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Synthesis of fused-ring systems and diarylmethane flavonoids *via ortho*-quinone methide intermediates†

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In this study, we investigated the ability of *ortho*-quinone methide (*o*-QM), an electron-poor diene, to undergo inverse electron-demand Diels–Alder (iEDDA) reaction with electron-rich dienophiles, resulting in fused-ring flavonoid systems. In addition, we explored the Michael-type addition using *o*-QM and various nucleophiles, providing access to diarylmethane products. The cycloaddition reactions proceeded in a highly regioselective way, depending on the charge distribution of the reacting partners. Overall, the electron-rich dienophiles or nucleophiles reacted with the intermediate *o*-QM to afford products in reduced to moderate yields. Electron-poor dienophiles or nucleophiles, on the other hand, failed to react with *o*-QM.

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Introduction

Flavonoids, the most commonly studied phytoestrogens, are natural compounds distributed in plants and have been part of the human diet such as in fruits and vegetables.¹ Different strategies have been developed to synthesize flavonoids, typically involving 2'-hydroxychalcones **A** as the intermediates. The chalcones can in turn be produced from the Claisen–Schmidt reaction between substituted acetophenone **B** and aromatic aldehyde **C** under basic or acidic conditions (Scheme 1a),² the Friedel–Crafts reaction, which involves the treatment of phenols **D** with cinnamoyl chloride **E** (Scheme 1b),^{2,3} and the Heck coupling reaction, which combines aryl vinyl ketone **F** and iodobenzene **G** (Scheme 1c).^{2–4} As the building blocks, the 2'-hydroxychalcones **A** can be transformed either into flavanones **H** *via* intramolecular cyclization under basic⁵ or acidic conditions⁶ or flavones **I** through oxidative cyclization³ (Scheme 1). The oxidative rearrangement of chalcones also can lead to the formation of isoflavones.⁷

Many naturally-occurring flavonoids contain fused-ring systems. For example, 4'-hydroxyisorottlerin and isorottlerin isolated from the kamala tree *Mallotus philippensis* possess a fused dimethyl-2H-pyran core.⁸ Hematoxylin, one of the main active components of *Hematoxylum campechianum* and

Caesalpinia sappan, contains a 3,4-cycloalkyl fused benzopyran unit. This compound has been shown to demonstrate anti-inflammatory activity and inhibit human immunodeficiency virus 1 (HIV-1) integrase.⁹

Many ring systems can be formed *via* Diels–Alder reactions of a highly reactive intermediate known as *ortho*-quinone methide (*o*-QM). Several natural products (Fig. 1) such as dependensin **1**,¹⁰ tocopherol **2**,¹¹ and hyperguinone **B** **3**,¹² were synthesized *via* cycloaddition of this intermediate.

Our group in particular has also utilized the *o*-QM intermediate for the synthesis of rottlerin **4**.¹³ The Mannich reaction has been reported as a viable method to generate *o*-QMs intermediates for synthesizing fused-ring systems starting from phenolic structures such as phenol,¹⁴ *p*-cresol,¹⁵ naphthol,¹⁶ and isoflavonoid structures.¹⁷ However, this methodology has not been previously reported on a flavanone system. There are very few reports on the synthesis of fused-ring flavanone analogues,^{18,19} however, the *o*-QM strategy can be adapted for the synthesis of complex scaffolds based on flavanones. Increasing the structural complexity of flavonoid structures can enhance their specificity for biological targets.^{20,21} The prevalence and significant biological roles of fused-ring flavonoids in nature provide a strong rationale for the synthesis and evaluation of a broader range of previously unexplored fused-ring analogues.^{8,9} Furthermore, fused-ring structures are more likely to have improved solubility, lipophilicity and selectivity towards target receptors.^{20–22}

o-QMs have been widely utilized for the construction of benzopyrans,²³ chromenes,²⁴ coumarins,²⁵ and xanthenes,²⁶ featured in many natural products. The reactivity of *o*-QMs can be explained by their resonance forms, where the neutral molecules are nonaromatic and resemble an α,β -unsaturated

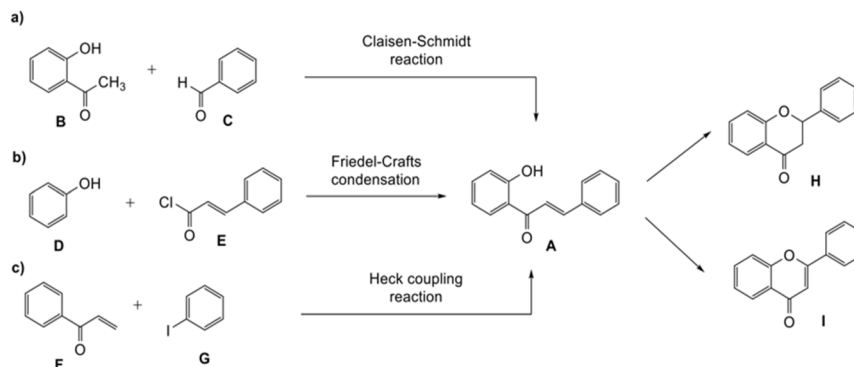
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Scheme 1 Synthetic approaches toward 2'-hydroxychalcones via Claisen–Schmidt reaction (a), Friedel–Crafts condensation (b), and Heck coupling reaction (c), as the precursors for flavonoid-class compounds.

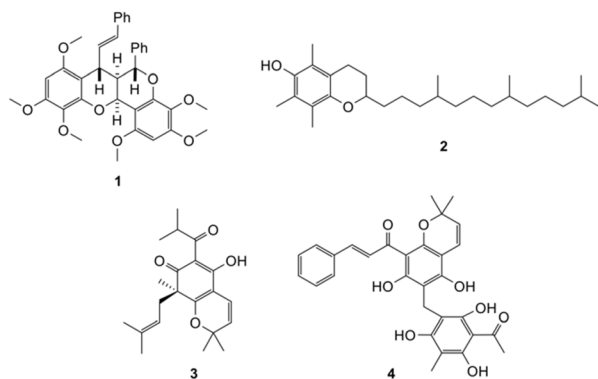


Fig. 1 Selected examples of natural products synthesized from *o*-QM intermediate.

ketone, while the zwitterionic resonance structures are aromatic and highly polarized.^{27,28} Therefore, *o*-QMs have two reactive sites, and their ambiphilic nature makes them very useful in many organic syntheses.^{24,29} *o*-QMs are considered to be electron-poor dienes that can react with electron-rich dienophiles following an inverse electron-demand Diels–Alder mechanism.^{30,31} Besides reacting with 2π partners *via* cycloaddition, *o*-QMs can also react with nucleophiles through Michael addition, or alternatively undergo oxa-6 π electrocyclization.^{27,28,32} In all three pathways, aromaticity is restored, providing a driving force for the reaction.²³ In this study, we discuss the synthesis of fused-ring systems that have never been explored on 6-hydroxyflavanone **5**, 7-hydroxyflavanone **6**, dihydrodaidzein **7** as an attempt to replace the free hydroxyl groups *via o*-QM intermediates. We also investigated the formation of diarylmethane products as a result of nucleophilic addition to the *o*-QM derived from isoflavanone, dihydrodaidzein **7**.

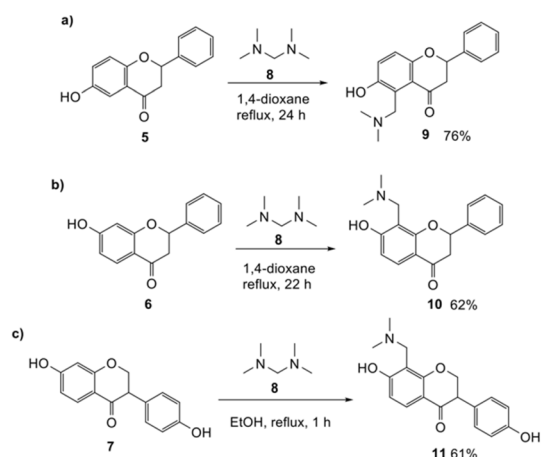
Results and discussion

Synthesis of Mannich base

Our approach for generating an *o*-QM was to use thermally-labile groups, such as the dimethylamine group, *via* the formation of Mannich bases. 6-Hydroxyflavanone **5** was refluxed

with a highly reactive aminal electrophile, (bis(*N,N*-dimethylamino)methane) **8**, in 1,4-dioxane for 24 h to give regioselectively the C5 *N,N*-dimethylaminomethyl-substituted flavanone **9** in 76% yield (Scheme 2a). Although there are two possible reactive sites in the 6-hydroxyflavanone **5**, including at C5 and C7 positions, the Mannich reaction took place exclusively at the C5 position. The structure of the Mannich base **9** was unambiguously established from the ¹H NMR spectrum, which showed the presence of two doublet signals at δ_{H} 7.02 ppm (d, *J* = 8.8 Hz, H7) and δ_{H} 6.92 ppm (d, *J* = 8.8 Hz, H8).

The use of aminals in the Mannich reaction has several advantages, including shorter reaction times and reduced side product formation.^{33,34} Additionally, formaldehyde is not required, unlike in conventional Mannich reactions.³⁴ In the case of 7-hydroxyflavanone **6**, the resultant product was confirmed as C8-substituted aminomethyl analogue **10** (Scheme 2b). This result was in line with the Mannich reaction of 7-hydroxyisoflavone, which also led to the corresponding C8-substituted Mannich base.¹⁷ For dihydrodaidzein **7**, the Mannich base was prepared by refluxing dihydrodaidzein **7** with aminal **8** in ethanol to give C8 substituted analogue **11** (Scheme 2c). The desired product **11** precipitated cleanly from the



Scheme 2 Regioselective Mannich reaction of 6-hydroxyflavanone **5** (a), 7-hydroxyflavanone **6** (b), and dihydrodaidzein **7** (c).

reaction mixture and was isolated as a white solid in moderate yield of 61%. Upon heating, these Mannich bases are proposed to form the *o*-QM intermediates *via* the loss of the dimethylamine moiety.

Diels–Alder reactions of *o*-QMs derived from 6-hydroxyflavanone

Our group previously worked on the *o*-QM reaction to synthesize fused-ring systems derived from phenoxodiol. The reaction was initially carried out by refluxing the Mannich base of phenoxodiol with ethyl vinyl ether as the dienophile in acetonitrile for 24 h. However, the reaction failed to give the expected product, and only the starting material was recovered. Changing the solvent to ethanol was also unsuccessful. While using the high boiling solvent, DMF did not yield a better outcome as the reaction mainly produced phenoxodiol and other degradation products. The reaction was then attempted by heating the Mannich base and ethyl vinyl ether in DMF at 140 °C using a pressure tube and an inert gas atmosphere for 24 h, resulting in the desired product in a low yield of 2%. Prolonged heating to 48 h increased the yield to 26%. Further attempts to improve the yield by extending the reaction time to 72 h did not give a better result. Hence, following this work, the cycloaddition reaction of 6-hydroxyflavanone was attempted by heating the mixture in DMF at a high temperature using a pressure vial under nitrogen gas.

Heating a solution of Mannich base **9** in DMF using a sealed tube at 155 °C with excess dienophiles under a nitrogen atmosphere afforded the cycloadducts containing a tetrahydropyran core (Scheme 3) with yields varied depending on the dienophile reactivity (Table 1). Different categories of dienophiles were investigated including enol ethers, conjugated diene, and enaminoketones. It was observed that all ethers (cyclic and acyclic) react in a regiospecific manner, forming acetal-type products **13a–c** (Table 1, entries 1–3) confirmed by the resonance of a downfield acetal proton at δ_{H} 5.25 ppm. The regioselectivity of Diels–Alder reactions can be explained based on the charge distribution of the reacting partners.^{35,36} The presence of the electron-donating oxygen atom in ether dienophiles increases the nucleophilicity of the β -carbon, thereby attacking the electrophilic methide carbon of *o*-QM **12**. Therefore, the cycloadducts **13a–c** were exclusively formed.

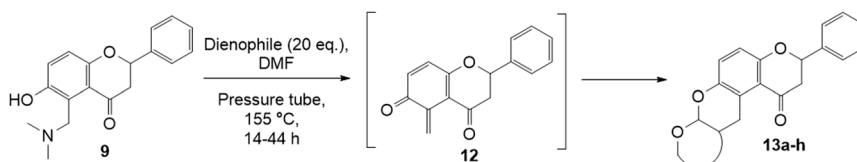
The fused-ring system **13a** adopted a *cis* configuration, which was concluded from the value of the coupling constant (*J*) between the acetal proton H9a (δ_{H} 5.25 ppm, d, *J* = 2.6 Hz) and the bridgehead proton H5a. Additionally, an important NOE correlation was observed between these two protons. The

relative stereochemistry of this adduct was in agreement with similar isoflavone cycloadducts as reported by Frasinyuk *et al.*¹⁷ As the Diels–Alder reactions enable the formation of new stereogenic centers, it is unavoidable to obtain diastereomeric mixtures unless the stereoselectivity is controlled. The diastereoisomerism caused the resonance of additional peaks for most of the carbon atoms in the ¹³C NMR spectrum of cycloadducts. These extra chemical shifts are shown in brackets in the ¹³C NMR data.

The reaction of *o*-QM **12** with *p*-methoxystyrene (36%), as expected, gave a higher yield of the cycloadduct than with styrene (25%) (Table 1, entries 4 and 5). This result indicated that the electron-rich dienophile favoured iEDDA reaction since the electron-donating substituent could raise the HOMO of the dienophile and thus reduce the activation energy.³⁷ The *o*-QM **12** was also treated with α -methyl styrene and 1,3-diisopropenylbenzene, providing cycloadducts **13f** (12%) and **13g** (21%), respectively (Table 1, entries 6 and 7). The low yields of these two products could be due to the low nucleophilicity of the dienophiles, as both styrenes lack electron-donating groups. With a more electron-rich dienophile, 3-(dimethylamino)-5,5-dimethylcyclohex-2-en-1-one, the reaction with *o*-QM **12** generated the cycloadduct **13h** in a higher yield of 79% (Table 1, entry 8). The reactivity of this dienophile is attributed to its ambiphilic properties, which combine the nucleophilicity of enamine with the electrophilicity of enone, leading to a push-pull nature of the double bond.²⁴ The β -carbon atom attached to the electron-withdrawing group, carbonyl group in this case, exhibits considerable nucleophilicity, reversely, the α -carbon atom displays considerable electrophilicity.³⁸ Hence, the β -carbon will attack the more electrophilic methide carbon of the *o*-QM **12**, favouring the formation of the hemi-aminal intermediate. Subsequent elimination of dimethylamine results in product **13h**. Attempts to use electron-poor benzofuran as the dienophile in the Diels–Alder reaction with *o*-QM **12**, as expected, failed to give the anticipated product **13i**, even after prolonged heating (Table 1, entry 9). This could be due to the inertness of benzofuran because of the high aromaticity, thus requiring a high energy barrier to dearomatize the five-membered ring.^{39,40}

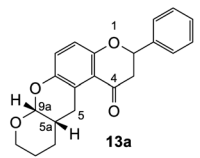
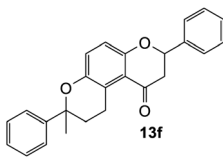
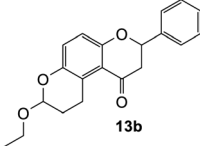
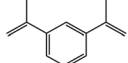
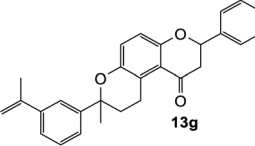
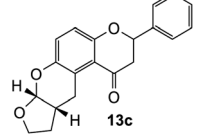
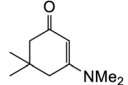
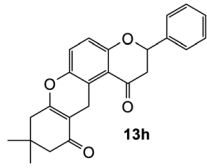
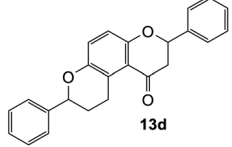
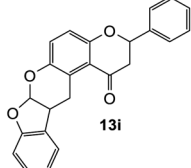
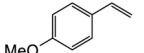
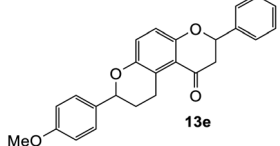
Diels–Alder reactions of *o*-QMs derived from 7-hydroxyflavanone

The Diels–Alder reaction of the 7-hydroxy Mannich base **10** proceeded through the *o*-QM **14** generated *in situ*, using similar conditions as described for the corresponding 6-substituted analogue **9** (Scheme 4). However, the reaction proved quite



Scheme 3 Cycloaddition reactions of Mannich base **9**.

