	_	_	-	+	-	methylated methyl gallate glucuronide isomer 3
M166	76.78	C9H10O8S	277.0014	-3.61	_	_	_	_	-	-	_	_	-	-	_	_	-	+	-	methylated methyl gallate sulfate isomer 1
M167	96.53	C9H10O8S	277.0015	-3.25	_	_	_	_	_	_	_	_	_	_	_	_	_	+	_	methylated methyl gallate sulfate isomer 2
M168	126.77	C9H10O8S	277.0018	-2.17	_	_	_	_	-	-	_	_	-	-	_	_	-	+	-	methylated methyl gallate sulfate isomer 3
M169	126.98	$C_{10}H_{12}O_8S$	291.0165	-5.15	_	-	_	_	-	-	_	_	_	-	_	_	_	+	-	dimethylated methyl gallate sulfate
M170	81.24	C7H8O4S	187.0064	-3.74	_	_	_	_	_	_	_	_	_	_	_	_	_	_	+	benzyl alcohol sulfate
M171	50.12	C8H8O6S	230.9969	0.00	_	_	_	_	-	-	_	+	-	-	+	_	-	-	-	3/4-hydroxy phenylacetic acid sulfate isomer 1
M172	45.80	C8H8O6S	230.9952	-5.19	_	_	_	_	-	-	_	+	-	-	_	_	-	-	-	3/4-hydroxy phenylacetic acid sulfate isomer 2
M173	46.27	C8H8O6S	230.9953	-6.93	_	_	_	_	-	-	_	_	-	-	+	_	-	-	-	3/4-hydroxy phenylacetic acid sulfate isomer 3
M174	53.30	C8H8O6S	230.9952	-7.36	_	_	_	_	-	-	_	_	+	+	_	_	-	-	-	3/4-hydroxy phenylacetic acid sulfate isomer 4
M175	48.15	C8H8O6S	230.9962	-3.03	_	_	_	_	-	-	_	_	-	-	_	_	-	_	+	3/4-hydroxy phenylacetic acid sulfate isomer 5
M176	43.92	C8H8O6S	230.9962	-3.03	_	_	_	_	-	-	_	_	-	-	_	_	-	-	+	3/4-hydroxy phenylacetic acid sulfate isomer 6
M177	127.45	C9H10O6S	245.0129	1.63	_	_	_	_	-	-	_	+	-	-	_	_	-	-	-	3/4-hydroxy phenylpropionic acid sulfate isomer 1
M178	74.60	C9H10O6S	245.0126	0.41	_	_	_	_	-	-	_	_	+	+	_	_	-	-	-	3/4-hydroxy phenylpropionic acid sulfate isomer 2
M179	70.97	C9H10O6S	245.0117	-3.27	_	_	_	_	_	_	_	_	_	_	+	_	_	_	_	3/4-hydroxy phenylpropionic acid sulfate isomer 3
M180	69.45	C9H10O6S	245.0109	-6.53	_	_	_	_	_	_	_	_	_	_	+	_	_	_	_	3/4-hydroxy phenylpropionic acid sulfate isomer 4
M181	55.54	C9H10O7S	261.0059	-5.75	_	_	_	_	_	_	_	+	_	_	_	_	_	_	_	3,4-dihydroxy phenylpropionic acid sulfate isomer 1
M182	58.99	C9H10O7S	261.0083	3.45	_	_	_	_	_	_	_	_	+	_	_	_	_	_	_	3,4-dihydroxy phenylpropionic acid sulfate isomer 2
M183	67.81	C9H10O7S	261.0079	1.92	_	_	_	_	_	_	_	_	+	_	_	_	_	_	_	3,4-dihydroxy phenylpropionic acid sulfate isomer 3
M184	64.31	C9H10O7S	261.0088	-5.36	_	_	_	_	_	_	_	_	_	_	+	_	_	_	_	3,4-dihydroxy phenylpropionic acid sulfate isomer 4
M185	40.17	C9H10O7S	261.0082	3.07	_	_	_	_	_	_	_	_	_	_	_	_	_	_	+	3,4-dihydroxy phenylpropionic acid sulfate isomer 5
M186	53.09	C9H10O7S	261.0062	-4.60	_	_	_	_	_	_	_	_	_	_	_	_	_	_	+	3,4-dihydroxy phenylpropionic acid sulfate isomer 6
M187	44.94	C ₈ H ₈ O ₇ S	246.9912	-2.43	_	_	_	_	_	_	_	+	_	_	_	_	_	_	_	3,4-dihydroxy phenylacetic acid sulfate isomer 1

TABLE 1 (Continued) Retention times (t_R) molecular formulae, and identities of 10 absorbed compounds and 195 metabolites of PRR and its 14 constituents in mice based on HPLC-ESI-IT-TOF-MSⁿ.

No.	t _R (min)	Formula	Meas. (Da)	Err (ppm)	Р	A	0	в	OB	BO	G	Г	ECG	CG	C	EA	DEA	MG	PRR	Identification
M188	49.91	$C_8H_8O_7S$	246.9918	0.00	I	I	I	I	I	I	I	I	+	+	I	I	1	I	I	3,4-dihydroxy phenylacetic acid sulfate isomer 2
M189	44.24	$C_8H_8O_7S$	246.9924	2.43	I	I	I	I	I	I	I	I	I	I	+	I	I	+	I	3,4-dihydroxy phenylacetic acid sulfate isomer 3
M190	51.43	$C_8H_8O_7S$	246.9925	2.83	I	I	I	I	I	I	I	I	+	I	I	I	I	I	I	3,4-dihydroxy phenylacetic acid sulfate isomer 4
M191	41.72	$C_8H_8O_7S$	246.9910	-3.24	I	I	I	I	I	I	I	Т	I	I	I	I	I	I	+	3,4-dihydroxy phenylacetic acid sulfate isomer 5
M192	125.83	$C_{12}H_{16}O_8S$	319.0488	-1.57	I	I	I	I	I	I	I	+	I	I	+	I	I	I	I	dihydroxylated methoxylated benzenepentanoic acid sulfate isomer 1
M193	90.36	$C_{12}H_{16}O_8S$	319.0507	4.39	I	I	I	I	I	I	I	I	+	I	I	I	I	I	I	dihydroxylated methoxylated benzenepentanoic acid sulfate isomer 2
M194	126.80	$C_{12}H_{16}O_8S$	319.0490	-0.94	I	I	I	I	I	I	I	I	I	I	+	I	I	I	I	dihydroxylated methoxylated benzenepentanoic acid sulfate isomer 3
M195	124.59	$C_{12}H_{16}O_8S$	319.0495	0.63	I	I	I	I	I	I	I	I	I	I	I	I	I	I	+	dihydroxylated methoxylated benzenepentanoic acid sulfate isomer 4
SUM					23	16	24	17	18	30	27	17	22	17	33	3	8	24	31	
Note: P,	paeoniflorin /-di-O-meth	; A, albiflorin; O, wellaoic acid: M	oxypaeoniflo IG methyl ga	rin; B, benzoylpaeo Ilate: PRR, Paeoni	niflori	in; OB, dix RuÌ	, hydrc hra: t _h	xybeni	zoylpaet tion tin	oniflorii Mea	n; BO,	benzoy	loxypaeo Exr. err	niflorin * · + d	; G, gal	loylpae.	niflorin; I	"lactiflori	in; ECG, eF	picatechin gallate; CG, catechin gallate; C, catechin; EA, ellagic acid;