Table 1 Cycloaddition adducts derived from 6-hydroxyflavanone 5

Entry	Dienophile	Time	Product	Yield [%]	Entry	Dienophile	Time	Product	Yield [%]
1		23 h		12	6		23 h		12
2		23 h		30	7		14 h		21
3		17 h		21	8		21 h		79
4		44 h		25	9		51 h		0
5		24 h		36					

challenging, as the flavanone core readily underwent ring-opening to form the chalcone. Hence, the chalcone cycloadducts **16a–e** were isolated in addition to the flavanone cycloadducts **15a–e** in low yields (Table 2). In the literature, Diels–Alder reactions on similar ring structures, such as dihydropyran and dihydrochromene favoured the ring-opened product, but could also induce ring-opening using PTSA as a catalyst.^{41–43} The formation of chalcone analogues was indicated by the appearance of two doublets at around δ_{H} 7.90 and 7.99 ppm with a large J value of 15.5 Hz, corresponding to the olefinic protons H_α and H_β , respectively. A sharp singlet further downfield at δ_{H} 13.99 ppm indicating a hydroxy group provided further evidence that the chalcone structure had formed. It was hypothesized that the dimethylamine group, which was

eliminated during the formation of the *o*-QM intermediate, facilitated the ring-opening reaction of the flavanone moiety.

However, this ring-opening reaction did not occur in 6-hydroxy analogue. The intermediate in the ring-opening reaction involves the formation of phenoxide ion. In 6-hydroxy analogues, the *para*-hydroxy group destabilizes the phenoxide ion by the electron donation, making this a high-energy intermediate. Meanwhile, in the 7-hydroxy analogues, the hydroxy group is in the *meta* position to the phenoxide ion thereby avoiding such destabilization. Hence, the ring-opening reaction is more likely to occur in the 7-hydroxy analogues.

As previously mentioned, the *o*-QM intermediate can undergo Michael-addition pathway with nucleophiles. Hence, the Michael addition reaction of the Mannich base **10** was

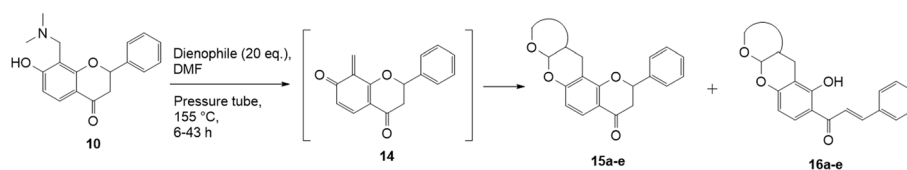
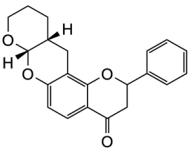
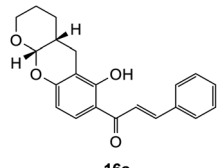
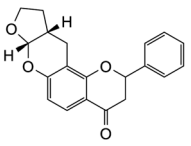
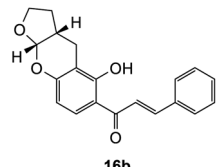
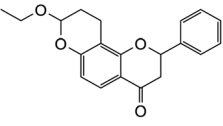
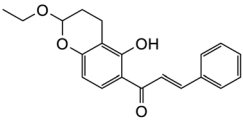
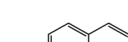
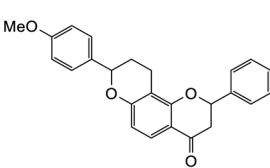
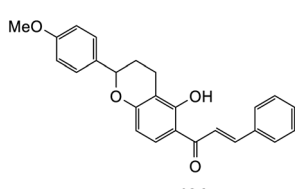
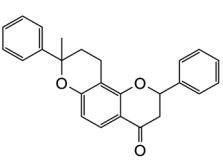
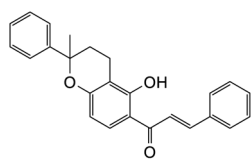
Scheme 4 Cycloaddition reactions of Mannich base **10** with various dienophiles.

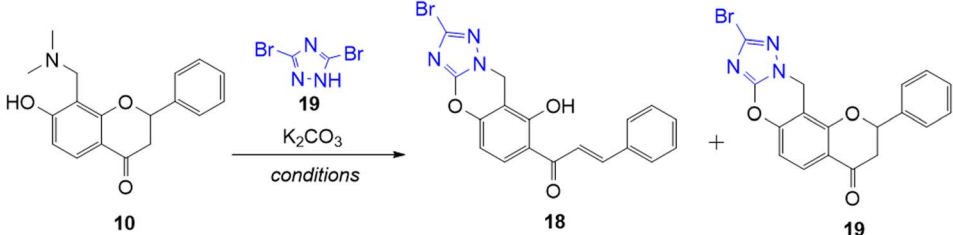
Table 2 Cycloaddition adducts derived from 7-hydroxyflavanone 6

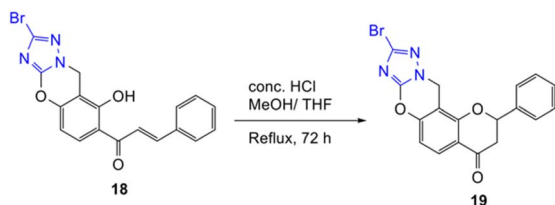
Entry	Dienophile	Time	Flavanone cycloadducts	Yield [%]	Chalcone cycloadducts	Yield [%]
1		43 h	 15a	12	 16a	8
2		43 h	 15b	6	 16b	5
3		6 h	 15c	7	 16c	5
4		6 h	 15d	7	 16d	8
5		16 h	 15e	4	 16e	7

attempted using 3,5-dibromo-1,2,4-triazole **17**. This triazole is an interesting nitrogen-based nucleophile because it allows the formation of fused nitrogen-containing heterocycles *via* aza-Michael addition reactions. Aza-Michael addition involves

a nucleophile, specifically an amine such as triazole in this case as a Michael-donor and an electron-poor diene such as *o*-QM as a Michael-acceptor, providing a useful method for constructing new C–N bond.⁴⁴

Table 3 Conditions for the reaction between the Mannich base **10** and 3,5-dibromo-1,2,4-triazole **17**

			
Entry	Conditions	18 [% yield]	19 [% yield]
1	Dienophile (1 eq.), DMF, K ₂ CO ₃ (3 eq.), pressure tube, 150 °C, 4.5 h	8	Not isolated
2	Dienophile (1 eq.), DMA, K ₂ CO ₃ (3 eq.), pressure tube, 150 °C, 4.5 h	58	8
3	Dienophile (1 eq.), NMP, K ₂ CO ₃ (3 eq.), pressure tube, 150 °C, 3.5 h	21	3



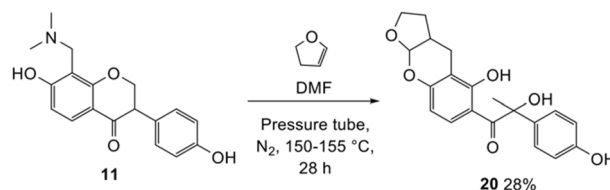
Scheme 5 Acid-catalyzed ring-closing reaction of chalcone **18**.

Initial attempts to obtain a successful Michael addition reaction between triazole **17** and *o*-QM precursor **14** were carried out by heating the reactants together in DMF in the presence of K_2CO_3 in a sealed tube for 4.5 h. The reaction led to complete consumption of the *o*-QM precursor **14**, giving the ring-opened chalcone product **18** a low yield of 8%, while the anticipated ring-closed flavanone **19** product was not observed (Table 3, entry 1). Changing the solvent to *N,N*-dimethylacetamide (DMA) led to the isolation of the desired product **19** in 8% yield, together with the ring-opened analogue **18** in 58% yield (Table 3, entry 2). The use of a slightly less polar solvent, *N*-methylpyrrolidine (NMP) did not result in a better outcome as the ring-opened product **18** was formed in a higher yield of 21%, while only producing 3% yield of the desired product **19** (Table 3, entry 3). The cyclization of chalcone product **18** was then affected by treating it with concentrated HCl in MeOH/THF under reflux for 72 h to afford the flavanone product **19**, which was purified by preparative HPLC (Scheme 5). However, the conversion was not complete and the reaction proceeded slowly, giving the flavanone product **19** in only 10%. The mechanism of the reaction is postulated to involve an aza-Michael addition reaction similar to that reported by Osipov *et al.*⁴⁵ The alkylation takes place at the more nucleophilic $N^{1(2)}$ position rather than at N^4 because of the presence of an adjacent atom with a lone pair of electrons (the α -effect). K_2CO_3 was needed to deprotonate the triazole, thereby facilitating attack of $N^{1(2)}$ at the electrophilic β -carbon of *o*-QM **14**.

Regeneration of the aromaticity of the triazole moiety is believed to be the driving force of this reaction.

Synthesis of fused-ring isoflavanones and diarylmethane analogues

The synthesis of fused-ring systems derived from dihydrodaidzein was first attempted by following the same method as previously described for the flavanones. Heating the Mannich base **11** with 2,3-dihydrofuran in DMF afforded a product in 28% yield. Surprisingly, 1H NMR analysis of the product revealed the disappearance of the characteristic oxygenated methylene group of isoflavanone moiety at *ca.* δ_H 4.56 ppm and the methine proton at *ca.* δ_H 3.87 ppm. An additional methyl peak was observed instead at δ_H 1.69 ppm. The ^{13}C NMR spectrum showed peaks for an oxygenated quaternary carbon at δ_C 90.4 ppm (C2) and a methyl group at δ_C 23.3 ppm (C3). Subsequently, the 2D HMBC analysis revealed that the methyl group was attached to a quaternary carbon (C2), and had three-bond length correlations with a carbonyl group at δ_C 198.9 ppm (C1). While the resonance of a doublet at δ_H 5.71 ppm ($J = 4.1$



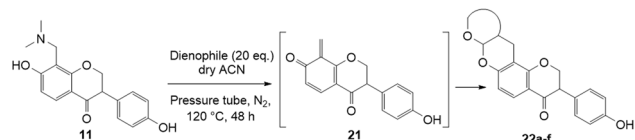
Scheme 6 Formation of ring-opened analogue **20** through Diels-Alder reaction of Mannich base **11** with 2,3-dihydrofuran.

Hz) corresponding to the acetal proton H6a and a multiplet at δ_H 2.69–2.77 ppm indicating a bridgehead proton H9a, confirmed that the cycloaddition reaction had taken place. Given these NMR data, the product was identified as the ring-opened analogue **20** (Scheme 6). In addition, HRMS identified a molecular mass of 357.1328, consistent with the protonated adduct of compound **20**, with an expected mass of 357.1333.

Using the same condition, the cycloaddition reactions were further repeated for other dienophiles, such as 3,4-dihydro-2*H*-pyran, ethyl vinyl ether, and *p*-methoxystyrene. However, the reactions were unsuccessful, causing the formation of degradation products and the 1H NMR spectra could not be identified. Subsequent cycloaddition reaction of Mannich base with 2,3-dihydrofuran was attempted by heating the mixture at 120 °C in a pressure tube using dry acetonitrile as the solvent under the nitrogen atmosphere. These conditions led to the isolation of the desired product **22a** in 18% yield after column chromatography.

Several test reactions were performed to explore the conditions required to form the ring-opened product. First, the Mannich base **11** was heated alone in DMF at a high temperature of 150–155 °C, which resulted in the cleavage of the dimethylamine group, forming the starting dihydrodaidzein **7** along with degradation products. This indicated that thermolysis of Mannich base **11**, but not ring-opening, occurred under these conditions. In the second attempt, the Mannich base **11** was treated with 2,3-dihydrofuran in dry acetonitrile at 150–155 °C for 22 h. This reaction yielded both the expected product **22a** and the ring-opened analogue **20**, in a ratio of 11:1. Lowering the temperature to 120 °C and extending the reaction time to 50 h did not alter the outcome. In the final trial, the cyclized product **22a** was heated alone in dry acetonitrile at 120 °C and monitored by TLC. At 48 h, the compound remained unchanged. However, after 69 h of heating, the adduct **22a** began to degrade, as evidenced by the appearance of several spots on the TLC plate. Notably, the ring-opened product did not form during this trial. Given these observations, it can be concluded that the ring-opening reaction only occurred when both Mannich base **11** and dienophile, in this case 2,3-dihydrofuran, were heated together. The ring-opening of the isoflavanone core is postulated to involve a retro-Michael-type ring-opening to give the propenone intermediate under basic conditions.⁴⁶ Subsequent nucleophilic addition by a water molecule generates the ring-opened product **20**.

Hence, the cycloaddition reactions of Mannich base **11** were carried out in dry acetonitrile at a lower temperature of 120 °C (Scheme 7), providing the desired cycloadducts **22a–f** in yields



Scheme 7 Formation of cycloadducts 22a–f from Mannich base 11.

varying between 10 and 33% (Table 4). Similar to cycloaddition adducts derived from flavanones, the Diels–Alder reactions between the Mannich base **11** and vinyl ether-type dienophiles regioselectively provided the acetal-type products (Table 4, entries 1–3). With *p*-methoxystyrene (Table 4, entry 4), the reaction also proceeded in high regioselectivity, where the resulting product bore the phenyl ring at C8 rather than C9. These results again suggested that charge distribution of the reacting partners determined the regioselectivity of the reactions.

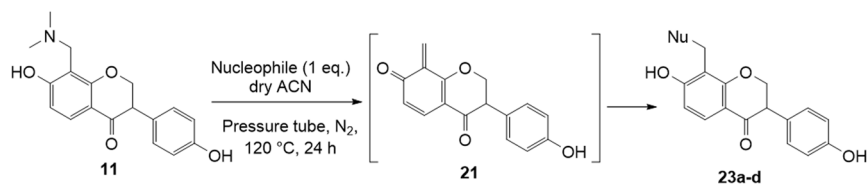
To further expand the substrate scope, silane-containing dienophiles were explored. Using the same conditions as depicted in Scheme 7, Mannich base **11** was treated with

(cyclohex-1-en-1-yloxy)trimethylsilane (Table 4, entry 5), providing the fused-ring system containing trimethylsiloxy group **22e**. However, with (cyclopent-1-en-1-yloxy)trimethylsilane (Table 4, entry 6), the cycloadduct **22f** obtained was suggested to undergo hydrolysis giving a hydroxyl group instead of the trimethylsiloxy group. The use of allyltrimethylsilane, ethynyltrimethylsilane and phenylacetylene as dienophiles failed to generate the desired cycloadducts **22g**, **22h**, and **22i** respectively (Table 4, entries 7–9) and only the starting Mannich base **11** was isolated from the reaction mixtures. The lack of nucleophilicity of these dienophiles again contributed to such observations, where the Diels–Alder reactions could not be facilitated.

We also investigated the reactivity of *o*-QM intermediates **21** derived from Mannich base **11** towards the Michael addition using various nucleophiles, providing diarylmethane products **23a–d** (Scheme 8). A key feature in the ^1H NMR spectra of the Michael adducts is the signal from the methylene bridge connecting C8 of the isoflavanone core with the nucleophilic moiety, which appeared as a singlet at around δ_{H} 3.70 ppm.

Table 4 Cycloaddition adducts derived from Mannich base 11

Entry	Dienophile	Time	Product	Yield [%]	Entry	Dienophile	Time	Product	Yield [%]
1		48 h		18	6		48 h		25
2		48 h		13	7		48 h		0
3		48 h		10	8		48 h		0
4		48 h		33	9		48 h		0
5		48 h		27					



Scheme 8 Michael addition of *in situ* generated *o*-QM **21** derived from Mannich base **11**.

The reaction between *o*-QM **20** and 3,5-dimethoxyphenol led to Michael adduct **23a**, where the 3,5-dimethoxyphenol reacted at the position between the two methoxy groups (Table 5, entry 1). Different regioselectivity was observed when the Mannich base **11** was treated with dimethoxyaniline (Table 5, entry 2). The ^1H NMR spectrum of the product suggests that the dimethoxyaniline moiety is unsymmetrical. However, another possible regioisomer, where the methylene bridge attached to the more sterically hindered position of the dimethoxyaniline moiety, was not observed. When thiophenol was used as the nucleophile, a thioether linkage was formed instead of the methylene bridge (Table 5, entry 3). As confirmed by the ^1H NMR spectrum, the methylene protons of the thioether linkage resonated at a more downfield region of δ_{H} 4.24 ppm compared with a typical methylene bridge (*ca.* δ_{H} 3.70 ppm). The thiol-Michael addition reaction is hypothesized to be catalyzed by dimethylamine, which was produced as a by-product during the *o*-QM formation. The amine deprotonates the thiol to generate the thiolate anion, a stronger nucleophile than the π electrons in the benzene ring that attacks the *o*-QM **20**. However, the use

of 4-chlorothiophenol generated the corresponding Michael adduct in lower yield compared to thiophenol (Table 5, entry 4). This was attributed to the presence of the electron-withdrawing chlorine substituent in the nucleophile, further enforcing the importance of using electron-rich nucleophiles for such reactions.

Further attempts using phloroglucinol and resorcinol had failed to generate the desired products **23e–f** respectively (Table 5, entries 5 and 6) as they are non-activated as compared to dimethoxyphenol and dimethoxyaniline even after prolonged heating or drying the reagent before use. In the two cases, dihydroadizzein **7** was isolated instead. This indicates that electron-rich nucleophiles are required for the Michael addition reaction to the *o*-QM intermediate. The [4 + 2] cycloaddition and Michael addition reaction proceeded with electron-rich alkenes and nucleophiles, which led to low overall yields for most *o*-QMs adducts. A possible reason for the low yields could be due to self-dimerization between the *o*-QM intermediate, leading to reduced efficiency of this reaction.⁴⁷ Overall, these methods pave the way for accessing fused-ring and diarylmethane

Table 5 Michael adducts derived from Mannich base **11**

Entry	Nucleophile	Product	Yield [%]	Entry	Nucleophile	Product	Yield [%]
1			8	4			16
2			17	5			0
3			41	6			0

Table 6 Anticancer activity of some cycloadducts against MDA-MB-231 triple negative breast cancer cell lines. Errors indicate SEM, $n = 3$

Compound	IC ₅₀ /μM
5	42.9 ± 1.7
13a	>100
13b	>100
13c	>100
6	>100
15a	66.5 ± 2.8
15b	43.2 ± 1.8
15c	88.6 ± 1.5
7	>100
22a	>100 (but ~100)
22b	>100
22c	>100
22d	35.1 ± 0.7

products from flavanones and isoflavanones that are not synthesized in nature, which might show interesting biological activity.

To assess the anti-proliferative properties, several cycloadditions adduct were tested *in vitro* against MDA-MB-231 triple-negative breast cancer cell lines using Alamar Blue assay. As shown in Table 6, the introduction of fused-ring system on 6-hydroxyflavanone **5** caused lower anti-proliferative activities compared to the parental compound. On the other hand, cycloadducts derived from 7-hydroxyflavanone **6** displayed slightly improved anti-proliferative activity against MDA-MB-231 breast cancer cell lines, though it was very weak. For cycloadducts synthesized from dihydrodaidzein **7**, only compound **22d** bearing 8-(*p*-methoxyphenyl) exhibited improved anti-proliferative activity compared to dihydrodaidzein **7** with the IC₅₀ value of 35.1 ± 0.7 μM. These findings pave the way for incorporating other dienophiles, which might lead to enhanced biological activity. Additional studies are also needed to further assess the anti-proliferative activity of Michael adducts.

Experimental

General procedures

All reactions requiring anhydrous conditions were performed in oven dried glassware under a nitrogen atmosphere unless otherwise stated. All the reagents were purchased from commercial sources and used without further purification.

Reactions were monitored using thin layer chromatography (TLC) using aluminum plates coated with silica gel GF₂₅₄. Compounds were detected by short and long wave-length ultraviolet light. Gravity column chromatography was carried out using 40–63 micron silica gel. Preparative reversed phase HPLC was performed using a PDA detector (254 nm) and C18 column (150 mm x 10 nm) on a gradient elution of 2–100% over 56 min with a flow rate of 5 mL min^{−1}. 0.1% Formic acid/MilliQ water was used as eluent A and 0.1% formic acid/acetonitrile was used as eluent B. ¹H NMR spectral data are reported as follows: chemical shifts are in parts per million (ppm) and internally referenced relative to the solvent nuclei (δ);

multiplicity; observed coupling constant (*J*) in hertz (Hz); proton count; as signment. Multiplicities are assigned as singlet (s), doublet (d), doublet of doublets (dd), doublet of triplets (dt), triplet, (t), doublet of doublet of doublets (ddd), triplet of doublets (td), multiplet (m), and broad singlet (br s) where appropriate. ¹³C NMR spectra were recorded in the designated solvents, and chemical shifts are reported in ppm internally referenced relative to the solvent nuclei. Melting points were measured using a Mel-Temp melting point apparatus and are uncorrected. Infrared spectra were recorded using an attenuated total reflection FTIR Spectrometer. High-resolution mass spectra were measured at 70 eV using a quadrupole analyzer and are reported with ion mass/charge (*m/z*) ratios as values in atomic mass units. All procedure involving use of a pressure tube at high temperature were performed following safety guidelines from the manufacturer.

Synthesis

General procedure 1 for the synthesis of Diels–Alder adducts from flavanones. To a solution of Mannich base **9** or **10** in DMF (10 mL) was added the appropriate dienophile (20 eq.) and the mixture was heated at 155 °C in a pressure tube held under a nitrogen atmosphere for 6–44 h. The solvent was evaporated *in vacuo*, and the crude product was subjected to column chromatography over silica gel G₆₀ (230–400 mesh) using *n*-hexane:EtOAc (95:5) as the eluent to afford the corresponding products. Further purification was performed using preparative reversed phase HPLC when required.

5-((Dimethylamino)methyl)-6-hydroxy-2-phenylchroman-4-one (9). To a solution of **5** (0.75 g, 3.13 mmol, 1 eq.) in 1,4-dioxane (30 mL) was added bis(*N,N*-dimethylamino)methane (0.85 mL, 6.25 mmol, 2.0 eq.). The mixture was heated to reflux for 24 h. After completion of reaction, the solvent was evaporated and the resulting crude product was chromatographed over silica gel (230–400 mesh) using DCM:MeOH (95:5) as eluent to afford **9** as an orange solid (0.71 g, 76%). M.p. 109–111 °C; IR (neat): $\nu_{\max}$ 3356, 2920, 1674, 1460, 1274, 1004, 903, 825, 759, 698 cm^{−1}; UV (MeOH): $\lambda_{\max}$ 348 nm (ϵ 56 259 cm^{−1} M^{−1}), 308 (9456), 230 (51 382); ¹H NMR (400 MHz, CDCl₃): δ 7.37–7.48 (m, 5H, H2', H3', H4', H5', H6'), 7.02 (d, *J* = 8.8 Hz, 1H, H7), 6.92 (d, *J* = 8.8 Hz, 1H, H8), 5.38 (dd, *J* = 2.8, 13.7 Hz, 1H, H2), 4.37 (d, *J* = 3.3 Hz, 2H, N-CH₂), 3.06 (dd, *J* = 13.7, 16.9 Hz, 1H, H3α), 2.84 (d, *J* = 2.8, 16.9 Hz, 1H, H3β), 2.38 (s, 6H, 2 × CH₃); ¹³C NMR (100 MHz, CDCl₃): δ 194.7 (C4), 156.2 (C8a), 154.5 (C6), 139.3 (C1'), 129.0 (C3', C5'), 128.8 (C4'), 126.3 (C2', C6'), 125.0 (C7), 121.5 (C4a), 119.5 (C5), 118.4 (C8), 79.0 (C2), 59.2 (CH₂-N), 46.8 (C3), 44.5 (2 × CH₃). HRMS (+ESI): (C₁₈H₁₉NO₃) calcd *m/z* 298.1438 [*M* + H]⁺, found *m/z* 298.1433.

2-Phenyl-2,3,10,11,11a,12-hexahydro-9H-dipyran[2,3-*b*:3',2'-*f'*]chromen-1(7aH)-one (13a). Flavanone **13a** was synthesized following general procedure 1 using the Mannich base **9** (207 mg, 0.70 mmol, 1 eq.) and 3,4-dihydro-2H-pyran (1.47 mL, 16.11 mmol, 23.0 eq.). The mixture was heated at 155 °C for 23 h. The product was obtained as a white solid (28 mg, 12%). M.p. 154–155 °C; IR (neat): $\nu_{\max}$ 2928, 1672, 1600, 1466, 1282, 1200, 1136, 1087, 999, 935, 831, 755, 699 cm^{−1}; UV

(MeOH): $\lambda_{\max}$ 348 nm (ϵ 47 430 cm⁻¹ M⁻¹), 308 (5719), 225 (17 156); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.56–7.58 (m, 2H, H2', H6'), 7.42–7.46 (m, 2H, H3', H5'), 7.36–7.40 (m, 1H, H4'), 7.05 (d, J = 8.7 Hz, 1H, H11), 6.91 (d, J = 8.7 Hz, 1H, H12), 5.51 (ddd, J = 3.1, 8.5, 13.2 Hz, 1H, H2), 5.25 (dd, J = 2.5, 12.8 Hz, 1H, H9a), 3.90–3.95 (m, 1H, H8 α), 3.63–3.67 (m, 1H, H8 β), 3.19–3.26 (m, 2H, H5), 3.09 (dd, J = 13.2, 16.4 Hz, 1H, H3 α), 2.81 (dd, J = 3.1, 16.4 Hz, 1H, H3 β), 2.13–2.17 (m, 1H, H5a), 1.61–1.70 (m, 2H, H7), 1.57–1.60 (m, 2H, H6); ¹³C NMR[‡] (100 MHz, acetone-*d*₆): δ 193.9 (193.8) (C4), 158.3 (158.2) (C12a), 148.5 (148.3) (C10a), 140.6 (140.5) (C1'), 129.5 (C3', C5'), 129.3 (129.2) (C4'), 127.2 (C2', C6'), 125.2 (125.1) (C11), 122.4 (122.3) (C4a'), 119.7 (119.5) (C4a), 118.1 (118.0) (C12), 96.9 (96.8) (C9a), 79.9 (79.8) (C2), 63.4 (62.9) (C8), 46.7 (46.6) (C3), 32.1 (C5a), 28.8 (29.4) (C5), 25.2 (25.4) (C6), 24.1 (24.4) (C7). HRMS (+ESI): (C₂₁H₂₀O₄) calcd m/z 359.1254 [M + Na]⁺, found m/z 359.1250.

7-Ethoxy-2-phenyl-2,3,9,10-tetrahydropyrano[3,2-*ff*]chromen-1(8*H*)-one (13*b*). Flavanone **13b** was synthesized following general procedure 1 using the Mannich base **9** (80 mg, 0.27 mmol, 1 eq.) and ethyl vinyl ether (0.52 mL, 5.43 mmol, 20.1 eq.). The mixture was heated at 155 °C for 23 h. The product was obtained as a yellow sticky solid (26 mg, 30%). IR (DCM): $\nu_{\max}$ 2920, 1679, 1467, 1318, 1267, 1200, 1062, 1010, 964, 823, 699 cm⁻¹. UV (MeOH): $\lambda_{\max}$ 343 nm (ϵ 11 743 cm⁻¹ M⁻¹), 308 (2206), 225 (2984); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.55–7.59 (m, 2H, H2', H6'), 7.38–7.47 (m, 3H, H3', H4', H5'), 7.04 (d, J = 8.9 Hz, 1H, H9), 6.89 (d, J = 8.9 Hz, 1H, H10), 5.49 (dd, J = 3.2, 13.1 Hz, 1H, H2), 5.25 (t, J = 2.6 Hz, 1H, H7), 3.77–3.83 (m, 1H, H1'' α), 3.60–3.66 (m, 1H, H1'' β), 3.25–3.36 (m, 1H, H5 α), 3.11–3.17 (m, 1H, H5 β), 3.09 (dd, J = 13.1, 16.1 Hz, 1H, H3 α), 2.78 (dd, J = 2.9, 16.1 Hz, 1H, H3 β), 1.95–2.02 (m, 1H, H6 α), 1.77–1.89 (m, 1H, H6 β), 1.10–1.16 (m, 3H, 2''-CH₃); ¹³C NMR[‡] (100 MHz, acetone-*d*₆): δ 193.8 (193.9) (C4), 158.0 (158.2) (C10a), 147.4 (147.5) (C8a), 140.7 (C1'), 129.5 (C3', C5'), 129.3 (C4'), 127.2 (C2', C6'), 125.8 (125.7) (C9), 124.8 (125.0) (C4a'), 119.3 (119.6) (C4a), 117.8 (117.8) (C10), 97.2 (97.1) (C7), 79.8 (79.7) (C2), 64.2 (C1''), 46.7 (46.8) (C3), 26.9 (27.1) (C6), 20.8 (21.0) (C5), 15.6 (C2''). HRMS (+ESI): (C₂₀H₂₀O₄) calcd m/z 347.1254 [M + Na]⁺, found m/z 347.1252.