3.1.1 Mass spectral features of eight paeoniflorins

The isomers paeoniflorin and albiflorin showed $[M+HCOOH-H]^-$ at m/z 525.16 with a molecular formula of $C_{23}H_{28}O_{11}$. The two isomers showed different relative abundances of the characteristic ions at m/z 449.14 and 479.15; for paeoniflorin, the relative abundance of the characteristic ion at m/z 449.14 was greater than that of the ion at m/z 479.15, whereas the opposite was observed for albiflorin. Oxypaeoniflorin showed $[M-H]^-$ at m/z 495.15, and its molecular formula was predicted to be $C_{23}H_{28}O_{12}$. The fragment ion at m/z 165.05 ($C_9H_{10}O_3$) was formed by the neutral loss of $C_{14}H_{18}O_9$ [150.04 Da ($C_8H_6O_3$) and 180.06 Da ($C_6H_{12}O_6$)].

Benzoylpaeoniflorin showed $[M+HCOOH-H]^-$ at m/z 629.18, and its molecular formula was $C_{23}H_{28}O_{12}$. The fragment ion at m/z 309.10 was formed by the neutral loss of 122.04 Da ($C_7H_6O_2$, benzoic acid) from the ion at m/z 431.13. The isomers benzoyloxypaeoniflorin and hydroxybenzoylpaeoniflorin showed $[M-H]^-$ at m/z 599.17 with a molecular formula of $C_{30}H_{32}O_{13}$. The two isomers showed differences in the relative abundances of the characteristic ions at m/z 477.14 and 447.13. For benzoyloxypaeoniflorin, the relative abundance of the characteristic ion at m/z 447.13, whereas the opposite was true for hydroxybenzoylpaeoniflorin.

Galloylpaeoniflorin showed $[M-H]^-$ at m/z 631.16 with a molecular formula of $C_{30}H_{32}O_{15}$. The fragment ion at m/z 509.12 was formed by a neutral loss of 122.04 Da ($C_7H_6O_2$).

Lactiflorin showed $[M-H]^-$ at m/z 461.13 with a molecular formula of $C_{23}H_{26}O_{10}$. The fragment ions at m/z 371.11 and 339.11 were formed by neutral losses of 90.03 Da ($C_3H_6O_3$) and 122.04 Da ($C_7H_6O_2$), respectively.

3.1.2 Mass spectral features of three catechins

Epicatechin gallate, catechin gallate, and catechin respectively showed $[M-H]^-$ at m/z 441.08, 441.08, and 289.06 with molecular formulae of $C_{22}H_{18}O_{10}$, $C_{22}H_{18}O_{10}$, and $C_{15}H_{14}O_6$. The cleavage pathways for these three catechins are the same as those described previously (Liu et al., 2009).

3.1.3 Mass spectral features of two ellagic acids

Ellagic acid showed $[M-H]^-$ at m/z 300.99 with a molecular formula of $C_{14}H_6O_8$. The fragment ion at m/z 284.00 was formed by a neutral loss of 17.00 Da (OH•).

3,3'-Di-O-methylellagic acid showed $[M-H]^-$ at m/z 329.03 with a molecular formula of $C_{16}H_{10}O_8$. The fragment ions at m/z 314.01 and 298.98 were formed by two sequential losses of 15.02 Da (CH₃•).

3.1.4 Mass spectral features of one gallate

Methyl gallate showed $[M-H]^-$ at m/z 183.03 with a molecular formula of $C_8H_8O_5$. The fragment ions at m/z 168.01 and 124.02 were formed by neutral losses of 15.02 Da (CH₃•) and 43.99 Da (CO₂), respectively.

3.2 Identification of the metabolites of eight paeoniflorins (P, A, O, B, OB, BO, G, L).

The metabolic pathways of benzoyloxypaeoniflorin are shown in Figure 1, and the metabolic pathways of other paeoniflorins are shown in Supplementary Figure S3.

3.2.1 Paeonimetabolin II ($C_{10}H_{16}O_4$) and its phase II metabolites (M3–M10 and M50–M54)

According to our previous study, paeonimetabolin II showed $[M-H]^-$ at m/z 199.09 with a molecular formula of $C_{10}H_{16}O_4$. M3-M6 showed $[M-H]^-$ at m/z 199.09 with a predicted molecular formula of C₁₀H₁₆O₄; thus, M3-M6 were tentatively identified as paeonimetabolin II and its isomers (Liang et al., 2013). M7-M10 showed [M-H]⁻ at *m/z* 279.05 with a predicted molecular formula of $C_{10}H_{16}O_7S$. The fragment ion at m/z 96.96 (HSO₄) was observed in the MS² spectra of M7, M9, and M10. Therefore, based on the mass spectral features and the literature (Shu et al., 1987; Liang et al., 2013), M7-M10 were tentatively identified as paeonimetabolin II sulfate isomers. M50-M54 showed [M-H]⁻ at m/z 361.15 with a predicted molecular formula of C16H26O9. The characteristic fragment at m/z 199.10 (C₁₀H₁₆O₄, paeonimetabolin II) was formed by a neutral loss of 162.05 Da in the MS² spectra of M54; therefore, M54 was tentatively identified as paeonimetabolin II glucoside. The other compounds, M50-M53, are M54 isomers and were tentatively identified as paeonimetabolin II glucoside isomers.

3.2.2 Phase II metabolites of 2,6-dihydroxycineol (_{C10H18O3}, epomediol or isomer) (M41–M49)

According to our previous identification of 2-hydroxy-1,8cineol ($C_{10}H_{18}O_2$) and our search of the SciFinder database, $C_{10}H_{18}O_3$ was tentatively identified as 2,6-dihydroxycineol or epomediol. According to the literature (Jose Cuevas et al., 2002; Xu et al., 2022), epomediol reduces cholestasis. M41–M49 showed [M–H]⁻ at m/z 265.07, and the molecular formula was predicted as $C_{10}H_{18}O_6S$. The characteristic fragment ion at m/z 183.09 ($C_{10}H_{16}O_3$) was formed by a neutral loss of 81.97 Da (H₂SO₃) in the MS² spectra of M45. Therefore, M45 was tentatively identified as 2,6-dihydroxycineol sulfate; M41–M44 and M46–M49 are isomers of M45 and were tentatively identified as 2,6-dihydroxycineol sulfate isomers.