2-Phenyl-2,3,9,10,10a,11-hexahydrofuro[2,3-*b*]pyrano[3,2-*ff*]chromen-1(7*aH*)-one (13*c*). Flavanone **13c** was synthesized following general procedure 1 using the Mannich base **9** (100 mg, 0.34 mmol, 1 eq.) and 2,3-dihydrofuran (0.51 mL, 6.75 mmol, 19.8 eq.). The mixture was heated at 155 °C for 17 h. The product was obtained as colourless needle-like crystals (23 mg, 21%). M.p. 148–149 °C; IR (neat): $\nu_{\max}$ 2950, 1665, 1455, 1312, 1268, 1202, 1066, 909, 829, 754, 692 cm⁻¹. UV (THF): $\lambda_{\max}$ 348 nm (ϵ 2514 cm⁻¹ M⁻¹), 239 (4352); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.56–7.59 (m, 2H, H2', H6'), 7.42–7.47 (m, 2H, H3', H5'), 7.36–7.40 (m, 1H, H4'), 7.03 (dd, J = 2.2, 8.9 Hz, 1H, H10), 6.91 (dd, J = 3.4, 8.9 Hz, 1H, H11), 5.54 (dd, J = 2.3, 13.6 Hz, 1H, H2), 5.52 (dd, J = 2.0, 15.9 Hz, 1H, H8a), 3.81–3.93 (m, 2H, H7), 3.67–3.76 (m, 1H, H5 α), 3.14–3.20 (m, 1H, H5 β), 3.07 (dd, J = 13.1, 17.3 Hz, 1H, H3 α), 2.80 (dd, J = 2.7, 16.3 Hz, 1H, H3 β),

2.72–2.76 (m, 1H, H5a), 2.07–2.12 (m, 1H, H6 α), 1.49–1.61 (m, 1H, H6 β). ¹³C NMR[‡] (100 MHz, acetone-*d*₆): δ 194.1 (193.9) (C4), 158.3 (158.2) (C11a), 148.8 (C9a), 140.5 (140.4) (C1'), 129.4 (C3', C5'), 129.2 (C4'), 127.1 (C2', C6'), 125.6 (125.5) (C10), 124.3 (124.1) (C4a'), 119.9 (119.5) (C4a), 117.9 (117.8) (C11), 102.0 (101.8) (C8a), 79.8 (79.7) (C2), 68.3 (68.2) (C7), 46.7 (46.6) (C3), 38.1 (38.0) (C5a), 29.4 (C6), 25.2 (24.9) (C5). HRMS (+ESI): (C₂₀H₁₈O₄) calcd m/z 323.1278 [M + H]⁺, found m/z 323.1278.

2,7-Diphenyl-2,3,9,10-tetrahydropyrano[3,2-*ff*]chromen-1(8*H*)-one (13*d*). Flavanone **13d** was synthesized following general procedure 1 using the Mannich base **9** (150 mg, 0.50 mmol, 1 eq.) and styrene (1.16 mL, 10.12 mmol, 20.2 eq.). The mixture was heated at 155 °C for 44 h. The product was obtained as a white solid (45 mg, 25%). M.p. 159–160 °C; IR (neat): $\nu_{\max}$ 2918, 1677, 1600, 1468, 1325, 1268, 1199, 1162, 1007, 939, 835, 750, 695 cm⁻¹. UV (MeOH): $\lambda_{\max}$ 342 nm (ϵ 5845 cm⁻¹ M⁻¹), 307 (1283), 231 (642); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.56–7.59 (m, 2H, H2', H6'), 7.46–7.49 (m, 2H, H2'', H6''), 7.43–7.46 (m, 2H, H3', H5'), 7.38–7.42 (m, 3H, ArH), 7.30–7.35 (m, 1H, ArH), 7.12 (d, J = 8.9 Hz, 1H, H9), 6.93 (d, J = 8.9 Hz, 1H, H10), 5.54 (dd, J = 3.1, 13.1 Hz, 1H, H2), 5.06 (t, J = 2.4 Hz, 1H, H7), 3.18–3.38 (m, 2H, H5), 3.13 (dd, J = 13.2, 16.2 Hz, 1H, H3 α), 2.79 (dd, J = 2.8, 16.2 Hz, 1H, H3 β), 2.22–2.30 (m, 1H, H6 α), 1.91–2.03 (m, 1H, H6 β); ¹³C NMR[‡] (100 MHz, acetone-*d*₆): δ 193.9 (193.7) (C4), 158.2 (157.9) (C10a), 150.7 (150.6) (C8a), 142.9 (142.8) (C1''), 140.6 (C1'), 129.5 (C3', C5'), 129.3 (C3'', C5''), 129.2 (C4'), 128.6 (128.5) (C4''), 127.2 (C2', C6'), 126.9 (126.8) (C2'', C6''), 125.7 (C9), 124.1 (124.0) (C4a'), 119.8 (119.5) (C4a), 118.0 (117.9) (C10), 79.8 (C2), 77.8 (77.6) (C7), 46.8 (46.6) (C3), 30.6 (30.5) (C6), 25.9 (25.5) (C5). HRMS (+ESI): (C₂₄H₂₀O₃) calcd m/z 379.1305 [M + Na]⁺, found m/z 379.1303.

7-(4-Methoxyphenyl)-2-phenyl-2,3,9,10-tetrahydropyrano[3,2-*ff*]chromen-1(8*H*)-one (13*e*). Flavanone **13e** was synthesized following general procedure 1 using the Mannich base **9** (90 mg, 0.30 mmol, 1 eq.) and *p*-methoxy styrene (0.81 mL, 6.05 mmol, 20.2 eq.). The mixture was heated at 155 °C for 24 h. The product was obtained as a yellow solid (42 mg, 36%). M.p. 163–164 °C; IR (neat): $\nu_{\max}$ 2920, 1670, 1613, 1513, 1455, 1378, 1245, 1172, 1024, 940, 820, 768; UV (MeOH): $\lambda_{\max}$ 348 nm (ϵ 3207 cm⁻¹ M⁻¹), 263 (8850), 230 (32 887), 201 (33 157); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.57–7.59 (m, 2H, H2', H6'), 7.43–7.46 (m, 2H, H3', H5'), 7.38–7.41 (m, 3H, H4', H2'', H6''), 7.09 (dd, J = 3.2, 8.9 Hz, 1H, H9), 6.95 (dd, J = 2.1, 8.9 Hz, 2H, H3'', H5''), 6.92 (dd, J = 2.6, 8.9 Hz, 1H, H10), 5.54 (dd, J = 3.0, 13.1 Hz, 1H, H2), 5.00 (t, J = 2.5 Hz, 1H, H7), 3.81 (s, 3H, -OCH₃), 3.32–3.39 (m, 1H, H5 α), 3.18–3.28 (m, 1H, H5 β), 3.13 (dd, J = 13.1, 16.1 Hz, 1H, H3 α), 2.79 (dd, J = 3.0, 16.1 Hz, 1H, H3 β), 2.20–2.26 (m, 1H, H6 α), 1.93–2.02 (m, 1H, H6 β); ¹³C NMR[‡] (100 MHz, acetone-*d*₆): δ 193.9 (193.7) (C4), 160.4 (160.3) (C4''), 158.1 (157.9) (C10a), 150.9 (150.8) (C8a), 140.7 (140.6) (C1'), 134.8 (134.8) (C1''), 129.5 (C3', C5'), 129.3 (129.2) (C4'), 128.3 (128.2) (C2'', C6''), 127.2 (127.2) (C2', C6'), 125.7 (125.7) (C9), 124.1 (124.0) (C4a'), 119.9 (119.6) (C4a), 118.0 (117.9) (C10), 114.6 (C3'', C5''), 79.8 (79.8) (C2), 77.6 (77.4) (C7), 55.6 (OMe), 46.8 (46.6) (C3), 26.1 (25.6) (C6), 23.4 (C5). HRMS (+ESI): (C₂₅H₂₂O₄) calcd m/z 409.1410 [M + Na]⁺, found m/z 409.1406.

[‡] Additional peaks are due to diastereomers.

7-Methyl-2,7-diphenyl-2,3,9,10-tetrahydropyrano[3,2-f]chromen-1(8H)-one (13f). Flavanone **13f** was synthesized following general procedure 1 using the Mannich base **9** (60 mg, 0.20 mmol, 1 eq.) and α -methyl styrene (0.53 mL, 4.08 mmol, 20.4 eq.). The mixture was heated at 155 °C for 23 h. The product was obtained as a yellow oil (9 mg, 12%). IR (DCM): $\nu_{\max}$ 2922, 1679, 1466, 1372, 1271, 1161, 1069, 937, 823, 763, 699 cm^{-1} ; UV (THF): $\lambda_{\max}$ 353 nm (ϵ 7891 $\text{cm}^{-1} \text{M}^{-1}$); ^1H NMR (400 MHz, acetone- d_6): δ 7.53–7.56 (m, 2H, H2', H6'), 7.41–7.45 (m, 4H, ArH), 7.28–7.39 (m, 4H, ArH), 7.22–7.24 (m, 1H, ArH), 7.20–7.21 (m, 1H, ArH), 6.93 (dd, J = 3.0, 9.0 Hz, 1H, H10), 5.46 (dd, J = 3.1, 13.5 Hz, 1H, H2), 3.23–3.29 (m, 1H, H5 α), 3.02 (dd, J = 13.2, 16.1 Hz, 1H, H3 α), 2.72 (dd, J = 3.1, 16.1 Hz, 1H, H3 β), 2.68–2.72 (m, 1H, H5 β), 2.52–2.60 (m, 2H, H6), 1.59 (s, 3H, CH₃); ^{13}C NMR $\ddagger$ (100 MHz, acetone- d_6): δ 193.6 (193.8) (C4), 157.9 (157.7) (C10a), 149.4 (149.5) (C8a), 146.3 (146.2) (C1''), 140.6 (C1'), 129.4 (C3', C5'), 129.2 (C3'', C5''), 129.2 (C4'), 127.6 (C4''), 127.1 (C2', C6'), 125.9 (C9), 125.8 (C2'', C6''), 123.8 (C4a'), 119.4 (119.2) (C4a), 118.0 (117.9) (C10), 79.8 (79.6) (C2), 78.4 (78.2) (C7), 46.6 (46.5) (C3), 32.8 (32.6) (C6), 31.1 (CH₃), 23.4 (23.3) (C5); HRMS (+ESI): (C₂₅H₂₂O₃) calcd m/z 393.1461 [M + Na]⁺, found m/z 393.1460.

7-Methyl-2-phenyl-7-(3-(prop-1-en-2-yl)phenyl)-2,3,9,10-tetrahydropyrano[3,2-f]chromen-1(8H)-one (13g). Flavanone **13g** was synthesized following general procedure 1 using the Mannich base **9** (60 mg, 0.20 mmol, 1 eq.) and 1,3-diisopropenylbenzene (0.69 mL, 4.03 mmol, 20.1 eq.). The mixture was heated at 155 °C for 14 h. The product was obtained as a yellow oil (17 mg, 21%). IR (DCM): $\nu_{\max}$ 2970, 1675, 1597, 1461, 1371, 1327, 1266, 1159, 1075, 933, 891, 799 cm^{-1} ; UV (MeOH) $\lambda_{\max}$ 348 nm (ϵ 17 816 $\text{cm}^{-1} \text{M}^{-1}$), 308 (2997), 230 (2381); ^1H NMR (300 MHz, acetone- d_6): δ 7.58–7.59 (m, 1H, ArH), 7.52–7.56 (m, 2H, H2', H6'), 7.40–7.45 (m, 2H, ArH), 7.34–7.39 (m, 3H, ArH), 7.28–7.32 (m, 1H, ArH), 7.24 (dd, J = 2.0, 9.0 Hz, 1H, H9), 6.93 (dd, J = 2.2, 9.0 Hz, 1H, H10), 5.45 (dd, J = 2.8, 13.2 Hz, 1H, H2), 5.34–5.36 (m, 1H, =CH₂), 5.05–5.07 (m, 1H, =CH₂), 3.22–3.33 (m, 1H, H5 α), 3.02 (dd, J = 13.1, 16.2 Hz, H3 α), 2.69 (dd, J = 3.0, 16.2 Hz, 1H, H3 β), 2.55–2.69 (m, 3H, H5, H6), 2.09–2.11 (m, 3H, CH₃), 1.61 (s, 3H, CH₃); ^{13}C NMR $\ddagger$ (100 MHz, acetone- d_6): δ 192.8 (C4), 157.1 (156.8) (C10a), 148.6 (C8a), 145.3 (C1''), 143.3 (C=CH₂), 141.1 (C3''), 139.7 (C1'), 128.5 (2 $\times$ ArCH), 128.4 (2 $\times$ ArCH), 126.2 (2 $\times$ ArCH), 125.0 (ArCH), 124.3 (ArCH), 123.9 (ArCH), 123.0 (ArC), 122.1 (ArCH), 118.6 (C4a), 117.2 (C10), 112.0 (C=CH₂), 78.9 (78.7) (C2), 77.6 (77.4) (C7), 45.7 (C3), 31.6 (31.8) (C6), 30.3 (29.9) (CH₃), 22.5 (22.3) (C5), 21.0 (CH₃); HRMS (+ESI): (C₂₈H₂₆O₃) calcd m/z 433.1774 [M + Na]⁺, found m/z 433.1771.

8,8-Dimethyl-2-phenyl-2,3,8,9,10,12-hexahydropyrano[3,2-a]xanthene-1,11-dione (13h). Flavanone **13h** was synthesized following general procedure 1 using the Mannich base **9** (32 mg, 0.11 mmol, 1 eq.) and 3-(dimethylamino)-5,5-dimethylcyclohex-2-en-1-one (18 mg, 0.11 mmol, 1 eq.). The mixture was heated at 155 °C for 21 h. The product was obtained as a pale-yellow solid (32 mg, 79%). M.p. 238–239 °C; IR (neat): $\nu_{\max}$ 2955, 1683, 1634, 1597, 1466, 1389, 1308, 1221, 1146, 1018, 832, 758 cm^{-1} ; UV (THF) $\lambda_{\max}$ 348 nm (ϵ 22 167 $\text{cm}^{-1} \text{M}^{-1}$), 287 (95 482), 255 (80 729), 236 (83 275); ^1H NMR (400 MHz, acetone- d_6): δ 7.57–7.59

(d, J = 8.1 Hz, 2H, H2', H6'), 7.43–7.47 (m, 2H, H3', H5'), 7.38–7.41 (m, 1H, H4'), 7.24 (d, J = 8.9 Hz, 1H, H11), 7.01 (d, J = 8.9 Hz, 1H, H12), 5.59 (dd, J = 2.7, 13.0 Hz, 1H, H2), 3.87 (d, J = 6.6 Hz, 2H, H5), 3.14 (dd, J = 13.1, 16.4 Hz, 1H, H3 α), 2.86 (dd, J = 2.7, 16.4 Hz, H3 β), 2.45 (s, 2H, H9), 2.28 (s, 2H, H7), 1.12 (s, 6H, 8-CH₃); ^{13}C NMR (100 MHz, acetone- d_6): δ 196.7 (C6), 193.4 (C4), 164.3 (C9a), 160.2 (C12a), 145.4 (C10a), 140.2 (C1'), 129.5 (C3', C5'), 129.4 (C4'), 127.2 (C2', C6'), 124.8 (C11), 123.6 (C4a'), 119.2 (C4a), 118.5 (C12), 109.0 (C5a), 79.8 (C2), 51.1 (C7), 46.3 (C3), 41.4 (C9), 32.6 (C8), 28.4 (2 $\times$ CH₃), 22.4 (C5); HRMS (+ESI): (C₂₄H₂₂O₄) calcd m/z 397.1410 [M + Na]⁺, found m/z 397.1408.

8-((Dimethylamino)methyl)-7-hydroxy-2-phenylchroman-4-one (10). To a solution of **6** (1.24 g, 5.18 mmol, 1 eq.) in 1,4-dioxane (50 mL), was added bis(*N,N*-dimethylamino)methane (1.41 mL, 10.34 mmol, 2.0 eq.). The mixture was heated at reflux for 22 h. After completion of reaction, the solvent was evaporated. The resulting crude product was chromatographed over silica gel using DCM : MeOH (95 : 5) as eluent to give **10** as a yellow solid (0.96 g, 62%). M.p. 144–145 °C; IR (neat): $\nu_{\max}$ 3011, 2956, 1654, 1583, 1440, 1361, 1310, 1210, 1103, 1050, 999, 819, 751, 698 cm^{-1} ; UV (THF) $\lambda_{\max}$ 305 nm (ϵ 16 622 $\text{cm}^{-1} \text{M}^{-1}$), 279 (29 794); ^1H NMR (400 MHz, CDCl₃): δ 7.80 (d, J = 8.7 Hz, 1H, H5), 7.44–7.46 (m, 3H, H3', H4', H5'), 7.36–7.43 (m, 2H, H2, H6'), 6.53 (d, J = 8.7 Hz, 1H, H6), 5.44 (dd, J = 3, 13 Hz, 1H, H2), 3.81 (s, 2H, CH₂), 2.98 (dd, J = 13.0, 16.8 Hz, 1H, H3 α), 2.82 (dd, J = 3.0, 16.8 Hz, 1H, H3 β), 2.38 (s, 1H, 2 $\times$ CH₃); ^{13}C NMR (100 MHz, CDCl₃): δ 190.6 (C4), 166.9 (C8a), 160.5 (C7), 139.4 (C1'), 129.0 (C3', C5'), 128.8 (C5), 128.2 (C4'), 126.2 (C2', C6'), 113.6 (C4a), 111.8 (C6), 108.0 (C8), 79.9 (C2), 55.5 (CH₂-N), 44.6 (2 $\times$ CH₃), 44.5 (C3); HRMS (+ESI): (C₁₈H₁₉NO₃) calcd m/z 298.1438 [M + Na]⁺, found m/z 298.1433.