3.2.3 Derivatives of phase II metabolites of 2,6dihydroxycineol (M55–M62)

M55 and M56 showed $[M-H]^-$ at m/z 267.09, and their molecular formula was predicted as $C_{10}H_{20}O_6S$. The characteristic fragment ion at m/z 169.09 ($C_{10}H_{17}O_2$) was formed by a neutral loss of 97.99 Da (H_2SO_4) in the MS² spectra of M56. Compared to M41–M49, M55 and M66 have two additional H atoms in their molecular formulae; thus, they were tentatively identified as hydrogenated 2,6-dihydroxycineol sulfate isomers.

M57–M62 showed $[M-H]^-$ at m/z 359.13, and their molecular formula was predicted as $C_{16}H_{24}O_{9}$. The characteristic ion at m/z 183.10 ($C_{10}H_{16}O_{3}$) was formed by a neutral loss of 176.03 Da ($C_{6}H_{8}O_{6}$) in the MS² spectra of M61; thus, M61 was tentatively identified as dehydrogenated 2,6-dihydroxycineol glucuronide. M57–M60 and M62 are M61 isomers and were thus identified as dehydrogenated 2,6-dihydroxycineol glucuronide isomers.

3.2.4 Lactiflorin-related metabolites (M69–M78)

M69 and M70 showed $[M+HCOOH-H]^-$ at m/z 509.16, and their molecular formula was predicted as $C_{23}H_{28}O_{10}$. The characteristic fragment ions at m/z 463.16 and 359.15 ($C_{16}H_{24}O_9$) were formed by sequential losses of 46.01 Da (CH_2O_2) and 104.01 Da (C_7H_4O) in the MS² spectra of M69. Based on the mass spectral features and origin of M69 (derived from lactiflorin), M69 was predicted as hydrogenated lactiflorin. M70 is an isomer of M69 and was predicted to be a hydrogenated lactiflorin isomer.

M71–M74 showed $[M+HCOOH-H]^-$ at m/z 525.15, and their molecular formula was predicted as $C_{23}H_{28}O_{11}$. The fragment ions at m/z 479.13 ($C_{23}H_{27}O_{11}$) and 461.13 ($C_{23}H_{25}O_{10}$) were formed by sequential losses of 46.01 Da (CH_2O_2) and 18.01 Da (H_2O) in the MS² spectra of M71. The ions at m/z 357.10 ($C_{16}H_{21}O_9$) and m/z 339.09 ($C_{16}H_{19}O_8$) were formed by sequential losses of 122.04 Da ($C_7H_6O_2$, benzoic acid) and 18.01 Da (H_2O) from the fragment at m/z479.13 ($C_{23}H_{27}O_{11}$). Based on the mass spectral features of M71 and its origin (derived from lactiflorin), M71 was predicted as a metabolite formed by the hydrogenation and hydroxylation of lactiflorin (i.e., M71 is a hydrogenated hydroxylated lactiflorin). M72–M74 are isomers of M71 and were thus predicted to be hydrogenated hydroxylated lactiflorin isomers.

M75–M78 showed $[M-H]^-$ at m/z 381.06 with a predicted molecular formula of $C_{17}H_{18}O_8S$. The fragment ions at m/z 259.01 ($C_{10}H_{11}O_6S$) and m/z 195.07 ($C_{10}H_{11}O_4$) were formed by neutral losses of 122.04 Da ($C_7H_6O_2$) and 186.00 Da [106.04 Da (C_7H_6O) and 79.96 Da (SO₃)] in the MS² spectra. The fragment ions at m/z 177.03 ($C_{10}H_9O_3$), m/z 165.04 ($C_9H_9O_3$), and m/z 147.05 ($C_9H_7O_2$) were formed by neutral losses of 18.01 Da (H_2O), 30.01 Da (CH_2O), and 48.02 (H_2O and CH_2O), respectively, from the ion at m/z 195.07 ($C_{10}H_{11}O_4$). M75–M78 were all derived from the metabolism of lactiflorin and were thus predicted to be metabolites formed by the hydrogenation, deglycosylation, and sulfation of lactiflorin (i.e., M75–M78 are hydrogenated deglycosylated lactiflorin sulfate isomers).

3.2.5 Other metabolites of paeoniflorins (M1–M2, M11–M40, M63–M68)

M1 and M2 showed $[M+HCOOH-H]^-$ and $[M-H]^-$ at m/z 525.16 and 495.15, respectively, and their molecular formulae



were predicted as $C_{23}H_{28}O_{11}$ and $C_{23}H_{28}O_{12}$, respectively. Their retention time and mass spectral features were consistent with those of the reference substances paeoniflorin and oxypaeoniflorin (see Table 1 and Supplementary Table S1). Thus, M1 and M2 were respectively identified as paeoniflorin and oxypaeoniflorin.

As reported previously (Liang et al., 2013), M11–M13 are isomers of desbenzoylpaeoniflorin ($C_{16}H_{24}O_{10}$) derived from albiflorin. Therefore, we identified M11–M13 as desbenzoyl albiflorin isomers.

According to our previous study (Liang et al., 2013), M14–M15, M16–M17, M23–M28, M35, M36–M40, M63–M64, M65, and M66–M67 were tentatively identified as desbenzoylpaeoniflorin isomers, methyl desbenzoylpaeoniflorin isomers, $C_{10}H_{14}O_3$ sulfate isomers, $C_{10}H_{20}O_4$ glucuronide, $C_8H_8O_3$ glucuronide isomers, gallic acid sulfates, hippuric acid, and hydroxyhipuric acid, respectively.

According to the literature, M18–M22 were tentatively identified as paeonimetabolin I glucuronide isomers (Wang et al., 2017b; Sun et al., 2018); M29–M34 were tentatively identified as $C_{10}H_{18}O_4$ glucuronide isomers (Liang et al., 2013; Sun et al., 2018), and M68 was tentatively identified as salicylic acid glucuronide (Shen et al., 2015).

3.3 Identification of the metabolites of three catechins (CG, ECG, C)

The metabolic pathway of catechin gallate is shown in Figure 2, and the metabolic pathways of other catechins are shown in Supplementary Figure S4.

3.3.1 Catechin (C $_{15}H_{14}O_6)$ and its phase II metabolites (M79–M91)

M79 showed $[M-H]^-$ at m/z 289.07, and its molecular formula was predicted as $C_{15}H_{14}O_6$. The retention time and mass spectral data of M79 are consistent with those of catechin's reference substance (see Table 1 and Supplementary Table S1). Therefore, M79 was identified as catechin.