2-Phenyl-7-((tetrahydro-2H-pyran-2-yl)oxy)chroman-4-one (15a). Flavanone **15a** was synthesized following general procedure 1 using the Mannich base **10** (100 mg, 0.34 mmol, 1 eq.) and 3,4-dihydro-2H-pyran (0.61 mL, 6.73 mmol, 19.8 eq.). The mixture was heated at 155 °C for 43 h. The product was obtained as a white solid (20 mg, 12%). M.p. 102–103 °C; IR (neat): $\nu_{\max}$ 2921, 2851, 1683, 1601, 1437, 1325, 1252, 1138, 1089, 937, 898 cm^{-1} ; UV (THF) $\lambda_{\max}$ 306 nm (ϵ 26 440 $\text{cm}^{-1} \text{M}^{-1}$), 278 (55 101), 239 (46 825), 232 (42 251), 217 (86 923); ^1H NMR (400 MHz, acetone- d_6): δ 7.67 (dd, J = 2.2, 8.7 Hz, 1H, H5), 7.62 (d, J = 8.7 Hz, 2H, H2', H6'), 7.46–7.50 (m, 2H, H3', H5'), 7.39–7.43 (m, 1H, H4'), 6.57 (dd, J = 2.1, 8.7 Hz, H6), 5.64 (dd, J = 3.1, 12.9 Hz, 1H, H2), 5.39 (dd, J = 2.7, 4.7 Hz, 1H, H7a), 3.90–3.97 (m, 1H, H9 α), 3.67–3.72 (m, 1H, H9, H9 β), 3.05 (dd, J = 12.8, 16.9 Hz, 1H, H3 α), 2.84 (dd, J = 3.1, 16.9 Hz, 1H, H3 β), 2.78–2.80 (m, 1H, H12 α), 2.64–2.71 (m, 1H, H12 β), 2.19–2.26 (m, 1H, H11a), 1.53–1.75 (m, 4H, H10, H11); ^{13}C NMR $\ddagger$ (100 MHz, acetone- d_6): δ 189.0 (189.0) (C4), 160.0 (160.0) (C12a'), 158.7 (158.6) (C6a), 139.2 (139.1) (C1'), 128.1 (128.1) (C3', C5'), 127.8 (C4'), 125.7 (ArCH), 125.6 (ArCH), 124.9 (124.9) (ArCH), 114.1 (114.0) (C4a), 110.0 (110.0) (C6), 108.1 (108.0) (C12a), 96.4 (96.3) (C7a), 79.1 (79.0) (C2), 61.8 (61.5) (C9), 43.2 (43.0) (C3), 30.1 (30.0) (C11a), 23.4 (23.3) (C11), 22.8 (22.6) (C10), 22.3 (22.1) (C12). HRMS (+ESI): (C₂₁H₂₀O₄) calcd m/z 359.1254 [M + Na]⁺, found m/z 359.1252.

The earlier fractions were confirmed as the corresponding ring-opened product **16a**.

(*E*)-1-(6-Hydroxy-3,4,4a,10a-tetrahydro-2H,5H-pyrano[2,3-*b*]chromen-7-yl)-3-phenylprop-2-en-1-one (**16a**). Yellow needle-like crystals (14 mg, 8%). M.p. 174–175 °C; IR (neat): $\nu_{\max}$ 2936, 1621, 1570, 1487, 1420, 1356, 1299, 1228, 1109, 1038, 975, 923, 853, 751 cm^{-1} ; UV (THF) $\lambda_{\max}$ 331 nm (ϵ 42 284 $\text{cm}^{-1} \text{M}^{-1}$), 212 (27 012); ^1H NMR (400 MHz, acetone- d_6): δ 13.92 (s, 1H, OH), 8.09 (d, J = 8.9 Hz, 1H, H10'), 7.99 (d, J = 15.5 Hz, 1H, H β), 7.89 (d, J = 15.5 Hz, 1H, H α), 7.86–7.88 (m, 2H, H2, H6), 7.47–7.48 (m, 3H, H3, H4, H5), 6.45 (d, J = 8.9 Hz, 1H, H9'), 5.42 (d, J = 2.4 Hz, 1H, H7'a), 3.88–3.96 (m, 1H, H6'a), 3.66–3.75 (m, 1H, H6'β), 2.76–2.78 (m, 1H, H3'α), 2.66 (dd, J = 4.2, 17.3 Hz, 1H, H3'β), 2.20–2.28 (m, 1H, H3'a), 1.53–1.75 (m, 4H, H4', H5'); ^{13}C NMR (100 MHz, acetone- d_6): δ 193.1 (CO), 165.0 (C2'), 160.9 (C8'a), 145.0 (C α), 135.9 (C1), 131.5 (C4), 130.5 (C10'), 129.8 (C3, C5), 129.7 (C2, C6), 121.7 (C β), 114.4 (C1'), 109.1 (C2'a), 109.0 (C9'), 98.0 (C7'a), 62.8 (C6'), 31.6 (C3'a), 24.6 (C4'), 24.4 (C5'), 23.5 (C3'). HRMS (+ESI): (C₂₁H₂₀O₄) calcd m/z 359.1254 [M + Na]⁺, found m/z 359.1256.

2-Phenyl-2,3,7a,9,10,10a-hexahydro-4H,11H-furo[2,3-*b*]pyrano[2,3-*ff*]chromen-4-one (**15b**). Flavanone **15b** was synthesized following general procedure 1 using the Mannich base **10** (100 mg, 0.34 mmol, 1 eq.) and 2,3-dihydrofuran (0.51 mL, 6.73 mmol, 19.8 eq.). The mixture was heated at 155 °C for 43 h. The product obtained (13 mg, 12%) after column chromatography was further purified by preparative HPLC using a 5 μM column (150 $\times$ 22 mm ID) at a flow rate of 5 mL min^{−1} with a gradient elution from 2% to 100% B (solvent A: 0.1% formic acid/MilliQ water) and solvent B (0.1% formic acid/acetonitrile) over 56 min to afford the expected product **15b** as a yellow solid (6 mg, 6%). M.p. 121 °C; IR (neat): $\nu_{\max}$ 2921, 1680, 1601, 1439, 1321, 1104, 1068, 939, 891 cm^{-1} ; ^1H NMR (400 MHz, acetone- d_6): δ 7.65 (dd, J = 2.1, 8.7 Hz, 1H, H5), 7.60 (d, J = 8.9 Hz, 2H, H2', H6'), 7.44–7.48 (m, 2H, H3', H5'), 7.37–7.41 (m, 1H, H4'), 6.50 (dd, J = 2.3, 8.7 Hz, 1H, H6), 5.64 (dd, J = 3.0, 14.2 Hz, 1H, H2), 5.61 (d, J = 4.2 Hz, 1H, H7a), 4.00–4.06 (m, 1H, H9α), 3.91–3.98 (m, 1H, H9β), 3.03 (dd, J = 12.6, 16.7 Hz, 1H, H3α), 2.90–2.95 (m, 2H, H11), 2.81 (dd, J = 3.1, 16.7 Hz, 1H, H3β), 2.72–2.80 (m, 1H, H10a), 2.10–2.18 (m, 1H, H10α), 1.56–1.68 (m, 1H, H10β); ^{13}C NMR† (100 MHz, acetone- d_6): δ 190.4 (190.3) (C4), 161.5 (161.5) (C11a'), 160.3 (C6a), 140.6 (140.5) (C1'), 129.5 (C3', C5'), 129.2 (C4'), 127.1 (C2', C6'), 126.5 (C5), 115.6 (115.5) (C4a), 111.7 (C6), 108.8 (108.7) (C11a), 101.9 (101.8) (C7a), 80.6 (80.4) (C2), 68.6 (68.6) (C9), 44.6 (44.5) (C3), 37.2 (37.1) (C10a), 28.4 (28.3) (C10), 20.2 (20.1) (C11). HRMS (+ESI): (C₂₀H₁₈O₄) calcd m/z 345.1097 [M + Na]⁺, found m/z 345.1094.

The earlier fractions were confirmed as the corresponding ring-opened product **16b**.

(*E*)-1-(5-Hydroxy-2,3,3a,9a-tetrahydro-4H-furo[2,3-*b*]chromen-6-yl)-3-phenylprop-2-en-1-one (**16b**). Yellow solid (6 mg, 5%). M.p. 153 °C; IR (neat): $\nu_{\max}$ 2894, 1609, 1590, 1488, 1419, 1357, 1219, 1111, 1062, 932, 852, 768 cm^{-1} ; UV (THF) $\lambda_{\max}$ 330 nm (ϵ 48 257 $\text{cm}^{-1} \text{M}^{-1}$), 237 (15 312), 213 (24 016); ^1H NMR (400 MHz, acetone- d_6): δ 13.93 (s, 1H, OH), 8.09 (d, J = 8.7 Hz, 1H, H9'), 7.99 (d, J = 15.4 Hz, 1H, H β), 7.90 (d, J = 15.4 Hz, 1H, H α), 7.87–7.88 (m, 2H, H2, H6), 7.47–7.48 (m, 3H, H3, H4, H5), 6.39 (d, J =

8.7 Hz, 1H, H8'), 5.64 (d, J = 4.1 Hz, 1H, H6'a), 4.02–4.05 (m, 1H, H5'α), 3.96–4.00 (m, 1H, H5'β), 2.87–2.99 (m, 2H, H3'), 2.11–2.18 (m, 1H, H3'a), 1.56–1.67 (m, 2H, H4'); ^{13}C NMR (100 MHz, acetone- d_6): δ 193.1 (CO), 165.1 (C2'), 161.0 (C7'a), 145.0 (C α), 135.9 (C1), 131.5 (C4), 130.7 (C9'), 129.8 (C3, C5), 129.7 (C2, C6), 121.6 (C β), 114.3 (C1'), 109.2 (C8'), 108.2 (C2'a), 102.1 (C6'a), 68.7 (C5'), 37.2 (C3'a), 28.2 (C4'), 19.6 (C3'); HRMS (+ESI): (C₂₀H₁₈O₄) calcd m/z 323.1278 [M + H]⁺, found m/z 323.1275.

8-Ethoxy-2-phenyl-2,3,9,10-tetrahydro-4H,8H-pyrano[2,3-*ff*]chromen-4-one (**15c**). Flavanone **15c** was synthesized following general procedure 1 using the Mannich base **10** (100 mg, 0.34 mmol, 1 eq.) and ethyl vinyl ether (0.64 mL, 6.73 mmol, 19.8 eq.). The mixture was heated at 155 °C for 6 h. The product was obtained as a yellow oil (8 mg, 7%). IR (DCM): $\nu_{\max}$ 2922, 1681, 1601, 1438, 1325, 1243, 1168, 1119, 1048, 948, 869 cm^{-1} ; UV (THF) $\lambda_{\max}$ 307 nm (ϵ 13 948 $\text{cm}^{-1} \text{M}^{-1}$), 276 (33 089); ^1H NMR (400 MHz, acetone- d_6): δ 7.64–7.66 (m, 1H, H5), 7.59–7.63 (m, 2H, H2', H6'), 7.44–7.48 (m, 2H, H3', H5'), 7.38–7.42 (m, 1H, H4'), 6.55 (dd, J = 4.0, 8.8 Hz, 1H, H6), 5.62 (dd, J = 3.2, 12.9 Hz, 1H, H2), 5.35–5.38 (m, 1H, H8), 3.80–3.89 (m, 1H, H1'α), 3.64–3.72 (m, 1H, H1'β), 3.04 (dd, J = 3.4, 16.6 Hz, 1H, H3α), 2.79–2.80 (m, 1H, H3β), 2.59–2.76 (m, 2H, H10), 2.01–2.03 (m, 1H, H9α), 1.82–1.96 (m, 1H, H9β), 1.15 ppm (t, J = 7.20 Hz, 3H, CH₃); ^{13}C NMR† (100 MHz, acetone- d_6): δ 190.6 (190.5) (C4), 161.0 (160.9) (C10a'), 159.3 (159.2) (C6a), 140.6 (140.6) (C1'), 129.5 (C3', C5'), 129.2 (129.1) (C4'), 127.1 (C2', C6'), 126.1 (C5), 115.4 (115.5) (C4a), 112.1 (C6), 111.9 (111.8) (C10a), 98.3 (98.1) (C8), 80.6 (80.3) (C2), 64.6 (64.5) (C1''), 44.6 (44.5) (C3), 26.3 (26.1) (C9), 15.5 (15.4) (C10), 15.3 (CH₃). HRMS (+ESI): (C₂₀H₂₀O₄) calcd m/z 347.1254 [M + Na]⁺, found m/z 347.1252.

The earlier fractions were confirmed as the corresponding ring-opened product **16c**.

(*E*)-1-(2-Ethoxy-5-hydroxychroman-6-yl)-3-phenylprop-2-en-1-one (**16c**). Yellow oil (6 mg, 5%); IR (DCM): $\nu_{\max}$ 2930, 1617, 1605, 1489, 1420, 1360, 1228, 1107, 1050, 947, 867, 773 cm^{-1} ; UV (THF) $\lambda_{\max}$ 327 nm (ϵ 16 770 $\text{cm}^{-1} \text{M}^{-1}$), 240 (4087); ^1H NMR (400 MHz, acetone- d_6): δ 13.88 (s, 1H, OH), 8.08 (d, J = 8.9 Hz, 1H, H8'), 7.99 (d, J = 15.5 Hz, 1H, H β), 7.89 (d, J = 15.5 Hz, 1H, H α), 7.86–7.89 (m, 2H, H2, H6), 7.47–7.48 (m, 3H, H3, H4, H5), 6.45 (d, J = 8.9 Hz, 1H, H7'), 5.38 (t, J = 3.1 Hz, 1H, H5'), 3.84–3.88 (m, 1H, H1'α), 3.65–3.71 (m, 1H, H1'β), 2.67–2.76 (m, 2H, H3'), 2.60–2.66 (m, 1H, H4'α), 1.87–1.96 (m, 1H, H4'β), 1.15 (t, J = 3.0 Hz, 3H, CH₃); ^{13}C NMR (100 MHz, acetone- d_6): δ 193.2 (CO), 164.4 (C2'), 160.0 (C6'a), 144.9 (C α), 135.9 (C1), 131.5 (C4), 130.2 (C8'), 129.8 (C3, C5), 129.7 (C2', C6'), 121.6 (C β), 114.3 (C1'), 111.6 (C2'a), 109.6 (C7'), 98.5 (C5'), 64.6 (C1''), 26.3 (C4'), 15.4 (C3'), 15.0 (CH₃); HRMS (+ESI): (C₂₀H₂₀O₄) calcd m/z 347.1254 [M + Na]⁺, found m/z 347.1253.