M80-M85 showed $[M-H]^-$ at m/z 369.02, and their molecular formula was predicted as C₁₅H₁₄O₉S. The fragment ion at m/z 289.07 (C₁₅H₁₃O₆) was formed by a neutral loss of 79.96 Da in the MS² spectra of M80-M85. Therefore, based on the mass spectral features and the literature (Gonzalez-Manzano et al., 2009; Rodriguez-Mateos et al., 2014), M80-M85 were predicted to be (epi)catechin sulfate isomers. In addition, in the MS^2 spectra of M81 and M84, the fragment ion at m/z 216.97 (C₇H₅O₆S) of the sulfate conjugate of the characteristic fragment ion at m/z 137.02 (C₇H₅O₃) formed after the RDA cleavage of the C-ring of catechin. Therefore, we presumed that the sulfate group bound to the hydroxyl group at C-5 or C-7. In the MS² spectra of M85, the ions at m/z 289.06 (C₁₅H₁₃O₆) and m/z 179.03 $(C_9H_7O_4)$ were formed by the sequential losses of 79.96 Da (SO_3) and 110.04 Da $(C_6H_6O_2)$ from the ion at m/z 369.02; however, no neutral loss of 82 Da (H₂SO₃) was observed. We concluded that the sulfate group did not bind to the hydroxyl group at C-3 due to the energy barrier required for bond breakage. No fragment ion was observed at m/z 216.97 (C₇H₆O₆S), and we concluded that the sulfate group did not bind to the hydroxyl group at C-5 or C-7. Hence, we assumed that the sulfate group bound to the hydroxyl group at C-3' or C-4'.

M86–M88 showed $[M-H]^-$ at m/z 465.10, and their molecular formula was predicted as $C_{21}H_{22}O_{12}$. The ion at m/z 289.06 ($C_{15}H_{13}O_6$) was formed by a neutral loss of 176.03 Da ($C_6H_8O_6$) in the MS² spectra. According to the literature (Gonzalez-Manzano et al., 2009; Liang et al., 2013; Rodriguez-Mateos et al., 2014), M86–M88 were predicted to be catechin glucuronide isomers.

M89–M91 showed $[M-H]^-$ at m/z 545.06, and their molecular formula was predicted as $C_{21}H_{22}O_{15}S$. The fragment ions at m/z 369.02 ($C_{15}H_{13}O_9S$) and 289.05 ($C_{15}H_{13}O_6$) were formed by the sequential losses of 176.03 Da

 $(\rm C_6H_8O_6)$ and 79.96 Da (SO_3) in the $\rm MS^2$ spectra. Therefore, M89–M91 were predicted to be catechin glucuronide sulfate isomers.

3.3.2 5-(3,4-Dihydroxyphenyl)-valeric acid ($C_{11}H_{14}O_4$) and its phase II metabolites (M97–M100)

M97 showed $[M-H]^-$ at m/z 209.08, and its molecular formula was predicted as $C_{11}H_{14}O_4$. The fragment ion at m/z 147.07 ($C_{10}H_{11}O$) was formed by a neutral loss of 62.01 Da (CO_2+H_2O) in the MS² spectrum. According to the literature (Scheline, 1970), M97 was predicted to be 5-(3,4-dihydroxyphenyl)-valeric acid.

M98–M100 showed $[M-H]^-$ at m/z 289.03, and their molecular formula was predicted as $C_{11}H_{14}O_7S$. The fragment ion at m/z 209.07 ($C_{11}H_{13}O_4$) was formed by a neutral loss of 79.96 Da (SO₃) in the MS² spectra. Therefore, M98–M100 were predicted to be 5-(3,4-dihydroxyphenyl)-valeric acid sulfate isomers.

3.3.3 Phase II metabolites of 5-(3,4-dihydroxyphenyl)- γ -valerolactone (C₁₁H₁₂O₄) (M103-M108)

M103–M106 and M107–M108 showed $[M-H]^-$ at m/z 287.02 and 383.09, respectively, and their molecular formulae were predicted as $C_{11}H_{12}O_7S$ and $C_{17}H_{20}O_{10}$, respectively. The fragment ion at m/z 207.05 ($C_{11}H_{11}O_4$) was formed by a neutral loss of 79.96 Da (SO₃) from the ion at m/z 287.02 and a neutral loss of 176.03 Da ($C_6H_8O_6$) from the ion at m/z 383.09 in the MS² spectra. Based on our previous report (Liang et al., 2013) and a report of 5-(3,4-dihydroxyphenyl)- γ -valerolactone (Kohri et al., 2003), we presumed that M103–M106 were 5-(3,4-dihydroxyphenyl)- γ -valerolactone sulfate isomers, while M107–M108 were 5-(3,4-dihydroxyphenyl)- γ -valerolactone glucuronide isomers.

3.3.4 Identification of methyl catechin gallate (M112)

M112 showed $[M-H]^-$ at m/z 455.09, and its molecular formula was predicted as $C_{23}H_{20}O_{10}$. The fragment ion at m/z303.08 ($C_{16}H_{15}O_6$) was formed by a neutral loss of 152.01 Da ($C_7H_4O_4$, galloyl) in the MS² spectrum, and the signal at m/z303.08 indicated an additional methyl group compared to catechin ($C_{15}H_{13}O_6$). Thus, M112 was predicted to having a methylcatechin skeleton. In addition, the fragment ion at m/z169.01 ($C_7H_5O_5$, gallic acid group) was observed; thus, its structure was presumed to contain a gallic acid group. Comparison of the molecular formula of M112 ($C_{23}H_{20}O_{10}$) with that of the original constituent (catechin gallate, $C_{22}H_{18}O_{10}$) suggested that M112 was a methylated product of catechin gallate. According to a previous report, M112 was predicted to be methyl catechin gallate (Kohri et al., 2003).



3.3.5 3,4-DiHPP-2-ol ($C_{15}H_{16}O_6$) and its phase II metabolites (M113–M119, M120)

M113–M119 showed $[M-H]^-$ at m/z 291.08, and their molecular formula was predicted as $C_{15}H_{16}O_6$. Based on the literature (Hara-Terawaki et al., 2017) and the origins of the metabolites (M113 was derived from catechin, while M114–M119 were derived from PRR decoction), M113 was tentatively identified as 3,4-diHPP-2-ol, while M114–M119 are isomers of M113 and were thus predicted to be 3,4-diHPP-2-ol isomers.

M120 showed $[M-H]^-$ at m/z 547.14, and its molecular formula was predicted as $C_{21}H_{24}O_{15}S$. The characteristic ion at m/z 371.11 ($C_{15}H_{15}O_9S$) was formed by a neutral loss of 176.03 Da ($C_6H_8O_6$) in the MS² spectrum. Therefore, M120 was tentatively identified as 3,4-diHPP-2-ol glucuronide sulfate.

3.3.6 Phase II metabolites of methyl catechin (C₁₆H₁₆O₆) (M122–M126, M127–M129, and M130–M132)

M122–M126 showed $[M-H]^-$ at m/z 383.04, and their molecular formula was predicted as $C_{16}H_{16}O_9S$. The ion at m/z 303.08 ($C_{16}H_{15}O_6$) of aglycone methylcatechin was formed by a neutral loss of 79.96 Da (SO₃) in the MS² spectra. In addition, the fragment ions at m/z 216.97 ($C_7H_5O_6S$) and 137.02 ($C_7H_5O_3$) were observed in all these metabolites except M124; hence, we concluded that the sulfate groups of M122, M123, M125, and M126 bind to the hydroxyl group at C-5 or C-7, and the methylation reaction occurs at the hydroxyl group at C-3, C-3', or C-4'. According to our previous study (Liang et al., 2013), M122–M126 were predicted to be methyl catechin sulfate isomers.