8-(4-Methoxyphenyl)-2-phenyl-2,3,9,10-tetrahydro-4H,8H-pyrano[2,3-*ff*]chromen-4-one (**15d**). Flavanone **15d** was synthesized following general procedure 1 using the Mannich base **10** (150 mg, 0.50 mmol, 1 eq.) and *p*-methoxystyrene (0.66 mL, 4.96 mmol, 9.9 eq.). The mixture was heated at 155 °C for 6 h. The product was obtained as a pale-yellow solid (14 mg, 7%). M.p. 173 °C; IR (neat): $\nu_{\max}$ 2935, 1673, 1583, 1511, 1437, 1327, 1242, 1167, 1105, 1067, 1030, 914, 825, 818, 742; UV (MeOH) $\lambda_{\max}$ 282 nm (ϵ 26 897 $\text{cm}^{-1} \text{M}^{-1}$), 222 (57 040), 202 (59

822); ^1H NMR (400 MHz, acetone- d_6): δ 7.66 (dd, J = 3.3; 8.8 Hz, 1H, H5), 7.59–7.62 (m, 2H, H2', H6'), 7.43–7.48 (m, 2H, H3', H5'), 7.37–7.41 (m, 3H, H4', H2'', H6''), 6.96 (d, J = 8.6 Hz, 2H, H3'', H5''), 6.58 (d, J = 8.8 Hz, 1H, H6), 5.63 (dd, J = 3.3, 13.9 Hz, 1H, H8), 5.13 (dd, J = 2.1, 19.4 Hz, 1H, H2), 3.81 (s, 3H, OMe), 2.98–3.08 (m, 1H, H3 α), 2.72–2.81 (m, 3H, H3 β , H10), 2.21–2.27 (m, 1H, H9 α), 1.96–2.21 (m, 1H, H9 β); ^{13}C NMR † (100 MHz, acetone- d_6): δ 190.5 (190.4) (C4), 162.4 (162.2) (C4''), 161.4 (161.3) (C10a'), 160.5 (160.4) (C6a), 140.7 (140.6) (C1'), 134.0 (C1''), 129.5 (C3', C5'), 129.2 (129.1) (C4'), 128.3 (128.2) (C2'', C6''), 127.1 (127.0) (C2', C6'), 126.2 (126.1) (C5), 115.2 (115.1) (C4a), 114.6 (C3'', C5''), 112.0 (C6), 111.3 (111.3) (C10a), 80.5 (80.4) (C8), 78.7 (78.5) (C2), 55.6 (OMe), 44.5 (C3), 29.5 (C9), 20.1 (19.9) (C10). HRMS (+ESI): (C₂₅H₂₂O₄) calcd m/z 387.1591 [M + Na] $^+$, found m/z 387.1590.

The earlier fractions were confirmed as the corresponding ring-opened product **16d**.

(*E*)-1-(5-Hydroxy-2-(4-methoxyphenyl)chroman-6-yl)-3-phenylprop-2-en-1-one (**16d**). Yellow solid (15 mg, 8%); M.p. 179–180 °C; IR (neat): ν_{max} 2932, 1584, 1498, 1359, 1219, 1111, 1027, 975, 856, 810, 768 cm $^{-1}$; UV (MeOH) λ_{max} 337 nm (ϵ 27 206 cm $^{-1}$ M $^{-1}$), 224 (30 336), 202 (45 292); ^1H NMR (400 MHz, acetone- d_6): δ 13.99 (s, 1H, OH), 8.09 (d, J = 8.9 Hz, 1H, H8'), 8.01 (d, J = 15.4 Hz, 1H, H β), 7.90 (d, J = 15.4 Hz, 1H, H α), 7.87–7.89 (m, 2H, H2, H6), 7.47–7.48 (m, 3H, H3, H4, H5), 7.41 (d, J = 8.5 Hz, 2H, H2'', H6''), 6.97 (d, J = 8.9 Hz, 2H, H3'', H5''), 6.47 (d, J = 8.9 Hz, 1H, H7'), 5.15 (dd, J = 2.3, 10.3 Hz, 1H, H5'), 3.82 (s, 3H, OCH₃), 2.82–2.83 (m, 1H, H3' α), 2.70–2.79 (m, 1H, H3' β), 2.25–2.30 (m, 1H, H4' α), 2.06–2.09 (m, 1H, H4' β); ^{13}C NMR (100 MHz, acetone- d_6): δ 193.1 (C=O), 165.0 (C2'), 163.0 (C4''), 160.5 (C6'a), 144.9 (C α), 135.9 (C1), 134.0 (C1''), 131.5 (C4), 130.2 (C8'), 129.8 (C3, C5), 129.7 (C2, C6), 128.4 (C2'', C6''), 121.7 (C β), 114.7 (C3'', C5''), 114.0 (C1'), 111.1 (C2'a), 109.5 (C7'), 78.9 (C5'), 55.6 (OCH₃), 29.3 (C4'), 19.6 (C3'); HRMS (+ESI): (C₂₅H₂₂O₄) calcd m/z 409.1410 [M + Na] $^+$, found m/z 409.1415.

8-Methyl-2,8-diphenyl-2,3,9,10-tetrahydro-4H,8H-pyrano[2,3-ff]chromen-4-one (**15e**). Flavanone **15e** was synthesized following general procedure 1 using the Mannich base **10** (100 mg, 0.34 mmol, 1 eq.) and α -methyl styrene (0.87, 6.73 mmol, 19.8 eq.). The mixture was heated at 155 °C for 16 h. The product obtained (12 mg, 10%) after column chromatography was further purified by preparative HPLC using a 5 μM column (150 $\times$ 22 mm ID) at a flow rate of 5 mL min $^{-1}$ with a gradient elution from 2% to 100% B (solvent A: 0.1% formic acid/MilliQ water) and solvent B (0.1% formic acid/acetonitrile) over 56 min to afford the expected product as a white solid (5 mg, 4%). M.p. 127–128 °C; IR (neat): ν_{max} 2969, 1670, 1580, 1434, 1329, 1258, 1193, 1099, 902, 817, 757 cm $^{-1}$; UV (THF) λ_{max} 305 nm (ϵ 13959 cm $^{-1}$ M $^{-1}$), 279 (35 526), 240 (39 045); ^1H NMR (400 MHz, acetone- d_6): δ 7.66–7.70 (dd, J = 3.8, 8.6 Hz, 1H, H5), 7.52–7.56 (m, 2H, H2', H6'), 7.37–7.44 (m, 5H, ArH), 7.31–7.36 (m, 4H, ArH), 7.23–7.27 (m, 2H, ArH), 6.72 (dd, J = 5.1, 8.7 Hz, 1H, H6), 5.54 (dd, J = 2.9, 13.0 Hz, 1H, H2), 2.98 (dd, J = 12.7, 16.7 Hz, 1H, H3 α), 2.78–2.79 (m, 1H, H3 β), 2.69–2.75 (m, 1H, H10 α), 2.52–2.63 (m, 1H, H10 β), 2.07–2.28 (m, 2H, H9), 1.64 (d, J = 2.2 Hz, 3H, CH₃). ^{13}C NMR † (100 MHz, acetone- d_6): δ 190.5 (190.3) (C4), 161.3 (161.2) (ArC), 161.1 (161.1) (ArC), 145.9

(145.7) (C1''), 140.5 (140.6) (C1'), 129.4 (129.4) (2 $\times$ ArCH), 129.4 (129.3) (2 $\times$ ArCH), 129.2 (ArCH), 127.8 (ArCH), 127.0 (2 $\times$ ArCH), 126.3 (126.2) (ArCH), 125.7 (125.6) (2 $\times$ ArCH), 115.1 (115.0) (C4a), 112.2 (112.1) (C6), 111.1 (111.0) (C10a), 80.6 (C2), 80.3 (80.2) (C8), 44.6 (44.4) (C3), 32.2 (32.0) (C9), 30.9 (CH₃), 17.7 (C10). HRMS (+ESI): (C₂₅H₂₂O₃) calcd m/z 371.1642 [M + H] $^+$, found m/z 371.1640.

The earlier fractions were confirmed as the corresponding ring-opened product **16e**.

(*E*)-1-(5-Hydroxy-2-methyl-2-phenylchroman-6-yl)-3-phenylprop-2-en-1-one (**16e**). Yellow oil (9 mg, 7%). IR (DCM): ν_{max} 2929, 1615, 1602, 1488, 1361, 1275, 1223, 1113, 1033, 763 cm $^{-1}$; UV (THF) λ_{max} 333 nm (ϵ 17 522 cm $^{-1}$ M $^{-1}$), 241 (4668); ^1H NMR (400 MHz, acetone- d_6): δ 13.88 (s, 1H, OH), 8.11 (d, J = 9.2 Hz, 1H, H8'), 7.99 (d, J = 15.5 Hz, 1H, H β), 7.88 (d, J = 15.5 Hz, 1H, H α), 7.86–7.88 (m, 2H, H2, H6), 7.46–7.48 (m, 3H, H3, H4, H5), 7.43 (d, J = 7.9 Hz, 2H, H2'', H6''), 7.31–7.35 (m, 2H, ArH), 7.22–7.25 (m, 1H, ArH), 6.62 (d, J = 9.2 Hz, 1H, H7'), 2.70–2.79 (m, 1H, H3' α), 2.54–2.60 (m, 1H, H3' β), 2.11–2.25 (m, 2H, H4'), 1.66 (s, 3H, CH₃); ^{13}C NMR (100 MHz, acetone- d_6): δ 193.0 (CO), 164.8 (C2'), 161.8 (C6'a), 145.8 (C1''), 144.8 (C α), 135.9 (C1), 131.5 (C4), 130.4 (C8'), 129.8 (C3, C5), 129.7 (C2, C6), 129.3 (C3'', C5''), 127.8 (C4''), 125.6 (C2'', C6''), 121.7 (C β), 114.0 (C1'), 110.9 (C2'a), 109.7 (C7'), 80.6 (C5'), 32.3 (C4'), 30.4 (CH₃), 17.2 (C3'); HRMS (+ESI): (C₂₅H₂₂O₄) calcd m/z 371.1642 [M + H] $^+$, found m/z 371.1644.

(*E*)-1-(2-Bromo-8-hydroxy-9H-benzo[e][1,2,4]triazolo[5,1-b][1,3]oxazin-7-yl)-3-phenylprop-2-en-1-one (**18**). To a mixture of the Mannich base **10** (150 mg, 0.50 mmol, 1 eq.) and K₂CO₃ (209 mg, 1.51 mmol, 3.0 eq.) in DMA (10 mL), 3,5-dibromo-1,2,4-triazole (115 mg, 0.51 mmol, 1.0 eq.) was added. The reaction mixture was heated to 150 °C in a pressure tube for 4 h. After completion of reaction, the reaction mixture was cooled to room temperature and poured into water (50 mL). The precipitated yellow solid was filtered, washed with water, and dried to give **18** as a yellow solid (117 mg, 58%). M.p. 219–221 °C; IR (neat): ν_{max} 1638, 1624, 1545, 1532, 1354, 1291, 1209, 1103, 984, 823, 770 cm $^{-1}$; UV (THF) λ_{max} 331 nm (ϵ 36 636 cm $^{-1}$ M $^{-1}$); ^1H NMR (400 MHz, DMSO- d_6): δ 13.82 (br s, 1H, OH), 8.51 (d, J = 8.6 Hz, 1H, H10'), 8.09 (d, J = 15.5 Hz, 1H, H α), 7.94–7.96 (m, 2H, H2, H6), 7.92 (d, J = 15.5 Hz, 1H, H β), 7.49–7.50 (m, 3H, H3, H4, H5), 7.01 (d, J = 8.6 Hz, 1H, H9'), 5.17 (s, 2H, CH₂-N); ^{13}C NMR (100 MHz, DMSO- d_6): δ 192.8 (CO), 160.8 (C2'), 152.6 (C8'a), 152.4 (C7'a), 145.9 (C β), 136.7 (C6'), 134.2 (C1), 132.0 (C10'), 131.3 (C4), 129.4 (C2, C6), 129.0 (C3, C5), 120.5 (C α), 116.3 (C1'), 107.8 (C9'), 104.8 (C2'a), 42.6 (C3'). HRMS (+ESI): (C₁₈H₁₂BrN₃O₃) calcd for ^{79}Br m/z 419.9954 [M + Na] $^+$, found m/z 419.9947; calcd for ^{81}Br m/z 421.9934 [M + Na] $^+$, found m/z 421.9926.

9-Bromo-2-phenyl-2,3-dihydro-4H,12H-chromeno[8,7-e][1,2,4]triazolo[5,1-b][1,3]oxazin-4-one (**19**). To a solution of **18** (30 mg, 0.075 mmol, 1 eq.) in MeOH (2 mL) was added concentrated HCl (0.13 mL, 1.58 mmol, 21 eq.). The reaction was heated to reflux for 72 h using a reflux setup. The reaction mixture was cooled to room temperature and poured into ice-water. The precipitated yellow solid (24 mg) was subjected to preparative reverse-phase HPLC using a 5 μM column (150 $\times$ 22 mm ID) at a flow rate of 5 mL min $^{-1}$ with a gradient elution from 2% to

100% B (solvent A: 0.1% formic acid/H₂O) and solvent B (0.1% formic acid/ACN) over 56 min to afford **19** as a pale-yellow solid (3 mg, 10%) and the starting material **18** (8 mg, 27%). M.p. 236 °C; IR (neat): ν_{max} 2919, 1699, 1606, 1532, 1521, 1442, 1270, 1232, 1094, 1060, 982, 886, 799 cm⁻¹; UV (THF) λ_{max} 313 nm (ϵ 15 610 cm⁻¹ M⁻¹), 269 (20 628); ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.85 (d, *J* = 8.8 Hz, 1H, H5), 7.56–7.58 (m, 2H, H2', H6'), 7.40–7.48 (m, 3H, H3', H4', H5'), 7.07 (d, *J* = 8.8 Hz, 1H, H6), 5.84 (dd, *J* = 3.1, 12.4 Hz, 1H, H2), 5.21 (s, 2H, CH₂-N), 3.26 (dd, *J* = 12.4, 16.9 Hz, 1H, H3 α), 2.98 (dd, *J* = 3.1, 16.9 Hz, 1H, H3 β); ¹³C NMR (100 MHz, DMSO-*d*₆): δ 189.8 (C4), 158.3 (C12b), 152.7 (C6a), 152.2 (C7a), 138.3 (C9), 136.7 (C1'), 128.7 (C3', C5'), 128.6 (C4'), 127.3 (C5), 126.3 (C2', C6'), 117.5 (C4a), 110.4 (C6), 105.4 (C12a), 79.7 (C2), 42.9 (CH₂-N), 42.7 (C3); HRMS (+ESI): (C₁₈H₁₂BrN₃O₃) calcd for ⁷⁹Br *m/z* 419.9954 [M + Na]⁺, found *m/z* 419.9952; calcd for ⁸¹Br *m/z* 421.9934 [M + Na]⁺, found *m/z* 421.9929.

General procedure 2 for the synthesis of Diels–Alder adducts and Michael adducts from isoflavanones. To a suspension of Mannich base **11** in dry acetonitrile (10 mL) was added the appropriate dienophile (20 eq.) or nucleophile (1 eq.). The reaction mixture was heated at 120 °C in a pressure tube held under a nitrogen atmosphere for 24–48 h. After completion of reaction, the solvent was evaporated *in vacuo*, and the crude product was subjected to column chromatography over silica gel G₆₀ (230–400 mesh) using *n*-hexane : EtOAc (95 : 5) as the eluent to afford the corresponding products. Further purification was performed using preparative reversed phase HPLC when required.