M127–M129 showed $[M-H]^-$ at m/z 479.11, and their molecular formula was predicted as $C_{22}H_{24}O_{12}$. The fragment ion at m/z 303.08 ([aglycone–H]⁻, $C_{16}H_{15}O_6$) was formed by a neutral loss of 176.03 Da ($C_6H_8O_6$) in the MS² spectra. The fragment ion at m/z 313.0573 ($C_{13}H_{13}O_9$) was observed in the MS² spectrum of M129, indicating that the glucuronide group binds to the hydroxyl group at C-5 or C-7, and the methylation reaction occurs at the hydroxyl group at C-3, C-3', or C-4'. According to our previous study (Liang et al., 2013), M127–M129 were predicted to be methyl catechin glucuronide isomers.

M130–M132 showed $[M-H]^-$ at m/z 559.07, and its molecular formula was predicted as $C_{22}H_{24}O_{15}S$. The fragment ions at m/z 383.04 ($C_{16}H_{15}O_9S$) and 303.08 ([aglycone–H]⁻, $C_{16}H_{15}O_6$) were formed by sequential losses of 176.03 Da ($C_6H_8O_6$) and 79.96 Da (SO_3) in the MS² spectra. Therefore, M130–M132 were predicted to be methyl catechin glucuronide sulfate isomers.

3.3.7 Other metabolites of catechins

M92-M94, M95-M96, M101-M102, M109, M110-M111, M121, M133, and M134-M135 showed [M-H]⁻ at *m/z* 355.04, 381.01, 305.03, 273.00, 271.02, 531.08, 242.99, and 216.98, respectively, and their molecular formulae were predicted as $C_{15}H_{16}O_8S$, $C_{12}H_{14}O_{12}S$, $C_{11}H_{14}O_8S$, $C_{10}H_{10}O_7S$, $C_{11}H_{12}O_6S$, C21H24O14S, C9H8O6S, and C7H6O6, respectively. According to our previous study (Liang et al., 2013), M92-M94, M95-M96, M101-M102, M109, M110-M111, M121, M133, and M134-M135 were predicted to be 3-HPP-2-ol sulfate pyrogallol-O-glucuronide isomers, sulfates, trihydroxy benzenepentanoic acid sulfate isomers, ferulic acid sulfate, 5-(3-hydroxyphenyl)-y-valerolactone sulfate isomers, 3-HPP-2-ol glucuronide sulfate, m-coumaric acid sulfate, and 3/4-hydroxy benzoic acid sulfate isomers, respectively.

3.4 Metabolites of two ellagic acid compounds (EA and DEA)

The metabolic pathway of 3,3'-di-O-methylellagic acid is shown in Figure 3, and the metabolic pathway of ellagic acid is shown in Supplementary Figure S5.

M136 and M137 showed $[M-H]^-$ at m/z 394.97, and their molecular formula was predicted as $C_{15}H_8O_{11}S$. The fragment ions at m/z 315.01 ($C_{15}H_7O_8$) and m/z 299.98 ($C_{14}H_4O_8$, ellagic acid) were formed by sequential losses of 79.96 Da (SO₃) and 15.02 Da (CH₃•) in the MS² spectra. Therefore, M136 and M137 were predicted to be methyl ellagic acid sulfate and isomers.

M138 and M139–M141 showed $[M-H]^-$ at m/z 306.99 and m/z 290.99, respectively, and their molecular formulae were predicted as $C_{13}H_8O_7S$ and $C_{13}H_8O_6S$, respectively. M138 was predicted to be urolithin A sulfate (Espin et al., 2013), while M139–M141 were predicted as urolithin B sulfate and isomers (Wang et al., 2017a).

Isomer (C₁₆H₁₀O₈, M142) and phase II metabolites (M143-M144, M145-M146) of 3,3'-di-O-methylellagic acid were tentatively identified as follows. M142 showed [M-H]at m/z 329.03, and its molecular formula was predicted as C16H10O8. The fragment ions in its MS2 spectrum were consistent with those of the 3,3'-di-O-methylellagic acid reference compound; however, the retention times were different. M142 was therefore predicted to be an isomer of 3,3'-di-O-methylellagic acid. M143-M144 showed [M-H]⁻ at m/z 408.98, and their molecular formula was predicted as $C_{16}H_{10}O_{11}S$. The fragment ion at m/z 329.02 ([aglycone-H]⁻, C₁₆H₉O₈) was formed by a neutral loss of 79.96 Da (SO₃) in the MS^2 spectra. The ions at m/z 314.00 ($C_{15}H_6O_8$) and m/z 298.98 (C14H3O8) were consistent with the 3,3'-di-O-methylellagic acid reference compound. Thus, M143-M144 were predicted to be 3,3'-di-O-methylellagic acid sulfates. M145-M146 showed [M-H]⁻ at m/z 505.06, and their molecular formula was predicted as C₂₂H₁₈O₁₄. The fragment ion at *m/z* 329.03 ([aglycone-H]⁻, $C_{16}H_9O_8$) was formed by a neutral loss of 176.03 Da ($C_6H_8O_6$) in the MS² spectra; hence, M145-M146 were predicted to be 3,3'-di-Omethylellagic acid glucuronides.

3.5 Identification of the metabolites of methyl gallate

The metabolic pathway of methyl gallate is shown in Figure 4.

3.5.1 Phase II metabolites of methyl gallate (C₈H₈O₅) (M147–M155, M156–M159, M160–M161, M162)

M147-M155 showed [M-H]⁻ at m/z 262.98, and their molecular formula was predicted as C8H8O8S. The fragment ion at m/z 183.02 ([aglycone-H]⁻, C₈H₇O₅) was formed by a neutral loss of 79.96 Da (SO₃) in the MS² spectra. Due to its chemical structure, methyl gallate is unlikely to produce the metabolites of these nine sulfate conjugates. Thus, we speculated that the aglycones of M147-M155 may be methyl gallate and methylgallic acid, which are isomers. These two isomers show different relative abundances of the characteristic ions at m/z168.00 and m/z 124.01; for methyl gallate, the relative abundance of the characteristic ion at m/z 124.01 is higher than that at m/z168.00, whereas the opposite is true for methylgallic acid. The MS² spectra of both M147 and M151 showed higher relative abundances of the characteristic ion at m/z 124.01 compared to that at m/z 168.00; thus, they were predicted to be methyl gallate sulfates. The other metabolites (M148-M150 and M152-M155) were predicted to be methyl gallate sulfate isomers and potentially originated from the metabolic reactions of methyl gallate (e.g., hydroxylation, dehydroxylation, hydrolysis, methylation, and sulfation).

M156–M159 showed $[M-H]^-$ at m/z 359.06, and their molecular formula was predicted as $C_{14}H_{16}O_{11}$. The fragment



ion at m/z 183.02 ([aglycone–H]⁻, C₈H₇O₅) was formed by a neutral loss of 176.03 Da (C₆H₈O₆) in the MS² spectra of M156–M158. Thus, M156–M158 were predicted to be methyl gallate glucuronides. M159 was predicted as a methyl gallate glucuronide isomer since it is an isomer of M156–M158. Because of its chemical structure, methyl gallate is unlikely to produce the four glucuronate conjugates. Thus, we speculated that the aglycones of M156–M159 may be methyl gallate and methylgallic acid. In the MS² spectrum of M157, the relative abundance of the characteristic ion at m/z 124.01 was higher than that at m/z 168.00; hence, M157 was tentatively identified as methyl gallate glucuronide, whereas M156, M158, and M159 had insufficient characteristic ions and could only be identified as methyl gallate glucuronide isomers.

M160 and M161 showed $[M-H]^-$ at m/z 535.09, and their molecular formula was predicted as $C_{20}H_{24}O_{17}$. The fragment ions at m/z 359.06 ($C_{14}H_{15}O_{11}$) and m/z 183.02 ([aglycone-H]⁻, $C_8H_7O_5$) were formed by sequential losses of two 176.03 Da ($C_6H_8O_6$) in the MS² spectra. Hence, M160–M161 were predicted to be methyl gallate diglucuronides.

M162 showed $[M-H]^-$ at m/z 439.01, and its molecular formula was predicted as $C_{14}H_{16}O_{14}S$. The fragment ions at m/z 262.97 ($C_8H_7O_8S$) and m/z 183.01 ([aglycone-H]⁻, $C_8H_7O_5$) were formed by sequential losses of 176.03 Da ($C_6H_8O_6$) and 79.96 Da (SO_3) in the MS² spectra. Therefore, M162 was predicted as methyl gallate sulfate glucuronide.

3.5.2 Phase II metabolites of methylated methyl gallate ($C_9H_{10}O_5$) (M163–M165, M166–M168)

M163–M165 showed $[M-H]^-$ at m/z 373.07, and their molecular formula was predicted as $C_{15}H_{18}O_{11}$. The fragment

ion at m/z 175.02 (C₆H₇O₆) was observed in the MS² spectra of M163–M165, suggesting the aglycone to be C₉H₁₀O₅. The fragment ion at m/z 183.01 (C₈H₇O₅, methyl gallate) was also observed, suggesting the aglycone to be methylated methyl gallate (C₉H₁₀O₅).

M166–M168 showed $[M-H]^-$ at m/z 277.00, and their molecular formula was predicted as $C_9H_{10}O_8S$. The fragment ion at m/z 197.04 ([aglycone–H]⁻, $C_9H_9O_5$) was formed by a neutral loss of 79.96 Da (SO₃) in the MS² spectra.

Therefore, M163–M165 and M166–M168 were predicted as methylated methyl gallate glucuronide isomers and methylated methyl gallate sulfate isomers, respectively.

3.5.3 Phase II metabolite of dimethylated methyl gallate (M169)

M169 showed $[M-H]^-$ at m/z 291.01, and its molecular formula was predicted as $C_{10}H_{12}O_8S$. The fragment ions at m/z 211.06 ($C_{10}H_{11}O_5$) and 196.03 ($C_9H_8O_5$) were formed by sequential losses of 79.96 Da (SO₃) and 15.02 Da (CH₃•) in the MS² spectrum. Therefore, M169 was predicted to be dimethylated methyl gallate sulfate.

3.6 Identification of metabolites derived from compounds with different structure types

M171–M176, M177–M180, M181–M186, M187–M191, and M192–M195 showed $[M-H]^-$ at m/z 230.99, 245.01, 261.00, 246.99, and 319.04, respectively, and their molecular formulae were predicted as $C_8H_8O_6S$, $C_9H_{10}O_6S$, $C_9H_{10}O_7S$, $C_8H_8O_7S$, and $C_{12}H_{16}O_8S$, respectively.



Based on our previous study (Liang et al., 2013), M171–M176, M177–M180, M181–M186, M187–M191, and M192–M195 were predicted as 3/4-hydroxy phenylacetic acid sulfate and its isomers, 3/4-hydroxy phenylpropionic acid sulfate and its isomers, 3,4-dihydroxy phenylpropionic acid sulfate and its isomers, and dihydroxy phenylacetic acid sulfate and its isomers, and dihydroxylated methoxylated benzenepentanoic acid sulfate and its isomers, respectively.

M171-M176, M177-M180, M181-M186, and M192-M195 are all derived from paeoniflorins and catechins, while M187-M191 are derived from paeoniflorins, catechins, and gallic acids.

3.7 Identification of the metabolites of PRR

A total of 31 PRR metabolites were identified (Supplementary Figure S6), the analyses of 30 metabolites were consistent with the above (except M170).

M170 showed $[M-H]^-$ at m/z 187.00, and its molecular formula was predicted as $C_7H_8O_4S$. The characteristic fragment ion at m/z 107.05 (C_7H_7O) was formed by a neutral

loss of 79.96 Da (SO₃) in its $\rm MS^2$ spectrum. Based on a previous report (Liu et al., 2020), M170 was tentatively identified as benzyl alcohol sulfate.

4 Discussion

4.1 Origins of PRR metabolites in mice

A total of 31 metabolites of PRR were identified in mice in this study. The metabolites of 14 PRR constituents were compared, and the relevant literature was analyzed to determine their possible origins. Finally, we identified 13 metabolites derived from catechins (M114–M119, M135, M175–M176, M185–M186, M191, and M195), 14 derived from paeoniflorins (M4–M5, M17, M34, M48–M51, M53, M56, M70, and M72–M74), two derived from ellagic acids (M140–M141), and two derived from other constituents (M65 and M170).

In rats, our previous research (Liang et al., 2013) identified 27, 27, six, 25, and five PRR metabolites derived from catechins,

gallic acids, catechins and gallic acids, paeoniflorins, and other constituents, respectively. In comparison, we identified 20 new metabolites formed by hydroxylation, hydrogenation, glucuronidation, and sulfation in mice: seven catechin-related metabolites (M114-M119, 3,4-diHPP-2-ol isomers; and M195, dihydroxylated methoxylated benzenepentanoic acid sulfate); paeoniflorin-related metabolites (M48-M49, 10 2.6dihydroxycineol sulfate isomers; M50, M51, and M53, paeonimetabolin II glucoside isomers; M56, hydrogenated 2,6dihydroxycineol sulfate; M70, hydrogenated lactiflorin isomer; and M72-M74, hydrogenated hydroxylated lactiflorin isomers); two ellagic acid-related metabolites (M140-M141, urolithin B sulfate isomers); and one other metabolite (M170, benzyl alcohol sulfate).