8-((Dimethylamino)methyl)-7-hydroxy-3-(4-hydroxyphenyl)chroman-4-one (11). To a solution of dihydrodaidzein **7** (1.20 g, 4.68 mmol, 1 eq.) in ethanol (60 mL) was added bis(*N,N*-dimethylamino)methane (1.28 mL, 9.37 mmol, 2.0 eq.). The reaction mixture was heated to reflux for 1 h. The product precipitated from the reaction mixture and was collected by filtration to give **11** as a white fine solid (0.92 g, 61%). M.p. 274–275 °C; IR (neat): ν_{max} 3280, 2962, 1651, 1582, 1439, 1384, 1111, 946, 805, 745 cm⁻¹. UV (MeOH): λ_{max} 329 nm (ϵ 15 863 cm⁻¹ M⁻¹), 260 (6509), 221 (19 014); ¹H NMR (300 MHz, DMSO-*d*₆): δ 7.59 (d, *J* = 8.7 Hz, 1H, H5), 7.03 (d, *J* = 8.7 Hz, 2H, H2', H6'), 6.69 (d, *J* = 8.7 Hz, 2H, H3', H5'), 6.41 (d, *J* = 8.7 Hz, 1H, H6), 4.56 (d, *J* = 6.6 Hz, 2H, H2), 3.87 (t, *J* = 6.6 Hz, 1H, H3), 3.74 (s, 2H, CH₂-N), 2.32 (s, 6H, N(CH₃)₂); ¹³C NMR (75 MHz, DMSO-*d*₆): δ 190.5 (C4), 166.8 (C7), 160.4 (C8a), 156.5 (C4'), 129.7 (C2', C6'), 127.6 (C5), 126.3 (C1'), 115.2 (C3', C5'), 112.4 (C8), 111.1 (C6), 107.7 (C4a), 71.4 (C2), 53.8 (CH₂-N), 50.0 (C3), 43.7 (2 × CH₃). HRMS (+ESI): (C₁₈H₁₉NO₄) calcd *m/z* 314.1387 [M + H]⁺, found *m/z* 314.1382.

2-Hydroxy-1-(5-hydroxy-2,3,3a,9a-tetrahydro-4H-furo[2,3-*b*]chromen-6-yl)-2-(4-hydroxyphenyl)propan-1-one (20). To a suspension of Mannich base **11** (121 mg, 0.38 mmol, 1 eq.) in DMF (10 mL) was added 2,3-dihydrofuran (0.58 mL, 7.71 mmol, 20.3 eq.). The reaction mixture was heated at 150–155 °C in a pressure tube for 28 h. After completion of reaction, the solvent was evaporated *in vacuo*, and the resulting crude was added methanol. The precipitated solid was filtered, washed with methanol, and air-dried to give product **20** as an off-white solid (39 mg, 28%). M.p. 252–253 °C; IR (neat): ν_{max} 3350, 2927, 1686, 1592, 1435, 1281, 1214, 1048, 935, 830 cm⁻¹; UV (THF): λ_{max} 317 nm (ϵ

8018 cm⁻¹ M⁻¹), 269 (11 083), 239 (9408); ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.55 (s, 1H, OH), 7.38 (d, *J* = 8.6 Hz, 1H, H4), 7.22 (d, *J* = 8.8 Hz, 2H, H2', H6'), 6.76 (d, *J* = 8.8 Hz, 2H, H3', H5'), 6.59 (d, *J* = 8.6 Hz, 1H, H5), 5.71 (d, *J* = 4.1 Hz, 1H, H6a), 3.99–4.04 (m, 1H, H8 α), 3.90–3.96 (m, 1H, H8 β), 2.88–3.02 (m, 2H, H10), 2.69–2.77 (m, 1H, H9a), 2.10–2.18 (m, 1H, H9 α), 1.69 (s, 3H, CH₃), 1.50–1.61 (m, 1H, H9 β); ¹³C NMR (100 MHz, DMSO-*d*₆): δ 199.1 (C1), 171.0 (C11), 161.5 (C5a), 157.6 (C4'), 128.5 (C1'), 126.0 (C2', C6'), 123.4 (C4), 115.4 (C3', C5'), 112.7 (C5), 111.5 (C1a), 103.9 (C10a), 101.1 (C6a), 90.4 (C2), 67.7 (C8), 35.4 (C9a), 27.1 (C9), 23.3 (C3), 18.3 (C10). HRMS (+ESI): (C₂₀H₂₀O₆) calcd *m/z* 357.1333 [M + H]⁺, found *m/z* 357.1328.

3-(4-Hydroxyphenyl)-2,3,7a,9,10,10a-hexahydro-4H,11H-furo[2,3-*b*]pyrano[2,3-*ff*]chromen-4-one (22a). Isoflavanone **22a** was synthesized following general procedure 2 using the Mannich base **11** (121 mg, 0.39 mmol, 1 eq.) and 2,3-dihydrofuran (0.58 mL, 7.67 mmol, 19.7 eq.). The product was obtained as a white solid (23 mg, 18%). M.p. 230–231 °C; IR (neat): ν_{max} 3265, 2953, 1679, 1587, 1515, 1438, 1357, 1228, 1166, 1096, 1025, 936, 888 cm⁻¹; UV (THF): λ_{max} 307 nm (ϵ 6429 cm⁻¹ M⁻¹), 283 (8425), 269 (12 282), 240 (9846); ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.36 (s, 1H, OH), 7.58 (d, *J* = 8.7 Hz, 1H, H5), 7.04 (d, *J* = 8.7 Hz, 2H, H2', H6'), 6.71 (d, *J* = 8.7 Hz, 2H, H3', H5'), 6.51 (d, *J* = 8.7 Hz, 1H, H6), 5.63 (d, *J* = 4.2 Hz, 1H, H7a), 4.62–4.66 (m, 2H, H2), 3.94–4.01 (m, 2H, H9), 3.86–3.91 (m, 1H, H3), 2.82 (d, *J* = 4.1 Hz, 2H, H11), 2.64–2.72 (m, 1H, H10a), 2.04–2.11 (m, 1H, H10 α), 1.43–1.54 (m, 1H, H10 β); ¹³C NMR† (100 MHz, DMSO-*d*₆): δ 191.1 (190.9) (C4), 160.5 (160.4) (C1a), 158.8 (158.7) (C6a), 156.6 (C4'), 129.8 (129.6) (C2', C6'), 126.0 (C1'), 125.9 (C5), 115.2 (C3', C5'), 114.4 (114.3) (C4a), 110.8 (110.6) (C6), 107.3 (C11a), 100.6 (C7a), 71.6 (71.5) (C2), 67.7 (C9), 50.1 (49.9) (C3), 35.6 (C10a), 27.1 (CH₂), 19.0 (CH₂). HRMS (+ESI): (C₂₀H₁₈O₅) calcd *m/z* 361.1046 [M + Na]⁺, found *m/z* 361.1045.

3-(4-Hydroxyphenyl)-2,3,7a,10,11,11a-hexahydro-4H,9H,12H-dipyran[2,3-*b*:2',3'-*ff*]chromen-4-one (22b). Isoflavanone **22b** was synthesized following general procedure 2 using the Mannich base **11** (150 mg, 0.48 mmol, 1 eq.) and 3,4-dihydro-2H-pyran (0.91 mL (97%), 9.64 mmol, 20.1 eq.). The product was obtained as a white solid (22 mg, 13%). M.p. 223–224 °C; IR (neat): ν_{max} 3380, 2947, 1675, 1603, 1516, 1435, 1356, 1230, 1038, 989 cm⁻¹; UV (THF): λ_{max} 307 nm (ϵ 6061 cm⁻¹ M⁻¹), 282 (8528), 270 (12 157), 239 (8880); ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.36 (s, 1H, OH), 7.59 (d, *J* = 8.2 Hz, 1H, H5), 7.03 (d, *J* = 8.6 Hz, 2H, H2', H6'), 6.71 (d, *J* = 8.6 Hz, 2H, H3', H5'), 6.57 (d, *J* = 8.6 Hz, 1H, H6), 5.39 (d, *J* = 2.9 Hz, 1H, H7a), 4.61–4.65 (m, 2H, H2), 3.91–3.97 (m, 1H, H3), 3.80–3.86 (m, 1H, H9 α), 3.64–3.66 (m, 1H, H9 β), 2.73 (dd, *J* = 2.4, 6.2 Hz, 1H, H12 α), 2.52–2.56 (m, 1H, H12 β), 2.15–2.16 (m, 1H, H11a), 1.44–1.66 (m, 4H, 10, 11-CH₂); ¹³C NMR† (100 MHz, DMSO-*d*₆): δ 191.1 (191.0) (C4), 160.3 (160.2) (C1a), 158.6 (158.5) (C6a), 156.6 (C4'), 129.8 (129.6) (C2', C6'), 126.0 (C1'), 125.9 (C5), 115.3 (C3', C5'), 114.4 (114.3) (C4a), 110.5 (C6), 108.2 (108.1) (C12a), 96.4 (C7a), 71.6 (71.5) (C2), 59.8 (C8), 50.1 (49.9) (C3), 29.9 (29.8) (C11a), 23.7 (23.6) (CH₂), 22.8 (22.7) (CH₂), 22.3 (22.2) (CH₂). HRMS (+ESI): (C₂₁H₂₀O₅) calcd *m/z* 353.1384 [M + H]⁺, found *m/z* 353.1385.

8-Ethoxy-3-(4-hydroxyphenyl)-2,3,9,10-tetrahydro-4H,8H-pyrano[2,3-*ff*]chromen-4-one (22c). Isoflavanone **22c** was synthesized

following general procedure 2 using the Mannich base **11** (150 mg, 0.48 mmol, 1 eq.) and ethyl vinyl ether (0.92 mL, 9.61 mmol, 20.0 eq.). The product obtained (32 mg, 20%) after column chromatography was further purified by preparative HPLC using a 5 μ m column (150 $\times$ 22 mm ID) at a flow rate of 5 mL min⁻¹ with an isocratic elution of 50% solvent B (solvent A: 0.1% formic acid/H₂O) and solvent B (0.1% formic acid/ACN) over 56 min to afford the expected product as a white solid (10 mg, 10%). M.p. 241–242 °C; IR (neat): $\nu_{\max}$ 3316, 2930, 1667, 1583, 1517, 1436, 1354, 1227, 1094, 921 cm⁻¹; UV (THF): $\lambda_{\max}$ 307 nm (ϵ 6705 cm⁻¹ M⁻¹), 281 (9394), 268 (14 875), 239 (9394); ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.41 (br s, 1H, OH), 7.59 (d, *J* = 8.8 Hz, 1H, H5), 7.04 (d, *J* = 8.7 Hz, 2H, H2', H6'), 6.71 (d, *J* = 8.7 Hz, 2H, H3', H5'), 6.55 (d, *J* = 8.8 Hz, 1H, H6), 5.38 (t, *J* = 3.6 Hz, 1H, H8), 4.59–4.69 (m, 2H, H2), 3.91–3.97 (m, 1H, H3), 3.73–3.81 (m, 1H, H1'' α), 3.61–3.69 (m, 1H, H1'' β), 2.62–2.68 (m, 1H, H10 α), 2.53–2.57 (m, 1H, H10 β), 1.96–2.02 (m, 1H, H9 α), 1.83–1.89 (m, 1H, H9 β), 1.12 (t, *J* = 7.1 Hz, 3H, H2''). ¹³C NMR $\ddagger$ (100 MHz, DMSO-*d*₆): δ 191.2 (C4), 159.8 (C1a), 157.9 (C6a), 156.6 (C4'), 129.7 (129.6) (C2', C6'), 125.9 (125.8) (C1'), 125.7 (C5), 115.2 (C3', C5'), 114.2 (C4a), 111.1 (C6), 110.3 (110.2) (C9a), 97.0 (96.9) (C8), 71.5 (71.4) (C2), 63.6 (63.5) (C1''), 50.0 (C3), 25.0 (24.9) (C9), 15.0 (CH₃), 14.3 (14.2) (C10). HRMS (+ESI): (C₂₀H₂₀O₅) calcd *m/z* 341.1384 [M + H]⁺, found *m/z* 341.1382.

3-(4-Hydroxyphenyl)-8-(4-methoxyphenyl)-2,3,9,10-tetrahydro-4H,8H-pyrano[2,3-*ff*]chromen-4-one (22d). Isoflavanone **22d** was synthesized following general procedure 2 using the Mannich base **11** (120 mg, 0.38 mmol, 1 eq.) and *p*-methoxystyrene (1.02 mL, 7.66 mmol, 20.2 eq.). The product was obtained as a white solid (51 mg, 33%). M.p. 263–264 °C; IR (neat): $\nu_{\max}$ 3285, 2918, 1657, 1580, 1511, 1435, 1357, 1175, 1029 cm⁻¹; UV (THF): $\lambda_{\max}$ 288 nm (ϵ 13 321 cm⁻¹ M⁻¹), 270 (15 011), 240 (12 758); ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.37 (s, 1H, OH), 7.59 (d, *J* = 8.6 Hz, 1H, H5), 7.36 (d, *J* = 8.6 Hz, 2H, H2'', H6''), 7.05 (d, *J* = 8.7 Hz, 2H, H2', H6'), 6.96 (d, *J* = 8.6 Hz, 2H, H3'', H5''), 6.71 (d, *J* = 8.5 Hz, 2H, H3', H5'), 6.57 (d, *J* = 8.7 Hz, 1H, H6), 5.10–5.14 (m, 1H, H8), 4.62–4.71 (m, 2H, H2), 3.89–3.98 (m, 1H, H3), 3.76 (s, 3H, OCH₃), 2.69–2.74 (m, 2H, H10), 2.16–2.19 (m, 1H, H9 α), 1.94–2.05 (m, 1H, H9 β); ¹³C NMR $\ddagger$ (100 MHz, DMSO-*d*₆): δ 191.1 (191.0) (C4), 160.9 (C4''), 160.3 (160.2) (C1a), 159.0 (C6a), 156.6 (C4'), 132.6 (ArC), 129.7 (129.6) (C2', C6'), 127.6 (127.5) (C2'', C6''), 126.1 (C1'), 126.0 (C1''), 125.7 (C5), 115.2 (115.3) (C3', C5'), 113.8 (C3'', C5''), 109.9 (111.0) (C6), 77.2 (77.1) (C8), 71.5 (C2), 55.1 (OCH₃), 50.0 (49.9) (C3), 27.7 (27.8) (C9), 18.8 (18.7) (C10). HRMS (+ESI): (C₂₄H₂₂O₇) calcd *m/z* 403.1574 [M + H]⁺, found *m/z* 403.1541.