In our previous research (Liang et al., 2013), the administered dosage in male Sprague-Dawley rats was 9.96 g PRR crude drug/kg rat body weight (equivalent to 18.99 g PRR crude drug/kg of mouse body weight). Pre-experiments revealed that the administration of a high dosage of PRR to ICR mice may lead to diarrhea. Therefore, the administered dosage of PRR in this study was 200 mg crude drug/kg mouse body weight, a much lower dosage than that administered in rats. Due to the lower PRR dose used in this study, we did not expect to discover more metabolites of PRR than that previously reported in rats. In addition, significant species differences have been reported in phase I and phase II metabolism (Qin et al., 2021); hence, the 20 new metabolites of PRR identified in mice might be explained by species differences. PRR produces 11 identical metabolites in rats and mice: paeonimetabolin II and its isomers (M4 and M5); C10H18O4 glucuronide (M34); methyl dibenzoylpaeoniflorin isomer (M17); 3/4-hydroxy phenylacetic acid sulfate isomers (M175-M176); 3,4-dihydroxy phenylpropionic acid sulfate isomers (M185-M186); 3,4-dihydroxy phenylacetic acid sulfate (M191); 3/4-hydroxy benzoic acid sulfate isomer (M135); and hippuric acid (M65).

4.2 New metabolites found in this study

4.2.1 New metabolites of PRR and its 14 constituents in mice

This was the first study on the *in vivo* metabolism of PRR and its 14 major constituents in mice. Thus, all the metabolites identified were newly found in mice.

4.2.2 New metabolites of PRR and each of its 14 constituents in the whole animal kingdom and microorganism

For 14 constituents of PRR, based on comparison with the literature, the following new metabolites were found in this study: eight new metabolites of paeoniflorin (M42–M47, M52, and M55) (Doyle and Griffiths, 1980; Hattori et al., 1985; Shu et al., 1987; Liang et al., 2013; Cao et al., 2015; Zhu et al.,

2018); 11 new metabolites of albiflorin (M7, M9-M10, M36-M39, M45, M47, M55, and M65) (Cao et al., 2015; Wang et al., 2017b); 24 new metabolites of oxypaeoniflorin (Ml, M14-M15, M18-M19, M22, M25-M26, M28-M29, M31, M33, M41-M47, M55, M57-M58, and M60-M61) (Hattori et al., 1985); and 17 new metabolites of benzoylpaeoniflorin (Ml, M14-M15, M18-M19, M23-M24, M27, M31, M33, M43-M45, M47, M57-M58, and M61) (Hattori et al., 1985). No metabolites of hydroxybenzoylpaeoniflorin, benzoyloxypaeoniflorin, galloylpaeoniflorin, or lactiflorin had been reported in vivo or in vitro. Thus, we identified 18, 30, 27, and 17 new metabolites of hydroxybenzoylpaeoniflorin, benzoyloxypaeoniflorin, galloylpaeoniflorin, or lactiflorin, respectively, in this study.

Considering the literature (Kohri et al., 2003), we found 21 new metabolites of epicatechin gallate: M80, M92–M93, M95, M98, M109, M101, M103–M104, M107, M110, M122, M133, M134, M174, M178, M182–M183, M188, M190, and M193. No metabolites of catechin gallate had yet been reported *in vivo* or *in vitro*. Hence, we identified 17 new metabolites of catechin gallate and one new metabolite of catechin (M121) (Liang et al., 2013).

Considering the literature (Doyle and Griffiths, 1980), we found two new metabolites of ellagic acid (M136 and M137). No *in vivo* or *in vitro* metabolites of 3,3'-di-O-methylellagic acid had been reported; hence, we identified eight new metabolites of 3,3'-di-O-methylellagic acid in this study.

Considering the literature (Jiamboonsri et al., 2016), we found 20 new metabolites of methyl gallate in this study (M147–M155, M160–M169, and M189).

Considering the literature (Liang et al., 2013), 20 new metabolites (M48–M49, M51–M52, M54, M56, M70, M72–M74, M114–M119, M140–M141, and M170) of PRR decoction were found in this study.

4.2.3 New metabolites of paeoniflorins, catechins, ellagic acids, and gallic acids in the whole animal kingdom and microorganism

A total of 26 new metabolites of paeoniflorin-type compounds were found in this study. Specifically, we found eight new metabolites of paeoniflorin, six new metabolites of albiflorin, 13 new metabolites of oxypaeoniflorin, seven new metabolites of benzoylpaeoniflorin, six new metabolites of hydroxybenzoylpaeoniflorin, seven new metabolites of new metabolites benzoyloxypaeoniflorin, nine of galloylpaeoniflorin, and 12 new metabolites of lactiflorin. The 26 new metabolites were hydrogenated 2,6-dihydroxycineol sulfate isomer (M55), 2,6-dihydroxycineol sulfate isomers (M41-M47), paeonimetabolin II sulfate isomers (M7-M10), paeonimetabolin II glucoside isomers (M52 and M54), hydrogenated lactiflorin and its isomer (M69 and M70), lactiflorin hydrogenated hydroxylated isomer (M71), hydrogenated deglycosylated lactiflorin sulfates (M75-M78), 3/4-hydroxy phenylacetic acid sulfate isomers (M171 and M172), 3/4-hydroxy phenylpropionic acid sulfate (M177), 3,4-dihydroxy phenylpropionic acid sulfate (M181), and 3,4-dihydroxy phenylacetic acid sulfate (M187).

One new metabolite of each of catechin gallate and catechin was found, namely, 3,4-diHPP-2-ol glucuronide sulfate (M120) and 3-HPP-2-ol glucuronide sulfate (M121), respectively.

Eight new metabolites of ellagic acids were found in this study. Specifically, we found two new metabolites of ellagic acid [methylellagic acid sulfates (M136 and M137)] and six new metabolites of 3,3'-di-O-methylellagic acid [3,3'-di-O-methylellagic acid sulfates (M142), 3,3'-di-O-methylellagic acid sulfates (M143 and M144), and 3,3'-di-O-methylellagic acid glucuronides (M145, and M146)].

Eight new metabolites of gallic acids were found in this study: methyl gallate diglucuronides (M160 and M161); methyl gallate sulfate glucuronide (M162); methylated methyl gallate glucuronides (M163–M165); dimethylated methyl gallate sulfate (M169); and 3, 4-dihydroxy phenylacetic acid sulfate isomer (M189).

4.3 Isomer metabolites produced by multiple original constituents of PRR

4.3.1 Isomer metabolites produced by compounds with the same structure type

The eight paeoniflorin constituents all produced desbenzoylpaeoniflorin isomers (M11-M15). Seven paeoniflorin constituents (all except lactiflorin) produced 2,6dihydroxycineol sulfate isomers (M41-M47). Seven paeoniflorin constituents (all except albiflorin) produced paeonimetabolin I glucuronide isomers (M18-M22) and $C_{10}H_{18}O_4$ glucuronide isomers (M29-M34). Six paeoniflorin constituents (all except albiflorin and lactiflorin) produced C10H14O3 sulfate isomers (M23-M28), hydrogenated 2,6-dihydroxycineol sulfate isomer (M55), and dehydrogenated 2,6-dihydroxycineol glucuronide isomers (M57-M62). Albiflorin, benzoyloxypaeoniflorin, and galloylpaeoniflorin produced C8H8O3 glucuronide isomers (M36-M40) and hippuric acid (M65). Paeoniflorin, benzoyloxypaeoniflorin, and galloylpaeoniflorin produced paeonimetabolin II glucoside isomers (M50-M54). Albiflorin, hydroxybenzoylpaeoniflorin, and galloylpaeoniflorin produced paeonimetabolin II sulfate isomers (M7-M10). Albiflorin, hydroxybenzoylpaeoniflorin, benzoylpaeoniflorin, and galloylpaeoniflorin all produced paeoniflorin (M1). Paeoniflorin, hydroxybenzoylpaeoniflorin, and benzoyloxypaeoniflorin all produced oxypaeoniflorin (M2). Both benzoyloxypaeoniflorin and galloylpaeoniflorin produced paeonimetabolin II isomers (M3 and M6).

All three catechin constituents produced (epi)catechin sulfate isomers (M80–M85), 5-(3,4-dihydroxyphenyl)- γ -valerolactone sulfate isomers (M103–M106), and methyl-catechin sulfates

(M122–M126). Both catechin gallate and catechin produced catechin glucuronide sulfates (M89–M91). Both epicatechin gallate and catechin produced 3-HPP-2-ol sulfate isomers (M92–M94), 5-(3,4-dihydroxyphenyl)-valeric acid sulfate isomers (M98–M100), trihydroxy-benzenepentanoic acid sulfate isomers (M101–M102), and 5-(3-hydroxyphenyl)- γ -valerolactone sulfate isomers (M110–M111). Both epicatechin gallate and catechin gallate produced pyrogallol-O-glucuronide sulfate isomers (M95–M96) and 5-(3,4-dihydroxyphenyl)- γ -valerolactone glucuronide isomers (M107–M108).

Both ellagic acid constituents produced urolithin B sulfate isomers (M139-M141).

4.3.2 Isomer metabolites produced by compounds with different structure types

Lactiflorin and the three catechin constituents produced 3hydroxy phenylacetic acid sulfate isomers (M171–M176) and 3/ 4-hydroxy phenylpropionic acid sulfate isomers (M177–M180).

Lactiflorin, epicatechin gallate, and catechin produced 3,4dihydroxy phenylpropionic acid sulfate isomers (M181–M186) and dihydroxy-methoxyl-benzenepentanoic acid sulfate isomers (M192–M195).

Lactiflorin, epicatechin gallate, catechin gallate, catechin, methyl gallate, and PRR produced 3,4-dihydroxy phenylacetic acid sulfate isomers (M187–M191).

4.4 Origins of the effective forms of PRR

In a previous study, we found 21 effective forms of PRR that account for its effects of clearing away heat, cooling the blood, and dissipating blood stasis (Xu et al., 2022). In this study, we elucidated some possible origins of the 10 known effective forms of PRR and their isomers:

- Desbenzoylpaeoniflorin isomer (C3) can be derived from eight original constituents: paeoniflorin, albiflorin, oxypaeoniflorin, benzoylpaeoniflorin, hydroxybenz oylpaeoniflorin, benzoyloxypaeoniflorin, galloylpae oniflorin, and lactiflorin.
- Paeoniflorin (C1) can be derived from benzoylpaeoniflorin, hydroxybenzoylpaeoniflorin, and galloylpaeoniflorin.
- Oxypaeoniflorin (C2) can be derived from paeoniflorin, hydroxybenzoylpaeoniflorin, and benzoyloxypaeoniflorin.
- 3/4-Hydroxy benzoic acid sulfate (C8) isomer can be derived from epicatechin gallate.
- C₁₀H₁₈O₄ glucuronide (C16) isomer can be derived from seven prototype paeoniflorin constituents (all except albiflorin).
- C₈H₈O₃ glucuronide isomers (C19 and C20) can be derived from albiflorin, benzoyloxypaeoniflorin, and galloylpa eoniflorin.
- 3'-O-methyl (epi)catechin 5-O-glucuronide (C5) isomer can be derived from catechin.

- 3-Hydroxy phenylpropionic acid sulfate (C6) isomer can be derived from lactiflorin, epicatechin gallate, catechin gallate, and catechin.
- 3,3'-Di-O-methylellagic acid isomer can be isomerized from 3,3'-di-O-methylellagic acid.

According to our previous study (Liang et al., 2013), $C_{10}H_{18}O_2$ (C9-C15) glucuronides and $C_{10}H_{14}O_3$ glucuronide (C17), two effective forms of PRR, can be derived from paeoniflorin in rats; however, their origins could not be determined in this study. Other effective forms of PRR including 3-hydroxy-4-methoxyphenylpropionic acid sulfate (C7), 3-methoxy-4-hydroxyphenylpropionic acid sulfate (C18), and benzoyl glucuronide (C21) can be derived from catechins in rats (Liang et al., 2013), but their specific origins need further investigation.

4.5 Insights for the quality control of PRR

Seventeen of the effective forms of PRR are metabolites that are not present in PRR and cannot be used for quality control. This study identified some of the precursors of the 10 effective forms of PRR, which can be used as indicators for the quality control of PRR.

5 Conclusion

This was the first study on the in vivo metabolism of PRR and its 14 constituents in mice. The metabolites were identified by the HPLC-DAD-ESI-IT-TOF-MSⁿ. In total, we identified 23, 16, 24, 17, 18, 30, 27, 17, 22, 17, 33, 3, 8, 24, and 31 metabolites of paeoniflorin, albiflorin, oxypaeoniflorin, benzoylpaeoniflorin, hydroxybenzoylpaeoniflorin, benzoyloxypaeoniflorin, galloylpaeoniflorin, lactiflorin, epicatechin gallate, catechin gallate, catechin, ellagic acid, 3,3'-di-Omethylellagic acid, methylgallate, and PRR, respectively, in mice. The main metabolic reactions included methylation, hydrogenation, hydrolysis, hydroxylation, glucuronidation, and sulfation. We elucidated the metabolites and metabolic pathways of the 14 constituents of PRR, including paeoniflorins, catechins, ellagic acids, and gallic acids, in mice, and clarified the possible origins of the 10 known effective forms of PRR and their isomers. The findings will facilitate further studies on the effective forms of PRR in vivo and are of great significance for exploring the mechanisms of action and quality control of PRR.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material, further inquiries can be directed to the corresponding authors.

Ethics statement

The animal study was reviewed and approved by the Biomedical Ethical Committee of Peking University (approval no. LA2019117).

Author contributions

FX and S-QC designed the study. JZ (first author), YL, and JZ (third author) performed the experiment. X-YG, J-JX, P-PW, X-TC, and L-HX participated in animal experiments. JZ (first author) and W-JS analyzed the data. JZ (first author), FX, XW, and S-QC revised the manuscript.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fphar. 2022.995641/full#supplementary-material

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