3-(4-Hydroxyphenyl)-7a-((trimethylsilyl)oxy)-2,3,7a,8,10,11,11a,12-octahydro-4H,9H-pyrano[2,3-*a*]xanthen-4-one (22e). Isoflavanone **22e** was synthesized following general procedure 2 using the Mannich base **11** (152 mg, 0.48 mmol, 1 eq.) and (cyclohex-1-en-1-yloxy)trimethylsilane (0.94 mL, 4.83 mmol, 10.1 eq.). The product was obtained as an off-white solid (57 mg, 27%). M.p. 197–198 °C; IR (neat): $\nu_{\max}$ 3350, 2934, 1657, 1584, 1516, 1435, 1362, 1245, 1174, 1093, 913 cm⁻¹; UV (THF): $\lambda_{\max}$ 268 nm (ϵ 14 649 cm⁻¹ M⁻¹), 238 (9386); ¹H NMR (400 MHz, acetone-*d*₆): δ 8.31 (s, 1H, OH), 7.69 (d, *J* = 8.9 Hz, 1H, H5), 7.13 (d, *J* = 8.5 Hz, 2H, H2', H6'), 6.79 (d, *J* = 8.5 Hz, 2H,

H3', H5'), 6.53 (d, *J* = 8.9 Hz, 1H, H6), 4.63–4.70 (m, 2H, H2), 3.84–3.88 (dd, *J* = 5.4, 7.3 Hz, 1H, H3), 2.43–2.48 (m, 2H, H8), 2.19–2.27 (m, 1H, H12 α), 1.97–2.02 (m, 1H, H11a), 1.67–1.73 (m, 1H, H11 α), 1.65–1.67 (m, 1H, H12 β), 1.62–1.63 (m, 1H, H10 α), 1.53–1.57 (m, 1H, H9 α), 1.49–1.52 (m, 1H, H11 β), 1.33–1.42 (m, 1H, H10 β), 1.06–1.15 (m, 1H, H9 β), 0.02 (s, 9H, 3 $\times$ CH₃); ¹³C NMR $\ddagger$ (100 MHz, acetone-*d*₆): δ 191.9 (191.5) (C4), 161.8 (161.7) (C1a), 158.8 (158.7) (C6a), 157.6 (C4'), 130.8 (130.6) (C2', C6'), 128.1 (127.8) (C1'), 126.7 (126.6) (C5), 116.2 (C3', C5'), 115.8 (115.7) (C12a), 112.1 (112.0) (C6), 110.0 (109.9) (C4a), 100.4 (100.4) (C7a), 72.9 (73.0) (C2), 51.9 (51.7) (C3), 39.5 (C12), 38.6 (38.5) (C11a), 29.6 (C9), 25.5 (C10), 24.4 (C11), 22.6 (22.5) (C8), 1.83 (3 $\times$ CH₃). HRMS (+ESI): (C₂₅H₃₀O₅Si) calcd *m/z* 461.1755 [M + Na]⁺, found *m/z* 461.1749.

7a-hydroxy-3-(4-hydroxyphenyl)-2,3,7a,8,9,10,10a,11-octahydro-4H-cyclopenta[*b*]pyrano[2,3-*ff*]chromen-4-one (22f). Isoflavanone **22f** was synthesized following general procedure 2 using the Mannich base **11** (120 mg, 0.38 mmol, 1 eq.) and (cyclopent-1-en-1-yloxy)trimethylsilane (0.68 mL, 7.66 mmol, 20.2 eq.). The product was obtained as a light-yellow sticky solid (34 mg, 25%). IR (DCM): $\nu_{\max}$ 3287, 2962, 1715, 1586, 1512, 1438, 1362, 1263, 1093 cm⁻¹; UV (THF): $\lambda_{\max}$ 288 nm (ϵ 169 500 cm⁻¹ M⁻¹), 269 (179 366), 241 (187 471); ¹H NMR (400 MHz, acetone-*d*₆): δ 7.64 (d, *J* = 8.7 Hz, 1H, H5), 7.13 (d, *J* = 8.5 Hz, 2H, H2', H6'), 6.79 (d, *J* = 8.5 Hz, 2H, H3', H5'), 6.62 (d, *J* = 8.7 Hz, 1H, H6), 4.63–4.66 (m, 2H, H2), 3.81–3.86 (m, 1H, H3), 2.98–3.04 (m, 1H, H11 α), 2.61–2.68 (m, 1H, H11 β), 2.44–2.52 (m, 1H, H10a), 2.20–2.27 (m, 1H, H10 α), 2.08–2.18 (m, 1H, H10 β), 1.98–2.02 (m, 1H, H9 α), 1.70–1.80 (m, 2H, H8), 1.60–1.73 (m, 1H, H9 β); ¹³C NMR $\ddagger$ (100 MHz, acetone-*d*₆): δ 191.6 (C4), 162.5 (C11a'), 162.2 (162.1) (C6a), 157.5 (C4'), 130.5 (130.5) (C2', C6'), 128.0 (128.0) (C1'), 127.4 (C5), 116.1 (C3', C5'), 115.4 (C7a), 115.0 (C11a, C4a), 110.7 (110.6) (C6), 72.6 (72.5) (C2), 51.6 (51.5) (C3), 49.4 (C10a), 38.1 (C10), 30.0 (C9), 23.2 (C11), 21.1 (C8). HRMS (+ESI): (C₂₁H₂₀O₅) calcd *m/z* 375.1203 [M + Na]⁺, found *m/z* 375.1199.

7-Hydroxy-8-(4-hydroxy-2,6-dimethoxybenzyl)-3-(4-hydroxyphenyl)chroman-4-one (23a). Isoflavanone **23a** was synthesized following general procedure 2 using the Mannich base **11** (120 mg, 0.38 mmol, 1 eq.) and 3,5-dimethoxyphenol (60 mg, 0.38 mmol, 1 eq.). The product was obtained as an orange sticky solid (13 mg, 8%). IR (DCM): $\nu_{\max}$ 3157, 2926, 1590, 1512, 1437, 1241, 1195, 1109 cm⁻¹; UV (THF): $\lambda_{\max}$ 288 nm (ϵ 9505 cm⁻¹ M⁻¹), 269 (11 701), 241 (18 629); ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.87 (s, 1H, OH-7), 9.31 (s, 1H, OH-4'), 9.17 (s, 1H, OH-4''), 7.45 (d, *J* = 8.5 Hz, 1H, H5), 7.00 (d, *J* = 8.5 Hz, 2H, H2', H6'), 6.67 (d, *J* = 8.5 Hz, 2H, H3', H5'), 6.45 (d, *J* = 8.5 Hz, 1H, H6), 6.00 (s, 2H, H3'', H5''), 4.48–4.53 (m, 2H, H2), 3.76–3.78 (m, 1H, H3), 3.72 (s, 2H, CH₂), 3.59 (s, 6H, 2 $\times$ OCH₃); ¹³C NMR (100 MHz, DMSO-*d*₆): δ 191.2 (CO), 162.1 (C7), 161.2 (C8a), 158.7 (C2'', C6''), 156.7 (C4''), 156.4 (C4'), 129.7 (C2', C6'), 126.7 (C1'), 125.3 (C5), 115.3 (C8), 115.2 (C3', C5'), 113.2 (C4a), 109.8 (C6), 107.1 (C1''), 92.2 (C3'', C5''), 71.2 (C2), 55.5 (2 $\times$ OCH₃), 50.0 (C3), 16.4 (CH₂). HRMS ESI (C₂₄H₂₂O₇) calcd *m/z* 423.1438 [M + H]⁺, found *m/z* 423.1438.

8-(2-Amino-4,6-dimethoxybenzyl)-7-hydroxy-3-(4-hydroxyphenyl)chroman-4-one (23b). Isoflavanone **23b** was synthesized following general procedure 2 using the Mannich

base **11** (120 mg, 0.38 mmol, 1 eq.) and 3,5-dimethoxyaniline (59 mg, 0.38 mmol, 1 eq.). The product was obtained as a brown sticky solid (28 mg, 17%). IR (DCM): $\nu_{\max}$ 3368, 3236, 2929, 1590, 1513, 1436, 1241, 1203, 1090, 1023, 998 cm^{-1} ; UV (THF): $\lambda_{\max}$ 288 nm (ϵ 8429 $\text{cm}^{-1} \text{M}^{-1}$), 268 (10 494), 241 (16 310); ^1H NMR (400 MHz, DMSO- d_6): δ 9.32 (s, 1H, OH), 7.50 (d, J = 8.7 Hz, 1H, H5), 6.99 (d, J = 8.4 Hz, 2H, H2', H6'), 6.65 (d, J = 8.4 Hz, 2H, H3', H5'), 6.52 (d, J = 8.7 Hz, 1H, H6), 5.87 (d, J = 2.2 Hz, 1H, H3''), 5.86 (d, J = 2.2 Hz, 1H, H5''), 4.55 (d, J = 6.2 Hz, 2H, H2), 3.80 (t, J = 6.2 Hz, 1H, H3), 3.70 (s, 3H, OMe), 3.67 (s, 2H, CH₂), 3.64 (s, 3H, OMe); ^{13}C NMR (100 MHz, DMSO- d_6): δ 191.1 (C4), 161.7 (C7), 161.1 (C8a), 158.9 (C4''), 158.7 (C6''), 156.4 (C4'), 147.3 (C2''), 129.5 (C2', C6'), 126.4 (C1'), 125.8 (C5), 115.2 (C3', C5'), 114.1 (C4a), 113.5 (C4a), 110.0 (C6), 104.5 (C1''), 93.2 (C3''), 88.1 (C5''), 71.2 (C2), 55.5 (OCH₃), 54.6 (OCH₃), 49.9 (C3), 16.7 (CH₂). HRMS (+ESI): (C₂₄H₂₃NO₆) calcd m/z 422.1598 [M + H]⁺, found m/z 422.1597.

7-Hydroxy-3-(4-hydroxyphenyl)-8-((phenylthio)methyl)chroman-4-one (23c). Isoflavanone **23c** was synthesized following general procedure 2 using the Mannich base **11** (150 mg, 0.48 mmol, 1 eq.) and thiophenol (0.05 mL, 0.49 mmol, 1.0 eq.). The product was obtained as colourless crystals (75 mg, 41%). M.p. 159–160 °C; IR (neat): $\nu_{\max}$ 3300, 2974, 1648, 1589, 1515, 1439, 1373, 1215, 1092, 1045 cm^{-1} ; UV (THF): $\lambda_{\max}$ 288 nm (ϵ 9612 $\text{cm}^{-1} \text{M}^{-1}$), 265 (15 138), 243 (18 960); ^1H NMR (400 MHz, acetone- d_6): δ 7.67 (d, J = 8.5 Hz, 1H, H5), 7.41–7.44 (m, 2H, H2'', H6''), 7.27–7.32 (m, 2H, H3'', H5''), 7.17–7.23 (m, 1H, H4''), 7.12 (d, J = 8.5 Hz, 2H, H2', H6'), 6.79 (d, J = 8.7 Hz, 2H, H3', H5'), 6.65 (d, J = 8.7 Hz, 1H, H6), 4.50–4.53 (m, 2H, H2), 4.24 (s, 2H, CH₂-S), 3.80 (t, J = 6.9 Hz, 1H, H3); ^{13}C NMR (100 MHz, acetone- d_6): δ 191.3 (C4), 162.6 (C7), 162.1 (C8a), 157.6 (C4'), 138.4 (C1''), 131.1 (C2'', C6''), 130.7 (C2', C6'), 129.6 (C3'', C5''), 128.5 (C5), 128.0 (C1'), 127.0 (C4''), 116.2 (C3', C5'), 115.4 (C4a), 112.9 (C8), 110.7 (C6), 72.8 (C2), 51.6 (C3), 27.5 (S-CH₂). HRMS (+ESI): (C₂₂H₁₈O₄S) calcd m/z 401.0818 [M + Na]⁺, found m/z 401.0819.

8-(((4-Chlorophenyl)thio)methyl)-7-hydroxy-3-(4-hydroxyphenyl)chroman-4-one (23d). Isoflavanone **23d** was synthesized following general procedure 2 using the Mannich base **11** (121 mg, 0.39 mmol, 1 eq.) and *p*-chlorothiophenol (56 mg, 0.39 mmol, 1.0 eq.). The product obtained (40 mg, 25%) after column chromatography was further purified by preparative HPLC using a 5 μM column (150 $\times$ 22 mm ID) at a flow rate of 5 mL min⁻¹ with a gradient elution of 2–100% solvent B (solvent A: 0.1% formic acid/H₂O) and solvent B (0.1% formic acid/ACN) over 56 min to afford the expected product as a white solid (26 mg, 16%). M.p. 199–200 °C; IR (neat): $\nu_{\max}$ 3101, 2925, 1672, 1590, 1513, 1440, 1377, 1277, 1092, 1002 cm^{-1} ; UV (THF): $\lambda_{\max}$ 288 nm (ϵ 15 607 $\text{cm}^{-1} \text{M}^{-1}$), 264 (25 598), 242 (28 406); ^1H NMR (400 MHz, DMSO- d_6): δ 9.39 (br s, 1H, OH), 7.57 (d, J = 8.7 Hz, 1H, H5), 7.38 (d, J = 8.9 Hz, 2H, H2'', H6''), 7.34 (d, J = 8.9 Hz, 2H, H3'', H5''), 7.01 (d, J = 8.7 Hz, 2H, H2', H6'), 6.69 (d, J = 8.7 Hz, 2H, H3', H5'), 6.54 (d, J = 8.7 Hz, 1H, H6), 4.47–4.49 (m, 2H, H2), 4.11 (s, 2H, CH₂-S), 3.79–3.82 (m, 1H, H3); ^{13}C NMR (100 MHz, DMSO- d_6): δ 190.7 (C4), 162.1 (C7), 160.7 (C8a), 156.5 (C4'), 136.1 (C1''), 130.7 (C2'', C6''), 130.6 (C4''), 129.6 (C2', C6'), 128.7 (C3'', C5''), 127.5 (C5), 126.1 (C1'), 115.2 (C3', C5'), 113.5 (C4a), 110.7 (C6), 109.8 (C8), 71.4 (C2), 50.2 (C3), 26.2 (S-CH₂).

HRMS (+ESI): (C₂₂H₁₇ClO₄S) calcd m/z 435.0428 [M + Na]⁺, found m/z 435.0429.

Cell viability assays

The cell viability assay was conducted following literature method.⁴⁸ MDA-MB-231 cells were seeded at 3000 cells per well in 96-well plates to ensure sustained exponential growth for 4 days. Cells were treated 24 h after seeding with 1 to 500 μM of compounds. After 72 h drug incubation, the treatment-media was replaced with 10% Alamar Blue in fresh media and the cells were incubated for another 6 h. The metabolic activity was detected by spectrophotometric analysis by assessing the absorbance of Alamar blue. Cell proliferation was determined and expressed as a percentage of untreated control cells. The determination of IC₅₀ values was done using GraphPad Prism 6.

Conclusions

In conclusion, *o*-QM intermediates derived from 6-hydroxyflavanone, 7-hydroxyflavanone, and dihydrodaidzein had been successfully utilized to generate fused-ring systems and diarylmethane products. The resultant yields of the analogues varied depending on the reactivity of the dienophiles or nucleophiles. With *o*-QMs derived from 6-hydroxyflavanone, the cycloadducts were formed in relatively higher yield when electron-rich enaminone was used as a dienophile. Reactions involving *o*-QMs derived from 7-hydroxyflavanone generated a mixture of both ring-closed and ring-opened products in comparatively lower yields. A thioether bridge was formed instead of a methylene bridge when sulfur-containing nucleophiles were used in the Michael addition reaction of Mannich base derived from dihydrodaidzein. A rationale has been postulated, in which the thiolate anion behaves as a stronger nucleophile than the π electrons in the benzene ring. Several cycloadducts exhibited slightly improved anti-proliferative activity against MDA-MB-231 breast cancer cell lines as compared to their corresponding parent compounds, although the activity was low. Overall, the reactions involving *o*-QM intermediates remain a useful method to generate a wide range of cycloaddition adducts and diarylmethane products.

Data availability

The data supporting this article have been included as part of the ESI.†

Author contributions

Conceptualization: N. K. and D. B., methodology: V. A. N., software N/A, validation: V. A. N., formal analysis: V. A. N., investigation: V. A. N., resources: N. K. and D. B., data curation, writing – original draft: V. A. N., writing – review & editing: V. F., D. S. W., N. K. and D. B., visualization: V. A. N. and V. F. supervision: N. K., D. B. and D. S. W., project administration: N. K. and D. B., funding acquisition: N. K. and D. B.

Conflicts of interest

There are no conflicts to declare.

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